



## REFERENCES

- Arsham, H. (1996) *Deterministic Modeling: Linear Optimization with Applications*. <http://home.ubalt.edu/ntsbarsh/index.html>, 12 December 2003.
- Aseeri, A. and Bagajewicz, M.J. (2003) New measures and procedures to manage financial risk with applications to the planning of gas commercialization in Asia. *Computer & Chemical Engineering*, Submitted.
- Baird, C.T. (1989) *Guide to Petroleum Product Blending*. Austin, TX: HPI Consultants Inc.
- Barbaro, A. and Bagajewicz, M.J. (2003) Managing financial risk in planning under uncertainty. *AIChE Journal*, Submitted.
- Beale, E.M.L. (1955) On minimizing a convex function subject to linear inequalities. *Journal of Royal Statistical Society, Series B*, 17, 173-184.
- Birge, J.R. and Louveaux, F. (1997) *Introduction to Stochastic Programming*. New York: Springer.
- Bonfill, A., Cantón, J., Bagajewicz, M.J., Espuña, A., and Puigjaner, L. (2003) *Managing financial risk in scheduling of batch plants*. ESCAPE 13 Meeting, Lappeenranta, Finland, June 1-4, 2003.
- Bopp, A.E., Kannan, V.R., Palocsay, S.W., and Stevens, S.P. (1996) An optimization model for planning natural gas purchases, transportation, storage and deliverability. *Omega*, 24(5), 511-522.
- Brooke, A., Kendrick, D., Meeraus A., and Raman, R. (1998) *GAMS: A User's Guide*. Washington D.C.: GAMS Development Corporation.
- Chevron Research Company, (1971) *31.0° API Iranian Heavy Crude Oil*. Chevron Oil Trading Company.
- Edgar, T.F., Himmelblau, D.M., and Lasdon, L.S. (2001) *Optimization of Chemical Processes 2<sup>nd</sup> edition*. New York: McGraw-Hill.
- EPPO (2003) *Energy Data Notebook Quarterly Report*. Energy Policy and Planning Office, Ministry of Energy, Thailand.

- Escudero, L.F., Quintana, F.J., and Salmerón, J. (1999) CORO, a modeling and an algorithmic framework for oil supply, transformation and distribution optimization under uncertainty. European Journal of Operational Research, 114, 638-656.
- Favennec, J.P. (Ed.). (2001) Refinery operation and management. Petroleum Refining Vol. 5. Paris: Technip.
- Floudas, C.A. (1995) Nonlinear and Mixed-Integer Optimization Fundamentals and Applications. Oxford: Oxford University Press.
- Gary, J.H. and Handwerk, G.E. (2001) Petroleum Refining Technology and Economics 4<sup>th</sup> edition. New York: Marcel Dekker.
- Glismann, K. and Gruhn, G. (2001) Short-term scheduling and recipe optimization of blending processes. Computers & Chemical Engineering, 25, 627-634.
- Göthe-Lundgren, M., Lundgren, J.T., and Persson, J.A. (2002) An optimization model for refinery production scheduling. International Journal of Production Economics, 78, 255-270.
- Guldmann, J.M. and Wang, F. (1999) Optimizing the natural gas supply mix of local distribution utilities. European Journal of Operational Research, 112, 598-612.
- Hsieh, S. and Chiang, C.C. (2001) Manufacturing-to-sale planning model for fuel oil production. Advanced Manufacturing Technology, 18, 303-311.
- Jia, Z. and Ierapetritou, M. (2003) Mixed-integer linear programming model for gasoline blending and distribution scheduling. Industrial & Engineering Chemistry Research, 42, 825-835.
- Jia, Z., Ierapetritou, M., and Kelly, J.D. (2003) Refinery short-term scheduling using continuous time formulation: crude-oil operations. Industrial & Engineering Chemistry Research, 42, 3085-3097.
- Joly, M., Moro, L.F.L., and Pinto, J.M. (2002) Planning and scheduling for petroleum refineries using mathematical programming. Brazilian Journal of Chemical Engineering, 19(2), 207-228.

