

REFERENCES

- Arusingkarat, K., Rirksomboon, T., Meeyoo, V. and Pengpanich, S. (2012) Catalytic Partial Oxidation of Methane over NiO-MgO/Ce_{0.75}Zr_{0.25}O₂ Catalysts: Effects of Low Mg Content and Incorporated Sequences. Unpublished.
- Bampenrat, A., Meeyoo, V., Kitiyanan, B., Rangsunvigit, P., and Risksomboon, T. (2008) Catalytic oxidation of naphthalene over CeO₂-ZrO₂ mixed oxide catalysts. Catalysis Communications, 9, 2349–2352.
- Bharadwaj, S.S. and Schmidt, L.D. (1995) Fuel Processing Technology, 42,109.
- Bokhimi, X., Morales, A., Novaro, O., Portilla, M., Lopez, T., Tzompantzi, F. and Gomez R. (1998) Tetragonal Nanophase Stabilization in Nondoped Sol-Gel Zirconia Prepared with Different Hydrolysis Catalysts. Journal of solid state chemistry, 135, 28-35.
- Bradford, M.C.J. and Vannice, M.A. (1999) Catalysis Reviews-science and Engineering, 41, 1–42.
- Chen, H., Xue, M., Hu, S. and Shen, J. (2012) The effect of surface acidic and basic properties on the hydrogenation of laurionitrile over the supported nickel catalysts. Chemical Engineering Journal, 181– 182, 677– 684.
- Chubb, T.A. (1980) Solar Energy Materials and Solar Cells, 24, 341.
- Fischer, V.F. and Tropsch, H. (1928) Brennst. Chemistry, 3, 39.
- Gamman, J.J., Millar, G.J., Rose, G. and Drennan, J. (1998) Journal of the Chemical Society, Faraday Transactions, 94, 701.
- Ghiotti, G. and Boccuzzi, F. (1987) Catalysis Reviews: Science and Engineering, 29, 151.
- Hori, C.E., Permana, H., Ng, K.Y.S., Brenner, A., More, K., Rahmoeller, K.M. and Belton, D.N. (1998) Applied Catalysis B: Environmental, 16, 105.
- Jang, W-J., Jeong, D-W., Shim, J-O., Roh, H-S., Son, I. H. and Jae Lee, S. (2013) H₂ and CO production over a stable Ni/MgO/Ce_{0.8}Zr_{0.2}O₂ catalyst from CO₂ reforming of CH₄. International Journal of Hydrogen Energy, 1-5.

- Mazzieri, V.A., Pieck, C.L., Vera, C.R., Yori, J.C., and Grau J.M. (2009) Effect of Ge content on the metal and acid properties of Pt-Re-Ge/Al₂O₃-Cl catalysts for naphtha reforming. Applied Catalysis A: General, 353, 93–100.
- Meirovitch, E. and Segal, A. (1991) Solar Energy Materials and Solar Cells, 46, 219.
- Pompeo, F., Nichio, N.N., Ferretti, O.A. and Resasco, D. (2005) Study of Ni catalysts on different supports to obtain synthesis gas. International Journal of Hydrogen energy, 30, 1399–1405.
- Richardson, J.T. and Paripatyadar, S.A. (1990) Applied Catalysis A: General, 61, 293.
- Rostrup-Nielsen, J.R., Anderson, R., and Boudart, M. (1984) in Catalysis: Science and Technology, 5, 1.
- Sato, K., Sago, F., Sago, F., Nagaoka, K. and Takita, Y. (2010) Preparation and characterization of active Ni/MgO in oxidative steam reforming of n-C₄H₁₀. International Journal of Hydrogen Energy, 11, 5393–5399.
- Sukkaeo, R., Rirksomboon, T., and Meeyoo, V. (2013) Catalytic Partial Oxidation of Methane over NiO/Ce_{0.75}Zr_{0.25-x}Mg_{2x}O₂-based catalysts. Unpublished.
- Thaicharoensutcharitthama, S., Meeyoo, V., Kitiyanan, B., Rangsunvigit, P., and Risksomboon, T. (2011) Hydrogen production by steam reforming of acetic acid over Ni-based catalysts. Catalysis Today, 164, 257–261.
- Thammachart, M., Meeyoo, V., Risksomboon, T., and Osuwan, S. (2001) Catalytic activity of CeO₂-ZrO₂ mixed oxide catalysts prepared via sol-gel technique: CO oxidation. Catalysis Today, 68, 53–61.
- Wang, J.A., Bokhimi, X., Novaro, O., Lo'pez, T., Tzompantzi, F., Go'mez, R., Navarrete, J., Llanos, M.E. and Lo'pez-Salinas, E. (1999) Effects of structural defects and acid-basic properties on the activity and selectivity of isopropanol decomposition on nanocrystallite sol-gel alumina catalyst. Journal of Molecular Catalysis A: Chemical, 137, 239–252.
- Wang, Y.H., Liu H.M. and Xu, B.Q. (2009) Durable Ni/MgO catalysts for CO₂ reforming of methane: Activity and metal-support interaction. Journal of Molecular Catalysis A: Chemical, 299, 44–52.
- Wender, I. (1996) Fuel Processing Technology, 48, 189.
- Zhang, G., Hattori, H. and Tanabe K. (1988) Applied Catalysis A: General, 36, 189.

