

การเติมหมู่ฟังก์ชันบนกราฟีนที่ได้จากการตกเคลือบด้วยไอเคมี



บทคัดย่อและแฟ้มข้อมูลฉบับเต็มของวิทยานิพนธ์ตั้งแต่ปีการศึกษา 2554 ที่ให้บริการในคลังปัญญาจุฬาฯ (CUIR)
เป็นแฟ้มข้อมูลของนิสิตเจ้าของวิทยานิพนธ์ ที่ส่งผ่านทางบัณฑิตวิทยาลัย

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วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต
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คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย
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CHULALONGKORN UNIVERSITY

FUNCTIONALIZATION OF CHEMICAL VAPOR DEPOSITION-GROWN GRAPHENE

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A Thesis Submitted in Partial Fulfillment of the Requirements
for the Degree of Master of Science Program in Chemistry

Department of Chemistry

Faculty of Science

Chulalongkorn University

Academic Year 2017

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จุฬาลงกรณ์มหาวิทยาลัย
CHULALONGKORN UNIVERSITY

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ภพนิตินุ ฉัตรมณีรุ่งเจริญ : การเติมหมู่ฟังก์ชันบนกราฟีนที่ได้จากการตกเคลือบด้วยไอเคมี (FUNCTIONALIZATION OF CHEMICAL VAPOR DEPOSITION-GROWN GRAPHENE) อ.ที่ปรึกษา
วิทยานิพนธ์หลัก: ดร.สกุลสุข อุ๋นอรุณทัย, หน้า.

กราฟีนได้กลายเป็นวัสดุที่น่าอัศจรรย์ตั้งแต่ที่มีการค้นพบในปี ค.ศ. 2004 เนื่องจากมีสมบัติที่น่าสนใจ เช่น มีค่าการนำความร้อนและการนำไฟฟ้าที่เป็นเลิศ มีความยืดหยุ่น และมีความแข็งแรงสูง นักวิจัยทั่วโลกได้หันมาสนใจศึกษาอัญรูปคาร์บอนนี้กันมากขึ้น ในการสังเคราะห์กราฟีนให้มีคุณภาพสูง การตกเคลือบด้วยไอเคมีเป็นวิธีที่มีความเป็นไปได้ที่จะทำการผลิตในเชิงอุตสาหกรรม โดยทั่วไป แก๊สมีเทนความบริสุทธิ์สูงซึ่งนิยมใช้เป็นสารตั้งต้นในการสังเคราะห์กราฟีนมีราคาสูง ผู้วิจัยจึงมีแนวคิดที่จะใช้แก๊สอะเซทิลีนแทนแก๊สมีเทน นอกจากนั้น การตกเคลือบด้วยไอเคมีที่ความดันบรรยากาศน่าจะสามารถลดต้นทุนการผลิตได้อย่างมาก อย่างไรก็ตาม การสังเคราะห์จะต้องทำการหาสภาวะที่เหมาะสมใหม่สำหรับแหล่งคาร์บอนชนิดใหม่โดยอาศัยการออกแบบการทดลอง ขั้นแรก ทำการตรวจสอบปัจจัยที่มีผลต่อการสังเคราะห์กราฟีนด้วยเทคนิคการตกเคลือบด้วยไอเคมีด้วยการออกแบบการทดลองแบบแฟกต์เค็ด-เบอร์มาน โดยพิจารณาจากพารามิเตอร์ 6 ชนิดที่ได้จากสเปกตรัมรามานของกราฟีน พบว่าอุณหภูมิที่ทำการตกเคลือบและผลรวมอัตราการไหลของแก๊สอาร์กอนและแก๊สไฮโดรเจนมีนัยสำคัญในการสังเคราะห์กราฟีนด้วยระดับความเชื่อมั่น 70 ถึง 80 เปอร์เซ็นต์ จากนั้นทำการศึกษาปัจจัยทั้งสองประกอบกับอัตราการไหลของแก๊สอะเซทิลีนโดยการสร้างแบบจำลองพื้นผิวตอบสนองด้วยการออกแบบการทดลองแบบบ็อกซ์-เบ็นเคน หลังจากการวิเคราะห์การถดถอยพหุคูณจึงได้ผลการแสดงความสัมพันธ์ระหว่างพารามิเตอร์ทั้งหมด 4 ชนิดที่ได้จากสเปกตรัมรามานของกราฟีนกับปัจจัยทั้งสามชนิด ซึ่งมีสัมประสิทธิ์การกำหนดระหว่าง 0.987 – 0.942 และทำให้ได้สภาวะการสังเคราะห์ที่เหมาะสมที่สุด ที่อุณหภูมิ 1050 องศาเซลเซียส อัตราการไหลของแก๊สอาร์กอน แก๊สไฮโดรเจน และแก๊สอะเซทิลีนเป็น 900, 100 และ 0.4 ลูกบาศก์เซนติเมตรต่อนาที ตามลำดับ จนถึงตอนนี้ กราฟีนมักจะถูกนำไปใช้เป็นตัวนำไฟฟ้าโปร่งใส ในขณะที่การนำกราฟีนไปใช้เป็นสารกึ่งตัวนำถูกจำกัดเนื่องจากกราฟีนไม่มีช่องว่างพลังงาน การตัดแปรงกราฟีนด้วยพันธะโคเวเลนต์เป็นวิธีหนึ่งซึ่งสามารถสร้างช่องว่างพลังงานให้กราฟีนได้ ถึงแม้ว่าการเติมหมู่ฟังก์ชันบนกราฟีนด้วยเกลือไดออกไซด์แอมโมเนียม โดยการแช่กราฟีนในสารละลายเกลือไดออกไซด์แอมโมเนียมที่มีสารลดแรงตึงผิวจะทำได้ง่ายสำหรับกราฟีนชั้นเดียว จำนวนโมเลกุลที่เติมลงไปมีจำนวนไม่มากสำหรับกราฟีนที่หนา มากกว่าหนึ่งชั้น ในที่นี้ ผู้วิจัยทำการศึกษาผลของวัสดุรองรับและแสง พบว่ากราฟีนที่ทำปฏิกิริยาเติมหมู่ฟังก์ชันบนวัสดุรองรับที่มีขั้วอย่างซิลิกอนไดออกไซด์บนซิลิกอนถูกบดปนเปื้อนด้วยเกลือไดออกไซด์แอมโมเนียมที่เหลือจากปฏิกิริยา ในขณะที่การเติมหมู่ฟังก์ชันให้กับกราฟีนบนวัสดุรองรับที่ไม่มีขั้วอย่างทองแดงได้ผลิตภัณฑ์ที่สะอาดกว่า ผู้วิจัยยังพบว่าแสงในช่วงยูวี-เอจากหลอดแบล็กไลท์หรือดวงอาทิตย์มีบทบาทสำคัญในการเติมหมู่ฟังก์ชัน โดยสันนิษฐานว่าแสงในช่วงดังกล่าวทำให้เกิดอิเล็กตรอนและฟิโนลแคตไอออนจากกราฟีนและเกลือไดออกไซด์แอมโมเนียม ตามลำดับ กระนั้น ปฏิกิริยาการเติมหมู่ฟังก์ชันสามารถเกิดได้ในที่มืดอย่างช้า ๆ โดยสรุป แสงสามารถเร่งปฏิกิริยาการเติมหมู่ไดออกไซด์แอมโมเนียมบนกราฟีนได้

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5772094923 : MAJOR CHEMISTRY

KEYWORDS: GRAPHENE / CHEMICAL VAPOR DEPOSITION / FUNCTIONALIZATION / DIAZONIUM / ACETYLENE / ATMOSPHERIC PRESSURE

BHOBNIBHIT CHATMANEERUNGCHAROEN: FUNCTIONALIZATION OF CHEMICAL VAPOR DEPOSITION-GROWN GRAPHENE. ADVISOR: SAKULSUK UNARUNOTAI, Ph.D., pp.

Graphene has been a wonder material since its discovery in 2004. Due to its interesting properties such as excellent thermal and electrical conductivities, flexibility and high strength, many researchers around the globe have turned their attention to this carbon allotrope. To synthesize graphene with high quality, chemical vapor deposition (CVD) is an industrially feasible method. Generally, methane is the most widely used carbon precursor. However, high purity grade is expensive in Thailand. Instead, acetylene should be a possible replacement. Moreover, operating CVD at atmospheric pressure would also dramatically reduce the cost. Still, growth conditions need to be optimized for new carbon sources using experimental design. First, seven factors in CVD process were screened by Plackett–Burman design and six Raman parameters of graphene were dependent variables. The results suggested that growth temperature and Ar-H₂ total flow rates were significant at confidence intervals of 70-80%. The two significant factors together with acetylene flow rate were then subject to constructing response surface model with Box–Behnken design. Four relationships were developed by multiple linear regression (MLR) with $R^2 = 0.987 - 0.942$ and the best quality graphene could be obtained at growth temperature of 1050 °C with Ar/H₂/C₂H₂ flow rates of 900/100/0.4 sccm. Until now, graphene has been mostly used as a transparent conductor. Its application as a semiconductor has been limited due to its zero bandgap. Covalent modification was suggested as a possible way to open the bandgap in graphene. Although functionalization of graphene with diazonium salts using surfactant in aqueous solution was trivial for single-layer graphene, degree of functionalization was relatively low in case of multilayer graphene. Herein, substrate and photoinduced effects were investigated. While functionalized graphene on polar SiO₂/Si substrate was contaminated with unreacted diazonium salts, functionalization of graphene on Cu growth substrate afforded the cleaner product. We also found that UVA light from a blacklight lamp or sunlight played important roles in functionalization presumably by generation of hot electrons and phenyl cations from graphene and diazonium salts, respectively. Nonetheless, the reaction could occur slowly without light. In conclusion, diazonium grafting could be enhanced under irradiation of light.

Department: Chemistry

Student's Signature

Field of Study: Chemistry

Advisor's Signature

Academic Year: 2017

ACKNOWLEDGEMENTS

First of all, I would like to express my deepest gratitude to my supervisor, Dr. Sakulsuk Unarunotai, who always believes in me. It is my pleasure to be his first graduate student. Throughout four years of master's degree, I am really thankful to his kindness and tireless heart. He is a patient and open-minded teacher. His positive thought and tranquility got me through cold and heat. Without his guidance and support, my master thesis could not be completed. Furthermore, I am grateful to my thesis committee members for their helpful advice during proposal examination.

As a small research group whose works are related to multidisciplinary fields, nothing could be succeeded without collaborations. Therefore, I would like to take the opportunity to thank Professor Dr. Mongkol Sukwattanasinitt for his aspect and support on upgrading our CVD system, Assistant Professor Dr. Kanet Wongravee for his invaluable suggestion on chemometrics and constant encouragement, Professor Dr. Sanong Ekgasit for accessing Raman spectrometer, Dr. Monpichar Srisa-ard for accessing spin-coater, Assistant Professor Dr. Soamwadee Chaianansucharit and Dr. Prasert Sinsermuksamkul for sharing their knowledge and experience about fittings.

For financial support, I would like to thank Ratchadaphiseksomphot Endowment Fund, Chulalongkorn University under Grant No. CU56-558-AM and Teaching Assistant Scholarship from Department of Chemistry, Faculty of Science, Chulalongkorn University.

Indispensably, I would like to offer my heartfelt thanks to my family, my friends, my colleagues, members of Material Advancement through Proficient Synthesis research group (MAPS), members of Sensor Research Unit (SRU), department staffs and everyone who comes to my life.

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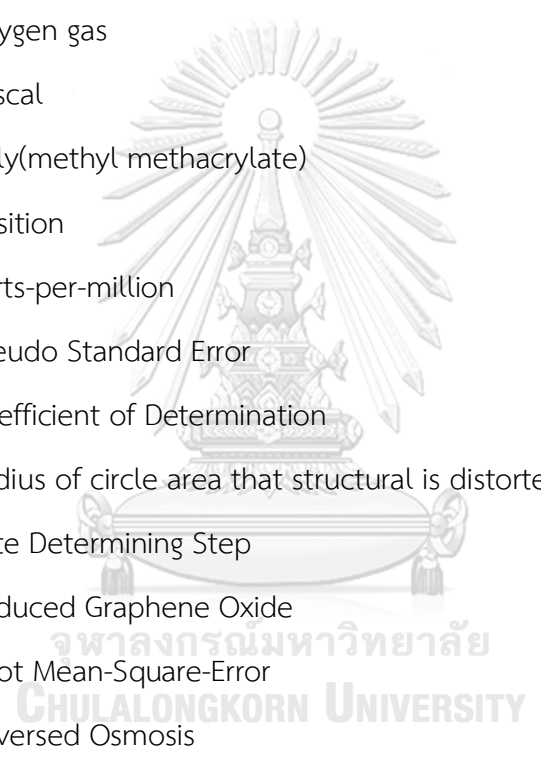
LIST OF ABBREVIATIONS AND SYMBOLS

%	: Percentage
2D	: Two-dimensional
3D	: Three-dimensional
4NBDT	: 4-Nitrobenzenediazonium tetrafluoroborate
α	: The probability of type I error
ANOVA	: Analysis of Variance
APCVD	: Atmospheric Pressure Chemical Vapor Deposition
Ar	: Argon
Ar ⁺	: Argon ion
Au	: Gold
β	: Model coefficient of
BN	: Hexagonal boron nitride
C ₂ H ₂	: Acetylene
C _A	: Raman scattering constant
Ca ²⁺	: Calcium ion
CCD	: Central Composite Design
CCF	: Central Composite Face-centered design
CH ₃ CN	: Acetonitrile
CH ₄	: Methane
cm	: Centimeter
cm ⁻¹	: Reciprocal centimeter
CNT	: Carbon Nanotube
Co	: Cobalt
C _S	: Raman scattering constant
Cu	: Copper



CuO	: Copper(II) Oxide
CV	: Cyclic Voltammetry
CV	: Cross Validation
CV-RMSE	: Root-mean-square error of cross validation
DOE	: Design of Experiment
EDS	: Energy dispersive spectroscopy
EG	: Epitaxy growth
E_L	: Laser energy
Et_2O	: Diethyl ether
eV	: Electron volt
Fe	: Iron
Fe^{3+}	: Iron(III) ion
FeCl_3	: Ferric chloride
FET	: Field Effect Transistor
FWHM()	: Full width at half maximum
g	: Gram
GaN	: Gallium nitride
GC	: Glassy Carbon
GO	: Graphene oxide
H_2	: Hydrogen gas
H_2O	: Water
H_2SO_4	: Sulfuric acid
HBF_4	: Tetrafluoroboric acid
HCl	: Hydrochloric acid
HfO_2	: Hafnium(IV) oxide
HNO_3	: Nitric acid
HOPG	: Highly Ordered Pristine Graphite

hr	: Hour
I()	: Intensity
iLO	: In-plane longitudinal optical phonon mode
I_{on}/I_{off}	: On-to-off current ratio
IR	: Infrared light
iTO	: In-plane transverse optical phonon mode
k Ω	: Kiloohm
L_D	: Interdefect distance
LOO	: Leave-one-out
LOOCV	: Leave-one-out cross validation
LPCVD	: Low Pressure Chemical Vapor Deposition
LUMO	: Lowest Unoccupied Molecular Orbital
m	: Meter
MANOVA	: Multivariate Analysis of Variance
MeOH	: Methanol
MFC	: Mass Flow Controller
min	: Minute
mK	: Millikelvin
ml	: Milliliter
MLR	: Multiple Linear Regression
mm	: Millimeter
μm	: Micrometer
mM	: Millimolar
mm^2	: Square micrometer
mS	: Millisiemen
mW	: Milliwatt
N_2	: Nitrogen gas



NaNO_2	: Sodium Nitrite
NBu_4BF_4	: Tetrabutylammonium tetrafluoroborate
Ni	: Nickel
nm	: Nanometer
n	: Number of experiments
NMR	: Nuclear Magnetic Resonance
$^\circ\text{C}$: Degree Celsius
O_2	: Oxygen gas
Pa	: Pascal
PMMA	: Poly(methyl methacrylate)
Pos()	: Position
ppm	: Parts-per-million
PSE	: Pseudo Standard Error
R^2	: Coefficient of Determination
r_A	: Radius of circle area that structural is distorted by point defect
RDS	: Rate Determining Step
rGO	: Reduced Graphene Oxide
RMSE	: Root Mean-Square-Error
RO	: Reversed Osmosis
rpm	: Round per minute
r_s	: Radius of D-band scattering area
RSM	: Response Surface Methodology
S	: Siemens
s	: Second
sccm	: Standard Cubic Centimeter per Minute
SD	: Standard deviation
SDS	: Sodium dodecylsulfate

SET	: Single Electron Transfer
Si	: Silicon
Si ₃ N ₄	: Silicon nitride
SiC	: Silicon carbide
SiO ₂	: Silicon dioxide
SiO ₂ /Si	: Silicon wafer with silicon dioxide insulating layer
sq	: Square area
TEM	: Transmission electron microscope
UV	: Ultraviolet
UV-A	: Ultraviolet A
V	: Volt
W	: Watt
w/v	: Weight by volume
\bar{X}	: Mean value
XPS	: X-ray Photoelectron Spectroscopy
XRD	: X-ray Diffraction
Ω	: Ohm

CHAPTER I

INTRODUCTION

1.1 Graphene

Graphene is an allotrope of carbon which consists of sp^2 -hybridized carbon atoms forming a 2D honeycomb structure [1]. Although it has been investigated since 1940 [2], this monolayer sheet was predicted to be thermodynamically unstable under ambient condition [3-4]. In 1960, thin lamellae of graphitic carbon, currently named reduced graphene oxide, were prepared by chemical reduction of graphite oxide (GO) [5]. However, such lamellae are not equivalent to graphene. It was not until 2004 that the existence of graphene was proven.

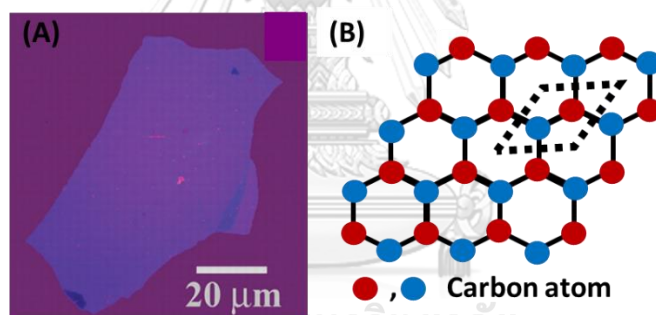


Figure 1.1 (A) Optical micrograph of graphene exfoliated from graphite by Scotch[®] tape method [6] (Copyright © 2004 American Association for the Advancement of Science) (B) Schematic top view of graphene lattice

Graphene was realized by mechanical exfoliation of graphite using Scotch[®] tape [6] (Figure 1.1(A)). The structure of exfoliated graphene is a single-atom-thick sheet of carbon atoms rearranged into a honeycomb pattern (Figure 1.1(B)). It is considered to be the thinnest material nowadays. Graphene has unique properties which outperform many of the preexisting materials. For example, electrical conductivity of graphene could reach 10^6 S cm^{-1} with extremely high mobility up to $200,000 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ [7]. Meanwhile thermal conductivity of graphene can exceed 3000 W mK^{-1} , which is higher

than that of graphite [8]. Graphene absorbs only 2.3% of visible light making it almost transparent [9]. Due to the nature of strong covalent bonds between carbon atoms in graphene, the material is considered as the strongest one [10]. Nonetheless, it is flexible and has low density. Moreover, it also has large specific surface area of about $2630 \text{ m}^2 \text{ g}^{-1}$ [11]. With some of the aforementioned properties, graphene is capable of being used in electronic and optoelectronic devices.

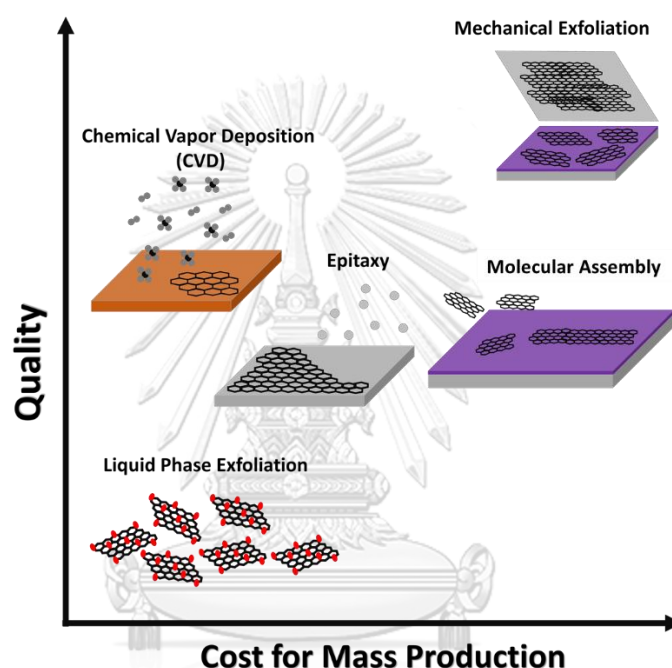


Figure 1.2 Comparison of graphene prepared by different methods in terms of quality and production cost

Despite having excellent properties, graphene sheets prepared by Scotch[®] tape method can only be produced in tiny amount without any control of film thickness and specific location of deposition. Moreover, the production process is not scalable, so it is not practical for industrial use. Alternative methods have been developed to realize graphene production in industrial scale. Chemical reduction of graphene oxide (GO) is one of the promising approaches because it can be carried out in large scale. According to Hummers' method [12], graphite as a starting material, is first oxidized in strongly acidic and oxidative condition to form graphite oxide which contains many

oxygen-containing functional groups on its basal plane and edge and is later exfoliated by ultrasonication to obtain graphene oxide (GO). GO is then chemically reduced to reduced graphene oxide (rGO) but some functional groups still remain in the structure [13]. Hence, rGO has lower quality, higher defect, smaller grain size and lower electrical conductivity than graphene obtained by Scotch[®] tape method. rGO can be applied to applications on which large surface area or further functionalization are required.

Due to the fact that graphene synthesized via Hummers' method, which is a top-down approach results in tiny flakes with low conductivity, many researchers has devoted their efforts to synthesize higher quality graphene from smaller precursors. Epitaxial growth (EG) of graphene from silicon carbide (SiC) was proposed as an alternative way [14]. In this method, graphene is formed at high temperature of 1200 – 1500 °C [14]. under ultrahigh vacuum when Si atoms evaporate from the surface of SiC and the remaining carbon atoms form a hexagonal graphitic structure. The size of epitaxial graphene is as large as that of the growth substrate. The electron mobility of this graphene could reach $5000 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ which is relatively high compared to rGO but still low compared to exfoliated graphene because of its polycrystallinity [14]. Although, this method affords high quality graphene, SiC is very expensive, and the process needs vacuum pumps.

Chemical vapor deposition (CVD) was then introduced in 2008 to synthesize graphene by feeding small carbonaceous precursor into high temperature tube furnace where metal substrate is heated inside [15]. The gaseous precursor decomposes on metal surface to form carbon film. When the reactor is cooled down, carbon atoms rearrange into a hexagonal pattern to form graphene. The process could be performed under both atmospheric and low pressures. Methane and copper are extensively used as a precursor and a substrate, respectively [16]. The size of CVD graphene is also as large as that of the substrate. The electron mobility ranges from 5000 to $12000 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ [17]. Moreover, this method is industrially scalable. Unfortunately, methane is relatively expensive for using as the precursor. As a result, other precursors were tested, for example, propane [18], acetylene [19] and ethanol [20] .

As illustrated in Figure 1.2, the best quality graphene can be obtained from Scotch[®] tape method but production cost of this method is relatively high. On the

other hand, liquid phase exfoliation is most economical, but the quality is not good. While CVD and Epitaxy can produce graphene with similar quality. However, CVD has less manufacturing cost than that of epitaxy. Therefore, CVD is most promising approach for industrial production of graphene.

1.2 Chemical Vapor Deposition

Chemical vapor deposition (CVD) is a well-known technique to produce a thin solid film. The film is formed by chemical reaction of vapor-phase precursors on the surface of substrate. The rate of reaction can be accelerated by heat (thermal CVD), UV light (photo-assisted CVD) or plasma (plasma-enhanced CVD). Generally, CVD is used to deposit metal oxide thin film by feeding volatile organometallic compounds on a reactive substrate. The deposited film is widely used in microelectronics and optoelectronics as protective, dielectric or coatings.

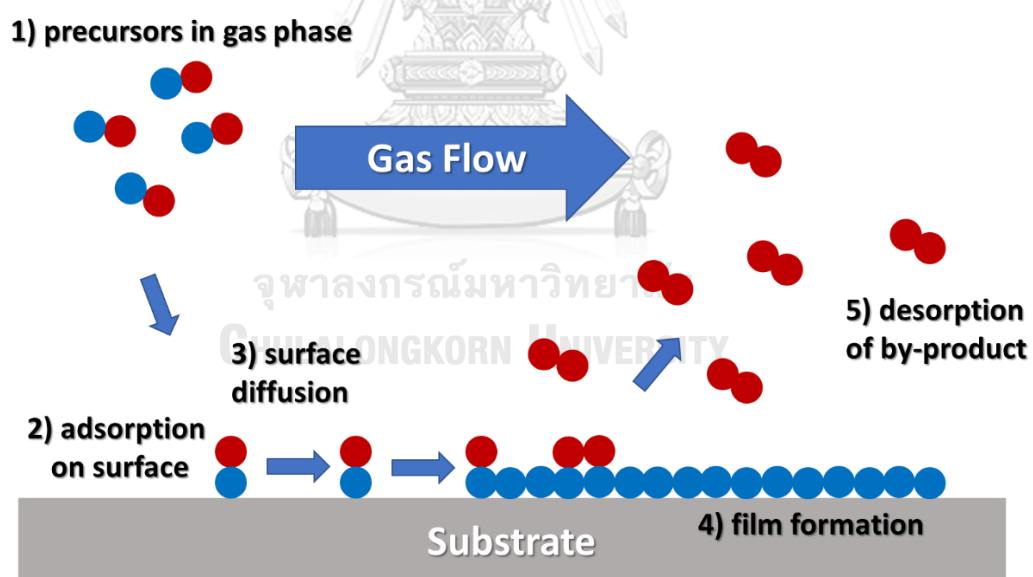


Figure 1.3 General mechanism of CVD processes

In CVD processes, volatile precursors approach and then adsorb onto the surface. The adsorbed precursor could diffuse and transport on surface. At an optimum state, reaction occurs and cleaves some fractions of the precursor molecule which

later desorb from the surface to vapor phase (Figure 1.3). Nucleation and growth of film depend on various factors such as temperature, purity of precursor and pressure.

Graphene can also be synthesized by CVD. This method could synthesize high quality graphene with large area by feeding carbonaceous compounds into the reactor. The precursors are usually hydrocarbon gases such as methane [10], acetylene [19, 21], propylene [22] or even volatile organic compound [23]. Substrate also plays an important role in the reaction. Copper and nickel are widely used because of low cost [24].

First success on CVD-grown graphene production was reported in 2008 [15]. Methane was fed into furnace at 1000 °C into which a nickel foil was placed. After graphene growth, the furnace was quickly cooled with a cooling rate 10 °C s⁻¹. The film was single layer.

Nowadays, methane and copper are the most popular carbon precursor and substrate, respectively, in graphene synthesis. However, methane is relatively expensive, and the gas needs to be imported. On the other hand, acetylene is a lot cheaper than methane. Thus, using acetylene, which is a by-product from petroleum manufacturing, instead of methane would benefit in cost reduction and help promoting graphene production. Therefore, acetylene is recognized as another promising candidate for synthesis of CVD-grown graphene.

1.3 Functionalization of graphene

Generally, defect-free exfoliated graphene has no band gap, thus it cannot be directly utilized as an active material in transistors. Also, even though the structure of CVD-grown graphene is not perfect, the band gap of as-grown graphene is still too small, and so it is commonly used as transparent electrodes

Development of graphene-based electronic devices has been impeded due to the zero band gap nature of the material. To be used especially for electronic and optoelectronic applications, opening band gap is a prerequisite [25].

In order to open the band gap of graphene, breaking symmetry or intervalley mixing [26] are some possibilities. Sublattice symmetry breaking [27], magnetic effect

[28] and quantum confinement effect [29] are considered as causes of band gap opening. Chemical modifications are practical approaches to achieve band gap opening [25, 30]. Two approaches have been proposed including physisorption of organic compounds and chemisorption of reactive species.

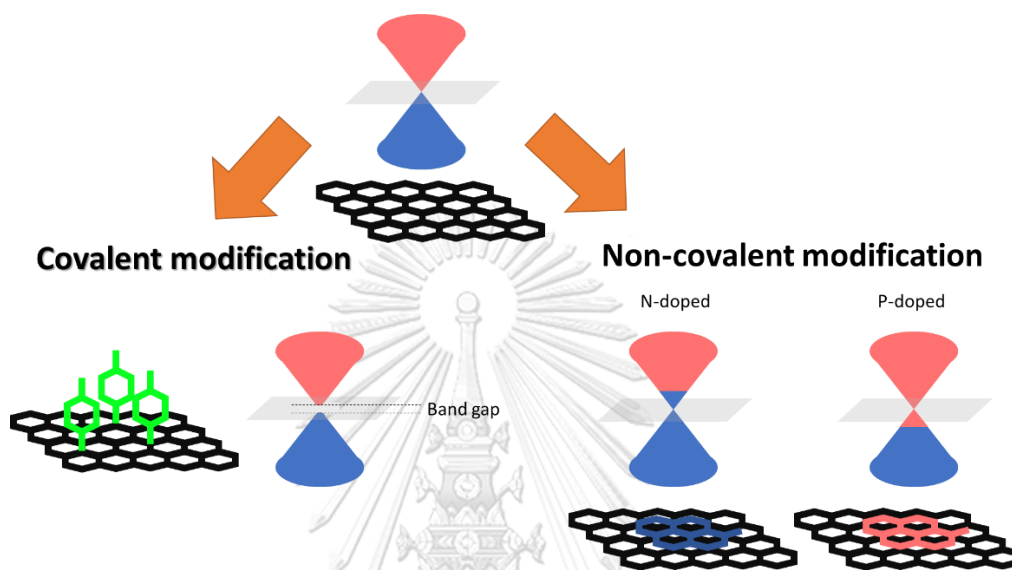


Figure 1.4 Schematic illustration of graphene modification methods: non-covalent modification and covalent modification

The physisorption of organic building blocks on graphene is considered as a non-covalent modification by which π - π interaction between graphene and organic dopants is prominent. Highly π -conjugated compounds such as perylene-3,4,9,10-tetracarboxylic dianhydride [31] and Co-phthalocyanines [32] or long-chained hydrocarbons such as alkylphosphonic acids [33] and pentacosadiynoic acid [34] were applied to graphene surface. According to strong van der Waals interactions, organic building blocks are periodically aligned on the surface. The charge transfer process between graphene and dopants shifting doping level in electronic structure changes charge carrier density resulting in band gap opening. Unfortunately, this method can achieve only a band gap of ~ 0.1 eV at maximum dopant concentration which is much lower than the desired band gap of 0.4 eV [30].

