

การศึกษาทางพุทธเคมีของลำต้นพญารากดำ



นางสาว ชลลดา ไพธิศรีทอง

สถาบันวิทยบริการ

จุฬาลงกรณ์มหาวิทยาลัย

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PHYTOCHEMICAL STUDY OF *DIOSPYROS RUBRA* LEC. STEM

Miss Shollada Posritong

สถาบันวิทยบริการ
จุฬาลงกรณ์มหาวิทยาลัย

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จากส่วนลำต้นของพญารากดำ (*Diospyros rubra* Lec.) สามารถแยกสารในกลุ่มไตรเทอร์ปีนอยด์ได้ 3 ชนิด คือ lupeol, betulin และ ursolic acid รวมทั้งได้สารผสมของ β -sitosterol และ stigmasterol การพิสูจน์เอกลักษณ์ของสารเหล่านี้ ทำโดยการวิเคราะห์ข้อมูล IR, MS, $^1\text{H-NMR}$ และ $^{13}\text{C-NMR}$ ร่วมกับการเปรียบเทียบกับค่าที่ได้มีรายงานไว้แล้ว



สถาบันวิทยบริการ
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ภาควิชา...เภสัชพฤกษศาสตร์.....ลายมือชื่อนิสิต.....
สาขาวิชา..เภสัชพฤกษศาสตร์.....ลายมือชื่ออาจารย์ที่ปรึกษา.....
ปีการศึกษา 2545ลายมือชื่ออาจารย์ที่ปรึกษาร่วม.....

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SHOLLADA POSRITONG : PHYTOCHEMICAL STUDY OF *DIOSPYROS RUBRA*

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From the stem of *Diospyros rubra* Lec., three triterpenoids including lupeol, betulin and ursolic acid, together with a mixture of β sitosterol and stigmasterol, have been isolated. Identification of these compounds was accomplished by analysis of IR, MS, $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ data, as well as the comparison with reported values.



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ABBREVIATION

br	=	broad (for NMR spectra)
°C	=	degree celsius
CC	=	column chromatography
CDCl ₃	=	deuterated chloroform
CHCl ₃	=	chloroform
cm	=	centimeter
¹³ C-NMR	=	Carbon-13 Nuclear Magnetic Resonance
δ	=	Chemical shift
<i>d</i>	=	doublet (for NMR spectra)
<i>dd</i>	=	doublet of doublet (for NMR spectra)
DEPT	=	Distortionless Enhancement by Polarization Transfer
DMSO- <i>d</i> ₆	=	deuterated dimethylsulfoxide
EIMS	=	Electron Impact Mass Spectroscopy
EtOH	=	ethanol
eV	=	electron volt
g	=	gram
¹ H-NMR	=	Proton Nuclear Magnetic Resonance
Hz	=	Hertz
IR	=	Infrared Spectroscopy
<i>J</i>	=	coupling constant
KBr	=	potassium bromide
L	=	liter
λ _{max}	=	wavelength at maximum absorption (nm)
<i>m</i>	=	multiplet (for NMR spectra)
m	=	meter
M ⁺	=	molecular ion
MeOH	=	methanol
mg	=	milligram

MHz	=	Megahertz
ml	=	milliliter
mm	=	millimeter
MS	=	Mass Spectrum
m/z	=	mass-to-charge ratio
ϵ	=	Molar absorptivity
nm	=	nanometer
NMR	=	Nuclear Magnetic Resonance
ppm	=	part per million
<i>q</i>	=	quartet (for NMR spectra)
rel. int.	=	relative intensity
<i>s</i>	=	singlet (for NMR spectra)
sp.	=	species
<i>t</i>	=	triplet (for NMR spectra)
TLC	=	Thin-Layer Chromatography
UV	=	Ultraviolet
var.	=	variety
ν_{\max}	=	wavenumber at maximum absorption

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CHAPTER I

INTRODUCTION

Diospyros rubra Lec. (Figure 1) is a tree which belongs to the genus *Diospyros* of the family Ebenaceae. The plant is an evergreen tree that can grow up to 5 m and very commonly found on limestone hill in dry evergreen forest at altitudes of 10-500 m (Phengkklai, 1981). Its leaves are (ovate-) oblong to lanceolate, 6-16 by 2-7 cm, with acute, obtuse or rounded base. The leaf apex is cuspidate with blunt tip, sometimes acute or obtuse. The leaf texture is subcoriaceous to coriaceous and the upper surface is glabrous while the lower surface is pubescent then glabrescent. The leaf has 8-12 pairs of secondary nerves, arched and anastomosing well away from the margin, more or less impressed on the upper surface but prominent on the lower surface. The scalariform veins are conspicuous. The petiole is 0.5-1 cm long, pubescent or glabrescent.

The male flowers are in cymose, inflorescence, 4-merous, sessile or subsessile. The calyx is broadly campanulate, 2-3 mm long, divided to two-third or to the base, sericeous outside but pubescent inside. The corolla is urceolate, 3-4 mm long, divided to one-third, glabrous on both sides except outside along the mid-line of the lobes down to the tube. The number of the stamens are 16-18, glabrous. The rudimentary ovary is hirsute. The female flowers are solitary, 4(-5)-merous, sessile or subsessile. The calyx and corolla are the same as those in male flowers but larger and the corolla is divided to half way. The ovary is globose, sericeous, 4-locular with single, sericeous style. The staminodes are absent.

The fruit of the plant is sessile, ellipsoid, orange or red when ripe, 1-2.5 by 1-2 cm, woody, with glabrous epicarp which is at least 2 mm thick, mostly singly seeded with faint line on seed coat. The base of the fruit is rounded, whereas the apex is acute or obtuse with short apiculus. The fruiting-calyx is divided to the base, pubescent outside, pubescent or woolly inside, accrescent. The lobes are reflexed, not plicate nor undulate, with inconspicuous nerves. The endosperm is ruminant (Phengkklai, 1981).

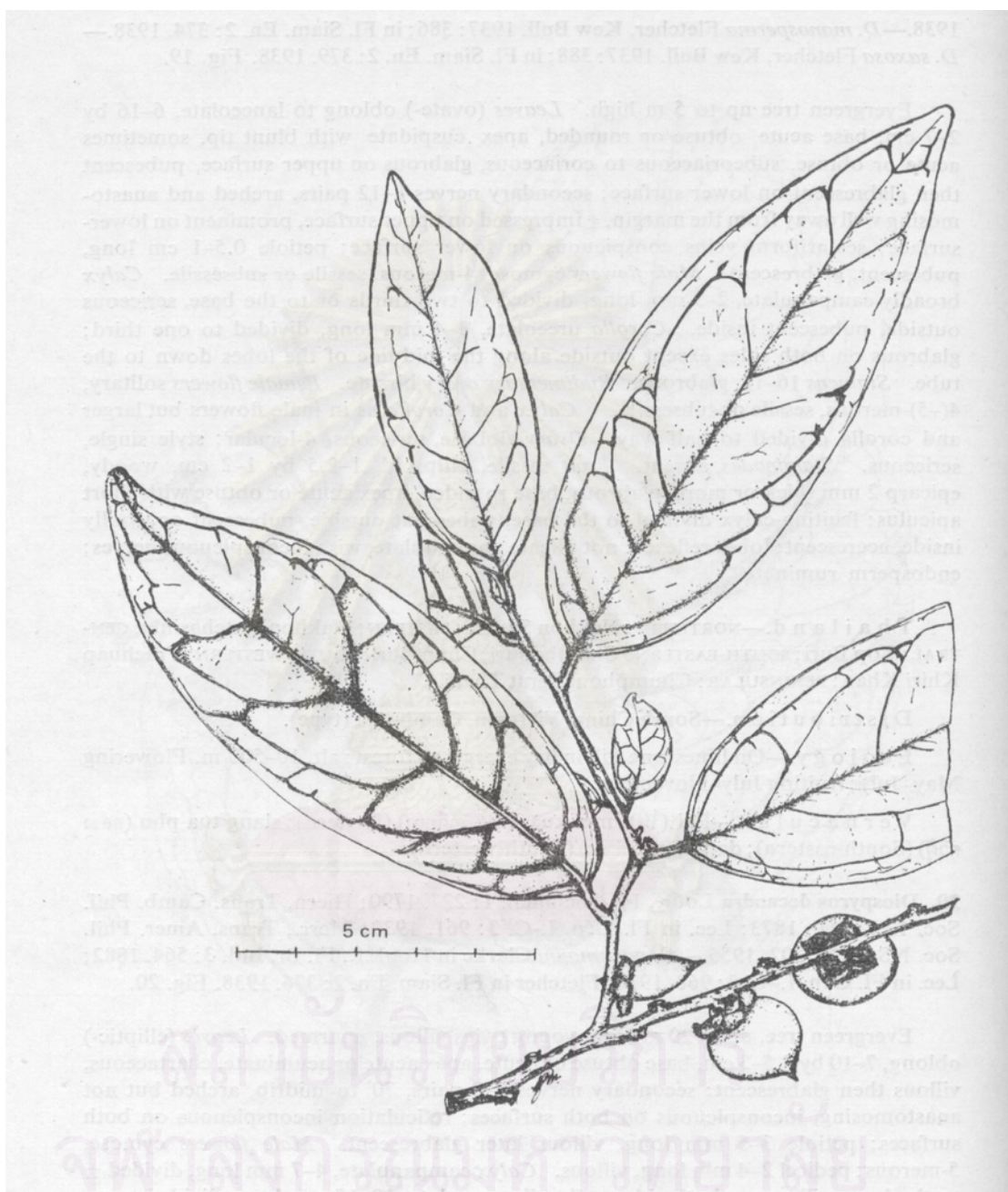


Figure 1. *Diospyros rubra* Lec. (from Flora of Thailand, volume two part four, October

1981.)

The plant can be found throughout Thailand. Its Thai vernacular names are “Phaya rak dam” (General), “Khlai” (Nakhon Sawan), “Dam dong” (Southwestern), “Di mi” (Prachuap Khiri Khan), “Fai”, “Muai dam khao”, “Mak kuea ka” (Eastern), “Salang tua phu” (Southeastern) (ส่วนพฤกษศาสตร์ป่าไม้ สำนักวิชาการป่าไม้ กรมป่าไม้, 2544). It is also found in South China, Vietnam, and Cambodia (Phengkai, 1981).

Plants in the genus *Diospyros* are mostly found in the tropics. A few of them are found in the subtropics (Heywood, 1978).

In Thailand 62 species of *Diospyros* can be found, as follows (ส่วนพฤกษศาสตร์ป่าไม้ สำนักวิชาการป่าไม้ กรมป่าไม้, 2544)

1. *Diospyros andamanica* (Kurz) Bakh. var. *aequabilis* Bakh.
2. *D. apiculata* Hiern
3. *D. areolata* King & Gamble
4. *D. bambuseti* Fletcher
5. *D. bejaudii* Lec.
6. *D. borneensis* Hiern
7. *D. brandisiana* Kurz
8. *D. buxifolia* (Bl.) Hiern
9. *D. castanea* Fletcher
10. *D. cauliflora* Bl.
11. *D. coetanea* (Craib) Fletcher
12. *D. collinsae* Craib
13. *D. confertiflora* (Hiern) Bakh.
14. *D. curranii* Merr.
15. *D. curraniopsis* Bakh.
16. *D. dasyphylla* Kurz
17. *D. decandra* Lour.
18. *D. dictyoneura* Hiern
19. *D. diepenhorstii* Miq.
20. *D. dumetorum* W.W. Smith.
21. *D. ehretioides* Wall. ex G. Don

22. *D. ferrea* (Willd.) Bakh.
a. *D. ferrea* (Willd.) Bakh. var. *ferrea* (Willd.) Bakh.
b. *D. ferrea* (Willd.) Bakh. var. *littorea* (R.Br.) Bakh.
23. *D. filipendula* Pierre ex Lec.
24. *D. frutescens* Bl.
25. *D. fulvopilosa* Fletcher
26. *D. glandulosa* Lace
27. *D. gracilis* Fletcher
28. *D. hasseltii* Zoll.
29. *D. insidiosa* Bakh.
30. *D. kaki* L.*
31. *D. kerrii* Craib
32. *D. kurzii* Hiern
33. *D. lanceifolia* Roxb.
34. *D. latisepala* Ridl.
35. *D. longipilosa* Phengklai
36. *D. malabarica* (Desr.) Kostel.
a. *D. malabarica* (Desr.) Kostel. var. *malabarica* Kostel.
b. *D. malabarica* (Desr.) Kostel. var. *siamensis* (Hochr.) Phengklai
37. *D. martabanica* C.B. Clarke
38. *D. mollis* Griff.
39. *D. montana* Roxb.
40. *D. oblonga* Wall. ex G. Don
41. *D. pendula* Hasselt ex Hassk.
42. *D. philippensis* A. DC.*
43. *D. pilosanthera* Blanco
44. *D. pilosula* (A.DC.) Hiern
45. *D. pubicalyx* Bakh.
46. *D. pyrrhocarpa* Miq.
47. *D. rhodocalyx* Kurz
48. *D. rubra* Lec.

49. *D. scalariformis* Fletcher
50. *D. scortechinii* King & Gamble
51. *D. sumatrana* Miq.
52. *D. tahanensis* Bakh.
53. *D. thaiensis* Phengklai
54. *D. toposia* Ham.
- a. *D. toposia* Ham. var. *toposia* Ham.
- b. *D. toposia* Ham. var. *toposioides* (King & Gamble) Phengklai
55. *D. transitoria* Bakh.
56. *D. trianthos* Phengklai
57. *D. truncata* Zoll. ex Moritzi
58. *D. undulata* Wall. ex G. Don
- a. *D. undulata* Wall. ex G. Don var. *undulata*
- b. *D. undulata* Wall. ex G. Don var. *cratericalyx* (Craib) Bakh.
59. *D. variegata* Kurz
60. *D. venosa* Wall. ex A.DC.
61. *D. wallichii* King & Gamble
62. *D. winitii* Fletcher
- * exotic plant

The importance of *Diospyros* species in traditional medicines have been known for a long time. In Thailand the fruit of *D. mollis* is used as anthelmintic (นันทวัน บุญยะประภัศร และ อรุณช ไชคชัยเจริญพร, 2542). The fruit of *D. rhodocalyx* is used as astringent to control bleeding and in the treatment of renal diseases (Sutthivaiyakit *et al.*, 1995). The bark of *D. montana* is used as a remedy for vomiting, high fever and jaundice, while its gum is used to cure tuberculosis (Pardhasaradhi *et al.*, 1990). The wood of *D. rubra* is used in the treatment of skin diseases, tuberculosis, renal disorders and urinary discharge (นันทวัน บุญยะประภัศร และ อรุณช ไชคชัยเจริญพร, 2542).

Out of 350 identified *Diospyros* species, more than 150 species have been investigated (Mallavadhani, Panda and Rao, 1998). Triterpenoids, steroids,

naphthoquinones, tannins and other groups of phytochemicals have been found. The triterpenoids and naphthoquinones are the groups of compounds which are found widespread and in almost all parts of the plants. Several compounds of these two groups have been shown to exert interesting bioactivities.

This study has been conducted in order to isolate and identify the chemical components from the stem of *D. rubra*, which is one of *Diospyros* species with no previous phytochemical study. The result obtained might add to the knowledge on chemical nature of the genus *Diospyros*, and provide useful information in the field of phytochemistry and chemotaxonomy.



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CHAPTER II

HISTORICAL

Chemical constituents of *Diospyros* species

A variety of chemical compounds have been isolated from plants in the genus *Diospyros* : naphthoquinones, triterpenoids, steroids, tannins, coumarins, etc. Distribution of these constituents in various parts of the plants is shown in Table 1. Two groups of compounds, naphthoquinones and triterpenoids, are found widespread and present in almost all parts of *Diospyros* plants. These compounds can be used as chemical markers of the genus for taxonomic study. The occurrence of triterpenoids and steroids in the genus *Diospyros*, is summarized as follows.

Table 1. Distribution of chemical constituents in various parts of *Diospyros* species.

Class of Compounds	Plant part
Carotenoids	Fruit
Tannins	Fruit, leaf
Sugars	Fruit, seed, root
Hydrocarbons	Fruit, seed, leaf
Lipids	Fruit, seed, bark
Aromatics	Fruit, root, bark
Flavonoids/coumarins	Fruit, leaf, root, sapwood
Terpenoids	Fruit, leaf, calyx, seed, root, bark, heartwood
Steroids	Leaf, root, bark, heartwood
Naphthoquinones	Fruit, leaf, root, bark, heartwood

Triterpenoids

Triterpenoids can be found widespread in the genus *Diospyros*. These metabolites are detected in almost all parts of the plants. *Diospyros* triterpenoids isolated so far are all with pentacyclic core and belong to the lupane, ursane, oleanane, taraxerane or friedelane types. The first three types are more prevalent in the genus.

The most common group of triterpenoids found in *Diospyros* is the lupanes. These compounds accumulate in bark and heartwood. Major metabolites of this type are betulin, betulinic acid and lupeol.

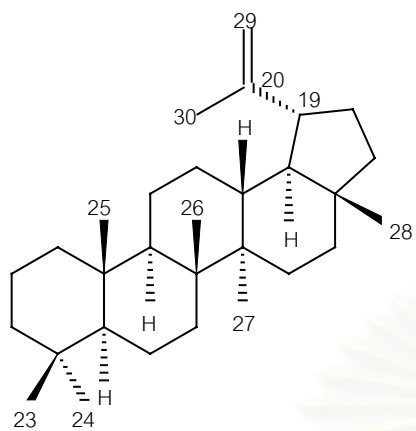
Another group of triterpenoids which are widely distributed in the genus *Diospyros* is the ursanes. Major metabolites of this type include α -amyrin, ursolic acid and baurenol. Ursolic acid accumulates in significant quantities in a number of *Diospyros* species and co-exists mostly with α -amyrin.

In the entire *Diospyros* genus, triterpene glycosides with oleanane skeleton were the only type being isolated. The aglycone of these glycosides is oleanolic acid which is the most abundant oleanane triterpenoid of the genus *Diospyros*.

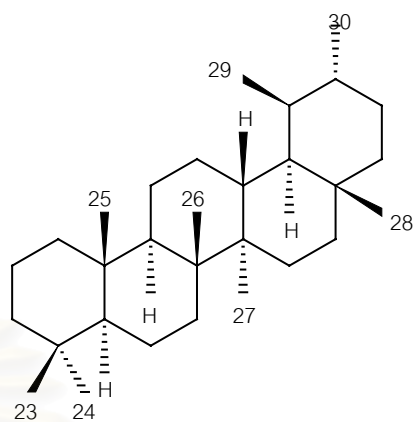
Only three metabolites of the taraxerane type have been found in the genus *Diospyros*. These metabolites are taraxerol, taraxerol acetate and taraxerone. It is interesting to note that further hydroxylation or unsaturation did not take place in this class of compounds and no significant biological activity has been reported for these metabolites.

Friedelanes do not seem to be widely represented in the genus *Diospyros*. There are only three reports on the isolation of friedelanes and related pentacyclic triterpenes (Mallavadhani, Panda and Rao, 1998).

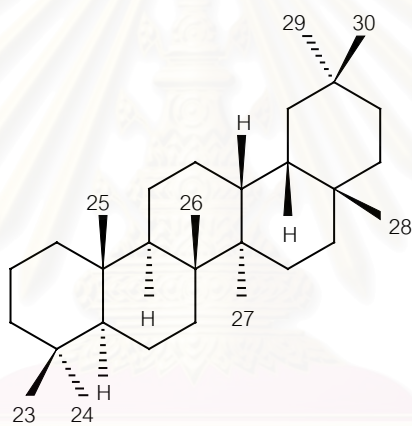
The occurrence of triterpenoids in the genus *Diospyros* is summarized in Table 2.



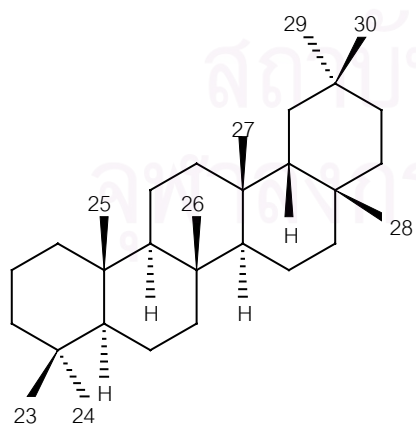
Lupane (A)



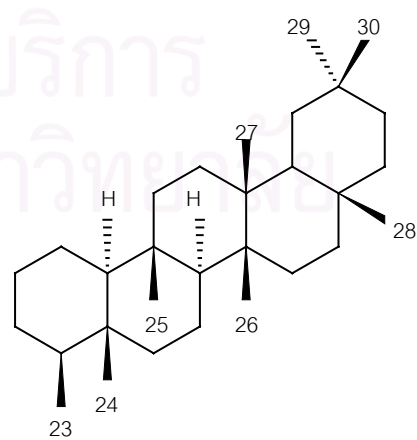
Ursane (B)



Oleanane (C)



Taraxerane (D)



Friedelane (E)

Table 2. Distribution of triterpenoids in the genus *Diospyros*.

Compounds	Sources	References
1. Friedelane type		
Friedelin (1)	<i>D. eriantha</i>	Chen, Yu and Huang,1992
	<i>D. ferrea</i>	Tiwarri, Masood and Minocha, 1979
	<i>D. maritima</i>	Higa, Orihara and Yogi,1998
	<i>D. undulata</i> var. <i>cratericalyx</i> Aoonpakh, 2001	
Friedelin-3-ol (2)	<i>D. eriantha</i>	Chen <i>et al.</i> , 1992
	<i>D. ferrea</i>	Chandler and Hooppe.,1979 Tiwari <i>et al.</i> , 1979
2 α -Hydroxyfriedelin (3)	<i>D. iturensis</i>	Zhong, Waterman and Jeffreys,1984
	<i>D. sanza-minika</i>	Zhong <i>et al.</i> ,1984
2. Lupane type		
28-Acetyl-3-(<i>E</i>)-coumaroylbetulin (4)	<i>D. maritima</i>	Kuo, Chang and Kuo,1997b
Allobetulin (5)	<i>D. montana</i>	Lillie, Musgrave and Skoyles,1976a
Betulin (6)	<i>D. abyssinica</i>	Zhong <i>et al.</i> ,1984
	<i>D. argentea</i>	Zakaria <i>et al.</i> ,1984
	<i>D. bipindensis</i>	Waterman and Mbi,1979
	<i>D. buxifolia</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. canaliculata</i>	Zhong <i>et al.</i> ,1984
	<i>D. candolleana</i>	Desai <i>et al.</i> ,1970
	<i>D. castanea</i>	Musgrave and Skoyles,1974
	<i>D. cauliflora</i>	Musgrave and Skoyles,1974
	<i>D. chevalieri</i>	Zhong <i>et al.</i> ,1984

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulin (6)	<i>D. chloroxylon</i>	Matsura <i>et al.</i> ,1971
	<i>D. cinnabarina</i>	Waterman and Mbi, 1979 ; Zhong <i>et al.</i> ,1984
	<i>D. consolatae</i>	Khan,Nkunya and Wevers, 1979, 1980; Paris and Prista, 1954
	<i>D. cornii</i>	Khan <i>et al.</i> ,1980; Paris and Prista, 1954
	<i>D. crassiflora</i>	Zhong <i>et al.</i> ,1984
	<i>D. curranii</i>	Musgrave and Skoyles,1974
	<i>D. dendo</i>	Zhong <i>et al.</i> ,1984
	<i>D. discolor</i>	Zakaria <i>et al.</i> ,1984
	<i>D. ebenaster</i>	Dominguez <i>et al.</i> ,1979
	<i>D. ebenum</i>	Gupta and Mahadevan, 1967, 1968
	<i>D. elliptifolia</i>	Musgrave and Skoyles,1974
	<i>D. embryopteris</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. eriantha</i>	Chen <i>et al.</i> ,1992
	<i>D. evena</i>	Musgrave and Skoyles,1974
	<i>D. exsculpta</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. fragrans</i>	Zhong <i>et al.</i> ,1984
	<i>D. gabunensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. gracilescens</i>	Zhong <i>et al.</i> ,1984
	<i>D. guianensis</i>	Braneton and Moretti, 1979
	<i>D. hirsuta</i>	Herath <i>et al.</i> ,1978
<i>D. hoyleana</i>	Zhong <i>et al.</i> ,1984	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulin (6)	<i>D. indica</i>	Sundar Ramaiah <i>et al.</i> ,1976
	<i>D. ismailii</i>	Zakaria <i>et al.</i> ,1984
	<i>D. iturensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. kaki</i>	Andriamasy and Fouraste, 1978 ; Matsura <i>et al.</i> ,1971
	<i>D. kaki</i> var. <i>sylvestris</i>	Tezuka <i>et al.</i> ,1972
	<i>D. kamerunensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. kirkii</i>	Maria <i>et al.</i> ,1980
	<i>D. leucomelas</i>	Recio <i>et al.</i> ,1995a, 1995b
	<i>D. longifolia</i>	Zhong <i>et al.</i> , 1984
	<i>D. lotus</i>	Yoshihira,Tezuka and Natori, 1971a; Zakaria <i>et al.</i> ,1984
	<i>D. maingayi</i>	Musgrave and Skoyles,1974; Zakaria <i>et al.</i> ,1984
	<i>D. malanonilau</i>	Singh and Prakash,1988
	<i>D. mannii</i>	Jeffreys, Zakaria and Waterman, 1983
	<i>D. maritima</i>	Tezuka <i>et al.</i> ,1973
	<i>D. melanoxylon</i>	Gupta and Roa,1964; Rao, Rao and Sundar Ramaiah, 1964, 1966; Sankaram and Sidhu, 1964
<i>D. mespiliformis</i>	Zhong <i>et al.</i> ,1984	
<i>D. microphylla</i>	Bhakuni <i>et al.</i> ,1971	
<i>D. mollis</i>	Musgrave and Skoyles, 1974 ; Yoshihira <i>et al.</i> ,1971b	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulin (6)	<i>D. monobuttensis</i>	Zhong <i>et al.</i> , 1984
	<i>D. montana</i>	Dutta, Dutta and Chakrarti, 1972 ; Misra, Nigam and Mitra 1972 ; Musgrave and Skoyles, 1974
	<i>D. moonii</i>	Herath <i>et al.</i> , 1978
	<i>D. morrisiana</i>	Yoshihira <i>et al.</i> , 1971a
	<i>D. obliquifolia</i>	Waterman and Mbi, 1979
	<i>D. oblongifolia</i>	Herath <i>et al.</i> , 1978
	<i>D. peregrina</i>	Bhaumik <i>et al.</i> , 1981 ; Dinda <i>et al.</i> , 1995 ; Misra <i>et al.</i> , 1971
	<i>D. pseudo-malabarica</i>	Musgrave and Skoyles, 1974
	<i>D. quaesita</i>	Herath <i>et al.</i> , 1978
	<i>D. rhodocalyx</i>	Sutthivaiyakit <i>et al.</i> , 1995
	<i>D. rotundifolia</i>	Gupta and Roa, 1964
	<i>D. sanza-minika</i>	Musgrave and Skoyles, 1974; Zhong <i>et al.</i> , 1984
	<i>D. siamang</i>	Zakaria <i>et al.</i> , 1984
	<i>D. siamensis</i>	Musgrave and Skoyles, 1974
	<i>D. siderophylla</i>	Li <i>et al.</i> , 1982
	<i>D. singaporensis</i>	Zakaria <i>et al.</i> , 1984
	<i>D. spinescens</i>	Herath <i>et al.</i> , 1978
<i>D. sumatrana</i>	Zakaria <i>et al.</i> , 1984	
<i>D. sylvatica</i>	Gupta and Roa, 1964; Roa, Roa and Sundar Ramaiah, 1966	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulin (6)	<i>D. thwaitesii</i>	Herath <i>et al.</i> ,1978
	<i>D. tomentosa</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. undulata</i> var. <i>cratericalyx</i>	Aoonpakh, 2001
	<i>D. variegata</i>	Musgrave and Skoyles,1974
	<i>D. verrucosa</i>	Khan, Kishimba and Lockslay, 1987a
	<i>D. virginiana</i>	Shukla and Kapadia,1989
	<i>D. walkeri</i>	Herath <i>et al.</i> ,1978
	<i>D. wallichii</i>	Zakaria <i>et al.</i> ,1984
	<i>D. zenkeri</i>	Zhong <i>et al.</i> ,1984
	Betulinic acid (7)	<i>D. abyssinica</i>
<i>D. alboflavescena</i>		Bouquet,1972
<i>D. argentea</i>		Zakaria <i>et al.</i> , 1984
<i>D. bipindensis</i>		Waterman and Mbi,1979
<i>D. buxifolia</i>		Bhakuni <i>et al.</i> ,1971
<i>D. canaliculata</i>		Zhong <i>et al.</i> ,1984
<i>D. candolleana</i>		Desai <i>et al.</i> ,1970
<i>D. castanea</i>		Musgrave and Skoyles, 1974
<i>D. cauliflora</i>		Musgrave and Skoyles, 1974
<i>D. chevalieri</i>		Zhong <i>et al.</i> ,1984
<i>D. chloroxylon</i>		Matsura <i>et al.</i> ,1971
<i>D. cinnabarina</i>		Waterman and Mbi,1979 ; Zhong <i>et al.</i> ,1984
<i>D. consolatae</i>		Khan <i>et al.</i> , 1987a, 1987b, 1979, 1980
<i>D. crassiflora</i>		Zhong <i>et al.</i> ,1984
<i>D. curranii</i>		Musgrave and Skoyles,1974

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulinic acid (7)	<i>D. dendo</i>	Zhong <i>et al.</i> ,1984
	<i>D. discolor</i>	Lin, 1978 ; Zakaria <i>et al.</i> , 1984
	<i>D. ebenum</i>	Brown and Thomson ,1965
	<i>D. elliptifolia</i>	Musgrave and Skoyles,1974
	<i>D. embryopteris</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. eriantha</i>	Chen <i>et al.</i> , 1992
	<i>D. evena</i>	Musgrave and Skoyles, 1974
	<i>D. exsculpta</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. ferrea</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. fragrans</i>	Zhong <i>et al.</i> ,1984
	<i>D. gabunensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. gilleti</i>	Bouquet , 1972
	<i>D. gracilescens</i>	Waterman and Mbi,1979 ; Zhong <i>et al.</i> ,1984
	<i>D. greeniwayi</i>	Khan and Rwekika, 1992
	<i>D. guaianensis</i>	Braneton and Moretti, 1979
	<i>D. hirsuta</i>	Herath <i>et al.</i> ,1978
	<i>D. hoyleana</i>	Bouquet,1973 ; Zhong <i>et al.</i> ,1984
	<i>D. ismailii</i>	Zakaria <i>et al.</i> ,1984
	<i>D. iturensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. kaki</i>	Andriamasy and Fouraste, 1978 ;Matsura <i>et al.</i> ,1971
<i>D. kaki</i> var. <i>sylvestris</i>	Tezuka <i>et al.</i> ,1972	
<i>D. kameerunensis</i>	Zhong <i>et al.</i> ,1984	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulinic acid (7)	<i>D. leucomelas</i>	Recio <i>et al.</i> ,1995a, 1995b
	<i>D. longiflora</i>	Zhong <i>et al.</i> ,1984
	<i>D. lotus</i>	Yoshihira <i>et al.</i> , 1971a ; Zakaria <i>et al.</i> ,1984
	<i>D. mafiensis</i>	Khan and Rwekika , 1999
	<i>D. maingayi</i>	Musgrave and Skoyles,1974; Zakaria <i>et al.</i> ,1984
	<i>D. mannii</i>	Jeffreys <i>et al.</i> ,1983
	<i>D. maritima</i>	Tezuka <i>et al.</i> ,1973
	<i>D. mespiliformis</i>	Zhong <i>et al.</i> ,1984
	<i>D. monobuttensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. montana</i>	Musgrave and Skoyles,1974; Likhitwitayawuid <i>et al.</i> ,1999 ; Narayan <i>et al.</i> ,1978
	<i>D. moonii</i>	Herath <i>et al.</i> , 1978
	<i>D. morrisiana</i>	Yoshihira <i>et al.</i> ,1971a
	<i>D. obliquifolia</i>	Waterman and Mbi , 1979
	<i>D. palmeri</i>	Dominguez <i>et al.</i> ,1979
	<i>D. peregrina</i>	Dinda <i>et al.</i> ,1995 ; Misra <i>et al.</i> ,1971
	<i>D. pseudo-malabarica</i>	Musgrave and Skoyles,1974
	<i>D. quaesita</i>	Herath <i>et al.</i> ,1978
	<i>D. rhodocalyx</i>	Sutthivaiyakit <i>et al.</i> , 1995
	<i>D. sanza-minika</i>	Musgrave and Skoyles,1974; Zhong <i>et al.</i> ,1984
	<i>D. siamang</i>	Zakaria <i>et al.</i> ,1984

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Betulinic acid (7)	<i>D. siamensis</i>	Musgrave and Skoyles,1974
	<i>D. siderophylla</i>	Li <i>et al.</i> ,1981
	<i>D. singaporensis</i>	Zakaria <i>et al.</i> , 1984
	<i>D. spinescens</i>	Herath <i>et al.</i> ,1978
	<i>D. sumatrana</i>	Zakaria <i>et al.</i> ,1984
	<i>D. sylvatica</i>	Rao <i>et al.</i> , 1966; Gupta and Rao, 1964
	<i>D. thwaitesii</i>	Herath <i>et al.</i> ,1978
	<i>D. tomentosa</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. verrucosa</i>	Khan <i>et al.</i> ,1980 ; Khan <i>et al.</i> ,1987a
	<i>D. virginiana</i>	Shukla and Kapadia, 1989
	<i>D. walkeri</i>	Herath <i>et al.</i> ,1978
	<i>D. wallichii</i>	Zakaria <i>et al.</i> ,1984
	<i>D. zenkeri</i>	Zhong <i>et al.</i> ,1984
	Betulinaldehyde (8)	<i>D. canaliculata</i>
<i>D. eriantha</i>		Chen <i>et al.</i> , 1992
3-(<i>E</i>)-Coumaroylbetulinaldehyde (9)	<i>D. maritima</i>	Chang and Kuo,1999
3-(<i>Z</i>)-Coumaroyllupeol (10)	<i>D. maritima</i>	Chang and Kuo,1998
3-(<i>E</i>)-Coumaroyl-28-palmitoyl betulin (11)	<i>D. maritima</i>	Chang and Kuo,1999
3-(<i>Z</i>)-Coumaroyl-28-palmitoyl betulin (12)	<i>D. maritima</i>	Chang and Kuo,1998

