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#### **APPENDICES**

## Appendix A The Catalytic Activity of 1-Hexyne, 1-Hexene Selectivity and *n*-Hexane Selectivity of 0.3 wt% Pd and Pd-Mn Supported on Alumina Catalysts

 Table A1
 The peak area and mole of components for calibration curve

Wt% of		Peak area	1	Mole of Components			
components	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	
0.5	343.5	318.3	356.4	2.04E-8	1.98E-8	1.97E-8	
1	645.7	673.6	693.4	4.08E-8	3.96E-8	3.93E-8	
2.5	1747.4	1720.0	1747.7	1.02E-7	9.91E-8	9.83E-8	

 Table A2
 The slope of calibration curve

Components	Slope of calibration curve
Hexyne	1.69E10
Hexene	1.73E10
Hexane	1.78E10



Figure A1 The calibration curve of 1-hexyne.



Figure A2 The calibration curve of 1-hexene.



Figure A3 The calibration curve of *n*-hexane.

**Table A3** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40  $^{\circ}$ C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub>

	Peak area	1	Mole	of Comp	onents	nents Conv $\begin{array}{c c} C_6H_{12} & C_6H_{14} \\ Sel & Sel \end{array}$			C <sub>6</sub> H <sub>12</sub> VId
C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	$C_{6}H_{12}$	$C_{6}H_{14}$	(%)	(%)	(%)	(%)
815.2	498.6	24.7	4.8E-8	2.9E-8	1.4E-9	41.8	83.5	2.6	34.9
385.9	764.4	27.5	2.3E-8	4.4E-8	1.6E-9	72.5	73.9	4.0	53.5
27.3	806.7	28.4	1.6E-9	4.7E-8	1.6E-9	98.1	57.6	4.4	56.5
0.0	705.8	38.7	0.0	4.1E-8	2.2E-9	100.0	49.5	5.2	49.4
0.0	625.1	65.1	0.0	3.6E-8	3.7E-9	100.0	43.8	7.9	43.8

**Table A4** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40 °C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 0.5

	Peak area	1	Mole of Components			Conv	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>
C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	(%)	(%)	(%)	(%)
515.2	708.8	0.0	3.0E-8	4.1E-8	0.0	58.4	96.0	0.0	56.1
20.2	1068.2	10.3	1.2E-9	6.2E-8	5.8E-10	98.4	86.0	0.9	84.5
0.0	926.9	201.8	0.0	5.4E-8	1.1E-8	100.0	73.3	15.5	73.3
0.0	852.7	345.9	0.0	4.9E-8	2.0E-8	100.0	67.5	26.6	67.5
0.0	774.5	421.9	0.0	4.5E-8	2.4E-8	100.0	61.3	32.4	61.3

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**Table A5** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40 °C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 0.75

	Peak area	l	Mole of Components			Conv	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	$C_6H_{12}$
C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	(%)	(%)	(%)	(%)
517.8	856.3	3.4	3.1E-8	5.0E-8	1.9E-10	63.0	95.1	-0.4	60.0
23.6	1208.6	9.4	1.4E-9	7.0E-8	5.3E-10	98.3	86.1	0.7	84.6
0.0	1064.2	235.7	0.0	6.2E-8	1.3E-8	100.0	74.5	16.0	74.5
0.0	979.2	400.1	0.0	5.7E-8	2.3E-8	100.0	68.6	27.2	68.6
0.0	863.2	522.6	0.0	5.0E-8	2.9E-8	100.0	60.4	35.6	60.4

**Table A6** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40  $^{\circ}$ C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 1.0

