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ศูนย์วิทยทรัพยากร
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APPENDIX

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX A

RELATIONSHIP BETWEEN AXIAL DISPERSION
COEFFICIENT AND VELOCITY AND PARTICLE SIZE

Let us follow the viewpoint that packed beds consist of series of randomly arrayed void cells through which fluid flows with some mixing . So a mixing efficiency is a function of turbulence and retention time for a given fluid , it follows that fluid velocity and cell size controls mixing efficiency . Cell size is a function of particle diameter . Thus it may be written (25) as

$$E_z = k \cdot d_p \cdot u \quad \text{or} \quad d_p \cdot u / E_z = 1/k = Pe \quad \text{A.1}$$

For a given velocity , a mixing time , provided by the cell dimension and the number of such cells , is required for perfect mixing . However , the situation which is characterized by imperfect mixing will have a cell retention time θ_n less than that of a perfect cell θ_p . Therefore more than one cell is required to achieve the perfect mixing time . For a bed of n_p void cells , the effect of imperfect mixing is to endow the bed with dispersion characteristics of a bed of n mixers . Since $n < n_p$, a greater dispersion and thus larger E results .

Imperfect mixing may be viewed as the result of by-passing or short circuiting and dead-space retention . This states that a fraction of particles entering void cells passes through in a much shorter time than the holding time θ_n , while other fractions retained for much greater periods than this holding time . Obviously a distribution of retention times results is characterized by a dispersion coefficient E . The similarity of this process to molecular diffusion offers an opportunity for a more quantitative treatment .

Consider a field of particles and a zone (x_2-x_1) bound by line P_1 and P_2 as shown in figure A1 .

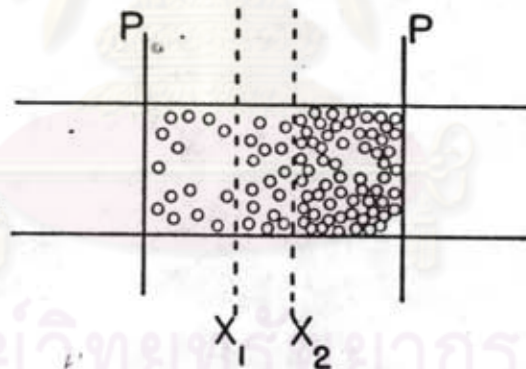


Fig. Einstein's kinetic-diffusion model.

Fig. A.1 Einstein's kinetic-diffusion model

The transport of U molecules from left to right per unit area can be expressed as

$$U_a = 1/2 * c_a * (x_2 - x_1)$$

A.2

Imperfect mixing may be viewed as the result of by-passing or short circuiting and dead-space retention . This states that a fraction of particles entering void cells passes through in a much shorter time than the holding time θ_n , while other fractions retained for much greater periods than this holding time . Obviously a distribution of retention times results is characterized by a dispersion coefficient E . The similarity of this process to molecular diffusion offers an opportunity for a more quantitative treatment .

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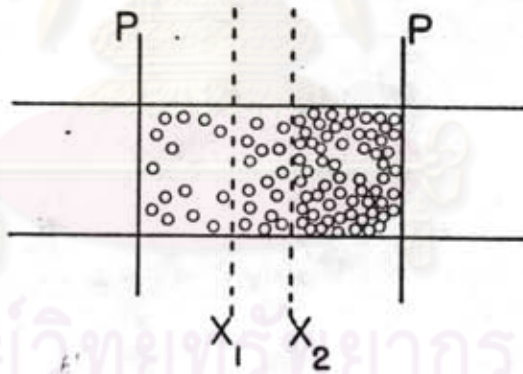


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A.2

If a real packed bed of length L is visualized as n mixing units , then the height of one mixer is L/n . By measuring upstream and downstream dispersion or diffusion , the mixing coefficient E is obtained . Therefore the time constant for axial mixing can be expressed in a similar manner as equation A.6

$$\theta_a = (L/n)^2/2E \quad \text{A.7}$$

As the holding time per mixing length for a bed velocity u is

$$\theta_n = \frac{L/n}{u}$$

since a perfect mixer is one in which retention time equals mixing time then

$$\frac{L/n}{u} = (L/n)^2/2E$$

or

$$n = Lu/2E \quad \text{A.9}$$

To relate this expression with the ideal perfect mixer formed by particles of diameter d_p , a mixing efficiency e is introduced here and defined as a ratio of actual mixers n to the limiting

number in the ideal perfect mixer (n_p) ; $e = n/n_p$. Since the length of each perfect mixing cell will be some fraction (F) of particle diameter d_p , the constant fraction F will be about unity or less depending on the packing arrangements , therefore the number of perfect mixers is

$$n_p = L/(F*d_p) \quad \text{A.10}$$

From equation A.9 and A.10 , mixing efficiency becomes

$$e = (Lu/2E)/(L/Fd_p) = (F/2)(d_p u/E)$$

or

$$e/F = 1/2*(d_p u/E) \quad \text{A.11}$$

For diffusion behaviour in the void cell approaches the ideal perfect mixer , $e \rightarrow 1$ and F is approximately 1 , thus e/F called mixing length for the bed becomes unity and equation A.11 can be expressed as

$$E = 1/2*d_p*u \quad \text{A.12}$$

APPENDIX B

DERIVATION OF THE MODEL'S TRANSFER FUNCTION

Consider the following system

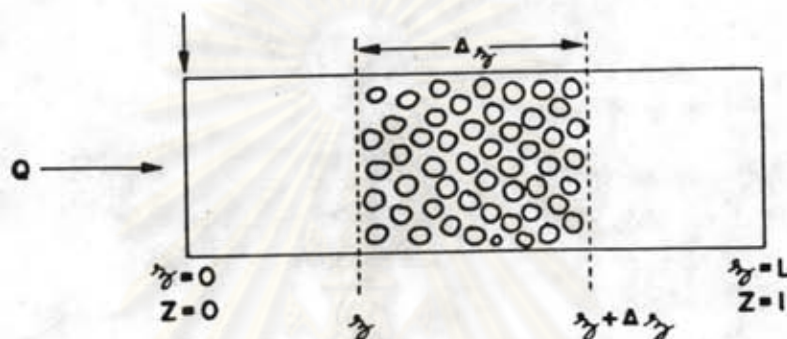


Fig. B1 Packed Bed System

Assumptions

1. No reactions occur
2. Fluid velocity remains constant
3. Radial velocity gradients do not exist

Then the material balance around this system is

$$\epsilon \cdot A \cdot \Delta z \cdot \frac{\partial c(z,t)}{\partial t} = Q[c(z,t) - c(z+\Delta z,t)] + E_z \cdot A \cdot \left[\frac{\partial c(z+\Delta z,t)}{\partial z} - \frac{\partial c(z,t)}{\partial z} \right]$$

dividing the above equation by \$A \cdot \Delta z\$ we obtain



$$\epsilon \frac{\partial c}{\partial t} = \frac{Q}{A} \left[\frac{c(z,t) - c(z+\Delta z,t)}{\Delta z} \right] + E_z \left[\frac{\partial(c(z+\Delta z,t)/\partial z) - \partial c(z,t)/\partial z}{\partial z} \right]$$

It ultimately yields

$$\frac{\partial c}{\partial t} = -u_y \frac{\partial c}{\partial z} + E_z \left[\frac{\partial^2 c}{\partial z^2} \right] \quad \text{B.1}$$

Accumulation convection dispersion
term term term

Let $Z = z/L$, $\tau = \epsilon L/u_y$, $Pe = L u_y/E_z$

Therefore equation B.1 can be rewritten in dimensionless form as

$$1/Pe \left(\frac{\partial^2 c}{\partial Z^2} \right) - \frac{\partial c}{\partial Z} = \tau \left(\frac{\partial c}{\partial \tau} \right) \quad \text{B.2}$$

Before applying a closed-closed boundary condition to equation B.2 , let us focus on mass conservation at the inlet point and the outlet point , respectively .