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
3-(<i>E</i>)- Coumaroylbetulin-28-yl ethylnonanedioate (13)	<i>D. maritima</i>	Kuo and Chang, 2000
3-(<i>E</i>)- Coumaroylbetulin-28-yl ethyl(2 <i>R</i>)-2-hydroxysuccinate (14)	<i>D. maritima</i>	Kuo and Chang, 2000
3-(<i>E</i>)- Coumaroylbetulin-28-yl ethyl succinate (15)	<i>D. maritima</i>	Kuo and Chang, 2000
<i>Epi</i> -lupeol (16)	<i>D. ebenaster</i>	Dominguez <i>et al.</i> ,1979
	<i>D. palmeri</i>	Dominguez <i>et al.</i> ,1979
3-(<i>E</i>)-Feruloyl-28-palmitoylbetulin (17)	<i>D. maritima</i>	Chang and Kuo,1998
3-(<i>E</i>)-Feruloylbetulin (18)	<i>D. maritima</i>	Kuo <i>et al.</i> ,1997b
Lupenone (19)	<i>D. mollis</i>	Yoshihira <i>et al.</i> , 1971b
Lupeol (20)	<i>D. abyssinica</i>	Zhong <i>et al.</i> ,1984
	<i>D. acuta</i>	Herath <i>et al.</i> ,1978
	<i>D. argentea</i>	Zakaria <i>et al.</i> ,1984
	<i>D. bipindensis</i>	Waterman and Mbi,1979
	<i>D. buxifolia</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. canaliculata</i>	Zhong <i>et al.</i> ,1984
	<i>D. candollenana</i>	Desai <i>et al.</i> ,1970
	<i>D. castanea</i>	Musgrave and Skoyles,1974
	<i>D. cauliflora</i>	Musgrave and Skoyles,1974
	<i>D. chevalicri</i>	Zhong <i>et al.</i> ,1984
<i>D. cinnabarina</i>		Waterman and Mbi,1979 ;
		Zhong <i>et al.</i> ,1984
<i>D. consolatae</i>		Khan <i>et al.</i> , 1987b

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Lupeol (20)	<i>D. cordifolia</i>	Chandra and Shastry, 1989
	<i>D. cornii</i>	Khan <i>et al.</i> ,1980
	<i>D. crassiflora</i>	Zhong <i>et al.</i> ,1984
	<i>D. curranii</i>	Musgrave and Skoyles,1974
	<i>D. dendo</i>	Zhong <i>et al.</i> ,1984
	<i>D. diepenhorstii</i>	Balza <i>et al.</i> ,1989
	<i>D. discolor</i>	Zakaria <i>et al.</i> ,1984
	<i>D. ebenum</i>	Gupta and Mahadevan,1967; Gupta and Mahadevan,1968
	<i>D. ehretioides</i>	Musgrave and Skoyles,1974
	<i>D. elliptifolia</i>	Musgrave and Skoyles,1974
	<i>D. embryopteris</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. eriantha</i>	Chen <i>et al.</i> , 1992
	<i>D. evena</i>	Musgrave and Skoyles,1974
	<i>D. exsculpta</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. fragrans</i>	Zhong <i>et al.</i> ,1984
	<i>D. gabunensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. gracilescens</i>	Waterman and Mbi,1979 ; Zhong <i>et al.</i> ,1984
	<i>D. greeniwayi</i>	Khan and Rwekika ,1998
	<i>D. guaianensis</i>	Braneton and Moretti, 1979
	<i>D. hirsuta</i>	Herath <i>et al.</i> ,1978
<i>D. hoyleana</i>	Zhong <i>et al.</i> ,1984	
<i>D. ismailii</i>	Zakaria <i>et al.</i> ,1984	
<i>D. iturensis</i>	Zhong <i>et al.</i> ,1984	
<i>D. kaki</i> var. <i>sylvestris</i>	Tezuka <i>et al.</i> ,1972	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Lupeol (20)	<i>D. kamerunensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. kirkii</i>	Maria <i>et al.</i> ,1979
	<i>D. longiflora</i>	Zhong <i>et al.</i> , 1984
	<i>D. lotus</i>	Yoshihira <i>et al.</i> , 1971a ; Zakaria <i>et al.</i> ,1984
	<i>D. mafiensis</i>	Khan and Rwekika, 1992
	<i>D. maingayi</i>	Musgrave and Skoyles,1974; Zakaria <i>et al.</i> ,1984
	<i>D. mannii</i>	Jeffreys <i>et al.</i> , 1983
	<i>D. maritima</i>	Tezuka <i>et al.</i> ,1973
	<i>D. melanoxylon</i>	Rao <i>et al.</i> , 1964; Rao <i>et al.</i> , 1966; Gupta and Rao, 1964
	<i>D. mespiliformis</i>	Zhong <i>et al.</i> ,1984
	<i>D. microphylla</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. mollis</i>	Yoshihira <i>et al.</i> , 1971b ; Musgrave and Skoyles,1974
	<i>D. monobuttensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. montana</i>	Musgrave and Skoyles,1974; Marayan, Row and Satyanarayana, 1978 ; Raj and Agrawal, 1979
	<i>D. moonii</i>	Herath <i>et al.</i> ,1978
	<i>D. morrisiana</i>	Yoshihira <i>et al.</i> ,1971a
	<i>D. obliquifolia</i>	Waterman and Mbi,1979
	<i>D. oblongifolia</i>	Herath <i>et al.</i> ,1978
<i>D. oppositifolia</i>	Herath <i>et al.</i> ,1978	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Lupeol (20)	<i>D. peregrina</i>	Bhaumik <i>et al.</i> ,1981 ; Dinda <i>et al.</i> , 1995
	<i>D. pseudo-malabarica</i>	Musgrave <i>et al.</i> ,1974
	<i>D. quaesita</i>	Herath <i>et al.</i> ,1978
	<i>D. quiloensis</i>	Harper, Kemp and Tanock, 1970
	<i>D. rheophytica</i>	Herath <i>et al.</i> ,1978
	<i>D. rhodocalyx</i>	Musgrave and Skoyles,1974; Sutthivaiyakit <i>et al.</i> , 1995
	<i>D. rotundifolia</i>	Gupta and Rao, 1964
	<i>D. sanza-minika</i>	Musgrave and Skoyles,1974; Zhong <i>et al.</i> ,1984
	<i>D. siamang</i>	Zakaria <i>et al.</i> ,1984
	<i>D. siamensis</i>	Musgrave and Skoyles,1974
	<i>D. siderophylla</i>	Li <i>et al.</i> ,1981
	<i>D. singaporensis</i>	Zakaria <i>et al.</i> ,1984
	<i>D. spinescens</i>	Herath <i>et al.</i> ,1978
	<i>D. sumatrana</i>	Zakaria <i>et al.</i> ,1984
	<i>D. thwaitesii</i>	Herath <i>et al.</i> ,1978
	<i>D. tomentosa</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. toposia</i>	Musgrave and Skoyles,1974
	<i>D. variegata</i>	Musgrave and Skoyles,1974
	<i>D. walkeri</i>	Herath <i>et al.</i> ,1978
<i>D. zenkeri</i>	Zhong <i>et al.</i> ,1984	
Oxyallobetulin (21)	<i>D. lotus</i>	Bhakuni <i>et al.</i> ,1971 ; Yoshihira <i>et al.</i> ,1971a

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Oxyallobetulin (21)	<i>D. montana</i>	Lillie, Musgrave and Skoyles, 1976a
	<i>D. morrisiana</i>	Yoshihira <i>et al.</i> , 1971a
Peregrinol (22)	<i>D. peregrina</i>	Jain and Yadav, 1994
3. Oleanane type		
β -Amyrin (23)	<i>D. lotus</i>	Yoshihira <i>et al.</i> , 1971a
	<i>D. morrisiana</i>	Yan <i>et al.</i> , 1989
Oleanolic acid (24)	<i>D. castanea</i>	Musgrave and Skoyles, 1974
	<i>D. cauliflora</i>	Musgrave and Skoyles, 1974
	<i>D. curranii</i>	Musgrave and Skoyles, 1974
	<i>D. evena</i>	Musgrave and Skoyles, 1974
	<i>D. kaki</i>	Matsura <i>et al.</i> , 1977
	<i>D. montana</i>	Dutta <i>et al.</i> , 1972 ; Misra., 1972; Musgrave and Skoyles, 1974
	<i>D. moonii</i>	Herath <i>et al.</i> , 1978
	<i>D. oblongifolia</i>	Herath <i>et al.</i> , 1978
	<i>D. tomentosa</i>	Bhakuni <i>et al.</i> , 1971
	<i>D. zombensis</i>	Gafner <i>et al.</i> , 1987 ; Gafner and Rodriguez, 1988
Olean-12-ene-3-one (25)	<i>D. morrisiana</i>	Yan <i>et al.</i> , 1989
Oleanolic acid glycosides (26, 27, 28, 29)	<i>D. peregrina</i>	Gupta and Tiwari, 1964
	<i>D. zombensis</i>	Gafner <i>et al.</i> , 1987 ; Gafner and Rodriguez, 1988

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Oleanolic acid acetate (30)	<i>D. eriantha</i>	Chen <i>et al.</i> ,1992
	<i>D. lotus</i>	Zakaria <i>et al.</i> ,1984
Oleanolic acid palmitate (31)	<i>D. montana</i>	Misra <i>et al.</i> ,1972
Oleanolic acid stearate (32)	<i>D. montana</i>	Misra <i>et al.</i> ,1972
4. Taraxerane type		
Taraxerol (33)	<i>D. cordifolia</i>	Chandra and Shastry, 1989
	<i>D. ferrea</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. hirsuta</i>	Herath <i>et al.</i> ,1978
	<i>D. kaki</i>	Zhong and Feng, 1987
	<i>D. lotus</i>	Bhakuni <i>et al.</i> ,1971;
		Yoshihira <i>et al.</i> ,1971a;
		Zakaria <i>et al.</i> ,1984
	<i>D. mollis</i>	Yoshihira <i>et al.</i> ,1971b
	<i>D. morisiana</i>	Yoshihira <i>et al.</i> ,1971a
	<i>D. nicaraguensis</i>	Hasbun <i>et al.</i> , 1988
Taraxerone (34)	<i>D. acuta</i>	Herath <i>et al.</i> ,1978
	<i>D. ferrea</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. lotus</i>	Zakaria <i>et al.</i> ,1984
	<i>D. maritima</i>	Kuo <i>et al.</i> ,1997c
	<i>D. moonii</i>	Herath <i>et al.</i> ,1978
	<i>D. oblongifolia</i>	Herath <i>et al.</i> ,1978
	<i>D. oppositifolia</i>	Herath <i>et al.</i> ,1978
	<i>D. quaesita</i>	Herath <i>et al.</i> ,1978
	<i>D. rheophytica</i>	Herath <i>et al.</i> ,1978
	<i>D. rhodocalyx</i>	Sutthivaiyakit <i>et al.</i> , 1995

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

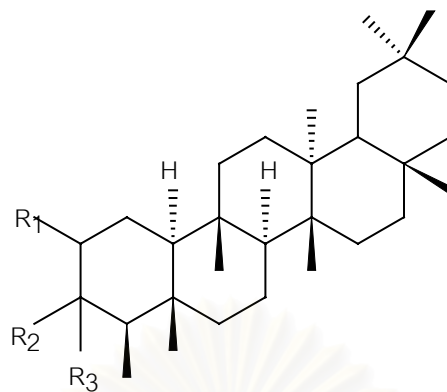
Compounds	Sources	References
Taraxerone (34)	<i>D. thwaitesii</i>	Herath <i>et al.</i> ,1978
Taraxeryl acetate (35)	<i>D. maingayi</i>	Zakaria <i>et al.</i> ,1984
	<i>D. singaporensis</i>	Zakaria <i>et al.</i> ,1984
5. Ursane type		
3 - β - Acetoxy-urs-11-ene-28, 13-olide (36)	<i>D. eriantha</i>	Chen <i>et al.</i> ,1992
α - Amyrenone (37)	<i>D. ebenum</i>	Gupta and Mahadevan, 1967; Sharma and Gupta, 1985
α - Amyrin (38)	<i>D. cornii</i>	Khan <i>et al.</i> ,1980 ; Gafner <i>et al.</i> ,1987
	<i>D. ebenum</i>	Brown and Thomson, 1965
	<i>D. kaki</i>	Andriamasy and Fouraste, 1978
	<i>D. kirkii</i>	Khan <i>et al.</i> ,1980 ; Khan <i>et al.</i> , 1987
	<i>D. mafiensis</i>	Khan and Rwekika, 1999
	<i>D. maingayi</i>	Zakaria <i>et al.</i> ,1984
	<i>D. melanoxylon</i>	Choudhury, 1973
	<i>D. mespiliformis</i>	Khan <i>et al.</i> ,1980
	<i>D. montana</i>	Misra <i>et al.</i> ,1972
	<i>D. natalensis</i>	Khan and Rwekika, 1992
<i>D. sylvatica</i>	Rao and Rao, 1968	

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Baurenol (39)	<i>D. ebenum</i>	Gupta and Mahadevan, 1967 ; Sharma and Gupta, 1985
	<i>D. kirkii</i>	Khan <i>et al.</i> , 1987b ; Khan <i>et al.</i> , 1979
	<i>D. melanoxylon</i>	Rao <i>et al.</i> , 1969
	<i>D. mespiliformis</i>	Khan <i>et al.</i> , 1979
	<i>D. sylvatica</i>	Rao and Rao, 1968
<i>Epi-uvaol</i> (40)	<i>D. montana</i>	Dutta <i>et al.</i> , 1972
19 α -Hydroxyursolic acid (41)	<i>D. kaki</i>	Matsura and linuma, 1977
Marsformosanone (42)	<i>D. peregrina</i>	Bhaumik <i>et al.</i> , 1981
Ursolic acid (43)	<i>D. castanea</i>	Musgrave and Skoyles, 1974
	<i>D. cauliflora</i>	Musgrave and Skoyles, 1974
	<i>D. curranii</i>	Musgrave and Skoyles, 1974
	<i>D. ebenum</i>	Sharma and Gupta, 1985
	<i>D. evena</i>	Musgrave and Skoyles, 1974
	<i>D. ferrea</i>	Bhakuni <i>et al.</i> , 1971
	<i>D. hirsuta</i>	Hearth <i>et al.</i> , 1978
	<i>D. kaki</i>	Matsura <i>et al.</i> , 1971 ; Matsura and linuma, 1977
	<i>D. leucomelas</i>	Recio <i>et al.</i> , 1995a ; 1995b
	<i>D. lotus</i>	Yoshihira <i>et al.</i> , 1971a ; Zakaria <i>et al.</i> , 1984

Table 2. Distribution of triterpenoids in the genus *Diospyros* (continued).

Compounds	Sources	References
Ursolic acid (43)	<i>D. montana</i>	Misra <i>et al.</i> ,1972 ; Musgrave And Skoyles,1974; Zafar, Singh and Khan, 1991
	<i>D. morrisiana</i>	Yoshihira <i>et al.</i> ,1971a
	<i>D. quaesita</i>	Herath <i>et al.</i> ,1978
	<i>D. tomentosa</i>	Bhakuni <i>et al.</i> ,1971
Ursolic acid acetate (44)	<i>D. eriantha</i>	Chen <i>et al.</i> ,1992
	<i>D. lotus</i>	Yoshihira <i>et al.</i> ,1971a
Ursolic acid palmitate (45)	<i>D. montana</i>	Misra <i>et al.</i> ,1972
Ursolic acid stearate (46)	<i>D. montana</i>	Misra <i>et al.</i> ,1972
Uvaol (47)	<i>D. lotus</i>	Zakaria <i>et al.</i> ,1984
	<i>D. maingayi</i>	Zakaria <i>et al.</i> ,1984
6. Miscellaneous		
Glut-5(6)-ene-3- β -ol (48)	<i>D. iturensis</i>	Zhong <i>et al.</i> ,1984
	<i>D. sanza-minika</i>	Zhong <i>et al.</i> ,1984



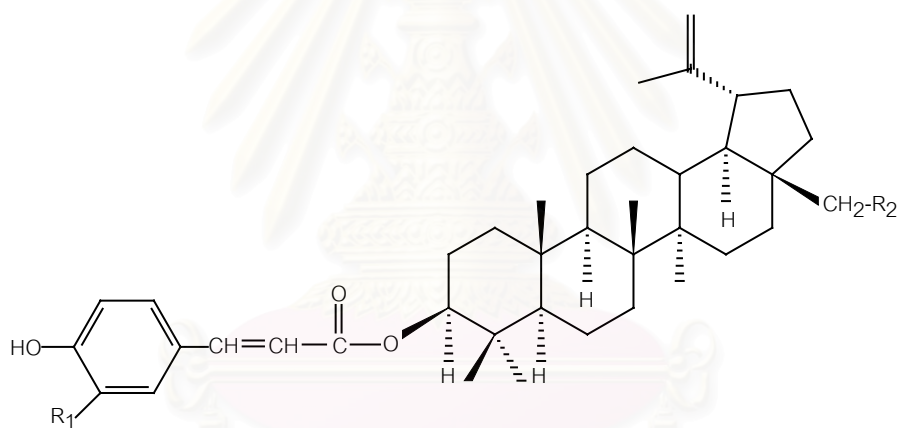
Friedelin (1)

$R_1 = H ; R_2, R_3 = OH$

Friedelin-3-ol (2)

$R_1 = H ; R_2 = \beta - H ; R_3 = \alpha - OH$

2 α -hydroxyfriedelin (Cerin) (3) $R_1 = \alpha - OH ; R_2, R_3 = =O$

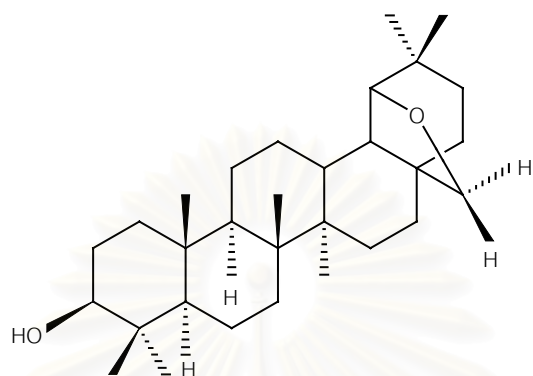


28-Acetyl-3-(*E*)-coumaroylbetulin (4)

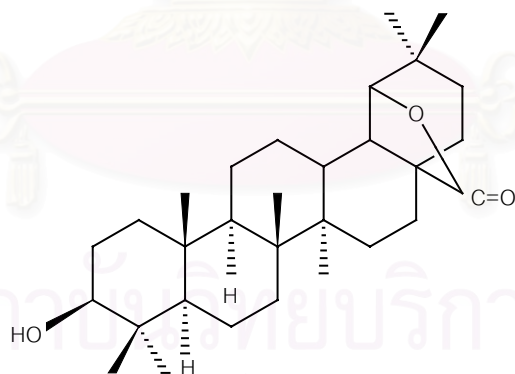
$R_1 = H ; R_2 = -O-C(=O)-CH_3$

3-(*E*)-Feruloylbetulin (18)

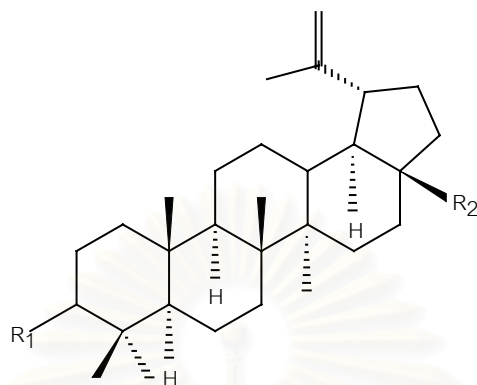
$R_1 = OCH_3 ; R_2 = -OH$



Allobetulin (5)



Oxyallobetulin (21)



Betulin (6) $R_1 = \beta\text{-OH}$; $R_2 = \text{CH}_2\text{OH}$

Betulinic acid (7) $R_1 = \beta\text{-OH}$; $R_2 = \text{COOH}$

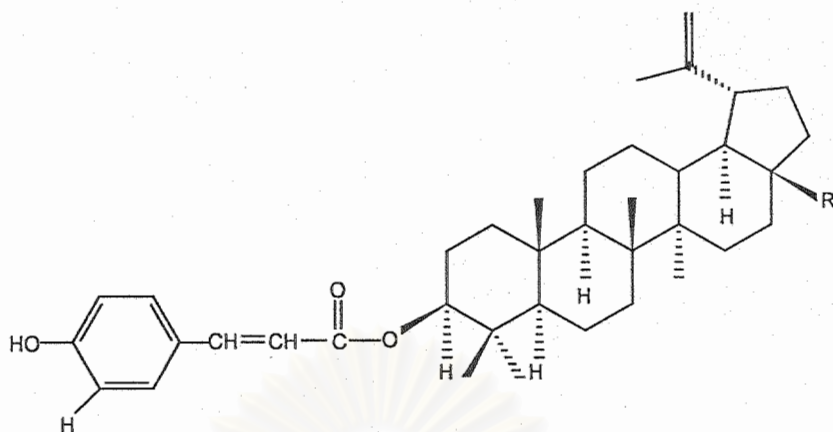
Betulinaldehyde (8) $R_1 = \beta\text{-OH}$; $R_2 = \text{CHO}$

Epi-lupeol (16) $R_1 = \alpha\text{-OH}$; $R_2 = \text{CH}_3$

Lupenone (19) $R_1 = =\text{O}$; $R_2 = \text{CH}_3$

Lupeol (20) $R_1 = \beta\text{-OH}$; $R_2 = \text{CH}_3$

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3-(*E*)-Coumaroylbetulinaldehyde (9) R = CHO

3-(*Z*)-Coumaroyllupeol (10) R = CH₃

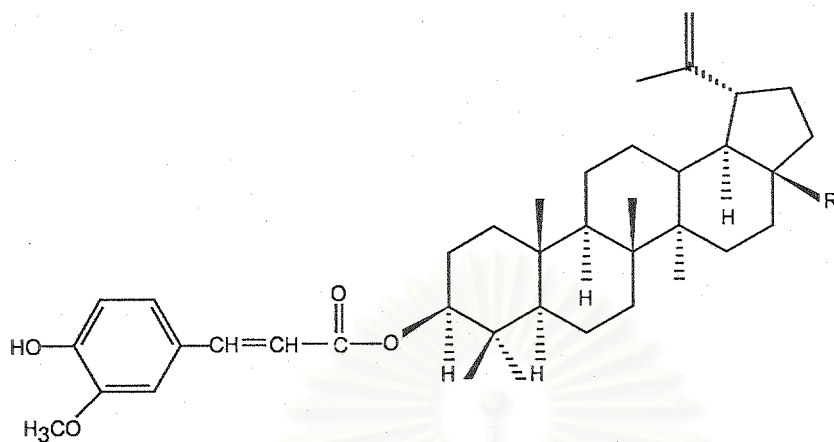
3-(*E*)-Coumaroyl-28-palmitoylbetulin (11) R = $\text{CH}_2\text{-O-C(=O)-CH}_2\text{-CH}_2\text{-(CH}_2\text{)}_{10}\text{CH}_2\text{-CH}_2\text{CH}_3$

3-(*Z*)-Coumaroyl-28-palmitoylbetulin (12) R = $\text{CH}_2\text{-O-C(=O)-CH}_2\text{-CH}_2\text{-(CH}_2\text{)}_{12}\text{CH}_3$

3-(*E*)-Coumaroylbetulin-28-yl ethylnonanedioate (13) R = $\text{CH}_2\text{-O-C(=O)-CH}_2\text{-(CH}_2\text{)}_5\text{CH}_2\text{-CO}_2\text{CH}_2\text{CH}_3$

3-(*E*)-Coumaroylbetulin-28-yl ethyl(2*R*)-2-hydroxysuccinate (14) R = $\text{CH}_2\text{-O-C(=O)-CH(OH)-CH}_2\text{-CO}_2\text{-CH}_2\text{CH}_3$

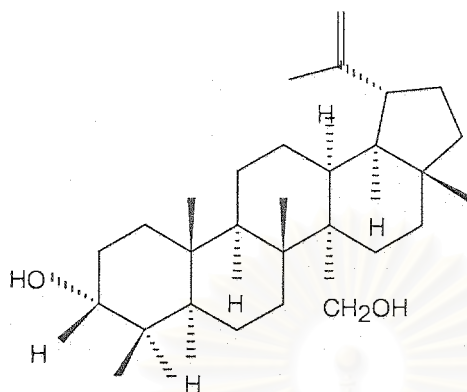
3-(*E*)-Coumaroylbetuliri-28-yl ethyl succinate (15) R = $\text{CH}_2\text{-O-C(=O)-CH}_2\text{-CH}_2\text{-CO}_2\text{-CH}_2\text{CH}_3$



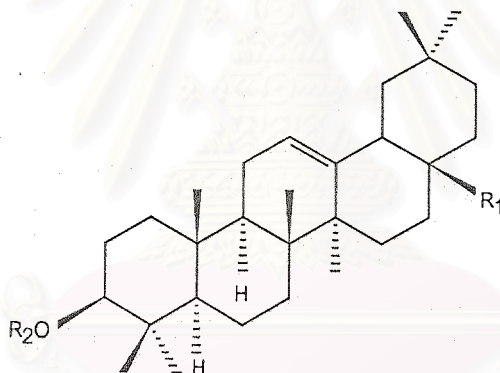
3-(*E*)-Feruloyl-28-palmitoylbetulin (17) $R = \text{CH}_2\text{-O-C(=O)-CH}_2\text{-(CH}_2\text{)}_{13}\text{CH}_3$

3-(*E*)-Feruloylbetulin (18) $R = \text{CH}_2\text{-OH}$

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Peregrinol (22)



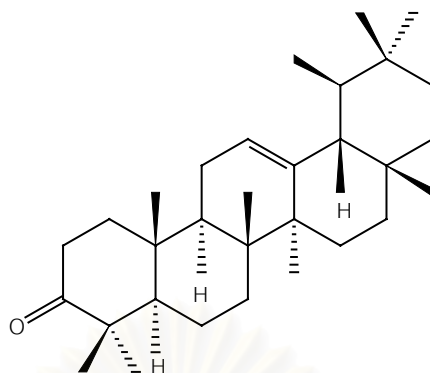
β -Amyrin (23) $R_1 = \text{CH}_3$; $R_2 = \text{H}$

Oleanolic acid (24) $R_1 = \text{COOH}$; $R_2 = \text{H}$

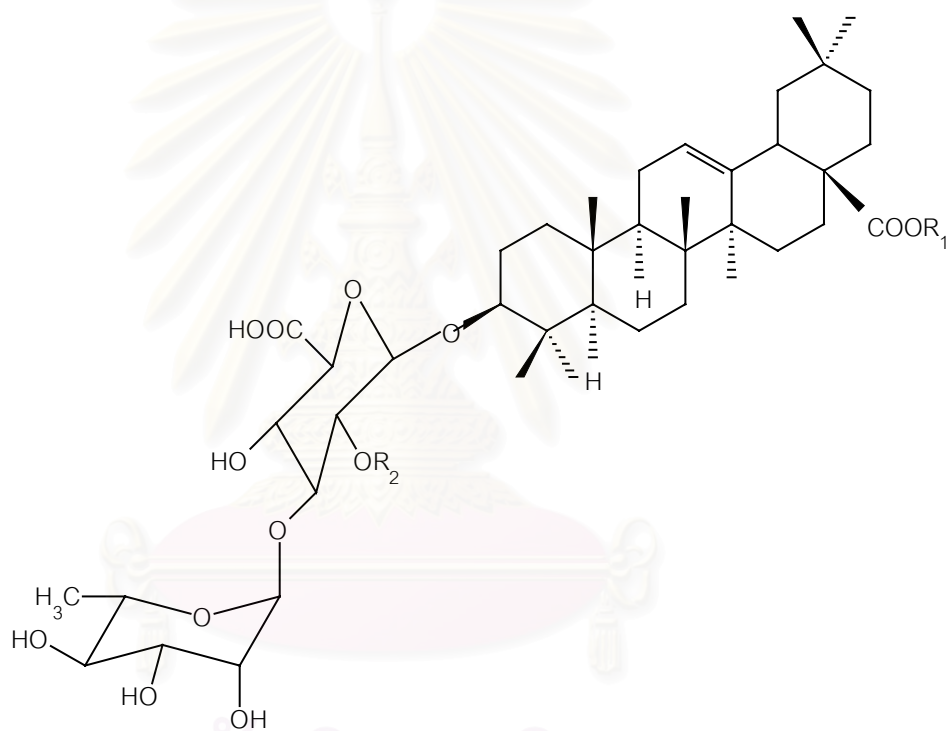
Oleanolic acid acetate (30) $R_1 = \text{COOH}$; $R_2 = \text{COCH}_3$

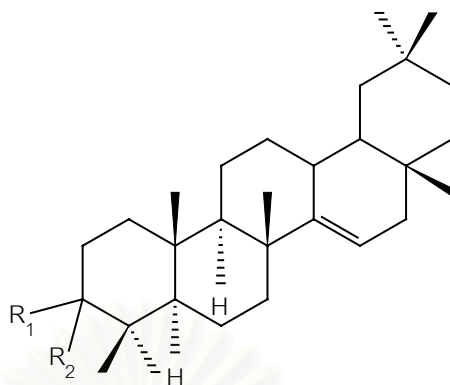
Oleanolic acid palmitate (31) $R_1 = \text{COOH}$; $R_2 = \text{CO}(\text{CH}_2)_{14}\text{CH}_3$

Oleanolic acid stearate (32) $R_1 = \text{COOH}$; $R_2 = \text{CO}(\text{CH}_2)_{16}\text{CH}_3$



Olean-12-ene-3-one (25)

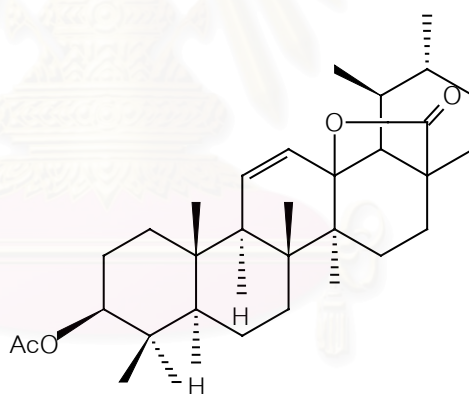
Oleanolic acid glycosides (26) $R_1 = R_2 = H$ (27) $R_1 = \text{Glucosyl}$; $R_2 = H$ (28) $R_1 = \text{Glucosyl}$; $R_2 = \text{Xylosyl}$ (29) $R_1 = H$; $R_2 = \text{Xylosyl}$



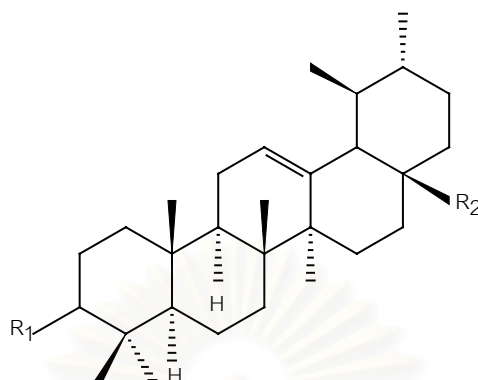
Taraxerol (33) $R_1 = \alpha\text{-H}$; $R_2 = \beta\text{-OH}$

Taraxerone (34) $R_1, R_2 = =O$

Taraxeryl acetate (35) $R_1 = \alpha\text{-H}$; $R_2 = \beta\text{-OCOCH}_3$



3β-Acetoxy-urs-11-ene-28,13-olide (36)



α -Amyrenone (37) $R_1 = =O$; $R_2 = -CH_3$

α -Amyrin (38) $R_1 = \beta$ -OH ; $R_2 = CH_3$

Epi-uvaol (40) $R_1 = \alpha$ -OH ; $R_2 = CH_2OH$

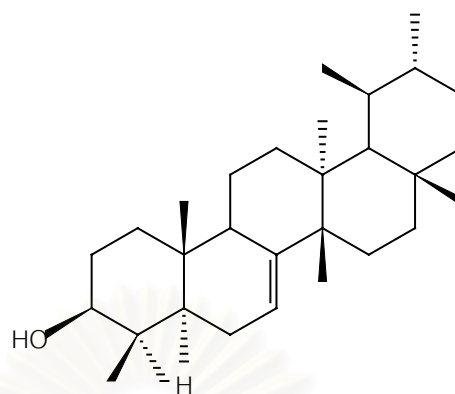
Ursolic acid (43) $R_1 = \beta$ -OH ; $R_2 = COOH$

Ursolic acid acetate (44) $R_1 = \beta$ -OCOCH₃ ; $R_2 = COOH$

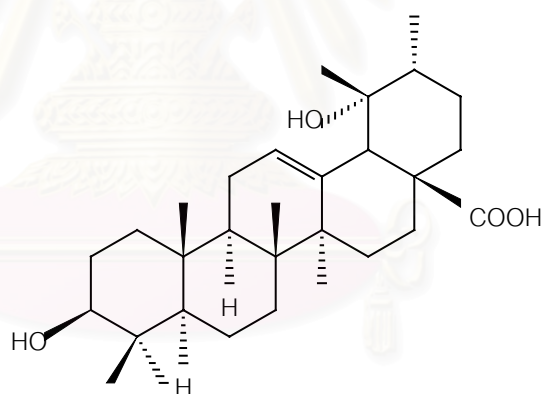
Ursolic acid palmitate (45) $R_1 = \beta$ -OCO(CH₂)₁₄CH₃ ; $R_2 = COOH$

Ursolic acid stearate (46) $R_1 = \beta$ -OCO(CH₂)₁₆CH₃ ; $R_2 = COOH$

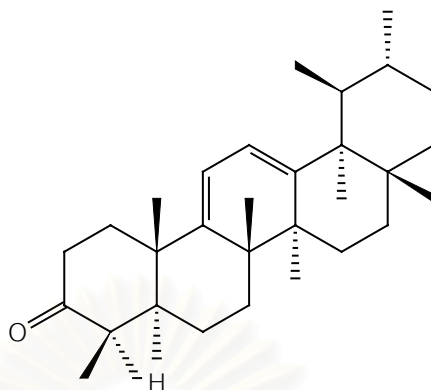
Uvaol (47) $R_1 = \beta$ -OH ; $R_2 = CH_2OH$



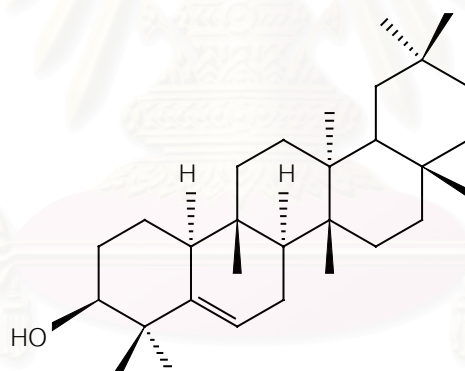
Baurenol (39)

19 α -Hydroxyursolic acid (41)

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Marsformosanone (42)

Glut-5(6)-ene-3- β -ol (48)

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Steroids

Only six steroid compounds have been detected and isolated from *Diospyros* species. The most common steroid found is β -sitosterol, which occurs in both free and glycosidic forms. β -Sitosterol was found to accumulate in a number of species and so far it was detected in more than 30 *Diospyros* plants. Campesterol, Stigmasterol, Stigmasta-4-ene-3-one and Stigmasta-5,6-dihydro-22-en-3 β -ol are the other steroids found in the *Diospyros* genus.

