

REFERENCES

- Ansel, H. C., Allen, L. V., and Popovich, N. G. 1999. Suppositories and Inserts. Pharmaceutical dosage form and drug delivery system. 7th ed. Philadelphia: Awolters Kluwer Company. pp. 279 - 283.
- Ansel, S., and Christel, C. 1997. In vitro and in vivo diclofenac sodium evaluation after rectal application of soft gelatin capsules enabling application induced transformation (AIT) into a semisolid system of liquid crystal (SSLC) for controlled release. Pharm. Res. 14(12): 1726 - 1729.
- Babar, A., Bellete, T., and Plakogiannis, F. M. 1999. Ketoprofen suppository dosage form: In vitro release and in vivo absorption studies in rabbits. Drug. Dev. Ind. Pharm. 25(2): 241 - 245.
- Cade, D., Cole, E. T., Mayer, J. P., and Witter, F. 1986. Liquid filled sealed hard gelatin capsules. Drug. Dev. Ind. Pharm. 12(11-13): 2289 - 2300.
- Cole, S. K., Story, M. J., Attwood, D., Laudanski, T., Robertson, J., and Barnwell, S. G. 1992. Studies using a non-ionic surfactant-containing drug delivery system designed for hard gelatin capsule compatibility. Int. J. Pharm. 88: 211 - 220.
- Dredan, J., Zelko, R., Bahari, E., Racz, I., and Gondar, E. 1998. Effect of polysorbates on drug release from wax matrices. Drug. Dev. Ind. Pharm. 24(6): 573 - 576.
- Eerikainen, S., Leino, J., Harjula, M., Klinge, E., and Marvola, M. 1996. Use of hard gelatin capsule as a rectal dosage form. S.T.P. Pharm. Sci. 6(6): 435 - 440.
- Gilbert, S. B., and Christopher, T. R. 1992. Modern pharmaceuticals. New York: Marcel Dekker, Inc. pp. 571 - 579.

- Hawley, A. R., Rowley, G., Lough, W. J., and Chatham, S. 1992. Physical and chemical characterization of thermosoftened bases for molten filled hard gelatin capsule formulations. Drug. Dev. Ind. Pharm. 18(16): 1719 - 1739.
- Jamali, F., and Dion, R. B. 1990. Clinical pharmacokinetic of ketoprofen and its enantiomers. Clin. Pharmacokinet. 19(3): 197 - 217.
- Jones, B. J. 1985. Hard gelatin capsules and the pharmaceutical formulator. Pharm. Tech. 106 - 108.
- Ken. Y., Shah. A. C., and Toshiaki. N. 1993. Enhanced rectal absorption of itazagel formulated with polysorbate 80 micelle vehicle in rat: role of co-administered esterase. J. Pharm. Pharmacol. 46: 608 - 611.
- Kenneth, J. S., Anthony, S. R., and Jamali, F. 1993. Ketoprofen pharmacokinetics in the elderly: Influence of rheumatic disease, renal function and dose. J. Clin. Pharmacol. 33: 1052 - 1059.
- Leino, J., Haavisto, H., Tomminen, T., Heinila, K., Eerikainen, S., Klinge, E., and Marvola, M. 1997. Development of rectally administered prolong release hard gelatin capsules using different polymers as diluent. S.T.P. Pharm. Sci. 7(5): 348 - 353.
- Maffione, G., Lamartino P., Guglielmini, G., and Gazzaniga, A. 1993. High viscosity HPMC as a film coating agent. Drug. Dev. Ind. Pharm. 19(16):2043 - 2053.
- Nilufer, T. and Ermis, D. 1996. Sustained released characteristics and pharmacokinetic parameters of ketoprofen suppositories using chitosan. Int. J. Pharm. 147:71 - 77.
- Nygqvist, H. 1983. Saturated salt solutions for maintaining specified relative humidities. Int. J. Pharm. Tech. & Prod. Mfr. 4(2): 47 - 48.

