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APPENDIX

silica gel G / 4% methanol in chloroform

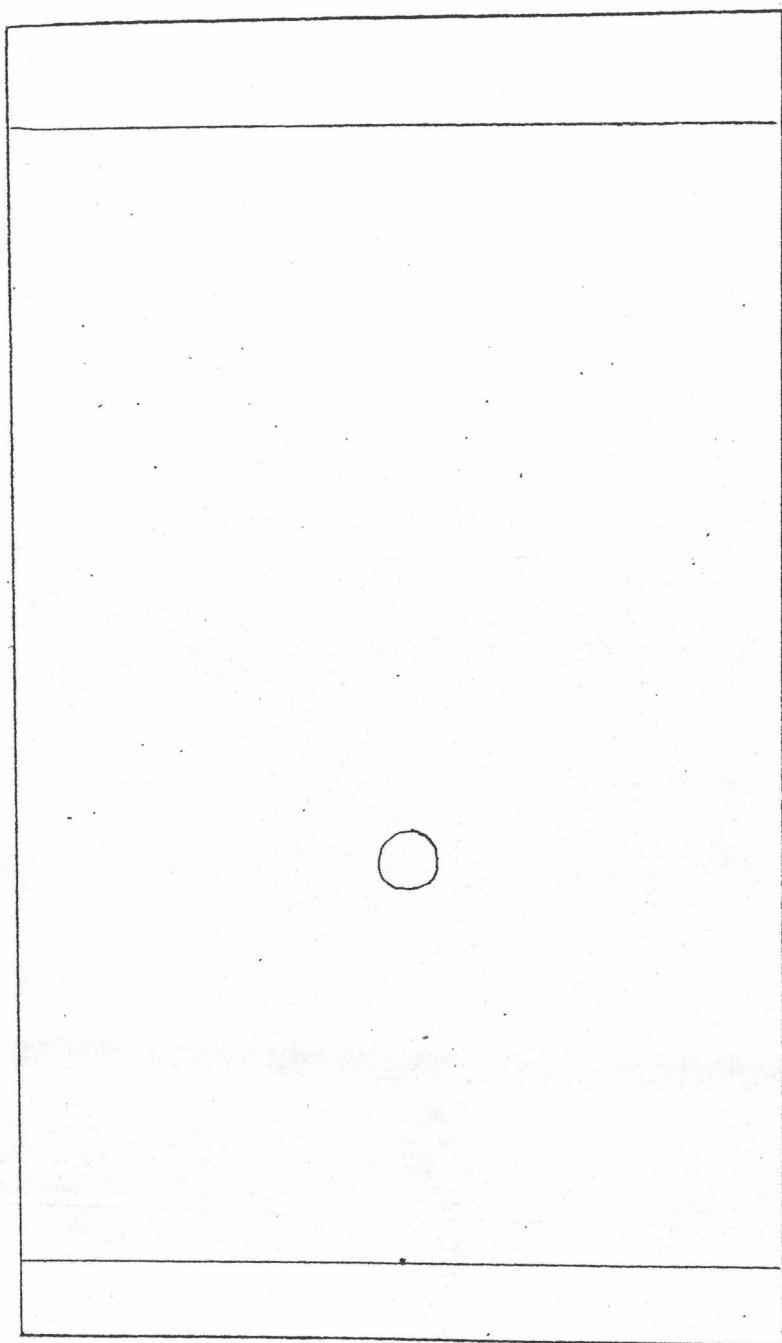


Fig. 4 Thin layer chromatogram of alkaloid Py

silica gel G / chloroform

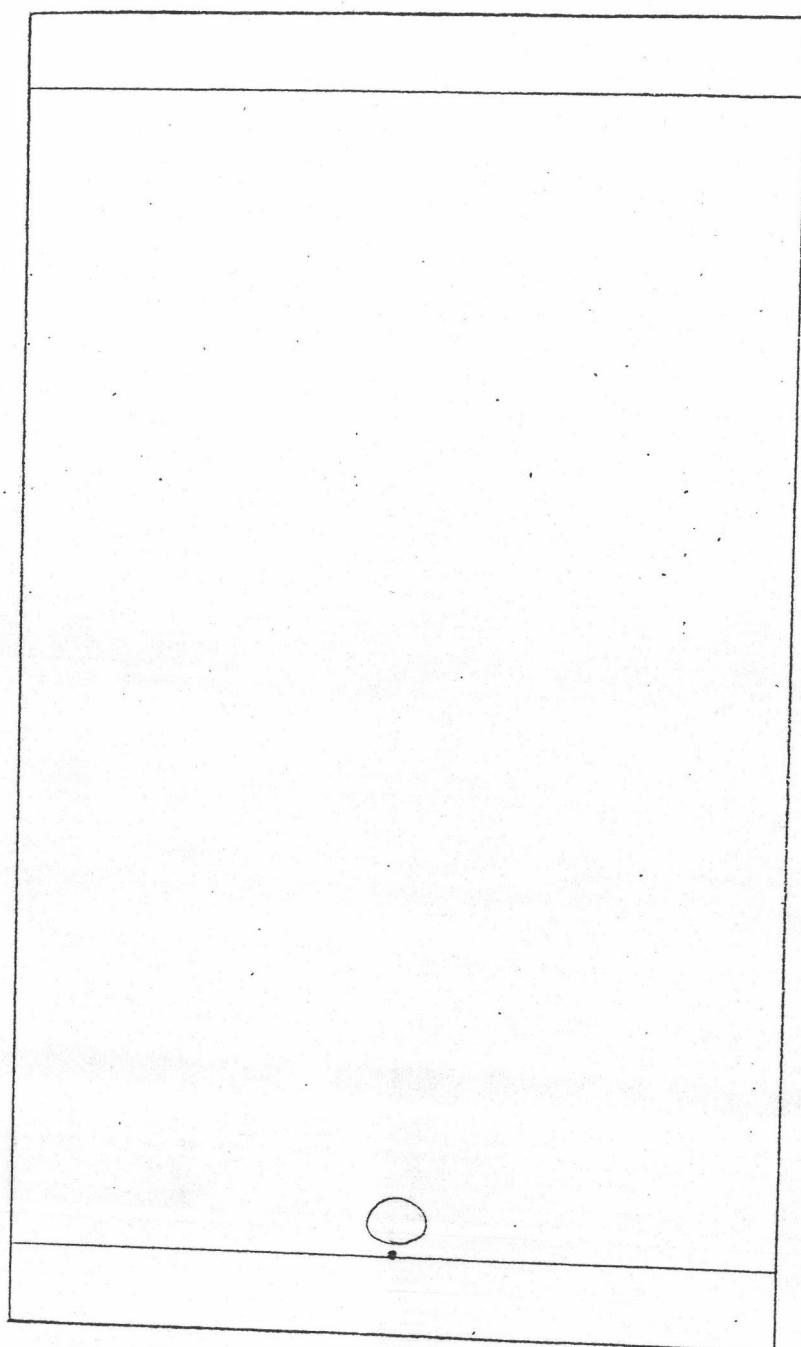


Fig. 5 Thin layer chromatogram of alkaloid Py

silica gel G / 10% acetone in chloroform

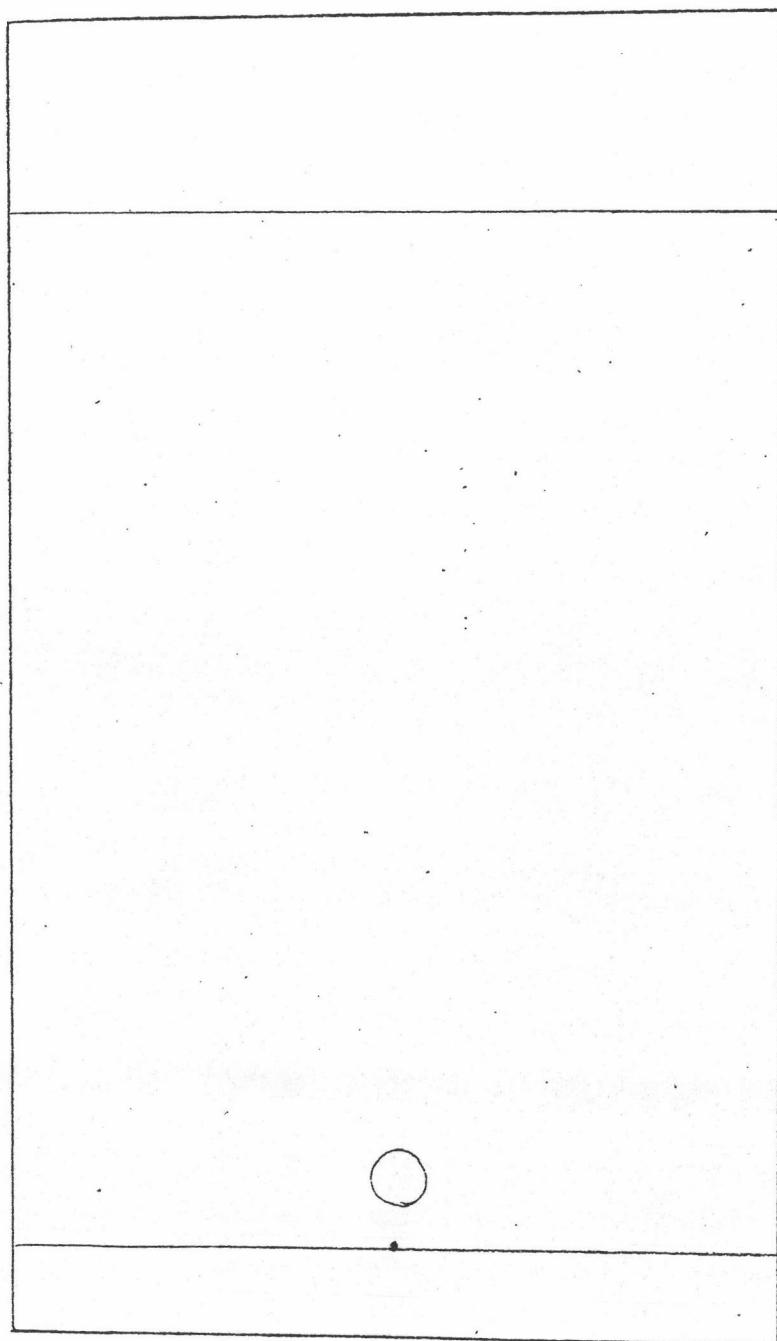


Fig. 6 Thin layer chromatogram of alkaloid Py

silica gel G / 30% benzene in chloroform

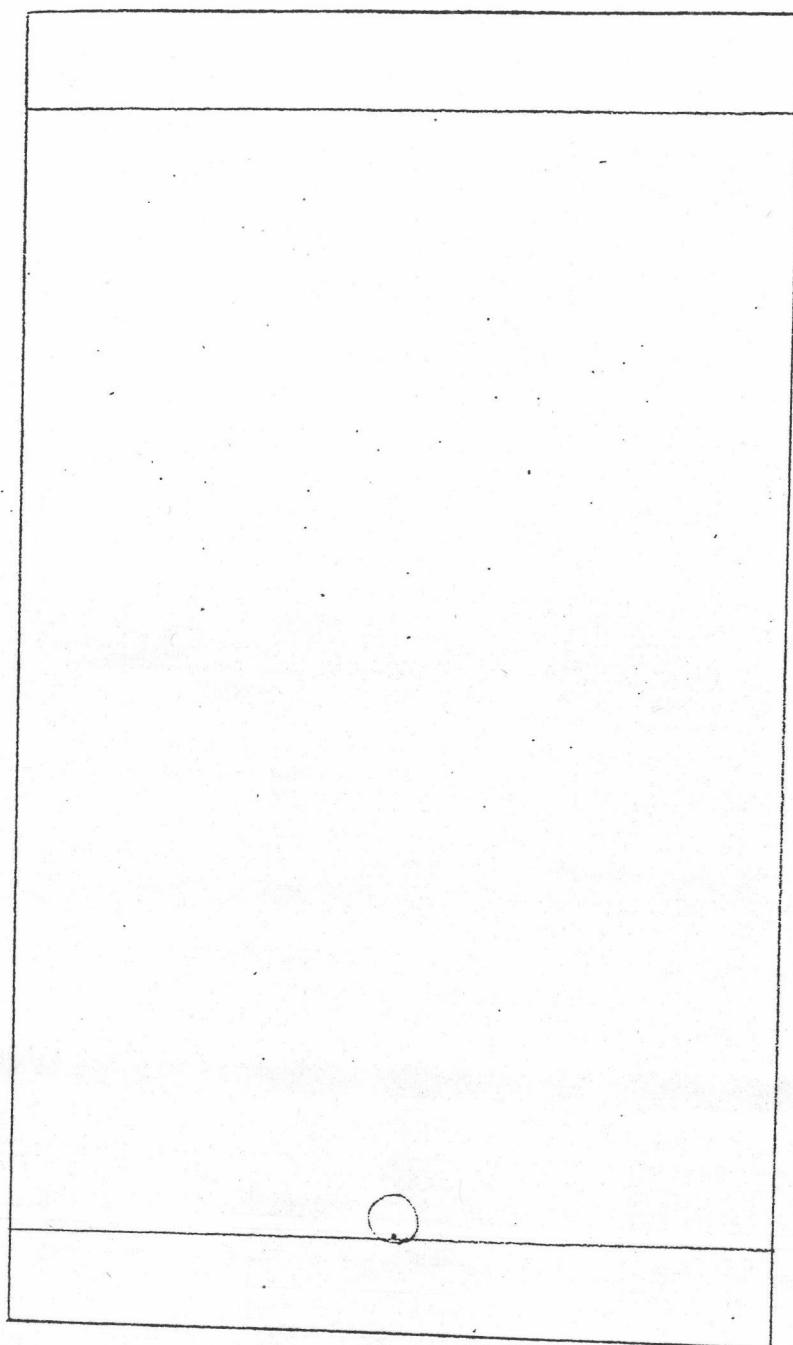


Fig. 7 Thin layer chromatogram of alkaloid Py

silica gel G / 10% ethyl acetate in chloroform

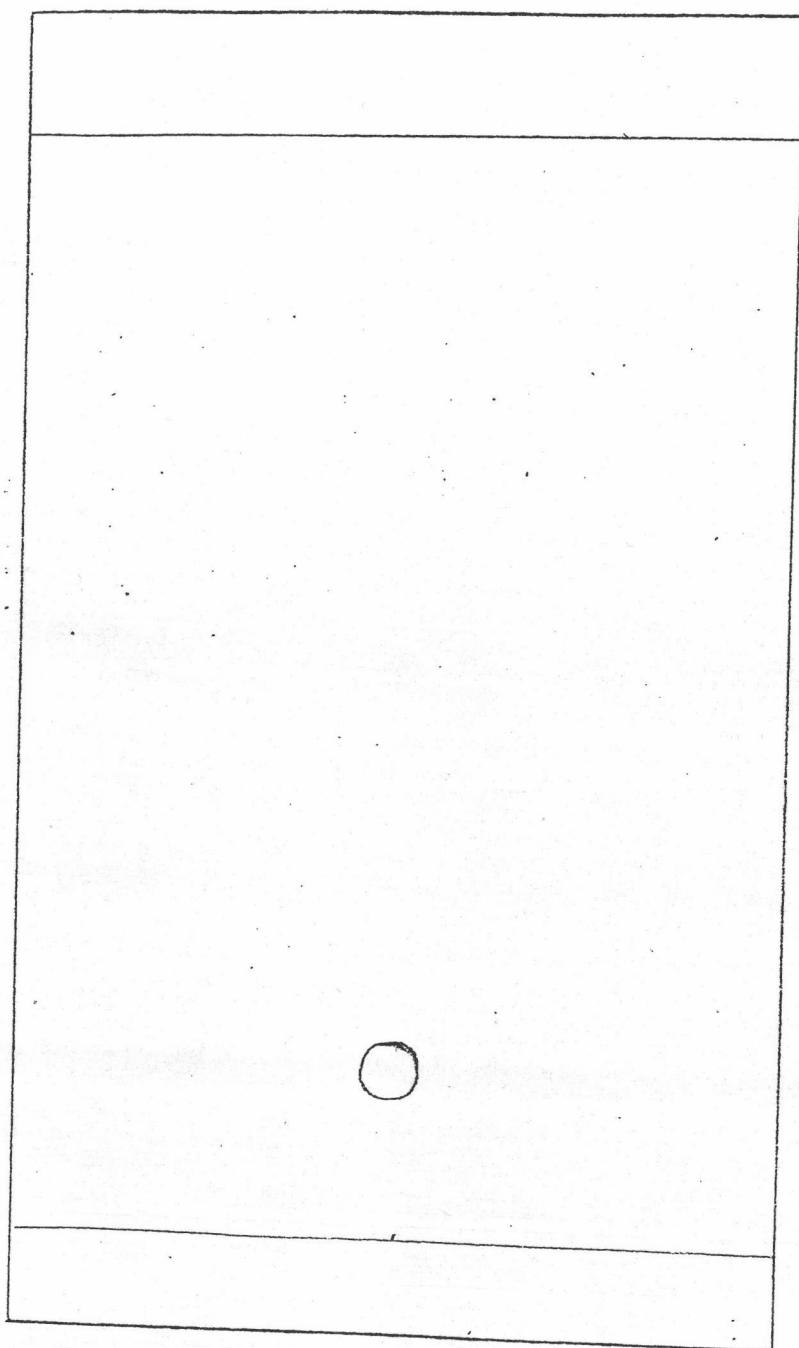


