

## REFERENCES

1. Buchholz, F. L. Preparation and structure of polyacrylate. In R. S. Harland, (ed.), Absorbency Polymer Technology. pp. 23-43. Amsterdam: Elsevier Science, 1990.
2. Masuda, F. Trends in the development of superabsorbent polymers for diaper. In F. L. Buchholz; and N. A. Peppas (eds.), Superabsorbent Polymers: Science and Technology; ACS Symposium Series # 573. pp. 88-98. Washington DC: The American Chemical Society, 1994.
3. Kiremitci, M., and Cukurova, H. Production and highly cross-linked hydrophilic polymer beads: effect of polymerization conditions on partial size and size distribution. Polymer 33(1992): 3257-3261.
4. Shimura, T., and Namba, T. Preparation and application of high-performance superabsorbent polymers. In F. L. Buchholz; and N. A. Peppas (eds.), Superabsorbent Polymers: Science and Technology; ACS Symposium Series # 573. pp.112-127. Washington DC: The American Chemical Society, 1994.
5. Gao, D., and Heimann, R. B. Structure and Mechanical Properties of Superabsorbent Poly(acrylamide)-Montmorillonite Composite Hydrogels. Polymer Gels and Networks 1(1993): 225-246.

6. Kim, U. Y. Recent developments in high water absorbing polymers and applications, In the Proceedings of the International Conference on Recent Developments in Petrochemical and Polymer Technologies. December 12-16, 1989, pp. 6-1 to 6-4. Bangkok, Thailand: Chulalongkorn University Press.
7. Mark, H. F., Bikales, N. M., Overberger, C. G., and Mengel, G. Encyclopedia of Polymer Science and Engineering. Volume 7, 2<sup>nd</sup> ed., pp. 783-802. New York: John Wiley & Sons, 1987.
8. Buchholz, F. L. Superabsorbent Polymer. J. of Chem. Ed. 73(1996): 512-515.
9. Polle, H. Filler for longitudinal sealing of electrical and/or optical cables. EP 357,685 (1991).
10. Kobayashi, H., Okamura, K., Sano, Y., and Shimomura, T. Water stopping agent for cables, water, stopping method using same. and water stopping tape containing same. WO 88/10001 (Aug. 06, 1988).
11. Kinney, A. B., and Scranton, A. B. Formation and structure of cross-linked polyacrylates methods for modeling network formation. In F. L. Buchholz; and N. A. Peppas (eds.), Superabsorbent Polymers: Science and Technology; ACS Symposium Series # 573. pp. 2-29. Washington DC: The American Chemical Society, 1994.

12. Peppas, L. B. Preparation and characterization of cross-linked hydrophilic networks. In R. S. Harland (ed.), Absorbency Polymer Technology. pp. 45-63. Amsterdam: Elsevier Science, 1990.
13. Stanley, F. W., Lamphere, J. C., and Wilson, L. R. Wrinkled absorbent particles of high effective surface area having fast absorption rate. WO 92/16565 (Oct. 1, 1992).
14. Ichikawa, T., and Nakajima, T. Superabsorptive polymers (from natural polysaccharides and peptides). In J.C. Salamone (ed.), Polymeric Materials Encyclopedia (Vol.10). pp. 8051-8059. New York: CRC Press, 1996.
15. Buchholz, F. L. Applications of superabsorbent polymers. In F. L. Buchholz, and A. T. Graham (eds.), Modern Superabsorbent Polymer Technology. pp.251. New York: Wiley-VCH, 1997.
16. Buchholz, F. L. Preparation and structure of polyacrylates. In R. S. Harland, (ed.), Absorbency Polymer Technology. pp. 29-33. Amsterdam: Elsevier Science, 1990.
17. Yoshinaga, K., Nakamura, T. and Itoh, K. Process for producing highly water-absorbent polymers. U.S. Pat 5,185,413 (feb. 9,1993).
18. Chen, C-Y., Chen, C-S., and Kuo, J-F. The multiple-component initiator technique for constant rate radical homopolymerization. Polymer 28(1987): 1396-1402.

19. Irie, Y., Iwasaki, K., Hatsuda, T., Kimura, K., Harada, N., Ishizaki, K., Shimomura, T., and Fujiwara, T. Method for production of hydrophilic polymer from hydrated gel polymer. U.S. Pat 4,920,202 (Apr. 24, 1990).
20. Yano, K., Kajikawa, K., Nagasuna, K., and Irie, Y. Method for the production of absorbent resin. EP 559476 (Aug. 9, 1993).
21. Stevens, M.P. Polymer Chemistry An Introduction. 2<sup>nd</sup> ed., New York: Oxford University Press, 1990, pp. 98-100.
22. Ding, Z. Y.; Aklonis, J. J.; and Salovey R. Model filled polymers. VI. determination of the crosslink density of polymeric beads by swelling. J Polym. Sci., Part B: Polym. Phys. 29 (1991): 1035-1038.
23. Chmelir, M., and Pauen, J. Process for the continuous product of polymers and copolymers of water-soluble monomers. U.S. Pat 4,857,610 (Aug. 15, 1989).
24. Nagasuna, K., Namba, T., Miyake, K., Kimura, K., and Shimomura, T. Production process for water-absorbent resin. U.S. Pat 4,973,632 (Nov. 27, 1990).
25. Masuda, F., Nishida, K., and Nakamura, A. Water absorbing starch resin. U.S. Pat 4,076,663 (Feb. 28, 1978).
26. Heidel, K. Method of manufacturing polysaccharide graft polymers which absorb water and are capable of swelling. U.S. Pat 4,777,232 (Feb. 13, 1987).

