

Reference

1. Moore, E.S., ASTM Classification in Coal, 3 rd ed., pp. 131-132, John Willey & Sons, 1950.
2. Huges, H.W., in Text-Book of Coal-Mining, pp. 4-7.
3. Perey, Metallurgy (Fuel), & C., 1875, p. 208.
4. Rachadawong S., Fundamental Properties and Utilization Potential of Thailand Lignite, paper contributed to the 7th International Conference on Coal since, June 10-14, 1968, Prague, Czechoslovakia.
5. Ukkimapant Y., Coal, pp. 3, in the Progress of Surveying and Producing in Thailand, paper contributed to the 2nd Academic Conference, September 12-13, 1983.
6. Department of Mineral Resources Thailand, Lignite-Thai Coal, June, 1981.
7. Poovatananuwong P., Master Thesis, Chemical Engineering, Faculty of Engineering, Chulalongkorn University, 1978.
8. Friederick, F., "Fluidized Bed Combustion and Emerging Technology", Canmet Report 79-39, October 1979, pp. 1-3.
9. Glicksman, L., Lord, W., Valenzuela, J., Bar-cohen, A., and Hughes, R., "A Model of the Fluid Mechanics in Fluidized Bed Combustors", AICHe Symp. Series 205, volume 77, pp. 139, 1981.
10. Baron, R.E., Hodges, J.L., Sarofim, A.F. "Mathematical Model for Predicting Efficiency of Fluidized Bed Steam Generators", AICHe Symp. Series 75 (176), 1979 : pp. 121-122.
11. Levenspiel, O., Kunii, D., Fitzgerald, T., Power Technology, in Press.

12. Levenspiel, O., Kunii, D., in Fluidization Engineering, pp. 336-342, John Willey & Sons, Inc., 1968.
13. Guha, S.K., Kanae, A., and Gupta, P.S., Can. J. Chem. Eng., 50, 602, 1972.
14. Botterill, J.S.M., Fluid-Bed-Heat Transfer, Academic Press, New York, 1975.
15. Wen, C.Y. and Hashinges, AIChE. J., 6, 220, (1960).
16. Merrick D., and Highley J., "Particle Size Reduction and Elutriation in a Fluidized Bed Process", AIChE Symp. Series No. 137, Vol. 70, 1974 : pp. 366-378.
17. Colakyan M., Catipovic N., Jovanovic G., and Fitzgerald T., "Elutriation from a Large Particle Fluidized Bed with and without Immersed Heat Transfer Tubes", AIChE Symp. Series 225, Vol. 77, 1981 : pp. 237.
18. Avadesian M.M., and Davidson J.F., "Combustion of Carbon Particles in a Fluidized Bed", Trans. Instn. Chem, Engrs. Vol. 51, 1973 : pp. 121-131.
19. Tield, M.A. Gill, D.W., Morgan, B.B. and Hawksley, P.G.W., "Combustion of Pulverized Coal", 1967. (Leatherhead : The British Coal Utilization Research Association)
20. McCabe, W.L., and Smith, J.C., "Coefficient for Mass transfer through Known Area", Unit Operation in Chemical Engineering, 2nd. ed., McGraw-Hill, pp. 631-632.
21. Davidson J.F., and Harrison D., in Fluidization, p. 397-398, Academic Press, 1971.
22. Levenspiel, O., Kunii, D., in Fluidization Engineering, pp. 180-181.

23. Mori S., Wen C.Y., "Estimation of Bubble Diameter in Gaseous Fluidized Beds", AICHE.J. No.1, Vol. 21, 1975 : pp. 109-115.
24. Cranfield, R., R., Geldart, D., "Large Particle Fluidization", Chem. Eng. Sci., 1974 : pp. 935-947.
25. Davidson J.F., and Harrison D., in Fluidization, pp. 630, Academic Press, 1971.
26. Pata, J., and Hartman M., "Minimum Fluidization Velocities of Lime and Limestone Particles", Ind. Eng. Chem. Progress. De., Vol. 17, No. 3, 1978 : pp. 231-236.
27. Ergun, S., Chemical Engineering Progress, 48, 89, 1952.
28. The short note of "Minimum Fluidization Velocity of a Binary Mixture of Different Sized Particles", in Chem. Eng. Sci., Vol. 29, 1974 : pp. 1301-1303.
29. Levenspiel O., Kunii, D., in Fluidization Engineering, pp. 64-79, John-Willey & Sons, Inc. 1968.
30. Milles G., Kolar A., Zakkay V., and Hakim S., "Bed Expansion Studies and Slugging characteristics in a Pressurized Fluidized Bed of Large Particle", AICHE Symp. Series, 205, Vol. 77, 1981 : pp. 166-173.
31. Geldart, D., Power Technology 7, 285, 1973.
32. Geldart, D., and Cranfield RR., Chem. Eng. J., 3, 211, 1972.
33. Catipovic, N.M., Jananovic G.N., and Fitzgerald J.J., Am. Inst. Chem. Eng. J. 24, 243, 1978.
34. Horio M., Wen C.Y., "An Assesment of Fluidized-bed Modelling" AICHE Symp., No. 161, Vol. 73, pp. 9-21.

35. Adel F., Sarofim and Beer, J.M., "Modelling of Fluidized Bed Combustion", Paper for Presentation at Seventeenth Symposium (International) on Combustion, Leeds, England, August 1978.
36. Tanaku Isama, "A Model for Fluidized Bed Combustion of Carbon Particles", Memoirs of the Faculty of Engineering, Kyushu University, Vol. 34, No. 2, June 1979 : pp. 53-60.
37. Perry R.H., Chilton C.H., Chemical Engineering Handbook, Fifth ed., pp. 3-211, International Student Edition.
38. Sherwood, K.T., Pigford R.L., and Wilke C.R., ed., Mass Transfer, pp. 18-24, International Student Edition, 1975.
39. Vanichseni S., Prasertdham P., Chongvisanr V., "Research and Development of the Energy from Oil shale for Electricity Generation," a Progress Report, July-December 1981.

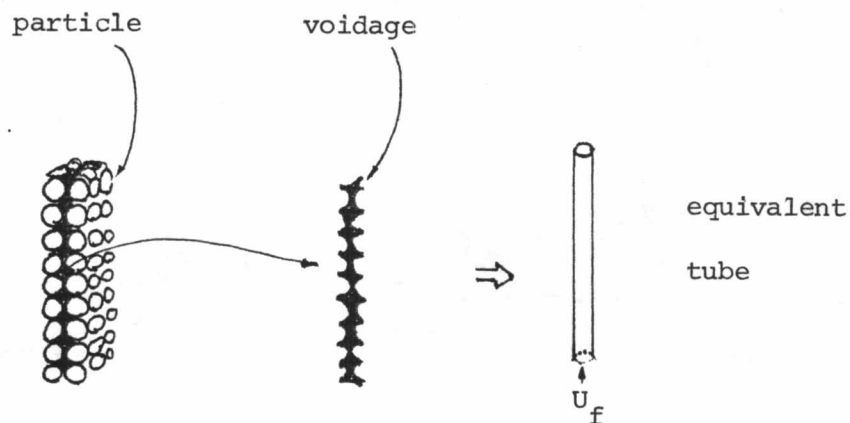
APPENDICES

Apeendix A

Minimum Fluidization Velocity (U_{mf})

The Ergun equation derivation for predicting minimum fluidization velocity (U_{mf}) is shown followingly.

In fixed bed, the particles of the same size are assumed. The small tubes are assumed to be the equivalent tubes for the voidages between particles. The surface area of the equivalent tube is assumed to be equal to the surface area of the particles.



Let the equivalent tube diameter be d_c and U_f is the air velocity passing through the tube.

From the Bernoulli equation

$$\frac{\Delta P}{\rho_g} + \Delta z \frac{g}{g_c} + \frac{\Delta(U_f^{-2})}{2g_c} + \Sigma F = W_f \quad \dots\dots (1)$$

where ΔP = pressure drop across the tube

ρ_g = density of gas

F = force

W_f = work

$$\frac{\Delta P}{\rho_g} + 0 + 0 + F = 0$$

$$\frac{\Delta P}{\rho_g} = F_{\text{friction}} + F_{\text{kinetic}} \quad \dots\dots (2)$$

where F_{friction} = friction force

F_{kinetic} = kinetic force

$$F_{\text{friction}} = f \frac{L}{d_c} \frac{\bar{U}_f^2}{2g_c} \quad \dots\dots (3)$$

$$F_{\text{kinetic}} = \frac{k \bar{u}_f^2}{2g_c} \quad \dots\dots (4)$$

where f, k = constant

$$\frac{\Delta P}{\rho_g} = f \frac{L}{d_c} \frac{U_f^2}{2g_c} + \frac{K U_f^2}{2g_c} \quad \dots\dots (5)$$

$$d_c = 2r = 2r \cdot \frac{2\pi r}{2\pi r} = \frac{4\pi r^2}{2\pi r} \cdot \frac{L}{L} \cdot \frac{N_c}{N_c} \quad \dots\dots (6)$$

where r = radius of the tube
 d_c = tube diameter
 N_c = numbers of tube
 L = length of tube

hence
$$d_c = \frac{4 \cdot \pi r^3 L N_c}{2 \pi r L N_c} = 4 \frac{\text{total void volume}}{\text{total surface of particles}}$$

$$d_c = \frac{4 \cdot N \cdot \frac{4}{3} \pi \frac{d_p^3}{8} \cdot \frac{\epsilon_m}{1-\epsilon_m}}{N \cdot \frac{4 \pi d_p^2}{4}} = \frac{2}{3} \cdot \frac{\epsilon_m}{1-\epsilon_m} d_p \dots\dots (7)$$

where N = numbers of particles
 d_p = particle diameter
 ϵ_m = void fraction in fixed bed

substitute Eq. (7) in Eq. (5)

$$-\frac{\Delta P}{\rho_g} = f \cdot \frac{3}{2} \frac{(1-\epsilon_m)}{\epsilon_m} \frac{L \bar{U}_f^2}{d_p} + \frac{K \bar{U}_f^2}{2g_c} \dots\dots (8)$$

from $U_o A_t = \bar{U}_f \epsilon_m A_t \dots\dots (9)$

A_t = bed cross section area

U_o = superficial velocity

$$K \propto \frac{L}{d_c} \dots\dots\dots (10)$$

$$f = \frac{64}{Re} = \frac{64 \mu}{\rho_g \bar{U}_f d_c} \dots\dots\dots (11)$$

where Re = Reynolds number

substitute Eq. (9), (10) and (11) into Eq. (8)

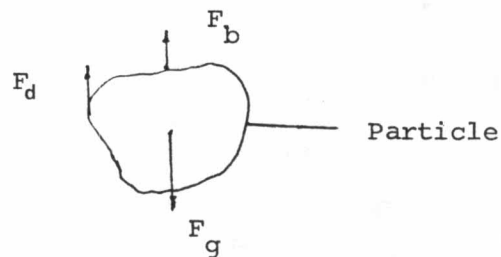
$$\left(-\frac{\Delta P}{L}\right) g_c = \frac{K_1 (1-\epsilon_m)^2 \mu U_o}{\epsilon_m^3 d_p^2} + K_2 \frac{(1-\epsilon_m) \rho_g U_o^2}{d_p \epsilon_m^3} \dots\dots\dots (12)$$

Ergun found that $K_1 = 150$, $K_2 = 1.75$

$$\therefore \left(-\frac{\Delta P}{L}\right) g_c = 150 \frac{(1-\epsilon_m)^2}{\epsilon_m^3} \cdot \frac{\mu U_o}{d_p^2} + \frac{1.75 (1-\epsilon_m) \rho_g U_o^2}{d_p \epsilon_m^3} \dots\dots\dots (13)$$

In a fluidized bed, a particle is floating and follows the equation

$$\Sigma F = 0 \dots\dots\dots (14)$$



$$F_b + F_d = F_g \quad \dots\dots (15)$$

where F_b = bouyancy force

F_d = drag force

F_g = gravitational force

$$F_g = (1 - \epsilon_{mf}) A_t L \rho_s \frac{g}{g_c} \quad \dots\dots (16)$$

$$F_b = \rho_g (1 - \epsilon_{mf}) A_t L \frac{g}{g_c} \quad \dots\dots (17)$$

$$F_d = \Delta P A_t \quad \dots\dots (18)$$

where ρ_s = solid density

ϵ_{mf} = minimum fluidization void fraction

substitute Eq. (16), (17) and (18) in Eq. (15)

$$(1 - \epsilon_{mf}) A_t L \rho_s \frac{g}{g_c} = \rho_g (1 - \epsilon_{mf}) A_t L \frac{g}{g_c} + (-\Delta P A_t)$$

$$\left(\frac{-\Delta P}{L} \right) g_c = (1 - \epsilon_{mf}) (\rho_s - \rho_g) g \quad \dots\dots (19)$$

From Eq. (13) and Eq. (19) and letting $\epsilon_m = \epsilon_{mf}$, $U_o = U_{mf}$

$$\text{hence } 150 \frac{(1-\epsilon_{mf})^2 \mu U_{mf}}{\epsilon_{mf}^3 (\varphi d_p)^2} + 1.75 \frac{(1-\epsilon_{mf}) \rho_g U_{mf}^2}{(\varphi d_p) \epsilon_{mf}^3} \dots\dots\dots (20)$$

$$= (1-\epsilon_{mf}) (\rho_s - \rho_g) g$$

$$\text{or } G_a = 150 \frac{1-\epsilon_{mf}}{\varphi^2 \epsilon_{mf}^3} Re_{mf} + 1.75 \frac{Re_{mf}^2}{\epsilon_{mf}^3} \dots\dots\dots (21)$$

where $Re_{mf} = \frac{U_{mf} d_p \rho_g}{\mu}$; particle Reynolds number at fluidization velocity

$G_a = g d_p^3 \rho_g \cdot (\rho_s - \rho_g) / \mu^2$; Galilei number

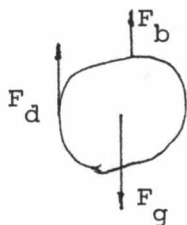
φ = sphericity of particle

Appendix B

Terminal Velocity of Particles (U_t)

The gas flow rate through a fluidized bed is limited on one hand by U_{mf} and on the other by entrainment of solids by the gas. This upper limit to gas flow rate is approximated by the terminal or free-fall velocity of the particles, (U_t).