At inlet point

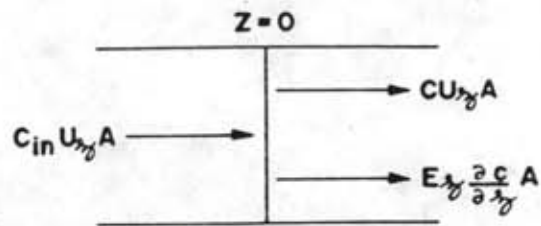


Fig.b2 Mass Conservation at The Inlet Point

$$A \cdot u_z \cdot c_{in}(z=0, t) = A \cdot u_z \cdot c(z=0, t) - A \cdot E_z \cdot \left[\frac{\partial c(z=0, t)}{\partial z} \right]$$

or

$$c_{in}(z=0, t) = c(z=0, t) - (E_z/u_z) \cdot \left[\frac{\partial c(z=0, t)}{\partial z} \right]$$

now let $Z = z/L$, and $u_z \cdot L/E_z = Pe$

so

$$c_{in}(z=0, t) = c(z=0, t) - (1/Pe) \cdot \left[\frac{\partial c(z=0, t)}{\partial z} \right]$$

B.3

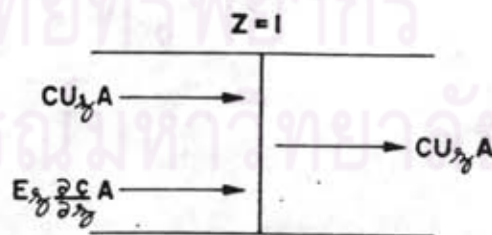
At outlet point

Fig.B3 Mass Conservation at The Outlet point

$$A u_s c_s(z=L,t) - A E_z [\partial c(z=L,t) / \partial z] = A u_s c_e(z=L,t)$$

or

$$c_s(z=L,t) - (E_z / u_s) [\partial c(z=L,t) / \partial z] = c_e(z=L,t)$$

let $Z = z/L$, and $u_s L / E_z = Pe$

hence

$$c_s(z=1,t) - (1/Pe) [\partial c(z=1,t) / \partial z] = c_e(z=1,t)$$

Since there exists a discontinuity at this position ,thus

$$c_s(z=1,t) = c_e(z=1,t)$$

So the equation becomes

$$\partial c(z=1,t) / \partial z = 0 \quad \text{B.4}$$

Let $\hat{c}(z,p) = \mathcal{L}[c(z,t)]$; therefore equation B.2 becomes

$$[Z\{p\hat{c}(z,p)\} - \{c(z,t=0)\}] = (1/Pe) [\partial^2 \hat{c} / \partial z^2] - \{\partial \hat{c} / \partial z\}$$

Since the initial condition is defined as shown below

$$c(z, t=0) = 0$$

thus

$$\mathcal{Z} * p * \hat{c}(z, p) = (1/Pe) * [\partial^2 \hat{c}(z, p) / \partial z^2] - [\partial \hat{c}(z, p) / \partial z]$$

Let $\hat{c}(s, p) = \mathcal{L}[\hat{c}(z, p)]$; so

$$\begin{aligned} \mathcal{Z} * p * \hat{c}(s, p) &= (1/Pe) * [s^2 \hat{c}(s, p) - s \hat{c}(z=0, p) - \{\partial \hat{c}(z=0, p) / \partial z\}] \\ &\quad - [s \hat{c}(s, p) - \hat{c}(z=0, p)] \end{aligned}$$

$$\frac{1 * s * \hat{c}(z=0, p)}{Pe} + \frac{1 * \partial \hat{c}(z=0, p)}{Pe \partial z} - \hat{c}(z=0, p) = \frac{1 * s^2 * \hat{c}(s, p)}{Pe} - \frac{s * \hat{c}(s, p)}{Pe} - \mathcal{Z} * p * \hat{c}(s, p)$$

$$\frac{s^2 - s - \mathcal{Z} * p}{Pe} * \hat{c}(s, p) = \left(\frac{s-1}{Pe} \right) * \hat{c}(z=0, p) + \frac{(1/Pe) * \partial \hat{c}(z=0, p)}{\partial z}$$

$$\hat{c}(s, p) = \frac{Pe}{(s^2 - s - \mathcal{Z} * p * Pe)} * \frac{-\hat{c}(z=0, p) + s * \hat{c}(z=0, p) + (1/Pe) \partial \hat{c}(z=0, p)}{Pe \partial z} \quad B.5$$

Equation B.3 can be transformed into the Laplace domain as follows

Let $\hat{c}(z, p) = \mathcal{L}[c(z, 0)]$

hence

$$\hat{c}_{in}(z=0, p) = \hat{c}(z=0, p) - (1/Pe) * [\partial \hat{c}(z=0, p) / \partial z]$$

or

$$-\hat{c}_{in}(z=0, p) = -\hat{c}(z=0, p) + (1/Pe) * [\partial \hat{c}(z=0, p) / \partial z] \quad B.6$$

Substituting equation B.6 into B.5

$$\hat{c}(s,p) = \frac{Pe * \{s\hat{c}(z=0,p)/Pe\} - \hat{c}_{in}(z=0,p)}{(s-\lambda_1)(s-\lambda_2)} \quad B.7$$

where

$$\lambda_1 = Pe/2 + \sqrt{Pe^2/4 + Pe*p*\tau}$$

$$\lambda_2 = Pe/2 - \sqrt{Pe^2/4 + Pe*p*\tau}$$

Taking the inverse Laplace transform of equation B.7

Let $\hat{c}(z,p) = \mathcal{L}^{-1}[c(s,p)]$

$$\hat{c}(z,p) = Pe * \frac{s}{(s-\lambda_1)(s-\lambda_2)} * \hat{c}(z=0,p) - Pe * \hat{c}_{in}(z=0,p) * \frac{1}{(s-\lambda_1)(s-\lambda_2)} \quad B.8$$

Applying the cover rule and taking the partial fraction to the first term of equation B.8

$$s/[(s-\lambda_1)(s-\lambda_2)] = a/(s-\lambda_1) + b/(s-\lambda_2)$$

$$a = \lambda_1/(\lambda_1-\lambda_2) \quad \text{and} \quad b = -\lambda_2/(\lambda_1-\lambda_2)$$

so

$$s/[(s-\lambda_1)(s-\lambda_2)] = [\lambda_1/(\lambda_1-\lambda_2)]/[s-\lambda_1] - [\lambda_2/(\lambda_1-\lambda_2)]/[s-\lambda_2] \quad B.9$$

Proceeding similarly as above for the second term we obtain

$$1/[(s-\lambda_1)(s-\lambda_2)] = c/(s-\lambda_1) + d/(s-\lambda_2)$$

$$c = 1/(\lambda_1-\lambda_2) \quad \text{and} \quad d = -1/(\lambda_1-\lambda_2)$$

hence

$$1/[(s-\lambda_1)(s-\lambda_2)] = (1/(\lambda_1-\lambda_2))/(s-\lambda_1) + [1/(\lambda_1-\lambda_2)]/(s-\lambda_2) \quad \text{B.10}$$

Applying B.9 and B.10 to B.8, the ultimate equation is

$$\begin{aligned} \hat{c}(z,p) = & \hat{c}(z=0,p) * [(\lambda_1/(\lambda_1-\lambda_2))\exp(\lambda_1*z) - (\lambda_2/(\lambda_1-\lambda_2))\exp(\lambda_2*z)] \\ & - Pe * \hat{c}_{in}(z=0,p) * [(1/(\lambda_1-\lambda_2))\exp(\lambda_1*z) - (1/(\lambda_1-\lambda_2))\exp(\lambda_2*z)] \end{aligned}$$

or

$$\begin{aligned} \hat{c}(z,p) = & \frac{Pe}{\{\lambda_1-\lambda_2\}} * (\lambda_1 * \exp(\lambda_1*z) - \lambda_2 * \exp(\lambda_2*z)) * \hat{c}(z=0,p) \\ & - \frac{Pe}{\{\lambda_1-\lambda_2\}} * [\hat{c}_{in}(z=0,p)] * [\exp(\lambda_1*z) - \exp(\lambda_2*z)] \quad \text{B.11} \end{aligned}$$

Differentiate B.11 with respect to z

$$\begin{aligned} \frac{\partial \hat{c}(z,p)}{\partial z} = & \frac{Pe}{\{\lambda_1-\lambda_2\}} * [\lambda_1^2 * \exp(\lambda_1*z) - \lambda_2^2 * \exp(\lambda_2*z)] * [\hat{c}(z=0,p)/Pe] \\ & - \frac{Pe}{\{\lambda_1-\lambda_2\}} * [\hat{c}_{in}(z=0,p)] * [\lambda_1 * \exp(\lambda_1*z) - \lambda_2 * \exp(\lambda_2*z)] \quad \text{B.12} \end{aligned}$$

Transforming B.4 into the Laplace domain and letting $\hat{c}(z,p) = \mathcal{L}\{c(z,t)\}$

$$\partial \hat{c}(z=1,p) / \partial z = 0 \quad \text{B.13}$$

Applying B.13 to B.12 at point $z=1$ we obtain

$$[\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)] * [\hat{c}(z=0,p) / Pe] = [\hat{c}_{in}(z=0,p)] * [\lambda_1 \exp(\lambda_1 z) - \lambda_2 \exp(\lambda_2 z)]$$

or

$$\hat{c}(z=0,p) = \frac{[\lambda_1 \exp(\lambda_1 z) - \lambda_2 \exp(\lambda_2 z)] * [\hat{c}_{in}(z=0,p)] * Pe}{\{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}}$$

B.14

Substituting B.14 into B.11 results in the following

$$\begin{aligned} \hat{c}(z,p) &= [Pe / \{\lambda_1 - \lambda_2\}] * \{[\lambda_1 \exp(\lambda_1 z) - \lambda_2 \exp(\lambda_2 z)] * [\hat{c}_{in}(z=0,p)] * \\ &\quad [(\lambda_1 \exp(\lambda_1 z) - \lambda_2 \exp(\lambda_2 z)) / \{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}] \\ &\quad - [Pe / \{\lambda_1 - \lambda_2\}] * [\hat{c}_{in}(z=0,p)] * [\exp(\lambda_1 z) - \exp(\lambda_2 z)]\} \end{aligned}$$

or

$$\begin{aligned} \hat{c}(z,p) &= [Pe / \{\lambda_1 - \lambda_2\}] * [\hat{c}_{in}(z=0,p)] * \{[-\exp(\lambda_1 z) - \exp(\lambda_2 z)] + \\ &\quad \{\lambda_1 \exp(\lambda_1 z) - \lambda_2 \exp(\lambda_2 z)\}^2 / \{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}\} \end{aligned}$$

