REFERENCES

- Adani, K. G., Barley, R. W., and Pascoe, R. D. 2005. Silver recovery from synthesis photographic and medical x-ray process effluents using activated carbon. <u>Miner. Eng.</u> 18: 1269-1276.
- Ajiwe, V. I. E., and Anyadiegwu, I. E. 2000. Recovery of Silver from Industrial Waste, Cassava Solution Effects. <u>Sep. Purif. Technol.</u> 18: 89-92.
- Association of Metropolotan Sewerage Agencies (AMSA). 2005. <u>The Silver Council.</u>

 <u>Code of Management Practice for Silver Dischargers.</u> Available from: http://www.envisioncompliance.com/pdf/framework.pdf.
- Atia, A. A. 2005. Adsorption of silver (I) and gold (III) on resins derived from bisthiourea and application to retrieval of silver ions from processed photo films. <u>Hydrometallurgy</u> 80: 98-106.
- Bourges, J., Broudic, J., Dauby, J., Leboucher, I., Leclerc, O., and Baticle, P. 2002.

 <u>Electrolytic method for the recovery and recycling of silver from a nitric acid solution</u>. United States Patent. 6,428,679.
- Buser, K. R., and Rettig, T. A. 1988. <u>Recovery of Silver from Photographic Films by</u>
 <u>High Shear Conditions</u>. Canada Patent. 1,238,790.
- Cerjan-Stefanovic, S., Briski, F., Kastelan-Macan, M. 1991. <u>Separation of Silver from waste water by ion-exchange resins and concentration by microbial cells</u>. <u>Fresenius</u>. J. Anal. Chem. 339: 636-639.
- Colorado Industrial Pretreatment Coordinators Association (CIPCA). 2004. Report on Silver Discharge to WWTP. Available from: http://cipca.org/forms/silver/SILVERRPT.DOC.
- Dartmouth Toxic Metals Research Program, Center for environmental Health Sciences at Dartmouth, Dartmouth Toxic Metal Research. 2005. <u>Toxic Metals "Silver"</u>. Available from: http://www.dartmouth.edu/~toxmetal/TXQAag.shtml.
- Department of Industrial Works, Ministry of Industry. 1992. <u>The Factory Act B.E.</u> 2535.
- Department of Industrial Works, Ministry of Industry. 1997. The Ministerial Notification of the Ministry of Industry "Industrial Hazardous Waste Disposal". No. 6 B.E. 2540 (1997).

- Department of Industrial Works, Ministry of Industry. 1992. <u>The Hazardous</u> Substance Act B.E. 2535.
- Department of Industrial Works, Ministry of Industry. 2005. <u>The Ministerial Notification of the Ministry of Industry "Industrial Waste Disporsal"</u> B.E. 2548 (2005).
- Eastman Kodak Company. 2006. The Technology of Silver Recovery for

 Photographic Processing Facilities. KODAK Publication No. J-212.

 Available from:

 http://www.kodak.com/eknec/documents/01/0900688a800f8101/J212ENG.p

 df#search=%22silver%20recovery%20technology%22.
- Eaton, G. T. <u>Photographic chemistry in black-and-white and color photography 4th</u> <u>ed.</u> New York: Morgan & Morgan, Inc., 1986.
- Eisler, R. 1996. Silver Hazards to Fish, Widelife, and Invertebrates: a Synoptic Review. Patuxent Wildlife Research Center, U.S. National Biological Service.
- Fujiwara, N., Yamamoto, K., and Masui, A. 1991. Utilization of a thermostable alkaline protease from an *Alkalophilic Thermophile* for the recovery of silver from used X-ray film. J. Ferment. Bioeng. 72: 306-308.
- Gambhir, S. S. 2004. <u>Orientation: Photographic Emulsions, Pharmacology M248:</u>

 <u>Introduction to Biological Imaging</u>. Available from: ...[2004, December 20].
- Gelatin Manufacturers Association of Asia Pacific. 2005. <u>About gelatin</u>. Available from: http://www.gmap-gelatin.co/about_gelatin_gel.html.
- Gercia, R. M. 1986. The recovery of silver from photographic film: a study on the leaching reaction with cyanide solution for industrial use. <u>Hydrometallurgy</u> 16: 395-400.
- Guevara, S. R., Arribere, M., Bubach, D., Vigliano, P., Rizzo, A., Alonso, M., and Sanchez, R. 2005. Silver contamination on abiotic and biotic compartments of Huapi National Park Lakes, Patagonia, Argentina. <u>Science of the Total.</u> <u>Environment</u> 336: 119-134.
- Handerson, K. 2003. The contemporary silver cycle for CIS countries: using industrial ecology to evaluate silver flows. <u>J. Young Investigators</u>. Available from: http://www.jyi.org/volumes/volume9/issue1/articles/henderson.html.

- Hilliard, H. E. 2000. Silver Recycling in the United States in 2000. U.S. <u>Geological Survey Circular 1196-N. U.S. Department of the Interior</u>. U.S. Geological Survey.
- Hochberg, J. 1989. Recovery of Silver from Photographic Film with High Shear and Caustic. United States Patent. 4,799,954.
- Ishikawa, H., Ishimi, K., Sugiura, M., Sowa, A., and Fujiwara, N. 1993. Kinetics and mechanism of enzymatic hydrolysis of gelatin layers of X-ray film and release of silver particles. <u>J. Ferment. Bioeng.</u> 76: 300-305.
- JACQUES, B. 2002. Electrolytic method for the recovery and recycling of silver from a nitric acid solution. US Patent 6,428,679. Available from: http://v3.espacenet.com/results?sf=a&DB=EPODOC&PN=us6428679&PG S=10&CY=ep&LG=en&ST=advanced.
- James, T. K., Woo, D. E. G., and Joseph, S. T. 1976. Recovery of Materials from Photographic Film. United States Patent. 1,432,000.
- Johnson, J., Bertram, M., Henderson, K., Jirikowic, J., and Graedel, T. E. 2005. The contemporary Asian silver cycle: 1-year stocks and flows. <u>J Mater Cycles</u> <u>Waste. Mang.</u> 7: 93-103.
- Kapur, A. 2006. The future of the red metal a developing country perspective from India. <u>Res. Cons. Recycling</u> 47: 160-182.
- Kawkasikum, T. 1995. Power Plant Engineering. 2,000, 1st ed. Technological Promoting Association (Thai-Japan). pp: 1-11.
- Lanzano, T., Bertram, M., De Palo, M., Wagner, C., Zyla, K., and Graedel, T. E. 2006. The contemporary European silver cycle. Res. Cons. Recycling 46: 27-43.
- Laungchonlatan, S. 1988. Feasibility Studies of Synthesis the Silver-Sulfadiazine from Used X-Ray Films. Master Degree Thesis. Department of Pharmaceutical Chemistry, Faculty of Pharmaceutical Science, Chulalongkorn University.
- McClearn, P. B. 2000. <u>Pollution Prevention Opportunities for Chemicals in Colorado Hospitals</u>. Available from: http://www.p2pays.org/ref/06/05847.pdf.
- Messerschmidt, H. 1988. <u>Method for Recovering Silver from Waste Photographic</u> Film and paper. United States Patent. 4,759,914.
- Ministry of Public Health. 1992. The Public Health Act B.E. 2535 (1992).

- Ministry of Public Health. 1993. <u>The Ministerial Notification of the Ministry of Public Health "Business Detrimental to Health"</u>. No. 5 B.E. 2538 (1993).
- Ministry of Public Health. 1997. The Ministerial Notification of the Ministry of Public Health "Business Detrimental to Health (Version II)". No. 12 B.E. 2542 (1997).
- Ministry of Public Health. 1998. <u>The Ministerial Notification of the Ministry of Public Health "Business Detrimental to Health (Version III)".</u> No. 13 B.E. 2543 (1998).
- Ministry of Public Health. 2001. <u>The Ministerial Notification of the Ministry of Public Health "Business Detrimental to Health (Version IV)".</u> B.E. 2546 (2001).
- Minnesota Pollution Control Agency. 2003. <u>Managing Photographic and X-ray Waste</u>. Waste/Hazardous Waste #4,46, 2003.
- Moreno, G. R. 1986. The Recovery of Silver from photographic film: a study of the leaching reaction with cyanide solution for industrial use. <u>Hydrometallurgy</u>. 16 (1986): 395-400.
- Nopakaew, A. 1996. <u>Heavy Metals Adsorption from Leachate Storage Pond Effluent</u>
 by Activated Carbon. <u>Master Degree Thesis</u>. Department of Environmental Engineering, Graduate School, Chulalongkorn University.
- Office of Energy Policy and Planning, Ministry of Energy. 2006. <u>Journal of Energy</u>

 <u>Policy</u>: Available from: Energywww.eppo.go.th/vrs/VRS73.pdf
- Peantumdee, W. 2000. Development of Silver Halide Emulsion for Macro Grain.

 <u>Senior Project</u>. Faculty of Science Chulalongkorn University 2000.
- Seelsaen, N. 2001. Management of Used X-ray Fixer Solution for Northeastern

 Hospitals. Master Degree Thesis. Department of Environmental
 Engineering, Faculty of Engineering, Chulalongkorn University.
- Silawatchananai, C. 1985. <u>Physics of Diagnostics X-rays 1st edition</u>. Bangkok: O.S. Printing House Co.,Ltd.
- Smarntarn, V. 2001. Recovery of silver of used X-ray film by using alkaline protease from Aspergillus oryzae U1521. <u>Master Degree Thesis</u>. Faculty of Science Naresuan University.
- Solomons, T. W. G., and Fryhle, C. B. 2000. <u>Organic Chemistry 7th ed</u>. John Wiley & Sons, Inc., 2000.

- Songkroah, C., and Nakbanpote, W. 2003. Silver recovery from fixer solution.

 <u>Journal of Printing, Pagkaging and Imaging Technology</u>. 73: 37-43.
- Songkroah, C., Nakbanpote, W., and Thiravetyan, P. 2004. Recovery of silverthiosulphate complexes with chitin. <u>Process Biochem.</u> 39: 1553-1559.
- Syed, S., Sureaha, S., Sharma, L. M. and Syed, A. A. 2002. Clean Technology for the Recovery of Silver from Processed Radiographic Films. <u>Hydrometallurgy</u>. 63: 277-280.
- Tameerak, N. 1979. Recovery of Silver from Photographic Solution Waste. <u>Master Degree Thesis</u>. Department of Chemical Engineering, Faculty of Engineering, Chulalongkorn University.
- Thai Customs Department. 2005. <u>Import/ Export Statistics</u>. Available from: http://www.customs.go.th/Statistic/StatisticIndex.jsp.
- The Encyclopaedia of Medical Imaging Volume I. 2004. Film Base. Available from: http://www.amershamhealth.com/medcyclopaedia.
- Tientanacom, W. 1979. Drying Characteristics and the Effect of Drying Conditions on the Properties of Gelatin. <u>Master Degree Thesis</u>. Chemical Engineering, Chulalongkora University 1979.
- United States Environmental Protection Agency (U.S. EPA). 1990. <u>Guides to Pollution Prevention: Selected Hospital Waste Streams</u>. Risk Reduction Engineering Laboratory Center for Environmental Research Information, Office of Research and Development, Cincinnati, Ohio.
- United States Environmental Protection Agency (U.S. EPA). 2004. Unit Conversions, Emissions Factors, and Other Reference Data. November 2004.
- United States Environmental Protection Agency (U.S. EPA). 2006. The Code of Federal Regulations (CFR) Title 40: Protection of Environment. Available from:

 http://ecfr.gpoaccess.gov/cgi/t/text/text-idx?c=ecfr&sid=b7e5aa404c1ec2ecd3261ef741b1acb3&rgn=div6&view=text&node=40:25.0.1.1.2.3&idno=40.
- Wade, L. G. 1995. Organic chemistry 3rd ed. New Jersey: Prentice-Hall, Inc. 1995.
- Washington State Department of Ecology. Environmental Management and Pollution Prevention. 1993. A guide for dental programs. Ecology Publication#97-413.
- World Health Organization. 2002. Silver and Silver Compounds: Environmental Aspects. Concise International Chemical Assessment Document 44. Geneva.

