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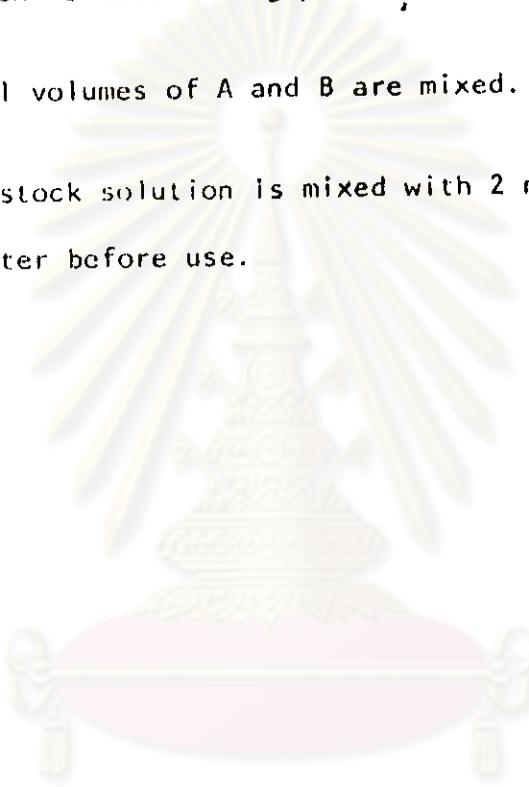
Dragendorff's spray reagent

Solution A : 0.85 g basic bismuth nitrate is dissolved in a mixture of 10 ml acetic acid and 40 ml water.

Solution B : A solution is made of 8 g potassium iodide in 20 ml water.

Stock solution : Equal volumes of A and B are mixed.

Spray reagent : 1 ml stock solution is mixed with 2 ml acetic acid and 10 ml water before use.



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Key to the Fig. XIX - LSolvent systems

- a = aluminium oxide G / chloroform  
b = aluminium oxide G / diethyl ether  
c = kieselgel G / chloroform  
d = kieselgel G / chloroform : acetone, 5 : 4  
e = kieselgel G / chloroform : ethyl alcohol, 97 : 3  
f = kieselgel G / chloroform : ethyl alcohol, 95 : 5  
g = kieselgel G / cyclohexane : ethyl acetate, 1 : 1  
h = kieselgel G / diethyl ether  
i = kieselgel G / ethyl acetate  
j = kieselgel G / ethyl acetate : diethyl ether, 1 : 1  
k = kieselgel G / ethyl acetate : isopropyl alcohol :  
strong solution of ammonium hydroxide,  
100 : 2 : 1

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Key to the Fig. XIX - L (continued)

- T<sub>a</sub> = alkaloids crystallised from concentrated ethyl alcohol  
solution of total crude alkaloids
- T<sub>b</sub> = purified alkaloidal extract
- T<sub>c</sub> = alkaloids fraction from aluminium oxide column
- T<sub>d</sub> = alkaloids fraction from aluminium oxide column
- T<sub>e</sub> = total alkaloids from isomerisation of TS<sub>2</sub>
- T<sub>f</sub> = total alkaloids from isomerisation of TS<sub>1</sub>
- TS<sub>1</sub> = isolated tetrahydroalstonine
- TS<sub>2</sub> = isolated rauniticine
- TS<sub>3</sub> = isolated 14-hydroxy-3-isorauniticine
- TS<sub>4</sub> = isolated oxindole (M<sup>+</sup> 368)
- TS<sub>5</sub> = isolated oxindole (M<sup>+</sup> 368)
- TS<sub>6</sub> = 3-isorauniticine (isomerisation product of TS<sub>2</sub>)
- TS<sub>7</sub> = akuammigine (isomerisation product of TS<sub>1</sub>)

Key to the Fig. XIX - L (continued)

Authentic alkaloids

1 = tetrahydroalstonine

2 = rauniticine

3 = akuammigine

4 = 3-isorauniticine

5 = isopteropodine

6 = pteropodine

7 = isorhynchophylline

8 = rhynchophylline

9 = rotundifoline

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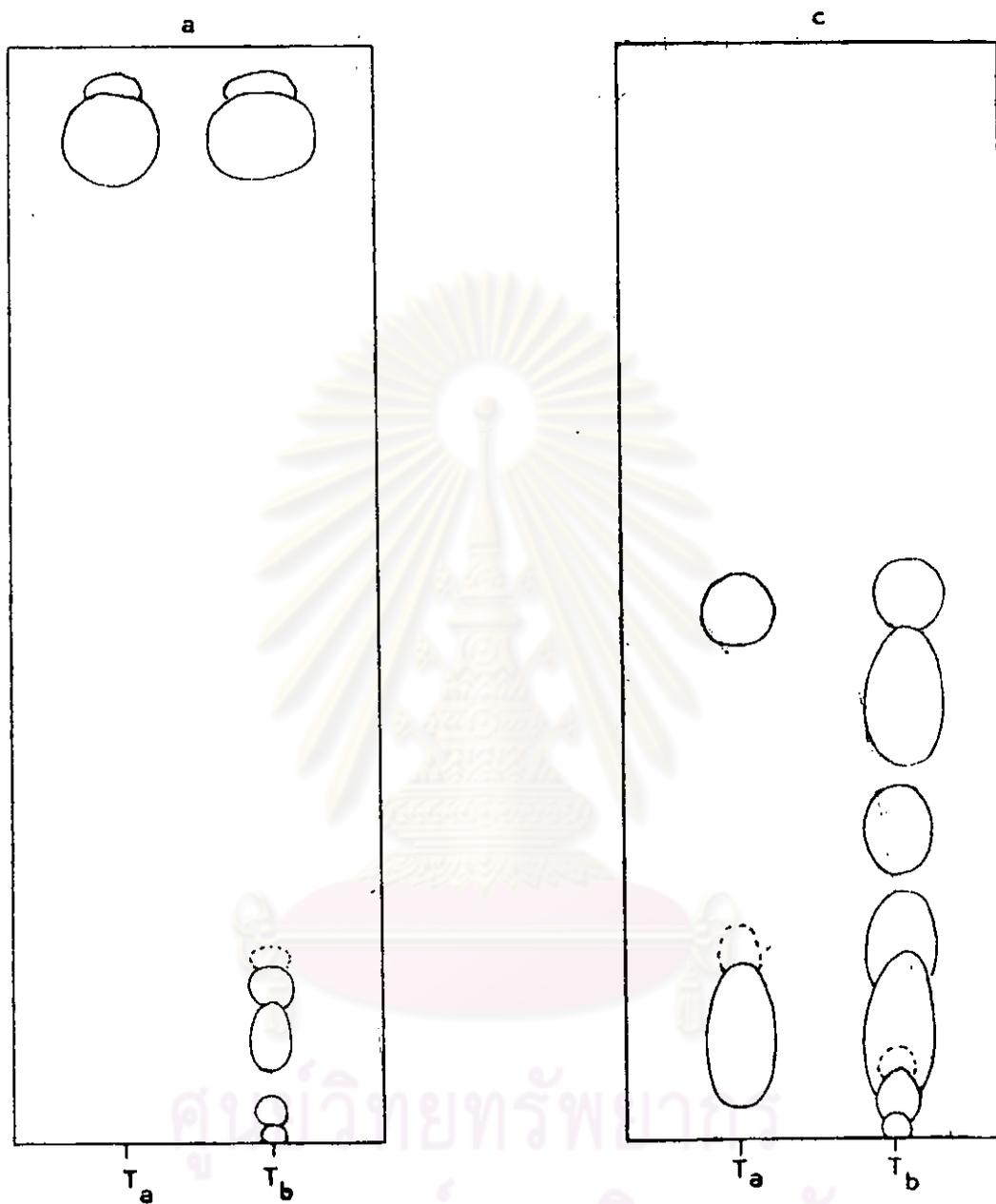


Fig. XIX Thin layer chromatograms of alkaloids.

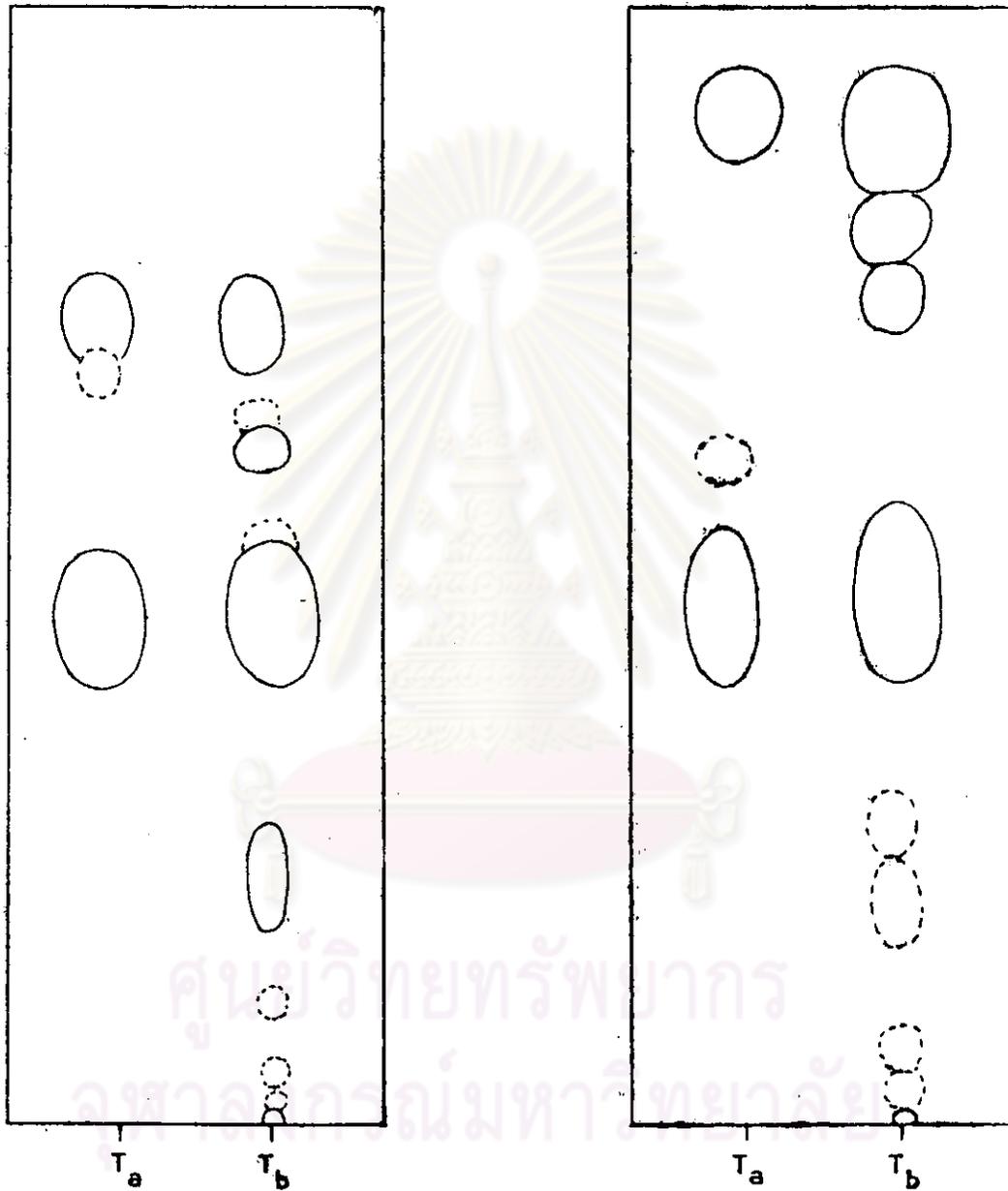


Fig. XX Thin layer chromatograms of alkaloids.

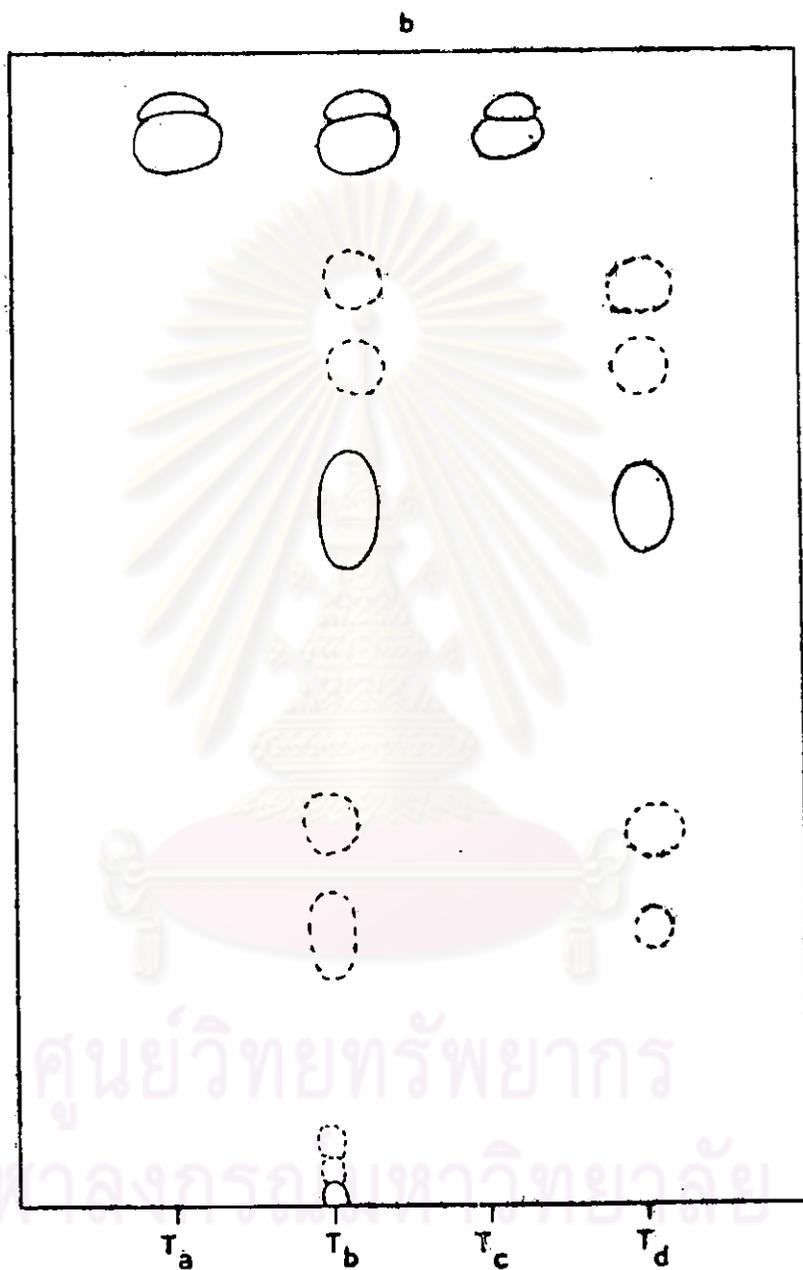


Fig. XXI Thin layer chromatogram of alkaloids.