Table 3. Distribution of steroids in the genus *Diospyros*.

Compounds	Sources	References
β - Sitosterol (49)	<i>D. acuta</i>	Herath <i>et al.</i> ,1978
	<i>D. buxifolia</i>	Bhakuni <i>et al.</i> ,1971; Musgrave and Skoyles,1974
	<i>D. chaetocarpa</i>	Herath <i>et al.</i> ,1978
	<i>D. chloroxylon</i>	Rao and Sunder, 1964 ; Sidhu <i>et al.</i> ,1968
	<i>D. cordifolia</i>	Chandra and Shastry, 1989
	<i>D. discolor</i>	Rao <i>et al.</i> , 1964 ; Sunder <i>et al.</i> ,1976
	<i>D. ebenaster</i>	Dominguez <i>et al.</i> ,1979
	<i>D. ebum</i>	Sharma and Gupta, 1985
	<i>D. eriantha</i>	Chen <i>et al.</i> , 1992
	<i>D. ferrea</i>	Bhakuni <i>et al.</i> , 1971
	<i>D. hirsuta</i>	Herath <i>et al.</i> , 1978

Table 3. Distribution of steroids in the genus *Diospyros* (continued).

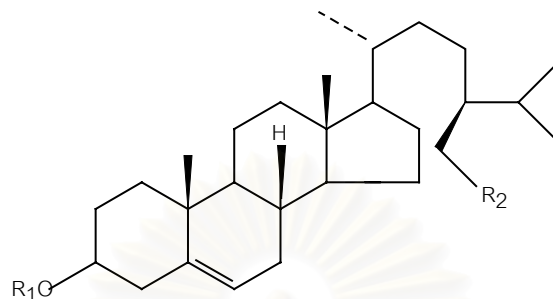
Compounds	Sources	References
β - Sitosterol (49)	<i>D. indica</i>	Sunder <i>et al.</i> ,1976
	<i>D. kaki</i>	Rao <i>et al.</i> , 1964 ; Matsura and linuma, 1977 ; Lin <i>et al.</i> ,1989
	<i>D. kirkii</i>	Maria <i>et al.</i> ,1980
	<i>D. lotus</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. malanonilau</i>	Bhakuni <i>et al.</i> ,1971 ; Srivastava and Kharya, 1980
	<i>D. melanoxylon</i>	Gupta and Roa, 1964 ; Sankaram and Sidhu, 1964 ; Sidhu <i>et al.</i> , 1968
	<i>D. moonii</i>	Dutta,Dutta and Chakrarti, 1972
	<i>D. montana</i>	Dutta, Dutta and Chakrarti, 1972 ; Goutum and Purohit, 1977 ; Misra, Nigam and Mitra, 1972 ; Raj and Agrawal, 1979
	<i>D. morrisiana</i>	Chen <i>et al.</i> , 1989 ; 1992
	<i>D. oblongifolia</i>	Herath <i>et al.</i> ,1978
	<i>D. oppositifolia</i>	Herath <i>et al.</i> ,1978
	<i>D. peregrina</i>	Gupta and Roa, 1964 ; Gupta and Tiwari,1964 ; Misra <i>et al.</i> , 1971 ; Misra, Nigam,and Mitra, 1972
	<i>D. quaseita</i>	Herath <i>et al.</i> ,1978

Table 3. Distribution of steroids in the genus *Diospyros* (continued).

Compounds	Sources	References
β -Sitosterol (49)	<i>D. rheophytica</i>	Herath <i>et al.</i> ,1978
	<i>D. spinescens</i>	Herath <i>et al.</i> ,1978
	<i>D. texana</i>	Dominguez <i>et al.</i> ,1979
	<i>D. tomentosa</i>	Bhakuni <i>et al.</i> ,1971
	<i>D. thwaitesii</i>	Herath <i>et al.</i> ,1978
	<i>D. walkeri</i>	Herath <i>et al.</i> ,1978
β -Sitosterol glucoside (50)	<i>D. kaki</i>	Matsura and Inuma, 1977
	<i>D. montana</i>	Dutta,Dutta and Chakrarti, 1972 ; Misra, Nigam and Mitra, 1972
	<i>D. morrisiana</i>	Chen <i>et al.</i> ,1989
	<i>D. peregrina</i>	Misra <i>et al.</i> , 1971 ; Misra, Nigam and Mitra, 1972
Stigmasterol (51)	<i>D. buxifolia</i>	Musgrave and Skoyles, 1974
	<i>D. castanea</i>	Musgrave and Skoyles, 1974
	<i>D. cauliflora</i>	Musgrave and Skoyles, 1974
	<i>D. curranii</i>	Musgrave and Skoyles,1974
	<i>D. dipenhorstii</i>	Musgrave and Skoyles, 1974
	<i>D. ebenum</i>	Sharma and Gupta.,1985
	<i>D. evena</i>	Musgrave and Skoyles, 1974
	<i>D. kaki</i>	Lin, Chou and Chen, 1988
	<i>D. mollis</i>	Musgrave and Skoyles, 1974

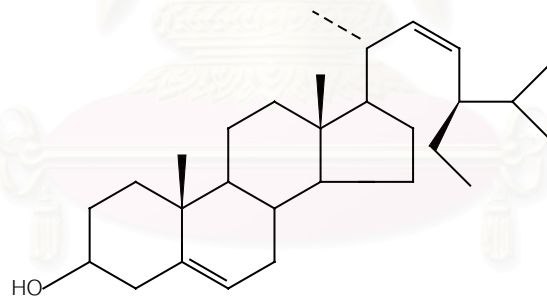
Table 3. Distribution of steroids in the genus *Diospyros* (continued).

Compounds	Sources	References
Stigmasterol (51)	<i>D. montana</i>	Musgrave and Skoyles,1974; Dutta, Dutta and Chakrarti, 1972 ; Goutum and Purohit, 1977
	<i>D. morrisiana</i>	Chen <i>et al.</i> ,1992
	<i>D. sanza-minika</i>	Musgrave and Skoyles,1974
Stigmasta-4-ene-3-one (52)	<i>D. morrisiana</i>	Musgrave and Skoyles,1974
Stigmasta-5,6-dihydro-22-en- 3 β -ol (53)	<i>D. morrisiana</i>	Musgrave and Skoyles,1974
Campesterol (54)	<i>D. discolor</i>	Sidhu and Prasad, 1971
	<i>D. kaki</i>	Lin, Chou and Chen, 1988

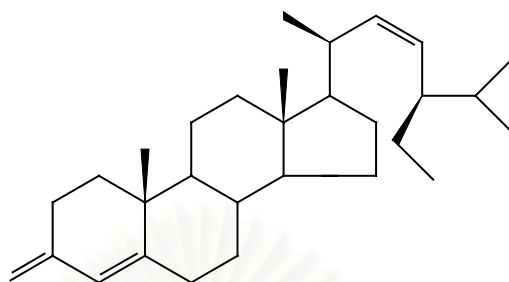


β -Sitosterol (49) R₁ = H ; R₂ = CH₃

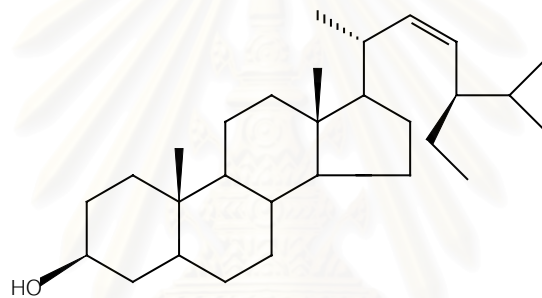
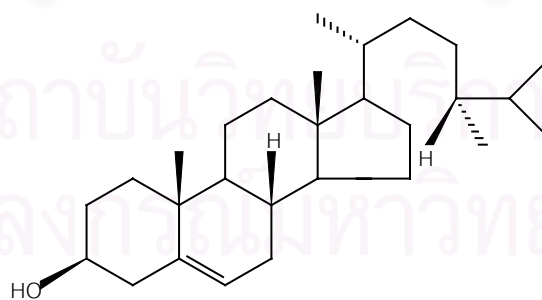
β -Sitosterol glucoside (50) R₁ = glucose ; R₂ = CH₃



Stigmasterol (51)



Stigmasta-4-en-3-one (52)

Stigmasta-5,6-dihydro-22-en-3 β -ol (53)

Campesterol (54)

Ethnomedicinal uses of *Diospyros* species

Plants in the genus *Diospyros* have been known for their medicinal uses since older times. Almost all parts of these plants have been used as medicines. Mallavadhani *et al.* (1998) have reviewed their uses in traditional medicines of various countries. In Thailand, medicinal uses of about 10 *Diospyros* species, have been recorded, as presented in Table 4.

Table 4. Uses of *Diospyros* species in Thai traditional medicine.

Species	Part used	Medicinal uses	References
<i>D. decandra</i> จันทขวาว	Wood	Antipyretic, anthelmintic, antiperspirant	นันทวัน บุญยะประภัศร และ อรรนุช โชคชัยเจริญพร, 2539
	Heart wood	Antipyretic, tonic	
	Fruit	Antidiarrhoeal	
<i>D. rhodocalyx</i> ตะโกนา	Root	Antipyretic, lactigenous	นันทวัน บุญยะประภัศร และ อรรนุช โชคชัยเจริญพร, 2541
	Stem	Antipyretic, antipruritic, tonic, diuretic, lactigenous	
	Stem bark	Diuretic, antitoothache	
	Bark	Digestive, diuretic, antitoothache	
	Wood	Tonic, antitoothache, aphrodisiac	
<i>D. peregrina</i> ตะโก	Root	Anthelmintic, antidiarrhoeal	นันทวัน บุญยะประภัศร และ อรรนุช โชคชัยเจริญพร, 2541
	Bark	Anthelmintic, antidiarrhoeal, antidyentery and digestive	
	Stem bark	Astringent, tonic, anti-emetic	
	Fruit	Antidiarrhoeal, anthelmintic	

Table 4. Uses of *Diospyros* species in Thai traditional medicine (continued).

Species	Part used	Medicinal uses	References
<i>D. peregrina</i> ตะโก	Fruit	Astringent	นันทวัน บุญยะประภัศร และ อรนุช ไชคชัยเจริญพร, 2541
	Flower	Anthelmintic	
<i>D. ehretioides</i>	Root	Antituberculosis, mucolytic	นันทวัน บุญยะประภัศร และ อรนุช ไชคชัยเจริญพร, 2541
<i>D. transitoria</i>	Root	Anthelmintic	นันทวัน บุญยะประภัศร และ อรนุช ไชคชัยเจริญพร, 2541
	Wood	Anthelmintic	
<i>D. areolata</i>	Root	Antidysentery, anthelmintic	นันทวัน บุญยะประภัศร และ อรนุช ไชคชัยเจริญพร, 2542
	Stem bark	Antidiarrhoeal, carminative	
	Wood	Antidiarrhoeal	
	Gum	Antidysentery, antidiarrhoeal	
	Flower	Anthelmintic	
	Fruit	Astringent, anthelmintic, antidiarrhoeal	
<i>D. malabarica</i> var. <i>siamensis</i>	Root	Anthelmintic, antidiarrhoeal	นันทวัน บุญยะประภัศร และ อรนุช ไชคชัยเจริญพร, 2542
	Stem bark	Antidiarrhoeal, carminative	
	Flower	Anthelmintic	
	Fruit	Astringent	
<i>D. mollis</i>	Root	Anthelmintic and antiemetic	นันทวัน บุญยะประภัศร และ อรนุช ไชคชัยเจริญพร, 2542
	Fruit, seed	Anthelmintic	

Table 4. Uses of *Diospyros* species in Thai traditional medicine (continued).

Species	Part used	Medicinal uses	References
<i>D. rubra</i> และ	Root, bark	Antitumor	นันทวัน บุญยะประภัศร
	Heart wood	Antituberculosis (duodenal & lung)	อรนุช ไชคชัยเจริญพร, 2542
	Stem	For backache	
	Leaf	Anti-inflammatory	
	Wood, gum	Antituberculosis	
<i>D. variegata</i>	Root, bark	Antitumor	นันทวัน บุญยะประภัศร และ
	Heart wood	Antituberculosis	อรนุช ไชคชัยเจริญพร, 2542
	Stem	Analgesic	
	Leaf, wood	For cure wounds	

Pharmacological Activity of *Diospyros* species

Many *Diospyros* species have been reported as exhibiting interesting bioactivity. The pharmacological activities of extracts and isolated compounds from these plants are summarized in Table 5 and Table 6, respectively.

Table 5. Pharmacological activities of *Diospyros* extracts.

Species	Part	Extract	Pharmacological activity	Refereces
<i>D. chloroxylon</i>	Aerial parts	50% EtOH	Antiviral	Dhar <i>et al.</i> , 1973
<i>D. cordifolia</i>	NS	Alcohol	Anti-inflammatory, antipyretic, analgesic, CNS depressant, spasmolytic, produces bradycardia and hypotension	Singh <i>et al.</i> , 1971 ; Kohli <i>et al.</i> , 1972
<i>D. embryopteris</i>	Leaves	80% EtOH	Showed abolition of libido	Choudhary <i>et al.</i> , 1990
<i>D. exsculpta</i>	Aerial parts	50% EtOH	Showed activity on cardiovascular system	Bhakuni <i>et al.</i> , 1971
	Seeds	Unsaponified matter	Produced fall in blood pressure and increase in respiration, also showed anorexia, CNA depressant and antibacterial activities	

Table 5. Pharmacological activities of *Diospyros* extracts (continued).

Species	Part	Extract	Pharmacological activity	References
<i>D. insignis</i>	Aerial parts	50% EtOH	Antifertility	Dhawan <i>et al.</i> , 1977
<i>D. kaki</i>	Leaves	Tannin	(i) Increases life span and decreases brain haemorrhage and infraction (ii) Showed scavenging action forwards active oxygen free radicals (iii) Inhibited lipid peroxidation	Uchida <i>et al.</i> , 1990
	Leaves	MeOH	Showed hypotensive activity against urethane anaesthetised rats	Funayama and Hikino, 1979
	Fruit	---	(i) Showed strong detoxifying activity on various snake venoms	Fukami <i>et al.</i> , 1979
	Fruit	---	(ii) Inactivated bacterial toxins	Fukami <i>et al.</i> , 1979
<i>D. leucomelas</i>	Leaves	CH ₂ Cl ₂ & MeOH	Showed anti-inflammatory activity	Recio <i>et al.</i> , 1995a Recio <i>et al.</i> , 1995b

Table 5. Pharmacological activities of *Diospyros* extracts (continued).

Species	Part	Extract	Pharmacological activity	References
<i>D. melanoxylon</i>	Seed	Unsaponified matter	Antibacterial	Aloskar <i>et al.</i> , 1992
<i>D. mespiliformis</i>	Seed	----	Antibacterial	Lajubutu <i>et al.</i> , 1995
<i>D. montana</i>	Leaves	Pet.ether, CCl ₄ ,C ₆ H ₆	Antibacterial	Goutam and Purohit, 1973
	Bark	90% EtOH	Inhibited the growth of Ehrlich ascites carcinoma in mice	Hazra <i>et al.</i> , 1981
	---	---	Showed potent anti-inflammatory and antipyretic activities in rats and analgesic activity in mice	Singh <i>et al.</i> , 1973

Table 5. Pharmacological activities of *Diospyros* extracts (continued).

Species	Part	Extract	Pharmacological activity	References
<i>D. montana</i>	Bark	Alcohol	Showed CNS depressant activity, decrease locomoter activity and loss of righting reflex in mice and rats, spasmolytic activity on rabbit and guinea pig ileum and produced hypotension in anaesthetised dogs	Singh et al., 1971
<i>D. morrisiana</i>	Stem	Hexane	Showed significant cytotoxicity against of human KB and A-549 lung carcinoma, HCT-8 colon tumor and murine P-338 and L-1210 lymphocytic leukaemia	Yan et al., 1989
<i>D. peregrina</i>	Fruit	Ether	Antibacterial	Joshi and Magar, 1952
	NS	Alcohol	Anti-amoebic, anti-viral and hypoglycaemic activities	Dhar et al., 1968
	Aerial parts	50% EtOH	Showed activity on human epidermoid carcinoma of nasopharynx in tissue culture and diuretic activity	Dhawan et al., 1980

Table 5. Pharmacological activities of *Diospyros* extracts (continued).

Species	Part	Extract	Pharmacological activity	References
<i>D. peregrina</i>	---	EtOAc	Significantly prevented rats from stress, gastric ulcers and hepatotoxicity	Singh et al., 1988
<i>D. virginiana</i>	Fruit	---	Produced tumors at the injection site in 66% or more of the treated rats	Kapadia et al., 1976
	Fruit	---	Showed cholesterol lowering activity	Ebihara et al., 1979 ; 1980
<i>D. zombensis</i>	Root bark	Petrol & CHCl ₃	Showed cytotoxicity against human colon carcinoma cells	Gafner et al., 1987 Gafner et al., 1988 ; 1989

Remark ; NS : not specified.

Table 6. Pharmacological activities of compounds isolated from *Diospyros* species.

Compounds	Pharmacological activity	References
β -Amyrin	Moderately cytotoxic	Yan <i>et al.</i> , 1989
Betulin	(i) Anti-inflammatory activity	Recio <i>et al.</i> , 1995
	(ii) Active against the Walker-Carcinoma-256 tumor system	Misra and Pandey, 1989
Betulinic acid	(i) Showed potent anti-inflammatory activity against TPA-induced edema	Recio <i>et al.</i> , 1995a Recio <i>et al.</i> , 1995b
	(ii) Active against the Walker-Carcinoma-256 tumor system	Misra and Pandey, 1989
	(iii) Inhibited P-388 leukaemia growth	Chen <i>et al.</i> , 1989
	(iv) Highly selective activity against human melanoma <i>in vitro</i> and <i>in vivo</i> .	Pisha <i>et al.</i> , 1995

Table 6. Pharmacological activities of compounds isolated from *Diospyros* species (continued).

Compounds	Pharmacological activity	References
Lupeol	Active against the Walker-Carcinoma-256 tumor system	Misra and Pandey, 1989
Lupeol acetate	Inhibited stress induce ulcers in rats,also decreases incidence of gastric ulceration	Gupta <i>et al.</i> , 1981
2 α ,3 β -Dihydroxy-olean-12-ene-28-oic acid (maslinic acid)	(i) Showed anti-inflammatory effect and inhibitory effect on histamine induced ileum contraction (ii) Showed potent inhibitory activity against HIV-1 protease	Shimizu <i>et al.</i> , 1986 Xu <i>et al.</i> , 1996
Oleanolic acid	(i) inhibited 12-O-tetradecanoyl-phorbol-13-acetate induced Epstein-Barr virus activation (ii) Showed potent antiarthritic and anti-inflammatory activities	Ohigashi <i>et al.</i> , 1986 Singh <i>et al.</i> , 1994

Table 6. Pharmacological activities of compounds isolated from *Diospyros* species (continued).

Compounds	Pharmacological activity	References
Taraxerol	Inhibited stress induced ulcers in rats, decreases incidence of gastric ulceration	Gupta <i>et al.</i> , 1981
Ursolic acid	(i) Anti-inflammatory activity	Recio <i>et al.</i> , 1995
	(ii) inhibited 12-O-tetradecanoyl-phorbol-13-acetate induced Epstein-Barr virus activation	Ohigashi <i>et al.</i> , 1986
	(iii) Suppressed tumor promoter induced inflammation of mouse ear	Hirota <i>et al.</i> , 1990
	(iv) Inhibited stress induced ulcers in rats, also decreases incidence of gastric ulceration induced by pyloric ligation	Gupta <i>et al.</i> , 1981
	(v) Increases blood sugar concentration, glycogen and ATP contents in muscles, heart and uterus on intergastric administration into rats	Golovina and Vasilenko, 1976 Golovina and Vasilenko, 1978

Table 6. Pharmacological activities of compounds isolated from *Diospyros* species (continued).

Compounds	Pharmacological activity	References
Ursolic acid	(vi) Showed potent inhibitory activity against HIV-1 protease	Singh <i>et al.</i> , 1994
Plumbagin	(i) stains the skin, produces blisters and effect mucous membrane (ii) inhibits the growth of all gram +ve and -ve test bacteria	Roy <i>et al.</i> , 1955 Apandi <i>et al.</i> , 1994
7-Methyljuglone	Showed cytotoxic activity against human colon carcinoma cells	Gafner <i>et al.</i> , 1987 ; Gafner <i>et al.</i> , 1988 ; Gafner <i>et al.</i> , 1989 ; Marston <i>et al.</i> , 1986
Diospyrin	(i) Inhibited the <i>in vivo</i> growth of Ehrlich Ascites Carcinoma (EAC). in Swiss Albino mice	Hazra <i>et al.</i> , 1984

Table 6. Pharmacological activities of single isolated metabolites of *Diospyros* species (continued).

Compounds	Pharmacological activity	References
Diospyrin	(ii) Showed <i>in vitro</i> activity against the protozoan <i>L. donovani</i>	Hazra <i>et al.</i> , 1986
Isodiospyrin	Showed cytotoxic activity against HCT-8 colon tumor and P-388 lymphocytic leukaemia	Gafner <i>et al.</i> , 1987 ; Gafner <i>et al.</i> , 1988 ; Gafner <i>et al.</i> , 1989 ; Yan <i>et al.</i> , 1989
Diospyrol	Showed anthelmintic activity	Fukami <i>et al.</i> , 1978 ; Fukami <i>et al.</i> , 1979 ; Sen <i>et al.</i> , 1974. Sen <i>et al.</i> , 1975.
Astragalin	Inhibited the angiotensin converting enzyme activity	Kameda <i>et al.</i> , 1987
Kaempferol-3-O-(2''-O-galloyl)-glucoside	Inhibited the angiotensin converting enzyme activity	Kameda <i>et al.</i> , 1987

Table 6. Pharmacological activities of single isolated metabolites of *Diospyros* species (continued).

Compounds	Pharmacological activity	References
Isoquereitrin	Inhibited the angiotensin converting enzyme activity	Kameda <i>et al.</i> , 1987
Quercetin-3-O-(2''-O-galloyl)-glucoside	Inhibited the angiotensin converting enzyme activity	Kameda <i>et al.</i> , 1987

CHAPTER III

EXPERIMENTAL

Source of Plant Material

The stem of *Diospyros rubra* Lec. were collected from Kao Soi-Dao National Park, Chanthaburi Province, Thailand, in April 2000. The plant was identified by comparison with the voucher specimen (BKF NO. 13260) at the Botanical Section, Royal Forest Department, Ministry of Agriculture and Co-operative, Thailand.

General Techniques

1. Chromatographic Techniques

1.1 Thin-Layer Chromatography (TLC)

- Technique : one way ascending
- Stationary phase : Silica gel 60F 254, precoated plate layer thickness 0.2 mm.
- Solvent systems : Various solvent systems depending on materials.
- Distance : 10 cm.
- Temperature : 28-35°C (room temperature)
- Detection : 1) UV light (254 and 365)
2) 10% sulfuric acid in ethanol and heating at 110°C

1.2 Column Chromatography (CC)

- Column : Flat bottom glass column (various diameter)
- Stationary phase : Silica gel 60 (No. 9385, E. Merck) particle size 0.040-0.063 mm. (230-400 mesh ASTM)

- Packing method : Dry and wet packing
- Sample loading : Dry packing - The sample was dissolved in a small amount of suitable organic solvent, mixed with a small quantity of adsorbent, triturated, dried and then loaded on the top of the column.
- Wet packing - The sample was dissolved in a small amount of the eluent, then loaded on the top of the column.
- Technique : Long and short column chromatography.
- Solvent system : Various solvent systems depending on materials.
- Detection : Fractions were examined by TLC observing under UV light at the wavelengths of 254 and 365 nm. The TLC plate was then sprayed with 10% sulfuric acid in ethanol and heated at 110°C. Fractions of similar chromatographic pattern were combined.

2. Spectroscopy

2.1 Infrared (IR) Absorption Spectra

IR spectra (KBr disc and thin film) were obtained on a Perkin Elmer Infrared Spectrophotometer Model 283 (Pharmaceutical Research Equipment Center, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

2.2 Mass Spectra (MS)

The electron impact mass spectra (EIMS) were obtained on a Fisons VG Trio 2000 quadrupole mass spectrometer (Department of Chemistry, Faculty of Science, Mahidol University) operating at 70 eV.

2.3 Proton and Carbon-13 Nuclear Magnetic Resonance (^1H and ^{13}C NMR) Spectra

The ^1H and ^{13}C NMR spectra were obtained either on a Bruker Avance 400 Ultra Shield™ 400 MHz NMR spectrometer (Department of Chemistry, Faculty

of Science, Burapha University) or a Bruker Avance DPX-300 300 MHz NMR spectrometer (Faculty of Pharmaceutical Sciences, Chulalongkorn University).

NMR solvents used in this study were deuterated chloroform (CDCl_3) and deuterated dimethylsulfoxide ($\text{DMSO}-d_6$). Chemical shifts were reported in ppm scale using the chemical shift of the solvent as the reference signal.

3. Melting Points

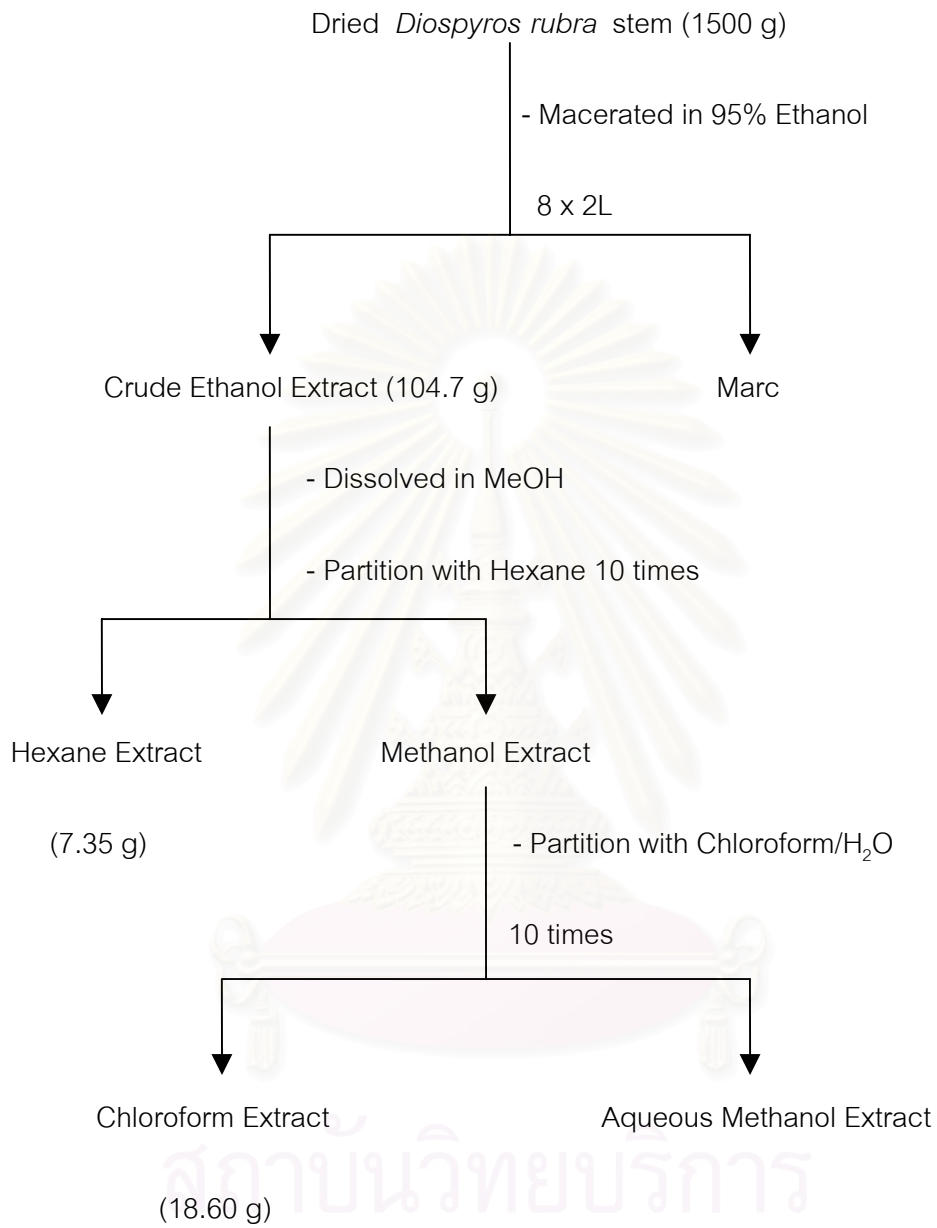
Melting points were obtained on a Gallenkamp Melting Point Apparatus Model MFB 595 (Department of Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

4. Solvents

Throughout this work, all organic solvents were of commercial grade and were redistilled prior to used.

Extraction Procedure

The dried stem of *Diospyros rubra* Lec. (1500 g) were ground into small pieces, macerated eight times in 95% ethanol (2 liters 2 days each) and then filtered. The filtrate of each batch was combined and concentrated under reduced pressure to yield 104.7 g of dried crude extract (6.98% of dried weight). The ethanol extract was dissolved in methanol/ H_2O , then partitioned (10 times) with 1 liter of hexane and chloroform, respectively. Each fraction was evaporated to dryness under reduced pressure to give 7.35 g of the hexane extract (0.49% of dried weight) and 18.60 g of the chloroform extract (1.24% of dried weight).



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Scheme 1. Extraction of *Diospyros rubra* Lec. stem.

Isolation Procedure

1. Fractionation of the hexane extract

The hexane extract (7.35 g) was subjected to silica gel short column chromatography using hexane : chloroform (1 : 1) as the eluent. The extract was dissolved in a small volume of the eluent and blended with silica gel (60.53 g) until dried, then loaded to the top of a glass column (10.0 x 15 cm) already packed with a slurry of silica gel (500 g), and eluted with the eluent. One hundred and seventy-two 30-ml fractions were collected and combined according to their TLC patterns into seven major fractions, DRH 01 – DRH 07 (Table 7). The column was then washed down with methanol and the eluate was combined as fraction DRH 08.

Table 7. Combined fractions from the hexane extract.

Fraction	Number of eluates	Weight (g)
DRH 01	1 - 21	0.12
DRH 02	22 - 30	0.19
DRH 03	31 - 37	1.21
DRH 04	38 - 40	0.88
DRH 05	41 - 51	0.95
DRH 06	52 - 84	2.98
DRH 07	85 - 125	0.44
DRH 08	126 - 172	0.56

1.1 Isolation of compounds DR 1 and DR 2

Both fractions DRH 03 and DRH 04, which gave a major violet-blue spot on TLC plate, were combined and purified by recrystallization in methanol to give compound DR 1 as colorless needles (1.4025 g).

Fraction DRH 06 (2.98 g) was separated by column chromatography using silica gel (200 g, 5.0 x 26 cm) with hexane : chloroform (1 : 2) as the eluent. Each 30 ml fraction was collected and detected by TLC, using hexane : chloroform (2 : 3) as the developing solvent system. One hundred and two fractions were collected and combined according to their TLC patterns into six major fractions (DRH 09 – DRH 14) as shown in Table 8.

Table 8. Combined fractions from DRH 06.

Fraction	Number of eluates	Weight (g)
DRH 09	1 – 12	0.48
DRH 10	13 – 29	0.55
DRH 11	30 – 47	0.26
DRH 12	48 – 65	0.58
DRH 13	66 – 87	0.47
DRH 14	88 – 102	0.45

Fraction DRH 10 (0.55 g) was separated by column chromatography using silica gel (30 g, 2.0 x 20 cm) with hexane : chloroform (2 : 3) as the eluent. The fractional volume was about 30 ml. The eluates were collected and combined following TLC examination, with hexane : chloroform (2 : 3) as the developing solvent system, into ten fractions (DRH 15 - DRH 24) as shown in Table 9.

