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APPENDICES

Appendix A Thermal Gravimetric Analysis (TGA)

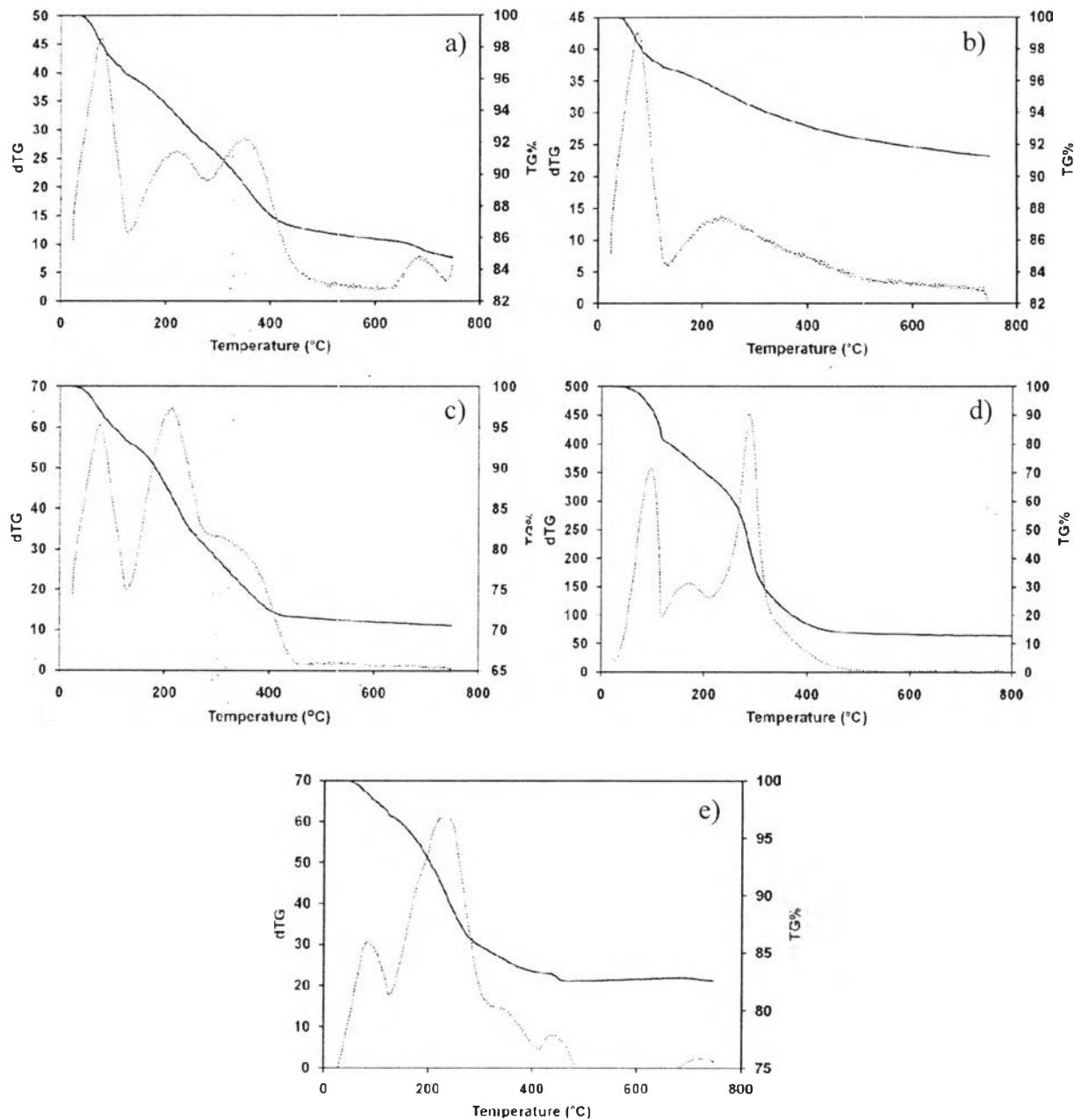


Figure A1 TGA profiles of prepared catalysts a) Pd/Al₂O₃, b) Pt/Al₂O₃, c) Cu/Al₂O₃, d) NiCu/Al₂O₃, and e) NiMo/Al₂O₃; dTG(···), TG(—).

