

# **C<sub>8</sub> AROMATICS ADSORPTION: EFFECTS OF ZEOLITE ACIDITY**



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A Thesis Submitted in Partial Fulfilment of the Requirements  
for the Degree of Master of Science  
The Petroleum and Petrochemical College, Chulalongkorn University  
in Academic Partnership with  
The University of Michigan, The University of Oklahoma,  
and Case Western Reserve University

2003

ISBN 974-17-2307-5

**Thesis Title:** C<sub>8</sub> Aromatics Adsorption: Effects of Zeolite Acidity  
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Accepted by the Petroleum and Petrochemical College, Chulalongkorn University, in partial fulfillment of the requirement for the Degree of Master of Science.

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## บทคัดย่อ

สุวรรณา ลิ้มสมุทรชัยกุล: การศึกษาผลของค่าความเป็นกรดของซีโอไลท์ต่อการดูดซับของสารอะโรมาติกคาร์บอน 8 อะตอม ( $C_8$  Aromatics Adsorption: Effects of Zeolite acidity) อ.ที่ปรึกษา: ดร.สันติ กุลประทีปปัญญา ผศ.ดร. ปราโมช รังสรรค์วิจิตร ผศ.ดร. ปมทอง มาลากุล ณ อยุธา 67 หน้า ISBN 974-17-2307-5

เทคนิคที่ดีที่สุดในการแยกพาราไซลีนออกจากสารผสมอะโรมาติกคาร์บอน 8 อะตอม คือเทคนิคการดูดซับด้วยซีโอไลท์ ดังนั้นการเข้าใจในปัจจัยที่มีผลต่อการแยกพาราไซลีนโดยใช้เทคนิคการดูดซับจึงมีความสำคัญ ในงานวิจัยนี้ศึกษาถึงผลของค่าความเป็นกรดของซีโอไลท์ต่อการดูดซับสารอะโรมาติกคาร์บอน 8 อะตอม ค่าความเป็นกรดของซีโอไลท์ถูกปรับเปลี่ยนโดยการแลกเปลี่ยนกับโลหะประจุบวกหมู่สอง ซึ่งได้แก่ แมกนีเซียม แคลเซียม สตรอนเซียม และแบเรียม โดยทั่วไปแล้วเมื่อประจุบวกขนาดใหญ่ขึ้น ค่าความเป็นกรดจะลดลง ผลการทดลองจากทั้งเทคนิคพัลซ์เทส และเบรคทรูพบว่า เมื่อค่าความเป็นกรดของซีโอไลท์เพิ่มขึ้น ค่าซีเล็กติวิตีของพาราไซลีนจะลดลงจนถึงจุดหนึ่ง ค่าซีเล็กติวิตีจะไม่ขึ้นกับค่าความเป็นกรดอีกต่อไป ผลของขนาดของประจุบวกมีความสำคัญมากกว่า ดังนั้นในการศึกษากลไกของการดูดซับของสารอะโรมาติกบนซีโอไลท์จึงต้องคำนึงถึงผลของค่าความเป็นกรดของซีโอไลท์ ขนาดประจุบวก และชนิดของซีโอไลท์

## ABSTRACT

4471031063: PETROCHEMICAL TECHNOLOGY PROGRAM

Suwanna Limsamutchaikul: C<sub>8</sub> Aromatics Adsorption: Effects of Zeolite Acidity.

Thesis Advisors: Dr. Santi Kulprathipanja, Asst. Prof. Pramoch Rangsunvigit, and Asst. Prof. Pomthong Malakul, 67 pp. ISBN 974-17-2307-5

Keywords: Xylene Separation/ Adsorption/ Zeolite/ Faujasite/ Acidity

The most attractive industrial technique for the separation of *p*-xylene is selective adsorption with zeolites. An understanding of parameters that influence the effectiveness of the separation is needed in order to achieve the best *p*-xylene separation. In this work, roles of zeolite acidity on liquid-phase adsorption of C<sub>8</sub> aromatics and *p*-xylene selectivity on 2.0X, 2.5X and Y zeolites were investigated at a constant temperature and pressure. Besides zeolite inherited acidity, acidity was also modified by exchanged di-valence cations, Mg, Ca, Sr, and Ba. Higher ionic radius exchanged cations are known to have lower acidity. The results from Pulse Test and Breakthrough techniques indicated that while the selectivity of *p*-xylene with respect to some C<sub>8</sub> aromatics decreased with the increase of zeolite acidity, there was a turning point where the effects of an exchanged cation size were more pronounced than those of the acidity. The effects of zeolite acidity, cation size and zeolite type must be taken into consideration for *p*-xylene selectivity.

## ACKNOWLEDGEMENT

This work is one of my great experiences that I never found in the classroom. Without these following people, this work cannot be successful.

Dr. Santi Kulprathipanja, US advisor, is my inspiration. Out of a sense of gratefulness, I would like to express my deepest gratitude to him for his helpful suggestions, creative discussions, patience and encouragement throughout my graduate work. I would also like to thank his wife, Ms. Apinya Kulprathipanja for her generous help in many aspects.

I would like to sincerely thank Dr. Pramoch Rangsunvigit, Thai advisor, who gave intensive suggestions, guidance, encouragement and patiently in proof reading my thesis.

I would also like to thank Dr. Pomthong Malakul, my co-advisor, for his excellent support and suggestion. He also proofed reading the thesis.

A special thank goes to the UOP LLC for the financial support and convenient while I have worked there. Additionally, I would like to take this chance to thank James W. Priegnitz, James E. Rekoske, Darryl M. Johnson and Dr. Linda Cheng for theirs greatly support.

I am grateful for the partial scholarship and partial funding of the thesis work provided by Postgraduate Education and Research program in Petroleum and Petrochemical Technology (PPT Consortium).

I would like to forward my appreciation to the Petroleum and Petrochemical College faculty and staff for their help and friendliness.

Finally, I am profoundly thankful to my family and all my friends for the love, cheerfulness and understanding.

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