Table 9. Combined fractions from DRH 10.

Fraction	Number of eluates	Weight (g)
DRH 15	1 – 15	0.035
DRH 16	16 – 27	0.078
DRH 17	28 – 33	0.021
DRH 18	34 – 45	0.032
DRH 19	46 – 66	0.045
DRH 20	67 – 92	0.012
DRH 21	93 – 96	0.008
DRH 22	97	0.003
DRH 23	98 – 104	0.041
DRH 24	105 – 113	0.056

Fraction DRH 16 was recrystallized in methanol to give compound DR 1 (0.023 g), and fraction DRH 18 which gave a major violet-red spot upon TLC detection was recrystallized in methanol to give compound DR 2 as colorless needles (0.0074 g).

1.2 Isolation of compound DR 3

Fraction DRH 14 (0.45 g) was separated by silica gel column chromatography (50 g, 2.0x35 cm) with chloroform as the eluent. The fractional volume was about 15 ml. The eluates were collected and combined following TLC examination, with chloroform as the developing solvent system, into four fractions (DRH 25 – DRH 28) as shown in Table 10.

Table 10. Combined fractions from DRH 14.

Fraction	Number of eluates	Weight (g)
DRH 25	1 – 7	0.12
DRH 26	8 – 16	0.08
DRH 27	17 – 23	0.07
DRH 28	24 – 41	0.13

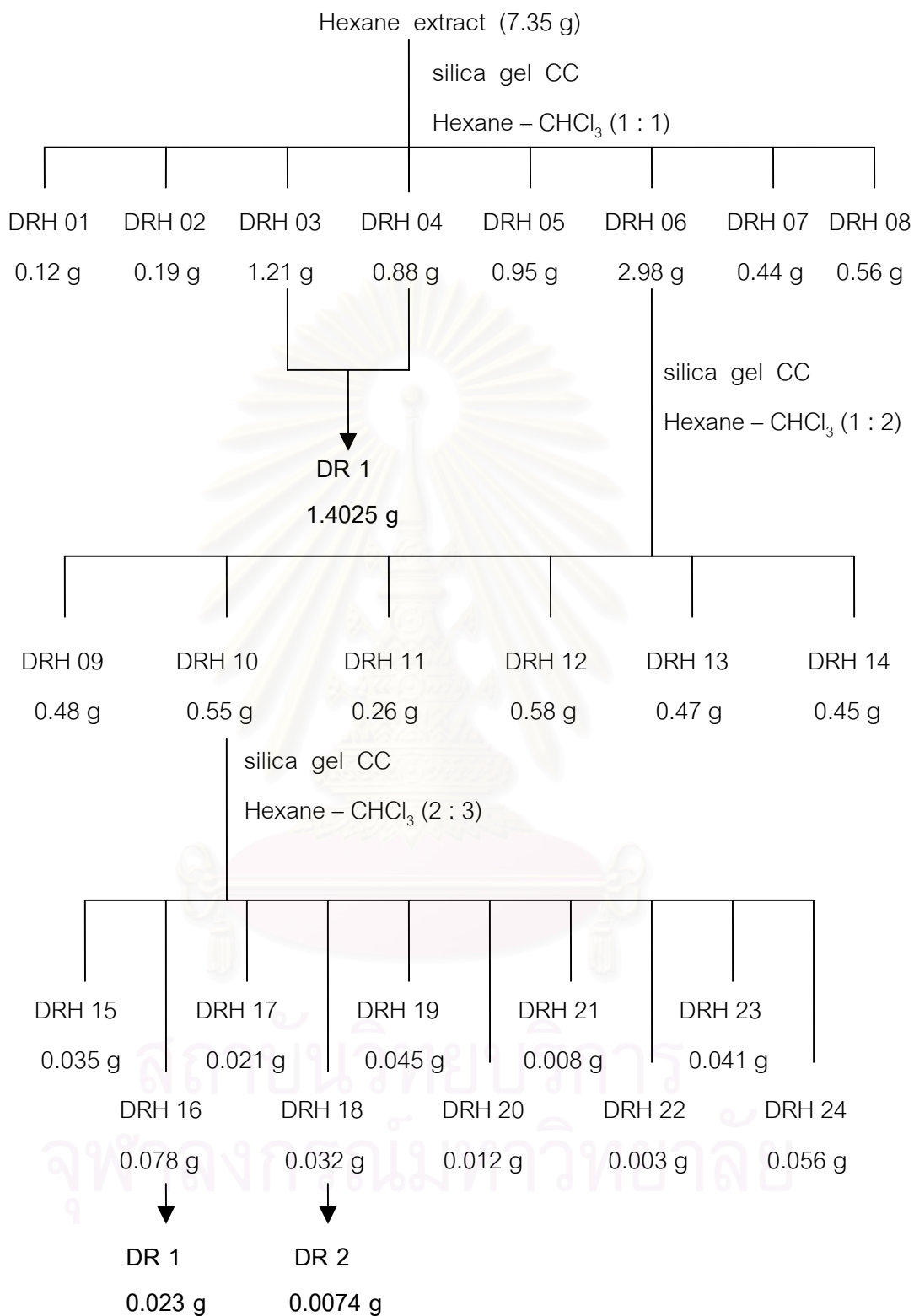
Fraction DRH 25 was separated by silica gel column chromatography (20 g, 2.0x20 cm) with chloroform as the eluent. The fractional volume was about 10 ml. The eluates were collected and combined following TLC examination, with chloroform as the developing solvent system, into five fractions (DRH 29 – DRH 33) as shown in Table 11.

Table 11. Combined fractions from DRH 25.

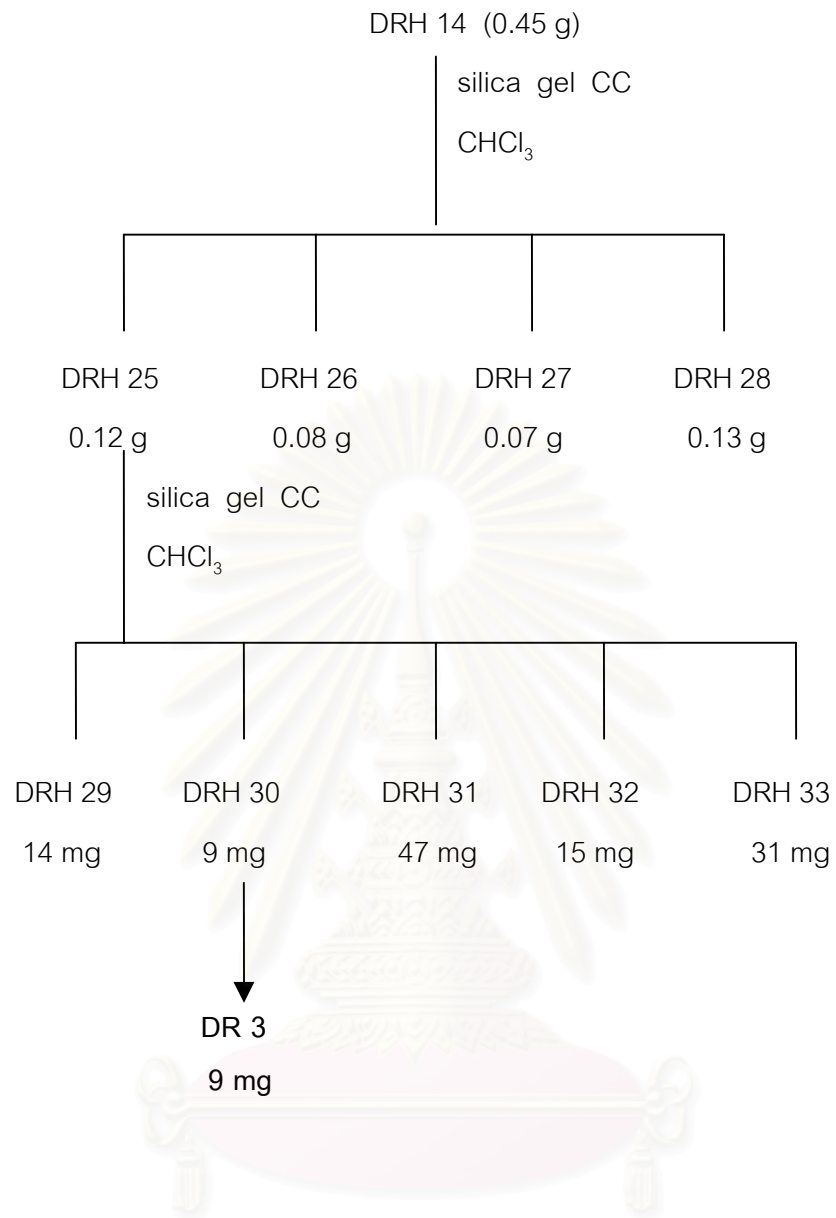
Fraction	Number of eluates	Weight (g)
DRH 29	1 – 12	0.014
DRH 30	13 – 18	0.009
DRH 31	19 – 31	0.047
DRH 32	32 – 37	0.015
DRH 33	38 – 62	0.031

Fraction DRH 30 gave a major violet-blue spot upon TLC detection. It was therefore recrystallized in methanol to give compound DR 3 as white amorphous powder (9 mg).

The fractionation of the hexane extract is summarized in Scheme 2.



Scheme 2. Fractionation of hexane extract.



Scheme 2. Fractionation of hexane extract (continued).

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2. Fractionation of the chloroform extract

The chloroform extract (18.60 g) was subjected to silica gel column chromatography (500 g, 6.0x40 cm) using chloroform : methanol (98 : 2) as the eluent. The extract was dissolved in the eluent and blended with silica gel (65 g) until dried, then loaded on top of the column and eluted with the eluent. One hundred and eighty 30-ml fractions were collected and combined according to their TLC patterns into eight major fractions (DRC 01 – DRC 08) as shown in Table 12.

Table 12. Combined fractions from the chloroform extract.

Fraction	Number of eluates	Weight (g)
DRC 01	1 – 26	1.7450
DRC 02	27 – 59	2.4864
DRC 03	60 – 74	0.9560
DRC 04	75 – 91	1.5400
DRC 05	92 – 105	1.0073
DRC 06	106 – 138	2.0581
DRC 07	139 – 150	0.9850
DRC 08	151 – 180	6.2146

Fraction DRC 07 was separated by silica gel column chromatography (50 g, 2.0x20 cm) with chloroform : methanol (95 : 5) as the eluent. The fractional volume was about 30 ml. The eluates were collected and combined following TLC examination, with chloroform : methanol (95 : 5) as the developing solvent system, into four fractions (DRC 09 – DRC 12) as shown in Table 13.

Table 13. Combined fractions from DRC 07.

Fraction	Number of eluates	Weight (g)
DRC 09	1 – 6	0.0485
DRC 10	7 – 19	0.2378
DRC 11	20 – 26	0.0540
DRC 12	27 – 45	0.3710

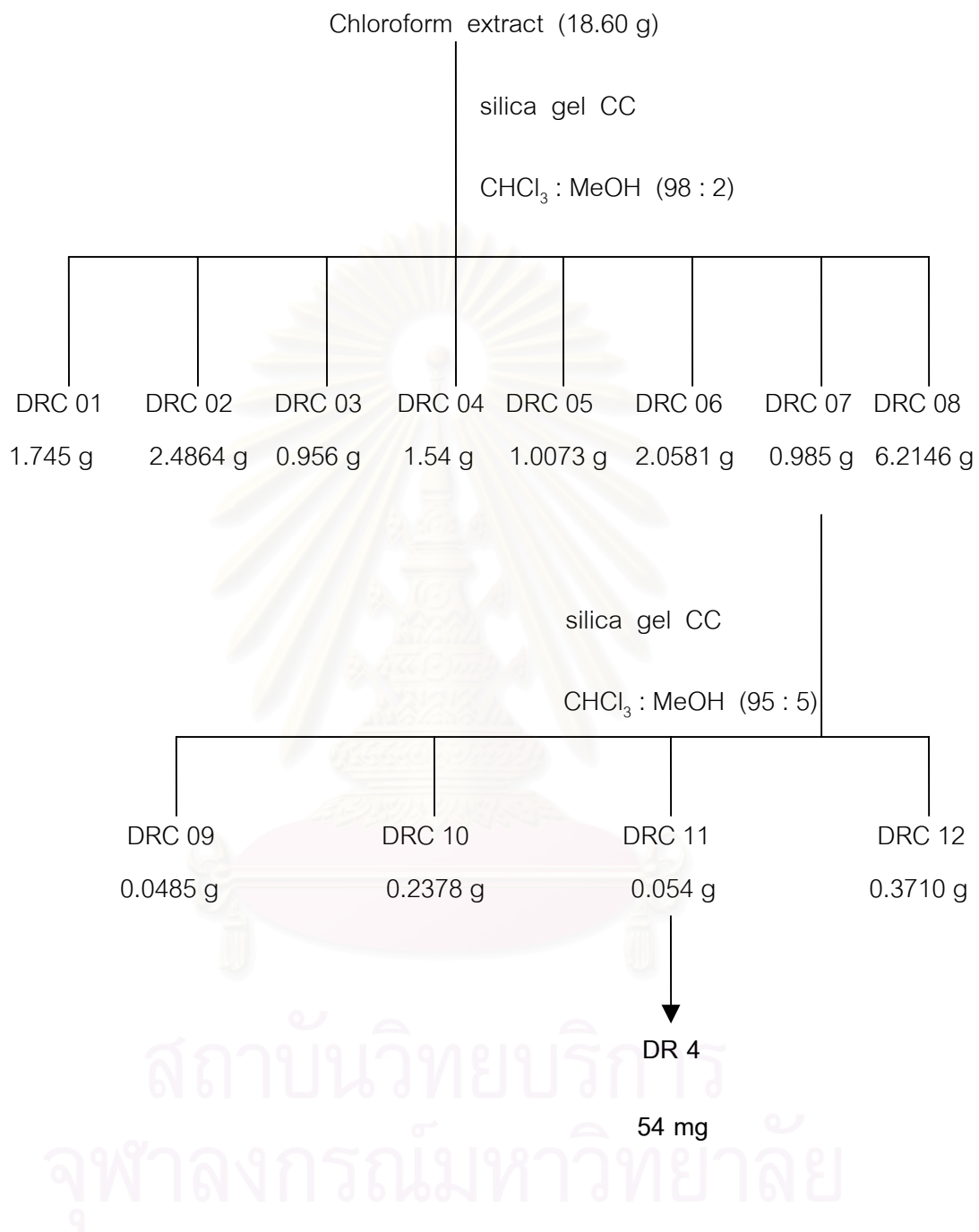
Fraction DRC 11 gave a major violet-red spot upon TLC detection with 10% sulfuric acid. It was therefore recrystallized in methanol to give compound DR 4 as colorless needles (0.054 g).

The fractionation of the chloroform extract is summarized in Scheme 3.

All isolated compounds were test with Liebermann-Burchard reagent and the results are shown in Table 14. The total amount of each compounds is also shown in the same table.

Table 14. Result of Liebermann-Burchard test and the total amount of isolated compounds.

Compound	Color with Liebermann-Burchard test	Total amount	
		Weight (g)	% yield
DR 1	Violet	1.4260	0.0950
DR 2	Blue	0.0074	0.0005
DR 3	Red – violet	0.0090	0.0006
DR 4	Red – violet	0.0540	0.0040



Scheme 3. Fractionation of chloroform extract.

Characterization of isolated compounds

1. Compound DR 1 (1.426 g, 0.095% yield)

Appearance : Colorless needles (methanol)

Solubility : Soluble in chloroform

Melting point : 215 – 216 °C

EIMS m/z : 426(16), 411(20), 409(32), 393(11), 383(5), 257(23),
(% relative intensity) 218(81), 205(46), 191(47), 189(100), 176(15),
148(31), 124(49), 108(37), 95(19), 80(27), 67(14) and
56(8) (Figure 2, page 79)

IR ν_{\max} (thin film) cm^{-1} : 3436, 2934, 2863, 1644, 1454, 1378, 1040, 881, 758,
476 (Figure 3, page 80)

^1H – NMR (δ ppm, 400 MHz, CDCl_3) (Figures 6a-6c, pages 84-86)

4.70 (1H, s), 4.60 (1H, s), 3.20 (1H, *dd* $J = 5.2, 5.2$ Hz), 2.40 (1H, *dt*), 1.91
(1H, *m*), 1.71 (3H, s), 1.05 (3H, s), 0.99 (3H, s), 0.97 (3H, s), 0.85 (3H, s), 0.81 (3H,
s), 0.79 (3H, s)

^{13}C – NMR (δ ppm, 100 MHz, CDCl_3) (Figures 4a-4b, pages 81-82)

150.9 (s), 109.3 (t), 79.0 (d), 55.3 (d), 50.5 (d), 48.3 (d), 48.0 (d), 43.0 (s),
42.8 (s), 40.9 (s), 40.0 (t), 38.9 (s), 38.7 (t), 38.1 (d), 37.2 (s), 35.6 (t), 34.3 (t), 29.9
(t), 28.0 (q), 27.4 (t), 27.5 (t), 25.2 (t), 20.9 (t), 19.3 (q), 18.3 (t), 18.0 (q), 16.1 (q),
16.0 (q), 15.3 (q), 14.6 (q)

2. Compound DR 2 (0.0074 g, 0.0005% yield)

Appearance : Colorless needles (methanol)

Solubility : Soluble in hexane, chloroform

Melting point : 140 –141 °C

EIMS m/z : 414(73), 396(64), 381(42)379(17), 329(75), 273(34),
(% relative intensity) 255(58), 231(46), 213(100), 199(56), 173(45), 159(75),
108(57), 95(57), 91(39), 81(34) and 56(10) (Figure
7, page 90)

IR ν_{\max} (thin film) cm^{-1} : 3436, 2933, 2868, 2371, 2289, 1639, 1465, 1372,
1101, 1050, 876, 476 (Figure 8, page 91)

^1H – NMR (δ ppm, 400 MHz, CDCl_3) (Figures 9a-9b, pages 92-93)

5.36 (1H, d $J = 5.2$ Hz), 5.14 (1H, dd $J = 15.2, 8.8$ Hz), 5.01 (1H, dd $J = 15.2,$
8.8 Hz), 3.54 (1H, m), 1.02 (3H, s), 0.93 (3H, d), 0.88 (3H, t), 0.85 (3H, d), 0.82 (3H,
 d), 0.70 (3H, s)

^{13}C – NMR (δ ppm, 100 MHz, CDCl_3) (Figures 10a-10b, pages 94-95)

140.7 (s), 121.7 (d), 71.8 (d), 56.7 (d), 56.0 (d), 50.1 (d), 45.8 (d), 42.3 (t),
42.3 (s), 39.7 (t), 37.2 (t), 36.5 (s), 36.1 (d), 33.9 (t), 31.9 (d), 31.9 (t), 31.6 (t), 29.1
(d), 28.2 (t), 26.0 (t), 24.3 (t), 23.0 (t), 21.0 (t), 19.8 (q), 19.4 (q), 19.0 (q), 18.8 (q),
11.9 (q), 11.8 (q)

3. Compound DR 3 (9 mg, 0.0006% yield)

Appearance : White amorphous powder (methanol)

Solubility : Soluble in chloroform

Melting point : 260 - 261 °C

EIMS m/z : 443(5), 442(12), 427(10), 411(32), 393(16), 207(47),
(% relative intensity) 203(100), 189(98), 95(48), 79(27), 67(24) and 55(11)
(Figure 13, page 102)

IR ν_{\max} (thin film) cm^{-1} : 3388, 2939, 2868, 1645, 1458, 1369, 1028, 987,
883, 472 (Figure 14, page 103)

^1H – NMR (δ ppm, 400 MHz, CDCl_3) (Figures 17a-17b, pages 107-108)

4.65 (1H, s), 4.55 (1H, s), 3.75 (1H, d $J = 10.8$ Hz), 3.29 (1H, d $J = 10.8$ Hz),
3.14 (1H, dd $J = 4.8, 4.8$ Hz), 1.65 (3H, s), 1.00 (3H, s), 0.95 (3H, s), 0.94 (3H, s),
0.80 (3H, s) and 0.73 (3H, s)

^{13}C – NMR (δ ppm, 100 MHz, CDCl_3) (Figures 15a-15b, pages 104-105)

150.9 (s), 110.1 (t), 78.7 (d), 61.0 (t), 56.4 (d), 51.4 (d), 48.8 (d), 48.6 (s),
48.2 (d), 42.8 (s), 40.8 (s), 39.2 (s), 38.4 (t), 37.6 (d), 37.0 (s), 34.7 (t), 34.4 (t), 30.2
(t), 29.6 (t), 27.8 (q), 27.5 (t), 26.6 (t), 25.7 (t), 21.2 (t), 20.0 (q), 18.7 (t), 17.1 (q),
16.4 (q), 15.8 (q) and 14.6 (q)

4. Compound DR 4 (54 mg, 0.004% yield)

Appearance : Colorless needles (methanol)
Solubility : Soluble in chloroform, methanol
Melting point : 259 °C
EIMS m/z : 438(2), 300(3), 248(100), 219(24), 203(68), 189(20),
(% relative intensity) 147(11), 133(61), 119(15), 95(6), 67(5) and 55(4)
(Figure 18, page 114)

IR ν_{\max} (thin film) cm^{-1} : 3460, 2925, 1689, 1637, 1261, 1089, 1029, 802
(Figure 19, page 115)

^1H – NMR (δ ppm, 400 MHz, $\text{DMSO-}d_6$) (Figures 22a-22b, pages 120-121)
5.11 (1H, *t*), 2.98 (1H, *m*), 1.02 (3H, *s*), 0.88 (3H, *d* $J = 7.2$ Hz), 0.88 (3H, *s*),
0.85 (3H, *s*), 0.79 (3H, *d* $J = 6.4$ Hz), 0.73 (3H, *s*), 0.66 (3H, *s*)

^{13}C – NMR (δ ppm, 75 MHz, $\text{DMSO-}d_6$) (Figures 20a-20b, pages 116-117)
178.4 (*s*), 138.3 (*s*), 124.7 (*d*), 77.0 (*d*), 54.9 (*d*), 52.5 (*d*), 47.1 (*s*), 46.9 (*d*),
41.8 (*s*), 38.8 (*s*), 38.6 (*d*), 38.5 (*d*), 38.5 (*s*), 38.3 (*t*), 36.6 (*s*), 36.4 (*t*), 32.8 (*t*), 30.3
(*t*), 28.4 (*q*), 27.6 (*t*), 27.1 (*t*), 23.9 (*t*), 23.4 (*q*), 23.0 (*t*), 21.2 (*q*), 18.1 (*t*), 17.1 (*q*),
17.0 (*q*), 16.2 (*q*), 15.3 (*q*)

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CHAPTER IV

RESULTS AND DISCUSSION

Chromatographic separation of the hexane and chloroform extracts of the stem of *Diospyros rubra* Lec. led to the isolation of four chemical constituents. The identification of these compounds was based on analysis of their spectroscopic data (IR, NMR and mass spectra) and also confirmed by comparison with those values previously reported in the literature. The details can be discussed as follows.

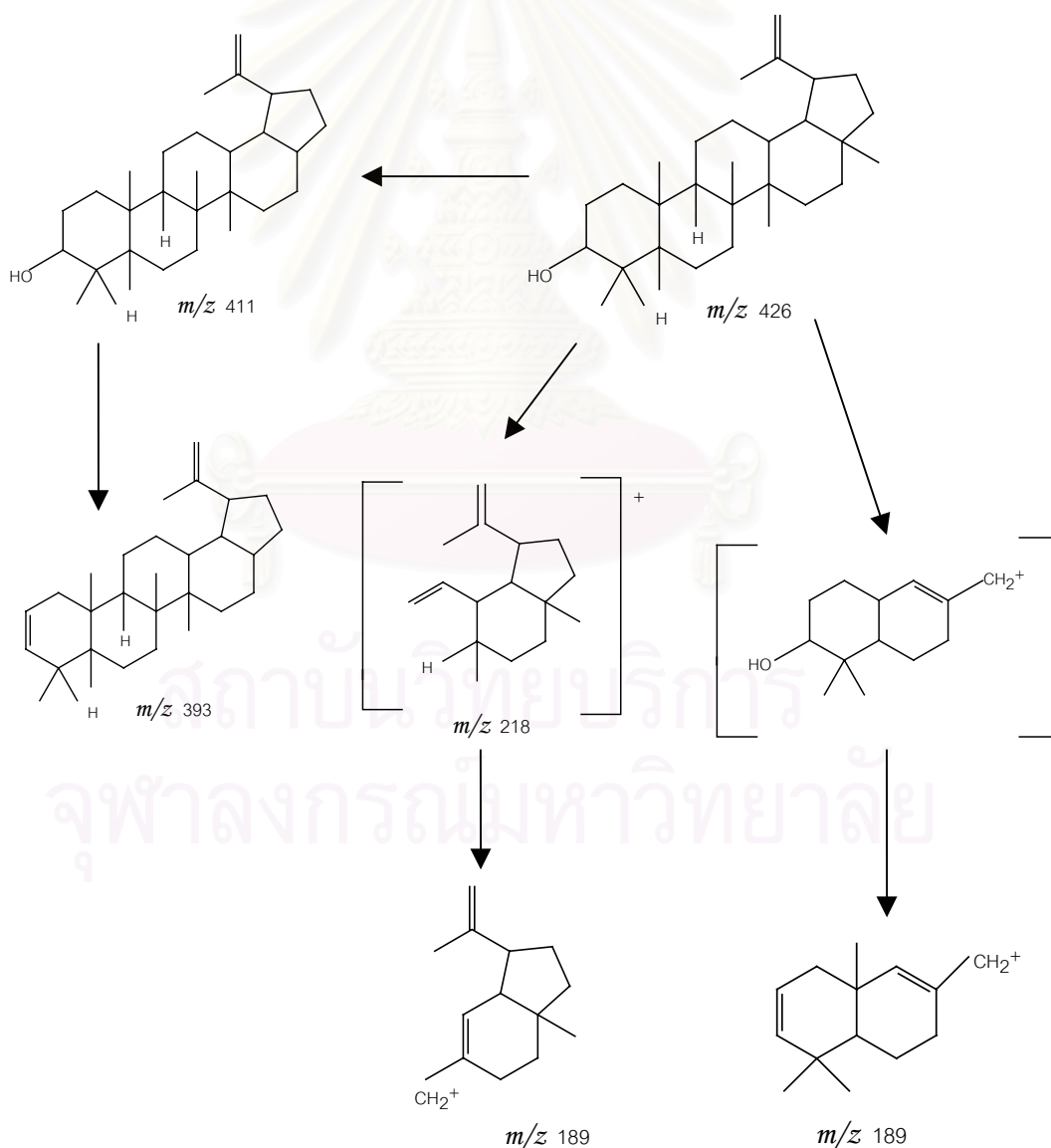
1. Identification of compound DR 1

Compound DR 1 was obtained as colorless needles (1.426 g, 0.095 % yield) from fractions DRH 03, DRH 04 and DRH 16 of the hexane extract. The compound gave violet color to Lieberman-Burchard reagent, suggesting that it is a triterpenoid. The EIMS spectrum of this compound (Figure 2) showed a molecular ion peak at m/z 426 which corresponded to the molecular formula of $C_{30}H_{50}O$. Successive losses of methyl and water produced fragment peaks at m/z 411 and 393, respectively. Intense fragment peaks at m/z 189 and 218 were suggestive of a pentacyclic triterpenoid with the lupane skeleton (Budzikiewicz, Wilson and Djerassi, 1963 ; Ogunkoya, 1981). These prominent peaks were the results of cleavage at different positions across the C-ring of the lupane skeleton as shown in Scheme 4. The presence of the alcohol functionality in the molecule was indicated by an IR absorption band at 3436 cm^{-1} (Figure 3).

The ^{13}C – NMR spectrum of DR 1 (Figure 4a – 4b) showed 30 carbon signals, supportive of a triterpenoid structure. The DEPT-90 and DEPT-135 experiments (Figure 5) were employed to classify these signals into those of six quaternary carbons at δ 37.2, 38.9, 40.9, 42.8, 43.0 and 150.9 ppm, six methine carbons at 38.1, 48.0, 48.3, 50.5, 55.3 and 79.0 ppm, eleven methylene carbons at

δ 18.3, 20.9, 25.2, 27.4, 27.5, 29.9, 34.3, 35.6, 38.7, 40.0 and 109.3 ppm, and seven methyl carbons at δ 14.6, 15.3, 16.0, 16.1, 18.0, 19.3 and 28.0 ppm. The two most downfield carbon signals at 150.9 and 109.3 ppm represents the disubstituted double bond at C-20 and C-29 in the lupane skeleton.

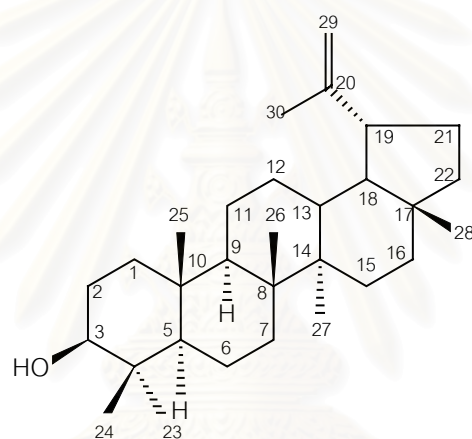
The ^1H -NMR spectrum (Figure 6a – 6c) showed seven singlets of tertiary methyls at δ 0.79 (H₃-24), 0.81 (H₃-28), 0.85 (H₃-25), 0.97 (H₃-27), 0.99 (H₃-23), 1.05 (H₃-26) and 1.71 ppm (H₃-30). The presence of exomethylene protons (H₂-29) could be observed as two downfield singlets (br) at δ 4.60 and 4.70 ppm. A double doublet at δ 3.20 ppm was attributable to the carbinolic proton (H-3).



Scheme 4. Mass fragmentation of compound DR 1

All spectroscopic data of DR 1 are in accordance with the structure of lupeol, a known triterpenoid of the lupane type. Comparison of the ^{13}C -NMR data of this compound with those previously reported for lupeol (Mahato and Kundu, 1994), is shown in Table 15.

Therefore, it was concluded that compound DR 1 is lupeol, the structure of which is shown below.



Lupeol ($\text{C}_{30}\text{H}_{50}\text{O}$)

Lupeol was previously isolated from several *Diospyros* species. It could be found in almost all parts of the plants, especially in the bark and heartwood. The compound have been reported as active against the Walker-Carcinoma-256 tumor system (Misra and Pandey, 1989). Lupeol was also revealed to possess antifungal and germination inhibitory activities (Higa *et al.*, 1998).

Table 15. Comparison of ^{13}C -NMR data of lupeol (in CDCl_3) and compound DR 1 (in CDCl_3).