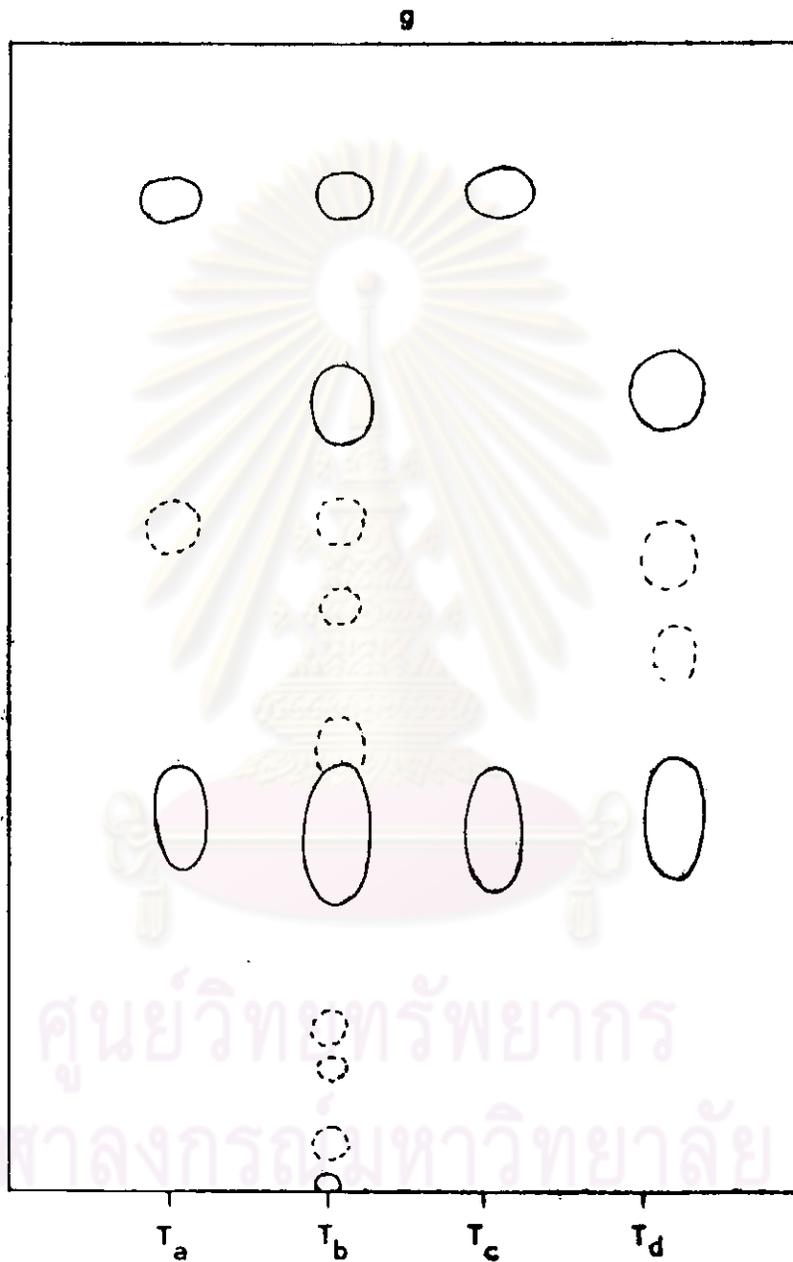


Fig. XXII Thin layer chromatogram of alkaloids.



Fig. XXIII Thin layer chromatogram of alkaloids.

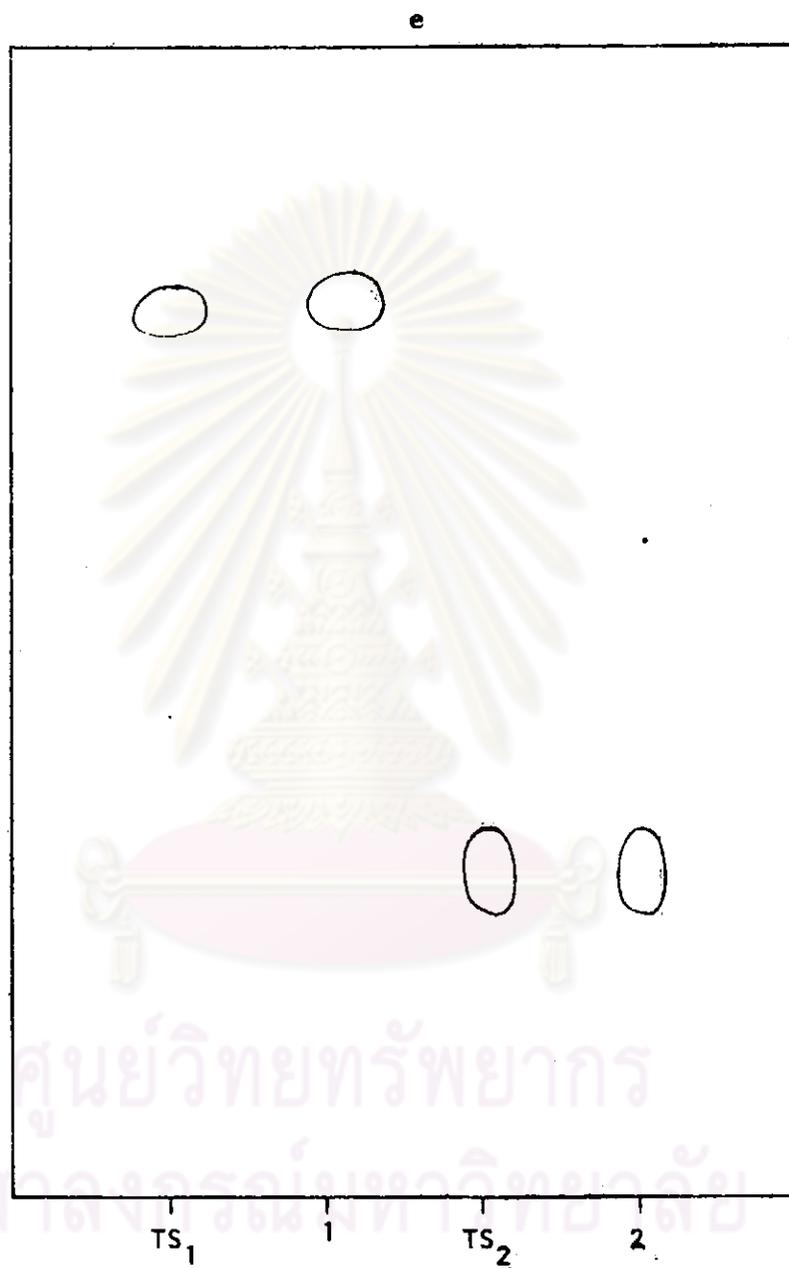


Fig. XXIV Thin layer chromatogram of alkaloids.

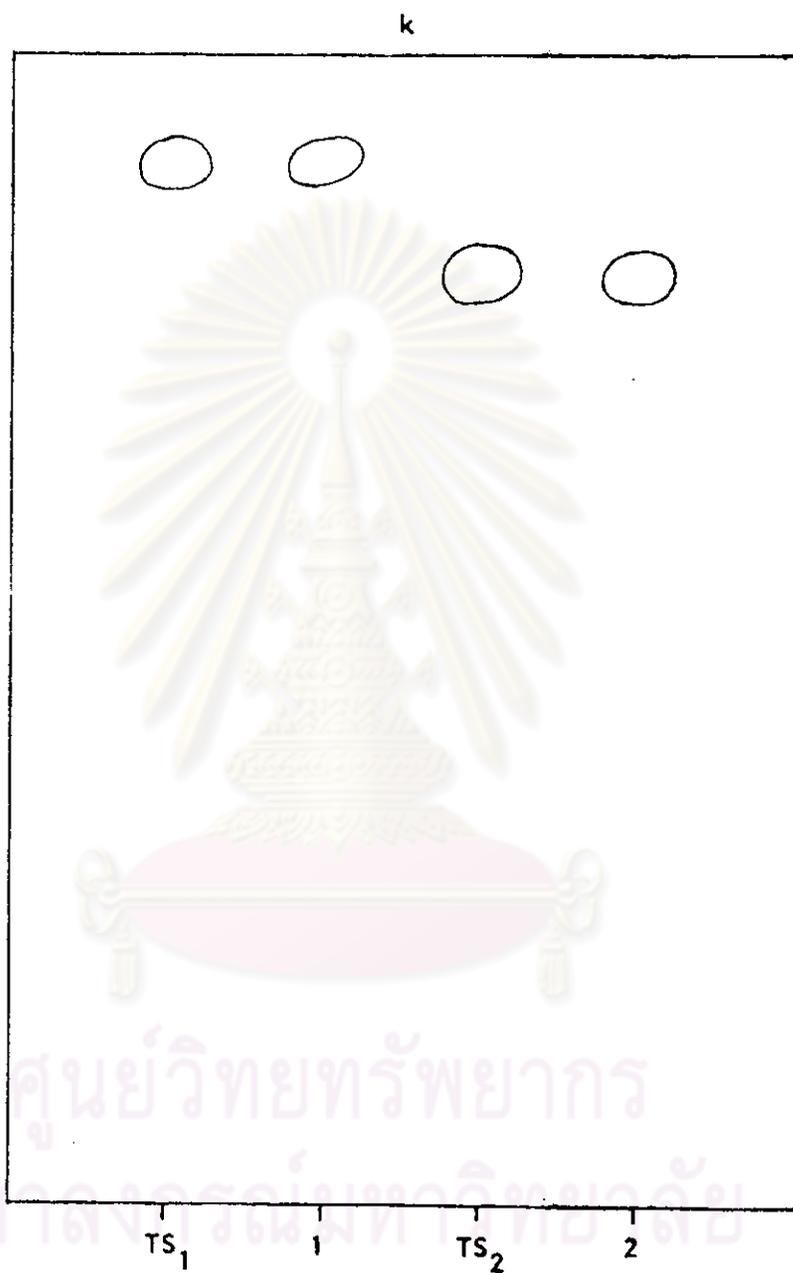


Fig. XXV Thin layer chromatogram of alkaloids.

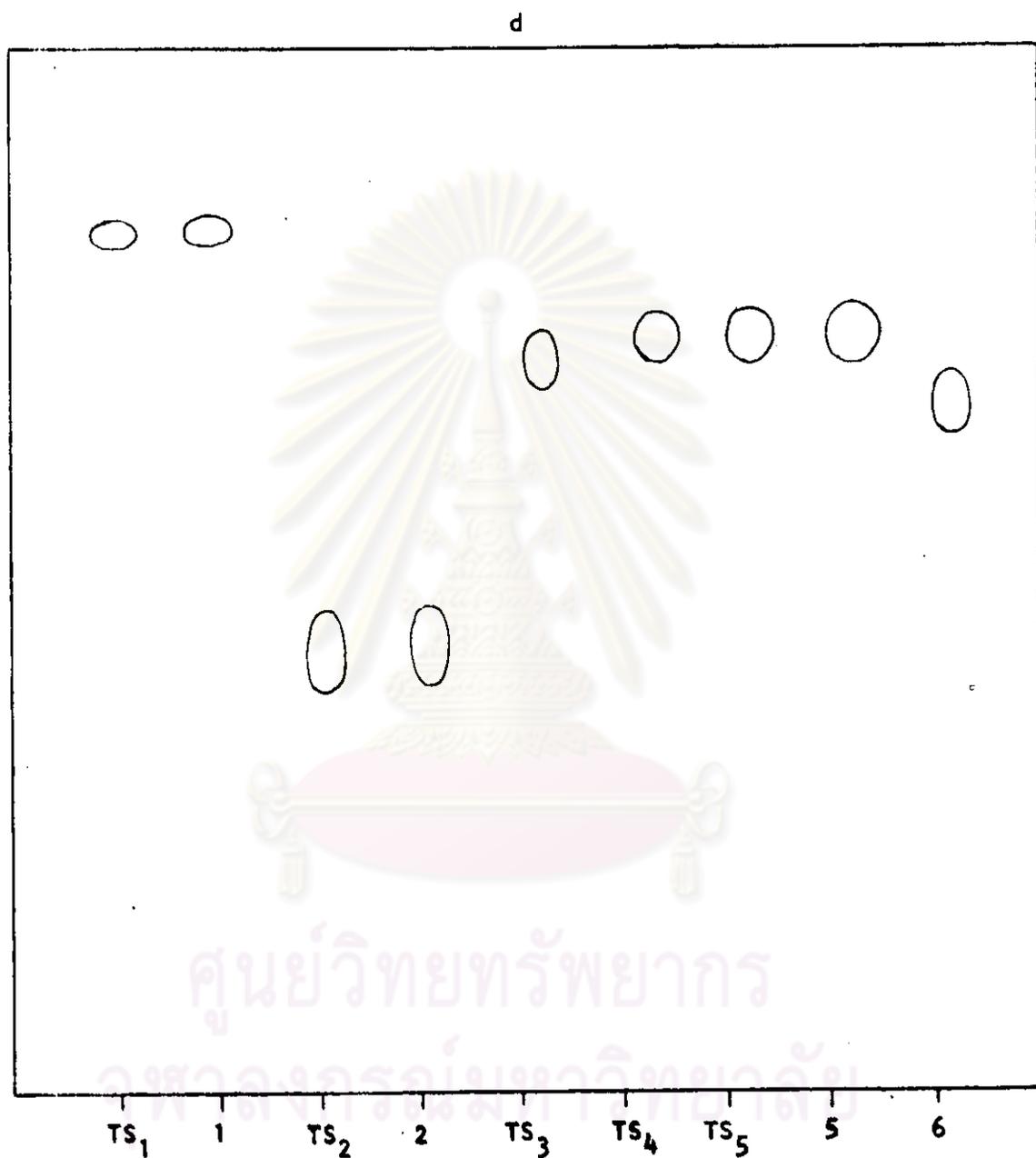


Fig. XXVI Thin layer chromatogram of alkaloids.



Fig. XXVII Thin layer chromatogram of alkaloids.

