REFERENCES

- Adelodun, A.A., Lim, Y.H., and Jo, Y.M. (2013) Surface oxidation of activated carbon pellets by hydrogen peroxide for preparation of CO₂ adsorbent. Journal of Industrial and Engineering Chemistry, 20(4), 2130-2137.
- Ania, C.O., Parra, J.B., and Pis, J.J. (2002) Influence of oxygen-containing functional groups on active carbon adsorption of selected organic compounds. <u>Fuel Processing Technology</u>, 79(3), 265-271.
- Arenillas, A., Smith, K.M., Drage, T.C., and Snape, C.E. (2005a) CO₂ capture using some fly ash-derived carbon materials. <u>Fuel</u>, 84(17), 2204-2210.
- Arenillas, A., Drage, T.C., Smith, K., and Snape, C.E. (2005b) CO₂ removal potential of carbons prepared by co-pyrolysis of sugar and nitrogen containing compounds. <u>Journal of Analytical and Applied Pyrolysis</u>, 74(1-2), 298-306.
- Auer, E., Freund, A., Pietsch, J., and Tacke, T. (1998) Carbon as supports for industrial precious metal catalysts. <u>Applied Catalysis</u>, 173(2), 259-271.
- Bergman, J. "Carbon Dioxide CO₂." Windows2universe. 15 Oct 2011. 26 Apr 2013 http://www.windows2universe.org/physical_science/chemistry/carbon_di-oxide.html
- Boehm, H.P. (2002) Surface oxides on carbon and their analysis: a critical assessment. <u>Carbon</u>, 40(2), 145-149.
- Burg, P., Fydrych, P., Cagniant, D., Nanse, G., Bimer, J., and Jankowska, A. (2002) The characterization of nitrogen-enriched activated carbons by IR, XPS and LSER methods. <u>Carbon</u>, 40(9), 1521-1531.
- Choi, S., Drese, J.H., and Jones, C.W. (2009) Adsorbent materials for carbon dioxide capture from large anthropogenic point sources. <u>ChemSusChem</u>, 2(9), 796-854.
- Datsyuk, V., Kalyva, M., Papagelis, K., Parthenios, J., Tasis, D., Siokou, A., Kallitsis, I., and Galiotis, C. (2008) Chemical oxidation of multiwalled carbon nanotubes. <u>Carbon</u>, 46(6), 833-840.

Ø

- Daud, W.M.A.W. and Houshamnd, A.H. (2010) Textural characteristics, surface chemistry and oxidation of activated carbon. <u>Journal of Natural Gas</u> <u>Chemistry</u>, 19(3), 267-279.
- Drage, T.C., Snape, C.E., Stevens, L.A., Wood, J., Wang, J., Cooper, A.I., Dawson, R., Guo, X., Satterley, C., and Irons, R. (2012) Materials challenges for the development of solid sorbents for post-combustion carbon capture. <u>Journal</u> <u>of Materials Chemistry</u>, 22(7), 2815.
- Fan, X., Zhang, L., Zhang, G., Shu, Z., and Shi, J. (2013) Chitosan derived nitrogen-doped microporous carbons for high performance CO₂ capture. <u>Carbon</u>, 61(0), 423-430.
- Febo, S., Leonardo, F., Mario, P., and Luca, P. (2002) Chemical modification of carbon fiber surfaces. <u>Carbon</u>, 40(5), 735–741.
- Gardiner, L. "The Biosphere: An Integral Part of the Planet and its Climate." Windows2universe. 25 May 2007. 24 Apr 2013. 24 Apr 2013 http://www.windows2universe.org/teacher_resources/online_courses/biosphere_intro.html
- Gerber, I., Oubenali, M., Bacsa, R., Durand, J., Goncalves, A., Pereira, M.F., Jolibois, F., Perrin, L., Poteau, R., and Serp, P. (2011) Theoretical and experimental studies on the carbon-nanotube surface oxidation by nitric acid: interplay between functionalization and vacancy enlargement. <u>Chemistry</u>, 17(41), 11467-11477.
- Gui, M.M., Yap, Y.X., Chai, S.-P., and Mohamed, A.R. (2013a) Multi-walled carbon nanotubes modified with (3-aminopropyl)triethoxysilane for effective carbon dioxide adsorption. <u>International Journal of Greenhouse</u> <u>Gas Control</u>, 14, 65-73.
- Gui, M.M., Yap, Y.X., Chai, S.-P., and Mohamed, A.R. (2013b) Aminefunctionalization of multi-walled carbon nanotubes for adsorption of carbon dioxide. <u>Asia-Pacific Journal of Chemical Engineering</u>, 8(2), 262-270.
- Huang, M.C. and Teng, H. (2003) Nitrogen-containing carbons from phenolformaldehyde resins and their catalytic activity in NO reduction with NH₃. <u>Carbon.</u> 41(5), 951-957.
- Iijima, S. (1991) Helical microtubules of graphitic carbon. <u>Nature</u>, 354(6348), 56-58.

