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นางสาว สุจิตรา ต้องกระโทก



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ELECTROCHEMICAL AND CATALYTIC PROPERTIES OF TRANSITION METAL-SCHIFF BASE COMPLEXES

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ได้ศึกษาสมบัติทางเคมีไฟฟ้าและสมบัติการเร่งปฏิกิริยาของสารประกอบเชิงซ้อนโลหะชิฟ เบสที่ใช้เป็นตัวเร่งปฏิกิริยาในปฏิกิริยาออกซิเคชันของไซโคลเฮกเซน สมบัติทางเคมีไฟฟ้าตรวจ สอบโดยเทคนิคไซคลิกโวลแทมเมทรี โคยใช้กลาสซีคาร์บอน และขั้วไฟฟ้าฟ้าฟิล์มบางของเพชรซึ่ง โคปด้วยบอรอนเป็นขั้วทำงาน ลวดเงินเป็นขั้วอ้างอิงและลวดแพลทินัมเป็นขั้วช่วย จากการศึกษา พบว่าสารประกอบเชิงซ้อนที่เกิดปฏิกิริยารีดอกซ์แบบผันกลับได้และกึ่งผันกลับได้จะให้ปริมาณ ผลิตภัณฑ์สูงกว่าสารประกอบเชิงซ้อนโลหะชิฟเบสชนิดอื่นที่ได้ศึกษา นอกจากนี้ผลของปฏิกิริยา รีดอกซ์ที่ได้จากกลาสซีการ์บอน และขั้วไฟฟ้าฟ้าฟิล์มบางของเพชรซึ่งโดปด้วยบอรอน มีความแตก ต่างกัน เนื่องจากการถ่ายโอนอิเล็กตรอนที่ผิวหน้าของขั้วไฟฟ้าฟ้าฟิล์มบางของเพชรซึ่งโดปด้วย บอรอนเกิดช้ากว่า เมื่อพิจารณาผลการศึกษาทางจลน์ศาสตร์ของปฏิกิริยาออกซิเคชันของไซโคลเฮก เซน พบว่าผลที่ได้สอดกล้องกับผลที่ได้จากเทคนิกไซคลิกโวลแทมเมทรี

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Electrochemical and catalytic properties of metal Schiff base complexes as catalyst in cyclohexane oxidation were thoroughly investigated. Cyclic voltammetry was performed using either glassy carbon and boron-doped diamond electrode, silver wire as a reference electrode and platinum wire as an auxillary electrode. The results of cyclic voltammetry revealed that metal Schiff base complexes which exhibited reversible and quasi-reversible reactions provided better yield of the desire products than other transition metal Schiff base complexes studied. In addition, the redox reaction employing glassy carbon and boron-doped diamond electrodes was different because the electrode-reaction kinetics of boron-doped diamond electrode was slow. The kinetic study of cyclohexane oxidation was found in good agreement with the results that gained from cyclic voltammetry.

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LIST OF ABBREVIATIONS

Fig	figure
°C	degree celsius
atm	atmosphere
mmol	millimole
NMR	nuclear magnetic resonance
IR	Infrared spectroscopy
g	gram (s)
mL	milliliter (s)
min	minute (s)
h	hour (s)
cm ⁻¹	unit of wavenumber
ppm	part per million
J	coupling constant
m	multiplet (NMR)
dd	doublet of doublet (NMR)
d	doublet (NMR)
S	singlet (NMR)
m.p.	melting point
dec	decomposed
lit	literature
%	percent
R _f	retardation factor

LIST OF ABBREVIATIONS (cont.)

haen	Bis(2-hydroxyacetophenone)-N, N'-ethylenediimine
sac	N-salicylalidene-2-anthranilic acid
sap	N-salicylalidene-2-aminophenol
salen	Bis(salicylaldehyde)-N,N'-ethylenediimine
salophen	Bis(salicylaldehyde)-N,N'-trimethylenediimine
H_2O_2	hydrogen peroxide
TBHP	tert-butylhydroperoxide