

# การสร้างสารบัญชิตยภูมิในเซลล์เพาะเลี้ยงของมะหาด

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วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรดุษฎีบัณฑิต  
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SECONDARY METABOLITE PRODUCTION IN CELL CULTURES OF  
*ARTOCARPUS LAKOOCHA*

Mrs. Suthira Maneechai

A Thesis Submitted in Partial Fulfillment of the Requirements  
for the Degree of Doctor of Philosophy Program in Pharmacognosy  
Department of Pharmacognosy and Pharmaceutical Botany  
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สุธีรา มณีฉาย : การสร้างสารทุติยภูมิในเซลล์เพาะเลี้ยงของมะหาด.

(SECONDARY METABOLITE PRODUCTION IN CELL CULTURES OF *ARTOCARPUS LAKOOCHA*) อ. ที่ปรึกษาวิทยานิพนธ์หลัก : ศ.ภก.ดร. กิตติศักดิ์ ลิขิตวิทยาวุฒิ, อ. ที่ปรึกษาวิทยานิพนธ์ร่วม : รศ.ดร. วันชัย ดีเอโกนามกุล, 173 หน้า.

การศึกษาการสร้างสารทุติยภูมิในเซลล์เพาะเลี้ยงของมะหาด โดยทำการศึกษาในแคลลัส ที่มาจากการชักนำให้เกิดเนื้อเยื่อเพาะเลี้ยงโดยใช้ส่วนลำต้น ใบเลี้ยง และใบอ่อนของพืชที่มาจาก การงอกโดยการเพาะเมล็ดในสภาพปราศจากเชื้อ ภายใต้สภาวะของอาหารสูตร Woody Plant Medium ที่ประกอบด้วย 2,4-D 1 มิลลิกรัมต่อลิตร BA 1 มิลลิกรัมต่อลิตร และน้ำตาล 20 กรัมต่อ ลิตรสามารถแยกสารบริสุทธิ์ได้ 13 ชนิด โดยเป็นสารกลุ่ม flavonoids and stilbenoids ชนิดใหม่ 4 ชนิด คือ 5,7,4'-trihydroxy-2'-methoxy-3-( $\gamma,\gamma$ -dimethylallyl)-flavone, 7,2',4'-trihydroxy-3'',4''- dehydropyrano [1'',4'':5,6] flavone, *trans*-2,4,5'-trihydroxy-3'-methoxy-4'-(3-methyl-*E*-but-1-enyl)-stilbene และ 3-geranyl-2,4,3',5'-tetrahydroxystilbene และอีก 9 ชนิดเป็นสารที่มีการรายงาน มาแล้วคือ aesculin, (-)-catechin, albanin A, isoartocarpesin, norartocarpin, cudraflavone B, cudraflavone C, artotonkin และ chlorophorin การพิสูจน์โครงสร้างทางเคมีของสารที่แยกได้นี้ อาศัย การวิเคราะห์สเปกตรัม ร่วมกับการเปรียบเทียบข้อมูลของสารที่รายงานมาแล้ว เมื่อนำสารบริสุทธิ์ที่ แยกได้ไปทำการทดสอบฤทธิ์ในการจับสารอนุมูลอิสระและฤทธิ์ยับยั้งเอนไซม์ tyrosinase พบว่า สารในกลุ่ม stilbenoids มีฤทธิ์แรงในการจับสารอนุมูลอิสระและฤทธิ์ยับยั้งเอนไซม์ tyrosinase ส่วนสารชนิดอื่นมีฤทธิ์อ่อนข้างต่ำ นอกจากนี้ได้พัฒนาวิธีวิเคราะห์สารด้วย TLC densitometry เพื่อวิเคราะห์ปริมาณสาร oxyresveratrol ในแก่นมะหาดและปวกหาด และเพื่อนำไปใช้ในการ วิเคราะห์สารทุติยภูมิในเซลล์เพาะเลี้ยงของมะหาด เมื่อทำการศึกษาความสัมพันธ์ของการเจริญ ของแคลลัสและการสร้างสารทุติยภูมิพบว่า แคลลัสมีการสร้างสารทั้ง 13 ชนิดในทุกช่วงของการ เจริญ มีรูปแบบการสร้างสารเหมือนกันแตกต่างกันเพียงปริมาณสารสำคัญที่สร้าง โดยมีการสร้างสาร ทุติยภูมิที่คู่ขนานกับการเจริญเติบโตของแคลลัส ซึ่งมีปริมาณสูงสุดในระยะก่อนเข้า stationary phase (ระยะ 15 - 17 วัน) และลดลงอย่างรวดเร็วเมื่อนำเนื้อเยื่อเข้าสู่ระยะ stationary phase (ระยะ 20 - 25 วัน)

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SUTHIRA MANEECHAI : SECONDARY METABOLITE PRODUCTION  
IN CELL CULTURES OF *ARTOCARPUS LAKOOCHA*

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173 pp.

A study on the secondary metabolite production in cell cultures of *Artocarpus lakoocha* Roxb. was initiated by inducing callus formation from aerial part explants on Woody Plant Medium (WPM), which contained 1 mg/l 2,4-dichlorophenoxyacetic acid (2,4-D), 1 mg/l benzyladenine (BA) and 20 g/l sucrose. A chemical study of the callus cultures of *A. lakoocha* led to the isolation of thirteen pure compounds, including four new compounds, namely, 5,7,4'-trihydroxy-2'-methoxy-3-( $\gamma,\gamma$ -dimethylallyl)-flavone, 7,2',4'-trihydroxy-3'',4''-dehydropyrano [1'',4'':5,6] flavone, *trans*-2,4,5'-trihydroxy-3'-methoxy-4'-(3-methyl-*E*-but-1-enyl)-stilbene and 3-geranyl-2,4,3',5'-tetrahydroxystilbene together with nine known compounds including aesculin, (-)-catechin, albanin A, isoartocarpesin, norartocarpin, cudraflavone B, cudraflavone C, artotonkin and chlorophorin. Their structures were elucidated through analysis of their spectroscopic data and by comparison with previously reported data. The isolated compounds were evaluated for free radical scavenging and antityrosinase activities. Stilbenoids showed potent free radical scavenging activity and tyrosinase inhibitory activity. A thin-layer chromatography (TLC) densitometric method was also developed for the determination of oxyresveratrol content in *A. lakoocha* heartwood and in the traditional drug 'Puag-Haad', and the secondary metabolites in plant cell cultures of *A. lakoocha*. A study on the growth-product relationship during the 25-day growth of *A. lakoocha* callus cultures showed that all the isolated compounds were formed over the period of the culture growth. The obtained chemical profiles appeared to proceed parallelly to the growth curve of the callus, with the maximum content in the period of progressive deceleration phase (day 15 - day 17) followed by a rapid decline in the stationary phase (day 20 - day 25).

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Field of Study : Pharmacognosy ..... Advisor's Signature .....

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**LIST OF ABBREVIATIONS AND SYMBOLS**

$[\alpha]_D^{20}$	= Specific rotation at 20°C and Sodium D line (589 nm)
$\alpha$	= Alpha
Acetone- $d_6$	= Deuterated acetone
ax	= Axial
$\beta$	= Beta
br	= Broad (for NMR spectra)
C	= Concentration
°C	= Degree Celsius
calcd	= Calculated
CD	= Circular Dichroism
$CDCl_3$	= Deuterated chloroform
$CH_2Cl_2$	= Dichloromethane
cm	= Centimeter
$^{13}C$ NMR	= Carbon-13 Nuclear Magnetic Resonance
$^1H$ - $^1H$ COSY	= Homonuclear (Proton-Proton) Correlation Spectroscopy
1-D	= One dimensional (for NMR spectra)
2-D	= Two dimensional (for NMR spectra)
d	= Doublet (for NMR spectra)
dd	= Doublet of doublets (for NMR spectra)
DPPH	= 1,1-Diphenyl-2-picrylhydrazyl
$\delta$	= Chemical shift
ED <sub>50</sub>	= 50% Effective Dose
eq	= Equatorial
ESI-MS	= Electrospray Ionization Mass Spectrometry
EtOAc	= Ethyl acetate
FCC	= Flash Column Chromatography
g	= Gram
GF	= Gel Filtration Chromatography
hr	= Hour
$^1H$ -NMR	= Proton Nuclear Magnetic Resonance



