## DEVELOPMENT OF POLYANILINE SENSOR FOR ETHANOL DETECTION

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#### ABSTRACT

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The nonconductive form of polyaniline, emeraldine base, was synthesized by the chemical oxidative polymerization of aniline using ammonium peroxydisulfate as an oxidant. Emeraldine base was converted to emeraldine salt (conductive form) by an acid doping process. Three different types of acid dopant; hydrochloric acid (HCl), acetic acid (CH<sub>3</sub>COOH), and hexanoic acid ( $C_5H_{11}COOH$ ) were used to study the effect of acid dopants and acid/polymer concentration ratio  $(N_a/N_p)$  on the specific conductivity of polyaniline films. The specific conductivity gradually increased with acid concentration. For HCl-doped polyaniline film, after  $N_{\mu}/N_{p}=9.8E+01$  $(C_a/C_p=10:1)$  the specific conductivity decreased due to over-doping. Whereas the specific conductivities of the CH<sub>3</sub>COOH and C<sub>5</sub>H<sub>11</sub>COOH-doped polyaniline films did not depend on acid concentration beyond the mole ratios of  $N_{AcOH}/N_p=5.9E+03$  and  $N_{Hexanoic}/N_p=3.1E+03$  ( $C_a/C_p=1000:1$ ). Polyaniline films were exposed to water and ethanol to study the effect of water and ethanol on the specific conductivity. The change in specific conductivity when the polyaniline films were exposed to water was greater than that when exposed to 100% ethanol. Mechanisms for the change in specific conductivity to water and ethanol are proposed.

### บทคัดย่อ

ถักษณาพร ธาราชีวิน : การพัฒนาสารพอถิอะนิถีนเพื่อใช้ในการตรวจวัดเอทานอล (Development of Polyaniline Sensor for Ethanol Detection) อ. ที่ปรึกษา : Prof. Johannes Schwank และ รศ. ดร. อนุวัฒน์ ศิริวัฒน์ 102 หน้า ISBN 974-334-176-5

้อเมอราลดีนเบส (Emeraldine base) ซึ่งเป็นสารที่ไม่นำไฟฟ้าสถานะหนึ่งของพอลิอลิ อะนิลีน (Polyaniline) ถูกสังเคราะท์ทางเคมีโดยใช้แอมโมเนียมเปอร์ออกซิไดซัลเฟต (Ammonium peroxydisulfate) เป็นสารออกซิแคนซ์ เอเมอราลดีนซอลท์ (Emeraldine salt) ซึ่งเป็น สถานะที่นำไฟฟ้าของพอลิอะนิลีนถูกเตรียมโดยกระบวนการโด๊ปด้วยกรดไฮโดรคลอริก (Hydrochloric acid), กรดอะซิติก (Acetic acid) และกรดเฮกซาโนอิก (Hexanoic acid) เพื่อศึกษา ผลของชนิดของสารโด๊ปและอัตราส่วนปริมาณของกรดต่อปริมาณพอลิอะนิลึนที่มีต่อค่าการนำไฟ ้ฟ้า จากการทุคลองพบว่าค่าการนำไฟฟ้าเพิ่มขึ้นเมื่อปริมาณกรดเพิ่มขึ้น ค่าการนำไฟฟ้าของฟิล์ม พอลิอะนิลีนที่ถูกโค๊ปด้วยกรดไฮโครคลอริกจะมีค่าลดลงเมื่ออัตราส่วนของกรดไฮโครคลอริกต่อ พอลิอะนิลีนมากกว่า 98 เนื่องจากสาร โค๊ปมีปริมาณมากเกินพอ ในขณะที่ค่าการนำไฟฟ้าของฟิล์ม พอลิอะนิลีนที่โด๊ปด้วยกรดอะซิติกและกรดเฮกซาโนอิกเพิ่มขึ้นเมื่อความเข้มข้นของกรดเพิ่มขึ้น และค่าการนำไฟฟ้าไม่ขึ้นกับความเข้มข้นของกรคเมื่ออัตราส่วนของกรคอะซิติกต่อพอลิอะนิลีน ้เท่ากับ 5900 และอัตราส่วนของกรดเฮกซาโนอิกต่อพอถิอะนิลีนเท่ากับ 3100 เมื่อทดสอบฟิล์มพอ ้ลิอะนิถีนในน้ำและเอทานอล พบว่าค่าการนำไฟฟ้าที่ทุดสอบในน้ำเกิดการเปลี่ยนแปลงมากกว่า ้เมื่อทดสอบในเอทานอลเนื่องจากโมเลกุลของน้ำมีขนาดเล็กกว่าเอทานอล กลไกการเปลี่ยนแปลง ้ค่าการนำไฟฟ้าของฟิล์มพอลิอะนิลีนที่ทำการทดสอบในน้ำและเอทานอลได้ถูกเสนอแนะในงาน วิจัยครั้งนี้

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