APPENDICES

Appendix A Experimental Data of Gas Calibration of GC-8A

Condition :Detector Current 140 mA

Temperature	°C	Column	Model
Column	25	Alltech	CTR I
Detector	120		
TCD-T	120		
Pressure	kPa		
Carrier Pressure (Primary)	600		
Carrier Pressure (1)	50		
TCD-Ref	10		

1. Methane

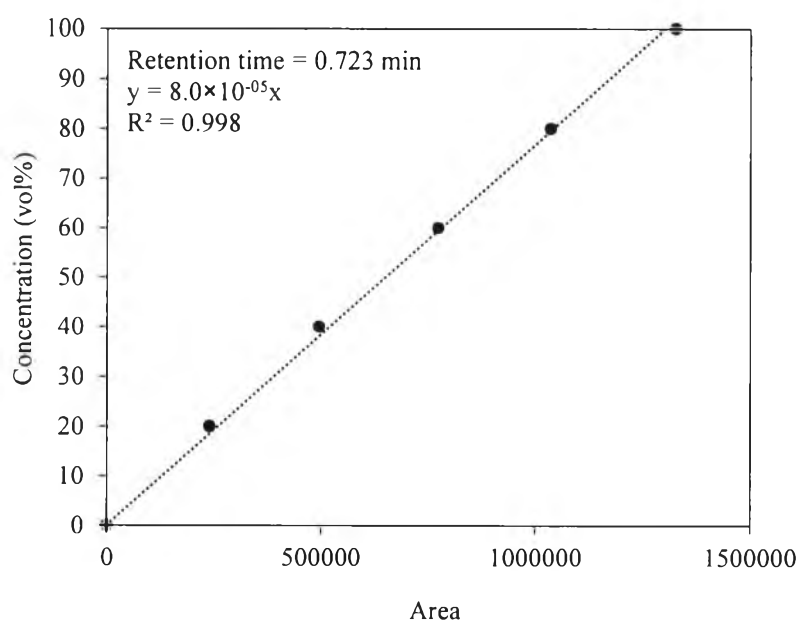


Figure A1 Relationship between area and concentration of methane.

2. Carbon Dioxide

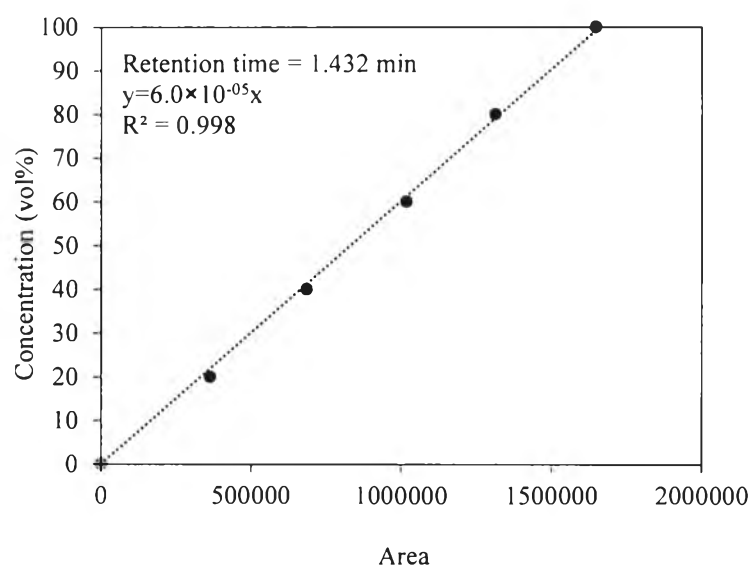


Figure A2 Relationship between area and concentration of carbon dioxide.