Rewriting this in the form of a transfer function

$$\hat{c}(z,p)/\hat{c}_{in}(z=0,p) = [(-\{\exp(\lambda_1 z) - \exp(\lambda_2 z)\} * \{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z) + \{\lambda_1 \exp(\lambda_1 z) - \lambda_2 \exp(\lambda_2 z)\}^2\}) / \{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}] * [Pe / \{\lambda_1 - \lambda_2\}]$$

or

$$\hat{c}(z,p)/\hat{c}_{in}(z=0,p) = [(-\lambda_1^2 \{\exp(2\lambda_1 z) + \lambda_2^2 \exp(\{\lambda_1 + \lambda_2\} z) + \lambda_1^2 \exp(\{\lambda_1 + \lambda_2\} z) - \lambda_2^2 \exp(2\lambda_2 z) + \lambda_1^2 \exp(2\lambda_1 z) + \lambda_2^2 \exp(2\lambda_2 z) - \lambda_2 \lambda_1 \lambda_2 \exp(\{\lambda_1 + \lambda_2\} z)) / \{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}] * [Pe / \{\lambda_1 - \lambda_2\}]$$

Rewriting the above transfer function yields

$$\frac{\hat{c}(z,p)}{\hat{c}_{in}(z=0,p)} = \frac{\{\lambda_1 - \lambda_2\}^2 \exp(\{\lambda_1 + \lambda_2\} z)}{\{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}} * \frac{Pe}{\{\lambda_1 - \lambda_2\}}$$

or

$$\begin{aligned} \frac{\hat{c}(z,p)}{\hat{c}_{in}(z=0,p)} &= Pe * \frac{\{\lambda_1 - \lambda_2\} \exp(\{\lambda_1 + \lambda_2\} z)}{\{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}} \\ &= \frac{Pe * \{1 - 2\}}{\exp(-\{\lambda_1 + \lambda_2\} z) * \{\lambda_1^2 \exp(\lambda_1 z) - \lambda_2^2 \exp(\lambda_2 z)\}} \end{aligned}$$

Thus

$$\frac{\hat{c}(z,p)}{\hat{c}_{in}(z=0,p)} = \frac{Pe^{*\{\lambda_1-\lambda_2\}}}{\{\lambda_1^2 \exp(-\lambda_2 * z) - \lambda_2^2 \exp(-\lambda_1 * z)\}} \quad \text{B.15}$$

At the end of the bed column at $Z=1$ the transfer function eventually becomes

$$H(p) = \frac{\hat{c}(z=1,p)}{\hat{c}_{in}(z=0,p)} = \frac{Pe^{*\{\lambda_1-\lambda_2\}}}{\{\lambda_1^2 \exp(-\lambda_2) - \lambda_2^2 \exp(-\lambda_1)\}} \quad \text{B.16}$$

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APPENDIX C

SIMPLIFICATION OF MODEL'S TRANSFER
FUNCTION TO COMPLEX FORM

$$H(j\omega) = \frac{Pe \begin{bmatrix} \lambda_1 & -\lambda_2 \\ 1 & 2 \end{bmatrix}}{\begin{matrix} -\lambda_1 & -\lambda_2 \\ 2 & 2 \\ \lambda_1 \cdot e & -\lambda_2 \cdot e \\ 1 & 2 \end{matrix}} \quad \text{C.1}$$

$$\lambda_{1,2} = \frac{1}{2} \left[Pe + \sqrt{Pe^2 + 4 \cdot Pe \cdot \tau \cdot j \cdot \omega} \right]$$

$$\text{Let } a = Pe$$

$$b = 4 \cdot \omega \cdot \tau$$

$$r = \sqrt{a^2 + a \cdot b}$$

$$\theta = \tan^{-1} \frac{b}{a}$$

Therefore

$$\lambda_1 = \frac{1}{2} \left[a + \sqrt{a^2 + a \cdot b \cdot j} \right]$$

$$= \frac{1}{2} \left[a + \left[\sqrt{r \cdot \cos \left[\frac{\theta}{2} \right]} + j \cdot \sqrt{r \cdot \sin \left[\frac{\theta}{2} \right]} \right] \right]$$

$$= \left[\frac{1}{2} a + \frac{1}{2} \sqrt{r \cdot \cos \left[\frac{\theta}{2} \right]} \right] + j \cdot \sqrt{r \cdot \sin \left[\frac{\theta}{2} \right]}$$

$$\lambda_2 = \frac{1}{2} \left[a - \sqrt{a^2 + a \cdot b \cdot j} \right]$$

$$= \frac{1}{2} \left[a - \left[\sqrt{r \cdot \cos \left[\frac{\theta}{2} \right]} + j \cdot \sqrt{r \cdot \sin \left[\frac{\theta}{2} \right]} \right] \right]$$

$$= \left[\frac{1}{2} a - \frac{1}{2} \sqrt{r \cdot \cos \left[\frac{\theta}{2} \right]} \right] + j \cdot \sqrt{r \cdot \sin \left[\frac{\theta}{2} \right]}$$

Now defining the following term as

$$c = \frac{1}{2} \left[a + \sqrt{r \cdot \cos \left[\frac{\theta}{2} \right]} \right]$$

$$d = \frac{1}{2} \left[a - \sqrt{r \cdot \cos \left[\frac{\theta}{2} \right]} \right]$$

$$l = \frac{1}{2} \left[\sqrt{r \cdot \sin \left[\frac{\theta}{2} \right]} \right]$$

Hence

$$\lambda_1 = c + j \cdot l \quad \text{C.2}$$

$$\lambda_2 = d - j \cdot l \quad \text{C.3}$$

$$\text{Let } f = \begin{bmatrix} 2 & 2 \\ c & -1 \end{bmatrix} ; \quad h = \begin{bmatrix} 2 & 2 \\ d & +1 \end{bmatrix}$$

$$g = 2c1 \quad ; \quad k = 2d1$$

Thus

$$\lambda_1^2 = (c + j1)^2 = \begin{bmatrix} 2 & 2 \\ c & -1 \end{bmatrix} + j \cdot 2c1 = f + j \cdot g \quad \text{C.4}$$

$$\lambda_2^2 = (d + j1)^2 = \begin{bmatrix} 2 & 2 \\ d & +1 \end{bmatrix} + 2 \cdot d \cdot 1 \cdot j = h + j \cdot k \quad \text{C.5}$$

$$\text{for } m = e^{-c} \cos(1) \quad ; \quad n = e^{-c} \sin(1)$$

$$o = e^{-d} \cos(1) \quad ; \quad q = e^{-d} \sin(1)$$

Then

$$e^{-\lambda_1 t} = e^{-(c+j1)t} = e^{-c} (\cos(1) - j \sin(1)) = m - j \cdot n \quad \text{C.6}$$

$$e^{-\lambda_2 t} = e^{-(d+j1)t} = e^{-d} (\cos(1) + j \sin(1)) = o + j \cdot q \quad \text{C.7}$$

Substituting c.2, c.3, c.4, c.5, c.6, c.7 in c.1

$$H(j\omega) = Pe \cdot \frac{c - d - j \cdot 2 \cdot 1}{(f + j \cdot g) \cdot (o + j \cdot q) - (h + j \cdot k) \cdot (m - j \cdot n)}$$

$$= Pe \cdot \left[\frac{c - d + j \cdot 2 \cdot 1}{(f \cdot o - g \cdot q - h \cdot m - n \cdot k) + j(f \cdot q + g \cdot o + h \cdot n - m \cdot k)} \right]$$

$$\text{Let } t = f \cdot o - g \cdot q - h \cdot m - n \cdot k$$

$$s = f \cdot q + g \cdot o + h \cdot n - m \cdot k$$

therefore equation c.1 eventually becomes

$$H(j\omega) = Pe \cdot \left[\frac{(c - d) + 2 \cdot j \cdot 1}{t + j \cdot s} \right] \cdot \left[\frac{t - i \cdot s}{t - i \cdot s} \right]$$

$$= Pe \cdot \left[\frac{(c - d) \cdot t + 2 \cdot s \cdot 1}{t^2 + s^2} \right] + j \cdot \left[\frac{-(c - d) \cdot s + 2 \cdot 1 \cdot t}{t^2 + s^2} \right]$$

$$= \text{Re} + j \cdot \text{Im}$$

$$\text{Where } \text{Re} = \left[\frac{Pe}{t^2 + s^2} \right] \cdot ((c - d) \cdot t + 2 \cdot 1 \cdot s)$$

$$\text{Im} = \left[\frac{Pe}{t^2 + s^2} \right] \cdot (-(c - d) \cdot s + 2 \cdot 1 \cdot t)$$

APPENDIX D

CURVE FITTING BETWEEN THEORETICAL RELATION
AND EXPERIMENTAL DATA

Let the following notations be defined as follows :

Pe	=	Peclet Number
$\bar{\tau}$	=	mean residence time
N	=	number of sampling points
ω	=	frequency
i	=	sampling points in the frequency domain
H_i	=	transfer function at the i^{th} point
H_i	=	amplitude of transfer function at point i^{th}
$\arg[H_i]$	=	argument of transfer function at point i^{th}
k	=	sampling point in time domain
F	=	normalized theoretical function in time domain obtained from inversion of transfer function by Inverse Fourier transformation
T	=	theoretical time
r	=	sampling point of experimental data
g_r	=	experimental data at r^{th}
f	=	normalized experimental function
t	=	experimental time

The following curve fitting exercises for three different flow rates illustrate how suitable the proposed model based on Pe and τ obtained from experiment is .

