ETHYLENE EPOXIDATION IN A LOW-TEMPERATURE DIELECTRIC BARRIER DISCHARGE SYSTEM: EFFECT OF ELECTRODE GEOMETRY



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ABSTRACT

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Ethylene oxide is a valuable chemical feedstock in producing many industrial chemicals, such as ethylene glycol, solvents, antifreezes, and adhesives. Hence, the partial oxidation of ethylene to ethylene oxide, so-called ethylene epoxidation, has been of great interest in many global research studies. In this work, the epoxidation of ethylene under a cylindrical dielectric barrier discharge (DBD) reactor was initially studied to find the optimum operating conditions and then was compared with that under a parallel DBD reactor. For the cylindrical DBD system, it was found that the ethylene oxide yield increased with decreasing O_2/C_2H_4 molar ratio, under the O₂-lean condition, and decreasing feed flow rate; however, there were optimum applied voltage and input frequency to obtain the highest ethylene oxide yield. The highest ethylene oxide yield of 2.41% was achieved when an O_2/C_2H_4 molar ratio of 0.25:1 (1:4), an applied voltage of 15 kV, an input frequency of 500 Hz, a feed flow rate of 50 cm³/min, and electrode gap distance of 5 mm were used. Under these optimum conditions, the power consumption was found to be 12.72×10^{-10} ¹⁶ Ws/molecule of ethylene oxide produced. The optimum conditions were used to comparatively investigate the epoxidation performance with the parallel DBD system. It was found that at the optimum conditions, the cylindrical DBD system still exhibited higher epoxidation performance. Therefore, the cylindrical DBD system was found to exhibit a high potential to produce ethylene oxide from ethylene epoxidation reaction.

บทคัดย่อ

ณัฐวรนันท์ เพิ่มสิน: การอีพอกซิเคชันของเอธิลีนภายใต้ระบบพลาสมาอุณหภูมิต่ำชนิด ใดอิเล็กทริกแบริเออร์ดิสชาร์จ: ผลของการวางตัวของขั้วไฟฟ้า (Ethylene Epoxidation in a Low-Temperature Dielectric Barrier Discharge System: Effect of Electrode Geometry) อ. ที่ปรึกษา : ผศ. คร. ธรรมนูญ ศรีทะวงศ์ และ รศ. คร. สุเมธ ชวเดช 70 หน้า

เอธีลีนออกไซค์เป็นผลิตภัณฑ์ปิโตรเคมีขั้นกลางที่มีบทบาทสำคัญอย่างยิ่งในการผลิต ผลิตภัณฑ์อุตสาหกรรมหลากหลายชนิด เช่น เอธิลีนไกลคอล, ตัวทำละลาย, สารต้านการเยือกแข็ง, และสารที่ใช้สำหรับการเชื่อมติด ด้วยเหตุนี้กระบวนการอีพอกซิเดชันของเอธีลืนไปเป็นเอธีลืนอ อกไซค์ซึ่งเป็นกระบวนการออกซิเคชันที่ไม่สมบูรณ์ของสารอินทรีย์ จึงเป็นกระบวนการที่ ้น่าสนใจอย่างยิ่งสำหรับการศึกษาเพื่อผลิตเอธีลีนออกไซค์อย่างกว้างขวาง ในงานวิจัยนี้ ้ กระบวนการอีพอกซิเคชันของเอธิลีนไปเป็นเอธิลีนออกไซค์ถูกทำการทคลองในเครื่องปฏิกรณ์ พลาสมาชนิด ใดอิเล็คทริคแบริเออร์ดิสชาร์จแบบทรงกระบอกเพื่อศึกษาสภาวะต่างๆที่เหมาะสม ในการเกิดปฏิกิริยาเป็นลำดับแรก และทำการเปรียบเทียบประสิทธิภาพกับเครื่องปฏิกรณ์พลาสมา ชนิดใดอิเล็กทริกแบริเออร์ดิสชาร์จแบบทรงสี่เหลี่ยมด้านขนาน จากการทดลองพบว่า สำหรับ ระบบพลาสมาชนิคไคอิเล็คทริคแบริเออร์คิสชาร์จแบบทรงกระบอก ผลได้ของเอธีลีนออกไซค์ เพิ่มขึ้นเมื่อทำการลดอัตราส่วนของออกซิเงนต่อเอธีลีนภายใต้สภาวะขาดแคลนออกซิเงน และลด อัตราการไหลของสารตั้งต้น อย่างไรก็ตาม จากผลทดลองพบอีกว่าก่าความต่างศักย์และก่าความถึ่ ที่เหมาะสมที่ทำให้ได้ค่าผลผลิตของเอธีลีนออกไซค์สูงที่สุด ซึ่งผลได้ของเอธีลีนออกไซค์มาก ที่สุดคือ 2.41 เปอร์เซ็นต์ เมื่อใช้อัตราส่วนของออกซิเจนต่อเอธีลินเป็น 0.25:1 (1:4), ความต่างศักย์ 15 กิโลโวลต์ ค่าความถี่ 500 เฮิรตซ์ ด้วยอัตราใหลของสารตั้งต้นเป็น 50 ถูกบาศก์เซนติเมตรต่อ ้นาที และระยะห่างระหว่างขั้วไฟฟ้า 5 มิลลิเมตร ภายใต้สภาวะที่เหมาะสมคังกล่าว พลังงานที่ใช้ ในการผลิตเอธีลีนออกไซด์ เท่ากับ 12.72x10⁻¹⁶ วัตต์วินาทีต่อโมเลกุล นอกจากนี้ เมื่อทำการ เปรียบเทียบประสิทธิภาพการเกิดปฏิกิริยาอีพอกซิเคชันกับระบบพลาสมาชนิดไคอิเล็กทริกแบริ เออร์คิสชาร์จแบบทรงสี่เหลี่ยมค้านขนาน ผลการทคลองพบว่า ภายใต้สภาวะที่เหมาะสมเครื่อง ู่ปฏิกรณ์พลาสมาอุณหภูมิต่ำชนิคไคอิเล็คทริคแบริเออร์คิสชาร์จแบบทรงกระบอกให้ประสิทธิภาพ การเกิดปฏิกิริยาอีพอกซิเคชันดีกว่า ดังนั้น ระบบพลาสมาชนิดไดอิเล็คทริคแบริเออร์ดิสชาร์จแบบ ทรงกระบอกจึงมีศักยภาพสูงสำหรับใช้ในการผลิตเอธีลินออกไซค์จากปฏิกิริยาอีพอกซิเคชันของ เอริลิน

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