**LIST OF ABBREVIATIONS AND SYMBOLS (continued)**

HMBC	= $^1\text{H}$ -detected Heteronuclear Multiple Bond Correlation
HMQC	= $^1\text{H}$ -detected Heteronuclear Multiple Quantum Coherence
HR- ESI-MS	= High Resolution Electrospray Ionization Mass Spectrometry
HSQC	= Heteronuclear Single Quantum Coherence
HSV-1	= Herpes Simplex Virus type 1
HSV-2	= Herpes Simplex Virus type 2
Hz	= Hertz
IC <sub>50</sub>	= Concentration showing 50% inhibition
IR	= Infrared
<i>J</i>	= Coupling constant
KBr	= Potassium bromide
Kg	= Kilogram
L	= Liter
L-DOPA	= L-3,4-dihydroxyphenylalanine
μg	= Microgram
μl	= Microliter
μM	= Micromolar
λ <sub>max</sub>	= Wavelength at maximal absorption
ε	= Molar absorptivity
M <sup>+</sup>	= Molecular ion
<i>m</i>	= Meta
m	= Multiplet (for NMR spectra)
MeOH	= Methanol
mg	= Milligram
[M+H] <sup>+</sup>	= Protonated molecular ion
MHz	= Megahertz
min	= Minute
ml	= Milliliter
mm	= Millimeter

**LIST OF ABBREVIATIONS AND SYMBOLS (continued)**

mM	= Millimolar
<i>m/z</i>	= Mass to charge ratio
MS	= Mass spectrum
mult.	= Multiplicity
MW	= Molecular weight
NaH <sub>2</sub> PO <sub>4</sub>	= Sodium dihydrogen phosphate
Na <sub>2</sub> HPO <sub>4</sub>	= Disodium hydrogen phosphate
nm	= Nanometer
NMR	= Nuclear Magnetic Resonance
NOESY	= Nuclear Overhauser Effect Spectroscopy
<i>o</i>	= Ortho
<i>p</i>	= Para
ppm	= Part per million
q	= Quartet (for NMR spectra)
s	= Singlet (for NMR spectra)
spp.	= Species
t	= Triplet (for NMR spectra)
TLC	= Thin Layer Chromatography
UV	= Ultraviolet
UV-VIS	= Ultraviolet and Visible spectrophotometry
VLC	= Vacuum Liquid Column Chromatography
$\nu_{\max}$	= Wave number at maximal absorption
2,4-D	= 2,4-Dichlorophenoxy acetic acid
BA or BAP	= <i>N</i> <sub>6</sub> -benzyladenine or <i>N</i> <sub>6</sub> -benzylaminopurine
DW	= Dry weight
FW	= Fresh weight

## CHAPTER I

### INTRODUCTION

The genus *Artocarpus* belongs to the family Moraceae of the order Urticales. This genus consists of 50 species, widely distributed in Sri Lanka, India, Pakistan, Myanmar, Thailand, Indo-china, South-China, Malaysia and Solomon Islands (Kochummen, 1978; Gardner, Sidisunthorn and Anusarnsunthorn, 2000).

Plants in the genus *Artocarpus* are often large trees. They are mostly evergreen trees with abundant white latex in all parts. The leaves are alternate, coriaceous, entire or pinnately lobe rarely pinnate. They have only one main vein from the base. The flowers are minute, tightly packed in oblong or globose heads. Male and female flowers are packed in separate heads. The male flowers have 2-4 sepals and 1 stamen. On the other hand, the female flowers have narrow tubular calyx fused with neighbouring flowers at the base. They consist of 1 slender style and 1 or 2 stigmas that are equal or unequal. The bracts are fused with the flowers to form a syncarp (fruit). The syncarps are irregularly globose or fist-shaped with many oblong seeds (J.D. Hooker, 1890; Gardner *et al.*, 2000; Wu *et al.*, 2003).

The species of *Artocarpus* in Thailand according to Smitinand (2001) are as follows.

<i>Artocarpus altilis</i> (Parkinson) Fosberg ( <i>A. communis</i> J.R. & G. Forst., <i>A. incisa</i> Linn. f.)	ขนุนลำปะลอส Khanun sampalo (Central); สาเก Sake (Central); Bread fruit tree; Bread nut tree.
<i>A. altissimus</i> J.J. Smith	ไสน Sanai (Surat Thani).
<i>A. chaplasha</i> Roxb.	หาดसान Haat san (Chiang Rai).
<i>A. dadah</i> Miq.	หังกัน Thang khan; ม่วงกวาง Muang kwang, (Yala); หาดรัม Hat rum, หาดลูกใหญ่, Hat luk yai (Trang); หาดขน Hat khon (Narathiwat).
<i>A. elasticus</i> Reinw. ex Blume	กะออก Ka ok, กะเอาะ Ka-o (Peninsular); ตีอกะ Tue-ka (Malay-Yala); เอาะ O (Trang, Ranong).
<i>A. gomezianus</i> Wall. Ex Trécul	ตะปั้ง Ta pang, ต้าปั้ง Tam-pang (Malay- Peninsular); หาดหนูน Hat nun (Northern); อีโป I po (Trang).
<i>A. heterophyllus</i> Lamk.	ขนุน Khanun (General); ขะนู Kha-nu (Chong-

<i>(A. integrifolia</i> Linn. f.)	Chanthaburi); ะเนอ Kha-noe (Khmer); ซึคีย Si-khue, ะหน้อย Pa-noi (Karen-Maehongson); นะยวชะ Na-yuai-sa (Karen-Kanchanaburi); นากอ Na-ko (Malay-Pattani); เนน Nen (Chaobon-Nakhonratchasima); มะหนุน Manun (Northern, Peninsular); ล้าง, ลาง Lang (Shan-Northern); หมักหมี่ Makmi (Northeastern); หมากกลาง Mak lang (Shan-Maehongson); Jack fruit tree.
<i>A. kemando</i> Miq.	ขนุนป่า Khanun pa (Narathiwat); ยาดู Yatu (Malay-Narathiwat)
<i>A. integer</i> (Thunb.) Merr.	จำปาตะ Champada (General); จำปาเดาะ Champado (Peninsular); Champedak
<i>A. lacucha</i> Roxb. ( <i>A. lakoocha</i> Roxb.)	กาเย Kaa-yae, ตาเป Ta-pae, ตาเปง Ta-pang (Malay-Narathiwat); มะหาด Mahat (Peninsular); มะหาดใบใหญ่ Mahat bai yai (Trang); หาด
<i>A. lanceifolius</i> Roxb.	ขนุนป่า Khanun pa (Peninsular); นังกาปีโต Nang-ka-pi-to, นังกาปีปิต Nang-ka-pi-pit (Malay-Peninsular); นังกาปีเป๊ะ Nang-ka-pi-pae (Malay-Narathiwat).
<i>A. nitidus</i> Tréc	มะหาดข่อย Mahat khoi (Surat Thani).
subsp. <i>lingnanensis</i> Jarrett ( <i>A. parva</i> Gagnep.)	
<i>A. rigidus</i> Blume subsp. <i>rigidus</i>	ขนุนป่า Khanun pa (Peninsular)
<i>A. rigidus</i> Bl. subsp. <i>asperulus</i> Jarrett. ( <i>A. calophyllus</i> Kurz)	ขนุนปาน Khanun pan (Surat Thani)

*Artocarpus lakoocha* Roxb. is an indigenous plant known in Thai as Mahaad. It is a large deciduous tree reaching 15-18 m. The bark is red brown to dark brown, which becomes rough and scaly with age. The branchlets are 3-6 mm. thick. The stipules are ovate-lanceolate, 4-5 cm. in length. The leaves (5-15 cm. x 10-15 cm.) are alternate. Their shape is oval or obovate. The margin of leaves are untoothed or with minute teeth. They have 10-18 secondary veins on each side of the midvein. The leaf texture is leathery. The young red brown shoots are hairy. The stalks are finely hairy, 1.4-3.3 cm. long. The twigs are rather stout without ring scars. The flower heads, which grow from leaf axils, are dirty yellow to pale pink or orange. The male heads (0.8-2 cm.) with globular stalks, while the female heads (1.2-2.3 cm.) are oval or oblong, with 2.5-3 cm. stalks. The fruits are 2.5-8 cm. with smooth velutin surface. Their fresh are pale yellow or orange that change to reddish

brown when dried. The seeds are ellipsoid with bent bristles (Hooker, 1890; Gardner *et al.*, 2000; Wu *et al.*, 2003).