- Panvipa Tuntisuk. 1993. The development of sustained release ketoprofen solid dispersion and in vivo evaluation. Doctoral Dissertation, Mahidol University, p. 65.
- Peeracha Thanawnttanawanich. 1999. Development of dimenhydrinate liquid filled coated hard gelatin capsule for rectal application: An alternative for tropical zone. (Master's Thesis, Pharmaceutical Sciences, Chulalongkorn University), pp. 41- 42.
- Plaxco, J. M., 1984. Suppositories. In: King, R. E. Ed. Dispensing of medication. 9th ed. Easton, Mack Publishing Company. p. 88.
- Reed, K. W., and Yalkowsky, S. H. 1985. Lysis of human red blood cells in the presence of various cosolvents. Bull. Parenter. Drug. Assoc. 39: 64 - 69.
- Reynolds, J. E. F. 1993. Martindale the extra pharmacopoeia. London: The Pharmaceutical Press. pp. 21 - 22.
- Ridgway, K. 1987. Hard gelatin capsules development and technology. London: The Pharmaceutical Press. pp. 165 - 167.
- Chang, R. K., Krishnaswamy, S., Raghavan S., and Munir, A. H. 1998. A study on gelatin capsule brittleness: Moisture transfer between the capsule shell and its content. J. Pharm. Sci. 87(5): 556 - 558.
- Shah, V. P., et al. 1992. Analytical methods validation: bioavailability, bioequivalence and pharmacokinetic studies. Int. J. Pharm. 82: 1 - 7.
- Shimpoo, K., Takeuchi, M., Iwata, M., Takahashi, M., Asai, I., Hiyoshi, K., Matsuura, K., Tsuchiya, J., and Tanabe, K. 1981. Toxicological study on the anorectal irritation and systemic organ toxicity evoked by long-term intrarectal administration of ketoprofen in rabbits. J. Toxicol. Sci. 6(3): 217 - 224.

The British Pharmacopoeia. 1993. London: HMSO. p. 977.

The British Pharmacopoeia. 1999. London: HMSO. pp. A 197 - 198.

Kibbe, A.H. 2000. Handbook of pharmaceutical excipients. London: The Pharmaceutical Press. pp. 392 - 397.

The United States Pharmacopoeia 24. 2000. Rockville: The United States Pharmacopoeial Convention. Inc. pp. 2084 - 2085.

US FDA. 1999. Waiver of *in vivo* bioavailability and bioequivalence studies for immediate release solid oral dosage forms contain certain active moieties/ active ingredient based on a biopharmaceutics classification system. pp. 1 - 14.

Wainer, I. W. 1985. Liquid chromatography in pharmaceutical development. Oregon: Aster publishing corporation. p. 438.

Weiner, D. J., and Yuh, L. 1994. Bioavailability studies. In Buncher, C. R., and Tsay, J. Y. (eds.). Statistics in the pharmaceutical industry, 2nd ed. New York: Marcel Dekker, Inc. p. 237.

Yie, W. C. 1992. Novel drug delivery system. New York: Marcel Dekker, Inc. pp. 177 - 207.

Zia, H., M, J. K., Donnell, J. P., and Luzzi, L. 1991. Cosolvency of dimethyl isosorbide for steroidal solubility. Pharm. Res. 8: 502 - 504.



APPENDICES

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX A

REAGENT PREPARATION

Phosphate buffer pH 7.4

Dissolved 27 g of potassium dihydrogen phosphate in water and adjust to 1 liter. Take 50 mL of this solution to mix with 39.5 mL of 0.2 M sodium hydroxide solution and diluting to 200 mL with water to a pH 7.4 ± 0.02 .

Sodium acetate buffer pH 4.2

Dissolved 1.6256 g of sodium acetate trihydrate in water, mix with 2.4 mL of glacial acetic acid, adjust with water to 500 ml and to pH of 4.2 ± 0.02 .



ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX B

VALIDATION OF ANALYTICAL METHOD FOR *IN VITRO* STUDIES

1. Accuracy

Table 28. Accuracy of analytical method for determination of ketoprofen in phosphate buffer pH 7.4 at $\lambda = 260$ nm

| Concentration ($\mu\text{g/mL}$) | Inversely estimated concentration ($\mu\text{g/mL}$) | % Recovery |
|---------------------------------------|-----------------------------------------------------------|------------|
| 4 | 4.03 | 100.80 |
| 9 | 8.93 | 99.23 |
| 15 | 15.03 | 100.22 |

Mean % recovery = 100.09, S.D. = 0.79, C.V. = 0.79%

* Each data was determined using three determinations per concentration.

