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**APPENDICES**

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## APPENDIX A

### The average thickness of multilayers

**Table A-1.** The average thickness of PSS-CHI bilayer calculated from ellipsometric data of Si-NH<sub>2</sub>-(PSS-CHI)<sub>5</sub> assemblies as a function of chitosan concentration in the presence of 0.25 M NaCl.

Chitosan concentration (mg mL <sup>-1</sup> )	Average thickness of bilayer (Å)
0.25	30.32 ± 3.85
0.5	29.9 ± 6.66
0.75	29.4 ± 2.45
1	31.32 ± 4.21

**Table A-2.** The average thickness of PSS-CHI bilayer calculated from ellipsometric data of Si-NH<sub>2</sub>-(PSS-CHI)<sub>10</sub> assemblies as a function of NaCl concentration.

Concentration of NaCl (M)	Average thickness of bilayer (Å)
0	16.28 ± 6.85
0.25	32.18 ± 8.23
0.5	35.76 ± 6.30
1	44.68 ± 7.99

**Table A-3.** Ellipsometric thickness of Si-NH<sub>2</sub>-(PSS-CHI)<sub>10</sub> assemblies as a function of the number of layer.

Number of layers	Average Thickness (Å)			
	Without NaCl	0.25 M NaCl in CHI solution	0.25 M NaCl in PSS solution	0.25 M NaCl in both CHI and PSS solutions
5	56.60 ± 1.52	64.80 ± 5.97	89.80 ± 2.56	119.20 ± 6.57
6	71.20 ± 7.12	87.00 ± 5.83	91.00 ± 3.42	145.00 ± 5.00
9	77.20 ± 4.04	131.80 ± 7.26	157.40 ± 4.57	246.80 ± 4.55
10	88.80 ± 5.54	149.50 ± 6.66	185.20 ± 3.56	297.80 ± 6.42
15	126.00 ± 6.12	228.60 ± 3.27	-	-
16	140.40 ± 5.41	248.50 ± 6.14	-	-
19	150.80 ± 2.86	303.60 ± 3.40	-	-
20	162.75 ± 6.85	321.80 ± 8.23	-	-

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## APPENDIX B

### Stratification of multilayered film

**Table B-1.** Zeta-potential of Si-NH<sub>2</sub>-(PSS-CHI)<sub>n</sub> assemblies, 0.25 M NaCl was added to chitosan solution.

Top layer	Number of layer	Zeta-potential (mV)
Si-NH <sub>2</sub>	0	10.15
Poly(styrene sulfonate)	1	-18.27
Chitosan	2	0.06
Poly(styrene sulfonate)	5	-5.57
Chitosan	6	8.07
Poly(styrene sulfonate)	9	-16.09
Chitosan	10	7.84
Poly(styrene sulfonate)	15	-16.51
Chitosan	16	11.53

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**Table B-2** Water contact angle of Si-NH<sub>2</sub>-(PSS-CHI)<sub>n</sub> assemblies, 0.25 M NaCl was added to chitosan solution

Top layer	Number of layer	Water contact angle (°)
Si-NH <sub>2</sub>	0	70±1.52
Poly(styrene sulfonate)	5	78±1.71
Chitosan	6	83±0.82
Poly(styrene sulfonate)	9	77±1.71
Chitosan	10	83±1.25
Poly(styrene sulfonate)	15	76±0.85
Chitosan	16	80±1.71
Poly(styrene sulfonate)	19	77±1.46
Chitosan	20	82±1.25

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## APPENDIX C

### Surface topography of multilayered films

**Table C-1.** The surface roughness of Si-NH<sub>2</sub>-(PSS-CHI)<sub>n</sub> prepared in the absence of NaCl and in the presence of 1 M NaCl.

Top layer	Number of layer	0 M NaCl		1 M NaCl	
		R <sub>a</sub> (nm)	R <sub>Z</sub> (nm)	R <sub>a</sub> (nm)	R <sub>Z</sub> (nm)
Si-NH <sub>2</sub>	0	0.5	4.9	0.5	4.9
Poly(styrene sulfonate)	1	1.2	24.3	9.1	80.8
Chitosan	2	0.9	36.1	9.1	40.1
Chitosan	10	7.2	148.1	11	127.3
Poly(styrene sulfonate)	15	3.0	22.8	9.9	46.2
Chitosan	16	7.5	104.7	10.1	47.2
Chitosan	20	9.3	109.3	15.7	158.0

R<sub>a</sub> : The arithmetic average of the deviations from the center plan.

