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## APPENDICES

Appendix A: Properties of materials in this experiment.

### 1. High-density polyethylene (RR1760)

MFI 2.16 kg/190 °C	3	g/10min
Density	0.957	g/cm <sup>3</sup>
Tensile strength at break	>18	N/mm <sup>2</sup>
Tensile strength at yield	>30	N/mm <sup>2</sup>
Elongation at break	>800	%
Notched impact strength	>4.5	mJ/mm <sup>2</sup>
Ball indentation hardness	>54	N/mm <sup>2</sup>

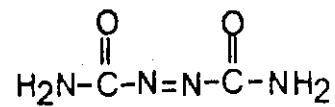
### 2. Natural Rubber (TTR5L)

### 3. Compatibilizer : Epolene wax

Ring and ball softening point	157 °C
Density	0.934 g/cm <sup>2</sup> at 25 °C
Acid Number	47
Brook field Thermosel Viscosity (cP)	
at 140 °C	solid
at 150 °C	solid
at 190 °C	400
Color, Gardner scale	11
Molecular Weight (GPC)	
Mw	9100
Mn	3900

**4. Blowing agent: Azodicarbonamide**

Chemical Structure



Molecular Weight	116.1
Specific Gravity	1.65
Specific Heat	0.25 kcal/kg °C
Combustion Heat	217.2 kcal/mol
Appearance	Fine, Pale Yellow Powder
Decomposition Point	195 - 202 °C
Gas Volume Evolved	240 ml/g

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**Appendix B: TGA thermograms of the blowing agent activated by blowing agent activator**

