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APPENDICES

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Appendix A

The calculation of the percentage of hydrogenation

The percentage of hydrogenation was calculated as follows:

(For example, from Figure B-11 in Appendix B)

1	A(5.2ppm)	B(1ppm)
2	6.57	682.1
3	m	n
4	63.611	0.09615
5	%Hydrogenation	
6	90.63849	

Where :

A2 = Olefinic proton (5.2 ppm); peak area

B2 = Total proton of saturated and unsaturated proton (0-2 ppm)

****A4 = Saturated proton****

$A4 = B2 - (7 \times A2) / 10$; Unsaturated proton/unit polyisoprene = 7

Saturated proton/unit polyisoprene = 10

****B4 = Unsaturated proton****

$B4 = A2 / (A2 + A4)$; [proton of unsaturate / (proton of unsaturate + proton of saturate)]

%Hydrogenation = $(1 - B4) \times 100$

APPENDIX B

¹H-NMR Spectra of Hydrogenated Natural Rubber

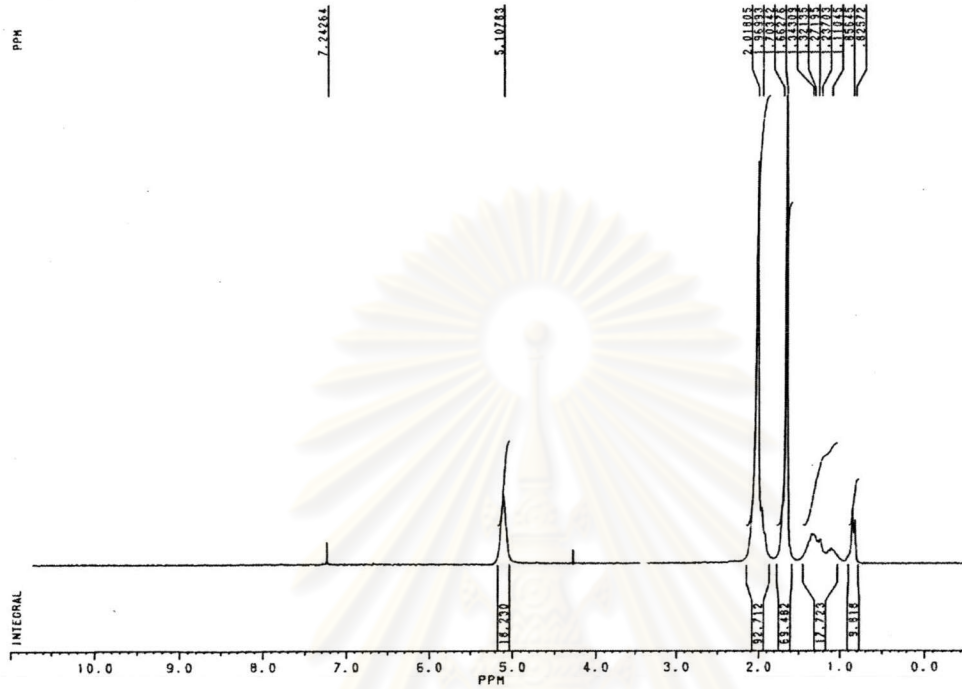


Figure B-1 ¹H-NMR Spectra of Hydrogenated Natural Rubber 22% (CDCl₃).