Another way to modify graphene is covalent modification by which sp^2 -hybridized carbon atoms are converted to sp^3 -hybridized carbon atoms. Developing methods for graphene functionalization is challenging for chemists, physicists and materials scientists. These methods require reactive species to react with chemically stable graphene. Many dopants which have successfully modified other carbon allotropes such as carbon nanotubes or fullerenes were also tried on graphene. Some of the most successful candidates are diazonium salts. Diazonium salt could also react with reduced graphene oxide (rGO), CVD-grown graphene or even exfoliated graphene [35]. Until now, the band gap of 0.36 eV has been achieved by functionalization of graphene [25]. Unfortunately, only few diazonium salts have been studied.

1.4 Scope of this research

1. Parametric study and optimization of CVD processes for graphene synthesis using acetylene as precursor under atmospheric pressure
2. Chemical functionalization of CVD-grown graphene with reactive diazonium salts

1.5 Objectives of this research

1. To optimize CVD processes for graphene synthesis using acetylene as precursor under atmospheric pressure
2. To develop methods for functionalization of graphene using diazonium salts

CHAPTER II

THEORY AND LITERATURE REVIEW

2.1 Graphene growth using acetylene in Chemical Vapor Deposition

Acetylene (C_2H_2), systemically called ethyne, is colorless, odorless and flammable gas consisting of two carbon atoms bonded together with a triple bond while each atom is terminated with a hydrogen atom (Figure 2.1(A)). It has been produced by cracking of petroleum products as a side product or by hydrolysis of calcium carbide (CaC_2). Generally, it is used as a fuel in a cutting torch because its flame can reach about $3100\text{ }^\circ\text{C}$ when mixed with oxygen (O_2). Acetylene can be kept in the form of calcium salt which is easier and safer to handle. This ionic salt could generate acetylene by adding water (H_2O). However, acetylene cannot be kept in copper (Cu) or Cu alloy container because copper (Cu) can catalyze decomposition of acetylene.

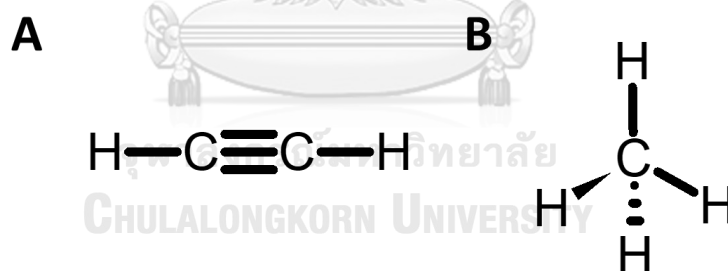


Figure 2.1 Chemical structures of (A) acetylene and (B) methane

Acetylene is a potential replacement of methane (Figure 2.1(B)) as a carbon precursor for graphene production because acetylene could reduce defect density in as-grown graphene [36]. However, acetylene has higher pyrolysis rate compared to methane. Therefore, graphene produced from acetylene tends to have high defect and small grain size.

Table 2.1 List of publications related to graphene synthesized from acetylene during 2009 – 2017 [19, 37-47]

Year	Authors	Type of CVD	Pressure	Substrate	Growth Temperature (°C)	Quality	No. Layer
2009	Kumar et al.	Thermal	LP	Ni film on SiO ₂ /Si	750	Defect-free	1
2010	Rümmelli et al.	Thermal	LP	MgO	325	High defect	2-10
2010	Sato et al.	Thermal	LP	Fe film on SiO ₂ /Si	650	Defect-free	Few
2011	Banerjee et al.	Thermal	LP	Co film on SiO ₂ /Si	800	Low defect	1
2011	Sun et al.	Thermal	LP	Si ₃ N ₄	1000	High defect	Mono – few
2011	Hofmann et al.	Thermal	LP	Au-Ni film on SiO ₂ /Si	600	Low defect	1
2013	Qi et al.	Thermal	AP	Cu foil	1000	Low defect	2
2015	Cortés et al.	Thermal	LP	CuO film on SiO ₂ /Si	1000	High defect	1
2015	Hseih and Chen	Thermal	LP	Ni foil	600	Low defect	Few
2016	Yang et al.	Thermal	LP	CuO on Cu	1035	Low defect	1
2016	Yang et al.	Thermal	LP	CuO on Cu	1035	Low defect	1
2017	Muñoz et al.	Plasma-Enhanced	LP	SiO ₂ /Si	700	High defect	1

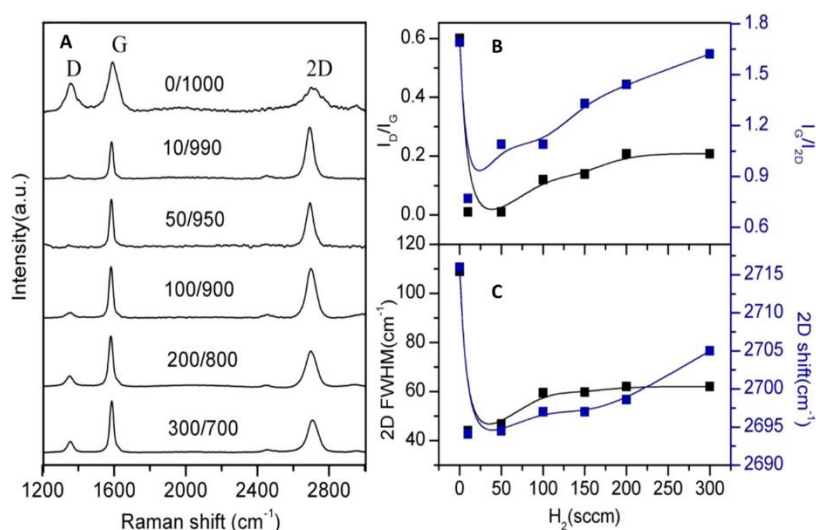


Figure 2.2 (A) Raman spectra, (B) $I(D)/I(G)$ and $I(G)/I(2D)$ and (C) FWHM(2D) and Pos(2D) of graphene synthesized with different H_2 flow rate [42] (Copyright © 2013, American Chemical Society)

CVD-grown graphene synthesized from this precursor has been developed since 2009 [19]. Thermal CVD under low pressure was the most extensively studied method to produce graphene with good-to-medium quality [19, 37, 39-41]. Metals with high carbon solubility such as Ni and Co were popular choices of substrate [19, 39] because they can retain carbon atoms on their bulk phase [48]. Moreover, growth step can perform at lower temperature (< 1000 °C). Metal alloys were alternative substrates to limit over-growth of graphene by blending the aforementioned metals with other metals with low carbon solubility [41]. Growing graphene on metal oxide, an insulating layer, is ideal for applications in device fabrication [38] because a transfer step is not required. Still, graphene grown on the insulating layer usually has high defect.

Despite the fact that low-pressure CVD (LPCVD) can produce high quality graphene, it takes a long reaction time and requires an expensive high-vacuum pump. Atmospheric pressure CVD (APCVD), on the other hand, is considered as it shortens the reaction time. Moreover, performing reaction under atmospheric pressure does not require any vacuum pump. Nevertheless, APCVD needs high operating temperature in order to synthesize high quality graphene as in LPCVD. APCVD process

for graphene synthesis was first achieved in 2013 on Cu substrate by Mei Qi et al. [42]. In the beginning, they investigated effects of growth temperature (850–1000 °C) and acetylene flow rate (1-12 sccm). They found that high temperature and low acetylene flow rate could reduce defect density. They also observed an effect of H₂/Ar flow rate ratio (Figure 2.2). Bilayer graphene could be synthesized under the flow rate ratio between 10/990 and 100/900. At higher ratios, graphene became multilayer. Consequently, they suggested that H₂ acts as an activator of the surface bound carbon and graphene etchant controlling morphology, nucleation density and nucleation size.

In addition to LPCVD and APCVD, plasma-enhanced CVD (PECVD) is another alternative way to reduce growth temperature. In 2017, Span research teams successfully synthesized graphene on SiO₂/Si through this approach [47]. However, as-grown graphene still had high defect. Furthermore, expensive and complicated instruments were needed.

Most of the CVD processes using acetylene as a precursor were performed at low pressure to reduce its activity and obtain the thinnest graphene. Substrate also played a crucial role on the quality of graphene. Although there was an effort to produce graphene directly on metal oxide, growing graphene on metal usually gave better results. Thermal CVD was more popular than other kind of CVD because of its lower cost. Nonetheless, it required high temperature. The presence of plasma could reduce the operating temperature.

2.2 Covalent functionalization of graphene

Covalently functionalized carbon materials have been tremendously utilized in sensing application [25]. Many methods for modification of carbon nanotubes (CNT), highly oriented pyrolytic graphite (HOPG) and glassy carbon (GC) have been studied. However, unlike other allotropes of carbon, graphene is not quite reactive because of its planarity. Reactive intermediates are required for forming adducts with

graphene due to its low reactivity. Converting sp^2 -hybridized carbons to sp^3 -hybridized carbons can alter graphene properties.

Free radicals are well known as very reactive intermediates. They can react with a variety of functional groups, for example, alkane, alkene, alkyne and carbonyl. Nevertheless, free-radical reactions cannot be controlled easily. The propagation reaction usually occurs and attributes to formation of oligomers or polymers. Generation of radical is also challenging because of its instability.

One of the most well-known compounds for radical formation is diazonium salt which undergoes electron transfer reaction. Then, the labile radical forms a covalent bond with graphene. Diazonium salt can be prepared from amine. Unfortunately, most of diazonium salts are not stable. It can decompose at room temperature. So, the diazonium salt should be kept under 5 °C or it should be generated in situ and used immediately. However, tetrafluoroborate salts of diazonium compounds are quite stable. They can be isolated in solid form and their solutions can be kept at room temperature.

Another possible method is decomposition of benzoyl peroxide. The grafting is activated by an unstable photogenerated benzoyl radical which then undergoes decarboxylation to form a phenyl radical. The radical can also form adduct with graphene as well as the radical generated from diazonium salt. Nonetheless, benzoyl peroxide is not persistent to light. Therefore, it is difficult to be stored without decomposition.

2.3 Diazonium-mediated functionalization

Surface functionalization with diazonium salts has been extensively studied for many purposes such as protection against corrosion or sensing application. The modified surfaces of carbon, silicon or metal have been achieved. This method could establish a strong C—C bond between surface of substrate and organic layer. Common carbon materials modified by diazonium salt are glassy carbon (GC), highly

Table 2.2 List of publications related to diazonium functionalization of graphene during 2009 – 2014 [49-61]

Year	Authors	Type of Graphene	Substrate	Diazonium salt	Solvent	Additive	Technique	Atmosphere	Temperature	Time
2009	Haddon et al.	EG	Si	4NBDT	CH ₃ CN	0.1 M NBu ₄ PF ₆	Solution Chemistry	Ar	rt	20 hr
2010	Strano et al.	ME	SiO ₂ /Si	4NBDT	H ₂ O	1 % SDS	Solution Chemistry	Air	35 °C	7 hr
2010	Tour et al.	GNR	SiO ₂ /Si	4NBDT	CH ₃ CN	N/A	Solution Chemistry	N ₂	rt	min to hr
2010	Choi et al.	ME	SiO ₂ /Si	4BrBDT	50 % v/v CH ₃ OH/H ₂ O	N/A	Solution Chemistry	N/A	rt	30 min
2011	Strano et al.	CVD	SiO ₂ /Si	4PPOBDT	H ₂ O	0.5 % SDS	Solution Chemistry	N/A	45 °C	8 hr
2011	Nouchi et al.	ME	HDMS-coated SiO ₂ /Si	4NBDT	H ₂ O	1 % SDS	Solution Chemistry	N/A	36 - 40 °C	7 hr
2011	Gao et al.	ME	SiO ₂ /Si	4NBDT	CH ₃ CN	NBu ₄ PF ₆	Solution Chemistry	Ar	rt	10 hr
2012	Guo et al.	CVD	SiO ₂ /Si	4NBDT	H ₂ O	0.1 M H ₂ SO ₄	Electrochemistry	Ar	rt	5 s
2012	Strano et al.	CVD	BN, Al ₂ O ₃ , OTS, SiO ₂	4NBDT	H ₂ O	0.5 % SDS	Solution Chemistry	N/A	35 °C	16.5 hr
2013	Tour et al.	CVD	SiO ₂ /Si	4NBDT, 4DMABDT, 4M CBDT	CH ₃ CN	N/A	Solution Chemistry	N/A	N/A	5 – 60 min
2013	Ago et al.	CVD	PDMS	4NBDT	H ₂ O	N/A	Mechanics	N/A	N/A	10 s – 180 min
2013	Ruoff et al.	CVD	Si	4NBDT	CH ₃ CN	N/A	Solution Chemistry	Ar	N/A	180 min
2014	Daasbjerg et al.	CVD	Ni and Cu	AQDT	CH ₃ CN	0.1 M NBu ₄ BF ₄ 0.01 M NBu ₄ I	Electrochemistry	N/A	N/A	N/A

Table 2.3 List of publications related to diazonium functionalization of graphene during 2015 – 2017 [62-67]

Year	Authors	Type of Graphene	Substrate	Diazonium salt	Solvent	Additive	Technique	Atmosphere	Temperature	Time
2015	Mooste et al.	CVD	Ni	4NBDT, 4BrBDT, ABD, Fast Garnet GBC, Fast Black K	CH ₃ CN	0.1 M NBU ₄ BF ₄	Electrochemistry	N/A	23 ± 1 °C	N/A
2015	Siaj et al.	CVD	SiO ₂	4CBDCI	H ₂ O	0.5 M HCl	Electrochemistry	N/A	N/A	N/A
2016	Assaud et al.	CVD	Pt film on SiO ₂ /Si	4TPBDCl	H ₂ O	0.1 M HCl	Electrochemistry	N/A	N/A	N/A
2016	Palermo et al.	CVD	SiO ₂ /Si, PET and Quartz	4DCOBDT	H ₂ O	0.1 M H ₂ SO ₄	Patterning and Electrochemistry	N/A	rt	N/A
2016	Siaj et al.	CVD	Cu	4NBDT	CH ₃ CN	N/A	Electrochemistry	N/A	N/A	N/A
2017	Kalbač et al.	CVD	Cu	4NBDCl, 4CBDCI, 4ECBDCI	30 % v/v H ₂ O/CH ₃ CN	N/A	Solution Chemistry	N/A	N/A	2 hr



oriented pyrolytic graphite (HOPG), carbon fiber, carbon blacks, carbon nanotubes and diamond. [68]

The grafting mechanism was studied by observing electrochemical behavior of diazonium salts. Most of grafting experiments were carried out in 0.1 M NBU_4BF_4 in CH_3CN or 0.1 M H_2SO_4 in H_2O . Diazonium salts have particularly low cathodic reduction potential. The aryl radical was considered as a reactive intermediate produced directly via concerted electron transfer from the electrode with dinitrogen cleavage. Then, the electrode surface was electro-grafted by the aryl radical to obtain C-surface bond. However, for most efficient grafting, surface must be clean and free from oxide. Deoxygenated solutions must be used in all processes to prevent oxide formation.

To characterize a modified surface, many methods have been used to proof the existence of organic layer. Fourier transform infrared spectroscopy (FTIR) is very useful in observing vibrational modes of grafted organic layers for example two strong bands of nitro group at ~ 1530 and 1350 cm^{-1} . This technique can also detect the adsorbed diazonium salts which $\text{N}\equiv\text{N}$ bond stretching locates between 2300 and 2130 cm^{-1} . X-ray photoelectron spectroscopy (XPS) is one of the most powerful techniques in surface characterization. The information obtained from XPS spectra shows atomic composition at the surface and its oxidation state. After modification, surface morphology is certainly changed. Atomic force microscopy can reveal the formation of organic layer with variable height and roughness. Furthermore, energy dispersive spectroscopy (EDS) could show the dispersion and density of atom on the surface.

In the past decade, methods for diazonium functionalization on graphene were developed. Various types of graphene such as mechanical exfoliated graphene [51, 53-54], epitaxial graphene [49], graphene nanoribbons [52] and CVD graphene [55-67], were functionalized. Since 2009, high quality graphenes such as mechanical exfoliated graphene or epitaxial graphene were employed. After 2011, CVD graphene was extensively applied because it holds considerable promise for the next generation of electronics. Various diazonium salts have also been functionalized. However, the structure of diazonium salts has been limited to only benzene core structure with different substituents.

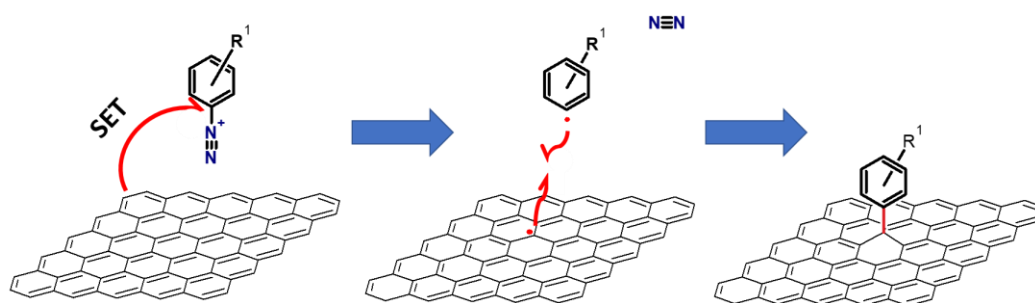


Figure 2.3 Schematic mechanism of diazonium grafting on graphene

The proposed reaction mechanism of diazonium grafting on graphene consists of two steps as shown in Figure 2.3. First, graphene transfers an electron to diazonium salt resulting in radical formation and nitrogen gas cleavage. This step was considered as rate determining step (RDS). Thereafter, it immediately forms a covalent bond with graphene radical.

Many approaches have been developed to enhance reaction rate of the first step including solution chemistry [49-55, 57, 59-60, 67], electrochemistry [56, 61-66] and mechanochemistry [58]. Solution chemistry is simple but it usually takes a long time. For electrochemical approach, it is the quickest way to perform functionalization within minute. However, this approach required expensive instruments with expertise in the field. In case of mechanochemical approach only one work was published, and this method was not favorable. Basically, the reactivity of graphene was enhanced by stretching graphene on flexible substrate.

The model chemical for developing methods for graphene functionalization method is 4-nitrobenzenediazonium salt. Its tetrafluoroborate salt was frequently used due to its high stability.

Reactivity studies are usually carried out by solution chemistry approach because there is no enhancement factor from external sources. The reactivity of functionalization solely depends on graphene and diazonium salt themselves. The single-layer graphene is more reactive than multilayer graphene. As the number of

graphene layer increases, the reactivity turns low [51]. Moreover, diazonium salt prefers reacting at the edge to basal plane [51]. In addition to the thickness of graphene, the substrate also plays an important role in the functionalization. Polar substrates such as SiO_2/Si and sapphire were found to enhance reactivity. On the other hand, functionalization on non-polar substrates, for instance, boron nitride, were nearly impossible [57]. Interestingly, the electrical conductivity of graphene deposited on non-polar substrates could be enhanced better than that deposited on polar substrates [69].

Raman parameters including $I(\text{D})/I(\text{G})$, $I(2\text{D})/I(\text{G})$, $\text{Pos}(\text{G})$, $\text{FWHM}(\text{G})$, $\text{Pos}(2\text{D})$ and $\text{FWHM}(2\text{D})$ frequently change after functionalization. $I(\text{D})/I(\text{G})$ increases and $I(2\text{D})/I(\text{G})$ decreases as shown in Figure 2.4(A) and 2.4(E), respectively, while $\text{Pos}(\text{G})$ and $\text{Pos}(2\text{D})$ shift to higher wavenumbers (Figure 2.4(C)). Moreover, G and 2D peaks usually broaden (Figure 2.4(B) and 2.4(D)).

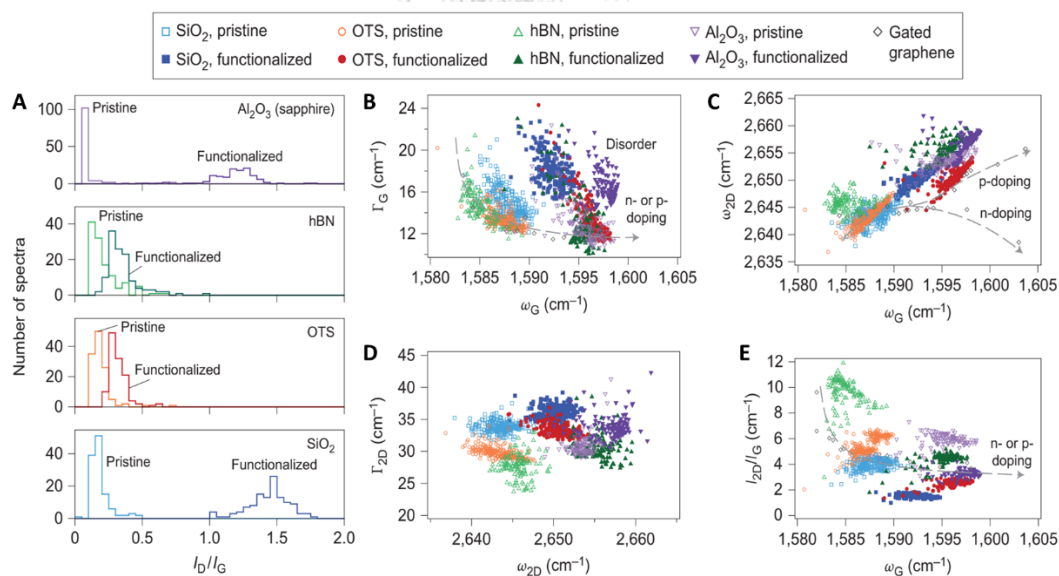


Figure 2.4 (A) Histograms of $I(\text{D})/I(\text{G})$ ratios graphene before and after functionalization on different substrates (B-E) scatter plots of Raman peak parameters of graphene before and after functionalization on different substrates [57] (Copyright © 2012, Springer Nature)

Although solution chemistry approach exhibits low reactivity, it is simple and does not require any complicated instrument. According to the studies by Haddon [49] and Gao [54], O_2 -free CH_3CN and NBu_4PF_6 were used as solvent and organic electrolyte, respectively. In addition, the reaction had to be performed in O_2 -free atmosphere to prevent O_2 adsorption on graphene surface [49, 54]. Later in 2010, Strano and co-workers [51] reported that the reaction could perform in H_2O in the presence of sodium dodecyl sulfate (SDS) [51]. The surfactant molecules could bring diazonium species closer to graphene such that electron transfers could occur more easily. The use of mixed solvents was also interesting, however, it has not been well studied.

Electrochemical approach is very effective in functionalizing graphene. Basically, graphene is functionalized by cyclic voltammetry (CV) where graphene is utilized as the working electrode in a three-electrode system. The functionalization could be carried out in aqueous solution [56, 62, 64]. Due to its fast process, it has expanded the horizon of diazonium salts used in graphene functionalization from small benzene ring derivatives to large polyaromatic derivatives. The diazonium salt could be readily prepared in situ or used in the form of stable tetrafluoroborate salt. In addition to H_2O -based electrochemistry, electrochemical process in organic solvent was also possible [61, 63, 65].

The last approach is mechanochemical approach first introduced in 2013 [58]. This approach is the least studied one in spite of its simplicity. This requires a stretching tool and graphene needs to be transferred to a flexible substrate such as a PDMS stamp. So far, the only diazonium salt tested with this approach was 4-nitrobenzenediazonium tetrafluoroborate (4NBDT).

2.3 Raman spectroscopy for graphene characterization

Raman spectroscopy is one of the most powerful tools for characterization of graphene. The characterization is both fast and non-destructive. Moreover, Raman spectrum provides structural information as well as electronic information. However,

to obtain accurate information, spectral interpretation needs to be carefully done. Raman spectroscopy is well-known method for distinguishing carbon allotropes. While Raman spectrum of graphite was first measured over 40 years ago, it was not until 2006 for the first Raman spectrum of graphene to be recorded.

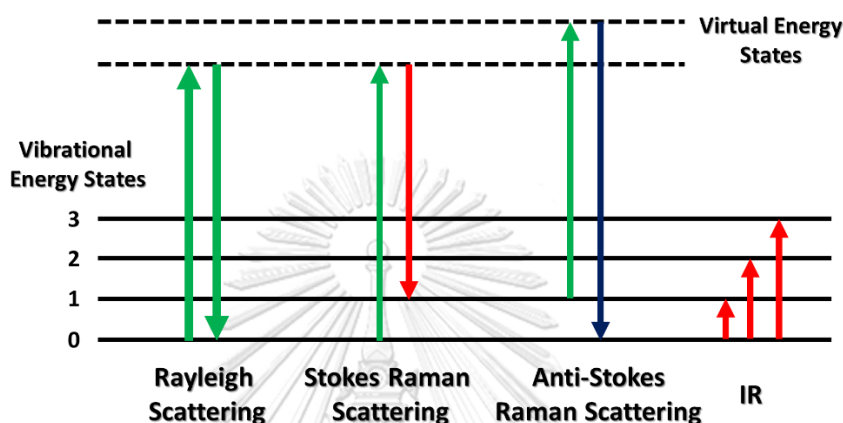


Figure 2.5 Schematic mechanism of Rayleigh scattering, Raman scattering and IR

In principle, Raman scattering occurs after an incident photon is absorbed by a Raman active material. The incident photon excited an electron to a virtual excited state and creates an electron-hole pair. The electron is scattered by phonon and loses or gain some energy before electron-hole recombination. The recombination generates a photon with lower or higher energy than the incident photon as depicted in Figure 2.5. Nonetheless, the scattering process rarely takes place. Therefore, the high intensity laser is required for Raman measurement. If the emitted photon has higher energy than the incident photon, the process is called Stoke scattering. On the other hand, if the emitted photon has lower energy than the incident photon, the process is called anti-Stoke scattering. However, most incident photons undergo elastic scattering process called Rayleigh process resulting in no energy shift. For general measurement, spectrum is recorded in Stoke region.

Two characteristic bands found in graphene include G and 2D (G') peaks as shown in Figure 2.6(B). G peak (graphitic peak), located approximately at 1600 cm^{-1} , is

generally observed in carbon materials consisting of sp^2 -hybridized carbon such as graphite, graphene, fullerene and carbon nanotube (Figure 2.6(A)), while the 2D peak appears around 2700 cm^{-1} . The intensity ratio between the 2D peak and G peak ($I(2D)/I(G)$) reflects the number of graphene layers.

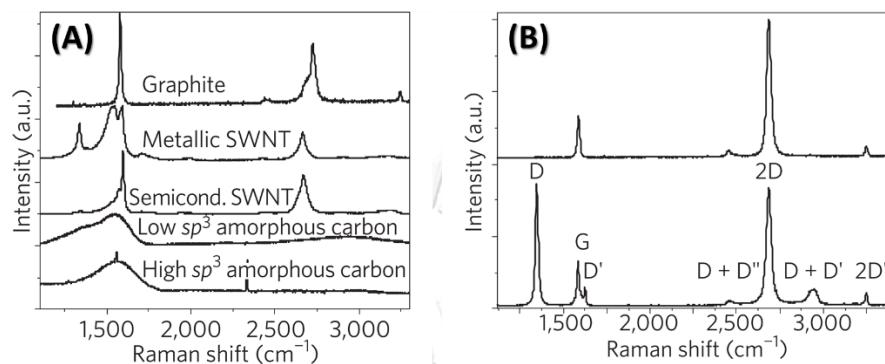


Figure 2.6 Raman spectra of (A) various carbon allotropes, (B) pristine graphene (top) and defective graphene (bottom) [70] (Copyright © 2013, Springer Nature)

The G band is attributed to first-order Raman scattering involving in-plane transverse optical phonon mode (iTO) or in-plane longitudinal optical phonon mode (iLO) at Γ point (Figure 2.7). The G band is sensitive to doping and the number of layers. When graphene gets thicker, the position of the G band shifts to a lower Raman shift. It is more complicated to interpret the spectrum when doping is involved since the position shift is also dependent on the number of graphene layers. Nonetheless, the doping level could be possibly indicated by observing $\text{FWHM}(G)$.

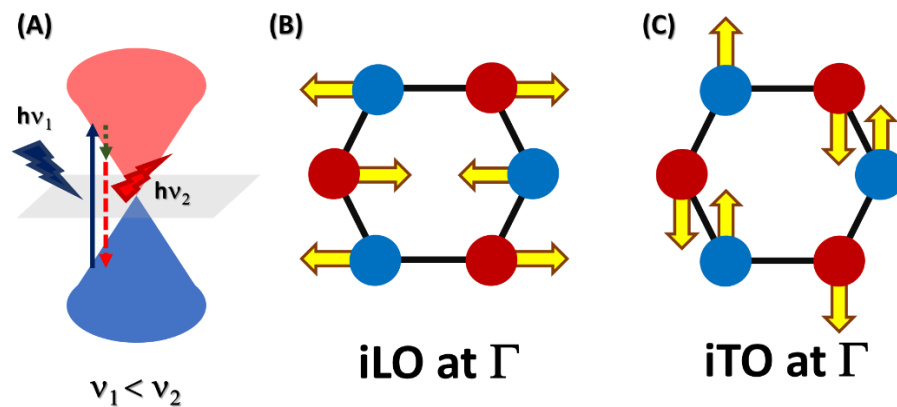


Figure 2.7 (A) Raman process of G band, sketch of phonon vibration of (B) inverse longitudinal optical mode at Γ and (C) inverse transverse optical mode at Γ

2D band is associated with double resonance process. It is a two-phonon scattering between K and K' points (Figure 2.8). After forming electron-hole pair, electron is inelastically scattered by in-plane transverse optical phonon mode (iT0) at K point and then inelastically scattered again by in-plane transverse optical phonon mode (iT0) before recombination.