Fig. 8 Thin layer chromatogram of alkaloid Py

SCIENTIFIC AND TECHNOLOGICAL RESEARCH EQUIPMENT CENTER
CHULALONGKORN UNIVERSITY

ANALYSIS REPORT

SAMPLE	จันทน์ชามด		
SAMPLE OWNER	นายจินดาพร ภูวิพัฒนาวงศ์		
INSTRUMENT	ELEMENTAL ANALYZER (PERKIN ELMER 240C)		
ANALYSIS CONDITION	COMBUSTION TEMP.	950 °C	
	REDUCTION TEMP.	650 °C	
	HELIUM PRESSURE	18.5 psi	
	OXYGEN PRESSURE	17.5 psi	

RESULTS

SAMPLE NAME	%N	%C	%H
compound X (1)	9.82	49.76	7.97
(2)	9.70	49.67	8.18
alkaloid Py (1)	8.66	73.74	6.93
(2)	8.52	73.97	6.89

ดร. อุษราวดี - ANALYST
๒๘ ก.พ. ๕๔ DATE

นายเหศุ ผลการทดสอบที่ได้รับนี้ เป็นผลการทดสอบเฉพาะส่วนย่างที่ทำการทดสอบจาก ศูนย์เครื่องมือวิจัยวิทยาศาสตร์และเทคโนโลยีจลย.เท่านั้น
AU/WS

