

## CHAPTER VI

### CONCLUSIONS AND RECOMMENDATION

#### 6.1 Conclusions

The main purpose of the present research is to investigate the effect of elevated temperature and coexisting gases on the individual and simultaneous removal efficiency of acetaldehyde ( $\text{CH}_3\text{CHO}$ ), ammonia ( $\text{NH}_3$ ) and trimethyl amine  $[(\text{CH}_3)_3\text{N}]$  from  $\text{N}_2$  using electron attachment reaction. The coexisting gases investigated are carbon dioxide gas, oxygen gas, and water vapor. From the results, it may be concluded as follows:

##### 6.1.1 Effect of $\text{CO}_2$ coexisting gas on separate and simultaneous $\text{CH}_3\text{CHO}$ , $(\text{NH}_3)$ and $(\text{CH}_3)_3\text{N}$ removal

The presence of  $\text{CO}_2$  substantially enhances the separate and simultaneous removal efficiency of  $\text{CH}_3\text{CHO}$ ,  $(\text{NH}_3)$  and  $(\text{CH}_3)_3\text{N}$  because of the effect of  $\text{CO}_3^-$  and  $\text{O}^-$  anions at low temperatures and various radicles at high temperature.

##### 6.1.2 Effect of $\text{O}_2$ coexisting gas on separate and simultaneous $\text{CH}_3\text{CHO}$ , $(\text{NH}_3)$ and $(\text{CH}_3)_3\text{N}$ removal

The presence of  $\text{O}_2$  drastically enhances the separate and simultaneous removal efficiency of  $\text{CH}_3\text{CHO}$ ,  $(\text{NH}_3)$  and  $(\text{CH}_3)_3\text{N}$  because of the effect of  $\text{O}_3$  and  $\text{O}^-$  anion at low temperatures and N, O radicals at high temperatures. The co - presence of  $\text{CO}_2$  has significant enhancement effect on the removal efficiency of  $\text{CH}_3\text{CHO}$ ,  $(\text{NH}_3)$  and  $(\text{CH}_3)_3\text{N}$  from  $\text{N}_2 - \text{O}_2$ .

### **6.1.2 Effect of H<sub>2</sub>O on the removal of separate and simultaneous CH<sub>3</sub>CHO, (NH<sub>3</sub>) and (CH<sub>3</sub>)<sub>3</sub>N removal**

The presence of H<sub>2</sub>O in N<sub>2</sub> slightly enhances the separate and simultaneous removal efficiency of CH<sub>3</sub>CHO, (NH<sub>3</sub>) and (CH<sub>3</sub>)<sub>3</sub>N at low to moderate temperatures because of the effect of H<sup>•</sup>, OH<sup>•</sup> and O<sup>•</sup> anions but slightly retards the removal efficiency at high temperatures because at a low discharge current, the relatively much smaller number of electrons tend to attach mostly to H<sub>2</sub>O. The presence of water vapor in N<sub>2</sub> - O<sub>2</sub> - CO<sub>2</sub> mixed gas generally has favorable effect on the removal efficiency of CH<sub>3</sub>CHO, NH<sub>3</sub> and (CH<sub>3</sub>)<sub>3</sub>N.

### **6.1.3 Two corona-discharge reactors in series for minimizing generation of byproducts and/or enhancing removal efficiency**

Two independently operated corona discharge reactors in series have shown a good promise for minimizing the generation of byproducts and enhancing the removal efficiency. The first reactor should aim at the complete removal of (CH<sub>3</sub>)<sub>3</sub>N at 300°C, whereas the second should aim at the complete removal of CH<sub>3</sub>CHO and NH<sub>3</sub> at 200°C while minimizing the generation of byproduct CO, O<sub>3</sub> and NO<sub>x</sub> down to 100 ppm, 0 ppm and 0 ppm, respectively. As the discharge current increases, the byproduct CO increases.

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### 6.1.4 Suitable Conditions for the removal of target gases from N<sub>2</sub> - CO<sub>2</sub>

#### Single component

Target Gas	Acetaldehyde	Ammonia	Trimethyl amine
CO <sub>2</sub> (%)	20	20	20
T (°C)	100	100	100
$\psi'$ (-)	1.0	0.21	0.97

#### Binary components

Target Gas	Acetaldehyde+Ammonia		Acetaldehyde+Trimethyl amine		Ammonia+Trimethyl amine	
	Acetaldehyde	Ammonia	Acetaldehyde	Trimethyl amine	Ammonia	Trimethyl amine
T (°C)	100	100	100	100	100	100
$\psi'$ (-)	0.91	0.64	0.93	0.95	-0.23	1.00

#### Tertiary components

Target Gas	Acetaldehyde + Ammonia + Trimethyl amine		
	Acetaldehyde	Ammonia	Trimethyl amine
T (°C)	100	100	100
$\psi'$ (-)	0.91	1.00	1.00

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## 6.2 Recommendation for future work

From the experimental results, CO is still detected, and its concentration generally increases as the discharge current increases. As in the case of the exhaust gas from a combustion process, it may be impossible to completely get rid of the byproduct CO in the presence of CO<sub>2</sub> because of the chemical equilibrium between CO and CO<sub>2</sub>. Nevertheless, more investigation of the applications of two independently operated corona discharge reactors in series for minimizing generation of byproducts should be carried out. To minimize the operating costs, the operating temperature and current discharge, which affect the energy - based efficiency,  $\psi_{ener}$  should be optimized.



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