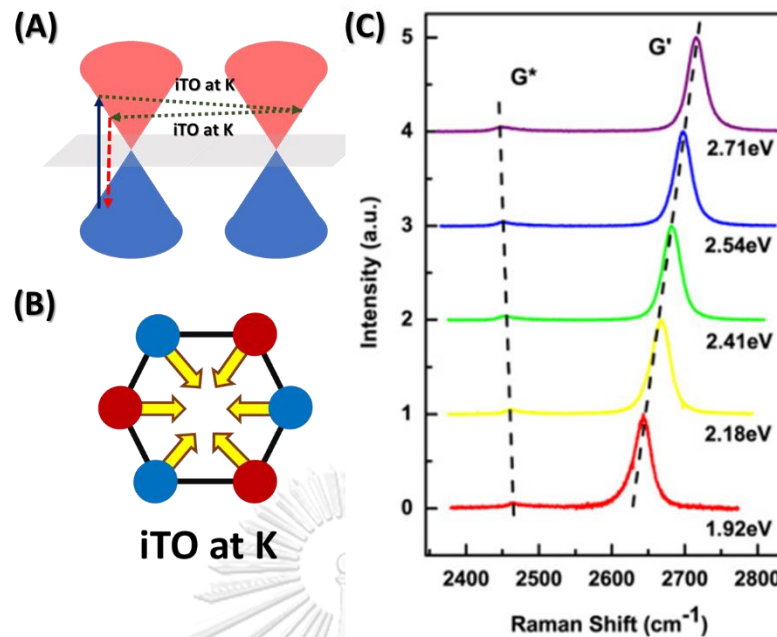


Figure 2.8 (A) Raman process of 2D peak, (B) sketch of phonon vibration of inverse transverse optical mode at K point and (C) Raman spectra showing shifts of 2D band according to different laser wavelengths [71] (Copyright © 2009 Elsevier B.V. All rights reserved.)

2D band is an important feature to estimate the number of graphene layers. When $I(2D)/I(G)$ ratio is greater than 1, it suggests that graphene is single-layer. If $I(2D)/I(G)$ ratio is approximately 1, it is bilayer. In case that $I(2D)/I(G)$ is much lower than 1, the film is very thick such that its properties become similar to those of graphite. However, $I(2D)/I(G)$ ratio can be easily disturbed by other factors such as doping and strain. The position and FWHM of 2D band ($Pos(2D)$ and $FWHM(2D)$) are alternative choices for determining the number of graphene layers. However, the position is less sensitive to the number of layers than $I(2D)/I(G)$. Therefore $FWHM(2D)$ is a good choice because it is not only sensitive to the number of graphene layers, but it is also rarely affected by other factors. In addition to parametric variables such as $I(2D)/I(G)$, $FWHM(2D)$ and $Pos(2D)$, shape of 2D band is a useful information to determine the number of graphene layers. By deconvoluting 2D band with Lorentzian

peaks, the number of peaks needed for fitting 2D band could support the estimation of number of graphene layers from FWHM(2D).

In case of pristine graphene obtained from mechanical exfoliation, only G and 2D bands are present in its Raman spectrum. However, graphene synthesized by other methods usually yields defective graphene. Raman spectrum of graphene with defects shows an additional peak around 1300 cm^{-1} called D band.

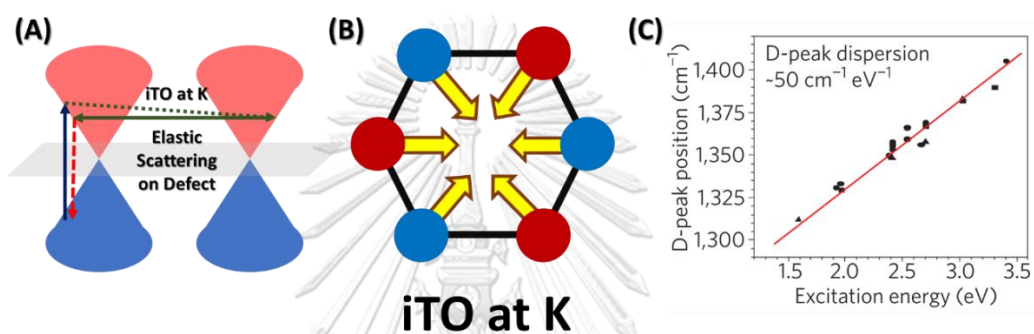


Figure 2.9 (A) Raman process of D band, (B) sketch of phonon vibration of inverse transverse optical mode at K point and (C) D peak dispersion with excitation energy [70] (Copyright © 2013, Springer Nature)

D band only appears when defects are present in graphene. The Raman process behind D band is like that behind 2D band. The electron scattered by iTO phonon in the first step as in 2D band is elastically scattered by defect before recombination (Figure 2.9). The energy loss is half of that in case of 2D band. That is the reason why G' band is alternatively called 2D band. D band is very dispersive to excitation energy as well as 2D band. The intensity ratio of D band to G band ($I(D)/I(G)$) is extensively used as the indicator to determine the amount of defects in graphene. In 2010, Lucchese and co-workers [72] successfully expressed the relationship between ($I(D)/I(G)$) and interdefect distance (L_D) as shown in Equation 2.1

$$\frac{I_D}{I_G} = C_A \frac{(r_A^2 - r_s^2)}{(r_A^2 - 2r_s^2)} \left[e^{-\pi r_s^2 / L_D^2} - e^{-\pi(r_s^2 - r_A^2) / L_D^2} \right] + C_s \left[e^{-\pi r_s^2 / L_D^2} \right] \quad \dots \text{(Equation 2.1)}$$

where $C_A = 4.2$, $C_S = 0.87$, $r_A = 3$ nm and $r_S = 1$ nm. r_S is the radius of circle area that graphene structure is distorted by point defects while r_A is the radius of D-band scattering area. The other 2 parameters were fitted from experimental data (Figure 2.10(A)).

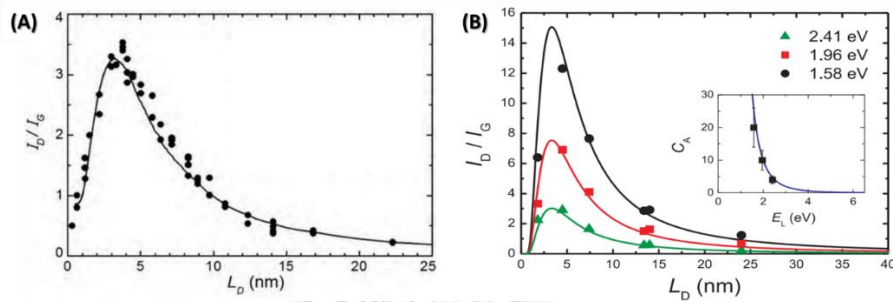


Figure 2.10 (A) The plot of $I(D)/I(G)$ ratio against interdefect distance (L_D) for samples exposed to distinct Ar^+ doses [72] (Copyright © 2010 Elsevier Ltd. All rights reserved.) and **(B)** $I(D)/I(G)$ ratio for monolayer graphene exposed to distinct Ar^+ doses obtained using different excitation lasers [73] (Copyright © 2011, American Chemical Society)

According to the dispersive property of D band, $I(D)/I(G)$ ratios measured from different laser wavelengths are not comparable. As the laser energy increases, C_A , obtained by fitting, decreases. The relations between C_A and laser energy (E_L) was derived from experimental data (Figure 2.10(B)) as shown in equation (2.2)

$$C_A = \frac{160}{E_L^4} \quad \dots \text{(Equation 2.2)}$$

By considering $L_D > 10$ nm and $L_D \gg r_A$ and r_S , equation (2.1) can be simplified as

$$\frac{I_D}{I_G} \approx C_A \frac{\pi(r_A^2 - r_S^2)}{L_D^2} \quad \dots \text{(Equation 2.3)}$$

By substituting $r_A = 3.1$ nm, $r_S = 1$ nm and equation (2.2) into equation (2.3), the equation (2.3) can be written as

$$L_D^2 (nm^2) = \frac{4.3 \times 10^3}{E_L^4} \left(\frac{I_D}{I_G} \right)^{-1} \quad \dots \text{(Equation 2.4)}$$

By converting laser energy to laser wavelength, we have

$$L_D^2 (nm^2) = \frac{1.8 \times 10^{-9} \lambda_t^4}{E_L^4} \left(\frac{I_D}{I_G} \right)^{-1} \quad \dots \text{(Equation 2.5)}$$

Those three major bands are commonly observed in graphene. However, there are some other bands. For example, in pristine graphene, 2D' band could be observed around 3250 cm^{-1} . This band shows a weak signal compared to three major bands. Moreover, some phonon modes are activated by defects introduced on graphene including D band, D' band, D + D' band and D + D'' band. Their phonon modes related to inelastic or elastic scattering of electrons on defects. Upon introduction of high defects, these peaks could appear apparently.

2.4 Experimental design for process optimization

Generally, “One-factor-at-a-time” method is carried out to observe effect of the factor and optimize condition. The experiments are performed by varying single factor while other factors are fixed. The optimum condition can be achieved in case of no interaction between factors. The method is favored by scientists and engineers. However, in some systems for example a CVD system, interactions is present and affects responses.

As mentioned above, there are interactions between factors in CVD systems. One-factor-at-a-time method might not give the best condition. Experimental design and response surface methodology should be used instead.

Experimental design, also called design of experiments (DOE) is an experimental plan carefully designed in advance. The aim of experimental design is to explain or describe variation of data under certain conditions. Many designs have been developed, for example, Box-Behnken design, central composite design (CCD) and central composite face-centered design (CCF). Each design has its own advantages and disadvantages. Hence, one design should be carefully selected for each study.

Response surface methodology (RSM) is utilized to explore the relationships between factors. Moreover, maximum or minimum of responses under investigating

space could be acquired by plotting response surface: 3D-contour of response of one factor against another factor.

At the beginning, mathematical model is constructed from data obtained from experimental design by a technique called multiple linear regression (MLR). The appropriate mathematical model could predict response well and suggest precise effect of factors to response.

In brief, a function of response depending to factors is guessed. It is usually composed of linear terms, quadratic terms and interaction terms as shown in Equation (2.6).

$$y = \beta_0 + \sum_{i=1}^n \beta_i x_i + \sum_{i=1}^n \beta_{ii} x_i^2 + \sum_{i=1}^n \sum_{i < j} \beta_{ij} x_i x_j \quad \dots \text{(Equation 2.6)}$$

where β_0 , β_i , β_{ii} , β_{ij} are coefficients, x_i and x_j are factors, and y is response. After obtaining a model equation, coefficient of determination (R^2) can be calculated. The coefficient indicates the proportion of variance in the response that could be predicted by factors. Thereby, good model must have coefficient of determination (R^2) close to 1.

However, the number of terms in function is limited by the number of experiments in design. Moreover, each term has different impacts on response. Significance of factors could be evaluated by analysis of variance (ANOVA) or student's t-test.

Some insignificant terms could be eliminated to improve predictive ability. Root mean-square-error (RMSE) is a tool to grade predictive ability. It can be calculated as shown in Equation (2.7), where y_i is actual response, \hat{y}_i is predicted response and n is total number of experiments

$$RMSE = \sqrt{\frac{\sum_{i=1}^n (y_i - \hat{y}_i)^2}{n}} \quad \dots \text{(Equation 2.7)}$$

However, good prediction in the model does not mean that it could accurately predict other experiments. Generally, further experiments are conducted, and actual response and predicted response are compared. If the mathematical model precisely predicts a response but poorly predicts responses from other

experiments, the model is overfitted. The overfit usually occurs when too many terms are added into the model.

There is another method for observing overfit, called cross validation. The data are divided into test set and training set. A model is developed from training set and RMSE are calculated from test set. The RMSE is specifically called root mean square error of cross-validation (CV-RMSE). K-fold cross validation (k-fold CV) is a very popular method. It divides data into k sets. One set is selected to be the test set. However, it requires large number of experiments. Leave-one-out cross validation (LOOCV) is an interesting method when the number of experiments is small. The test set consists of only one experiment. The disadvantage of this CV is excessive computation.

2.4.1 Plackett-Burman design

This design was introduced in 1946 by 2 British statisticians. The design is popular and economical for ruggedness testing. The main feature is that $4n-1$ factors can be included in the study with only $4n$ experiments, where n is a positive integer. For example, 12 experiments are needed for screening the effect of 11 factors. The advantage of this design is that the number of experiments is less than other designs with equal factors. In case that the number of factors does not match to $4n-1$, dummy factor can be integrated in the experiments. More dummy factors could improve estimation of error. Furthermore, each factor must be divided into two levels including high level and low level denoted as “+” and “-”, respectively.

Table 2.4 The first set of factors for Plackett-Burman design

Number of factors	Set of factors
3	- + +
7	-- + + + + -
11	----- + + + - + + +
15	----- + + + + + + + - - - -
19	----- + - + - + + + + - - + + - + + -

The set of factors for each experiment combines $2n-1$ factors at low level with $2n$ factors at high level. The first sets of factors are available in the literature as shown in Table 2.4. Other experiments are assigned in a cyclical manner from the first experiment. For last experiment, all factors are set to low level. In case of 7 factors, set of factors for 8 experiments are listed in Table 2.5.

After all, there are 4 experiments with high level and 4 experiments with low level for each factor. The effect of each parameter was calculated as shown in Equation (2.8)

$$\text{Effect} = \frac{2[\sum y_{(+)} - \sum y_{(-)}]}{N} \dots \text{(Equation 2.8)}$$

where $y_{(+)}$ is a response when a given factor is at high level, $y_{(-)}$ is a response when a given factor is at low level and N is total number of experiments.

Table 2.5 Examples of the experimental design using Box-Behnken design with 3 factors

Run	X_1	X_2	X_3	X_4	X_5	X_6	X_7
1	-	-	+	+	+	+	-
2	-	-	-	+	+	+	+
3	+	-	-	-	+	+	+
4	+	+	-	-	-	+	+
5	+	+	+	-	-	-	+
6	+	+	+	+	-	-	-
7	-	+	+	+	+	-	-
8	-	-	-	-	-	-	-

There are many ways to decide which factor has a significant impact on response. The most effective method is F-test. However, this method requires the presence of dummy factor in the design. When no dummy factor is included, Lenth's pseudo standard error (PSE) is an alternative method. However, it assumes that the factor with least effect has no impact on the experiment. The variance of factor with least effect is presumably pure error variance.

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2.4.2 Box-Behnken design

Box-Behnken Design introduced in 1960 by George E. P. Box and Donald Behnken is a three-level fractional factorial design. The advantage of this design is the limit of sample size as the number of factors grows. The sample size is sufficient to make regression including linear terms, interaction terms and quadratic terms. Three levels are denoted as “-1”, “0”, “+1” for low level, medium level and high level, respectively. This design is nearly rotatable because distances from the center point to other points are equal.

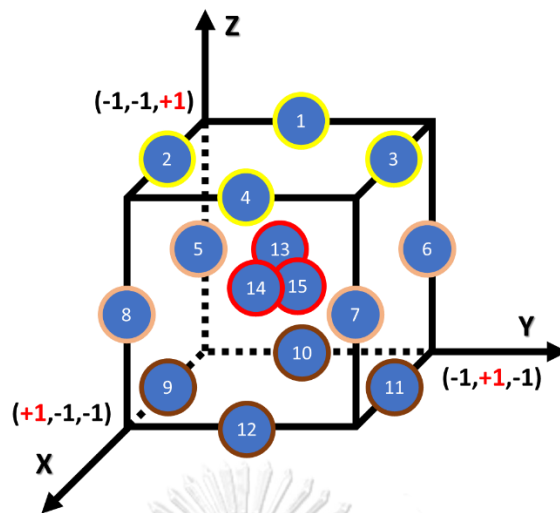


Figure 2.11 Graphical representation of Box-Behnken design

In case of 3-factored Box-Behnken model, 15 experiments are needed to carry out. As graphically illustrated in Figure 2.11, one point represents one experiment and three coordinates of each point indicate the value of three factors. 12 points are at the edge of the box while 3 points are at the center of the box. All 12 points have the same distance from the center while points at the corner are relatively far comparing to those 12 points. This might cause inaccuracy in prediction around that area. This problem could be solved by central composite design (CCD). However, due to the requirement of 5 levels per factor, CCD might not be suitable for our work.

CHAPTER III EXPERIMENTAL

3.1 CVD instrumentation

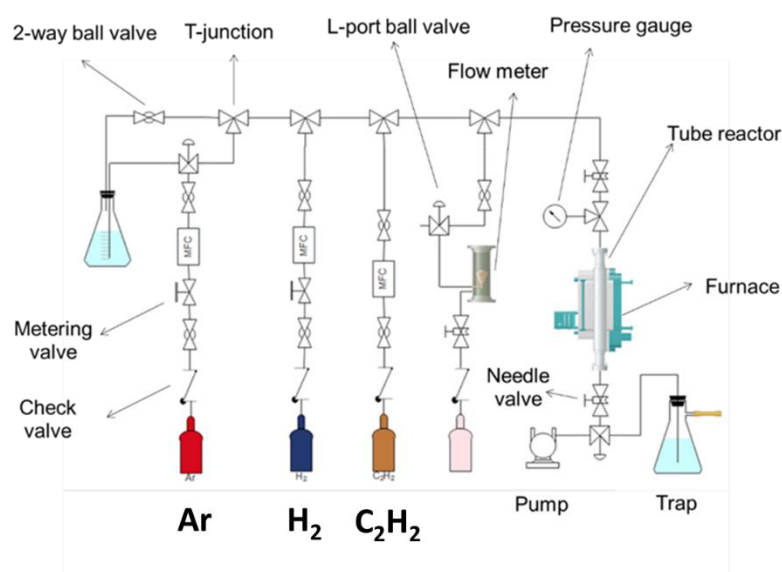


Figure 3.1 Schematic design of our CVD reactor

Our CVD reactor was built by assembling two components including a gas control panel and a furnace reactor (Figure 3.1). Three gases including argon (Ar), hydrogen gas (H_2) and acetylene (C_2H_2) were used for graphene synthesis. Acetylene serves as carbon precursor while H_2 and Ar act as catalyst and diluting gas, respectively. In addition to a gas regulator attached to each gas cylinder, check valves were installed to prevent backflow and contamination. 2-way ball valves were used to allow and stop gas flows. Gas flow rates were regulated by Thermal Mass Flow Controllers (MFCs). Metering valves between 2-way ball valves and MFCs have no function in the process. Before gas mixing at T-junction, other 2-way ball valves were installed to avoid backflow of the mixed gases. A photograph of our gas control panel is shown in Figure 3.2. Then, the mixed gases flowed through a needle valve, a pressure gauge and the tube reactor, respectively. At the exit end of the tube reactor,

there were another needle valve and L-port ball valve to selectively direct leftover gases to either a water trap or a vacuum pump.

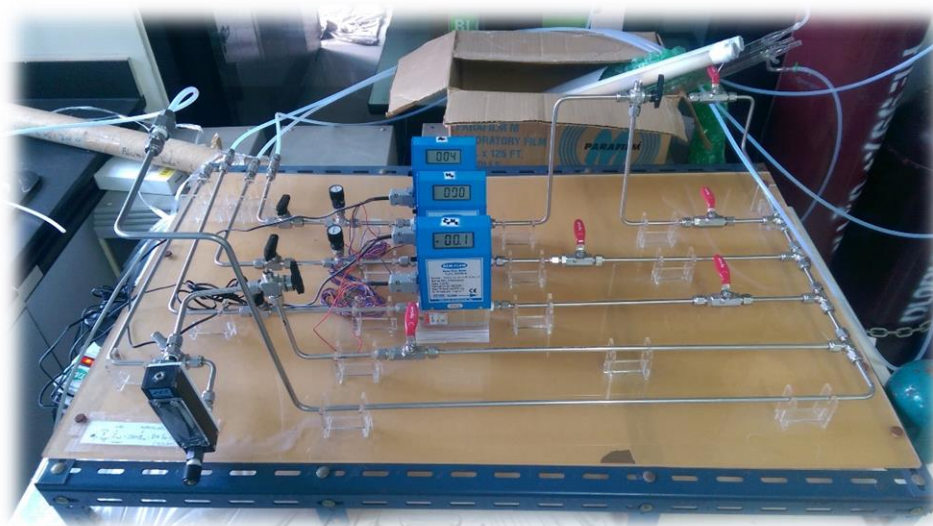


Figure 3.2 Our gas control panel

All gases were purchased from Praxair. Argon and hydrogen gases were obtained as ultra-high-purity (99.999 %), while, acetylene gas was of 99.6 % purity. The mixed gases were fed into a mullite tube which was placed in a commercial horizontal tube furnace (Lindberg/Blue M™) as shown in Figure 3.3. Leftover gases after the process were bubbled into water prior to be released into the atmosphere.



Figure 3.3 A CVD reactor

3.2 Optimization of CVD conditions

The process to synthesize graphene via CVD consists of 4 main steps including heating, annealing, growth and cooling. The temperature profile of each step is displayed in Figure 3.4. We note that the last 3 steps are very crucial to the quality of graphene. During annealing step, substrate is cleaned by hydrogen and its surface is reconstructed. Clean specific facets are important to obtaining high quality graphene. During growth and cooling steps, carbon atoms are deposited and rearranged on the substrate, respectively. Here, controlling nucleation rate and growth rates is the key to success.

In our work, a copper foil (purchased from Alfa Aesar as 99.8 % purity with a thickness of 25 μm) was cut into 1.5 x 1.5 cm^2 pieces and used as our graphene growth substrate. The substrate was extensively pre-cleaned before performing CVD process by sonicated in isopropanol for 5 min and acetone for 5 min, respectively. The substrate was then blow-dried by an Ar gun. The copper substrate was inserted into a mullite tube and positioned at center of the tube. After connecting the reactor to the gas lines, Ar was flushed into the tube at the flow rate of 1000 sccm for 5 min to purge air from the tube. Finally, composition of the gas mixture was adjusted before starting a CVD process.

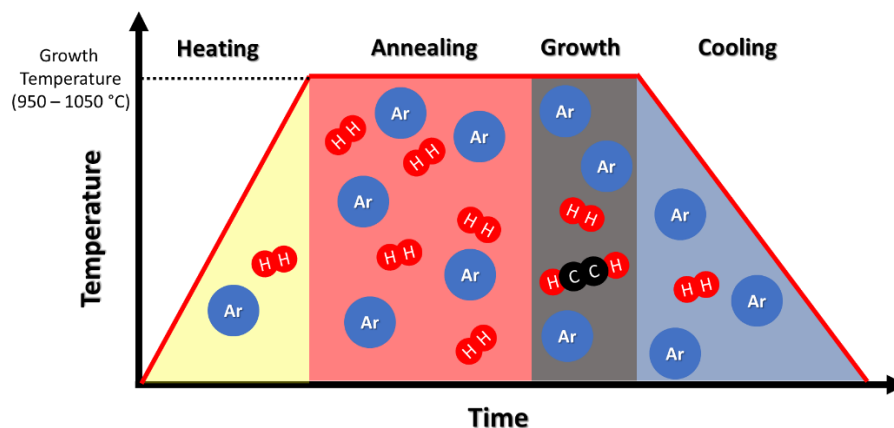


Figure 3.4 The temperature profile of a CVD process for graphene synthesis

Quality of graphene was estimated from Raman spectra. Every Raman spectrum was processed by Python script (appendix) to obtain 6 Raman parameters of graphene sample. They included intensity ratio of D peak to G peak ($I(D)/I(G)$), intensity ratio of 2D peak to G peak ($I(2D)/I(G)$), position of G peak ($Pos(G)$), full width at half maximum of G peak ($FWHM(G)$), position of 2D peak ($Pos(2D)$) and full width at half maximum of 2D peak ($FWHM(2D)$). In optimization process, all statistical calculations in Plackett-Burman screening experiment and response surface modeling were performed in Microsoft Excel and SPSS licensed by Chulalongkorn University.

3.2.1 Plackett-Burman screening experiments

Many parameters (or factors) in CVD process could affect the quality of as-grown graphene. Screening experiment was conducted to eliminate insignificant factors and focus on important factors.

Plackett-Burman design was used in this screening process due to its high performance with fewer experiments. 7 factors as listed in Table 3.1 were selected. The high and low levels were determined by our preliminary study and previous literatures. The coded factors were summarized in Table 3.2.

Table 3.1 CVD Factors for screening experiments

Label	Factors	Low Level (-)	High Level (+)	Unit
X ₁	pre-growth composition (Ar/H ₂)	150/150	300/300	sccm/sccm
X ₂	annealing time	20	60	min
X ₃	growth temperature	950	1050	°C
X ₄	post-annealing composition (Ar/H ₂)	450/50	900/100	sccm/sccm
X ₅	acetylene flow rate	0.4	1.0	sccm
X ₆	growth time	10	20	min
X ₇	cooling time	5	10	°C/min

For each experiment, heating rate was set to 25 °C/min, while flow rate ratios of Ar to H₂ in pre-growth stage (heating and annealing steps) and post-annealing stage (growth and cooling steps) were set to 1:1 and 9:1, respectively. Once the temperature was below 500 °C, it was dropped uncontrollably under Ar atmosphere with a flow rate of 150 sccm. Each experiment was repeated thrice.

Table 3.2 Coded factors for Plackett-Burman screening experiments

Run	X ₁	X ₂	X ₃	X ₄	X ₅	X ₆	X ₇
1	-	-	-	-	-	-	-
2	-	-	+	+	+	+	-
3	-	-	-	+	+	+	+
4	+	-	-	-	+	+	+
5	+	+	-	-	-	+	+
6	+	+	+	-	-	-	+
7	+	+	+	+	-	-	-
8	-	+	+	+	+	-	-

3.2.2 Response surface methodology

After screening experiments, 3 significant factors were selected for studying their effects and interaction effects on dependent variables. Box-Behnken design was used in this section because the design required less number of experiments (15 runs) and instrument limitation. Each factor was divided into 3 levels (-1, 0 and +1). -1 and +1 represented low and high level, respectively, from the screening experiments whereas 0 represented middle point between low and high level. The coded parameters in each experiment were listed in Table 3.3. The other factors in the previous experiments were determined by considering their effects on the 6 dependent variables obtained from Raman spectra.

3.3 Transfer method of graphene

For Raman measurement, graphene on Cu growth substrate had to be transferred to a SiO₂/Si (285 or 300 nm SiO₂, B-doped Si) substrate by PMMA-mediated transfer method (Figure 3.5). First, graphene on Cu was attached to a glass slide by Scotch[®] tape in order to prevent PMMA coating on back side of copper. As-grown or modified graphene on copper was coated with 4 % w/w PMMA (purchased from Aldrich with average molecular weight ~996000 Da) in toluene by spin-coating at 500 rpm for 5 s followed by 3000 rpm for 30 s. The PMMA-coated graphene on Cu was baked in the oven at 100 °C for at least 30 min to remove remaining solvent. The coated copper was detached from glass slide and 4 edges of the coated substrate were then trimmed. In addition, graphene on back side of substrate was removed by diluted HNO₃ and Cu substrate was etched by floating on 1 M FeCl₃. The floating PMMA/graphene hybrid film was subsequently transferred to beakers filled with RO water, 0.25 % w/w HCl, and milli-Q water, respectively, in order to remove contaminated Fe³⁺ ions. The film was finally taken up on SiO₂/Si substrate. The water between PMMA/graphene film and SiO₂/Si substrate was removed by heating in an oven at 80 °C for at least 1 hr. PMMA was removed with acetone 5 times and isopropanol once. The wet graphene film on the new substrate was blow-dried by an Ar gun.

Table 3.3 Coded factors for 15 experiments of 3-level Box-Behnken design

Run	X	Y	Z
1	-1	0	+1
2	0	-1	+1
3	0	+1	+1
4	+1	0	+1
5	-1	-1	0
6	-1	+1	0
7	+1	+1	0
8	+1	-1	0
9	0	-1	-1
10	-1	0	-1
11	0	+1	-1
12	+1	0	-1
13	0	0	0
14	0	0	0
15	0	0	0

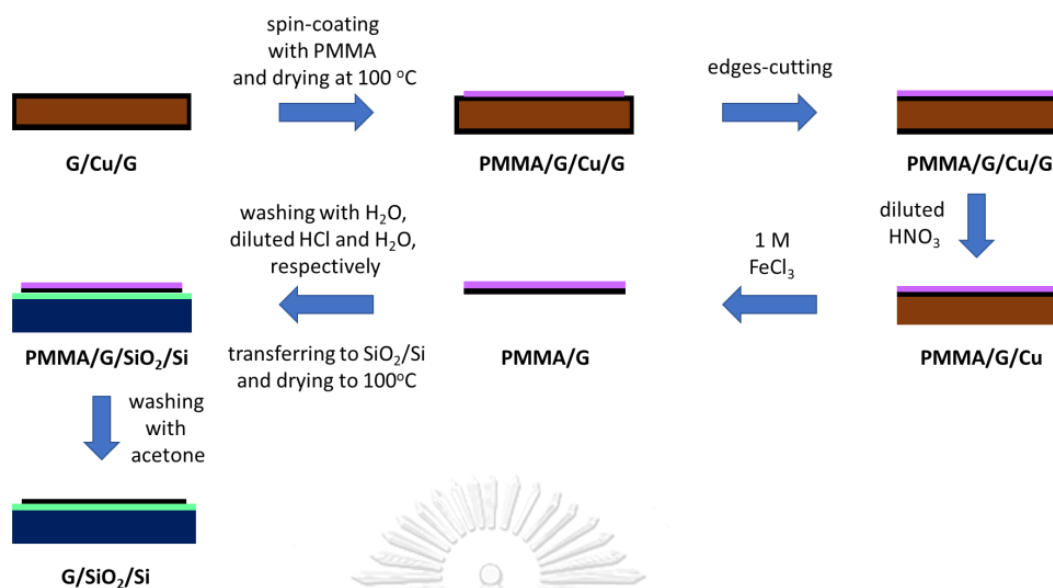


Figure 3.5 Schematic process of PMMA-mediated transfer method

3.4 Raman measurement

Raman spectra were recorded using DXR Raman microscope (Thermo scientific) equipped with a 532-nm excitation laser. All samples were analyzed under 100X-objective lens with laser spot size of 1 μm . The typical laser power was 10 mW with a pinhole aperture of 50 μm . The exposure time was set to 2 s with 2 accumulations. 250 spectra were measured on each sample.