D.1 Curve fitting at high velocity

The following presentation shows plots of magnitude , phase angle and transfer function against frequencies .

$$Pe := 208$$

$$\tau := 50.53148$$

$$N := 64$$

$$d\omega := 0.038$$

$$i := 0 \dots N$$

$$\omega_i := i \cdot d\omega$$

$$\lambda_{1i} := 0.5 \cdot \left[Pe + \sqrt{Pe^2 + 4 \cdot Pe \cdot \tau \cdot \omega_i \cdot j} \right]$$

$$\lambda_{2i} := Pe - \lambda_{1i}$$

$$H_i := \frac{Pe \cdot \left[\lambda_{1i} - \lambda_{2i} \right]}{\left[\left[\lambda_{1i} \cdot \lambda_{1i} \right] \exp \left[-\lambda_{2i} \right] \right] - \left[\left[\lambda_{2i} \cdot \lambda_{2i} \right] \cdot \exp \left[-\lambda_{1i} \right] \right]}$$

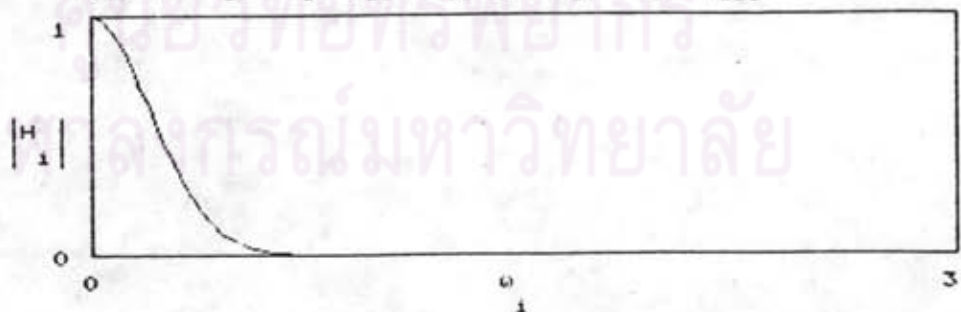


Fig. D1 Curve showing magnitude of transfer function vs frequency for $Pe = 208$ and $\tau = 50.53148$

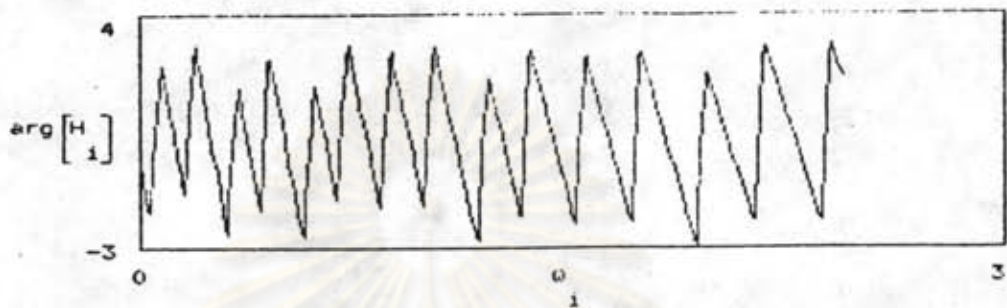


Fig. D2 The Argument of Transfer Function vs
Frequency for $Pe = 208$ and $\zeta = 50.53148$

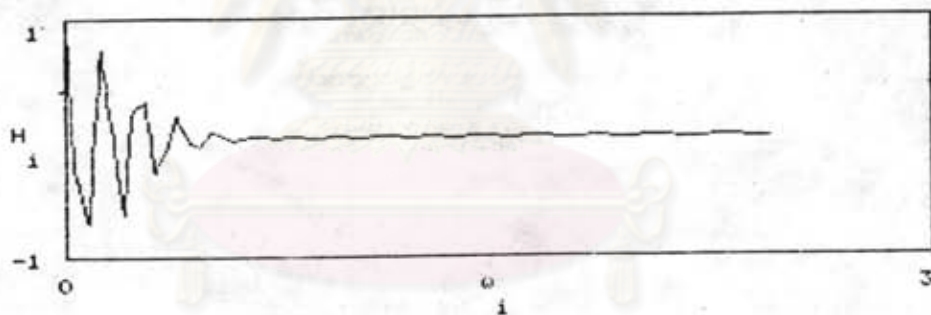


Fig. D3 Curve of Transfer Function vs Frequency
for $Pe = 208$ and $\zeta = 50.53148$

The transformation of the functions from the frequency domain to the time domain can be performed in the following manner .

$$k := 0 \dots 127$$

$$F := 0.0598 \cdot \text{fft}[H]$$

$$T_k := k \cdot \frac{2 \cdot \pi}{2 \cdot N \cdot d\omega}$$

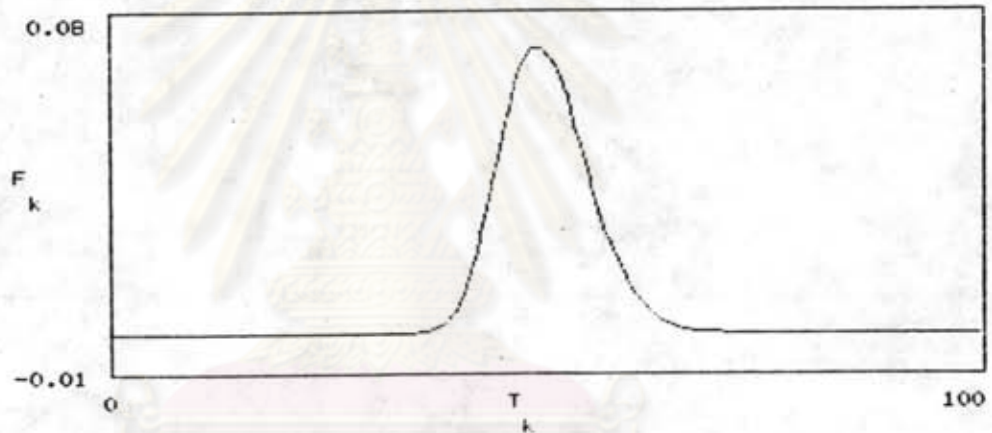


Fig. D4 Theoretical Normalized Concentration-time

Curve for $Pe = 208$ and $\tau = 50.53148$

The experimental concentration distribution data and the comparison of the experimental response curve with the theoretical curve are presented as follows :-

$r = 1 \dots 42$

q	t
0	32.4
2	36.285
3.2	37.17
5	38.055
7.2	38.94
10.2	39.825
19	40.71
26	41.595
36	42.48
42	43.365
60.2	44.25
82.2	45.135
97	46.02
109	46.905
120	47.79
132	48.675
137	49.56
139	50.445
136.2	51.33
132.2	52.215
119.2	53.1
109.2	53.985
96.2	54.87
84.2	55.755
72	56.64
58	57.525
45	58.41
31	59.295
18	60.18
8	61.065
3	61.95
2.2	62.835
2	63.72
1.2	64.605
1	65.49
1	66.375
1	67.26
0.2	68.145
0.2	69.03
0.2	69.915
0.2	70.8
	71.685

$r = 0.00051164 q$

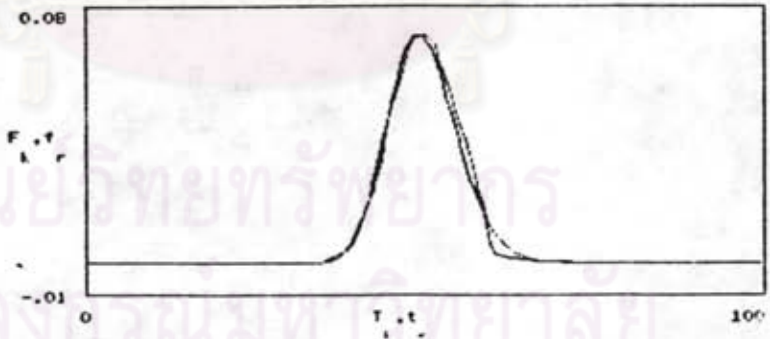


Fig. D5 Comparison of experimental response curve and theoretical response for $Pe = 208$ and $\tau = 50.53148$

D.2 Curve fitting at medium velocity

The following presentation shows the plots of magnitude, phase angle and transfer function ^a against frequencies.