- Lababidi, H.M.S., Ahmed, M.A., Alatiqi, I.M., and Al-Enzi, A.F. (2004) Optimizing the supply chain of a petrochemical company under uncertain operating and economic conditions. Industrial & Engineering Chemistry Research, 43, 63-73.
- Lee, H., Pinto, J.M., Grossmann, I.E., and Park, S. (1996) Mixed-integer linear programming model for refinery short-term scheduling of crude oil unloading with inventory management. Industrial & Engineering Chemistry Research, 35, 1630-1641.
- Linsmeier, T.J. and Pearson, N.D. (2000) Value at Risk. Financial Analysts Journal, 56(2), 2000.
- Liu, M.L. and Sahinidis, N.V. (1996) Optimization in process planning under uncertainty. Industrial & Engineering Chemistry Research, 35, 4154-4165.
- Liu, M.L. and Sahinidis, N.V. (1997) Process planning in a fuzzy environment. European Journal of Operational Research, 100, 142-169.
- Maurin, H. (1967) Programmation Linéaire Appliquée. Paris: Éditions Technip.
- Maples, R.E. (2000) Petroleum Refinery Process Economics 2<sup>nd</sup> edition. Tulsa, OK: Pennwell Corporation.
- Moro, L.F.L., Zanin, A.C., and Pinto, J.M. (1998) A planning model for refinery diesel production. Computers & Chemical Engineering, 22, Suppl., S1039-1042.
- Moro, L.F.L. and Pinto, J.M. (2004) Mixed-integer programming approach for short-term crude oil scheduling. Industrial & Engineering Chemistry Research, 43, 85-94.
- Neiro, S.M.S. and Pinto, J.M. (2003) Supply chain optimization of petroleum refinery complexes. FOCAPO proceedings, Coral Springs, FL, 59-72.
- Pelham, R. and Pharris, C. (1996) Refinery operations and control: a future vision. Hydrocarbon Processing, 75(7), 89-94.
- Pinto, J.M., Joly, M., and Moro, L.F.L. (2000) Planning and scheduling models for refinery operations. Computers & Chemical Engineering, 24, 2259-2276.
- Pinto, J.M. and Moro, L.F.L. (2000) A planning model for petroleum refineries. Brazilian Journal of Chemical Engineering, 17(4-7), 575-586.

- Shah, N. (1998) Single and multisite planning and scheduling: Current status and future challenges. FOCAPO AIChE symposium series, 94(320), 91-110.
- Shah, N. (1996) Mathematical programming techniques for crude oil scheduling. Computers & Chemical Engineering, 20, Suppl., S1227-S1232.
- Simons, R. (1995) Planning and scheduling BP's oil refineries – New and old approaches. MP in Action, February.
- Simons, R. (1997) MIMI brings OR tools together. MP in Action, December.
- Swift, A.O. (2000) Control and optimization. In A.G. Lucas (Ed.), Modern Petroleum Technology: Volume 2 Downstream. 6th ed. Chichester (UK): John Wiley & Sons.
- Tsiakis, P., Shah, N., and Pantelides, C.C. (2001) Design of multi-echelon supply chain networks under demand uncertainty. Industrial & Engineering Chemistry Research, 40, 3585-3604.
- Wenkai, L., Hui, C.W., Hua, B., and Tong, Z. (2002) Scheduling crude oil unloading, storage, and processing. Industrial & Engineering Chemistry Research, 41, 6723-6734.
- Zhang, J., Zhu, X.X., and Towler, G.P. (2001) A level-by-level debottlenecking approach in refinery operation. Industrial & Engineering Chemistry Research, 40, 1528-1540.
- Zhang, J., Zhu, X.X., and Towler, G.P. (2001) A simultaneous optimization strategy for overall integration in refinery planning. Industrial & Engineering Chemistry Research, 40, 2640-2653.
- Zhang, N. and Zhu, X.X. (2000) A novel modeling and decomposition strategy for overall refinery optimization. Computers & Chemical Engineering, 24, 1543-1548.

