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## APPENDICES

### Appendix A Surface Tension of Surfactant Solutions

**Table A1** Surface tension for solution of CPB, OP(EO)<sub>10</sub>, and their mixtures

Total surfactant concentration (μM)	Surface tension (mN/m)				
	α = 0	α = 0.25	α = 0.50	α = 0.75	α = 1.0
20	71.46	68.47	66.66	64.00	60.80
40	69.27	64.38	59.71	55.22	49.66
60	67.25	56.41	55.16	49.56	45.92
80	65.58	53.79	50.55	47.44	43.35
100	62.45	52.12	46.96	44.39	41.22
200	57.13	45.19	41.52	37.86	30.66
300	52.25	39.87	37.47	33.78	30.73
400	48.15	38.61	35.78	33.52	30.67
500	44.17	36.16	35.66	34.02	30.49
600	39.69	36.30	35.78	34.16	31.40
700	39.18	36.52	35.76	33.78	30.48
800	38.59	36.46	35.86	34.00	31.60
900	38.64	36.57	35.91	33.75	30.51
1,000	38.66	36.16	36.01	33.80	31.44
3,000	38.47	36.36	36.53	33.86	30.81
5,000	38.56	36.28	35.84	33.73	30.89
7,000	38.22	36.12	36.13	33.76	30.79
10,000	38.00	36.36	35.99	33.92	30.86

## Appendix B Adsorption Isotherm of Surfactant Solution

**Table B1** Adsorption isotherm on HDPE of 1:0 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> HDPE)
20	10.81	0.51
40	26.49	0.74
60	45.14	0.82
80	62.70	0.96
100	81.89	1.00
200	179.46	1.13
300	274.32	1.42
400	372.43	1.53
500	470.54	1.63
600	567.03	1.83
700	663.24	2.03
800	762.70	2.06
900	863.78	2.00
1000	963.51	2.02
3000	2962.16	2.10
5000	4962.16	2.10
7000	6964.86	1.94
10000	9962.16	2.09

**Table B2** Adsorption isotherm on HDPE of 3:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ HDPE)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ HDPE)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ HDPE)
20	9.80	0.56	0.42	0.14
40	23.13	0.93	0.70	0.23
60	40.00	1.10	0.83	0.28
80	56.33	1.32	0.99	0.32
100	73.20	1.48	1.11	0.37
200	159.73	2.23	1.68	0.55
300	257.14	2.37	1.77	0.60
400	352.97	2.60	2.05	0.55
500	450.07	2.76	2.07	0.69
600	546.40	2.97	2.23	0.74
700	647.34	2.92	2.18	0.74
800	746.67	2.97	2.23	0.74
900	845.70	3.01	2.22	0.79
1000	943.40	3.13	2.35	0.78
3000	2947.48	2.90	2.16	0.74
5000	4941.44	3.22	2.30	0.92
7000	6941.44	3.24	2.32	0.92
10000	9942.79	3.16	2.24	0.92

**Table B3** Adsorption isotherm on HDPE of 1:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Total amount of surfactant adsorbed (μmole/m <sup>2</sup> HDPE)	Cationic surfactant adsorbed (μmole/m <sup>2</sup> HDPE)	Nonionic surfactant adsorbed (μmole/m <sup>2</sup> HDPE)
20	9.32	0.59	0.31	0.28
40	20.00	1.10	0.55	0.55
60	35.07	1.38	0.69	0.69
80	52.32	1.53	0.75	0.79
100	68.49	1.75	0.87	0.88
200	160.54	2.19	1.08	1.11
300	255.07	2.50	1.25	1.25
400	350.70	2.72	1.39	1.34
500	442.75	3.16	1.60	1.57
600	543.02	3.17	1.59	1.57
700	643.02	3.17	1.59	1.58
800	742.75	3.17	1.60	1.57
900	842.75	3.18	1.61	1.57
1000	943.58	3.13	1.61	1.53
3000	2947.97	2.88	1.50	1.39
5000	4947.97	2.89	1.50	1.39
7000	6936.94	3.50	1.65	1.85
10000	9939.64	3.34	1.49	1.84

**Table B4** Adsorption isotherm on HDPE of 1:3 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ HDPE)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ HDPE)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ HDPE)
20	6.35	0.76	0.21	0.55
40	11.60	1.57	0.42	1.15
60	24.55	1.96	0.49	1.47
80	42.21	2.09	0.52	1.57
100	54.86	2.49	0.56	1.93
200	140.43	3.30	0.85	2.44
300	233.09	3.68	0.88	2.80
400	321.60	4.32	0.97	3.36
500	425.07	4.15	0.97	3.19
600	521.87	4.31	0.96	3.36
700	622.03	4.30	0.99	3.31
800	721.35	4.35	1.03	3.32
900	822.57	4.29	0.97	3.32
1000	922.73	4.27	1.00	3.27
3000	2924.10	4.20	0.97	3.23
5000	4924.10	4.19	0.97	3.22
7000	6924.10	4.21	0.97	3.23
10000	9917.12	4.59	0.90	3.69

