#### **CHAPTER V**

# THERMODYNAMIC FUNCTIONS FROM SELECTED EQUATIONS OF STATE

#### Derived Functions Of EOS.

Expressions for enthalpy departure, entropy departure, and component fugacity coefficient of a mixture are derived below from Equations (3-10), (3-13) and (3-17) for the five equations of state discussed in the previous sections. The thermodynamic property expressions for a pure substance are not separately derived here, because the mixture expressions also apply to pure substances.

#### 5.1 SRK Equation Of State.

#### 5.1.1 Enthalpy Departure.

Expressions for isothermal enthalpy departure for the Soave equations are derived from Equations (3-10) and (4-1).

$$\frac{H-H^*}{RT} = Z - 1 + \frac{1}{RT} \int_{\infty}^{V} \left[ T \left( \frac{\partial P}{\partial T} \right)_{V} - P \right] dV.$$
(3-10)

Equations of state provide the P-V-T relation required to evaluate the right side of the above equation. The values of the ideal gas state enthalpy,  $H^*$ , in the left side of Equation (3-10) for pure substances, and can be calculated from Equation (3-11) for mixtures.

Differentiating Equation (4-1) with respect to T, at constant , and multiplying by T, give

$$T\left(\frac{\partial P}{\partial T}\right)_{V} = \frac{RT}{V-b} - \frac{T}{V(V+b)}\left(\frac{da}{dT}\right).$$
(5-1)

The integrands in Equation (3-10) then becomes

$$T\left(\frac{\partial P}{\partial T}\right)_{V} - P = \frac{1}{V(V+b)} \left[a - T\left(\frac{da}{dT}\right)\right].$$
(5-2)

Combining Equation (3-10) with (5-2) and then integrating gives

$$\frac{H-H^*}{RT} = Z - 1$$
$$-\frac{1}{bRT} \left[ a - T \left( \frac{da}{dT} \right) \right] \ln \left( 1 + \frac{b}{V} \right)$$
(5-3)

Making use of (a/bRT) = (A/B) and (b/V) = (B/Z) gives

$$\frac{H-H^*}{RT} = Z-1$$

$$-\frac{A}{B}\left[1-\frac{T}{a}\left(\frac{da}{dT}\right)\right]\ln\left(1+\frac{B}{Z}\right).$$
(5-4)

Differentiating Equation (4-7) with respect to T, at constant compositions gives

$$\frac{da}{dT} = \sum_{i}^{N} \sum_{j}^{N} x_{i} x_{j} \left[ a_{i}^{0.5} \left( \frac{da_{j}^{0.5}}{dT} \right) + a_{j}^{0.5} \left( \frac{da_{i}^{0.5}}{dT} \right) \right] (1 - k_{ij})$$
(5-5)

$$= 2\sum_{i}^{N}\sum_{j}^{N}x_{i}x_{j}a_{i}^{0.5}\left(\frac{da_{j}^{0.5}}{dT}\right)(1-k_{ij}).$$
(5-6)

The temperature derivative of  $a_j^{0.5}$  is obtained by combining Equations (4-8) and (4-10) and (4-11) and the differentiating the resulting equation with respect to T as follows:

$$\frac{da_{j}^{0.5}}{dT} = \frac{d}{dT} \left\{ a_{ci}^{0.5} \left[ 1 + m_{j} \left( 1 - T_{rj}^{0.5} \right) \right] \right\}$$
$$= -\frac{1}{2T} m_{j} \left( a_{cj} T_{rj} \right)^{0.5}.$$
(5-7)

Combining Equation (5-6) and (5-7) gives

$$T\left(\frac{da}{dT}\right) = -\sum_{i}^{N} \sum_{j}^{N} x_{i} x_{j} m_{j} \left(a_{i} a_{cj} T_{rj}\right)^{0.5} \left(1 - k_{ij}\right)$$
(5-8)

where  $m_j$  is given by Equation (4-11) and  $a_{cj}$  is given by Equation (4-9). Combining Equation (5-8) with Equations (5-3) and (5-4) separately gives

$$\frac{H-H^*}{RT} = Z - 1 - \frac{1}{bRT} \left[ a + \sum_{i}^{N} \sum_{j}^{N} x_i x_j m_j \right]$$