Fig. 9 Elemental analysis of alkaloid Py and compound X

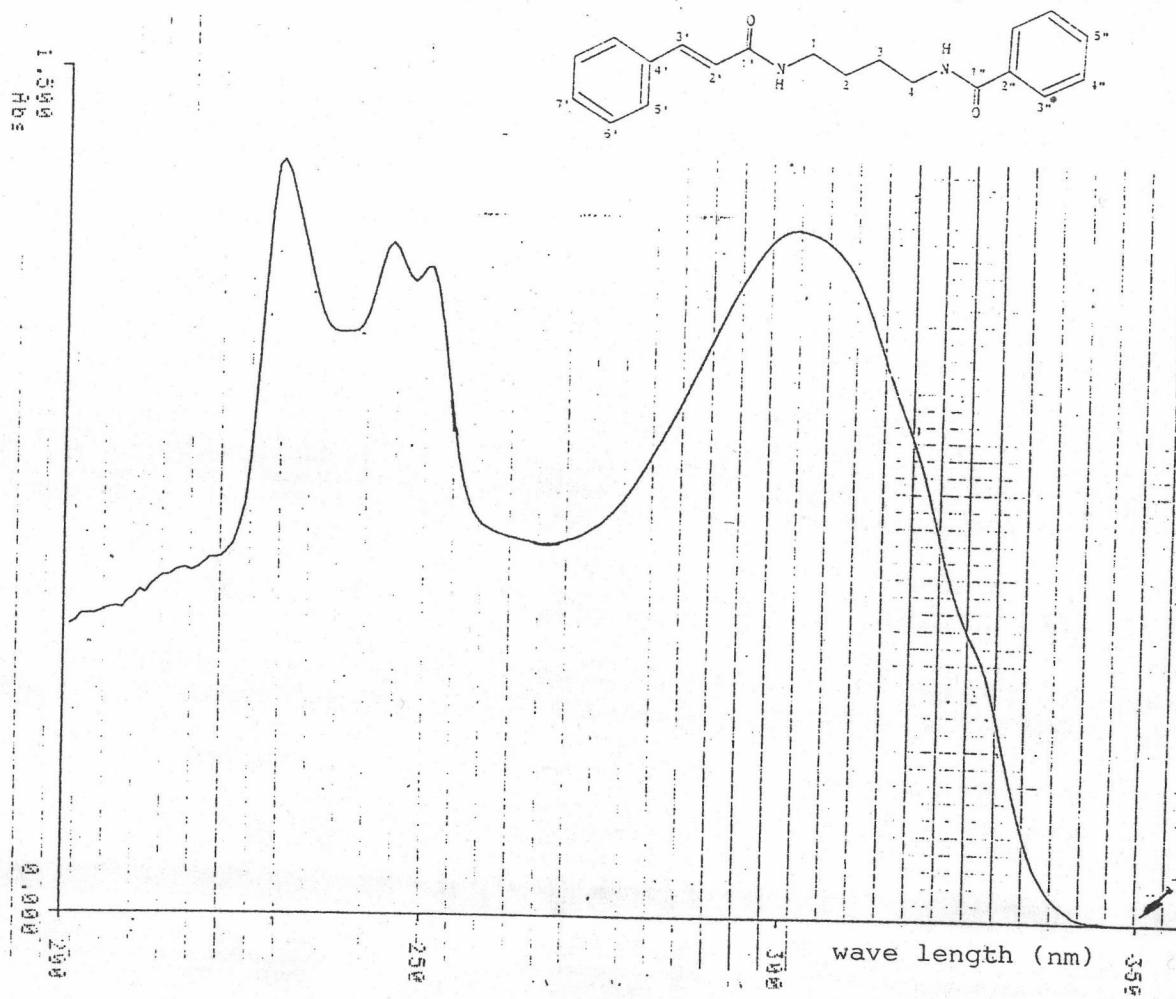


Fig. 10 Ultraviolet absorption spectrum of alkaloid Py

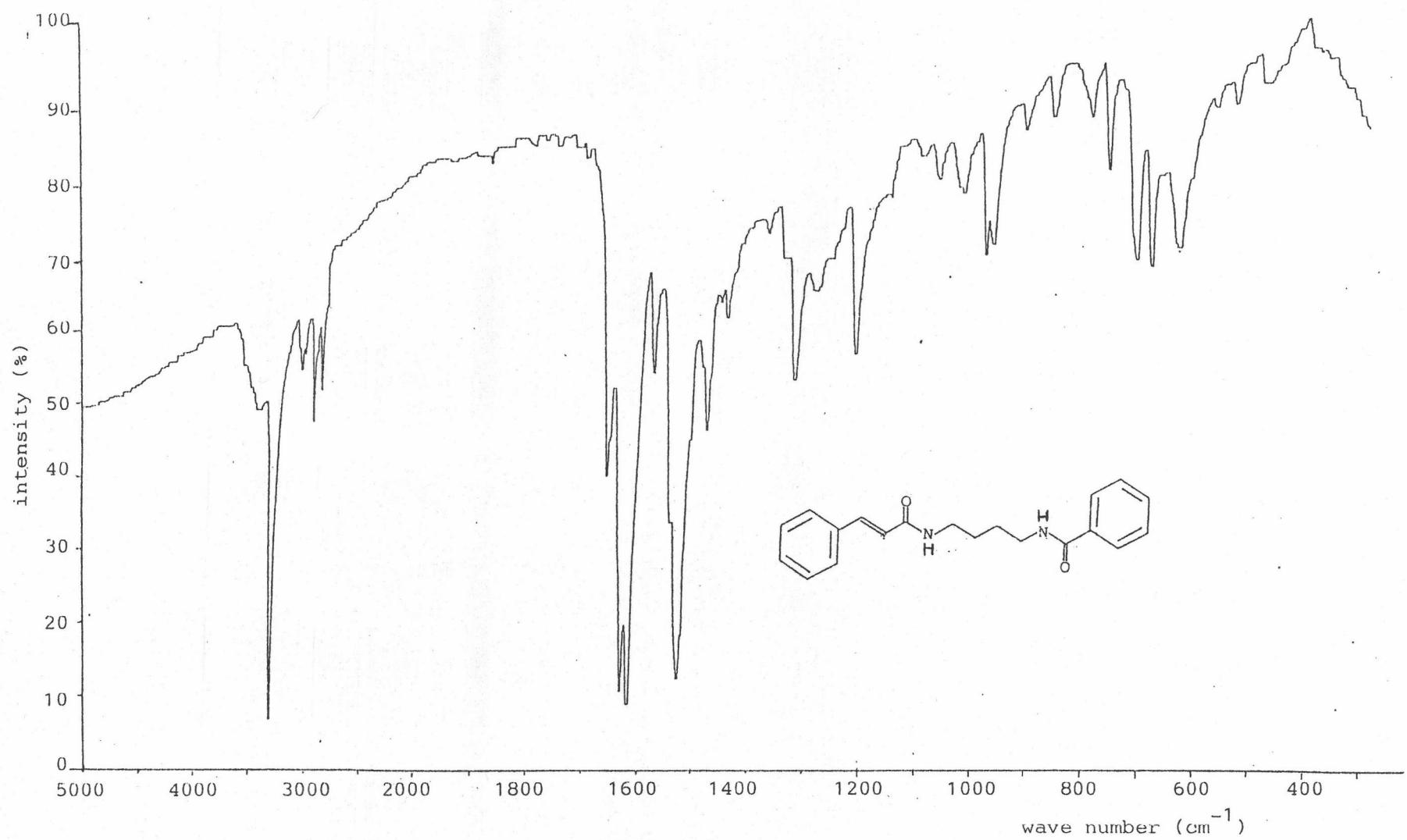


Fig. 11 Infrared absorption spectrum of alkaloid Py

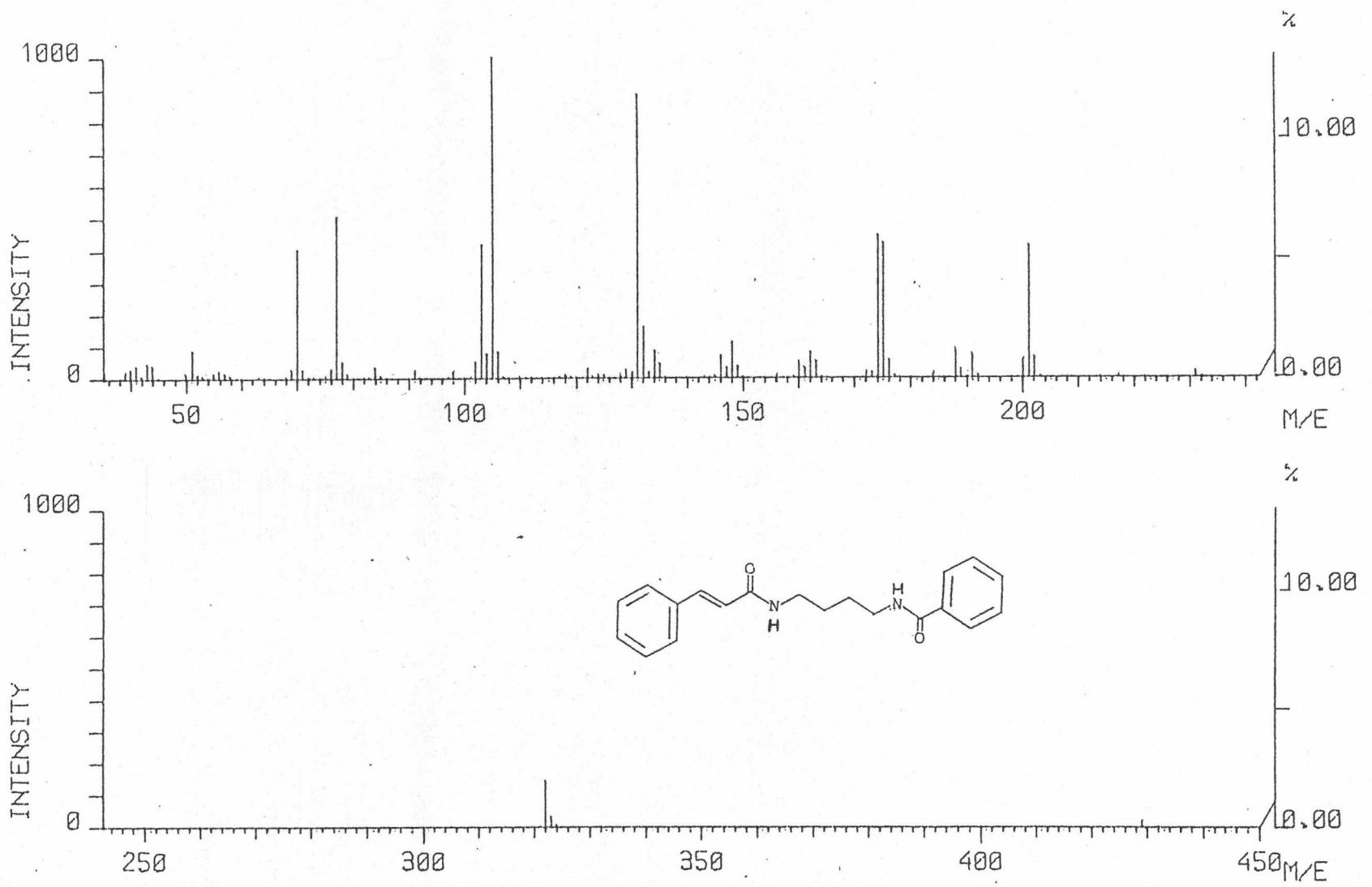


Fig. 12 Mass spectrum of alkaloid Py

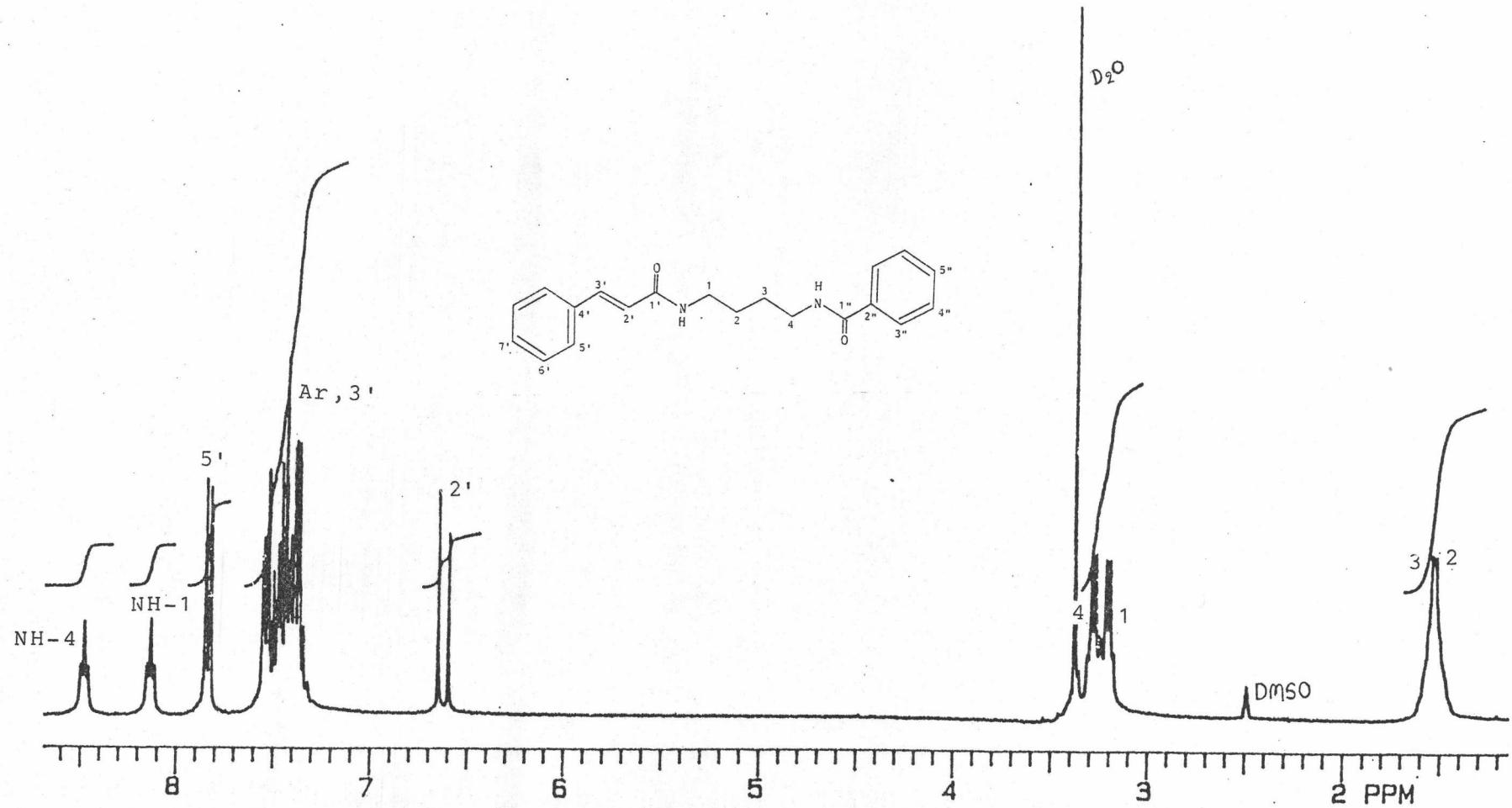


Fig. 13 ¹H Nuclear magnetic resonance spectrum (300 MHz) of alkaloid Py in DMSO-*d*₆