Appendix B X-ray Diffraction (XRD)

The XRD characterization was performed to identify unique metal oxide species of the studied catalysts. The XRD patterns of the catalysts were illustrated in Figure B1-B7.

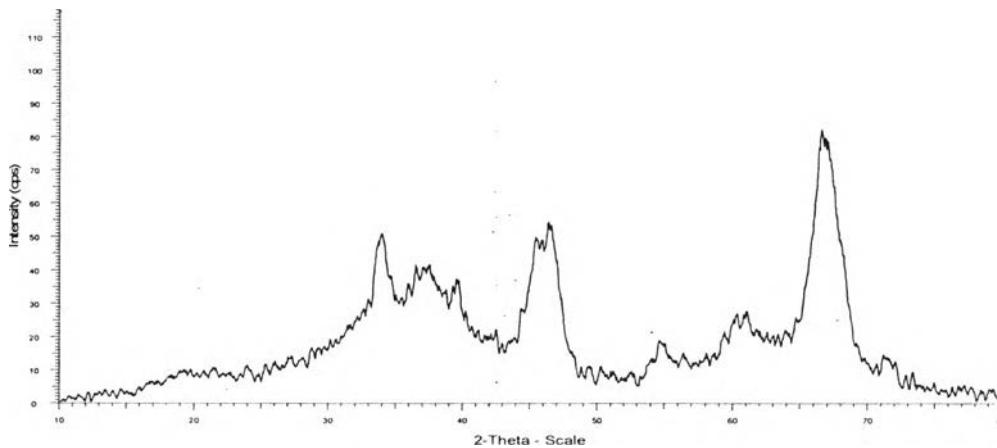


Figure B1 The XRD patterns of the Pd/Al₂O₃.

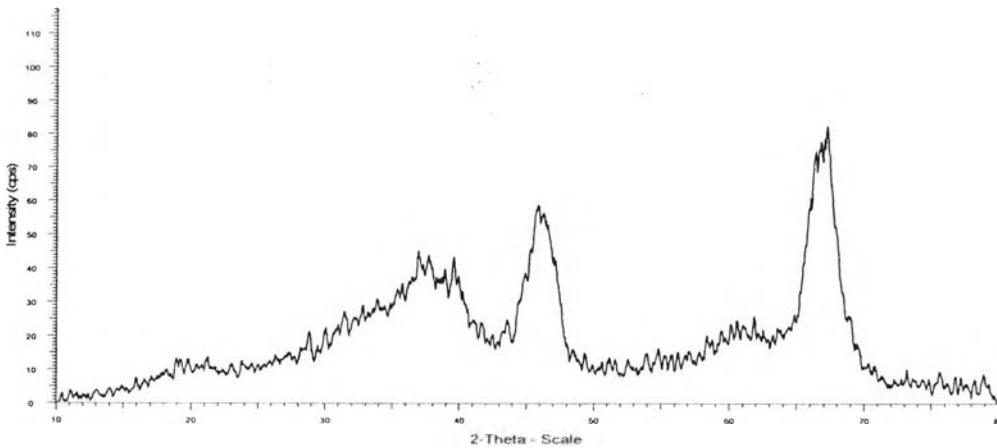


Figure B2 The XRD patterns of the Pt/Al₂O₃.