σ

- Jaramillo, J., Álvarez, P.M., and Gómez-Serrano, V. (2010) Oxidation of activated carbon by dry and wet methods: Surface chemistry and textural modifications. <u>Fuel Processing Technology</u>, 91(11), 1768-1775.
- Jin, Y.Z., Gao, C., Hsu, W.K., Zhu, Y., Huczko, A., Bystrzejewski, M., Roe, M., Lee, C.Y., Acquah, S., Kroto, H., and Walton, D.R.M. (2005) Large-scale synthesis and characterization of carbon spheres prepared by direct pyrolysis of hydrocarbons. <u>Carbon</u>, 43(9), 1944-1953.
- Kang, J., Li, O.L., and Saito, N. (2013) Synthesis of structure-controlled carbon nano spheres by solution plasma process. <u>Carbon</u>, 60, 292-298.
- Kang, J., Li, O.L., and Saito, N. (2014) Hierarchical meso-macro structure porous carbon black as electrode materials in Li-air battery. <u>Journal of Power</u> <u>Sources</u>, 261, 156-161.
- Khalili, S., Ghoreyshi, A.A., Jahanshahi, M., and Pirzadeh, K. (2013) Enhancement of carbon dioxide capture by amine-functionalized multi-walled carbon nanotube. <u>CLEAN - Soil, Air, Water</u>, 41(10), 939-948.
- László, K., Tombácz, E., and Josepovits, K. (2001) Effect of activation on the surface chemistry of carbons from polymer precursors. <u>Carbon</u>, 39(8), 1217-1228.
- Lee, J., Kim. J., and Hyeon, T. (2006) Recent progress in the synthesis of porous carbon materials. <u>Advanced Materials</u>, 18(16), 2073-2094.
- Lee, Z.H., Lee, K.T., Bhatia, S., and Mohamed, A.R. (2012) Post-combustion carbon dioxide capture: Evolution towards utilization of nanomaterials. <u>Renewable and Sustainable Energy Reviews.</u> 16(5), 2599-2609.
- Li, L., Zhao, N., Wei, W., and Sun, Y. (2013) A review of research progress on CO₂ capture, storage, and utilization in Chinese Academy of Sciences. <u>Fuel</u>,108, 112-130.
- Macías-García, A., Díaz-Díez, M.A., Cuerda-Correa, E.M., Olivares-Marín, M., and Gañan-Gómez, J. (2006) Study of the pore size distribution and fractal dimension of HNO3-treated activated carbons. <u>Applied Surface Science</u>, 252(17), 5972-5975.

Manocha, S.M. (2003) Porous carbons. Sadhana, 28(1-2), 335-348.