Table 29. Accuracy of analytical method for determination of ketoprofen in 75% methanol at $\lambda = 258$ nm

| Concentration ($\mu\text{g/mL}$) | Inversely estimated concentration ($\mu\text{g/mL}$) | % Recovery |
|---------------------------------------|-----------------------------------------------------------|------------|
| 4 | 3.95 | 98.80 |
| 10 | 10.01 | 100.11 |
| 14 | 14.04 | 100.27 |

Mean % recovery = 99.73, S.D. = 0.81, C.V. = 0.81%

* Each data was determined using three determinations per concentration.

2. Precision

2.1 Within Run Precision

Table 30. Within run precision of analytical method for determination of ketoprofen in phosphate buffer pH 7.4 at $\lambda = 260$ nm

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 4 | 4.044 \pm 0.048 | 1.18 |
| 9 | 8.954 \pm 0.053 | 0.59 |
| 15 | 15.003 \pm 0.031 | 0.21 |

* Each data was determined using three determinations per concentration.

Table 31. Within run precision of analytical method for determination of ketoprofen in 75% methanol at $\lambda = 258$ nm

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 4 | 4.008 \pm 0.058 | 1.46 |
| 10 | 10.104 \pm 0.081 | 0.80 |
| 14 | 14.055 \pm 0.077 | 0.55 |

* Each data was determined using three determinations per concentration.

2.2 Between Run Precision

Table 32. Between run precision of analytical method for determination of ketoprofen in phosphate buffer pH 7.4 at $\lambda = 260$ nm

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 4 | 4.029 \pm 0.071 | 1.76 |
| 9 | 8.973 \pm 0.063 | 0.70 |
| 15 | 15.040 \pm 0.076 | 0.51 |

* Each data was determined using three determinations per concentration.

Table 33. Between run precision of analytical method for determination of ketoprofen in 75% methanol at $\lambda = 258$ nm

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 4 | 3.950 \pm 0.073 | 1.86 |
| 10 | 9.967 \pm 0.052 | 0.52 |
| 14 | 14.068 \pm 0.085 | 0.61 |

* Each data was determined using three determinations per concentration.

3. Calibration curve

Table 34. Typical calibration curve data for determination of ketoprofen in phosphate buffer pH 7.4 estimated using linear regression¹

| Concentration ($\mu\text{g/mL}$) | Absorbance ($\lambda = 260 \text{ nm}$) | Inversely estimated concentration ($\mu\text{g/mL}$) ² . | % Recovery ³ |
|---------------------------------------|----------------------------------------------|--------------------------------------------------------------------------|-------------------------|
| 3 | 0.182 | 3.02 | 100.53 |
| 4.6 | 0.284 | 4.59 | 99.82 |
| 6.2 | 0.384 | 6.16 | 99.32 |
| 7.8 | 0.491 | 7.82 | 100.28 |
| 9.4 | 0.594 | 9.42 | 100.25 |
| 11.0 | 0.697 | 11.02 | 100.18 |
| 12.6 | 0.801 | 12.63 | 100.26 |
| 14.2 | 0.901 | 14.26 | 99.99 |
| 15.8 | 1.004 | 15.79 | 99.93 |
| | | Mean | 100.06 |
| | | S.D. | 0.37 |
| | | % C.V. ⁴ | 0.37 |