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$$\left(a_{i}a_{cj}T_{rj}\right)^{0.5}\left(1-k_{ij}\right)\left[\ln\left(1+\frac{b}{V}\right)\right]$$
(5-9)

or

$$\frac{H-H^{*}}{RT} = Z - 1 - \frac{A}{B} \left[ 1 + \frac{1}{a} \sum_{i}^{N} \sum_{j}^{N} x_{i} x_{j} m_{j} \right] \\ \cdot \left( a_{i} a_{cj} T_{jj} \right)^{0.5} \left( 1 - k_{ij} \right) \left[ \ln \left( 1 + \frac{B}{Z} \right) \right].$$
(5-10)

# 5.1.2 Entropy Departure.

Expressions for isothermal entropy departure are derived from Equation (3-13), which is

$$\frac{S-S_0^*}{R} + \ln\frac{P}{P_0} = \ln Z + \int_{\infty}^{\nu} \left[\frac{1}{R}\left(\frac{\partial P}{\partial T}\right)_{\nu} - \frac{1}{V}\right] dV.$$
(5-11)

As is the case of enthalpy departure derivations. Equation (4-1) provides the P-V-T relation required to evaluate the right side of Equation (3-13) for the Redlich-Kwong, the Wilson and the Soave equations. Thus, Equation (5-1), which is based on Equation (4-1), is also valid for the entropy departure expressions. Dividing Equation (5-1) through by RT and substracting /V from both sides of the equation gives the integrand in Equation (3-13):

$$\frac{1}{R} \left( \frac{\partial P}{\partial T} \right)_{V} - \frac{1}{V} = \frac{b}{V(V-b)} - \frac{1}{RTV(V+b)} \left( T \frac{da}{dT} \right).$$
(5-12)

Combining Equation (3-13) with (5-12) and integrating the resulting equation gives

$$\frac{S-S_0^*}{R} + \ln\frac{P}{P_0} = \ln\left(Z - \frac{Pb}{RT}\right) + \frac{1}{bRT}\left(T\frac{da}{dT}\right)\ln\left(1 + \frac{b}{V}\right).$$
(5-13)

Making use of (a/bRT) = (A/B), (b/V) = (B/Z), and Pb/RT = B gives

$$\frac{S-S_0^*}{R} + \ln\frac{P}{P_0} = \ln(Z-B) + \frac{A}{B} \left[\frac{T}{a}\frac{da}{dT}\right] \ln\left(1 + \frac{B}{Z}\right).$$
(5-14)

Combination of Equations (4-7) and (5-8) with Equations (5-13) and (5-14) gives the following Soave entropy departure expressions:

$$\frac{S-S_{0}^{*}}{R} + \ln \frac{P}{P_{0}} = \ln \left( Z - \frac{Pb}{RT} \right) - \frac{1}{bRT} \left[ \sum_{i}^{N} \sum_{j}^{N} x_{i} x_{j} m_{j} \right]$$

$$\left( a_{i} a_{cj} T_{ij} \right)^{0.5} \left( 1 - k_{ij} \right) \left[ \ln \left( 1 + \frac{b}{V} \right) \right]$$

$$\frac{S-S_{0}^{*}}{R} + \ln \frac{P}{P_{0}} = \ln(Z-B) - \frac{A}{B} \left[ \sum_{i}^{N} \sum_{j}^{N} x_{i} x_{j} m_{j} \right]$$
(5-15)

$$(a_i a_{cj} T_{rj})^{0.5} (1 - k_{ij}) \left| \ln \left( 1 + \frac{B}{Z} \right) \right|.$$
 (5-16)

# 5.1.3 Fugacity Coefficient.

The fugacity coefficient expressions for the Soave equations are derived from Equations (3-17) and (4-1). Equation (3-17) is

$$\ln \phi_i = -\ln \left( Z - \frac{Pb}{RT} \right) + (Z - 1)B'_i - \frac{a}{bRT} \left( A'_i - B'_i \right) \ln \left( 1 + \frac{b}{V} \right).$$
(5-17)