The derivation of U_t is as follow.



$$F_d = \text{drag force}$$

$$F_b = \text{buoyancy force}$$

$$F_g = \text{gravitational force}$$

$$F = ma$$

$$\therefore a = 0$$

$$F = 0 \quad \dots\dots\dots (1)$$

$$F_d = F_g - F_b \quad \dots\dots\dots (2)$$

$$F_d = A_t K C_d$$

where A_t = cross sectional area of bed

C_d = drag coefficient

K = characteristic kinetic energy

$$K = \frac{1}{2} \rho_g U_t^2 \quad \dots\dots\dots (3)$$

$$F_g = \frac{4}{3} \frac{\pi d_p^3}{8} \rho_s \frac{g}{g_c}$$

$$= \frac{\pi d_p^3}{6} \rho_s \frac{g}{g_c} \quad \dots\dots\dots (4)$$

$$F_b = \frac{\pi d_p^3}{6} \rho_g \frac{g}{g_c} \quad \dots\dots\dots (5)$$

substitute Eq. (3), (4) and (5) in Eq. (2)

$$\frac{1}{2} \rho_g U_t^2 \frac{\pi d_p^2}{4} C_d = \frac{\pi d_p^3}{6} (\rho_s - \rho_g) \frac{g}{g_c}$$

$$U_t = \frac{\frac{4}{3} d_p (\rho_s - \rho_g) \frac{g}{g_c}^{\frac{1}{2}}}{\rho_g C_d} \quad \dots\dots\dots (6)$$

Eq. (6) is used for spherical particles

An alternate way of finding U_t for spherical particles uses analytic expressions for the drag coefficient C_d . Unfortunately,

no single simple expression can represent the experimental findings in the flow regime of interest, so for the particle Reynolds number ranges shown we have

$$C_d, \text{ spherical} = \frac{24}{Re_p} \quad \text{for } Re_p < 0.4 \quad \dots\dots (7)$$

$$C_d, \text{ spherical} = \frac{10}{Re_p^{1/2}} \quad \text{for } 0.4 < Re_p < 500 \quad \dots\dots (8)$$

$$C_d, \text{ spherical} = 0.43 \quad \text{for } 500 < Re_p < 200,000 \quad \dots\dots (9)$$

substituting these values of C_d in Eq. (6) gives analytic expressions for U_t , or

$$U_t, \text{ spherical} = \frac{g(\rho_s - \rho_g) d_p^3}{18} \quad \text{for } Re_p < 0.4 \quad \dots\dots (10)$$

$$U_t, \text{ spherical} = \frac{4}{225} \frac{(\rho_s - \rho_g)^2 g^2}{\rho_g} \frac{1}{3} d_p \quad \text{for } 0.4 < Re_p < 500 \quad \dots\dots (11)$$

$$U_t, \text{ spherical} = \frac{3.1g (\rho_s - \rho_g) d_p^{1/2}}{\rho_g} \quad \text{for } 500 < Re_p < 200,000 \quad \dots\dots (12)$$

where $Re_p = \frac{d_p \rho_g U_t}{\mu}$

For non-spherical particles, Pettyjohn and Christiansen (1948) have proposed that the terminal velocity can be estimated by multiplying the above U_t values by a correction factor.

$$\xi = 0.843 \log \left(\frac{\psi}{0.065} \right) \dots\dots\dots (13)$$

where ψ = sphericity of particle.

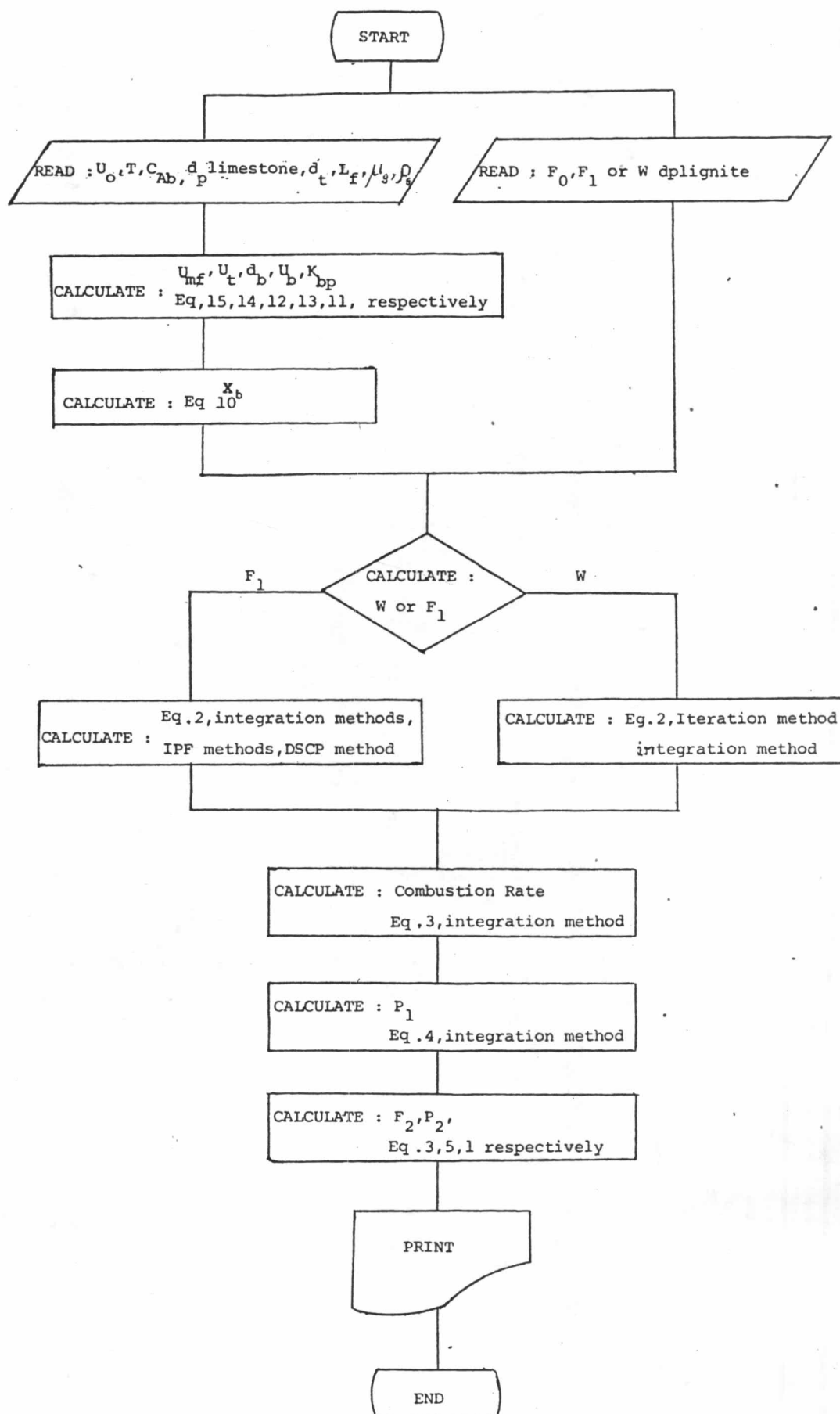
Appendix C

System Model Computation

The system model is computed by a numerical digital computation method.

In computation, the Simpson rule and the Trapezoidal rule are used for integration, an Iteration method is used for finding the weight of Char in bed W, Davies-Swann-Campey and Powell's method (DSCP) for optimization in order to find the value of the overflow rate F_1 . Interpolation uses Newton Forward, Newton Backward, Everett method, IPF.

The flow chart and computer programs for system model calculation are shown followingly. The numbers of equations in the flow chart are from section 4.4.8.



```

C THESE PROGRAM WRITTEN BY MR.NUTTAWUT WACHARAKUNDILOKE
C IT IS SYSTEM MODELLING PROGRAM OF COMBUSTION EFFICIENCY
C DEPARTMENT OF CHEMICAL ENGINEERING,FACULTY OF ENGINEERING
C CHULALONGKORN UNIVERSITY
C H=FINITE DIFFERENCE OF PARTICLE DIAMETER
C HZR=FINITE DIFFERENCE OF PARTICLE .LT.DPCMIN
CHF1TR=FINITE DIFFERENCE OF TRIAL FEED RATE
C THE VALUE OF F1STRT IS .LT. FIEND,THE VALUE OF FIX IS FROM LARGE TO SMALL
C TK=TEMPERATLFE CF AIR IN BED IN KELVIN UNIT
C MC=MOLECULAR WEIGHT OF CARBON,MS=MOLECULARWEIGHT OF SULPHER
C P=ATMOSPHERIC PRESSURE
C R=GAS CONSTANT
C EXC=EXCESS AIR
C PCTCCL=PERCENT OF FIXED CARBON IN COAL
C PCCL=PEFCENT OF ALL CARBON IN COAL
C PSCL=PERCENT OF SULPHER IN COAL
C THE PERCENT ABOVE IS NOT THE REAL PERCENT MEANING,OTHERWISE HAVE TO *100.
C DG=DENSITY CF AIR AT 35 C
C N=NUMBERS OF PARTICLE DIAMETER VALUES OF FEED
C EMEU=VI SCOUSITY OF AIR
C
C NN=NUMBERS CF SUPERFICIAL VELOCITY(LO)
C LET PCTCCL=FCCL
C
1     DIMENSICN Y1(999),Y11(999),Y1ZF(999),Y2(999),Y11ZR(999)
2     DIMENSICN DPC(999),DPC12(999),PC(999),P1(999),P2(999)
3     DIMENSICN UT(999),II(999),SF(999),RR(999),KK(999)
4     DIMENSICN ERRCF(999),FIXX(999),YSUM2(999),ERJPDN(999)
5     DIMENSICN LC(50),DAI(50),CAFI(50),DGI(50),EMEUI(50),F1I(50)
6     DIMENSICN TKI(50)
7     DIMENSICN UMF1(50)
8     DIMENSICN FOI(50)
9     DIMENSICN WI(50)
10    INTEGEF TRY,KNOWF1
11    REAL II,KK,IIIX,KKX
12    REAL LOWLMT
13    READ(5,10) DPCMAX,DPCMIN,H,DCL,G,DT,DCP,N
14    10 FORMAT(7F10.4,I10)
C F1STRT IS STARTING POINT OF TRIAL F1,AND IS UPPER LIMIT
C FIEND IS END POINT OF TRIAL F1, AND IS LOWER LIMIT
15    READ(5,11) F1STRT,FIEND,HF1TR,EPS,HZR
16    11 FORMAT(5F10.4)
C SPH=SPHERICITY OF COAL PARTICLE
17    READ(5,20) PCTCCL,SPH,NN
18    20 FORMAT(2F10.4,I10)
19    REAC(5,21) (UO(I),I=1,NN)
20    REAC(5,21) (CAI(I),I=1,NN)
21    REAC(5,21) (FCI(I),I=1,NN)
22    REAC(5,21) (WI(I),I=1,NN)
23    REAC(5,21) (F1I(I),I=1,NN)
24    21 FORMAT(8F10.4)

```

```

25      READ(5,22)   (CAPI(I),I=1,NN)
26      READ(5,22)   (DGI(I),I=1,NN)
27      READ(5,22)   (EMEI(I),I=1,NN)
28      22 FORMAT(EE10.4)
      C FOF BED HEIGHT MORE THAN TDH,A=C.
29      READ(5,23) A
30      23 FORMAT(F10.4)
      C*****
      C IF WE KNOW F1, THEN HAVE TO TRY FOR W, LET DATA BELOW IN A READ STATEMENT BE
      C W=0.0,WINIT=.....,F1=.....,KNOWF1=1,MAX=.....
      C IF WE DON'T KNOW F1,BUT WE KNOW THE VALUE OF W, LET DATA BE
      C W=.....,WINIT=0.,F1=0.,KNOWF1=0,MAX=0
      C*****
      C IF F1 KNOWN,W TO BE FOUND,AND F1 .EQ.0.,W FOUND IS THE MAX. ONE
      C THAT MAKES THE CARBON IN COAL COMBUSTED ALL
      C*****
31      READ(5,25) W,WINIT,KNOWF1,MAX
32      25 FORMAT(2F10.4,2I1)
      C IF DESIRE TO USE OPTIMIZATION FOR FINDING F1, LET KYDSCP=1
      C KYDSCP=KEY TO OPEN TO DSCP METHOD, KYDSCP=1 MEANS 'OPEN',=0 MEANS 'CLOSE'
      C IMAX=ALLOWABLE NUMBERS OF ITERATION IN DSCP
      C HDSCP=FINITE DIFFERENCE IN DSCP
      C XODSCP=THE INITIAL VALUE OF F1 TRIAL FROM WHICH THE SEARCH START
33      READ(5,26) KYDSCP,IMAX,HDSCP,XODSCP
34      26 FORMAT(2I10,2F10.4)
35      READ(5,30) (PC(I),I=1,N)
36      30 FORMAT(8F10.4)
37      READ(5,32) P,R,PCCL,MC
38      32 FORMAT(3F10.4,I10)
39      READ(5,33) (TKI(I),I=1,NN)
40      READ(5,33) (UMFI(I),I=1,NN)
41      33 FORMAT(8F10.4)
42      DO 100 I=1,N
43      DPC(I)=DFCMIN+FLOAT(I-1)*H
44      100 CONTINUE
45      DPMX=DFCMIN+HZR
46      DPZR=HZF
47      NL=(DPMX-DPZR)/HZR+1.
      C NL NOT BE EQUAL TO ZERO
48      NZR=N+NL
      C****
      C IF KNOWING F1, INSERT THESE PROGRAM(WITHOUT C) , BUT ALREADY KNOWN W
      C F1=.....
      C KEY=1.
      C IF(KEY.EQ.1.) GO TO 460
      C
49      DO 1000 ITR=1,NN
50      W=W1(ITR)
51      FOF=FOI(ITR)
52      UMF=UMFI(ITR)
53      TK=TKI(ITR)
54      UOQ=UO(ITR)
55      DA=DAI(ITR)
56      CAP=CAPI(ITR)
57      DG=DGI(ITR)
58      EMELE=EMELI(ITR)

```

```

59         F1=F11(I1R)
60         IF(KNOWF1.EQ.1) GO TO 190
61         L=(F1STRT-F1END)/HF1TR+1.1
62         WRITE(6,189) L
63     189  FORMAT(/T10,'L=',I10)
64         GO TO 191
65     190  CONTINUE
66         W=WINIT
67         WO=W
68         L=MAX
69     191  CONTINUE
70         DO 400 TRY=1,L
71         IF(KNOWF1.EQ.1) GO TO 192
72         Z=TRY-1
73         FIX=F1STRT-Z*HF1TR
74         GO TO 192
75     192  CONTINUE
76         FIX=F1
77     193  CONTINUE
78         DO 200 I=1,N
79         N1=N-I+1
80         DPCX=DPCMAX-FLCAT(I-1)*H
81         CALL IIDF(DPCMAX,H,FIX,IIX,DCL,G,DT,DDP,DA,PCTCCL,SPH,
      * U00,w,CAP,DC,EMEU,DPCX,UMF,A,UTX,SHX,RRX,KKX)
82         Y2(I)=PC(N1)/((DPCX**3)*IIX)
83         CALL INTGT(H,I,Y2,SUM2)
84         Y11(N1)=(DPCX**3)*IIX/ABS(RFX)
85         Y1(N1)=Y11(N1)*SUM2
86     200  CONTINUE
87         DO 300 I=1,NL
88         NL1=NL-I+1
89         DPX=DPCMIN-FLOAT(I)*HZR
90         CALL IIDP(DPCMAX,HZR,FIX,IIX,DCL,G,DT,DDP,DA,PCTCCL,SPH,
      * U00,w,CAP,DC,EMEU,DPX,UMF,A,UTX,SHX,RRX,KKX)
91         Y11ZR(NL1)=(DPX**3)*IIX/ABS(RFX)
92         Y1ZR(NL1)=Y11ZR(NL1)*SUM2
93     300  CONTINUE
94         CALL INTGT(H,N,Y1,SUM11)
95         CALL INTGT(HZR,NL,Y1ZR,SUM12)
96         SUM1=SUM11+SUM12
97         IF(KNOWF1.EQ.1) GO TO 331
98         DIF=(W/FCO)-SUM1
99         YDIF=ABS(DIF)
100        IF(YDIF.LE.EPS) GO TO 440
101        EFRCF(TFY)=YDIF
102        FIXX(TRY)=FIX
103        GO TO 400
104     331  W1=FCO*SLM1
105        DIF=ABS(WO-W1)
106        IF(DIF.LE.EPS) GO TO 459
107        WO=W1
108        W=WC
109     400  CONTINUE
110        IF(KNOWF1.EQ.1) GO TO 451
111        DIV=1
112        WRITE(6,410)