					14	
		A				
			ADDES	DIGEO		
			APPEN	DICES		
					£	
					3 .	
	*					
	OC.					

APPENDIX A ANALYTICAL EQUIPMENTS

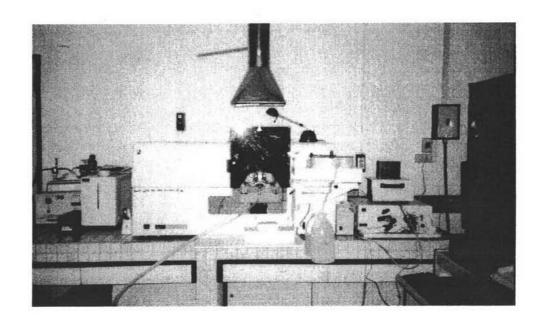


Figure A-1 Atomic Absorption Spectroscopy (Perkin Elmer instruments Model AA analyst 800)

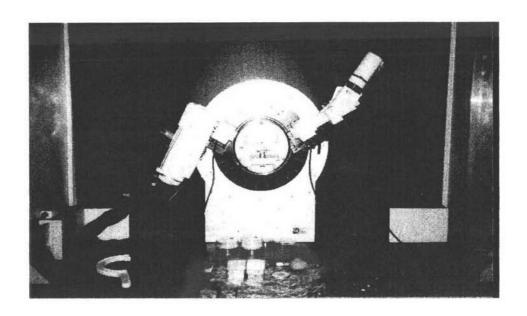


Figure A-2 X-Ray Diffractometer (Bruker Model D8 Advance)

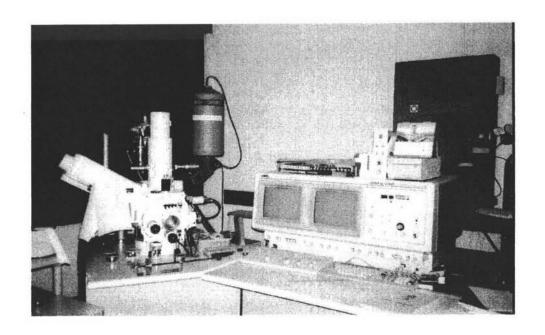


Figure A-3 Scanning Electron Microscope (JEOL Model JSM 6400) equipped with Energy Dispersive Spectroscopy

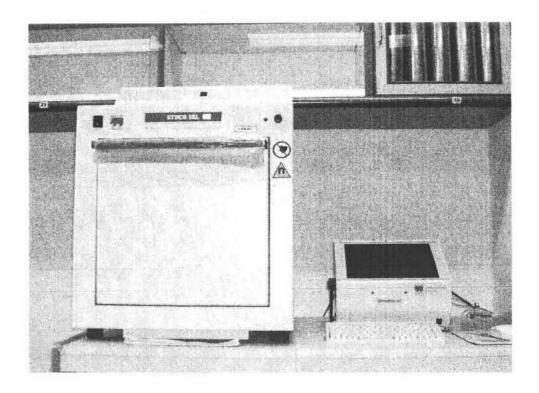


Figure A-4 Microwave for digestion (Milestone Model ETHOS SEI)

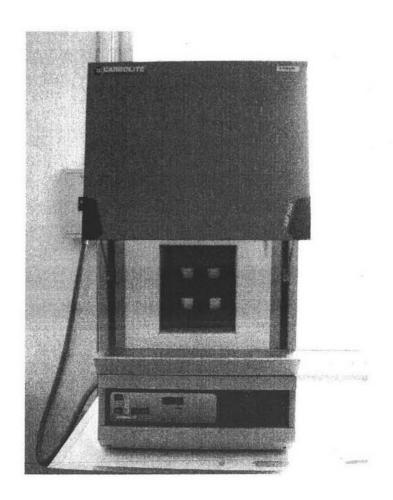


Figure A-5 Ashing Furnace (Cabolite Model AAF 1100)

APPENDIX B CO₂ EMISSION CALCULATION

CO₂ EMISSION CALCULATION

B1. CO₂ Emission Factors

1) CO₂ emission factor of lignite

$$CO_2(kg/kW - hr) = 210.47 \frac{lb}{mmBtu} \times \frac{kg}{2.2046 lb} \times \frac{mmBtu}{10^6 Btu} \times \frac{10,107Btu}{kW - hr}$$

= 0.96 kg/kW - hr

(Use primary kW-hr to Btu Number = 10,107 which is based on ca' 34% Efficiency)

2) CO2 emission factor of natural gas

$$CO_2(kg/kW - hr) = 116.39 \frac{lb}{mmBtu} \times \frac{kg}{2.2046 \, lb} \times \frac{mmBtu}{10^6 \, Btu} \times \frac{10,107Btu}{kW - hr}$$

= 0.53 kg/kW - hr

3) CO2 emission factor of fuel oil

$$CO_2(kg/kW - hr) = 159.66 \frac{lb}{mmBtu} \times \frac{kg}{2.2046 \, lb} \times \frac{mmBtu}{10^6 \, Btu} \times \frac{10,107Btu}{kW - hr}$$

= 0.73 kg/kW - hr

B2. Fuel used in Electricity Plant in Thailand.

For the calculation, natural gas was considered as a sole fuel, as it represented 68% of the fuel sources used for generating electricity Thailand (**Table B-1**).

Table B-1 Fuels used in Electricity Plant in Thailand in 2005.

Fuel source	%
Natural gas	68
Coal (Lignite)	15
Fuel oil	8
Hydro energy	5
Import and others	4

Source: Office of Energy Policy and Planning, Ministry of Energy, 2006

B3. CO₂ Emission from Leaching Process

Condition 1

Temperature: 100°C

Time: 20 min

Used film = 5 g in 50 mL of 5% oxalic acid solution

Equipment

1. Heater

1.1 Thermal energy required to heat up solution from 25°C to 100°C in 7 minutes

From

$$Q = mC_p \Delta T$$

[joule]

m = Mass

[g]

C_p = Specific Heat Capacity

[J/g-°C]

 $T = Temperature [^{\circ}C]$

Substitute for:

m = 55g,

 $C_p = 4.186 \text{ J/g}^{\circ}\text{C},$

 $\Delta T = 100-25 = 75 \, {}^{\circ}C$

So

$$Q = (55)(4.186)(75)$$

= 17.27 kJ

1.2 Electrical heater used in the heat-up 7-minute period

= 58.80 kJ

1.3 Energy to compensate heat loss during 20 min retention time to maintain temperature at 100°C (Temp. control turned on ca. 10 minutes out of 20 minutes).

$$= 84 \text{ kJ}$$

Total electrical energy used for leaching of 5 g used film @ 98%leaching efficiency =
$$58.80 + 84 = 142.8 \text{ kJ}$$

efficiency
$$= 58.80 + 84 = 142.8 \text{ kJ}$$

$$= \frac{142.8 \text{ kJ}}{5 \text{ g}} \times \frac{1,000,000 \text{ g}}{ton}$$

$$= \frac{28,560,000 \text{ kJ}}{ton} \times \frac{hr}{3600 \text{ s}}$$

$$= 7.933 \text{ kW-hr}$$

$$CO_2$$
 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO_2 emission = 4,204.49 kg

2. Stirring

Power to mix 1 ton of used film in 2.5 m³ acid solution is estimated to 5 Hp

=
$$5 \text{ Hp} \times 0.7457 \frac{[kW]}{[Hp]} \times \frac{20 \text{ min}}{60 \text{ min}}$$

= $1.24 \text{kW} - \text{hr}$

$$CO_2$$
 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO_2 emission = 0.66 kg

3. Cooling

Power to supply a cooling unit is estimated to 20 Hp

=
$$20 \text{ Hp} \times 0.7457 \frac{[kW]}{[Hp]} \times \frac{20 \text{ min}}{60 \text{ min}}$$

= $4.97 \text{ kW} - \text{hr}$

$$CO_2$$
 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO_2 emission = 2.63 kg

4. Wastewater treatment

- 4.1 Assumption: leaching 1,000 kg of used film in 2.5 m³ in acid solution
- 4.2 Amount of oxalic acid used = 125 kg in one batch = 12.5 kg when reuse 10 times

$$1 \text{ kg oxalic acid} = \frac{32 \text{ kg COD}}{184} = 0.174 \text{kg COD}$$

$$12.5 \text{ kg oxalic acid} = 0.174 \text{ kg COD x } 12.5 = 2.175 \text{ kg COD}$$

4.3 Amount of leached gelatin = 20% by weight of used film = 200 kg

1 kg gelatin =
$$\frac{6,208 \text{ kg COD}}{4,866}$$
 = 1.276kg COD

200 kg gelatin = 1.276 kg COD x 200 = 255.2 kg COD

- 4.4 Total COD = 2.175 + 255.2 = 257.375 kg COD
- 4.5 Energy for aeration in biological treatment = 1.5-2.5 kg O₂/kW-hr Use average = $2 \text{ kg O}_2/\text{kW-hr}$

4.6 kW - hr needed for wastewater treatment =
$$\frac{257.375}{2}$$
 = 128.787

4.7 CO₂ emission

 CO_2 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO₂ emission

= 68.26 kg

5. Total amount of CO_2 emission = 4,204.49 + 0.66 + 2.63 + 68.26 =4,276 kg

Condition 2

Temperature: 90°C

Time: 60 min

Used film = 5 g in 50 mL of 5% oxalic acid solution

Equipment

1. Heater

1.1 Thermal energy required to heat up solution from 25°C to 90°C in 5 minutes

 $Q = mC \Delta T$ [joule] From

$$m = Mass$$
 [g]

$$C_p$$
 = Specific Heat Capacity [J/g- $^{\circ}$ C]

$$T = Temperature [^{\circ}C]$$

Substitute for:

$$m = 55g$$

$$C_p = 4.186 \text{ J/g}^{\circ}\text{C},$$

$$\Delta T = 90-25 = 65 \, ^{\circ}C$$

So

$$Q = (55)(4.186)(65)$$

$$= 15 \text{ kJ}$$

1.2 Electrical heater used in the heat-up 5-minute period

$$=42 \text{ kJ}$$

1.3 Energy to compensate heat loss during 60 min retention time to maintain temperature at 90°C (Temp. control turned on ca. 10 minutes out of 60 minutes).

Total electrical energy used for leaching of 5 g used film @ 98%leaching

efficiency

$$= 42 + 84 = 126 \text{ kJ}$$

$$= \frac{126 \text{ kJ}}{5 \text{ g}} \times \frac{1,000,000 \text{ g}}{ton}$$

$$= \frac{25,200,000 \text{ kJ}}{ton} \times \frac{hr}{3600 \text{ s}}$$

$$= 7,000 \text{ kW-hr}$$

 CO_2 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO_2 emission =

= 3,710 kg

2. Stirring

Power to mix 1 ton of used film in 2.5 m³ acid solution is estimated to 5 Hp

unit = 5 Hp × 0.7457
$$\frac{\text{[kW]}}{\text{[Hp]}}$$
 × 1hr

$$= 3.73 kW - hr$$

 CO_2 emission factor of natural gas = 0.53 kg/kW-hr

= 1.98 kgThe amount of CO₂ emission

3. Wastewater treatment

- 3.1 Assumption leaching 1,000 kg of used film in 2.5 m³ in acid solution
- 3.2 Amount of oxalic acid used = 125 kg in one batch

= 12.5 kg when reuse 10 times

1 kg oxalic acid=32/184 kg COD = 0.174 kg COD

12.5 kg oxalic acid = 0.174 kg COD x 12.5 = 2.175 kg COD

3.3 Amount of leached gelatin = 20% by weight of used film = 200 kg

1 kg gelatin = 6,208/4,866 kg COD = 1.276 kg COD

200 kg gelatin = 1.276 kg COD x 200 = 255.2 kg COD

$$3.4 \text{ Total COD} = 2.175 + 255.2 = 257.375 \text{ kg COD}$$

- 3.5 Energy for aeration in biological treatment = $1.5-2.5 \text{ kg } \text{O}_2/\text{kW-hr}$ Use average = $2 \text{ kg O}_2/\text{kW-hr}$
- 3.6 kW-hr needed for wastewater treatment = 257.375/2 = 128.787
- 3.7 CO₂ emission