1. $r^2 = 1$, $Y = 0.0643x - 0.0116$

2. Inversely estimated concentration = (Absorbance + 0.0116)/ 0.0643

3. % Recovery = (Inversely estimated concentration / Known concentration) x 100

4. % C.V. = (S.D./ Mean) X 100

* Each data was determined triplicately

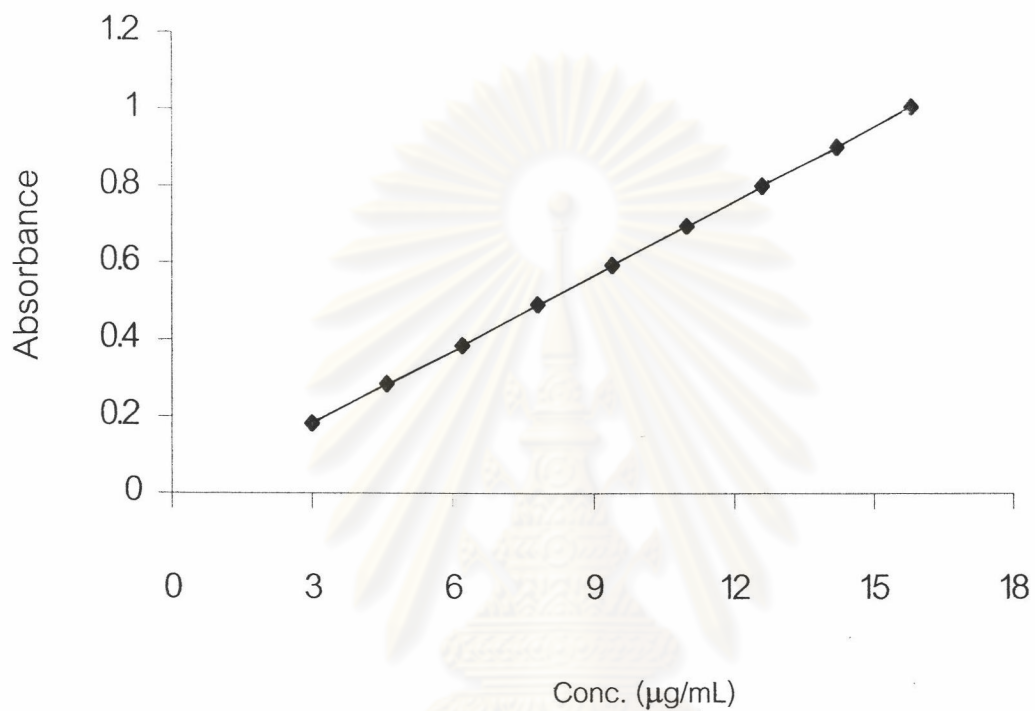


Figure 26. Typical calibration curve for determination of ketoprofen in phosphate buffer pH 7.4 at $\lambda = 260$ nm

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

Table 35. Typical calibration curve data for determination of ketoprofen in 75%methanol estimated using linear regression¹

| Concentration ($\mu\text{g/mL}$) | Absorbance ($\lambda = 260 \text{ nm}$) | Inversely estimated concentration ($\mu\text{g/mL}$) ² . | % Recovery ³ |
|---------------------------------------|----------------------------------------------|--------------------------------------------------------------------------|-------------------------|
| 3 | 0.193 | 3.02 | 100.83 |
| 5 | 0.322 | 4.98 | 99.50 |
| 7 | 0.453 | 6.98 | 99.65 |
| 9 | 0.588 | 9.03 | 100.30 |
| 11 | 0.719 | 11.01 | 100.08 |
| 13 | 0.848 | 12.97 | 99.76 |
| 15 | 0.982 | 15.02 | 100.10 |
| | | Mean | 100.03 |
| | | S.D. | 0.45 |
| | | % C.V. ⁴ | 0.45 |