```

```

113 410 FORMAT('1',T1C,'TRYING FOR F1 IS DIVERGE'/
      * T10,'TRY AGAIN BY HAND CHECKING AND THEN GO TO NEXT STEP'/)
114 WRITE(6,42C)
115 420 FORMAT(110,7('-----')/125,'F1 TRIAL',T45,'|',
      * T60,'ERRCR'/T10,7('-----'))
116 WRITE(6,43C)(F1XX(I),ERROR(I),I=1,L)
117 430 FORMAT(125,F1C.4,T45,'|',T55,F13.8)
118 WRITE(6,434)
119 434 FORMAT(11C,7('-----'))
120 IF(KYDSCP.NE.1) GO TO 437
C FIND FIXX THAT ITS ERROR IS EQUAL TO ZERO BY OPTIMIZATION
C USING DAVIES-SWANN-CAMPEY AND POWELL'S METHOD(DSCP)
121 LOWLMT=F1XX(L)
122 UPLMT=F1XX(1)
123 DO 435 I=1,L
124 LII=L-(I-1)
125 ERUPDN(LII)=EFRDR(I)
126 435 CONTINUE
127 CALL DSCF(HDSCP,XDSCP,F1,EFF1,EPS,IMAX,LOWLMT,HF1TR,ERLPCN,L)
128 WRITE(6,438) F1,ERF1
129 438 FORMAT(11C/2CX,'MINIMIZATION BY DSC-POWELL METHOD'/ 25X,6HX = ,
      * F12.5,25X,6HPM = ,F12.5)
130 WRITE(6,436) F1,ERF1
131 436 FORMAT('/',T10,'BUT AFTER HAND CHECKING AND DESIRING TO USE OPTIMIZ
      *ATION NUMERICAL CALCULATION',/,T10,'OVERFLOW,F1,=',F10.4,
      * T45,'BUT ERROR FROM CALCULATION =',F10.4)
132 GO TO 460
133 437 STOP
134 440 DIV=C
135 F1=F1X
136 WRITE(6,45C) F1
137 450 FORMAT('1',T1C,'TRYING FOR F1 IS CONVERGE,',T4), 'F1=',F1C.4)
138 GO TO 46C
139 451 CONTINUE
140 WRITE(6,452) W
141 452 FORMAT('1',T1C,'FINDING W IS DIVERGE, LAST TRIAL OF W IS',F1C.4)
142 STOP
143 459 CONTINUE
144 W=W1
145 460 CONTINUE
C****
C FIND TABLE OF DPC(I),PD(I),SH(I),KK(I),RR(I),II(I)
C FROM NEAR ZERO TO DPCMAX
C INPUT F1 FROM HAND CHECKING IF TRYING FOR F1 IS DIVERGE
146 DO 480 I=1,NL
147 DPC12(I)=FZR*FLOAT(I)
148 480 CONTINUE
149 DO 490 I=1,N
150 NLI=NL+I
151 DPC12(NLI)=DPCMIN+FLOAT(I-1)*H
152 490 CONTINUE
153 DO 500 I=1,NZF
154 DPX=DPC12(I)
155 IF(I.LE.NL) GO TO 495
156 HHZR=H
157 GO TO 496

```

```

158 495 CCNTINUE
159 HHZR=HZF
160 496 CONTINUE
161 CALL IICF(DPCMAX,HHZF,F1,IIX,DCL,G,CT,DDP,DA,PCTCCL,SPH,
* UOO,k,CAP,CG,E4EU,DPX,UMF,A,UTX,SFX,PRX,KKX)
162 UT(I)=UTX
163 II(I)=IIX
164 SH(I)=SFX
165 RR(I)=PRX
166 KK(I)=KKX
167 500 CONTINUE
168 WRITE(6,510) UOO,F1,W
169 510 FORMAT('1',T1C,'UO =',F10.4,',',T30,'F1 =',F10.4,',',T50,'W =',
* F10.4/)
170 WRITE(6,520)
171 520 FORMAT(110,8('-----')/115,'DPC',T21,'|',T26,'UT',T33,'|',
* T38,'SF',T44,'|',T52,'KK',T58,'|',T65,'RR',T72,'|',T78,'II'/
* T1C,E('-----'))
172 WRITE(6,530) (DPC12(I),UT(I),SH(I),KK(I),RR(I),II(I),I=1,NZF)
173 530 FORMAT(110,F10.4,T21,'|',T22,F10.4,T33,'|',T34,F10.4,T44,'|',
* T45,F10.4,T58,'|',T59,F10.4,T72,'|',T73,F10.4)
174 WRITE(6,540)
175 540 FORMAT(11C,8('-----'))
C***FIND P1(I)
C***FIND P1(CFC1),SIZE DISTRIBUTION OF COAL PARTICLE IN F1
176 DO 600 I=1,N
177 N1=N-I+1
178 NZF1=NZF-I+1
179 YSUM2(I)=PC(N1)/((DPC(N1)**3)*II(NZF1))
180 CALL INTGT(H,I,YSJ42,SUM2)
181 A=F00*(CFC(N1)**3)*II(NZF1)/(W*ABS(FP(NZF1)))
182 P1(NZF1)=A*SUM2
183 600 CONTINUE
184 DO 700 I=1,N1
185 NZR1=NZF-I
186 DPCX=DPCMIN-FLCAT(I)*HZR
187 P1(NZR1)=FCC*(DPCX**3)*II(NZR1)*SUM2/(W*ABS(RR(NZR1)))
188 700 CONTINUE
189 DO 701 I=1,NZF
190 NT=NZR-(I-1)
191 PIT=P1(NT)
192 IF(PIT.EC.C.) GO TO 702
193 701 CONTINUE
194 GO TO 704
195 702 NT1=NT-1
196 DO 703 I=1,NT1
197 P1(I)=C.
198 703 CONTINUE
199 704 CONTINUE
200 WRITE(6,705)
201 705 FORMAT('1',11C,7('-----'))
202 WRITE(6,706)
203 706 FORMAT(11C,'PARTICLE DIAMETER OF COAL,CM',T40,'|',T51,
* 'SIZE DISTRIBUTION ,CM**(-1)')
204 WRITE(6,707)
205 707 FORMAT(11C,7('-----'))

```

```

206      WRITE(6,7C6) (DPC(I),P0(I),I=1,N)
207      708 FORMAT(118,F13.4,T40,'|',T51,F13.4)
208      WRITE(6,7C7)
209      WRITE(6,710) FCO,4,U00,F1
210      710 FORMAT('1',T1C,'F) =',F10.4,',',T25,'k=',F10.4,',',T40,'UC=',
      * F10.4/T1C,'F1=',F10.4)
211      WRITE(6,720)
212      720 FORMAT(11C,7('-----'))/
      * T25,'DPC1,CM',T45,'|',T60,'P1,CM**--1'/T10,7('-----'))
213      WRITE(6,720) (DPC12(I),P1(I),I=1,NZF)
214      730 FORMAT(125,F10.4,T45,'|',T60,F10.4)
215      WRITE(6,740)
216      740 FORMAT(11C,7('-----'))
      C***
      C**FIND ELUTRIATION RATE,F2
217      DPCX=HZF
218      DO 800 I=1,NL
219      YSUM2(I)=3.*W*P1(I)*RR(I)/DFCX
220      DPCX=DPCX+HZF
221      800 CONTINUE
222      CALL INTGT(HZF,NL,YSUM2,SMF21)
223      DO 850 I=1,N
224      NLI=NL+I
225      DPCX=DPCMIN+FLOAT(I-1)*H
226      YSUM2(I)=3.*W*P1(NLI)*RP(NLI)/DPCX
227      850 CONTINUE
228      CALL INTGT(H,N,YSUM2,SMF22)
229      SMF2=SMF21+SMF22
230      WRITE(6,860) SMF2
231      860 FORMAT('1',T1C,'RATE OF COAL COMBUSTED=',F10.4)
232      F2=FCO-F1-SMF2
      C**
      C* FIND P2(DPC2),SIZE DISTRIBUTION OF COAL PARTICLE IN F2
233      DO 900 I=1,NZF
234      P2(I)=W*KK(I)*P1(I)/F2
235      900 CONTINUE
236      WRITE(6,905) F2
237      905 FORMAT('1',T1C,'F2 =',F10.4)
238      WRITE(6,910)
239      910 FORMAT(11C,7('-----'))/
      * T25,'DPC2,CM',T45,'|',T60,'P2,CM**--1'/T10,7('-----'))
240      WRITE(6,730) (DPC12(I),P2(I),I=1,NZF)
241      WRITE(6,740)
      C*
      C CONCLUSION
242      EFF=(SMF2/FCC)*100.
243      IF(F2.GT.C.) GO TO 915
244      F2=C.C
245      915 CONTINUE
246      WRITE(6,920) SMF2
247      920 FORMAT(11C,'RATE OF COAL COMBUSTED=',F10.4)
248      WRITE(6,930) FCO,F1,F2,EFF
249      930 FORMAT(110,'FC =',F10.4,T25,'F1 =',F10.4,T40,'F2 =',F10.4,
      * T55,'EFF =',F10.4)
      C FIND PERCENT OF EXCESS AIR IN BED
250      AA=0.21*F*(3.1416*(DT+DDP)*(DT-DDP)/4.)*U00
251      BB=R*TK*FCC*FCCL/4C
252      EXC=AA/BB-1.
253      PCTEXC=EXC*100.0
254      WRITE(6,940) TK,PCTEXC
255      940 FORMAT(//T10,'TEMPERATURE OF AIR IN BED =',F10.4,3X,'K'/
      * T10,'EXCESS AIR=',F10.4)
256      1000 CONTINUE
257      STOP
258      END

```

```

C SUBROUTINE FOR FINDING II,SH,RR,KK
C H= FINITE DIFFERENCE OF COAL DIAMETER
C F1= FEED RATE OF COAL
C DPCMAX=MAXIMUM PARTICLE DIAMETER
C DCL=DENSITY OF COAL PARTICLE
C DT=DIAMETER OF BED
C DDP=DIAMETER OF DRAIN PIPE IN BED
C DA=Coefficient DIFFUSIVITY
C PCTCCL=PERCENT CARBON IN COAL PARTICLE.(IN FRACTION)
C SPH=SPHERICITY
C W=WEIGHT OF COAL IN BED
C CAP=CONCENTRATION OF OXYGEN IN PARTICULATE PHASE
C
1   SUBROUTINE IIDP(DPCMAX,H,F1,II,DCL,G,DT,DDP,DA,PCTCCL,SPH,
    *   UO,W,CAP,DG,EMEU,DPC,UMF,A,UT,SH,RR,KK)
2   DIMENSION Y(1000)
3   REAL II,KK,KKX
4   N=(DPCMAX-DPC)/H+1.
5   DO 10 I=1,N
6   DPX=DPC+FLOAT(I-1)*H
7   CALL TMV(DCL,G,DPX,DG,EMEU,SPH,UTX)
8   CALL SRK(DCL,G,DT,DDP,DA,PCTCCL,UO,h,CAP,DG,EMEU,DPX,UMF,A,
    *   UTX,SHX,RRX,KKX)
9   Y(I)=(F1/W+KKX)/FRX
10  CONTINUE
11  30 CALL INTGT(H,N,Y,SLM)
12  IF(SUM.GE.70) GO TO 50
13  40 II=1./EXP(SUM)
14  CALL TMV(DCL,G,DPC,DG,EMEU,SPH,UT)
15  CALL SRK(DCL,G,DT,DDP,DA,PCTCCL,UO,W,CAP,DG,EMEU,DPC,UMF,A,
    *   UT,SH,FR,KK)
16  RETURN
17  50 WRITE(6,60) SUM,DPC
18  60 FORMAT('1',T10,'SUM IS MORE THAN 70,=',F10.4,'DPC= ',F10.4)
19  STOP
20  END

```



```

C SUBROUTINE FOR CALCULATING INTEGRATION WITH FINITE DIFFERENCE
C SUBROUTINE INTGT(F,N,Y,SLM)
C H=FINITE DIFFERENCE OF X
C Y=FUNCTION OF X
1  SUBROUTINE INTGT(H,N,Y,SUM)
2  DIMENSION Y(N)
3  REAL ITRAP,ISIMP
4  IF(N.GE.7) GO TO 5
5  SUM1=0
6  DO 3 I=1,N
7  SUM1=SUM1+Y(I)*H
8  3 CONTINUE
9  SUM=SUM1
10 GO TO 60
11 5 CONTINUE
12 KT1=FLOAT(N)/2.+0.6
13 KT2=N/2
14 IF(KT1-KT2) 1C,4C,1C
C SIMPSON RULE
15 10 N1=N-1
16 YM1=Y(2)
17 DO 20 I=4,N1,2
18 YM1=YM1+Y(I)
19 20 CONTINUE
20 N2=N-2
21 YM2=Y(3)
22 DO 30 I=5,N2,2
23 YM2=YM2+Y(I)
24 30 CONTINUE
25 ISIMP=(H/3.)*(Y(1)+4.*YM1+2.*YM2+Y(N))
26 SUM=ISIMP
27 GO TO 60
C TRAPEZOIDAL RULE
28 40 YMID=Y(2)
29 N1=N-1
30 DO 50 I=3,N1
31 YMID=YMID+Y(I)
32 50 CONTINUE
33 ITRAP=(H/2.)*(Y(1)+2.*YMID+Y(N))
34 SUM=ITRAP
35 60 RETURN
36 END