	Peak area	1	Mole of Components			Conv	$C_6H_{12}$	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>
C <sub>6</sub> H <sub>10</sub>	$C_{6}H_{12}$	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	(%)	Sel (%)	(%)	(%)
233.0	1007.3	8.6	1.4E-8	5.8E-8	4.8E-10	81.2	98.2	0.8	79.7
0.0	1128	81.5	0.0	6.5E-8	4.6E-9	100.0	89.3	6.3	89.3
0.0	998.4	224.1	0.0	5.8E-8	1.3E-8	100.0	79.0	17.2	79.0
0.0	895.3	363.2	0.0	5.2E-8	2.0E-8	100.0	70.8	27.9	70.8
0.0	820.4	447.6	0.0	4.8E-8	2.5E-8	100.0	64.9	34.4	64.9

**Table A7** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40  $^{\circ}$ C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 1.5

	Peak area Mole			of Comp	onents	Conv	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>
C <sub>6</sub> H <sub>10</sub>	$C_{6}H_{12}$	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	(%)	(%)	(%)	(%)
768.9	463.4	5.6	4.5E-8	2.7E-8	3.2E-10	38.0	96.6	1.1	36.7
401.8	772.1	10.6	2.4E-8	4.5E-8	6.0E-10	67.6	90.4	1.2	61.1
35.8	985.2	29.7	2.1E-9	5.7E-8	1.7E-9	97.1	80.3	2.4	78.0
0.0	935.8	194.8	0.0	5.4E-8	1.1E-8	100.0	74.0	15.0	74.0
0.0	845.3	267.5	0.0	4.9E-8	1.5E-8	100.0	66.9	20.6	66.9

**Table A8** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40 °C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 2.0

	Peak area	1	Mole	of Comp	Conv	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	$C_6H_{12}$	
C <sub>6</sub> H <sub>10</sub>	$C_{6}H_{12}$	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	(%)	(%)	Sel (%)	(%)
842.8	395.7	7.8	5.0E-8	2.3E-8	4.4E-10	32.0	97.8	1.9	31.3
535.4	661.1	17.3	3.2E-8	3.8E-8	9.7E-10	56.8	92.1	2.3	52.3
90.2	953.9	35.7	5.3E-9	5.5E-8	2.0E-9	92.7	81.4	3.0	75.5
0.0	948.6	160.2	0.0	5.5E-8	9.0E-9	100.0	75.1	12.3	75.1
0.0	850.2	200.0	0.0	4.9E-8	1.1E-8	100.0	67.3	15.4	67.3

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**Table A9** The peak area, mole of components, catalytic activity, 1-hexene selectivity, *n*-hexane selectivity and 1-hexene yield for the hydrogenation of 1-hexyne at 40  $^{\circ}$ C and 1.5 bar over 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 5.0

]	Peak area		Mole	of Comp	onents	Conv	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>
C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	(%)	(%)	(%)	(%)
1013.1	384.8	8.1	6.0E-8	2.2E-8	4.6E-10	27.7	97.4	2.0	27.0
645.7	701.4	20.2	3.8E-8	4.1E-8	1.1E-9	53.9	91.1	2.6	49.1
193.7	1008.3	38.6	1.1E-8	5.8E-8	2.2E-9	86.2	81.9	3.1	70.6
0.0	1082.1	189.5	0.0	6.3E-8	1.1E=8	100.0	75.8	12.9	75.8
0.0	951.2	294.7	0.0	5.5E-8	1.7E-8	100.0	66.6	20.1	66.6

Appendix B The Mole Fraction of 1-Hexyne, 1-Hexene and *n*-Hexane for 1-Hexyne Hydrogenation of 0.3 wt% Pd and Pd-Mn Supported on Alumina Catalysts

Table B1 The mole fraction for the hydrogenation of 1-hexyne at 40  $^{\circ}$ C and 1.5 bar over 0.3 %wt Pd/Al<sub>2</sub>O<sub>3</sub>

Time (h)	Mo	ole fracti	on
Time (n)	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>
0	1.00	0.00	0.00
1	0.61	0.37	0.02
2	0.33	0.65	0.02
3	0.03	0.94	0.03
4	0.00	0.95	0.03
5	0.00	0.91	0.09

