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APPENDICES

Appendix A Adsorption of Cetyl Trimethyl Ammonium Bromide (CTAB) on Homoionic clinoptilolite

Weight of clinoptilolite = 0.2 g

Volume of CTAB solution = 20 ml

Molecular weight of CTAB = 364.45 g/mol

Table A1 Adsorption data of CTAB on clinoptilolite

No.	Initial concentration (μM)	Equilibrium concentration (μM)	Amount of CTAB adsorbed ($\mu\text{mol/g}$ of clinoptilolite)
1	54.8772	12.2817	4.2596
2	164.6316	19.8104	14.3786
3	274.3860	47.7440	22.5876
4	548.7721	130.7666	44.5893
5	823.1582	329.9884	53.8687
6	1097.5442	486.8819	63.8558
7	1371.9303	670.3396	69.7830
8	1646.3163	916.4488	72.9868
9	1920.7024	1050.9661	86.9736
10	2195.0885	1054.4813	123.6373
11	2469.4745	1152.7612	131.6713
12	2743.8606	1178.2141	159.4238
13	3429.8258	1392.9165	197.7110
14	4115.7909	1877.7285	223.5611
15	4801.7560	2567.7375	223.4019
16	5487.7212	3097.5402	239.0181

Table A2 Zeta potential data for CTAB adsorption on clinoptilolite

No.	Amount of CTAB adsorbed ($\mu\text{mol/g}$ of clinoptilolite)	Zeta potential (mV)
1	4.2596	-45.25
2	14.3786	-42.79
3	22.5876	-35.85
4	44.5893	-21.10
5	53.8687	19.14
6	63.8558	30.02
7	69.7830	37.28
8	72.9868	42.38
9	86.9736	52.58
10	123.6373	45.77
11	131.6713	39.79
12	159.4238	49.76
13	197.7110	44.76
14	223.5611	48.02
15	223.4019	45.29
16	239.0181	51.59

Appendix B Adsorption of Cetyl Trimethyl Ammonium Bromide (CTAB) on Bentonite clay

Weight of bentonite clay = 0.2 g

Volume of CTAB solution = 20 ml

Molecular weight of CTAB = 364.45 g/mol

Table B1 Adsorption data of CTAB on bentonite clay

No.	Initial concentration (μM)	Equilibrium concentration (μM)	Amount of CTAB adsorbed ($\mu\text{mol/g}$ of bentonite clay)
1	27.4386	2.7439	17.0000
2	68.5965	19.2070	67.0000
3	137.1930	41.1579	128.7500
4	274.3861	63.1088	336.0000
5	548.7721	68.5965	360.0000
6	685.9652	72.7123	448.5000
7	1097.5440	100.1509	489.7500
8	1371.9300	103.3612	532.5000
9	3429.8260	109.7544	573.5000
10	4527.3700	112.4983	655.0000
11	4938.9490	116.1476	677.6700
12	5350.5280	137.1930	692.5000
13	5556.3180	150.9123	734.0000
14	5762.1070	152.7507	851.0000
15	6173.6860	174.2351	981.0000
16	6585.2650	178.3509	1019.6700
17	6791.0550	185.2106	1098.0000
18	6859.6520	257.0174	1221.0000
19	6996.8450	373.1650	1359.6700
20	7408.4240	411.5791	1400.0000
21	8643.1610	768.281	1580.0000
22	9877.8980	960.3512	1650.0000
23	10289.4800	1371.93	1700.0000

Table B1 Adsorption data of CTAB on bentonite clay (con't)

No.	Initial concentration (μM)	Equilibrium concentration (μM)	Amount of CTAB adsorbed ($\mu\text{mol/g}$ of bentonite clay)
24	11112.6400	1838.3870	1786.0000
25	12347.3700	2195.0880	1790.0000
26	13719.3000	2892.0290	1792.0000
27	15434.2200	3347.5100	1792.5000
28	17149.1300	3621.8960	1793.0000