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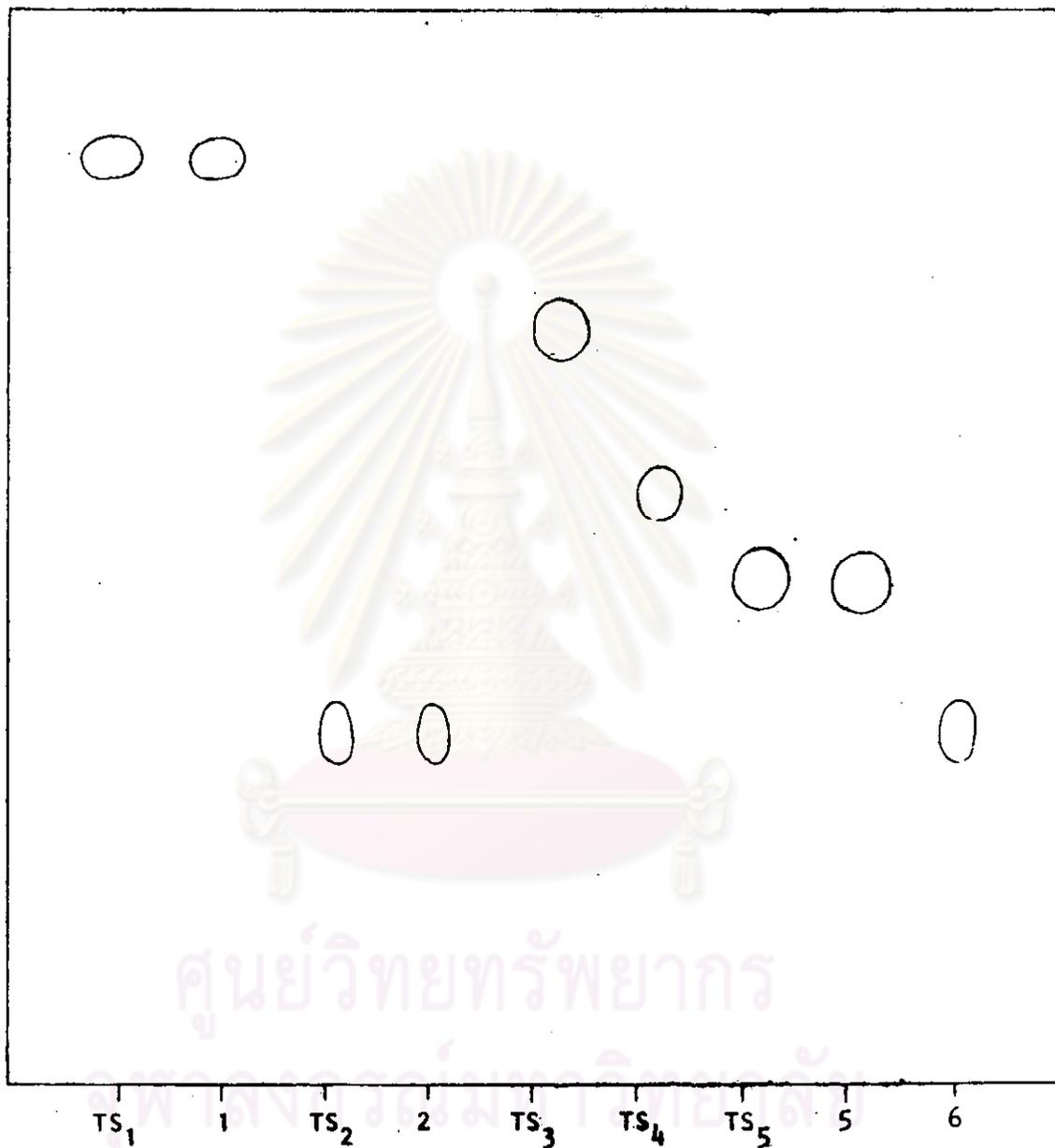


Fig. XXVIII Thin layer chromatogram of alkaloids.

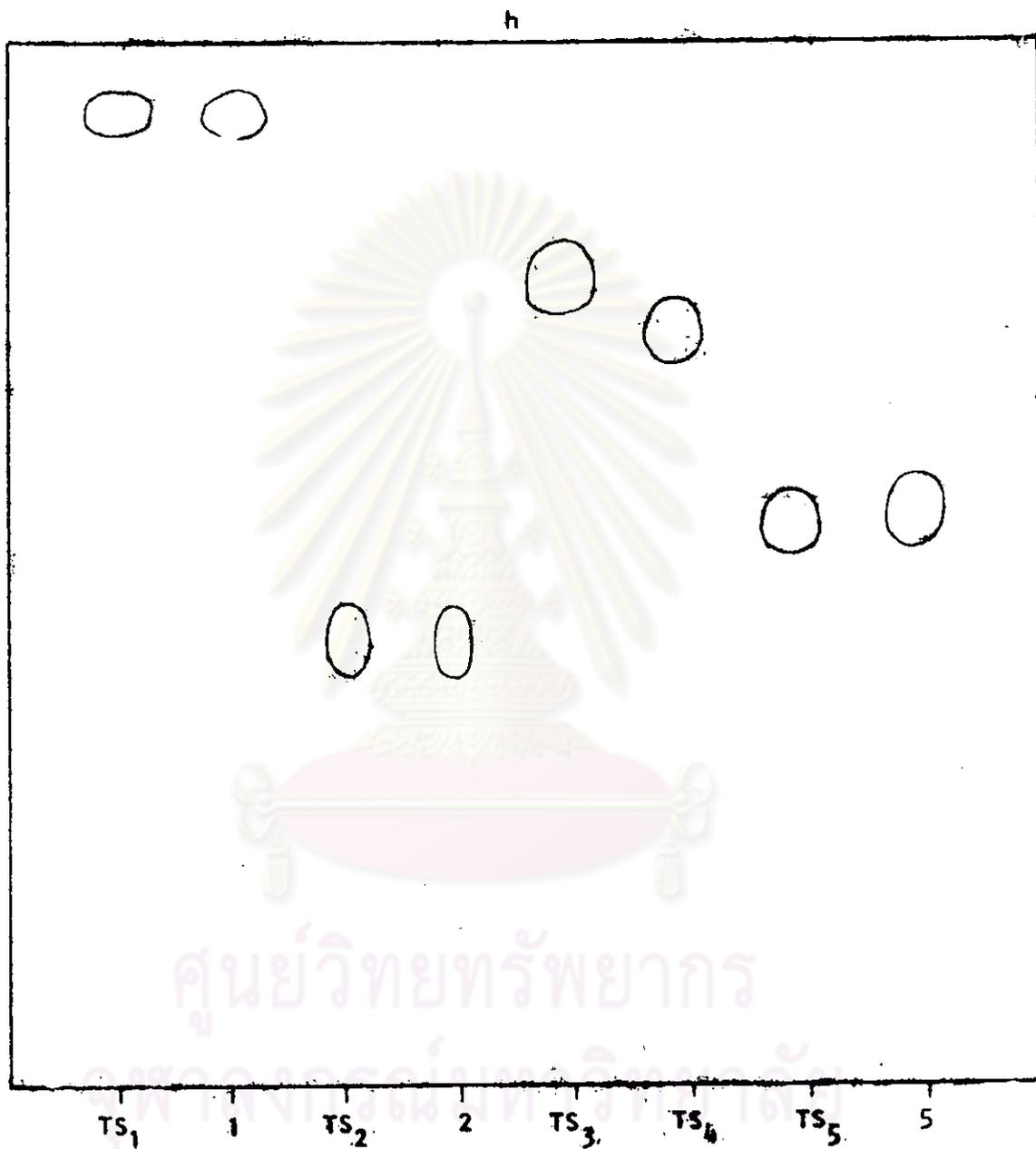


Fig. XXIX Thin layer chromatogram of alkaloids.

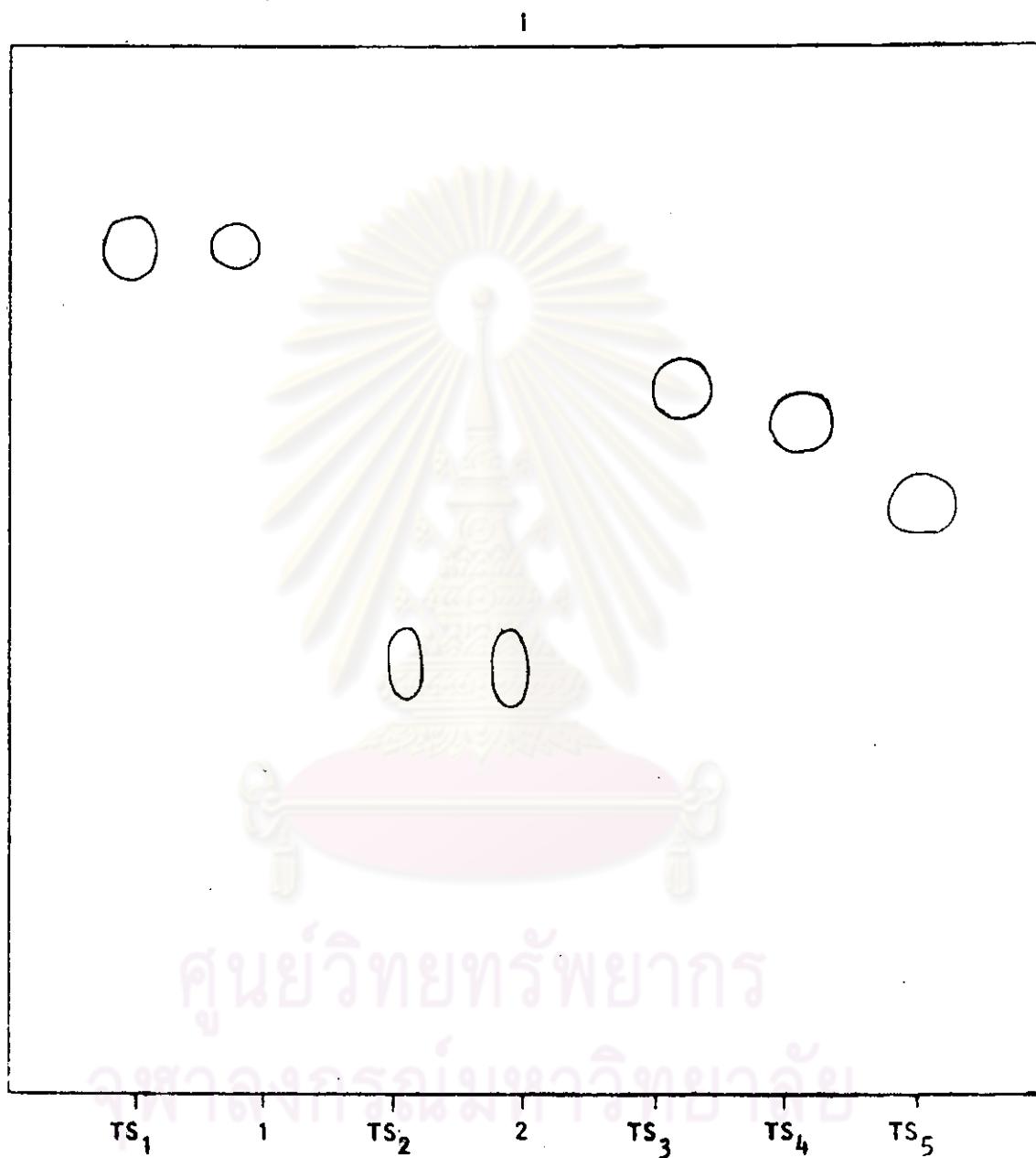


Fig. XXX Thin layer chromatogram of alkaloids.

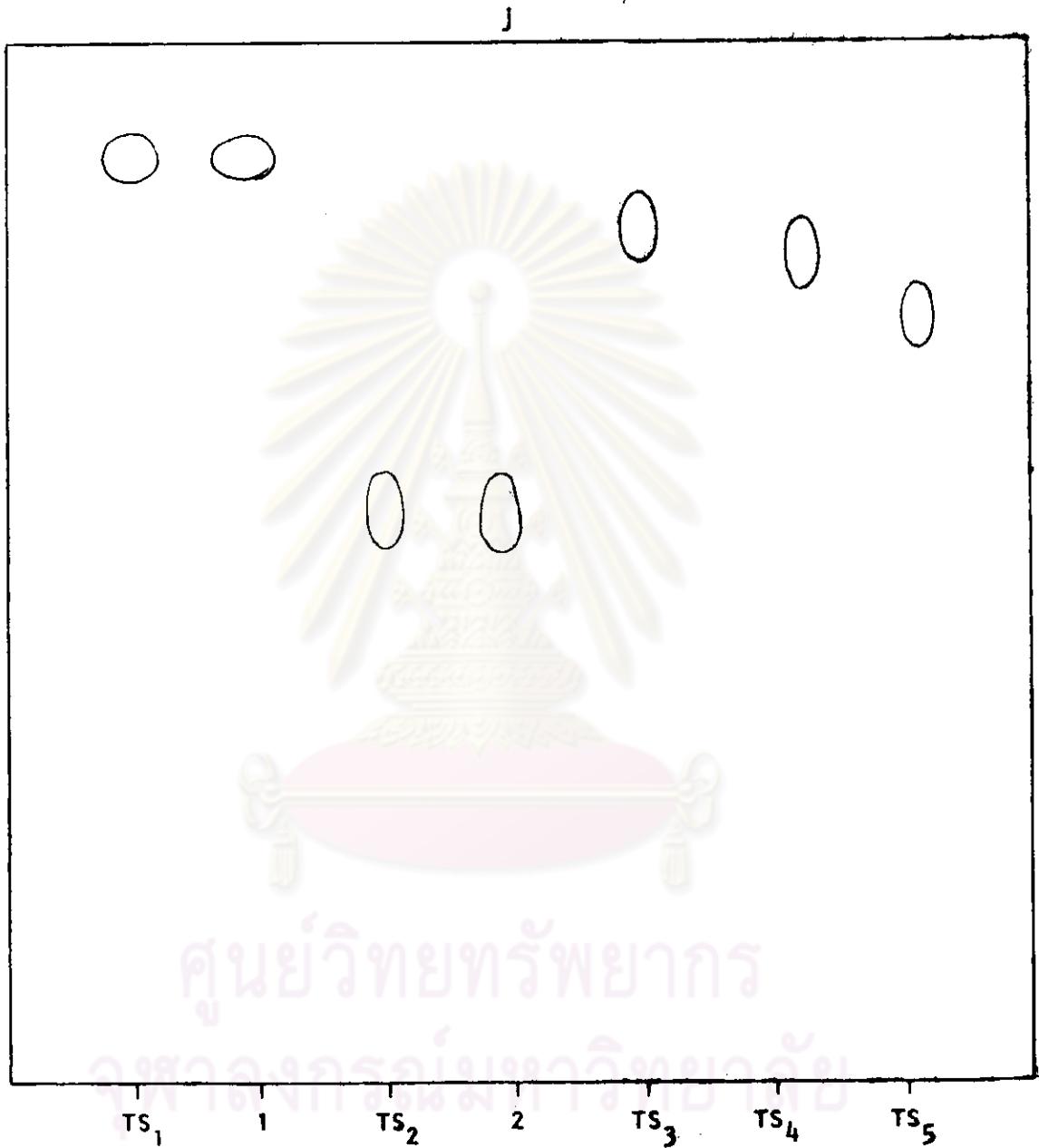


Fig. XXXI Thin layer chromatogram of alkaloids.



