REFERENCES

- Ballegooie, P. V., and Rudin, A. (1988). Reactive extrusion of polystyrene / polyethylene blends, <u>Polymer Engineering and Science</u>, 28(21), 1434-1442.
- Barentsen, W. M., and Heikens, D. (1973). Effect of graft copolymer of LDPE with PS (PS-g-LDPE) on LDPE/PS blends, <u>Polymer</u>, 14, 579-586.
- Boutevin, B., Lusinchi, J. M., Pietrasanta, Y., and Robin, J. J. (1996, March). Improving poly(ethylene terephthalate)/high-density polyethylene blends by using grafted copolymer, <u>Polymer Engineering and Science</u>, 36(6), 879-884.
- Boutevin, B., Pietrasanta, Y., Taka, M., and Sarraf, T. (1985). Influence of grafted copolymer P(E-g-MMA) on the mechanical properties of PE-PVC alloy, <u>Polymer Bulletin</u>, 14, 25-30.
- Di Lorenzo, M. L. and Frigiane, M. (1997). Compatibilization criteria and procedures for binary blends: A review, <u>Journal of Polymer</u>
 <u>Engineering</u>, 17(6), 429-459.
- Dompas, D., Croaeninckx, G., Isogawa, M., Haasegawa, T., and Kadokura, M. (1997). Miscibility behavior of blends of poly(vinyl chloride) with poly (methyl methacrylate-co-styrene) copolymers, <u>Polymer</u>, 36(2), 421-429.
- Folkes, M. J., and Hope, P. S. (Eds.) (1993). <u>Polymer Blends and Alloy</u>, London, England: Cambridge University Press.
- Gallucci, R. R., and Going, R. C. (1988). Preparation and reactions of epoxymodified polyethylene, <u>Journal of Applied Polymer Science</u>, 27, 425-437.

- He, P., Huang, H., Xiao, W., Huang, S., and Chang, S. (1997). A study on PVC/LLDPE blends with solid state chlorinated polyethylene as compatibilizer, <u>Journal of Applied Polymer Science</u>, 64, 2535-2541.
- Hojabr, S., Barker, W. E., Russell, K. E., McLellan, P. J., and Huneault, M. A. (1998). Melt grafting glycidyl methacrylate onto polyethylene,

 <u>International Polymer Processing</u>, 2, 118-128.
- Hu, G. H., Sun, Y. J., and Lambla, M. (1996). Effects of processing parameters on the *in situ* compatibilization of polypropylene and poly (buthylene terephthalate) blends by one-step reactive extrusion, <u>Journal of Applied Polymer Science</u>, 61, 1039-1047.
- Kalfoglou, N. K., Skafidas, D. S., Kallistsis, J. K., Lambert, J. C., and Stappen, L. V. (1995). Comparison of compatibilizer effectiveness for PET/HDPE blends, <u>Polymer</u>, 36(23), 4453-4462.
- Kalfoglou, N. K., Skafidas, D. S., and Sofiripoulou, D. (1994).
 Compatibilization of blends of poly(ethylene terephthalate) and linear low density polyethylene with ionomer of poly(ethylene-comethacrylate), Polymer, 36(17), 3624-3630.
- Kroeze, E., Brinke, G., and Hadziioannou, G. (1997). Compatibilization of low density polyethylene and poly(vinyl chloride) by segmented EB (SAN-block-EB)_n block copolymers, <u>Polymer</u>, 38(2), 379-389.
- Kulshreshtha, A. K. (1993). A review of commercial polymer blends based on PVC, ABS, and PC, Polymer and Plastic Technology and Engineering, 32(6), 5571-578.
- Lee, Y. D., and Chang, K. C. (1985). Effects of poly(methyl methacrylate-con-butyl acrylate) on the properties and processing of poly(vinyl chloride), Journal of Applied Polymer Science, 30, 3605-3616.