Carbon	Chemical shift (δ) ppm	
	Lupeol	DR 1
1	38.7	38.7
2	27.4	27.5
3	78.9	79.0
4	38.8	38.9
5	55.3	55.3
6	18.3	18.3
7	34.2	34.3
8	40.8	40.9
9	50.4	50.5
10	37.1	37.2
11	20.9	20.9
12	25.1	25.2
13	38.0	38.1
14	42.8	42.8
15	27.4	27.4
16	35.5	35.6
17	43.0	43.0
18	48.2	48.3
19	47.9	48.0
20	150.9	150.9
21	29.8	29.9
22	40.0	40.0
23	28.0	28.0
24	15.4	15.3
25	16.1	16.1
26	15.9	16.0
27	14.5	14.6
28	18.0	18.0
29	109.3	109.3
30	19.3	19.3

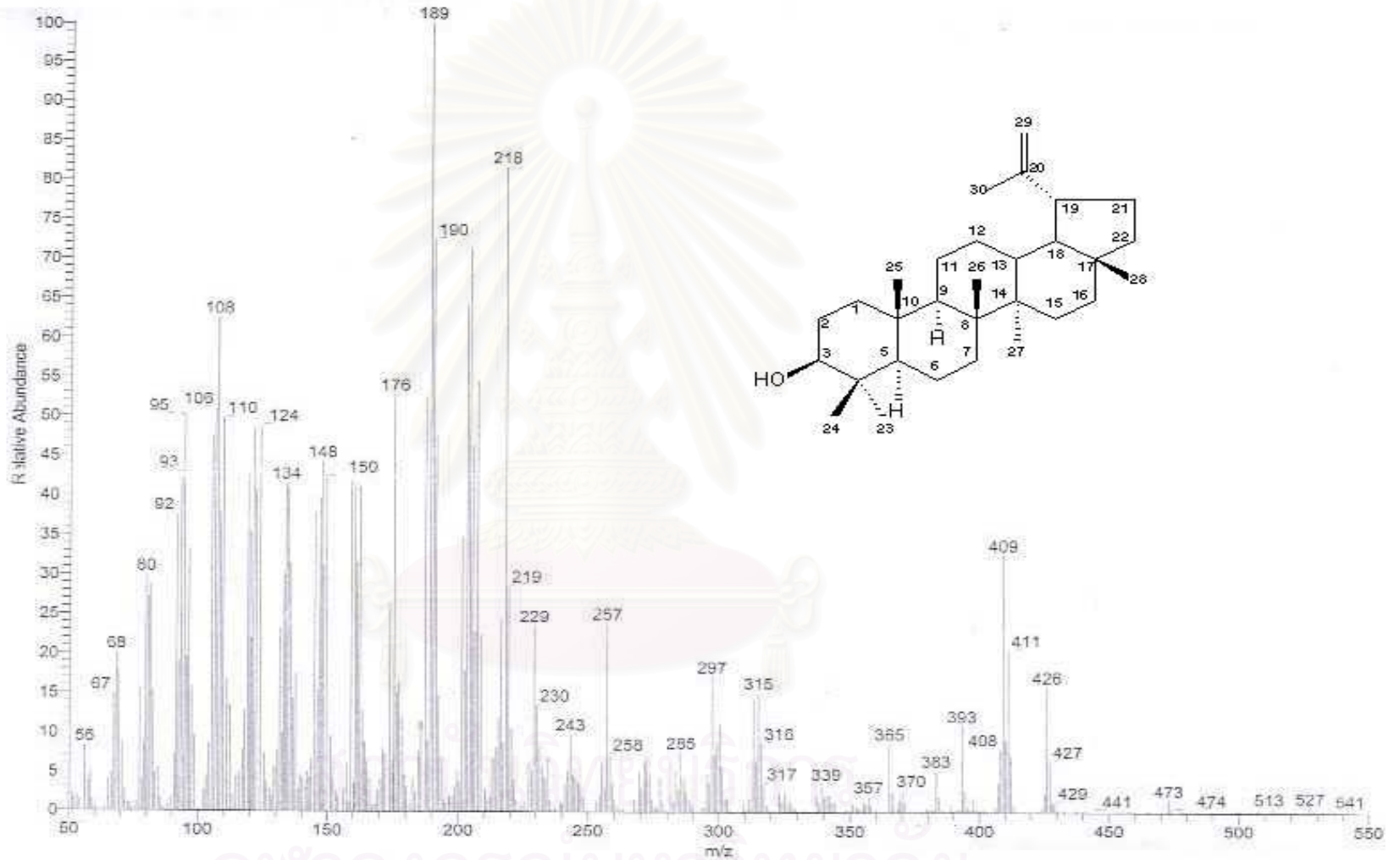


Figure 2. EIMS of compound DR 1

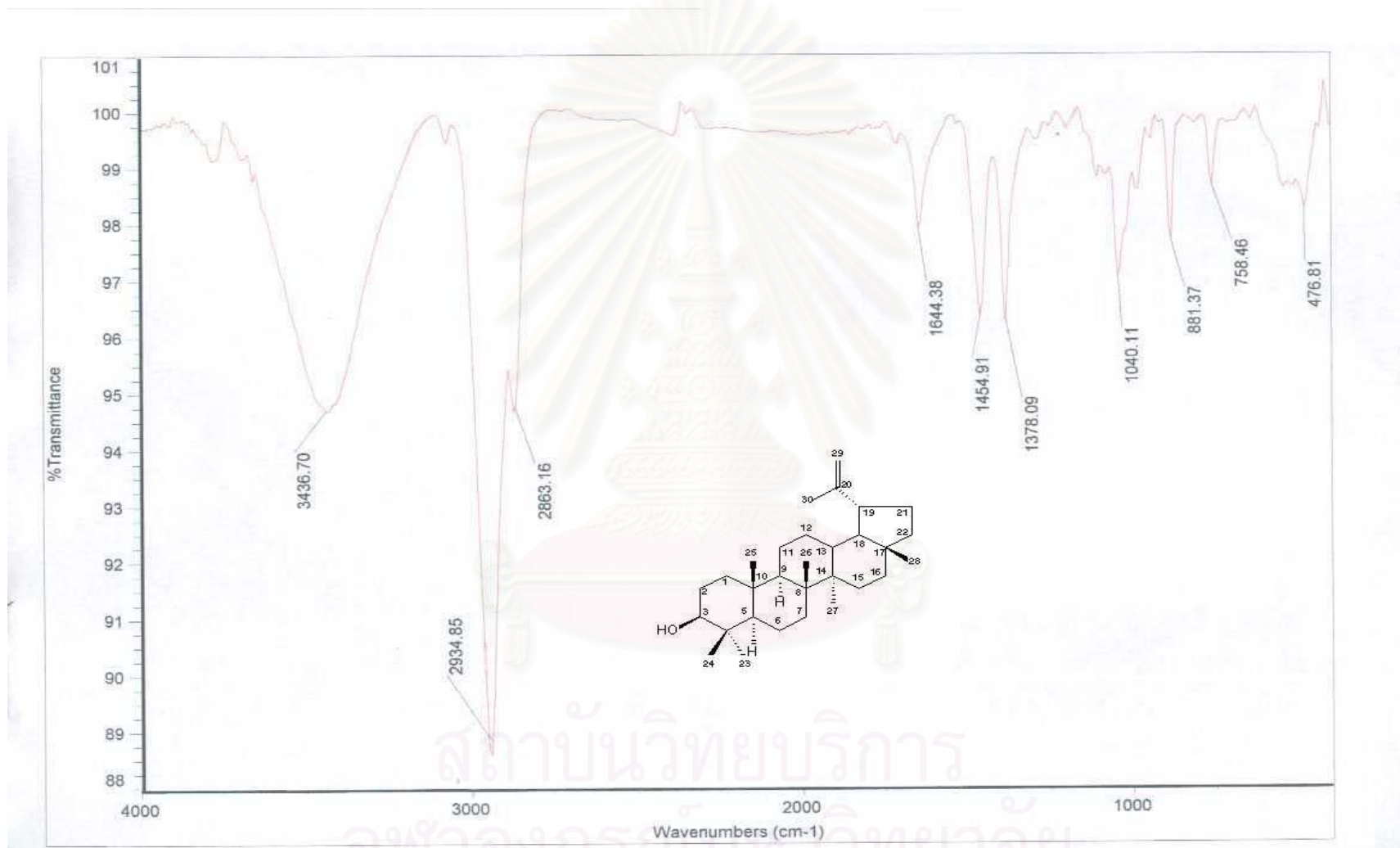


Figure 3. IR spectrum of compound DR 1

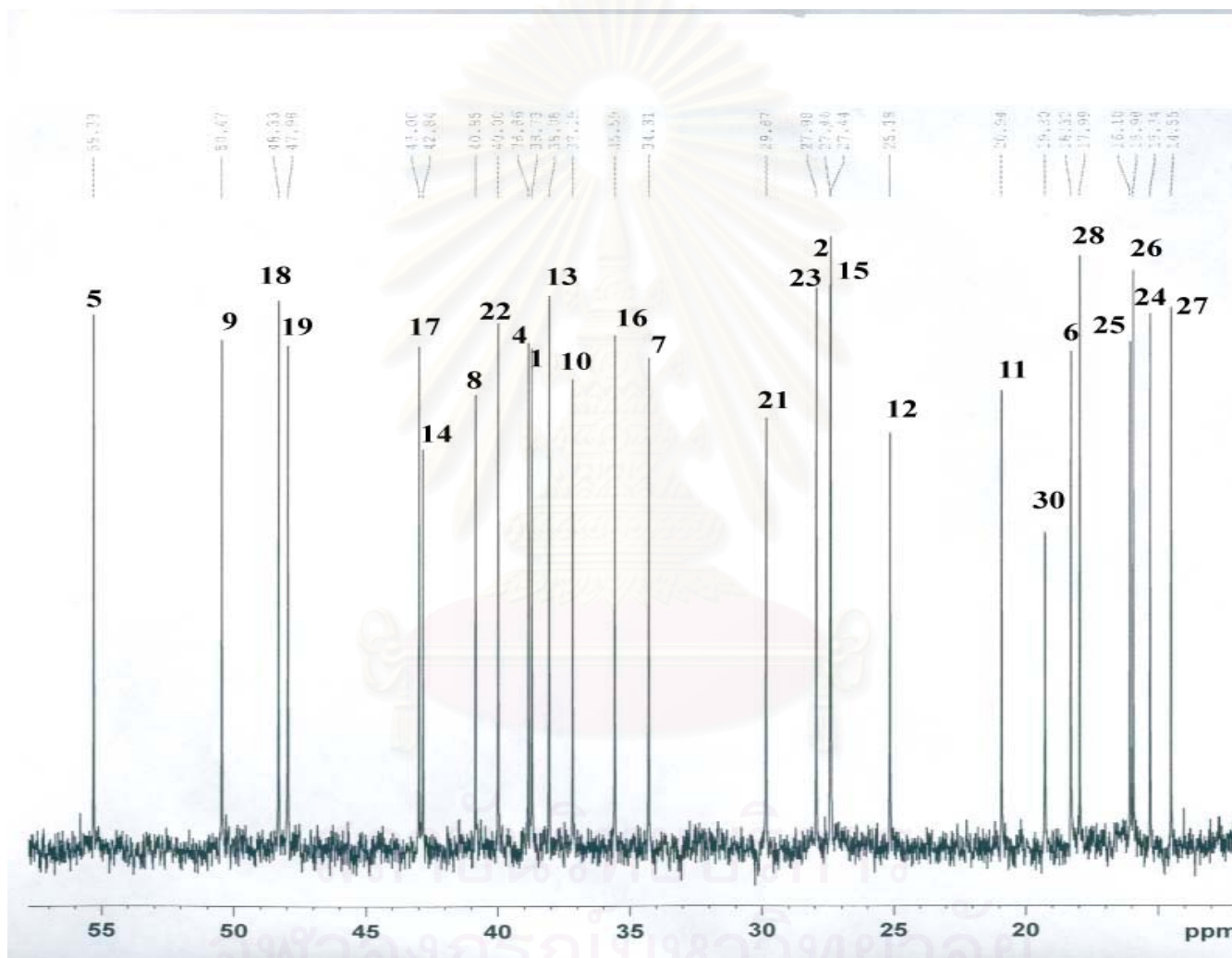


Figure 4b. The 100 MHz ^{13}C -NMR spectrum of compound DR 1 (expanded)

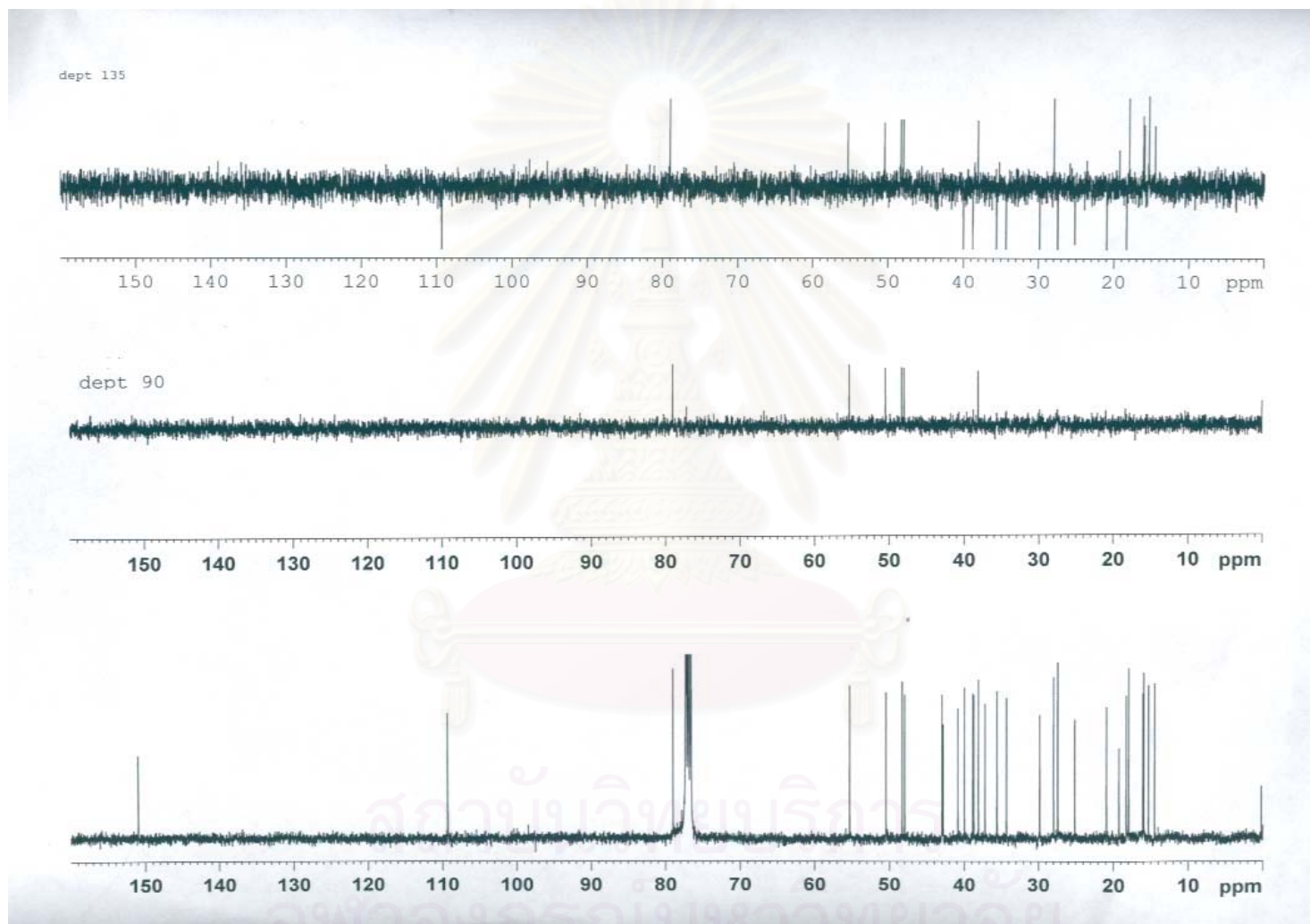


Figure 5. The 100 MHz ^{13}C -DEPT NMR spectra of compound DR 1

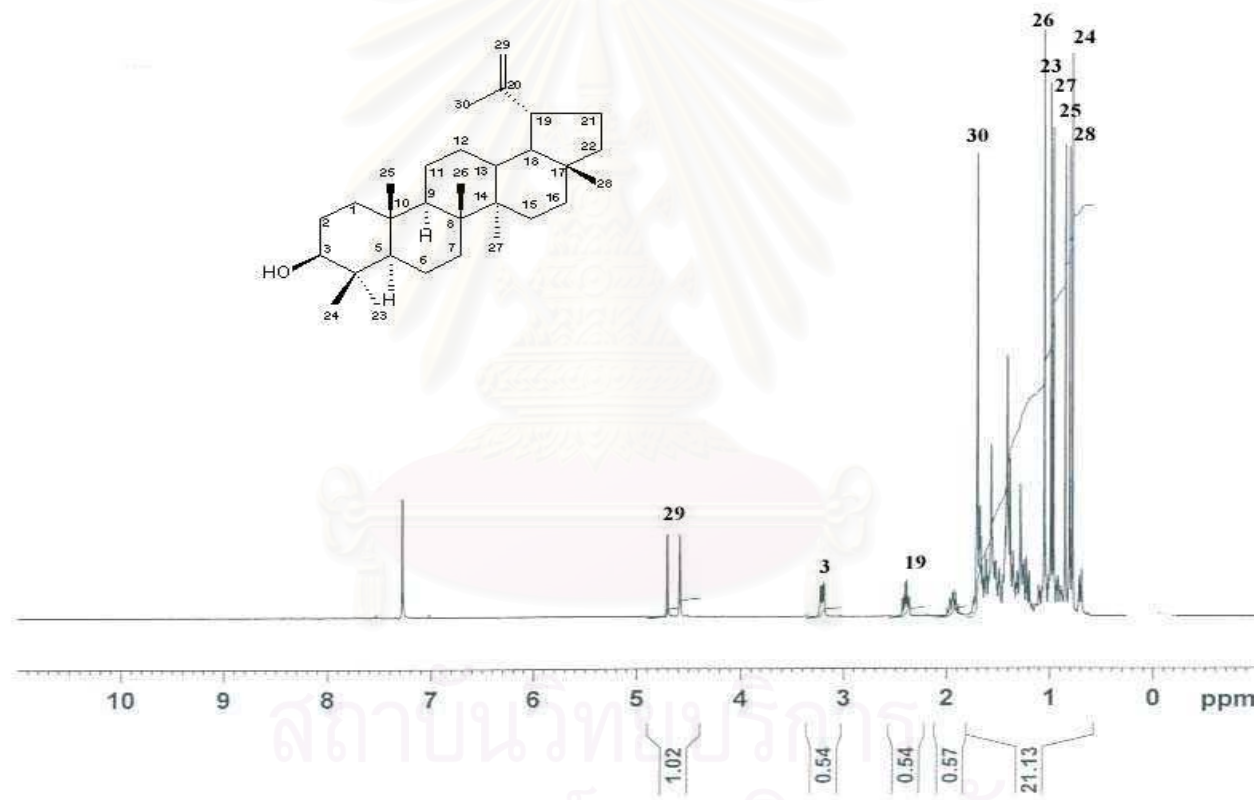


Figure 6a. The 400 MHz ¹H-NMR spectrum of compound DR 1

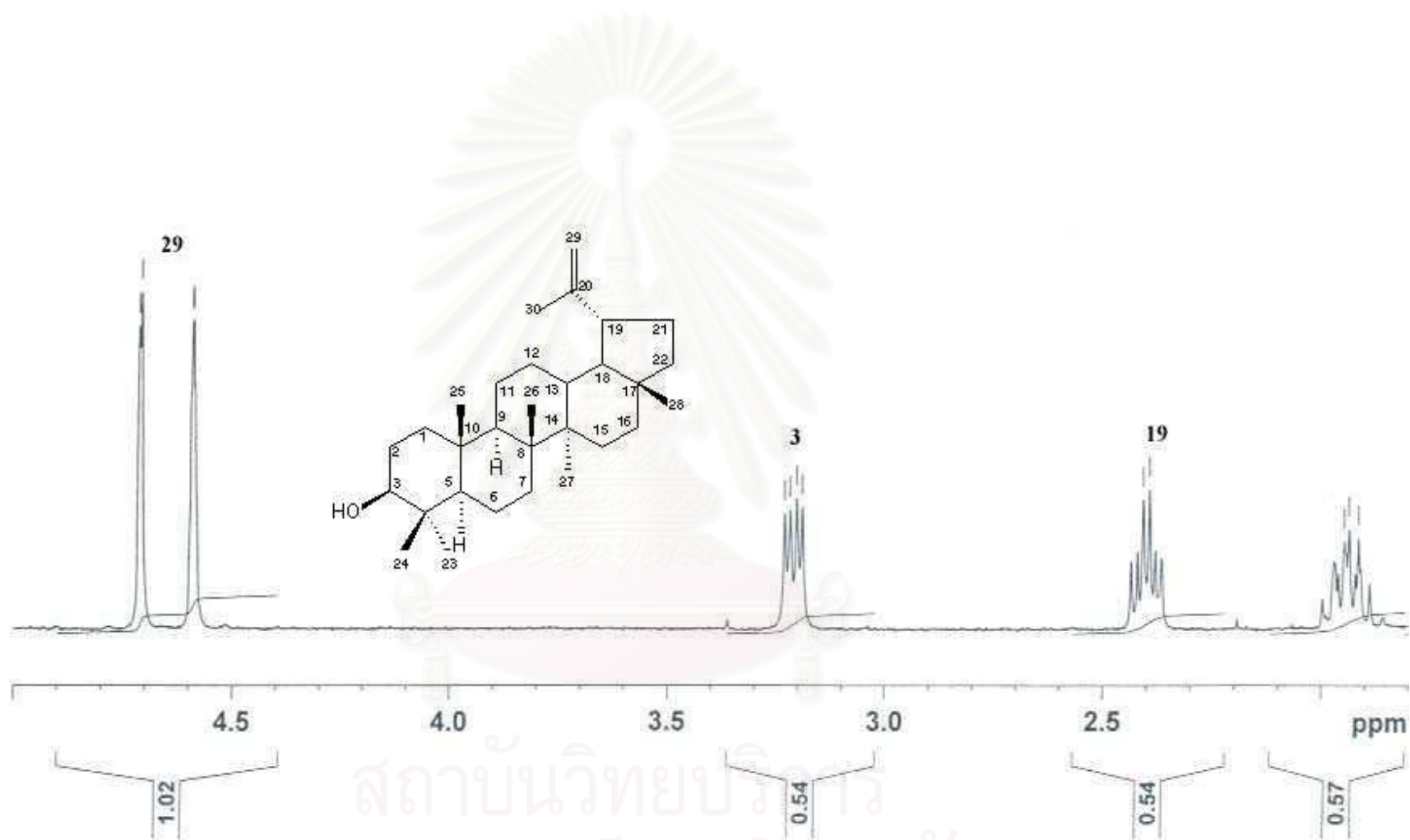


Figure 6b. The 400 MHz $^1\text{H-NMR}$ spectrum of compound DR 1 (expanded)

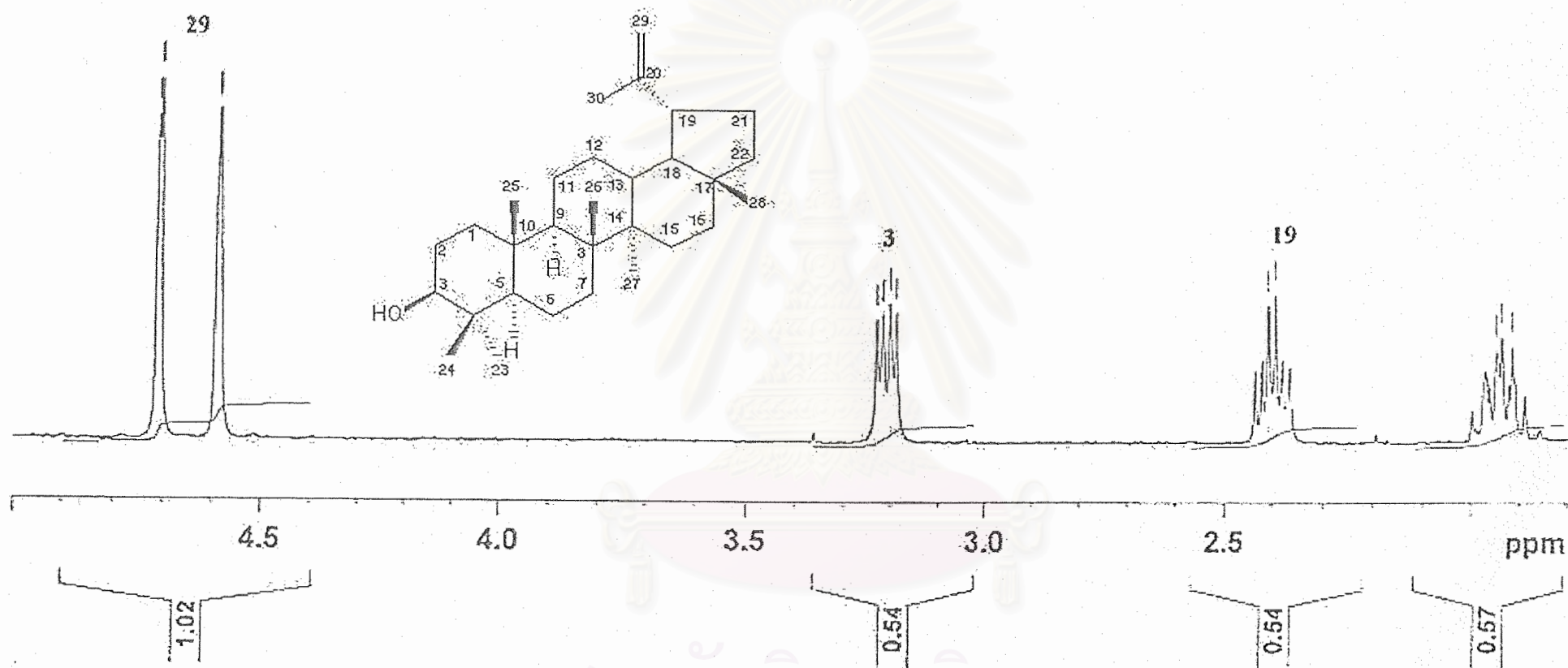


Figure 6c. The 400 MHz ¹H-NMR spectrum of compound DR1 (expanded)

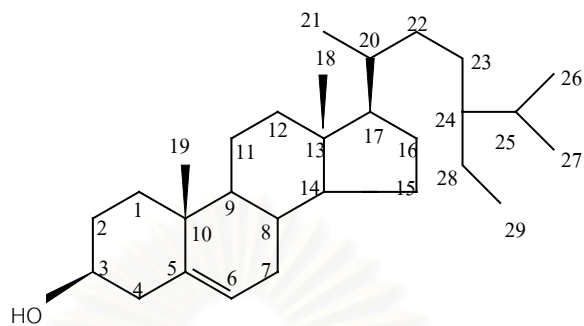
2. Identification of compound DR 2

Compound DR 2 was obtained as colorless needles (74 mg, 0.0005 % yield) from fraction DRH 16 of the hexane extract. The compound gave blue color to Lieberman-Burchard reagent, suggesting the presence of the steroid nucleus. The EIMS spectrum of this compound (Figure 7) showed a molecular ion peak at m/z 414, which corresponded to molecular formula of $C_{29}H_{50}O$. The IR spectrum displayed the OH absorption at 3436 cm^{-1} (Figure 8).

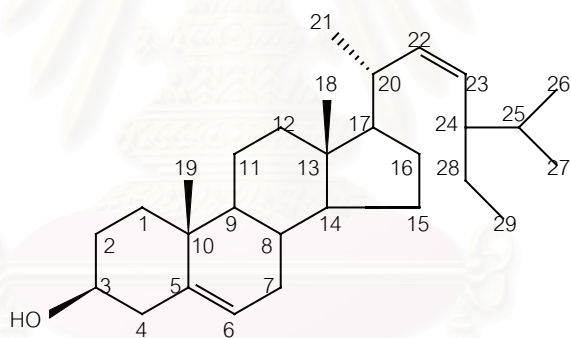
The $^1\text{H-NMR}$ spectrum (Figure 9a-9b) gave evidences which suggested that DR 2 is a mixture of β -sitosterol and stigmasterol. Two double doublets at δ 5.01 ($J = 15.2, 8.8\text{ Hz}$) and 5.14 ($J = 15.2, 8.8\text{ Hz}$) ppm were attributable to H-22 and H-23 of stigmasterol, respectively, while a doublet ($J = 5.2\text{ Hz}$) at δ 5.36 ppm was assignable to H-6 of both β -sitosterol and stigmasterol. A multiplet at δ 3.54 ppm was attributable to the methine proton of hydroxy-substituted position 3. The ratio of β -sitosterol and stigmasterol in the mixture, as deduced from the integration of peak areas, is 3 : 2.

In the $^{13}\text{C-NMR}$ spectrum (Figures 10a-10b), 29 carbon signals of β -sitosterol were evident, while the signals of stigmasterol were hardly observed. However, the signals for C-22 and C-23 of stigmasterol could be observed at δ 137.8 and 129.1 ppm, respectively, in the DEPT-90 NMR spectrum (Figure 11).

Therefore, it was concluded that DR 2 is a mixture of β -sitosterol and stigmasterol, both of which are common phytosterols widely distributed in the plant kingdom. Comparison of $^{13}\text{C-NMR}$ data of DR 2 with the reported data of β -sitosterol (Rubinstein *et al.*, 1976) and stigmasterol (Rubinstein *et al.*, 1976) is shown in Table 16. The structures of β -sitosterol and stigmasterol are shown below.



β - Sitosterol



Stigmasterol

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Table 16. Comparison of ^{13}C -NMR data of β -sitosterol (in CDCl_3), stigmasterol (in CDCl_3) and compound DR 2 (in CDCl_3).

Carbon	Chemical shift (δ) ppm		
	β -sitosterol	Stigmasterol	DR 2
1	37.2	37.2	37.3
2	31.7	31.7	31.7
3	71.7	71.8	71.8
4	42.3	42.4	42.3
5	140.8	140.8	140.8
6	121.7	121.7	121.7
7	31.9	32.0	31.9
8	31.9	32.0	31.9
9	50.1	50.2	50.2
10	36.5	36.6	36.5
11	21.1	21.1	21.1
12	39.8	39.7	39.8
13	42.3	42.4	42.3
14	56.8	56.9	56.8
15	24.3	24.4	24.3
16	28.2	29.0	28.3
17	56.0	56.1	56.1
18	11.9	12.1	12.0
19	19.4	19.4	19.4
20	36.1	40.5	36.2
21	18.8	21.1	18.8
22	33.9	138.0	34.0 ^a (137.8) ^b
23	26.1	129.3	26.1 ^a (129.1) ^b
24	45.8	51.3	45.9
25	29.1	32.0	29.2
26	19.8	21.3	19.8
27	19.0	19.0	19.0
28	23.1	25.4	23.1
29	11.9	12.3	11.9