 CO_2 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO₂ emission

= 68.26 kg

4. Total amount of CO_2 emission = 3,710 + 1.98 + 68.26 = 3,780 kg

B4 CO₂ Emission from Existing Technology

Condition

Temperature: 1200°C

Time: 6.67 hr

Used film = 1,000 kg

Equipment

1. Furnace

1.1 Power of blower = 253.5 Hp

unit = 253.5 Hp × 0.7457
$$\frac{[kW]}{[Hp]}$$
 × 6.67 hr
= 1,260.863 kW - hr

 CO_2 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO₂ emission = 668.26 kg

1.2 Fuel oil

Fuel oil = 80 l/hr
Fuel oil =
$$80 \frac{[1]}{[hr]} \times 6.67[hr] = 533.6l$$

unit = $533.6[1] \times \frac{0.985[kg]}{[1]} \times \frac{10^7[cal]}{[kg]} \times \frac{4.186[W-s]}{[cal]} \times \frac{1[hr]}{3600[s]}$
= $6,111.513kW$ - hr

 CO_2 emission factor of fuel oil = 0.73 kg/kW-hr

The amount of CO₂ emission

= 4.461.40 kg

2. Cooling

unit = 8 Hp × 0.7457
$$\frac{\text{[kW]}}{\text{[Hp]}}$$
 × 6.67 hr

$$= 39.791 \,\mathrm{kW - hr}$$

 CO_2 emission factor of natural gas = 0.53 kg/kW-hr

The amount of CO₂ emission = 21.08 kg

3. Bag filter

Power of an equipment = 61.9 Hp
unit = 61.9 Hp × 0.7457
$$\frac{[kW]}{[Hp]}$$
 × 6.67 hr
= 307.879 kw - hr

$$CO_2$$
 emission factor of natural gas = 0.53 kg/kW-hr
The amount of CO_2 emission = 163.18 kg

4. Scrubber

Power of an equipment = 21.8 Hp
unit = 21.8 Hp × 0.7457
$$\frac{[kW]}{[Hp]}$$
 × 6.67 hr
= 108.429 kW - hr

CO2 emission factor of natural gas =
$$0.53 \text{ kg/kW-hr}$$

The amount of CO₂ emission = 57.47 kg

APPENDIX C QUESTIONNAIRS

Radiographic Films Importer Questionnaire

Address	Moo	Soi	Road
			Province
Postal code		Tel	
. The objectiv	es of radiograpl	hic films importin	ıg
•		A. 100 KBART IN 100 - 100 (100 H)	
Obje	ctive		Proportion (%)
□ F	or sell		
□ Ir	dustrial use		
Total			
2. Amount of r	adiographic film	n selling in Thaila	and (average during last year)
2. Amount of r		m selling in Thaila	and (average during last year) Selling amount (%)
	Types	s of buyer	Selling amount (%)
□но	Types	s of buyer	Selling amount (%)
□Ho	Types	s of buyer	Selling amount (%)
□Ho	Types ospital ctory hers (Please des	s of buyer	Selling amount (%)
□ Ho □ Fa	Types ospital ctory hers (Please des	s of buyer	Selling amount (%)
□ Ho □ Fa	Types ospital ctory hers (Please des	s of buyer scribe)	Selling amount (%)
□ Ho □ Fa	Types ospital ctory hers (Please des	s of buyer scribe)	Selling amount (%)
□ Ho □ Fa	Types ospital ctory hers (Please des	s of buyer scribe) Name (Sig	Selling amount (%) gnature) Block Letter

Radiographic Films, Developer and Fixer Using Questionnaire

II		•	2 Dec 24/2000 (1977) (177		
	amount of beds				
	i Road				
Tambol					
Postal Code					
Name)v	Position			
Using and Manager 1. Radiographic Fil 1.1 Radiographic	ms		hic Film	s, Develope	r and Fixe
	-	2002		20	03
1) Amount of consum (Boxes)	nption per year				
2) Film expense per y	year (Baht)				
3) Name of suppliers		3.1		3.1	
		3.2		3.2	
		3.3		3.3	
1.2 Radiographic f	films consumtion	according to the	eir size		
Size Number of sheet per boxes		Net weight (gram/box)	Using amount/year (boxes)		Remark
	sheet per boxes	(Branning)	2002	2003	
☐ 8 inc. X 7 inc.					
☐ 8 inc. X 10 inc.					
□ 10 inc. X 12 inc.					
☐ 12 inc. X 15 inc.					
☐ 14 inc. X 17 inc.					
Total					

1.3 Film	n processing	
	Automatic processing	
	Manual processing	
1.4 Use	d Radiographic films management (after collected for	years)
	Send to the supplier	
	Burn in hospital's incinerator	
	Burn in other organization's incinerator	
	Sell (please fill in 1.5)	
	others (please describe)	

1.5 Used radiographic films selling

7	2002	2003
1) Amount of selling per year (Tons)		
2) Income from selling (Baht)		
3) Buyer's name	NameAddress	NameAddress
	Tel	Tel

2. Developer

2.1 Developer consumption

	2002	2003
Amount of consumption per year (Litre, gallon*)		
2) Developer cost per year (Baht)		
3) Name of suppliers	3.1	3.1
	3.2	3.2
	3.3	3.3

<u>Remark</u>: * in case of gallon, please inform "l gallon = _____ litre"

2.2 Spen	t Developer Management
	Send to the supplier
	Send to hospital wastewater treatment plant
	Send to domestic wastewater treatment plant
	sell (please fill in 2.3)
	Others (please describe)

2.3 Spent Developer Selling

	2002	2003
1) Amount of selling per year (Litre, gallon*)		
2) Income from selling (Baht)		
3) Buyer's name	NameAddress	NameAddress
	Tel	Tel

<u>Remark</u>: * in case of gallon, please inform "1 gallon = litre"

3 Fixer

3.1 Fixer consumption

	2002	2003
1) Amount of consumption per year (Litre, gallon*)		
2) Fixer cost per year (Baht)		
3) Name of suppliers	3.1	3.1
	3.2	3.2
	3.3	3.3

<u>Remark</u>: * in case of gallon, please inform "1 gallon = ______ litre"

	☐ Send to the supplier☐ Send to hospital wa	stewater treatment plant	
	☐ Send to domestic w	astewater treatment plant	
	☐ Sell (please fill in 3.	.3)	
	☐ Others (please descri	ribe)	
3.3	Spent Fixer Selling		
		2002	2003
) Amo	ount of selling per year		
(Litre	e, gallon*)		
2) Inco	me from selling (Baht)		
) Buye	er's name	Name	Name
		Address	Address
		Tel	Tel
Remarl	\underline{k} : * in case of gallon, ple		
Remari	₹: * in case of gallon, ple		
	k: * in case of gallon, ple	ase inform "1 gallon =	litre'
low to		ase inform "1 gallon =	litre'
low to	select the buyer/waste d	ase inform "1 gallon =	litre'
low to	select the buyer/waste do	ase inform "1 gallon =	litre'
Iow to) Kind □	select the buyer/waste do of waste dealer (you can a General person	ase inform "1 gallon = lealer of used films and s answer more than 1 item)	litre'
How to) Kind □ □	select the buyer/waste do of waste dealer (you can a General person Juristic person Both of general and juris	ase inform "1 gallon = lealer of used films and some series and the series are than 1 item)	litre'
How to) Kind □ □	of waste dealer (you can a General person Juristic person Both of general and juris	ase inform "1 gallon = lealer of used films and somewhat makes than 1 item) stic person er more than 1 item)	litre'
Iow to Kind	of waste dealer (you can a General person Juristic person Both of general and juris umentation (you can answe	lealer of used films and somewhat the stic person ter more than 1 item)	litre'
low to) Kind	of waste dealer (you can a General person Juristic person Both of general and juris umentation (you can answe Copy of national identifi Copy of business establish	lealer of used films and somewhat the person than 1 item) er more than 1 item) cation card shment license	litre'
How to) Kind □ □	of waste dealer (you can a General person Juristic person Both of general and juris umentation (you can answe	lealer of used films and somewhat the person er more than 1 item) cation card shment license tion certificate	litre'

) Crite	ria of selection (you can answer mor	re than 1 item)	
	Price		
	Waste treatment permit		
	Others (please describe)		
		Date	

APPENDIX D UNIFORM OF HAZARDOUS WASTE MANIFEST

						м	นายเลขในกำกับการ	งนฮ่งของเสียกัน สราย	: Munifest Nu.
			1	บกำกับกา	รขนส่งของเลี	เยอันตราย			
			. (Un	iform H:	azardous Was	ste Manifest)		
		1. ฮ่วนขอ			This section must				
เม ชื่อ : nam	e				2) เถษประจำด้ว	ยู้ก่อกำเนิดของเสีย	อันคราย : Generato	r's ID	
สถานก่อกำเน็ด : Generator address					ไทรศัพท์ : Phone				
3) ค้าบส่วน	องเสียอันคราย : Transport	er							
าายที่ ! ชื่อมริษัท : First company name					เถขาไระจำด้าผู้ขนส่งของเสียกันตราย รายที่ t : Transporter's ID				
		ame			เลขประจำตัวผู้ขา	นส่งของเสียอันคร	าย รายที่ 2 : Transp	orter's ID	
		เสียอันคราย : Treatment Stor		ilines (TSDF	Fs)				
						บรวบราม บำบัด	และกำรัดของเสียอัว	เพราช ราชที่ I Dispose	r's ID
		ine	V-10-11					เคราช ราชที่ 2 Dispose	
	ยลของของเสียลันสรายที่ขา								
ลำดับ	รายละเอียด	รหัสของเสีย	หมวดข้อวัสดุ	ที่ไม่ใช้แล้ว	การนะบรรจ	: Cuntainers	ปริบาครสุทธิ	หน่วยน้ำหนัก :	รายละเอียดเพิ่มเดิน :
Nu.	(Description)	ลับตราช Waste ID.	иили	võ	จำนวน : No.	ชนิด : Type	Quantity	Unit Wt / Vol	Additional Information
-	(Ocacinpinon)						150.20		
-		-							
	: แลมสีมลับละ พ.ทั้งนาว	Total Quantity VONMAT:	lionid	ก็ตร/กดา	ายกัฒลร : Literate	เต ของเซ็ม	: solid	โลกรีบ / ตับ : Kgs. / ก	ons
	กิจางเลสสนหรายการมหา คิที่มีลักษณะพิเศษ และจักร		nquiu	on ingine	THINDING CHEST				
		-							
	andling Instructions and a		4		2 4 4		y .		
7) คำรับรอง	ะทักเหล้าขอรับรองว่าใต้ส	ทั้งมอบของเสียอันคราชแ ล้ วค	ามที่ระบุข้างต้น แ	เละมีการบรร	รุติคปี เขหรือฉลากเ	ยขางเหมาะสมคร	เลามข้อถ้าหนดของ	ບປືນກາຄມົບຖວະບາວ	A MESTURIOSA.
Generator	Certificate: I hereby declare	that the contents of this consign	ment are accurately	y desembed ab	ove and have been pa	cked and labeled an	d are in proper conditi	on for transport according	to regulation
as vo Gener	ator's name	a						onth	M.W : Year
		2. ส่วนขอ	งผู้ขนส่งของเสีย	อันคราย: Th	is section must be	1			T 4
() ซื้อผู้ขนส่งราชที่ 1 : Transporter's name					2) พาหนะที่ใช้	, 100		8004	11 เครื่องบิน
ากงประจำด้าผู้ชนส่ง : Transponee's ID โทรสัทท์ : Phone โทรสาร : Fax อุณจิน : Emergency					Vehicle	Tru	ck Tra	in Ship	Plane
โทรสิทท์ : P	hone Insert	ว : Faxอุกเจ็น :	Emergency		3) เสขทะเบียน	CALCO I		1	
			no como reservo		Mmuz : Vehicle				
		ับของเสียกับครายแล้วตามที่							
		clare that I have received the							
-		nc							
5) ชื่อผู้บนสำเวอที่ 2 : Transporter's name				10.340,40.440,40.400	6) พาหนะที่ใช้ โรกบร		ไฟ ีเรือ	่ ⊓ เครื่องบิน	
เลขประจำคับผู้ขนส่ง : Transporter's ID					Vehicle	Tru	ck Tra	in Ship	Plane
โทรศัพท์: Phone ใทรสาร: Fax กุกเฉ็น: Emergency					7) เลขทะเบียน				i
					พาหนะ : Velnel				
		ับของเสียอันครายแล้วตามที่							
Transporter	Certification: I hereby de-	clare that I have received the	type and quantity	y of waste as	described above by	the generator and	d that waste has bee	n transported according	g to regulations.
โดยขนส่งจ	ากจังหวัด : From		"ไปยังจังหวัด T	n	0.000	ใช้ระบะเวล	าประมาณ Time M	ending	## fiu : hours/day
ลงที่อผู้ขบก่	41 1011 2 Transporter's nar	nuc	mannen !	อายเจ็น : sig	nature	วันก่	i . Datc เทีย	34 : Month	
		3. ส่วนของผู้ประกอบการเ	สถานเก็บรวบรวม	น บำบัด และ	ะกำจัดของเสียอันดา	110: This section	must be complete	d hyTSDFs	
เ) ชื่อผู้วันกำ	าจัด TSDF's name								
สถานที่กำจั	≥: TSDF's address		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		ไทรศัพท์ Phon	e	ไทรสาร : Fax	กรณีลุกเฉิน - E	mergency
3) คำรับรถเ	· จำพเจ้าขอรับรองว่าได้รั	ับของเสียอันตรายเล้าคามป่	ริบานที่ระบุจ้างค้า	นนี้	-				
TSDF certif	ficate of arrival : I hereby d	declare that I have received the	ne reference load.						
		ภายในระยะเวลา Treatment			L เคียน :month	Ll îl : year û	บรากวันที่ใต้รับขอ	เลีย : since the day tha	i received waste
กาซื้อด้รับกั	าพัด : TSDF 's name		อายเชิ้น : Signat	urc		วันที่ : Date	เพื่อน : Month	и.н.	: Ycar
		14 Discrepancy Nutification							
		ste			1 July Country				
		รเธ สำคัญ Returned จัดป						tweeta - Reason of	action
านหล่งคืน	Date returned/	./ (วัน/เดียน/ปี :dd / เกเ	п/уу) иштогач	ะเนกากับกา	สมสาของเสริยม	a maamay : Re	turneu manifest no		w-11400000000000000000000000000000000000