Cell culture technology has been developed as a possible tool to both study and produce plant secondary metabolites (Zhong, J.J., 2001). Previous reports of *Artocarpus* plants have been focused mainly on micropropagation, for example *A. lakoocha* (Janick, J and Whipkey, A., eds. 2002), *A. altilis* (Murch, S.J. et al., 2008) and *A. heterophyllus* (Lee, C.L. and Keng, C.L., 2005). No reports of secondary metabolite production from cell cultures of *A. lakoocha* have been discussed. Previous phytochemical studies of stem heartwood and root of this plant have revealed the presence of triterpenoids, flavonoids and stilbenes, some of which possessed biological activities such as antiherpetic, antioxidant and tyrosinase inhibitory activities (Jagtap and Bapat, 2010).

The purpose of this research is to study the secondary metabolite production from *A. lakoocha* cell cultures. The objectives of the research are:

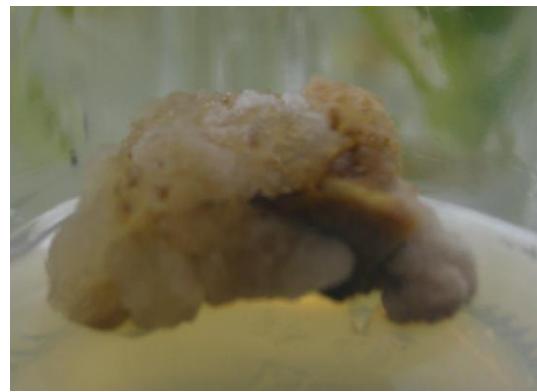
1. To develop a simple, reliable and efficient method for the quantitative analysis of oxyresveratrol in plant extracts and for quantitative analysis in plant cell cultures.
2. To study methods for producing cell cultures of *A. lakoocha*.
3. To isolate and purify compounds from cell cultures of *A. lakoocha*.
4. To determine the chemical structure of each isolated compound.
5. To evaluate each isolated compound for its free radical scavenging activity, and tyrosinase inhibition potential.



A



B



C

**Figure 1** *Artocarpus lakoocha* Roxb.

A) Tree

B) Fruit

C) Callus culture

## CHAPTER II

### HISTORICAL

#### 1. Chemical Constituents of *Artocarpus* spp.

A number of compounds have been isolated from the genus *Artocarpus*. They can be classified as flavonoids, triterpenoids, steroids, stilbenoids and miscellaneous substances (Tables 1-4).

**Table 1 Distribution of flavonoids in the genus *Artocarpus***

Plant and chemical compound	Plant part	Reference
<i>Artocarpus altilis</i>		
Apigenin [1]	Heartwood	Shimizu <i>et al.</i> , 1998
Artobiloxanthone [2] (artocarpus flavone KB-1)	Bark	Aida <i>et al.</i> , 1997
Artocarpesin [3]	Heartwood	Shimizu <i>et al.</i> , 1998
Artocarpin [4]	Heartwood	Venkataraman, 1972
Artocarpus chalcone AC-3-1 [5]	Flower Leaves	Fujimoto <i>et al.</i> , 1987a Wang <i>et al.</i> , 2007b
Artocarpus chalcone AC-3-3 [6]	Flower	Fujimoto <i>et al.</i> , 1987a
Artocarpus chalcone AC-5-1 [7]	Flower Leaves	Fujimoto <i>et al.</i> , 1987a Wang <i>et al.</i> , 2007b
Artocarpus chalcone I [8]	Flower	Fujimoto, Agusutein, and Made, 1987b
Artocarpus flavanone AC-3-2 [9]	Flower	Fujimoto <i>et al.</i> , 1987a
Artocarpus flavanone AC-5-2 [10]	Flower Leaves	Fujimoto <i>et al.</i> , 1987a Wang <i>et al.</i> , 2007b
Artocarpus flavone KB-1 [2] (Artobiloxanthone)	Bark	Fujimoto <i>et al.</i> , 1990
Artocarpus flavone KB-2 [11]	Bark	Fujimoto <i>et al.</i> , 1990
Artocarpus flavone KB-3 [12] (Artonin E)	Bark	Fujimoto <i>et al.</i> , 1990
Artochamin B [13]	Root cortex	Lin <i>et al.</i> , 2006
Artochamin D [14]	Root cortex	Lin <i>et al.</i> , 2006

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Artocommunol CB [15]	Root cortex	Chan, Ko, and Lin, 2003
Artocommunol CC [16]	Root cortex	Lin <i>et al.</i> , 2006
Artocommunol CD [17]	Root cortex	Chan <i>et al.</i> , 2003
Artocommunol CE [18]	Root cortex	Chan <i>et al.</i> , 2003
Artoflavone A [19]	Root cortex	Lin <i>et al.</i> , 2009
Artomunoflavanone [20]	Root cortex	Lin <i>et al.</i> , 2006
Artomunoisoxanthone [21]	Root cortex	Lin <i>et al.</i> , 2006
Artomunoxantrione [22]	Root bark	Shieh and Lin, 1992
Artomunoxanthone [23]	Root bark	Shieh and Lin, 1992
Artomunoxanthotrione epoxide [24]	Root bark	Lin <i>et al.</i> , 1992
Artonin E [12] (Artocarpus flavone KB-3)	Bark	Hano <i>et al.</i> , 1990d
Artonin F [25]	Bark	Hano <i>et al.</i> , 1990d
Artonin K [26]	Bark	Aida <i>et al.</i> , 1997
Artonin V [27]	Root bark	Hano, Inami and Nomura, 1994
Artonol A [28]	Bark	Aida <i>et al.</i> , 1997
Artonol B [29]	Bark	Aida <i>et al.</i> , 1997
Artonol C [30]	Bark	Aida <i>et al.</i> , 1997
Artonol D [31]	Bark	Aida <i>et al.</i> , 1997
Artonol E [32]	Bark	Aida <i>et al.</i> , 1997
Cudraflavone A [33] (Isocyclomorusin)	Root bark Heartwood	Shieh and Lin, 1992 Han <i>et al.</i> , 2006
Cudraflavone C [34]	Heartwood	Han <i>et al.</i> , 2006
Cycloaltilislin [35]	Stem Leaves	Chen <i>et al.</i> , 1993 Wang <i>et al.</i> , 2007b



**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Cycloaltilisin 6 [36]	Bud cover	Patil <i>et al.</i> , 2002
Cycloaltilisin 7 [37]	Bud cover	Patil <i>et al.</i> , 2002
Cycloartobiloxanthone [38]	Bark	Hano <i>et al.</i> , 1990d
Cycloartocarpin [39]	Heartwood	Venkataraman, 1972
Cycloartomunin [40]	Root bark	Lin and Shieh, 1991
Cycloartomunoxanthone [41]	Root bark	Lin and Shieh, 1991
Cyclocommunin [42]	Root bark	Lin and Shieh, 1991
Cyclocommunol [43]	Root bark	Lin and Shieh, 1991
Cyclocommunomethonol [44]	Root cortex	Lin <i>et al.</i> , 2006
Cyclogeracommunin [45]	Root cortex	Lin <i>et al.</i> , 2009
Cyclomorusin [46]	Stem Root	Chen <i>et al.</i> , 1993 Lin and Shieh, 1991
Cyclomulberrin [47]	Stem Root bark	Chen <i>et al.</i> , 1993 Lin and Shieh, 1992
Dihydroartomunoxanthone [48]	Root cortex	Lin <i>et al.</i> , 2006
Dihydrocycloartomunin [49]	Root bark	Lin and Shieh, 1991
Dihydroisocycloartomunin [50]	Root bark	Lin and Shieh, 1992
Dihydromorin [51]	Heartwood	Shimizu <i>et al.</i> , 1998
Engeletin [52]	Stem	Chen <i>et al.</i> , 1993
(2 <i>S</i> )-Euchrenone a7 [53]	Heartwood	Han <i>et al.</i> , 2006
Gemichalcone B [54]	Heartwood	Han <i>et al.</i> , 2006
Gemichalcone C [55]	Heartwood	Han <i>et al.</i> , 2006
3'-Geranyl-2',3,4,4'- tetrahydroxy-chalcone [56]	Leaves	Shimizu <i>et al.</i> , 2000
Hydroxyartocarpin [57]	Stem bark	Shamaun <i>et al.</i> , 2010