σ

- Maroto-Valer, M.M., Lu, Z., Zhang, Y., and Tang, Z. (2008) Sorbents for CO₂ capture from high carbon fly ashes. <u>Waste Manag</u>, 28(11), 2320-2328.
- Maroto-Valer, M.M., Tang, Z., and Zhang, Y. (2005) CO₂ capture by activated and impregnated anthracites. <u>Fuel Processing Technology</u>, 86(14–15), 1487-1502.
- Mondal, M.K., Balsora, H.K., and Varshney, P. (2012) Progress and trends in CO₂ capture/separation technologies: A review. <u>Energy</u>, 46(1), 431-441.
- Montes-Morán, M.A., Suárez, D., Menéndez, J.A., and Fuente, E. (2004) On the nature of basic sites on carbon surfaces: an overview. <u>Carbon</u>, 42(7), 1219-1225.
- Moreno-Castilla, C., López-Ramón, M.V., and Carrasco-Marín, F. (2000) Changes in surface chemistry of activated carbons by wet oxidation. <u>Carbon</u>, 38(14), 1995-2001.
- Pevida, C., Plaza, M.G., Arias, B., Fermoso, J., Rubiera, F., and Pis, J.J. (2008a) Surface modification of activated carbons for CO₂ capture. <u>Applied Surface</u> <u>Science</u>, 254(22), 7165-7172.
- Pevida, C., Drage, T.C., and Snape, C.E. (2008b) Silica-templated melamineformaldehyde resin derived adsorbents for CO₂ capture. <u>Carbon</u>, 46(11), 1464-1474.
- Pietrzak, R. (2009) XPS study and physico-chemical properties of nitrogenenriched microporous activated carbon from high volatile bituminous coal. <u>Fuel</u>, 88(10), 1871-1877.

a

- Plaza, M.G., Pevida, C., Arenillas, A., Rubiera, F., and Pis, J.J. (2007) CO₂ capture by adsorption with nitrogen enriched carbons. <u>Fuel</u>, 86(14), 2204-2212.
- Plaza, M.G., Pevida, C., Arias, B., Fermoso, J., Casal, M.D., Martín, C.F., Rubiera,
 F., and Pis, J.J. (2009a) Development of low-cost biomass-based adsorbents for postcombustion CO₂ capture. <u>Fuel</u>, 88(12), 2442-2447.
- Plaza, M.G., Pevida, C., Arias, B., Fermoso, J., Rubiera, F., and Pis, J.J. (2009b) A comparison of two methods for producing CO₂ capture adsorbents. <u>Energy</u> <u>Procedia</u>, 1(1), 1107-1113.

- Plaza, M.G., Pevida, C., Arias, B., Casal, M.D., Martin, C.F., Fermoso, J., Rubiera, F., and Pis, J.J. (2009c) Different approaches for the development of lowcost CO₂ adsorbents. Journal of Environmental Engineering-Asce, 135(6), 426-432.
- Plaza, M.G., Rubiera, F., Pis, J.J. and Pevida, C. (2010) Ammoxidation of carbon materials for CO₂ capture. <u>Applied Surface Science</u>, 256(22), 6843-6849.

.

- Plaza, M.G., Thurecht, K.J., Pevida, C., Rubiera, F., Pis, J.J., Snape, C.E., and Drage,
 T.C. (2013) Influence of oxidation upon the CO₂ capture performance of a phenolic-resin-derived carbon. <u>Fuel Processing Technology</u>, 110, 53-60.
- Przepiórski, J. (2006) Enhanced adsorption of phenol from water by ammoniatreated activated carbon. Journal of Hazardous Materials, 135(1-3), 453-456.
- Przepiórski, J., Skrodzewicz, M., and Morawski, A.W. (2004) High temperature ammonia treatment of activated carbon for enhancement of CO₂ adsorption. Applied Surface Science, 225(1–4), 235-242.
- Qian, H.S., Han, F.M., Zhang, B., Guo, Y.C., Yue, J., and Peng, B.X. (2004) Noncatalytic CVD preparation of carbon spheres with a specific size. <u>Carbon</u>, 42(4), 761-766.
- Raymundo-Piñero, E., Cazorla-Amorós, D., Linares-Solano, A., Find, J., Wild, U., and Schlögl, R. (2002) Structural characterization of N-containing activated carbon fibers prepared from a low softening point petroleum pitch and a melamine resin. <u>Carbon</u>, 40(4), 597-608.
- Riebeek, H. and Simmon, R. "Global Warming." EOS Project Science Office. 3 Jun 2010. 21 Apr 2013 ">http://earthobservatory.nasa.gov/Features/GlobalWarming/>
- Rubin, E.S., Mantripragada, H., Marks, A., Versteeg, P., and Kitchin, J. (2012) The outlook for improved carbon capture technology. <u>Progress in Energy and</u> <u>Combustion Science</u>, 38(5), 630-671.
- Rupasinghe, J. "Differences between Physisorption and Chemisorption." Jscienceclass. 24 Jun 2012. 24 Jun 2015. http://jscienceclass.blogspot.com/2012/06/ /physisorption-and-chemisorption.html>

