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UTILIZATION OF BAGASSE AND BAGASSE FLY ASH AS ADSORBENTS FOR REMOVAL OF LEAD AND CHROMIUM AND SUBSEQUENT UTILIZATION AS CONSTRUCTION MATERIALS

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โรงงานผลิตน้ำตาลจากอ้อยจะใช้ชานอ้อยเป็นเชื้อเพลิงในการผลิตไอน้ำ และกระแสไฟฟ้า ซึ่งหลังจากเผาไหม้แล้วชานอ้อยบางส่วนจะกลายเป็นเถ้าลอยชานอ้อย ดังนั้นวัตถุประสงค์ของ งานวิจัยนี้คือการศึกษาศักยภาพในการนำชานอ้อย (B) และเถ้าลอยชานอ้อย (BFA) มาใช้ ประโยชน์ ผลการศึกษาพบว่า ประสิทธิภาพสูงสุดในการกำจัดตะกั่วและโครเมียม เกิดขึ้นที่ค่า pH ของสารละลายเท่ากับ 6 และ 1 ตามลำดับ BFA มีประสิทธิภาพในการดูดซับตะกั่วมากกว่า B การ กำจัดเฮกซะวาเลนท์โครเมียม (Cr(VI)) เกิดทั้งการรีดักชันและการดูดซับ พบว่า การรีดักชันเป็นกล ไกหลักในการกำจัด Cr(VI) สำหรับการกำจัดด้วย B และให้ผลตรงกันข้ามเมื่อกำจัดด้วย BFA หมู่ ไฮดรอกซีในเซลลูโลสของ B น่าจะเป็นแหล่งให้อิเล็กตรอนในปฏิกิริยารีดักซัน และพบว่า B และ BFA ทั้งที่ผ่านและไม่ผ่านการกำจัดโลหะหนักมาแล้ว เป็นตัวขัดขวางการเกิดปฏิกิริยาไฮเดรชัน ของปูนชีเมนต์ โดยทำให้มีค่ากำลังรับแรงอัดลดลง เวลาบ่มเพียง 3 วัน เพียงพอสำหรับการกำจัด ทิ้งแบบฝังกลบ (มากกว่า 10 กก./ลบ.ม.) ส่วนความเข้มข้นของโลหะหนักทั้งสองชนิดจากน้ำชะ ของก้อนหล่อแข็ง มีค่าต่ำกว่า 5 มก./ล. ซึ่งเป็นค่ามาตรฐานของประกาศกระทรวงอุตสาหกรรม ฉบับที่ 6 (พ.ศ. 2540) และยังพบว่า BFA มีศักยภาพที่จะใช้แทนที่ปุ่นซีเมนต์ในการผลิตคอนกรีต บล็อกประสานปูพื้น และ คอนกรีตบล็อกก่อผนังได้สูงสุดถึงร้อยละ 30 การใช้ประโยชน์จากของ เหลือทิ้งในลักษณะนี้ เป็นทางเลือกหนึ่งในการกำจัดตะกั่ว และโครเมียม ซึ่งเป็นการใช้ของเหลือ ทั้งจากโรงงานหนึ่งมาบำบัดของเสียจากอีกโรงงานหนึ่ง จัดเป็นตัวอย่างในการจัดการของเหลือทิ้ง เพื่อการพัฒนาที่ยั่งยืน

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SIRAWAN RUANGCHUAY: UTILIZATION OF BAGASSE AND BAGASSE FLY ASH AS ADSORBENTS FOR REMOVAL OF LEAD AND CHROMIUM AND SUBSEQUENT UTILIZATION AS CONSTRUCTION MATERIALS. THESIS ADVISOR: MANASKORN RACHAKORNKIJ, Ph.D., THESIS CO-ADVISOR: PROF. METHI WECHARATANA, Ph.D., 268 pp. ISBN 974-17-4398-8.

Cogeneration of steam and electricity is a standard practice in the sugarcane industry worldwide. For Thailand, 22% of the small power producers (SPPs) use bagasse as a fuel, thus generating a large quantity of fly ash and residues. The purpose of this research is to convert bagasse (B) and bagasse fly ash (BFA) into inexpensive and effective removal materials for lead (Pb) and chromium (Cr) and to assess the efficiencies of cementitious solidification of spent materials for disposal in landfill and to develop construction materials namely; interlocking concrete paving blocks and hollow non-load-bearing concrete blocks.

The batch study results indicated that removal efficiency of Pb, total Cr, and Cr (VI) increased with increasing contact time, and amount of material applied. The maximum removals of Pb and Cr were obtained at pH 6 and pH 1, respectively. BFA was a better adsorbent for Pb than B. The removal of Cr(VI) may involve two processes. namely, reduction of Cr(VI) to Cr(III) and adsorption of Cr(VI) and Cr(III). Reduction was likely the major Cr(VI) removal mechanism than adsorption in B while the reverse was true for BFA. The hydroxyl groups in cellulose structure of B were found to be the potential major reduction sites. Portland cement hydration was inhibited by the increase in the amounts of B and BFA, resulting in poor strength performance. The compressive strengths increased with curing time and a 3-day curing period was enough to meet the strength requirement of landfill standards (3.5 ksc) for 5 and 10% replacement with B and BFA, respectively. Concentrations of lead and chromium in leachate form all solidified products were below the limits (5 mg/L) established by the Ministry of Industries No.6, B.E. 2540 (1997). BFA has high efficiency for partial cement replacement up to 30% in interlocking paving concrete block and hollow non-load-bearing concrete block construction. This alternative treatment process may very well be the choice for Pb or Cr removal. It utilizes wastes from other factory to treat waste. This is an example of comprehensive waste management suitable for sustainable development.

Co-advisor's signature

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NOMENCLATURE

AAS = Atomic Absorption Spectrophotometer

ASTM = American Society for Testing and Materials

B = bagasse

B-Cr = chromium adsorbed bagasse

B-Pb = lead adsorbed bagasse

BFA = bagasse fly ash

BFA-Cr = chromium adsorbed bagasse fly ash

BFA-Pb = lead adsorbed bagasse fly ash

 C_2S = dicalcium silicate C_3S = tricalcium silicate C_3A = tricalcium aluminite

 C_3AF = tetracalcium aluminoferrite

Cr = chromium

Cr(III) = trivalent chromium

Cr(VI) = hexavalent chromium

C-S-H = calcium silicate hydrate

DI = deionized water

ICP = inductively coupled plasma optical emission spectroscopy

ksc = kilogram per square centimeter

LOI = loss on iginition(%) defined by ASTM C311 as the weight

fraction of materials that is lost by heating the oven dried

sample at 750 °C

Pb = lead

SEM = scanning electron microscope S/S = Solidification/Stabilization

Total Cr = total chromium, include of hexavalent chromium and

trivalent chromium

w/c ratio = water-to-cement ratio, the weight ratio of water to cement

XRD = X-ray diffraction spectrometer

XRF = X-ray fluorescence spectroscope