3.5 Synthesis of 4-nitrobenzenediazonium tetrafluoroborate

4-Nitrobenzenediazonium tetrafluoroborate (4NBDT) was prepared from 4-nitroaniline (Figure 3.6). First, 4-nitroaniline (1.38 g, 10 mmol) was added into 50 % w/w HBF₄ solution (3.5 ml) and then, RO water (3.5 ml) was added into the mixture. Once the mixture became homogeneous, the solution was cooled down to 0 °C. The solution of NaNO₂ (0.695 g, 10.1 mmol) in water (1.5 ml) was added dropwise.

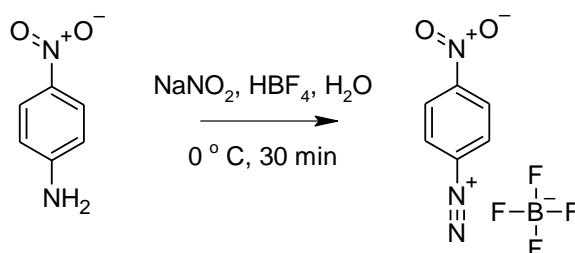


Figure 3.6 Synthesis of 4-nitrobenzenediazonium tetrafluoroborate (4NBDT) from 4-nitroaniline

The mixture was stirred for 30 min. The precipitate was filtrated through filter paper. The precipitate was washed with small amounts of cold methanol (MeOH) and copious amounts of diethylether (Et₂O). The product was recrystallized in acetone/Et₂O and then dried in vacuum. The diazonium salt was kept in refrigerator to prevent decomposition.

3.6 Functionalization of graphene

3.6.1 Conventional method

4-Nitrobenzenediazonium tetrafluoroborate (31 mg, 0.125 mmol) was dissolved in 5 ml of 1 % w/v SDS solution. The mixture was sonicated for 5 min. The diazonium solution was dropped into the 100-ml beaker which a magnetic bar and graphene on substrate (SiO₂/Si or Cu) were placed on the opposite sides. The solution was gently stirred at room temperature. After 7 hr of continuous stir, the solution was carefully removed by dropper. The functionalized graphene was washed thrice with Milli-Q water. Then, the film was soaked in Milli-Q water overnight in order to completely remove surfactant molecules from the surface. Finally, the film was dried by an Ar gun.

3.6.2 Photoinduced method



Figure 3.7 A fluorescence box installed with 2 blacklights on the top of the box

The same diazonium solution as in 3.6.1 was dropped into beaker in which graphene film was placed. Then the solution was irradiated inside a fluorescence box under blacklight (364 nm) for 30 min. The solution was poured out and the film was washed thrice with Milli-Q water and then soaked in Milli-Q water overnight. Finally, the film was dried by an Ar gun.

CHAPTER IV

RESULTS AND DISCUSSION

As aforementioned in chapter II, copper (Cu), nickel (Ni) and cobalt (Co) are promising substrates for graphene synthesis. At high temperature, carbon atoms can dissolve in these metals. Because Cu has lowest carbon solubility as stated in Table 4.1, self-limiting growth can be achieved easier than other metal. Hence, Cu is widely used as a substrate for graphene synthesis.

Table 4.1 Carbon solubility in nickel (Ni), cobalt (Co) and copper (Cu) [48]

Metal Substrate	% C atom	Temperature (°C)
Ni	0.19	600
Co	0.13	700
Cu	0.0007–0.0280	1000

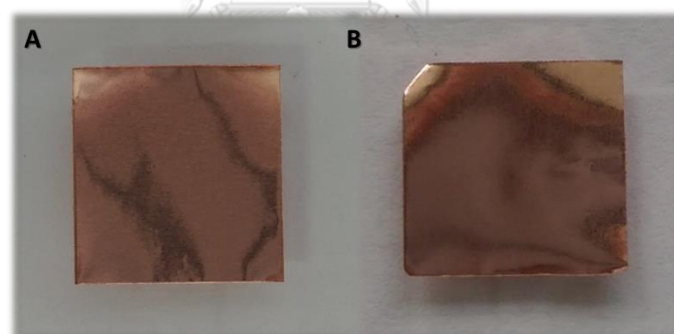


Figure 4.1 Photographs of 25- μm -thick Cu foils (A) before and (B) after graphene growth by CVD

Cu from Alfa Aesar is bright reddish brown as shown in Figure 4.1(A). After graphene growth there was no noticeable change when observing by naked eyes (Figure 4.1(B)). Optical micrographs of Cu surface revealed that, Cu surface became smoother and formed microscopic grains after CVD process (Figure 4.2). However, the quality of graphene formed could not be verified by those micrographs.

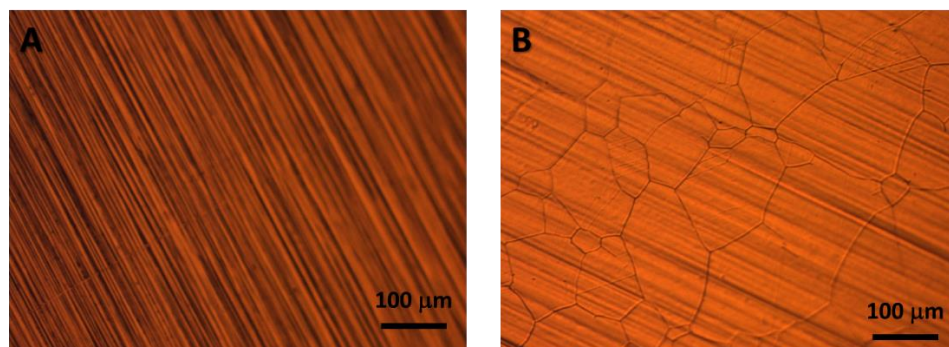


Figure 4.2 Optical micrographs of Cu foils (A) before and (B) after graphene growth by CVD

Due to the fact that Raman measurement of graphene on Cu gives low signal with strong fluorescent background, graphene needs to be transferred to SiO₂/Si substrate prior to the Raman measurement.

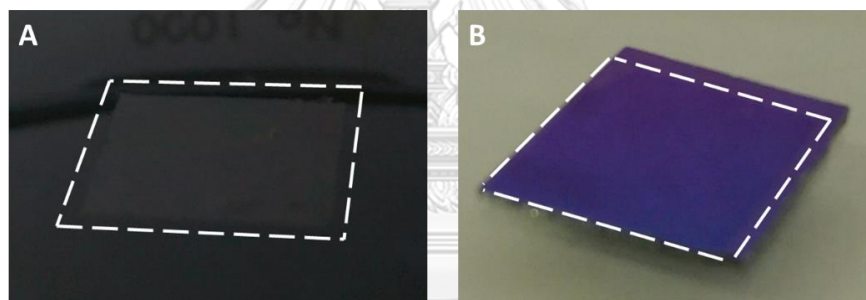


Figure 4.3 (A) PMMA/graphene hybrid film floating on 1 M FeCl₃ solution (B) graphene film transferred on SiO₂/Si substrate (285 nm SiO₂)

As described in chapter III, graphene was transferred by PMMA-mediated method. PMMA acts as a support for graphene after Cu-etching (Figure 4.3(A)). Finally, PMMA was removed by acetone after being taken up onto SiO₂/Si substrate. As illustrated in Figure 4.3(B), graphene can be clearly discriminated from SiO₂/Si substrate. On 285 nm SiO₂ wafer, graphene can be seen more easily than that on 300 nm SiO₂ wafer by naked eyes. However, graphene can be clearly observed on both wafers under microscope (Figure 4.4)

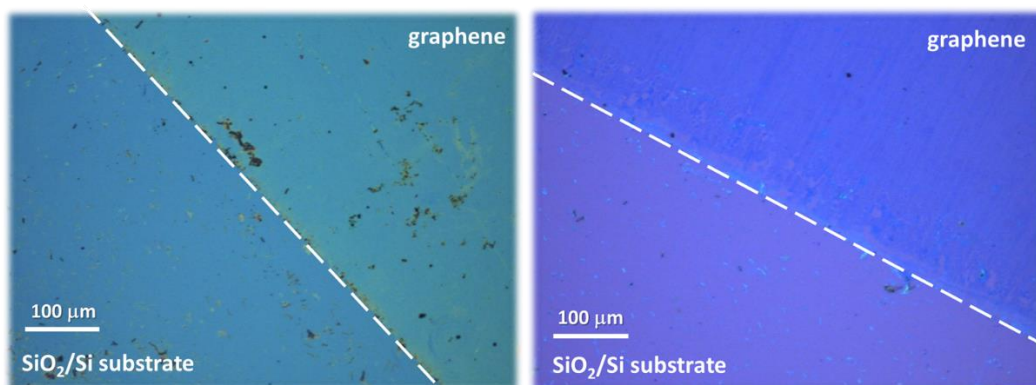


Figure 4.4 Graphene on (A) 300 nm and (B) 285 nm SiO₂ wafer

4.1 Plackett-Burman screening experiment

First, optimizing direction of 6 dependent variables, whether a higher or lower value is preferable, needed to be decided. I(D)/I(G) ratio indicating the amount of defect on graphene should be minimized since the quality of graphene strongly depends on this variable. On the contrary, I(2D)/I(G) ratio should be maximized because graphene with fewer layers is more desirable. Positions of G and 2D peaks are shifted by 2 reasons including increase in the number of layer and doping. As the number of layer increases, the red shift of G peak and the blue shift of 2D peak are observed. For doping, the position of G always goes red shift while the position of 2D can possibly go red or blue shift depending on the type of doping. Despite the fact that Ar, H₂ and C₂H₂ or combinations of them were continuously fed throughout the process, trace amount of air could possibly remain in the reactor. Therefore, as-grown graphene could probably be doped by N₂ or O₂. As a result, high quality pristine graphene usually shows G and 2D peak at higher and lower wavenumber, respectively, compared to our graphene. Moreover, 2D peak broadens as the number of layer increases while G peak broadens with increasing level of doping. Thus, the full width at half maximum of G and 2D peaks should be lessened. Favorable trends for each dependent variable are summarized in Table 4.2.

Table 4.2 Dependent variables and their optimizing directions

Dependent Variable	Optimizing Direction
I(D)/I(G)	Minimize
I(2D)/I(G)	Maximize
Pos(G)	Maximize
FWHM(G)	Minimize
Pos(2D)	Minimize
FWHM(2D)	Minimize

The experiments were conducted according to parameters in previous chapter. Each run was repeated thrice. Each sample was characterized by a confocal Raman spectrometer and 250 spectra were collected. The representative value of each parameter as shown in Table 4.3 was averaged from 750 spectra.

To decide which factors significantly influenced dependent variable, Lenth's Pseudo Standard Error (Lenth's PSE) was chosen because it could justify our decision without the need of a dummy factor (a factor without any effect on dependent variable). This method assumes that a factor with the least effect is the dummy.

Table 4.3 Summary of parameters from Raman spectra for Plackett-Burman experiments

Run	I(D)/I(G)			I(2D)/I(G)			Pos(G)			FWHM(G)			Pos(2D)			FWHM(2D)		
	\bar{x}	SD	n	\bar{x}	SD	n	\bar{x}	SD	n	\bar{x}	SD	n	\bar{x}	SD	n	\bar{x}	SD	n
1	0.94	0.17	3	0.41	0.05	3	1586.3	1.0	3	50.31	10.18	3	2696.9	0.3	3	86.22	4.39	3
2	0.27	0.05	3	1.05	0.10	3	1587.3	1.8	3	24.90	0.74	3	2691.7	1.5	3	56.88	1.02	3
3	1.05	0.04	3	1.05	0.04	3	1587.3	2.5	3	24.90	0.78	3	2691.7	2.0	3	56.88	0.13	3
4	1.10	0.07	3	0.30	0.07	3	1586.6	3.3	3	69.70	7.31	3	2694.2	5.2	3	98.67	4.55	3
5	1.03	0.05	3	0.44	0.05	3	1587.7	1.6	3	57.40	4.08	3	2694.7	1.8	3	88.28	2.28	3
6	0.19	0.04	3	0.99	0.28	3	1585.0	5.3	3	23.78	2.18	3	2691.0	6.3	3	56.79	3.79	3
7	0.21	0.07	3	1.21	0.05	3	1586.1	2.4	3	22.91	1.73	3	2689.3	2.6	3	52.90	3.96	3
8	0.28	0.05	3	0.94	0.15	3	1588.2	3.4	3	26.99	2.02	3	2690.5	2.4	3	59.15	6.52	3

As stated in Table 4.3, $I(D)/I(G)$ ratios range from 0.19 to 1.10. This indicates that inter-defect distances vary from 11.5 to 27.6 nm. While $I(2D)/I(G)$ ratios, interpreting the number of graphene layer, imply that only bilayer graphene up to multilayer graphene were produced under various growth conditions. The interpretation according to $I(2D)/I(G)$ ratios was also consistent with that according to $FWHM(2D)$ values, which can also be used to determine the thickness of graphene. When considering $Pos(G)$ values, all of samples synthesized under these conditions were not doped by trace amount of nitrogen remained in the reactor. This is in consistent with a previous report [74] in that $Pos(G)$ of N-doped graphene is lower than 1583 cm^{-1} . Interestingly, there is a remarkable relationship between $I(D)/I(G)$ ratios and $FWHM(G)$ values. In high defect samples, their G peaks are usually broadened. This suggests that as defect density increases, graphene becomes closer to amorphous carbon.

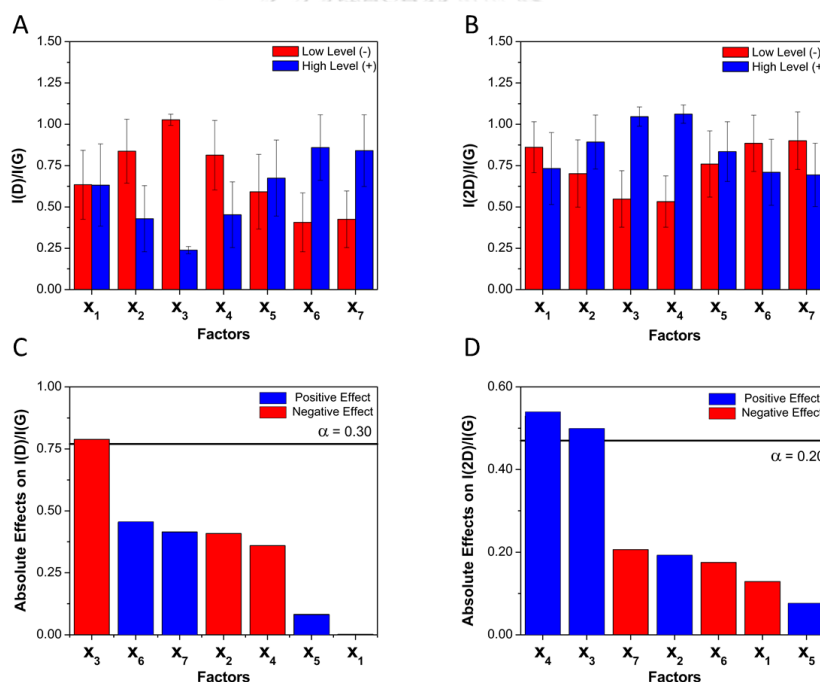


Figure 4.5 Bar charts showing average (A) $I(D)/I(G)$ and (B) $I(2D)/I(G)$ ratios at high and low levels of independent variables and Pareto charts of absolute effect of independent variables on (C) $I(D)/I(G)$ and (D) $I(2D)/I(G)$ values

After acquiring data from Raman spectra, they were processed as described in the literature [75]. As shown in Figure 4.5(A), annealing time (X_2), growth temperature (X_3) and Ar-H₂ composition during post-annealing steps (X_4) showed negative effect on I(D)/I(G). On the other hand, growth time (X_5) and cooling rate (X_6) positively affected I(D)/I(G). While Ar-H₂ composition before growth (X_1) and acetylene flow rate (X_5) had no much effect on I(D)/I(G). The Pareto chart in Figure 4.5(C) showed sorted absolute effect on I(D)/I(G). The most effective factor was growth temperature (X_3) followed by growth time (X_6), cooling rate (X_7), annealing time (X_2), Ar-H₂ composition during post-annealing steps (X_4), acetylene flow rate (X_5) and Ar-H₂ composition before growth (X_1), respectively. At 70% level of confidence, growth temperature (X_3) was the only significant factor affecting I(D)/I(G).

As seen in Figure 4.5(B), I(2D)/I(G) could be enhanced by increasing annealing time (X_2), growth temperature (X_3) and Ar-H₂ composition during post-annealing steps (X_4). On the contrary, increasing cooling rate (X_7), growth time (X_6) and Ar-H₂ composition before growth (X_1) resulted in decrease of I(2D)/I(G), while acetylene flow rate (X_5) did not make any noticeable change in I(2D)/I(G). We also found that growth temperature (X_3) and Ar-H₂ composition during post-annealing steps (X_4) had much higher impact on I(2D)/I(G) than the other affecting factors had at 80% level of confidence (Figure 4.5(D)).

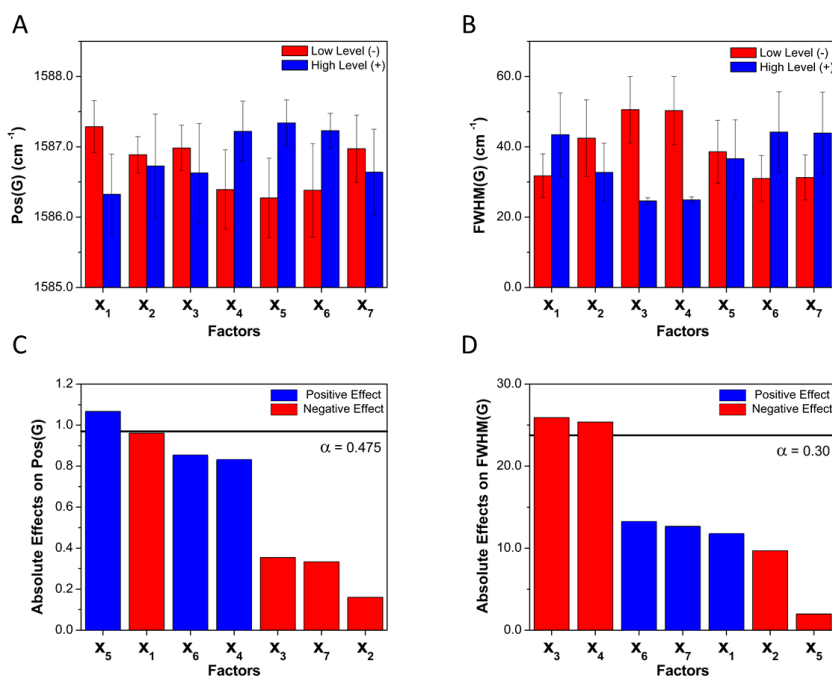


Figure 4.6 Bar charts showing average (A) Pos(G) and (B) FWHM(G) values at high and low levels of independent variables and Pareto charts of absolute effect of independent variables on (C) Pos(G) and (D) FWHM(G) values

Figure 4.6(A) shows that the growth temperature (X₃), cooling rate (X₇) and annealing time (X₂) were clearly insignificant to Pos(G). While acetylene flow rate (X₅), Ar-H₂ composition before growth (X₁), growth time (X₆) and Ar-H₂ composition during post-annealing step (X₄) showed small effects on Pos(G). Nonetheless, only acetylene flow rate (X₅) was predicted to be significant at 52.5% level of confidence (Figure 4.6(C)). However, effects of all factors on Pos(G) were less than the resolution of Raman spectrometer (7 cm⁻¹). Therefore, the results were not valid.

FWHM(G), suggesting doping level on graphene, was negatively affected by annealing time (X₂), growth temperature (X₃) and Ar-H₂ composition during post-annealing steps (X₄) as shown in Figure 4.6(B). On the contrary, Ar-H₂ composition before growth (X₁), growth time (X₆) and cooling rate (X₇) had positive influences on FWHM(G), whereas acetylene flow rate (X₅) did not show any significant effect on FWHM(G). Comparing absolute effects of independent variables, growth temperature

(X_3) and Ar- H_2 composition during post-annealing steps (X_4) were anticipated as significant factors at 70% level of confidence as shown in Figure 4.6(D). Growth time (X_6), cooling rate, (G) Ar- H_2 composition before growth (X_1) and annealing time (X_2) had similar absolute effects.

Information on peak 2D is very useful for indicating the thickness of graphene. $I(2D)/I(G)$ ratio is usually considered first because it is the factor that can be easily observed. Still, $I(2D)/I(G)$ may be remarkably disturbed by charge particles adsorbed on graphene. Therefore, many researchers commonly scrutinize Pos(2D) as well as FWHM(2D) due to their less charge effects.

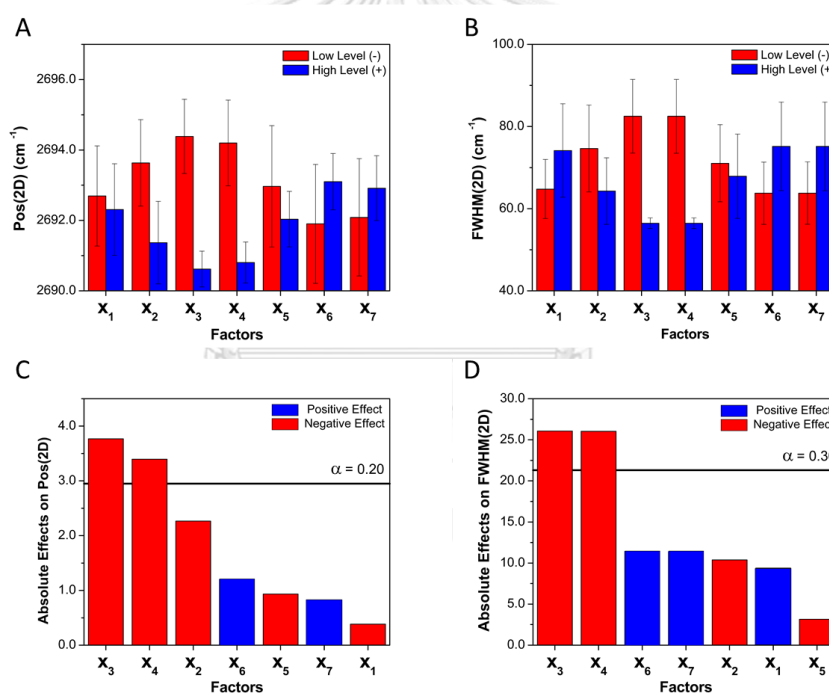


Figure 4.7 Bar charts showing average (A) Pos(2D) and (B) FWHM(2D) values at high and low levels of independent variables and Pareto charts of absolute effect of independent variables on (C) Pos(2D) and (D) FWHM(2D) values

As demonstrated in Figure 4.7, only growth time (X_6) and cooling rate (X_7) had positive effect on Pos(2D). Meanwhile, the other factors negatively affected Pos(2D). However, the effective factors for tuning Pos(2D) were growth temperature (X_3) and

Ar-H₂ composition during post-annealing steps (X₄) at 80% level of confidence. As well as Pos(2D), growth temperature (X₃) and Ar-H₂ composition during post-annealing steps (X₄) were considered as significant factors for reducing FWHM(2D). In addition to Pos(G), Pos(2D) showed variations which were less than the resolution of Raman spectrometer. Hence, this information was also not reliable to describe the variation of Pos(2D).

According to the above studies as summarized in Table 4.4, the factors having significant on various dependent variables were growth temperature (X₃) and Ar-H₂ composition during post-annealing steps (X₄). Both of them played important roles in tuning I(2D)/I(G), FWHM(G), Pos(2D) and FWHM(2D). In addition, growth temperature (X₃) took an important part in decreasing defect density estimated by I(D)/I(G). Therefore, growth temperature (X₃) and Ar-H₂ composition during post-annealing steps (X₄) were selected for further study based on response surface methodology.

Nevertheless, optimization of Pos(G) was still unclear. The most effective parameter was acetylene flow rate (X₅), which was significant at 52.5% level of confidence. This low level of confidence meant it was possible that none of the factors studied here was a significant factor affecting Pos(G). Still, acetylene flow rate is one of the crucial factors in graphene synthesis, because it controls nucleation rate and growth rate [47]. In addition, there was a report [42] that studied the effect of acetylene flow rate on quality of as-grown graphene. Thus, acetylene flow rate (X₅) was also selected for further study using response surface methodology.

Table 4.4 Summary of optimum direction of independent factors to achieve optimizing direction of dependent variables (P = Positive, N = Negative)

Dependent variable	Optimizing direction	Optimum direction						
		X ₁	X ₂	X ₃	X ₄	X ₅	X ₆	X ₇
I(D)/I(G)	N	P	P	P	P	N	N	N
I(2D)/I(G)	P	N	P	P	P	P	N	N
FWHM(G)	N	N	P	P	P	P	N	N
FWHM(2D)	N	N	P	P	P	P	N	N

Apart from selecting three factors for constructing response surfaces, the remaining factors that would not be included in the study needed to be fixed by considering optimum direction, as in Table 4.4. Firstly, Ar-H₂ composition before growth (X₁) was fixed to 150/150 sccm/sccm because setting the factor at low level satisfied most of dependent variables except I(D)/I(G). Furthermore, using such flow rates could reduce gas cost. In addition, low total flow rates could reduce retention time which makes cleaning process more efficient. Secondly, annealing time (X₂) was set at 60 min. The results were in consistent with previous reports in that extension of annealing time could improve the quality of graphene [76-78]. Prolonging annealing time could afford cleaner reconstructed surface. Thirdly, growth time (X₆) was specified to 10 min as reported by Chinese research team [42]. Normally, CVD process using acetylene is not self-limited, due to high pyrolysis rate of acetylene, so long growth time could directly affect graphene thickness. Finally, cooling rate was set at 5 °C, to satisfy all dependent variables except Pos(2D). This results is in consistent with previous report [15] that low cooling rate could reduce both defect density and thickness of graphene. As temperature drops, carbon atoms could probably form 5-, 6-, 7- or 8-membered rings. However, only 6-membered ring is thermodynamically favored. Other ring types could form but they are not stable. Slow cooling provides more time for rearrangement and more energy to overcome energy barrier to form stable graphene.

Noticeably, for these experiments, high value of α was used to determine whether or not each factor was significant. Generally, α was set at 0.05 to 0.10, which corresponded to 90 to 95% level of confidence. This also meant that there was high possibility that type I error could occur. This might be because the dependent variables were quite sensitive to those independent factors and dummy factors were not added in the experiments. To guarantee correctness of collected data, a great number of Raman spectra had to be recorded and interpreted.

4.2 Response surface methodology

The results from screening experiments suggested that growth temperature (X_3) and Ar-H₂ composition during post-annealing steps (X_4) be used for response surface modeling.

According to previous report [42], growth temperature (X_3) was already known as the main factor controlling the quality of graphene. For Ar-H₂ flow rates (X_4), they did not only dilute acetylene gas but also controlled residence time of acetylene gas. Last but not least, acetylene flow rate (X_5) also directly affected nucleation and growth rates. Thus, acetylene flow rate (X_5) was selected as another factor for response surface modeling.

Normally, highest growth temperature of graphene synthesized via CVD process from methane is 1050 °C. However, the other processes operate below 1000 °C. Also, some researchers anticipated that CVD graphene synthesized from acetylene could perform at lower temperature. Thus, the effect of growth temperature (X_3) in our experiments was observed between 950 to 1050 °C

Furthermore, the total flow rates of input gases in previous reports [10, 19, 42] were very different depending on operating pressure. At atmospheric pressure, Ar-H₂ total flow rates of 1000 sccm were confirmed to produce low defect bilayer graphene. Nonetheless, H₂/Ar flow rate ratio must be 0.010-0.111. Hence, the flow rate ratio of H₂/Ar in experiments was kept at 0.111 and total flow rates were investigated.

As stated earlier that acetylene, as carbon precursor, has high pyrolysis rate, so small amounts of acetylene were needed in the process. The lowest flow rate of acetylene fed into the reactor was 0.24 sccm under low pressure. While at atmospheric pressure, it was a little bit higher at 1.0 sccm. Therefore, in these experiments, acetylene flow rate (X_5) was varied lower than previous reports [42, 47].

Table 4.5 Three factors for response surface modelling (RSM)

Factor	Unit	Level (-1)	Level (0)	Level (+1)
C	°C	950	1000	1050
D	sccm/sccm	450/50	675/75	900/100
E	sccm	0.4	0.7	1.0

Box-Behnken design was employed in this study. Although this model does not cover all over spaces, the required number of experiments for this design is less than that of other designs such as central composite design (CCD) or full factorial design. Also, high level of temperature expected to be a main factor for controlling the quality of graphene was limited due to furnace specification.

According to Box-Behnken design, three levels of each factor had to be specified. Thus, middle point between high and low level in Plackett-Burman design was set as Level (0) as shown in Table 4.5.

15 experiments were conducted with variable parameters as specified in Table 4.6 and fixed parameters as discussed in previous section. The as-grown graphene on Cu was transferred to SiO₂/Si before Raman measurement. 250 Raman spectra were collected for each run and 6 Raman parameters were extracted for modeling as have been done in the screening experiments.