$$Pe := 230$$

$$\tau := 82.39171$$

$$N := 64$$

$$d\omega := 0.019$$

$$i := 0 \dots N$$

$$\omega_i := i \cdot d\omega$$

$$\lambda_{1i} := 0.5 \cdot \left[Pe + \sqrt{Pe^2 + 4 \cdot Pe \cdot \tau \cdot \omega_i \cdot j} \right]$$

$$\lambda_{2i} := Pe - \lambda_{1i}$$

$$H_i := \frac{Pe \cdot [\lambda_{1i} - \lambda_{2i}]}{[[\lambda_{1i} \cdot \lambda_{1i}] \cdot \exp[-\lambda_{2i}]] - [[\lambda_{2i} \cdot \lambda_{2i}] \cdot \exp[-\lambda_{1i}]]}$$

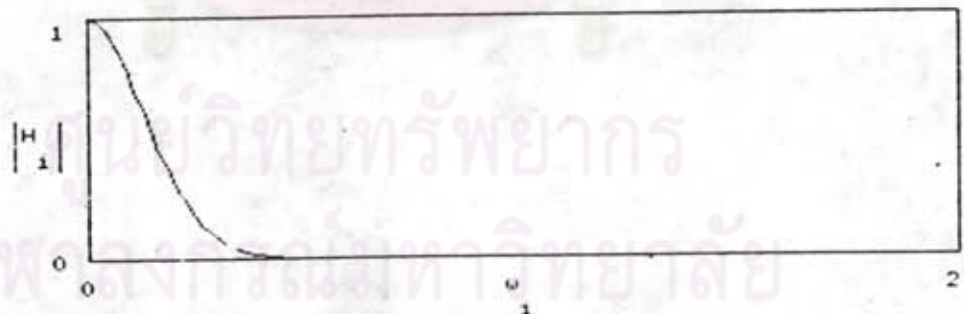


Fig. D6 Curve showing magnitude of transfer function vs frequency for $Pe = 230$ and $\tau = 82.39171$

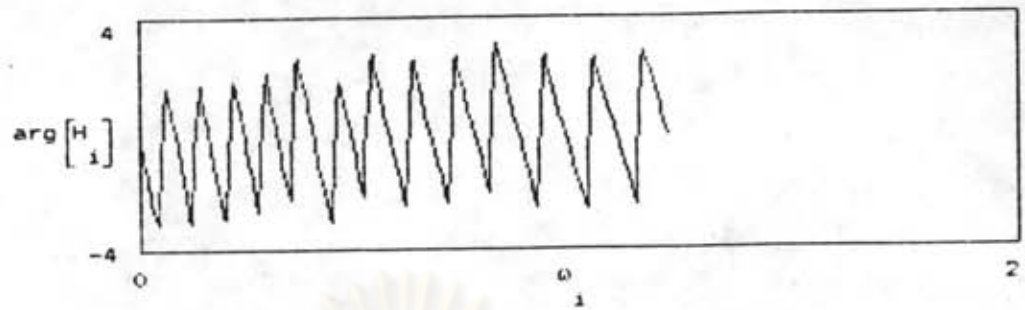


Fig. D7 The Argument of Transfer Function vs Frequency for $Pe = 230$ and $\zeta = 82.39171$

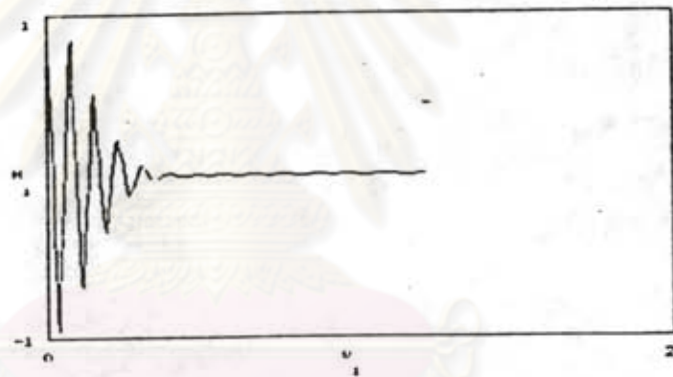


Fig. D8 Curve of Transfer Function vs Frequency for $Pe = 208$ and $\zeta = 50.53148$

The transformation of the functions from the frequency domain to the time domain can be performed in the following manner .

$$k := 0 \dots 127$$

$$F := 0.0723 \cdot \text{ifft}[H]$$

$$T_k := k \cdot \frac{2 \cdot \pi}{2 \cdot N \cdot d_0}$$

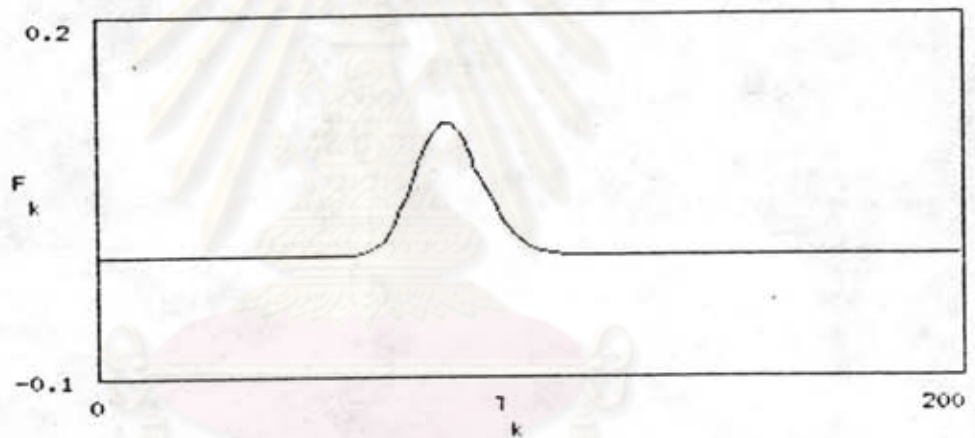


Fig. D9 Theoretical Normalized Concentration-time

Curve for $Pe = 230$ and $\tau = 82.39171$

The experimental concentration distribution data and the comparison of the experimental response curve with the theoretical curve are presented as follows :-



D.3 Curve fitting at low velocity

The following presentation shows the plots of magnitude , phase angle and transfer function gainst frequencies .

Pe := 260
 τ := 209.0635
 N := 64
 dω := 0.012
 i := 0 ..N
 ω_i := i · dω

$$\lambda_{1i} := 0.5 \cdot \left[Pe + \sqrt{Pe^2 + 4 Pe \cdot \tau \cdot \omega_i \cdot j} \right]$$

$$\lambda_{2i} := Pe - \lambda_{1i}$$

$$H_i := \frac{Pe \left[\lambda_{1i} - \lambda_{2i} \right]}{\left[\left[\lambda_{1i} \cdot \lambda_{1i} \right] \exp \left[-\lambda_{2i} \right] \right] - \left[\left[\lambda_{2i} \cdot \lambda_{2i} \right] \exp \left[-\lambda_{1i} \right] \right]}$$

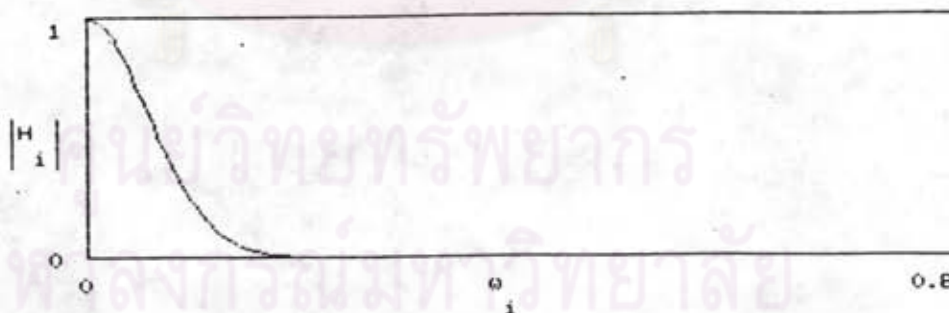


Fig. D11 Curve showing magnitude of transfer function vs frequency for Pe = 260 and τ = 209.0635

D.3 Curve fitting at low velocity

The following presentation shows the plots of magnitude , phase angle and transfer function gainst frequencies .

$$Pe := 260$$

$$\tau := 209.0635$$

$$N := 64$$

$$d\omega := 0.012$$

$$i := 0 \dots N$$

$$\omega_i := i \cdot d\omega$$

$$\lambda_{1i} := 0.5 \cdot \left[Pe + \sqrt{Pe^2 + 4 Pe \cdot \tau \cdot \omega_i \cdot j} \right]$$

$$\lambda_{2i} := Pe - \lambda_{1i}$$

$$H_i := \frac{Pe \cdot \begin{bmatrix} \lambda_{1i} & -\lambda_{2i} \end{bmatrix}}{\left[\begin{bmatrix} \lambda_{1i} & \lambda_{1i} \end{bmatrix} \cdot \exp[-\lambda_{2i}] \right] - \left[\begin{bmatrix} \lambda_{2i} & \lambda_{2i} \end{bmatrix} \cdot \exp[-\lambda_{1i}] \right]}$$

