

## REFERENCES

1. J.K. Mitchell. Fundamentals of Soil Behavior. New York : Wiley, 1993.
2. A.C. Palmer. Ice Lancing, Thermal Diffusion and Water Migration in Freezing Soil . J. Glaciol 47, 6 (1967) : 681-694.
3. D.M. Aderson and N.R. Morgenstern. Physics, Chemistry, and Mechanics of Frozen Ground. A review, Proceedings of the Second International Permafrost Conference, 257-288. Washington : National Academy Press, 1973.
4. J.K. Agnieszka and K. Piotr. The Effect of the Alternate Freezing and Thawing Cycles on Pore Structure of Cementitious Composites Modified by MHEC and PVA . Building and Environment 32 (1997) : 509-512.
5. West, Robert C.. Handbook of Chemistry and Physics. 61<sup>st</sup> Edition. Florida : CRC, 1981.
6. Cutnell, Joha D. and W.J. Kenneth. Physics. 3<sup>rd</sup> Edition. New York : Wiley, 1995.
7. T.C. Powers and T.L. Brownyard. J. Amer. Concr. Inst. 18 (1947) : 549-602.
8. T.C. Powers and T.L. Brownyard. J. Amer. Concr. Inst. 18 (1947) : 933-969.
9. T.C. Powers and R.A. Helmuth. Proc. Highway Res. Board. 32 (1953) : 285-297.
10. JICA. Ceramic Engineering. (1980): 241-243.
11. DIN 52252 Part1. Testing the Frost Resistance of facing Bricks and Clinker Blocks: Freezing of Single Bricks on All Sides . (December 1986).
12. G.C. Robinson et al. Frost-Resistance Testing. Am. Ceram. Soc. Bull. 74, 8 (1995) : 57-61.
13. S.N. Monteiro and C.M.F. Vieira. Solid State Sintering of Red Ceramics at Lower Temperatures. Ceramic International 30 (2004) : 381-387.
14. C.M.F. Vieira et al. Incorporation of Granite Waste in Red Ceramics. Material Science and Engineering 373 (2004) : 115-121.
15. Loran S. O'Bannon. Dictionary of Ceramic Science and Engineering. New York and Lodon : Plenum Press, 1984.
16. W.D. Kingery et al. Introduction to Ceramics. 2<sup>nd</sup> Ediion. New York : Wiley, 1975.
17. G.P. Emiliani and F. Corbara. Tecnologia Ceramica-La Lavorazione. Faenza : Gruppo Editiriale Faenza Editrice, 1999.

18. Ch. Schmidt-Reinholz and H. Schmidt. Tile Brick Int 11, 79 (1995).
19. F.H. Norton. Elements of Ceramics. 2<sup>nd</sup> Edition : Addison-Wesley, 1974.
20. W.E. Worral. Clays and Ceramic Raw Materials. 2<sup>nd</sup> Edition. London & New York : Elsevier Applied Science Publishers, 1986.
21. W. Ryan. Properties of Ceramic Raw Materials. 2<sup>nd</sup> Edition. Great Britain : Biddles Ltd, 1978.
22. W.G. Lawrence and R.R. West. Ceramic Science for The Potter. 2<sup>nd</sup> Edition. Pennsylvania : Chilton Book, 1982.
23. R. Dieckmann. J. Electrochem. Soc 116 (1969) : 1409.
24. W. E. Brownell. Structural Clay Products. New York : Springer-Verlag, 1976.
25. W.E. Brownell. Subsolidus Reactions Between Mullite and Iron Oxide . J. Am. Ceram. Soc. 41 (1958) : 226-230.
26. J.E. Houseman and C.J. Koenig. Influence of Kiln Atmospheres in Firing Structural Clay Products: II, Color Development and Burnout. J. Am. Ceram. Soc. 54 (1971) : 82-89.
27. W. Wanie. Firing in Reducing Atmosphere. Ziegelindustrie 20, 2-3 (1967) : 21-25, 89-95.
28. D.F. Dailly. Ceramic Thermochemistry: II Chemical Reduction in Bricks . Ceramics 18, 216 (1967) : 18-22.
29. M.F. Chaplin. Water; its importance to life. Biochem. Mol. Biol. Educ. 29, 2 (2001) : 54-59.
30. C. Martin. The Phase Diagram of Water. Scilinks from NSTA : Science Educators, 2003.
31. Kenneth G. Libbrecht. Snow Crystals. Physics department, California Institute of Technology , 1999.
32. F.A.L. Dullien . Porous Media : Fluid Transport and Pore Structure. New York : Academic Press, 1979.
33. M. Mamillan. Durabilite des pierres tendres. 102 : Centre d'Etude Batiment Travaux Publics, 1984.