**Table B5** Adsorption isotherm on HDPE of 0:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> HDPE)
20	10.83	0.51
40	23.33	0.92
60	34.17	1.43
80	45.83	1.88
100	55.83	2.43
200	126.67	4.05
300	215.83	4.64
400	315.83	4.63
500	415.00	4.67
600	515.83	4.64
700	615.00	4.71
800	715.83	4.66
900	815.83	4.66
1000	915.83	4.67
3000	2916.67	4.60
5000	4916.67	4.59
7000	6916.67	4.61
10000	9916.67	4.60

**Table B6** Adsorption isotherm on PC of 1:0 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> PC)
20	15.14	0.24
40	32.70	0.35
60	47.57	0.60
80	63.51	0.80
100	80.00	0.97
200	173.78	1.28
300	266.76	1.62
400	363.78	1.76
500	460.54	1.92
600	558.11	2.04
700	652.70	2.31
800	752.16	2.32
900	853.24	2.28
1000	951.89	2.34
3000	2954.05	2.24
5000	4954.05	2.24
7000	6951.35	2.37
10000	9954.05	2.24

**Table B7** Adsorption isotherm on PC of 3:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PC)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PC)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PC)
20	13.33	0.32	0.24	0.08
40	26.10	0.68	0.47	0.20
60	39.73	0.99	0.74	0.24
80	56.60	1.14	0.85	0.28
100	73.47	1.29	0.97	0.32
200	163.00	1.80	1.36	0.45
300	250.07	2.44	1.83	0.61
400	343.13	2.76	2.07	0.69
500	436.73	3.08	2.31	0.77
600	533.60	3.23	2.42	0.81
700	634.28	3.20	2.39	0.81
800	733.33	3.25	2.44	0.81
900	832.64	3.28	2.43	0.85
1000	933.60	3.23	2.42	0.81
3000	2933.33	3.25	2.44	0.81
5000	4933.33	3.25	2.43	0.81
7000	6933.33	3.24	2.43	0.81
10000	9931.98	3.21	2.50	0.81

**Table B8** Adsorption isotherm on PC of 1:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PC)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PC)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PC)
20	8.49	0.56	0.28	0.28
40	26.58	0.65	0.33	0.32
60	43.56	0.80	0.39	0.41
80	61.64	0.89	0.45	0.45
100	73.15	1.31	0.66	0.65
200	158.09	2.04	1.03	1.01
300	247.95	2.53	1.28	1.26
400	335.34	3.15	1.57	1.58
500	430.14	3.41	1.70	1.71
600	529.86	3.41	1.71	1.70
700	628.47	3.49	1.70	1.79
800	730.14	3.40	1.70	1.70
900	830.14	3.40	1.70	1.70
1000	930.14	3.40	1.70	1.70
3000	2934.23	3.21	1.58	1.63
5000	4931.53	3.33	1.71	1.62
7000	6931.53	3.33	1.71	1.62
10000	9931.53	3.33	1.71	1.62

**Table B9** Adsorption isotherm on PC of 1:3 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Total amount of surfactant adsorbed (μmole/m <sup>2</sup> PC)	Cationic surfactant adsorbed (μmole/m <sup>2</sup> PC)	Nonionic surfactant adsorbed (μmole/m <sup>2</sup> PC)
20	10.83	0.73	0.10	0.73
40	29.17	0.86	0.15	0.86
60	45.00	1.20	0.27	1.20
80	61.67	1.46	0.33	1.46
100	74.17	2.06	0.45	2.06
200	154.17	3.66	0.71	3.66
300	235.83	5.11	1.03	5.11
400	337.50	4.99	1.22	4.99
500	435.83	5.13	1.23	5.13
600	535.83	5.11	1.18	5.11
700	636.67	5.07	1.17	5.07
800	736.67	5.06	1.19	5.06
900	835.83	5.13	1.21	5.13
1000	934.17	5.25	1.23	5.25
3000	2933.33	5.31	1.25	5.31
5000	4933.33	5.30	1.25	5.30
7000	6941.67	4.65	1.25	4.65
10000	9933.33	5.32	1.18	5.32

