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CHAPTER 6

DISCUSSIONS

In this section the following items are discussed;

- 1. The assumptions implied in the mathematical model .
- 2. The influence of controlled variables .
- 3. The prediction of axial dispersion coefficient .
- Comparative analysis for experimental axial dispersion coefficient .

6.1 Assumptions implied in the mathematical model

The derivation of equation 2.6 involves various assumptions such as ;

- 1. The absence of radial gradients .
- 2. The existence of a uniform velocity profile .

6.1.1 Radial gradients

Although McHenry (8) mentioned that radial diffusivity is about sixfold less than axial diffusivity for turbulent flow of gases among particles, Smith et al (30) demonstrated that unusual radial velocity profiles can be encountered in the flow of gases through packed bed at every flow rate. They attributed these apparent anomalies to variation of void fraction with radial position in the packed bed. Smith indicated that radial variations of velocities



exist in packed columns of various sizes and operating conditions. However large radial variations of velocities would be expected to result in a bed curve fitting between theoretical and experimental curves which is not the case as shown in appendix D .

A good fit between a theoretical curve to an experimental curve does not completely imply that there are no radial gradients. It may simply mean that the axial dispersion coefficient can describe whatever contribution radial gradients have on axial dispersion.

6.1.2 Velocity profile

Volkov et al (9) indicated that the fluctuation of the velocity field arised from the influence of packed bed parameter such as particle size distribution, mean particle size, the compaction method and the tube diameter. In case where the average particle sizes is large, where size distributions are wide, where particles are packed loosely, and where tube diameters are large, it has been found that interstitial velocity varies to some extent depending on the velocity measuring point within the bed. Otherwise the uniform velocity assumption can be valid.

6.2 Influence of controlled variables

The attention of this study is confined to the influence of three controlled variables on the axial dispersion phenomena, particle size, tube diameter and flow rate.

6.2.1 Particle size

The graphical representation shown in figure 5.1, 5.2, 5.3, 5.4 and 5.5 infers that at a given tube diameter and a specified Reynolds number (or velocities for figure 5.5), an increase in particle size gives rise to lesser axial dispersion. Volkov et al (9) presented a study for the elution of hydrogen by nitrogen in a 100 mm diameter tube and using three mean particle sizes of 0.36, 0.52 and 0.75 mm as shown in figure 6.1.

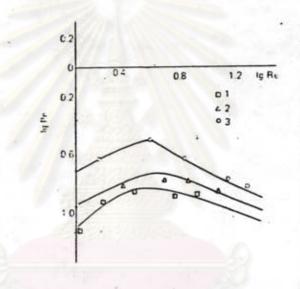


Fig. 6.1 Peclet numbers vs Reynolds numbers for dp equal to 0.36 (1), 0.52 (2) and 0.75 (3).

Their result shows a similar particle size effect with that of this work . In contrast, the plots of $E_{\rm Z}({\rm cm^2/s})$ versus $d_{\rm P}$ as in figure 6.2 using Hsu & Hyne's prediction shows a relationship opposite to that of ours . This controversial aspect reflects that the dispersion behaviour of small particle size does not obay the relationship given in equation 2.3 , especially , the particle size diameter term . Therefore some corrections are required to provide a more reliable prediction for small particles .

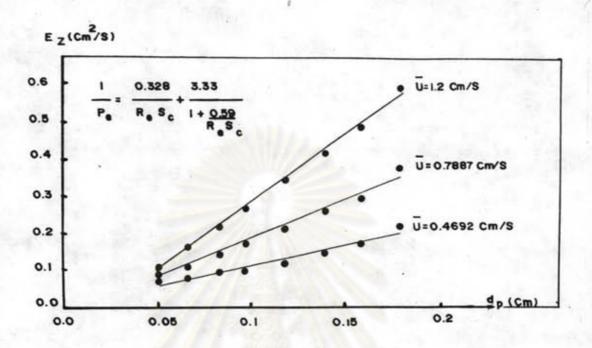


Fig. 6.2 Relationship between particle size and axial dispersion coefficientbased on Hsu & Hyne's equation .

6.2.2 Tube diameter

This study uses small diameter beds packed with very small sized particles; data in the literature is generally devoted to larger diameter beds with particle sizes larger than those used in this study.

According to Gun's suggestion (20), we selected tube to particle diameter ratios above ten so as to eliminate wall effects. In addition to ensure the elimination of end effects as suggested by Scott et al (34), bed length to particle diameter ratios larger than 50 were also used in this study.

Figures 5.6 , 5.7 , 5.8 , 5.9 and 5.10 indicated effects of variations in tube diameter . As smaller bed diameter are used , the lesser the axial dispersion .