Table B2 Zeta potential data for CTAB adsorption on bentonite clay

No.	Amount of CTAB adsorbed ($\mu\text{mol/g}$ of bentonite)	Zeta potential (mV)
1	14.0000	-56.61
2	67.0000	-49.45
3	110.0000	-48.42
4	128.7500	-41.80
5	200.0000	-35.50
6	280.0000	-23.94
7	336.0000	-19.75
8	360.0000	-15.20
9	448.5000	8.18
10	489.7500	24.45
11	532.5000	31.12
12	573.5000	38.00
13	655.0000	44.00
14	677.6700	50.00

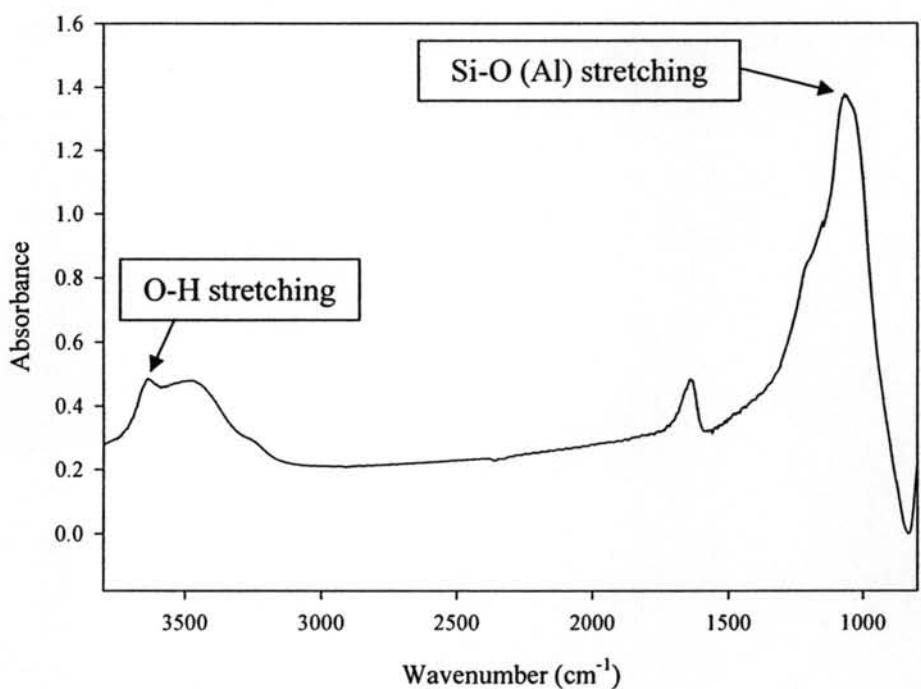


Figure A1 FTIR spectra of clinoptilolite

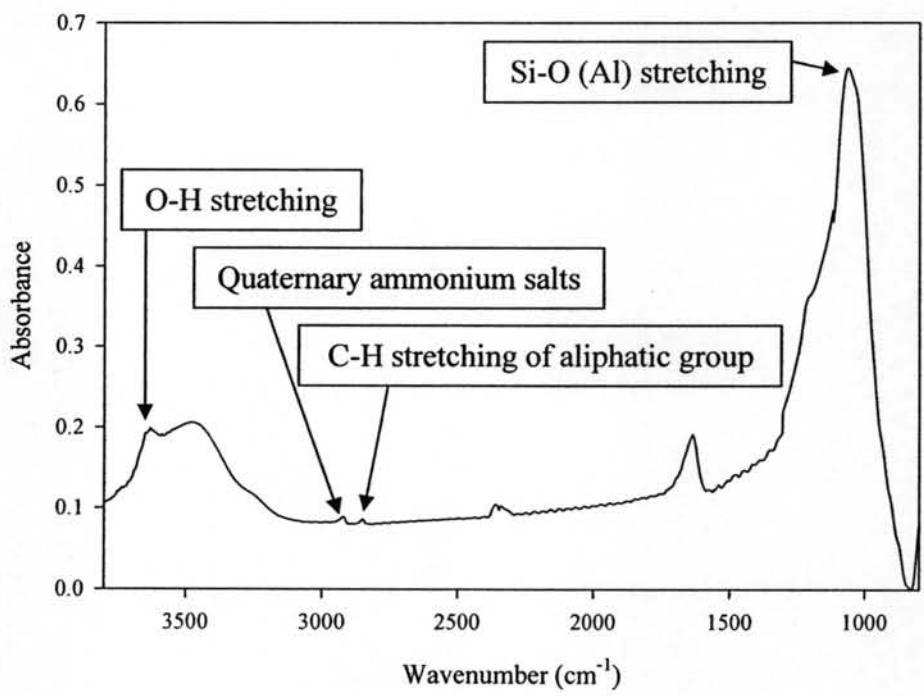


Figure A2 FTIR spectra of CTAB-modified clinoptilolite

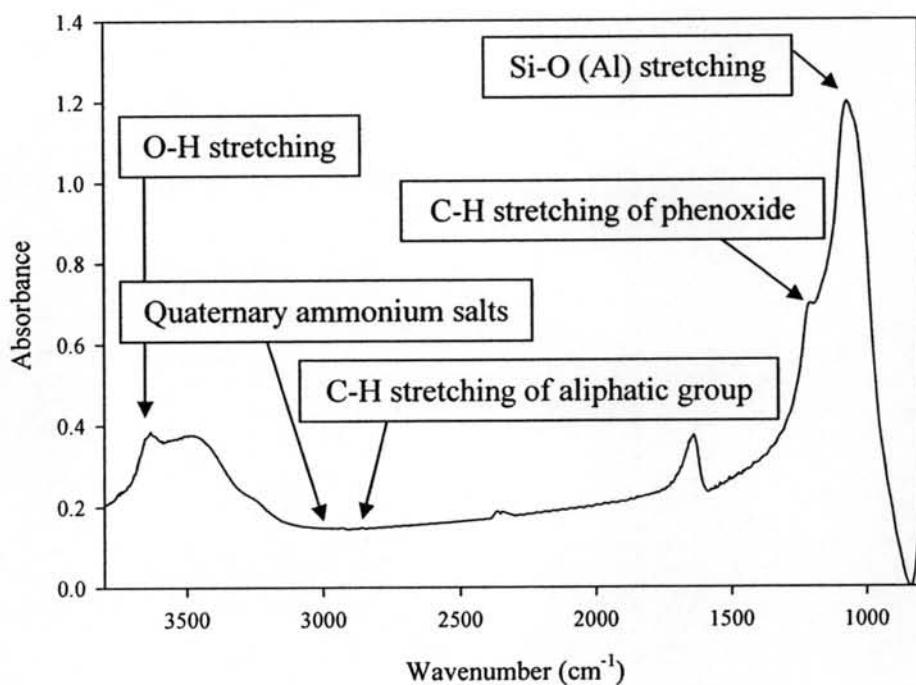


Figure A3 FTIR spectra of SMZ

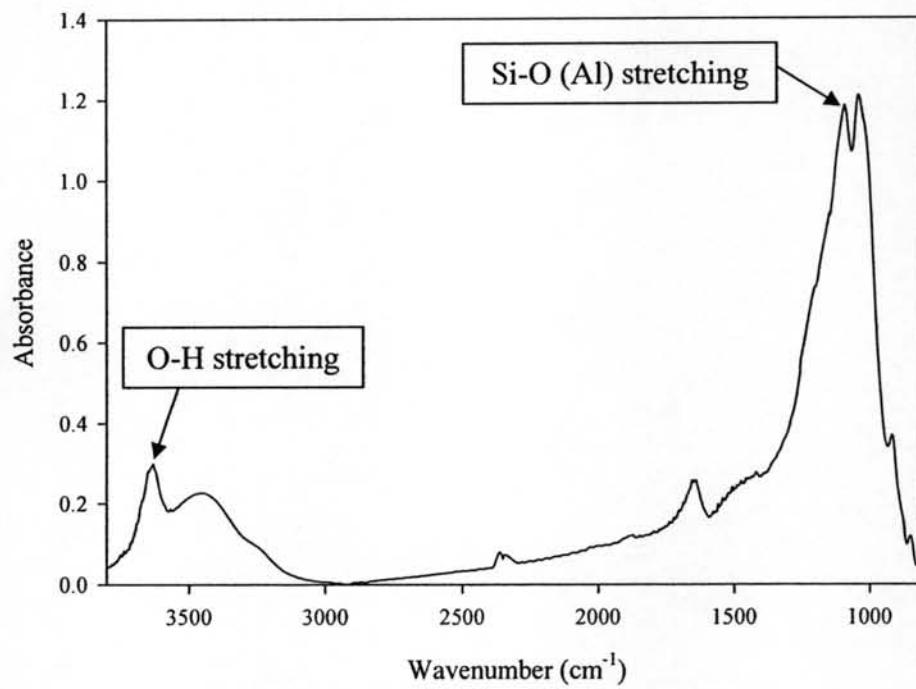


Figure A4 FTIR spectra of bentonite

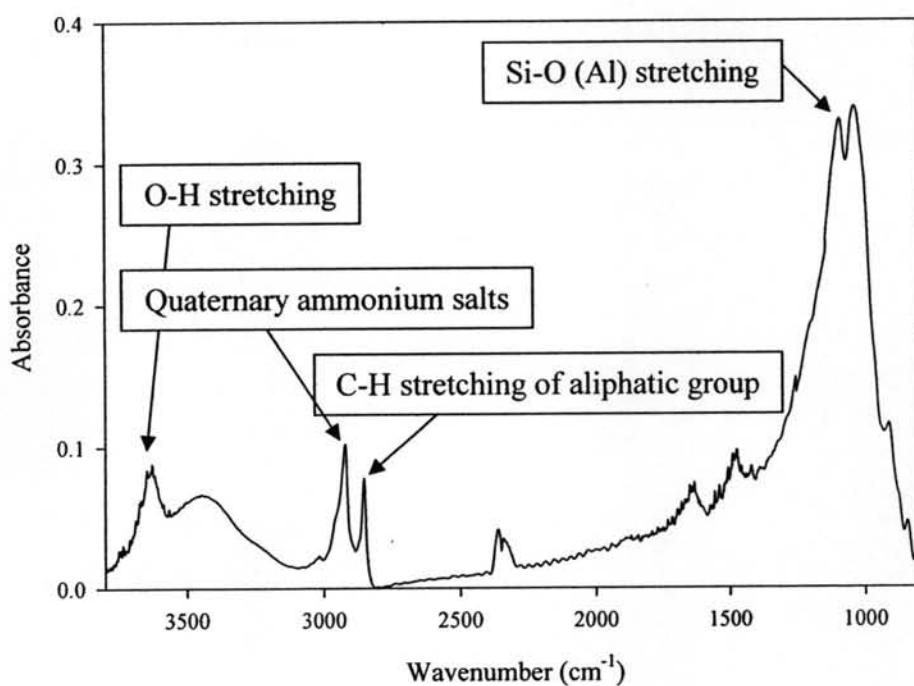