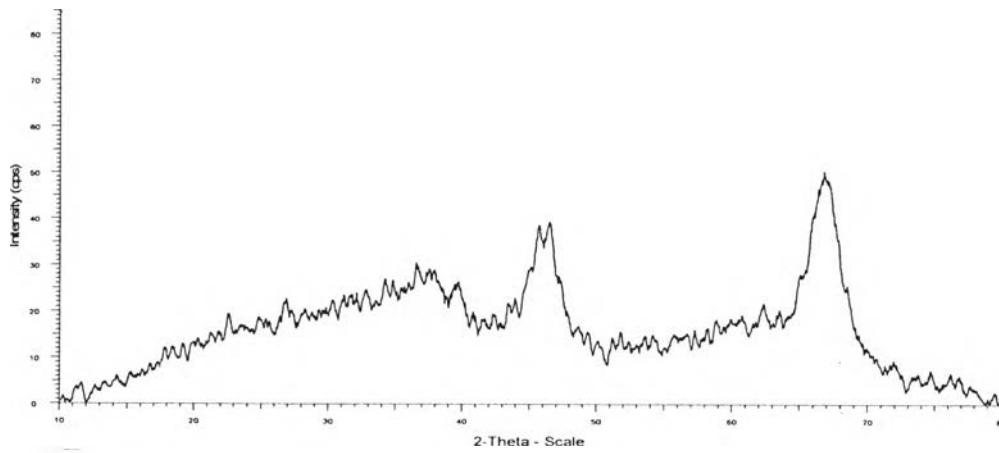


Figure B3 The XRD patterns of the Ni-W/Al₂O₃.

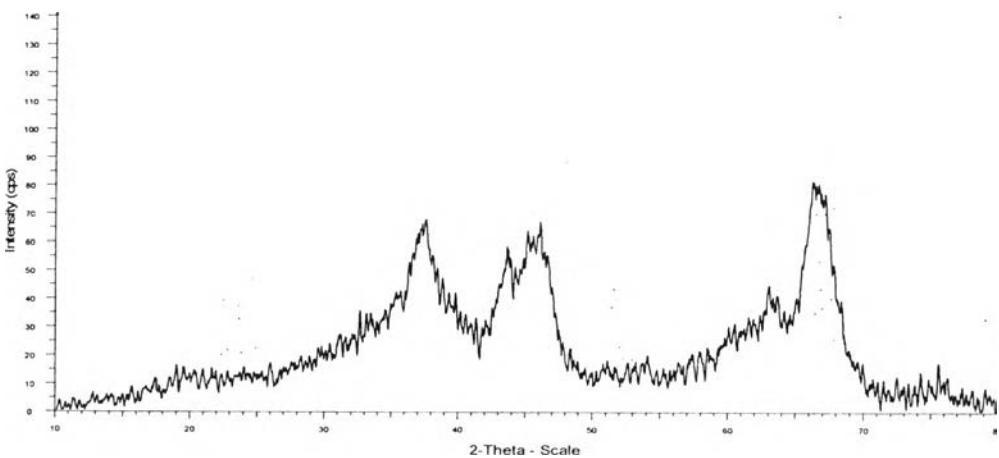


Figure B4 The XRD patterns of the Ni-Cu/Al₂O₃.

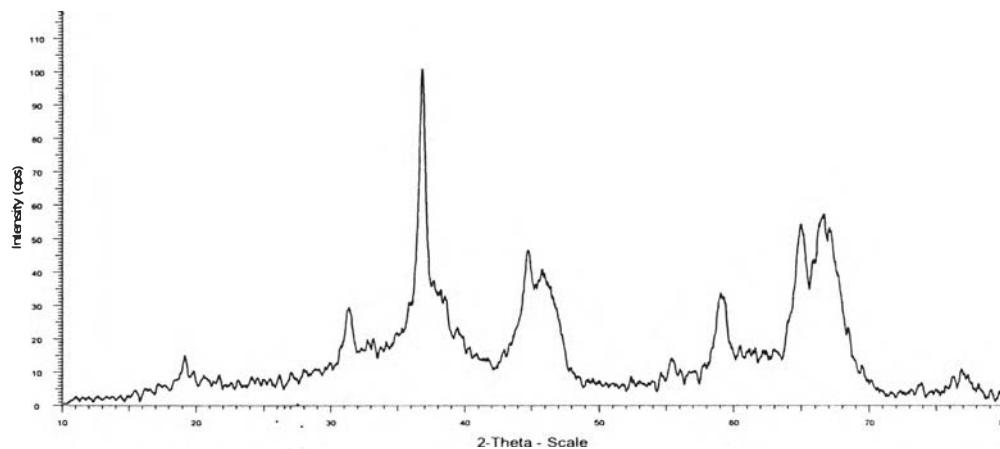


Figure B5 The XRD patterns of the Ni-Co/Al₂O₃.