Fig. XXXII Thin layer chromatogram of alkaloids.



Fig. XXXIII Thin layer chromatogram of alkaloids.

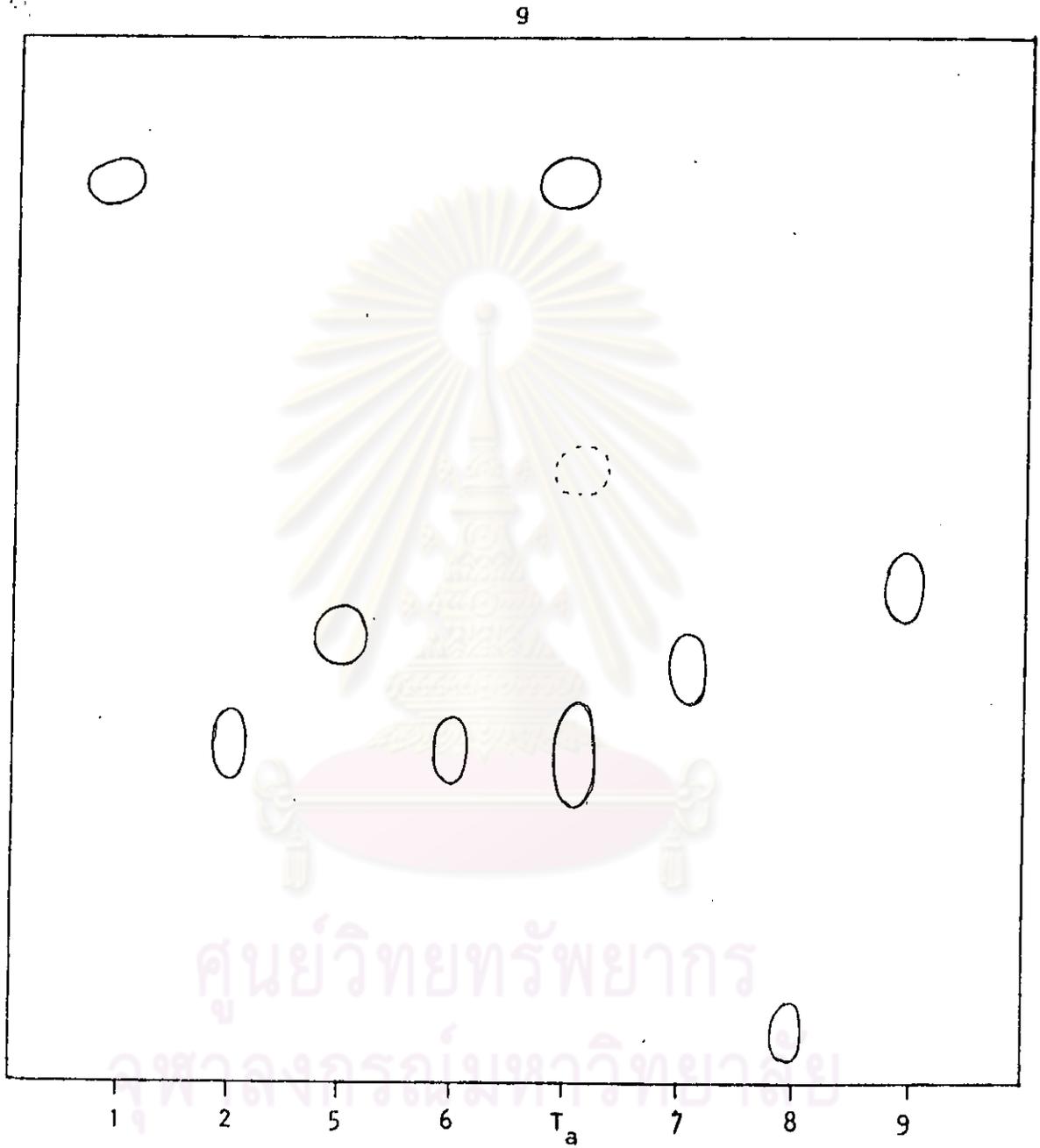


Fig. XXXIV Thin layer chromatogram of alkaloids.

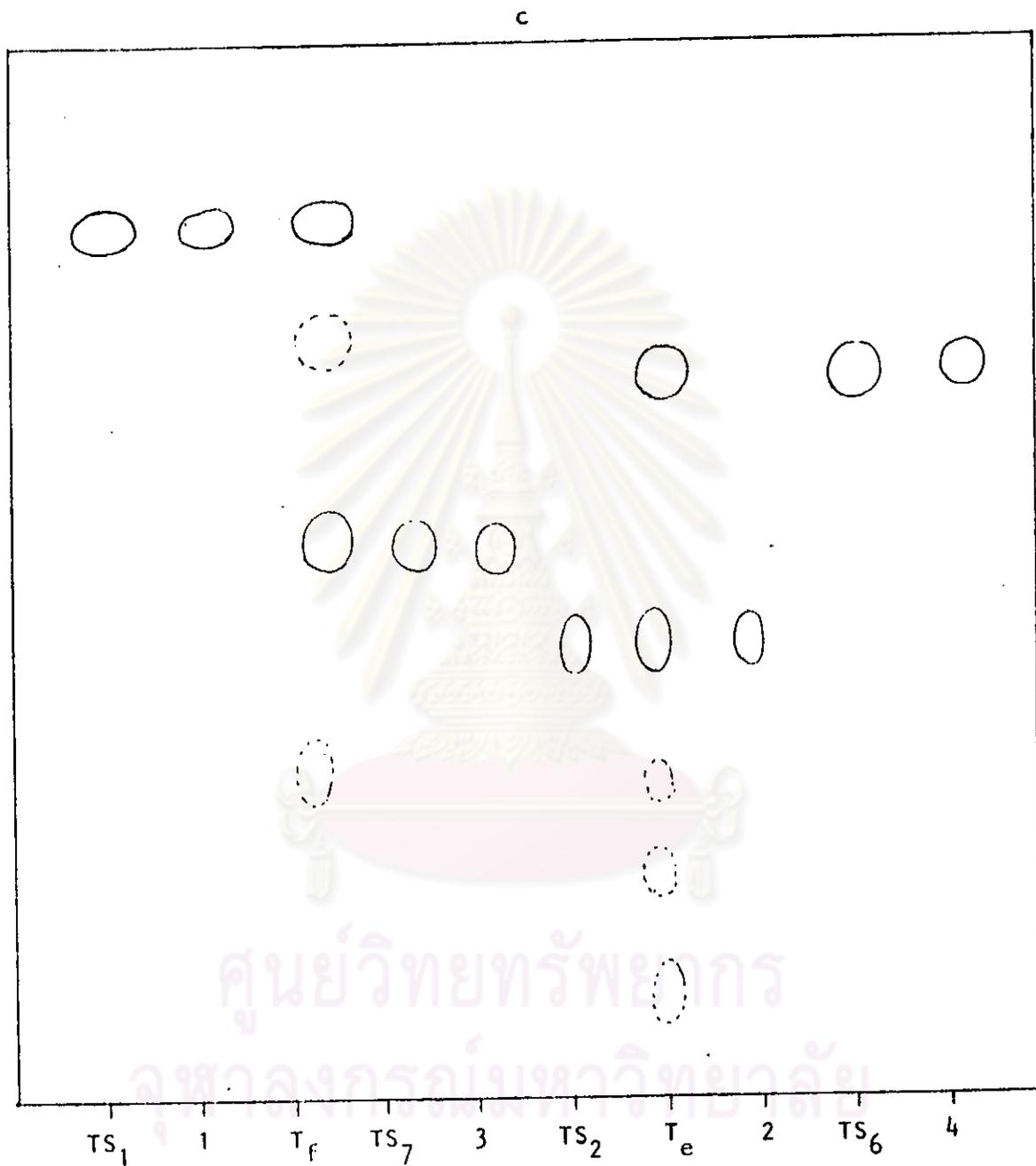


Fig. XXXV Thin layer chromatogram of alkaloids.

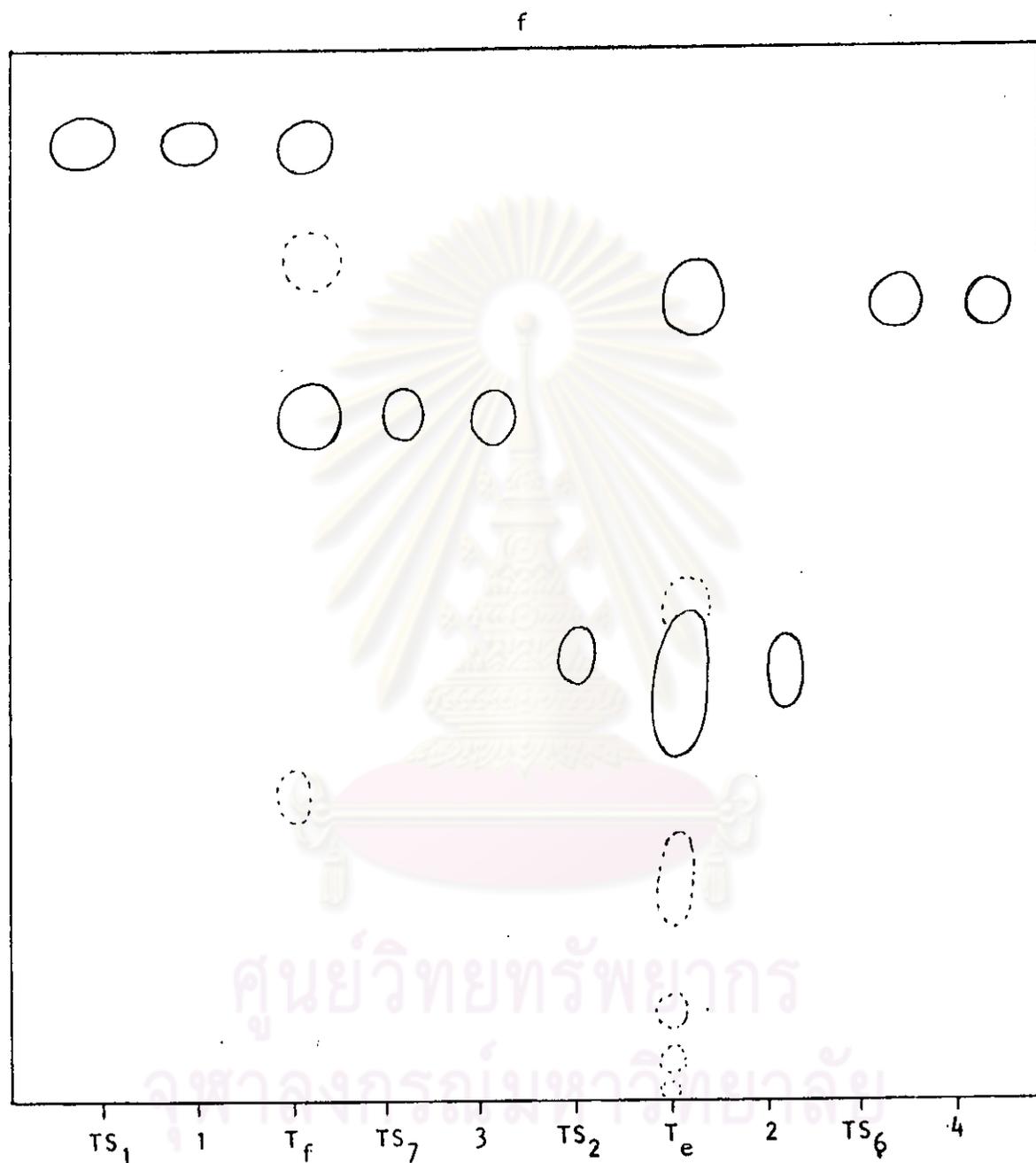


Fig. XXXVI Thin layer chromatogram of alkaloids.