Using the notations of A and B instead of a and b gives

$$\ln \phi_i = -\ln(Z - B) + (Z - 1)B'_i - \frac{A}{B}(A'_i - B'_i)\ln\left(1 + \frac{B}{Z}\right)$$
(5-18)

where

$$A_{i}^{\prime} = \frac{1}{an} \left[ \frac{\partial (n^{2}a)}{\partial n_{i}} \right]_{T,n_{j}}$$
(5-19)

$$B_i' = \frac{1}{b} \left( \frac{\partial(nb)}{\partial n_i} \right)_{n_i}.$$
(5-20)

 $B'_i$  is obtained from Equation (4-5):

$$B_i' = \frac{b_i}{b}.\tag{5-21}$$

From Equation (4-7):

$$A_{i}' = \frac{1}{a} \left[ 2a_{i}^{0.5} \sum_{j}^{N} x_{j} a_{j}^{0.5} \left( 1 - k_{ij} \right) \right].$$
(5-22)

5.2 Peng-Robinson EOS.

Equation (4-12) can be written as follows:

$$P = \frac{RT}{V - b} - \frac{a}{\left[V + \left(2^{0.5} + 1\right)b\right]\left[V - \left(2^{0.5} - 1\right)b\right]}.$$
(5-23)

# 5.2.1 Enthalpy Departure.

The integrand in Equation (3-10) is obtained from Equation (4-12), by using the same procedure used in deriving Equation (5-2):

$$T\left(\frac{\partial P}{\partial T}\right)_{V} = \frac{RT}{V-b} - \frac{T}{\left[V + \left(2^{0.5} + 1\right)b\right]\left[V - \left(2^{0.5} - 1\right)b\right]}\left(\frac{da}{dT}\right)$$
(5-24)

where T(da/dT) is given by Equation (5-8) in which  $m_j$ ,  $a_i$ , and  $a_{cj}$  are represented by Equations (4-16), (4-8), and (4-15) respectively. Substracting Equation (4-12) from Equation (5-24) and combining the result equation with Equation (3-10) gives

$$\frac{H-H^*}{RT} = Z-1$$

$$+\frac{1}{RT} \left[ a - T \frac{da}{dT} \right]_{\infty}^{V} \frac{dV}{\left[ V + (2^{0.5} + 1)b \right] \left[ V - (2^{0.5} - 1)b \right]}.$$
(5-25)

Integrating gives the following Peng-Robinson enthalpy departure expressions:

$$\frac{H-H^*}{RT} = Z-1 - \frac{1}{2^{1.5}bRT} \left[ a - T\frac{da}{dT} \right] \ln \left( \frac{V + (2^{0.5} + 1)b}{V - (2^{0.5} - 1)b} \right)$$
(5-26)

or

$$\frac{H-H^*}{RT} = Z - 1 - \frac{A}{2^{1.5}B} \left[ 1 - \frac{T}{a} \frac{da}{dT} \right] \ln \left( \frac{Z + (2^{0.5} + 1)B}{Z - (2^{0.5} - 1)B} \right)$$
(5-27)

where 
$$T\left(\frac{da}{dT}\right) = -\sum_{i}^{N} \sum_{j}^{N} x_{i} x_{j} m_{j} \left(a_{i} a_{cj} T_{rj}\right)^{0.5} \left(1 - k_{ij}\right).$$
 (5-8)

5.2.2 Entropy Departure.

From Equations (5-12) and (5-7):

$$\frac{S - S_0^*}{R} + \ln\frac{P}{P_0} = \ln\left(Z - \frac{Pb}{RT}\right) + \frac{1}{2^{1.5}bRT}\left(T\frac{da}{dT}\right)\ln\left(\frac{V + (2^{0.5} + 1)b}{V - (2^{0.5} - 1)b}\right)$$
(5-28)

or

$$\frac{S-S_0^*}{R} + \ln\frac{P}{P_0} = \ln(Z-B) + \frac{A}{2^{1.5}B} \left[\frac{T}{a}\frac{da}{dT}\right] \ln\left(\frac{Z+(2^{0.5}+1)B}{Z-(2^{0.5}-1)B}\right)$$
(5-29)

where T(da/dT) is given by Equation (5-8) in which  $m_j, a_i$ , and  $a_{cj}$  are represented by Equations (4-16), (4-8), and (4-15) respectively.