## APPENDICES

### Appendix A Nomenclature.

#### Indices

$t$	Set of time periods
$c$	Set of all commodities
$o$	Set of crude oils
$p$	Set of products
$i$	Set of intermediates
$u$	Set of productive units
$q$	Set of properties
$s$	Set of scenarios

#### Parameters

$pro_{u,c,q}$	Property $q$ of commodity $c$ from unit $u$	
$px_{p,q}$	Maximum property $q$ of product $p$	
$pn_{p,q}$	Minimum property $q$ of product $p$	
$cyield_{o,c}$	Percent of component $c$ in crude oil $o$	(%)
$yield_{u,c}$	Percent yield of commodity $c$ from unit $u$	(%)
$dem_{p,t}$	Demand of product $p$ in time period $t$	( $m^3$ )
$ux_u$	Maximum capacity of unit $u$	( $m^3$ )
$un_u$	Minimum capacity of unit $u$	( $m^3$ )
$ox_o$	Maximum monthly purchase of crude oil $o$	( $m^3$ )
$on_o$	Minimum monthly purchase of crude oil $o$	( $m^3$ )
$stox_p$	Maximum storage capacity of product $p$	( $m^3$ )
$cp_{p,t}$	Unit sale price of product $p$ in time period $t$	(\$/ $m^3$ )
$co_{o,t}$	Unit purchase price of crude oil $o$ in time period $t$	(\$/ $m^3$ )
$ci_{i,t}$	Unit purchase price of intermediate $i$ in time period $t$	(\$/ $m^3$ )
$cl_{p,t}$	Unit cost of lost demand penalty for product $p$ in time period $t$	(\$/ $m^3$ )
$\rho_s$	Probability of scenario $s$	
$density_u$	Density of feed to unit $u$	( $ton/m^3$ )

$fuel_u$	Percent energy consumption for unit $u$ based on tFOE	(%)
$disc$	Percent discount from normal price	(%)

### Variables

$PO_{u,c,q,t}$	Property $q$ of commodity $c$ from unit $u$ in time period $t$	
$AF_{u,t}$	Amount of feed to unit $u$ in time period $t$	( $m^3$ )
$AO_{u,c,t}$	Amount of outlet commodity $c$ from unit $u$ in time period $t$	( $m^3$ )
$A_{u,c,u',t}$	Amount of commodity $c$ flow between unit $u$ and unit $u'$ in time period $t$	( $m^3$ )
$MANU_{p,t}$	Amount of product $p$ produced in time period $t$	( $m^3$ )
$AC_{o,t}$	Amount of crude oil $o$ refined in time period $t$	( $m^3$ )
$AI_{i,t}$	Amount of intermediate $i$ added in time period $t$	( $m^3$ )
$AS_{p,t}$	Amount of product $p$ stored in time period $t$	( $m^3$ )
$AL_{p,t}$	Amount of lost demand for product $p$ in time period $t$	( $m^3$ )
$AD_{p,t}$	Amount of discount product sold in time period $t$	( $m^3$ )
$Burnt_{p,t}$	Amount of product $p$ burnt in time period $t$	( $m^3$ )
$Used_t$	Amount of fuel used in time period $t$	(tFOE)
$TP_{p,t}$	Income from selling product $p$ in time period $t$	(\$)
$TO_{o,t}$	Expense from purchasing crude oil $o$ in time period $t$	(\$)
$TI_{i,t}$	Expense from purchasing intermediate in time period $t$	(\$)
$TS_{p,t}$	Expense from storage product $p$ in time period $t$	(\$)
$TL_{p,t}$	Expense from lost demand of product $p$ in time period $t$	(\$)
$TD_{p,t}$	Expense from discount sales of product $p$ in time period $t$	(\$)
$sales_{p,t}$	Sales of product $p$ in time period $t$	( $m^3$ )

## Appendix B Blending equation for each property.

The product properties used in this model can be calculated by based on the Equation (3.5). The following equations show how each property is calculated.