**Table B10** Adsorption isotherm on PC of 0:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Amount of surfactant adsorbed ( $\mu\text{mole}/\text{m}^2 \text{PC}$ )
20	10.83	0.73
40	29.17	0.86
60	45.00	1.20
80	61.67	1.46
100	74.17	2.06
200	154.17	3.66
300	235.83	5.11
400	337.50	4.99
500	435.83	5.13
600	535.83	5.11
700	636.67	5.07
800	736.67	5.06
900	835.83	5.13
1000	934.17	5.25
3000	2933.33	5.31
5000	4933.33	5.30
7000	6941.67	4.65
10000	9933.33	5.32

**Table B11** Adsorption isotherm on PVC of 1:0 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> PVC)
20	17.57	0.11
40	34.86	0.23
60	52.70	0.32
80	70.00	0.44
100	87.30	0.56
200	172.16	1.22
300	257.57	1.87
400	348.92	2.25
500	444.86	2.43
600	540.81	2.61
700	634.59	2.88
800	734.05	2.91
900	832.70	2.96
1000	931.89	3.00
3000	2937.84	2.73
5000	4935.14	2.85
7000	6935.14	2.86
10000	9940.54	2.62

**Table B12** Adsorption isotherm on PVC of 3:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PVC)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PVC)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PVC)
20	10.07	0.44	0.33	0.11
40	19.86	0.89	0.67	0.22
60	33.47	1.17	0.87	0.29
80	49.53	1.34	1.01	0.33
100	59.32	1.79	1.35	0.44
200	145.59	2.39	1.80	0.59
300	236.46	2.80	2.10	0.70
400	330.34	3.13	2.34	0.79
500	424.21	3.33	2.49	0.84
600	516.73	3.64	2.73	0.91
700	616.60	3.69	2.77	0.92
800	717.03	3.68	2.79	0.89
900	817.45	3.65	2.80	0.85
1000	916.19	3.68	2.76	0.91
3000	2922.52	3.46	2.71	0.74
5000	4919.82	3.53	2.80	0.73
7000	6911.49	3.90	2.80	1.10
10000	9921.17	3.47	2.74	0.73

**Table B13** Adsorption isotherm on PVC of 1:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Total amount of surfactant adsorbed (μmole/m <sup>2</sup> PVC)	Cationic surfactant adsorbed (μmole/m <sup>2</sup> PVC)	Nonionic surfactant adsorbed (μmole/m <sup>2</sup> PVC)
20	8.49	0.51	0.25	0.26
40	24.66	0.67	0.34	0.33
60	40.00	0.88	0.44	0.44
80	56.98	1.01	0.50	0.51
100	69.86	1.33	0.67	0.66
200	137.00	2.77	1.42	1.36
300	224.64	3.32	1.63	1.69
400	313.96	3.78	1.88	1.91
500	406.89	4.10	2.12	1.98
600	512.32	3.87	1.92	1.95
700	607.97	4.06	2.07	1.98
800	704.71	4.19	2.18	2.02
900	808.27	4.05	2.10	1.95
1000	908.51	4.03	2.05	1.98
3000	2909.58	3.98	2.14	1.84
5000	4901.35	4.34	2.14	2.02
7000	6909.68	3.98	2.14	1.84
10000	9915.09	3.74	1.91	1.84

**Table B14** Adsorption isotherm on PVC of 1:3 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PVC)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PVC)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PVC)
20	6.62	0.59	0.15	0.44
40	9.10	1.36	0.33	1.02
60	19.84	1.77	0.41	1.36
80	26.73	2.34	0.55	1.80
100	45.38	2.48	0.61	1.87
200	138.49	2.71	0.69	2.02
300	217.70	3.61	0.98	2.64
400	293.51	4.70	1.17	3.53
500	383.45	5.13	1.28	3.85
600	483.67	5.13	1.42	3.71
700	580.70	5.27	1.33	3.94
800	682.21	5.19	1.30	3.89
900	781.51	5.22	1.29	3.93
1000	880.81	5.25	1.29	3.97
3000	2888.29	4.92	1.25	3.67
5000	4888.29	4.93	1.25	3.68
7000	6885.59	5.05	1.37	3.68
10000	9878.60	5.34	1.31	4.03

**Table B15** Adsorption isotherm on PVC of 0:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> PVC)
20	9.17	0.48
40	17.50	0.99
60	27.50	1.43
80	35.00	1.98
100	45.83	2.39
200	106.67	4.12
300	181.67	5.23
400	260.00	6.17
500	359.17	6.21
600	461.67	6.10
700	557.50	6.28
800	659.17	6.22
900	755.83	6.34
1000	867.50	5.83
3000	2858.33	6.24
5000	4858.33	6.26
7000	6866.68	5.86
10000	9858.33	6.25