2. $r^2 = 1$, $Y = 0.0658x - 0.0057$

2. Inversely estimated concentration = (Absorbance + 0.0057) / 0.0658

3. % Recovery = (Inversely estimated concentration / Known concentration) x 100

4. % C.V. = (S.D./ Mean) X 100

* Each data was determined triplicately

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

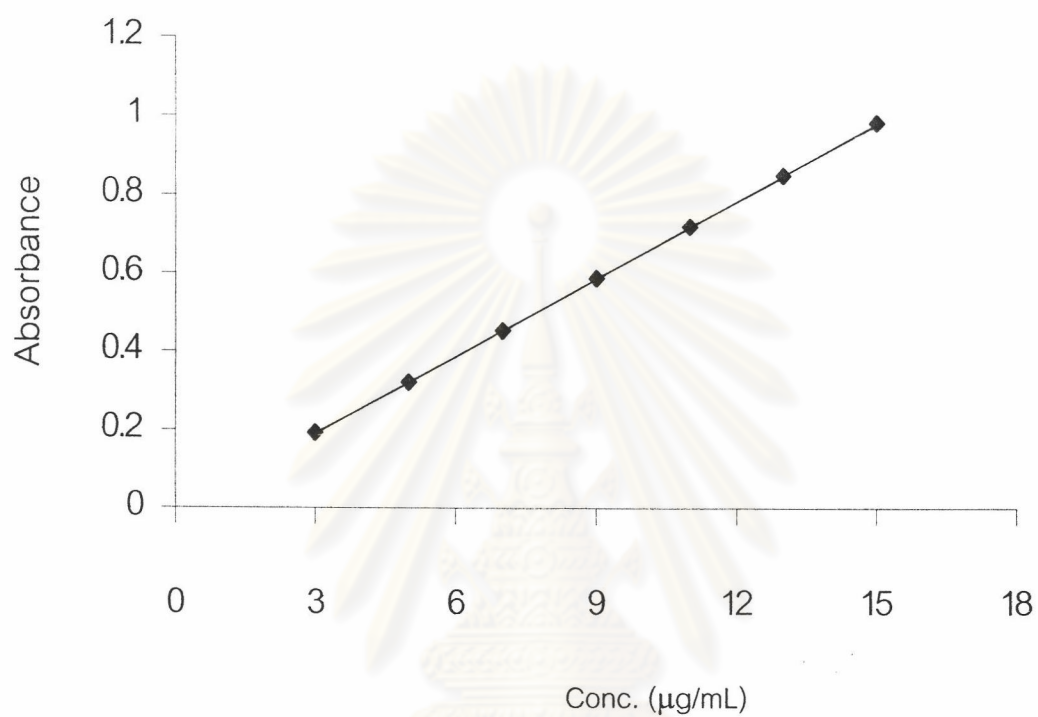


Figure 27. Typical calibration curve for determination of ketoprofen in 75% methanol at 258 nm

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX C

VALIDATION OF ANALYTICAL METHOD FOR STABILITY STUDIES

1. Accuracy

Table 36. Accuracy of analytical method for determination of ketoprofen in preparation

| Concentration ($\mu\text{g/mL}$) | Inversely estimated concentration ($\mu\text{g/mL}$) | % Recovery |
|---------------------------------------|-----------------------------------------------------------|------------|
| 15 | 14.64 | 97.59 |
| 35 | 34.59 | 98.83 |
| 65 | 64.53 | 99.28 |

Mean % recovery = 98.57, S.D. = 0.88, C.V. = 0.89%

* Each data was determined using three determinations per concentration.

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

2. Precision

2.1 Within Run Precision

Table 37. Within run precision of analytical method for determination of ketoprofen in preparation

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 15 | 15.382 \pm 0.130 | 0.85 |
| 35 | 34.883 \pm 0.229 | 0.66 |
| 65 | 64.751 \pm 0.514 | 0.79 |

* Each data was determined using three determinations per concentration.

2.2 Between run precision

Table 38. Between run precision of analytical method for determination of ketoprofen in preparation

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 15 | 15.049 \pm 0.246 | 1.63 |
| 35 | 34.769 \pm 0.303 | 0.87 |
| 65 | 64.510 \pm 0.766 | 1.19 |

* Each data was determined using three determinations per concentration.

