

## CHAPTER II

### THE GENERALIZED DIFFUSIVITY-MOBILITY RATIO

The diffusivity-mobility ratio or the relation between the diffusion coefficient  $D$  and mobility  $\mu$  of the carriers with charge  $q$  of a semiconductor at finite temperature  $T$  in simple case or completely non-degenerate case is found to be<sup>5</sup>

$$\frac{D}{\mu} = \frac{k_B T}{|q|}. \quad (2.1)$$

This relation is found by Einstein in the context of Brownian motion.

For the generalized diffusivity-mobility ratio, Landsberg<sup>6</sup> applied a concept of diffusion of charged particles in a semiconductor in 1952, as we show later. In this chapter we shall denote the electrochemical potential for charged particles by  $\lambda$ . However, it is convenient to introduce also the chemical potential  $\lambda_0$  of a component in a system. It will be recalled from thermodynamics that  $\lambda_0$  plays the same role as regards equilibrium with respect to particle diffusion or chemical equilibrium, as temperature,  $T$ , does with respect to heat conduction or thermal equilibrium. That  $T$ ,  $\lambda_0$  and also pressure  $P$  are on the same footing may be seen from the Gibb's equation for the mean energy  $U$  of a system of  $N$  neutral particles, which is

$$dU = TdS - PdV + \lambda_0 dN, \quad (2.2)$$

where  $S$ ,  $V$  are entropy and volume, respectively. For a system of particles of charge  $q$  (positive or negative) a change in the number of particles lead to additional energy term  $q\phi dN$ , where  $\phi(\vec{r})$  is the electrostatic potential at position  $\vec{r}$  being considered.

Hence, the analogue of (2.2) for a system of charged particles is

$$dU = TdS - PdV + (\lambda_0 + q\phi)dN. \quad (2.3)$$

Many thermodynamic functions need to be amended by virtue of this change, and  $\lambda_0$  has to be replaced by the electrochemical potential

$$\lambda = \lambda_0 + q\phi. \quad (2.4)$$

In semiconductor work, we deal with electronic states in the conduction band and vacancies in electronic states in valence bands. At equilibrium, they can be treated in terms of a single electrochemical potential  $\lambda$ . This chemical potential pertains to electronic states so that  $q$  is negative. Consequently the change of  $\lambda$  with  $\phi$  is given by

$$\lambda = \lambda_0 - |q|\phi. \quad (2.5)$$

Suppose now that  $\lambda_0$  depends on the position  $\vec{r}$  through the  $\vec{r}$ -dependence of a general carrier concentration  $n$  and the temperature  $T$ . We then have, using (2.5), that the electric field is given by

$$\begin{aligned}\vec{E} &= -\nabla\phi \\ &= \frac{1}{|q|}\nabla(\lambda - \lambda_0) \\ &= \frac{1}{|q|}\left\{\nabla\lambda - \left(\frac{\partial\lambda_0}{\partial n}\right)_{T,V}\nabla n - \left(\frac{\partial\lambda_0}{\partial T}\right)_{n,V}\nabla T\right\}.\end{aligned}\quad (2.6)$$

Now, the current density  $\vec{j}$  due to the current carriers of concentration  $n$  is the sum of a conduction current density  $|q|\mu n\vec{E}$  and a diffusion current density of  $-qD\nabla n$ .

For uniform temperature,

$$\begin{aligned}\vec{j} &= |q|\mu n\vec{E} - qD\nabla n \\ &= \mu n\left[\nabla\lambda - \frac{\partial\lambda_0}{\partial n}\nabla n\right] - qD\nabla n.\end{aligned}\quad (2.7)$$

We also note that from irreversible thermodynamics

$$\vec{j} = \mu n\nabla\lambda.\quad (2.8)$$

so that the coefficient of  $\nabla n$  in (2.7) must vanish. One needs not appeal to (2.8) if one observes that in thermal equilibrium  $\nabla \lambda = 0$  from thermodynamics, that  $\vec{j} = 0$  by hypothesis, so that the coefficient of  $\nabla n$  in (2.7) must vanish. The relation obtained may then be expected to be valid also away from but in the neighborhood of equilibrium. Hence

$$\frac{|q| D}{n \mu} = -\frac{|q|}{q} \left( \frac{\partial \lambda_0}{\partial n} \right)_{T,V}. \quad (2.9)$$

Since  $\lambda$  and  $\lambda_0$  differ by a quantity  $(|q|\phi)$  which can normally be regarded as independence of  $n$ , one finally finds

$$\frac{D}{\mu} = -\frac{n}{q} \left( \frac{\partial \lambda}{\partial n} \right)_{T,V}, \quad (2.10)$$

this is the generalized diffusivity-mobility ratio. In the semiconductor work, the electrochemical potential  $\lambda$  is the same as the Fermi energy  $E_f$  and in our work we deal with n-type semiconductors, majority carriers are electrons with charge  $e$ , so (2.10) can be rewritten as

$$\frac{D}{\mu} = \frac{n}{e} \left( \frac{\partial E_f}{\partial n} \right)_{T,V}, \quad (2.11)$$

equation (2.11) is an important relation which is used to derive the diffusivity-mobility ratio for n-type heavily doped semiconductors in Chapter IV, and to derive by Van Cong and Debiais an empirical diffusivity-mobility ratio.

Note that for Maxwell-Boltzmann distribution of electrons or holes of concentration  $n$  and  $p$ , respectively, and with  $A_e, A_h$  the appropriate functions of temperature and volume

$$n \rightarrow \begin{cases} n = A_e \exp(\lambda/k_B T) \\ p = A_h \exp(-\lambda/k_B T) \end{cases}, \quad (2.12)$$

so that one has

$$\frac{D}{\mu} = \frac{k_B T}{|q|},$$

as the diffusivity-mobility ratio in completely non-degenerate case as (2.1).

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