- Sayari, A., Belmabkhout, Y., and Serna-Guerrero, R. (2011) Flue gas treatment via CO₂ adsorption. <u>Chemical Engineering Journal</u>, 171(3), 760-774.
- Shafeeyan, M.S., Daud, W.M.A.W., Houshmand, A., and Arami-Niya, A. (2011) Ammonia modification of activated carbon to enhance carbon dioxide adsorption: Effect of pre-oxidation. <u>Applied Surface Science</u>, 257(9), 3936-3942.
- Shafeeyan, M.S., Daud, W.M.A.W., Houshmand, A., and Shamiri, A. (2010) A review on surface modification of activated carbon for carbon dioxide adsorption. <u>Journal of Analytical and Applied Pyrolysis</u>, 89(2), 143-151.
- Sing, K.S.W., Everett, D.H., Haul, R.A.W., Moscou, L., Pierotti, R.A., Rouquerol, J., and Siemieniewska, T. (1985) Reporting physisorption data for gas/solid systems with special reference to the determination of surface area and porosity. <u>Pure and Applied Chemistry</u>, 57(4), 603-619.
- Stobinski, L., Lesiak, B., Kövér, L., Tóth, J., Biniak, S., Trykowski, G., and Judek, J. (2010) Multiwall carbon nanotubes purification and oxidation by nitric acid studied by the FTIR and electron spectroscopy methods. <u>Journal of Alloys</u> <u>and Compounds</u>, 501(1), 77-84.
- Su, F., Lu, C., Cnen, W., Bai, H., and Hwang, J.F. (2009) Capture of CO₂ from flue gas via multiwalled carbon nanotubes. <u>Science of the Total Environment</u>, 407(8), 3017-3023.
- Sujata Carbons Pvt. Ltd. "Introduction: Activated Carbon." Sujatacarbons. 2011. 30 Apr 2013. http://sujatacarbons.com/activated-carbon.html
- Surin, T. (2013) Influence of oxidation and nitrogen treatment on activated carbon for CO₂ adsorption. M.S. Thesis, The Petroleum and Petrochemical College, Chulalongkorn University, Bangkok, Thailand.
- Tanada, S., Kawasaki, N., Nakamura, T., Mamiko, A., and Isomura, M. (1999) Removal of formaldehyde by activated carbons containing amino groups. <u>Colloid and Interface Science</u>, 214(1), 106–108.
- The American Ceramic Society. "DOE Awards Millions for Carbon Capture, Storage and for Solar Grid Integration." Ceramics. 7 Sep 2010. 28 Apr 2013 <http://ceramics.org/ceramic-tech-today/doe-awards-millions-for-carboncapture-storage-and-for-solar-grid-integration>