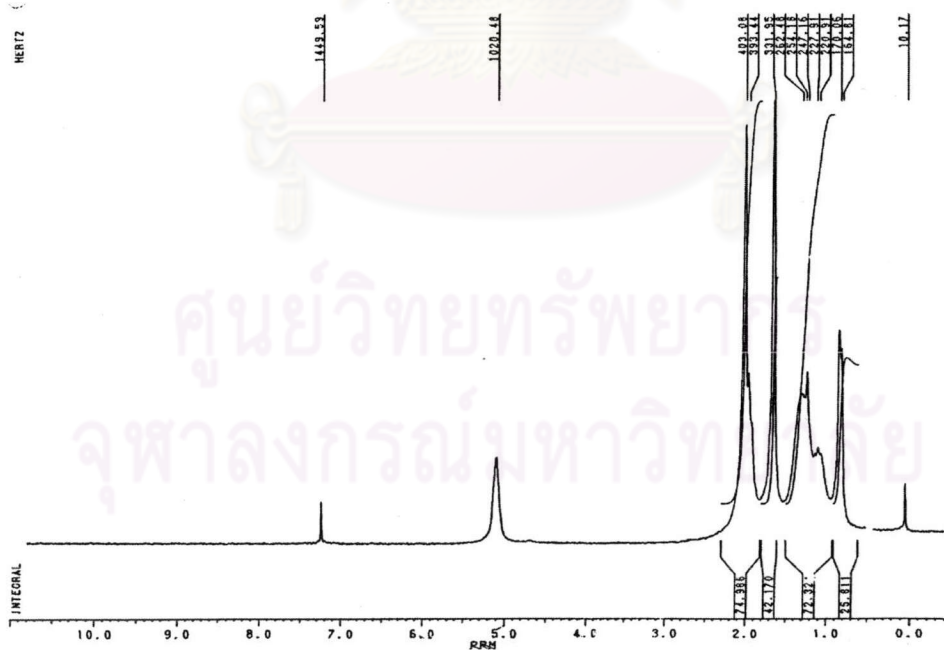


Figure B-2 ¹H-NMR Spectra of Hydrogenated Natural Rubber 28% (CDCl₃).

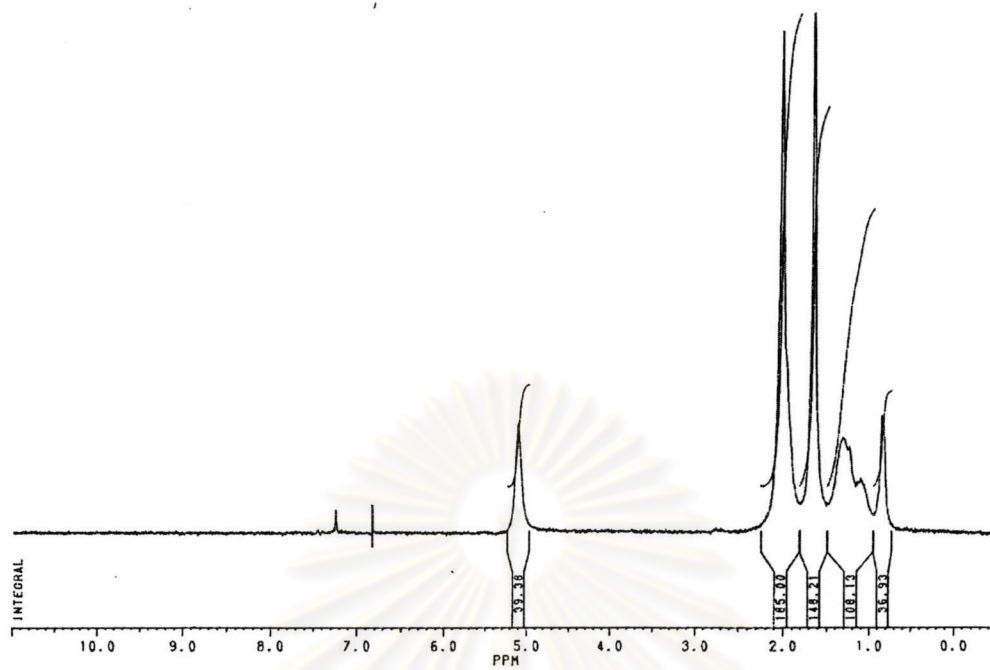


Figure B-3 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 31% (CDCl_3).