**Table B16** Adsorption isotherm on ABS of 1:0 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)
20	17.57	0.06
40	28.11	0.31
60	41.08	0.49
80	55.95	0.62
100	65.14	0.90
200	138.38	1.58
300	208.38	2.36
400	293.24	2.74
500	377.30	3.15
600	462.43	3.54
700	546.22	3.96
800	647.30	3.92
900	741.89	4.06
1000	843.78	4.02
3000	2843.24	4.04
5000	4845.95	3.97
7000	6843.24	4.03
10000	9837.84	4.17

**Table B17** Adsorption isotherm on ABS of 3:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)
20	6.53	0.34	0.26	0.09
40	13.06	0.69	0.52	0.17
60	19.05	1.05	0.80	0.26
80	29.39	1.30	0.98	0.32
100	36.46	1.63	1.23	0.41
200	102.32	2.51	1.89	0.62
300	186.13	2.92	2.19	0.73
400	244.64	4.00	3.01	0.99
500	337.97	4.17	3.14	1.03
600	416.87	4.73	3.54	1.18
700	517.82	4.73	3.54	1.19
800	617.70	4.71	3.55	1.16
900	714.28	4.74	3.55	1.19
1000	816.33	4.72	3.54	1.18
3000	2819.14	4.66	3.59	1.07
5000	4821.85	4.57	3.50	1.07
7000	6821.85	4.59	3.51	1.07
10000	9812.16	4.83	3.54	1.28

**Table B18** Adsorption isotherm on ABS of 1:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Total amount of surfactant adsorbed (μmole/m <sup>2</sup> ABS)	Cationic surfactant adsorbed (μmole/m <sup>2</sup> ABS)	Nonionic surfactant adsorbed (μmole/m <sup>2</sup> ABS)
20	4.93	0.38	0.19	0.19
40	9.86	0.78	0.39	0.39
60	16.71	1.12	0.56	0.56
80	21.64	1.50	0.75	0.75
100	33.42	1.72	0.86	0.86
200	98.09	2.63	1.32	1.31
300	154.23	3.76	1.87	1.89
400	227.70	4.43	2.24	2.18
500	303.58	5.06	2.55	2.51
600	403.58	5.06	2.55	2.51
700	503.58	5.07	2.55	2.52
800	603.58	5.06	2.55	2.51
900	705.20	5.02	2.51	2.51
1000	803.31	5.07	2.56	2.51
3000	2802.70	5.08	2.51	2.58
5000	4808.33	4.92	2.57	2.35
7000	6811.04	4.88	2.51	2.37
10000	9802.70	5.08	2.51	2.58

**Table B19** Adsorption isotherm on ABS of 1:3 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)
20	2.48	0.45	0.11	0.34
40	5.52	0.89	0.22	0.66
60	11.31	1.26	0.31	0.95
80	15.45	1.67	0.42	1.25
100	28.13	1.85	0.46	1.39
200	61.53	3.55	0.90	2.65
300	119.44	4.66	1.15	3.50
400	203.92	5.06	1.32	3.74
500	281.96	5.60	1.43	4.18
600	370.52	5.90	1.51	4.39
700	470.92	5.89	1.50	4.39
800	571.89	5.87	1.50	4.38
900	672.03	5.88	1.49	4.38
1000	771.33	5.90	1.49	4.41
3000	2764.19	6.06	1.56	4.49
5000	4766.89	6.01	1.50	4.51
7000	6766.89	6.01	1.50	4.51
10000	9762.84	6.09	1.60	4.50

**Table B20** Adsorption isotherm on ABS of 0:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ ABS)
20	1.64	0.57
40	2.46	0.97
60	4.10	1.44
80	5.74	1.91
100	7.38	2.38
200	34.43	4.26
300	96.72	5.24
400	169.67	5.93
500	246.72	6.53
600	342.62	6.61
700	444.26	6.60
800	539.34	6.71
900	636.89	6.78
1000	739.34	6.71
3000	2737.70	6.74
5000	4737.70	6.74
7000	6737.70	6.76
10000	9737.70	6.75

**Table B21** Adsorption isotherm on PMMA of 1:0 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> PMMA)
20	17.30	0.21
40	30.81	0.72
60	41.89	1.43
80	55.14	1.96
100	69.73	2.39
200	157.57	3.35
300	252.70	3.73
400	344.05	4.42
500	440.27	4.71
600	535.95	5.06
700	625.41	5.90
800	724.59	5.94
900	827.30	5.73
1000	924.86	5.93
3000	2924.32	5.96
5000	4927.03	5.75
7000	6924.32	5.97
10000	9924.32	5.96