APPENDIX E PUBLISHED PAPERS

ORIGINAL PAPER

Policy concept applied to X-ray waste management in Thailand

P. Khunprasert · N. Grisdanurak · J. Thaveesri · V. Danutra · W. Puttitavorn

Received: 4 April 2006/Accepted: 17 July 2006 © Springer-Verlag 2006

Abstract In Thailand, 98.5% of the imported X-ray films are used for medical services. After the developing process, approximately 55-65% of metallic silver still remains on the developed X-ray films. The used films are practically kept at the hospitals for record-keeping purposes for 5-10 years, while the developing agents are reused at minimal twice. The discarded films and the spent (fixer) solution are normally sold to waste dealers prior to their delivery to waste processors for silver recovery. To recover silver, typically, the used films are chemically leached; the spent solution is processed by electrolysis technique. Nevertheless, the recovery processes also generate toxic chemical wastes, which are commonly spilled or discharged into the environment without any proper treatment and disposal. The dealers for

X-ray waste are one of the major stakeholders who are apparently exempted from state control and regulations; but, depending on its size and capacity, the processors must have a factory-operating permit. It appeared that the key regulators of X-ray waste are the Ministry of Industry and the Ministry of Public Health. In this study, the monitoring and tracking of silver-contaminated wastes were strictly carried out from the generation points to their ultimate disposals. A fact sheet of the management of X-ray waste has been proposed.

Keywords X-ray waste · Waste management · Fact sheet

P. Khunprasert - V. Danutra National Research Center for Environmental and

Hazardous Waste Management, Chulalongkorn University,

Bangkok 10330, Thailand

N. Grisdanurak (🖂)
Department of Chemical Engineering,
Faculty of Engineering, Thammasat University,
Klong-Luang, Pathumtani 12120, Thailand
e-mail: gnurak@engr.tu.ac.th

J. Thaveesri

Department of Industrial Works, Ministry of Industry, Bangkok 10400, Thailand

W. Puttitavorn

Environmental Research Institute, Chulalongkorn University, Bangkok 10330, Thailand

Abbreviations

 Ag^0 Metallic silver Ag+ Silver ion Silver halide AgX F Material flow into and out for each hospital inside the control volume F&M Fabrication and manufacturing HA Hospital accreditation MOI Ministry of Industry MOPH Ministry of Public Health PWG Primary waste generator 5 The stock of material retained or depleted from the reservoir over a time period SWG Secondary waste generator WD Waste dealer

WG Waste generator
WP Waste processor
WR Waste regulator

Introduction

Silver is a precious metal, which can be feasibly recycled from its contaminated wastes. With an increasing demand for silver in the world, the secondary production of silver from scraps and other forms of wastes has become a new supply in the market (Handerson 2003). Owing to its considerable silver content, X-ray waste (i.e., used films and spent fixer) is one of secondary silver sources, which has recently been focused on.

To date, there have been few articles reported on the cycles of silver (Johnson et al. 2005; Lanzano et al. 2006). These articles have described the stock and flow of the silver cycle in an overall depiction focused on the industrial usage of silver from its ore extraction to its final disposal. Therefore, an investigation on the flow patterns of silver originated from X-ray waste was undertaken in this study in order to improve its management system in Thailand.

X-ray films contain greater amounts of silver than that in any other types of film due to the radiation-sensitive silver halide (AgX), which is scattered on the gelatin of the emulsion layer on both sides of a polyester sheet (i.e., the film base). Traditionally, 94–98% of X-ray films are used in medical services, e.g., chest X-rays, mammograms, CT scans, etc. Annually, X-ray films of ca. 2 billions are commercialized around the world (Gambhir 2004).

In the developing process of X-ray films, light-exposed silver ions (Ag⁺) are reduced by a developer solution to metallic silver (Ag⁰) and retained on the film. The unexposed silver is then dissolved in the form of silver thiosulfate into the fixer solution (Colorado Industrial Pretreatment Coordinators Association 2004). The overall development process of an X-ray film is illustrated Fig. 1. Wastes such as film scraps,

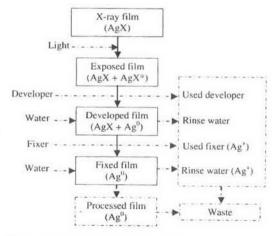


Fig. 1 Development process of X-ray films

spent fixer, spent developer, rinse-water, and spilled chemicals are mainly generated within hospital radiology departments' premise.

With respect to recovery technologies, there are several techniques for recycling of silver from certain photographic wastes such as developed films and spent (developer or fixer) solution. There are four common methods being used to leach silver metal from used X-ray films, which are thermal (Hochberg 1989), biological (Fujiwara et al. 1991; Ishikawa et al. 1993), mechanical or physical (Hochberg 1989; Buser and Retting 1988), and chemical treatments (Moreno 1986: Messerschmidt 1988; Laungchonlatan 1988; Ajiwe and Anyadiegwu 2000; Syed et al. 2002). The first method relies on combustion for the recovery of silver and energy; while the rest not only recover silver, but also recycle the film base. The silver sludge, obtaining from the latter three methods, is smelted and further refined to form silver ingot.

Silver can be recovered from the spent fixer solution by electrolysis (Bourges et al. 2002), metallic replacement, chemical precipitation, reverse osmosis and ion exchange (Atia 2005; Cerjan-Stefanovic et al. 1991). The last two methods provide a way of increasing silver concentrations in a waste solution, so they are suitable for solutions with low silver content, i.e., rinse water and spent developer solution. Electrolysis and metallic replacement are effective methods that lead in a straightforward way to the retrieval of silver ingot. A simple depiction of these recovery methods is shown in Fig. 2. Recently, some researchers have confirmed the

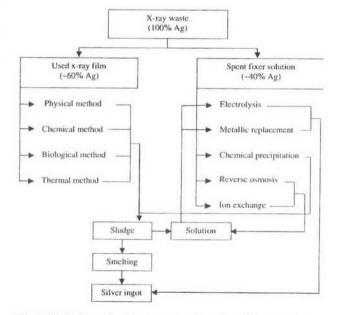


Fig. 2 Techniques for the recovery silver from X-ray waste

effectiveness of a technique for the recovery of silver from spent fixer solution by the silver's adsorption to activated carbons (Adani et al. 2005) and to chitin (Songkroah et al. 2004).

X-ray waste poses potential hazards towards public health and the environment because it typically contains significant concentrations of silver. Silver, Ag+ and its compounds have been reported to have higher bioconcentration factors in marine flora and fauna than in the fresh water organisms. Though it seems that the accumulation of silver by terrestrial plants from soils is low; silver (as silver nitrate) showed an adverse effect on sensitive laboratory mammals at total concentrations as low as 250 µg/l in drinking-water (brain histopathology), 6 mg/kg in diet (high accumulation in kidneys and liver), or 13.9 mg/kg body weight (lethality) (World Health Organization 2002). X-ray waste is generally considered hazardous when contaminated with silver to a certain extent. The United States Environmental Protection Agency (USEPA 2006) defines a solid waste containing silver in excess of 5 mg/l, determined by the Toxicity Characteristic Leaching Procedure (TCLP), as category D011 hazardous material.

Of importance, there is little information on how X-ray waste is manipulated in Thailand. Most researches (Tameerak 1979; Seelsaen 2001; Songkroah 2004) focused on technology for silver recovery from the waste more than on management policy. This paper reports on the present situation of X-ray waste in Thailand with emphasis on material flow analysis of silver and proposes future prospects for its management.

Present situation of X-ray waste management in Thailand

One important tool of industrial ecology based upon the law of conservation of mass to monitor the flow of specific substances is a Substance Flow Analysis (SFA). SFA is used for identifying environmental problems and proposing remedial/prevention strategies (Kapur 2006). In this study, the existing situation of X-ray waste management in Thailand is explained through a SFA to determine how much silver is disposed of into the environment. The flow of silver consists of four stages which are (1) production, (2) fabrication and manufacturing (F&M), (3) use, and (4) waste management (WM). In this analysis, the system boundary only limits the flow of silver in the production and F&M stages. A flow diagram of the silver waste life cycle is presented in Fig. 3.

A mass balance equation has been formulated to access the flow patterns of the silver waste cycle. It determines the recovered and lost amounts and establishes the amount entering the material reservoirs.

The mass balance equation is as follows:

$$dF/dt = \sum F_{\text{input}} - \sum F_{\text{output}} + \sum (S_t - S_{t-1}), \quad (1)$$

where F represents the material flow into and out for each hospital inside the control volume and S represents the stock of material retained or depleted from the reservoir over a time period.

Production of X-ray waste

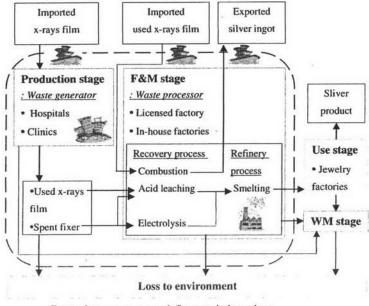
In a production phase, it was found that there is currently no film manufacturer in Thailand and X-ray film products are all imported. Silver flows in an active form (AgX), coated on the X-ray film, into the study boundary. The application of the film has been mainly used for medical purposes. It was found that 98.5% of imported films are distributed to hospitals, medical institutes, and clinics; these utilizers produce wastes at their premise and become the primary waste generators (PWGs).

The stocks of silver did not change from year to year; this was due to the fact that a procurement amount of new X-ray films is carried out on an annual basis and the amount of used films, i.e., retained silver primarily depends upon the space of the used film collection area in each hospital. For example, for the year 2003–2004, the accumulation term approaches zero under steady-state conditions. The net accumulation of material equals to the summation of silver influx minus silver outflow in a defined control volume. Hence, Eq. 2 can be written as follows:

$$F_{\text{silver}} = \sum_{i \text{nput}(2004)} F_{\text{input}(2004)} - \sum_{i \text{output}(2004)} F_{\text{output}(2004)} + \sum_{i \text{output}(2004)} S_{(2004)} - \sum_{i \text{output}(2004)} S_{(2003)}.$$
(2)

All X-ray films are imported by the suppliers, and their quantity is recorded by the Department of Customs, under the Ministry of Finance. Figure 4 indicates the amount of imported film from 2001–2005. The annual imported amount of film has increased in 2002 and then remains stable at a range of 630–640 ton. According to ca. 2% silver content (by weight) in X-ray film, this amounts to 12.7 ton of silver as the main material input in the defined system.