Table 4.6 Coded and actual parameters in Box-Behnken Design

Run	Coded Value			Actual Value		
	C	D	E	C (°C)	D (sccm/sccm)	E (sccm)
1	-1	0	+1	950	675/75	1.0
2	0	-1	+1	1000	450/50	1.0
3	0	+1	+1	1000	900/100	1.0
4	+1	0	+1	1050	675/75	1.0
5	-1	-1	0	950	450/50	0.7
6	-1	+1	0	950	900/100	0.7
7	+1	+1	0	1050	900/100	0.7
8	+1	-1	0	1050	450/50	0.7
9	0	-1	-1	1000	450/50	0.4
10	-1	0	-1	950	675/75	0.4
11	0	+1	-1	1000	900/100	0.4
12	+1	0	-1	1050	675/75	0.4
13	0	0	0	1000	675/75	0.7
14	0	0	0	1000	675/75	0.7
15	0	0	0	1000	675/75	0.7

As shown in Table 4.7, the quality of graphene varied from medium to low. The best sample was from Run 12 with I(D)/I(G) of 0.28, I(2D)/I(G) of 1.45, Pos(G) of 1587.42 cm^{-1} , FWHM(G) of 23.21 cm^{-1} , Pos(2D) of 2687.85 cm^{-1} and FWHM(2D) of 47.40 cm^{-1} . Even though, defect density of the best sample in these experiments was greater than that of the best sample from Plackett-Burman screening experiments, the thickness of graphene from Run 12 as indicated by I(2D)/I(G) and FWHM(2D) were much better than that from the screening experiments. Unfortunately, ranges of Pos(G) and Pos(2D) were less than the resolution of Raman spectrometer. Consequently, models developed from these data for predicting Pos(G) and Pos(2D) were not reliable.

Table 4.7 Summary of Raman parameters from experiments of Box-Behnken Design

Run	I(D)/I(G)		I(2D)/I(G)		Pos(G)		FWHM(G)		Pos(2D)		FWHM(2D)	
	\bar{X}	SD	\bar{X}	SD	\bar{X}	SD	\bar{X}	SD	\bar{X}	SD	\bar{X}	SD
1	1.05	0.13	0.35	0.07	1586.45	2.41	60.93	14.35	2696.33	4.20	89.39	7.58
2	0.94	0.33	0.46	0.12	1587.05	2.25	47.56	15.36	2690.67	3.50	74.08	9.77
3	0.72	0.28	0.70	0.15	1586.66	2.38	39.29	13.42	2693.71	3.65	71.00	7.77
4	0.41	0.33	0.83	0.27	1586.44	2.34	32.83	9.36	2693.47	5.19	66.50	6.56
5	1.06	0.14	0.37	0.07	1582.38	2.37	62.07	13.70	2691.41	3.75	91.60	7.14
6	1.09	0.16	0.41	0.07	1582.90	2.07	56.94	13.95	2691.05	4.25	87.24	6.17
7	0.34	0.27	1.19	0.36	1587.34	2.73	26.07	4.27	2688.60	4.07	50.68	5.13
8	0.74	0.49	1.25	0.45	1584.57	1.98	34.57	12.31	2683.88	5.37	57.25	7.92
9	0.82	0.34	0.74	0.20	1585.67	2.57	41.44	15.74	2689.05	3.93	69.80	9.42
10	1.04	0.17	0.42	0.09	1584.97	2.44	55.00	14.57	2693.53	4.43	85.72	7.41
11	0.64	0.31	0.91	0.25	1585.62	2.11	36.49	11.86	2686.54	4.34	64.67	8.14
12	0.28	0.29	1.45	0.57	1587.42	2.13	23.21	3.77	2687.85	3.94	47.40	5.28
13	0.86	0.32	0.76	0.20	1585.73	2.60	44.45	14.78	2687.33	4.58	72.06	9.08
14	0.91	0.30	0.71	0.18	1584.86	2.26	44.98	14.34	2687.19	4.60	72.05	7.44
15	0.94	0.38	0.66	0.19	1584.86	2.78	47.03	17.19	2688.72	4.52	74.26	10.01

The average parameters including I(D)/I(G), I(2D)/I(G), FWHM(G) and FWHM(2D) were then analyzed by multiple linear regression (MLR) using SPSS software provided by Chulalongkorn University. The full quadratic model was selected for every dependent variable because linear terms, interaction terms and quadratic terms were included. The calculated regression coefficients were summarized in Table 4.8.

Coefficient of determination was calculated for investigating fitness of model. All models had high coefficient of determination up to 0.97. This implied that the mathematical model could predict dependent variables well.

In addition to coefficient of determination, analysis of variance (ANOVA) and lack-of-fit test were performed to observe degree of fit. ANOVA results were shown in Table 4.9. P-values of the corrected models of I(D)/I(G), I(2D)/I(G), FWHM(G) and FWHM(2D) were less than 0.05 indicating that those models were significant and well-described by the factors.

Table 4.8 Summary of coefficients calculated from multiple linear regression (MLR)

Independent variable	Coefficients			
	I(D)/I(G)	I(2D)/I(G)	FWHM(G)	FWHM(2D)
Intercept	0.903	0.708	45.485	72.792
X_3	-0.31	0.395	-14.782	-16.515
X_4	-0.095	0.049	-3.356	-2.391
X_5	0.044	-0.146	3.059	4.172
$X_3 * X_4$	-0.107	-0.025	-0.842	-0.551
$X_3 * X_5$	0.03	-0.135	0.92	3.86
$X_4 * X_5$	-0.01	0.019	-0.831	0.511
X_3^2	-0.09	0.08	0.613	0.634
X_4^2	-0.006	0.016	-1.183	-1.733
X_5^2	-0.117	-0.024	-3.104	-1.174

Then, the effect of each term was considered. Growth temperature (X_3) apparently affected all Raman parameters as p-value was less than 0.05. While Ar-H₂ flow rates (X_4) had influences only on I(D)/I(G), FWHM(G) and FWHM(2D). Interestingly, acetylene flow rate (X_5), which seemed to have no effect on any dependent variable according to the screening experiments, showed effects on I(2D)/I(G) and Pos(2D).

Three factors (X_3 , X_4 and X_5) have three interaction terms including X_3X_4 , X_3X_5 and X_4X_5 . The interaction between growth temperature (X_3) and Ar-H₂ flow rates (X_4) played important roles in I(D)/I(G) and Pos(2D) whereas the interaction between growth temperature (X_3) and acetylene flow rate (X_5) affected I(2D)/I(G) and FWHM(2D).

For quadratic terms (X_3^2 , X_4^2 and X_5^2), they can suggest non-linear relationships. In models of I(2D)/I(G) and FWHM(2D), the quadratic terms were not required for good

fit. On the contrary, the quadratic terms of growth temperature (X_3^2) and acetylene flow rate (X_5^2) showed significant roles on I(D)/I(G) models. Meanwhile, only quadratic term of acetylene flow rate (X_5^2) significantly described changes in FWHM(G).

The results from ANOVA, as stated in Table 4.9, suggested that each model needed different sets of terms to describe the dependent variable, even though coefficient of determination were close to 1 and there were no lack-of-fit in every model, as shown in Table 4.10. This might indicate that our current models were overfitted. So, leave-one-out cross validation (LOOCV) was carried out to observe the overfit. As illustrated in Figure 4.8, there were variations in every model.

Due to unnecessary terms in the full quadratic models, new models were developed by removing terms with p-value higher than 0.05 according to ANOVA results.

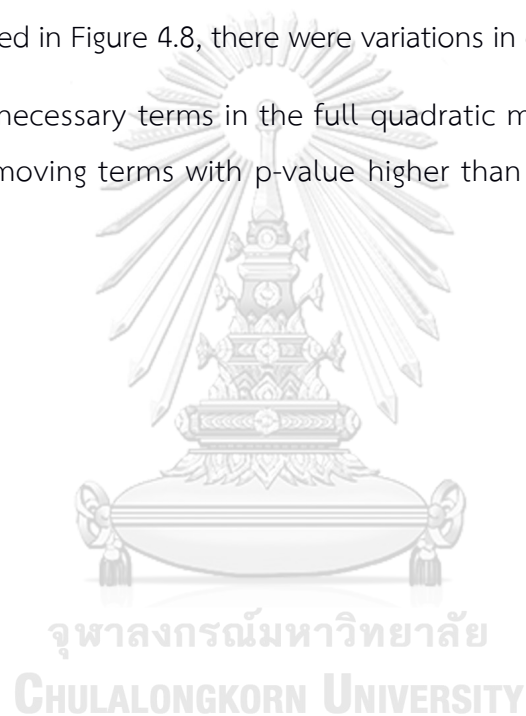


Table 4.9 ANOVA of full quadratic models for I(D)/I(G), I(2D)/I(G), FWHM(G), and FWHM(2D)

Source	Sum of Squares	df	Mean Square	F	p-value	
Corrected Model	I(D)/I(G)	.981 ^a	9	0.109	29.861	0.001
	I(2D)/I(G)	1.543 ^b	9	0.171	24.636	0.001
	FWHM(G)	1964.070 ^d	9	218.23	114.668	0
	FWHM(2D)	2446.634 ^f	9	271.848	39.505	0
Intercept	I(D)/I(G)	2.445	1	2.445	669.523	0
	I(2D)/I(G)	1.505	1	1.505	216.217	0
	FWHM(G)	6206.602	1	6206.602	3261.219	0
	FWHM(2D)	15896.02	1	15896.02	2309.986	0
X ₁	I(D)/I(G)	0.769	1	0.769	210.661	0
	I(2D)/I(G)	1.248	1	1.248	179.299	0
	FWHM(G)	1748.104	1	1748.104	918.53	0
	FWHM(2D)	2182.075	1	2182.075	317.096	0
X ₂	I(D)/I(G)	0.072	1	0.072	19.784	0.007
	I(2D)/I(G)	0.019	1	0.019	2.792	0.156
	FWHM(G)	90.102	1	90.102	47.343	0.001
	FWHM(2D)	45.729	1	45.729	6.645	0.05

Table 4.9 ANOVA of full quadratic models for I(D)/I(G), I(2D)/I(G), FWHM(G), and FWHM(2D) (continued)

Source		Sum of Squares	df	Mean Square	F	p-value
X_5	I(D)/I(G)	0.015	1	0.015	4.187	0.096
	I(2D)/I(G)	0.171	1	0.171	24.601	0.004
	FWHM(G)	74.836	1	74.836	39.322	0.002
	FWHM(2D)	139.228	1	139.228	20.232	0.006
$X_3 * X_4$	I(D)/I(G)	0.045	1	0.045	12.454	0.017
	I(2D)/I(G)	0.002	1	0.002	0.349	0.58
	FWHM(G)	2.834	1	2.834	1.489	0.277
	FWHM(2D)	1.213	1	1.213	0.176	0.692
$X_3 * X_5$	I(D)/I(G)	0.004	1	0.004	0.993	0.365
	I(2D)/I(G)	0.073	1	0.073	10.528	0.023
	FWHM(G)	3.384	1	3.384	1.778	0.24
	FWHM(2D)	59.593	1	59.593	8.66	0.032
$X_4 * X_5$	I(D)/I(G)	0	1	0	0.118	0.746
	I(2D)/I(G)	0.001	1	0.001	0.198	0.675
	FWHM(G)	2.763	1	2.763	1.452	0.282
	FWHM(2D)	1.045	1	1.045	0.152	0.713

Table 4.9 ANOVA of full quadratic models for I(D)/I(G), I(2D)/I(G), FWHM(G), and FWHM(2D) (continued)

Source		Sum of Squares	df	Mean Square	F	p-value
X_3^2	I(D)/I(G)	0.03	1	0.03	8.16	0.036
	I(2D)/I(G)	0.024	1	0.024	3.401	0.124
	FWHM(G)	1.387	1	1.387	0.729	0.432
	FWHM(2D)	1.486	1	1.486	0.216	0.662
X_4^2	I(D)/I(G)	0	1	0	0.037	0.856
	I(2D)/I(G)	0.001	1	0.001	0.131	0.732
	FWHM(G)	5.164	1	5.164	2.714	0.16
	FWHM(2D)	11.086	1	11.086	1.611	0.26
X_5^2	I(D)/I(G)	0.05	1	0.05	13.829	0.014
	I(2D)/I(G)	0.002	1	0.002	0.303	0.606
	FWHM(G)	35.584	1	35.584	18.697	0.008
	FWHM(2D)	5.085	1	5.085	0.739	0.429
Error	I(D)/I(G)	0.018	5	0.004		
	I(2D)/I(G)	0.035	5	0.007		
	FWHM(G)	9.516	5	1.903		
	FWHM(2D)	34.407	5	6.881		

Table 4.10 Lack-of-fit analysis of I(D)/I(G), I(2D)/I(G), FWHM(G), and FWHM(2D) quadratic models

Dependent Variable		Sum of Squares	df	Mean Square	F	p-value
I(D)/I(G)	Lack of Fit	.015	3	.005	3.566	.227
	Pure Error	.003	2	.001		
I(2D)/I(G)	Lack of Fit	.030	3	.010	4.056	.204
	Pure Error	.005	2	.002		
FWHM(G)	Lack of Fit	5.788	3	1.929	1.035	.526
	Pure Error	3.728	2	1.864		
FWHM(2D)	Lack of Fit	31.168	3	10.389	6.415	.138
	Pure Error	3.239	2	1.620		

The coefficients of reduced models were shown in Table 4.11. The coefficients had slight variations from the full model, but the sign of each term remained the same. Because Box-Behnken design is near rotatable. Although terms were added or removed, the coefficients would not be significantly varied.

ANOVA was performed to observe the significances of models and each term (Table 4.12 – 4.15). After removing inactive terms, all models still remained significant. Coefficient of determination of every reduced model was lower than those of full models. Nevertheless, all coefficients of determination were greater than 0.90. Moreover, no lack-of-fit was found for every model.

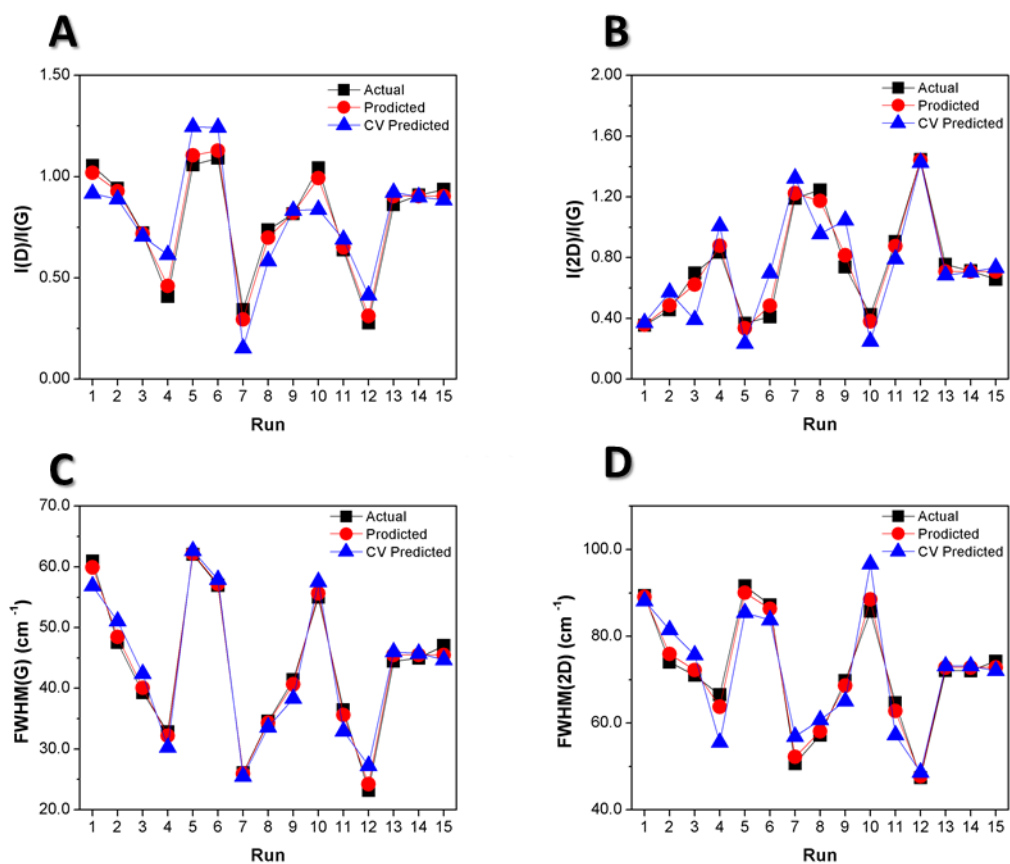


Figure 4.8 Leave-one-out cross validation of (A) $I(D)/I(G)$, (B) $I(2D)/I(G)$, (C) FWHM(G) and (D) FWHM(2D) models

Table 4.11 Summary of coefficients calculated from multiple linear regression (MLR) after removing unnecessary terms

Variables	Coefficients			
	I(D)/I(G)	I(2D)/I(G)	FWHM(G)	FWHM(2D)
Constant	0.899	0.747	45.159	71.58
X_3	-0.31	0.395	-14.782	-16.515
X_4	-0.095	-	-3.356	-2.391
X_5	-	-0.146	3.059	4.172
$X_3 * X_4$	-0.107	-	-	-
$X_3 * X_5$	-	-0.135	-	3.86
$X_4 * X_5$	-	-	-	-
X_3^2	-0.089	-	-	-
X_4^2	-	-	-	-
X_5^2	-0.116	-	-3.064	-

The remaining terms in most models significantly influenced the dependent variables. The cross validations of reduced models are shown in Figure 4.9. The predicted values became closer to the actual values. LOOCV-RMSE would be utilized to observe predictive ability of the models.

By comparing LOOCV-RMSE of full quadratic models to reduced models, the RMSE became smaller. This indicated that the predictive ability increased after removing unnecessary terms.

Table 4.12 ANOVA and Lack-of-fit analysis for reduced model of I(D)/I(G)

Source	Type III Sum of Squares	df	Mean Square	F	p-value
Corrected Model	.962 ^a	5	.192	45.881	.000
Intercept	3.502	1	3.502	835.318	.000
X ₃	.769	1	.769	183.462	.000
X ₄	.072	1	.072	17.230	.002
X ₃ *X ₄	.045	1	.045	10.846	.009
X ₃ ²	.030	1	.030	7.076	.026
X ₅ ²	.050	1	.050	12.019	.007
Error	.038	9	.004		
Total	10.343	15			
Corrected Total	1.000	14			
Lack of Fit	.015	3	.005	1.307	.356
Pure Error	.023	6	.004		

a. R Squared = .962

Table 4.13 ANOVA and Lack-of-fit analysis for reduced model of I(2D)/I(G)

Source	Type III Sum of Squares	df	Mean Square	F	p-value
Corrected Model	1.492 ^a	3	.497	63.939	.000
Intercept	8.360	1	8.360	1074.643	.000
X ₃	1.248	1	1.248	160.393	.000
X ₅	.171	1	.171	22.007	.001
X ₃ *X ₅	.073	1	.073	9.418	.011
Error	.086	11	.008		
Total	9.938	15			
Corrected Total	1.578	14			
Lack of Fit	.035	5	.007	.841	.566
Pure Error	.050	6	.008		

a. R Squared = .946

Table 4.14 ANOVA and Lack-of-fit analysis for reduced model of FWHM(G)

Source	Type III Sum of Squares	df	Mean Square	F	p-value
Corrected Model	1948.083 ^a	4	487.021	190.974	.000
Intercept	14275.467	1	14275.467	5597.790	.000
X ₃	1748.104	1	1748.104	685.478	.000
X ₄	90.102	1	90.102	35.331	.000
X ₅	74.836	1	74.836	29.345	.000
X ₅ ²	35.042	1	35.042	13.741	.004
Error	25.502	10	2.550		
Total	30390.261	15			
Corrected Total	1973.585	14			
Lack of Fit	21.774	8	2.722	1.460	.469
Pure Error	3.728	2	1.864		

a. R Squared = .987

Table 4.15 ANOVA and Lack-of-fit analysis for reduced model of FWHM(2D)

Source	Type III Sum of Squares	df	Mean Square	F	p-value
Corrected Model	2426.625 ^a	4	606.656	111.485	.000
Intercept	76856.147	1	76856.147	14123.813	.000
X ₃	2182.075	1	2182.075	400.999	.000
X ₄	45.729	1	45.729	8.404	.016
X ₅	139.228	1	139.228	25.586	.000
X ₃ *X ₅	59.593	1	59.593	10.951	.008
Error	54.416	10	5.442		
Total	79337.189	15			
Corrected Total	2481.041	14			
Lack of Fit	51.177	8	6.397	3.950	.218
Pure Error	3.239	2	1.620		

a. R Squared = .978

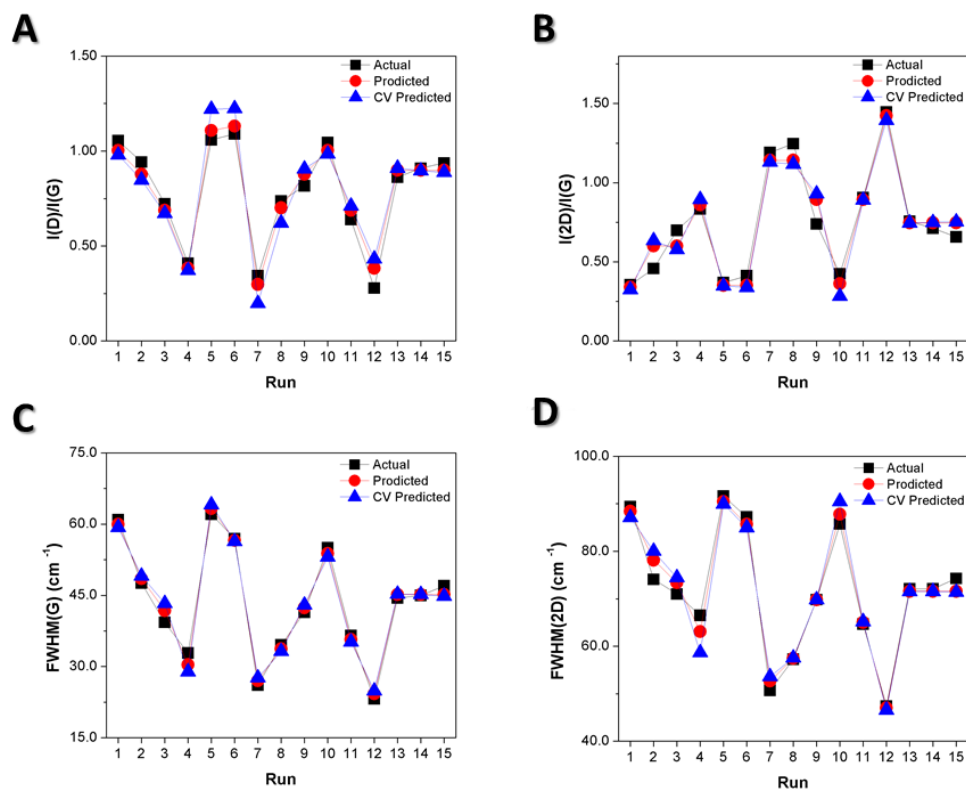


Figure 4.9 Leave-one-out cross validation of (A) $I(D)/I(G)$, (B) $I(2D)/I(G)$, (C) $FWHM(G)$ and (D) $FWHM(2D)$ reduced models

Table 4.16 Comparison of root-mean-square error (RMSE) of leave-one-out cross validation (LOOCV), coefficient of determination

Model	LOOCV-RMSE		Coefficient of determination	
	Full	Reduced	Full	Reduced
I(D)/I(G)	0.130	0.097	0.982	0.962
I(2D)/I(G)	0.181	0.099	0.978	0.946
FWHM(G)	2.595	2.015	0.995	0.987
FWHM(2D)	5.808	3.295	0.986	0.978

In summary, the reduced models were good tools for approximating I(D)/I(G), I(2D)/I(G), FWHM(G) and FWHM(2D).

After obtaining the mathematical models, contour plots (Figure 4.10 – 4.14) were constructed to observe effect of each parameter. In models of I(2D)/I(G), only growth temperature (X_3) and acetylene flow rate (X_5) were considered while all three parameters were considered in the other models. Furthermore, the parameters would be optimized to achieve highest quality graphene under investigating space.

Obviously, growth temperature (X_3) was the most effective factor to control I(D)/I(G) as seen in Figure 4.10(A) and 4.10(B). This is in consistent with previous report [42]. I(D)/I(G) gradually increased as temperature dropped. While Ar-H₂ flow rates (X_4) had small negative effect on I(D)/I(G), however, the interaction between growth temperature (X_3) and Ar-H₂ flow (D) had more effect than only Ar-H₂ flow rates (X_4). At high Ar-H₂ flow rate (X_4) and high temperature (X_3), I(D)/I(G) reached minimum within investigating space.

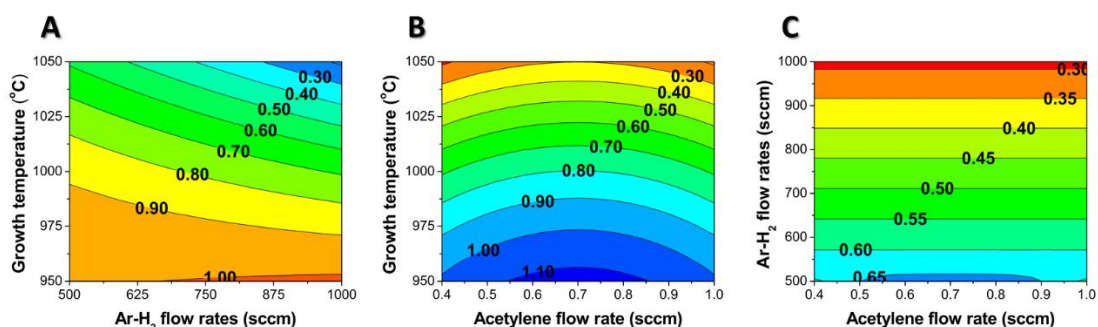


Figure 4.10 Response surface plots of $I(D)/I(G)$ for (A) growth temperature and Ar-H₂ flow rates, (B) growth temperature and acetylene flow rate and (C) Ar-H₂ flow rates and acetylene flow rate

Although the linear term of acetylene flow rate (X_5) was not considered as active component in the model, the quadratic term of acetylene flow rate (X_5^2) was an important element for better approximation. Interestingly, both high and low acetylene flow rates could afford graphene with low defect. On the other hand, medium acetylene flow rate (X_5) could increase $I(D)/I(G)$. The interaction between acetylene flow rate (X_5) and Ar-H₂ flow rates (X_4) was suppressed at high Ar-H₂ flow rates (X_4) and parabolic change was observed at very low Ar-H₂ flow rates (X_4) (Figure 4.10(C)).

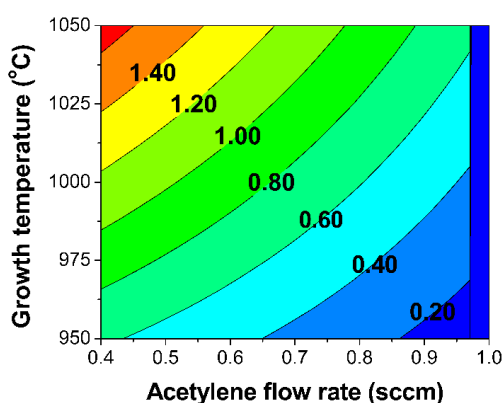


Figure 4.11 Response surface plots of $I(2D)/I(G)$ for growth temperature and acetylene flow rate

Model of $I(2D)/I(G)$ concerned only to growth temperature (X_3) and acetylene flow rate (X_5). Figure 4.11(A) shows that growth temperature (X_3) was still a dominant factor. Interestingly, $I(2D)/I(G)$ rapidly changed with acetylene flow rate (X_5) when it was greater than 0.95 sccm but the change became gradual when acetylene flow rate (X_5) was lower than 0.975 sccm. Such observation was more obvious at higher growth temperature (X_3). The maximum $I(2D)/I(G)$ was observed at lowest acetylene flow rate (X_5) and highest growth temperature within investigating space.

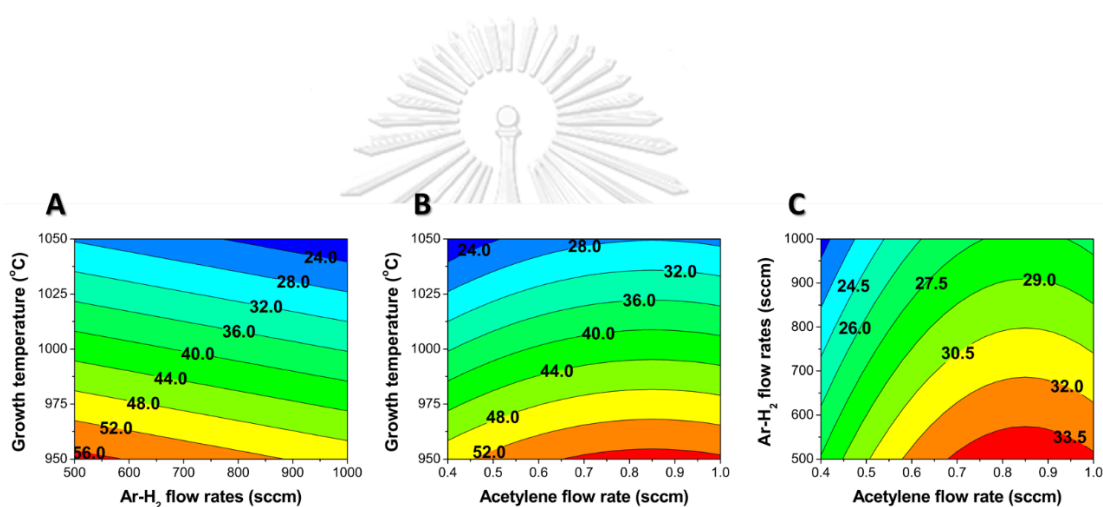


Figure 4.12 Response surface plots of FWHM(G) for (A) growth temperature and Ar-H₂ flow rates, (B) growth temperature and acetylene flow rate and (C) Ar-H₂ flow rates and acetylene flow rate

For FWHM(G), no interaction term was present in the model. There were only three linear terms (X_3 , X_4 and X_5) and a quadratic term of acetylene flow rate (X_5^2). Growth temperature was once again a key factor to tune FWHM(G). The lowest FWHM(G) was obtained at high Ar-H₂ flow rates (X_4) and growth temperature (X_3) (Figure 4.12(A)). Decreasing acetylene flow rate (X_5) can also decrease FWHM(G) (Figure 4.12(B)). This observation was more obvious when acetylene flow rate (X_5) was lower than 0.6 sccm.