Table 39. Linear regression for determination of ketoprofen in preparation at 0, 0.5, 1, 1.5, 2, 2.5 and 3 months

| Time (month) | Linear regression | |
|--------------|------------------------|------------------|
| 0 | $Y = 0.0175X + 0.0075$ | ; $r^2 = 1$ |
| 0.5 | $Y = 0.0169X + 0.0237$ | ; $r^2 = 0.9996$ |
| 1 | $Y = 0.0168X - 0.0074$ | ; $r^2 = 0.9999$ |
| 1.5 | $Y = 0.0151X + 0.0108$ | ; $r^2 = 0.9996$ |
| 2 | $Y = 0.0157X - 0.0008$ | ; $r^2 = 1$ |
| 2.5 | $Y = 0.0162X + 0.0010$ | ; $r^2 = 0.9997$ |
| 3 | $Y = 0.0160X + 0.0029$ | ; $r^2 = 1$ |

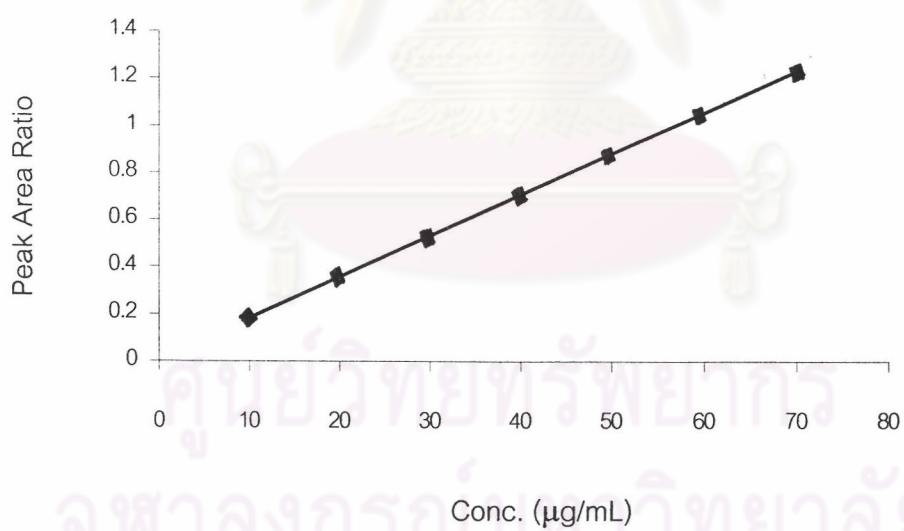


Figure 28. Typical calibration curve for determination of ketoprofen in formulation at 2 months

APPENDIX D

VALIDATION OF ANALYTICAL METHOD FOR *IN VIVO* STUDIES

1. Accuracy

Table 40. Accuracy of analytical method for determination of ketoprofen in rabbit plasma

| Concentration ($\mu\text{g/mL}$) | Inversely estimated concentration ($\mu\text{g/mL}$) | % Recovery |
|---------------------------------------|-----------------------------------------------------------|------------|
| 10 | 10.89 | 108.94 |
| 150 | 156.38 | 104.26 |
| 270 | 276.74 | 102.50 |

Mean % recovery = 105.23, S.D. = 3.33, C.V. = 3.16%

* Each data was determined using three determinations per concentration.

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

2. Precision

2.1 Within run precision

Table 41. Within run precision of analytical method for determination of ketoprofen in rabbit plasma

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 10 | 10.756 \pm 0.508 | 4.73 |
| 150 | 152.896 \pm 3.065 | 2.01 |
| 270 | 273.537 \pm 5.870 | 2.14 |

* Each data was determined using three determinations per concentration.

2.2 Between run precision

Table 42. Between run precision of analytical method for determination of ketoprofen in rabbit plasma

| Concentration ($\mu\text{g/mL}$) | Estimated concentration | % C.V. |
|---------------------------------------|-------------------------|--------|
| | Mean \pm S.D. | |
| 10 | 10.643 \pm 0.400 | 3.76 |
| 150 | 153.308 \pm 3.964 | 2.59 |
| 270 | 267.646 \pm 8.691 | 3.25 |

* Each data was determined using three determinations per concentration.