# 5.2.3 Fugacity Coefficient.

The fugacity coefficient expression for Peng-Robinson equation may be derived from Equations (3-17) and (4-12), by using the same procedure used for deriving Equation (5-17). The expression is

$$\ln \phi_{i} = -\ln \left( Z - \frac{Pb}{RT} \right) + (Z - 1)B_{i}^{\prime}$$
$$-\frac{a}{2^{1.5}bRT} \left( A_{i}^{\prime} - B_{i}^{\prime} \right) \ln \left( \frac{V + (2^{0.5} + 1)b}{V - (2^{0.5} - 1)b} \right)$$
(5-30)

$$\ln \phi_i = -\ln(Z-B) + (Z-1)B'_i$$

$$-\frac{A}{2^{1.5}B} \left(A_i' - B_i'\right) \ln\left(\frac{Z + (2^{0.5} + 1)B}{Z - (2^{0.5} - 1)B}\right)$$
(5-31)

where  $B'_i$  and  $A'_i$  are given by Equations (5-21) and (5-22), but with the  $b_i$  and  $a_i$  being given by Equations (4-14) and (4-8).

#### 5.3 ALS Equation Of State.

#### 5.3.1 Enthalpy Departure.

Expressions for isothermal enthalpy departure for the ALS equations are derived from Equations (3-10) and (4-17). Differentiating Equation (4-17) with respect to T, at constant , and multiplying by T, give

$$T\left(\frac{\partial P}{\partial T}\right)_{V} = \frac{RT}{V - b_{1}} - \frac{T}{\left(V - b_{2}\right)\left(V + b_{3}\right)}\left(\frac{da}{dT}\right).$$
(5-32)

The integrand in Equation (3-10) then becomes

$$T\left(\frac{\partial P}{\partial T}\right)_{\nu} - P = \frac{1}{\left(V - b_2\right)\left(V + b_3\right)} \left[a - T\left(\frac{da}{dT}\right)\right].$$
(5-33)

Combining Equation (3-10) with (5-33) and then integrating gives

or

$$\frac{H-H^*}{RT} = Z-1$$

$$-\frac{1}{(b_2+b_3)RT} \left[a-T\left(\frac{da}{dT}\right)\right] \ln\left(\frac{V+b_3}{V-b_2}\right)$$
(5-34)

or

$$\frac{H-H^*}{RT} = Z-1$$

$$-\frac{A}{\left(B_2+B_3\right)RT} \left[1-\frac{T}{a}\left(\frac{da}{dT}\right)\right] \ln\left(\frac{Z+B_3}{Z-B_2}\right)$$
(5-35)

where

$$B_k = \frac{Pb_k}{RT} \qquad k = 1, 2, 3.$$
(5-36)

# 5.3.2 Entropy Departure.

Expressions for isothermal entropy departure are derived from Equation (3-13). Dividing Equation (5-32) through by RT and substracting /V from both sides of the equation gives the integrand in Equation (3-13):

$$\frac{1}{R} \left( \frac{\partial P}{\partial T} \right)_{V} - \frac{1}{V} = \frac{b_1}{V(V - b_1)} - \frac{1}{RT(V - b_2)(V + b_3)} \left( T \frac{da}{dT} \right).$$
(5-37)