### (1) Octane blending (Gary and Handwerk, 2001)

Octane numbers are blended on a volumetric basis using the blending octane numbers of the components. In practice true octane numbers do not blend linearly and it is necessary to use blending octane numbers in making calculations.

$$ON_{blend} = \sum V_i(ON)_i \quad (B.1)$$

where:

$$\begin{aligned} ON_{blend} &= \text{blending octane numbers of total blend} \\ ON_i &= \text{blending octane numbers of component } i \\ V_i &= \text{volume fraction of component } i \end{aligned}$$

### (2) RVP blending (Baird, 1989)

The theoretical method for blending to the desired Reid vapor pressure requires that the average molecular weight of each of the streams be known. Although there are accepted ways of estimating the average molecular weight of a refinery stream from boiling point, gravity, and characterization factor, none of these methods lend themselves to use in linear programming models. A more convenient way is to use the empirical method developed by Chevron Research Company. Reid Vapor Pressure Blending Indices (RVPI) have been compiled as a function of the RVP of the blending streams. The Reid vapor pressure of the blend is closely approximated by the sum of all the products of volume fraction (V) times the RVPI for each component. The Chevron equation for RVP follows:

$$RVPI = RVP^{1.25} \quad (B.2)$$

$$RVPI_{blend} = \sum V_i(RVPI)_i \quad (B.3)$$

where:

$RVP$	= Reid vapor pressure, psia
$RVPI_{blend}$	= Reid vapor pressure blending indices of total blend
$RVPI_i$	= Reid vapor pressure blending indices of component i
$V_i$	= volume fraction of component i

### (3) Aromatic blending (Baird, 1989)

Aromatic blending can be calculated by the sum of all products of volume fraction (V) times the aromatic content for each component.

$$\%Aro_{blend} = \sum V_i(\%Aro)_i \quad (B.4)$$

where:

$\%Aro_{blend}$	= percent of aromatic content in total blend
$\%Aro_i$	= percent of aromatic content in component i
$V_i$	= volume fraction of component i

### (4) Freezing point (Baird, 1989)

Maurin (1967) presents a direct method for the weight blending of freeze points. The freezing point of the blend can be approximated by the sum of all the products of weight fraction (W) times the freezing point index (FPI) for each component. The blending index numbers for this system are presented in Table B1.

$$FPI_{blend} = \sum W_i(FPI)_i \quad (B.5)$$

$$W_i = \frac{V_i \times S.G._i}{\sum V_i \times S.G._i} \quad (B.6)$$

where:

$FPI_{blend}$	= freezing point indices of total blend
$FPI_i$	= freezing point indices of component i
$W_i$	= weight fraction of component i

$V_i$  = volume fraction of component i  
 $S.G._i$  = specific gravity of component i

(5) Cetane index blending (Baird, 1989)

Cetane number blending can be estimated linearly on the volume basis by the sum of all products of volume fraction (V) times the cetane blending index for each component. Although this is an approximate method, it can give the satisfactory result.

$$CI_{blend} = \sum V_i(CI)_i \quad (B.7)$$

where:

$CI_{blend}$  = cetane indices of total blend  
 $CI_i$  = cetane indices of component i  
 $V_i$  = volume fraction of component i

(6) Viscosity blending (Baird, 1989)

Viscosity is not an additive property and it is necessary to use special techniques to estimate the viscosity of a mixture from the viscosities of its components. The use of blending index systems provides a much simpler approach to viscosity blending and is most often used in linear programming models to represent the viscosity blending of distillate and residual fuels. It is usually true to a satisfactory approximation that the viscosity blending index numbers (VBN) of the blend will be the sum of all the products of the weight fraction times the VBN for each component. In equation form:

$$VBN_{blend} = \sum (W_i \times VBN_i) \quad (B.8)$$

where:

$VBN_{blend}$  = viscosity blending index numbers of total blend  
 $VBN_i$  = viscosity blending index numbers of component i

$W_i$  = weight fraction of component i

Blending of kinematic viscosities (centistokes) may be done at any temperature, but the viscosities of all components of the blend must be expressed at the same temperature. The Refutas system can be used to calculate VBN. The equations for the Refutas blending index numbers are as follows:

$$VBN = 10.975 + 14.535(\ln(\ln(CST + 0.8))) \quad (B.9)$$

$$CST = \exp(\exp((VBN - 10.975)/14.535)) - 0.8 \quad (B.10)$$

where:

$VBN$  = viscosity blending index number

$CST$  = viscosity in centistokes

#### (7) Sulfur blending (Maples, 2000)

Sulfur blending can be calculated linearly on the mass basis by the sum of all products of weight fraction ( $W$ ) times the sulfur content for each component.

$$\%Sulfur_{blend} = \sum W_i (\%Sulfur)_i \quad (B.11)$$

$$W_i = \frac{V_i \times S.G._i}{\sum V_i \times S.G._i} \quad (B.12)$$

where:

$\%Sulfur_{blend}$  = percent of sulfur content in total blend

$\%Sulfur_i$  = percent of sulfur content in component i

$W_i$  = weight fraction of component i

$V_i$  = volume fraction of component i

$S.G._i$  = specific gravity of component i

## (8) Specific gravity blending (Maples, 2000)

The specific gravity of a blend can be estimated very accurately from the sum of all products of volume fraction (V) times the specific gravity for each component.

$$SG_{blend} = \sum V_i(\%SG)_i \quad (B.13)$$

where:

$$\begin{aligned} SG_{blend} &= \text{specific gravity of total blend} \\ SG_i &= \text{specific gravity of component } i \\ V_i &= \text{volume fraction of component } i \end{aligned}$$

## (9) Pour point blending (Baird, 1989)

Hu and Burns (1970) presented equations for estimating these properties of blends. They employed the concept of substituting blending index values for given properties. Pour point can be determined from the following equation:

$$PPI = 10,000 \frac{\left[ (PP + 459.69)^{1/x} \right]}{\left[ (140 + 459.69)^{1/x} \right]} \quad (B.14)$$

$$PP = \left[ (PPI) \frac{(459.69 + 140)^{(1/x)}}{10000} \right]^x - 459.69 \quad (B.15)$$

where:

$$\begin{aligned} PP &= \text{Pour point, } ^\circ\text{F} \\ PPI &= \text{Pour point blending index} \\ x &= \text{Constant} \end{aligned}$$

Although the optimum value for the constant x must be determined experimentally for a given refinery, Hu and Burns determined that x values of 0.08 for pour points gave the best results on an industry-wide basis.

The pour point of a blend can be estimated very accurately from the sum of all products of volume fraction (V) times the pour point index for each component.

$$PPI_{blend} = \sum V_i(PPI)_i \quad (B.16)$$

where:

- $PPI_{blend}$  = pour point indices of total blend
- $PPI_i$  = pour point indices of component i
- $V_i$  = volume fraction of component i

**Table B1** Maurin freeze point blending indices for weight blending (Maurin, 1967)

Freeze Point, °C	Blending Index	Freeze Point, °C	Blending Index	Freeze Point, °C	Blending Index
-50	11.2	-9	28.3	16	65.3
-45	12.2	-8	29.0	17	68.2
-40	14.0	-7	29.8	18	71.0
-35	15.8	-6	30.7	19	74.2
-30	17.6	-5	31.8	20	77.4
-29	17.9	-4	32.9	21	80.7
-28	18.3	-3	34.0	22	83.9
-27	18.7	-2	35.0	23	87.1
-26	19.0	-1	36.1	24	90.4
-25	19.4	0	37.2	25	94.0
-24	19.8	1	38.3	26	97.6
-23	20.3	2	39.5	27	101.4
-22	20.8	3	40.9	28	105.8
-21	21.3	4	42.4	29	110.1
-20	21.8	5	43.8	30	115.4
-19	22.3	6	45.2		
-18	22.9	7	46.7		
-17	23.4	8	48.4		
-16	24.0	9	50.2		
-15	24.5	10	52.0		
-14	25.0	11	54.2		
-13	25.6	12	56.3		
-12	26.2	13	58.5		
-11	26.9	14	60.6		
-10	27.6	15	62.8		

**Appendix C Data of commodities and productive units.****Table C1** Fuel used in processing unit (expressed in fuel oil equivalence)  
(Favenneec, 2001)