3. Hydrogen

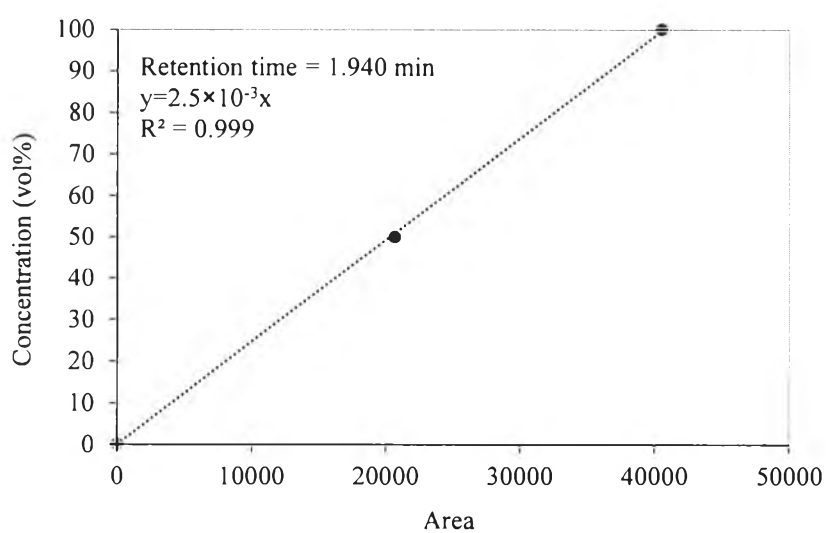


Figure A3 Relationship between area and concentration of hydrogen.

4. Carbon monoxide

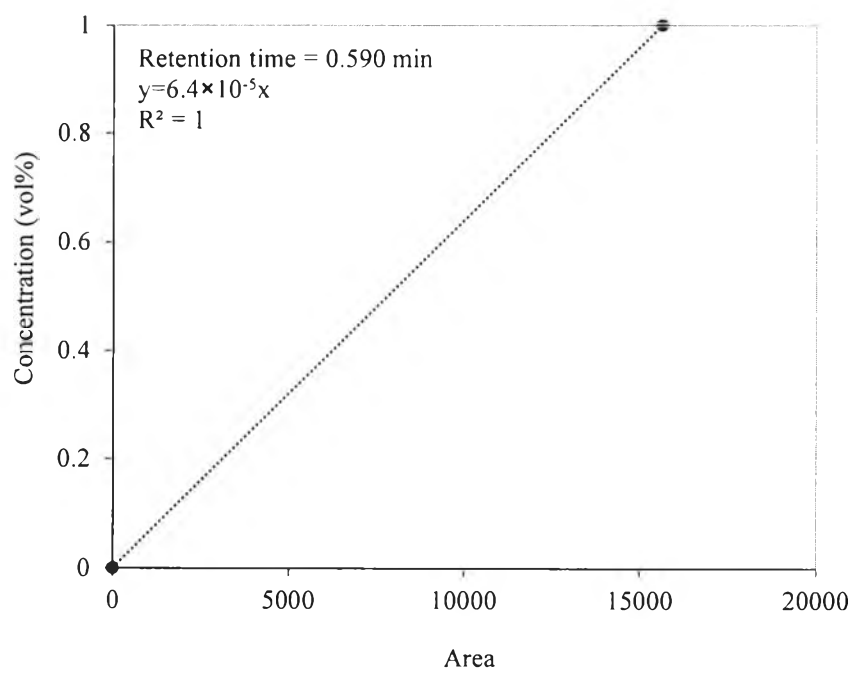


Figure A4 Relationship between area and concentration of carbon monoxide.