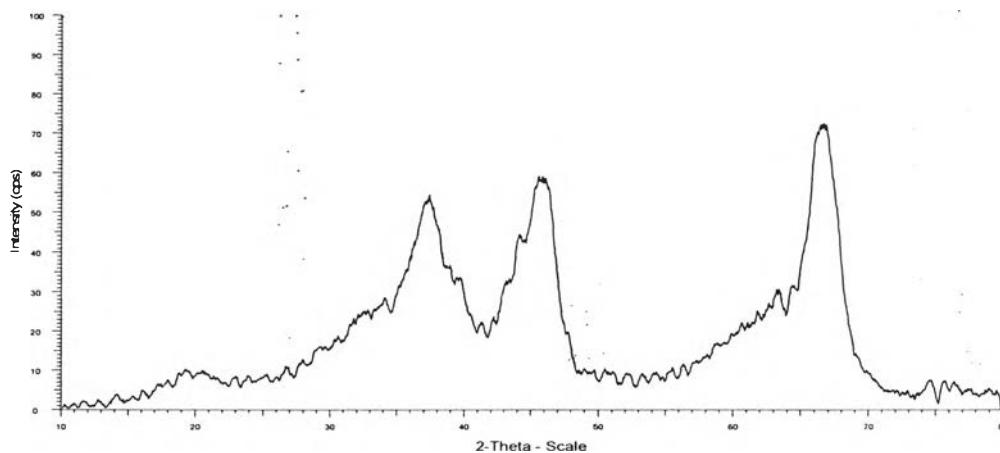


Figure B6 The XRD patterns of the Ni/Al₂O₃.

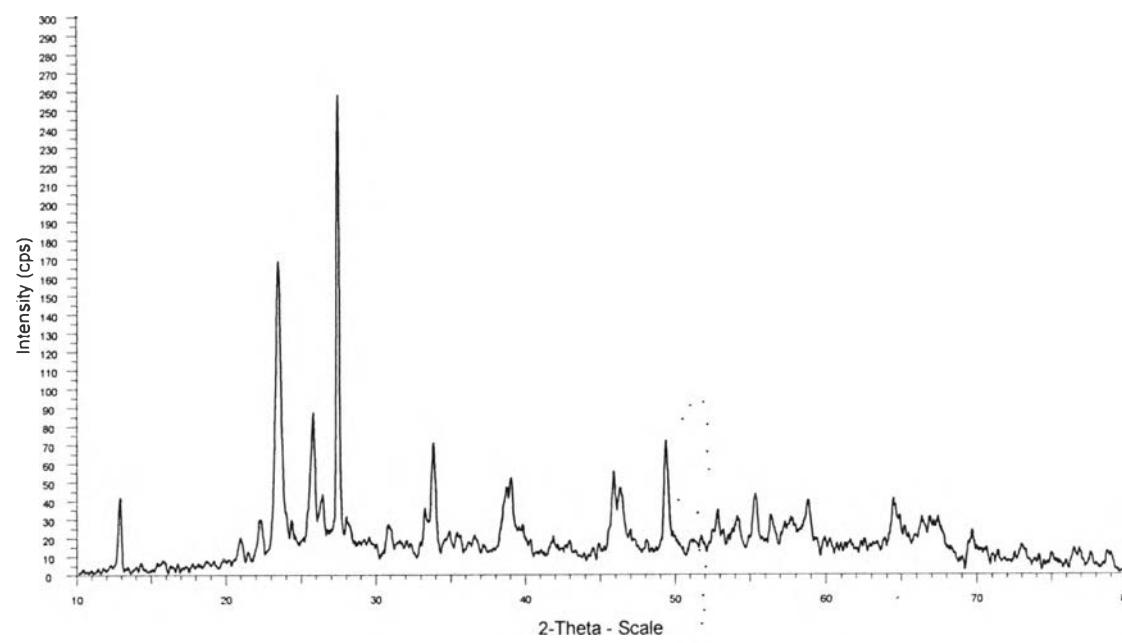


Figure B7 The XRD patterns of the Cu-Mo/Al₂O₃.