Figure A5 FTIR spectra of CTAB-modified bentonite

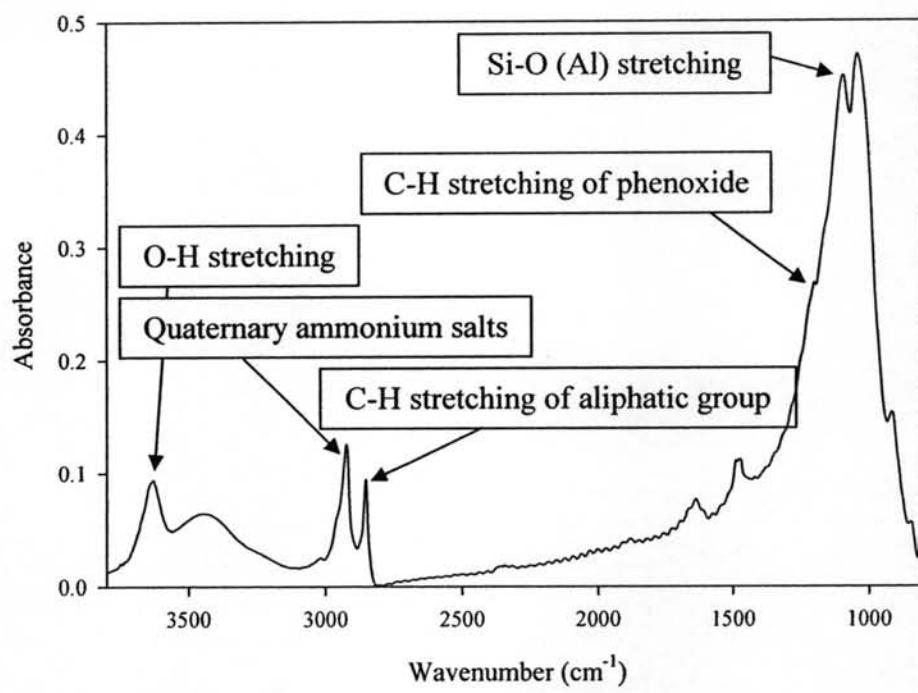


Figure A6 FTIR spectra of SMB

Appendix C Adsorption of Cadmium on Surfactant-Modified Adsorbents (SMADs) in Single-Solute System

Homoionic clinoptilolite and SMZ

Weight of adsorbent = 0.2 gram

Volume cadmium solution = 20 ml

Table C1 Adsorption data of cadmium on homoionic clinoptilolite and SMZ

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)
1	0.4448	0.0082	0.0036	0.0022	0.0201
2	0.8897	0.0164	0.0072	0.0145	0.0435
3	1.3345	0.0848	0.1163	0.0714	0.0824
4	1.7793	0.2069	0.1626	0.1659	0.1211
5	2.2242	0.3652	0.1924	0.3707	0.1484
6	2.6690	0.6028	0.2092	0.5144	0.1766
7	3.1139	0.9378	0.2249	0.7954	0.1963
8	3.5587	1.3710	0.2395	1.9395	0.1925
9	4.0036	1.7933	0.2484	2.7376	0.2067