a - the signal of β -sitosterol

b - the signal of stigmasterol

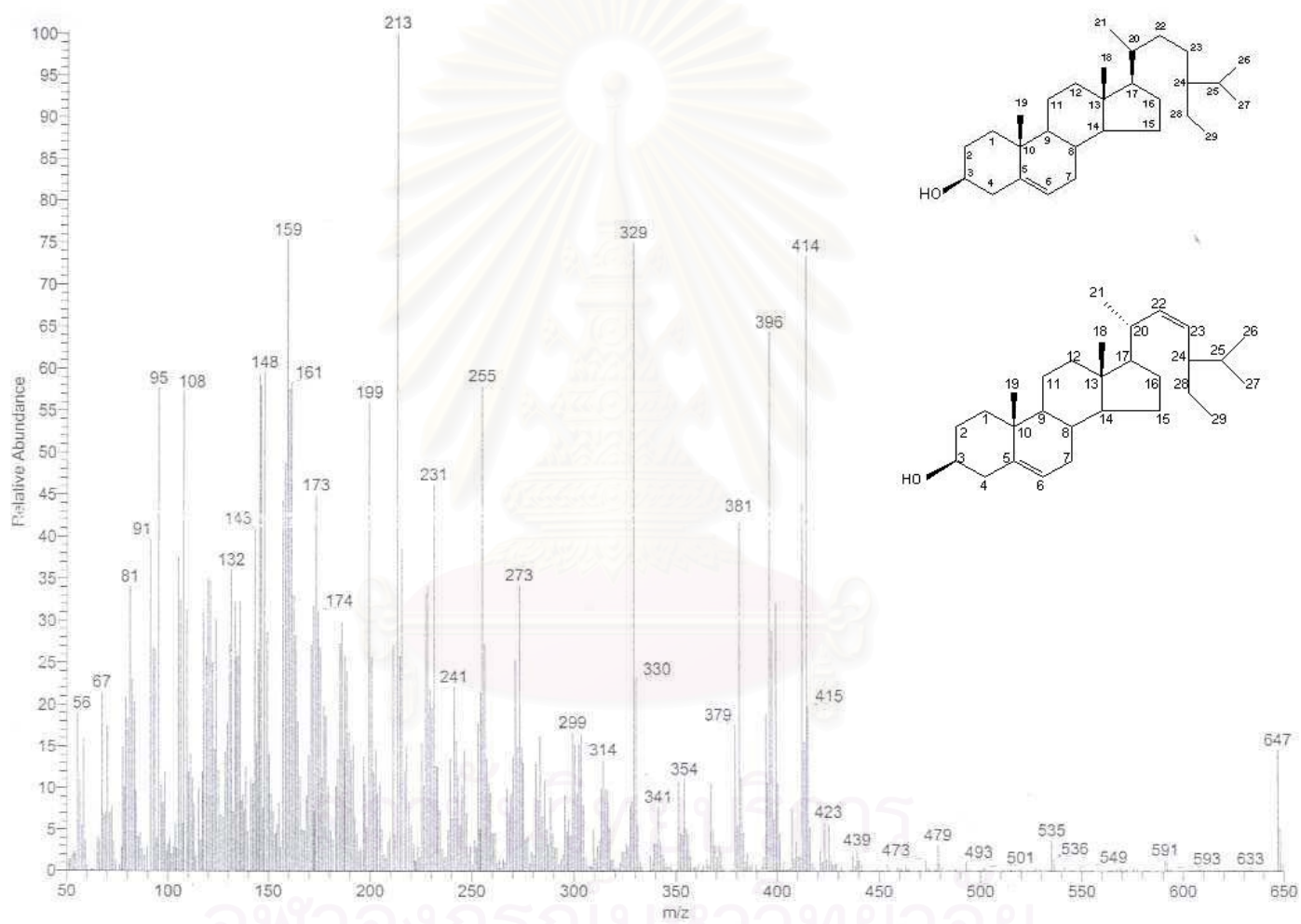


Figure 7. EIMS of compound DR 2

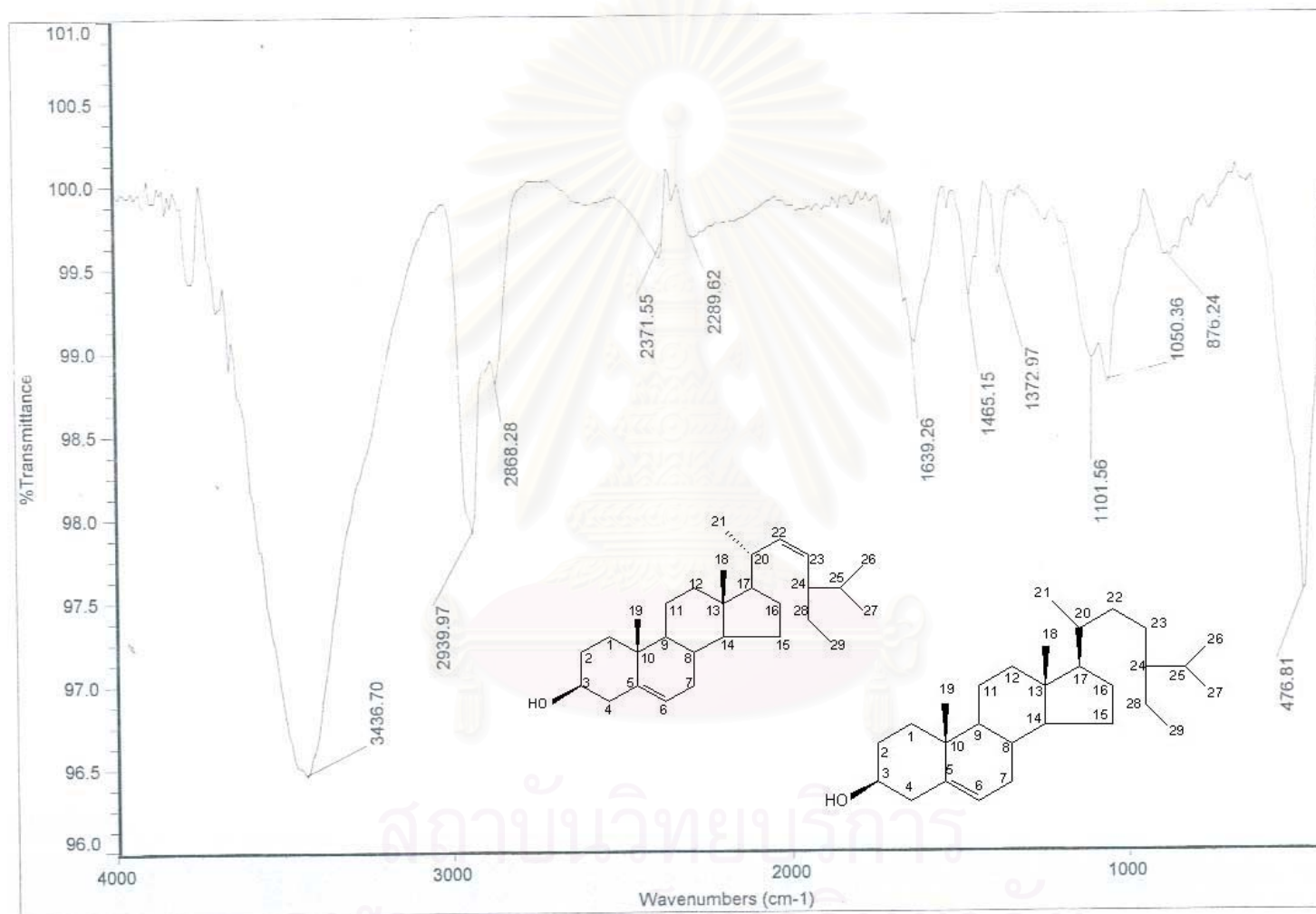


Figure 8. IR spectrum of compound DR 2

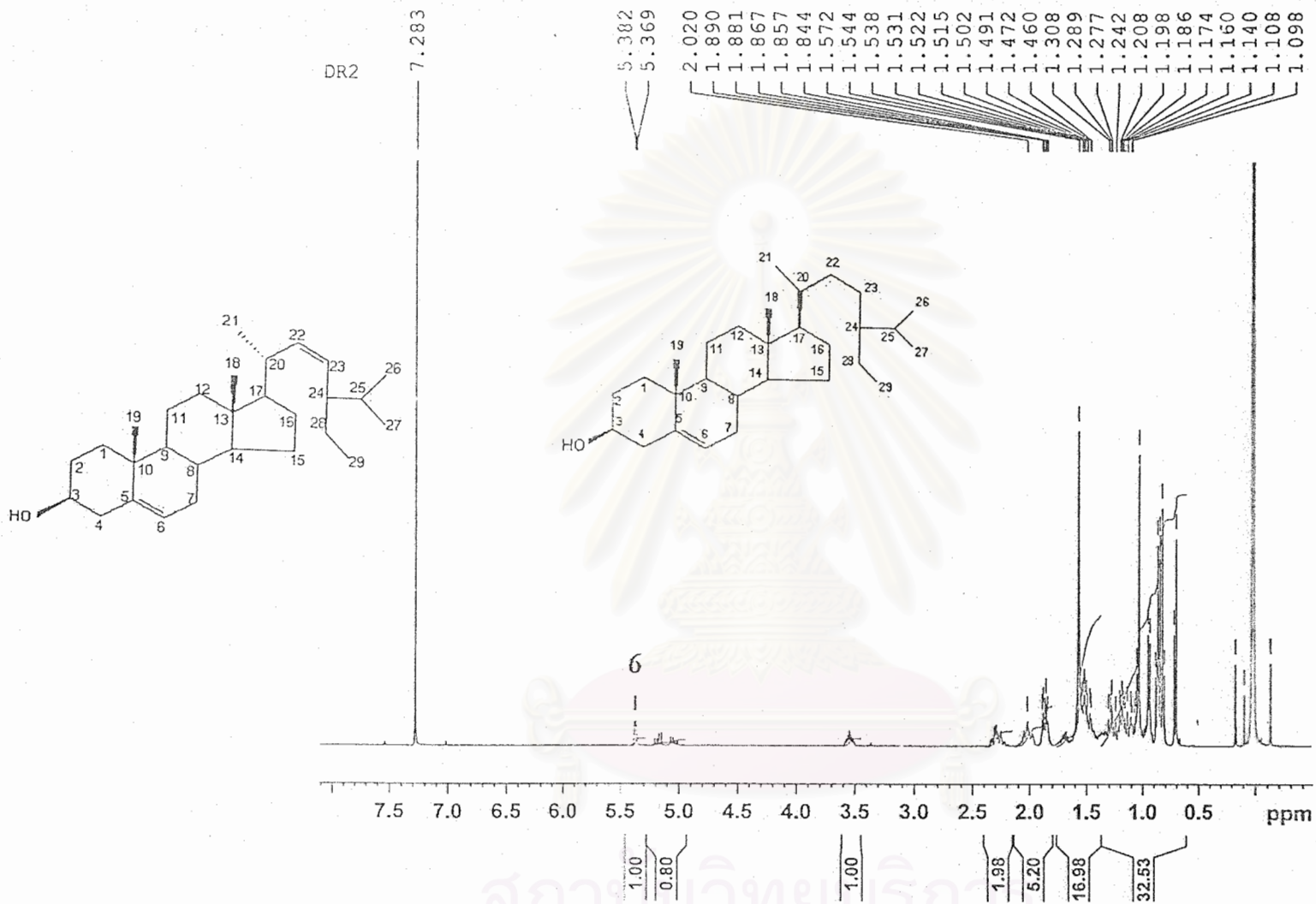


Figure 9a. The 400 MHz $^1\text{H-NMR}$ spectrum of compound DR2

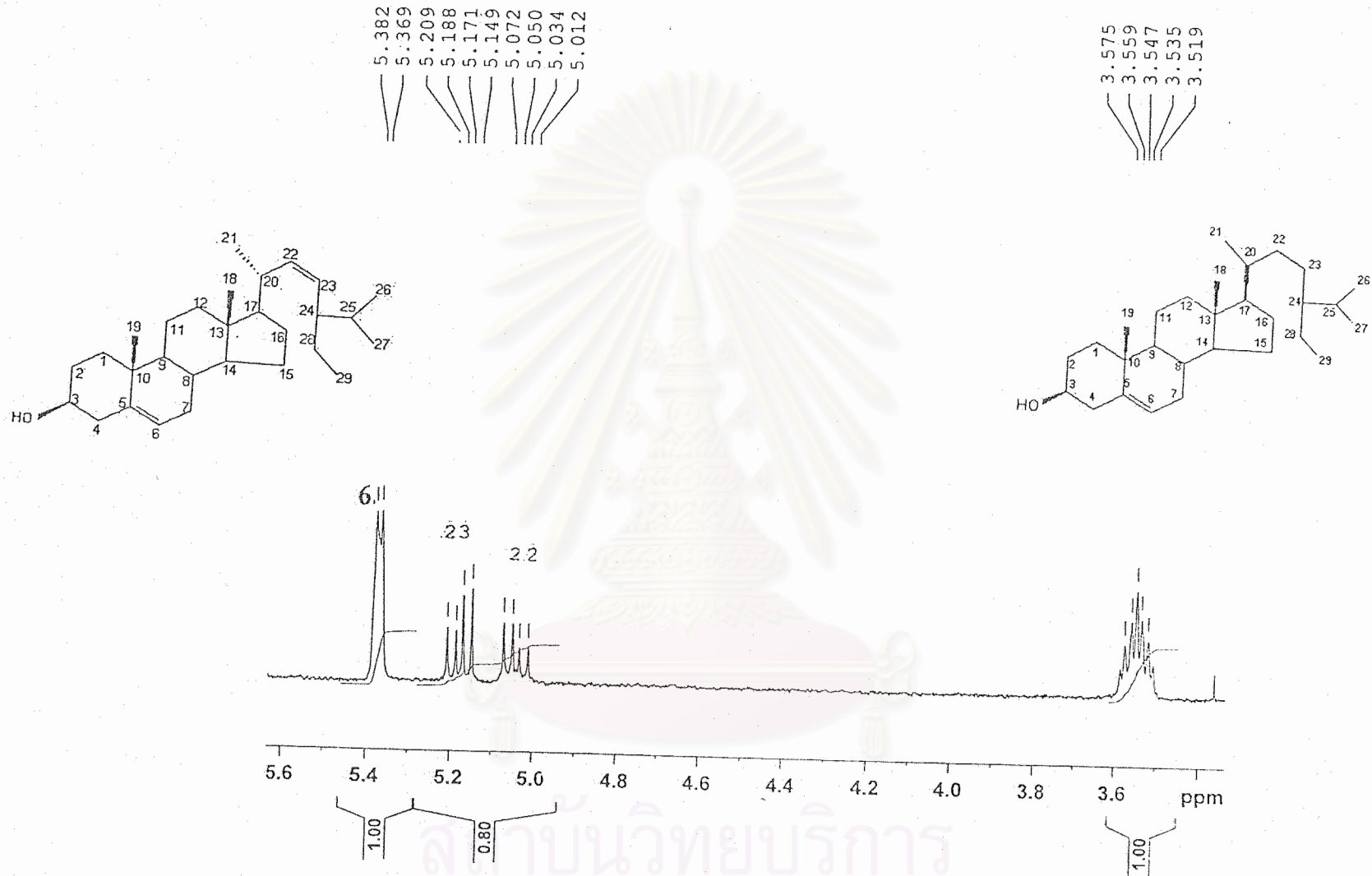


Figure 9b. The 400 MHz ¹H-NMR spectrum of compound DR 2 (expanded)

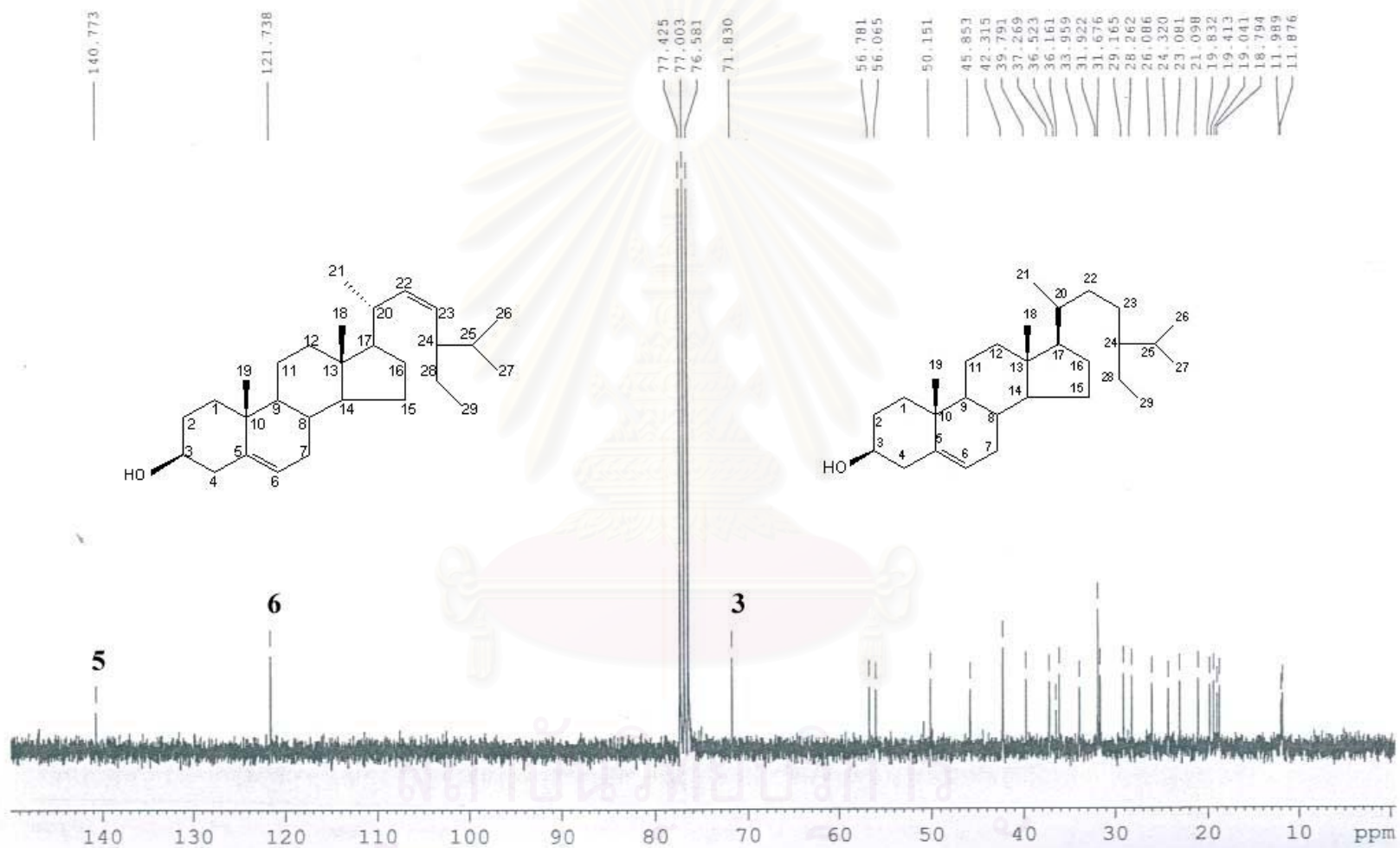


Figure 10a. The 100 MHz ^{13}C -NMR spectrum of compound DR 2

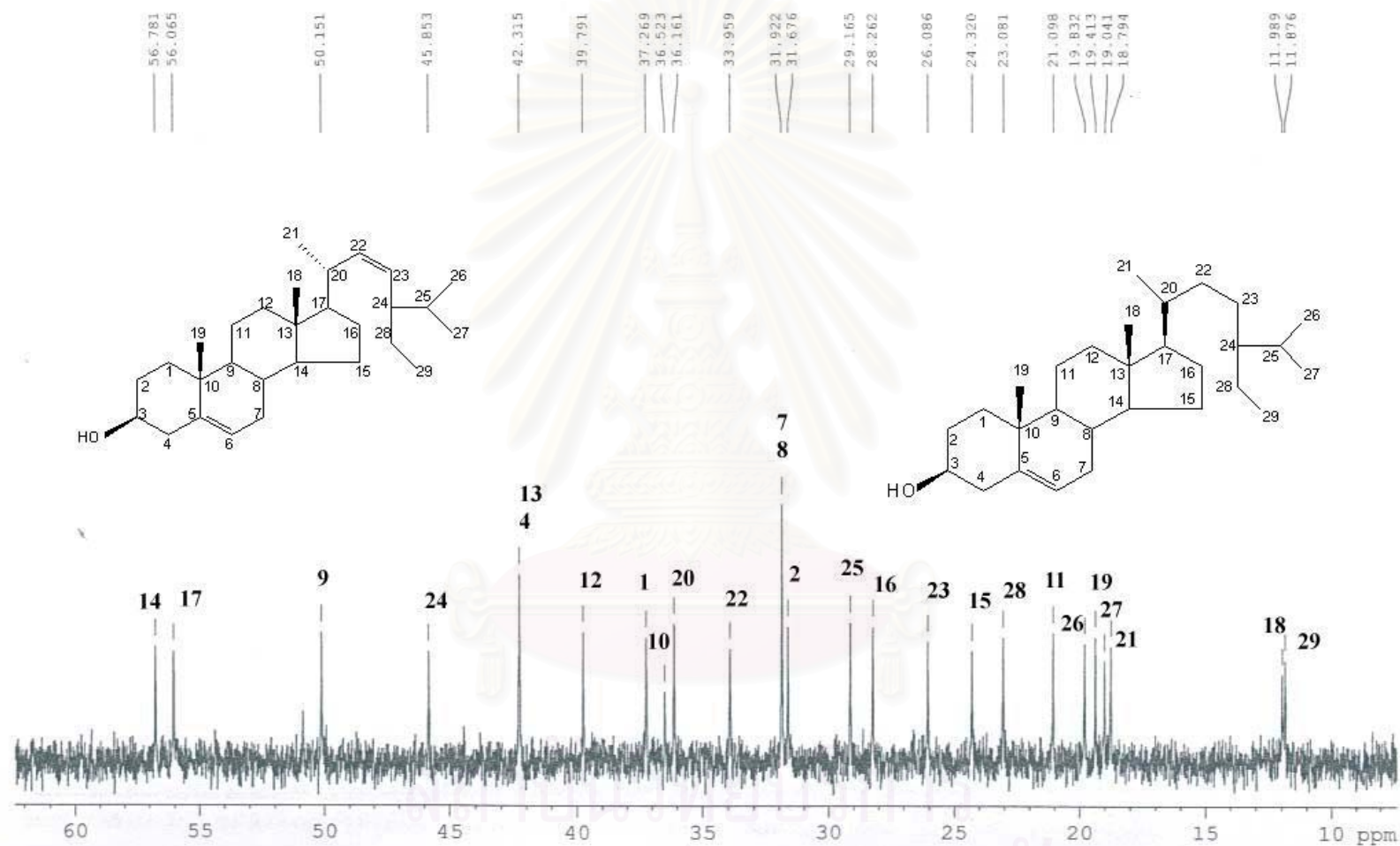
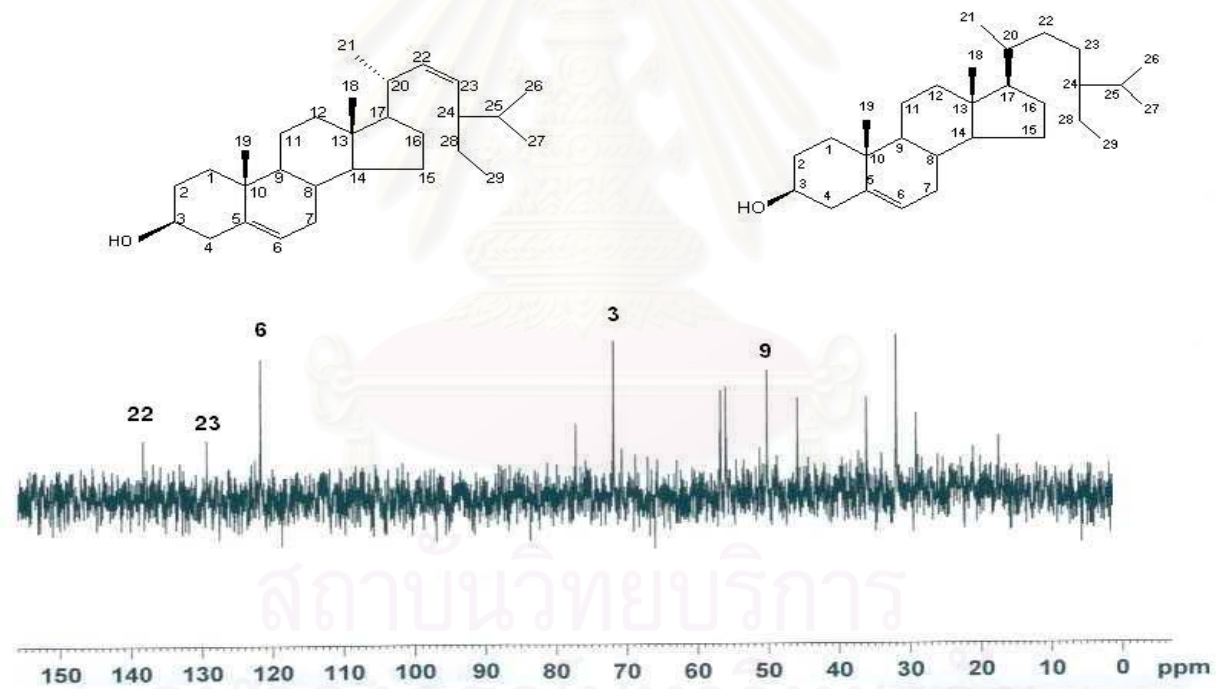


Figure 10b. The 100 MHz ^{13}C -NMR spectrum of compound DR 2 (expanded)

DEPT 90



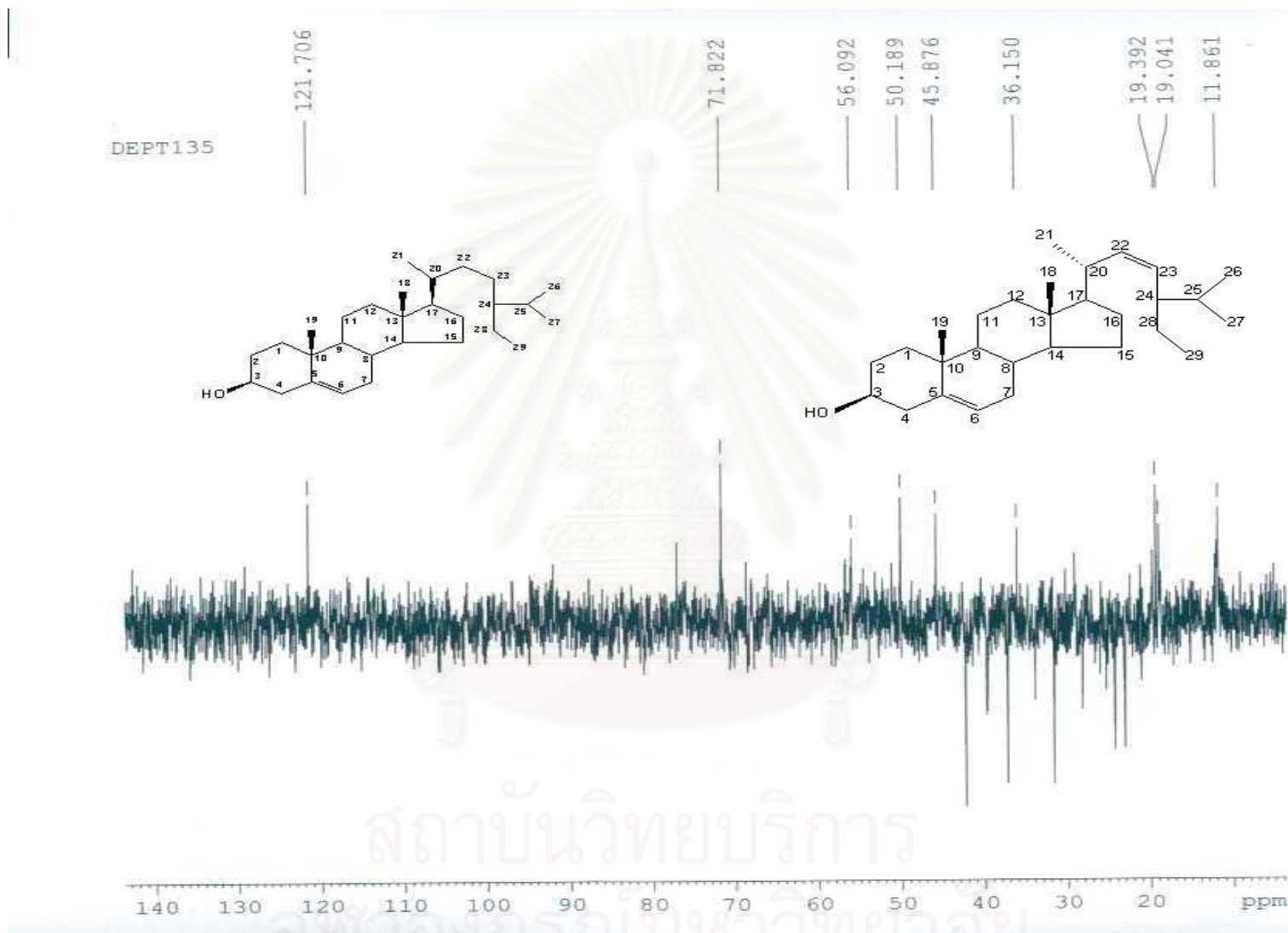


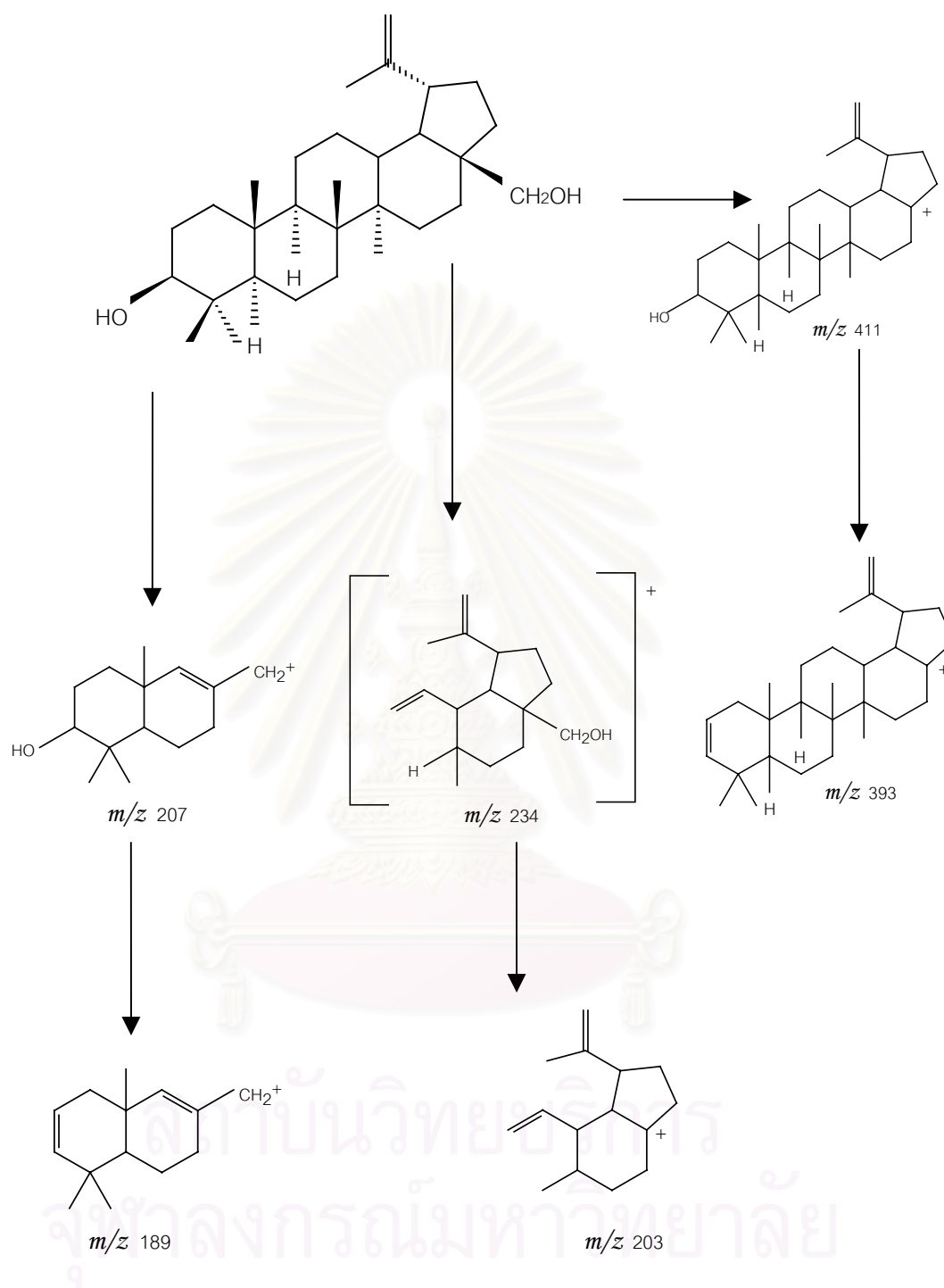
Figure 12. The 100 MHz ^{13}C -DEPT 135 NMR spectra of compound DR 2

3. Identification of compound DR 3

Compound DR 3 was obtained as white amorphous powder from fraction DRH 30 of the hexane extract (9 mg, 0.0006% yield). The compound gave red-violet color to Liebermann-Burchard reagent, suggesting its triterpenoid nature. The EIMS spectrum of this compound (Figure 13) showed a molecular ion peak at m/z 442, corresponding to the molecular formula $C_{30}H_{50}O_2$. Mass fragment peaks at m/z 411 (M^+-CH_2OH) and 393 ($M^+-CH_2OH-H_2O$) suggested the presence of a primary alcoholic group and a hydroxyl group, respectively. An intense fragment peak at m/z 189 was suggestive of a lupane-type triterpenoid (Ogunkoya, 1981). This peak, as well as other peaks at m/z 203, 207 and 234, was the results of cleavage across the C-ring of the lupane skeleton (Scheme 5). The presence of a hydroxyl group in the molecule was also confirmed by an absorption band at 3388 cm^{-1} in the IR spectrum (Figure 14).

The $^{13}\text{C-NMR}$ spectrum (Figure 15a – 15b) showed the signals of 30 carbon atoms, supportive of a triterpenoid structure. The DEPT-90 and DEPT-135 experiments (Figure 16) helped in classifying the signals into those of six quaternary carbons at δ 37.0, 39.2, 40.8, 42.8, 48.6 and 150.9 ppm, six methine carbons at δ 37.6, 48.2, 48.8, 51.4, 56.4 and 78.7 ppm, twelve methylene carbons at δ 18.7, 21.2, 25.7, 26.6, 27.5, 29.6, 30.2, 34.4, 34.7, 38.3, 61.0 and 110.1 ppm, and six methyl carbons at δ 14.6, 15.8, 16.4, 17.1, 20.0 and 27.8 ppm.

The $^1\text{H-NMR}$ spectrum of DR 3 (Figure 17a-17b) displayed six methyl singlets at δ 0.73 (H_3 -24), 0.80 (H_3 -25), 0.94 (H_3 -27), 0.95 (H_3 -23), 1.00 (H_3 -26) and 1.65 ppm (H_3 -30). The presence of exomethylene protons could be observed as two downfield singlets (br) at δ 4.55 and 4.65 ppm (H_2 -29), while a pair of doublets at δ 3.29 ($J=10.8$) and 3.75 ($J=10.8$) ppm could be attributed to hydroxy methylene protons (H_2 -28). Another one proton double doublet at δ 3.14 ($J=4.8$) ppm could be assigned to the carbinylic proton (H -3).



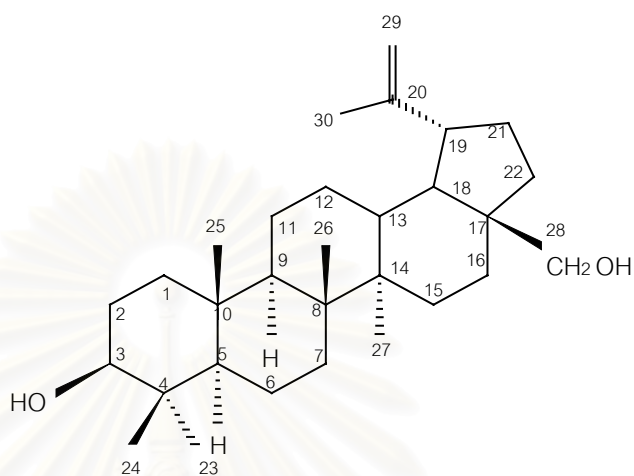
Scheme 5. Mass fragmentation of compound DR 3

All spectroscopic data of DR 3 are in accordance with betulin, a known triterpenoid of the lupane type. Comparison of the ^{13}C -NMR data of this compound with those previously reported for betulin (Tinto *et al.*, 1992) is shown in Table 17.

Table 17. Comparison of ^{13}C -NMR data of betulin (in CDCl_3) and compound DR 3 (in CDCl_3).

Carbon	Chemical shift (δ) ppm		Carbon	Chemical shift (δ) ppm	
	Betulin	DR 3		Betulin	DR 3
1	38.8	38.4	16	29.2	29.6
2	27.2	27.5	17	47.8	48.6
3	78.9	78.7	18	48.8	48.8
4	38.9	39.2	19	47.8	48.2
5	55.3	56.4	20	150.6	150.9
6	18.3	18.7	21	29.8	30.2
7	34.3	34.7	22	34.0	34.4
8	40.9	40.8	23	28.0	27.8
9	50.4	51.4	24	15.4	15.8
10	37.2	37.0	25	16.1	17.1
11	20.9	21.2	26	16.0	16.4
12	25.3	25.7	27	14.8	14.6
13	37.3	37.6	28	60.2	61.0
14	42.7	42.8	29	109.6	110.1
15	27.0	26.6	30	19.1	20.0

Therefore, it was concluded that DR 3 is betulin, the structure of which is shown below.



Betulin ($C_{30}H_{50}O_2$)

Similar to lupeol, betulin has been isolated from several plants of the genus *Diospyros*. The compound has been used as antiseptic (Batta and Rangaswami, 1973). It was demonstrated as having inhibitory effect against Epstein – Barr virus activation (Konoshima *et al.*, 1987), *in vitro* antitumor activity against human epidermoid carcinoma of nasopharynx (Miles *et al.*, 1974) and Walker-256 tumor system (Sheth *et al.*, 1973).