Appendix C Scanning Electron Microscope (SEM)

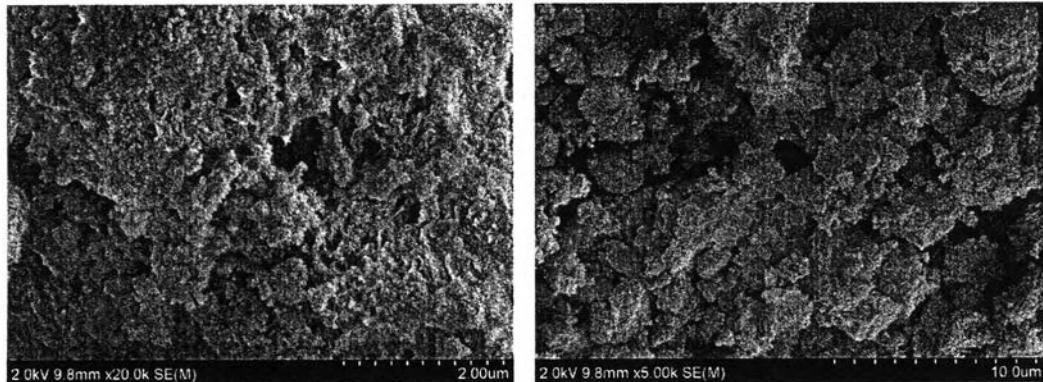


Figure C1 SEM image of Al₂O₃

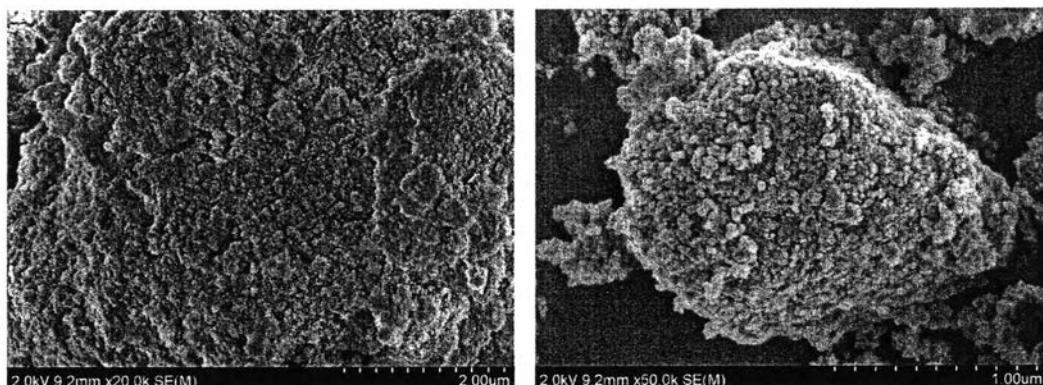


Figure C2 SEM image of Ni/Al₂O₃ (Fresh).

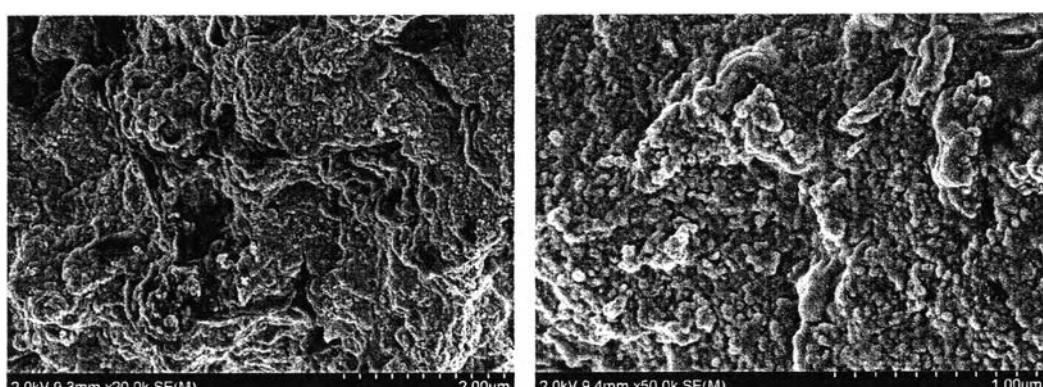


Figure C3 SEM image of Ni/Al₂O₃ (Spent).