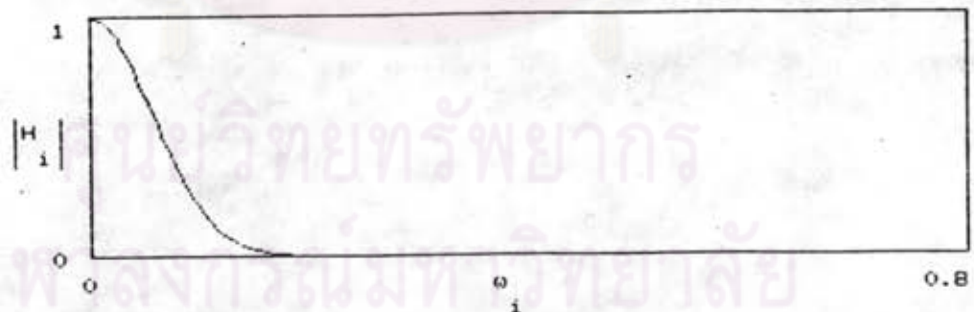


Fig. D11 Curve showing magnitude of transfer function vs frequency for $Pe = 260$ and $\tau = 209.0635$

$$t := 0 \dots 127$$

$$F := 0.034 \text{ ifft}[H]$$

$$T_k := k \frac{2\pi}{2 \cdot N \cdot \Delta\omega}$$

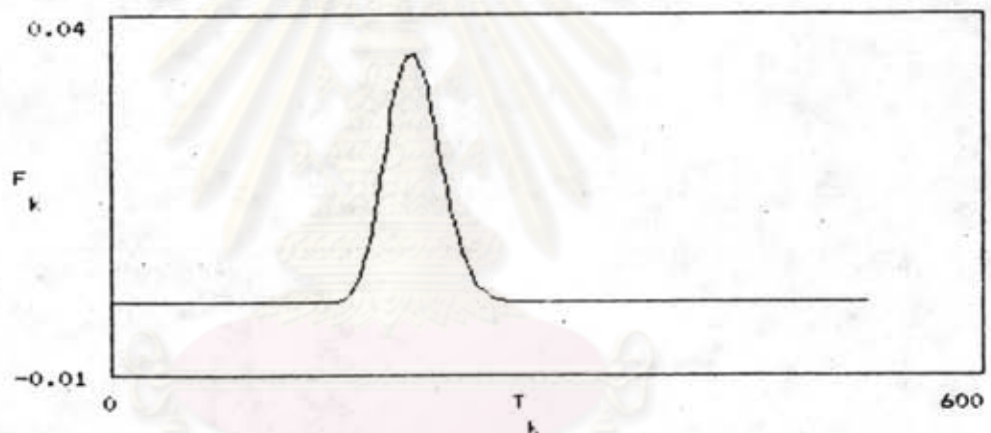


Fig. D14 Theoretical Normalized Concentration-time
Curve for $Pe = 260$ and $\zeta = 209.0635$

The experimental concentration distribution data and the comparison of the experimental response curve with the theoretical curve are presented as follows :-

$F = 0.77$

q_0	r
0.5	68
0.5	63
0.5	56
0.5	52
1	50
1	43
1.5	39
1.5	34
2	29
2.5	25
4.5	22
6	18
8.5	15
10	11
13.5	8
17	6
18.5	4
22	2.5
27	2
35	2
40	2
46.5	1.5
53	1
60	1
66	0.5
72	0.5
78	0.5
86	0.5
91	0.5
97.5	
103	
105.5	
109.5	
112	
113.5	
115	
115.5	
116.5	
115.5	
114	
112	
109.5	
106	
102	
95	
89	
85	
80	
75	

q_{50}	r
68	
63	
56	
52	
50	
43	
39	
34	
29	
25	
22	
18	
15	
11	
8	
6	
4	
2.5	
2	
2	
2	
1.5	
1	
1	
0.5	
0.5	
0.5	
0.5	
0.5	

t_0	r
146.4	
148.0827	
149.7652	
151.4482	
153.1310	
154.8137	
156.4965	
158.1793	
159.8620	
161.5448	
163.2275	
164.9103	
166.5931	
168.2758	
169.9586	
171.6413	
173.3241	
175.0068	
176.6896	
178.3724	
180.0551	
181.7379	
183.4206	
185.1034	
186.7862	
188.4689	
190.1517	
191.8344	
193.5172	
195.2	
196.8827	
198.5655	
200.2482	
201.9310	
203.6137	
205.2965	
206.9793	
208.6620	
210.3448	
212.0275	
213.7103	
215.3931	
217.0758	
218.7586	
220.4413	
222.1241	
223.8068	
225.4896	
227.1724	

t_{50}	r
228.8551	
230.5379	
232.2206	
233.9034	
235.5862	
237.2689	
238.9517	
240.6344	
242.3172	
244	
245.6827	
247.3655	
249.0482	
250.7310	
252.4137	
254.0965	
255.7793	
257.4620	
259.1448	
260.8275	
262.5103	
264.1931	
265.8758	
267.5586	
269.2413	
270.9241	
272.6068	
274.2896	
275.9724	

$$q_r = 1/r \left[r \pm 48, q_0, q_{50} \right]_{r-49}$$

$$t_r = 1/r \left[r \pm 48, t_0, t_{50} \right]_{r-49}$$

$$f_r = 0.00029477 q_r$$

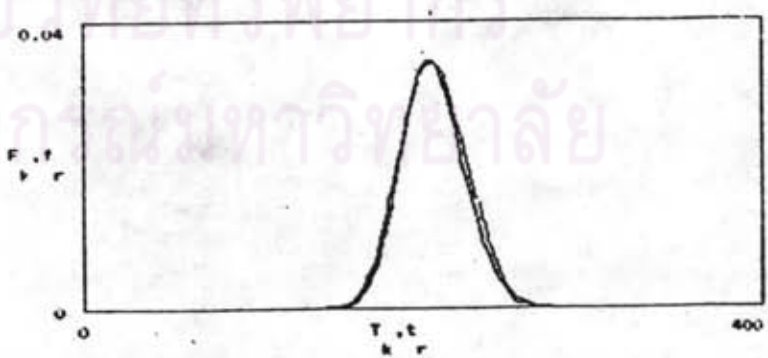


Fig.D15 Comparison of experimental response curve and theoretical response for $Pe = 260$ and $\tau = 209.0635$

APPENDIX E

MOMENT'S METHOD FOR DERIVATION OF
MEAN RESIDENCE TIME AND VARIANCE

From the transfer function given in equation 2.6

$$H(s) = \frac{Pe*(\lambda_1 - \lambda_2)}{[\lambda_1^2 * \exp(-\lambda_2) - \lambda_2^2 * \exp(-\lambda_1)]}$$

let $q = \sqrt{1 + 4s/Pe}$, $U = (Pe/2)*(1+q)$, $V = (Pe/2)*(1-q)$

$$\begin{aligned} H(s)_{z=1} &= \frac{Pe*[(Pe/2)*(2*q)]}{\frac{Pe^2*(1+q)^2 * e^{-(Pe/2)*(1-q)}}{4} - \frac{Pe^2*(1-q)^2 * e^{-(Pe/2)*(1+q)}}{4}} \\ &= \frac{4*q*e^{[(Pe/2)*(1-q)]}}{[(1+q)^2 - (1-q)^2 * e^{-\{(Pe/e)*\{1+q\} + \{(Pe/2)*\{1-q\}\}}]} \\ &= \frac{4*q*e^V}{[(1+q)^2 - (1-q)^2 * e^{(-U+V)}]} \end{aligned}$$

E.1

Therefore equation E.1 eventually becomes

$$H(s) = \frac{4q \exp\left(\frac{Pe}{2}\right) \{1-q\}}{[(1+q)^2 - (1-q)^2 \exp(-Peq)]} \quad E.2$$

since

$$\frac{dH}{ds} = \frac{dH \cdot dq}{dq \cdot ds}$$

and

$$\frac{dq}{ds} = \frac{1}{2} (1 + 4s/Pe)^{-1/2} (4/Pe) = \frac{2\mathcal{Z}}{Pe \sqrt{1+4s\mathcal{Z}/Pe}} = 2\mathcal{Z}/Pe$$

$$\frac{dH}{dq} = 4 \left[\frac{[(1+q)^2 - (1-q)^2 e^{-Peq}] [-q \{Pe/2\} e^{\{Pe/2\}(1-q)} + e^{\{Pe/2\}(1-q)}]}{[(1+q)^2 - (1-q)^2 e^{-Peq}]^2} \right]$$

$$- 4 \left[\frac{[q e^{\{Pe/2\}(1-q)}] [2(1+q) - \{-2(1-q)e^{-Peq} + (1-q)^2 (-Pe) e^{-Peq}\}]}{[(1+q)^2 - (1-q)^2 e^{-Peq}]^2} \right]$$