- Tseng, H.H. and Wey, M.Y. (2006) Effects of acid treatments of activated carbon on its physiochemical structure as a support for copper oxide in DeSO₂ reaction catalysts. <u>Chemosphere</u>, 62(5), 756-766.
- United States Environmental Protection Agency. "U.S. Greenhouse Gas Inventory Report: 1990-2013." Epa. 6 Apr 2015. 25 Apr 2015. http://www.epa.gov/climatechange/ghgemissions/usinventoryreport.html
- Universal Industrial Gases, Inc. "Carbon Dioxide (CO₂) Properties, Uses, Applications." Uigi. 2003. 27 Apr 2013 http://www.uigi.com/carbondioxide.html
- Wang, J., Senkovska, I., Oschatz, M., Lohe, M.R., Borchardt, L., Heerwig, A., Liu, Q., and Kaskel, S. (2013) Imine-linked polymer-derived nitrogen-doped microporous carbons with excellent CO₂ capture properties. <u>ACS Appl</u> <u>Mater Interfaces</u>, 5(8), 3160-3167.
- Wang, L. and Yang, R.T. (2012) Significantly increased CO₂ adsorption performance of nanostructured templated carbon by tuning surface area and nitrogen doping. <u>The Journal of Physical Chemistry C</u>, 116(1), 1099-1106.
- Wang, Z., Yu, L., Zhang, W., Zhu, Z., He, G., Chen, Y., and Hu, G. (2003) Carbon spheres synthesized by ultrasonic treatment. <u>Physics Letters A</u>, 307(4), 249-252.
- Wanga, M., Lawala, A., Stephensonb, P., Siddersb, J., Ramshawa, C., and Yeunga,
 H. (2011) Post-combustion CO₂ capture with chemical absorption: A Satate-of-the-art Review. <u>Chemical Engineering Journal</u>, 89(9), 1609-1624.
- Xia, Y., Mokaya, R., Walker, G.S., and Zhu, Y. (2011) Superior CO₂ adsorption capacity on N-doped, high-surface-area, microporous carbons templated from zeolite. <u>Advanced Energy Materials</u>, 1(4), 678-683.
- Xing, W., Liu, C., Zhou, Z., Zhou, J., Wang, G., Zhuo, S., Xue, Q., Song, L. and Yan, Z. (2014) Oxygen-containing functional group-facilitated CO₂ capture by carbide-derived carbons. <u>Nanoscale Research Letters</u>, 9(1), 1-8.
- Yang, H., Xu, Z., Fan, M., Gupta, R., Slimane, R.B., Bland, A.E., and Wright, I.
 (2008) Progress in carbon dioxide separation and capture: A review.
 Journal of Environmental Sciences, 20(1), 14-27.

σ

- Ye, Q., Jiang, J., Wang, C., Liu, Y., Pan, H., and Shi, Y. (2012) Adsorption of lowconcentration carbon dioxide on amine-modified carbon nanotubes at ambient temperature. <u>Energy & Fuels</u>, 26(4), 2497-2504.
- Yu, C.H. (2012) A Review of CO₂ capture by absorption and adsorption. <u>Aerosol</u> <u>and Air Quality Research</u>, 12(5), 745-769.
- Zhu, Z.H., Radovic, L.R., and Lu, G.Q. (1999) Effects of acid treatments of carbon on N₂O and NO reduction by carbon-supported copper catalysts. <u>Carbon</u>, 38(3), 451–464.

APPENDICES

Appendix A Calculation for Molar Concentration of Nitric Acid

From;
$$M = \frac{\% \times 10 \times d}{Mw}$$
 (A1)

Where,

70

M = molar concentration, M

% = percentage of nitric acid, %

 $d = density of nitric acid, g/cm^3$

Mw = molecular weight of nitric acid, g/mol

70 % of nitric acid;

$$M = \frac{\frac{\% \times 10 \times d}{Mw}}{\frac{70 \times 10 \times 1.41}{63.01}}$$

$$= 15.66$$
From;

$$M_1V_1 = M_2V_2$$
(A2)
where,

 V_1 , V_2 = volume of nitric acid, cm³

 $10 \text{ cm}^3 \text{ of } 10 \text{ M} \text{ nitric acid}; \quad 15.66 \times \text{V}_1 = 10 \times 10$ 0

 $V_1 = 6.39$

 6.39 cm^3 of 70 % (15.66 M) nitric acid was mixed with 3.61 cm³ of deionized water to obtain 10 cm³ of 10 M nitric acid.