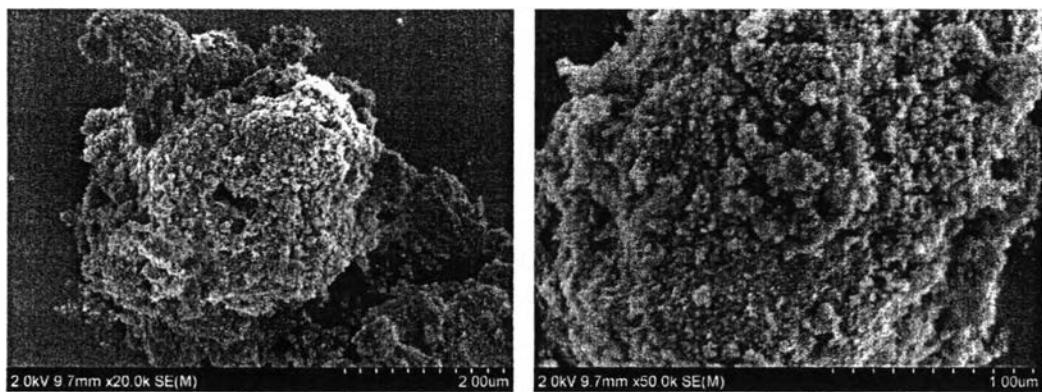


Figure C4 SEM image of Ni-Co/Al₂O₃ (Fresh).

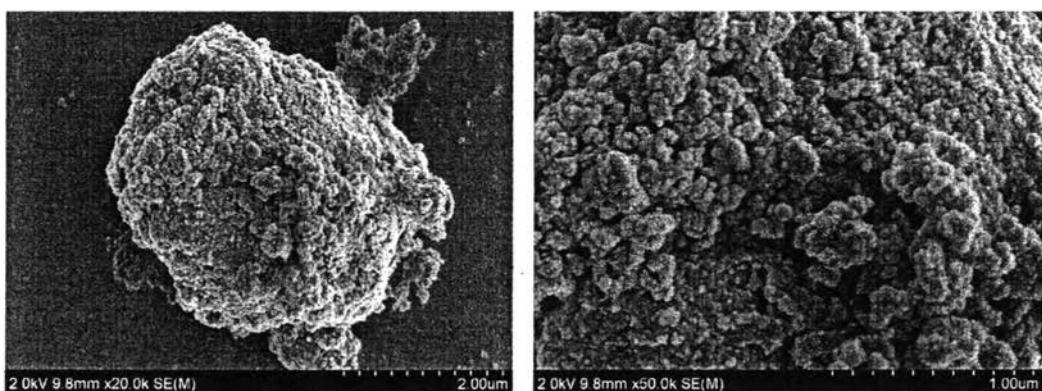


Figure C5 SEM image of NiMo/Al₂O₃ (Fresh).

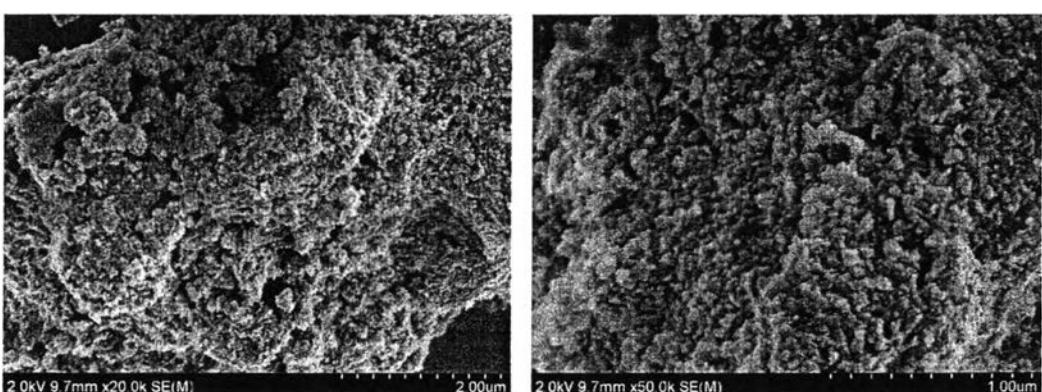


Figure C6 SEM image of NiMo/Al₂O₃ (Spent).