34. J.F. Daian. Processus de Condensation et de Transfert d'eau Dans un Matériau Meso et Macroporeux, Etude Experimentale du Mortier de Cement, 319. Natl : Polytechnique de Grenoble, 1986.
35. S.S. Efimov. Vlaga Gigroskopiceskih Materialov. Nauka : Novosibirsk, 1986.
36. J.R. Blachere and J.E. Young. Failure of Capillary Theory of Frost Damage as Applied to Ceramics. J. Am. Ceram. Soc. 57, 5 (1974) : 212-216.
37. R. Defay and I. Prigogine. Tension Superficielle et Adsorption (Surface Tension and Adsorption). Belgium : S.A. Liege, 1951.
38. S. chatterji. Freezing of Aqueous Solution in a Porous Medium. Part I. Freezing of Air-Entraining Agent Solutions . Cement and Concr. Research 15 (1985) :13-20.
39. G. Fagerlund. Studies of the Destruction Mechanism at Freezing of Porous Materials. J. Aguirre-Puente, Nordiques Problem Raised by Frost Action, 167-197. Paris : Centre d'Etude Arctique, 1979.
40. B. Butterworth. Frost Resistance of Bricks and Tiles: A Review. J. Brit. Ceram. Soc. 1, 2 (1964) : 202-236.
41. M. Bellanger et al. Water Behavior in Limestones as a function of pores structure: Application to Frost Resistance of Some Lorraine Limestones . Engineering Geology 36 (1993) : 99-108.
42. G.G. Litvan. Phase Transitions of Adsorbates: IV, Mechanism of Frost Action in Hardened Cement Paste. J. Am. Ceram. Soc. 55, 1 (1972) : 38-42.
43. N. Masahiko. Automatic Unidirectional Freeze-Thaw Test for Frost Durability of Building Materials. Am. Ceram. Soc. Bull. 67, 12 (1988) : 1966-1968.
44. Mikulas Sveda. Frost Resistance of Brick. Am. Ceram. Soc. Bull. 80, 9 (2001) : 46-48.
45. Prof. George Miller. Dictionary and Encyclopedia. The Cognitive Science Department, Princeton University, 2003.
46. Jander, W. and E. Hoffmann. Reaction in the solid state at high temperatures. XI. Reaction between calcium oxide and silicon dioxide. Z. anorg. allg. Chem. 218, (1934) : 211-223.



Appendices

ศูนย์วิทยทรัพยากร  
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## Appendix A

Table A-1 Particle size distribution of raw clay

particle sizes ( $\mu\text{m}$ )	raw clay wt. %	raw clay wt. % finer than
0.226	0.163	0.163
0.259	0.284	0.447
0.296	0.453	0.900
0.339	0.691	1.591
0.389	1.027	2.618
0.445	1.353	3.971
0.510	1.695	5.666
0.584	1.957	7.623
0.669	2.176	9.799
0.766	2.355	12.154
0.877	2.529	14.683
1.005	2.808	17.491
1.151	3.064	20.555
1.318	3.588	24.143
1.510	4.103	28.246
1.729	4.596	32.842
1.981	5.191	38.033
2.269	5.691	43.724
2.599	6.141	49.865
2.976	6.514	56.379
3.409	6.866	63.245
3.905	7.082	70.327
4.472	7.009	77.336
5.122	6.688	84.024
5.867	5.454	89.478
6.720	4.075	93.553
7.697	2.738	96.291
8.816	2.301	98.592
10.097	0.920	99.512
11.565	0.366	99.878
13.246	0.122	100.000
15.172	0.000	100.000
17.377	0.000	100.000
19.904	0.000	100.000

Table A-2 Particle size distribution of sand and grog

sieve no.	particle size ( $\mu\text{m}$ )	sand wt. %	grog wt. %
over 20 mesh	> 850	2.7	11.4
20 - 35 mesh	850 - 500	33.5	24.9
35 - 50 mesh	500 - 300	37.7	22.2
50 - 100 mesh	300 - 150	21.6	15.1
100 - 140 mesh	150 - 106	2.2	4.1
140 - 200 mesh	106 - 75	0.9	4.0
200 - 230 mesh	75 - 63	0.3	2.0
230 - 325 mesh	63 - 45	0.3	3.2
under 325 mesh	< 45	0.8	13.0

Table A-3 Water content of raw materials

materials	wt. (g)	dried wt. (g)	water content (%)	ave	stdev
raw clay 1	108.1	88.9	17.8	20.7	2.0
raw clay 2	114.2	92.4	19.0		
raw clay 3	110.4	84.4	23.5		
raw clay 4	104.6	82.0	21.6		
raw clay 5	107.7	84.9	21.1		
raw clay 6	104.4	82.4	21.1		
sand	100.0	94.0	6.0	6.0	-
grog	100.0	95.7	4.3	4.3	-