<b>Units</b>	<b>Fuel used (wt%)</b>
CDU2	1.8
CDU3	1.8
NPU2	2
NPU3	2
ISOU	4
CRU2	2.5
CRU3	2.5
KTU	2
GO-HDS	2
DGO-HDS	2

**Table C2** Oman crude specification

<b>API Gravity</b>	34.80		<b>% Sulfur</b>	1.160	
Methane	vol%	0.00	Iso-butane	vol%	0.30
Ethane	vol%	0.02	N-butane	vol%	0.92
Propane	vol%	0.33			

Description		Component Fraction							
		FG	LPG	LN	MN	HN	IK	DO+GO	FO
Vol. yield on crude	lv%	0.02	1.55	5.33	2.70	6.30	13.80	22.40	46.30
Aromatics content	lv%	-	-	1.20	4.25	8.24	11.94	20.94	-
Cetane index		-	-	-	-	30.10	46.40	54.10	-
Freeze point	°C	-	-	-	-85.50	-74.60	-53.50	-8.80	-
RONC		-	-	69.50	49.20	40.60	27.60	-	-
RVP	kg/cm <sup>2</sup>	-	-	0.70	0.16	0.04	0.00	-	-
Specific gravity		-	-	0.6517	0.7119	0.7385	0.7844	0.8447	0.9367
Sulfur	wt%	-	-	0.012	0.027	0.030	0.108	0.687	1.938
Viscosity @ 50 °C	cSt	-	-	-	0.41	0.54	1.01	3.64	609.00
Viscosity @ 100 °C	cSt	-	-	-	0.34	0.40	0.62	1.66	52.22
Pour point	°C	-	-	-	-	-	-77.90	-	7.00

**Table C3** Tapis crude specification

<b>API Gravity</b>	44.50		<b>% Sulfur</b>	0.025	
Methane	vol%	0.00	Iso-butane	vol%	0.82
Ethane	vol%	0.54	N-butane	vol%	1.21
Propane	vol%	0.66			

Description		Component Fraction							
		FG	LPG	LN	MN	HN	IK	DO+GO	FO
Vol. yield on crude	lv%	0.54	2.69	3.27	5.70	10.70	21.90	30.40	21.50
Aromatics content	lv%	-	-	1.78	5.11	13.09	16.82	17.41	-
Cetane index		-	-	-	-	20.90	45.10	59.30	33.30
Freeze point	°C	-	-	-	-	-83.50	-51.10	6.00	-
RONC		-	-	81.70	76.00	68.20	60.30	-	-
RVP	kg/cm <sup>2</sup>	-	-	0.66	0.15	0.05	0.00	-	-
Specific gravity		-	-	0.6713	0.7247	0.7557	0.7857	0.8271	0.9175
Sulfur	wt%	-	-	0.000	0.000	0.001	0.004	0.034	0.056
Viscosity @ 50 °C	cSt	-	-	-	0.43	0.55	0.96	2.88	15.26
Viscosity @ 100 °C	cSt	-	-	-	0.31	0.37	0.58	1.37	4.59
Pour point	°C	-	-	-	-	-	-63.40	-	58.40

**Table C4** Labuan crude specification

<b>API Gravity</b>	31.80		<b>% Sulfur</b>	0.080	
Methane	vol%	0.00	Iso-butane	vol%	0.18
Ethane	vol%	0.02	N-butane	vol%	0.36
Propane	vol%	0.22			

Description		Component Fraction (%vol)							
		FG	LPG	LN	MN	HN	IK	DO+GO	FO
Vol. yield on crude	lv%	0.02	0.76	2.42	3.20	9.00	20.30	42.70	20.70
Aromatics content	lv%	-	-	7.05	0.64	16.13	26.36	41.67	-
Cetane index		-	-	-	-	11.30	30.20	39.20	-
Freeze point	°C	-	-	-	-	-	-67.80	-8.10	-
RONC		-	-	83.20	76.40	73.60	50.20	-	-
RVP	kg/cm <sup>2</sup>	-	-	0.64	0.16	0.04	0.00	-	-
Specific gravity		-	-	0.6898	0.7402	0.7759	0.8280	0.8911	0.9530
Sulfur	wt%	-	-	0.001	0.001	0.002	0.017	0.083	0.175
Viscosity @ 50 °C	cSt	-	-	-	0.53	0.62	1.04	3.31	132.07
Viscosity @ 100 °C	cSt	-	-	-	0.35	0.41	0.63	1.46	14.50
Pour point	°C	-	-	-	-	-	-86.50	-	45.10