Fig. 3 Material flow analysis of silver in X-ray waste



Remark: - - defines study boundary

In addition to domestic supply, used films are imported by a waste processor (WP). The price of imported used films is apparently cheaper than the domestic used film. At present, the amount of imported used films is legally limited as it is allowed for a factory test-run operation. In average, approximately 1% silver content (by weight) is found in the used X-ray film, this amount to 0.1 ton of silver as the other material input in our defined system.

After films being used for internal effect diagnosis assistance, silver becomes (primary) waste as spent photographic solution and solid sheet material. It was found that, silver, in both its metallic and compound forms, remains on the film after being developed and in the spent fixer at 55–65 and 35–45%, respectively. Finally, these wastes are generally sold to waste dealers

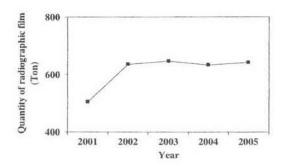


Fig. 4 Quantities of imported X-rays film from 2001 to 2005 (Thai Customs Department 2005)

(WDs) rather than directly to WPs for the further recovery of silver.

However, a portion of the silver sinks into the environment, through rinse water and used developer solution which are contaminated with silver during the developing process. These solutions are usually drained directly into sewerage system with silver concentrations in the range of 12–100 mg/l, which is, according to our best estimation, ca. 0.04 ton of silver.

F&M of X-ray waste

The used X-ray films and spent fixer enter silver recovery factories as secondary raw material. In Thailand, these kinds of factories are categorized as a typed-106 factory since they are engaged in the recycling or recovery of materials from industrial wastes and/or discarded industrial products. With a defined number of workers or horse powers, the 106-type activity becomes a factory and is controlled under certain regulations promulgated by the Factory Act B.E. 2535 (1992). To date, only a single silver recovery factory has been officially registered with the Department of Industrial Works, under the Ministry of Industry (MOI). This factory imports used X-ray films from abroad and uses them as raw materials in a combustion recovery process. Approximately, 0.09 ton of silver is recoverable by this method.

The majority of the silver recovered from used X-ray films and spent fixer solution is processed by the so-called "house" factories. These small-scale WPs

conduct their businesses without any facilities to treat their wastes and become a large group of secondary waste generators (SWGs).

Electrolysis and acid-leaching are the most commonly used silver recovery techniques. Electrolysis is an effective method for extracting silver from solution, whereas combustion and acid leaching are suitable for retrieving silver from used films. Approximately, 95–96% of the silver contained in the used film and fixer solution, which is equivalent to 12.1 ton, is recoverable.

Silver ash produced by the combustion treatment and sludge generated from the chemical leaching process are smelted and refined to obtain silver ingots, while other silver-contaminated chemicals cross the boundary into the environment or enter a waste treatment plant before release into the environment. The silver ingots are sold to domestic jewelry and silverware manufacturers.

The residual wastes, including wastewater and leached plastic sheets, which are contaminated by traces of silver, are sent for external waste management outside the system boundary. Only 40% of the wastewater has been treated in-house by neutralization or dilution prior to discharge into the environment. It was found that the amount of silver loss in the effluent is approximately 0.55 ton. Furthermore, the analysis of leached film showed that an average of 0.08% of silver by weight still retains on the film sheet.

Silver flow of X-ray waste throughout the production and F&M stages is illustrated in Fig. 5. Even total losses of 0.6 ton of silver through the biosphere in our system boundary are not comparable to the material available in the system outside. The wastes, especially the emissions from the "in-house" WPs, are classified into hazardous material categories (USEPA 2006).

Waste management

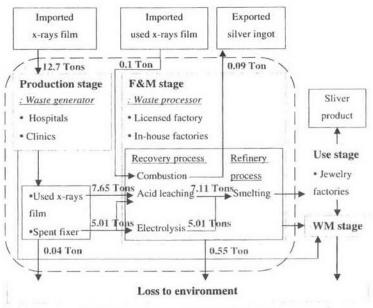
Handling waste between waste generators and waste processors

One hundred out of the entire 424 hospitals in the country, including both private and public hospitals throughout the country, were selected as the target group (National Statistic Office 2004). The data obtained from our integrated questionnaires, which focused on the hospital's management of X-ray waste, indicated that management practices differ among the hospitals. This was mainly due to the different hospitals, clinics and medical centers are under different authorities. Approximately, 87% of hospitals keep their used X-rays films for 5–10 years for their medical records; while spent fixer is reused minimal twice and collected for a number of years until it reaches a sufficient amount to be sold.

In 2002 and 2003, used X-ray films were sold approximately at a price of 839 and 777 US dollars per metric ton, respectively. The price of spent fixer was approximately 680 US dollars per cubic meter.

The X-ray WDs can be either general or juristic persons. A dealer who offered the highest price gets

Fig. 5 Silver flow of X-ray waste in production and F&M stages. All units measured in kg Ag (from 2003 to 2004)



Remark: - - defines study boundary

the X-ray films and fixer solution. In general, a bidding process is employed at a hospital where enormous amount of X-ray waste is produced to gain maximum income. These WDs are not required to provide information on their recovery processes and, especially, how the recovery of the waste is managed. Figreveals the percentages of document requirements which are a national identification card, a business establishment license, a resident registration certificate, and combinations of the afore-mentioned documents. The data showed that 44% of the studied hospitals requires merely a national identification card, and 39% of the samples requires both a national identification card and a business establishment license. Yet, only 4% of them does require all these documents.

At present, there has been no specific state notification for management of X-ray waste; as a result, it depends on individual hospitals to set up their own requirements. It appears that most hospitals have little concerns for the wastes they sell. In prequalification process of the WD, the hospitals neither take into consideration the reliability of the WP, who actually process the waste, nor require the WD to disclose their WP and the processors' operating permits. This means that no information about these WP is officially available. Therefore, the concern of hospitals for their X-ray waste seems to end up after the bidding process end because the hospitals have no clues to which their X-ray wastes are transported and how the wastes are subsequently processed.

Problems of X-ray waste management

Based on our analysis, it was found that there are intrinsic problems occurred in a system. Figure 7 shows

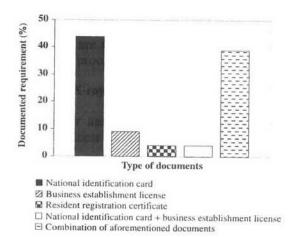


Fig. 6 Percentages of required documents

the relationship between an X-ray material flow and key stakeholders. The figure depicts five major stakeholders: (i) X-ray film and fixer suppliers, (ii) primary X-ray WGs, (iii) X-ray WDs/vendors, (iv) X-ray WPs or secondary X-ray WGs, and (v) X-ray waste regulators (WRs).

First of all, for the supply side, there is adequate statistics from the Department of Customs, under the Ministry of Finance; as the materials are all imported into the country. The mass quantity and the names of importing firms are well-documented and the information is available for public.

Secondly, the PWGs, which can be divided into two groups: Large PWG and Small PWG. The Large PWG, i.e., hospitals or medical services, account for a large portion of X-ray waste quantity. In general, they have been subjected to some regulations endorsed by the Public Health Act B.E. 2535 (1992) and issued by the Ministry of Public Health (MOPH). Nevertheless, up to the present, there has not been any specific rule for the control of X-ray waste from Large PWG issued by MOPH.

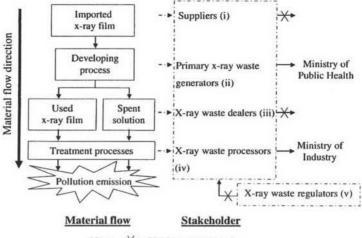
The Small PWG, i.e., factories and some photo shops that use X-ray films in their production processes, are small X-ray waste producers. X-ray waste from these factories is currently under the control of MOI by use of the Factory Act B.E. 2535 (1992). Yet, there is no control for the photo shops.

Thirdly, the X-ray WPs, i.e., used film and spent fixer recycling factories, which are, on the other hand, the SWGs. As it was previously described, only a single WP is registered and controlled by MOI via the enforcement of the Factory Act B.E. 2535 (1992); while a number of "house" factories remains unknown.

It appears that the major feed raw materials for a silver recovery factory would be collectable from the medical services within the country. But, actually, the X-ray waste of domestic origin has hardly ever processed at the only one licensed factory in Thailand. This factory has to import used X-ray films for its recycling operation. It is obvious that a loophole exists in the state control system.

The X-ray waste from industrial operations, both as PWG and SWG (or WP) is legally defined as "industrial hazardous waste" according to the Ministerial Notification No. 6 B.E. 2540 (1997) of the Factory Act B.E. 2535 (1992). This Act requires a factory that produces and/or recycles its X-ray waste to apply for an operating permit for their factories and a waste evacuation license from the MOI. However, this law can only be enforced on a factory; but not applicable to the other major WG (i.e., hospitals), which are regulated by the Public Health Act B.E. 2535 (1992) and to a

Fig. 7 Relationship between X-ray material flow and stakeholders



Note: X Unclear control unit.

"house" factory. The "house" factory is not considered as "a factory" under the Factory Act B.E. 2535 (1992), unless it utilizes power equal or more than 5 horse powers and/or employs equal or more than seven workers.

Fourthly, the X-ray WDs, who actually transfer wastes from the generators to the processors, are free waste traders; no government body is designated to overlook these dealers. As a result, it is very difficult to identify where the waste commodity is distributed. Thus, it is not possible to locate where the waste is delivered and to specify who manages it.

Finally, the WRs, there occurs both regulatory gaps and overlapping jurisdictions among government authorities; this results in a non-systematic management of X-ray in Thailand. Hospitals as Large PWG, which are affiliated to and/or under a control different state agencies, are lacking of close cooperation to set up measures to control their X-ray wastes, such as defining of waste (to be hazardous or non-hazardous), separation and collection procedures, rules to select the WD and WP, and monitoring measures.

At present, it seems that there is one direction of material flow without any monitoring system. Pollution originated from silver recovery processes continues to release into the environment without awareness or proper treatment. Official records of the X-ray waste are not sufficient for planning of a proper management of the waste.

Future prospects for X-ray waste management

In Thailand, the existing measures are insufficient for controlling X-ray waste. In order to achieve the goal of

complete waste management, silver as a hazardous element should be strictly monitored and tracked from the generation points to their ultimate disposal sites as proposed in Fig. 8. Pollution prevention guidelines in the form of a fact sheet were then developed to manage silver-contaminated wastes for businesses that produce the wastes. The main subjects covered in the fact sheet are type of X-ray waste, lists of management and disposal options, manifest system, and name lists of WD and silver WP in Thailand.

Type of X-ray waste

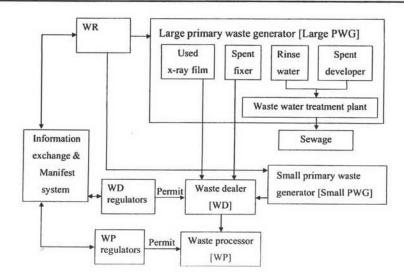
X-ray waste consists of used X-ray films, spent fixer, spent developer, water rinse and chemical vapors. The used X-ray films and spent fixer are considered as hazardous wastes due to their high silver content.

Lists of management and disposal options for spent photographic solution and used films from waste generators to waste processors

Because silver content in both spent fixer and used film can be recycled, the WGs should handle these wastes separately from other wastes. For instance, the fixer should not be mixed with spent developer.

Of importance is the selection of the X-ray waste management company because their liability should not end up when the waste is evacuated from the hospital facility. Guidelines have been established for choosing a company for an off-site disposal of X-ray waste. For example, the recyclers have to apply for a permit from the MOI. In addition, they would have to treat the wastes generated from the recycling process prior to release it into the environment.

Fig. 8 Proposed X-ray waste management system



Manifest system for X-ray waste

A manifest system for X-ray waste is needed to control and to monitor the waste. A manifest paper is a management tool to keep track on the silver-contaminated hazardous waste during transportation. The manifest serves as a legal document for hazardous wastes that would not be able to be shipped off-site without it. The WR should establish the linkage with each other to jointly and uniformly monitor the waste.