```

```

C SUBROUTINE FOR CALCULATING SHERWOOD NUMBER(SH),
C SHRINKING RATE OF PARTICLE (FR)
C ELUTRIATION CONSTANT (KK)
C CAP=CONCENTRATION OF OXYGEN IN PARTICULATE PHASE,GMMCLE/CM**3
C PCTCC= PERCENT OF CARBON IN CCAL
C DCL = DENSITY OF COAL, DPC = DIAMETER OF CCAL PARTICLE
C UO USED IN CALCULATING KK NOT BE LESS THAN UMF
1  SUBROUTINE SRK(DCL,C,CT,CDP,DA,FCTCCL,UO,W,CAP,CG,EMEU,DPC,UMF,A,
   * UT,SH,FR,KK)
2  REAL KK,K,K1
3  SH = 2.0+0.6*((UO*DG/EMEU)**0.5)*((EMEL/(CG*DA))**(1./3.))
   * *(DPC**0.5)
4  RR=57.6*CAP*ST*DA/(DPC*PCTCCL*DCL)
5  IF(UT.GT.UO) GO TO 10
6  K1=130.*EXP(-10.4*((LT/UO)**0.5)*((UMF/(UO-UMF))**0.25))+A
7  K=3.1416*(CT-CDP)*(DT+DDP)/4.
8  KK=(CG*LC*K/K1)*K1
9  RETURN
10 10 KK=0
11  RETURN
12  END

```

```

C SUBROUTINE FOR CALCULATING TERMINAL VELOCITY (UT)
C DS= DENSITY OF PARTICLE
C G= GRAVITATIONAL ACCELERATION
C DP=PARTICLE DIAMETER
C DG=DENSITY OF GAS
C EMEU=VISCOSITY OF GAS
C SPH=SPHERICITY
1  SUBROUTINE TMV(DS,G,DP,DG,EMEU,SPH,LT)
2  ANU=C.843*ALOG10(SPF/C.C65)
3  UTSPHR=C.153*(DP**1.14)*(G**0.71)*((DS-DG)**0.711)/((EMEU**0.43)
   * *(DG**0.29))
4  UT=ANU*UTSPHR
5  RETURN
6  END

```

```

C DSC POWEL METHOD USING THE COMBINED DAVIES-SWANN CAMPEY AND POWEL
C *****
C THIS SUBROUTINE USING THE COMBINED DAVIES-SWANN CAMPEY AND POWELL
C METHOD(DSCP),TC COMPUTE THE MINIMUM OF A CONTINUOUS FUNCTION
C THE FOLLOWING PROVISIONS FOR USERS TO SPECIFY,
C XO-THE INITIAL VALUE OF X FROM WHICH THE SEARCH START
C DX-THE SEARCH STEP SIZE
C EPS-THE PRECISION INDEX
C MAX-THE MAXIMUM ALLOWABLE NUMBER OF ITERATIONS
C X00,HH,YY,NN ARE ARGUMENTS IN SUBROUTINE SUBPROGRAM IPF,
C LOWER LIMIT ,FINITE DIFFERENCE,VALUES OF Y,NUMBERS OF Y RESPECTIVELY
C *****WRITEN BY NUTTAWUT WACHARAKUNDILOK*****
1   SUBROUTINE DSCP(CX,XO,XM,FM,EPS,MAX,XCC,HH,YY,NN)
2   DIMENSION YY(NN)
3   DOUBLE PRECISION X1,X2,X3
4   DATA ITAP,JTAF/5,6/
5   H=DX
6   2 CONTINUE
C START THE SEARCH THE BOUND MINIMUM
7   ITR=0.
8   M=0
9   M1=0
10  CALL IPF(XOO,HH,YY,NN,XO,FA)
11  X=XO+H
12  CALL IPF(XOO,HH,YY,NN,X,FB)
13  IF(FB.LE.FA) GO TO 4
C REVERSE THE SEARCH DIRECTION TO OBTAIN DECREASING IN FUNCTION VALUE
C
14  X=XO
15  H=-2.*H
16  GO TO 6
C THE FUNCTION STILL DECREASING ,INCREASE THE STEP SIZE BY DOUBLE
C THE PREVIOUS INCREASE IN STEP SIZE
17  4 H=2.*H
18  6 CONTINUE
19  X=X+H
20  CALL IPF(XCC,HH,YY,NN,X,FC)
21  IF (FC.GT.FB) GO TO 6
22  FB=FC
23  ITR=ITR+1
24  IF(ITR.GT.MAX) GO TO 8
25  GO TO 4
C MINIMUM IS BOUND
26  8 CONTINUE
27  X2=X-H
28  IF(X2.LT.X) GO TO 9
29  X1=X
30  X3=X2-H/2.
31  GO TO 10
32  9 CONTINUE
33  X3=X
34  X1=X2-H/2.
35  10 H=-H/2.

```

```

36      X=X+H
37      X2S=X2
38      CALL IPF(XCC,HH,YY,NN,X2S,FX2)
39      CALL IPF(XCC,HH,YY,NN,X,FXX)
40      IF(FX2.LT.FXX) GO TO 12
-----
41      X1=X2
42      X2=X
43      GO TO 14
44      12 X3=X
45      14 CONTINUE
46      X1S=X1
-----
47      X2S=X2
48      X3S=X3
49      CALL IPF(X00,HH,YY,NN,X1S,FA)
50      CALL IPF(X00,HH,YY,NN,X2S,FB)
51      CALL IPF(X00,HH,YY,NN,X3S,FC)
52      B=2.*(FA-2.*FB+FC)
-----
53      IF(B) 15,24,15
54      15 X=X2-ABS(H)*(FA-FC)/B
55      WRITE(6,111) X
56      111 FORMAT(//,T10,'X=',F10.4)
57      ITR=C
58      16 CONTINUE
-----
59      XX1=X-X1
60      XX2=X-X2
61      XX3=X-X3
62      IF(ABS(XX1).LE.EPS) GO TO 40
63      IF(ABS(XX2).LE.FPS) GO TO 40
64      IF(ABS(XX3).LE.FPS) GO TO 40
-----
65      WRITE(6,222) X
66      222 FORMAT(T10,'X1EST=',F10.4)
67      ITR=ITR+1
68      IF(ITR.GT.MAX) GO TO 38
69      CALL IPF(X00,HH,YY,NN,X,FP)
70      XP=X
-----
71      XP2=XP-X2
72      IF(ABS(XP2).LE.EPS) GO TO 40
73      IF(FB.L1.FP) GO TO 18
74      IF(X.LT.X2) GO TO 17
75      X1=X2
76      FA=FB
-----
77      X2=X
78      FB=FP
79      GO TO 20
80      17 X3=X2
81      FC=FB
82      X2=X
-----
83      FB=FP
84      GO TO 20
85      18 CONTINUE
86      IF(X.L1.X2) GO TO 19
87      X3=X
88      FC=FP
-----
89      GO TO 20
90      19 X1=X
91      FA=FP

```



```

92      20 CONTINUE
      C NOW PERFORM THE PARABOLIC INTERPOLATION
93      A=(X2-X3)*FA+(X3-X1)*FB+(X1-X2)*FC
94      IF(A) 21,28,21
95      21 X=(1./2.)*((X2**2-X3**2)*FA+(X2**2-X1**2)*FB+(X1**2+X2**2)*FC)/A
96      IF(X.GT.X3) GO TO 22
97      IF(X.LT.X1) GO TO 22
98      GO TO 23
99      22 X312=(X3-X1)/2.
100     B=2.*(FA-2.*FB+FC)
101     IF(B.EQ.C.) GO TO 24
102     X=X2-ABST(X312)*(FA-FC)/B
103     23 CONTINUE
104     GO TO 16
105     24 CONTINUE
106     IF(M) 26,26,30
107     26 M=M+1
108     GO TO 8
109     28 CONTINUE
110     IF(M1) 29,29,32
111     29 M1=M1+1
112     GO TO 8
113     30 WRITE(JTAP,281)
114     GO TO 40
115     32 WRITE(JTAP,282)
116     GO TO 40
117     38 CONTINUE
118     WRITE(JTAP,280) ITR
      C
119     40 CONTINUE
      C
120     XM=X
121     CALL IFF(XCC,HH,YY,NN,XM,FM)
122     280 FORMAT(1HC,2CX,' SEARCH FAILED WITHIN',I6,' ITERATIONS')
123     281 FORMAT(1HC/2CX,' THE DSC DIVISOR BEING ZERO SECOND TIME')
124     282 FORMAT(1HC/2CX,' THE POWELL DIVISOR BEING ZERO SECOND TIME')
      C
125     RETURN
126     END

```

```

C SUBROUTINE IPF(INTERPOLATION) INCLUDES
C NEWTON FORWARD INTERPOLATION FORMULAR
C NEWTON BACKWARD INTERPOLATION FORMULAR
C AND EVERETT INTERPOLATION FORMULAR
C M=DEGREE OF THE INTERPOLATION POLYNOMIAL,M=N-1
C N= THE NUMBER OF TABULATED DATA POINTS
C X1=INITIAL VALUE OF X,t=FINITE DIFFERENCE,CY=THE DIFFERENCE OF Y OR ANY DEGREE
C XP=THE POINT OF X WHICH IS USED TO FIND YP OR Y-REQUIRED
C N NOT BE LESS THAN 7
1     SUBROUTINE IPF(X1,H,Y,N,XP,YP)
2     DIMENSION Y(N)
3     DIMENSION CY(200,200)
4     DOUBLE PRECISION DY,PF,CQ,A,P,Q
5     M=N-1
6     N1=N-1
7     DO 2 I=1,N1
8     2 DY(I,1)=Y(I+1)-Y(I)
9     DO 6 K=2,M
10    L=N-K
11    DO 4 I=1,L
12    4 DY(I,K)=CY(I+1,K-1)-CY(I,K-1)
13    6 CONTINUE
14    XN=X1+t*FLCAT(N1)
15    SECT=(XN-X1)/3.
16    SECT1=X1+SECT
17    SECT2=X1+2*SECT
18    IF(XP.GT. SECT2) GO TO 30
19    IF(XP.GE. SECT1) GO TO 20
C XP=X1+p*H
C NFIF
20    10 P=(XP-X1)/H
21    PP=1.
22    YP=Y(1)
23    DO 12 K=1,4
24    K1=K-1
25    PP=PF*(F-FLOAT(K1))/FLCAT(K)
26    12 YP=YP+PP*DY(1,K)
27    RETURN
C XP=XN-p*t
C NBIF
28    30 P=(XN-XP)/H
29    PP=1.
30    YP=Y(N)
31    NN=N
32    DO 33 K=1,4
33    K1=K-1
34    PP=-PF*(F-FLCAT(K1))/FLCAT(K)
35    NN=NN-1
36    33 YP=YP+PP*CY(NN,K)
37    RETURN
C EIF
38    20 K=FLOAT(N)/2.+0.6
39    KI=N/2+C.6

```

```
40      XO=X1+(FLOAT(K)-1.)*H
41      P=(XP-XC)/H
42      C=1.-P
43      A=1.
44      PP=P
45      QQ=Q
46      YP=Q*Y(K)+P*Y(K+1)
47      IF(K-KI) 22,21,22
48  21  L=M/2
49      GO TO 25
50  22  L=M/2-1
51  25  CONTINUE
52      DO 23 I=1,3
53      II=2*I
54      A=A*II*(II+1)
55      QQ=(Q+I)*CC*(Q-I)
56      PP=(P+I)*PP*(P-I)
57      KK=K-I
58  23  YP=YP+(QQ*DY(KK,II)+PP*DY(KK+1,II))/A
59      RETURN
60      END
```

```

C SUBROUTINE FOR FINDING CAP,X,KBP
C CAP= CONCENTRATION OF OXYGEN IN PARTICULATE PHASE
C CAB= CONCENTRATION OF OXYGEN IN BUBBLE AT SPECIFIED HEIGHT
C CAO= CONCENTRATION OF OXYGEN IN BUBBLE AT INLET OF BED(GAS INLET)
C UC= SUPERFICIAL VELOCITY, UB= BUBBLE VELOCITY, UMF= MINIMUM FLUIDISATION VEL.
C DB= BUBBLE DIAMETER , LF= HEIGHT OF BED
C G= GRAVITATIONAL ACCELERATION, DA= OXYGEN DIFFUSION COEFFICIENT
C KBP= GAS INTERCHANGE COEFFICIENT FROM BUBBLE TO PARTICULATE PHASE
C X= INTERPHASE CHANGE
C LB NOT EQUAL TO ZERO
C THIS SUBROUTINE DOUBLE PRECISION CAB,CAB,B,C

```

```

SUBROUTINE DXYC(CAB,CAO,UMF,UC,DB,DA,G,LF,UB,KBP,X,CAP)
DOUBLE PRECISION CAB,CAO,B,C
REAL LF,KBP
A=(DA**0.5)*(G**0.25)/(DB**1.25)
KBP=4.5*(UMF/DB)+5.85*A
UB=0.711*((G*DB)**0.5)+(UC-UMF)
X=KBP*LF/UB
IF(X.GT.50.) GO TO 10
B=EXP(-X)
GO TO 20
10 B=0.0
20 CONTINUE
C=1.
CAP=(CAB-CAO*B)/(C-B)
RETURN
END

```

```

C SUBROUTINE FOR FINDING BUBBLE DIAMETER IN FBC
C LF=HEIGHT OF FBC BED
C UC=SUPEFFICIAL VELOCITY
C UMF=MINIMUM FLUIDIZATION VELOCITY
C DB= BUBBLE DIAMETER