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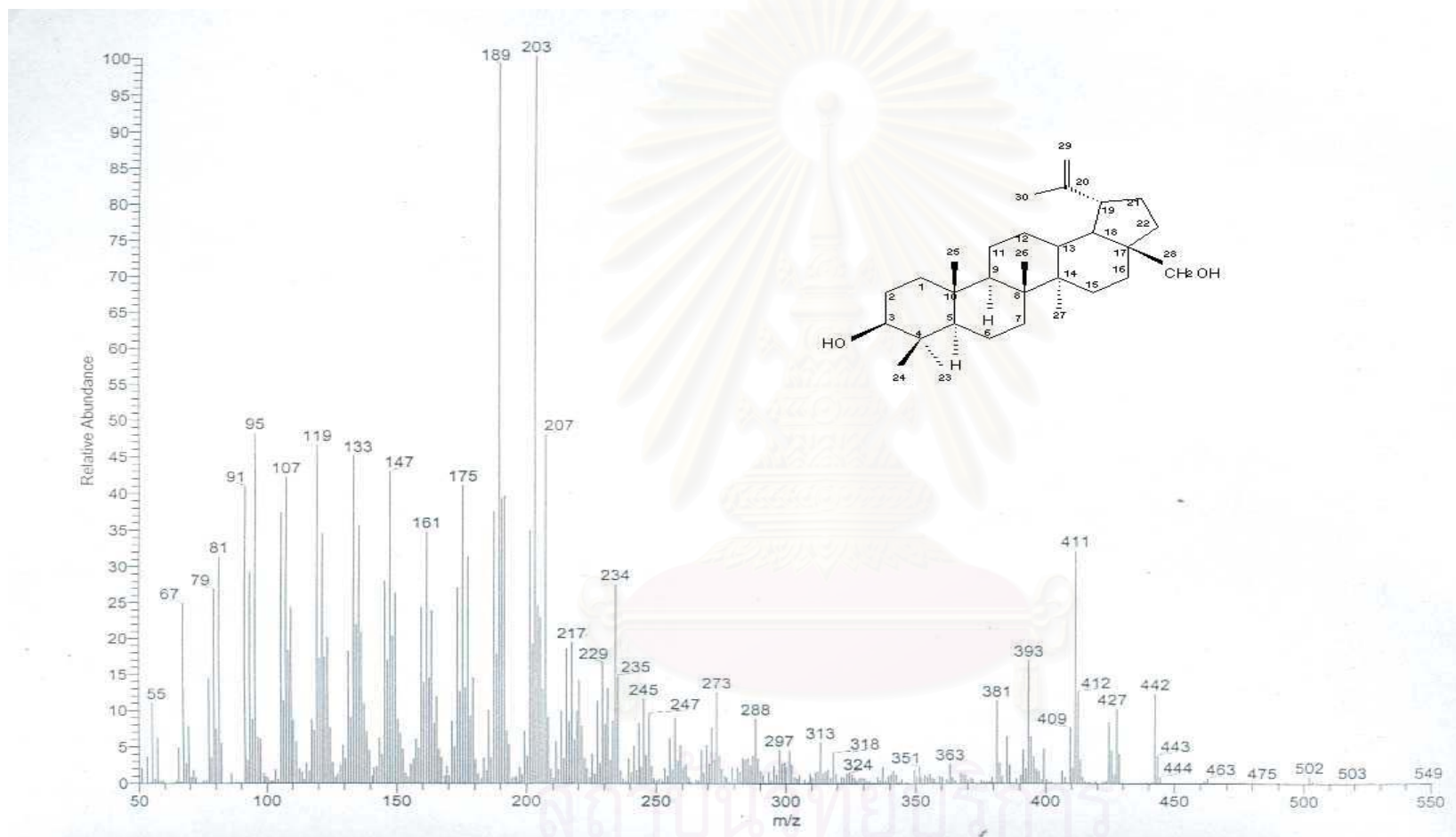


Figure 13. EIMS of compound DR 3

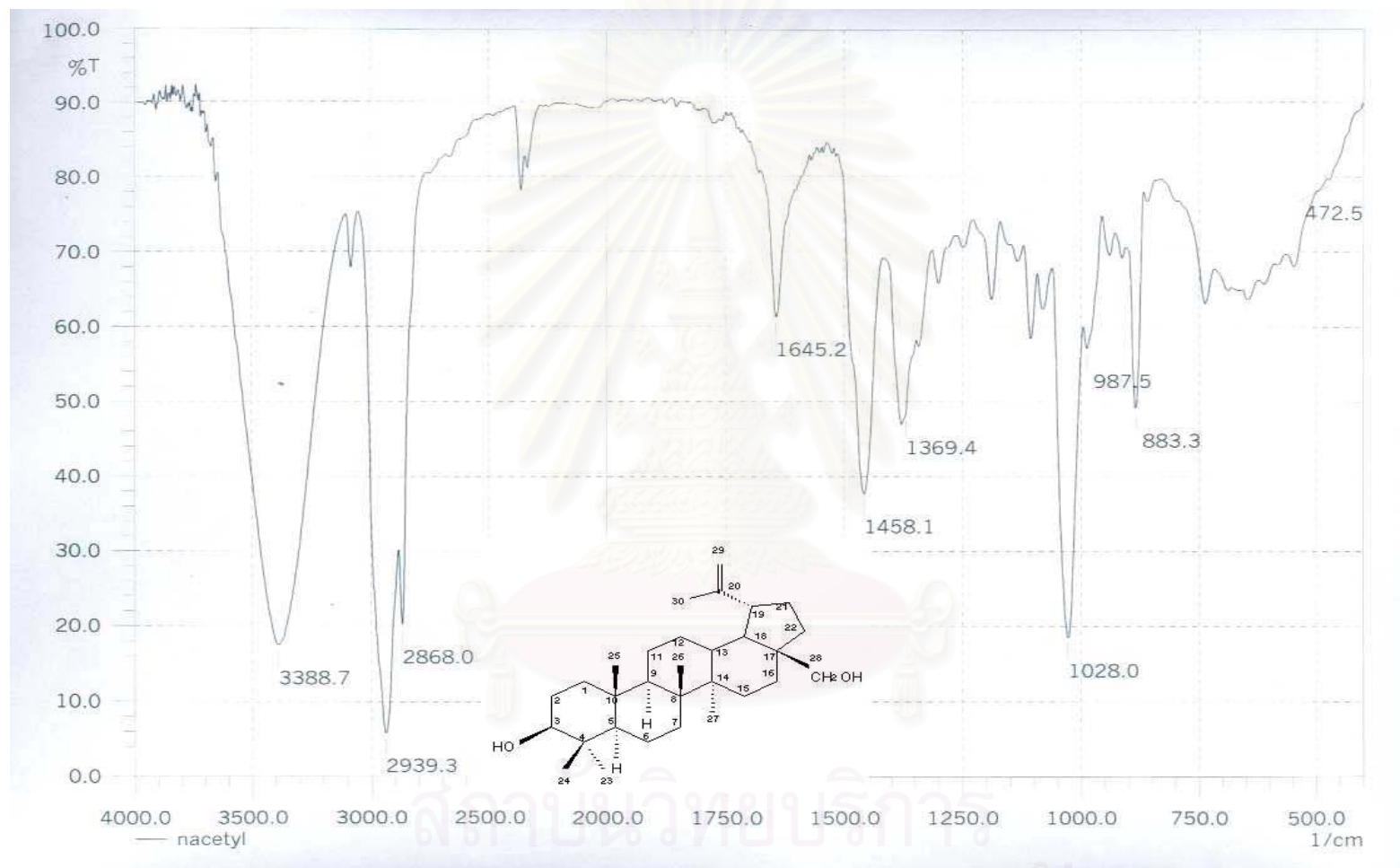


Figure 14. IR spectrum of compound DR 3

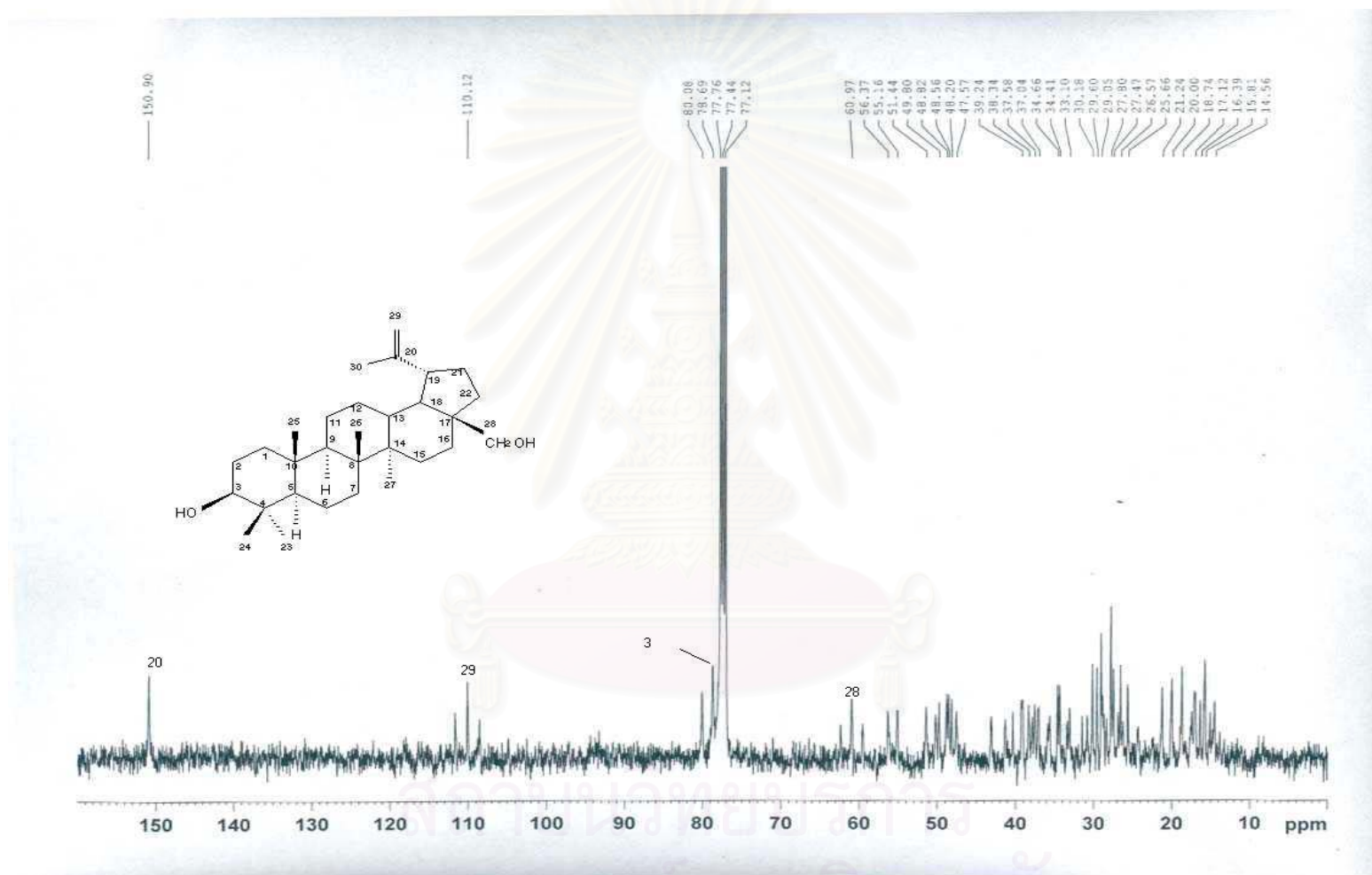


Figure 15a. The 100 MHz ^{13}C -NMR spectrum of compound DR 3 (in CDCl_3)

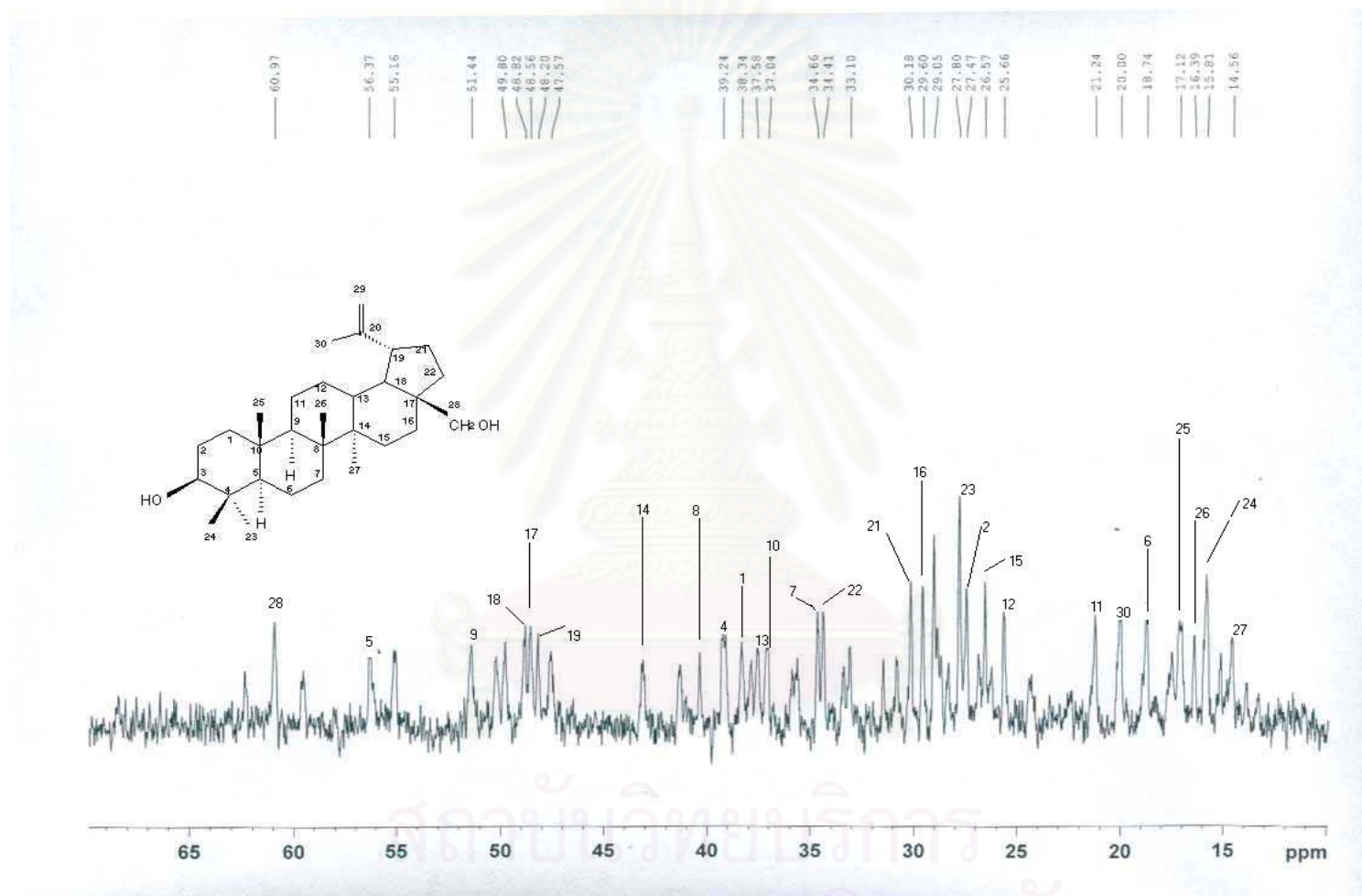


Figure 15b. The 100 MHz ^{13}C -NMR spectrum of compound DR 3 (in CDCl_3) (expanded)

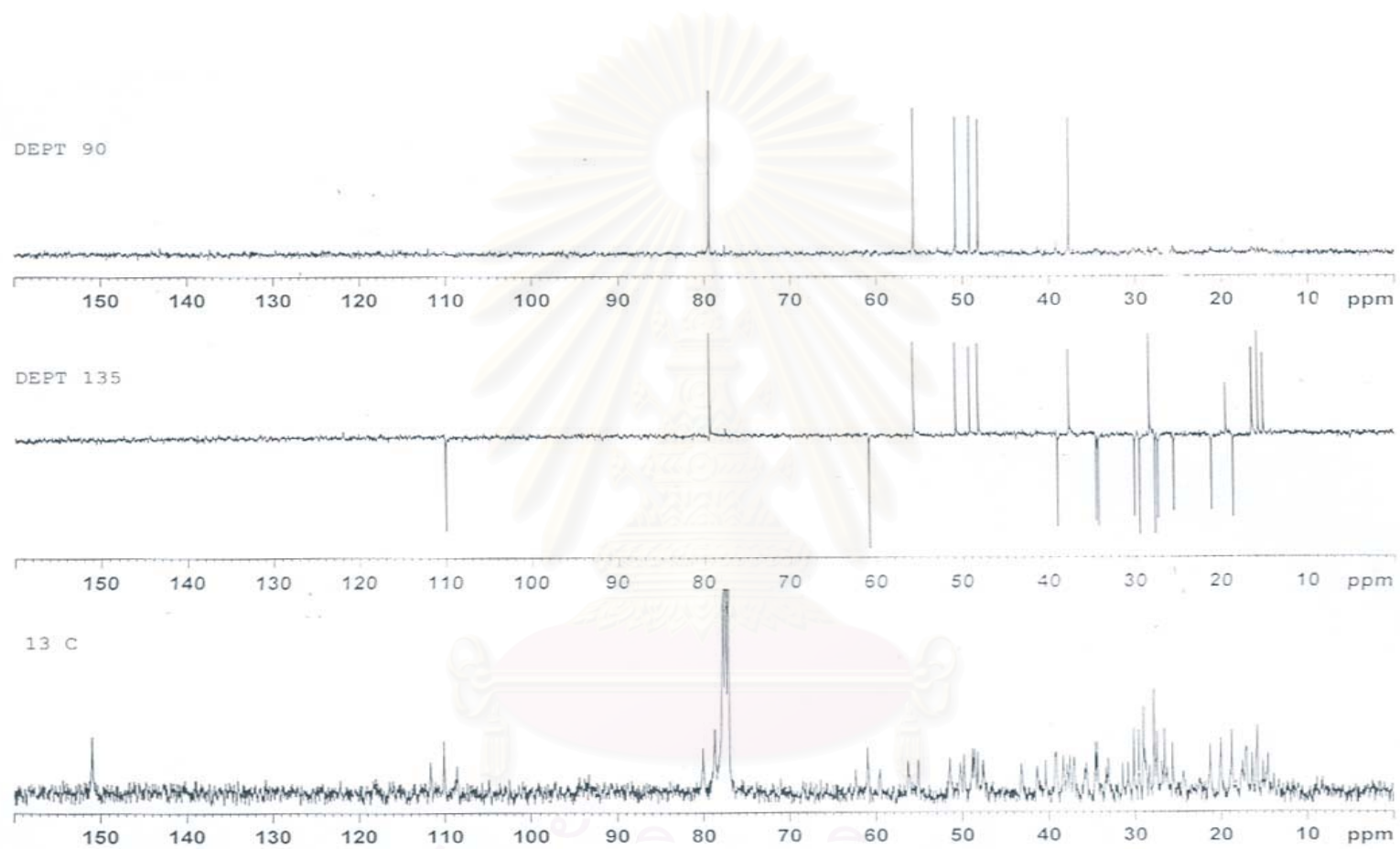


Figure 16. The 100 MHz ^{13}C -DEPT NMR spectra of compound DR 3 (in CDCl_3)

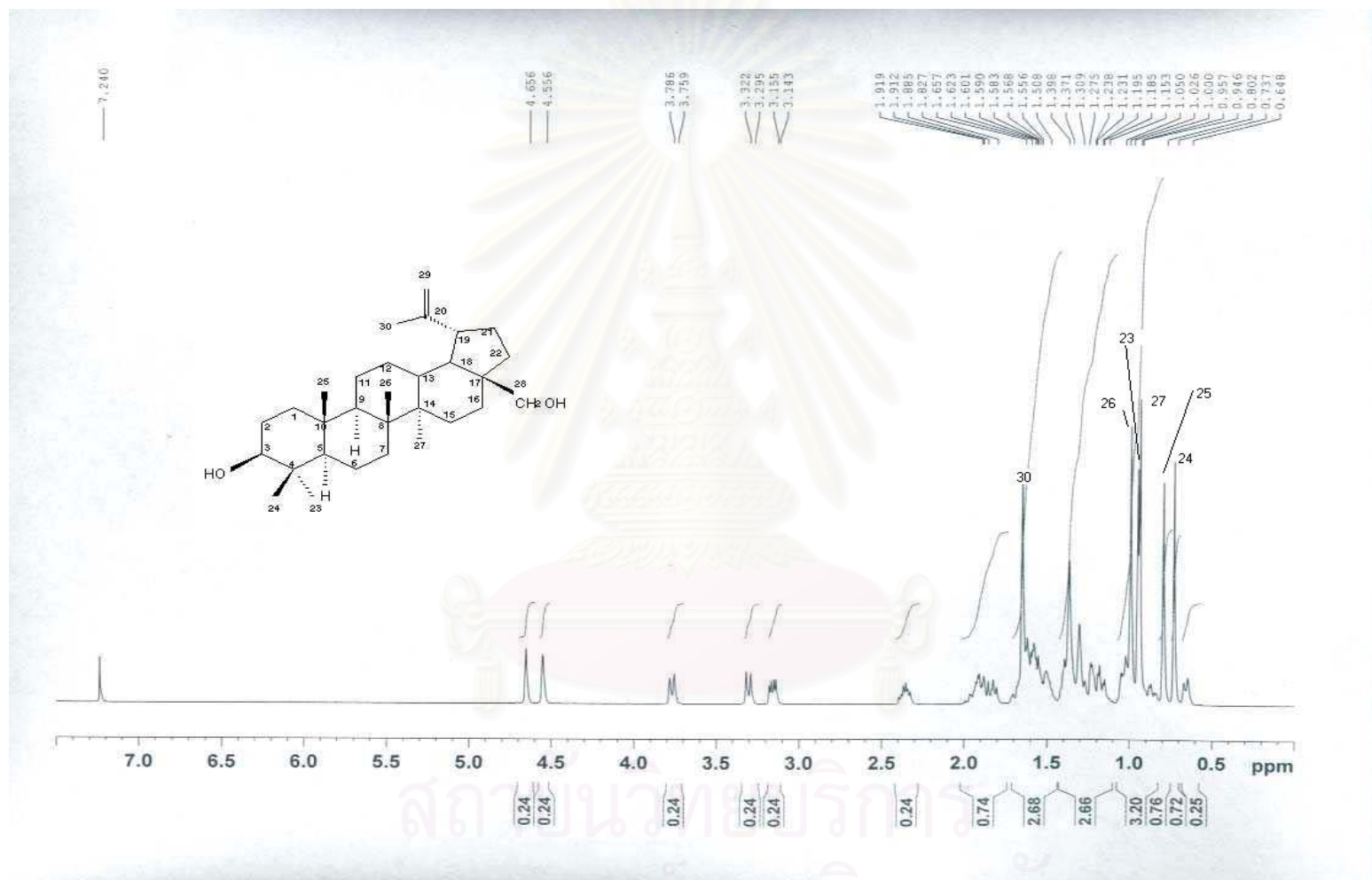


Figure 17a. The 400 MHz ^1H NMR spectrum of compound DR 3 (in CDCl_3)

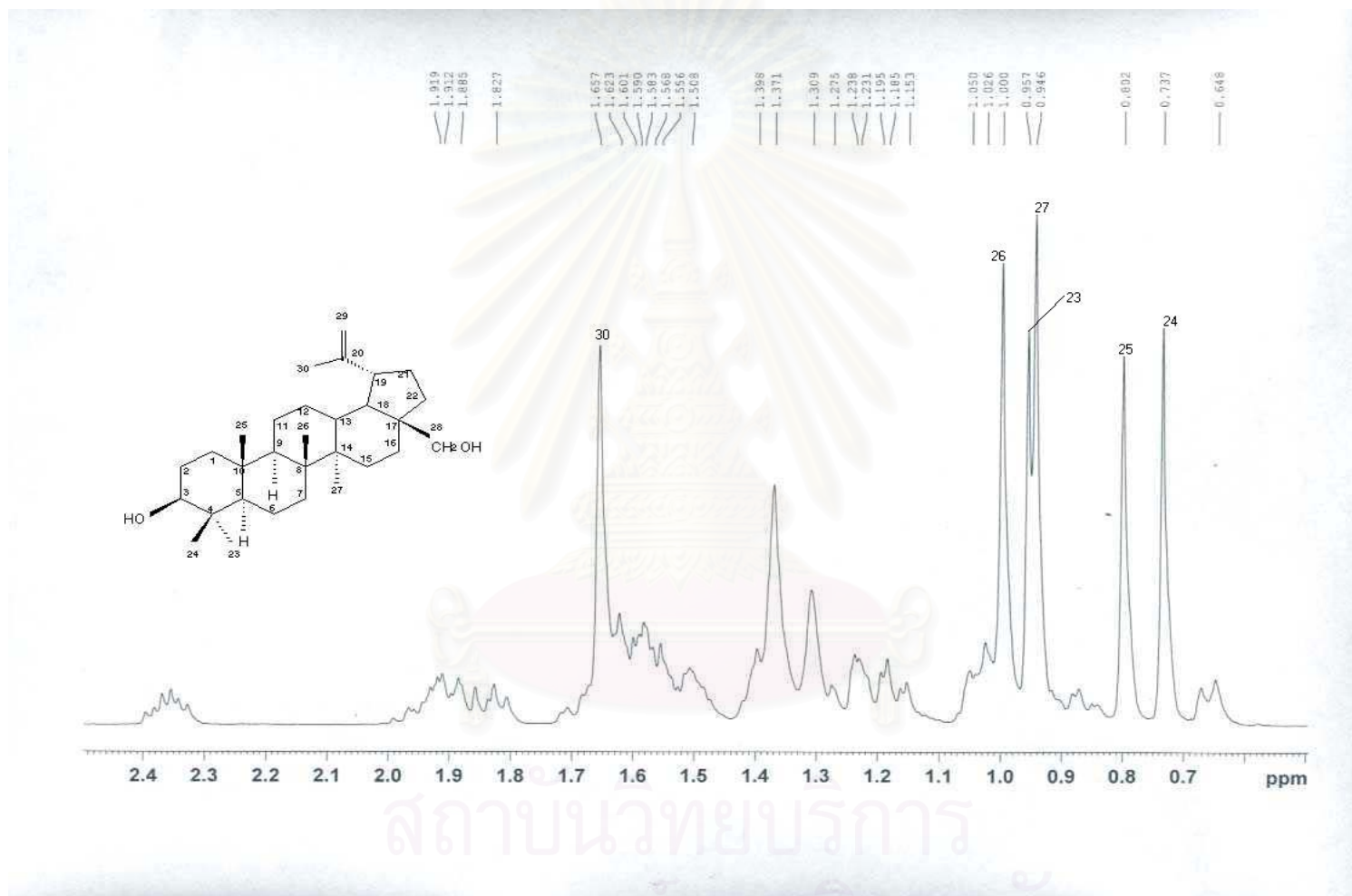


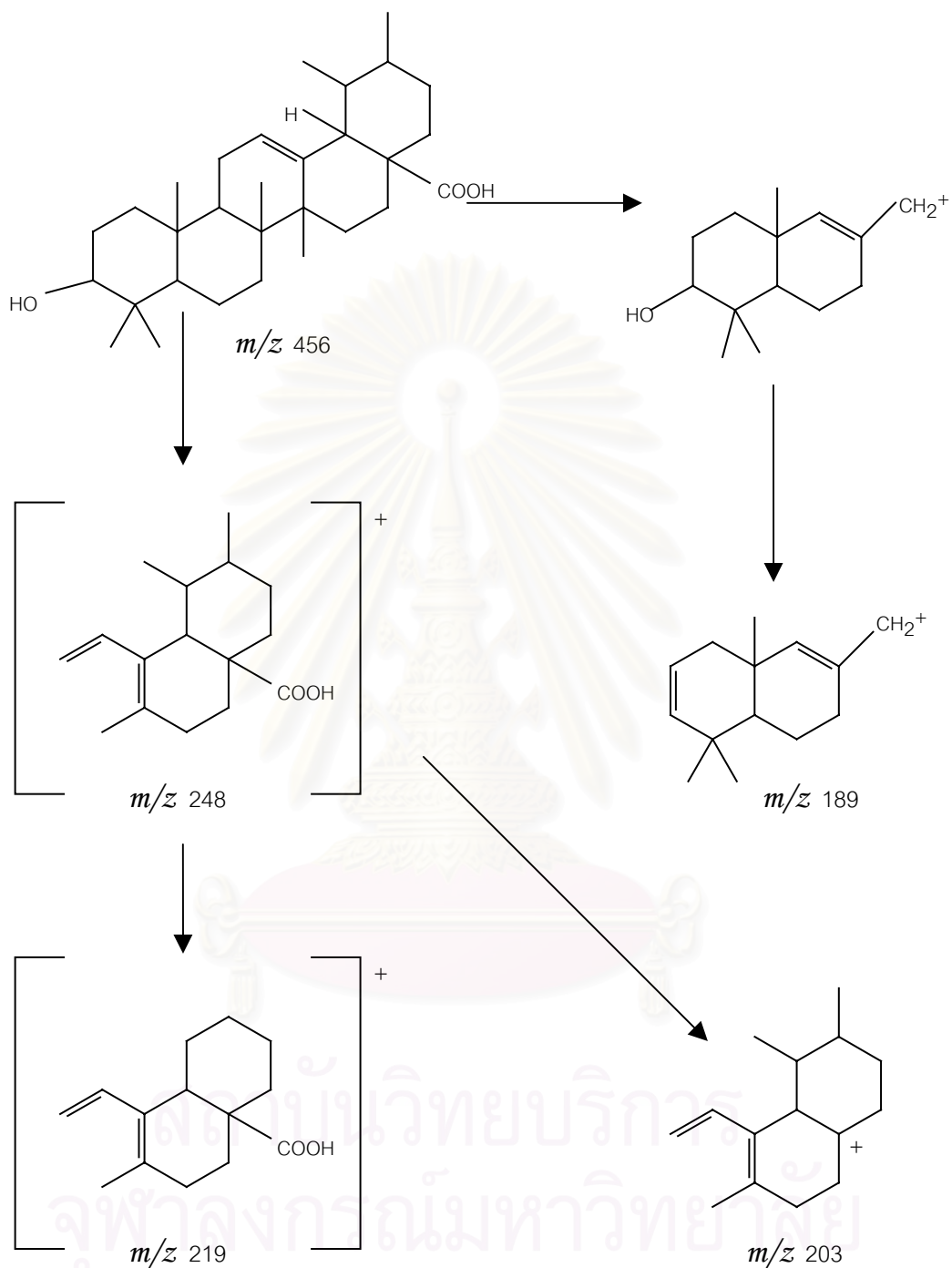
Figure 17b. The 400 MHz ^1H NMR spectrum of compound DR 3 (in CDCl_3) (expanded)

4. Identification of compound DR 4

Compound DR 4 was obtained as colorless needles (54 mg, 0.0004% yield) from fraction DRC 11 of the chloroform extract. The compound gave red-violet color to Liebermann-Burchard reagent, suggesting its triterpenoid nature. The base peak at m/z 248 in the EIMS (Figure 18), resulting from the cleavage through a retro-Diels-Alder reaction, is characteristic of a C-12 unsaturated triterpenoid with the oleanane or ursane skeleton containing carboxylic group in ring D or E (Ogunkoya, 1981). The further losses of the methyl group and the carboxylic group led to fragment peaks at m/z 219 and 203, respectively. A peak at m/z 189 was produced by the loss of water from the other product of the retro-Diels-Alder fragmentation (Scheme 6). The presence of the carboxylic functionality in the molecule was confirmed by an IR absorption band at 1689 cm^{-1} (Figure 19). A broad band at 3460 cm^{-1} indicated the presence of the hydroxyl group.

The ^{13}C -NMR spectrum of DR 4 (Figure 20a-20b) showed 30 carbon signals, supportive of a triterpenoid structure. The DEPT-90 and DEPT-135 (Figure 21a-21b) experiments were employed to classify these signals into those of seven quaternary carbons at δ 178.4 (C-28), 138.3 (C-13), 47.1 (C-17), 41.8 (C-14), 38.8 (C-8), 38.5 (C-4) and 36.6 ppm (C-10), seven methine carbons at δ 124.7 (C-12), 77.0 (C-3), 54.9 (C-5), 52.5 (C-18), 46.9 (C-9), 38.6 (C-19) and 38.5 ppm (C-20), nine methylene carbons at δ 38.3 (C-1), 36.4 (C-22), 32.8 (C-7), 30.3 (C-21), 27.6 (C-15), 27.1 (C-2), 23.9 (C-16), 23.0 (C-11) and 18.1 ppm (C-6), and seven methyl carbons at δ 28.4 (C-23), 23.4 (C-27), 21.2 (C-30), 17.1 (C-29), 17.0 (C-26), 16.2 (C-25) and 15.3 ppm (C-24). The most downfield carbon signal at δ 178.4 ppm represents the carbonyl carbon of the carboxylic group and two downfield carbon signals at δ 138.3 and 124.7 ppm represents the double bond between C-12 and C-13 of the compound.

The ^1H -NMR spectrum of DR 4 (Figure 22a-22b) showed the signals of methyl protons at δ 0.66 (H_3 -25, s), 0.73 (H_3 -26, s), 0.79 (H_3 -29, d $J=6.4$ Hz),



Scheme 6. Mass fragmentation of compound DR 4

0.85 (H₃-24, s), 0.88 (H₃-23, s), 0.88 (H₃-30, *d* *J*=7.2 Hz) and 1.02 ppm (H₃-27, s). The signal at δ 5.11 ppm could be assigned to the olefinic proton (H-12) and the signal at δ 2.98 ppm could be assigned to the methine proton of hydroxy – substituted position 3 (H-3).

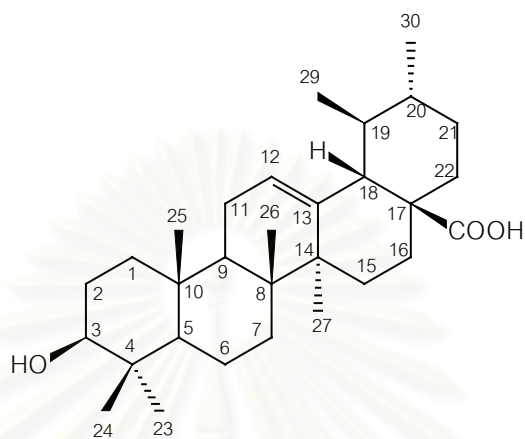
Comparison of ¹³C-NMR data of DR 4 with the reported data of ursolic acid (Lin *et al.*, 1987) and methyl ursolate (Zhang and Yu, 1990) suggested that DR 4 is ursolic acid. Almost all carbon assignments of DR 4 are in agreement with the reported data of ursolic acid except for the assignments of signals for C-11 and C-29 at δ 23.0 and 17.1 ppm (DR 4) instead of δ 17.1 and 23.2 ppm (ursolic acid), respectively. However, the assignments of C-11 and C-29 of DR 4 were found to be in agreement with the reported data of methyl ursolate (δ 23.3 and 16.9 ppm, respectively), and the DEPT experiment of DR 4 also indicated that the carbon signal at δ 17.1 ppm is due to a methyl carbon (C-29), while the signal at δ 23.0 ppm is due to a methylene carbon (C-11).