Name lists of dealers and silver waste processors in Thailand

Both WD and WP have to register with the regulatory bodies. Fundamental information of them such as name, address, a silver recycling method and a waste treatment process, should be publicized to facilitate and to normalize a selection for WD. Only a dealer associated with licensed silver recycler should be allowed to enter the bidding process of the WG.

The WR might use the Hospital Accreditation (HA) standard as an incentive tool to help controlling X-ray waste by adopting the fact sheet as a specification for X-ray waste management. HA certification system plays a key role for improvement of hospital quality by increasing both efficiency and competitiveness. Waste management is mentioned as one aspect of this standard; however, its specifications have not set up yet.

Conclusion

X-ray waste management in Thailand involves a number of parties: the WGs, the WDs, the WPs and the WRs. Inadequate regulatory attention is paid on X-ray waste management. A material flow analysis was used for investigation of the silver loss to the environment. It was found that a considerable amount of silver has been released into the environment. The discharge mostly comes from the small WPs, which are not registered and licensed. These "house" factories exempted environmental rules and regulations. The concrete rules for the WPs are yet to be formulated. The fact sheet for management X-ray waste proposed includes necessary measures to keep the X-ray waste from on-going deterioration of the environment.

Acknowledgments Special thanks are extended to the Cleaner Technology Internship Program 2006: Thammasat University Node for financial support.

References

Adani KG, Barley RW, Pascoe RD (2005) Silver recovery from synthesis photographic and medical x-ray process effluents using activated carbon. Miner Eng 18:1269–1276

Ajiwe VIE, Anyadiegwu IE (2000) Recovery of silver from industrial waste, Cassava solution effects. Sep Purif Technol 18:89–92

Atia AA (2005) Adsorption of silver (I) and gold(III) on resins derived from bisthiourea and application to retrieval of silver ions from processed photo films. Hydrometallurgy 80:98–106

Buser KR, Rettig TA (1988) Recovery of silver from photographic films by high shear conditions. Canada Patent, 1238790

Bourges J, Broudic J, Dauby J, Leboucher I, Leclerc O, Baticle P (2002) Electrolytic method for the recovery and recycling of silver from a nitric acid solution. United States Patent 6,428,679

Cerjan-Stefanovic S, Briski F, Kastelan-Macan M (1991) Separation of Silver from waste water by ion-exchange resins and concentration by microbial cells. Fresenius J Anal Chem 339:636-639

- Colorado Industrial Pretreatment Coordinators Association (CIPCA) (2004) Report on silver discharge to WWTP (online). http://www.cipca.org/forms/silver/SILVERRPT.DOC. Cited 27 May 2004
- Department of Industrial Works, Ministry of Industry (1997) The Notification of the Ministry of Industry no. 6 B.E. 2540
- Department of Industrial Works, Ministry of Industry (1992) The Factory Act B.E. 2535
- Fujiwara N, Yamamoto K, Masui A (1991) Utilization of a thermostable alkaline protease from an alkalophilic thermophile for the recovery of silver from used X-ray film. J Ferment Bioeng 72:306-308
- Gambhir SS (2004) Orientation: photographic emulsions, pharmacology M248. Introd Biol Imaging (online)
- Handerson K (2003) The contemporary silver cycle for CIS countries: using industrial ecology to evaluate silver flows. J Young Investigators (online)
- Hochberg J (1989) Recovery of silver from photographic film with high shear and caustic. United States Patent 4,799,954
- Ishikawa H, Ishimi K, Sugiura M, Sowa A, Fujiwara N (1993) Kinetics and mechanism of enzymatic hydrolysis of gelatin layers of X-ray film and release of silver particles. J Ferment Bioeng 76:300–305
- Johnson J, Bertram M, Henderson K, Jirikowic J, Graedel TE (2005) The contemporary Asian silver cycle: 1-year stocks and flows. J Mater Cycles Waste Mang 7:93-103
- Kapur A (2006) The future of the red metal—a developing country perspective from India. Res Cons Recycling 47:160– 182
- Lanzano T, Bertram M, De Palo M, Wagner C, Zyla K, Graedel TE (2006) The contemporary European silver cycle. Res Cons Recycling 46:27-43
- Laungchonlatan S (1988) Feasibility studies of synthesis the silver-sulfadiazine from used X-ray films. Department of

- Pharmaceutical Chemistry, Faculty of Phamaceutical Science, Chulalongkorn University
- Messerschmidt H (1988) Method for recovering silver from waste photographic film and paper. United States Patent 4,759,914
- Moreno GR (1986) The recovery of silver from photographic film: a study of the leaching reaction with cyanide solution for industrial use. Hydrometallurgy 16:395-400
- Seelsaen N (2001) Management of used X-ray fixer solution for Northeastern Hospitals, Department of Environmental Engineering, Faculty of Engineering, Chulalongkorn University
- Songkroah C, Nakbanpote W, Thiravetyan P (2004) Recovery of silver-thiosulphate complexes with chitin. Process Biochem 39:1553–1559
- Syed S, Sureaha S, Sharma LM, Syed AA (2002) Clean technology for the recovery of silver from processed radiographic films. Hydrometallurgy 63:277-280
- Tameerak N (1979) Recovery of silver from photographic solution waste. Department of Chemical Engineering, Faculty of Engineering, Chulalongkorn University
- Thai Customs Department (2005) Import/export statistics (online). http://www.customs.go.th/Statistic/StatisticIndex.jsp. Cited 17 Jun 2005
- United States Environmental Protection Agency (USEPA) (2006) The code of federal regulations (CFR) title 40: protection of environment (online). http://www.ecfr.gpoaccess.gov/cgi/t/text/text-idx?c=ecfr&sid=b7e5aa404c1ec2ecd3261ef741b1acb3&rgn=div6&view=text&no-de=40:25.0.1.1.2.3&idno=40. Cited 10 Jan 2006
- World Health Organization Geneva (2002) Silver and silver compounds: environmental aspects, Concise International Chemical Assessment Document 44





Journal of Cleaner Production xx (2006) 1-9



Radiographic film waste management in Thailand and cleaner technology for silver leaching

P. Khunprasert a, N. Grisdanurak b,*, J. Thaveesri c, V. Danutra a, W. Puttitavorn d

* National Research Center for Environmental and Hazardous Waste Management, Chulalongkorn University, Bangkok 10330, Thailand b Department of Chemical Engineering, Faculty of Engineering, Thammasat University, Klong-Luang, Pathumtani 12120, Thailand c Department of Industrial Works, Ministry of Industry, Bangkok 10400, Thailand d The Institute of Environmental Research, Chulalongkorn University, Bangkok 10330, Thailand

Received 26 February 2006; accepted 16 June 2006

Abstract

In Thailand, a large amount of radiographic film waste is generated in hospitals. Fixer and developer solutions removed approximately 35–45% of the sensitive silver compounds from the films, depending on the object exposed. Solutions are reused twice, while used films are kept for the records for more than 5–10 years. After that, these wastes are sold to dealers. More than 90% of the dealers are not officially registered. Wastes are delivered to in-house factories to recover silver either by electrolysis, acid leaching (HNO₃) or combustion process. Most waste processors are not really concerned with handling the wastes. Highly toxic chemicals are generally spilled and discharged into the environment. The cleaner technology concept of silver leaching from radiographic film was investigated using weak acids such as acetic, oxalic and malonic acids. The tests were carried out under low temperature conditions.

An oxalic acid solution at 5% (w/v) provided the best leaching condition at 100 °C for 20 min. This achieved 100% silver leaching from the films. The obtained silver was in its metal form and ready for ingot production. Following this, a fact sheet on the waste management of radiographic film waste was created to reduce the risks of spills and leakages of hazardous silver into the environment.

© 2006 Elsevier Ltd. All rights reserved.

Keywords: Radiographic waste; Silver removal; Waste management

1. Introduction

To date, only a few articles have been reported on silver cycles [1,2]. They have discussed the presence of silver cycles in commodities of the industrial sectors rather than those from medical usage. The stock and flow of silver with regard to photographic film processing have been a concern in Thailand's effort to improve its waste management. Radiographic film is a polyester sheet coated on both sides by radioactive material, which is sensitive to light. It is used for industrial purposes and medical and dental services for some investigations. Approximately 2 billion radiographs are taken around

the world each year, including chest X-rays, mammograms, CT scans, etc. Traditionally, 94–98% of X-ray films are used in medical services [3]. Several waste streams are generated in hospital radiology departments, such as wastewater containing photographic chemicals, scrap films such as those purged from old files, or chemicals and silver removed from the film generated from poor photographs during processing, fixer solutions, and spoiled chemicals.

The wastes described previously are classified as "hazardous" due to their heavy metal contents. Like other countries, Thailand has recognized that the wastes from the production and formulation of photographic chemicals, and the processing of the material are defined as hazardous wastes by the Notification of the Ministry of Industry No. 6 [B.E. 2540 (1997)] issued pursuant to the Factory Act B.E. 2535 (1992) in Chapter 4, Section 20 in Ref. [4]. Even though it has been legally

0959-6526/\$ - see front matter © 2006 Elsevier Ltd. All rights reserved. doi:10.1016/j.jclepro.2006.06.010

^{*} Corresponding author.

E-mail address: gnurak@engr.tu.ac.th (N. Grisdanurak).

recognized as a hazardous waste for several years, silver concentration amounts from 2000 to 6000 mg/l have been found in fixer solutions [5]. In addition, more than 60% of the silver remains coated on used film after it is developed.

Existing technologies used for silver recovery are precipitation, ion exchange [6,7], and electrolysis [8]. Some hospitals provide an electrolysis silver recovery service for the on-site treatment of silver recovery. Many researchers have studied options for silver recovery from developing solutions, such as adsorption by activated carbons [9], chitin [10], etc. Up to our knowledge, there is little information on retrieving silver from used films. In Thailand, in particular, used film sheets have been managed carelessly. The main subjects covering radiographic silver waste in this paper are as follows:

- The present situation and a material flow analysis.
- · Silver recovery techniques.
- · Problems and a fact sheet on waste management.

2. Current situation of radiographic silver waste in Thailand

A Substance Flow Analysis (SFA) [2] was used to apply the concepts of industrial ecology to study how materials and energy flow into, throughout, and out of the system. This tracking system assists the evaluations of effects to the society, the economy, and the environment from extraction through production, consumption, and finally disposal. An equation of the conservation of mass is used to describe the silver substance flow in a limited boundary. This is to access the flow patterns of the silver waste cycle; to determine the amounts that are recovered and lost as waste; and to establish the amounts entering the reservoirs.

The conservation of mass is described by a basic substance flow equation:

$$\frac{\mathrm{d}F}{\mathrm{d}t} = \sum F_{\mathrm{input}} - \sum F_{\mathrm{output}} + \sum (S_t - S_{t-1}) \tag{1}$$

where F refers to material flow into and out of each hospital, inside the control volume, and S designates the stock of material retained or depleted from the reservoir over the time period. Film consumption data for each hospital are documented, however, the stocks are not documented; but there should not be much difference between each year since the stocks depend upon the limited space in each hospital.

Fig. 1 illustrates the silver waste life cycle as treated in this work, including the sub-processes, and the material flows linking them together. Silver used in medical services appears on the X-ray film. Currently, there are no film manufacturers in Thailand. Therefore, all the film used in Thailand is imported. Film is distributed to several kinds of hospitals, such as general, private and university hospitals, which are managed by different policies and stakeholders, including the Ministry of Industry, Education, and Public Health. After in-house image

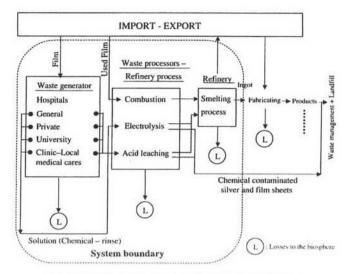


Fig. 1. Material flow of silver in radiographic hospital waste.

developing, silver becomes waste in both the solution and solid sheet material. Waste collection procedures differ from hospital to hospital, and between the waste types. The wastes are usually kept for a certain amount of years, and later sold to waste processors for further recovery. At this point, some used film is imported to Thailand to recover its silver. Several techniques, such as electrolysis, acid leaching, and combustion, are utilized to obtain silver complex. Silver in chemicals is collected by electrolysis, while silver in used film is retrieved by combustion and acid leaching. Silver complex is sold to a smelting process to obtain ingots, while contaminated silver chemicals cross a boundary for other treatments. Ingots are then ready to be sent out from the studied boundary for fabricating and manufacturing.