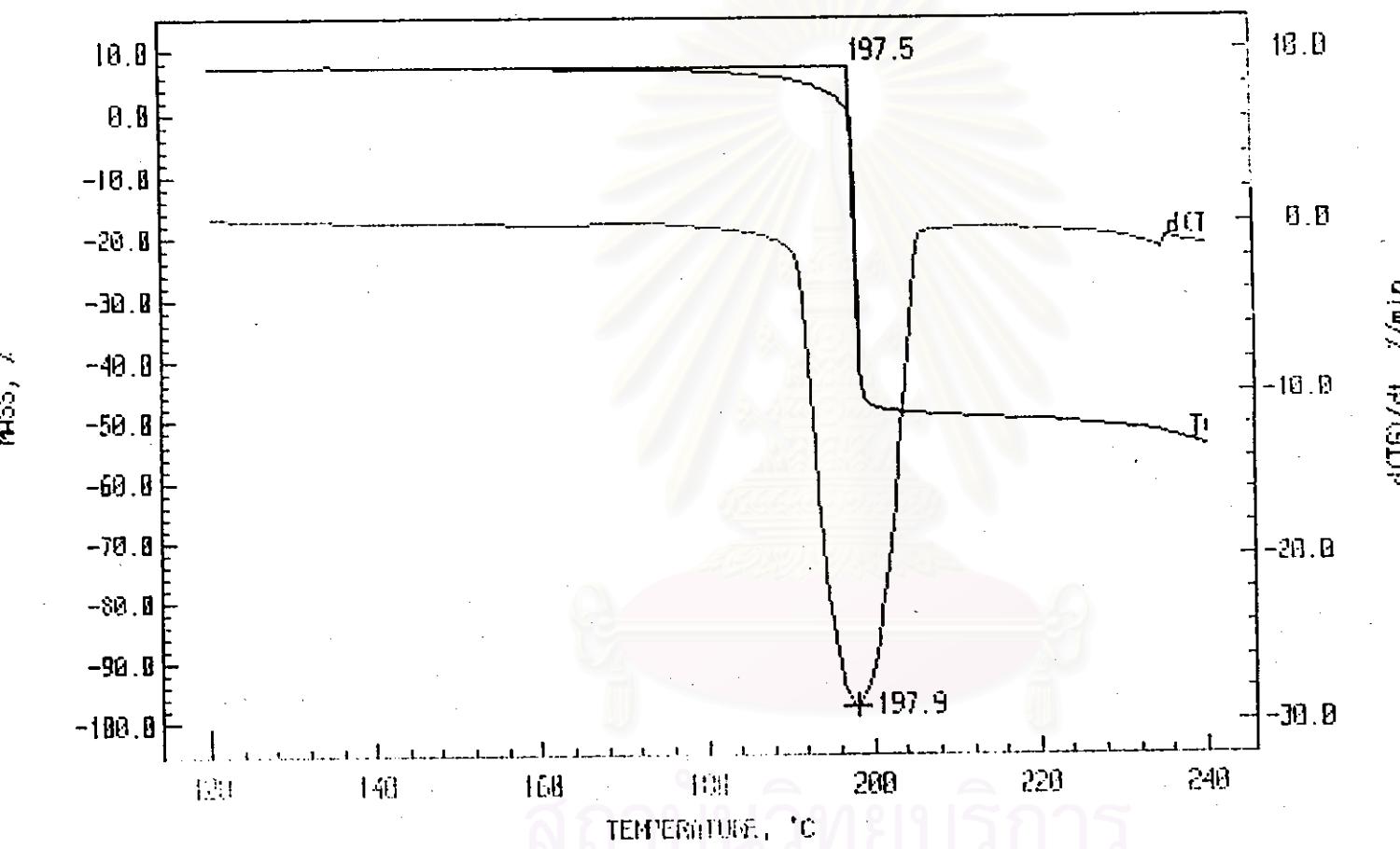


Figure B1 TGA thermogram of azodicarbonamide activated by 10% zinc oxide

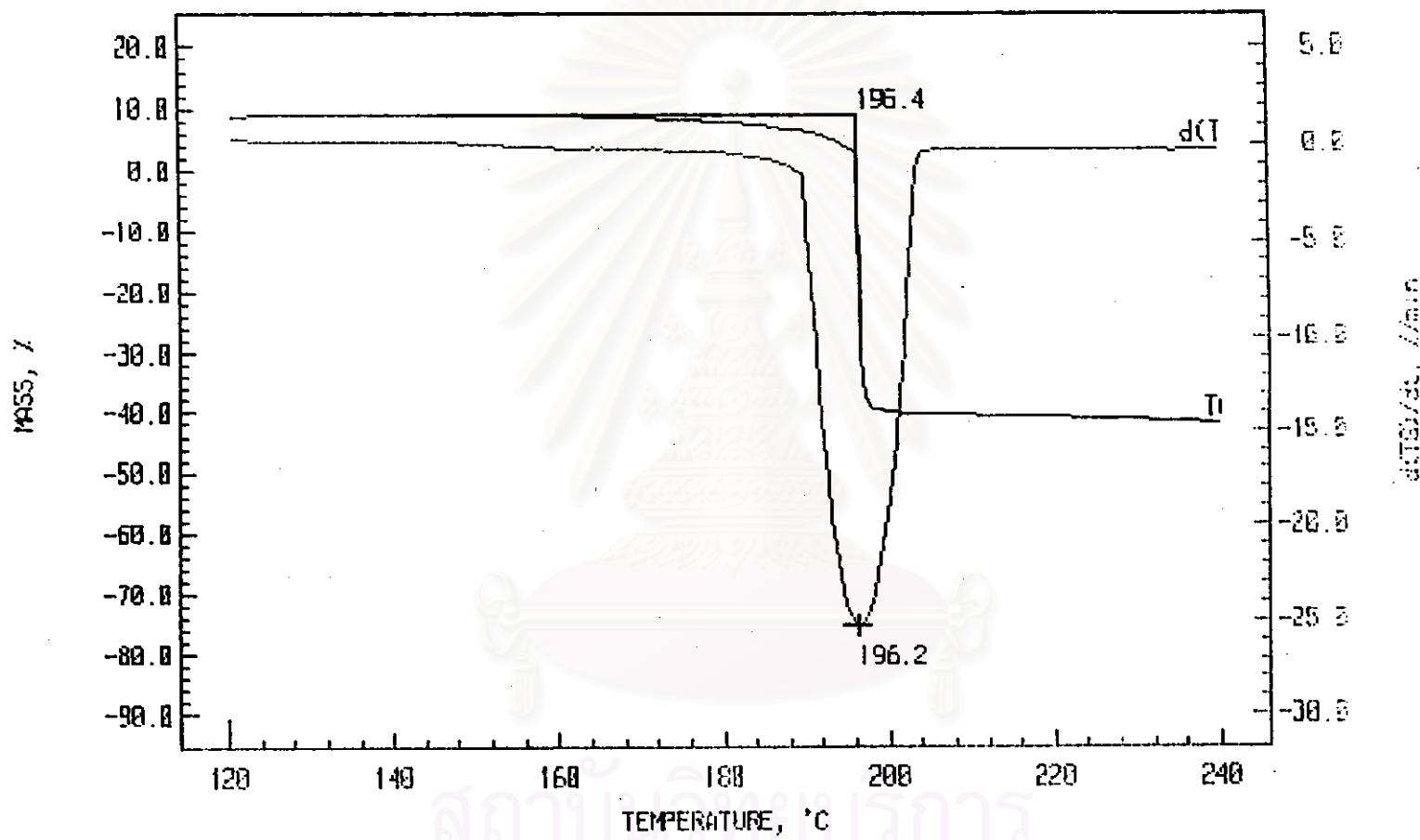


Figure B2 TGA thermogram of azodicarbonamide activated by 25% zinc oxide

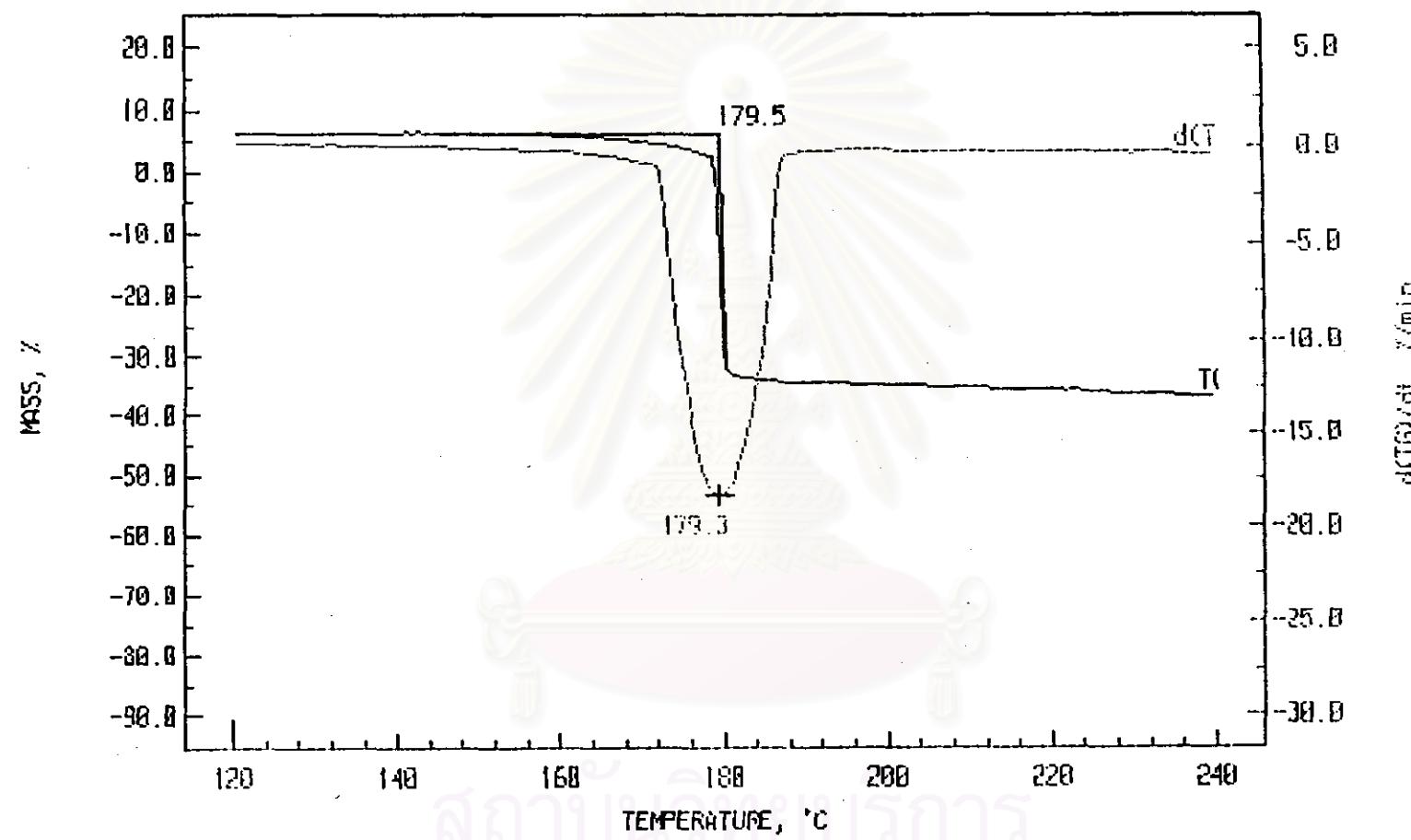