6.2.3 Flow rate

Since velocity is directly proportional to flow rate, so figure 5.11 presents the effect of flow rate upon axial dispersion . We find that increasing velocity yields a greater dispersion . Hsu & Hyne (37) modified general equation 2.3 ($E_{\rm Z} = rD_{\rm m} + 0.5*u*d_{\rm p}$ or 1/Pe = r/(Re*Sc) + 0.5) and with their constansts , the predicted curves are plotted as illustrated in figure 6.3 . A similar tendency of this predicted curve plotting in $E_{\rm Z}$ against u confirms our results .

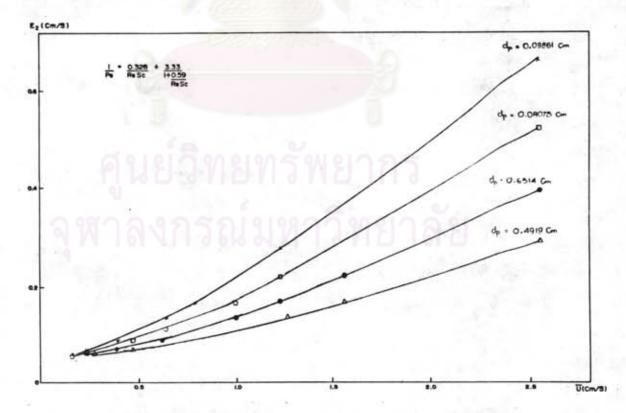


Fig. 6.3 Relationship between velocity and axial dispersion coefficient based on Hsu & Hyne's equation .

6.3 The prediction of axial dispersion coefficient

Since our experimental axial dispersion coefficient does not follow the relationship given in equation 2.3. We hereby introduce a general power law to relate axial dispersion coefficient and the three factors discussed in the last section.

$$E_z/D_m = a(U)^b(d_p)^c(d_t)^d$$
 6.1

All constants (a,b,c,d) will be evaluated by graphical methods using following equations:

$$\ln [E_z/D_m] = b \ln U + \ln [a(d_p)^c(d_t)^d]$$
 6.2

$$ln [E_z/D_m] = clnd_p + ln [a(U)^b(d_t)^d]$$
 6.3

$$ln [E_z/D_m] = dlnd_t + ln [a(U)^b(d_p)^c$$
 6.4

The logarithmic plot of E_Z (cm 2 /s) VS u (cm/s) as shown in figure 6.4 yields points scattering around a linear regression . A similar scatter arises from the other two plots (E_Z versus d_p and E_Z versus d_t) as shown in figures 6.5 and 6.6 , respectively . The slopes of these regressions equal to constants b , c and d . Constant a can be obtained by optimization using the following criteria

$$\varepsilon = 1/n \Sigma [E_z(experimental) - E_z(predicted)]^2$$

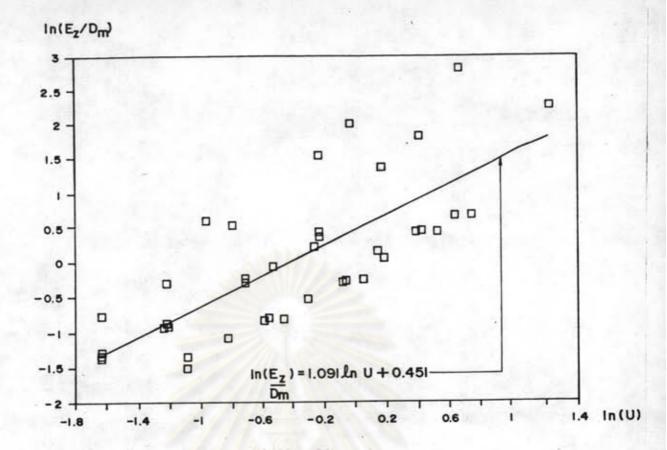


Fig. 6.4 Axial dispersion coefficient vs velocity for entire set of data showing linear regression line.

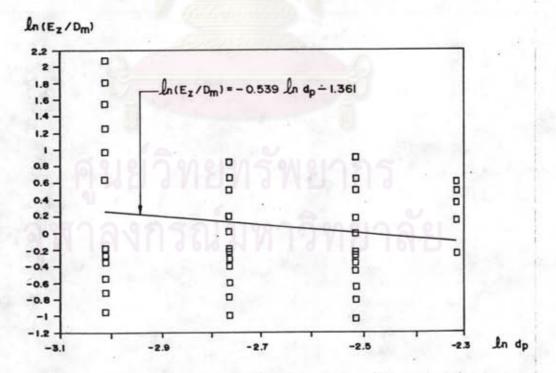


Fig. 6.5 Axial dispersion coefficient vs particle size for entire set of data at same fixed velocities showing linear regression line .