10	4.4484	2.6563	0.2695	3.8022	0.2095

Bentonite clay and SMB

Weight of adsorbent = 0.2 gram

Volume of cadmium solution = 20 ml

Table C2 Adsorption data of cadmium on bentonite clay and SMB

No.	Initial concentration (mM)	Bentonite clay		SMB	
		Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)
1	0.8897	0.0882	0.01213	0.1800	0.0018
2	1.3345	0.1324	0.0182	0.2699	0.0028
3	1.7794	0.1765	0.0243	0.6581	0.0072
4	2.2242	0.2206	0.0303	0.8432	0.0093
5	2.6690	0.4840	0.0421	1.4526	0.0142
6	3.1139	0.8027	0.0672	1.8617	0.0161
7	3.5587	1.4842	0.1374	2.2946	0.0189
8	4.0036	2.5884	0.1928	2.7475	0.0204
9	4.4484	2.8896	0.2255	3.0257	0.0217
10	5.3381	3.4522	0.2456	4.5114	0.0239
11	6.2278	4.4036	0.2938	4.9738	0.0262
12	7.1174	5.4063	0.3278	5.8829	0.0282
13	8.0071	6.0435	0.3529	6.7593	0.0308
14	8.8968	6.5172	0.3662	7.8174	0.0337

**Appendix D Adsorption of Organic Compounds on Surfactant-Modified
Adsorbents (SMADs) in Single-Solute System**

Homoionic clinoptilolite and SMZ

Weight of adsorbent = 0.2 gram

Volume of organics solution = 20 ml

Table D1 Adsorption data of benzene on homoionic clinoptilolite and SMZ

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)
1	0.4796	0.3342	0.0065	0.2759	0.0155
2	0.9592	0.7044	0.0043	0.6295	0.0262
3	1.9185	1.7057	0.0058	1.1350	0.0740
4	3.8370	3.5901	0.0116	2.3671	0.1322
5	5.7554	5.3984	0.0174	3.8160	0.2002
6	7.6740	6.3596	0.0345	5.1803	0.2314
7	9.5924	9.1211	0.0453	6.1971	0.3147
8	11.9906	10.0307	0.0540	8.4412	0.3614
9	14.3887	11.9418	0.0648	10.6256	0.4593
10	19.1849	18.0175	0.0722	14.2707	0.5906