Under steady-state conditions within the one year period of consideration, the accumulation term approaches zero. The net accumulation of material would be equal to the summation of silver flowing into the system minus silver flowing out from a defined control volume. If one were to consider the years 2003–2004, Eq. (1) could be written as follows:

$$F \text{ silver} = \sum_{i \text{input } (2004)} - \sum_{i \text{foutput } (2004)} + \sum_{i \text{S} (2004)} S_{(2004)} - \sum_{i \text{S} (2003)} S_{(2003)}$$
(2)

2.1. Radiographic silver waste generator

The Customs Department (Thailand) has the duty of recording the amount of X-ray film that is imported; data on this are presented in Fig. 2. The annual imported amount increased and then remained stable at around 630–640 metric tons of film. According to the composition of silver content in the film as shown in Table 1, an approximate amount of 13 tons of silver enters the country's environmental cycle each year.

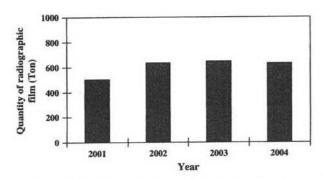


Fig. 2. Quantity of imported X-ray film from 2001-2004 [17].

According to our analysis, an average of 35–45% of the silver compounds are removed from the film after it is developed. Similar to our results, the Colorado Industrial Pretreatment Coordinators Association, 2004, disclosed that the remaining silver on film was about 60% [11]. Rinse water, usually discharged directly into hospital sewage, is found to have silver contamination, in the range of 12–100 mg/l, from the radiographic laboratory effluents of many hospitals. According to the concentration which is higher than 5 mg/l [12], it falls into the hazardous material category.

2.2. Radiographic silver waste processor

Used film and developing solution waste are collected and kept in the hospital area for their annual sale to waste dealers. Wastes in solution form (fixer and developer) are kept for a one-year period; while approximately 87% of hospitals keep used X-rays films for 5–10 years depending on the medical treatment criteria (for personal records) and the size of the collection area. Waste processors of photographic waste are categorized as type 106 — recycle, reuse, or recovery factories — based on the attached list of the Ministerial Regulation B.E. 2535 (1992), according to the Factory Act B.E. 2535 (1992) [4]. The factories that accept these wastes must be registered officially.

Table 2 shows the number of waste processors for silver recovery. Recently, only one silver recovery factory has been authorized by the Department of Industrial Works (DIW). This factory uses a burning method followed by a melting process.

Table 1 Silver content in materials

Control Comment III Interest Control		
Material	Silver content per unit material (g Ag/m ²)	Total silver (kg Ag)
Original X-ray film	4-6°	12 700°
Used X-ray film	3.0-3.5 ^a	6980-8320 ^d
Solution (fixer + developer)	around 2000-6000 mg/lh	

- " Material sample is digested with HNO3 and determined by AAS.
- h Solution sample is directly measured by AAS.
- ^c Total silver is estimated from 2% Ag on amount of imported X-ray film in 2004.
- ^d Total silver is estimated from 1.1 to 1.3% Ag on amount of imported X-ray film in 2004.

Table 2 Factory lists managing radiographic wastes

Process	Number of factory	Remark
Combustion/burning	1	In import-export zone
Electrolysis	17	
Acid leaching	8	

However, this plant is located in an import—export zone, meaning that only imported used film can be managed by it. The remaining 25 waste receivers are not registered. They all carry out in-house operations, either electrolysis or acid leaching processes. Factories with the electrolysis process recover silver from the developing agent waste, while factories with acid leaching operations recover silver from used film. However, they lack proper waste management systems. Fig. 3 shows the accessible areas of several in-house waste processors.

Leakage and spills were generally observed during the walk-in process. The rinse water chemical after silver recovery might contain traces of silver, therefore further treatment would be required. About 60% of the rinse water has been delivered to the treatment plant outside the boundary, while the rest is treated in-house by the neutralization or diluting process and then dumped into the environment. Our investigation found that approximately 5400–8700 mg of Ag/l was present in the processor effluents.

Silver complex from the recovery process is passed through a smelting step for remolding into an ingot. Afterwards, it is sold to fabricating manufacturers or exporters. Ingots refined from the import—export zone are entirely exported, while the ingots refined in in-house factories are sent to be fabricated for hand-made domestic products.

Losses to the biosphere can be calculated from the mass balance equation. Fig. 4 presents silver flows in medical care services and waste processors. Approximately 91.8% has been recovered from the radiographic film and solution process. From the analysis of processed film after acid leaching and electrolysis, some silver was found on the film sheet at an average of 0.08% wt. However, these chemical contaminated silver and film sheets were sent for waste management outside the system boundary. Even though the losses through the biosphere in our system boundary are not significantly compared to the system outside, the concentrations from each step seem to be over the standard concentration [12]. The emission from the in-house waste processors was found to be high and should be taken into consideration for the purpose of pollution prevention.

3. Silver recovery techniques

3.1. Present recovery techniques

Two typical ways to recover silver from used radiographic films are combustion technology and the acid leaching process. The used film is incinerated at high temperatures

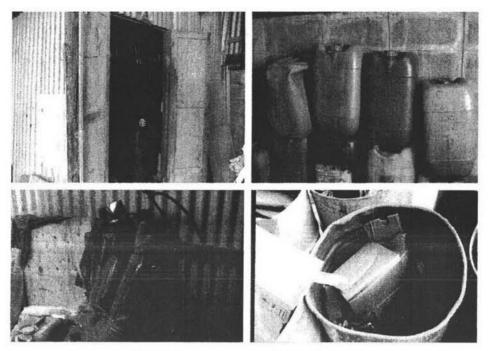


Fig. 3. In-house waste processor for silver recovery.

(>980 °C) in the combustion process and silver is recovered from ash by smelting and refining processes. Generally, the yields from the burning process are less than 95% [13,14]. In the acid leaching process, used film is submerged into a strong acid solution. Several chemicals, such as cyanide solution, ferric chloride, hydroxide, nitric acid, etc., have been used. A yield for the chemical leaching method is 99% [15]. It is operated as an open system at 70–80 °C, with a retention time of 30 min. Both processes have some environmental disadvantages.

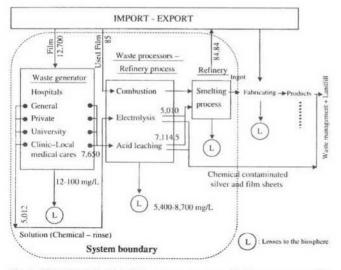


Fig. 4. Silver flows in a hospital waste generator and in-house recovery. All units measured in kg Ag (in the year 2003–2004).

3.2. Proposed cleaner technology approach for silver recovery

In order to reduce the usage of high temperature conditions and highly toxic solutions, we investigated how to use a weak acid as a leaching agent to remove silver. Acetic, oxalic, and malonic acids were used in this work. The investigation was performed through the leaching process on the effects of acid, temperature, and aging time. Used radiographic film chips of the same size were weighed and extracted under refluxing at 60–100 °C with 1–5% (w/v) acid solutions for 10–120 min. Afterwards, the films and sludge were separated out of the solution by filtration with 42 filter paper, Ashless (Whatman). The entire procedure is summarized as shown in Fig. 5.

3.2.1. Analytical method

Sludge was dried at 105 °C for 2 h, cooled down to room temperature in a desiccator and burnt in a furnace at 500 °C for 3.5 h. The ash was digested with microwave digestion by 10 ml of HNO₃ until the solution became clear. The silver concentration in the solution and sludge solution was measured by Atomic Absorption Spectroscopy (AAS). The silver concentration in the solution after the leaching process was measured by AAS. Silver sludge morphology was tested using Scanning Electron Microscopy (SEM) equipped with Energy Dispersive X-ray (EDX) Linkisis Series and X-ray Diffraction (XRD).

3.2.2. Results and discussion

A size of 1×1 cm² of radiographic film chip was used to avoid the attachment problem during the test run. In this study, increasing acid concentrations of 1-5% (w/v) were used. It

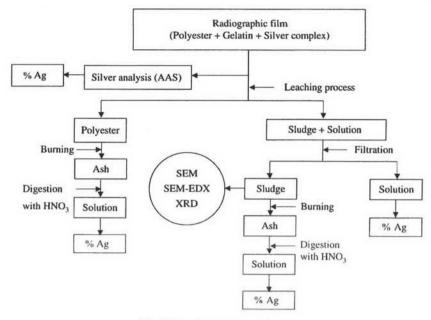


Fig. 5. Experiment schematic flow.

was observed that the removal efficiency could be enhanced by increasing the concentration of all the acids, as shown in Fig. 6(a). Oxalic acid provided the highest silver removal compared to malonic and acetic acids. Several duplication tests were done for each of the conditions; it was found that at 100 °C, 20 min and 5% oxalic acid concentration, there was nearly 100% removal. The effect of temperature on the extraction method for 10 min at 5% (w/v) acid solution is shown in Fig. 6(b). The efficiency of the extraction rate for each chemical increased strengthening of the temperature. However, some of the constraints should be addressed in order to recycle both film residue and solution. An example is that the temperature should not be higher or close to the melting point of the film. Therefore, as a further suggestion, 100 °C should be sufficient. As shown, there was not much extraction at temperatures less than 80 °C. It was then expected that the temperature was not high enough to dissolve the gelatin. However, at 90 °C or higher, the reaction worked well.

From the experimental results, the extreme conditions in which silver is leached out of film, as shown in Fig. 7, are

at temperatures of 90 and 100 °C. The effect of the heating time on the extraction process was investigated at 10-120 min. This factor is related to the temperature. Certainly, the higher temperature provided more removal efficiency. However, the condition at 100 °C, supported by results of the leaching process, can be carried out with a short retention time. The best removal efficiency conditions, which provide the highest potential of leaching (>99% efficiency removal), are at 5% (w/v) acid solution at 100 °C for about 20 min.

A characterization of residue of the leaching process at the optimum condition was performed. XRD, SEM and EDX were used. The XRD patterns of silver sludge from the leaching process were inspected. The major diffraction line of the residue sludge was presented at 38 (111), 44 (200), 64 (220), 77 (311), and 81 (222) degree 2θ , corresponding to the silver element fingerprint. Other kinds of silver complexes such as silver oxalate, and explosive materials could not be found. Fig. 8 shows the silver particles with a magnification of 3000, loaded onto the plastic film layer (a) and silver waste after leaching (b), respectively. Using an EDX, the particles present on the film were detected as

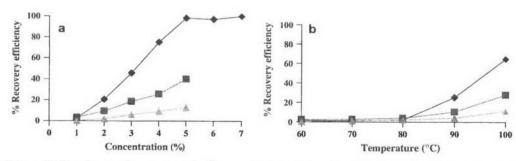


Fig. 6. Recovery efficiency (%) from the extraction process of 1 cm² processed film by (♠) oxalic acid, (■) malonic acid, and (▲) acetic acid. (a) Extraction process at 100 ° for about 20 min by refluxing with various acid concentrations. (b) Extraction process by refluxing with 5% (w/v) acid concentration for about 10 min at various temperatures.

P. Khunprasert et al. | Journal of Cleaner Production xx (2006) 1-9

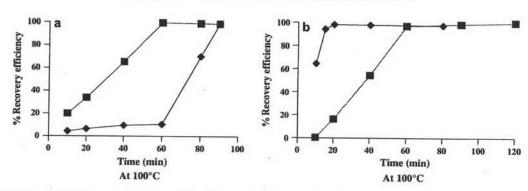


Fig. 7. Recovery efficiency (%) from the extraction process of 1 cm² processed film by refluxing, with various retention times (�) 1% (w/v) oxalic acid, (■) 5% (w/v) oxalic acid.

silver element. Silver particles first appeared in a clearly spherical form covered with gelatin. After treatment, acid dissolved the gelatin by breaking down the cross-link network of the gelatin. As a result the gelatin became flaky. However, the silver remained in the same shape covered with the flaky gelatin. These results confirmed that oxalic acid solution leaches silver out from the sheet without any reaction with the silver. In addition, this acid solution can be reused for this process several times.