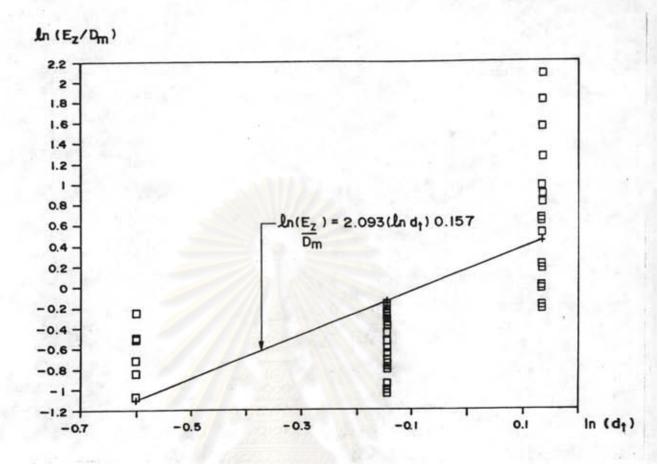


Fig. 6.6 Axial dispersion coefficient vs tube diameter for entire set of data at some fixed velocities showing linear regression line .

The optimization curve is given in figure 6.7 by plotting against a . Eventually , we get the relationship for predicting the axial dispersion coefficient as

$$E_z = 2.757(U)^{1.188}(d_p)^{-0.478}(d_t)^{1.921}(D_m)$$
 6.5

Figure 6.8 depicts the graphical comparison between experimental dispersion coefficient and the predicted dispersion coefficient based on equation 6.5.



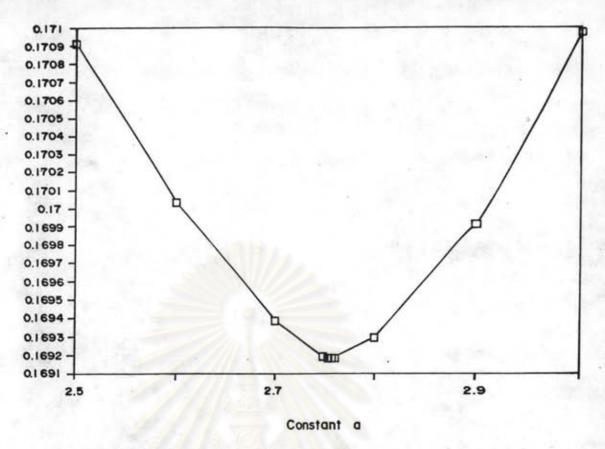


Fig. 6.7 Average minimal deviation of theoretical axial dispersion coefficient from experimental values versus constant a .

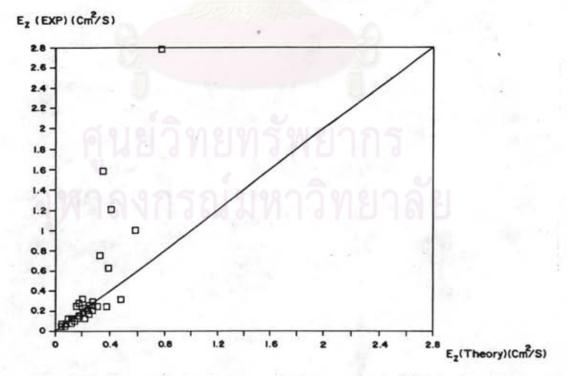


Fig. 6.8 Comparison of measured dispersion coefficient with that obtained with emperical formula .

6.4 Comparative analysis for experimental axial dispersion coefficient

As no propane - methane molecular diffusivity information was available , here is reported data for similar pairs of hydrocarbons as given in table 6.1

Table 6.1 DIFFUSION COEFFICIENTS IN BINARY GAS SYSTEMS

DisP is the product of D in square centimeters per second and the pressure in atmospheres

System	т. *к	Did'	DisP (calc. using force con- stants from viscosity'	Error.	Cealr, using estimated force	Error.	D.P Arnold method	Error.	DnP Gilliand method	Error.	DisP Nattery method	Error.
Ethane-methane Ethane-propane	293 293	0.0850 0.163	0 079 0 157	=7	0.0811	-5 +2	0.0935 0.162	+10	0.00K	+15	0.152 0.098	+15

^{*} K error - (calculated - experimental) /experimental × 100.

Source : The properties of Gases and Liquids , Reid , pp.275

axial dispersion coefficients (E_Z) obtained for Reynolds numbers between 0.14-1.50 as shown in table 5.2 indicate the same order of magnitude as molecular diffusion coefficient (D_m) of the ethane-propane system . This shows that our experimental axial dispersion coefficients (E_Z) are very small and probably can be neglected for engineering purposes .

^{8/} See all formula of calculated methods shown in table 6.1 in reference no. 36