Table D2 Adsorption data of benzoic acid on homoionic clinoptilolite and SMZ

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of benzoic acid adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of benzoic acid adsorbed (mmol/g)
1	0.0893	0.1118	9E-05	0.0781	0.0011
2	0.1786	0.1911	0.0002	0.1602	0.0018
3	0.3571	0.3544	0.0003	0.3052	0.0052
4	0.5357	0.5274	0.0008	0.4745	0.0061
5	0.7143	0.7200	0.0016	0.6259	0.0088
6	0.8929	0.8670	0.00091	0.7969	0.0096
7	1.0714	1.0346	0.0014	0.9549	0.0116
8	1.2500	1.2261	0.0024	1.1331	0.0117
9	1.4286	1.3774	0.0026	1.2823	0.0146
10	1.7857	1.7369	0.0049	1.6064	0.0179

Table D3 Adsorption data of phenol on homoionic clinoptilolite and SMZ

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of phenol adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of phenol adsorbed (mmol/g)
1	2.1224	2.0822	0.0035	1.9834	0.0137
2	4.2449	4.1644	0.0070	3.8882	0.0354
3	8.4899	8.3884	0.0081	7.9258	0.0558
4	12.7349	12.4823	0.0061	11.9065	0.0819
5	16.9799	17.0331	0.0087	15.8987	0.1069
6	21.2248	21.4919	0.0109	19.7296	0.1480
7	26.5310	26.3278	0.0140	24.8533	0.1658
8	31.8372	31.6035	0.0158	29.5009	0.2313
9	37.1434	37.3050	0.0186	34.5764	0.2540
10	42.4496	42.1558	0.0193	39.4229	0.2995

Table D4 Adsorption data of chlorobenzene on homoionic clinoptilolite and SMZ

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of chlorobenzene sorbed (mmol/g)	Equilibrium concentration (mM)	Amount of chlorobenzene sorbed (mmol/g)
1	0.0778	0.0751	0.0023	0.0690	0.0034
2	0.1555	0.1626	0.0006	0.1374	0.0081
3	0.3887	0.3743	0.0013	0.3360	0.0243
4	0.6997	0.6640	0.0031	0.628	0.0403
5	1.0885	1.0573	0.0024	0.9679	0.0654
6	1.4772	1.4225	0.0046	1.2896	0.0901
7	1.7882	1.7628	0.0074	1.4491	0.1075
8	2.2547	2.2677	0.0093	1.9684	0.1356
9	2.7212	2.6204	0.0084	2.3756	0.1636
10	3.1099	2.9617	0.0096	2.7150	0.1876

Table D5 Adsorption data of naphthalene on homoionic clinoptilolite and SMZ

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of naphthalene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of naphthalene adsorbed (mmol/g)
1	0.0012	0.0012	9.01364E-06	0.0011	2.9410E-05
2	0.0050	0.0048	2.82524E-05	0.0044	0.0001
3	0.0087	0.0086	3.42276E-05	0.0077	0.0002
4	0.0124	0.0123	3.55214E-05	0.0110	0.0003
5	0.0150	0.0148	3.0142E-05	0.0132	0.0004
6	0.0175	0.0173	3.2565E-05	0.0154	0.0004
7	0.0187	0.0185	4.1579E-05	0.0165	0.0004
8	0.0212	0.0211	2.8398E-05	0.0187	0.0005
9	0.0225	0.0222	5.3016E-05	0.0198	0.0005

Table D5 Adsorption data of naphthalene on homoionic clinoptilolite and SMZ
(con't)

No.	Initial concentration (mM)	Homoionic clinoptilolite		SMZ	
		Equilibrium concentration (mM)	Amount of naphthalene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of naphthalene adsorbed (mmol/g)
10	0.0250	0.0248	3.2032E-05	0.0222	0.00054

Bentonite clay and SMB

Weight of adsorbent = 0.2 gram

Volume of organics solution = 20 ml

Table D6 Adsorption data of benzene on bentonite clay and SMB

No.	Initial concentration (mM)	Bentonite clay		SMB	
		Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)
1	37.4100	0.1164	0.0198	0.1189	0.0747
2	74.8200	0.2329	0.0040	0.2258	0.1430
3	149.6400	0.4658	0.0079	0.4503	0.2873
4	299.2800	3.7262	0.0209	0.9075	0.5847
5	448.9200	5.5893	0.0313	1.4321	0.8629
6	598.5600	7.4525	0.0417	1.8976	1.1529
7	748.2000	9.3156	0.0521	2.3720	1.4411
8	935.2500	11.6445	0.0652	2.9650	1.8014
9	1122.3000	13.9734	0.0782	3.5271	2.1678
10	1496.4000	18.6312	0.1043	4.4818	2.9470