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Isoartocarpesin [58]	Heartwood	Shimizu <i>et al.</i> , 1998
Isobacachalcone [59]	Heartwood	Han <i>et al.</i> , 2006
Isocyclomorusin [33] (Cudraflavone A)	Stem	Chen <i>et al.</i> , 1993
Isocyclomulberrin [42] (Cyclocommunin)	Stem	Chen <i>et al.</i> , 1993
Isolespeol [60]	Leaves	Fang <i>et al.</i> , 2008b
Morin [61]	Heartwood	Venkataraman, 1972
Morachalcone A [62]	Heartwood	Han <i>et al.</i> , 2006
Morusin [63]	Stem bark	Fujimoto <i>et al.</i> , 1990
(+)-Norartocarpanone [64]	Heartwood	Shimizu <i>et al.</i> , 1998
Norartocarpetin [65]	Heartwood	Venkataraman, 1972
1-(2,4-Dihydroxyphenyl)-3-[3,4-dihydro-3,8-dihydroxy-2-methyl-2-(4-methyl-3-pentenyl)-2H-1-benzopyran-5-yl]-1-propanone [66]	Leaves	Wang <i>et al.</i> , 2007b
1-(2,4-Dihydroxyphenyl)-3-[8-hydroxy-2-methyl-2-(3,4-epoxy-4-methyl-1-pentenyl)-2H-1-benzopyran-5-yl]-1-propanone [67]	Leaves	Wang <i>et al.</i> , 2007b
1-(2,4-Dihydroxyphenyl)-3-[8-hydroxy-2-methyl-2-(4-hydroxy-4-methyl-2-pentenyl)-2H-1-benzopyran-5-yl]-1-propanone [68]	Leaves	Wang <i>et al.</i> , 2007b
1-(2,4-Dihydroxyphenyl)-3-{4-hydroxy-6,6,9-trimethyl-6a,7,8,10a tetrahydro-6H-dibenzo [b,d]pyran-5-yl}-1-propanone [69]	Leaves	Wang <i>et al.</i> , 2007b
3'',3''-Dimethylpyrano[3',4']2,4,2'-trihydroxy chalcone [70]	Heartwood	Han <i>et al.</i> , 2006

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
5'-Geranyl-2',4',4'-trihydroxychalcone [71]	Leaves	Fang <i>et al.</i> , 2008
2-[6-Hydroxy-3,7-dimethylocta-2( <i>E</i> ),7-dienyl]-2',3,4,4'-tetrahydroxy dihydrochalcone [72]	Leaves	Wang <i>et al.</i> , 2007b
3,4,2',4'-Tetrahydroxy-3'-geranyldihydrochalcone [73]	Leaves	Fang <i>et al.</i> , 2008
<b><i>Artocarpus bracteata</i></b>		
Artoindonesianin J [74]	Root and stem bark	Ersam <i>et al.</i> , 2002
Carpachromene [75]	Root and stem bark	Ersam <i>et al.</i> , 2002
Kozonol C [76]	Root and stem bark	Ersam <i>et al.</i> , 2002
6-Prenylalpigenin [77]	Root and stem bark	Ersam <i>et al.</i> , 2002
<b><i>Artocarpus champeden</i></b>		
Artocarpone A [78]	Stem bark	Widyawaruyanti <i>et al.</i> , 2007
Artocarpone B [79]	Stem bark	Widyawaruyanti <i>et al.</i> , 2007
Artoindonesianin A [80]	Root	Hakim <i>et al.</i> , 1999
Artoindonesianin A-2 [81]	Stem bark Heartwood	Syah <i>et al.</i> , 2006b Widyawaruyanti <i>et al.</i> , 2007
Artoindonesianin A-3 [82]	Heartwood	Syah <i>et al.</i> , 2006b
Artoindonesianin B [83]	Root	Hakim <i>et al.</i> , 1999
Artoindonesianin E [84]	Stem bark	Widyawaruyanti <i>et al.</i> , 2007
Artoindonesianin M [85]	Heartwood	Syah <i>et al.</i> , 2002a
Artoindonesianin Q [86]	Heartwood	Syah <i>et al.</i> , 2002b

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Artoindonesianin R [87]	Heartwood	Syah <i>et al.</i> , 2002b
Artoindonesianin S [88]	Heartwood	Syah <i>et al.</i> , 2002b
Artoindonesianin T [89]	Heartwood	Syah <i>et al.</i> , 2002b
Artoindonesianin U [90]	Heartwood	Syah <i>et al.</i> , 2004
Artoindonesianin V [91]	Heartwood	Syah <i>et al.</i> , 2004
Artonin A [92]	Root Stem bark	Hakim <i>et al.</i> , 1999 Widyawaruyanti <i>et al.</i> , 2007
Artonin B [93]	Heartwood	Syah <i>et al.</i> , 2004
Artopeden A [94]	Bark	Wahyuni <i>et al.</i> , 2009
Cyclochampedol [95]	Stem bark	Achmad <i>et al.</i> , 1996 Parenti <i>et al.</i> , 1998
Cyclocommunin [42]	Heartwood	Syah <i>et al.</i> , 2004
Cycloheterophyllin [96]	Stem bark	Widyawaruyanti <i>et al.</i> , 2007
Heteroflavanone C [97]	Stem bark	Widyawaruyanti <i>et al.</i> , 2007
Heterophyllin [98]	Stem bark	Widyawaruyanti <i>et al.</i> , 2007
5'-Hydroxycudraflavone A [99]	Heartwood	Syah <i>et al.</i> , 2004
<b><i>Artocarpus chaplasha (A. chama)</i></b>		
Artocarpesin [3]	Heartwood	Rao, Rathi and Venkataraman, 1972
Artocarpin [4]	Heartwood Root	Rao <i>et al.</i> , 1972 Wang <i>et al.</i> , 2004
Artochamin A [100]	Root	Wang <i>et al.</i> , 2004
Artochamin B [13]	Root	Wang <i>et al.</i> , 2004
Artochamin C [101]	Root	Wang <i>et al.</i> , 2004

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Artochamin D [14]	Root	Wang <i>et al.</i> , 2004
Artochamin E [102]	Root	Wang <i>et al.</i> , 2004
Artonin A [92]	Root	Wang <i>et al.</i> , 2004
Artonin E [12] (Artocarpus flavone KB-3)	Root	Wang <i>et al.</i> , 2004
Artonin U [103]	Root	Wang <i>et al.</i> , 2004
Chaplashin [104]	Heartwood	Rao <i>et al.</i> , 1972
Cudraflavone A [33] (Isocyclomorusin)	Root	Wang <i>et al.</i> , 2004
Cycloartobiloxanthone [38]	Root	Wang <i>et al.</i> , 2004
Cycloartocarpesin [105]	Heartwood	Rao <i>et al.</i> , 1972
Cycloartocarpin [39]	Heartwood	Rao <i>et al.</i> , 1972
Cycloartocarpin A [106]	Root	Wang <i>et al.</i> , 2004
5'-Hydroxycudraflavone A [99]	Root	Wang <i>et al.</i> , 2004
<i>Artocarpus dadah</i> Afzelechin-3- <i>O</i> - $\alpha$ -L-rhamnoside [107]	Stem bark Twig	Su <i>et al.</i> , 2002
(+)-Catechin [108]	Stem bark Twig	Su <i>et al.</i> , 2002
Dihydromorin [51]	Stem bark	Su <i>et al.</i> , 2002
Engeletin [52]	Twig	Su <i>et al.</i> , 2002
(+)-Epiafzelechin [109]	Stem bark	Su <i>et al.</i> , 2002
(-)-Epiafzelechin-(4 $\beta$ →8)-epicatechin [110]	Stem bark	Su <i>et al.</i> , 2002
Gemichalcone B [54]	Twig	Su <i>et al.</i> , 2002
Isogemichalcone B [111]	Twig	Su <i>et al.</i> , 2002