**Table B22** Adsorption isotherm on PMMA of 3:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)
20	9.80	0.81	0.61	0.20
40	20.14	1.57	1.17	0.39
60	33.20	2.12	1.59	0.53
80	46.80	2.61	1.95	0.65
100	59.05	3.21	2.43	0.79
200	139.86	4.74	3.56	1.18
300	229.39	5.57	4.19	1.38
400	326.26	5.83	4.38	1.45
500	422.32	6.11	4.60	1.51
600	519.73	6.34	4.76	1.58
700	619.32	6.34	4.77	1.57
800	719.73	6.33	4.75	1.58
900	819.05	6.35	4.78	1.57
1000	918.63	6.39	4.76	1.64
3000	2922.52	6.09	4.78	1.31
5000	4914.19	6.68	4.73	1.95
7000	6919.82	6.30	4.99	1.31
10000	9915.54	6.67	4.69	1.97

**Table B23** Adsorption isotherm on PMMA of 1:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)
20	9.86	0.80	0.41	0.40
40	19.46	1.62	0.83	0.79
60	34.80	1.99	1.00	0.98
80	42.21	2.99	1.54	1.45
100	54.80	3.57	1.79	1.78
200	139.73	4.76	2.39	2.37
300	229.89	5.54	2.84	2.70
400	322.18	6.12	3.04	3.08
500	413.15	6.85	3.43	3.42
600	513.15	6.86	3.44	3.42
700	612.34	6.89	3.48	3.41
800	713.99	6.76	3.42	3.34
900	813.15	6.87	3.44	3.43
1000	912.05	6.93	3.45	3.48
3000	2915.09	6.70	3.41	3.29
5000	4915.09	6.67	3.40	3.27
7000	6915.09	6.71	3.42	3.29
10000	9915.09	6.71	3.42	3.29

**Table B24** Adsorption isotherm on PMMA of 1:3 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ PMMA)
20	9.93	0.79	0.20	0.59
40	20.68	1.52	0.34	1.18
60	31.17	2.27	0.56	1.70
80	41.37	3.05	0.75	2.30
100	53.51	3.66	0.90	2.76
200	131.04	5.42	1.36	4.06
300	217.66	6.49	1.63	4.86
400	311.46	6.98	1.79	5.19
500	411.60	6.97	1.78	5.19
600	511.73	6.96	1.77	5.19
700	611.06	7.02	1.82	5.20
800	712.03	6.94	1.81	5.13
900	812.41	6.92	1.72	5.20
1000	912.00	6.95	1.75	5.20
3000	2913.06	6.85	1.60	5.25
5000	4910.36	7.06	1.81	5.25
7000	6910.36	7.06	1.81	5.25
10000	9911.71	6.94	1.70	5.24

**Table B25** Adsorption isotherm on PMMA of 0:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> PMMA)
20	6.56	1.06
40	10.66	2.32
60	18.03	3.30
80	23.77	4.45
100	31.15	5.44
200	104.10	7.54
300	200.00	7.88
400	298.36	8.04
500	400.00	7.89
600	497.54	8.09
700	596.72	8.12
800	698.36	8.03
900	797.54	8.09
1000	898.36	8.00
3000	2901.64	7.77
5000	4901.64	7.74
7000	6901.64	7.74
10000	9901.64	7.76

**Table B26** Adsorption isotherm on Nylon66 of 1:0 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> Nylon66)
20	10.27	0.29
40	13.24	0.81
60	16.49	1.38
80	20.00	1.85
100	23.24	2.42
200	44.05	4.91
300	122.16	5.63
400	200.54	6.27
500	295.68	6.37
600	397.57	6.35
700	495.95	6.39
800	595.14	6.37
900	696.76	6.39
1000	811.35	6.40
3000	2797.30	6.37
5000	4800.00	6.29
7000	6794.59	6.35
10000	9797.30	6.37

**Table B27** Adsorption isotherm on Nylon66 of 3:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Total amount of surfactant adsorbed (μmole/m <sup>2</sup> Nylon66)	Cationic surfactant adsorbed (μmole/m <sup>2</sup> Nylon66)	Nonionic surfactant adsorbed (μmole/m <sup>2</sup> Nylon66)
20	9.80	0.32	0.24	0.08
40	13.33	0.99	0.75	0.25
60	16.60	1.41	1.06	0.35
80	18.78	2.17	1.64	0.53
100	21.78	2.49	1.88	0.61
200	62.59	4.02	3.02	1.00
300	148.85	5.05	3.80	1.25
400	222.59	5.50	4.13	1.37
500	308.31	6.21	4.67	1.54
600	388.02	6.46	4.83	1.63
700	477.03	7.06	5.32	1.74
800	573.11	7.02	5.32	1.70
900	672.68	7.06	5.33	1.74
1000	771.44	7.07	5.32	1.75
3000	2810.81	7.17	5.27	1.89
5000	4791.89	7.02	5.33	1.69
7000	6740.09	7.08	5.26	1.82
10000	9771.40	7.03	5.24	1.80