Thus

$$\frac{dH \cdot dq}{dq \cdot ds} = \frac{8\mathcal{Z}}{Pe^2 q} \left[\frac{[(1+q)^2 - (1-q)^2 e^{-Peq}] [q \{Pe/2\} e^{\{Pe/2\}(1-q)} + e^{\{Pe/2\}(1-q)}]}{[(1+q)^2 - (1-q)^2 e^{-Peq}]^2} \right]$$

$$- \frac{8\mathcal{Z}}{Pe^2 q} \left[\frac{[q e^{\{Pe/2\}(1-q)}] [2(1+q) - \{-2(1-q)e^{-Peq} + (1-q)^2 (-Pe) e^{-Peq}\}]}{[(1+q)^2 - (1-q)^2 e^{-Peq}]^2} \right]$$

or

$$\frac{dH}{ds} = \zeta \frac{[(1+q)^2 - (1-q)^2 e^{-Pe*q}] \{ \frac{4(8/Pe*q) - 4}{Pe*q} \} e^{Pe/2} \{1-q\}}{[(1+q)^2 - (1-q)^2 e^{-Pe*q}]^2}$$

$$- \zeta \left[\frac{[4*q*e^{Pe/2}*{1-q}] \frac{4*(1+q) - \{-2(1-q)e^{-Pe*q} + (1-q)^2*(2/q)*e^{-Pe*q}\}}{Pe*q}}{[(1+q)^2 - (1-q)^2 e^{-Pe*q}]^2} \right]$$

$$\begin{aligned} \lim_{s \rightarrow 0} \frac{-dH}{ds} &= \left[\frac{-(2)^2 \{ \frac{4(8/Pe) - 4}{Pe} \} + 4(8/Pe)}{(2)^4} \right] \\ &= \left[\frac{-32/Pe + 16 + 32/Pe}{16} \right] \\ &= \zeta \end{aligned}$$

E.3

According to equation 3.36

$$\mu_1 = \bar{\tau} = \zeta$$

let the following letters be defined as :

$$G = \frac{4*(1+q) - \{-2(1-q)e^{-Pe*q} + (1-q)^2*(2/q)*e^{-Pe*q}\}}{Pe*q}$$

$$H = e^{\{Pe/2\}(1-q)}$$

$$I = \{(8/Pe*q)-4\}$$

$$J = [(1+q)^2 - (1-q)^2 e^{-Pe*q}]$$

$$K = H*I*J$$

$$L = 4*q*H*G$$

$$M = e^{-Pe*q + \{Pe/2\}(1-q)}$$

Therefore the second order differentiation of the transfer function is

$$\frac{\partial^2 H}{\partial s^2} = \frac{(\partial H / \partial s)}{\partial s} = \frac{J^2 * \left[\frac{\tau \partial K}{\partial s} - \tau \partial L \right]}{(J^2)^2} - \left[\frac{K - L}{J^4} \right] \tau \frac{\partial J^2}{\partial s} \quad E.4$$

Now consider the following differentiations

$$\begin{aligned} \frac{\partial K}{\partial s} &= J * \frac{\partial I * H}{\partial s} + I * H * \frac{\partial J}{\partial s} \\ &= J * [I * H * (-\tau/q) - \{16\tau/Pe^2 * q^3\} * H] + I * H * G * \tau \end{aligned} \quad E.5$$

$$-\frac{\partial L}{\partial s} = -\frac{\partial\{16/Pe\}H}{\partial s} + \frac{\partial\{16*q/Pe\}H}{\partial s} - \frac{\partial 8*M}{\partial s} + \frac{\partial 16*q*M}{\partial s}$$

$$\frac{\partial 8*q^2*M}{\partial s} - \frac{\partial\{16/Pe\}M}{\partial s} + \frac{\partial\{16*q/Pe\}M}{\partial s}$$

$$= \{\tau*16/(Pe*q)\}H + \{\tau*16/Pe\}H - \{\tau*32/(Pe^2*q)\}H +$$

$$\{\tau*24/q\}M - \{48*\tau\}M + \{\tau*32/(Pe*q)\}M + 24*\tau*q*M -$$

$$\{\tau*32/Pe\}M + \{\tau*48/(Pe*q)\}M - \{\tau*48/Pe\}M + \{32/(Pe^2*q)\}M \quad E.6$$

$$\frac{\partial J^2}{\partial s} = 2*J*\tau*G \quad E.7$$

Substituting E.5 to E.7 into E.4 , then taking the limit of such a relationship . As s approaches zero , we obtain

$$\lim_{s \rightarrow 0} \frac{\partial^2 H}{\partial s^2} = \frac{\tau \partial J^2}{(J^2)^2} * \lim_{s \rightarrow 0} \frac{\partial K}{\partial s} - \frac{\tau J^2}{(J^2)^2} * \lim_{s \rightarrow 0} \frac{\partial L}{\partial s} -$$

$$\tau \frac{K - L}{J^4} * \lim_{s \rightarrow 0} \frac{\partial J^2}{\partial s}$$

$$= \frac{-4*\tau^2}{Pe} + \tau^2 + \frac{2*\tau^2}{Pe} - \frac{2*\tau^2}{Pe^2} + \frac{2*\tau^2*e^{-Pe}}{Pe^2} + \frac{4*\tau^2}{Pe} \quad E.8$$

By subtracting E.3 from E.8 , this yields equation 3.37 .

$$\zeta^2 = \frac{2\tau^2}{Pe} - \frac{2\tau^2}{Pe^2} + \frac{2\tau^2 e^{-Pe}}{Pe^2} \quad \text{E.9}$$



ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX F ·

DETERMINATION OF PARTICLE SIZES

The crushed particles were sieved with 30 - 50 mesh screens and determined by a microscope with a spectromicrometer . By using Ferret's method as depicted in figure F1, the particle size distributions appear to have the same shape as in figure F2. The surface mean diameters with the following relationship were utilized to obtain the average diameter .

$$\text{surface mean diameter} = d_{p\sum n} = \left(\frac{\sum n d_p^2}{\sum n} \right)^{0.5}$$



Fig.F1 Ferret's diameter , must be the longest dimension along the line parallel to the base of the field of view

The obtained surface mean diameters were 0.4919 , 0.6314 ,
0.8075 and 0.986 mm .



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Fig.F.2 Particle size distribution with an
average of 0.6314 mm .

APPENDIX G

SAMPLE CALCULATIONS

The following sample calculations are based on system H-4-0.9 at volumetric flow rate 0.81 cc/sec

1. Calculation for Density of Propane

$$\text{From} \quad PV = nRT$$

$$P = \left(\frac{m}{V}\right) \left(\frac{R}{MW}\right) * T = \rho * K * T$$

$$\text{At constant pressure :} \quad \rho_1 T_1 = \rho_2 T_2$$

$$\rho_{\text{propane}}(0^\circ\text{C}, 1 \text{ atm}) = 1.97 \times 10^{-3} \quad \text{g/cc(G.1)}$$

$$\text{so } \rho_{\text{propane}}(45^\circ\text{C}, 1 \text{ atm}) = \frac{(1.97 \times 10^{-3})(273.15)}{318.15}$$

$$= 1.691 \times 10^{-3} \quad \text{g/cc}$$

$$= 1.691 \quad \text{Kg/m}^3$$

G.1/ Data obtained from Handbook of tables for Applied Engineering Science , 2nd ed., The Chemical Rubber Co., 1973 , pp. 57 .

2. Calculation for Superficial Velocity

$$U = Q/A = \frac{Q}{(\pi/4) \cdot d_t^2}$$

$$\text{For } Q = 10/12.38 = 0.81 \text{ cc/sec}$$

$$d_t = 0.45 \text{ inch} = 1.14 \text{ cm}$$

$$\text{Thus } U = \frac{0.8078}{(0.7854)(1.14)^2} = 0.7936 \text{ cm/sec}$$

$$= 7.936 \times 10^{-3} \text{ m/sec}$$

3. Calculation for Reynolds Numbers

$$\text{From } Re = \frac{U \rho d_p}{\mu}$$

$$\mu_{\text{propane}(45^\circ\text{C}, 1 \text{ atm})} = 0.875 \times 10^{-2} \text{ cp(G.2)}$$

$$= 8.75 \times 10^{-6} \text{ Pa*sec}$$

$$d_p = 0.9861 \text{ mm}$$

$$= 9.861 \times 10^{-4} \text{ m}$$

$$\text{so } Re = \frac{(7.936 \times 10^{-3})(1.691)(9.861 \times 10^{-4})}{(8.75 \times 10^{-6})}$$

$$= 1.5124$$

4. Calculation for The Mean Residence Time

From equation 3.30 , 3,36

$$\begin{aligned} \mu_{ty} &= \frac{\sum t c_e(t) \Delta t}{\sum c_e(t) \Delta t} = \sum t \left(\frac{c_e(t)}{\sum c_e(t)} \right) \\ &= \sum t F_y(t) = -\lim_{s \rightarrow 0} \frac{dY(s)}{ds} \\ \mu_{ty} &= -\lim_{s \rightarrow 0} \frac{dH(s)}{ds} = \bar{t} \end{aligned}$$

From value in table G.1

$$\sum c_e(t) = 1954.5$$

and $\mu_{\text{packed bed}} = 50.53$

A similar calculation for μ_{blank} is done by using data in table G.2

so $\mu_{\text{blank}} = 2.56$

G.2/ Data from monograph in Perry Handbook pp.21 .