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Norartocarpetin [65]	Twig	Su <i>et al.</i> , 2002
Steppogenin [112]	Twig	Su <i>et al.</i> , 2002
<i>Artocarpus elasticus</i> Artelasticin [113]	Heartwood	Kijjoa <i>et al.</i> , 1996
Artelasticinol [114]	Root bark	Ko <i>et al.</i> , 2005
Artelastin [115]	Heartwood	Kijjoa <i>et al.</i> , 1996
Artelastinin [116]	Heartwood	Kijjoa <i>et al.</i> , 1998
Artelastocarpin [117]	Heartwood	Cidade <i>et al.</i> , 2001
Artelastochromene [118]	Heartwood	Kijjoa <i>et al.</i> , 1996
Artelastofuran [119]	Heartwood	Kijjoa <i>et al.</i> , 1998
Artelastoheterol [120]	Root bark	Ko <i>et al.</i> , 2005
Artelastoxanthone [121]	Root bark	Ko <i>et al.</i> , 2005
Artocarpesin [3]	Heartwood	Kijjoa <i>et al.</i> , 1996
Artocarpin [4]	Heartwood	Kijjoa <i>et al.</i> , 1996
Artoindonesianin E1 [122]	Wood	Musthapa <i>et al.</i> , 2009
Artonin F [25]	Root bark	Ko <i>et al.</i> , 2005
Artonin S [123]	Root bark	Ko <i>et al.</i> , 2005
Artonol A [28]	Root bark	Ko <i>et al.</i> , 2005
Artonol B [29]	Root bark	Ko <i>et al.</i> , 2005
Carpelastofuran [124]	Heartwood	Cidade <i>et al.</i> , 2001
Cycloartelastoxanthendiol [125]	Root bark	Ko <i>et al.</i> , 2005
Cycloartelastoxanthone [126]	Root bark	Ko <i>et al.</i> , 2005
Cycloartobiloxanthone [38]	Root bark	Ko <i>et al.</i> , 2005
Cycloartocarpesin [105]	Heartwood	Pendse <i>et al.</i> , 1976

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Cycloartocarpin [39]	Heartwood	Pendse <i>et al.</i> , 1976
Cyclomorusin [46]	Root bark	Ko <i>et al.</i> , 2005
Integrin [127]	Heartwood	Pendse <i>et al.</i> , 1976
Norartocarpin [128]	Heartwood	Pendse <i>et al.</i> , 1976
<i>Artocarpus fretessi</i>		
Afzelechin [129]	Stem bark	Soekamto <i>et al.</i> , 2003
Afzelechin-3-O- $\alpha$ -L-rhamnoside [107]	Stem bark	Soekamto <i>et al.</i> , 2003
Artonin A [92]	Stem bark	Soekamto <i>et al.</i> , 2003
Mulberrin [130]	Stem bark	Soekamto <i>et al.</i> , 2003
Mulberrochromene [131]	Stem bark	Soekamto <i>et al.</i> , 2003
Norartocarpetin [65]	Stem bark	Soekamto <i>et al.</i> , 2003
<i>Artocarpus glaucus</i>		
Cudraflavone C [34]	Root bark	Hakim <i>et al.</i> , 2006
Catechin [108]	Root bark	Hakim <i>et al.</i> , 2006
<i>Artocarpus gomezianus</i>		
Albanin A [132]	Root	Likhitwitayawuid, Sritularak and De-Eknamkul, 2000
Artocarpesin [3]	Heartwood	Venkataraman, 1972
Artocarpin [4]	Heartwood Root	Venkataraman, 1972 Likhitwitayawuid <i>et al.</i> , 2000
Cudraflavone C [34]	Root	Likhitwitayawuid <i>et al.</i> , 2000
Cycloartocarpin [39]	Heartwood Root	Venkataraman, 1972 Likhitwitayawuid <i>et al.</i> , 2000

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Isocyclomorusin [33]	Root	Likhitwitayawuid <i>et al.</i> , 2000
Morin [61]	Heartwood	Venkataraman, 1972
Norartocarpetin [65]	Heartwood Root	Venkataraman, 1972 Likhitwitayawuid <i>et al.</i> , 2000
Norcycloartocarpin [133]	Heartwood	Likhitwitayawuid <i>et al.</i> , 2006
<b><i>Artocarpus heterophyllus</i></b>		
Afzelechin-(4 $\alpha$ →8)-catechin [134]	Leaves	An <i>et al.</i> , 1992
Albanin A [132]	Wood	Arung, Shimizu and Kondo, 2006b
Artocarpanone [135]	Heartwood	Radhakrishnan, Rao and Venkataraman, 1965
Artocarpanone A [136]	Root bark	Lin, Lu and Huang, 1995
Artocarpesin [3]	Heartwood	Radhakrishnan <i>et al.</i> , 1965
Artocarpetin [137]	Heartwood	Venkataraman, 1972
Artocarpetin A [138]	Root bark	Lin <i>et al.</i> , 1995
Artocarpetin B [139]	Root	Chung <i>et al.</i> , 1995
Artocarpfuranol [140]	Wood	Zheng <i>et al.</i> , 2008
Artocarpin [4]	Heartwood	Radhakrishnan <i>et al.</i> , 1965
Artoflavanone [141]	Root	Dayal and Seshadri, 1974
Artoheterophyllin B [142]	Twig	Zheng <i>et al.</i> , 2009
Artoheterophyllin C [143]	Twig	Zheng <i>et al.</i> , 2009



**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Artoheterophyllin D [144]	Twig	Zheng <i>et al.</i> , 2009
Artonin A [92]	Root bark	Hano <i>et al.</i> , 1989
Artonin B [93]	Root bark	Hano <i>et al.</i> , 1989
Artonin C [145]	Root bark	Hano, Aida and Nomura, 1990a
Artonin D [146]	Root bark	Hano <i>et al.</i> , 1990a
Artonin I [147]	Root bark	Hano <i>et al.</i> , 1989
Artonin J [148]	Root bark	Aida <i>et al.</i> , 1993
Artonin K [26]	Root bark	Aida <i>et al.</i> , 1993
Artonin L [149]	Root bark	Aida <i>et al.</i> , 1993
Artonin Q [150]	Bark	Aida <i>et al.</i> , 1994
Artonin R [151]	Bark	Aida <i>et al.</i> , 1994
Artonin S [123]	Bark	Aida <i>et al.</i> , 1994
Artonin T [152]	Bark	Aida <i>et al.</i> , 1994
Artonin U [103]	Bark	Aida <i>et al.</i> , 1994
Artonin X [153]	Bark	Shinomiya <i>et al.</i> , 1995
Catechin [108]	Leaves	Yamazaki <i>et al.</i> , 1987
Cudraflavone A [33]	Root bark	Lin <i>et al.</i> , 1995
Cudraflavone C [34]	Wood	Arung <i>et al.</i> , 2006b
Cyanomaclurin [154]	Heartwood	Radhakrishnan <i>et al.</i> , 1965
Cycloartocarpesin [105]	Heartwood	Pathasarathy <i>et al.</i> , 1969
Cycloartocarpin [39]	Heartwood	Venkataraman, 1972
Cycloartocarpin A [106]	Root bark	Lu and Lin, 1994

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Cycloheterophyllin [96]	Bark	Rao, Varadan, and Venkataraman, 1971
	Root bark	Hano <i>et al.</i> , 1989
Dihydromorin [51]	Heartwood	Venkataraman, 1972
Heteroartonin A [155]	Root bark	Chung <i>et al.</i> , 1995
Heteroflavanone A [156]	Root bark	Lu and Lin, 1993
Heteroflavanone B [157]	Root bark	Lu and Lin, 1993
Heteroflavanone C [97]	Root bark	Lu and Lin, 1994
Heterophyllin [98]	Root bark	Hano <i>et al.</i> , 1989
Heterophyllol [158]	Root bark	Lu and Lin, 1993
Isocycloheterophyllin [159]	Bark	Rao, Varadan and Venkataraman, 1973
Kuwanon C [160]	Wood	Arung <i>et al.</i> , 2006b
Kuwanon R [161]	Root bark	Shinomiya <i>et al.</i> , 1995
Kuwanon T [162]	Root bark	Shinomiya <i>et al.</i> , 1995
Morin [61]	Heartwood	Radhakrishnan <i>et al.</i> , 1965; Pathasarathy <i>et al.</i> , 1969; Mu and Li, 1982
Norartocarpetin [65]	Heartwood	Radhakrishnan <i>et al.</i> , 1965
Norartocarpin [128]	Heartwood Wood	Venkataraman, 1972 Arung <i>et al.</i> , 2006b
Oxydihydroartocarpesin [163]	Heartwood	Pathasarathy <i>et al.</i> , 1969
6-Prenylapigenin [77]	Wood	Arung <i>et al.</i> , 2006b
3-Prenyl luteolin [164]	Wood	Arung <i>et al.</i> , 2010
Procyanidin B-3 [165]	Leaves	An <i>et al.</i> , 1992