- Lemieux, E., Prud'homme, R. E., Forle, R., Jerome, R., and Teyssie, P. (1988). Influence of the tacticity of poly(methyl methacrylate) on its miscibility with chlorinated polymers, <u>Macromolecules</u>, 21, 2148-2154.
- Lu, X., and Weiss, R. A. (1991). Development of miscible blends of polyamide-6 and manganese sulfonated polystyrene using specific interactions, <u>Macromolecules</u>, 24, 4381-4385.
- Lui, X. C., Baker, W. E., and Russell, K. E. (1990). Functionalization of polyethylene and their used in reactive blending, <u>Journal of Applied</u>
 <u>Polymer Science</u>, 41, 2285-2300.
- Nakamura, Y., Watanabe, A., Mori, K., and Tamura, K. (1987). Improvement in low-temperature toughness of poly(vinyl chloride) by co-
- crosslinking with polyethylene, <u>Journal of Polymer Science: Part C: Polymer</u>
 Letters, 25, 127-129.
- Park, S. H., Park, K. Y., and Suh, K. D. (1998). Compatibilizing effect of isocyanate function group on polyethylene terephthalate/low density polyethylene blends, <u>Journal of Polymer Science: Part B: Polymer Physics</u>, 36, 447-453.
- Parmer, J. F., Dickinson, L. C., Chein, J. C., and Porter, R. S. (1989).

 Miscibility of poly(methacylate)/poly(vinyl chloride) blends,

 Macromolecules, 22, 1078-1083.
- Paul, D. R., and Newman, S. (Eds.) (1978). <u>Polymer Blends</u>, New York: Academic Press.
- Sathe, S. N., Srinivasa, G. S., Rao, K. V., and Devi, S. (1996, Mid-October). The effect of composition on morphological, thermal, and mechanical properties of polypropylene/nylon-6/polypropylene-g-buthyl acrylate blends, Polymer Engineering and Science, 36(19), 2443-2450.

- Shing, J. B. W., Baker, W. E., Russell, K. E., and Whitney, R. A. (1994).

 Effect of reaction conditions on the grafting of 2-(dimethylamine-1-ethyl methacrylate) on hydrocarbon substrate, <u>Journal of Polymer Science</u>: Part A: Polymer Chemistry, 32, 1691-1702.
- Siemann, A., and Baker, W. E. (1989, August). Basic functionalization of polyethylene in the melt, <u>Polymer Engineering and Science</u>, 29(16), 1117-1123.
- Sun, Y. J., Hu, G. H., and Lambla, M. (1995). Free radical grafting glycidyl methacrylate onto polypropylene in a co-rotating twin screw extruder, <u>Journal of Applied Polymer Science</u>, 57, 1043-1054.
- Valenza, A., Sparado, G., Galderaro, E., and Acierno, D. (1993, Mid-October). Mechanical properties-structure relationship in blends of polyamide-6 with γ-irradiated LDPE, <u>Polymer Engineering and Science</u>, 33(20),1336-1340.
- Wang, X., Tzoganakis, C., and Rempel, G. L. (1996). Chemical modification of polypropylene with peroxide, pentaacrythritol triacrylate by reactive extrusion, <u>Journal of Applied Polymer Science</u>, 61, 1395-1404.
- Xanthos, M. (Eds.) (1992). <u>Reactive Extrusion</u>, New York: Oxford University Press.
- Xanthos, M., Young, M. W., and Biesenberger, J. A. (1990, March).

 Polypropylene/polyethylene blends compatibilized through
 functionalization, Polymer Engineering and Science, 30(6), 355-365.
- Yang, D., Zhang, B., Yand, Y., Fang, Z., Sun, G., and Feng, Z. (1984).

 Improving polypropylene/high density polyethylene blends by using EDPM as compatibilizer, <u>Polymer Engineering and Science</u>, 24(8), 612-624.

- Yoon, L. K., Choi, C. H., and Kim, B. K. (1995). Reactive extrusion of PP / natural rubber blends, <u>Journal of Applied Polymer Science</u>, 56, 239-246.
- Zhong, Z., Zheng, S., Yang, K., and Guo, Q. (1997). Blends of poly(vinyl chloride) with acrylonitrite-chlorinated polyethylene-styrene copolymer: I. Miscibility, phase behavior and thermal properties, <u>Journal of Macromolecular Science-Physics</u>, B36(4), 441-454.

APPENDICES

1. Calculation of percentage of grafting and total reaction efficiency

From FT-IR spectrometer: The integral area carbonyl (CO) peak (~1730 cm⁻¹) and CH bending peak (~1368 cm⁻¹) were obtained. The integral ratio was calculated by followed equation.