```

```

SUBROUTINE DRCG(UC,UMF,LF,DB)
REAL LF
DB1=0.0326*((UC-UMF)**1.11)*(LF**0.81)
DB=DB1/1.81
RETURN
END

```


An Example of Output from the Computer Simulation

UJ = 394.2300, F1 = 0.0259, W = 8.6584

CPC	UT	SH	KK	RR	II
0.0020	5.3542	2.3501	122.2946	0.5239	0.0001
0.0040	11.7956	2.4551	83.3734	0.2781	0.0001
0.0060	18.7332	2.6063	61.4450	0.1937	0.0002
0.0080	26.0041	2.7001	47.2101	0.1505	0.0004
0.0100	33.5367	2.7827	37.2730	0.1241	0.0007
0.0120	41.2845	2.8575	30.0110	0.1062	0.0013
0.0140	49.2160	2.9262	24.5308	0.0932	0.0022
0.0160	57.3083	2.9901	20.2547	0.0833	0.0037
0.0180	65.5437	3.0502	16.9581	0.0756	0.0060
0.0200	73.9086	3.1070	14.2896	0.0693	0.0092
0.0220	82.3915	3.1610	12.1284	0.0641	0.0137
0.0240	90.9833	3.2126	10.3591	0.0597	0.0196
0.0260	99.6760	3.2621	8.8977	0.0559	0.0274
0.0280	108.4627	3.3098	7.6797	0.0527	0.0371
0.0300	117.3381	3.3558	6.6584	0.0499	0.0492
0.0320	126.2966	3.4002	5.7963	0.0474	0.0633
0.0340	135.3339	3.4433	5.0645	0.0452	0.0803
0.0360	144.4461	3.4852	4.4400	0.0432	0.0993
0.0380	153.6292	3.5259	3.9047	0.0414	0.1213
0.0400	162.8804	3.5655	3.4439	0.0397	0.1449
0.0420	172.1967	3.6042	3.0456	0.0383	0.1718
0.0440	181.5752	3.6419	2.7001	0.0369	0.1994
0.0460	191.0136	3.6788	2.3994	0.0357	0.2304
0.0480	200.5058	3.7149	2.1369	0.0345	0.2612
0.0500	210.0615	3.7503	1.9070	0.0334	0.2953
0.0520	219.6669	3.7849	1.7053	0.0325	0.3283
0.0540	229.3241	3.8189	1.5276	0.0315	0.3647
0.0560	239.0313	3.8523	1.3709	0.0307	0.3988
0.0600	258.5903	3.9173	1.1095	0.0291	0.4990
0.0700	308.2705	4.0710	0.6707	0.0259	0.6631
0.0800	358.9575	4.2139	0.4181	0.0235	0.8507
0.0900	410.5413	4.3482	0.0000	0.0215	0.9040
0.1000	462.9353	4.4753	0.0000	0.0200	0.9053
0.1100	516.0693	4.5961	0.0000	0.0186	0.9067
0.1200	569.8855	4.7115	0.0000	0.0175	0.9082
0.1300	624.3323	4.8222	0.0000	0.0165	0.9060
0.1400	679.3696	4.9288	0.0000	0.0157	0.9077
0.1500	734.9612	5.0316	0.0000	0.0150	0.9095
0.1600	791.0750	5.1310	0.0000	0.0143	0.9113
0.1700	847.6821	5.2273	0.0000	0.0137	0.9171
0.1800	904.7559	5.3209	0.0000	0.0132	0.9191
0.1900	962.2759	5.4119	0.0000	0.0127	0.9174
0.2000	1020.2227	5.5005	0.0000	0.0123	0.9197
0.2100	1078.5762	5.5870	0.0000	0.0119	0.9219
0.2200	1137.3201	5.6714	0.0000	0.0115	0.9281
0.2300	1196.4395	5.7539	0.0000	0.0112	0.9306
0.2400	1255.9189	5.8347	0.0000	0.0108	0.9293
0.2500	1315.7473	5.9137	0.0000	0.0105	0.9319
0.2600	1375.9124	5.9912	0.0000	0.0103	0.9346
0.2700	1436.4011	6.0673	0.0000	0.0100	0.9373
0.2800	1497.2051	6.1419	0.0000	0.0098	0.9440
0.2900	1558.3122	6.2152	0.0000	0.0096	0.9431
0.3000	1619.7173	6.2873	0.0000	0.0093	0.9461
0.3100	1681.4102	6.3581	0.0000	0.0091	0.9491
0.3200	1743.3799	6.4279	0.0000	0.0090	0.9523
0.3300	1805.6230	6.4965	0.0000	0.0088	0.9594
0.3400	1868.1292	6.5642	0.0000	0.0086	0.9628
0.3500	1930.8962	6.6308	0.0000	0.0094	0.9622

DPC	UT	SH	KK	RR	II
C.3600	993.9114	6.6965	C.0C0C	0.0083	0.9656
C.3700	2057.1733	6.7613	C.0C0C	0.0081	0.9691
C.3800	2120.6768	6.8252	C.0900	0.0080	C.9767
C.3900	184.4124	6.8882	C.0000	0.0079	0.9766
0.4000	2248.3772	6.9505	C.0C0C	0.0077	0.9762
C.4100	2312.5674	7.0120	0.0000	0.0076	C.980C
0.4200	2376.9771	7.0728	0.0C00	0.0075	0.9838
0.4300	2441.6018	7.1328	C.000C	0.0074	0.9878
C.4400	2506.4380	7.1921	0.0000	0.0073	0.9959
C.4500	2571.4797	7.2508	0.0000	0.0072	0.9958

PARTICLE DIAMETER OF COAL, CM	SIZE DISTRIBUTION, CM** (1)
0.0600	0.0200
0.0700	0.0400
0.0800	0.5000
0.0900	0.1000
0.1000	0.1500
0.1100	0.3750
0.1200	0.7500
0.1300	1.1200
0.1400	1.5000
0.1500	1.8750
0.1600	2.2500
0.1700	2.8200
0.1800	3.2500
0.1900	3.8800
0.2000	4.3000
0.2100	4.9500
0.2200	5.3200
0.2300	5.5000
0.2400	5.6200
0.2500	5.6500
0.2600	5.6800
0.2700	5.6700
0.2800	5.6300
0.2900	5.5500
0.3000	5.3500
0.3100	4.8000
0.3200	3.9500
0.3300	3.0500
0.3400	2.3000
0.3500	1.7500
0.3600	1.4000
0.3700	1.1000
0.3800	0.8300
0.3900	0.4600
0.4000	0.3500
0.4100	0.2500
0.4200	0.1800
0.4300	0.1200
0.4400	0.0700
0.4500	0.0300

F0 = 1.6717, W= 8.6584, U0= 394.2300
 F1= 0.0259

DPC1, CM	F1, CM** 1
0.0020	0.0000
0.0040	0.0000
0.0060	0.0000
0.0080	0.0000
0.0100	0.0000
0.0120	0.0000
0.0140	0.0000
0.0160	0.0000
0.0180	0.0000
0.0200	0.0000
0.0220	0.0000
0.0240	0.0000
0.0260	0.0000
0.0280	0.0000
0.0300	0.0000
0.0320	0.0000
0.0340	0.0000
0.0360	0.0000
0.0380	0.0000
0.0400	0.0000
0.0420	0.0000
0.0440	0.0000
0.0460	0.0000
0.0480	0.0000
0.0500	0.0000
0.0520	0.0000
0.0540	0.0000
0.0560	0.0576
0.0600	0.0841
0.0700	0.2017
0.0800	0.3911
0.0900	0.6065
0.1000	0.8859
0.1100	1.2372
0.1200	1.6409
0.1300	2.0832
0.1400	2.5667
0.1500	3.0791
0.1600	3.6045
0.1700	4.1363
0.1800	4.6311
0.1900	5.0430
0.2000	5.4172
0.2100	5.6884
0.2200	5.8689
0.2300	5.9194
0.2400	5.8615
0.2500	5.7194
0.2600	5.5003
0.2700	5.1653
0.2800	4.7696
0.2900	4.2186
0.3000	3.6099
0.3100	2.9339
0.3200	2.3286
0.3300	1.7944
0.3400	1.3921
0.3500	1.0552

DPC1,CM	F1,CM**1
0.3600	0.7936
0.3700	0.5685
C.3800	0.3873
C.3900	0.2572
C.4000	0.2251
C.4100	0.1514
C.4200	0.0966
C.4300	0.0550
C.4400	0.0260
C.4500	0.0091

RATE OF COAL COMBUSTED= 1.5768

F2 = 0.0690

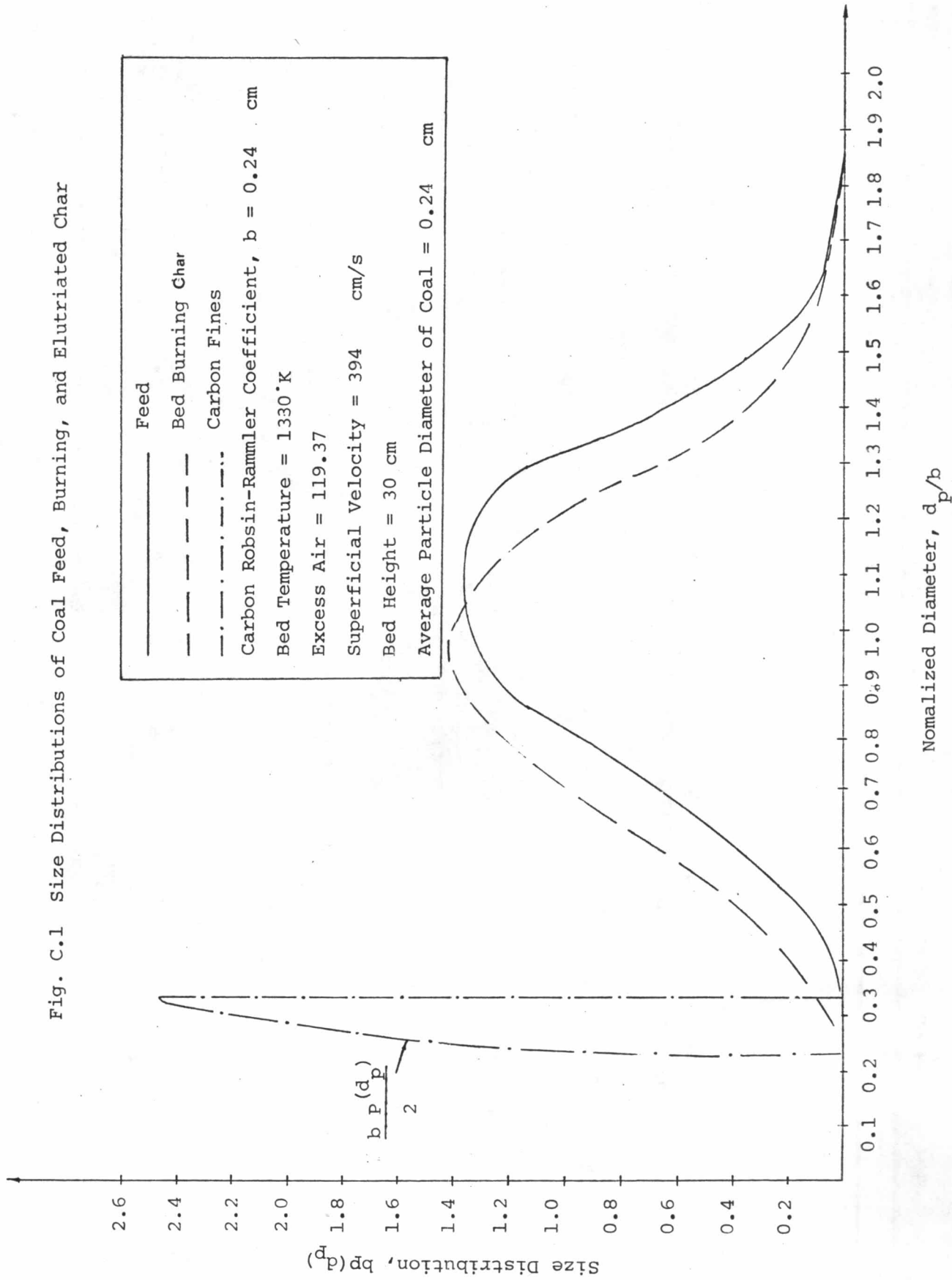
DPC2, CM	F2, CM** 1
0.0020	0.0000
0.0040	0.0000
0.0060	0.0000
0.0080	0.0000
0.0100	0.0000
0.0120	0.0000
0.0140	0.0000
0.0160	0.0000
0.0180	0.0000
0.0200	0.0000
0.0220	0.0000
0.0240	0.0000
0.0260	0.0000
0.0280	0.0000
0.0300	0.0000
0.0320	0.0000
0.0340	0.0000
0.0360	0.0000
0.0380	0.0000
0.0400	0.0000
0.0420	0.0000
0.0440	0.0000
0.0460	0.0000
0.0480	0.0000
0.0500	0.0000
0.0520	0.0000
0.0540	0.0000
0.0560	9.9063
0.0600	11.7023
0.0700	16.9619
0.0800	20.5056
0.0900	0.0000
0.1000	0.0000
0.1100	0.0000
0.1200	0.0000
0.1300	0.0000
0.1400	0.0000
0.1500	0.0000
0.1600	0.0000
0.1700	0.0000
0.1800	0.0000
0.1900	0.0000
0.2000	0.0000
0.2100	0.0000
0.2200	0.0000
0.2300	0.0000
0.2400	0.0000
0.2500	0.0000
0.2600	0.0000
0.2700	0.0000
0.2800	0.0000
0.2900	0.0000
0.3000	0.0000
0.3100	0.0000
0.3200	0.0000
0.3300	0.0000
0.3400	0.0000
0.3500	0.0000
0.3600	0.0000

DPC2,CM	F2,CM** - 1
0.3700	0.0000
0.3800	0.0000
0.3900	0.0000
0.4000	0.0000
0.4100	0.0000
0.4200	0.0000
0.4300	0.0000
0.4400	0.0000
0.4500	0.0000

RATE OF COAL COMBUSTED= 1.5766
 FO = 1.6717 F1 = 0.0259 F2 = 0.0690 EFF = 94.2203

TEMPERATURE OF AIR IN BED = 1330.0000 K
 EXCESS AIR= 119.3747

Fig. C.1 Size Distributions of Coal Feed, Burning, and Elutriated Char



Appendix D

Average Particles Diameter

```

C FIND THE AVERAGE DIAMETER OF LIGNITE PARTICLE
C DBP MAX=MAXIMUM DIAMETER OF BED PARTICLE
C DBP MIN=THE MINIMUM DIAMETER OF BED PARTICLE
C PBP(I)= SIZE DISTRIBUTION OF PARTICLE ,BETWEEN DP AND DP+D(DP)
C NM=(DBP MAX-DBP MIN)/H+1, THE NUMBER OF DATAS TO BE CALCULATE
C H=FINITE DIFFERENCE OF DBP
1  DIMENSION DBP(100),PBP(100),YFB(100)
2  DO 200 J=1,2
3  READ(5,21) DBP MAX,DBP MIN,H,NM
4  21 FORMAT(3F10.4,I10)
5  READ(5,31) (PBP(I),J=1,NM)
6  31 FORMAT(9F10.4)
7  DO 100 I=1,NM
8  DBP(I)=DBP MIN+(I-1)*H
9  YFB(I)=PBP(I)/DBP(I)
10 100 CONTINUE
11  CALL INTGT (N,NM,YFB,SUM)
12  DPA=1/SUM
13  WRITE(5,36)
14  36 FORMAT('1',T11,'FIND THE AVERAGE DIAMETER OF LIGNITE PARTICLE1)
15  WRITE (6,37)
16  37 FORMAT('0',T11,7('-----'))
17  WRITE(5,38)
18  38 FORMAT (T11,'DIAMETER OF LIGNITE PARTICLE ,CM',T45,'1',T51,'SIZE
  *DISTRIBUTION,CM**(-1)')
19  WRITE(6,39)
20  39 FORMAT(T11,7('-----'))
21  WRITE(5,41) (DBP(I),PBP(I),I=1,NM)
22  41 FORMAT(T18,F13.4,T45,'1',T58,F13.4)
23  WRITE(6,42)
24  42 FORMAT(T11,7('-----'))
25  WRITE(6,43) DBP MAX,DBP MIN,NM,H,DPA
26  43 FORMAT(T11,'MAXIMUM DIAMETER OF BED MATERIAL = ',T51,F13.4/
  * T11,'MINIMUM DIAMETER OF BED MATERIAL = ',T51,F13.4/
  * T11,'RANGE NUMBERS OF PARTICLE DIAMETER = ',T51,I4/
  * T11,'FINITE DIFFERENCE = ',T51,F13.4//
  * T11,'THE AVERAGE BED PAPTICLE DIAMETER =',T51,F13.4)
27 200 CONTINUE
28  STOP
29  END

C SUBROUTINE FOR CALCULATING INTEGRATION WITH FINITE DIFFERENCE
C SUBROUTINE INTGT(H,N,Y, SUM)
C H=FINITE DIFFERENCE OF X
C Y=FUNCTION OF X

30  SUBROUTINE INTGT(H,N,Y,SUM)
31  DIMENSION Y(N)
32  REAL ITRAP,ISIMP
33  KT1=FLJAT(N)/2.+0.6
34  KT2=V/2
35  IF(KT1-KT2) 10,40,10

C SIMPSON RULE
36  10 N1=N-1
37  YM1=Y(2)
38  DO 20 I=4,N1,2
39  YM1=YM1+Y(I)
40  20 CONTINUE
41  N2=N-2
42  YM2=Y(3)
43  DO 30 I=5,N2,2
44  YM2=YM2+Y(I)
45  30 CONTINUE
46  ISIMP=(4/3.)*Y(1)+4.*YM1+2.*YM2+Y(N)
47  SUM=ISIMP
48  GO TO 60

C TRAPEZOIDAL RULE
49  40 YMID=Y(2)
50  N1=N-1
51  DO 50 I=3,N1
52  YMID=YMID+Y(I)
53  50 CONTINUE
54  ITRAP=(H/2.)*(Y(1)+2.*YMID+Y(N))
55  SUM=ITRAP
56  RETURN
57  END