Combining Equation (3-13) with (5-37) and integrating the resulting equation gives

$$\frac{S - S_0^*}{R} + \ln \frac{P}{P_0} = \ln \left( Z - \frac{Pb_1}{RT} \right) + \frac{1}{\left( b_2 + b_3 \right) RT} \left( T \frac{da}{dT} \right) \ln \left( \frac{V + b_3}{V - b_2} \right)$$
(5-38)

or

$$\frac{S - S_0^*}{R} + \ln\frac{P}{P_0} = \ln(Z - B_1) + \frac{A}{(B_2 + B_3)} \left(\frac{T}{a}\frac{da}{dT}\right) \ln\left(\frac{Z + B_3}{Z - B_2}\right).$$
 (5-39)

# 5.3.3 Fugacity Coefficient.

The fugacity coefficient expressions for the ALS equation are derived from Equations (3-17) and (4-17).

$$\ln \phi_{i} = -\ln \left( Z - \frac{Pb_{1}}{RT} \right) + (Z - 1)B_{1i}' + \frac{a}{(b_{2} + b_{3})RT} \left( A_{i}' - B_{2i}' \right) \ln \left( V - b_{2} \right)$$
$$- \frac{a}{(b_{2} + b_{3})RT} \left( A_{i}' + B_{3i}' \right) \ln \left( V + b_{3} \right)$$
(5-40)

or

$$\ln \phi_{i} = -\ln(Z - B_{1}) + (Z - 1)B_{1i}' + \frac{A}{(B_{2} + B_{3})}(A_{i}' - B_{2i}')\ln(Z - B_{2})$$
$$-\frac{A}{(B_{2} + B_{3})}(A_{i}' + B_{3i}')\ln(Z + B_{3})$$
(5-41)

where

$$B'_{ki} = \frac{b_{ki}}{b_k} \qquad k = 1, 2, 3.$$
(5-42)

# 5.4 SBC Equation Of State.

# 5.4.1 Enthalpy Departure.

Expressions for isothermal enthalpy departure for the SBC equations are derived from Equations (3-10) and (4-30). Differentiating Equation (4-30) with respect to T, at constant , and multiplying by T, give

$$T\left(\frac{\partial P}{\partial T}\right)_{\nu} = \frac{RT}{\left(V - k_{0}\beta\right)} + \frac{k_{0}RT}{\left(V - k_{0}\beta\right)^{2}} \left(T\frac{d\beta}{dT}\right) + \frac{k_{1}\beta RT}{\left(V - k_{0}\beta\right)^{2}} + \frac{2k_{0}k_{1}\beta RT}{\left(V - k_{0}\beta\right)^{3}} \left(T\frac{d\beta}{dT}\right) + \frac{k_{1}RT}{\left(V - k_{0}\beta\right)^{2}} \left(T\frac{d\beta}{dT}\right) + \frac{1}{eV} \left(T\frac{dc}{dT}\right) - \frac{1}{\left(V + e\right)} \left[\frac{1}{e} \left(T\frac{dc}{dT}\right) - \frac{1}{\left(k_{0}\beta + e\right)} \left(T\frac{da}{dT} + T\frac{dc}{dT}\right) + \frac{k_{0}(a + c)}{\left(k_{0}\beta + e\right)^{2}} \left(T\frac{d\beta}{dT}\right)\right] - \frac{1}{\left(V - k_{0}\beta\right)} \left[\frac{1}{\left(k_{0}\beta + e\right)} \left(T\frac{da}{dT} + T\frac{dc}{dT}\right) - \frac{k_{0}(a + c)}{\left(k_{0}\beta + e\right)^{2}} \left(T\frac{d\beta}{dT}\right)\right] - \frac{k_{0}(a + c)}{\left(k_{0}\beta + e\right)^{2}} \left(T\frac{d\beta}{dT}\right).$$
(5-43)