**Table C5** Seria light crude specification

<b>API Gravity</b>	35.80		<b>% Sulfur</b>	0.068	
Methane	vol%	0.00	Iso-butane	vol%	0.26
Ethane	vol%	0.00	N-butane	vol%	0.62
Propane	vol%	0.25			

Description		Component Fraction (%vol)							
		FG	LPG	LN	MN	HN	IK	DO+GO	FO
Vol. yield on crude	lv%	0.00	1.13	4.77	4.00	11.30	23.10	35.00	19.50
Aromatics content	lv%	-	-	2.65	8.19	15.88	24.28	53.56	-
Cetane index		-	-	-	-	12.80	31.60	43.00	-
Freeze point	°C	-	-	-	-	-	-59.60	-6.40	-
RONC		-	-	79.50	68.00	60.70	49.60	-	-
RVP	kg/cm <sup>2</sup>	-	-	0.69	0.16	0.04	0.00	-	-
Specific gravity		-	-	0.6798	0.7415	0.7696	0.8200	0.8781	0.9506
Sulfur	wt%	-	-	0.000	0.001	0.003	0.020	0.080	0.155
Viscosity @ 50 °C	cSt	-	-	-	0.21	0.21	0.21	0.34	132.96
Viscosity @ 50 °C	cSt	-	-	-	0.21	0.21	0.21	0.27	16.87
Pour point	°C	-	-	-	-	-	-65.80	-	35.90

**Table C6** Phet crude specification

<b>API Gravity</b>	40.70		<b>% Sulfur</b>	0.050	
Methane	vol%	0.00	Iso-butane	vol%	0.37
Ethane	vol%	0.07	N-butane	vol%	1.04
Propane	vol%	0.37			

Description		Component Fraction (%vol)							
		FG	LPG	LN	MN	HN	IK	DO+GO	FO
Vol. yield on crude	lv%	0.07	1.78	3.05	3.70	8.50	15.00	28.10	38.00
Aromatics content	lv%	-	-	1.05	5.93	12.15	14.42	14.58	-
Cetane index		-	-	-	-	22.60	45.60	61.40	-
Freeze point	°C	-	-	-	-	-88.80	-48.90	13.40	-
RONC		-	-	70.00	61.40	53.50	41.60	-	-
RVP	kg/cm <sup>2</sup>	-	-	0.71	0.16	0.04	0.00	-	-
Specific gravity		-	-	0.6662	0.7200	0.7502	0.7840	0.8236	0.8941
Sulfur	wt%	-	-	0.000	0.000	0.001	0.006	0.047	0.087
Viscosity @ 50 °C	cSt	-	-	-	0.35	0.48	0.94	3.12	39.72
Viscosity @ 50 °C	cSt	-	-	-	0.22	0.26	0.52	1.50	10.59
Pour point	°C	-	-	-	-	-	-51.50	-	55.90

**Table C7** Murban crude specification

<b>API Gravity</b>	40.80		<b>% Sulfur</b>	0.867	
Methane	vol%	0.00	Iso-butane	vol%	0.45
Ethane	vol%	0.07	N-butane	vol%	1.32
Propane	vol%	0.52			