Table D7 Adsorption data of benzoic acid on bentonite clay and SMB

No.	Initial concentration (mM)	Bentonite clay		SMB	
		Equilibrium concentration (mM)	Amount of benzoic acid adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of benzoic acid adsorbed (mmol/g)
1	0.0818	0.0739	0.0016	0.0437	0.0076
2	0.1637	0.1606	0.0006	0.1003	0.0127
3	0.3273	0.3109	0.0033	0.1810	0.0293
4	0.4910	0.4626	0.0057	0.2729	0.0436
5	0.6547	0.6374	0.0035	0.4299	0.0549
6	0.8183	0.7966	0.0043	0.5281	0.0581
7	0.9820	0.9252	0.0520	0.6590	0.0646
8	1.1457	1.1359	0.0020	0.7975	0.0696
9	1.3093	1.2746	0.0069	0.9055	0.0807
10	1.6367	1.5934	0.0087	1.1120	0.0849

Table D8 Adsorption data of phenol on bentonite clay and SMB

No.	Initial concentration (mM)	Bentonite clay		SMB	
		Equilibrium concentration (mM)	Amount of phenol adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of phenol adsorbed (mmol/g)
1	2.1251	2.1248	7.9701E-05	1.7251	0.0800
2	4.2503	4.2499	8.2356E-05	3.4039	0.1693
3	8.5007	8.5002	8.7663E-05	7.0956	0.2810
4	12.7510	12.7506	9.2974E-05	10.9442	0.3614
5	17.0019	17.0009	9.8284E-05	14.8419	0.4319
6	21.2517	21.2512	0.0001	19.0313	0.4441
7	26.5647	26.5641	0.0001	23.9852	0.5159
8	31.8776	31.8770	0.0001	29.0490	0.5657
9	37.1905	37.1899	0.0001	33.9850	0.6411
10	42.5034	42.5028	0.0001	39.1195	0.6768

Table D9 Adsorption data of chlorobenzene on bentonite clay and SMB

No.	Initial concentration (mM)	Bentonite clay		SMB	
		Equilibrium concentration (mM)	Amount of chlorobenzene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of chlorobenzene adsorbed (mmol/g)
1	8.7500	0.0772	0.0083	0.0087	0.0104
2	17.5000	0.1544	0.0032	0.0241	0.0221
3	43.7500	0.3861	0.0080	0.0487	0.0489
4	78.7500	0.6950	0.0010	0.0761	0.1001
5	122.5000	1.0811	0.0016	0.1034	0.1394
6	166.2500	1.4672	0.0021	0.1604	0.2162
7	201.2500	1.7761	0.0025	0.1767	0.2371
8	253.7500	2.2394	0.0032	0.2280	0.3050
9	306.2500	2.7027	0.0039	0.3041	0.4642
10	350.0000	3.0888	0.0044	0.3250	0.5459

Table D10 Adsorption data of naphthalene on bentonite clay and SMB

No.	Initial concentration (mM)	Bentonite clay		SMB	
		Equilibrium concentration (mM)	Amount of naphthalene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of naphthalene adsorbed (mmol/g)
1	0.1600	0.0011	1.3456E-05	4.9168E-05	0.0001
2	0.6400	0.0047	3.0419E-05	0.0003	0.0005
3	1.1200	0.0084	3.1778E-05	0.0006	0.0008
4	1.6000	0.0120	5.2642E-05	0.0009	0.0012
5	1.9200	0.0142	7.5653E-05	0.0011	0.0014
6	2.2400	0.0170	4.4050E-05	0.0017	0.0016
7	2.4000	0.0182	4.9704E-05	0.0020	0.0017
8	2.7200	0.0208	4.5409E-05	0.0025	0.0019
9	2.8800	0.0219	5.8865E-05	0.0029	0.0020
10	3.2000	0.0244	5.8470E-05	0.0034	0.0022

Appendix E Adsorption Experiments of Cadmium on Surfactant-Modified Adsorbents (SMADs) in Single-solute and Mixed-Solute System