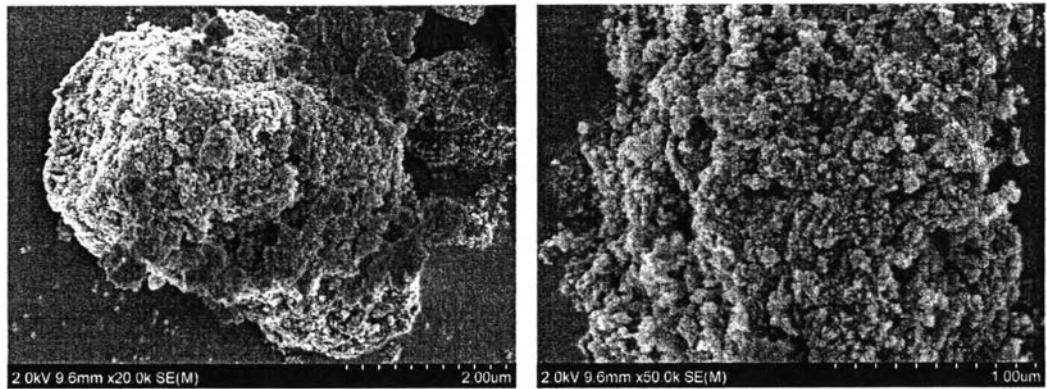


Figure C7 SEM image of Cu/Al₂O₃ (Fresh).

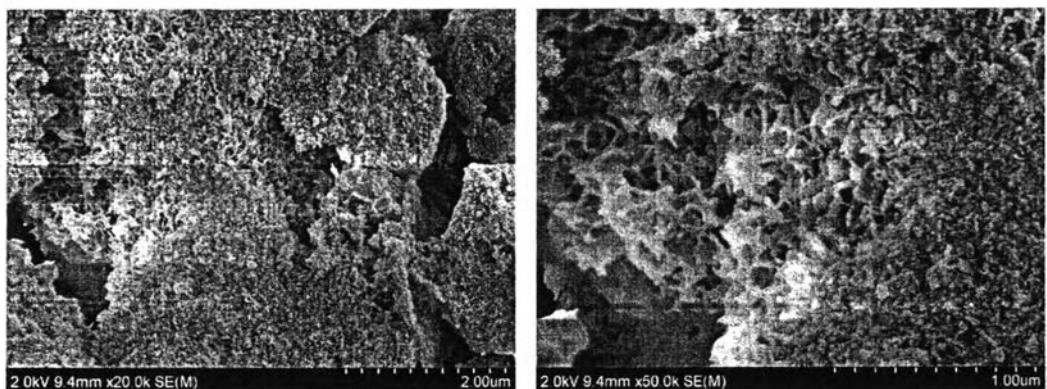


Figure C8 SEM image of Cu/Al₂O₃ (Spent).

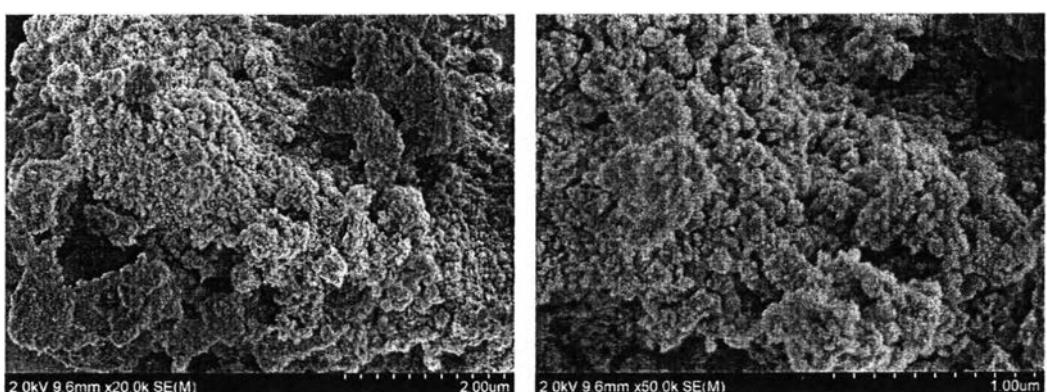


Figure C9 SEM image of CuMo/Al₂O₃ (Fresh).