```

Appendix E

Estimation of Air-to-fuel Ratio

$$\text{Air-to-fuel Ratio} = \frac{\text{air feed rate (by weight)}}{\text{coal feed rate}}$$

$$= \frac{P A_t U_o}{RT} M_{\text{air}} F_o$$

$$= \frac{P A_t U_o}{RT F_o} M_{\text{air}}$$

where P = air pressure

A_t = bed cross section area

U_o = superficial velocity

M_{air} = molecular weight of air

R = gas constant

T = air temperature

F_o = coal feed rate

Appendix F

Estimation of Excess Air

$$\begin{aligned}
 \text{Excess Air} &= \left(\left(\frac{\text{mole of air used}}{\text{mole of carbon}} \right) / \left(\frac{\text{mole of air}}{\text{mole of carbon}} \right)_{\text{theory}} - 1 \right) \times 100 \\
 &= \frac{\left(\frac{\text{mole of air used}}{\text{mole of Carbon}} \right)_{\text{actual}}}{\left(\frac{\text{mole of oxygen} \times \frac{1}{0.21}}{\text{mole of carbon}} \right)_{\text{theory}}} - 1 \quad \times 100 \\
 &= \left(\frac{PV}{RT \times \text{mole of carbon}} \times 0.21 - 1 \right) \times 100 \\
 &= \left(0.21 \frac{PA_t U_o}{RT} \times \frac{\text{molecular weight of carbon}}{\text{weight of carbon / time}} - 1 \right) \times 100 \\
 &= \left(0.21 \frac{PA_t U_o M_c}{RTYF_o} - 1 \right) \times 100
 \end{aligned}$$

where P = air pressure

A_t = bed cross section area

U_o = superficial velocity

M_c = carbon molecular weight

R = gas constant

T = air temperature

F_o = coal feed rate

Y = percentage of carbon in coal

Appendix G

Fuel Mixture Feed Rate



The fuel mixture in these experiments mentioned before is fed with screw feeder which is controlled by a variable speed motor.

The fuel mixture feed rates and the speed numbers are shown in Table G.1 and Table G.2

Table G.1

Fuel Mixture Feed Rate

Ratio of limestone and lignite in fuel mixture = 1:1 by weight

\bar{d}_p of coal = 1 mm.

\bar{d}_p of limestone = 0.5 mm.

motor speed No.	fuel mixture feed rate	
	gm/sec	gm/min.
1	1.5	87.3
2	2.0	118.1
3	2.7	163.4
4	3.6	214.0
5	4.6	277.1
6	5.2	311.3

\bar{d}_p = average diameter of particles

Table G.2

Fuel Mixture Feed Rate

Ratio of limestone and lignite in fuel mixture = 1:1 by weight

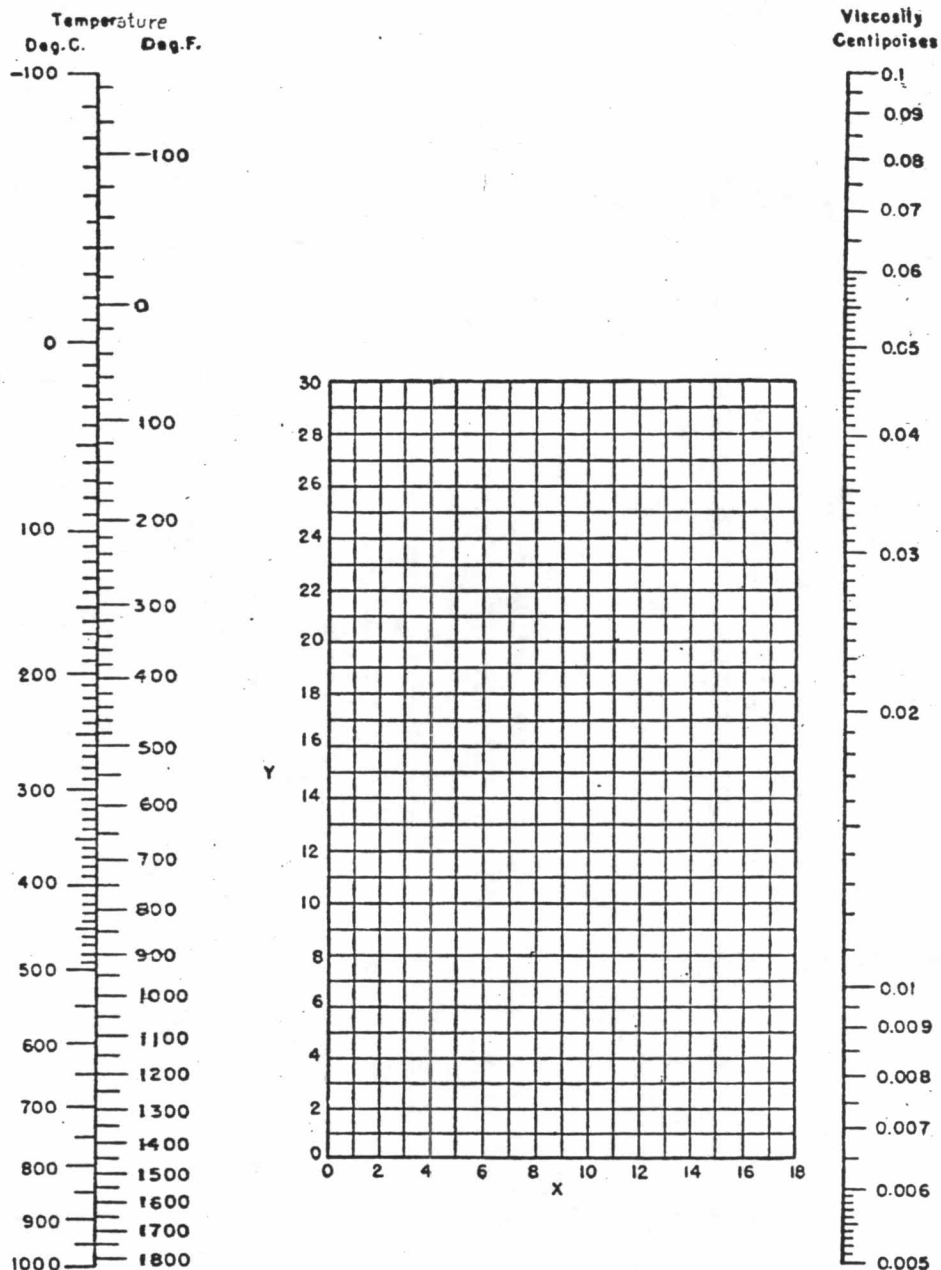
\bar{d}_p of coal = 2.4 mm.

\bar{d}_p of limestone = 1 mm.

motor speed No.	fuel mixture feed rate	
	gm/sec	gm/min.
1	1.2	71.6
2	1.6	98.2
3	2.4	144.6
4	3.3	200.6
5	4.3	259.4
6	5.1	303.3
7	5.4	325.6

\bar{d}_p = the average diameter of particles

Appendix H
Viscosity³⁷

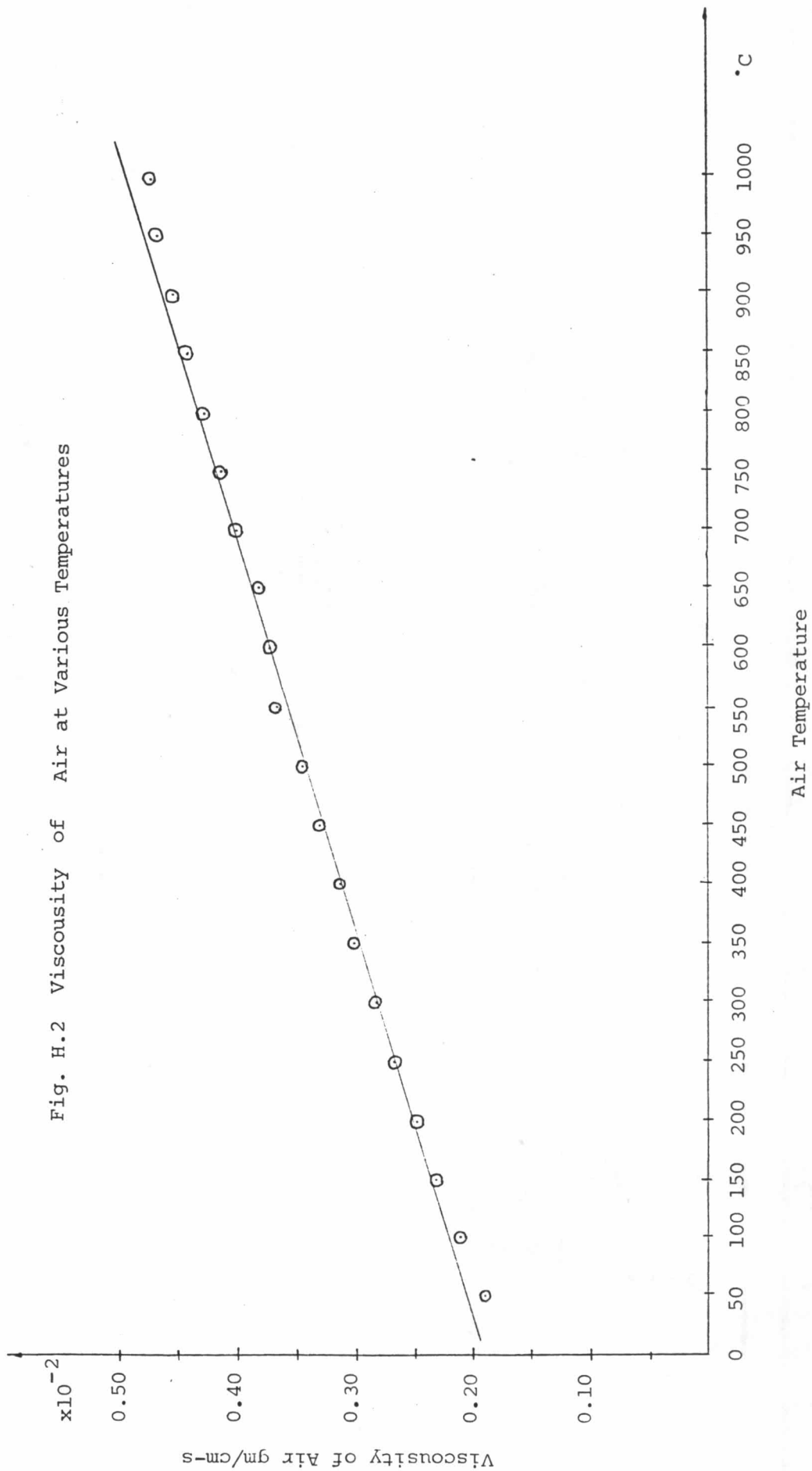


Viscosities of gases at 1 atm. For coordinates, see

Coordinates for use with Fig.															
No.	Gas	X	Y	No.	Gas	X	Y	No.	Gas	X	Y	No.	Gas	X	Y
1	Acetic acid	7.7	14.3	15	Chloroform	8.9	15.7	29	Freon-113	11.3	14.0	43	Nitric oxide	10.9	20.5
2	Acetone	5.9	13.0	16	Cyanogen	9.2	15.2	30	Helium	10.9	20.5	44	Nitrogen	10.6	20.0
3	Acetylene	9.5	14.9	17	Cyclohexane	9.2	12.0	31	Hexane	8.6	11.8	45	Nitrosyl chloride	8.0	17.0
4	Air	11.0	20.0	18	Ethane	9.1	14.5	32	Hydrogen	11.2	12.4	46	Nitrous oxide	8.8	19.0
5	Ammonia	5.4	16.0	19	Ethyl acetate	5.5	13.2	33	3H ₂ + 1N ₂	11.2	17.2	47	Oxygen	11.0	21.3
6	Argon	10.5	22.4	20	Ethyl alcohol	9.2	14.2	34	Hydrogen bromide	5.5	20.9	48	Pentane	7.0	12.9
7	Benzene	5.5	13.2	21	Ethyl chloride	8.5	15.6	35	Hydrogen chloride	5.8	18.7	49	Propane	9.7	12.9
8	Bromine	5.9	19.2	22	Ethyl ether	8.9	13.0	36	Hydrogen cyanide	9.5	14.9	50	Propyl alcohol	8.4	13.4
9	Butene	9.2	13.7	23	Ethylene	9.5	15.1	37	Hydrogen iodide	9.0	21.3	51	Propylene	9.0	15.8
10	Butylene	5.9	13.0	24	Fluorine	7.3	23.8	38	Hydrogen sulfide	8.6	18.0	52	Sulfur dioxide	9.6	17.0
11	Carbon dioxide	9.5	18.7	25	Freon-11	10.6	15.1	39	Iodine	9.0	18.4	53	Toluene	8.6	12.4
12	Carbon disulfide	5.0	16.0	26	Freon-12	11.1	16.0	40	Mercury	5.3	22.9	54	2, 3, 3-Trimethylbutane	9.5	10.5
13	Carbon monoxide	11.0	20.0	27	Freon-21	10.8	15.3	41	Methane	9.9	15.5	55	Water	6.0	16.0
14	Chlorine	9.0	18.4	28	Freon-22	10.1	17.0	42	Methyl alcohol	8.5	15.6	56	Xenon	9.3	22.0

Fig H.1 Viscosity of Gases

Fig. H.2 Viscosity of Air at Various Temperatures



Appendix I

Diffusivity³⁸

For low pressure of binary gas mixture and high temperature, the diffusivity of oxygen in air is expressed in the following equation^{32, 33}

$$D_{AB} = \frac{0.001858 T^{3/2} (1/M_A + 1/M_B)^{1/2}}{P \sigma_{AB}^2 \Omega_D} \text{ cm}^3/\text{s} \quad \dots\dots (1)$$

where D_{AB} = Mutual diffusion coefficient species A in binary of A and B, cm^2/s

P = total pressure = 1 atm.