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## Appendix B

Table B-1 Linear drying shrinkage (%)

conditions	L drying shrinkage (%)	conditions	L drying shrinkage (%)
mixed clay1	5.1	grog 10%	4.8
mixed clay2	5.5	grog 12%	4.3
sand 10%	5.0	grog #50	4.7
sand 15%	4.6	grog #100	4.4
sand #50	4.9	S50G5	6.2
sand #100	4.9	SGG	5.8

Table B-2 Linear firing shrinkage (%)

temperature°C	mixed clay 1	mixed clay 2		
900	1.1	1.1		
950	1.9	1.9		
1000	3.3	3.2		
1050	4.6	4.5		
1100	5.0	5.2		
	CaO 1%	S50G5	SGG	
850	1.0	0.5	0.7	
900	1.8	0.9	1.2	
950	3.0	2.0	2.4	
1000	4.3	3.4	3.7	
1050	5.2	4.7	4.7	
1100	5.3	4.9	4.9	
	sand 10%	sand 15%	sand #50	sand #100
900	0.8	0.5	0.9	1.0
950	1.5	1.2	1.8	1.8
1000	2.8	2.4	3.2	3.3
1050	4.1	3.4	4.6	4.8
1100	4.4	3.7	5.1	5.3
	grog 10%	grog 12%	grog #50	grog #100
900	1.1	1.0	1.0	1.1
950	1.8	1.7	1.9	1.8
1000	3.2	3.2	3.3	3.3
1050	4.5	4.5	4.6	4.7
1100	5.0	5.0	5.1	5.2
	glass2%	glass4%	glass6%	glass8%
850	0.7	0.8	1.0	1.1
900	1.3	1.4	1.7	2.1
950	2.3	2.7	2.9	3.2
1000	3.9	3.9	4.1	4.5
1050	5.0	5.1	5.1	5.1
1100	5.3	5.3	5.2	5.2

## Appendix C

Table C-1 Water absorption (w.a. %) and bulk density (b.d. g/cm<sup>3</sup>)

temperature °C	mixed clay 1 (hand mixing)		mixed clay 1 (machine mixing)		mixed clay 2			
	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.		
900	14.9	1.9	14.5	2.0	14.6	1.9		
950	13.7	2.0	13.6	2.0	13.6	2.0		
1000	11.6	2.0	11.2	2.1	11.5	2.0		
1050	9.1	2.1	7.6	2.2	9.0	2.1		
1100	7.4	2.2	5.4	2.3	7.4	2.2		
	Sand 10%		Sand 15%		Sand #50		Sand #100	
	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.
900	14.3	1.9	14.4	1.9	14.2	1.9	14.2	2.0
950	13.3	2.0	13.5	2.0	13.1	2.0	13.0	2.0
1000	11.2	2.1	12.1	2.0	10.3	2.1	10.1	2.1
1050	9.3	2.1	10.0	2.1	7.6	2.2	7.0	2.2
1100	7.7	2.2	7.8	2.1	5.8	2.2	5.5	2.2
	Grog 10%		Grog 12%		Grog #50		Grog #100	
	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.
900	14.7	1.9	15.1	1.9	14.4	1.9	14.1	2.0
950	14.1	2.0	14.1	2.0	13.7	2.0	13.1	2.0
1000	11.8	2.0	12.1	2.0	10.9	2.1	10.2	2.1
1050	9.3	2.1	9.5	2.1	8.4	2.2	7.6	2.2
1100	7.8	2.2	8.2	2.1	7.5	2.2	6.0	2.2
	NaO 3%		CaO 1%		S50G5		SGG	
	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.
850	11.7	2.0	14.1	1.9	13.3	2.0	12.4	2.0
900	9.6	2.1	13.3	2.0	13.2	2.0	11.9	2.0
950	8.0	2.1	10.4	2.1	11.7	2.0	9.6	2.1
1000	4.8	2.2	6.1	2.2	8.8	2.1	6.9	2.2
1050	4.0	2.2	4.5	2.3	5.9	2.2	4.8	2.3
1100			4.1	2.2	4.0	2.3	3.7	2.3
	glass2%		glass4%		glass6%		glass8%	
	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.	w.a.	b.d.
850	13.3	2.0	13.1	2.0	12.5	2.0	12.1	2.0
900	13.0	2.0	12.6	2.0	11.9	2.0	11.5	2.0
950	11.5	2.1	10.8	2.1	9.8	2.1	8.9	2.1
1000	8.4	2.2	7.7	2.2	7.3	2.2	6.4	2.2
1050	6.1	2.2	5.7	2.2	5.3	2.3	5.3	2.3
1100	4.8	2.3	4.6	2.3	4.7	2.3	4.7	2.3