Integral ratio = Integral area of CO peak/integral area of CH bending peak

For example: The integral area of CO peak, integral area of C-H bending peak and integral ratio of HDPE modified with 5 % wt MMA and DCP:MMA = 1:60 equivalent mole were shown as followed:

	Crude	Purified
Integral area of CO peak	1.8452	1.2665
Integral area of C-H bending peak	0.6898	0.9411
Integral ratio	2.6749	1.3458

Percent weight of MMA in sample was calculated from the equation that obtained from calibration curve (Figure 3.1).

Integral ratio = 0.9387 x percent weight of MMA in sample or Percent weight of MMA in sample = Integral ratio / 0.9387

Then:

Percent weight of MMA in crude sample = 2.6749/0.9387 = 2.87 %, and Percent weight of MMA in purified sample = 1.3458/0.9387 = 1.44 %

39

Percentage of grafting and total reaction efficiency were calculated by

used percent weight of MMA in purified sample and crude sample which are

weight of MMA grafted on HDPE and weight of MMA converted

respectively.

PG = Weight of MMA grafted on HDPE / Initial weight of MMA x 100

TE = Weight of MMA converted / Initial weight of MMA x 100

Then

 $PG = 1.44/5 \times 100 = 28.8 \%$

 $TE = 2.87/5 \times 100 = 57.33 \%$

2. Example of raw data from GPC analysis

Millenium Sample Information

Project Name:

PE

Sample Name:

HDPE H6240B

Vial: Injection: 11

AM

Channel:

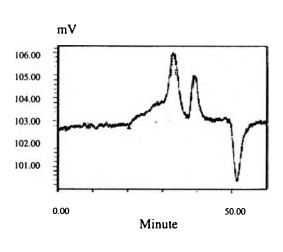
SATIN

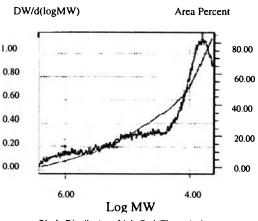
Date Acquired:

28/01/99 02:04:05 PM

Scale Factor:
Acq Melt Set:
Processing Method:

1.00 1.00 aun l





Sample Type:

Date Processed: 17/02/99

Volume:

Dilution:

Run Time:

Broad Unknown

11:17:26

100.00

60.0 min

1.00000

Black: Distribution Light Red: Theoretical Green: Observed Gyan: Fitted Blue: g' Magenta: Cumulative

Peak Results

#	Name	Ret Time (min)	% Area	Mn (Daltons)	Mw (Daltons)	Polydispersity
1	HDPE H6240B	33.200	100.00	12750	154870	12.141468

Broad Unknown

11:16:58

100.00

60.0 min

1.00000

Millenium Sample Information

Project Name:

PE

Sample Name:

5 % MMA 1:60 DCP

Vial:

12

Injection:

1

AM

Channel:

109.00

108.00

107.00

106.00

105.00

104.00

103.00

102.00

0.00

SATIN

Date Acquired:

28/01/99 03:07:44 PM

Scale Factor:

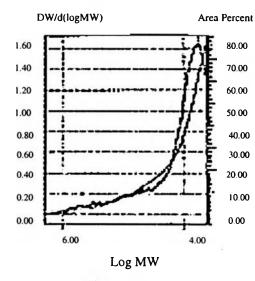
1.00 1.00

Acq Melt Set: Processing Method:

mV

aun l

Minute



Sample Type:

Date Processed: 17/02/99

Volume:

Dilution:

Run Time:

Black: Distribution Light Red: Theoretical Green: Observed Gyan: Fitted Blue: g' Magenta: Cumulative

Peak Results

50.00

#	Name	Ret Time (min)	% Area	Mn (Daltons)	Mw (Daltons)	Polydispersity
l	5% MMA 1:60 DCP	31.437	100.00	7567	97259	12.85

CURRICULUM VITAE

Name: Chaiwat Towichayathamrong

Date of Birth: August 14, 1973

Nationality: Thai

University Education:

1991-1994 Bachelor Degree of Science in Material Science,

Chulalongkorn University.

Working Experience:

1995-1996 Position: Sale and customer service engineer

Company name: Thai MFC Co., Ltd.

1996-1997 Position: Quality control engineer

Company name: Thai MFC Co., Ltd.