M_A = Molecular weight of A

here, the molecular weight of oxygen = 32

M_B = Molecular weight of B,

here, the molecular weight of air = 28.84

σ = Lennard-Jones potential parameter, Å (see Table

Table 1.1 LENNARD-JONES POTENTIAL PARAMETERS

Molecule	Compound	σ , Å	ϵ/k °K
A	Argon	3.542	93.3
He	Helium†	2.551	10.22
Kr	Krypton	3.655	178.9
Ne	Neon	2.820	32.8
Xe	Xenon	4.082	206.9
Air	Air	3.711	78.6
Br ₂	Bromine	4.296	507.9
CCl ₄	Carbon tetrachloride	5.947	322.7
CF ₄	Carbon tetrafluoride	4.662	134.0
CHCl ₃	Chloroform	5.389	340.2
CH ₂ Cl ₂	Methylene chloride	4.898	356.3
CH ₃ Br	Methyl bromide	4.118	449.2
CH ₃ Cl	Methyl chloride	4.182	350.0
CH ₃ OH	Methanol	3.626	481.8
CH ₄	Methane	3.758	148.6
CO	Carbon monoxide	3.690	91.7
COS	Carbonyl sulfide	4.130	336.0
CO ₂	Carbon dioxide	3.941	195.2
CS ₂	Carbon disulfide	4.483	467.0
C ₂ H ₂	Acetylene	4.033	231.8
C ₂ H ₄	Ethylene	4.163	224.7
C ₂ H ₆	Ethane	4.443	215.7
C ₂ H ₅ Cl	Ethyl chloride	4.898	300.0
C ₂ H ₅ OH	Ethanol	4.530	362.6
C ₂ N ₂	Cyanogen	4.361	348.6
CH ₃ OCH ₃	Methyl ether	4.307	395.0
CH ₂ CHCH ₃	Propylene	4.678	298.9
CH ₃ CCH	Methylacetylene	4.761	251.8
C ₃ H ₆	Cyclopropane	4.807	248.9
C ₃ H ₈	Propane	5.118	237.1
<i>n</i> -C ₃ H ₇ OH	<i>n</i> -Propyl alcohol	4.549	576.7
CH ₃ COCH ₃	Acetone	4.600	560.2
CH ₃ COOCH ₃	Methyl acetate	4.936	469.8
<i>n</i> -C ₄ H ₁₀	<i>n</i> -Butane	4.687	531.4
<i>iso</i> -C ₄ H ₁₀	Isobutane	5.278	330.1
C ₂ H ₅ OC ₂ H ₅	Ethyl ether	5.678	313.8
CH ₃ COOC ₂ H ₅	Ethyl acetate	5.205	521.3
<i>n</i> -C ₅ H ₁₂	<i>n</i> -Pentane	5.784	341.1
C(CH ₃) ₄	2,2-Dimethylpropane	6.464	193.4
C ₆ H ₆	Benzene	5.349	412.3
C ₆ H ₁₂	Cyclohexane	6.182	297.1
<i>n</i> -C ₆ H ₁₄	<i>n</i> -Hexane	5.949	399.3
Cl ₂	Chlorine	4.217	316.0
F ₂	Fluorine	3.357	112.6
HBr	Hydrogen bromide	3.353	449.0
HCN	Hydrogen cyanide	3.630	569.1
HCl	Hydrogen chloride	3.339	344.7
HF	Hydrogen fluoride	3.148	330.0
HI	Hydrogen iodide	4.211	288.7
H ₂	Hydrogen	2.827	59.7
H ₂ O	Water	2.641	809.1
H ₂ O ₂	Hydrogen peroxide	4.196	289.3
H ₂ S	Hydrogen sulfide	3.623	301.1
Hg	Mercury	2.969	750.0
I ₂	Iodine	5.160	474.2
NH ₃	Ammonia	2.900	558.3
NO	Nitric oxide	3.492	116.7
NOCl	Nitrosyl chloride	4.112	395.3
N ₂	Nitrogen	3.798	71.4
N ₂ O	Nitrous oxide	3.828	232.4
O ₂	Oxygen	3.467	106.7
PH ₃	Phosphine	3.981	251.5
SF ₆	Sulfur hexafluoride	5.128	222.1
SO ₂	Sulfur dioxide	4.112	335.4
SnBr ₄	Stannic bromide	6.388	563.7
UF ₆	Uranium hexafluoride	5.967	236.8

† Calculated from quantum-mechanical formulas.

σ_{AB} = Lennard-Jones potential parameter of the mixture
of A and B

Ω_D = collision integral (see Table 1.2 and Fig 1.1)

T = gas temperature

The value of σ_{AB} can be calculated from

$$\sigma_{AB} = \frac{1}{2} (\sigma_A + \sigma_B) \quad \dots\dots (2)$$

A = oxygen

B = air

hence, $\sigma_{AB} = \frac{1}{2} (3.711 + 3.467)$
 $= 3.589$ for binary of oxygen and air $\dots\dots (3)$

Using the Table 1.1, the value of ϵ/k can be found, and
the value of ϵ/k of the binary gas mixture can be calculated by

$$\frac{\epsilon_{AB}}{K} = \left(\frac{\epsilon_A}{K} \cdot \frac{\epsilon_B}{K} \right)^{\frac{1}{2}} \quad \dots\dots (4)$$

Knowing the value of ϵ_{AB}/K , the collision integral can be
found from Table 1.2

Table 1.2 VALUES OF THE COLLISION INTEGRAL Ω_D BASED ON THE LENNARD-JONES POTENTIAL†

kT/ϵ	Ω_D^\ddagger	kT/ϵ	Ω_D	kT/ϵ	Ω_D
0.30	2.662	1.65	1.153	4.0	0.8836
0.35	2.476	1.70	1.140	4.1	0.8788
0.40	2.318	1.75	1.128	4.2	0.8740
0.45	2.184	1.80	1.116	4.3	0.8694
0.50	2.066	1.85	1.105	4.4	0.8652
0.55	1.966	1.90	1.094	4.5	0.8610
0.60	1.877	1.95	1.084	4.6	0.8568
0.65	1.798	2.00	1.075	4.7	0.8530
0.70	1.729	2.1	1.057	4.8	0.8492
0.75	1.667	2.2	1.041	4.9	0.8456
0.80	1.612	2.3	1.026	5.0	0.8422
0.85	1.562	2.4	1.012	6	0.8124
0.90	1.517	2.5	0.9996	7	0.7896
0.95	1.476	2.6	0.9878	8	0.7712
1.00	1.439	2.7	0.9770	9	0.7556
1.05	1.406	2.8	0.9672	10	0.7424
1.10	1.375	2.9	0.9576	20	0.6640
1.15	1.346	3.0	0.9490	30	0.6232
1.20	1.320	3.1	0.9406	40	0.5960
1.25	1.296	3.2	0.9328	50	0.5756
1.30	1.273	3.3	0.9256	60	0.5596
1.35	1.253	3.4	0.9186	70	0.5464
1.40	1.233	3.5	0.9120	80	0.5352
1.45	1.215	3.6	0.9058	90	0.5256
1.50	1.198	3.7	0.8998	100	0.5130
1.55	1.182	3.8	0.8942	200	0.4644
1.60	1.167	3.9	0.8888	400	0.4170

† From J. O. Hirschfelder, C. F. Curtiss, and R. B. Bird, "Molecular Theory of Gases and Liquids," John Wiley & Sons, Inc., New York, 1954

‡ Hirschfelder uses the symbols T^* for kT/ϵ and $\Omega^{(1,1)*}$ in place of Ω_D .

Now the value of ε_{AB}/K of oxygen in air = ε_{AB}/K

$$= \frac{1}{2} \left(\frac{\varepsilon_A}{k} \cdot \frac{\varepsilon_B}{k} \right)^{\frac{1}{2}}$$

$$= \frac{1}{2} (78.6 + 106.7)^{\frac{1}{2}}$$

$$= 91.58 \quad \dots\dots (5)$$

The collision integral is also the function of temperature. The collision integral of oxygen in air as the function of temperature is shown in Fig. 1.1

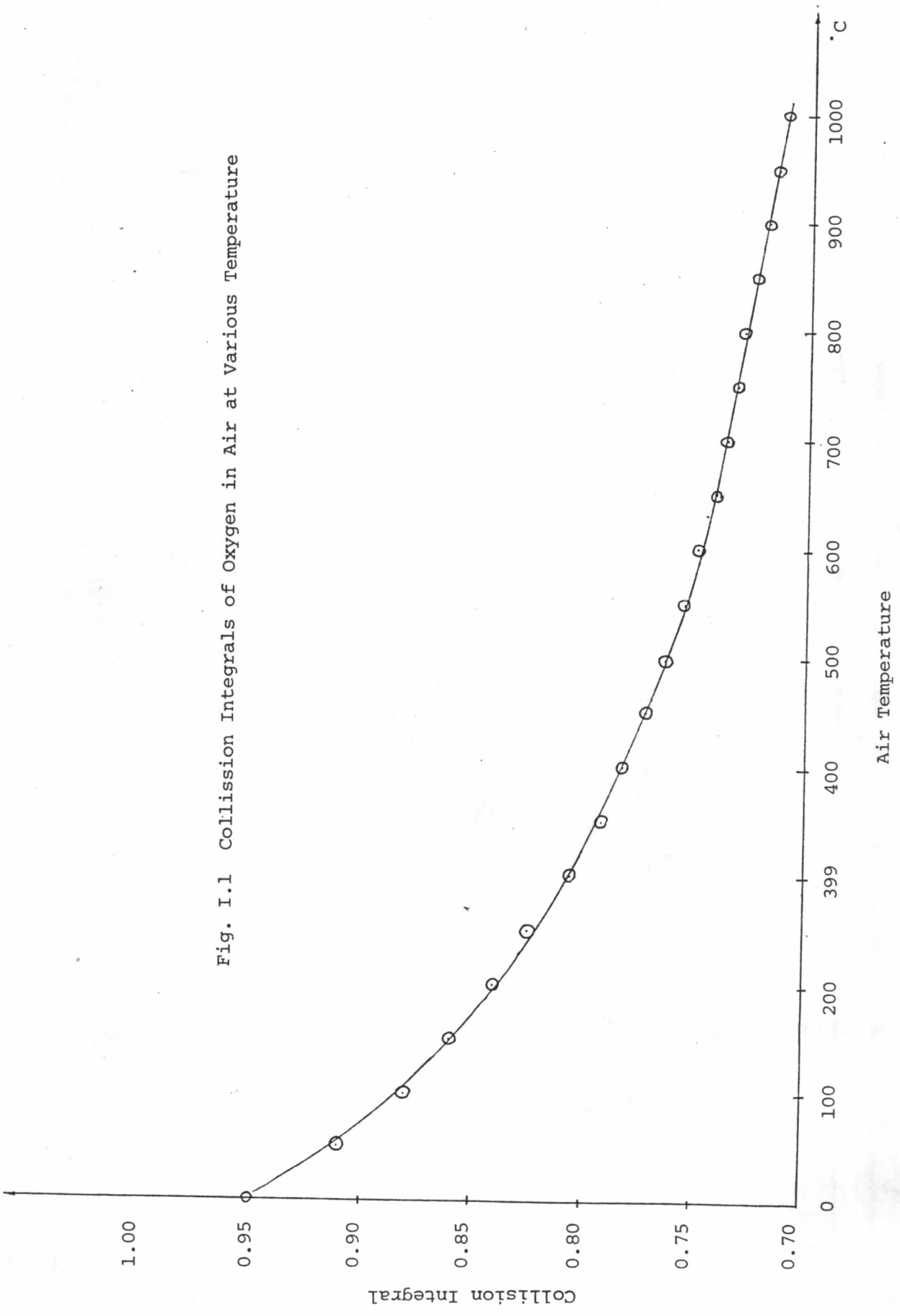
Hence, the diffusion of oxygen in air can be expressed in the following equation

$$D_{AB} = 3.73 \times 10^{-5} \frac{T^{3/2}}{\Omega_{AB}} \quad \dots\dots (6)$$