3. Calibration curve

Table 43. Typical calibration curve data for determination of ketoprofen in rabbit plasma estimated using linear regression¹

| Concentration ($\mu\text{g/mL}$) | PAR | Inversely estimated concentration ($\mu\text{g/mL}$). ² | % Recovery |
|---------------------------------------|--------|-------------------------------------------------------------------------|------------|
| 2 | 0.0156 | 2.29 | 114.54 |
| 40 | 0.315 | 40.70 | 101.75 |
| 80 | 0.617 | 79.39 | 99.24 |
| 120 | 0.928 | 119.30 | 99.42 |
| 160 | 1.250 | 160.52 | 100.33 |
| 200 | 1.557 | 199.89 | 99.95 |
| 240 | 1.883 | 214.71 | 100.71 |
| 280 | 2.181 | 279.97 | 99.99 |
| | | Mean | 102.07 |
| | | S.D. | 5.35 |
| | | % C.V. | 5.27 |

1. $r^2 = 0.9999$, $Y = 0.0078X - 0.0023$

2. Inversely estimated concentration = $(\text{PAR} + 0.0023) / 0.0078$

3. % Recovery = $(\text{Inversely estimated concentration} / \text{Known concentration}) \times 100$

4. % C.V. = $(\text{S.D.} / \text{Mean}) \times 100$

* Each data point was determined triplicately

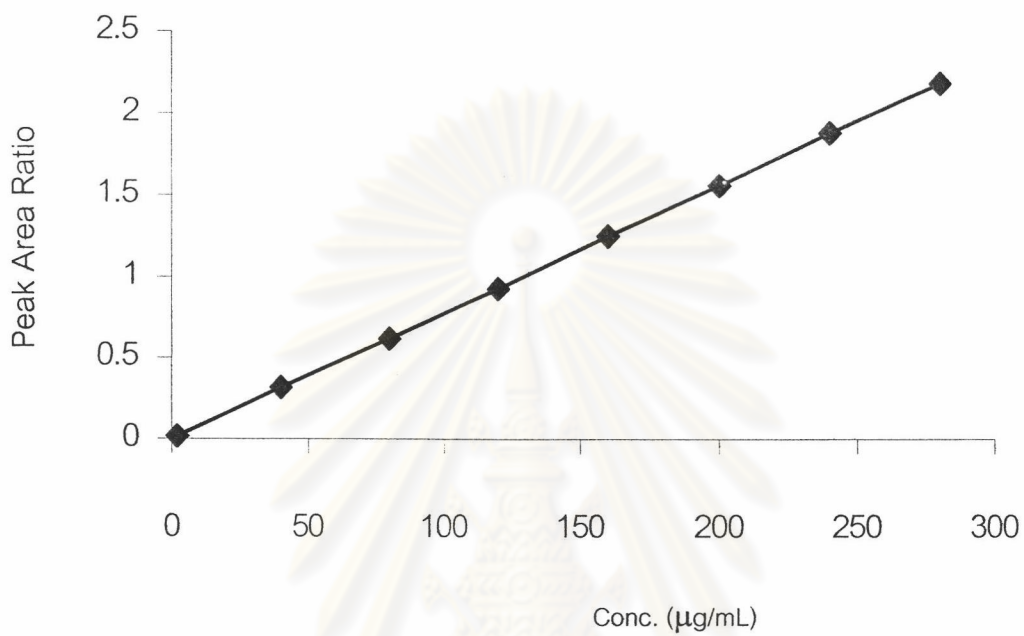


Figure 29. Typical calibration curve data for determination of ketoprofen in rabbit plasma

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

APPENDIX E

STATISTICS

1. Mean (\bar{X})

$$\bar{X} = \sum X/n$$

2. Standard deviation

$$S.D. = \sqrt{\sum (X-\bar{X})^2 / n-1}$$

3. Coefficient of variation (C.V.)

$$C.V. = S.D. / \text{Mean}$$

4. Non-compartment method.

In single dose pharmacokinetic study, blood sampling is stopped at some time (t^*) when drug concentration (C^*) is measurable. Pharmacokinetic parameters are calculated as follow:

4.1 Area under the concentration time curve (AUC).

$$AUC = AUC_{0-t^*} + AUC_{t^*-\infty}$$

$$AUC_{t^*-\infty} = C^*/\lambda$$

Where λ is the slope of the terminal exponential phase of a plot of natural log drug concentration versus time.