The integrand in Equation (3-10) then becomes

$$T\left(\frac{\partial P}{\partial T}\right)_{V} - P = \frac{RT}{\left(V - k_{0}\beta\right)^{2}} \left[T\left(\frac{d\beta}{dT}\right)\right] \left(k_{0} + k_{1} + \frac{2k_{0}k_{1}\beta}{\left(V - k_{0}\beta\right)}\right)$$

$$-\frac{1}{e}\left[\frac{1}{V}-\frac{1}{(V+e)}\left[c-T\left(\frac{dc}{dT}\right)\right]\right]$$
$$-\frac{1}{\left(k_{0}\beta+e\right)}\left[\frac{1}{\left(V+e\right)}-\frac{1}{\left(V-k_{0}\beta\right)}\right]\left\{\left[a-T\frac{da}{dT}\right]+\left[c-T\frac{dc}{dT}\right]\right\}$$
$$-\frac{k_{0}(a+c)}{\left(k_{0}\beta+e\right)^{2}}\left(T\frac{d\beta}{dT}\right)\left\{\frac{1}{\left(V+e\right)}\right.$$
$$\left.-\frac{1}{\left(V-k_{0}\beta\right)}+\frac{\left(k_{0}\beta+e\right)}{\left(V-k_{0}\beta\right)^{2}}\right\}.$$
(5-44)

Combining Equation (3-10) with (5-44) and then integrating gives

$$\frac{H-H^{*}}{RT} = Z - 1 - \frac{\left(k_{0} + k_{1}\right)}{\left(V - k_{0}\beta\right)} \left(T\frac{d\beta}{dT}\right) - \frac{k_{0}k_{1}\beta}{\left(V - k_{0}\beta\right)^{2}} \left(T\frac{dB}{dT}\right)$$
$$- \frac{1}{\left(k_{0}\beta + e\right)RT} \left\{ \left[a - T\left(\frac{da}{dT}\right)\right] + \left[c - T\left(\frac{dc}{dT}\right)\right] \right\} \ln\left(\frac{V+e}{V-k_{0}\beta}\right)$$
$$+ \frac{1}{eRT} \left[c - T\left(\frac{dc}{dT}\right)\right] \ln\left(1 + \frac{e}{V}\right)$$
$$- \frac{k_{0}(a+c)}{\left(k_{0}\beta + e\right)^{2}RT} \left(T\frac{d\beta}{dT}\right) \left\{ \ln\frac{\left(V+e\right)}{\left(V-k_{0}\beta\right)} - \frac{\left(k_{0}\beta + e\right)}{\left(V-k_{0}\beta\right)} \right\}.$$
(5-45)

Same procedure as Equation (5-8) gives

for  $T_r \ll 1$ ,

$$T\left(\frac{da}{dT}\right) = a_c \sqrt{\alpha} \left(-\sqrt{T_r}\right) \left(X_2 + 2X_3 \left(1 - \sqrt{T_r}\right) + 3X_4 \left(1 - \sqrt{T_r}\right)^2\right)$$
(5-46)

and for  $T_r > 1$ ,

$$T\left(\frac{da}{dT}\right) = a_c \sqrt{\alpha} \left(-\sqrt{T_r}\right) \left(X_2 + 2X_5 \left(1 - \sqrt{T_r}\right) + 3X_6 \left(1 - \sqrt{T_r}\right)^2\right)$$
(5-47)

and

$$T\left(\frac{d\beta}{dT}\right) = 3\beta \left[-0.03125 - 2*0.0054\ln(T_r)\right]$$
(5-48)

and finally,

$$T\left(\frac{dc}{dT}\right) = c_c \sqrt{\xi} \left(-\sqrt{T_r}\right) X_{7}.$$
(5-49)

#### 5.4.2 Entropy Departure.

Expressions for isothermal entropy departure are derived from Equation (3-13). Dividing Equation (5-43) through by RT and substracting /V from both sides of the equation gives the integrand in Equation (3-13):

$$\frac{1}{R} \left( \frac{\partial P}{\partial T} \right)_{V} - \frac{1}{V} = \frac{k_0 \beta}{V (V - k_0 \beta)} + \frac{k_1 \beta}{\left( V - k_0 \beta \right)^2} + \frac{\left( k_0 + k_1 \right)}{\left( V - k_0 \beta \right)^2} \left( T \frac{d\beta}{dT} \right) + \frac{2k_0 k_1 \beta}{\left( V - k_0 \beta \right)^3} \left( T \frac{d\beta}{dT} \right) + \frac{1}{e} \left[ \frac{1}{V} - \frac{1}{\left( V + e \right)} \left[ T \left( \frac{dc}{dT} \right) \right]$$