SMZ

Weight of adsorbent = 0.2 gram

Volume of cadmium solution = 20 ml

Table E1 Adsorption data of cadmium on SMZ in single-solute system and mixed-solute system

No.	Initial concentration (mM)	SMZ in single-solute system		SMZ in mixed-solute system	
		Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)
1	0.4448	3.8022	0.2095	2.5933	0.1771
2	0.8897	2.7376	0.2067	1.6137	0.1718
3	1.3345	1.9395	0.1926	1.2149	0.1678
4	1.7794	0.7954	0.1963	0.9231	0.1586
5	2.2242	0.5144	0.1766	0.6221	0.1552
6	2.6690	0.3707	0.1484	0.3946	0.1336
7	3.1139	0.1659	0.1211	0.2467	0.1140
8	3.5587	0.0714	0.0824	0.1567	0.1078
9	4.0036	0.0145	0.0435	0.0613	0.0958
10	4.4484	0.0022	0.0201	0.0146	0.0352

SMB

Weight of adsorbent = 0.2 gram

Volume of cadmium solution = 20 ml

Table E2 Adsorption data of cadmium on SMB in single-solute system and mixed-solute system

No.	Initial concentration (mM)	SMB in single-solute system		SMB in mixed-solute system	
		Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of cadmium adsorbed (mmol/g)
1	0.4448	3.8022	0.2095	2.5933	0.1771
2	0.8897	2.7376	0.2067	1.6137	0.1718
3	1.3345	1.9395	0.1926	1.2149	0.1678
4	1.7794	0.7954	0.1963	0.9231	0.1586
5	2.2242	0.5144	0.1766	0.6221	0.1552
6	2.6690	0.3707	0.1484	0.3946	0.1336
7	3.1139	0.1659	0.1211	0.2467	0.1140
8	3.5587	0.0714	0.0824	0.1567	0.1078
9	4.0036	0.0145	0.0435	0.0613	0.0958
10	4.4484	0.0022	0.0201	0.0146	0.0352

**Appendix F Adsorption of Organic Compounds on Surfactant-Modified
Adsorbent (SMADs) in Single-Solute and Mixed-Solute System**

SMZ

Weight of adsorbent = 0.2 gram

Volume of organics solution = 20 ml

Table F1 Adsorption data of benzene on SMZ in single-solute system and mixed-solute system

No.	Initial concentration (mM)	SMZ in single-solute system		SMZ in mixed-solute system	
		Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)
1	19.1849	14.2708	0.5906	13.4085	0.5823
2	14.3887	10.6256	0.4593	12.0643	0.5077
3	11.9906	8.4412	0.3614	10.6686	0.4195
4	9.5924	6.1971	0.3147	9.0357	0.3507
5	7.6740	5.1803	0.2314	7.5619	0.2774
6	5.7555	3.8160	0.2002	5.9654	0.2096
7	3.8370	2.3672	0.1322	3.7959	0.1315
8	1.9185	1.1350	0.0741	2.7342	0.1229
9	0.9592	0.6295	0.0262	1.4674	0.0336
10	0.4796	0.2759	0.0155	0.75841	0.0197

SMB

Weight of adsorbent = 0.2 gram

Volume of organics solution = 20 ml

Table F2 Adsorption data of benzene on SMB in single-solute system and mixed-solute system

No.	Initial concentration (mM)	SMB in single-solute system		SMB in mixed-solute system	
		Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)	Equilibrium concentration (mM)	Amount of benzene adsorbed (mmol/g)
1	19.1849	4.4818	2.9470	4.2448	2.6369
2	14.3887	3.5271	2.1678	3.6841	2.2942
3	11.9906	2.96503	1.8014	3.2177	1.7991
4	9.5924	2.3720	1.4411	2.7259	1.5466
5	7.6740	1.8976	1.1529	2.1452	1.1657
6	5.7555	1.4321	0.8629	1.6333	0.9272
7	3.8370	0.9075	0.5847	1.2871	0.8428
8	1.9185	0.4503	0.2873	0.8581	0.68051
9	0.9592	0.2258	0.1430	0.3341	0.1163
10	0.4796	0.1189	0.0747	0.1440	0.0634

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