Figure B3 TGA thermogram of azodicarbonamide activated by 50% zinc oxide

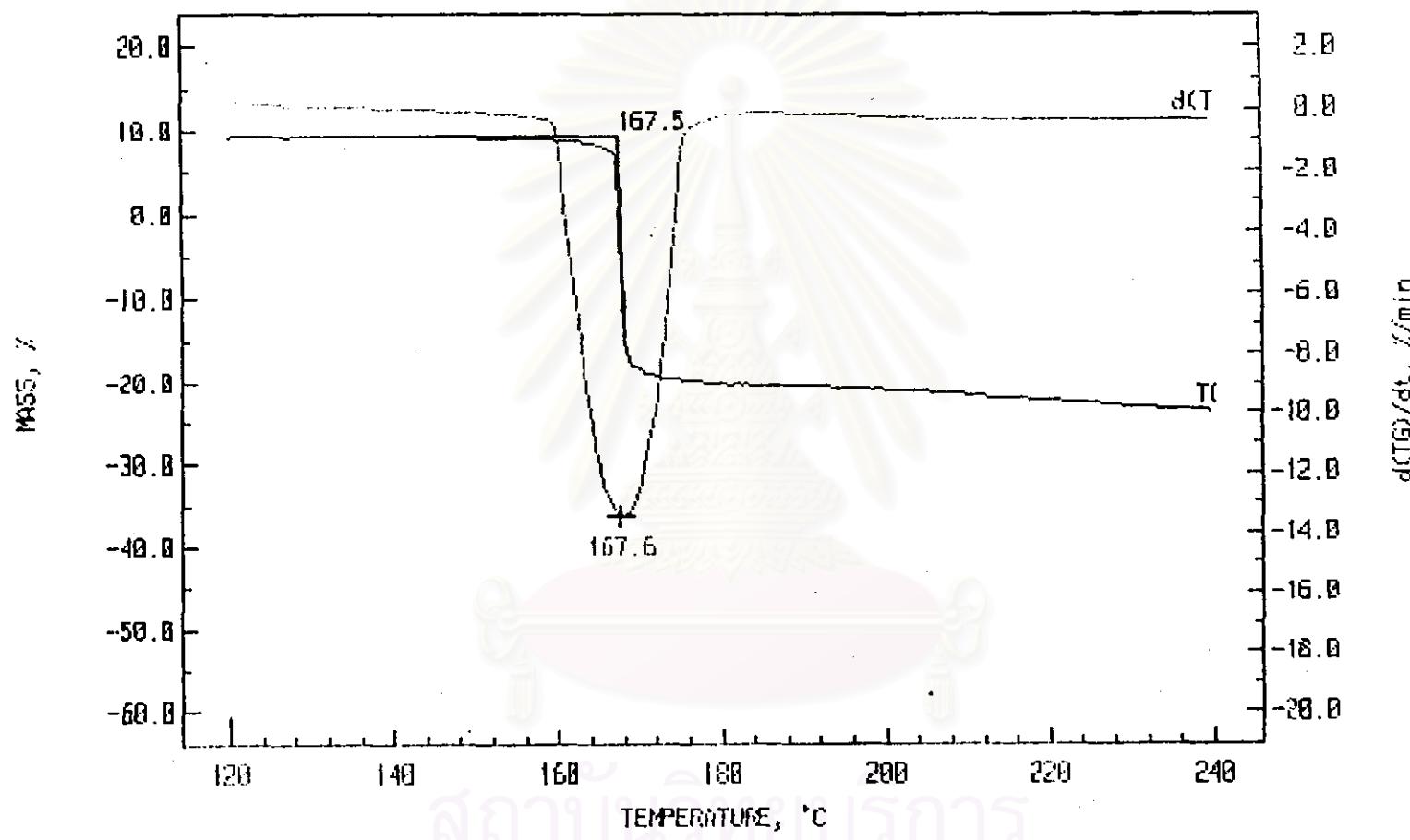


Figure B4 TGA thermogram of azodicarbonamide activated by 75% zinc oxide

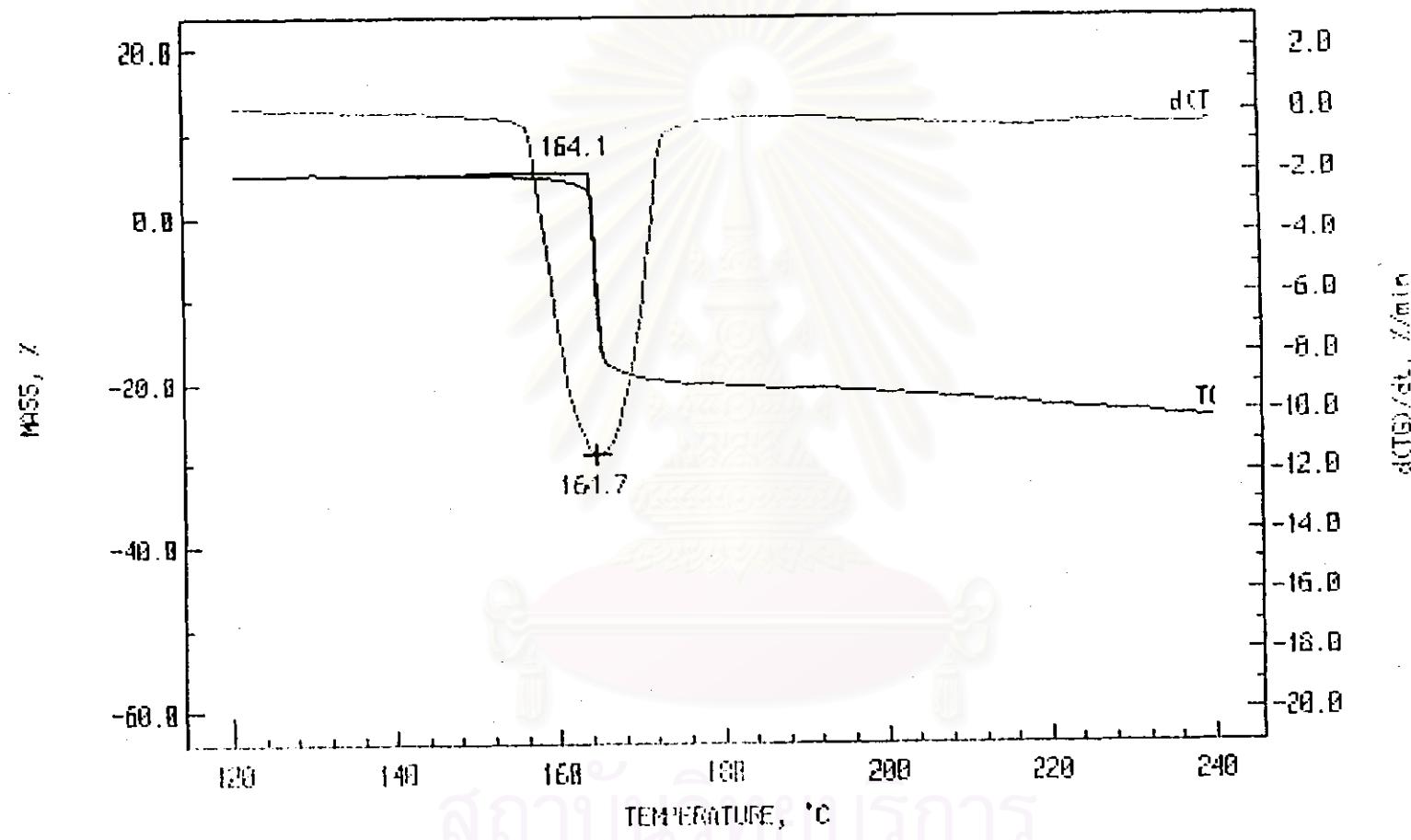
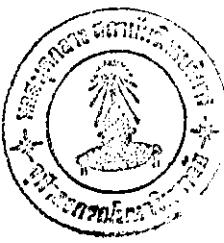


Figure B5 TGA thermogram of azodicarbonamide activated by 100% zinc oxide



VITA

Miss Suratchana Thinakom was born on June 9, 1973 in Ranong, Thailand. She earns the Bachelor of Science degree from the Department of Chemistry, Faculty of Science, Prince of Songkhla University, in 1995. She began her higher study in the Program of Petrochemistry and Polymer Science, Graduate School, Chulalongkorn University, in 1996, and completed the Masters degree of Petrochemistry and Polymer Science program in April 1999.

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