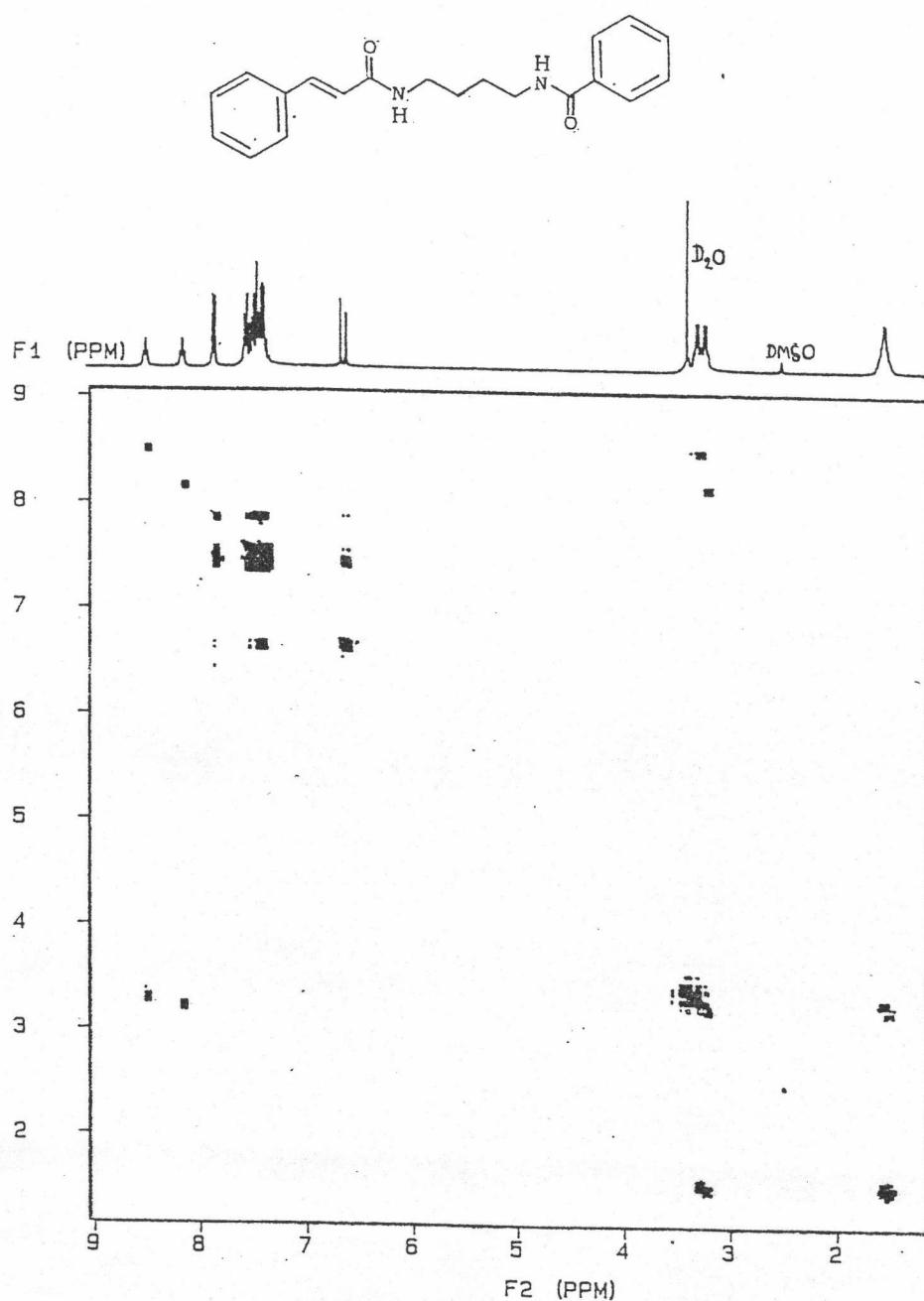


Fig. 14 2-D Homonuclear (COSY) nuclear magnetic resonance spectrum (300 MHz) of alkaloid Py

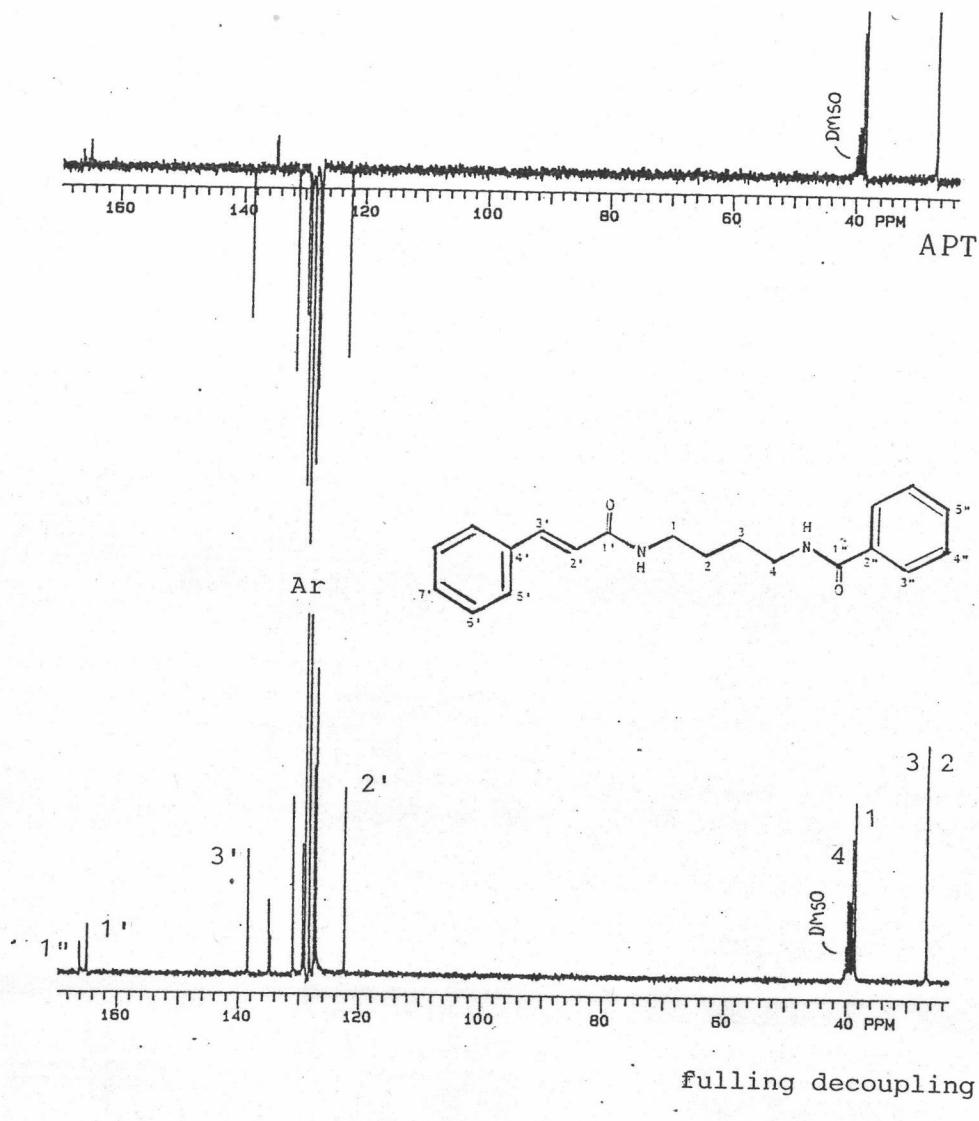


Fig. 15 ^{13}C Nuclear magnetic resonance spectrum (75 MHz) of alkaloid Py in $\text{DMSO}-\text{d}_6$

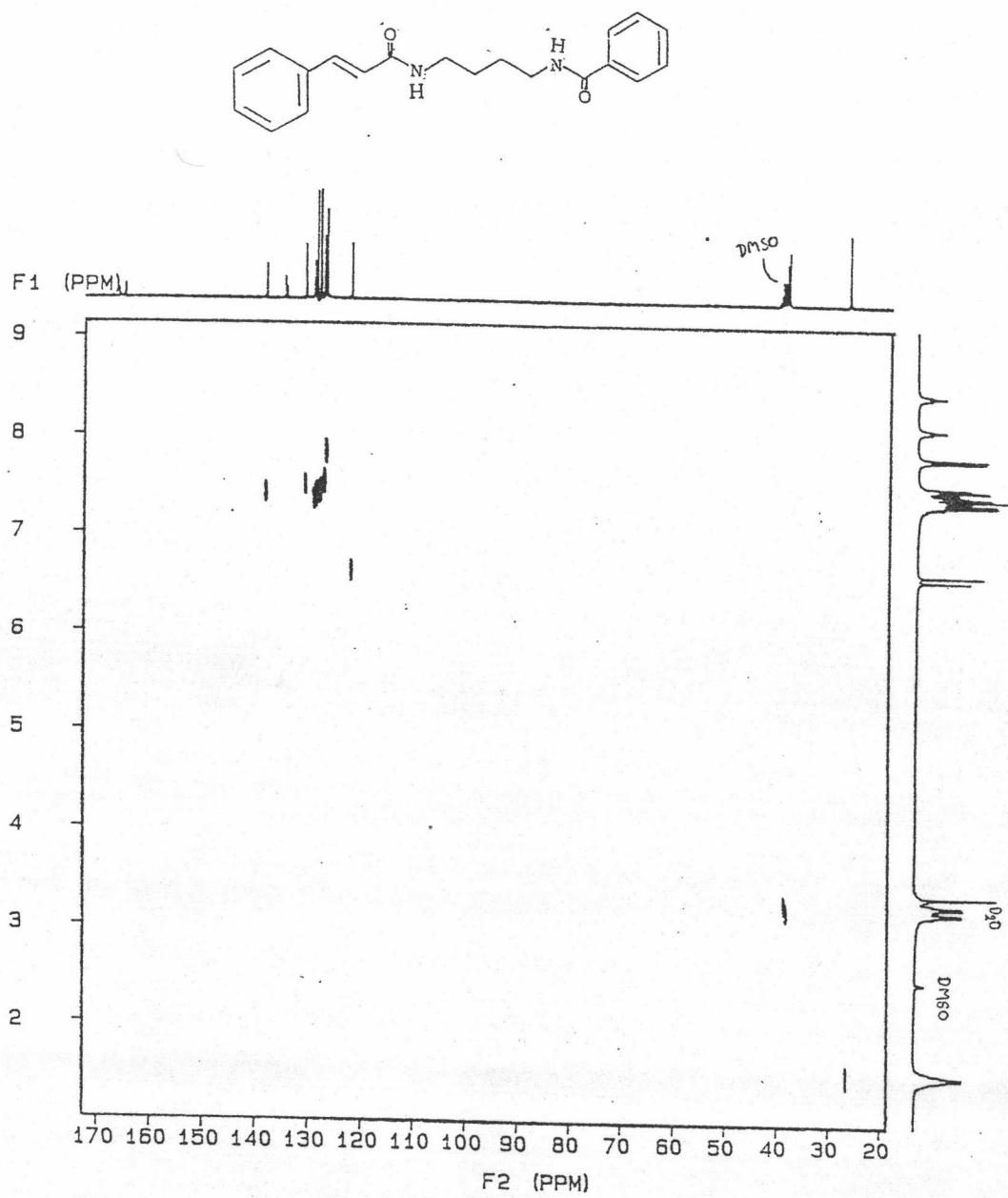


Fig. 16 2-D Heteronuclear (HETCOR) nuclear magnetic resonance spectrum (300 MHz) of alkaloid Py