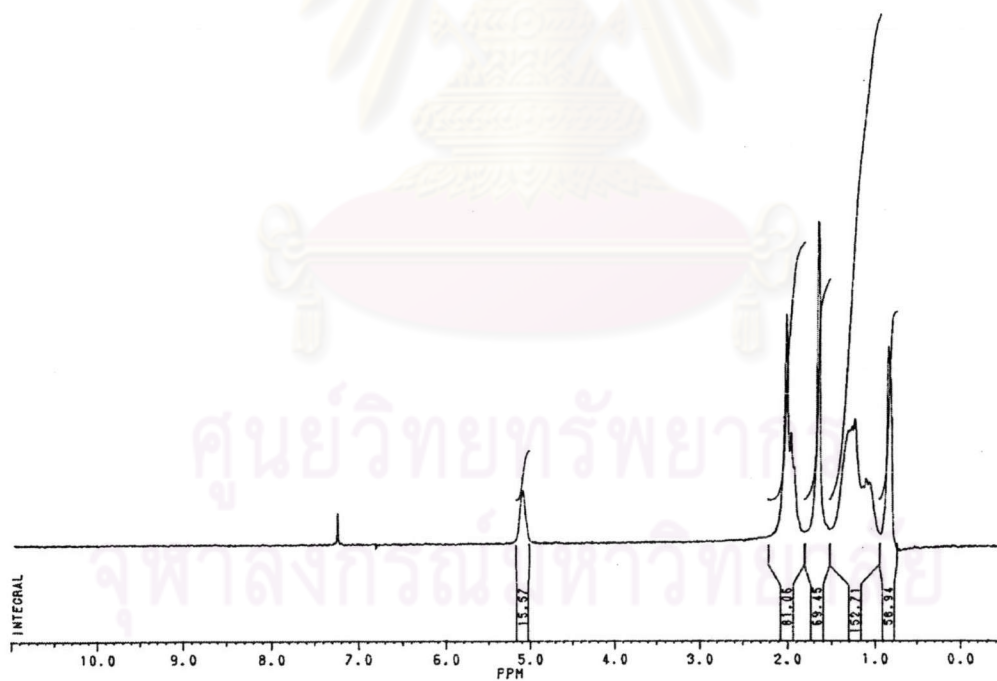


Figure B-4 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 38% (CDCl_3).

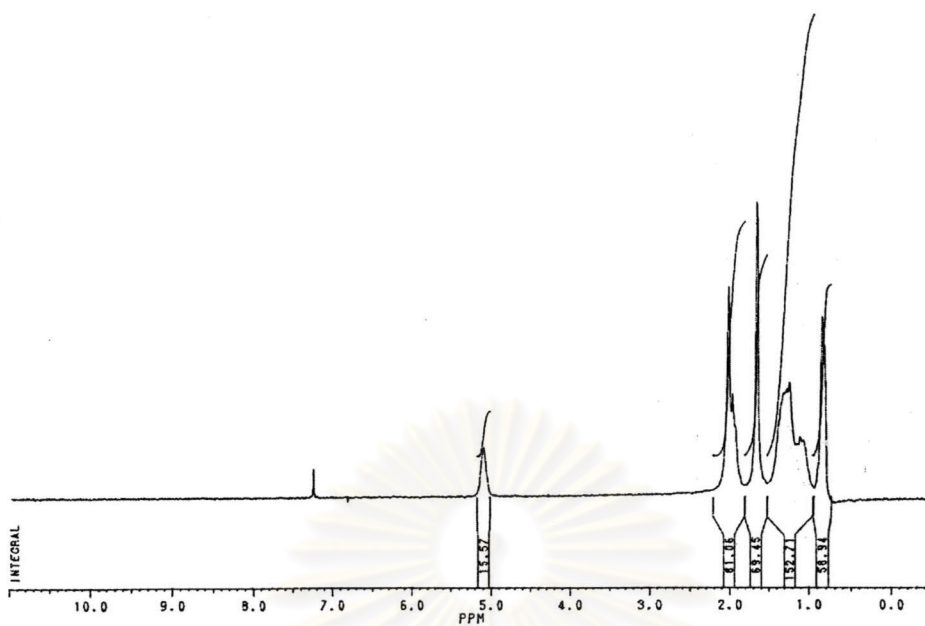


Figure B-5 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 55% (CDCl_3).