## Appendix D

Table D-1 Bending strength (MPa)

temperature °C	mixed clay 1 (hand mixing)		mixed clay 1 (machine mixing)		mixed clay 2			
	MPa	stdev	MPa	stdev	MPa	stdev		
900	19.6	3.2	25.2	3.0	23.1	2.1		
950	18.4	2.8	25.5	5.0	24.4	3.5		
1000	19.9	2.2	26.0	4.4	25.4	2.4		
1050	19.5	4.8	26.0	6.5	25.2	5.5		
1100	23.0	3.6	33.6	5.2	34.2	5.9		
	Sand 10%		Sand 15%		Sand #50		Sand #100	
	MPa	stdev	MPa	stdev	MPa	stdev	MPa	stdev
900	16.2	1.5	12.6	1.4	20.8	2.4	21.8	2.4
950	15.7	1.7	13.1	1.2	22.9	2.6	23.4	3.0
1000	18.7	2.4	12.8	1.7	23.6	3.4	24.9	3.0
1050	16.7	2.0	12.9	1.9	26.6	2.6	27.2	3.5
1100	18.2	2.9	14.0	1.9	31.0	5.9	32.0	7.9
	Grog 10%		Grog 12%		Grog #50		Grog #100	
	MPa	stdev	MPa	stdev	MPa	stdev	MPa	stdev
900	16.7	2.8	16.0	2.2	19.4	1.8	19.2	3.0
950	19.3	2.5	17.1	2.4	19.7	2.9	20.9	3.3
1000	20.7	2.2	17.3	1.7	22.0	3.0	21.1	4.9
1050	19.3	2.1	18.6	1.8	20.2	2.6	20.8	2.7
1100	22.3	2.8	20.4	2.2	23.5	3.4	25.1	3.1
	NaO 3%		CaO 1%		S50G5		SGG	
	MPa	stdev	MPa	stdev	MPa	stdev	MPa	stdev
850	13.3	1.3	19.7	3.4	22.7	2.6	23.4	3.1
900	12.8	1.3	21.9	2.9	23.8	3.2	23.9	3.8
950	12.8	1.4	21.0	3.8	26.8	4.8	28.9	3.2
1000	17.1	2.9	21.0	2.4	26.7	5.0	24.6	3.9
1050	14.3	3.3	25.7	5.8	27.4	5.8	25.8	2.8
1100			24.3	4.6	32.5	4.4	32.9	4.4
	glass2%		glass4%		glass6%		glass8%	
	MPa	stdev	MPa	stdev	MPa	stdev	MPa	stdev
850	18.4	4.4	20.2	3.9	21.8	3.0	20.4	4.3
900	22.0	4.5	22.7	3.7	24.2	4.0	23.1	3.0
950	24.1	3.7	24.7	3.7	25.0	3.6	23.7	2.4
1000	23.9	3.0	24.1	4.4	23.8	2.7	25.5	3.2
1050	24.3	6.6	24.8	4.9	25.2	5.4	25.6	4.1
1100	28.0	5.7	30.0	6.3	26.6	6.3	26.8	4.5

## Appendix E

Table E-1 Capillary pore volume (ml/g)

pore diameter (nm)	capillary pore volume (ml/g)						
	SM 1	SM 2	Portugal	German	S50G5	SGG	SGG 950°C
under 6	0.00269	0.00296	0.00049	0.00025	0.00042	0.00109	0.00047
6 - 8	0.00242	0.00270	0.00022	0.00011	0.00021	0.00078	0.00021
8 - 10	0.00299	0.00322	0.00017	0.00008	0.00033	0.00105	0.00021
10 - 12	0.00399	0.00424	0.00019	0.00007	0.00071	0.00146	0.00027
12 - 16	0.00691	0.00789	0.00028	0.00009	0.00128	0.00287	0.00050
16 - 20	0.00886	0.00868	0.00036	0.00008	0.00240	0.00407	0.00073
20 - 80	0.04716	0.04933	0.00560	0.00046	0.02401	0.02958	0.01055
over 80	0.00685	0.01137	0.00639	0.00030	0.05418	0.00825	0.00693
<b>total</b>	<b>0.08187</b>	<b>0.09039</b>	<b>0.01370</b>	<b>0.00144</b>	<b>0.08354</b>	<b>0.04915</b>	<b>0.01987</b>



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Mr. Punyoot Huantanom was born in Rayong on 31<sup>st</sup> July 1980. He received a Bachelor's Degree with 2<sup>nd</sup> Class Honors in Physics from Faculty of Science, Srinakharinwirot University in 2002. He continued studying for Master's Degree in the field of Ceramic Technology at Chulalongkorn University and graduated in December 2004.



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