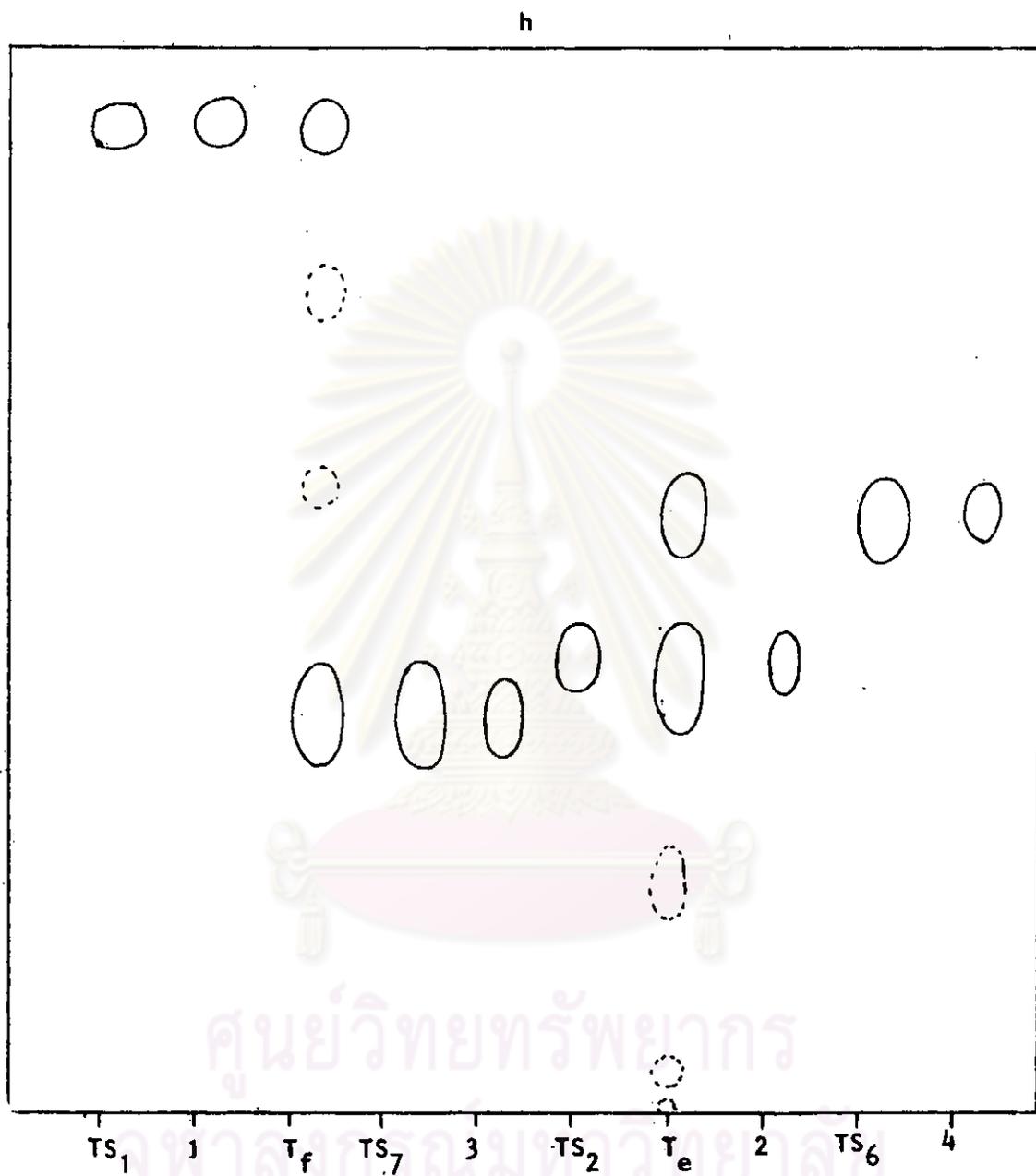


Fig. XXXVII Thin-layer chromatogram of alkaloids.

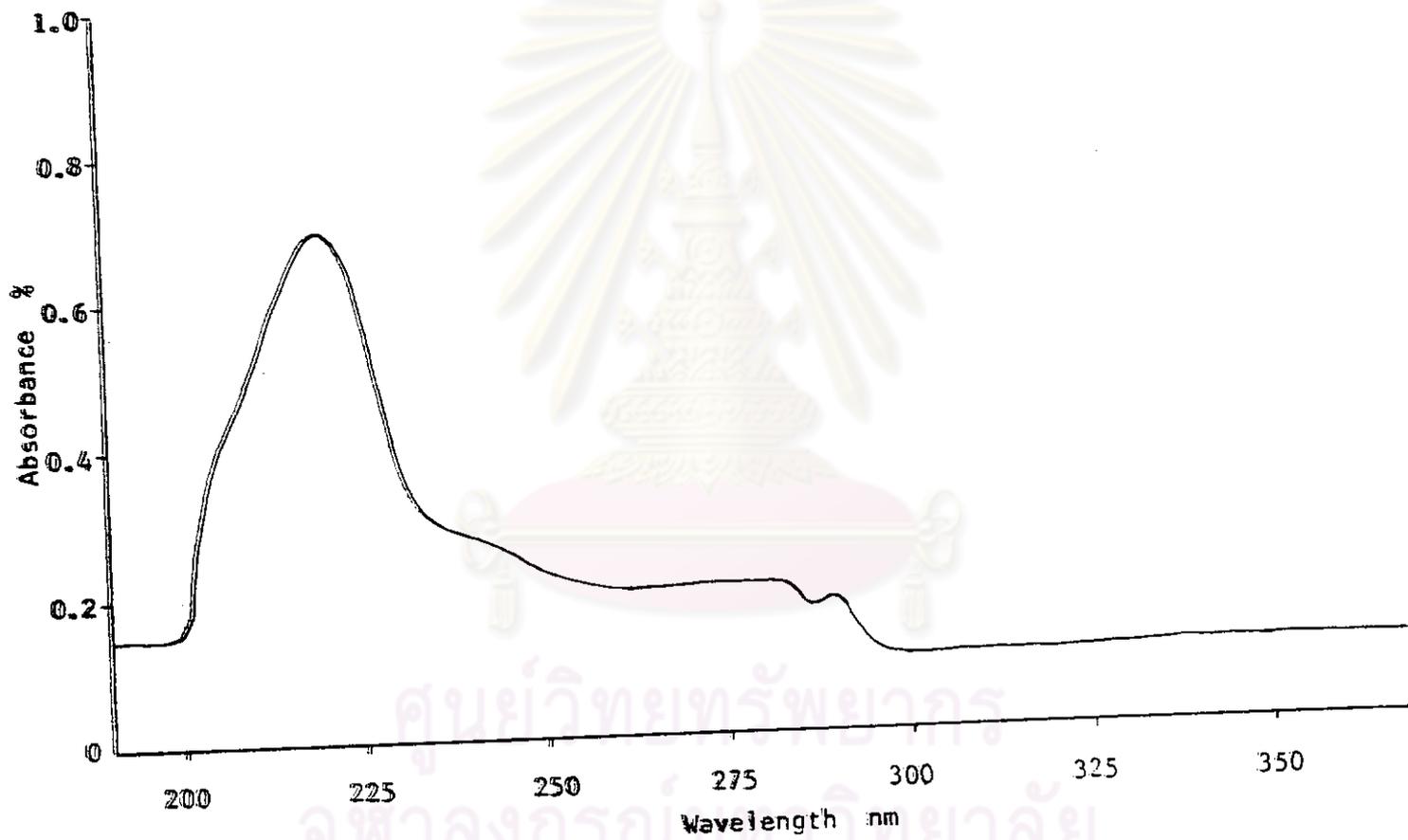


Fig. XXXVIII Ultraviolet absorption spectrum of alkaloid  $TS_1$  in methyl alcohol.

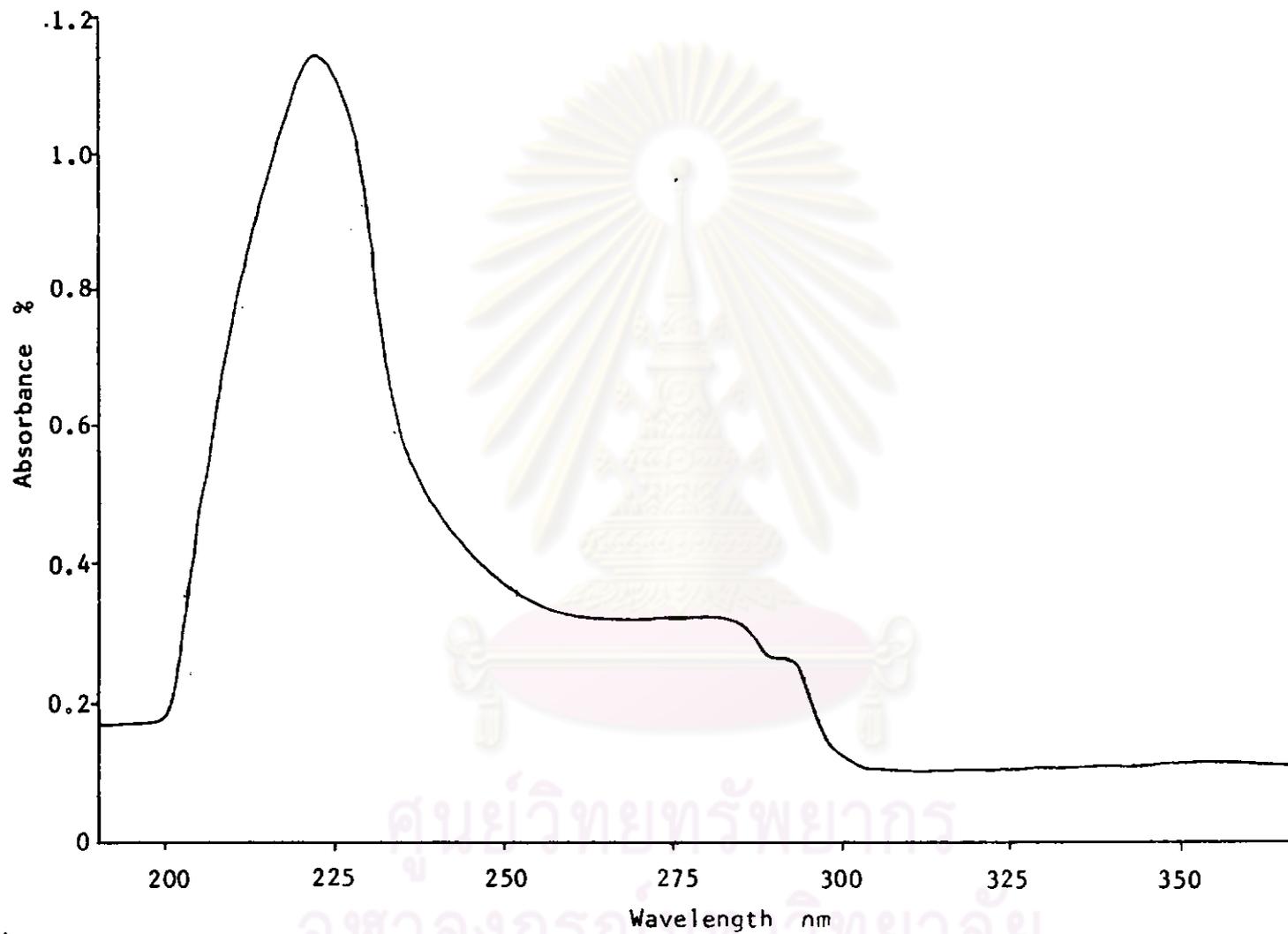


Fig. XXXIX Ultraviolet absorption spectrum of alkaloid TS<sub>2</sub> in methyl alcohol.

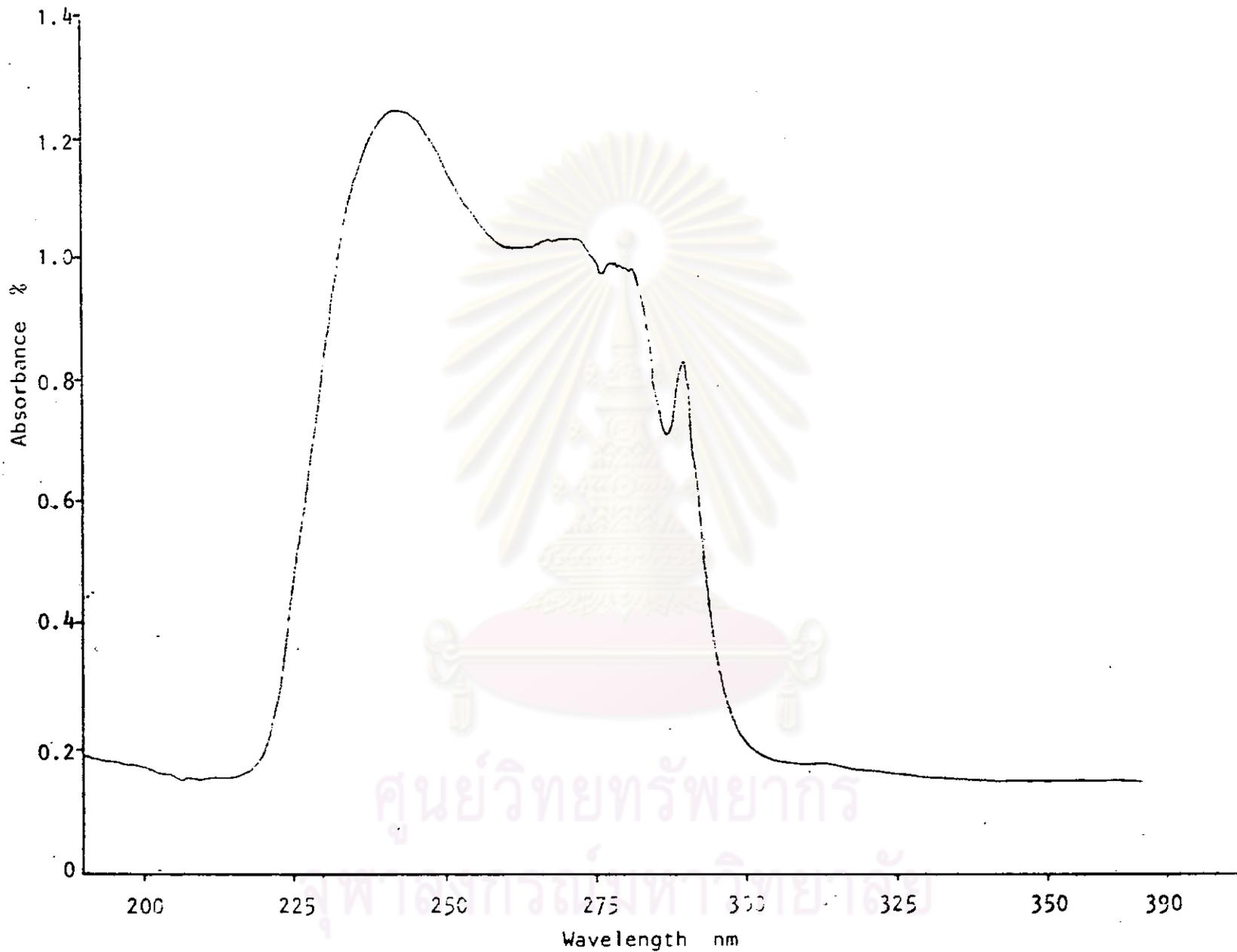


Fig. XL Ultraviolet absorption spectrum of alkaloid  $TS_3$  in methyl alcohol.