Appendix B Calibration of Brooks 5850E Mass Flow Controllers

1. Methane

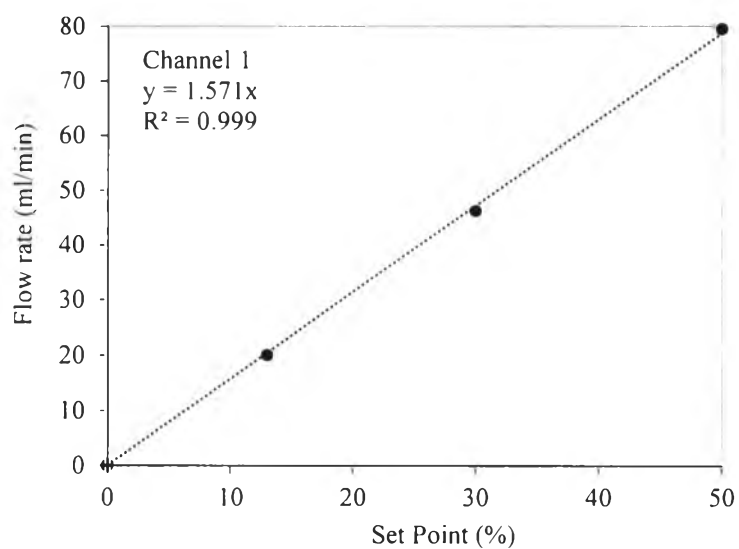


Figure B1 Relationship between SP and flow rate of methane.

2. Hydrogen

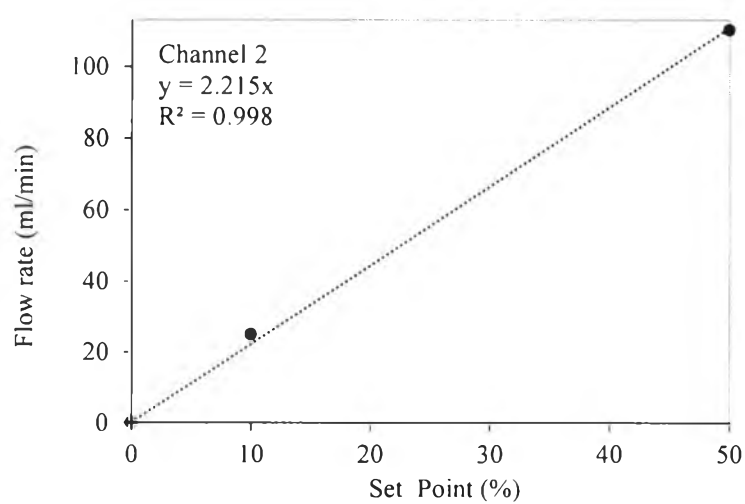


Figure B2 Relationship between SP and flow rate of hydrogen.

3. Carbon Dioxide

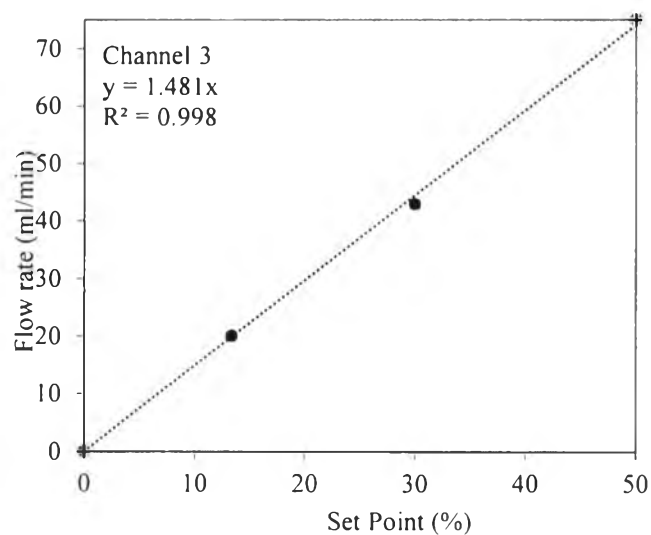


Figure B3 Relationship between SP and flow rate of carbon dioxide.