4.6 Elimination half life ($t_{1/2}$)

$$t_{1/2} = 0.693 / \lambda$$

5. Analytical of variance for complete randomized block design

In statistic terms the calculation to set up analysis of variance table are as follow:

| Souse of variation | d.f. | Sum of Square | Mean square |
|--------------------|------------------|---------------|---------------|
| Total | $rp - 1$ | SStotal | - |
| Block | $r - 1$ | SSblock | MSblock |
| Formulation | $p - 1$ | SSformulation | MSformulation |
| Error | $(r - 1)(p - 1)$ | SSerror | MSerror |

Where

$$\text{C.T.} = \text{Correction term} = (\sum x)^2 / rp$$

$$p = \text{number of formulation (} p = 2 \text{)}$$

$$r = \text{total number of subjects (} r = 12 \text{)} = \text{number of block}$$

ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

Data presented are individual subject of the C_{max} of ketoprofen after rectal administration of 100 mg ketoprofen coated rectat hard gelatin capsule.

| Subject | 20% Tween 80 [®] | 20% DMI | Sum | Mean |
|---------|---------------------------|----------|----------|---------|
| 1 | 236.123 | 139.864 | 375.987 | 187.993 |
| 2 | 166.063 | 173.430 | 339.492 | 169.746 |
| 3 | 246.541 | 135.863 | 382.405 | 191.202 |
| 4 | 288.993 | 256.093 | 545.086 | 272.543 |
| 5 | 312.352 | 237.603 | 549.955 | 274.978 |
| 6 | 218.767 | 180.144 | 398.911 | 199.456 |
| 7 | 203.094 | 286.025 | 489.119 | 244.560 |
| 8 | 229.586 | 227.073 | 456.659 | 228.330 |
| 9 | 247.637 | 133.734 | 381.371 | 190.685 |
| 10 | 233.435 | 212.183 | 445.618 | 222.809 |
| 11 | 235.907 | 180.811 | 416.718 | 208.359 |
| 12 | 244.240 | 164.678 | 408.919 | 204.459 |
| Sum | 2862.739 | 2327.501 | 5190.240 | |
| Mean | 238.562 | 193.958 | | 216.260 |

1. Correction term = $(5190.240)^2/24 = 1122441.30$
2. SStotal = $[(236.123)^2 + (166.063)^2 + \dots + (408.919)^2] - C.T. = 54370.28$
5. SSreplete = $[(375.987)^2 + (339.492)^2 + \dots + (408.919)^2] / 2 - C.T. = 24667.08$
6. SSformulation = $[(2862.739)^2 + (2327.501)^2] / 12 - C.T. = 11936.64$
7. SSresidual = $0.551 - (0.030 + 0.088 + 0.023 + 0.294) = 17766.56$
8. MS = SS/df

| Source of variation | d.f. | SS | MS | F ratio | F table | Sig.level $\alpha = 0.05$ |
|---------------------|------|----------|----------|---------|---------|------------------------------|
| Total | 23 | 54370.28 | - | - | - | |
| Block | 11 | 24667.08 | 2242.46 | 1.39 | 2.82 | NS |
| Formulation | 1 | 11936.64 | 11936.64 | 7.39 | 4.84 | S |
| Error | 11 | 17766.56 | 1615.14 | - | - | |

Where : F table obtained from the table of F ratio for 0.05 level of significance. The test showed that there are significant differences in C_{max} value in both formulations.

6. Relative bioavailability

$$\text{Relative bioavailability} = (AUC_{\text{test}} / AUC_{\text{ref}}) \times (\text{Dose}_{\text{ref}} / \text{Dose}_{\text{test}}) \times 100$$



ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย

VITA

Miss Mallika Laoweerathum was born on June 7, 1974 in Ayutthaya, Thailand. She received her Bachelor Degree of Science in Pharmacy from the faculty of Pharmacy, Mahidol University, Bangkok, Thailand in 1996. After graduation, she worked in Chaiyapoom Hospital, Bumnejnaron Hospital, Chaiyapoom and Bangpahun Hospital, Ayutthaya in 1996, 1997 and 1999, respectively before entering the Master's Degree program in Pharmacy at Chulalongkorn University.



ศูนย์วิทยทรัพยากร
จุฬาลงกรณ์มหาวิทยาลัย