**Table B28** Adsorption isotherm on Nylon66 of 1:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ Nylon66)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ Nylon66)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ Nylon66)
20	4.66	0.46	0.23	0.22
40	8.22	0.93	0.47	0.47
60	16.71	1.31	0.65	0.66
80	20.00	1.88	0.94	0.94
100	31.53	2.13	1.09	1.04
200	82.45	3.67	1.82	1.85
300	132.91	5.15	2.61	2.54
400	210.16	5.92	2.99	2.94
500	292.68	6.51	3.35	3.17
600	367.43	7.24	3.66	3.58
700	443.85	7.92	3.98	3.94
800	532.91	7.96	4.01	3.95
900	637.79	7.83	3.90	3.93
1000	743.33	8.01	4.06	3.95
3000	2736.94	8.09	3.99	4.10
5000	4742.57	8.01	4.12	3.89
7000	6742.57	8.10	4.16	3.93
10000	9736.94	8.10	3.99	4.10

**Table B29** Adsorption isotherm on Nylon66 of 1:3 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration ( $\mu\text{M}$ )	Equilibrium concentration ( $\mu\text{M}$ )	Total amount of surfactant adsorbed ( $\mu\text{mole/m}^2$ Nylon66)	Cationic surfactant adsorbed ( $\mu\text{mole/m}^2$ Nylon66)	Nonionic surfactant adsorbed ( $\mu\text{mole/m}^2$ Nylon66)
20	6.89	0.17	0.04	0.13
40	9.10	0.39	0.09	0.29
60	15.74	0.56	0.15	0.41
80	17.95	0.78	0.20	0.58
100	22.36	0.98	0.25	0.72
200	40.56	2.01	0.51	1.50
300	58.36	3.04	0.70	2.34
400	73.58	4.09	0.99	3.10
500	105.38	4.96	1.24	3.72
600	128.81	5.90	1.46	4.43
700	159.41	6.81	1.69	5.13
800	207.12	7.45	1.84	5.61
900	257.25	8.08	1.96	6.12
1000	300.70	8.76	2.19	6.57
1200	504.84	8.75	2.19	6.55
1500	810.27	8.68	2.20	6.48
2000	1302.12	8.78	2.22	6.56
2500	1798.96	8.80	2.23	6.57
3000	2309.01	8.73	2.20	6.53
5000	4300.68	8.80	2.19	6.61
7000	6309.01	8.70	2.20	6.51
10000	9304.73	8.74	2.14	6.60

**Table B30** Adsorption isotherm on Nylon66 of 0:1 molar ratio of CPB:OP(EO)<sub>10</sub>

Initial concentration (μM)	Equilibrium concentration (μM)	Amount of surfactant adsorbed (μmole/m <sup>2</sup> Nylon66)
20	5.00	0.19
40	6.67	0.42
60	10.83	0.62
80	12.50	0.85
100	17.50	1.04
200	26.67	2.17
300	45.00	3.20
400	54.17	4.46
500	78.33	5.53
600	84.17	6.73
700	132.50	7.68
800	210.83	10.27
900	301.67	10.18
1000	511.67	10.30
3000	2675.00	10.19
5000	4525.00	10.36
7000	6658.33	10.64
10000	9675.00	10.13

### Appendix C Contact Angle of Surfactant Solutions

**Table C1** Contact angle for solutions of CPB, OP(EO)<sub>10</sub>, and their mixtures on HDPE

Total surfactant concentration (μM)	Contact angle (degree)				
	α = 0	α = 0.25	α = 0.50	α = 0.75	α = 1.0
20	88.33	82.77	82.41	81.37	78.90
40	86.70	80.50	79.83	75.63	71.20
60	84.14	79.21	78.62	74.77	68.30
80	81.10	76.24	78.79	70.20	65.76
100	78.73	73.77	71.52	66.40	61.14
200	76.45	68.50	67.77	56.07	44.16
300	74.24	65.57	60.37	50.73	44.27
400	72.11	60.42	53.82	50.77	44.13
500	70.81	56.60	53.81	50.73	43.91
600	62.66	56.76	53.91	50.53	44.26
700	62.87	56.54	53.72	50.75	44.05
800	62.26	56.62	53.82	50.73	44.18
900	62.85	56.68	53.87	50.73	44.14
1,000	62.65	56.62	53.88	50.75	44.09
3,000	62.52	56.43	53.81	50.74	44.16
5,000	62.73	56.46	53.82	50.81	44.12
7,000	62.70	56.60	53.82	50.73	44.22
10,000	62.63	56.65	53.81	50.71	44.11