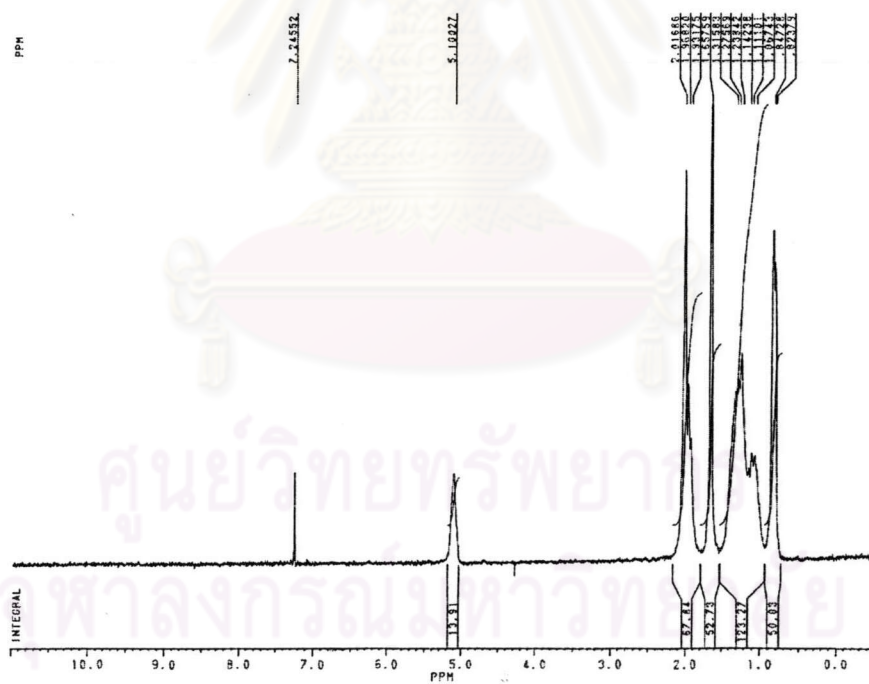


Figure B-6 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 59% (CDCl_3).

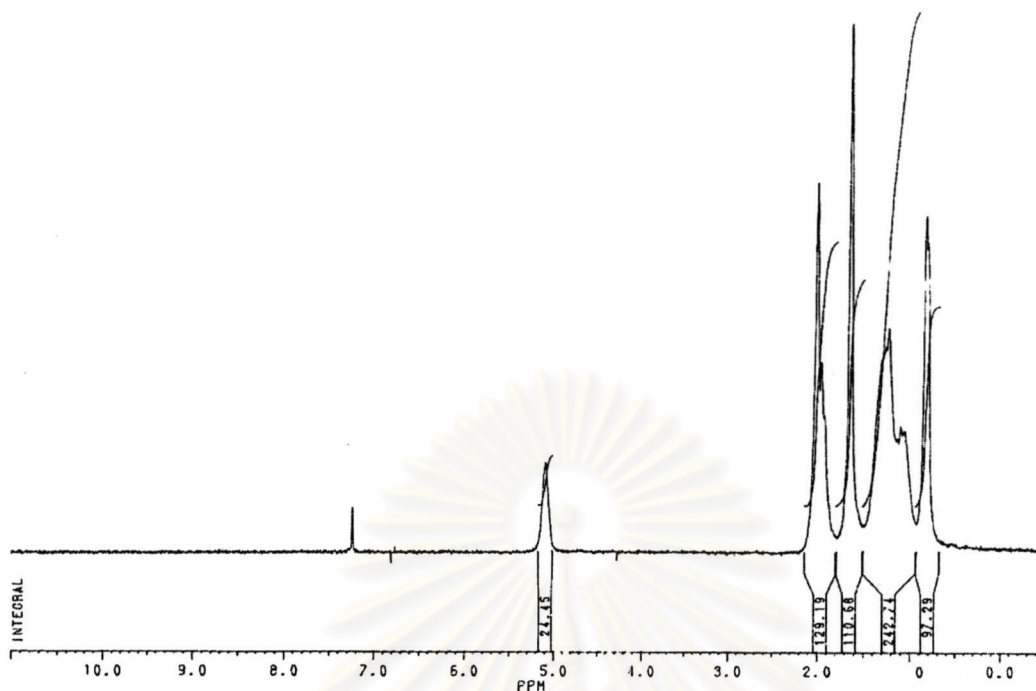


Figure B-7 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 63% (CDCl_3).