silica gel G / 50% methanol in chloroform

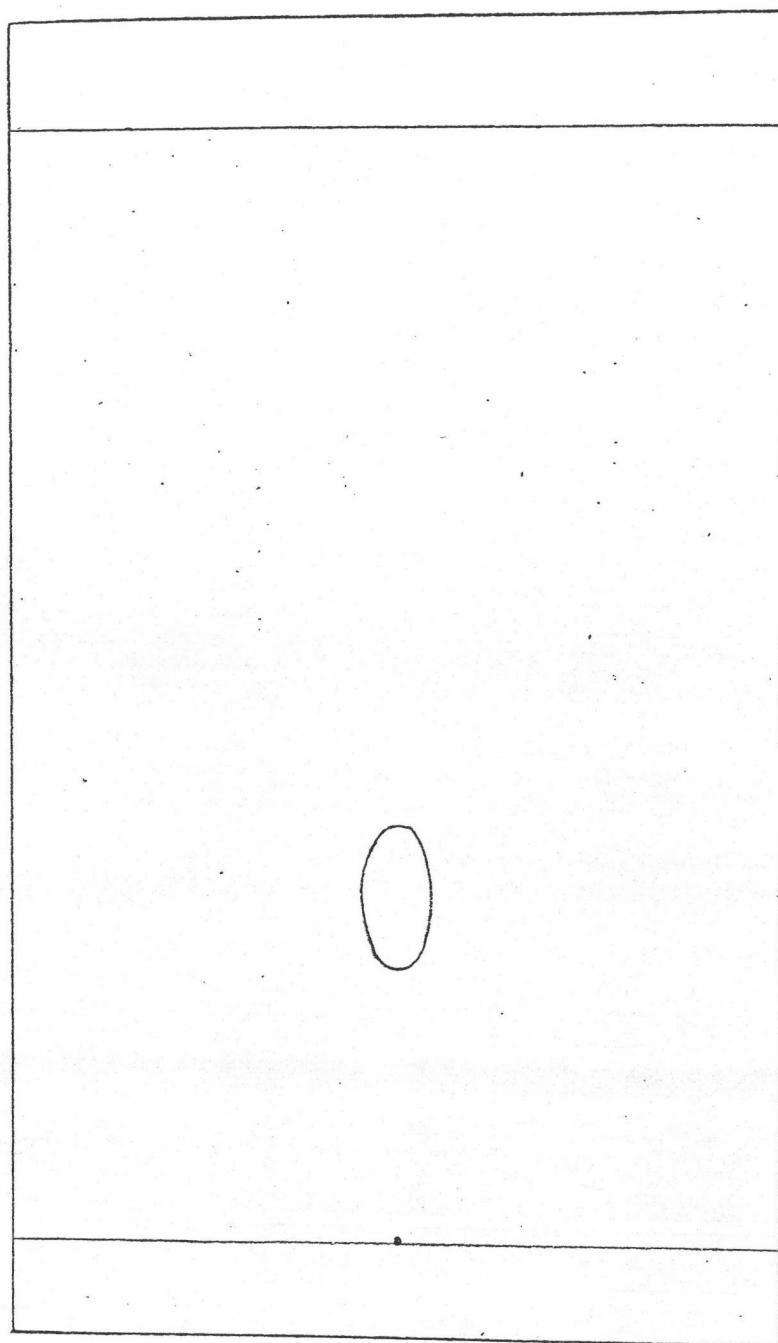


Fig. 17 | Thin layer chromatogram of compound X

silica gel G / 30% ethanol in chloroform

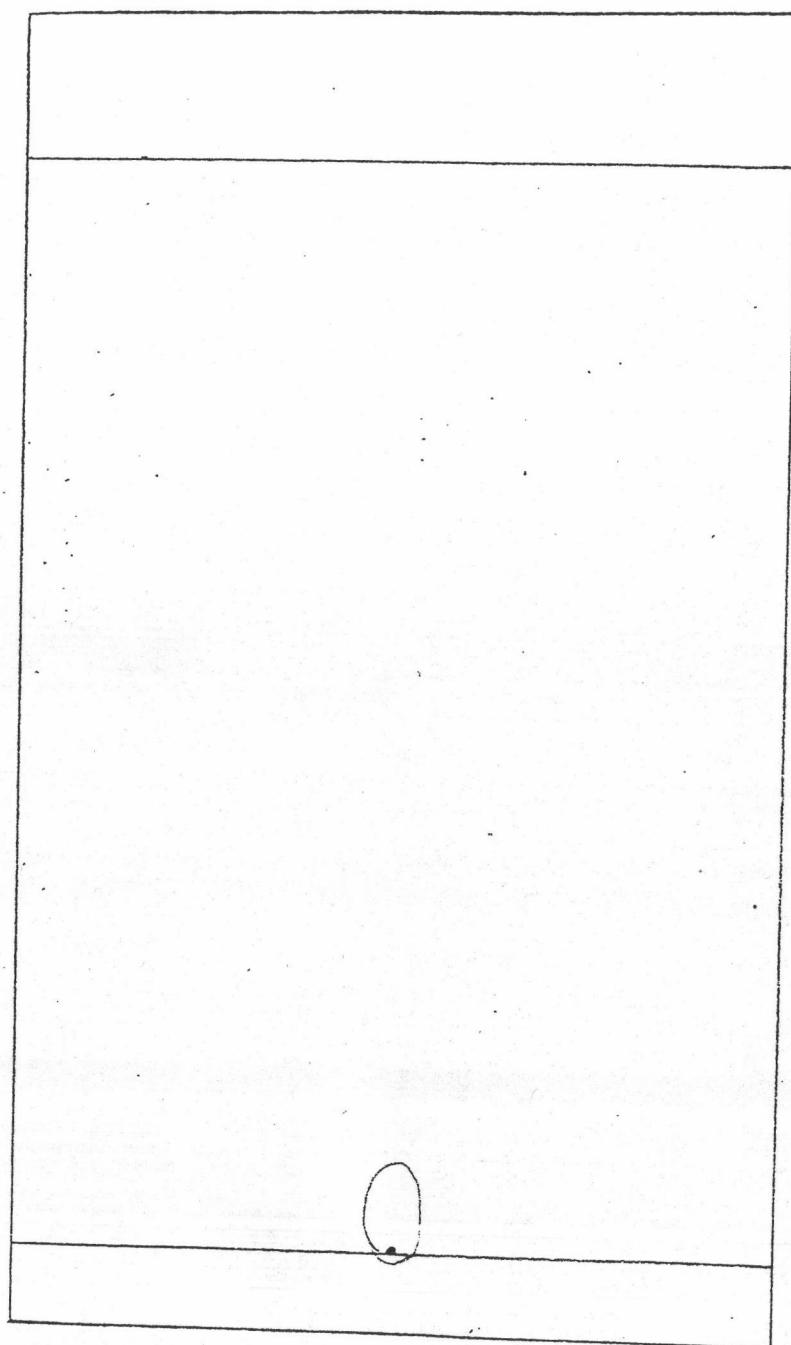


Fig. 18 Thin layer chromatogram of compound X

silica gel G / 50% acetone in methanol

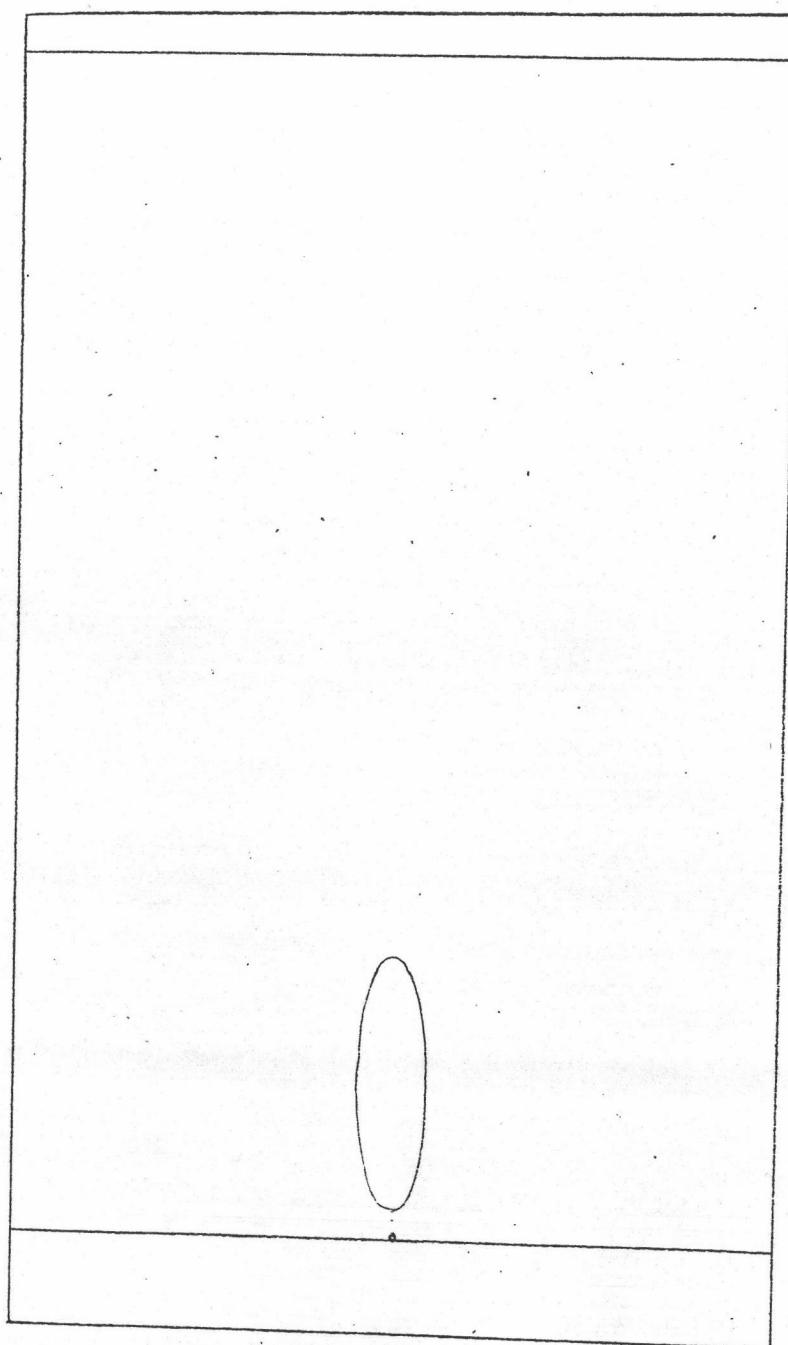


Fig. 19 | Thin layer chromatogram of compound X

silica gel G / 50% ethyl acetate in methanol