27. Buchholz, F. L. Preparation methods of superabsorbent polyacrylates. In F. L. Buchholz; and N. A. Peppas (eds.), Superabsorbent Polymers: Science and Technology; ACS Symposium Series # 573. pp. 27-38. Washington DC: The American Chemical Society, 1994.
28. Trommsdorff, E., and Munzer, M. Polymerization Process, High Polymers Vol. XXIX, Schildknecht, C. E., and Skeist, I. (eds.), pp. 106-124. New York: John Wiley & Sons, 1977.
29. Wagner, M. P. Natural and synthetic silicas in plastics, Additives for Plastics Volume 1, pp. 9-27. Academic Press, 1978.
30. Liu, Z. S., and Rempel, G. L. Preparation of superabsorbent polymer by crosslinking acrylic acid and acrylamide copolymer. J. Appl. Polym. Sci. 64 (1997): 1345-1353.
31. Rangaraj, A., Vangani, V., and Rakshit, A. K. Synthesis and characterization of some water soluble polymer. J. Appl. Polym. Sci. 66(1997): 45-56.
32. Dubrovskii, S. A., Afanaseva, M. V., Lagutina, M. A., and Kazanskii, K. S. Comprehensive characterization of superabsorbent polymer hydrogels. Polymer Bulletin 24 (1990): 107-113.
33. Yao, K-J., and Zhou, W-J. Synthesis and water absorbency of the copolymer of acrylamide with anionic monomer. J. Appl. Polym. Sci. 53 (1994): 1533-1538.

34. Smith, S. J., and Lind, E. F. Superabsorbent polymer having improved absorption rate and absorption under pressure. U.S. Pat 5,399,591 (Mar. 21, 1995).
35. Chen, J., Park, H., and Park, K. Synthesis of superporous hydrogels: hydrogels with fast swelling and superabsorbent properties. J. Biomed. Mater. Res. 44 (1999): 53-62.
36. Zhou, W-J., Yao, K-J., and Kurth, M.J. Studies of crosslinked poly(AM-MSAS-AA) gels. II. Effects of polymerization conditions on the water absorbency. J. Appl. Polym. Sci. 64 (1997): 1009-1014.
37. Karadag, E., Saraydin, D., and Guven, O. Removal of some cationic dyes from aqueous solutions by acrylamide/itaconic acid hydrogels. Water, Air, and Soil Pollution 106(1998): 369-378.
38. Chen, J., and Zhao, Y. Relationship between water absorbency and reaction condition in aqueous solution polymerization of acrylate superabsorbents. J. Appl. Polym. Sci. 75 (2000): 808-814.
39. Wu, J., Wei, Y., Lin, J., and Lin, S. Study on starch-graft-acrylamide/mineral powder superabsorbent composite. Polymer 44(2003): 6513-6520.
40. Kourosh, K. and Zohuriaan-Mehr, M. J. Superabsorbent Hydrogel Composites. Polym. Adv. Technol. 14 (2003): 438-444.

41. Chen, J. and Park, K. Synthesis and characterization of superporous hydrogel composites. J. Control. Rel. 65 (2000): 73-82
42. Uyanik, N., and Erbil, C. Monomer reactivity ratios of itaconic acid and acrylamide copolymers determined by using potentialmetric titration method. Eur. Polym. J. 36(2000): 2651-2654.
43. Brandrup, J., and Immergut, E.H. (eds), Polymer Handbook, 4nd ed., pp VI/201. New York: John Wiley&Sons, 1999.



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**APPENDIX**

ศูนย์วิทยทรัพยากร  
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## APPENDIX A

### Absorbency under load

Absorbency under load (AUL) measures the ability of a polymer to absorb fluid while under a static load and can be considered as a measurement of gel stability or gel strength. Similar information has been collected on water-swelling gels by measurements of the swelling pressure, which is the pressure generated by the gel in contact with an external source of fluid while being confined to a given volume. The test has been developed to measure under conditions that simulate those in the actual personal-use application. For example, the polymer in a diaper will occasionally be under a compressive load when a baby sits or lies on the diaper.

A porous filter plate is placed in a petri dish and fluid (such as 0.9% sodium chloride solution or water) is added so that the liquid level is equal to the top of the filter plate. A filter paper is placed on the filter plate and allowed to thoroughly wet with fluid. Polymer (0.16 gram) is carefully scattered onto the filter screen of the test device (a flexiglass cylinder with 100 mesh stainless steel cloth in the bottom: cylinder diameter = 26 mm, height = 35 mm). A piston assembly, including additional weight to achieve a load, is placed on top of the polymer. After weighing the assembled device, it is placed on the filter plate, and absorption is allowed for 1 hour. After 1 hr, the entire device is reweighed and the absorbency under load is calculated by the following formula:

$$\text{AUL} = \frac{(\text{mass of cylinder group after suction} - \text{mass of cylinder group dry})}{(\text{initial sample mass of the polymer})}$$

## VITAE

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