Appendix B⁻⁻N₂ Adsorption/desorption Isotherms at -196 °C



Figure B1 N₂ adsorption/desorption isotherm of AC.



Figure B2 N₂ adsorption/desorption isotherm of ACO10-6.



Figure B3 N₂ adsorption/desorption isotherm of ACO10-8.



Figure B4 N₂ adsorption/desorption isotherm of ACO10-12.



Figure B5 N_2 adsorption/desorption isotherm of ACN400.



Figure B6 N_2 adsorption/desorption isotherm of ACN600.



Figure B7 N_2 adsorption/desorption isotherm of ACN800.



Figure B8 N₂ adsorption/desorption isotherm of ACO10-6/N400.



Figure B9 N₂ adsorption/desorption isotherm of ACO10-8/N400.



Figure B10 N_2 adsorption/desorption isotherm of ACO10-12/N400.

o



Figure B11 N₂ adsorption/desorption isotherm of CB.



Figure B12 N₂ adsorption/desorption isotherm of ACB.



Figure B13 N₂ adsorption/desorption isotherm of ACBO10-12.



Figure B14 N_2 adsorption/desorption isotherm of ACBN400.



Figure B15 N₂ adsorption/desorption isotherm of ACBO10-12/N400.



Figure B16 N_2 adsorption/desorption isotherm of CP.



Figure B17 N₂ adsorption/desorption isotherm of ACP.



Figure B18 N₂ adsorption/desorption isotherm of ACPO10-12.



Figure B19 N_2 adsorption/desorption isotherm of ACPN400.



Figure B20 N₂ adsorption/desorption isotherm of ACPO10-12/N400.

σ





Figure C1 Pore size distribution of AC.



Figure C2 Pore size distribution of ACO10-6.

o



Figure C3 Pore size distribution of ACO10-8.



Figure C4 Pore size distribution of ACO10-12.



Figure C5 Pore size distribution of ACN400.



Figure C6 Pore size distribution of ACN600.



Figure C7 Pore size distribution of ACN800.



Figure C8 Pore size distribution of ACO10-6/N400.



Figure C9 Pore size distribution of ACO10-8/N400.



Ø





Figure C11 Pore size distribution of CB.



Figure C12 Pore size distribution of ACB.

-



Figure C13 Pore size distribution of ACBO10-12.



Figure C14 Pore size distribution of ACBN400.

-



Figure C15 Pore size distribution of ACBO10-12/N400.



Figure C16 Pore size distribution of CP.



Figure C17 Pore size distribution of ACP.



Figure C18 Pore size distribution of ACPO10-12.



Figure C19 Pore size distribution of ACPN400.

.



Figure C20 Pore size distribution of ACPO10-12/N400.

A	ppendix	D	The	Deconvol	lution of	C1s	XPS	Spectra	ł
---	---------	---	-----	----------	-----------	-----	-----	---------	---

_			
Region	Peak	Position (eV)	Assignment
Cls	1	284.3±0.1	Graphite
	2	285.0±0.1	Aliphatics
	3	285.4±0.2	Carbons linked to nitrogen
	4	285.8±0.1	Hydroxyl, Ether
	5	286.6±0.2	Carbonyl
	6	289.2	Esters, Lactones, Anhydrides, Acids, Hydroxyl

 Table D1
 Assigned binding energies for C1s XPS spectra

—— Peak 1	— · - Peak 2	Peak 3			
Peak 4	— — Peak 5	— — Peak 6			
Envelope peak					



σ

Figure D1 C1s XPS spectra of AC.



Figure D2 C1s XPS spectra of ACO10-6.



Figure D3 C1s XPS spectra of ACO10-8.



Figure D4 C1s XPS spectra of ACO10-12.



Figure D5 C1s XPS spectra of ACN400.



Figure D6 C1s XPS spectra of ACN600.



Figure D7 C1s XPS spectra of ACN800.



Figure D8 C1s XPS spectra of ACO10-6/N400.