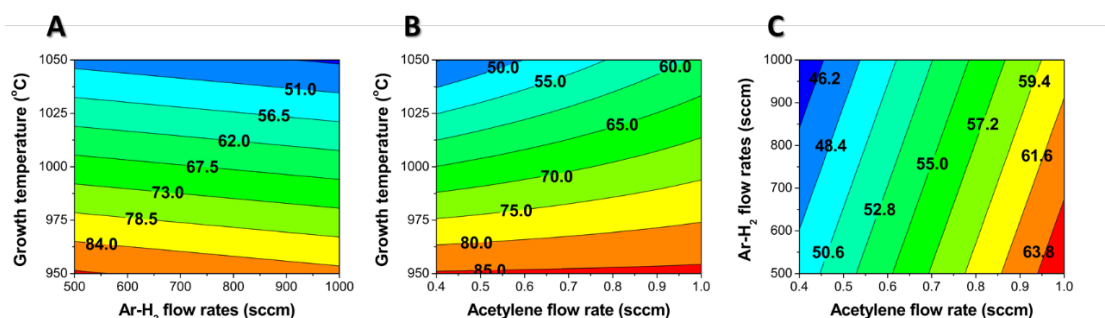


Figure 4.13 Response surface plots of FWHM(2D) for (A) growth temperature and Ar-H₂ flow rates, (B) growth temperature and acetylene flow rate and (C) Ar-H₂ flow rates and acetylene flow rate

FWHM(2D) is the most precise indicator for approximating the number of graphene layer because it is not disturbed efficiently by charge impurity. Figure 4.13(B) showed that the best quality graphene was produced at high temperature (X_3) with low acetylene flow rate (X_5). While high flow rates of Ar and H₂ (X_4) showed small negative effect on FWHM(2D) (Figure 4.13(A) and 4.13(B)). The thinnest graphene synthesized under investigating space was bilayer with FWHM(2D) of 40-50 cm⁻¹.

In summary, growth temperature (X_3), acetylene flow rate (X_5) and Ar-H₂ flow rates (X_4) are important factors to achieve the thinnest graphene with lowest defect. The response surface models suggested that the best quality graphene could be produced at 1050 °C with Ar-H₂-acetylene flow rate ratio of 900/100/0.4.

Nucleation and growth rates are kinetic parameters controlling growth of graphene. Low nucleation and high growth rates are preferred because they could lead to large graphene grain with low defect. Defects on graphene usually refer to graphene edges or non-6-membered ring structures. Defect density can be estimated by I(D)/I(G) ratio from Raman spectra. Due to high pyrolysis rate of acetylene, nucleation rate of the process is often high resulting in high edge defects. Furthermore, high nucleation rate could create additional graphene layer resulting in multilayer structure. The effects of CVD parameters must be investigated in order to reduce nucleation rate

According to this study, growth temperature (X_3), Ar-H₂ flow rates (X_4) and acetylene flow rate (X_5) were involved in the models. The most convenient way to

reduce nucleation rate is to dilute acetylene gas by increasing Ar-H₂ flow rates (X₄) and decreasing acetylene flow rate (X₅). In addition, increasing Ar-H₂ flow rates (X₄) could also reduce resident time of gases. Consequently, acetylene has a time limit to adsorb on Cu surface. However, those two factors (X₄ and X₅) had less effects than growth temperature (X₃). Based on Bertran's report [79], nucleation rate drops as temperature increases. Therefore, synthesizing CVD process at high temperature could afford graphene with low defect.

4.3 Functionalization of CVD-grown graphene

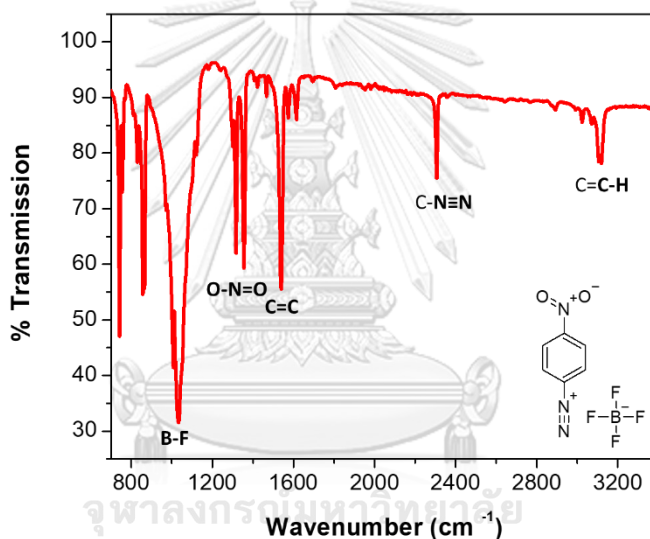


Figure 4.14 IR spectrum of 4-nitrobenzenediazonium tetrafluoroborate (4NBDT)

As grown graphene was functionalized with 4-nitrobenzenediazonium tetrafluoroborate (4NBDT). Firstly, 4NBDT was synthesized from 4-nitroaniline and the product was then characterized by IR spectroscopy and NMR spectroscopy. IR spectra provide information about the functional groups of substance while NMR spectra confirm the structure of organic compound.

The IR spectrum of 4NBDT as shown in Figure 4.14 exhibited 5 characteristic peaks. The diazonium vibration (N≡N) was clearly observed at 2307 cm⁻¹ while no

amine peak was observed around 3600 cm^{-1} . This indicated that the amine reactant was successfully converted to diazonium salt. Moreover, high intense peak of B–F vibration appeared at 1033 cm^{-1} suggesting that tetrafluoroborate ion was the counterion of diazonium salt. Meanwhile, the vibrational peak of O–N=O group was also observed at 1356 and 1317 cm^{-1} and the C=C and C–H vibrations of phenyl core structure also appeared at 1540 and 3118 cm^{-1} .

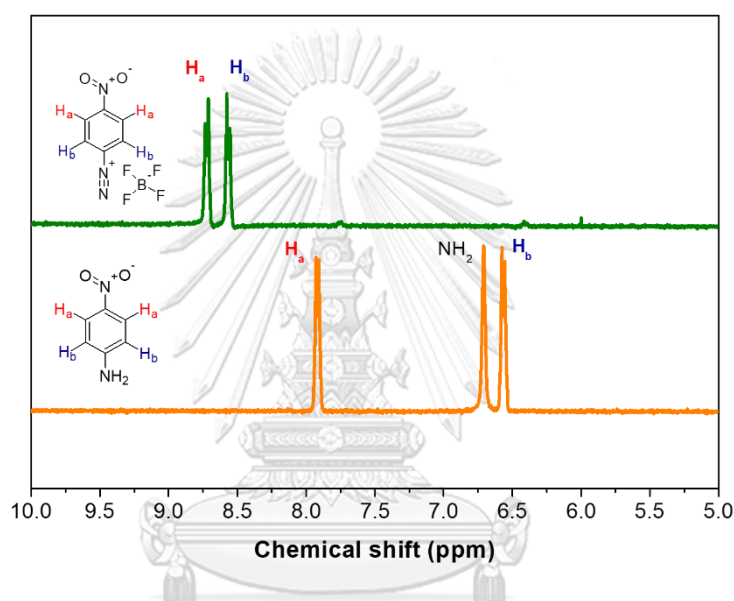


Figure 4.15 NMR spectra of 4-nitrobenzenediazonium tetrafluoroborate (4NBDT) (top) and 4-nitroaniline (bottom)

NMR spectra of 4NBDT and 4-nitroaniline were compared in Figure 4.15. After diazotization, the doublet amino protons around 8 ppm disappeared and phenyl protons were shifted to lower field for approximately 2 ppm. This suggested that after amine was converted to diazonium ion, protons on phenyl ring became more electron-deficient. These NMR results confirmed that 4NBDT was successfully synthesized.

4.3.1 Effect of substrate on functionalization

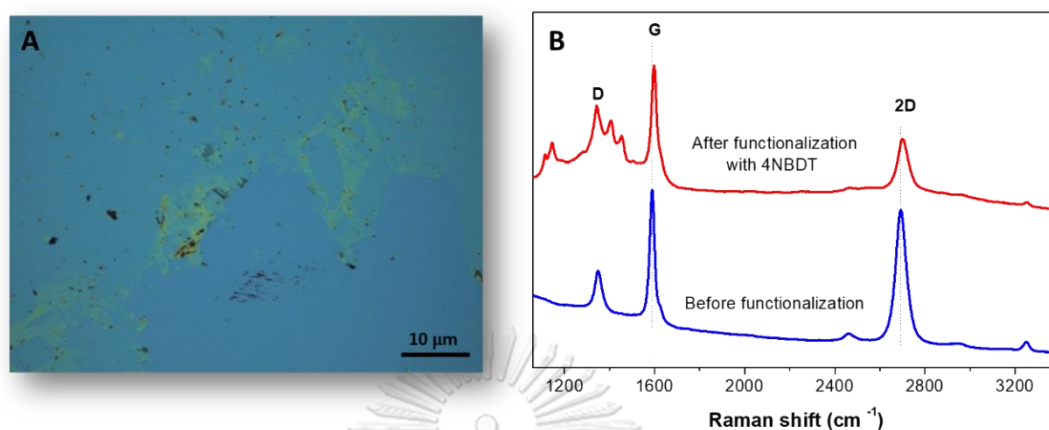


Figure 4.16 (A) Optical micrograph of modified graphene and (B) Raman spectra of graphene before functionalization (top) and after functionalization on SiO₂/Si (bottom)

The procedure reported by Strano and co-worker [51] was tested with graphene synthesized from acetylene. Graphene on SiO₂/Si was incubated in stirred 25 mM 4NBDT in 1 % w/v SDS/H₂O for 7 hr. The reaction was carried out in a dark fume hood. After the reaction was completed, the sample was washed with copious amounts of water before incubating in water overnight to make sure that all unreacted reactants were removed. However, in their publication, they used mechanical exfoliated single layer graphene while the graphene used in our experiment was CVD graphene. Although our graphene film was not entirely uniform, its average thickness was bilayer.

Interestingly, after modification, some parts of graphene film were lost as illustrated in Figure 4.16(A). This might be due to weak interactions between substrate and graphene.

Modified graphene remained on the substrate was subjected to Raman measurement. Figure 4.16(B) showed Raman spectra of graphene before and after functionalization where three characteristic peaks were found. Clearly, their positions

and intensity changed significantly. The intensity of D peak apparently increased indicating that some sp^2 -hybridized carbons were converted to sp^3 -hybridized carbons. This implied that functionalization was successful. However, there were messy peaks found between D and G peak and around 1200 cm^{-1} . These peaks were attributed to π - π interaction between unreacted diazonium salt and graphene. Moreover, the intensity of 2D peak significantly dropped due to doping. As well as Pos(G) and Pos(2D), they were also shifted whereas; FWHM(G) and FWHM(2D) had small increase as shown in Table 4.17.

Table 4.17 Average Raman parameters of graphene before functionalization and after functionalization with 4NBDT on SiO_2/Si

Raman parameter	Before functionalization	After functionalization
I(D)/I(G)	0.31 ± 0.01	0.44 ± 0.01
I(2D)/I(G)	1.03 ± 0.02	0.55 ± 0.01
Pos(G)	1589.5 ± 0.2	1596.9 ± 0.1
FWHM(G)	24.2 ± 0.3	25.1 ± 2.7
Pos(2D)	2693.4 ± 0.3	2701.7 ± 0.3
FWHM(2D)	56.5 ± 0.5	64.9 ± 0.6

The XPS spectra (Figure 4.17) also confirmed that the functionalization was successful as the N1s peak appeared at 401 and 407 eV where two types of nitrogen had similar amounts. The peak at higher binding energy was assigned to nitro group while the peak at lower binding energy was attributed to adsorbed nitrogen gas or diazene. Even though XPS measurement was performed at ultrahigh vacuum, some nitrogen or oxygen molecules can still adsorb and their peaks can be present in the spectra. The diazene might stem from diazo coupling resulting in multilayer functionalization. According to XPS spectra, a nitro group could be found each every 21 carbon atoms of graphene.

The existence of unreacted diazonium salt might affect conductivity according to doping. Strong interaction between diazonium salt and graphene and low reactivity of graphene were considered as cause of this problem. However, we could not produce single layer graphene from acetylene. To solve the problem, the strategy was changed from graphene to substrate.

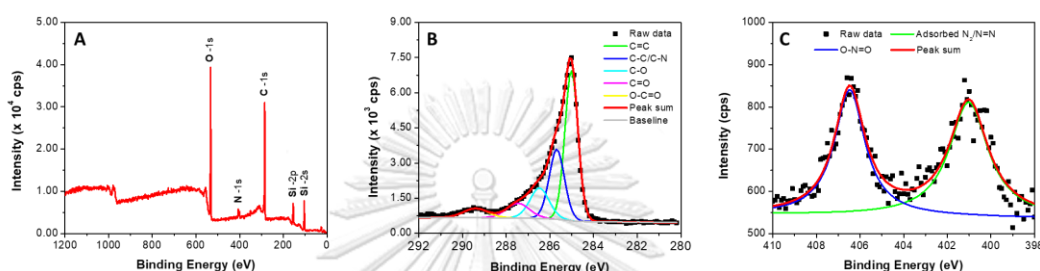


Figure 4.17 (A) Survey XPS spectrum (B) C1s XPS spectrum and (C) N1s XPS spectrum of functionalized graphene on SiO₂/Si

SiO₂/Si is a promising substrate for functionalization of graphene thank to its polarity. Since graphene is an extremely thin material, the substrate could possibly retain 4NBDT on the surface of graphene. Changing substrate to less polar was an alternative way.

Cu is another interesting candidate for graphene functionalization since no transfer step is required prior to functionalization. Moreover, graphene and Cu had lattice-matching because graphene was CVD grown directly on Cu. This could prevent loss of graphene during functionalization. More importantly, Cu is inert to diazonium functionalization since diazonium cannot react with Cu.

Graphene on Cu was functionalized under same condition as graphene on SiO₂/Si. After overnight of incubation in water, graphene was transferred to SiO₂/Si for Raman measurement. The image taken from an optical microscope as in Figure 4.18(A) revealed smooth and continuous surface of functionalized graphene.

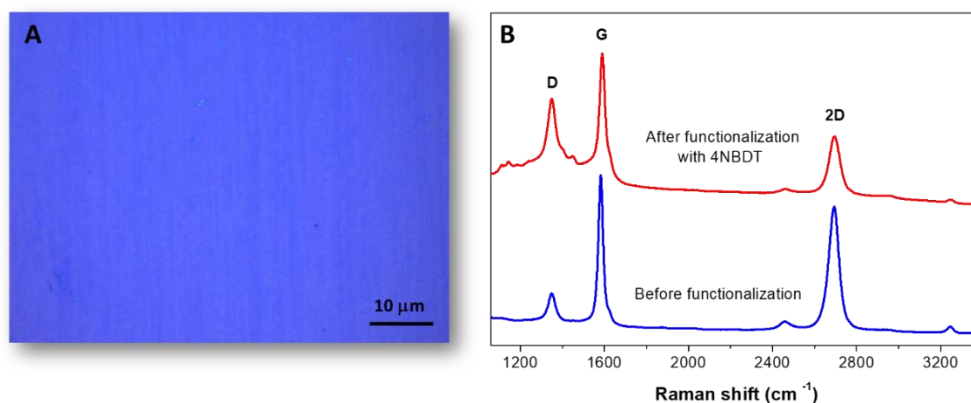


Figure 4.18 (A) Optical micrograph of modified graphene and (B) Raman spectra of graphene before functionalization (top) and after functionalization on Cu (bottom). All samples were transferred to SiO_2/Si substrate prior to taking an optical micrograph and measuring Raman spectra

After functionalization, the intensity of D peak distinctly increased while the intensity of 2D peak significantly declined as shown in Figure 4.18(B). In addition, both G and 2D peaks were red-shifted. Intriguingly, messy peaks found in graphene modified on SiO_2/Si were not observed on graphene modified on Cu. However, degree of functionalization estimated by increase of $I(\text{D})/I(\text{G})$ in case of functionalization on SiO_2/Si and Cu were not comparable because of different quality of starting graphene. Whilst the change of other Raman parameters (Table 4.18) corresponded to the modification on SiO_2/Si .

Table 4.18 Average Raman parameters of graphene before functionalization and after functionalization with 4NBDT on Cu

Raman parameter	Before functionalization	After functionalization
I(D)/I(G)	0.19 ± 0.01	0.57 ± 0.02
I(2D)/I(G)	0.99 ± 0.02	0.48 ± 0.01
Pos(G)	1585.0 ± 0.2	1590.4 ± 0.2
FWHM(G)	23.8 ± 0.2	31.4 ± 0.3
Pos(2D)	2691.0 ± 0.3	2695.2 ± 0.3
FWHM(2D)	56.8 ± 0.4	63.5 ± 0.5

XPS spectra of graphene functionalized on Cu were similar to that of graphene functionalized on SiO₂/Si as seen in Figure 4.19. However, most of nitrogen signal came from nitro group rather than adsorbed nitrogen gas with diazene. The calculated ratio of nitro group to carbon atoms of graphene was 1 to 27.

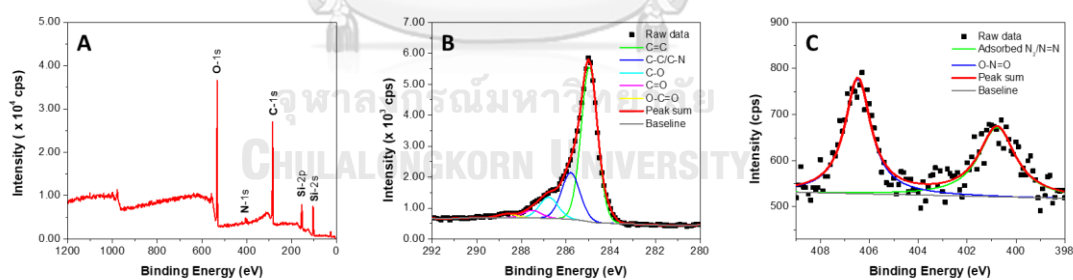


Figure 4.19 (A) Survey XPS spectrum (B) C1s XPS spectrum and (C) N1s XPS spectrum of graphene functionalized on Cu

The XPS results indicated that functionalization on SiO₂/Si could give higher degree of functionalization than on Cu. This observation is in consistent with those in other reports [57]. However, functionalization on SiO₂/Si also had drawbacks. First, it was difficult to remove unreacted diazonium salt from graphene surface by simple

washing. Second, some parts of graphene were torn off due to lattice-mismatching and weak interactions between graphene and substrate. Therefore, functionalization on Cu was a better choice.

4.3.2 Effect of light on functionalization

Diazonium salts are utilized in organic synthesis of aromatic compounds as versatile intermediates. Diazonium group can be substituted by hydrogen, halide, cyanide and hydroxyl groups or coupled with electron-rich aromatic ring. It has been taught in organic classes for a long time, however, a property of diazonium salt is usually neglected. That is sensitivity of diazonium salt to light [80]. Under irradiation of suitable wavelength (350-450 nm), nitrogen could cleave, and aryl cation are formed. The aryl cation can readily react with nucleophiles.

However, grafting mechanism of a diazonium salt to graphene is quite different. First, an electron from graphene transfers to a diazonium salt. Nitrogen gas is released, and aryl radical is formed. Second, the active radical reacts with graphene by forming a covalent bond. The first step is considered as the rate determining step and many approaches were made to push diazonium salt close to graphene surface. Another approach is to promote electron transfer from graphene to diazonium salt.

The conventional method uses SDS as surfactant to bring diazonium salt close to graphene surface so electron transfer takes place easily. However, the reaction takes 7 hr to complete with stirring. Additionally, there is risk that magnetic bar would damage graphene surface.

As mentioned earlier, diazonium salt could decompose under light irradiation. Nonetheless, there is no any clear report about the role of light in diazonium functionalization on graphene. Therefore, light would be employed in the following experiments in order to observe the effect of light on graphene functionalization.

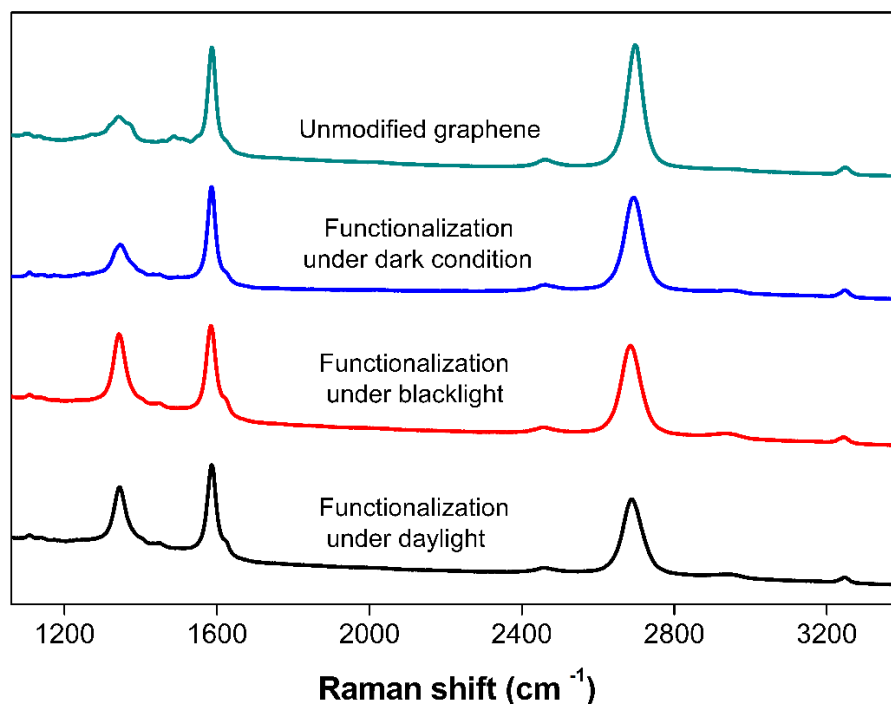


Figure 4.20 Raman spectra of unmodified graphene (cyan) and modified graphene under dark condition (blue), blacklight (red) and daylight (black)

Three experiments were conducted in order to compare modification methods under dark, daylight and blacklight conditions. As-grown graphene on Cu was incubated in 25 mM of 4NBDT under different light condition for 30 min. Then, all modified graphene was washed with copious amounts of water before incubating in water overnight. Afterwards, graphene was transferred to SiO₂/Si substrate for Raman measurement.

As illustrated in Figure 4.20, under dark condition, the intensity of D peak slightly increase, whereas the intensity of 2D peak marginally lessened. This implied that the grafting reaction could slowly take place without light.

As expected, under irradiation of light, D peak tremendously increased. However, unlike conventional method, 2D peak did not immensely decrease. This suggested that only covalent modification was dominant while diazonium salt rarely remained on graphene. Although the rise of D peak in case of irradiation under

blacklight was similar to that under daylight, $I(D)/I(G)$ of graphene modified under blacklight was slightly higher. This indicated that light took an important role in enhancement of diazonium functionalization.

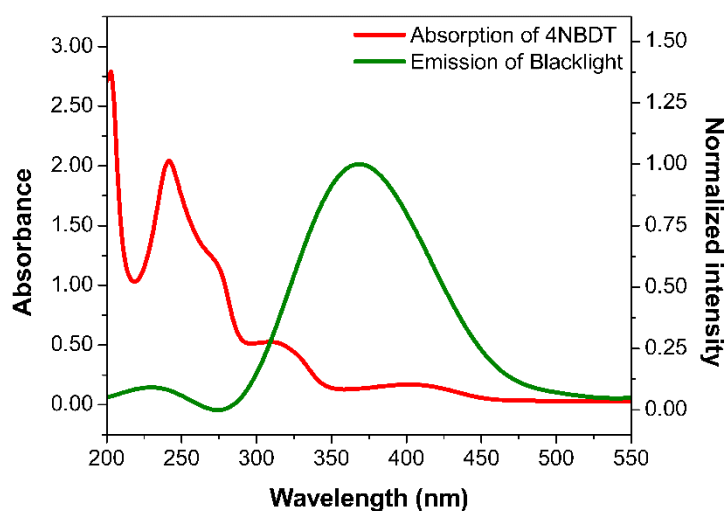


Figure 4.21 UV spectrum of 4-nitrobenzenediazonium tetrafluoroborate (4NBDT) and fluorescent spectrum of blacklight used in these experiments

For detailed discussion, the emission spectra of blacklight was recorded using a UV-Vis spectrophotometer. The blacklight emitted a broad spectrum from 275 to 525 nm as shown in Figure 4.21. The maximum emission wavelength of 368 nm with full width at half maximum of 101 nm made the light falling into a UV-A region. Since the UV spectrum of 4NBDT revealed an apparent absorption peak at 241 nm and shoulder peaks at 276 and 308 nm, the absorption spectrum of 4NBDT and the emission spectrum of blacklight were overlapping from 300 to 350 nm. Therefore, it was possible that 4NBDT could absorb photon energy from blacklight to release nitrogen gas and form 4-nitrophenyl cation. Then, the cation would accept an electron from graphene via single-electron transfer (SET) and become 4-nitrophenyl radical as displayed in Figure 4.22.

Additionally, it was also possible that blacklight could simultaneously interact with graphene. For example, photoinduced functionalization of graphene with

benzoyl peroxide was reported [81]. In that work, hot electrons generated from a 514.5 nm Raman laser were considered as activators for functionalization. Similarly, hot electrons could be generated from graphene by blacklight in our case. Therefore, SET from graphene to 4-nitrophenyl cation could be accelerated. Finally, a covalent bond between 4-nitrophenyl radical and graphene radical was formed.

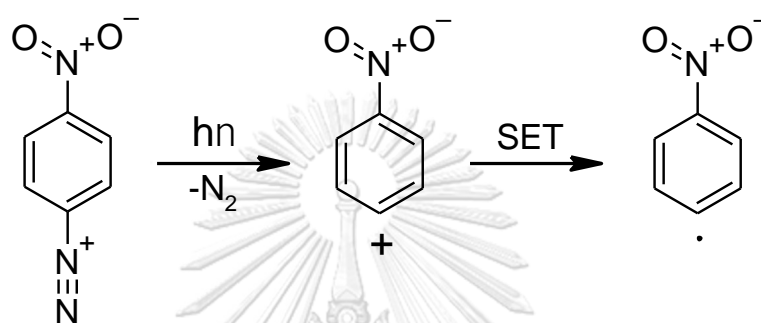


Figure 4.22 Schematic steps explaining the generation of 4-nitrophenyl radical under radiation of light

CHAPTER V

CONCLUSION

To optimize an atmospheric CVD process for graphene synthesis on copper substrate from acetylene precursor, six Raman parameters of graphene including I(D)/I(G), I(2D)/I(G), Pos(G), FWHM(G), Pos(2D) and FWHM(2D) were selected as dependent variables. According to Plackett-Burman screening experiments, growth temperature (X_3) and Ar-H₂ flow rates (X_4) were expected to be significant at moderate level of confidence (70 – 80%). Together with acetylene flow rate (X_5), three parameters of CVD process for graphene synthesis were subject to response surface modeling (RSM). Only four approximation functions were obtained by performing multiple linear regression (MLR) of data according to Box-Behnken design;

$$I(D)/I(G) = 0.899 - 0.310C - 0.095D - 0.107C*D - 0.089C^2 - 0.116E^2 \quad : R^2 = 0.962$$

$$I(2D)/I(G) = 0.747 + 0.395C - 0.146E - 0.135C*E \quad : R^2 = 0.946$$

$$FWHM(G)/cm^{-1} = 45.159 - 14.782C - 3.356D + 3.059E - 3.064E^2 \quad : R^2 = 0.987$$

$$FWHM(2D) /cm^{-1} = 71.580 - 16.515C - 2.391D + 4.172E + 3.860C*E \quad : R^2 = 0.978$$

where $C = 0.02*((\text{growth temperature})/^\circ\text{C}) - 20$, $D = 0.004*((\text{Ar flow rate} + \text{H}_2 \text{ flow rate})/\text{sccm}) - 3$ and $E = 3.333*((\text{acetylene flow rate})/\text{sccm}) - 2.333$. Growth temperature is a key factor to control both defect amount and number of graphene layers by adjusting nucleation rate. The best condition for synthesizing graphene under our investigating space was growth temperature of 1050 °C and Ar, H₂, C₂H₂ flow rates of 900, 100, 0.4 sccm, respectively.

Graphene samples prepared by above condition on different substrates including SiO₂/Si and Cu were then functionalized by 4-nitrobenzenediazonium tetrafluoroborate (4NBDT) and the functionalized graphene samples were characterized by Raman spectroscopy and X-ray photoelectron spectroscopy (XPS). Unreacted diazonium salts were remained on functionalized graphene samples on SiO₂/Si. On the other hand, functionalized graphene samples on Cu afforded cleaner surfaces. Moreover, in the presence of light (UV-A region), degree of functionalization was enhanced whereas the grafting reaction slowly occurred in the dark. Light in such

region was presumably an electron-transfer activator because it could generate hot electrons from graphene and they could be transferred to 4-nitrophenyl cations generated simultaneously from diazonium salt also accelerated by blacklight.