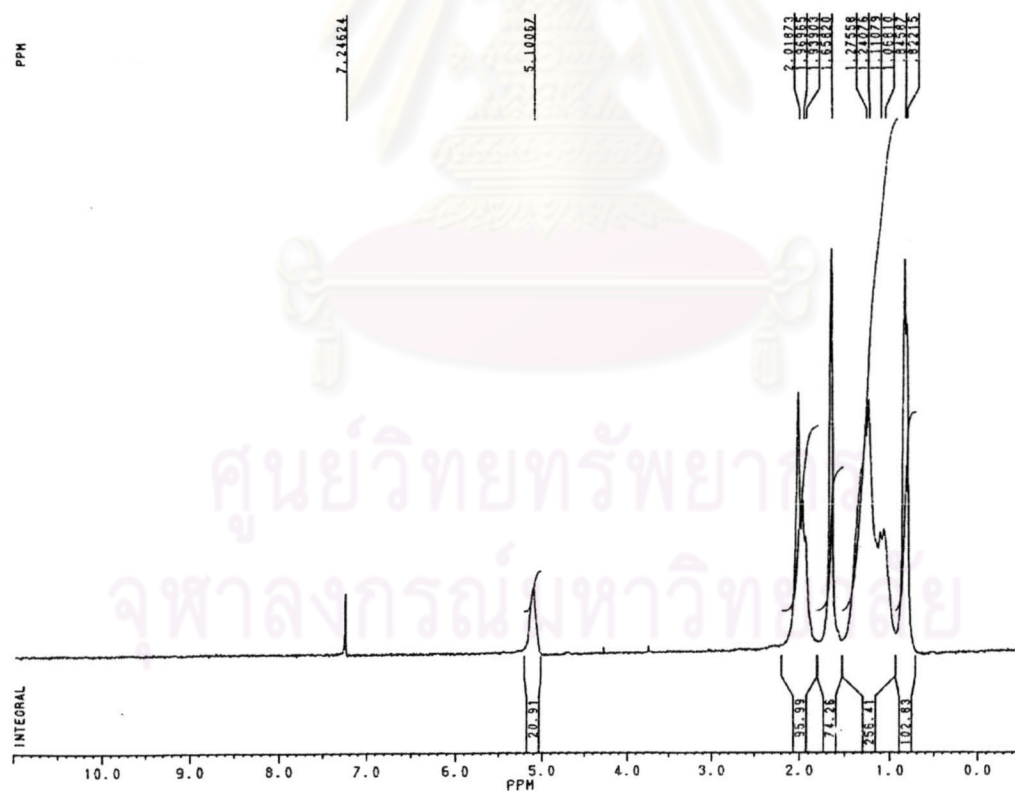


Figure B-8 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 68% (CDCl_3).