Description		Component Fraction (%vol)							
		FG	LPG	LN	MN	HN	IK	DO+GO	FO
Vol. yield on crude	lv%	0.07	2.29	5.94	3.30	10.10	20.40	25.90	29.70
Aromatics content	lv%	-	-	1.76	0.41	12.41	20.48	25.48	-
Cetane index		-	-	-	-	27.90	43.50	53.20	-
Freeze point	°C	-	-	-	-	-90.20	-56.10	-2.60	-
RONC		-	-	76.00	72.20	70.10	56.30	-	-
RVP	kg/cm <sup>2</sup>	-	-	0.75	0.16	0.04	0.00	-	-
Specific gravity		-	-	0.6609	0.7145	0.7438	0.7883	0.8455	0.9268
Sulfur	wt%	-	-	0.000	0.000	0.000	0.107	1.051	1.688
Viscosity @ 50 °C	cSt	-	-	-	0.39	0.52	0.93	3.09	88.41
Viscosity @ 50 °C	cSt	-	-	-	0.28	0.35	0.57	1.42	13.84
Pour point	°C	-	-	-	-	-	-73.00	-	33.40

**Table C8** Product specifications (x = maximum, n = minimum)

<b>Description</b>		<b>LPG</b>	<b>SUPG</b>	<b>ISOG</b>	<b>JP-1</b>	<b>HSD</b>	<b>FO #1</b>	<b>FO #2</b>	<b>FOVS</b>
RON			91n	95n					
RVP @ 37.8 °C	kPa		62x	62x					
Aromatic Content	vol%		35x	35x	25x				
Freezing Point	°C				-47x				
Cetane Index						47n			
Viscosity @ 50 °C	cSt						7-80	7-180	
Viscosity @ 100 °C	cSt								3-30
Sulfur Content	wt%					0.05x	2x	2x	0.5x
Pour Point	°C						24x	24x	57x

Product specifications are based on rules from Ministry of Commerce (MOC.).

**Table C9** Product storage data

<b>Description</b>		<b>LPG</b>	<b>SUPG</b>	<b>ISOG</b>	<b>JP-1</b>	<b>HSD</b>	<b>FO #1</b>	<b>FO #2</b>	<b>FOVS</b>
Product stored at initial	m <sup>3</sup>	1,500	14,100	8,400	15,400	54,000	-	-	-
Storage capacity	m <sup>3</sup>	5,000	16,000	14,000	28,000	80,000	5,000	15,000	35,000

### Appendix D Members of sets.

$aref \in \{RON, ARO, RVP\}$   
 $AV_q \in \{YD, RON, RVP, ARO, CI, SG\}$   
 $AW_q \in \{FP, S, V50, V100\}$   
 $C_{CDU} \in \{FG, LPG, LN, MN, HN, IK, DO, FO\}$   
 $C_{CRU} \in \{FG, LPG, REF\}$   
 $CDU \in \{CDU2, CDU3\}$   
 $C_{ia} \in \{MTBE, DCC\}$   
 $C_o \in \{OM, TP, LB, SLEB, PHET, MB\}$   
 $C_p \in \{LPG, SUPG, ISOG, JP-1, HSD, FO1, FO2, FOVS\}$   
 $CRU \in \{CRU2, CRU3\}$   
 $ctank \in \{OMT, TPT, LBT, SLEBT, PHETT, MBT\}$   
 $GSP \in \{GSP91, GSP95\}$   
 $HDS \in \{GO-HDS, DGO-HDS\}$   
 $in \in \{MTBE, DCC, REF, ISO, LN, HN\}$   
 $int \in \{MTBET, DCCT, REFT, ISOT, LNT, HNT\}$   
 $itank \in \{MTBET, DCCT\}$   
 $nap \in \{LN, MN, HN\}$   
 $ptank \in \{LPGT, GSP91, GSP95, JPT, DSP, FO1P, FO2P, FOVSP\}$   
 $UC_u \in \{(OM,OMT), (TP,TPT), (LB,LBT), (SLEB,SLEBT),$   
 $(PHET,PHETT), (MB,MBT)\}$   
 $UI_u \in \{(MTBE,MTBET), (DCC,DCCT)\}$   
 $UP_u \in \{(LPG,LPGT), (SUPG,GSP91), (ISOG,GSP95), (JP-1,JPT), (HSD,$   
 $DSP), (FO1,FO1P), (FO2,FO2P), (FOVS,FOVSP)\}$

## CURRICULUM VITAE

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