$$-\frac{1}{\left(k_{0}\beta+e\right)}\left[\frac{1}{\left(V+e\right)}-\frac{1}{\left(V-k_{0}\beta\right)}\right]\left\{-\left[T\frac{da}{dT}+T\frac{dc}{dT}\right]\right\}$$
$$-\frac{k_{0}(a+c)}{\left(k_{0}\beta+e\right)^{2}}\left(T\frac{d\beta}{dT}\right)\left\{\frac{1}{\left(V+e\right)}$$
$$-\frac{1}{\left(V-k_{0}\beta\right)}+\frac{\left(k_{0}\beta+e\right)}{\left(V-k_{0}\beta\right)^{2}}\right\}.$$
(5-50)

Combining Equation (3-13) with (5-50) and integrating the resulting equation gives

$$\frac{S-S_0^*}{R} + \ln \frac{P}{P_0} = \ln \left( Z - \frac{Pk_0\beta}{RT} \right) - \frac{k_1\beta}{(V-k_0\beta)}$$

$$- \frac{\left(k_0 + k_1\right)}{\left(V - k_0\beta\right)} \left( T \frac{d\beta}{dT} \right) - \frac{k_0k_1\beta}{(V-k_0\beta)^2} \left( T \frac{d\beta}{dT} \right)$$

$$+ \frac{1}{\left(k_0\beta + e\right)RT} \left\{ T \left( \frac{da}{dT} \right) + T \left( \frac{dc}{dT} \right) \right\} \ln \left( \frac{V+e}{V-k_0\beta} \right)$$

$$- \frac{1}{eRT} \left[ T \left( \frac{dc}{dT} \right) \right] \ln \left( 1 + \frac{e}{V} \right)$$

$$- \frac{k_0(a+c)}{\left(k_0\beta + e\right)^2 RT} \left( T \frac{d\beta}{dT} \right) \left\{ \ln \frac{(V+e)}{(V-k_0\beta)} - \frac{\left(k_0\beta + e\right)}{(V-k_0\beta)} \right\}.$$
(5-51)

5.4.3 Fugacity Coefficient.

The fugacity coefficient expressions for the SBC equation are derived from Equations (3-17) and (4-30).

$$\ln \phi_{i} = -\ln \left( Z - \frac{Pk_{0}\beta}{RT} \right) + \frac{(k_{0} + k_{1})\beta}{(V - k_{0}\beta)} + \frac{k_{0}k_{1}\beta^{2}}{(V - k_{0}\beta)^{2}} + \frac{c}{RT(V + e)} + \frac{c}{eRT} \ln \left( 1 + \frac{e}{V} \right) - \frac{(a + c)}{(k_{0}\beta + e)RT} \left[ \frac{e}{(V + e)} + \frac{k_{0}\beta}{(V - k_{0}\beta)} \right] - \frac{(a + c)}{(k_{0}\beta + e)RT} \ln \frac{(V + e)}{(V - k_{0}\beta)}.$$
(5-52)

#### 5.5 TCC Equation Of State.

# 5.5.1 Enthalpy Departure.

Expressions for isothermal enthalpy departure for the TCC equation are derived from Equations (3-10) and (4-68). Differentiating Equation (4-68) with respect to T, at constant , and multiplying by T, give

$$T\left(\frac{\partial P}{\partial T}\right)_{V} = \frac{RT}{V-b} - \frac{2}{(2V+4b+c+w)(2V+4b+c-w)}\left(T\frac{da}{dT}\right).$$
 (5-53)

The integrand in Equation (3-10) then becomes

. .

$$T\left(\frac{\partial P}{\partial T}\right)_{\nu} - P = \frac{2}{(2V+4b+c+w)(2V+4b+c-w)}\left[a - T\left(\frac{da}{dT}\right)\right].$$
 (5-54)