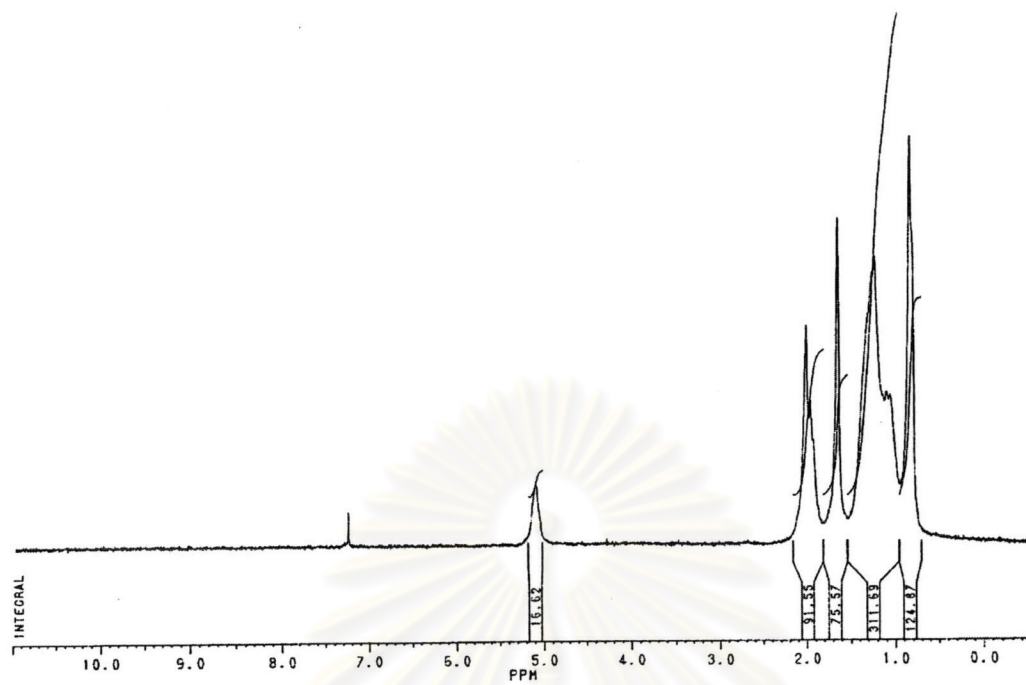


Figure B-9 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 75% (CDCl_3).

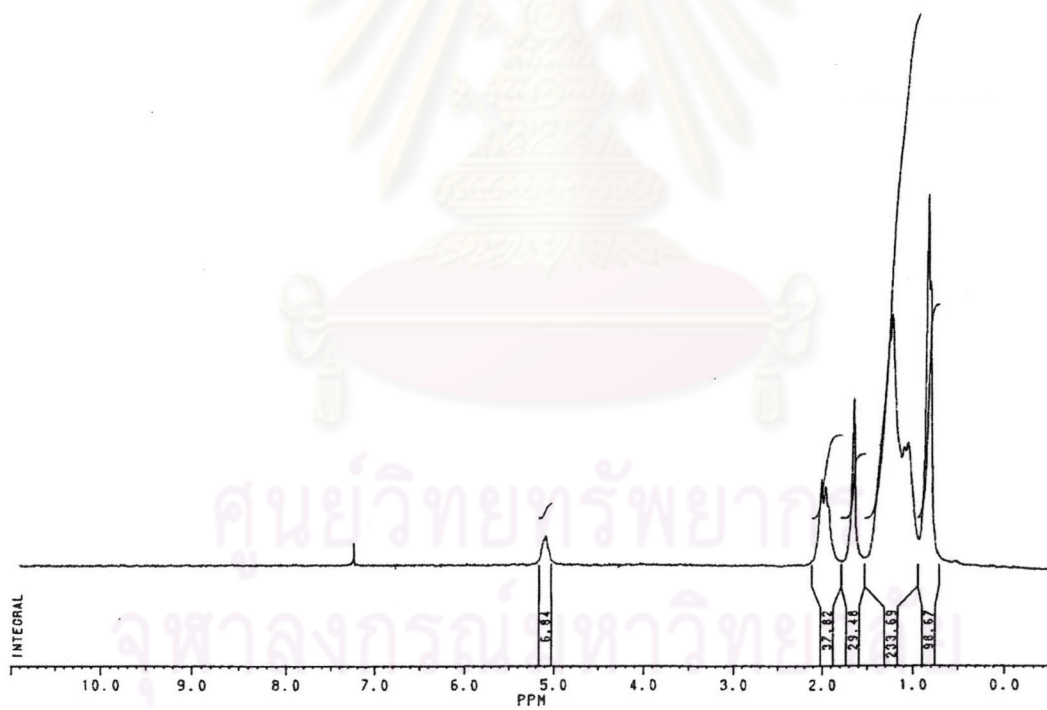


Figure B-10 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 84% (CDCl_3).