**Table B2** The mole fraction for the hydrogenation of 1-hexyne at 40  $^{\circ}$ C and 1.5 bar over 0.3 %wt Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 0.5, 0.75 and 1.0

	Pd	I/Mn = 0	).5	Pd	Pd/Mn = 0.75			Mn = 1	1.0		
Time (h)	Mole fraction										
	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>14</sub>	$C_{6}H_{12}$	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>		
0	1.00	0.00	0.00	1.00	0.00	0.00	1.00	0.00	0.00		
1	0.43	0.57	0.00	0.38	0.62	0.00	0.19	0.80	0.01		
2	0.02	0.97	0.01	0.02	0.97	0.01	0.00	0.93	0.07		
3	0.00	0.82	0.17	0.00	0.82	0.18	0.00	0.82	0.18		
4	0.00	0.72	0.28	0.00	0.72	0.28	0.00	0.72	0.28		
5	0.00	0.65	0.35	0.00	0.63	0.37	0.00	0.65	0.35		

	Pd	Mn = 1	1.5	Pd/Mn = 2.0			Pd	Pd/Mn = 5.0			
Time (h)	Mole fraction										
	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>10</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>	C <sub>6</sub> H <sub>12</sub>	C <sub>6</sub> H <sub>14</sub>		
0	1.00	0.00	0.00	1.00	0.00	0.00	1.00	0.00	0.00		
1	0.63	0.37	0.00	0.68	0.31	0.01	0.72	0.27	0.01		
2	0.34	0.65	0.01	0.45	0.54	0.01	0.48	0.51	0.01		
3	0.03	0.94	0.03	0.09	0.88	0.03	0.16	0.81	0.03		
4	0.00	0.83	0.17	0.00	0.86	0.14	0.00	0.85	0.15		
5	0.00	0.76	0.24	0.00	0.81	0.19	0.00	0.77	0.23		

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**Table B3** The mole fraction for the hydrogenation of 1-hexyne at 40  $\degree$ C and 1.5 barover 0.3 % wt Pd/Al<sub>2</sub>O<sub>3</sub> at Pd/Mn = 1.5, 2.0 and 5.0

# Appendix C Hydrogen Chemisorption of 0.3 %wt Pd and Pd-Mn supported on alumina catalysts

Table C1  $H_2$  chemisorption results of 0.3 wt% Pd/Al<sub>2</sub>O<sub>3</sub> catalysts at various Mn loadings

- Catalysts	H/Pd	
Pd/Al <sub>2</sub> O <sub>3</sub>	0.3381 0.1875 0.2015	
$Pd-Mn/Al_2O_3$ ( $Pd/Mn = 0.5$ )		
$Pd-Mn/Al_2O_3 (Pd/Mn = 0.75)$		
$Pd-Mn/Al_2O_3 (Pd/Mn = 1.0)$	0.2154	
$Pd-Mn/Al_2O_3 (Pd/Mn = 1.5)$	0.1738	
$Pd-Mn/Al_2O_3 (Pd/Mn = 2.0)$	0.1541	
$Pd-Mn/Al_2O_3 (Pd/Mn = 5.0)$	0.1439	

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### **Proceeding:**

1. Kijtithanont, N., and Kitiyanan, B. (2014, April 22) Selective Hydrogenation of 1-Hexyne Using Pd-Mn on Alumina Catalysts. <u>Proceedings of the 20<sup>th</sup> PPC</u> <u>Symposium on Petroleum, Petrochemicals, and Polymers 2014</u>, Bangkok, Thailand.

#### **Presentation:**

1. Kijtithanont, N., and Kitiyanan, B. (2014, April 22) Selective Hydrogenation of 1-Hexyne Using Pd-Mn on Alumina Catalysts. Poster presented at <u>The 20<sup>th</sup> PPC</u> <u>Symposium on Petroleum, Petrochemicals, and Polymers 2014</u>, Bangkok, Thailand.