**Table C2** Contact angle for solutions of CPB, OP(EO)<sub>10</sub>, and their mixtures on PC

Total surfactant concentration (μM)	Contact angle (degree)				
	α = 0	α = 0.25	α = 0.50	α = 0.75	α = 1.0
20	82.53	81.72	78.61	78.61	76.87
40	80.48	78.05	71.92	71.92	70.29
60	78.65	76.45	67.88	67.88	65.22
80	75.60	74.16	64.09	64.09	59.31
100	70.11	66.13	59.84	59.84	53.68
200	67.21	62.37	52.72	52.72	41.96
300	65.60	55.78	49.77	49.77	41.87
400	63.46	54.88	79.52	49.52	42.02
500	61.40	54.70	48.87	48.87	41.92
600	58.16	54.60	49.89	49.89	41.96
700	58.35	54.49	49.82	49.82	41.83
800	58.28	54.81	49.62	49.62	41.94
900	58.12	54.79	49.60	49.60	41.96
1,000	57.97	54.61	49.74	49.74	42.07
3,000	58.17	54.67	49.62	49.62	41.87
5,000	58.21	54.77	49.78	49.78	42.04
7,000	58.15	54.74	49.80	49.80	41.96
10,000	58.09	54.70	49.68	49.68	41.92

**Table C3** Contact angle for solutions of CPB, OP(EO)<sub>10</sub>, and their mixtures on PVC

Total surfactant concentration (μM)	Contact angle (degree)				
	α = 0	α = 0.25	α = 0.50	α = 0.75	α = 1.0
20	81.65	80.35	79.04	77.85	76.36
40	79.23	78.27	77.41	75.59	69.52
60	76.32	74.64	73.02	72.93	60.59
80	75.31	72.27	70.07	68.99	53.37
100	73.64	70.15	69.88	63.11	44.46
200	70.33	65.20	62.45	60.03	37.40
300	67.60	60.71	55.33	49.04	37.39
400	65.49	56.43	50.65	48.92	37.48
500	61.41	53.80	50.71	49.89	37.37
600	55.84	53.06	50.60	49.03	37.42
700	55.13	53.83	50.68	49.16	37.44
800	54.88	53.76	50.69	49.05	37.39
900	55.22	53.77	50.67	49.12	37.44
1,000	55.47	53.01	50.65	49.02	37.39
3,000	55.48	53.73	50.67	48.93	37.43
5,000	55.75	53.86	50.64	49.00	37.42
7,000	55.53	53.55	50.61	49.02	37.41
10,000	55.63	53.85	50.61	49.09	37.40

**Table C4** Contact angle for solutions of CPB, OP(EO)<sub>10</sub>, and their mixtures on ABS

Total surfactant concentration (μM)	Contact angle (degree)				
	$\alpha = 0$	$\alpha = 0.25$	$\alpha = 0.50$	$\alpha = 0.75$	$\alpha = 1.0$
20	81.42	79.57	77.57	76.89	75.42
40	80.11	78.59	77.10	75.05	67.59
60	78.05	76.24	74.96	74.83	58.40
80	76.42	74.63	72.61	69.57	51.43
100	75.40	74.90	70.26	68.40	43.23
200	72.52	66.10	64.78	55.40	36.50
300	72.41	60.37	52.05	47.77	36.38
400	68.62	54.36	49.15	47.66	36.56
500	61.37	52.24	49.16	47.60	36.29
600	55.60	52.22	49.14	47.77	36.77
700	55.54	51.69	49.18	47.41	36.64
800	55.53	52.02	49.15	47.85	36.42
900	55.67	52.78	49.22	47.66	36.51
1,000	55.53	52.33	48.91	47.74	36.19
3,000	55.66	52.26	49.21	47.77	36.49
5,000	55.55	52.06	49.17	47.71	36.49
7,000	55.64	52.32	49.16	47.74	36.65
10,000	55.59	52.08	49.14	47.75	36.48