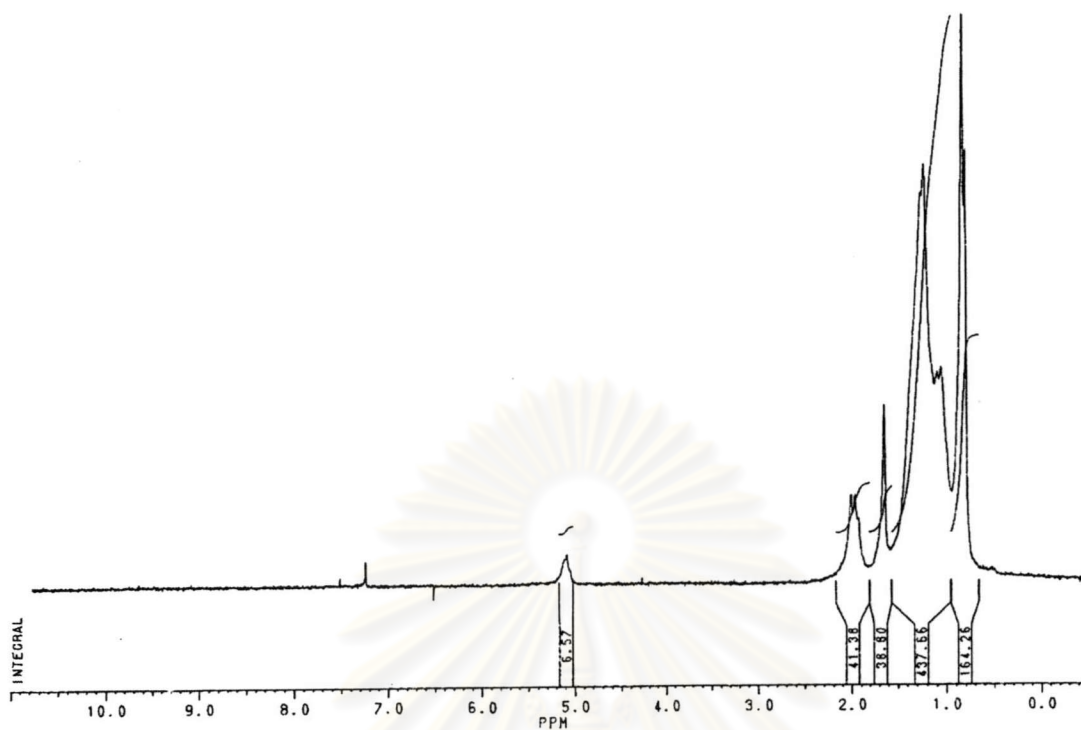


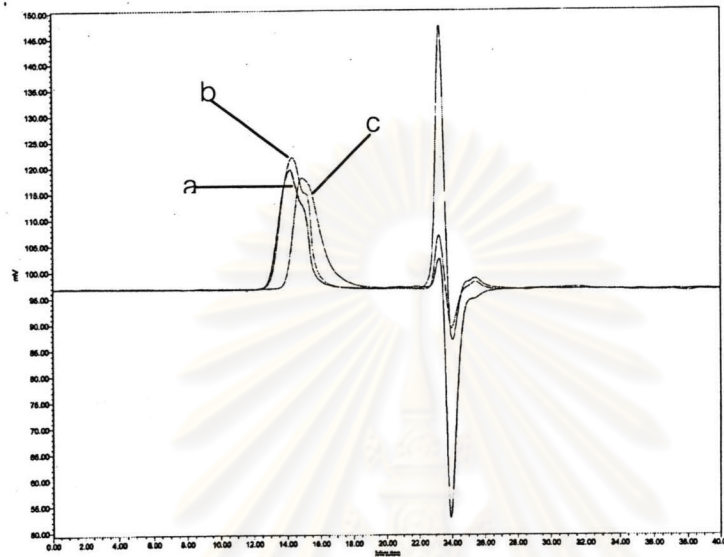
Figure B-11 $^1\text{H-NMR}$ Spectra of Hydrogenated Natural Rubber 91% (CDCl_3).

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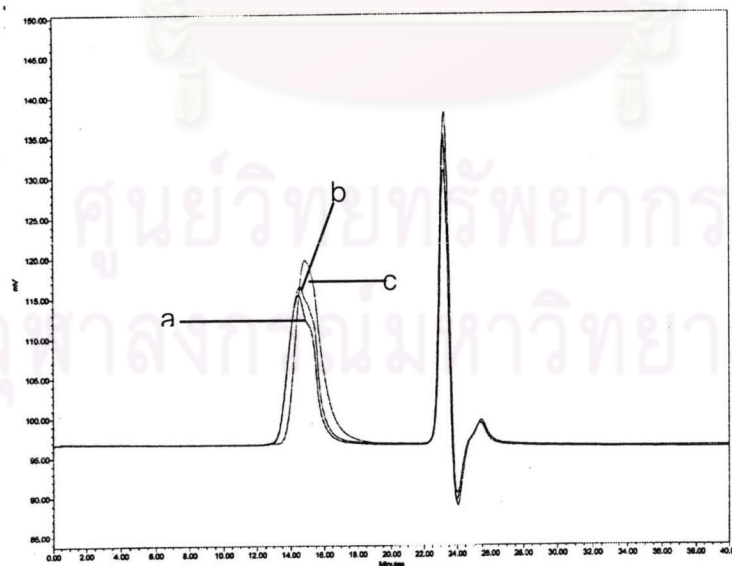
APPENDIX C