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Procyanidin C-1 [166]	Leaves	An <i>et al.</i> , 1992
<i>Artocarpus hirsuta</i> Artocarpanone [135]	Heartwood	Venkataraman, 1972
Artocarpesin [3]	Heartwood	Venkataraman, 1972
Artocarpetin [137]	Heartwood	Venkataraman, 1972
Artocarpin [4]	Heartwood	Venkataraman, 1972; Arung <i>et al.</i> , 2006b
Cyanomaclurin [154]	Heartwood	Venkataraman, 1972
Cycloartocarpesin [105]	Heartwood	Venkataraman, 1972
Cycloartocarpin [39]	Heartwood	Venkataraman, 1972
Dihydromorin [51]	Heartwood	Venkataraman, 1972
Morin [61]	Heartwood	Venkataraman, 1972
Norartocarpetin [65]	Heartwood	Venkataraman, 1972
Oxydihydroartocarpesin [163]	Heartwood	Venkataraman, 1972
<i>Artocarpus integer</i> Artocarpanone [135]	Heartwood	Pendse <i>et al.</i> , 1976
Artocarpesin [3]	Heartwood	Pendse <i>et al.</i> , 1976
Artocarpetin [137]	Heartwood	Pendse <i>et al.</i> , 1976
Catechin [108]	Leaves	Yamazaki <i>et al.</i> , 1987
Chaplashin [104]	Heartwood	Pendse <i>et al.</i> , 1976
Cycloartocarpesin [105]	Heartwood	Pendse <i>et al.</i> , 1976
Cycloartocarpin [39]	Heartwood	Pendse <i>et al.</i> , 1976
Cyclointegrin [167]	Heartwood	Pendse <i>et al.</i> , 1976
Cyanomaclurin [154]	Heartwood	Pendse <i>et al.</i> , 1976

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Dihydromorin [51]	Heartwood	Pendse <i>et al.</i> , 1976
Integrin [127]	Heartwood	Pendse <i>et al.</i> , 1976
Morin [61]	Heartwood	Pendse <i>et al.</i> , 1976
Norartocarpetin [65]	Heartwood	Pendse <i>et al.</i> , 1976
Oxydihydroartocarpesin [163]	Heartwood	Pendse <i>et al.</i> , 1976
Oxyisocyclointegrin [168]	Heartwood	Pendse <i>et al.</i> , 1976
<i>Artocarpus kemando</i> Artobiloxanthone [2] (artocarpus flavone KB-1)	Stem bark	Seo <i>et al.</i> , 2003
Artocarpin [4]	Root	Hakim <i>et al.</i> , 2006
Artoindonesianin B [83]	Root	Hakim <i>et al.</i> , 2006
Artoindonesianin D [169]	Root	Hakim <i>et al.</i> , 2006
Artonin E [12]	Stem bark	Seo <i>et al.</i> , 2003
Artonin O [170]	Stem bark	Seo <i>et al.</i> , 2003
Cycloartobiloxanthone [38]	Stem bark Root	Seo <i>et al.</i> , 2003 Hakim <i>et al.</i> , 2006
Cycloartocarpin [39]	Root	Hakim <i>et al.</i> , 2006
Norartocarpetin [65]	Root	Hakim <i>et al.</i> , 2006
<i>Artocarpus lakoocha</i> Artocarpin [4]	Heartwood	Venkataraman, 1972
Cycloartocarpin [39]	Heartwood	Venkataraman, 1972
5,7-Dihydroxyflavone-3- <i>O</i> - $\alpha$ -L-rhamnoside [171]	Root bark	Chauhan and Kumari, 1979
5-Hydroxy-7,2',4'-trimetroxyflavone [172]	Stemwood	Pavaro and Reutrakul, 1976
Galangin-3- <i>O</i> - $\alpha$ -L-rhamnoside [173]	Root bark	Chauhan and Kumari, 1979
Galangin-3- <i>O</i> - $\beta$ -D-galactopyranosyl-(1 $\rightarrow$ 4)- $\alpha$ -L-rhamnoside [174]	Root bark	Chauhan, Kumari and Saraswat, 1979

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Kaempferol-3- <i>O</i> - $\beta$ -D-xyloside [175]	Root bark	Chauhan <i>et al.</i> , 1982
Norartocarpin [128]	Heartwood	Venkataraman, 1972
Norcycloartocarpin [176]	Heartwood	Venkataraman, 1972
Quercetin-3- <i>O</i> - $\alpha$ -L-rhamnoside [177]	Root bark	Chauhan <i>et al.</i> , 1982
5,7,2',4',-tetrahydroxy-3-prenyl-6-geranylflavone [178]	Root bark	Tantrakarnsakul, 2010
<b><i>Artocarpus lanceifolius</i></b>		
Artelasticin [113]	Heartwood	Syah <i>et al.</i> , 2001
Artelastofuran [119]	Heartwood	Syah <i>et al.</i> , 2001
Artobiloxanthone [2]	Bark	Hakim <i>et al.</i> , 2002a
Artoindonesianin G [179]	Heartwood	Syah <i>et al.</i> , 2001
Artoindonesianin H [180]	Heartwood	Syah <i>et al.</i> , 2001
Artoindonesianin I [181]	Heartwood	Syah <i>et al.</i> , 2001
Artoindonesianin P [182]	Bark	Hakim <i>et al.</i> , 2002a
Artoindonesianin Z-1 [183]	Stem bark	Syah <i>et al.</i> , 2006a
Artoindonesianin Z-2 [184]	Stem bark	Syah <i>et al.</i> , 2006a
Artoindonesianin Z-3 [185]	Stem bark	Hakim <i>et al.</i> , 2006
Artoindonesianin Z-4 [186]	Stem bark	Musthapa <i>et al.</i> , 2009
Artoindonesianin Z-5 [187]	Stem bark	Musthapa <i>et al.</i> , 2009
Artonin E [12]	Stem	Cao <i>et al.</i> , 2003
Artonol B [29]	Bark	Hakim <i>et al.</i> , 2002a
Cycloartobiloxanthone [38]	Bark	Hakim <i>et al.</i> , 2002a
14-Hydroxyartonin E [188]	Stem	Cao <i>et al.</i> , 2003

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
<i>Artocarpus lowii</i> 2',4'-Dihydroxy-4-methoxy-3'-prenyldihydrochalcone [189]	Leaves	Jamil <i>et al.</i> , 2008
<i>Artocarpus maingayi</i> Artocarpin [4] Artoindonesianin D [169] Chaplashin [104] Cudraflavone A [33] Cycloartocarpin A [106]	Stem bark Stem bark Stem bark Stem bark Stem bark	Hakim <i>et al.</i> , 2006 Hakim <i>et al.</i> , 2006 Hakim <i>et al.</i> , 2006 Hakim <i>et al.</i> , 2006 Hakim <i>et al.</i> , 2006
<i>Artocarpus nobilis</i> Artobilochromen [190] Artobiloxanthone [2] Artonin E 2'-methylether [191] Artonin V 2'-methylether [192] Chromanoartobilochromen A [193] Chromanoartobilochromen B [194] Chromanoartobilochromene [195] Cycloartobiloxanthone [38] 1-(3,4-Dihydro-3,5-dihydroxy-2-methyl-2-(3-methyl-2-butenyl)-2H-1-benzopyran-6-yl-3-(4-hydroxy phenyl)-2(E)-propen-1-one [196]	Bark Bark Root bark Root bark Stem bark Stem bark Bark Bark Fruit	Pavanasasivam, Sultanbawa and Mageswaran, 1974; Kumar <i>et al.</i> , 1977; Sultanbawa and Surendrakumar, 1989 Sultanbawa and Surendrakumar, 1989 Jayasinghe <i>et al.</i> , 2008 Jayasinghe <i>et al.</i> , 2008 Kumar <i>et al.</i> , 1977 Pavanasasivam <i>et al.</i> , 1974; Kumar <i>et al.</i> , 1977 Pavanasasivam <i>et al.</i> , 1974 Pavanasasivam <i>et al.</i> , 1974 Jayasinghe <i>et al.</i> , 2006