5. Calculation for The Variance

From equation 3.31

$$\begin{aligned}\sigma_{ty}^2 &= \frac{\sum (t-\mu)^2 c_e(t) \Delta t}{\sum c_e(t) \Delta t} = \frac{\sum (t-\mu_{ty})^2 \left(\frac{c_e(t)}{\sum c_e(t)} \right)}{\sum c_e(t)} \\ &= \sum (t-\mu_{ty})^2 F_y(t) = \lim_{s \rightarrow 0} \left[\frac{d^2 Y(s)}{ds^2} - \left(\frac{dY(s)}{ds} \right)^2 \right]\end{aligned}$$

From table G.1

Hence $\sigma_{\text{packed bed}}^2 = 23.06$

A similar calculation for σ_{blank}^2 is done by using data obtained from table G.1

therefore $\sigma_{\text{blank}}^2 = 1.04$

6. Calculation for The Peclet Number

From equation 3.37

$$\frac{\sigma^2}{\bar{t}^2} = \frac{2}{Pe} - \frac{2}{Pe^2} + \frac{2 * e^{-Pe}}{Pe^2}$$

$$\sigma^2 = \sigma_{\text{packed bed}}^2 - \sigma_{\text{blank}}^2$$

$$= 23.06 - 1.04$$

$$= 22.02$$

$$\tau^2 = (t_{\text{packed bed}} - t_{\text{blank}})^2$$

$$= (50.53 - 2.56)^2$$

$$= 2301.12$$

Using the program shown in appendix H , the Peclet value obtained was 208

7. Calculation for Axial Dispersion

From $E_z = U \cdot L / Pe$

where $L = 6.0 \times 10^{-1} \text{ cm}$

Then

$$E_z = \frac{(7.936 \times 10^{-3})(6.0 \times 10^{-1})}{(208)}$$

$$= 2.289 \times 10^{-5} \text{ m}^2/\text{sec}$$

$$= 0.2289 \text{ cm}^2/\text{sec}$$

8. Calculation for Particle Peclet Number

$$Pe_p = \frac{U \cdot d_p}{E_z}$$

$$= \frac{(7.936 \times 10^{-3})(9.861 \times 10^{-4})}{(2.289 \times 10^{-5})}$$

$$= 0.3419$$

Table G.1 Data , mean and variance for sample calculation

t (sec)	$C_u(t)$	$F_y(t)$	$tF_y(t)$	$(t-\mu)^2 F_y(t)$
35.40	0	0	0.0371	0.2091
36.28	2	0.0010	0.0666	0.3220
37.17	3.5	0.0017	0.0974	0.4013
38.05	5	0.0025	0.1494	0.5199
38.94	7.5	0.0038	0.2139	0.6214
39.83	10.5	0.0053	0.3957	0.9469
40.71	19	0.0097	0.5533	1.0739
41.60	26	0.0133	0.7824	1.2084
42.48	36	0.0184	0.9984	1.1984
43.37	45	0.0230	1.370	1.2402
44.25	60.5	0.0309	1.9744	1.2968
45.14	85.5	0.0437	2.2839	1.0318
46.02	97	0.0496	2.6168	0.7531
46.91	109	0.0557	2.9342	0.4778
47.79	120	0.0613	3.3122	0.2469
48.68	133	0.0680	3.4739	0.0729
49.56	137	0.0700	3.5876	0.0013
50.45	139	0.0711	3.5848	0.0393
51.33	136.5	0.0698	3.5397	0.1813
52.22	132.5	0.0678	3.2468	0.3884
53.10	119.5	0.0611	3.0244	0.6497
53.99	109.5	0.0560	2.7091	0.9088
54.87	96.5	0.0494	2.4105	1.1580
55.76	84.5	0.0432	2.0865	1.3530
56.64	72	0.0368	1.7071	1.4314
57.53	58	0.0297	1.3448	1.4117
58.41	45	0.0230	0.9405	1.2047
59.30	31	0.0158	0.3387	0.5187

Table G.1 (continued)

t (sec)	$C_u(t)$	$F_y(t)$	$tF_y(t)$	$(t-\mu)^2 F_y(t)$
60.18	11	0.0056	0.3387	0.5187
61.07	5	0.0026	0.1562	0.2813
61.95	4	0.0020	0.1268	0.2646
62.84	3	0.0015	0.0964	0.2305
63.72	2.5	0.0013	0.0815	0.2209
64.61	2	0.0010	0.0661	0.2013
65.49	1.5	0.0007	0.0503	0.1706
66.38	1	0.0005	0.0340	0.1276
67.26	1	0.0005	0.0344	0.1424
68.15	1	0.0005	0.0349	0.1579
69.03	0.5	0.0003	0.0177	0.0871
69.92	0.5	0.0003	0.0179	0.0956
70.8	0.5	0.0003	0.0181	0.1046
71.69	0.5	0.0003	0.0183	0.1140
	1954.5		50.5315	23.0652

Therefore $\mu_{\text{packed bed}} = 50.53$
 $\sigma^2_{\text{packed bed}} = 23.07$

Table G.1 Data, mean and variance for blank

t (sec)	$C_u(t)$	$F_y(t)$	$tF_y(t)$	$(t-\mu)^2 F_y(t)$
1.8	16.2	0.6183	1.1130	0.3531
3.6	9	0.3435	1.2366	0.3746
5.4	1	0.0382	0.2061	0.3087
	26.2		2.5557	1.0365

Therefore $\mu_{\text{blank}} = 2.56$ and $\sigma^2_{\text{blank}} = 1.04$

The flow chart shown can be written in basic as follows :

```

5   CLS
10  INPUT " RATIO OF VARIANCE AND SQUARE OF MEAN RESIDENCE TIME "; R
20  IF R >= 0.0001 AND R < 0.019 THEN PE = 2000 : GOTO 40
30  PE = 200
40  GOSUB 250
50  IF PE > 200 THEN 60 ELSE 150
60  IF A < R THEN 70 ELSE 90
70  PE = PE - 100 : GOSUB 250
80  GOTO 60
90  PE = PE + 10
100 GOSUB 250
110 IF A > R THEN 90 ELSE 120
120 IF ABS(R-A) <= 6.5 x 10-5 THEN 240 ELSE 130
130 Pe = Pe - 1
140 GOSUB 250 : GOTO 120
150 IF A < R THEN 160 ELSE 180
160 Pe = Pe -10
170 GOSUB 250 : GOTO 150
180 Pe = Pe + 1
190 GOSUB 250
200 IF A > R THEN 180 ELSE 210
210 Pe = Pe - 0.1
220 GOSUB 250
230 IF ABS(R-A) <= 1.5 x 10-5 THEN 240 ELSE 210
240 PRINT A , R : END
250 A = (2/Pe) - (2/Pe^2) + (2/Pe^2)*EXP(-1*Pe)
260 RETURN

```

APPENDIX I

EXAMPLES OF CHROMATOGRAPHIC PEAKS



Fig. 11 Three impulse responses of methane tracer in propane at a flow rate of 0.81 cc/sec .

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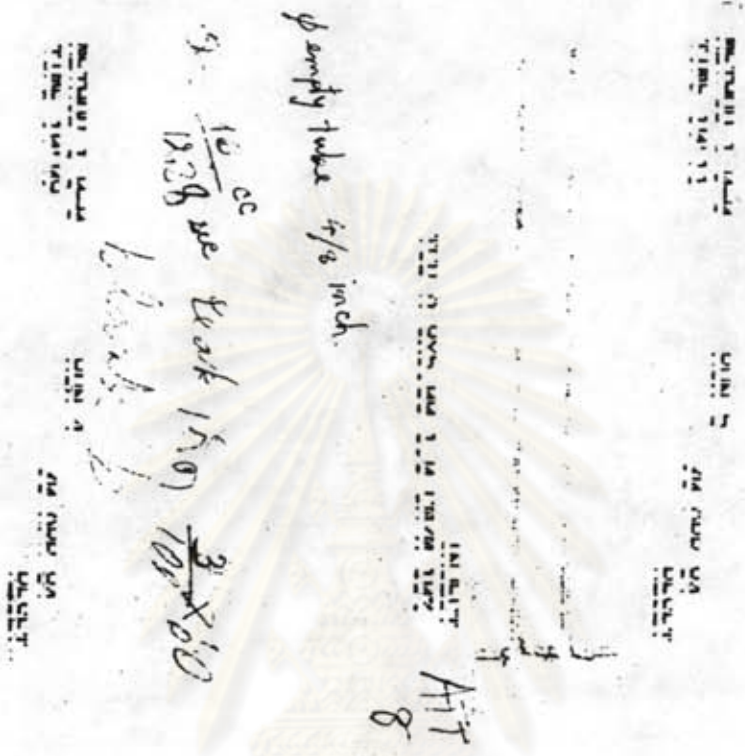


Fig. 12 Two chromatographic peaks of blank performances at flow rate of 0.81 cc/sec .

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METHOD 1 0-0
TIME 11:03
RESET

PIN 5 30 APP :4

INJECT
TCD 0 8x5.00 1.0 CM/M 95%

Ar + CO₂



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Fig. I3 The chromatogram showing adsorption of carbon dioxide by MSC-3A

INJECT
TCD 8x5.00 1.0 CM/M 95%

2.56 MIN/ML AT 1.00 CM/M 95%
1.40 ML



Fig. I4 Chromatograph showing that methane is not adsorbed by MSC-3A adsorbents

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AUTOBIOGRAPHY

Chukiat Chailitlerd received a Bachelor of Pharmacy from Chiangmai University in 1985 . He had worked with the Organon (Thailand) Co.,Ltd. He then applied to Chulalongkorn University's Program of Petrochemical Technology in 1987 .



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