A plot of refractive index (RI) versus elution volume obtained from GPC

The Overlaid chromatogram of natural rubber and hydrogenated rubber



FigureC.1 The effect of catalyst on average molecular weight (a) Natural rubber (b) Non-catalyst (c) Hydrogenated natural rubber (Entry No. 4).



FigureC.2 The effect of hydrogen pressure on average molecular weight (a) P 30 bar (b) P 35 bar (c) P 40 bar.

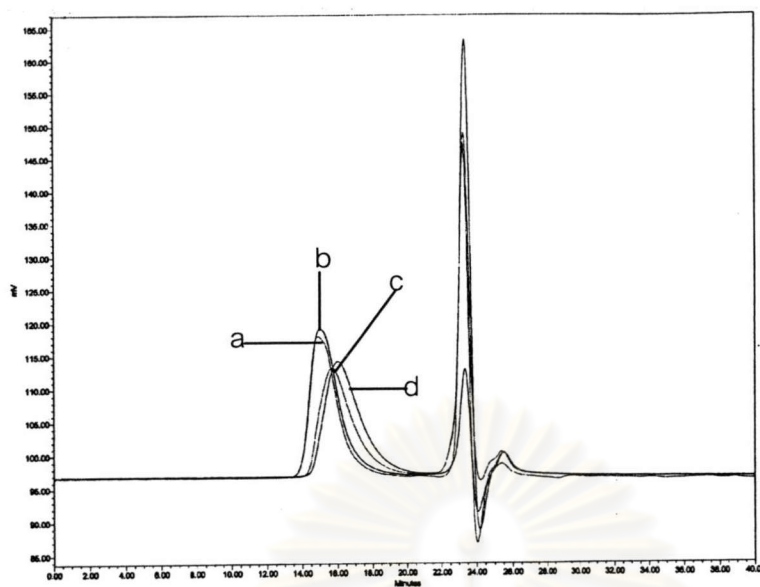


Figure C.3 The effect of temperature on average molecular weight (a) T 100 °C
(b) T 120 °C (c) T 130 °C (d) T 150 °C.

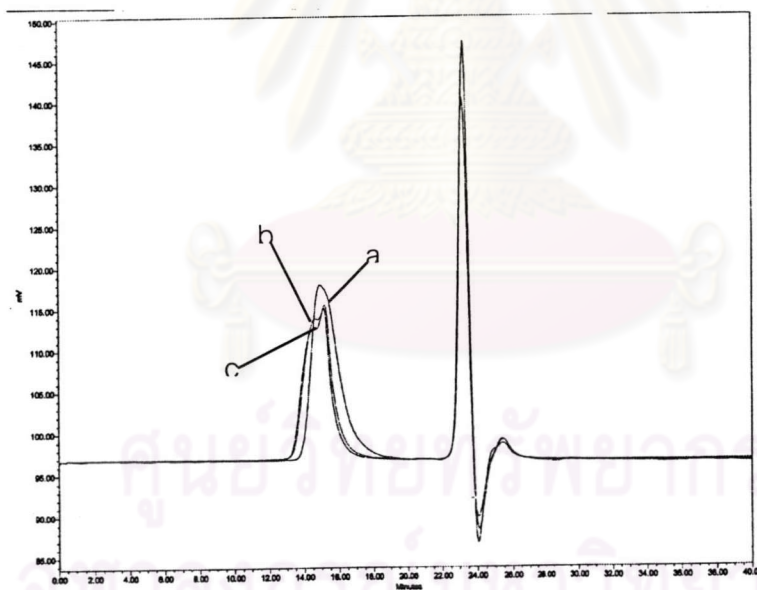


Figure C.4 The effect of catalyst amount on average molecular weight (a) catalyst 0.09 g
(b) catalyst 0.13 g (c) catalyst 0.15 g .



VITA

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