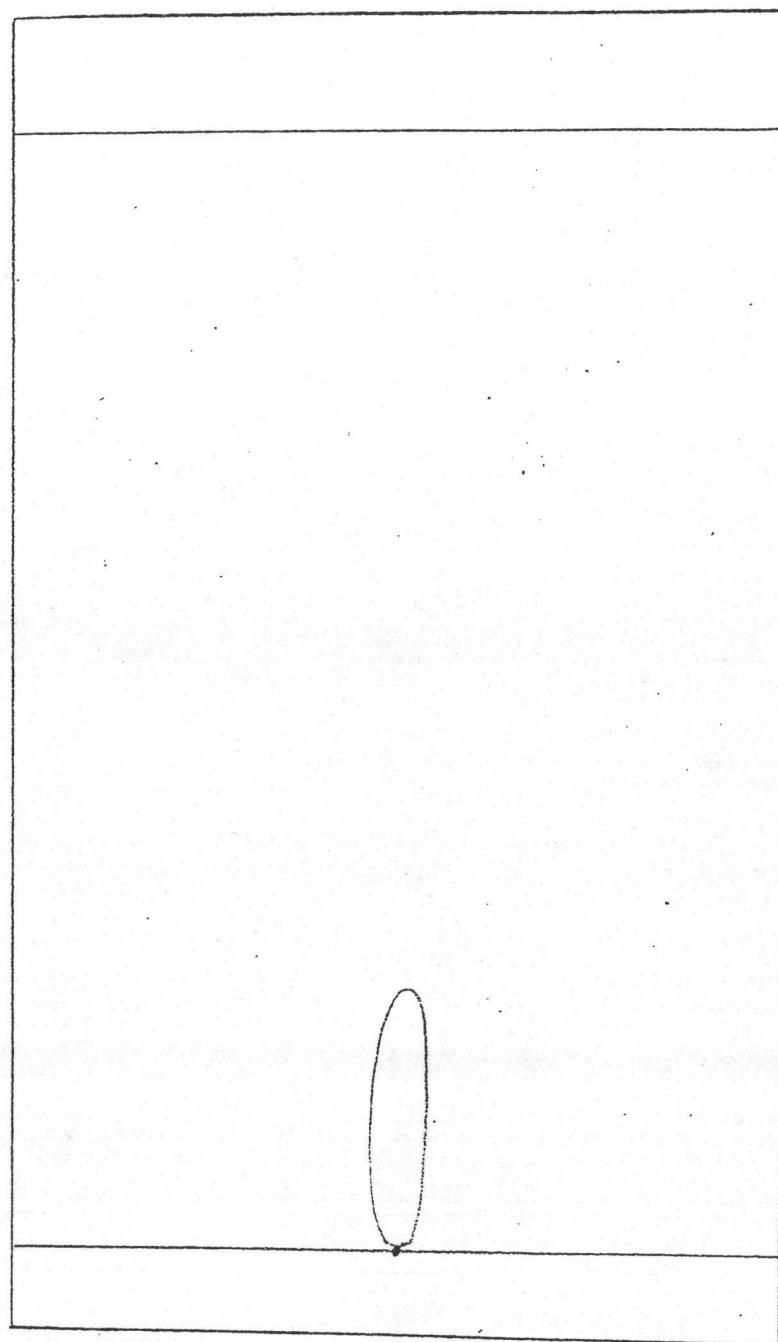


Fig. 20 Thin layer chromatogram of compound X

silica gel G / 30% ammonia in n-propyl alcohol

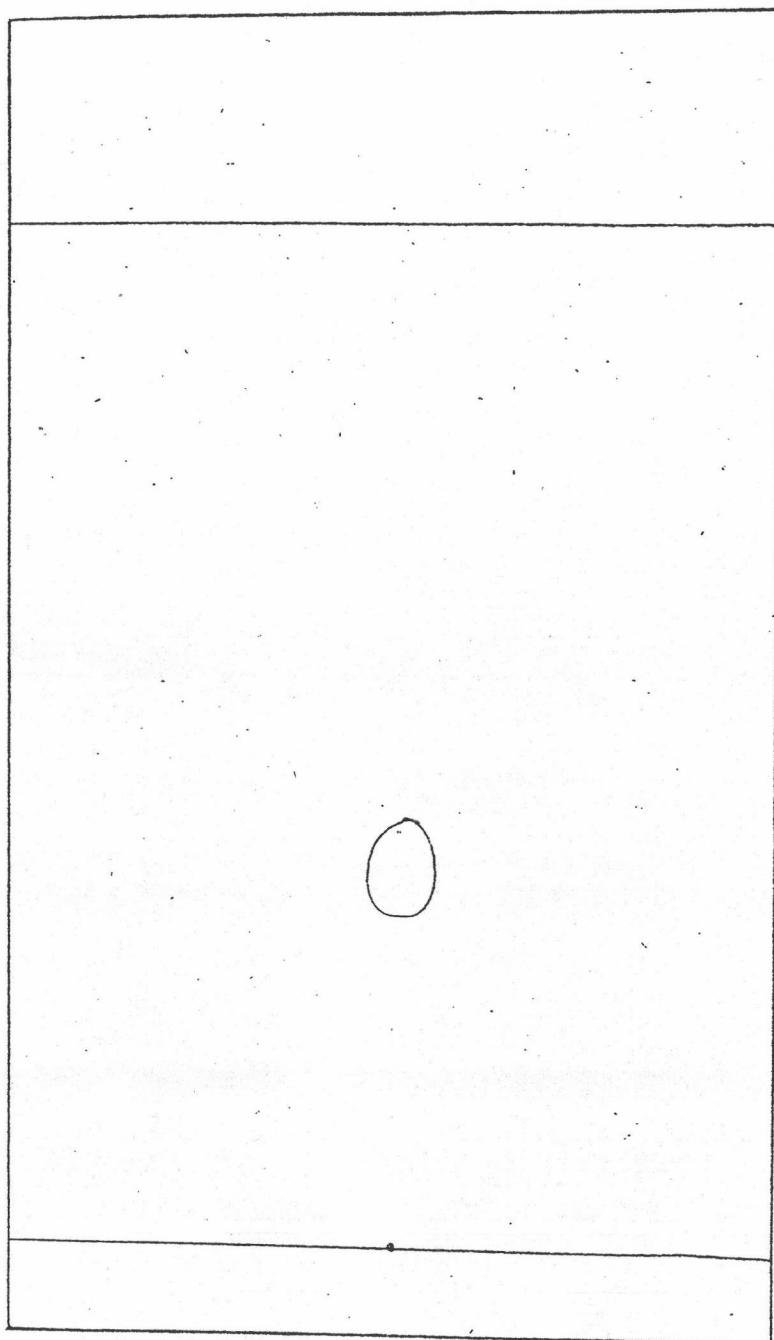


Fig. 21 Thin layer chromatogram of compound X

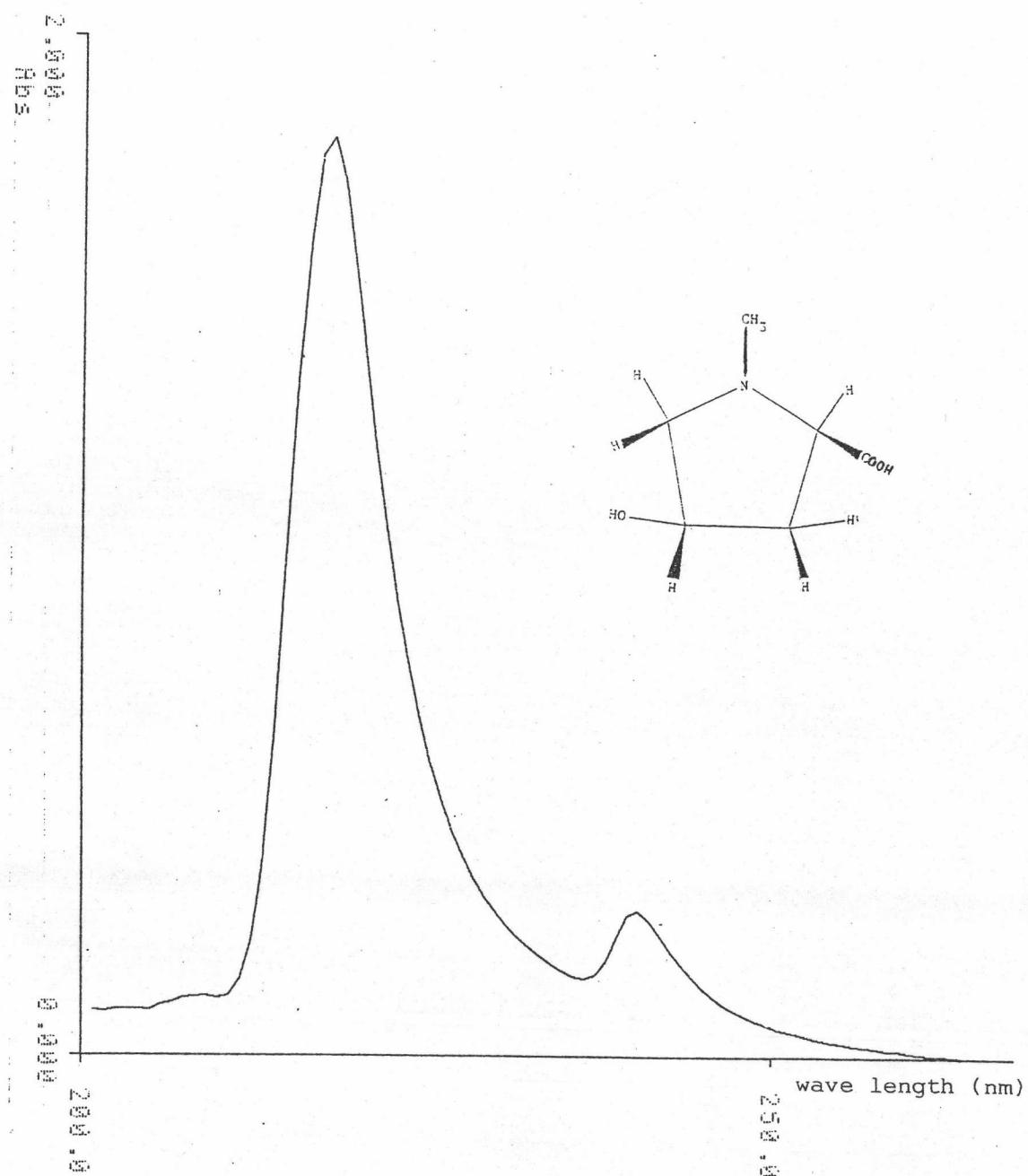


Fig. 22 Ultraviolet absorption spectrum of compound X

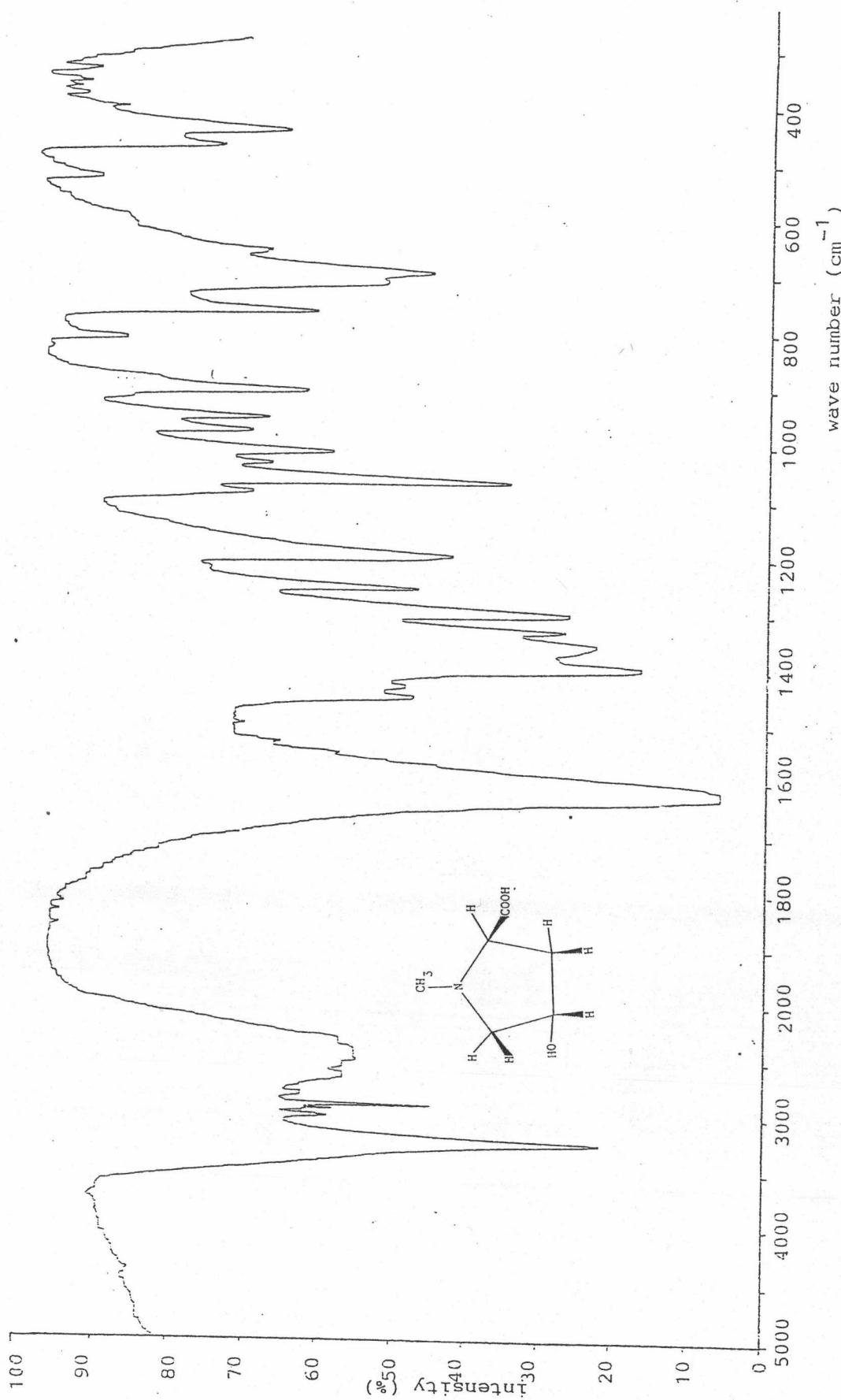


Fig. 23 Infrared absorption spectrum of compound X

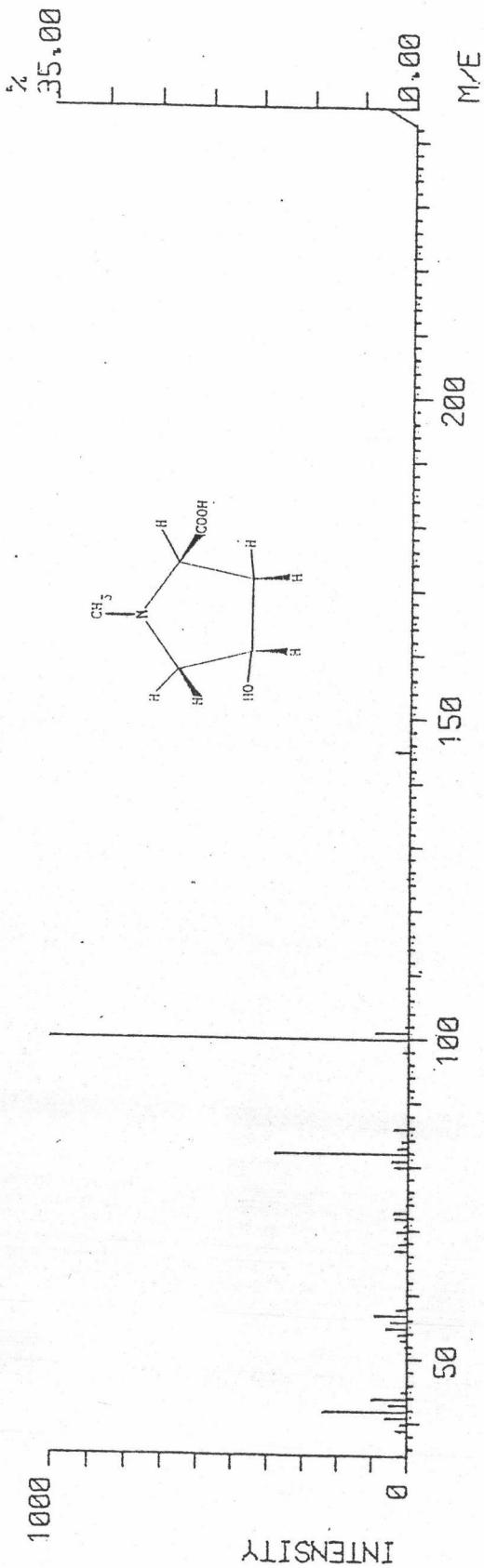