3.3. Comparison of technologies

In this section, the lists of advantages and disadvantages for available technologies and our study are tabulated in Table 3. Temperature, toxicity, and other operating conditions were considered. According to the operating phase, it is obvious that all of them provide similar excellent silver removal; however, oxalic acid leaching has a higher leaching potential than the other two in terms of toxicity and the final product.

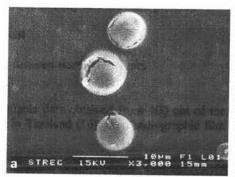
4. Waste management

4.1. Handling waste between waste generators and waste processors

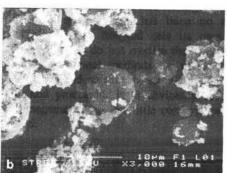
Based on the available data obtained from 100 out of the entire 424 hospitals in Thailand [16], used radiographic film

was sold at approximately 839 and 777 dollars per metric ton, for the years 2002 and 2003, respectively. For radiographic solutions, only used fixer was sold because it contained a high silver content, while used developer and water rinse were discharged into a hospital's wastewater treatment plant. The price spent on fixer in both years was approximately 680 dollars per cubic meter. A bidding process is generally used at each hospital to obtain the highest income.

The waste dealer of radiographic waste can be either a general or juristic person. Generally, hospitals have not required specific proof from waste dealers. The presence of an identification card might be enough. Fig. 9 discloses the percentages of required documents. A copy of an identification card, a copy of a business permit, and a copy of a resident registration are required at 45%, 9%, and 4%, respectively. Only 4% of the hospitals require all three documents. Table 4 summarized the present measurements and problems of each hospital with regard to the management of the radiographic waste. Because there has been no regulation announced officially, each hospital sets its own requirements. Some hospitals also do not require the waste dealers who disclose their operational methods for their waste processors. Most hospitals had no clue where all wastes were being delivered to and processed. It is obviously seen that hospital waste management staff has little concern for the waste after being sold.



Processed film (X 3,000)



Sludge (X 3,000)

Fig. 8. Silver particles on processed film and sludge from the extraction process by refluxing with 5% (w/v) oxalic acid at 100 °C.

Table 3 Lists of advantages and disadvantages for available silver removal techniques

Criteria	Combustion	Acid leaching	
		HNO ₃ , cyanide solution	This work: oxalic acid
Temperature (°C)	>980	~70-80	~90-100
Operating	High efficiency	High toxicity	Lower toxicity
condition Short retention time	Short retention time	High efficiency	High efficiency
	Short retention time	Short retention time	
	Able to be reused	Able to be reused	
Emission	High CO ₂ air emission	Low CO ₂ emission	Low CO ₂ emission
		Acid vapor emission	Acid vapor emission
Product	Sludge containing silver complex, and require a smelting process	Silver complex, such as silver nitrate and silver cyanate, requires electrolysis for a further process.	Sludge containing silver element with gelatin, and simple for a smelting process

Some waste dealers do not even have a commercial register authorized by the Ministry of Commerce, Thailand (MOC). For those who have been registered with the MOC and report to the office; however, the recorded information is not linked to the waste processor. Additionally, the management system after selling is not disclosed. These are some of the general misconceptions of radiographic waste management.

4.2. Problems of radiographic waste management

From the survey data about the radiographic waste management system, it was found that there are many drawbacks, they are as follows:

- There are too many waste dealers and waste processors to cope with the radiographic waste generated in Thailand; however, it is not clear where all the used radiographic waste goes and who manages them.
- (2) In addition, there are some unofficial waste processors of silver wastes, who are not concerned about whether the treatment is green or not. This presents significant potential hazards to humans, especially to the workers and the environment when untreated pollutants are emitted.
- (3) There is a lack of a manifest system of radiographic waste along the waste stream. It is complicated for the regulators to control and monitor the waste.
- (4) Radiographic waste is under the control of many authorities. They set different measurements to deal with the waste, such as different definitions of waste (hazardous and non-hazardous wastes), management procedures, etc. Moreover, the regulators of each section lack cooperation,

- which leads to a gap at the regulatory level, especially towards the waste dealers.
- (5) The only official waste processor imports all the raw materials, the used X-ray film, from another country to treat or recover silver. The transport of hazardous waste from foreign countries for treatment in Thailand seems to be a growing trend. That means all of the pollution that occurs after treatment remains in our nation.

5. Future prospects for radiographic waste management

In order to achieve the goal of used film waste management, silver hazardous wastes should be strictly monitored and tracked from the point of generation through to their ultimate disposal. In addition, better recovery techniques are required.

A management tool is to keep track of this hazardous waste through a shipping document called a hazardous waste manifest. Hazardous wastes may not be shipped off-site without a manifest and each manifest serves as a legal document. In general, this manifest does not cover the waste generated in hospitals. A single manifest typically consists of six copies, each designated for specific destination. This distribution of copies ensures that all parties involved (i.e. the hospital as a waste generator, waste dealer, waste processor, and the DIW) receive matching copies confirming the proper transport and disposition of hazardous waste.

The flow of material and the manifest is shown in Fig. 10. The hospital (the waste generator) retains copy 1 to be rechecked with manifest copy 5 after the wastes are delivered. The waste transporter retains copies 4–6 with the shipment. Upon arrival with the shipment at the treatment, storage, and

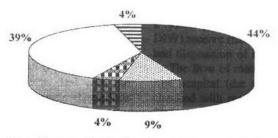


Fig. 9. Requirement percentage of each document type: ■ a copy of an identification card,
a copy of a business permit, a copy of a resident registration,
a copy of an identification card + a copy of a business permit + a copy of a resident registration.

Table 4

Present measurements and problem for radiographic waste management of waste generators

Type of hospitals	Regulator	Process	Problem
Ba Ad Mi	Ministry of Public Health	Transferring an administrative duty to finish in each hospital. The hospitals set their regulation for the management radiographic waste.	Lack of standard measurement to manage radiographic waste radiographic waste. Lack of direct authority to control.
	Bangkok Metropolitan	Hospitals set their regulations for the management	- The waste management policies enacted to
	Administration (BMA)	radiographic waste. They just report to the BMA, after finishing the process.	manage this waste are different.
	Ministry of University Affairs	Hospitals set their own rules but do not report to anyone.	2
	Other authorities	NA	
Private hospitals		The hospitals set their own regulation for the management radiographic waste.	

disposal facilities (TSDF), the transporter retains copy 4 and submits copy 5 to hospital waste generator and copy 6 to the DIW. The DIW would then be able to verify the amount of waste generated by the hospitals and waste processors by manifest copies 2 and 6. The DIW should provide the linkage with the Department of Public Health (DPH) to monitor the amount of imported film. Moreover, waste transporters and processors should be strictly registered under the law.

5.1. Conclusions

Used radiographic film waste management in Thailand involves many parties: the waste generators, waste dealers, waste processors and waste regulators. Recently, serious attention has not been placed on the management part of the equation. A material flow analysis was used to investigate the loss to the environment. It was found that a huge amount of silver has been distributed to the biosphere. The releases are mostly at the level of the waste processor units, which are in-house factories that are not officially registered. The use of weak acids to leach silver from the processed film was studied. A high efficiency of silver removal by oxalic acid could be obtained within moderate conditions. Oxalic acid dissolves gelatin without reacting with the silver to form its complex. This

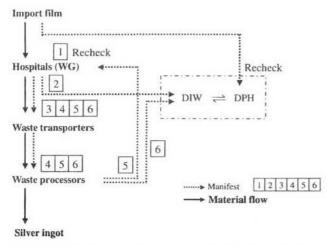


Fig. 10. The use of a manifest to track hospital radiographic waste flow.

provides for the simple separation of silver waste and allows for the benefit of reuse. Finally, prospective suggestions for the management of radiographic waste are proposed in the form of a simple manifest and proper monitoring system.

Acknowledgements

Special thanks are extended to CT Internship Program 2005 under the National Metal and Materials Technology Center, Thailand, for the financial support.

References

- Johnson J, Bertram M, Henderson K, Jirikowic J, Graedel TE. The contemporary Asian silver cycle: 1-year stocks and flows. J Mater Cycles Waste Manag 2005;7:93—103.
- [2] Lanzano T, Bertram M, De Palo M, Wagner C, Zyla K, Graedel TE. The contemporary European silver cycle. Resour Conserv Recycl 2006;46: 27–43.
- [3] Gambhir SS. Orientation: photographic emulsions, pharmacology M248: introduction to biological imaging [Online]. 2004.
- [4] Department of Industrial Works, Ministry of Industry, The Notification of the Ministry of Industry No. 6 B.E. 2540 (1997), 1997; and Department of Industrial Works, Ministry of Industrial, The Factory Act B.E. 2535 (1992), 1992.
- [5] Seelsaen N. Management of used X-ray fixer solution for Northeastern hospitals. Department of Environmental Engineering, Faculty of Engineering, Chulalongkorn University; 2001.
- [6] Atia AA. Adsorption of silver(I) and gold(III) on resins derived from bisthiourea and application to retrieval of silver ions from processed photo films. Hydrometallurgy 2005;80:98-106.
- [7] Cerjan-Stefanovic S, Briski F, Kastelan-Macan M. Separation of silver from waste water by ion-exchange resins and concentration by microbial cells. Fresenius J Anal Chem 1991;339:636—9.
- [8] US Patent 6,428,679 (2002).
- [9] Adani KG, Barley RW, Pascoe RD. Silver recovery from synthesis photographic and medical X-ray process effluents using activated carbon. Miner Eng 2005;(18):1269-76.
- [10] Songkroah C. Nakbanpote W, Thiravetyan P. Recovery of silver—thiosulphate complexes with chitin. Process Biochem 2004;39:1553—9.
- [11] Colorado Industrial Pretreatment Coordinators Association (CIPCA). Report on silver discharge to WWTP [Online]. Available from: http://cipca.org/forms/silver/SILVERRPT.DOC; 2004 [accessed 27.05.04].
- [12] United States Environmental Protection Agency (USEPA). The code of federal regulations (CFR) title 40: protection of environment [Online]. Available from: ; 2006 [accessed 10.01.06].

- [13] Ishikawa H, Ishimi K, Sugiura M, Sowa A, Fujiwara N. Kinetics and mechanism of enzymatic hydrolysis of gelatin layers of X-ray film and release of silver particles. J Ferment Bioeng 1993; 76:300-5.
- [14] Smarntarn V. 2001. Recovery of silver of used X-ray film by using alkaline protease from Aspergillus oryzae U1521. Faculty of Science, Naresuan University.
- [15] Gercia RM. The recovery of silver from photographic film: a study on the leaching reaction with cyanide solution for industrial use. Hydrometallurgy 1986;16:395-400.
- [16] National Statistical Office. Summary health statistic about hospitals and health care units [Online]. National Statistical Office; 2004.
- [17] The Customs Department. Impact/export statistics [Online]. The Customs Department; 2004.

BIOGRAPHY

Miss Pattamawan Khunprasert, was born on February 2, 1975 in Samutprakarn, Thailand. She finished her secondary school from Non-Formal Education Center, Bangkok. After that, she joined and graduated in major of Science-Chemistry, Faculty of Education at Srinakharinwirot University and graduated in 1994. Afterwards, she enrolled in Master's degree in Environmental Technology, Faculty of Energy and Material, King Mongkut's University of Technology Thonburi in 1998.

In 1999, she worked at Technical Provision and Planning Division, Department of Public Cleansing, Bangkok Metropolitan Administration. She moved to work at Technical Provision and Planning Division, City Planning Department, Bangkok Metropolitan Administration in 2000. In the same year till now, she has been Scientist at Hazardous substances Information Center, Hazardous Substances Control Bureau, Department of Industrial Works, Ministry of Industry.

In 2003, she started her Ph.D. degree in International Programs in Environmental Management, Chulalongkorn University and completed the program in May 2007.

During her Ph.D. study, she was accepted as an oral presentation in The First International Conference - Environmentally Sustainable Development, Pakistan, 2005 and Southeast Asia Regional Symposium on Chemical Engineering "New Trend in Chemical Engineering", Vietnam in the same year.

Before she accomplished her Ph.D. degree, her research work was published in the international journals; Clean Technology and Environmental Policy volume 9, no. 2, page 93-101, 2006, Journal of Cleaner Production, available online at http://www.dx.doi.org./10.1016/j.jclepro.2006.06.010...