Figure D9 C1s XPS spectra of ACO10-8/N400.



Figure D10 C1s XPS spectra of ACO10-12/N400.



Figure D11 C1s XPS spectra of CB.



Figure D12 C1s XPS spectra of ACB.



Figure D13 C1s XPS spectra of ACBO10-12.



Figure D14 C1s XPS spectra of ACBN400.



Figure D15 C1s XPS spectra of ACBO10-12/N400.



Figure D16 C1s XPS spectra of CP.



Figure D17 C1s XPS spectra of ACP.



Figure D18 C1s XPS spectra of ACPO10-12.



Figure D19 C1s XPS spectra of ACPN400.



÷,

Figure D20 C1s XPS spectra of ACPO10-12/N400.

Region	Peak	Position	Assignment
		(eV)	
Ols	1	530.3	Oxides
	2	531.3±0.1	Carbonyl, Carboxyl
100	3	532.1±0.1	Ketones, Lactones, Acids, Esters
	4	532.9±0.1	Hydroxyl, Carboxyl
	5	533.6±0.1	Hydroxyl attached to aromatic ring, oxygen in bridge
			(esters)
	6	534.9±0.1	Adsorbed water

 Table E1
 Assigned binding energies for O1s XPS spectra

—— Peak 1	— Peak 2	Peak 3		
Peak 4	— — Peak 5	— — Peak 6		
Envelope peak				



Figure E1 O1s XPS spectra of AC.

o



Figure E2 Ols XPS spectra of ACO10-6.



Figure E3 Ols XPS spectra of ACO10-8.



Figure E4 Ols XPS spectra of ACO10-12.



Figure E5 Ols XPS spectra of ACN400.



Figure E6 Ols XPS spectra of ACN600.



Figure E7 Ols XPS spectra of ACN800.



Figure E8 Ols XPS spectra of ACO10-6/N400.



Figure E9 Ols XPS spectra of ACO10-8/N400.



Figure E10 O1s XPS spectra of ACO10-12/N400.



Figure E11 O1s XPS spectra of ACB.



Figure E12 OIs XPS spectra of ACBO10-12.



Figure E13 Ols XPS spectra of ACBN400.



Figure E14 O1s XPS spectra of ACBO10-12/N400.



Figure E15 Ols XPS spectra of ACP.



Figure E16 O1s XPS spectra of ACPO10-12.



Figure E17 O1s XPS spectra of ACPN400.



Figure E18 O1s XPS spectra of ACPO10-12/N400.

Appendix F The Deconvolution of N1s XPS Spectra.

 Table F1
 Assigned binding energies for N1s XPS spectra

Region	Peak	Position (eV)	Assignment	
N1s	1	398.1±0.1	Pyridinic	
	2	398.9±0.1	Pyridinic, Imine	
	3	399.8±0.2	Imine, Amide, Amine	
	4	400.7±0.2	Pyrrolic, Pyridonic	
	5	401.4±0.3	Quaternary nitrogen	
	6	402.8±0.2	Pyridine-N-oxide	

Peak 1	— Peak 2	Peak 3
Peak 4	— — Peak 5	— — Peak 6

----- Envelope peak



Figure F1 N1s XPS spectra of ACO10-6.



Figure F2 N1s XPS spectra of ACO10-8.



Figure F3 N1s XPS spectra of ACO10-12.

÷



Figure F4 N1s XPS spectra of ACN400.



Figure F5 N1s XPS spectra of ACN600.

109



Figure F6 N1s XPS spectra of ACN800.



Figure F7 N1s XPS spectra of ACO10-6/N400.



Figure F8 N1s XPS spectra of ACO10-8/N400.



Figure F9 N1s XPS spectra of ACO10-12/N400.

Ø



Figure F10 N1s XPS spectra of ACBO10-12.



Figure F11 N1s XPS spectra of ACBN400.



Figure F12 N1s XPS spectra of ACBO10-12/N400.



Figure F13 N1s XPS spectra of CP.