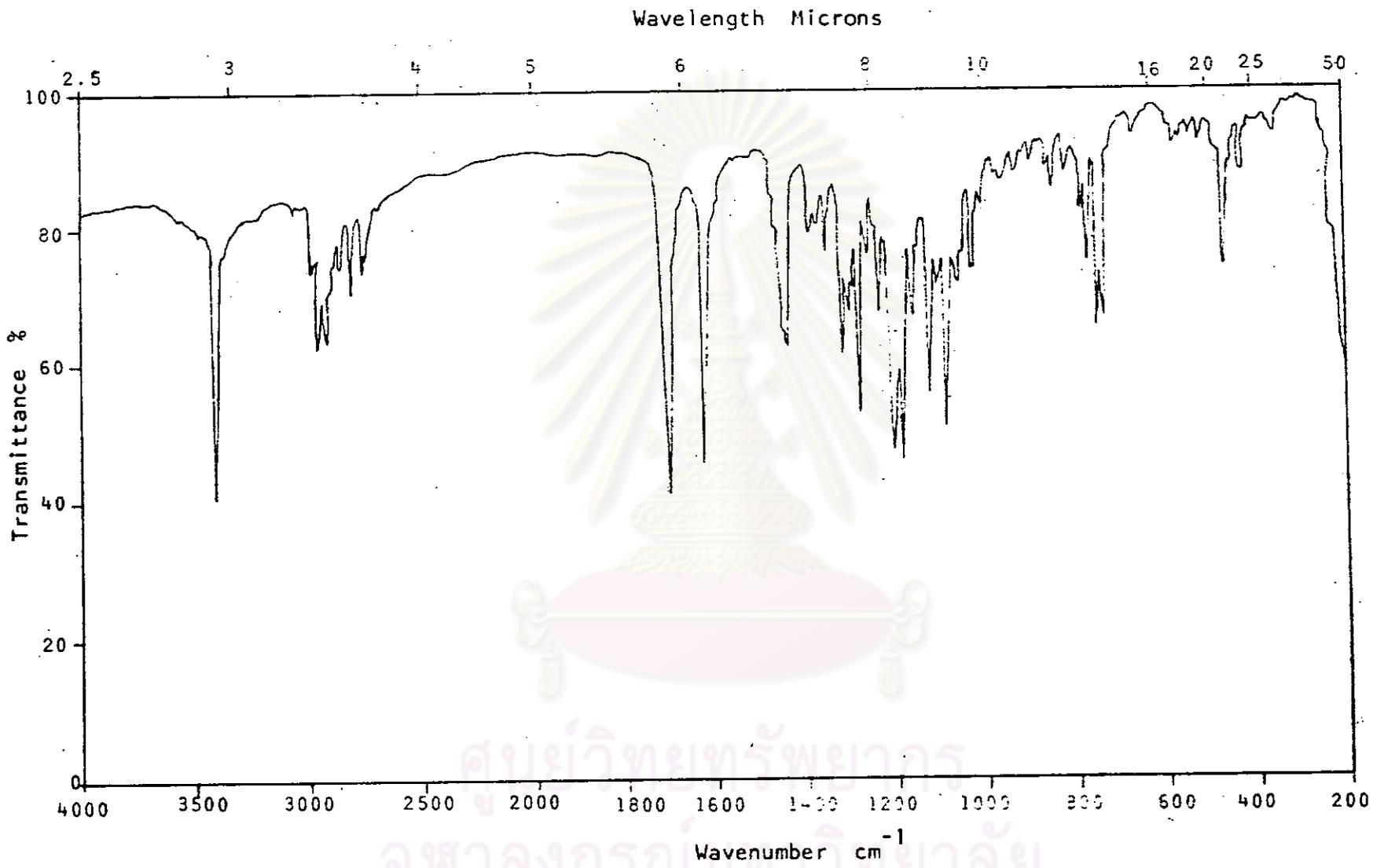


Fig. XLI Infrared absorption spectrum of alkaloid  $\text{TS}_1$  in potassium bromide disc.

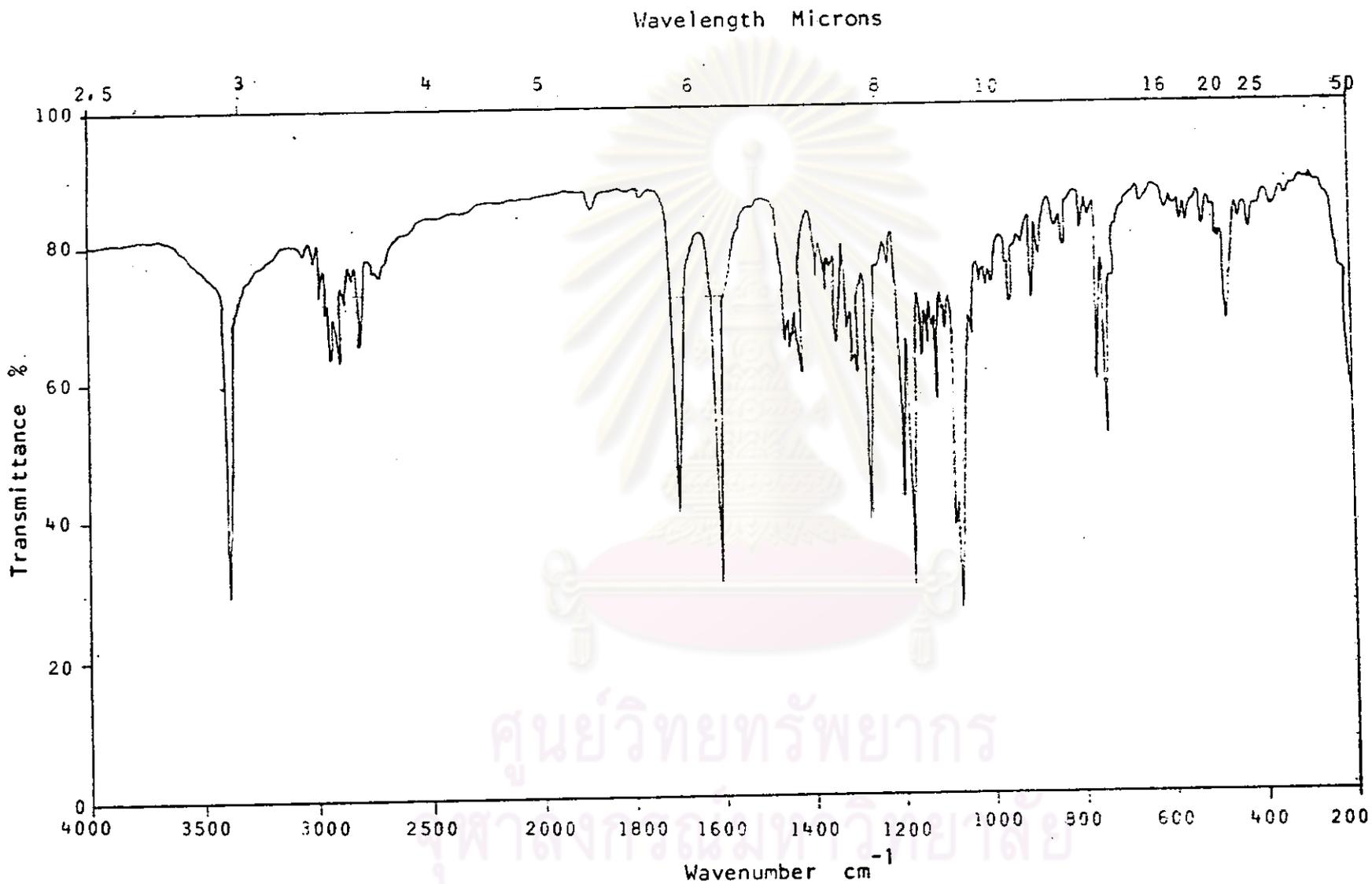


Fig. XLIII Infrared absorption spectrum of alkaloid  $\text{TS}_2$  in potassium bromide disc.

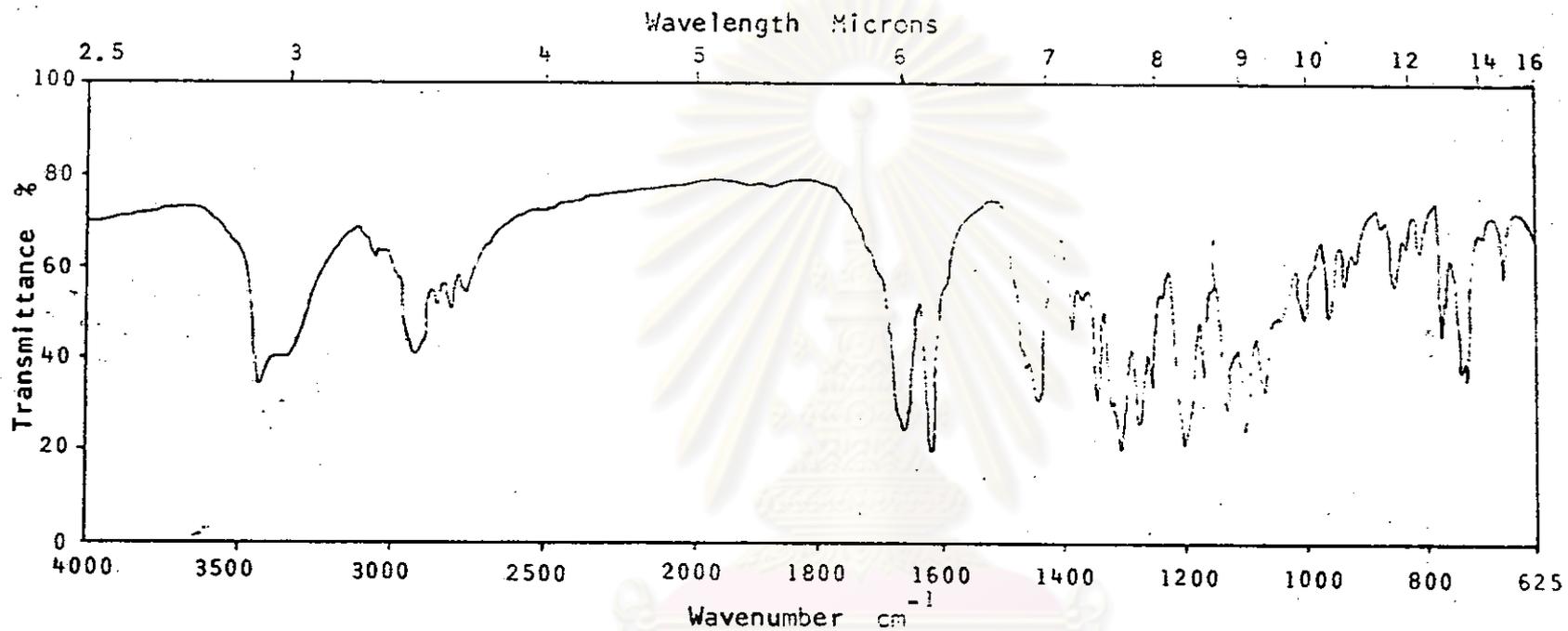


Fig. XLIII Infrared absorption spectrum of alkaloid TS<sub>3</sub> in potassium bromide disc.

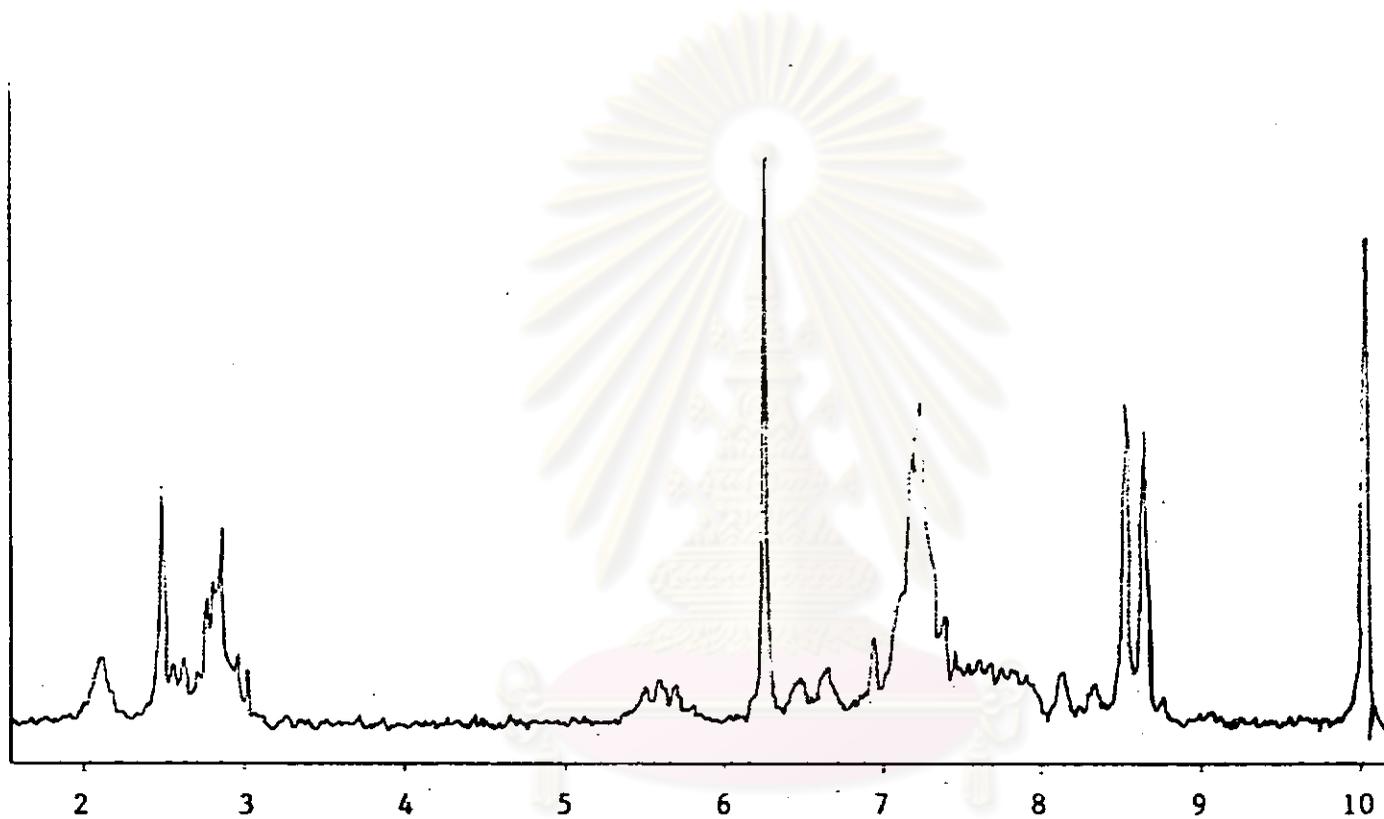


Fig. XLIV Nuclear magnetic resonance spectrum ( $\text{CDCl}_3$ ) of alkaloid  $\text{TS}_2$ .