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APPENDIX

จุฬาลงกรณ์มหาวิทยาลัย
CHULALONGKORN UNIVERSITY

Python script for analysis of Raman spectra

```

1  import os
2  import subprocess
3  import RI_DG_2D_version_1_22
4  import AG_ver_1_02
5  import AF_1_02
6  import SA_1_01
7  import shutil
8  import Histogram_version_1_02
9
10 #RI_DG_2D_version_1_22.interpretator('Prefix file', First number, Last number, 'output
    filename')
11 RI_DG_2D_version_1_22.interpretator('data1_', 1, 100, '00001')
12 RI_DG_2D_version_1_22.interpretator('data2_', 1, 100, '00002')
13 RI_DG_2D_version_1_22.interpretator('data3_', 1, 100, '00003')
14 #AG_ver_1_02.AverageGraph('Prefix file', First number, Last number)
15 AG_ver_1_02.AverageGraph('data1_', 1, 100)
16 AG_ver_1_02.AverageGraph('data2_', 1, 100)
17 AG_ver_1_02.AverageGraph('data3_', 1, 100)
18 #os.rename('Original name', 'new name')
19 shutil.move('Average_data1_from1to100.csv', 'Average_data_0001.csv')
20 shutil.move('Average_data2_from1to100.csv', 'Average_data_0002.csv')
21 shutil.move('Average_data3_from1to100.csv', 'Average_data_0003.csv')
22 AG_ver_1_02.AverageGraph('Average_data_', 1, 3)
23 shutil.move('Average Average_data_from1to3.csv', 'Average_graph.csv')
24 #AF_1_02.AssemblyFile(No file, No spectra collected, 'Filename1', 'Filename2',
    'Filename3', 'output filename')
25 AF_1_02.AssemblyFile(3, 250, '00001', '00002', '00003', 'StatRawdata')
26 Histogram_version_1_02.HistogramMaker('StatRawdata')
27 SA_1_01.StatAnalysis('StatRawdata')

```

Python function for calculating Raman parameters of graphene

```

1  def interpretator(Prefix, first_file_number, Last_file_number, Net_file_name):
2      import pandas as pd
3      import numpy as np
4      import csv
5      from math import factorial
6
7      print("Graphene's Raman Spectra Interpretator")
8      print("by Bhobnibhit C updated Feb 13th 2018 version 1.21")
9      print("\nThis program will calculate I(D)/I(G), I(2D)/I(G), Pos(G), FWHM(G),
10     Pos(2D) and FWHM(2D) from Raman spectra\n")
11     print("\nThis version, automatically save all files\n")
12
13     #create array for keeping all data
14     Net = []
15     for i in range(1000):
16         Net.append([])
17         for j in range(7):
18             Net[i].append(0)
19
20     h = 0
21
22     #Prefix = input('Enter input file prefix: ')
23     #first_file_number = Current_file_number = int(input('First file number: '))
24     #Last_file_number = int(input('Last file number: '))
25     Current_file_number = first_file_number
26     filename1 = 'love'
27
28     print("\n#####")
29     print("#####\n")
30     print("****Analysis in Progress****")
31     #print("\n")
32     #print(Prefix)
33     #print(first_file_number)
34     #print(Last_file_number)
35     #print(filename1)
36
37     #making file name
38     while first_file_number-1 < Current_file_number < Last_file_number +1:
39         if 0 < Current_file_number < 10:
40             filename1 = Prefix + "000" + str(Current_file_number) + ".csv"
41             filename_input = Prefix + "000" + str(Current_file_number)
42             #print(filename1)
43             Current_file_number = Current_file_number + 1
44         elif 9 < Current_file_number < 100:
45             filename1 = Prefix + "00" + str(Current_file_number) + ".csv"
46             filename_input = Prefix + "000" + str(Current_file_number)
47             #print(filename1)
48             Current_file_number = Current_file_number + 1
49         elif 99 < Current_file_number < 1000:
50             filename1 = Prefix + "0" + str(Current_file_number) + ".csv"
51             filename_input = Prefix + "000" + str(Current_file_number)
52             #print(filename1)
53             Current_file_number = Current_file_number + 1
54         c_reader = csv.reader(open(filename1, 'r'), delimiter=',')
55         RamanShift_Full_list = list(zip(*c_reader))[0]
56         c_reader = csv.reader(open(filename1, 'r'), delimiter=',')
57         Intensity_Full_list = list(zip(*c_reader))[1]
58
59         #print(RamanShift_Full)
60         #print("\n\n\n")
61         #print(Intensity_Full)
62
63         #df = pd.read_csv("point2_0001.csv")
64         #df = pd.read_csv(filename)
65         #RamanShift_Full = df.RamanShift #you can also use df['column_name']
66         #Intensity_Full = df.Intensity #you can also use df['column_name']

```



```

132         #print(RamanShift_Full[i],",",Intensity_Full[i])
133         x2 = RamanShift_Full[i]
134         y2 = Intensity_Full[i]
135         break
136
137     #print(x1,y1,x2,y2)
138     #calculating slope and intercept
139     slope_2D = (y2-y1)/(x2-x1)
140     intercept_2D = ((y2 - slope_2D*x2) + (y1 - slope_2D*x1))/2
141     #print(slope_2D,intercept_2D)
142     #create baseline array
143     i = 0
144     j = 0
145     k = 0
146     filename_reserved = "peak_2D"+ filename_input + ".csv"
147     with open(filename_reserved, "w") as sp:
148         for i in range (0, len(RamanShift_Full)):
149             while 2000 < RamanShift_Full[i] < 3200 :
150                 j = j +1
151                 #print(RamanShift_Full[i],",",Intensity_Full[i], file=sp)
152                 RamanShift_baseline_2D[k] = RamanShift_Full[i]
153                 Intensity_baseline_2D[k] = Intensity_Full[i]
154                 k = k + 1
155                 break
156     sp.close()
157     #print(RamanShift_baseline_2D)
158     #print(len(RamanShift_baseline_2D))
159     #creating fitting array
160     filename_reserved = "2DbaselineFit"+ filename_input + ".csv"
161     with open(filename_reserved, "w") as sp:
162         for i in range (0, len(RamanShift_baseline_2D)):
163             #print(baseline_2D[i])
164             baseline_2D[i] = float((slope_2D)*(RamanShift_baseline_2D[i]) +
165                                     (intercept_2D))
166             #print(RamanShift_baseline_2D[i],",",baseline_2D[i], file=sp)
167             #print(RamanShift_baseline_2D[i],",",baseline_2D[i])
168     sp.close()
169     #print(baseline_2D)
170     #print("\n\n\n\n")
171     #print(Intensity_baseline_2D)
172
173     #Substract baseline
174     filename_reserved = "2D_baseline_corrected"+ filename_input + ".csv"
175     with open(filename_reserved, "w") as sp:
176         for i in range (0, len(RamanShift_baseline_2D)):
177             baseline_2D_corrected[i] = Intensity_baseline_2D[i] - baseline_2D[i]
178             #print(RamanShift_baseline_2D[i],",",baseline_2D_corrected[i], file=sp)
179     sp.close()
180
181     #finding peak_2D
182     max_2D = []
183     for k in range(0,20):
184         max_2D.append(0)
185     k = 0
186     #finding maximum of 2D
187     for i in range (1, len(RamanShift_baseline_2D)-1):
188         if baseline_2D_corrected[i-1] < baseline_2D_corrected[i] and
189            baseline_2D_corrected[i] > baseline_2D_corrected[i+1] and 2650 <
190            RamanShift_baseline_2D[i] < 2750:
191             #print(RamanShift_baseline_2D[i],",",baseline_2D_corrected[i])
192             max_2D[k] = baseline_2D_corrected[i]
193             k = k +1
194     #print(max_2D)
195     int_2D = np.amax(max_2D)
196
197     for i in range (0, len(RamanShift_baseline_2D)):
198         if baseline_2D_corrected[i] == int_2D:

```



```

256                                     #print(RamanShift_baseline_2D[i],",",baseline_2D_correcte
257                                     d[i])
258                                     Pos_at_half_height_2D[1] = RamanShift_baseline_2D[i]
259                                     EF = EF + 0.0001
260 #print(Pos_at_half_height_2D[0],Pos_at_half_height_2D[1])
261 #print("\n", Pos_at_half_height_2D)
262 FWHM_2D = abs(float(Pos_at_half_height_2D[1]) -
263 float(Pos_at_half_height_2D[0]))
264 #print(FWHM_2D)
265 #print("Full width at half maximum of 2D peak is", FWHM_2D,"cm-1")
266 #print("\n\n\n\n\n\n\n\n\n\n\n\n\n\n\n\n")
267
268 baseline_DandG = []
269 RamanShift_baseline_DandG = []
270 Intensity_baseline_DandG = []
271 baseline_DandG_corrected = []
272
273 for k in range(0,726):
274     RamanShift_baseline_DandG.append(0)
275 for k in range(0,726):
276     Intensity_baseline_DandG.append(0)
277 for k in range(0,726):
278     baseline_DandG.append(0)
279 for k in range(0,726):
280     baseline_DandG_corrected.append(0)
281
282 #create baseline array for D and G peak
283 i = 0
284 j = 0
285 k = 0
286 filename_reserved = "peak_D_and_G"+ filename_input + ".csv"
287 with open(filename_reserved, "w") as sp:
288     for i in range (0, len(RamanShift_Full)):
289         while 1100 < RamanShift_Full[i] < 1800 :
290             j = j +1
291             #print(RamanShift_Full[i],",",Intensity_Full[i], file=sp)
292             RamanShift_baseline_DandG[k] = RamanShift_Full[i]
293             Intensity_baseline_DandG[k] = Intensity_Full[i]
294             k = k + 1
295             break
296 sp.close()
297 #print(RamanShift_baseline_DandG)
298 #print(j)
299 #print(len(RamanShift_baseline_DandG))
300 #print(Intensity_baseline_DandG)
301 #print(len(Intensity_baseline_DandG))
302
303 #marking 2 points for D and G
304 for i in range (0, len(RamanShift_Full)):
305     while 1100 < RamanShift_Full[i] < 1102 :
306         #print(RamanShift_Full[i],",",Intensity_Full[i])
307         x1 = RamanShift_Full[i]
308         y1 = Intensity_Full[i]
309         break
310 for i in range (0, len(RamanShift_Full)):
311     while 1798 < RamanShift_Full[i] < 1800 :
312         #print(RamanShift_Full[i],",",Intensity_Full[i])
313         x2 = RamanShift_Full[i]
314         y2 = Intensity_Full[i]
315         break
316
317 #print(x1,y1,x2,y2)
318 #calculating slope and intercept
319 slope_DandG = (y2-y1)/(x2-x1)
320 intercept_DandG = ((y2 - slope_DandG*x2) + (y1 - slope_DandG*x1))/2
321 #print(slope_DandG,intercept_DandG)

```



```

320 #creating fitting array
321 filename_reserved = "DandG_baselineFit"+ filename_input + ".csv"
322 with open(filename_reserved, "w") as sp:
323     for i in range (0, len(RamanShift_baseline_DandG)):
324         #print(baseline_DandG[i])
325         baseline_DandG[i] =
            float((slope_DandG)*(RamanShift_baseline_DandG[i]) +
            (intercept_DandG))
326         #print(RamanShift_baseline_DandG[i],",",baseline_DandG[i], file=sp)
327         #print(RamanShift_baseline_DandG[i],",",baseline_DandG[i])
328     sp.close()
329     #print(baseline_DandG)
330     #print("\n\n\n\n")
331     #print(Intensity_baseline_DandG)
332     #Substract baseline
333     filename_reserved = "DandG_baseline_corrected"+ filename_input + ".csv"
334     with open(filename_reserved, "w") as sp:
335         for i in range (0, len(RamanShift_baseline_DandG)):
336             baseline_DandG_corrected[i] = Intensity_baseline_DandG[i] -
            baseline_DandG[i]
337             #print(RamanShift_baseline_DandG[i],",",baseline_DandG_corrected[i],
            file=sp)
338     sp.close()
339
340
341
342
343 #finding peak_D
344 max_D = []
345 for k in range(0,20):
346     max_D.append(0)
347     k = 0
348 #finding maximum of D
349
350 for i in range (1, len(baseline_DandG_corrected)-1):
351     if baseline_DandG_corrected[i-1] < baseline_DandG_corrected[i] and
        baseline_DandG_corrected[i] > baseline_DandG_corrected[i+1] and 1300 <
        RamanShift_baseline_DandG[i] < 1400:
352         #print(RamanShift_baseline_D[i],",",baseline_D_corrected[i])
353         max_D[k] = baseline_DandG_corrected[i]
354         k = k +1
355     #print(max_D)
356     int_D = np.amax(max_D)
357
358 for i in range (0, len(RamanShift_baseline_DandG)):
359     if baseline_DandG_corrected[i] == int_D:
360         pos_D = RamanShift_baseline_DandG[i]
361         #elif int_D == 0:
362         #pos_D = 0
363     #print(pos_D,",",int_D)
364     #print("Position of D peak is at", pos_D,"cm-1")
365     #print("Intensity of D peak is", int_D)
366
367 #finding peak_G
368 max_G = []
369 for k in range(0,20):
370     max_G.append(0)
371     k = 0
372 #finding maximum of G
373 int_G = np.amax(max_G)
374 for i in range (1, len(RamanShift_baseline_DandG)-1):
375     if baseline_DandG_corrected[i-1] < baseline_DandG_corrected[i] and
        baseline_DandG_corrected[i] > baseline_DandG_corrected[i+1] and 1550 <
        RamanShift_baseline_DandG[i] < 1600:
376         #print(RamanShift_baseline_DandG[i],",",baseline_DandG_corrected[i])
377         max_G[k] = baseline_DandG_corrected[i]
378         k = k +1

```

```

379     #print(max_G)
380     int_G = np.amax(max_G)
381
382     for i in range(0, len(RamanShift_baseline_DandG)):
383         if baseline_DandG_corrected[i] == int_G:
384             pos_G = RamanShift_baseline_DandG[i]
385             #elif int_G == 0:
386                 #pos_G = 0
387             #print(pos_G, ",", int_G)
388             #print("Position of G peak is at", pos_G, "cm-1")
389             #print("Intensity of G peak is", int_G)
390
391         #finding FWHM_G
392         half_height_G = int_G/2
393         Pos_at_half_height_G = []
394         for k in range(0,20):
395             Pos_at_half_height_G.append(0)
396         k = 0
397         #print(half_height_G)
398         EF = 0.001
399         #define range of G searching
400         for i in range(0, len(RamanShift_baseline_DandG)):
401             if 1450 < RamanShift_baseline_DandG[i] < 1451:
402                 LL = i
403             elif 1669 < RamanShift_baseline_DandG[i] < 1700:
404                 UL = i
405         #print(LL,UL)
406         if int_G != 0:
407             while Pos_at_half_height_G[1] == 0:
408                 ErrorFWHM = float(EF)/100
409                 #print("FWHM_G 1st loop")
410                 #print(ErrorFWHM)
411                 k = 0
412                 for j in range(0,20):
413                     Pos_at_half_height_G.append(0)
414                 for i in range(LL, UL):
415                     #print(RamanShift_baseline_DandG[i])
416                     if half_height_G - half_height_G*ErrorFWHM <
417                         baseline_DandG_corrected[i] < half_height_G +
418                             half_height_G*ErrorFWHM:
419
420                         #print(RamanShift_baseline_DandG[i], ",", baseline_DandG_correc
421                             ted[i])
422                         Pos_at_half_height_G[k] = RamanShift_baseline_DandG[i]
423                         k = k + 1
424                         EF = EF + 0.0001
425                 #recheck that FWHM is correct
426                 EF = 0.0001
427                 if Pos_at_half_height_G[0] < pos_G and Pos_at_half_height_G[1] < pos_G:
428                     Pos_at_half_height_G[1] = 0
429                     #print("FWHM_G 2nd loop")
430                     while Pos_at_half_height_G[1] == 0:
431                         ErrorFWHM = float(EF)/100
432                         for i in range(LL, UL):
433                             if RamanShift_baseline_DandG[i] > pos_G:
434                                 if half_height_G - half_height_G*ErrorFWHM <
435                                     baseline_DandG_corrected[i] < half_height_G +
436                                         half_height_G*ErrorFWHM:
437
438                             #print(RamanShift_baseline_2D[i], ",", baseline_2D_corr
439                                 ected[i])
440                             Pos_at_half_height_G[1] =
441                                 RamanShift_baseline_DandG[i]
442                             EF = EF + 0.0001
443                 elif Pos_at_half_height_G[1] > pos_G and Pos_at_half_height_G[0] > pos_G:
444                     Pos_at_half_height_G[1] = 0
445                     #print("FWHM_G 3rd loop")

```

```

437     while Pos_at_half_height_G[1] == 0:
438         ErrorFWHM = float(EF)/100
439         #print(ErrorFWHM)
440         for i in range (LL, UL):
441             if RamanShift_baseline_DandG[i] < pos_G:
442
443                 #print(RamanShift_baseline_DandG[i],baseline_DandG_correc
444                     ted[i])
445                 if half_height_G - half_height_G*ErrorFWHM <
446                     baseline_DandG_corrected[i] < half_height_G +
447                     half_height_G*ErrorFWHM:
448                     #print("3rd record loop")
449
450                     #print(RamanShift_baseline_DandG[i],"",baseline_Dand
451                         G_corrected[i])
452                     Pos_at_half_height_G[1] =
453                     RamanShift_baseline_DandG[i]
454                     EF = EF + 0.1
455                     FWHM_G = abs(float(Pos_at_half_height_G[1]) -
456                         float(Pos_at_half_height_G[0]))
457                     #print(FWHM_G)
458                     if int_G != 0:
459                         ID_IG = int_D/int_G
460                         I2D_IG = int_2D/int_G
461                     elif int_G == 0:
462                         ID_IG = "N/A"
463                         I2D_IG = "N/A"
464                         FWHM_G = 0
465                     #print("Full width at half maximum of G peak is", FWHM_G,"cm-1")
466                     #print("\n")
467                     #print("****The end of Analysis****")
468                     #print("\n")
469                     #print("****Conclusions****")
470                     #print("\n")
471                     #print("I(D)/I(G) = ",ID_IG)
472                     #print("I(2D)/I(G) = ",I2D_IG)
473                     #print("Pos(G) = ",pos_G,"      cm-1")
474                     #print("FWHM(G) = ",FWHM_G,"      cm-1")
475                     #print("Pos(2D) = ",pos_2D,"      cm-1")
476                     #print("FWHM(2D) = ",FWHM_2D,"      cm-1")
477                     #print("\n")
478                     Response = "y"
479                     #Conclusions
480                     if 5 < FWHM_G < 100 and 10 < FWHM_2D < 120 and 1570 < pos_G < 1600 and 2650
481                         < pos_2D < 2725:
482                         #Response = input('Do you want to save the results? ("n" not to save)')
483                         filename_saved = "R" + filename_input + ".csv"
484                         if Response == "n":
485                             print("The result wasn't be saved")
486                         else:
487                             #filename = input('Enter filename of the results: ')
488                             with open(filename_saved, "w") as sp:
489                                 print("I(D)/I(G) , I(2D)/I(G) , Pos(G) , FWHM(G) , Pos(2D) , FWHM(2D) ",
490                                     file=sp)
491
492                                 print(ID_IG,"",I2D_IG,"",pos_G,"",FWHM_G,"",pos_2D,"",FWHM_2
493                                     D, file=sp)
494                                 Net[h][0] = Current_file_number - 1
495                                 Net[h][1] = ID_IG
496                                 Net[h][2] = I2D_IG
497                                 Net[h][3] = pos_G
498                                 Net[h][4] = FWHM_G
499                                 Net[h][5] = pos_2D
500                                 Net[h][6] = FWHM_2D
501                                 h = h + 1
502                                 #print("already saved output as", filename_saved)
503                             #else:

```

```

492         #print("####bad input data####\n####It won't be saved####")
493         sp.close()
494         print("@",Current_file_number-1)
495         #print("\n")
496
497         #print("#####")
498         #print("\n")
499
500     #printing all data in file
501     print("****The End of Analysis****")
502
503     print("\n#####")
504     #print("\n")
505     #Net_file_name = input('We would save all data in one file\n\nEnter output file
506     name: ')
507     Net_file_name_with_csv = Net_file_name + ".csv"
508     with open(Net_file_name_with_csv, "w") as sp:
509         print("No, I(D)/I(G), I(2D)/I(G), Pos(G), FWHM(G), Pos(2D), FWHM(2D)", file=sp)
510         for i in range(0, len(Net)):
511             if Net[i][1] != 0 or Net[i][2] != 0 or Net[i][3] != 0 or Net[i][4] != 0 or
512             Net[i][5] != 0 or Net[i][6] != 0:
513                 print(Net[i][0],",",Net[i][1],",",Net[i][2],",",Net[i][3],",",Net[i][4],",",
514                 ",Net[i][5],",",Net[i][6], file=sp)
515             else:
516                 break
517     sp.close()
518     print("\nThe",Net_file_name_with_csv,"has been saved.\n")
519
520     print("#####")
521     print("\nThank you, See you next time\n")

```


Python function for collecting data from files

```

1  def AssemblyFile(NoFile, NoSpectra, File1, File2, File3, Net_file_name):
2      import pandas as pd
3      import numpy as np
4      import csv
5      from math import factorial
6      import random
7
8      print("\nAutomatic Assembling File ")
9      print("by Bhojibhit C updated Mar 6nd 2018 version 1.02\n")
10
11     print("This program will assembly files (up to 10 files and 250 spectra per file)
12     into single file.\n")
13     print("Instructions\n****Input files must be .csv with commas as dividers****\n ")
14
15     #Getting file name
16     #NoFile = int(input("How many files do you want to assembly?\n====>"))
17     #NoSpectra = int(input("How many spectra do you want to record per file?\n====>"))
18     FileName = []
19     FileNamewocsv = []
20     Net = []
21     for i in range(1000):
22         Net.append(0)
23         for j in range(8):
24             Net[i].append(0)
25     for i in range(0,10):
26         FileName.append(0)
27         FileNamewocsv.append(0)
28     FileNamewocsv[0] = File1
29     FileNamewocsv[1] = File2
30     FileNamewocsv[2] = File3
31     for i in range(0, NoFile):
32         print("Please enter file name No.",str(i + 1),"without .csv")
33         #FileNamewocsv[i] = input("----->")
34         FileName[i] = str(FileNamewocsv[i]) + ".csv"
35         #Deleting First line of file
36         with open(FileName[i], 'r') as fin:
37             data = fin.read().splitlines(True)
38             FileName[i]
39             with open(FileNamewocsv[i] + "_DFL.csv", 'w') as fout:
40                 fout.writelines(data[1:])
41     #print(FileName)
42
43     print("\n#####\n")
44     print("*****We are transferring your data*****\n")
45     CountFile = 0
46     CountNoSpectra = 0
47     CountID_IG = 0
48     CountI2D_IG = 0
49     CountPosG = 0
50     CountFWHMG = 0
51     CountPos2D = 0
52     CountFWHM2D = 0
53
54     for i in range(0, NoFile):
55         #print(NoFile)
56         c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
57         Filename = np.asarray(list(zip(*c_reader))[0])
58         for k in range(0, len(Filename)):
59             Net[CountNoSpectra][1] = Filename[k]
60             CountNoSpectra = CountNoSpectra + 1
61         for k in range(0, len(Filename)):
62             Net[CountFile][0] = i + 1
63             CountFile = CountFile + 1
64         c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
65         Filename = np.asarray(list(zip(*c_reader))[1])

```

```

65     Filename = [float(i) for i in Filename]
66     for k in range(0, len(Filename)):
67         Net[CountID_IG][2] = Filename[k]
68         CountID_IG = CountID_IG + 1
69     c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
70     Filename = np.asarray(list(zip(*c_reader))[2])
71     Filename = [float(i) for i in Filename]
72     for k in range(0, len(Filename)):
73         Net[CountI2D_IG][3] = Filename[k]
74         CountI2D_IG = CountI2D_IG + 1
75     c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
76     Filename = np.asarray(list(zip(*c_reader))[3])
77     Filename = [float(i) for i in Filename]
78     for k in range(0, len(Filename)):
79         Net[CountPosG][4] = Filename[k]
80         CountPosG = CountPosG + 1
81     c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
82     Filename = np.asarray(list(zip(*c_reader))[4])
83     Filename = [float(i) for i in Filename]
84     for k in range(0, len(Filename)):
85         Net[CountFWHMG][5] = Filename[k]
86         CountFWHMG = CountFWHMG + 1
87     c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
88     Filename = np.asarray(list(zip(*c_reader))[5])
89     Filename = [float(i) for i in Filename]
90     for k in range(0, len(Filename)):
91         Net[CountPos2D][6] = Filename[k]
92         CountPos2D = CountPos2D + 1
93     c_reader = csv.reader(open(FileNamewocsv[i] + "_DFL.csv", 'r'), delimiter=',')
94     Filename = np.asarray(list(zip(*c_reader))[6])
95     Filename = [float(i) for i in Filename]
96     for k in range(0, len(Filename)):
97         Net[CountFWHM2D][7] = Filename[k]
98         CountFWHM2D = CountFWHM2D + 1
99
100     #random number
101     randoms = random.sample(range(CountNoSpectra), k = NoSpectra)
102     sorted_randoms = sorted(randoms)
103     print(sorted_randoms)
104
105     #print data on screen
106     for k in range(0, len(sorted_randoms)):
107         if Net[sorted_randoms[k]][1] != 0 or Net[sorted_randoms[k]][2] != 0 or
108             Net[sorted_randoms[k]][3] != 0 or Net[sorted_randoms[k]][4] != 0 or
109             Net[sorted_randoms[k]][5] != 0 or Net[sorted_randoms[k]][6] != 0 or
110             Net[sorted_randoms[k]][7] != 0:
111
112             print(Net[sorted_randoms[k]][0],",",Net[sorted_randoms[k]][1],",",Net[sorted_
113                 randoms[k]][2],",",Net[sorted_randoms[k]][3],",",Net[sorted_randoms[k]][4],",",
114                 ",Net[sorted_randoms[k]][5],",",Net[sorted_randoms[k]][6],",",Net[sorted_ran
115                 oms[k]][7])
116         else:
117             break
118     print("\n****The End of Transfer****")
119
120     print("\n#####")
121     #print("\n")
122     #print data as file
123     #Net_file_name = input('We would save all data in one file\n\nEnter output file
124     name: ')
125     Net_file_name_with_csv = Net_file_name + ".csv"
126     with open(Net_file_name_with_csv, "w") as sp:
127         for k in range(0, len(sorted_randoms)):
128             if Net[sorted_randoms[k]][1] != 0 or Net[sorted_randoms[k]][2] != 0 or
129                 Net[sorted_randoms[k]][3] != 0 or Net[sorted_randoms[k]][4] != 0 or
130                 Net[sorted_randoms[k]][5] != 0 or Net[sorted_randoms[k]][6] != 0 or

```

```

120         Net[sorted_randoms[k]][7] != 0:
121             print(Net[sorted_randoms[k]][0],",",Net[sorted_randoms[k]][1],",",Net[sor
122                 ted_randoms[k]][2],",",Net[sorted_randoms[k]][3],",",Net[sorted_randoms[k]
123                 ][4],",",Net[sorted_randoms[k]][5],",",Net[sorted_randoms[k]][6],",",Net
124                 [sorted_randoms[k]][7], file=sp)
125             else:
126                 break
127         sp.close()
128
129     print("\n#####\n")
130     print("The",Net_file_name_with_csv,"has been saved.\n")
131
132     print("#####\n")
133     print("C'est fini. Merci, au revoir")
134     #print(Net)

```

Python function for creating histograms

```

1  def HistogramMaker(Filenamelpre):
2      import pandas as pd
3      import numpy as np
4      import csv
5      import scipy.stats
6      import statistics
7      from math import factorial
8      from time import gmtime, strftime
9      from datetime import datetime
10
11     print("\nHistogram maker")
12
13     print("\n#####\n")
14     print("by Bhobnibhit C updated Mar 28rd 2018 version 1.01\n\n")
15
16     #Input Data
17     #Filenamelpre = input("Please enter first filename: ")
18     Filenamel = Filenamelpre + ".csv"
19
20     #Collect data file 1
21     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
22     NoSpectra_1 = np.asarray(list(zip(*c_reader))[0])
23     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
24     ID_IG_1 = np.asarray(list(zip(*c_reader))[2])
25     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
26     I2D_IG_1 = np.asarray(list(zip(*c_reader))[3])
27     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
28     PosG_1 = np.asarray(list(zip(*c_reader))[4])
29     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
30     FWHMG_1 = np.asarray(list(zip(*c_reader))[5])
31     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
32     Pos2D_1 = np.asarray(list(zip(*c_reader))[6])
33     c_reader = csv.reader(open(Filenamel, 'r'), delimiter=',')
34     FWHM2D_1 = np.asarray(list(zip(*c_reader))[7])
35     ID_IG_1 = [float(i) for i in ID_IG_1]
36     I2D_IG_1 = [float(i) for i in I2D_IG_1]
37     PosG_1 = [float(i) for i in PosG_1]
38     FWHMG_1 = [float(i) for i in FWHMG_1]
39     Pos2D_1 = [float(i) for i in Pos2D_1]
40     FWHM2D_1 = [float(i) for i in FWHM2D_1]
41
42     print("\n#####\n")
43     print("\nDescriptives\n\nVariables Max Min Average")
44     print("ID/IG", np.amax(ID_IG_1), np.amin(ID_IG_1), np.mean(ID_IG_1))
45     print("I2D/IG", np.amax(I2D_IG_1), np.amin(I2D_IG_1), np.mean(I2D_IG_1))
46     print("PosG", np.amax(PosG_1), np.amin(PosG_1), np.mean(PosG_1))
47     print("FWHMG", np.amax(FWHMG_1), np.amin(FWHMG_1), np.mean(FWHMG_1))
48     print("Pos2D", np.amax(Pos2D_1), np.amin(Pos2D_1), np.mean(Pos2D_1))
49     print("FWHM2D", np.amax(FWHM2D_1), np.amin(FWHM2D_1), np.mean(FWHM2D_1))
50
51     print("\n#####\n")
52     print("Please enter these informations\n")
53     print("ID/IG")
54     First_ID_IG = 0 #float(input("Start Bin: "))
55     Last_ID_IG = 2.4 #float(input("End Bin: "))
56     Bin_ID_IG = 0.15 #float(input("Bin Width: "))
57     Level_ID_IG = int((Last_ID_IG - First_ID_IG)/Bin_ID_IG)
58     print("I2D/IG")
59     First_I2D_IG = 0 #float(input("Start Bin: "))
60     Last_I2D_IG = 4 #float(input("End Bin: "))
61     Bin_I2D_IG = 0.2 #float(input("Bin Width: "))
62     Level_I2D_IG = int((Last_I2D_IG - First_I2D_IG)/Bin_I2D_IG)