Figure F14 N1s XPS spectra of ACP.



Figure F15 N1s XPS spectra of ACPO10-12.



Figure F16 N1s XPS spectra of ACPN400.



D

Figure F17 N1s XPS spectra of ACPO10-12/N400.





Figure G1 CO₂ adsorption/desorption profile of AC.



Figure G2 CO₂ adsorption/desorption profile of ACO10-6.



Figure G3 CO₂ adsorption/desorption profile of ACO10-8.



Figure G4 CO₂ adsorption/desorption profile of ACO10-12.



Figure G5 CO₂ adsorption/desorption profile of ACN400.



Figure G6 CO₂ adsorption/desorption profile of ACN600.



Figure G7 CO_2 adsorption/desorption profile of ACN800.



Figure G8 CO₂ adsorption/desorption profile of ACO10-6/N400.



Figure G9 CO₂ adsorption/desorption profile of ACO10-8/N400.



Figure G10 CO₂ adsorption/desorption profile of ACO10-12/N400.

σ



Figure G11 CO₂ adsorption/desorption profile of ACB.



Figure G12 CO₂ adsorption/desorption profile of ACBO10-12.

o.



Figure G13 CO₂ adsorption/desorption profile of ACBN400.



Figure G14 CO₂ adsorption/desorption profile of ACBO10-12/N400.

ο



Figure G15 CO₂ adsorption/desorption profile of ACP.



Figure G16 CO₂ adsorption/desorption profile of ACPO10-12.



Figure G17 CO₂ adsorption/desorption profile of ACPN400.



Figure G18 CO₂ adsorption/desorption profile of ACPO10-12/N400.

CURRICULUM VITAE

Name: Ms. Nanthiya Thongwichit

Date of Birth: January 18, 1991

Nationality: Thai

University Education:

2008–2012 Bachelor Degree of Petrochemical and Polymeric Materials, Faculty of Engineering and Industrial Technology, Silpakorn University, Bangkok, Thailand

Work Experience:

April 2010	Position:	Student Internship	
	Company name:	Khanom Gas Separation Plant PTT Co.,	
		Ltd., Nakhon Si Thammarat, Thailand	

Publication:

 Thongwichit, N.; Li, O.L.; Yaowarat, W.; Saito, N.; and Suriyapraphadilok, U. (2015, April 30) Adsorption of carbon dioxide by solution plasma synthesized hetero-atom doped carbon nano spheres. <u>Japanese Journal of Applied Physics</u> (Submitted).

Proceedings:

 Thongwichit, N.; and Suriyapraphadilok, U. (2015, April 21) Effects of surface treatment of activated carbon on its surface chemistry and textural characteristics. Proceeding of <u>The 6th Research Symposium on Petroleum, Petrochemicals, and Advanced Materials and The 21th PPC Symposium on Petroleum, Petrochemicals, and Polymers, Chulalongkorn University, Bangkok, Thailand.
</u>

Presentations:

 Thongwichit, N.; and Suriyapraphadilok, U. (2015, January 23-26) Modification of carbon surface by oxidation and ammonia treatment for carbon dioxide adsorption. Paper presented at <u>The 15th Internal Symposium on Biomimetic Materials Processing (BMMP-15)</u>, Nagoya University, Nagoya, Japan.

- Thongwichit, N.; Li, O.L.; Saito, N.; and Suriyapraphadilok, U. (2015, March 26-31) Carbon nano spheres by solution plasma process for CO₂ adsorption. Paper presented at <u>The 7th Internal Symposium on Advanced Plasma Science</u> and Its Applications for Nitrides and Nanomaterials and The 8th Internal Conference on Plasma-Nano Technology & Science, Nagoya University, Nagoya, Japan.
- Thongwichit, N.; and Suriyapraphadilok, U. (2015, August 16-10) Different approaches of surface treatment on activated carbon for CO₂ captured enhancement. Paper presented at <u>The 250th American Chemical Society National Meeting & Exposition</u>, Boston, United States of America.