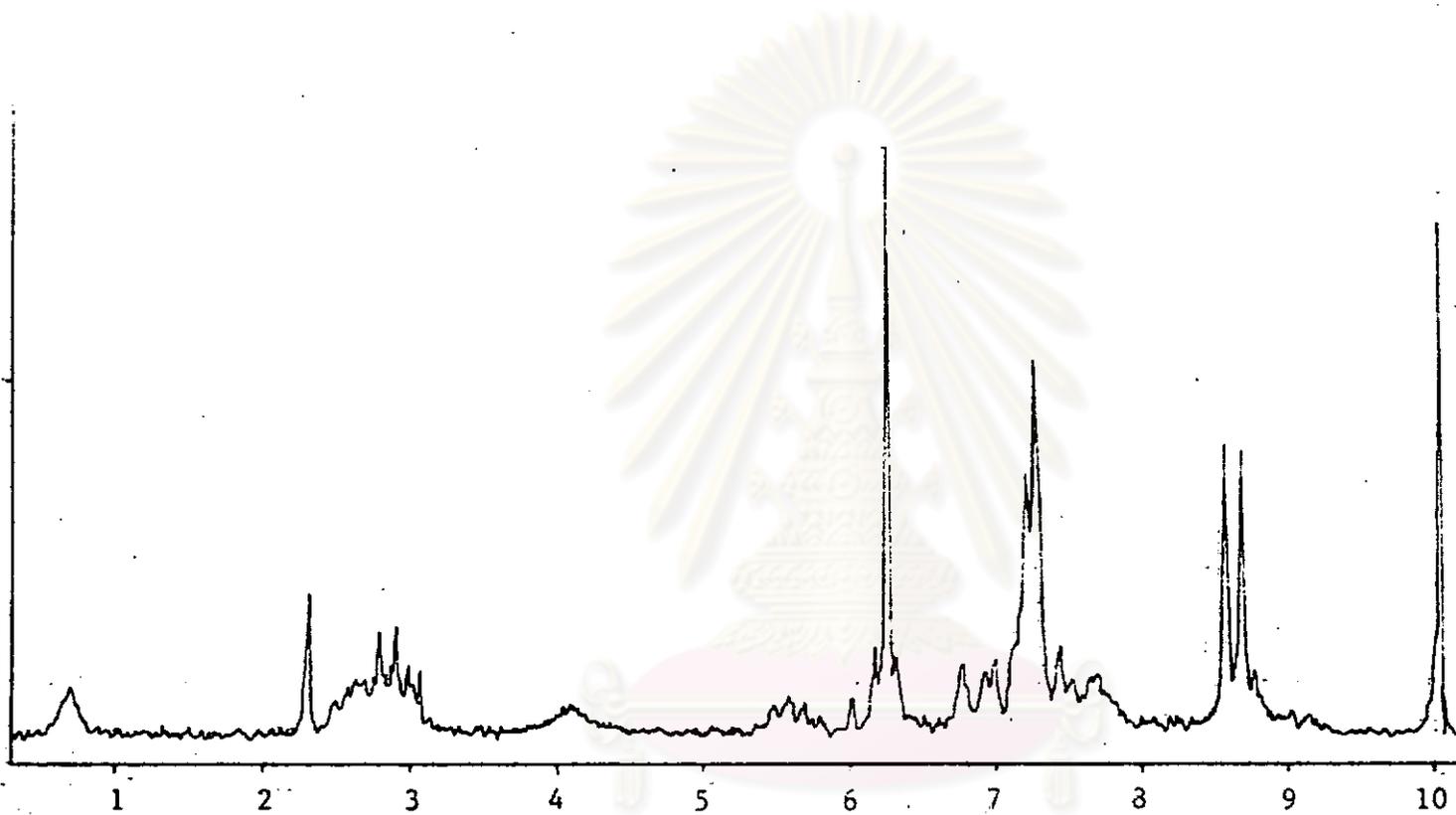


Fig. XLV Nuclear magnetic resonance spectrum ( $\text{CDCl}_3$ ) of alkaloid  $\text{TS}_3$ .

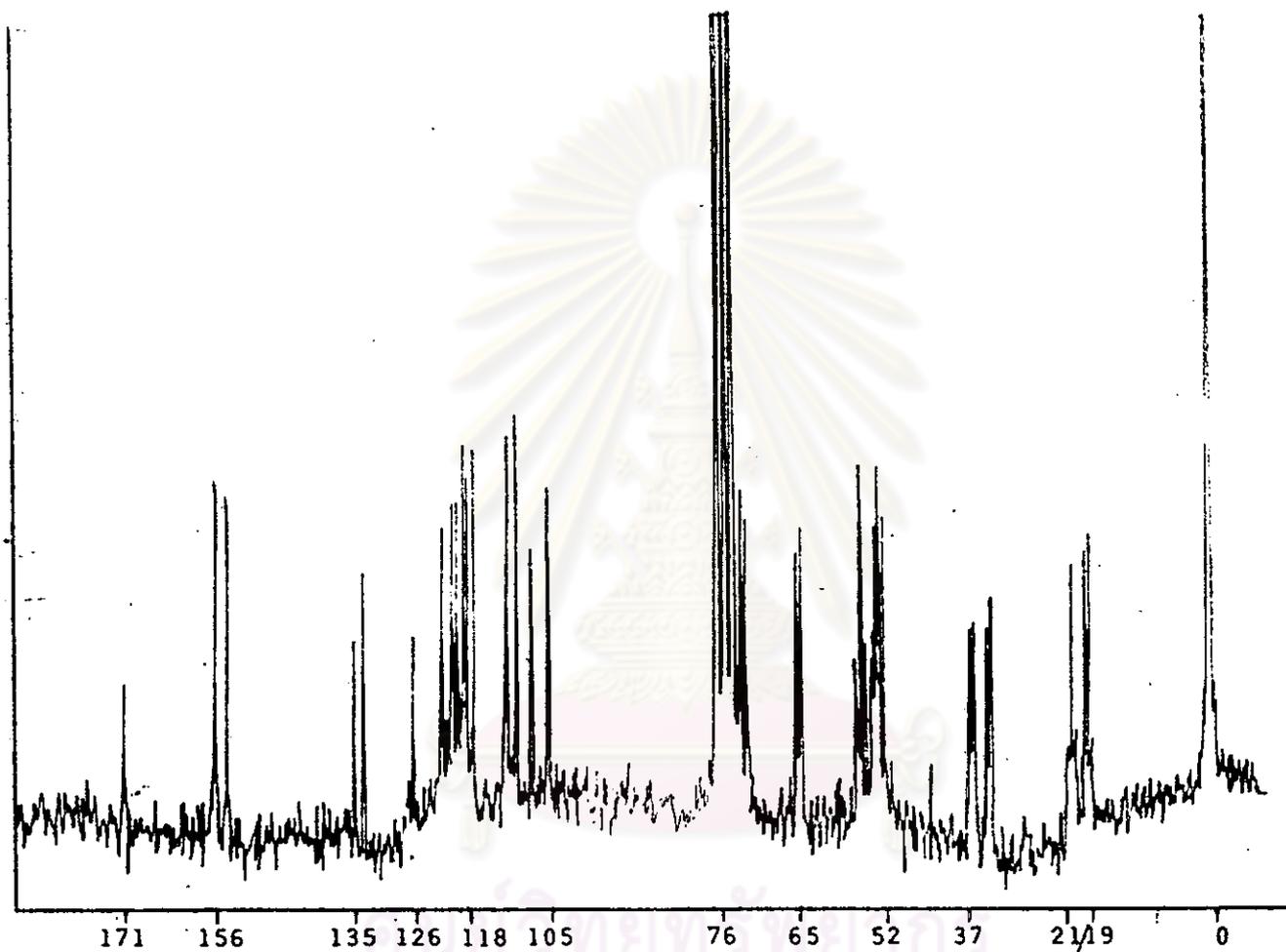


Fig. XLVI  $^{13}\text{C}$  Nuclear magnetic resonance spectrum ( $\text{CDCl}_3$ ) of alkaloid  $\text{TS}_3$ .

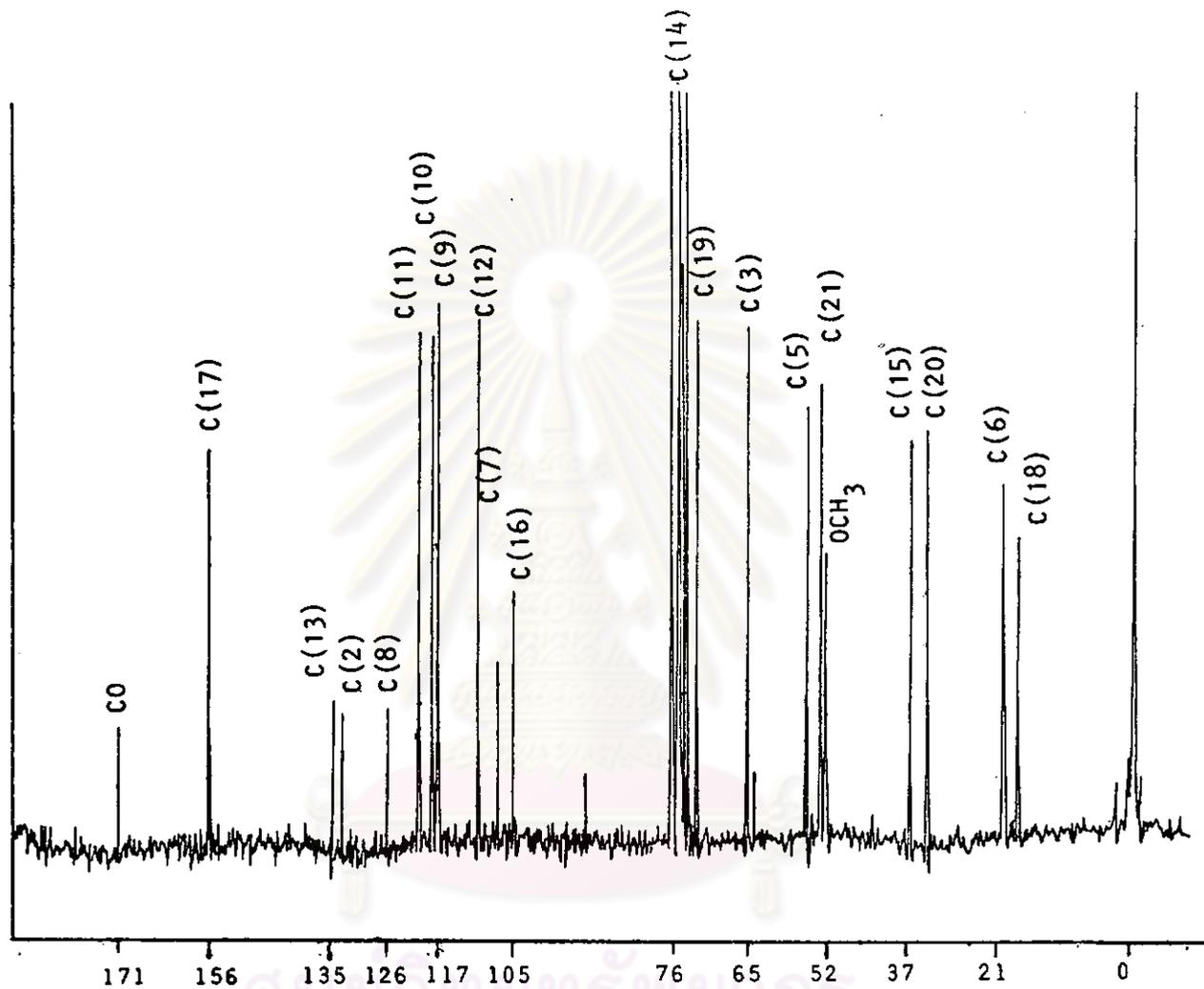


Fig. XLVII  $^{13}\text{C}$  Nuclear magnetic resonance spectrum ( $\text{CDCl}_3$ ) of alkaloid  $\text{TS}_3$ ,  
decoupled for clear single peaks.

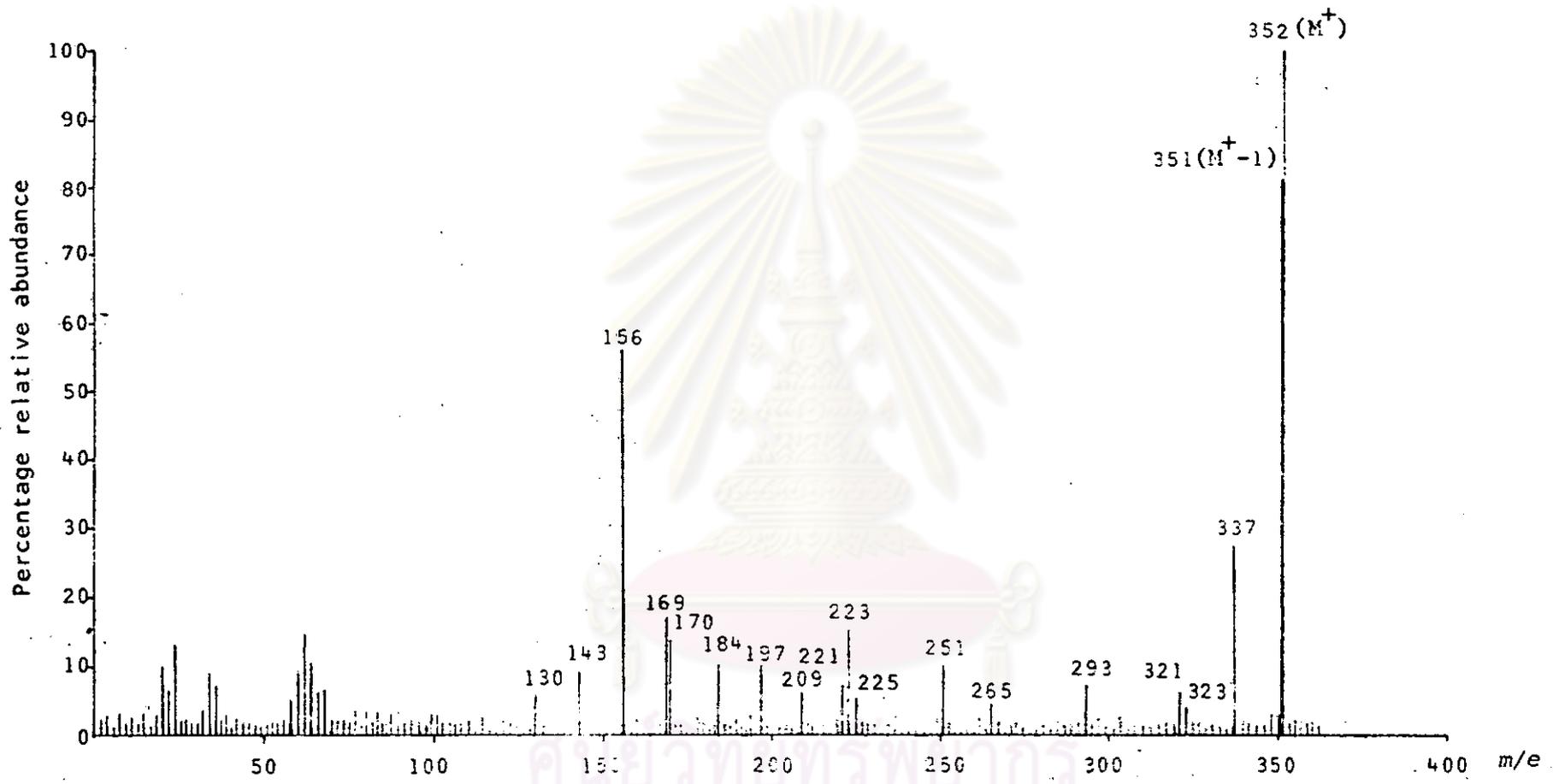


Fig. XLVIII Mass spectrum of alkaloid TS<sub>1</sub>.