```



```

62     print("PosG")
63     First_PosG = 1570 #float(input("Start Bin: "))
64     Last_PosG = 1600 #float(input("End Bin: "))
65     Bin_PosG = 5 #float(input("Bin Width: "))
66     Level_PosG = int((Last_PosG - First_PosG)/Bin_PosG)
67     print("FWHMG")
68     First_FWHMG = 0 #float(input("Start Bin: "))
69     Last_FWHMG = 100 #float(input("End Bin: "))
70     Bin_FWHMG = 5 #float(input("Bin Width: "))
71     Level_FWHMG = int((Last_FWHMG - First_FWHMG)/Bin_FWHMG)
72     print("Pos2D")
73     First_Pos2D = 2650 #float(input("Start Bin: "))
74     Last_Pos2D = 2720 #float(input("End Bin: "))
75     Bin_Pos2D = 10 #float(input("Bin Width: "))
76     Level_Pos2D = int((Last_Pos2D - First_Pos2D)/Bin_Pos2D)
77     print("FWHM2D")
78     First_FWHM2D = 20 #float(input("Start Bin: "))
79     Last_FWHM2D = 120 #float(input("End Bin: "))
80     Bin_FWHM2D = 10 #float(input("Bin Width: "))
81     Level_FWHM2D = int((Last_FWHM2D - First_FWHM2D)/Bin_FWHM2D)
82
83     row = [Level_ID_IG, Level_I2D_IG, Level_PosG, Level_FWHMG, Level_Pos2D, Level_FWHM2D]
84     row_sum = int(np.amax(row))
85
86     sum = []
87     for i in range(row_sum):
88         sum.append([])
89         for j in range(12):
90             sum[i].append(0)
91     #print(sum)
92
93     #ID_IG
94     Edge_ID_IG = []
95     for i in range(0,Level_ID_IG +1):
96         Edge_ID_IG.append(0)
97     for i in range(0, len(Edge_ID_IG)):
98         if i == 0:
99             Edge_ID_IG[i] = First_ID_IG
100        elif i == len(Edge_ID_IG):
101            Edge_ID_IG[i] = Last_ID_IG
102        else:
103            Edge_ID_IG[i] = Edge_ID_IG[i-1] + Bin_ID_IG
104
105    j = 0
106    for i in range(0, len(Edge_ID_IG)-1):
107        sum[j][0] = (Edge_ID_IG[i] + Edge_ID_IG[i+1])/2
108        j = j + 1
109    for i in range(0, len(ID_IG_1)):
110        for j in range(1, len(Edge_ID_IG)):
111            if Edge_ID_IG[j-1] < ID_IG_1[i] < Edge_ID_IG[j]:
112                sum[j-1][1] = sum[j-1][1] + 1
113
114    #max_ID_IG = []
115    #for i in range(0, row_sum):
116        #max_ID_IG.append(0)
117    #for i in range(0, row_sum):
118        #max_ID_IG[i] = sum[i][1]
119    #Vmax_ID_IG = np.amax(max_ID_IG)
120    #for i in range(0, row_sum):
121        #sum[i][1] = (sum[i][1])*100/Vmax_ID_IG
122
123    #for j in range(1, row_sum):
124        #print(sum[j-1][0],sum[j-1][1])
125
126    #I2D_IG
127    Edge_I2D_IG = []
128    for i in range(0,Level_I2D_IG +1):
129        Edge_I2D_IG.append(0)
130    for i in range(0, len(Edge_I2D_IG)):

```

```

129         if i == 0:
130             Edge_I2D_IG[i] = First_I2D_IG
131         elif i == len(Edge_I2D_IG):
132             Edge_I2D_IG[i] = Last_I2D_IG
133         else:
134             Edge_I2D_IG[i] = Edge_I2D_IG[i-1] + Bin_I2D_IG
135     j = 0
136     for i in range(0, len(Edge_I2D_IG)-1):
137         sum[j][2] = (Edge_I2D_IG[i] + Edge_I2D_IG[i+1])/2
138         j = j + 1
139     for i in range(0, len(I2D_IG_1)):
140         for j in range(1, len(Edge_I2D_IG)):
141             if Edge_I2D_IG[j-1] < I2D_IG_1[i] < Edge_I2D_IG[j]:
142                 sum[j-1][3] = sum[j-1][3] + 1
143
144     #for j in range(1, row_sum):
145         #print(sum[j-1][2],sum[j-1][3])
146
147     #PosG
148     Edge_PosG = []
149     for i in range(0,Level_PosG +1):
150         Edge_PosG.append(0)
151     for i in range(0, len(Edge_PosG)):
152         if i == 0:
153             Edge_PosG[i] = First_PosG
154         elif i == len(Edge_PosG):
155             Edge_PosG[i] = Last_PosG
156         else:
157             Edge_PosG[i] = Edge_PosG[i-1] + Bin_PosG
158     j = 0
159     for i in range(0, len(Edge_PosG)-1):
160         sum[j][4] = (Edge_PosG[i] + Edge_PosG[i+1])/2
161         j = j + 1
162     for i in range(0, len(PosG_1)):
163         for j in range(1, len(Edge_PosG)):
164             if Edge_PosG[j-1] < PosG_1[i] < Edge_PosG[j]:
165                 sum[j-1][5] = sum[j-1][5] + 1
166
167     #for j in range(1, row_sum):
168         #print(sum[j-1][4],sum[j-1][5])
169
170     #FWHMG
171     Edge_FWHMG = []
172     for i in range(0,Level_FWHMG +1):
173         Edge_FWHMG.append(0)
174     for i in range(0, len(Edge_FWHMG)):
175         if i == 0:
176             Edge_FWHMG[i] = First_FWHMG
177         elif i == len(Edge_FWHMG):
178             Edge_FWHMG[i] = Last_FWHMG
179         else:
180             Edge_FWHMG[i] = Edge_FWHMG[i-1] + Bin_FWHMG
181     j = 0
182     for i in range(0, len(Edge_FWHMG)-1):
183         sum[j][6] = (Edge_FWHMG[i] + Edge_FWHMG[i+1])/2
184         j = j + 1
185     for i in range(0, len(FWHMG_1)):
186         for j in range(1, len(Edge_FWHMG)):
187             if Edge_FWHMG[j-1] < FWHMG_1[i] < Edge_FWHMG[j]:
188                 sum[j-1][7] = sum[j-1][7] + 1
189
190     #for j in range(1, row_sum):
191         #print(sum[j-1][6],sum[j-1][7])
192
193     #Pos2D
194     Edge_Pos2D = []
195     for i in range(0,Level_Pos2D +1):

```

```

196     Edge_Pos2D.append(0)
197 for i in range(0, len(Edge_Pos2D)):
198     if i == 0:
199         Edge_Pos2D[i] = First_Pos2D
200     elif i == len(Edge_Pos2D):
201         Edge_Pos2D[i] = Last_Pos2D
202     else:
203         Edge_Pos2D[i] = Edge_Pos2D[i-1] + Bin_Pos2D
204 j = 0
205 for i in range(0, len(Edge_Pos2D)-1):
206     sum[j][8] = (Edge_Pos2D[i] + Edge_Pos2D[i+1])/2
207     j = j + 1
208 for i in range(0, len(Pos2D_1)):
209     for j in range(1, len(Edge_Pos2D)):
210         if Edge_Pos2D[j-1] < Pos2D_1[i] < Edge_Pos2D[j]:
211             sum[j-1][9] = sum[j-1][9] + 1
212
213 #for j in range(1, row_sum):
214     #print(sum[j-1][8],sum[j-1][9])
215
216 #FWHM2D
217 Edge_FWHM2D = []
218 for i in range(0,Level_FWHM2D +1):
219     Edge_FWHM2D.append(0)
220 for i in range(0, len(Edge_FWHM2D)):
221     if i == 0:
222         Edge_FWHM2D[i] = First_FWHM2D
223     elif i == len(Edge_FWHM2D):
224         Edge_FWHM2D[i] = Last_FWHM2D
225     else:
226         Edge_FWHM2D[i] = Edge_FWHM2D[i-1] + Bin_FWHM2D
227 j = 0
228 for i in range(0, len(Edge_FWHM2D)-1):
229     sum[j][10] = (Edge_FWHM2D[i] + Edge_FWHM2D[i+1])/2
230     j = j + 1
231 for i in range(0, len(FWHM2D_1)):
232     for j in range(1, len(Edge_FWHM2D)):
233         if Edge_FWHM2D[j-1] < FWHM2D_1[i] < Edge_FWHM2D[j]:
234             sum[j-1][11] = sum[j-1][11] + 1
235
236 #for j in range(1, row_sum):
237     #print(sum[j-1][10],sum[j-1][11])
238
239 #print(sum)
240
241 print("\n#####\n")
242 for i in range (0, len(sum)):
243
244     print(sum[i][0],",",sum[i][1],",",sum[i][2],",",sum[i][3],",",sum[i][4],",",sum[i]
245           ][5],",",sum[i][6],",",sum[i][7],",",sum[i][8],",",sum[i][9],",",sum[i][10],",",s
246           um[i][11])
247
248 print("\n#####\n")
249 Net_file_name_with_csv = "Histograms_" + Filenamelpre + ".csv"
250 with open(Net_file_name_with_csv, "w") as sp:
251     #print("I(D)/I(G),I(2D)/I(G),Pos(G),FWHM(G),Pos(2D),FWHM(2D)", file=sp)
252     for i in range (0, len(sum)):
253
254         print(sum[i][0],",",sum[i][1],",",sum[i][2],",",sum[i][3],",",sum[i][4],",",s
255               um[i][5],",",sum[i][6],",",sum[i][7],",",sum[i][8],",",sum[i][9],",",sum[i][1
256               0],",",sum[i][11], file=sp)
257 sp.close()
258 print(Net_file_name_with_csv," has been saved. I hope you enjoy this world.")
259 #K = input()
260 #if K != "bye":

```

```
253     print("Au revoir")
```


Python function for statistical analysis of data

```

1  def StatAnalysis(FileName1pre):
2      import pandas as pd
3      import numpy as np
4      import csv
5      import scipy.stats
6      import statistics
7      from math import factorial
8      from time import gmtime, strftime
9      from datetime import datetime
10
11     print("\nSTATISTICAL ANALYSIS")
12
13     print("\n#####\n")
14     print("by Bhobnibhit C updated Mar 28rd 2018 version 1.01\n")
15
16     print("This program will calculate statistical parameters for you.\n")
17     print("Instructions\n***Input files must be .csv with commas as dividers***")
18
19     #print("\n#####\n")
20     #print("List of Analysis Method\n1) Descriptives including means, median, standard
21     deviation, skewness, sample size and standard error of mean\n2) Levene's test of
22     variance equality\n3) One-way ANOVA\n4) t-test (both equal and unequal variance)\n")
23
24     print("#####\n")
25     #Input Data
26     #FileName1pre = input("Please enter first filename: ")
27     #FileName2pre = input("Please enter second filename: ")
28     FileName1 = FileName1pre + ".csv"
29     FileName2 = FileName2pre + ".csv"
30
31     #Collect data file 1
32     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
33     NoSpectra_1 = np.asarray(list(zip(*c_reader))[0])
34     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
35     ID_IG_1 = np.asarray(list(zip(*c_reader))[2])
36     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
37     I2D_IG_1 = np.asarray(list(zip(*c_reader))[3])
38     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
39     PosG_1 = np.asarray(list(zip(*c_reader))[4])
40     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
41     FWHMG_1 = np.asarray(list(zip(*c_reader))[5])
42     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
43     Pos2D_1 = np.asarray(list(zip(*c_reader))[6])
44     c_reader = csv.reader(open(FileName1, 'r'), delimiter=',')
45     FWHM2D_1 = np.asarray(list(zip(*c_reader))[7])
46     ID_IG_1 = [float(i) for i in ID_IG_1]
47     I2D_IG_1 = [float(i) for i in I2D_IG_1]
48     PosG_1 = [float(i) for i in PosG_1]
49     FWHMG_1 = [float(i) for i in FWHMG_1]
50     Pos2D_1 = [float(i) for i in Pos2D_1]
51     FWHM2D_1 = [float(i) for i in FWHM2D_1]
52
53     #print(ID_IG_1)
54     #print(I2D_IG_1)
55     #print(PosG_1)
56     #print(FWHMG_1)
57     #print(Pos2D_1)
58     #print(FWHM2D_1)
59
60     #Collect data file 2
61     #c_reader = csv.reader(open(FileName2, 'r'), delimiter=',')
62     #NoSpectra_2 = np.asarray(list(zip(*c_reader))[0])
63     #c_reader = csv.reader(open(FileName2, 'r'), delimiter=',')
64     #ID_IG_2 = np.asarray(list(zip(*c_reader))[2])

```

```

60 #c_reader = csv.reader(open(Filename2, 'r'), delimiter=',')
61 #I2D_IG_2 = np.asarray(list(zip(*c_reader))[3])
62 #c_reader = csv.reader(open(Filename2, 'r'), delimiter=',')
63 #PosG_2 = np.asarray(list(zip(*c_reader))[4])
64 #c_reader = csv.reader(open(Filename2, 'r'), delimiter=',')
65 #FWHM2D_2 = np.asarray(list(zip(*c_reader))[5])
66 #c_reader = csv.reader(open(Filename2, 'r'), delimiter=',')
67 #Pos2D_2 = np.asarray(list(zip(*c_reader))[6])
68 #c_reader = csv.reader(open(Filename2, 'r'), delimiter=',')
69 #FWHM2D_2 = np.asarray(list(zip(*c_reader))[7])
70 #ID_IG_2 = [float(i) for i in ID_IG_2]
71 #I2D_IG_2 = [float(i) for i in I2D_IG_2]
72 #PosG_2 = [float(i) for i in PosG_2]
73 #FWHM2D_2 = [float(i) for i in FWHM2D_2]
74 #Pos2D_2 = [float(i) for i in Pos2D_2]
75 #FWHM2D_2 = [float(i) for i in FWHM2D_2]
76
77 #print(ID_IG_2)
78 #print(I2D_IG_2)
79 #print(PosG_2)
80 #print(FWHM2D_2)
81 #print(Pos2D_2)
82 #print(FWHM2D_2)
83
84 #Descriptives
85 print("\nStatistical Analysis\n")
86 print("File 1:", Filename1)
87
88 print("\n#####")
89 print("\nDescriptives\n")
90 print("Mean")
91 print("Group
92 1, ", np.mean(ID_IG_1), np.mean(I2D_IG_1), np.mean(PosG_1), np.mean(FWHM2D_1), np.mean(Pos2D_1), np.mean(FWHM2D_1))
93 #print("Group
94 2, ", np.mean(ID_IG_2), np.mean(I2D_IG_2), np.mean(PosG_2), np.mean(FWHM2D_2), np.mean(Pos2D_2), np.mean(FWHM2D_2))
95 print("Median")
96 print("Group
97 1, ", np.median(ID_IG_1), np.median(I2D_IG_1), np.median(PosG_1), np.median(FWHM2D_1), np.median(Pos2D_1), np.median(FWHM2D_1))
98 #print("Group
99 2, ", np.median(ID_IG_2), np.median(I2D_IG_2), np.median(PosG_2), np.median(FWHM2D_2), np.median(Pos2D_2), np.median(FWHM2D_2))
100 print("Mode")
101 print("Group
102 1, ", "\n", scipy.stats.mode(ID_IG_1), "\n", scipy.stats.mode(I2D_IG_1), "\n", scipy.stats.mode(PosG_1), "\n", scipy.stats.mode(FWHM2D_1), "\n", scipy.stats.mode(Pos2D_1), "\n", scipy.stats.mode(FWHM2D_1))
103 #print("Group
104 2, ", "\n", scipy.stats.mode(ID_IG_2), "\n", scipy.stats.mode(I2D_IG_2), "\n", scipy.stats.mode(PosG_2), "\n", scipy.stats.mode(FWHM2D_2), "\n", scipy.stats.mode(Pos2D_2), "\n", scipy.stats.mode(FWHM2D_2))
105 print("Standard Deviation")
106 print("Group 1, ", np.std(ID_IG_1, ddof = 1), np.std(I2D_IG_1, ddof = 1), np.std(PosG_1, ddof = 1), np.std(FWHM2D_1, ddof = 1), np.std(Pos2D_1, ddof = 1), np.std(FWHM2D_1, ddof = 1))
107 #print("Group 2, ", np.std(ID_IG_2, ddof = 1), np.std(I2D_IG_2, ddof = 1), np.std(PosG_2, ddof = 1), np.std(FWHM2D_2, ddof = 1), np.std(Pos2D_2, ddof = 1), np.std(FWHM2D_2, ddof = 1))
108 print("Sample Size")
109 print("Group
110 1, ", len(ID_IG_1), len(I2D_IG_1), len(PosG_1), len(FWHM2D_1), len(Pos2D_1), len(FWHM2D_1))
111 #print("Group
112 2, ", len(ID_IG_2), len(I2D_IG_2), len(PosG_2), len(FWHM2D_2), len(Pos2D_2), len(FWHM2D_2))
113 print("Standard Error")

```

```

105 print("Group 1,", scipy.stats.sem(ID_IG_1, ddof = 1), scipy.stats.sem(I2D_IG_1, ddof =
1), scipy.stats.sem(PosG_1, ddof = 1), scipy.stats.sem(FWHMG_1, ddof =
1), scipy.stats.sem(Pos2D_1, ddof = 1), scipy.stats.sem(FWHM2D_1, ddof = 1))
106 #print("Group 2,", scipy.stats.sem(ID_IG_2, ddof = 1), scipy.stats.sem(I2D_IG_2, ddof =
1), scipy.stats.sem(PosG_2, ddof = 1), scipy.stats.sem(FWHMG_2, ddof =
1), scipy.stats.sem(Pos2D_2, ddof = 1), scipy.stats.sem(FWHM2D_2, ddof = 1))
107 print("Skewness")
108 print("Group
1,", scipy.stats.skew(ID_IG_1), scipy.stats.skew(I2D_IG_1), scipy.stats.skew(PosG_1), sci
py.stats.skew(FWHMG_1), scipy.stats.skew(Pos2D_1), scipy.stats.skew(FWHM2D_1))
109 #print("Group
2,", scipy.stats.skew(ID_IG_2), scipy.stats.skew(I2D_IG_2), scipy.stats.skew(PosG_2), sci
py.stats.skew(FWHMG_2), scipy.stats.skew(Pos2D_2), scipy.stats.skew(FWHM2D_2))
110
111 Net_file_name = "StatisticalReport_" + Filenamelpre + ".txt"
112 with open(Net_file_name, "w") as sp:
113     print("Statiscal Analysis Report\n", file = sp)
114     print("File 1:", Filenamel, file = sp)
115
116     #print("\n#####
#####", file = sp)
117
118     print("\n#####
#####", file = sp)
119     print("\nDescriptives\n", file = sp)
120     print("Mean", file = sp)
121     print("Group
1,", np.mean(ID_IG_1), np.mean(I2D_IG_1), np.mean(PosG_1), np.mean(FWHMG_1), np.mean(P
os2D_1), np.mean(FWHM2D_1), file = sp)
122     #print("Group
2,", np.mean(ID_IG_2), np.mean(I2D_IG_2), np.mean(PosG_2), np.mean(FWHMG_2), np.mean(P
os2D_2), np.mean(FWHM2D_2), file = sp)
123     print("Median", file = sp)
124     print("Group
1,", np.median(ID_IG_1), np.median(I2D_IG_1), np.median(PosG_1), np.median(FWHMG_1), n
p.median(Pos2D_1), np.median(FWHM2D_1), file = sp)
125     #print("Group
2,", np.median(ID_IG_2), np.median(I2D_IG_2), np.median(PosG_2), np.median(FWHMG_2), n
p.median(Pos2D_2), np.median(FWHM2D_2), file = sp)
126     print("Mode", file = sp)
127     print("Group
1,", "\n", scipy.stats.mode(ID_IG_1), "\n", scipy.stats.mode(I2D_IG_1), "\n", scipy.sta
ts.mode(PosG_1), "\n", scipy.stats.mode(FWHMG_1), "\n", scipy.stats.mode(Pos2D_1), "\n
", scipy.stats.mode(FWHM2D_1), file = sp)
128     #print("Group
2,", "\n", scipy.stats.mode(ID_IG_2), "\n", scipy.stats.mode(I2D_IG_2), "\n", scipy.sta
ts.mode(PosG_2), "\n", scipy.stats.mode(FWHMG_2), "\n", scipy.stats.mode(Pos2D_2), "\n
", scipy.stats.mode(FWHM2D_2), file = sp)
129     print("Standard Deviation", file = sp)
130     print("Group 1,", np.std(ID_IG_1, ddof = 1), np.std(I2D_IG_1, ddof =
1), np.std(PosG_1, ddof = 1), np.std(FWHMG_1, ddof = 1), np.std(Pos2D_1, ddof =
1), np.std(FWHM2D_1, ddof = 1), file = sp)
131     #print("Group 2,", np.std(ID_IG_2, ddof = 1), np.std(I2D_IG_2, ddof =
1), np.std(PosG_2, ddof = 1), np.std(FWHMG_2, ddof = 1), np.std(Pos2D_2, ddof =
1), np.std(FWHM2D_2, ddof = 1), file = sp)
132     print("Sample Size", file = sp)
133     print("Group
1,", len(ID_IG_1), len(I2D_IG_1), len(PosG_1), len(FWHMG_1), len(Pos2D_1), len(FWHM2D_1
), file = sp)
134     #print("Group
2,", len(ID_IG_2), len(I2D_IG_2), len(PosG_2), len(FWHMG_2), len(Pos2D_2), len(FWHM2D_2
), file = sp)
135     print("Standard Error", file = sp)
136     print("Group 1,", scipy.stats.sem(ID_IG_1, ddof =
1), scipy.stats.sem(I2D_IG_1, ddof = 1), scipy.stats.sem(PosG_1, ddof =
1), scipy.stats.sem(FWHMG_1, ddof = 1), scipy.stats.sem(Pos2D_1, ddof =
1), scipy.stats.sem(FWHM2D_1, ddof = 1), file = sp)

```



```

135     #print("Group 2,", scipy.stats.sem(ID_IG_2, ddof =
136     1), scipy.stats.sem(I2D_IG_2, ddof = 1), scipy.stats.sem(PosG_2, ddof =
137     1), scipy.stats.sem(FWHMG_2, ddof = 1), scipy.stats.sem(Pos2D_2, ddof =
138     1), scipy.stats.sem(FWHM2D_2, ddof = 1), file = sp)
139     print("Skewness", file = sp)
140     print("Group
141     1,", scipy.stats.skew(ID_IG_1), scipy.stats.skew(I2D_IG_1), scipy.stats.skew(PosG_1)
142     , scipy.stats.skew(FWHMG_1), scipy.stats.skew(Pos2D_1), scipy.stats.skew(FWHM2D_1),
143     file = sp)
144     #print("Group
145     2,", scipy.stats.skew(ID_IG_2), scipy.stats.skew(I2D_IG_2), scipy.stats.skew(PosG_2)
146     , scipy.stats.skew(FWHMG_2), scipy.stats.skew(Pos2D_2), scipy.stats.skew(FWHM2D_2),
147     file = sp)
148     print("\n#####For
149     Excel#####\n", file = sp)
150     print(np.mean(ID_IG_1), ", ", np.std(ID_IG_1, ddof =
151     1), ", ", len(ID_IG_1), ", ", np.mean(I2D_IG_1), ", ", np.std(I2D_IG_1, ddof =
152     1), ", ", len(I2D_IG_1), ", ", np.mean(PosG_1), ", ", np.std(PosG_1, ddof =
153     1), ", ", len(PosG_1), ", ", np.mean(FWHMG_1), ", ", np.std(FWHMG_1, ddof =
154     1), ", ", len(FWHMG_1), ", ", np.mean(Pos2D_1), ", ", np.std(Pos2D_1, ddof =
155     1), ", ", len(Pos2D_1), ", ", np.mean(FWHM2D_1), ", ", np.std(FWHM2D_1, ddof =
156     1), ", ", len(FWHM2D_1), file = sp)
157
158     print("\n#####
159     #####", file = sp)
160     print("\nRaw data\n", file=sp)
161     print("File, I(D)/I(G), I(2D)/I(G), Pos(G), FWHM(G), Pos(2D), FWHM(2D)", file=sp)
162     for i in range(0, len(NoSpectra_1)):
163
164         print("1, ", ID_IG_1[i], ", ", I2D_IG_1[i], ", ", PosG_1[i], ", ", FWHMG_1[i], ", ", Pos2D_
165         1[i], ", ", FWHM2D_1[i], file = sp)
166         #for i in range(0, len(NoSpectra_2)):
167
168             #print("2, ", ID_IG_2[i], ", ", I2D_IG_2[i], ", ", PosG_2[i], ", ", FWHMG_2[i], ", ", Pos2D
169             _2[i], ", ", FWHM2D_2[i], file = sp)
170
171     print("\n\n#####
172     #####\n\nThis file was recorded
173     on", str(datetime.now()), file = sp)
174
175     print("\n\n#####
176     #####\n\nEnd of File", file=sp)
177
178     sp.close()
179
180     print("\n\n#####
181     #####\n\nEnd of Analysis\nYour results has been saved as",
182     Net_file_name)
183
184     print("\n\n#####
185     #####\n\nC'est fini. Merci, au revoir")

```

Python function for averaging Raman spectra

```

1  def AverageGraph(Prefix, first_file_number, Last_file_number):
2      import pandas as pd
3      import numpy as np
4      import csv
5      from math import factorial
6
7      print("\nAutomatic Graph Averager")
8      print("by Bhobnibhit C updated Mar 2nd 2018 version 1.02\n")
9
10     print("This program will average your graphs for representing a set of data.\n")
11     print("Instructions\n****Input files must be .csv with commas as dividers****\n ")
12     print("****Normalization will be done before average****\n")
13
14     #create array for keeping all data
15     R_RamanShift = []
16     for i in range(10000):
17         R_RamanShift.append(0)
18     R_Intensity = []
19     Sum_Intensity = []
20     for i in range(10000):
21         R_Intensity.append(0)
22         Sum_Intensity.append(0)
23     NoData = 0
24
25
26     #Prefix = input('Enter input file prefix: ')
27     #first_file_number = Current_file_number = int(input('First file number: '))
28     #Last_file_number = int(input('Last file number: '))
29     Current_file_number = first_file_number
30     filename1 = 'love'
31
32     print("\n#####")
33     print("****Analysis in Progress****\n")
34     #print("\n")
35     #print(Prefix)
36     #print(first_file_number)
37     #print(Last_file_number)
38     #print(filename1)
39
40     filename1 = Prefix + "000" + str(Current_file_number) + ".csv"
41     c_reader = csv.reader(open(filename1, 'r'), delimiter=',')
42     R_RamanShiftpre = list(zip(*c_reader))[0]
43     for i in range(0, len(R_RamanShiftpre)):
44         R_RamanShift[i] = float(R_RamanShiftpre[i])
45
46     #print(R_RamanShift)
47     RamanShift_Full = []
48     Intensity_Full = []
49     for i in range(0,100000):
50         RamanShift_Full.append(0)
51         Intensity_Full.append(0)
52
53     #making file name
54     while first_file_number-1 < Current_file_number < Last_file_number +1:
55         if 0 < Current_file number < 10:
56             filename1 = Prefix + "000" + str(Current_file_number) + ".csv"
57             filename_input = Prefix + "000" + str(Current_file_number)
58             #print(filename1)
59             Current_file_number = Current_file_number + 1
60         elif 9 < Current_file_number < 100:
61             filename1 = Prefix + "00" + str(Current_file_number) + ".csv"
62             filename_input = Prefix + "000" + str(Current_file_number)
63             #print(filename1)
64             Current_file_number = Current_file_number + 1
65         elif 99 < Current_file_number < 1000:
66             filename1 = Prefix + "0" + str(Current_file_number) + ".csv"

```

```

66     filename_input = Prefix + "000" + str(Current_file_number)
67     #print(filename1)
68     Current_file_number = Current_file_number + 1
69     c_reader = csv.reader(open(filename1, 'r'), delimiter=',')
70     RamanShift_Fullpre = list(zip(*c_reader))[0]
71     c_reader = csv.reader(open(filename1, 'r'), delimiter=',')
72     Intensity_Fullpre = list(zip(*c_reader))[1]
73
74     #cut range into 1000 to 3400 and finding minimum
75     for i in range (0, len(RamanShift_Fullpre)):
76         RamanShift_Full[i] = float(RamanShift_Fullpre[i])
77         Intensity_Full[i] = float(Intensity_Fullpre[i])
78     #print(RamanShift_Full)
79     #print(Intensity_Full)
80     RamanShift_1000to3400 = []
81     Intensity_1000to3400 = []
82     for k in range(0,100000):
83         RamanShift_1000to3400.append(0)
84         Intensity_1000to3400.append(0)
85     k = 0
86     for i in range (0, len(RamanShift_Full)):
87         while 1000 < RamanShift_Full[i] < 3400 :
88             #print(RamanShift_Full[i],",",Intensity_Full[i])
89             RamanShift_1000to3400[k] = float(RamanShift_Full[i])
90             Intensity_1000to3400[k] = float(Intensity_Full[i])
91             k = k + 1
92             break
93     BaselineCorrector = np.amin(Intensity_1000to3400)
94
95     #finding multiplier
96     RamanShift_1450to1700 = []
97     Intensity_1450to1700 = []
98     for k in range(0,100000):
99         RamanShift_1450to1700.append(0)
100        Intensity_1450to1700.append(0)
101    k = 0
102    for i in range (0, len(RamanShift_Full)):
103        while 1450 < RamanShift_Full[i] < 1700 :
104            RamanShift_1450to1700[k] = float(RamanShift_Full[i])
105            Intensity_1450to1700[k] = float(Intensity_Full[i])
106            k = k + 1
107            break
108    SignalModulator = np.amax(Intensity_1450to1700)
109    #print(R_RamanShift)
110    #print(Intensity_Full)
111    #print(RamanShift_Full)
112    for i in range(0, len(R_RamanShift)):
113        #print("Check")
114        #print(RamanShift_Full[i],R_RamanShift[i])
115        if RamanShift_Full[i] == R_RamanShift[i]:
116            #print("Correct")
117            Sum_Intensity[i] = float(Sum_Intensity[i]) + (Intensity_Full[i] -
118                BaselineCorrector)/SignalModulator
119    print("@",str(Current_file_number-1))
120    NoData = NoData + 1
121    #print(NoData)
122    #print(Sum_Intensity)
123    for i in range(0, len(Sum_Intensity)):
124        R_Intensity[i] = Sum_Intensity[i]/NoData
125
126    filename_reserved = "Average_" + Prefix + "from" + str(first_file_number) + "to" +
127    str(Current_file_number - 1) + ".csv"
128    with open(filename_reserved, "w") as sp:
129        for i in range (0, len(R_RamanShift)):
130            if R_RamanShift != 0 and R_Intensity[i] != 0:
131                #print(R_RamanShift[i],",",R_Intensity[i], file=sp)
132                #print(R_RamanShift[i],",",R_Intensity[i])

```

```
131     sp.close()
132
133     print("\n#####\n")
134     print("Your result has been recorded as ",filename_reserved)
135     print("****C'est fini!!! Merci, Au revoir****")
136
```

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