## RESULTS

In selection of ointment bases, the study was investigated by measuring the hexamidine release from different ointment bases. (Hydrophilic ointment, Hydrophilic petrolatum, Polyethylene glycol ointment and White ointment). The ointment base that gave the maximum amount of hexamidine release was the base of choice.

Figure 2. Showed the release of hexamidine from different ointment bases containing the same concentration ( $0.1 \%$ $\mathrm{W} / \mathrm{W}$ ) at $37^{\circ} \mathrm{C}$ in 2 hours. It was found that the polyethylene glycol ointment gave the maximum release of hexamidine, the second was hydrophilic ointment, while hydrophilic petrolatum was the third, and white ointment became the fourth respectively.

Figure 3-8. Showed the release of hexamidine from polyethylene glycol ointment containing the same concentration ( $0.1 \% \mathrm{~W} / \mathrm{W}$ ) and the same concentration of each cationic surfactants (Benzalkonium chloride $1: 1000$ solution, Cetrimide 1:1000 solution and Cetylpyridinium chloride 1:1000 solution) beginning from $1 \%, 3 \%, 5 \%, 7 \%, 10 \%$ and finally $12 \%$ respectively at $37^{\circ} \mathrm{C}$. The results were pointed out that every ointment
preparation gave the release of hexamidine more than the one without cationic surfactant.

Figure 9. Showed the effect of various concentrations (1, 3, 5, 7 and $10 \%$ ) of benzalkonium chloride $1: 1000$ solution on the release of hexamidine from polyethylene glycol ointment. The curve showed that the amount of hexamidine increased as the concentration of the benzalkonium chloride solution increased.

Figure 10. Showed the effect of various concentrations $(1,3,5,7$ and $10 \%)$ of cetrimide $1: 1000$ solution on the release of hexamidine from polyethylene glycol ointment. The results also showed that the amount of hexamidine increased as the concentration of the cetrimide solution increased.

Figure 11. Showed the effect of various concentrations $(1,3,5,7$ and $10 \%$ ) of cetylpyridinium chloride $1: 1000$ solution on the release of hexamidine from polyethylene glycol ointment. The results also showed that the amount of hexamidine increased as the concentration of the cetylpyridinium chloride solution increased.

In the part of cationic surfactants, the results can be indicated that they had effect on the release of hexamidine from polyethylene glycol ointment (Figure 3-8) actively, especially when the concentration of each cationic surfactants increased, the amount of hexamidine releasing increased too.

The amount of each cationic surfactants that could be added to the polyethylene glycol ointment containing $0.1 \% \mathrm{~W} / \mathrm{W}$ hexamidine without loosing stability and viscosity was $10 \%$.

Finally the result showed that benzalkonium chloride 1:1000 solution was the best cationic surfactant for hexamidine.


จุฬาลงกรณ์มหาวิทยาลัย



Figure 3 Concentration V.S.time curves of tiexanidine releasing from $0.1 \%$ W/W PEG ointment with $1 \%$ of various cationic surfactants at $37^{\circ} \mathrm{C}$.

Keys : Control, $\Delta$ Cetylpyridinium chloride,
O Cetrimide, Benzalkonium chloride.


Figure 4 Concentration V.S. time curves of Hexaraidine releasing from 0.1 \% W/W PEG ointment with $3 \%$ of various cationic surfactants at $37^{\circ} \mathrm{C}$

Keys : - Control, $\triangle$ Cetylpyridinium chloride O Cetrimide, Benzalkonium chloride


Figure 5 Concentration V.S. time curves of Hexamidine releasing from $0.1 \%$ W/W PEG ointment with $5 \%$ of various cationic surfactants at $37^{\circ} \mathrm{C}$

Keys : Control, $\triangle$ Cetylpyridinium chloride O Cetrimide, A Benzalkonium chloride.


Figure 6 Concentration V.S. time curves of Eexamidine releasing from $0.1 \% \mathrm{~W} / \mathrm{W}$ PEG ointment with $7 \%$ of various cationic surfactants at $37^{\circ} \mathrm{C}$

Keys : Control, $\triangle$ Cetylpyridinium chloride,
O Cetrimide, A Benzalkonium chloride.



Figure 8 Concentration V.S. time curves of Hexamidine releasing from $0.1 \% \mathrm{~W} / \mathrm{W}$ PEG ointment with $12 \%$ of various cationic surfactants at $37^{\circ} \mathrm{C}$

Keys : Control, $\triangle$ Cetylpyridinium chloride, O Cetrimide, $\Delta$ Benzaikorium chloride.


Figure 9 Concentration V.S. time curves of Hexamidine releasing from $0.1 \% \mathrm{~W} / \mathrm{W}$ PEG ointment with various concentration of Benzalkorium chloride 1 : 1000 solution at $37^{\circ} \mathrm{C}$

Keys
$: 0$ $0 \%, \Delta$ $1 \%, \square 3 \%$, - $5 \%$ A 75 $10 \%$,


Figure 10 Concentration V.S. time curves of Hexamidine releasing from $0.1 \% \mathrm{~W} / \mathrm{W}$ PEG ointment with various concentrations of Cetrimide 1:1000 solution a.t $37^{\circ} \mathrm{C}$

Keys: $\quad 0 \%, \Delta 1 \%, \bigcirc 3 \%, 05 \%, \square 7 \%, \square 10 \%$.


Figure 11 Concentration V.S. time curves of Gexamidine releasine from $0.1 \% \mathrm{~W} / \mathrm{W}$ PEG ointment with various concentrations of Cetylpyridinium chloride $1: 1000$ solution at $37^{\circ} \mathrm{C}$

Keys : $0 \%, \triangle 1 \%, 03 \%, \bigcirc 5 \%, \square 7 \%, \square 10 \%$