4. Helium

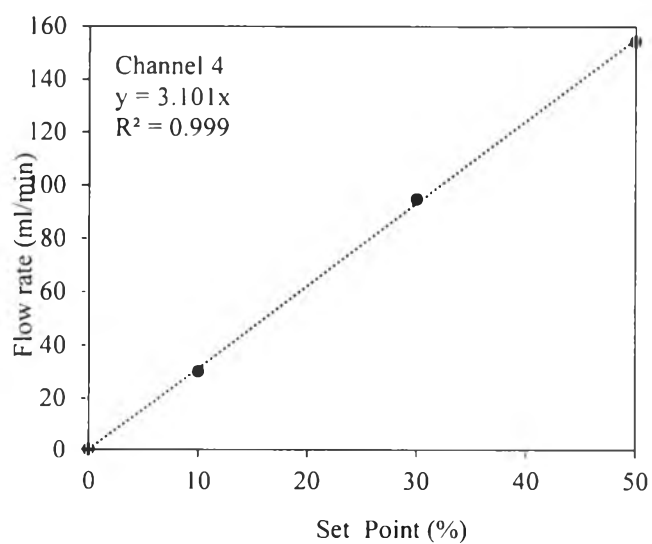


Figure B4 Relationship between SP and flow rate of helium.

Appendix C Experimental Data of Catalytic Activity Tests for Methane Dry Reforming

Table C1 Catalytic activity test of Ni/CZO-S catalyst at various temperature, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Temperature (°C)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
500	10.9	18.2	19.3	16.4	0.85
550	24.2	31.0	26.9	25.2	0.93
600	37.3	44.5	37.5	35.5	0.94
650	48.7	59.4	52.8	46.0	0.87
700	61.4	69.4	61.5	55.4	0.90
750	74.1	79.6	69.1	64.0	0.93
800	84.1	87.3	75.8	70.3	0.93
850	90.1	92.6	78.7	75.5	0.96
900	91.9	95.5	79.6	78.2	0.98

Table C2 Catalytic activity test of Ni/CZN-1 catalyst at various temperature, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Temperature (°C)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
500	12.6	22.9	25.6	19.1	0.75
550	28.2	35.8	35.1	28.7	0.82
600	46.3	51.3	52.3	40.0	0.76
650	62.5	66.8	66.6	51.7	0.78
700	73.8	76.6	74.3	61.6	0.83
750	82.4	84.2	77.5	69.2	0.89
800	88.0	89.3	80.5	74.0	0.92
850	91.7	94.4	82.6	78.1	0.94
900	92.3	95.0	83.6	79.5	0.95

Table C3 Catalytic activity test of Ni/CZN-2 catalyst at various temperature, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Temperature (°C)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
500	15.2	22.7	15.5	17.8	1.15
550	29.4	34.9	25.8	27.0	1.04
600	45.3	49.9	39.7	38.3	0.96
650	59.6	64.6	56.5	50.4	0.89
700	72.1	76.1	66.0	59.7	0.90
750	81.3	83.7	73.9	66.2	0.89
800	87.8	89.3	80.0	71.1	0.89
850	90.3	94.4	83.8	76.0	0.91
900	91.1	96.5	84.6	77.8	0.92

Table C4 Catalytic activity test of Ni/CZO-C catalyst at various temperature, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Temperature (°C)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
500	12.7	15.8	22.3	17.7	0.79
550	26.9	29.7	35.7	28.7	0.80
600	45.9	47.8	47.4	41.3	0.87
650	57.6	62.1	59.2	53.4	0.90
700	70.3	74.2	68.6	63.8	0.93
750	81.0	83.4	75.2	71.2	0.94
800	91.2	90.5	82.3	75.8	0.92
850	94.1	95.1	84.7	78.8	0.93
900	94.8	97.4	86.6	79.9	0.93

Table C5 Catalytic activity test of Ni/CZM-1 catalyst at various temperature, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Temperature (°C)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
500	18.7	24.0	27.5	23.9	0.87
550	40.4	41.6	37.0	36.2	0.98
600	65.4	61.4	59.8	52.6	0.88
650	79.8	79.0	76.7	67.8	0.88
700	89.2	87.9	82.7	74.7	0.90
750	93.9	92.4	86.7	78.4	0.90
800	94.8	95.1	87.5	81.3	0.93
850	95.3	97.5	89.9	84.0	0.93
900	95.7	98.2	89.4	84.4	0.94

Table C6 Catalytic activity test of Ni/CZM-2 catalyst at various temperature, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Temperature (°C)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
500	16.2	25.5	19.3	20.6	1.06
550	42.9	48.3	36.0	38.9	1.09
600	54.4	58.7	60.7	48.6	0.80
650	73.6	76.3	75.6	64.3	0.85
700	84.3	85.1	79.7	71.5	0.89
750	90.3	89.6	82.6	75.4	0.91
800	94.1	93.5	84.0	77.7	0.93
850	95.1	96.5	87.0	80.4	0.93
900	95.7	95.4	87.1	81.5	0.93