**Table C5** Contact angle for solutions of CPB, OP(EO)<sub>10</sub>, and their mixtures on PMMA

Total surfactant concentration (μM)	Contact angle (degree)				
	α = 0	α = 0.25	α = 0.50	α = 0.75	α = 1.0
20	74.32	73.56	71.43	70.11	64.66
40	69.26	68.33	68.46	69.41	60.04
60	68.90	65.51	67.51	65.36	57.14
80	66.52	59.85	61.84	60.47	50.84
100	65.23	58.32	58.28	55.97	42.89
200	64.12	55.72	52.11	45.47	35.19
300	61.16	53.42	50.60	42.74	35.01
400	56.66	51.68	44.04	42.34	35.10
500	53.43	46.40	44.00	42.66	35.07
600	50.60	46.27	44.17	42.13	35.14
700	50.98	46.33	44.12	42.61	35.22
800	50.30	46.23	44.08	42.59	35.34
900	50.29	46.20	43.96	42.60	35.01
1,000	50.89	46.63	44.03	42.76	35.34
3,000	50.43	46.43	44.00	42.71	35.22
5,000	50.67	46.41	43.98	42.84	35.13
7,000	50.63	46.40	44.27	42.73	35.11
10,000	50.61	46.28	44.11	42.81	35.26

**Table C6** Contact angle for solutions of CPB, OP(EO)<sub>10</sub>, and their mixtures on Nylon66

Total surfactant concentration (μM)	Contact angle (degree)				
	α = 0	α = 0.25	α = 0.50	α = 0.75	α = 1.0
20	61.42	67.52	65.51	63.58	62.29
40	58.59	59.81	59.37	56.44	54.64
60	56.56	57.34	56.72	53.78	52.14
80	54.77	55.55	54.45	49.60	48.09
100	53.91	53.48	52.01	46.00	41.69
200	51.57	47.22	45.78	43.35	33.91
300	48.14	46.15	43.69	39.56	33.69
400	46.76	44.26	41.31	39.30	33.83
500	43.81	41.70	40.86	39.72	34.09
600	43.53	41.63	41.19	39.42	34.24
700	43.92	41.65	41.48	39.17	33.59
800	43.80	41.62	41.26	39.45	33.70
900	44.04	41.68	41.30	39.64	33.52
1,000	43.19	41.60	41.22	39.26	33.65
3,000	43.99	41.66	41.26	39.68	34.13
5,000	43.95	41.64	41.27	39.47	33.89
7,000	43.70	41.67	41.64	39.54	33.92
10,000	43.97	41.68	41.65	39.52	33.88

## Appendix D Example of Calculation for Surfactant Adsorption Isotherms

### Adsorption for solution of CPB on HDPE

$$\text{Surfactant}_{\text{adsorb}} = \frac{[\text{Surfactant}]_I - [\text{Surfactant}]_E \times V_{\text{sol}}}{1,000 \times W_{\text{plastic}} \times a_s}$$

where

$\text{Surfactant}_{\text{adsorp}}$	=	Adsorption of surfactant, ( $\mu\text{mole}/\text{m}^2$ plastic)
$[\text{Surfactant}]_I$	=	Initial surfactant solution concentration, ( $\mu\text{M}$ )
$[\text{Surfactant}]_E$	=	Equilibrium surfactant solution concentration, ( $\mu\text{M}$ )
$V_{\text{sol}}$	=	Volume of solution, (mL)
$W_{\text{plastic}}$	=	Weight of plastic, (g)
$a_s$	=	Specific surface area of plastic, ( $\text{m}^2/\text{g}$ )

The adsorption isotherm was a plot between adsorption of surfactant on HDPE ( $\mu\text{mole}/\text{m}^2$  HDPE) and concentration of surfactant solution ( $\mu\text{M}$ ).

$$[\text{Surfactant}]_I = 1,000 \mu\text{M}$$

Equilibrium concentration of surfactant was converted from

$$\text{UV-Vis spectrophotometer (wavelength)} \longrightarrow \mu\text{M}.$$

Calibration equation for CPB solution from UV-Vis spectrophotometer,

$$Y = 0.0037X.$$

where

$$X = [\text{Surfactant}]_E, (\mu\text{M})$$

$$Y = \text{Wavelength} = 3.565$$

Substituting into calibration equation,

$$X = 3.565/0.0037 = 963.51 \mu\text{M}$$

Thus, surfactant adsorption for solution of CPB on HDPE at 1,000  $\mu\text{M}$ , initial concentration is

$$\text{Surfactant}_{\text{adsorb}} = \frac{(1,000 - 963.15) \times 20}{1,000 \times 0.2506 \times 1.443} = 2.02 \mu\text{mole}/\text{m}^2 \text{ HDPE}$$

Note: Calibration equation for OP(EO)<sub>10</sub> solution from UV-Vis spectrophotometer,

$$Y = 0.0012X$$

## CURRICULUM VITAE

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