Fig. 24 Mass spectrum of compound X

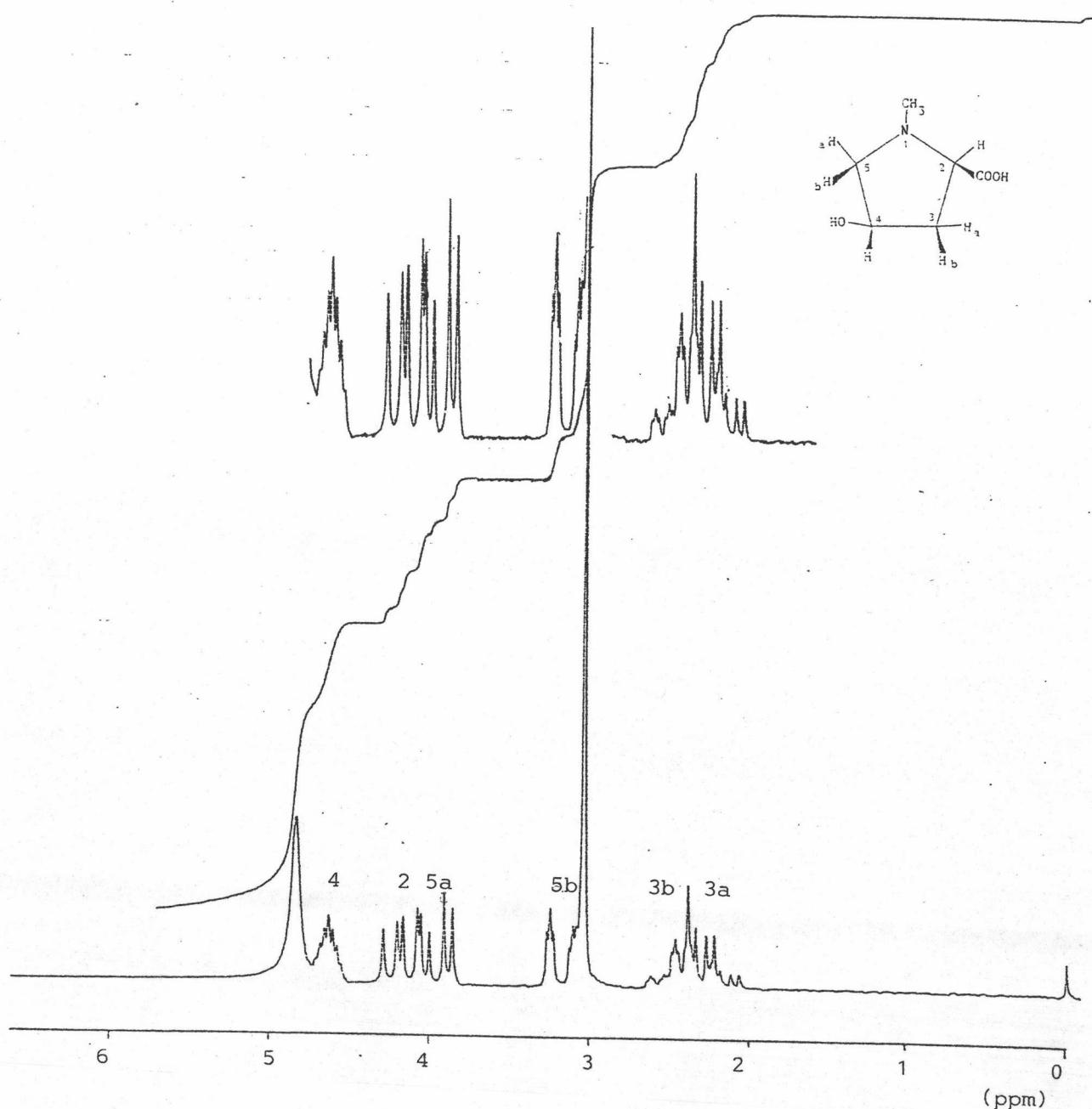


Fig. 25 ^1H Nuclear magnetic resonance spectrum (90 MHz)
of compound X in D_2O

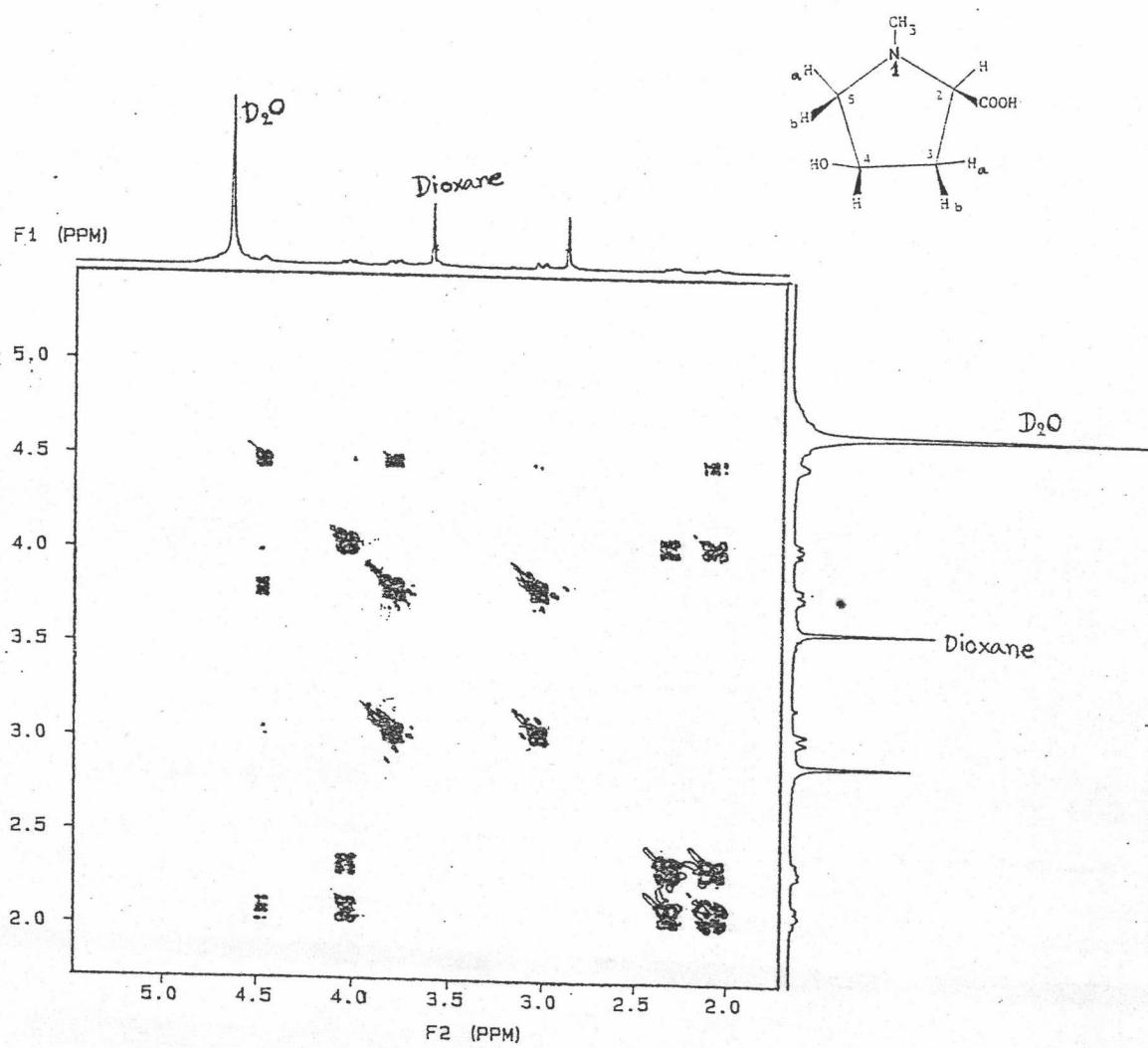


Fig. 26 2-D Homonuclear (COSY) nuclear magnetic resonance spectrum (300 MHz) of compound X

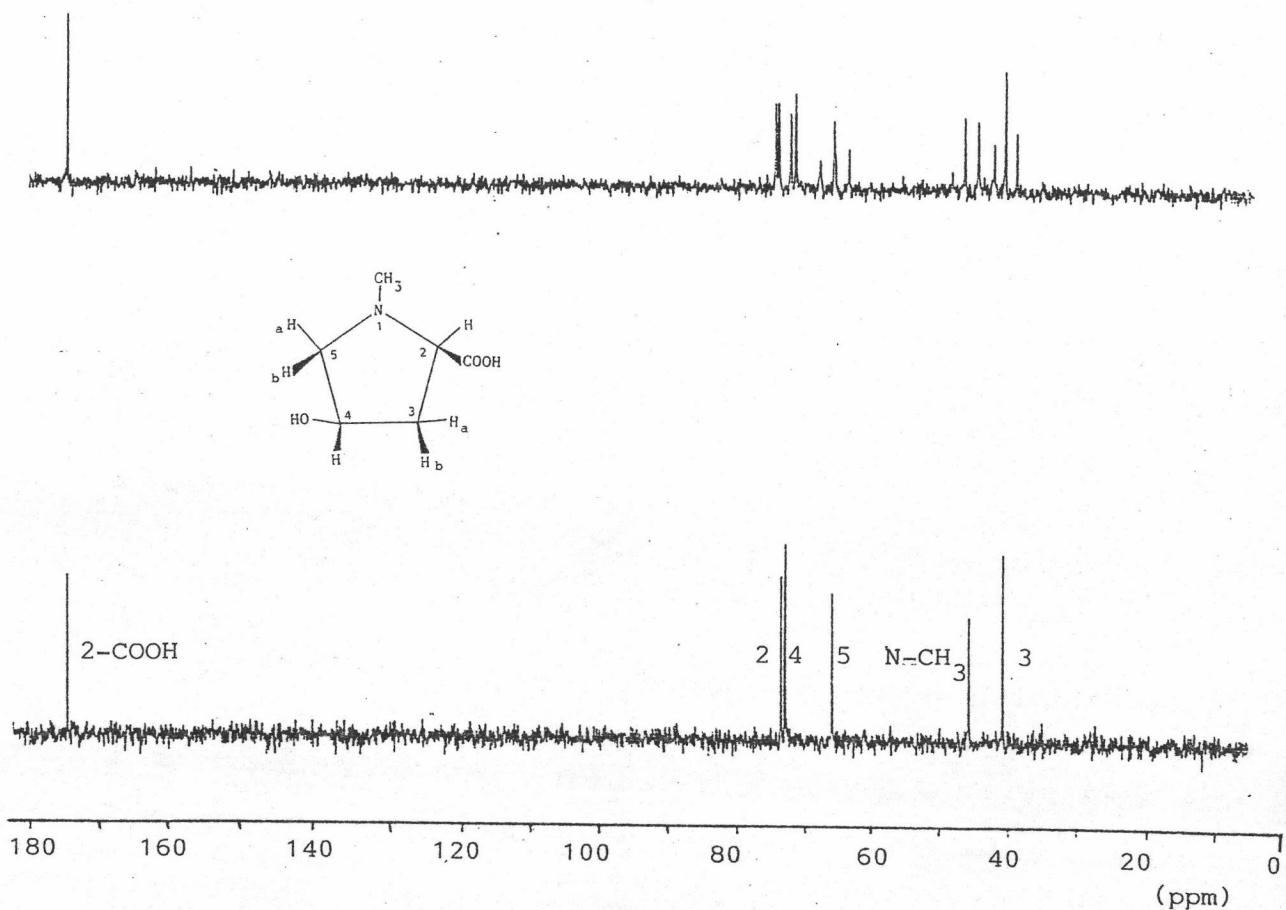


Fig. 27 | ^{13}C Nuclear magnetic resonance spectrum (22.5 MHz)
of compound X in D_2O

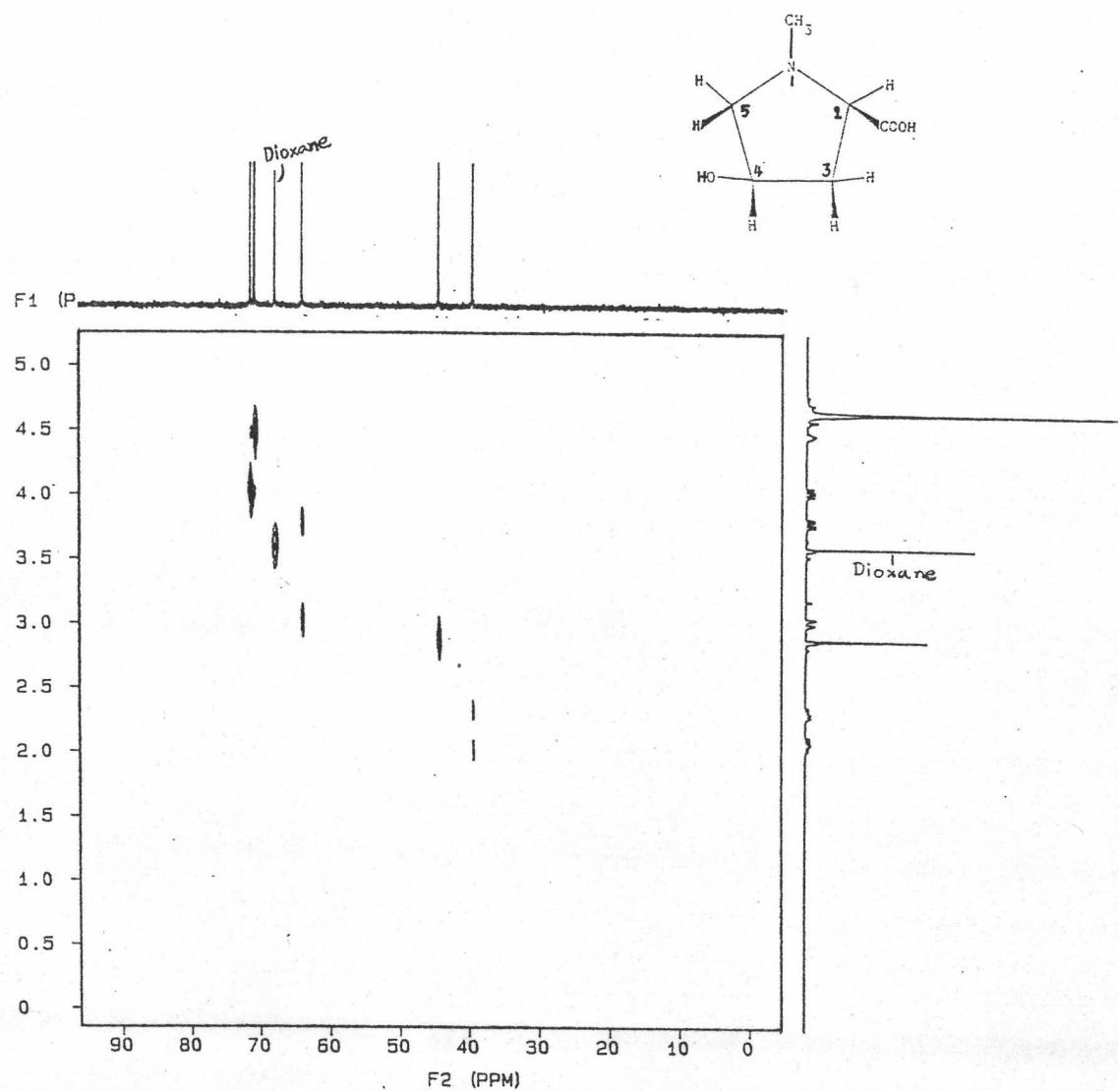


Fig. 28 2-D Heteronuclear (HETCOR) nuclear magnetic resonance spectrum (300 MHz) of compound X

VITA

Mr. Jindaporn Puripattanavong was born on January 17th, 1965 in Suratthani, Thailand. He obtained a B.Sc. in Pharm. from the Faculty of Pharmacy, Prince of Songkla University in 1988. Since graduation, he has been appointed as an instructor in the Department of Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmacy, Prince of Songkla University, Songkhla, Thailand.