Combining Equation (3-10) with (5-54) and then integrating gives

$$\frac{H-H^*}{RT} = Z-1$$

$$-\frac{1}{wRT} \left[ a - T \left(\frac{da}{dT}\right) \right] \ln \left(\frac{2V+4b+c+w}{2V+4b+c-w}\right)$$
(5-55)

or

$$\frac{H-H^*}{RT} = Z - 1 - \frac{A}{W} \left[ 1 - \frac{T}{a} \left( \frac{da}{dT} \right) \right] \ln(\Phi)$$
(5-56)

where

$$= \left(16B^2 + 4BC + C^2\right)^{1/2} \tag{5-57}$$

$$\Phi = \frac{2Z + 4B + C + W}{2Z + 4B + C - W}.$$
(5-58)

# 5.6.2 Entropy Departure.

Expressions for isothermal entropy departure are derived from Equation (3-13). Dividing Equation(5-53) through by RT and substracting /V from both sides of the equation gives the integrand in Equation (3-13):

$$\frac{1}{R} \left(\frac{\partial P}{\partial T}\right)_{V} - \frac{1}{V} = \frac{b}{V(V-b)}$$
$$-\frac{2}{RT(2V+4b+c+w)(2V+4b+c-w)} \left(T\frac{da}{dT}\right).$$
(5-59)

Combining Equation (3-13) with (5-59) and integrating the resulting equation gives

$$\frac{S-S_0^*}{R} + \ln\frac{P}{P_0} = \ln\left(Z - \frac{Pb}{RT}\right) + \frac{2}{wRT}\left(T\frac{da}{dT}\right)\ln\left(\frac{2V+4b+c+w}{2V+4b+c-w}\right)$$
(5-60)

or

$$\frac{S-S_0^*}{R} + \ln\frac{P}{P_0} = \ln(Z-B) + \frac{A}{W} \left(\frac{T}{a}\frac{da}{dT}\right) \ln(\Phi).$$
(5-61)

# 5.6.3 Fugacity Coefficient.

The fugacity coefficient expressions for the TCC equation of state are derived from Equations (3-17) and (4-68).

$$\ln \phi_{i} = \frac{B_{i}}{Z - B} - \ln(Z - B) + \frac{A}{W} \left\{ \Theta_{i} - \frac{1}{a} \left[ \sum_{j} x_{j} \left( a_{ij}^{0} + a_{ji}^{0} \right) + \epsilon_{i} \right] \right\} \ln(\Phi) + \left( \frac{1}{Z - B} - 1 \right) \left\{ \frac{1}{2} \left( 4B_{i} + C_{i} \right) - \Theta_{i} \left[ Z + \frac{1}{2} (4B + C) \right] \right\}$$
(5-62)

where

$$a_{ij}^{0} = \left(a_{i}a_{j}\right)^{1/2} \left(1 - \frac{k_{ij}}{T}\right)$$
(5-63)

$$\Theta_i = \frac{1}{W^2} \Big[ 2(8B+C)B_i + (2B+C)C_i \Big].$$
(5-64)

 $\in_i$  is due to the composition-dependent term in the mixing rule,

$$\epsilon_{i} = \frac{\left[\sum_{j} H_{ij}^{1/3} G_{ij}^{1/3} \left(a_{i}a_{j}\right)^{1/6} x_{j}\right]^{3}}{\sum_{j} G_{ij} x_{j}}$$

$$+ 3 \sum_{j} x_{j} \frac{\left[\sum_{k} H_{jk}^{1/3} G_{jk}^{1/3} \left(a_{j}a_{k}\right)^{1/6} x_{k}\right]^{2} \left[H_{ji}^{1/3} G_{ji}^{1/3} \left(a_{j}a_{i}\right)^{1/6}\right]}{\sum_{k} G_{jk} x_{k}}$$

$$- \sum_{j} x_{j} \frac{\left[\sum_{k} H_{jk}^{1/3} G_{jk}^{1/3} \left(a_{j}a_{k}\right)^{1/6} x_{k}\right]^{3}}{\sum_{k} G_{jk} x_{k}} \left[1 + \frac{G_{ji}}{\sum_{k} G_{jk} x_{k}}\right].$$

$$(5-65)$$