where A = oxygen

B = air

Fig. I.1 Collision Integrals of Oxygen in Air at Various Temperature



Appendix J

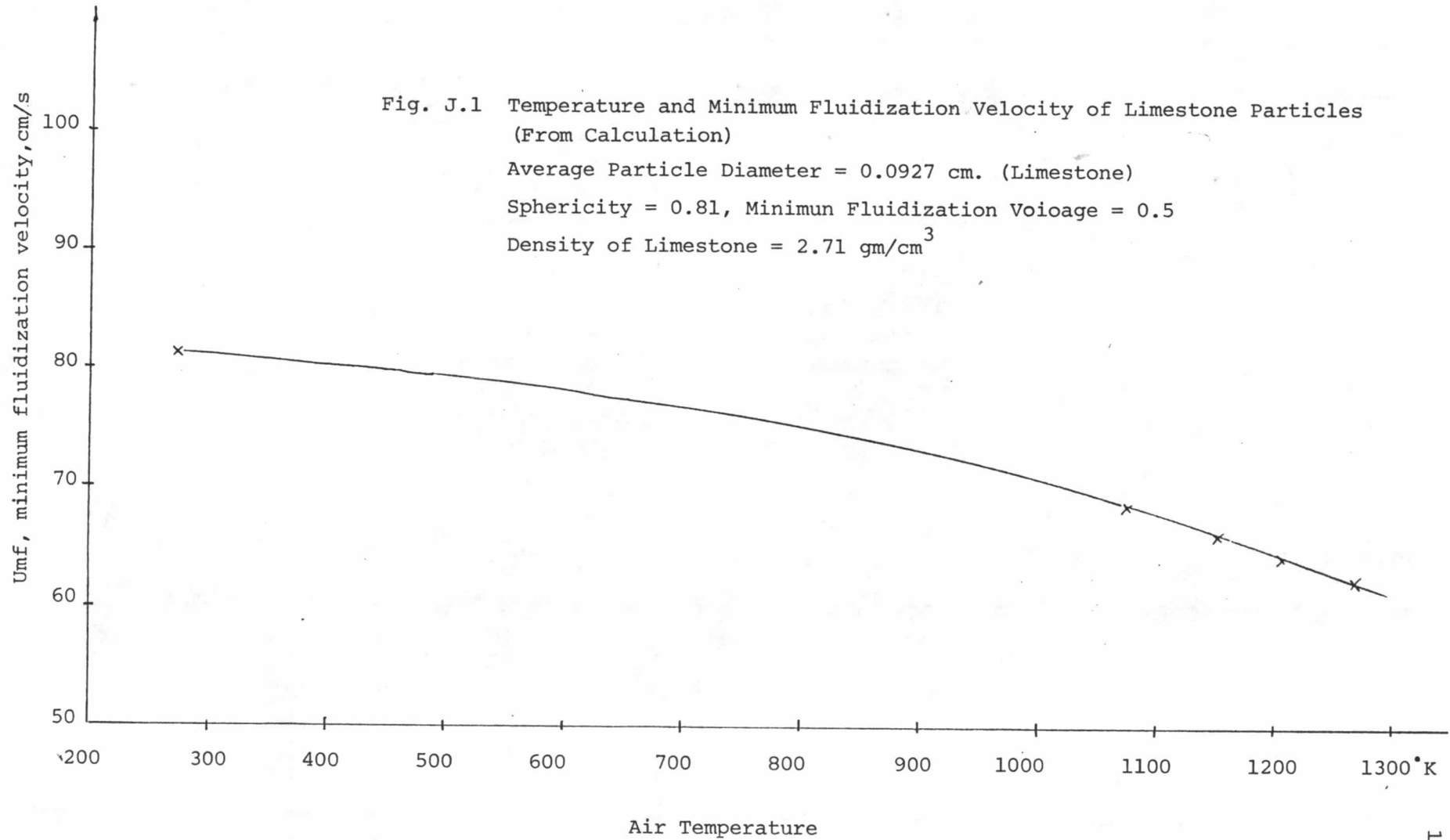
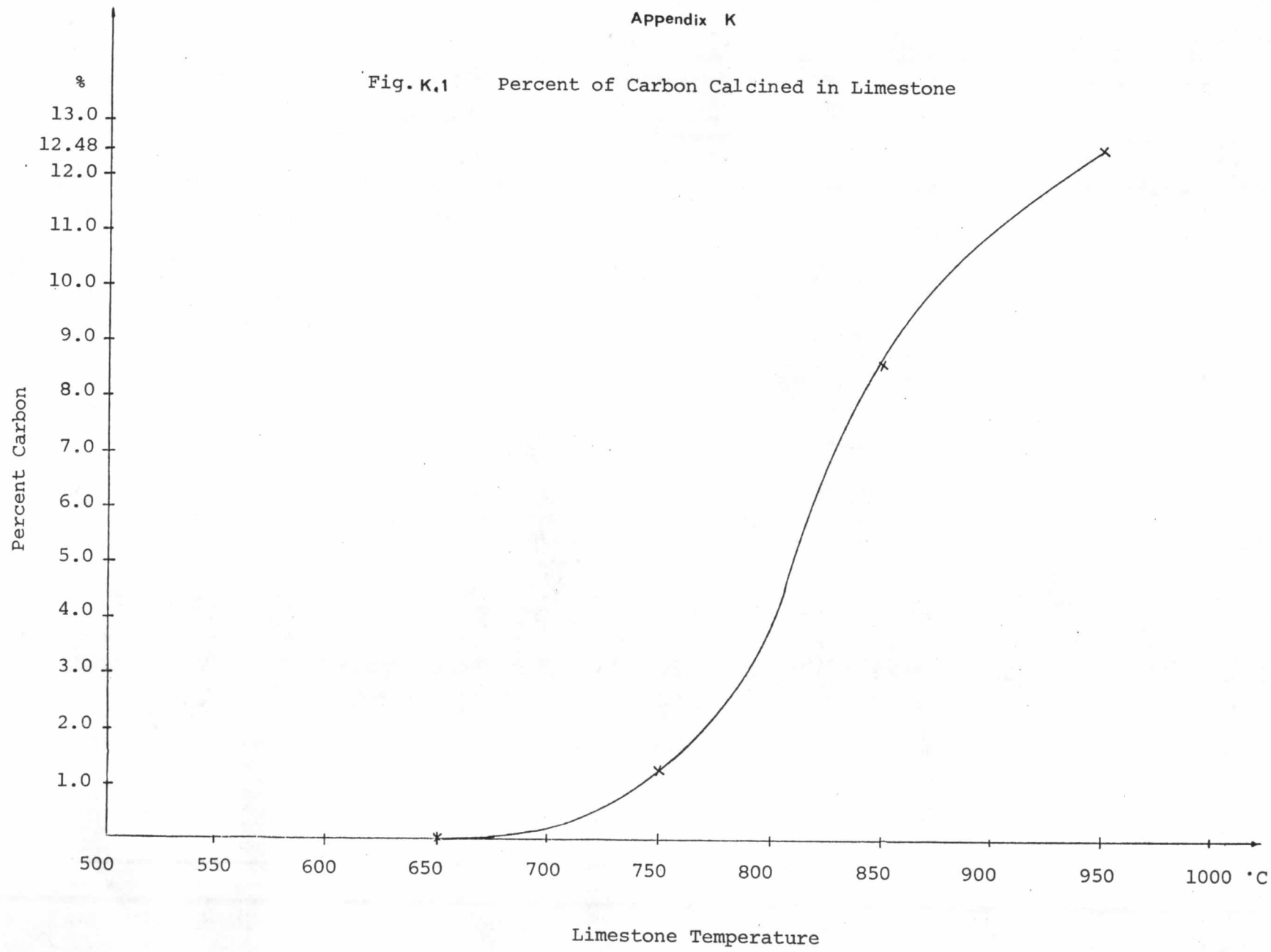


Fig. K.1 Percent of Carbon Calcined in Limestone



Appendix L

Approximate Heat Balance in the Combustor

The heat balance in the combustor can be estimated followingly.

Now an example of Calculation is shown

From data No. 3 of Table 6.1

average bed temperature = $\bar{T}_B = 995^\circ \text{C}$

average gas temperature = $\bar{T}_G = 418^\circ \text{C}$

inlet limestone temperature = 36°C

inlet lignite temperature = 30°C

inlet air temperature = 35°C

lignite feed rate = $F_{ocl} = 1.13 \text{ gm/s}$

limestone feed rate = $F_{ol} = 1.13 \text{ gm/s}$

specific heat of coal ash = $0.31 \text{ cal/gm}^\circ \text{C}$

specific heat of limestone = $0.217 \text{ cal/gm}^\circ \text{C}$

specific heat of air at \bar{T}_G of $418^\circ \text{C} = 0.26 \text{ cal/gm}^\circ \text{C}$

heating value of lignite = 4166 cal/gm

air density at STP = $\rho_g = 1.293 \times 10^{-3} \text{ gm/cm}^3$

inlet air velocity = 376 an/s

ash content in lignite = 10%

bed diameter = $.15 \text{ cm}$

Inlet

If carbon is all combusted, heat will

be generated = $4166 \times 1.13 \text{ cal/s}$

= 4708 cal/s

Outlet

Due to the specific heat of CO, CO₂, O₂, N₂ and air at the same temperature of 418°C is approximately the same.

Let the specific heat of outlet gas = $0.26 \frac{\text{cal}}{\text{gn}^\circ\text{C}}$

heat accepted by outlet gas and limestone

$$= (418-35) \times 1.293 \times 10^{-3} \times \frac{273}{(273+418)} \times 376 \times \frac{1}{4} \times 15 \times 15 \\ \times 0.26 + (995-30) \times 1.13 \times 0.217 \\ = 3617 \quad \text{cal/s}$$

heat accepted by coal ash

$$= 1.13 \times 0.10 \times (995-30) \times 0.31 \\ = 34$$

hence heat accepted by outlet solids and gas

$$= 365 \quad \text{cal/s}$$

the calculated outlet heat is less than the inlet potential

$$\text{heat} = 4708 - 3651 = 1057 \quad \text{cal/s}$$

This uncalculated heat can be lost by other way such as the conduction of heat through the refractory and stainless steel of combustor, heat consumed by the endothermic reaction of



Appendix M

Some Constant Values

density of limestone	$\pm 2.75 \text{ gm/cm}^3$
density of lignite	$= 1.50 \text{ gm/cm}^3$
density of air at STP	$= 1.293 \times 10^{-3} \text{ gm/cm}^3$
universal gas constant	$= 82.06 \text{ atm-cm}^3/\text{gmole} \cdot \text{K}$ $= 1.987 \text{ cal/gmole} \cdot \text{K}$
gravitational acceleration	$= 980 \text{ cm/s}^2$
sphericity of lime	$= 0.77$
void fraction of limestone of 0.5 mm. in average particles diameter	$= 0.513$
void fraction of limestone of 1.0 mm in average particles diameter	$= 0.45$
inlet oxygen concentration	$= 1.28 \times 10^{-3} \text{ kmole/m}^3$ at 1173°K and 1 atm. $= 2.18 \times 10^{-6} \text{ gmole/cm}^3$ at 1173°K and 1 atm.

Appendix N

Output from Computer Simulation

From chapter 7 the graphs of effects of operating variables on carbon combustibles loss are plotted from the following data:

N.1 The Effect of Superficial Velocity on Carbon Combustibles Loss

Let $\frac{W}{W_B} = 0.1$, $C_{Ab} = 91.3\%$, C_{Ao} , $\bar{T}_B = \text{constant} = 1173^\circ\text{K}$, $F_o = \text{constant} = 1.13 \text{ gm/s}$
 $\bar{d}_p \text{ limestone} = 0.1 \text{ cm}$, $\bar{d}_p \text{ lignite} = 0.24 \text{ cm}$, $\rho_g = 0.28 \times 10^{-3} \text{ gm/cm}^3$, $\mu_g = 4.6 \times 10^{-4} \text{ gm/cm-s}$
 $D_A = 2.08 \text{ cm}^2/\text{s}$, $C_{Ao} = 2.18 \times 10^{-6} \text{ gmole/cm}^3$, $U_{mf} = 65 \text{ cm/s}$ at 1173°K , $L_f = 30 \text{ cm}$
 $d_t = 15 \text{ cm}$, limestone : lignite = 1:1 by weight

NO	U_o cm/s	X	C_{Ap} gmole/cm ³	\bar{t} s	W gm	F_1 gm/s	F_2 gm/s	η %	EXC %
1	200	3.7	1.98×10^{-6}	6.2	7.1	0.17	0.03	82.7	91.5
2	250	2.9	1.97×10^{-6}	6.1	6.9	0.13	0.03	86.2	139.4
3	300	1.8	1.95×10^{-6}	6.0	6.8	0.10	0.03	89.0	187.3
4	350	1.7	1.948×10^{-6}	5.9	6.7	0.05	0.04	92.2	235.1
5	400	1.5	1.94×10^{-6}	5.8	6.7	0.02	0.05	94.2	283.0
6	450	1.4	1.92×10^{-6}	5.7	6.4	0.00	0.07	93.7	330.9

W = weight of burning char in the bed \bar{t} = average residence time of burning char particles in bed
 W_B = weight of bed material η = combustion efficiency
 C_{Ab} = concentration of oxygen in bubble EXC = excess air
 C_{Ao} = concentration of oxygen in inlet air $\bar{d}_p \text{ limestone}$ = average particles diameter of limestone
 C_{Ap} = concentration of oxygen in particulate phase $\bar{d}_p \text{ coal}$ = average particles diameter of coal
 \bar{T}_B = average bed temperature ρ_g = density of air
 F_o = coal feed rate μ_g = viscosity of air
 F_1 = overflow rate D_A = diffusivity of oxygen in air
 F_2 = elutriation rate U_{mf} = minimum fluidizing velocity
 U_o = superficial velocity L_f = expanded bed height
X = number of transfer unit d_t = bed diameter

N. 2 The Effect of Bed Temperature on Carbon Combustibles Loss

Let $\frac{W}{W_B} = 0.1 \%$, $C_{Ab} = 91.3 \%$, C_{Ao} , $U_o = \text{constant} = 350 \text{ cm/s}$, $F_o = 1.13 \text{ gm/s}$

$L_f = 30 \text{ cm}$, $d_t = 15 \text{ cm}$, $\bar{d}_p \text{ limestone} = 0.1 \text{ cm}$, $\bar{d}_p \text{ lignite} = 0.74 \text{ cm}$

limestone: lignite = 1:1 by weight

NO.	\bar{T}_B °K	D_A cm ² /s	μ_g gm/cm-s	ρ_g gm/cm ³	U_{mf} cm/s	X	C_{Ao} gmole/cm ³	C_{Ab} gmole/cm ³	C_{Ap} gmole/cm ³	W gm	F_1 gm/s	F_2 gm/s	η %	EXC. %
1	1273	2.39	4.7×10^{-4}	2.6×10^{-4}	62	1.63	2.0×10^{-6}	1.8×10^{-6}	1.8×10^{-6}	6.6	0.04	0.04	93.3	209
2	1223	2.24	4.7×10^{-4}	2.7×10^{-4}	64	1.67	2.1×10^{-6}	1.9×10^{-6}	1.9×10^{-6}	6.6	0.05	0.04	92.0	221
3.	1173	2.09	4.5×10^{-4}	2.8×10^{-4}	65	1.71	2.2×10^{-6}	2.0×10^{-6}	2.0×10^{-6}	6.5	0.06	0.04	91.0	235
4	1123	1.95	4.4×10^{-4}	2.9×10^{-4}	67	1.76	2.3×10^{-6}	2.1×10^{-6}	2.0×10^{-6}	6.5	0.07	0.04	90.5	250
5	1073	1.81	4.3×10^{-4}	3.1×10^{-4}	68	1.78	2.4×10^{-6}	2.2×10^{-6}	2.1×10^{-6}	6.5	0.09	0.04	89.1	266
6	1023	1.67	4.1×10^{-4}	3.2×10^{-4}	70	1.83	2.5×10^{-6}	2.3×10^{-6}	2.2×10^{-6}	6.4	0.10	0.04	88.4	274

N.3 The Effect of Coal Feed Rate on Carbon Combustibles Loss

Let $\frac{W}{W_B} = 0.1$, $C_{Ab} = 91.3\% C_{Ao}$, $\bar{T}_B = \text{constant} = 1173^\circ\text{K}$, $U_o = \text{constant} = 400 \text{ cm/s}$

\bar{d}_p limestons = 0.1 cm, \bar{d}_p lignite = 0.24 cm, $\rho_g = 0.28 \times 10^{-3}$, $\mu_g = 4.6 \times 10^{-4} \text{ gm/cm-s}$, $d_t = 15 \text{ cm}$

$D_A = 2.08 \text{ cm}^2/\text{s}$, $C_{Ao} = 2.18 \times 10^{-6} \text{ gmole/cm}^3$, $U_{mf} = 6.5 \text{ cm/s}$ at 1173°K , $L_f = 30 \text{ cm}$

limestone : lignite = 1:1 by weight

NO.	F_o gm/s	X	C_{Ap} gmole/cm ³	W gm	\bar{t} s	F_1 gm/s	F_2 gm/s	η %	EXC %
1	0.5	1.5	1.94×10^{-6}	2.87	13.0	0.00	0.03	94.8	767
2	1.0	"	"	5.77	6.6	0.00	0.05	95.4	333
3	1.5	"	"	6.56	4.4	0.43	0.03	69.3	188
4	2.0	"	"	"	3.3	0.99	0.01	50.4	117
5	2.5	"	"	"	2.6	1.53	0.00	40.0	73
6	3.0	"	"	"	2.2	2.04	0.00	33.5	44

Bibliography

Mr. Nuttawut Wacharakuldiloke graduated with a bachelor's degree in Chemical Engineering from the Department of Chemical Engineering, Faculty of Engineering, Chulalongkorn University in 1980. He is now working at the National Fertilizer Corp. Ltd.