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
(-)-Dihydrofuranoartobilochromen A [197]	Stem bark	Kumar <i>et al.</i> , 1977
(-)-Dihydrofuranoartobilochromen B-1 [198]	Stem bark	Kumar <i>et al.</i> , 1977
(-)-Dihydrofuranoartobilochromen B-2 [199]	Stem bark	Kumar <i>et al.</i> , 1977
Dihydroisoartonin E 2'-methylether [200]	Root bark	Jayasinghe <i>et al.</i> , 2008
Furanoartobilochromen A [201]	Bark	Pavanasasivam <i>et al.</i> , 1974
Furanoartobilochromen B-1 [202]	Bark	Pavanasasivam <i>et al.</i> , 1974
Furanoartobilochromen B-2 [203]	Bark	Pavanasasivam <i>et al.</i> , 1974
3'-Geranyl-2',3,4,4'-tetrahydroxychalcone [204]	Leaves Fruit	Jayasinghe <i>et al.</i> , 2004a Jayasinghe <i>et al.</i> , 2006
3'-Geranyl-4',5,7-trihydroxyflavanone [205]	Fruit	Jayasinghe <i>et al.</i> , 2006
8-Geranyl-3',4',7-trihydroxyflavanone [206]	Fruit	Jayasinghe <i>et al.</i> , 2006
8-Geranyl-4',7-dihydroxyflavanone [207]	Fruit	Jayasinghe <i>et al.</i> , 2006
Isoartonin E 2'-methylether [208]	Root bark	Jayasinghe <i>et al.</i> , 2008
Isonymphaeol-B [209]	Fruit	Jayasinghe <i>et al.</i> , 2006
Lespeol [210]	Fruit	Jayasinghe <i>et al.</i> , 2006
2,4,4'-Trihydroxy-3-[(2E)-5-methoxy-3,7-dimethylocta-2,6-dienyl]chalcone [211]	Fruit	Jayasinghe <i>et al.</i> , 2006
2',4',4'-Trihydroxy-3'-geranylchalcone [212]	Leaves	Fujimoto <i>et al.</i> , 2004
2',4',4'-Trihydroxy-3'-[6-hydroxy-3,7-dimethyl-2(E),7-octadiethyl] chalcone [213]	Leaves	Fujimoto <i>et al.</i> , 2004

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
2',4',4'-Trihydroxy-3'-[2-hydroxy-7-methyl-3-methylene-6-octaethyl] chalcone [214]	Leaves	Fujimoto <i>et al.</i> , 2004
2',3,4,4'-Tetrahydroxy-3'-geranyl chalcone [215]	Leaves	Fujimoto <i>et al.</i> , 2004
2',3,4,4'-Tetrahydroxy-3'-[6-hydroxy-3,7-dimethyl-2( <i>E</i> ),7-octadiethyl] chalcone [216]	Leaves	Fujimoto <i>et al.</i> , 2004
Oxydihydromorusin [217]	Stem bark	Kumar <i>et al.</i> , 1977; Fukai and Nomura, 1993
Xanthoangelol [218]	Leaves Fruit	Jayasinghe <i>et al.</i> , 2004a Jayasinghe <i>et al.</i> , 2006
Xanthoangelol B [219]	Leaves Fruit	Jayasinghe <i>et al.</i> , 2004a Jayasinghe <i>et al.</i> , 2006
<i>Artocarpus obtusus</i>		
Dihydroartoindonesianin C [220]	Stem bark	Hashim <i>et al.</i> , 2010
Pyranocycloartobiloxanthone A [221]	Stem bark	Hashim <i>et al.</i> , 2010
<i>Artocarpus odoratissimus</i>		
Artosimmin [222]	-	Ee <i>et al.</i> , 2010
<i>Artocarpus pithecofallus</i>		
Morin [61]	Heartwood	Mu and Li, 1982
<i>Artocarpus rigida</i>		
Artobiloxanthone [2]	Stem bark	Hano <i>et al.</i> , 1990b
Artocarpol B [223]	Root bark	Ko, Lin, and Yang, 2000
Artocarpol H [224]	Root bark	Lu <i>et al.</i> , 2002
Artonin E [12]	Stem bark	Hano <i>et al.</i> , 1990b
Artonin G [225]	Stem bark	Hano <i>et al.</i> , 1990b
Artonin H [226]	Stem bark	Hano <i>et al.</i> , 1990b



**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Artonin M [227]	Stem bark	Hano <i>et al.</i> , 1990b
Artonin N [228]	Stem bark	Hano <i>et al.</i> , 1990b
Artonin O [170]	Stem bark	Hano <i>et al.</i> , 1990b
Artonin P [229]	Stem bark	Hano <i>et al.</i> , 1990b
Cycloartobiloxanthone [38]	Stem bark	Hano <i>et al.</i> , 1990b
Rubraflavone C [230]	Root bark	Lu <i>et al.</i> , 2002
<b><i>Artocarpus rotundo</i></b>		
Artoindonesianin L [231]	Root bark	Suhartati <i>et al.</i> , 2001
Artonin E [12]	Root bark	Suhartati <i>et al.</i> , 2001
Artonin M [226]	Root bark	Suhartati <i>et al.</i> , 2001
Artonin O [170]	Root bark	Suhartati <i>et al.</i> , 2001
Cycloartobiloxanthone [38]	Root bark	Suhartati <i>et al.</i> , 2001
<b><i>Artocarpus scoteginii</i></b>		
Artobiloxanthone [2] (artocarpus flavone KB-1)	Stem bark	Hakim <i>et al.</i> , 2006
Artonin E [12] (artocarpus flavone KB-3)	Stem bark	Hakim <i>et al.</i> , 2006
Cycloartobiloxanthone [38]	Stem bark	Hakim <i>et al.</i> , 2006
Norartocarpetin [65]	Stem bark	Hakim <i>et al.</i> , 2006
5'-Hydroxycudraflavone A [99]	Stem bark	Hakim <i>et al.</i> , 2006
<b><i>Artocarpus sepicanus</i></b>		
Sepicanin A [232]	Leaves	Radwan <i>et al.</i> , 2009
<b><i>Artocarpus styracifolius</i></b>		
Styracifolin A [233]	Stem bark	Bourjot <i>et al.</i> , 2010

**Table 1 (continued)**

Plant and chemical compound	Plant part	Reference
Styracifolin B [234]	Stem bark	Bourjot <i>et al.</i> , 2010
<i>Artocarpus teysmanii</i>		
Artoindonesianin C [235]	Root bark	Makmur <i>et al.</i> , 2000
Artonin J [148]	Root bark	Makmur <i>et al.</i> , 2000
Cycloartobiloxanthone [38]	Root bark	Makmur <i>et al.</i> , 2000
<i>Artocarpus tonkinensis</i>		
Alphitonin-4- <i>O</i> -glucoside [236]	Leaves	Thuy <i>et al.</i> , 2004
Artokin-4'- <i>O</i> -glucoside [237]	Leaves	Dang <i>et al.</i> , 2009
Artotonin A [238]	Root	Ma <i>et al.</i> , 2010
Artotonin B [239]	Root	Ma <i>et al.</i> , 2010
Maesopsin-4- <i>O</i> -glucoside [240]	Leaves	Thuy <i>et al.</i> , 2004
<i>Artocarpus venenosa</i>		
Paratocarpin A [241]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin B [242]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin C [243]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin D [244]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin E [245]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin F [246]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin G [247]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin H [248]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin I [249]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin J [250]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin K [251]	Stem bark	Nomura <i>et al.</i> , 1998
Paratocarpin L [252]	Stem bark	Nomura <i>et al.</i> , 1998

**Table 2** Distribution of triterpenoids and steroids in the genus *Artocarpus*

Plant and chemical compound	Category	Plant part	Reference
<b><i>Artocarpus altilis</i></b>			
$\alpha$ -Amyrin [253]	Triterpene	Latex	Ultee, 1949
$\alpha$ -Amyrin acetate [254]	Triterpene	Fruit	Altman and Zito, 1976
$\beta$ -Amyrin acetate [255]	Triterpene	Latex	Ultee, 1949
Cycloart-23-ene-3 $\beta$ ,25-diol [256]	Triterpene	Fruit	Altman and Zito, 1976
Cycloart-24-ene-3 $\beta$ -ol [257] (Cycloartenol)	Triterpene	Fruit Stem bark	Altman and Zito, 1976 Pavanasasivum and Sultanbawa, 1973
Cycloart-25-ene-3 $\beta$ ,24-diol [258]	Triterpene	Fruit	Altman and Zito, 1976
Cycloartenone [259]	Triterpene	Stem bark	Pavanasasivum and Sultanbawa, 1973
Cycloartenyl acetate [260]	Triterpene	Stem bark	Pavanasasivum and Sultanbawa, 1973
Lupeol acetate [261]	Triterpene	Root bark	Shieh and Lin, 1992
$\beta$ -Sitosterol [262]	Steroid	Root bark	Shieh and Lin., 1992
<b><i>Artocarpus champeden</i></b>			
Cycloartenone [259]	Triterpene	Stem bark	Achmad <i>et al.</i> , 1996
Cycloeucaleanol [263]	Triterpene	Stem bark	Achmad <i>et al.</i> , 1996
Glutinol [264]	Triterpene	Stem bark	Achmad <i>et al.</i> , 1996
24-Methylenecycloartenone [265]	Triterpene	Stem bark	Achmad <i>et al.</i> , 1996
<b><i>Artocarpus chaplasha</i></b>			
Cycloartenyl acetate [260]	Triterpene	Stem bark	Mahato, Banerjee and Chakravarti, 1971