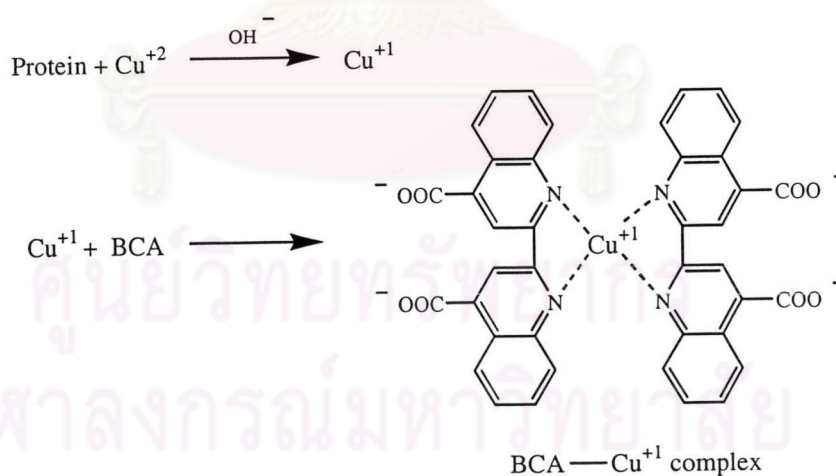
R<sub>Z</sub> : Standard deviation of the z values.

## APPENDIX D

### Bicinchoninic acid assay

Bicinchoninic acid assay is a method used for determination of the amount of proteins. The standard reagents used in this method are reagent A, reagent B and reagent C. Reagent A consists of an aqueous solution of  $\text{Na}_2\text{tartrate}$ ,  $\text{Na}_2\text{CO}_3$ ,  $\text{NaHCO}_3$  in 0.2 M  $\text{NaOH}$ , pH 11.25. Reagent B is 4% (W/V) bicinchoninic acid solution, pH 8.5. Reagent C is 4%  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  in deionized water.

The principle of the bicinchoninic assay (BCA) relies on the formation of a  $\text{Cu}^{2+}$ -protein complex under alkaline conditions, followed by reduction of the  $\text{Cu}^{2+}$  to  $\text{Cu}^{1+}$ . The amount of reduction is proportional to protein present. It has been shown that the peptide bond is able to reduce  $\text{Cu}^{2+}$  to  $\text{Cu}^{1+}$ . BCA forms a purple-blue complex with  $\text{Cu}^{1+}$  in alkaline environments, thus providing a basis to monitor the reduction of alkaline  $\text{Cu}^{2+}$  by proteins.<sup>30</sup> Figure D-1 shows complexation between bicinchoninic acid and  $\text{Cu}^{1+}$ .



**Figure D-1.** Formation of purple complex between BCA and cuprous ion generated from the biuret reaction.

**Table D-1.** The amount of plasma protein adsorption per surface area ( $\mu\text{g}/\text{cm}^2$ ) of Si-  $\text{NH}_2$ -(PSS-CHI) $_n$  assemblies prepared in the absence of NaCl and presence of 1 M NaCl.

Top layer	Number of layer	The amount of plasma protein adsorption ( $\mu\text{g}/\text{cm}^2$ )	
		0 M NaCl	1 M NaCl
Si-NH <sub>2</sub>	0	4.828±4.100	1.494±0.424
Poly(styrene sulfonate)	1	3.908±0.999	2.567±0.663
Chitosan	2	7.893±0.531	3.142±1.154
Poly(styrene sulfonate)	5	6.475±1.153	3.908±0.791
Chitosan	6	8.046±0.844	11.954±2.304
Poly(styrene sulfonate)	9	6.475±2.839	7.126±0.633
Chitosan	10	7.165±3.309	12.299±1.416
Poly(styrene sulfonate)	15	4.866±1.564	15.172±0.973
Chitosan	16	5.862±2.110	18.797±5.886
Poly(styrene sulfonate)	19	9.234±0.948	27.510±10.078
Chitosan	20	8.391±0.633	29.080±15.667

## VITAE

Miss Chamaiporn Mutchapato was born in Nongkhai, Thailand, on October 26<sup>th</sup>, 1976. She received Bachelor degree of science in 1998 from Department of Chemistry, Faculty of Science, Ubonratchathani University. She started as a master degree student with a major in Polymer Science, Program of Petrochemistry and Polymer Science, Chulalongkorn University in 1999 and completed program in 2003.



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