Appendix D Experimental Data of Stability Tests for Methane Dry Reforming

Table D1 Stability test of Ni/CZO-S catalyst at 750 °C, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Time (hr)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
1	70.3	69.3	72.5	66.8	0.93
2	65.5	67.0	67.9	64.5	0.95
3	62.8	65.6	66.4	63.9	0.96
4	61.3	64.9	66.0	62.6	0.95
5	59.9	64.2	61.4	61.5	1.00
6	59.2	63.7	61.4	60.7	0.99
7	59.7	64.0	61.4	60.8	0.99
8	58.9	63.3	61.4	59.7	0.97
9	58.7	63.2	59.8	59.0	0.99
10	57.2	61.9	56.0	61.9	0.93

Table D2 Stability test of Ni/CZN-1 catalyst at 750 °C, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Time (hr)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
1	75.2	75.5	74.0	62.7	0.85
2	74.6	76.0	68.4	63.4	0.93
3	73.1	75.5	67.2	63.5	0.95
4	73.8	76.6	70.0	64.2	0.93
5	73.6	76.6	67.7	62.7	0.93
6	73.1	75.5	67.2	63.5	0.95
7	72.3	75.6	67.7	63.4	0.94
8	73.8	76.6	70.0	64.2	0.93
9	72.3	75.6	67.7	63.4	0.94
10	72.3	75.6	67.7	63.4	0.85

Table D3 Stability test of Ni/CZN-2 catalyst at 750 °C, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Time (hr)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
1	73.4	75.5	73.4	66.1	0.90
2	72.9	75.4	72.8	66.1	0.91
3	67.0	69.7	67.5	61.2	0.91
4	66.2	69.8	68.8	60.8	0.88
5	66.5	70.7	68.1	61.2	0.90
6	-	-	-	-	-
7	-	-	-	-	-
8	-	-	-	-	-
9	-	-	-	-	-
10	-	-	-	-	-

Table D4 Stability test of Ni/CZO-C catalyst at 750 °C molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Time (hr)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
1	51.8	62.4	51.4	51.8	1.01
2	50.4	60.3	51.0	51.0	1.00
3	49.6	59.6	52.4	50.3	0.96
4	49.5	59.6	49.7	49.6	1.00
5	48.9	59.3	49.4	49.3	1.00
6	48.8	59.1	49.2	49.1	1.00
7	48.2	58.8	48.5	49.0	1.01
8	47.8	58.5	50.4	49.1	0.97
9	47.5	58.3	48.2	49.1	1.02
10	47.1	58.3	49.2	48.8	1.01

Table D5 Stability test of Ni/CZM-1 catalyst at 750 °C, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Time (hr)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
1	83.4	82.8	83.2	82.6	0.99
2	82.0	81.9	84.7	85.2	1.01
3	81.9	82.3	83.8	84.9	1.02
4	82.6	83.0	89.2	83.2	0.93
5	82.8	82.9	86.5	82.8	0.95
6	82.4	83.0	86.2	82.0	0.94
7	81.8	82.4	86.6	82.4	0.94
8	81.7	82.3	84.7	81.8	0.96
9	81.5	82.1	84.6	81.5	0.96
10	81.1	81.8	85.0	81.1	0.99

Table D6 Stability test of Ni/CZM-2 catalyst at 750 °C, molar ratio of CH₄/CO₂/He = 1:1:8 with GHSV of 10,600 h⁻¹

Time (hr)	X _{CH₄} (%)	X _{CO₂} (%)	Y _{CO} (%)	Y _{H₂} (%)	H ₂ /CO
1	74.9	77.8	72.9	65.7	0.90
2	74.1	77.3	71.6	65.3	0.91
3	74.2	77.7	71.2	65.7	0.93
4	72.8	76.7	70.6	64.4	0.91
5	72.4	76.3	70.3	64.2	0.92
6	72.1	76.2	70.2	63.6	0.91
7	71.4	75.6	68.1	63.0	0.93
8	72.1	75.9	69.3	62.7	0.91
9	71.9	75.9	71.5	62.2	0.87
10	71.8	75.6	69.6	62.5	0.90

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Proceedings:

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