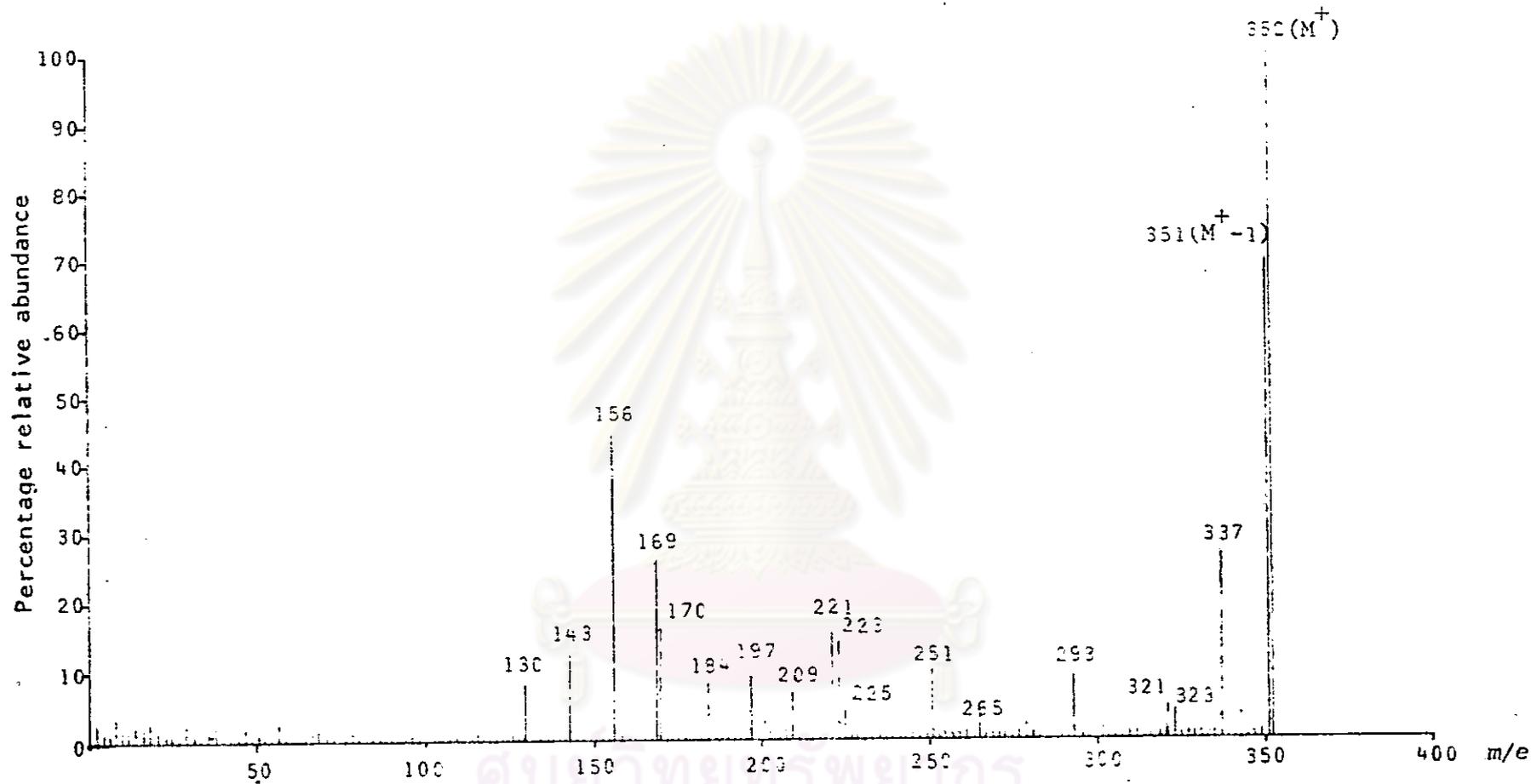


Fig. XLIX Mass spectrum of alkaloid TS<sub>2</sub>.

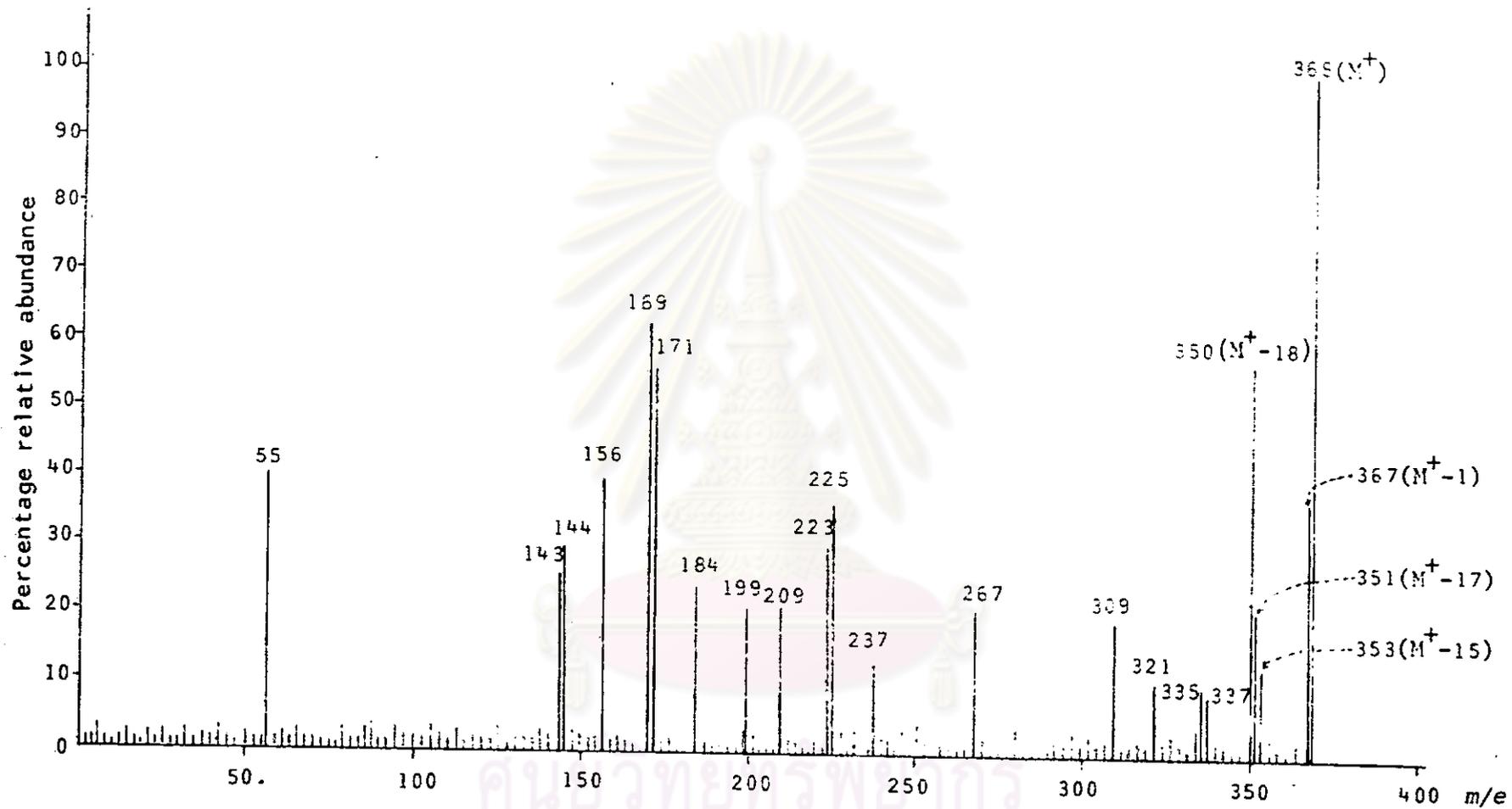
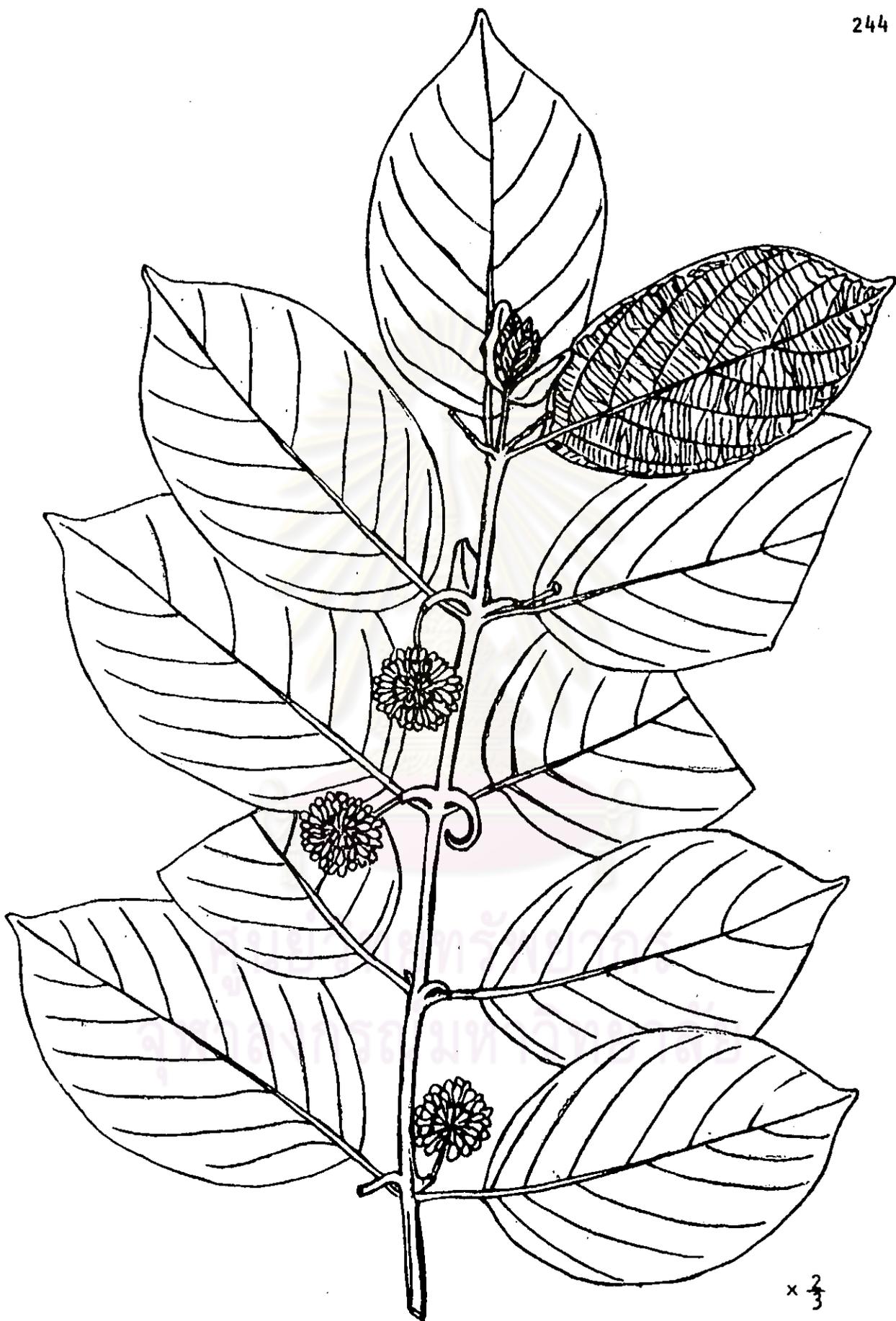


Fig. L Mass spectrum of alkaloid TS<sub>3</sub>.

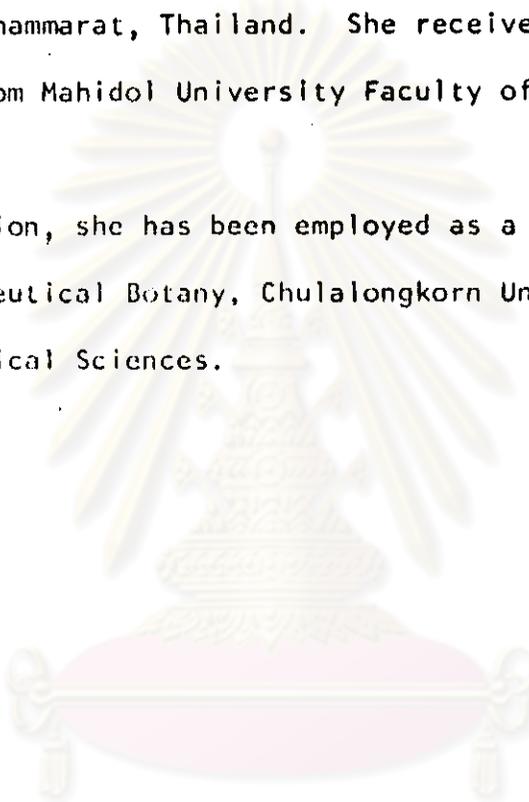


Ngōb (ໄຈຸ) *Uncaria attenuata* Korth. (RUBIACEAE)

## VITA

Mrs. Tanomjit Supavita was born on December 26, 1948 at Pakpanang, Nakornsrihammarat, Thailand. She received the degree of B.Sc. in Pharm. from Mahidol University Faculty of Pharmacy in 1972.

Since graduation, she has been employed as a staff in the Department of Pharmaceutical Botany, Chulalongkorn University Faculty of Pharmaceutical Sciences.



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