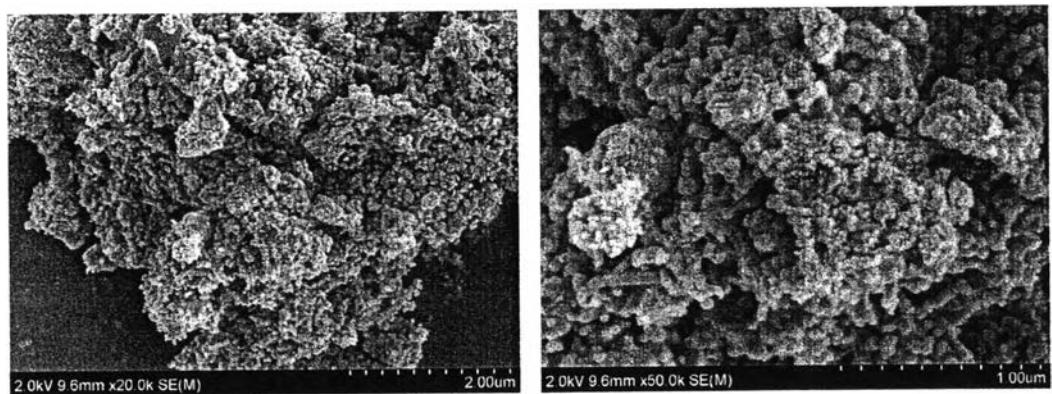


Figure C10 SEM image of CuMo/Al₂O₃ (Spent).

Appendix D Electronic Properties of Active Metals

Metal	atomic radius	Bonding radius	Ground state	Eletrone - gativity	Density	First outer shell	Second outer shell	Third outer shell	Ground state	Oxidation state	1st Ionization (ev)	Crystal
Mo	1.39	1.3	$7s^1$	2.16	10.2	13	1		$4d^5 5s$	6,3	7.0924	bcc
W	1.41	1.3	$5D^0$	2.36	19.35	32	12	2	$4f^{14} 5d^4 6s^2$	6	7.864	bcc
Cu	1.28	1.17	$2S^{0.5}$	1.9	8.92	18	1		$3d^{10} 4s$	2,1	7.7264	ccp
Ni	1.24	1.15	$3F^4$	1.91	8.9	16	2		$3d^8 4s^2$	2,3	7.6398	ccp
Pd	1.37	1.28	$1S^0$	2.2	12.02	18	0		$4d^{10}$	2,4	8.3369	fcc
Pt	1.39	1.3	$3D^5$	2.28	21.45	32	17	1	$4f^{14} 5d^9 6s$	4,2	8.9588	fcc
Co	1.25	1.16	$4F^{4.5}$	1.88	8.9	15	2		$3d^7 4s^2$	2,3	7.881	hcp
Rh	1.34	1.25	$4F^{4.5}$	2.28	12.4	16	1		$4d^8 5s$	3,4	7.4589	fcc
Ir	1.36	1.27	$4F^{4.5}$	2.2	22.421	32	15	2	$4f^{14} 5d^7 6s^2$	3,4	8.967	fcc
Ru	1.34	1.25	$5F^5$	2.2	12.3	15	1		$4d^7 5s$	3,4,8	7.3605	hcp
Zn	1.38	1.25	$1S^0$	1.65	7.14	18	2		$3d^{10} 4s^2$	2	9.3942	hcp

CURRICULUM VITAE

Name: Mr. Thawesak Parawan

Date of Birth: March 24, 1987

Nationality: Thai

University Education:

2005–2009 Bachelor Degree of Chemical Engineering, Faculty of Engineering, Prince of Songkla University, Songkhla, Thailand

Work Experience:

2008	<p>Position: Research Assistant (Student Internship)</p> <p>Company: Guangdong University of Technology (GDUT), Guangzhou, P.R China</p>
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Proceedings:

1. Parawan, T., Jongpatiwut, S., Sreethawong, T., Butnark, S., and Osuwan, S. (2011, January 5-7) Deoxygenation of Animal Fats for the Production of Hydrogenated Biodiesel: Effect of Active Metals. Proceedings of Pure and Applied Chemistry International Conference 2011, Bangkok, Thailand
2. Parawan, T., Jongpatiwut, S., Sreethawong, T., Butnark, S., and Osuwan, S. (2011, April 26) Deoxygenation of Beef Fat for the Production of Hydrogenated Biodiesel: Effect of Active Metals. Proceedings of The 2nd Research Symposium on Petroleum, Petrochemicals, and Advanced Materials and the 17th PPC Symposium on Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.