All spectroscopic data of DR 4 are in accordance with ursolic acid, a known triterpenoid of the ursane type. Comparison of the ¹³C-NMR data of this compound with the literature value of ursolic acid (Lin *et al.*, 1987) is shown in Table 18.

Table 18. Comparison of ^{13}C -NMR data of ursolic acid (in pyridine- d_5) and DR 4 (in DMSO- d_6).

Carbon	Chemical shift (δ) ppm		Carbon	Chemical shift (δ) ppm	
	Ursolic acid	DR 4		Ursolic acid	DR 4
1	38.7	38.3	16	24.2	23.9
2	27.2	27.1	17	47.5	47.1
3	78.2	77.0	18	52.7	52.5
4	38.8	38.5	19	39.1	38.6
5	55.2	54.9	20	38.8	38.5
6	18.3	18.1	21	30.7	30.3
7	33.0	32.8	22	36.7	36.4
8	39.5	38.8	23	28.0	28.4
9	47.5	46.9	24	15.7	15.3
10	36.9	36.6	25	15.4	16.2
11	17.1	23.0	26	17.0	17.0
12	125.2	124.7	27	23.5	23.4
13	138.3	138.3	28	179.9	178.4
14	42.0	41.8	29	23.2	17.1
15	28.2	27.6	30	21.2	21.2

Therefore, it was concluded that DR 4 is ursolic acid, the structure of which is shown below.



Ursolic acid ($C_{30}H_{48}O_3$)

Similar to lupeol and betulin, ursolic acid has been isolated from several plants of the genus *Diospyros*. The compound has been reported possessing anti-inflammatory activity (Recio *et al.*, 1995a ; Recio *et al.*, 1995b). It was also demonstrated as exhibiting inhibitory effect against Epstein-Barr virus activation (Ohigashi *et al.*, 1986), and inhibitory effect against HIV-1 protease (Singh, Singh and Bani, 1994).

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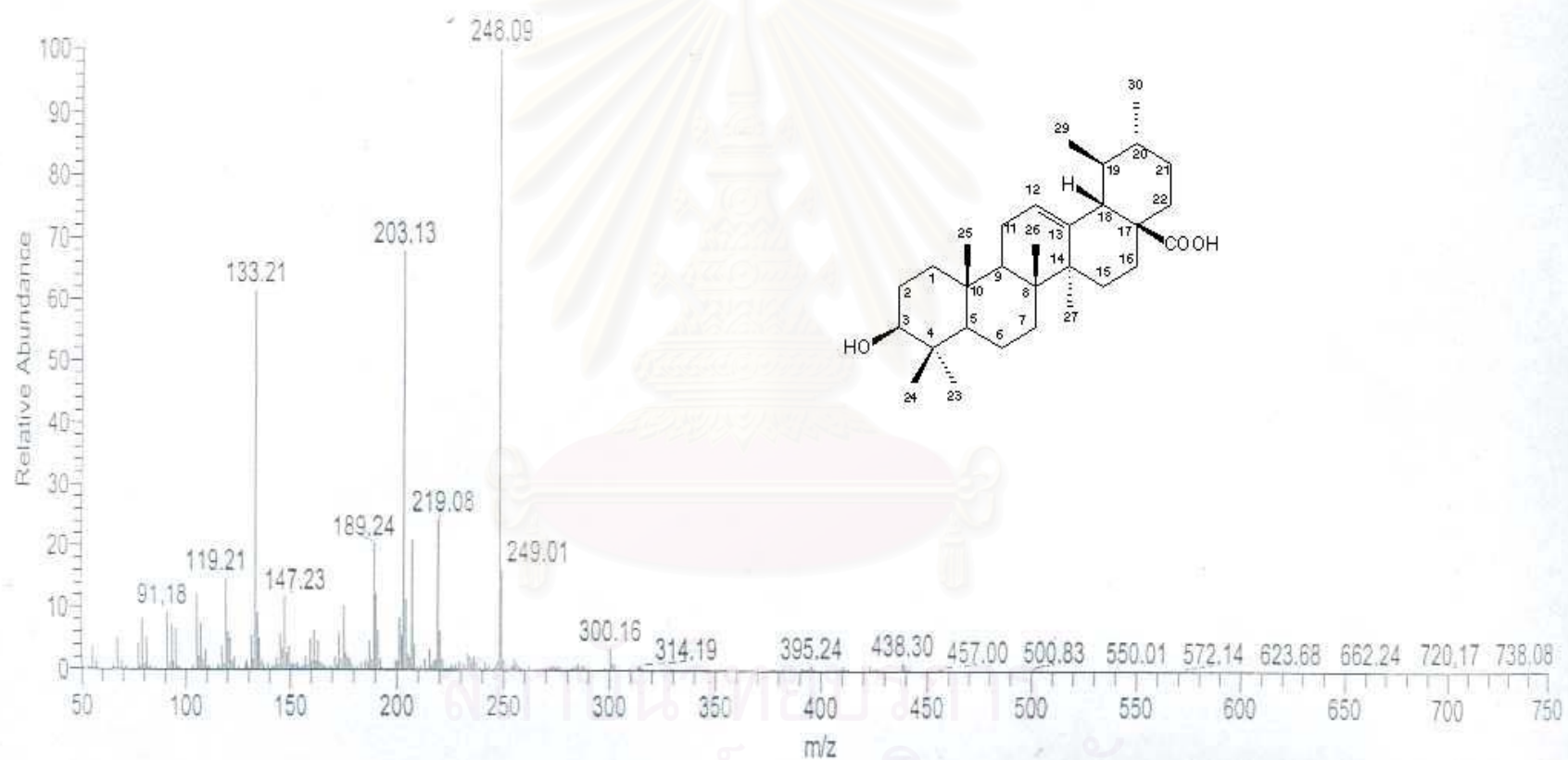


Figure 18. EIMS of compound DR 4

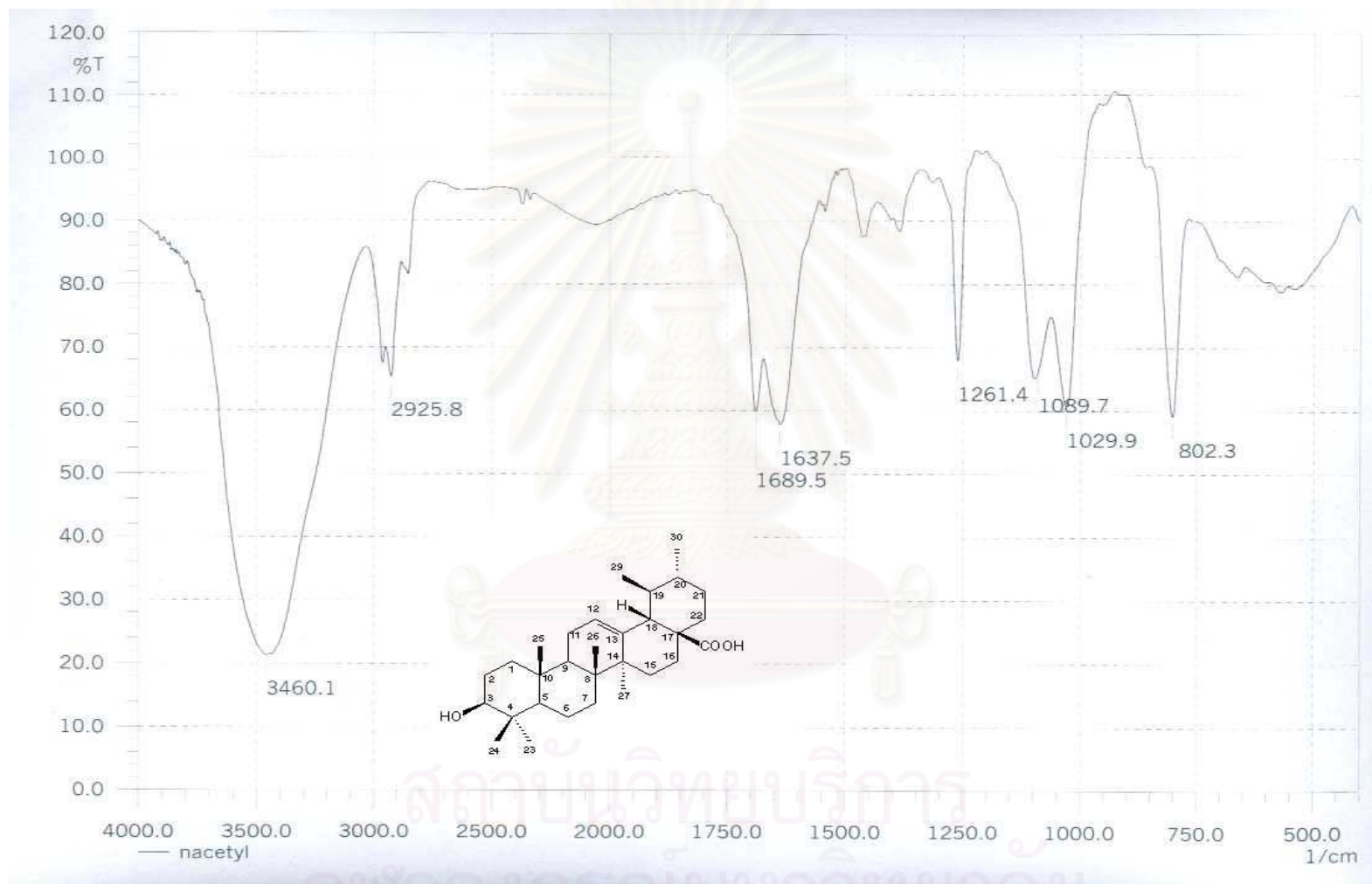


Figure 19. IR spectrum of compound DR4

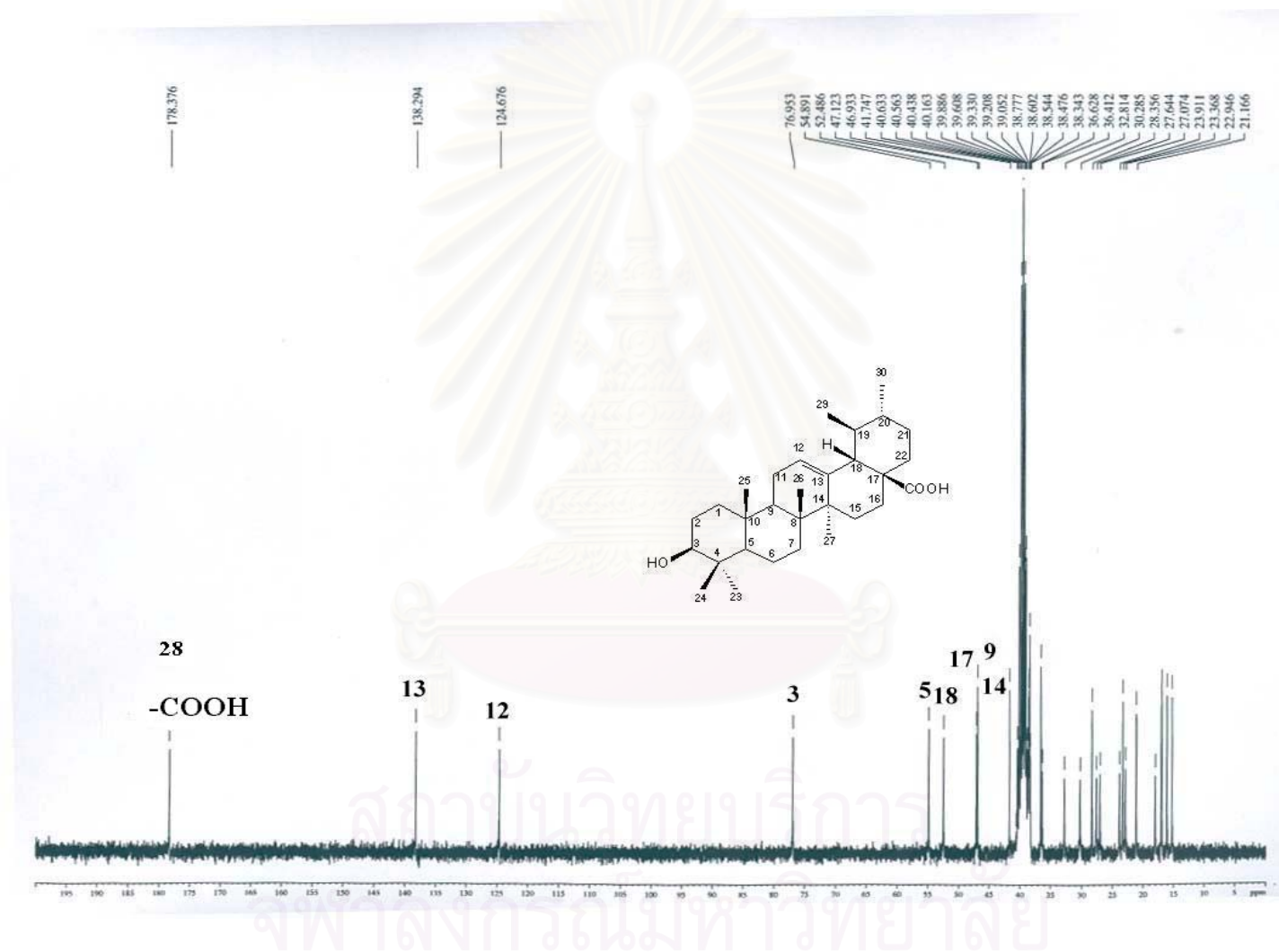


Figure 20a. The 100 MHz ^{13}C -NMR spectrum of compound DR 4 (in $\text{DMSO-}d_6$)

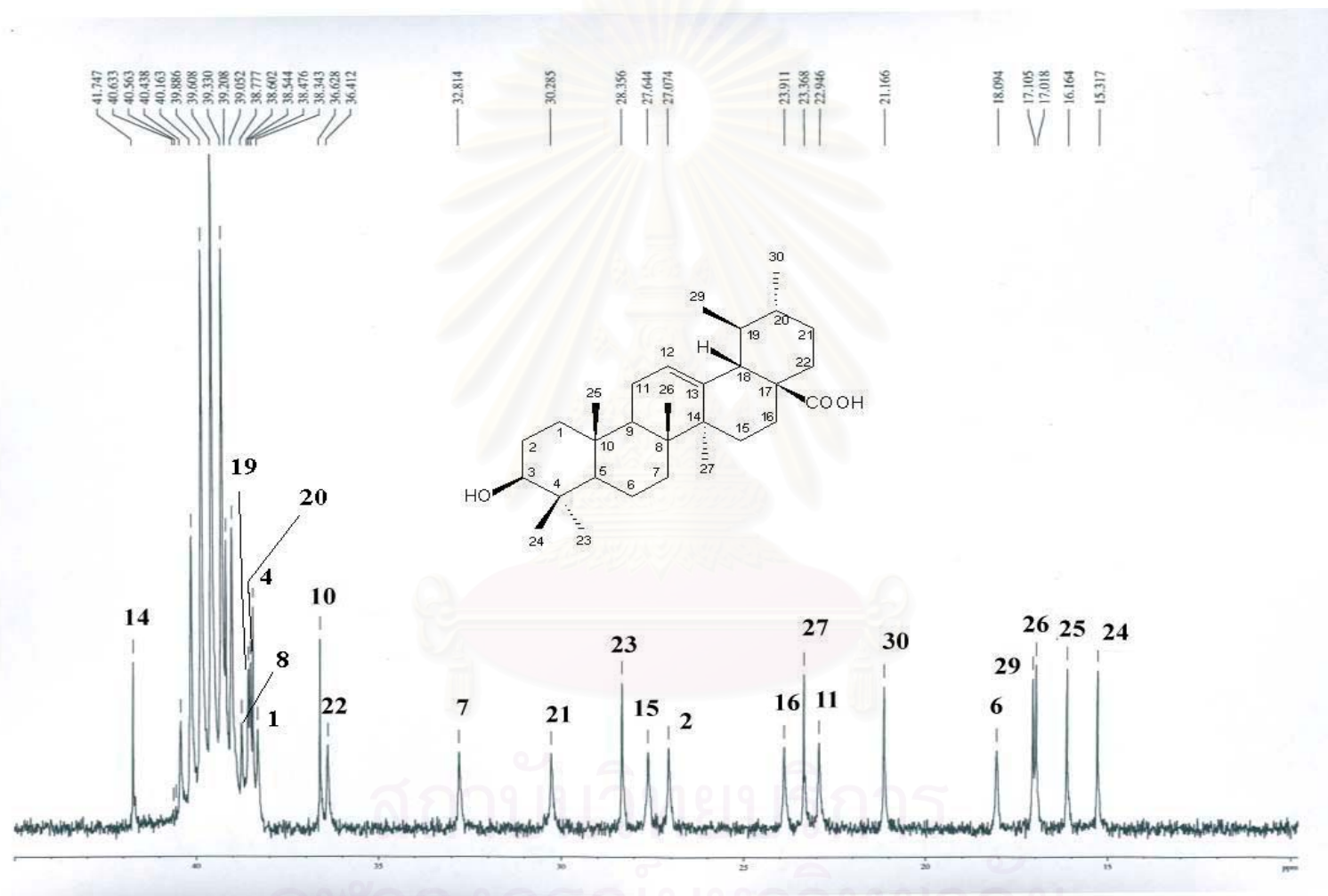


Figure 20b. The 100 MHz ^{13}C -NMR spectrum of compound DR 4 (in $\text{DMSO-}d_6$) (expanded)

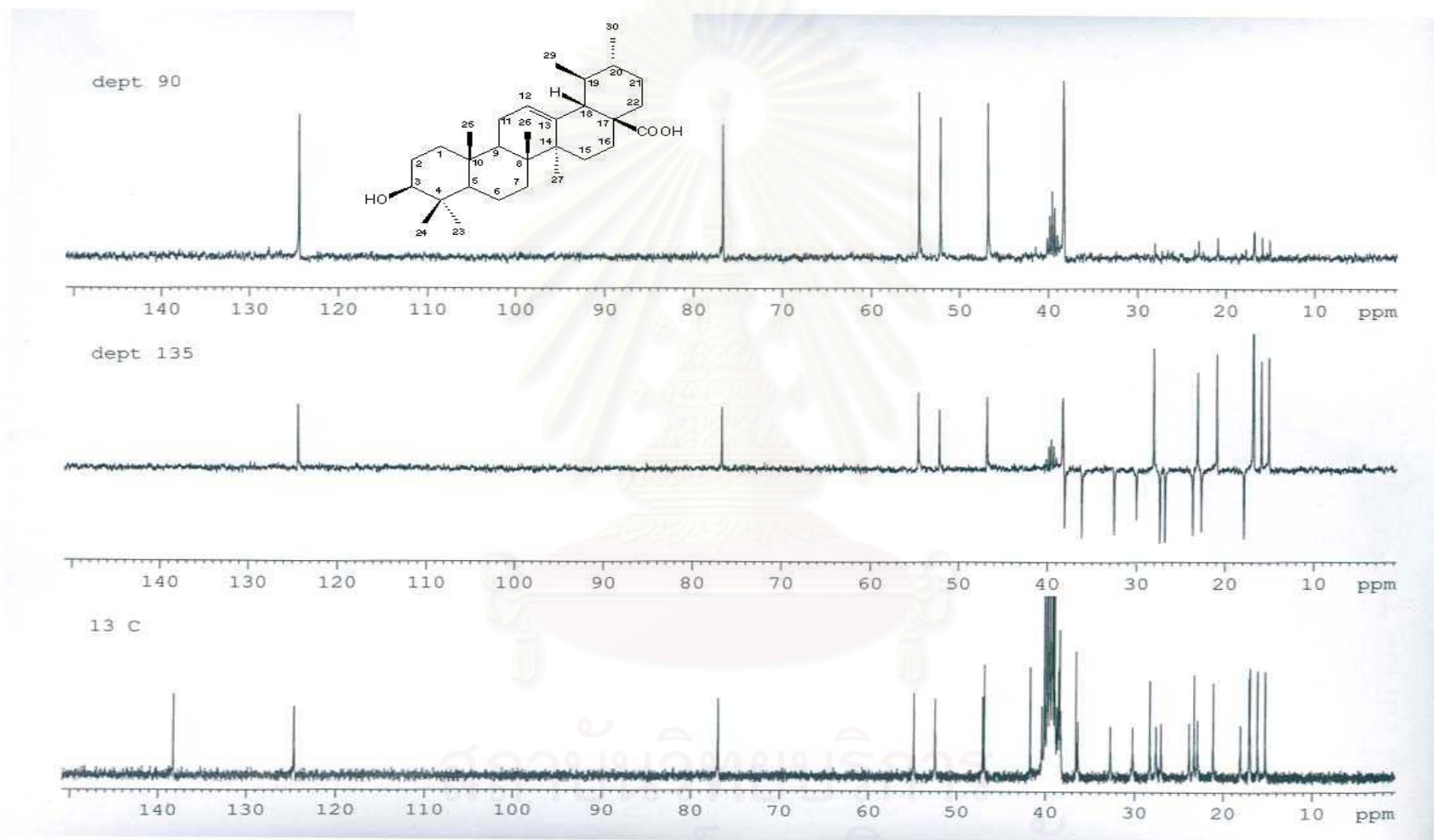


Figure 21a. The 100 MHz ^{13}C -DEPT NMR spectrum of compound DR 4 (in $\text{DMSO-}d_6$)

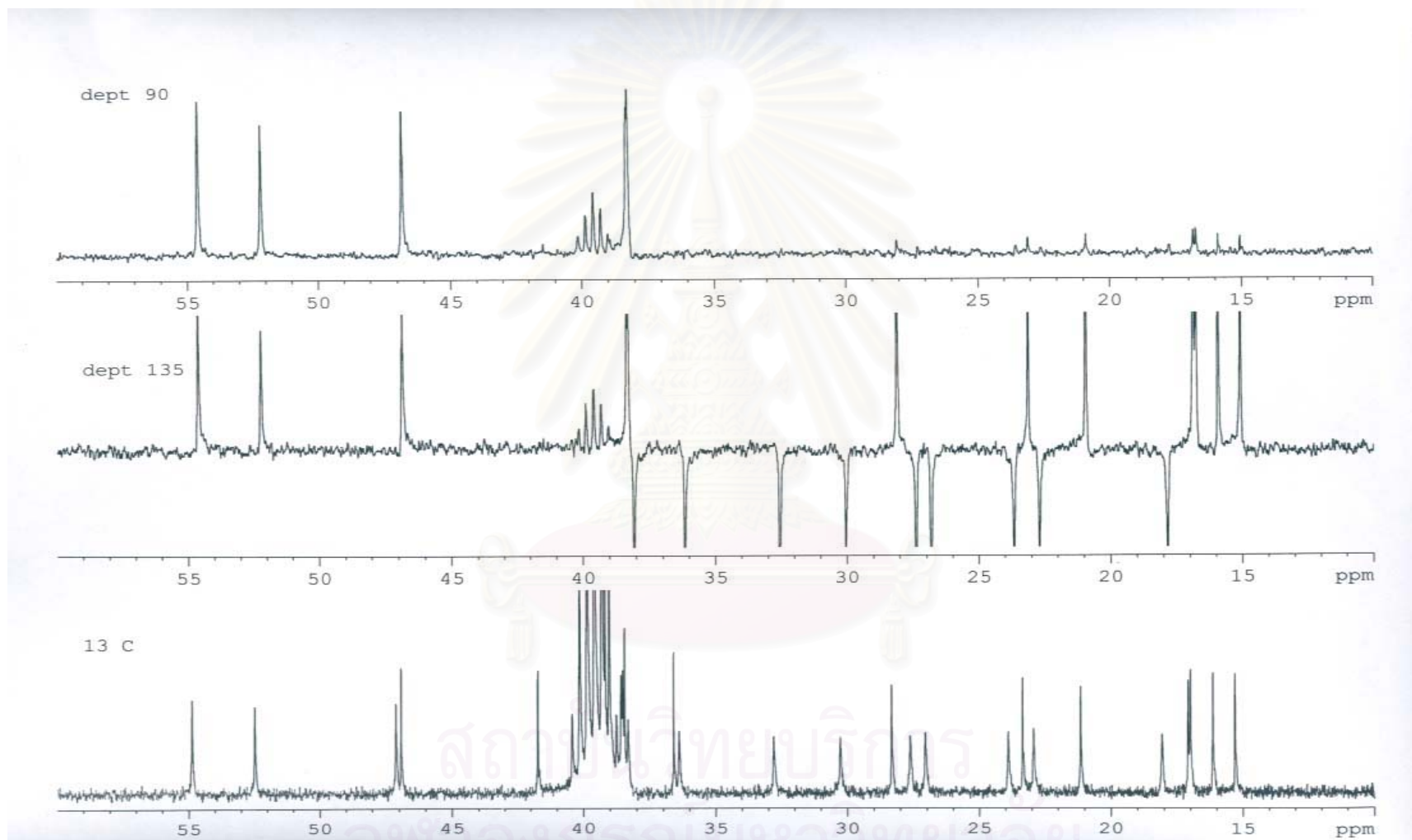


Figure 21b. The 100 MHz ^{13}C -DEPT NMR spectrum of compound DR 4 (in $\text{DMSO-}d_6$) (expanded)

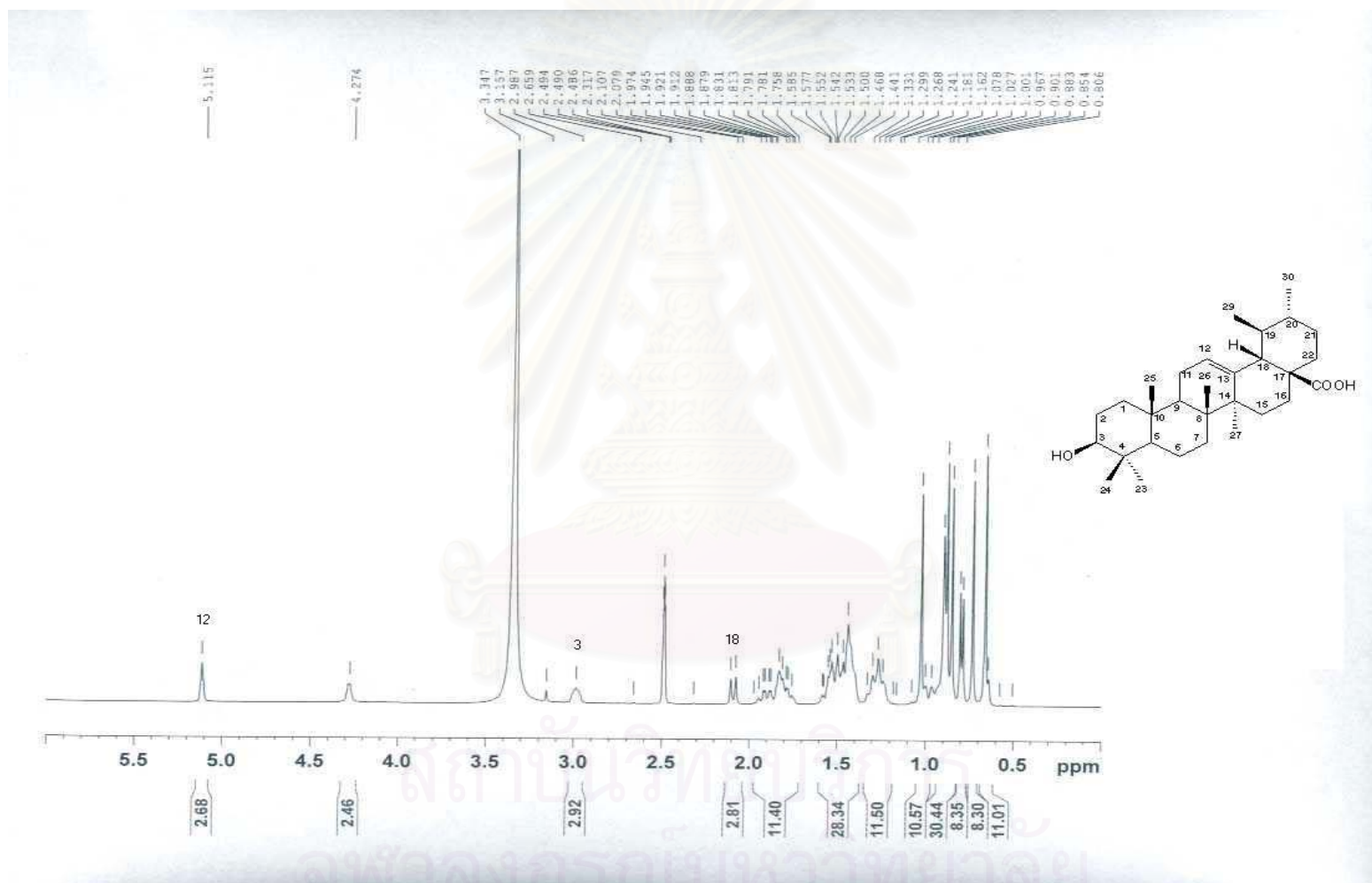


Figure 22a. The 400 MHz ^{13}C -NMR spectrum of compound DR 4 (in $\text{DMSO-}d_6$)

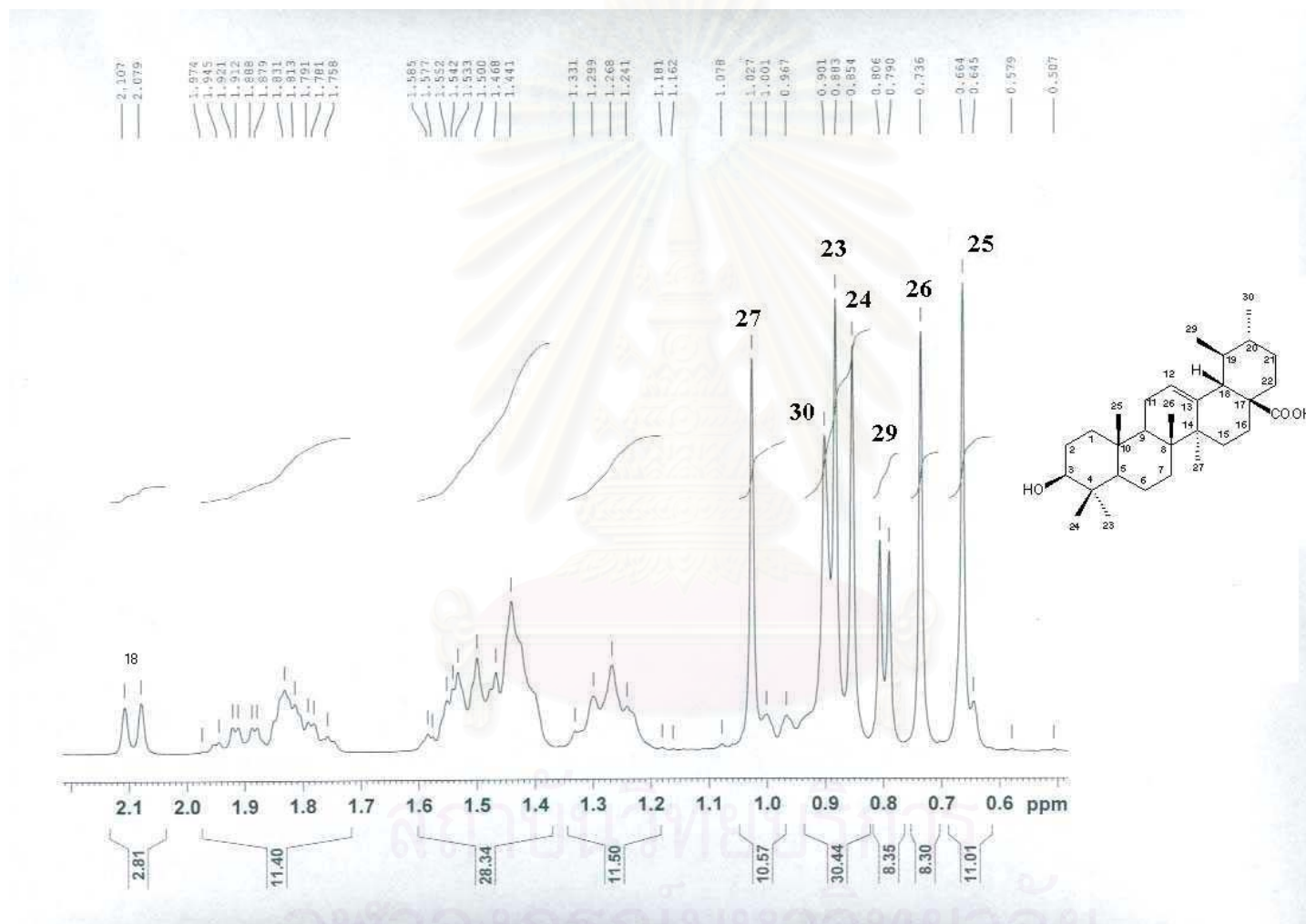


Figure 22b. The 400 MHz ^{13}C -NMR spectrum of compound DR 4 (in $\text{DMSO-}d_6$) (expanded)

CHAPTER V

CONCLUSION

In the present investigation of *Diospyros rubra* Lec. , three triterpenoids and a mixture of steroids were isolated from the stem of the plants by chromatographic techniques.

Two lupane – type triterpenoids, lupeol and betulin, together with a mixture of β -sitosterol and stigmasterol, were isolated from the hexane extract, while an ursane – type triterpenoids, ursolic acid, was isolated from the chloroform extract. The identification of isolated compounds was accomplished by analysis of their spectroscopic data.

This is the first report of the chemical constituents of this *Diospyros* species and the data obtained would be valuable in the chemotaxonomic and phytochemical studies of this plant genus.



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