**Table 2 (continued)**

Plant and chemical compound	Category	Plant part	Reference
Isocycloartenol acetate [266]	Triterpene	Stem bark	Mahato <i>et al.</i> , 1971
Lupeol acetate [261]	Triterpene	Stem bark	Mahato <i>et al.</i> , 1971
$\beta$ -Sitosterol [262]	Steroid	Root bark	Shieh and Lin., 1992
<i>Artocarpus elasticus</i>			
$\beta$ -Amyrin acetate [254]	Triterpene	Latex	Ultee, 1949
Lupeol acetate [261]	Triterpene	Latex	Ultee, 1949
$\beta$ -Sitosterol [262]	Steroid	Root bark	Shieh and Lin., 1992
<i>Artocarpus gomezianus</i>			
Lupeol acetate [261]	Triterpene	Leaves	Kingroungpet, 1994
Simiarenol [267]	Triterpene	Leaves	Kingroungpet, 1994
$\beta$ -Sitosterol [262]	Steroid	Root bark	Shieh and Lin., 1992
Stigmasterol [268]	Steroid	Leaves	Kingroungpet, 1994
<i>Artocarpus heterophyllus</i>			
Betulin [269]	Triterpene	Root bark	Lu and Lin, 1994
Betulinic acid [270]	Triterpene	Root Root bark	Dayal and Seshadri, 1974 Lu and Lin, 1994
Butyrospermol [271]	Triterpene	Fruit	Barton, 1951
Cycloartenone [257]	Triterpene	Stem Bark	Pavanasasivam and Sultanbawa, 1973
Cycloartenyl acetate [260]	Triterpene	Stem Bark	Pavanasasivam and Sultanbawa, 1973
9,19-Cyclolanost-23-ene-3 $\beta$ ,25-diol (Cycloart-23-ene-3 $\beta$ ,25-diol) [256]	Triterpene	Fruit	Kielland and Malterud, 1994
Cycloartenone [257]	Triterpene	Stem Bark	Pavanasasivam and Sultanbawa, 1973

**Table 2 (continued)**

Plant and chemical compound	Category	Plant part	Reference
9,19-Cyclolanost-25-ene-3 $\beta$ ,24-diol [258]	Triterpene	Fruit	Kielland and Malterud, 1994
9,19-Cyclolanost-3-one-24,25-diol [272]	Triterpene	Fruit	Kielland and Malterud, 1994
Ursolic acid [273]	Triterpene	Root	Dayal and Seshadri, 1974
$\beta$ -Sitosterol [262]	Steroid	Root bark Root bark	Lu and Lin, 1994 Shieh and Lin., 1992
<i>Artocarpus integer</i>			
Cycloartenone [259]	Triterpene	Latex	Pant and Chaturvedi, 1989
$\beta$ -Sitosterol [262]	Steroid	Root bark	Shieh and Lin., 1992
<i>Artocarpus lakoocha</i>			
$\beta$ -Amyrin acetate [254]	Triterpene	Stem Bark	Kapil and Joshi, 1960
Cycloartenol [257] (Cycloart-24-ene-3 $\beta$ -ol)	Triterpene	Stem Bark	Pavanasasivam and Sultanbawa, 1973
Cycloartenone [259]	Triterpene	Stem Bark	Pavanasasivam and Sultanbawa, 1973
Cycloartenyl acetate [260]	Triterpene	Stem Bark	Pavanasasivam and Sultanbawa, 1973
Lupeol [274]	Triterpene	Root Bark	Chauhan and Kumari, 1979
$\beta$ -Sitosterol [262]	Steroid	Root bark	Shieh and Lin., 1992
<i>Artocarpus nobilis</i>			
Cycloartenol [257] (Cycloart-24-ene-3 $\beta$ -ol)	Triterpene	Stem Bark Heartwood	Pavanasasivam and Sultanbawa, 1973
Cycloartenone [259]	Triterpene	Stem Bark Heartwood	Pavanasasivam and Sultanbawa, 1973
Cycloartenyl acetate [260]	Triterpene	Stem Bark Heartwood	Pavanasasivam and Sultanbawa, 1973

**Table 3** Distribution of stilbenoids in the genus *Artocarpus*

Plant and chemical compound	Plant part	Reference
<i>Artocarpus altilis</i>		
Artocarbene [275]	Heartwood	Shimizu <i>et al.</i> , 1997
4-Prenyloxyresveratrol [276]	Heartwood	Shimizu <i>et al.</i> , 1997
<i>Artocarpus chaplasha</i>		
Artochamin H [277]	Stem	Wang <i>et al.</i> , 2006
Artochamin I [278]	Stem	Wang <i>et al.</i> , 2006
Artochamin J [279]	Stem	Wang <i>et al.</i> , 2006
Artochamin K [280]	Stem	Wang <i>et al.</i> , 2006
Artochamin F [281]	Stem	Wang <i>et al.</i> , 2006
Artochamin G [282]	Stem	Wang <i>et al.</i> , 2006
Artostilbene A [283]	Stem	Wang <i>et al.</i> , 2007a
Artostilbene B [284]	Stem	Wang <i>et al.</i> , 2007a
Oxyresveratrol [285]	Heartwood	Rao <i>et al.</i> , 1972
Resveratrol [286]	Heartwood	Rao <i>et al.</i> , 1972
<i>Artocarpus dadah</i>		
3-(2,3-Dihydroxy-3-methylbutyl)-resveratrol [287]	Stem bark	Su <i>et al.</i> , 2002
3-( $\gamma,\gamma$ -Dimethylallyl)-oxyresveratrol [288]	Stem bark	Su <i>et al.</i> , 2002
3-( $\gamma,\gamma$ -Dimethylallyl)-resveratrol [289]	Stem bark	Su <i>et al.</i> , 2002
3-( $\gamma,\gamma$ -Dimethylpropenyl) moracin M [290]	Stem bark	Su <i>et al.</i> , 2002
Moracin M [291]	Twig	Su <i>et al.</i> , 2002
Oxyresveratrol [285]	Stem bark Twig	Su <i>et al.</i> , 2002
Resveratrol [286]	Twig	Su <i>et al.</i> , 2002

**Table 3 (continued)**

Plant and chemical compound	Plant part	Reference
<i>Artocarpus fretessi</i>		
Artoindonesianin X [292]	Root bark	Soekamto <i>et al.</i> , 2003
Artoindonesianin Y [293]	Root bark	Soekamto <i>et al.</i> , 2003
<i>Artocarpus gomezianus</i>		
Andalasin [294]	Root	Likhitwitayawuid <i>et al.</i> , 2001
Artogomezianol [295]	Root	Likhitwitayawuid <i>et al.</i> , 2001
Artoindonesianin N [296]	Bark	Hakim <i>et al.</i> , 2002b
Artoindonesianin O [297]	Bark	Hakim <i>et al.</i> , 2002b
Oxyresveratrol [285]	Heartwood	Likhitwitayawuid <i>et al.</i> , 2006
Resveratrol [286]	Root	Likhitwitayawuid <i>et al.</i> , 2000
<i>Artocarpus heterophyllus</i>		
Artoheterophyllin A [298]	Twig	Zheng <i>et al.</i> , 2009
<i>Artocarpus integer</i>		
Artocarbene [275]	Aerial part	Boonlaksiri <i>et al.</i> , 2000
4-Methoxy-2,2-dimethyl-6-(2-(2,4-dihydroxy)phenyl-trans-ethenyl)chromene [299]	Aerial part	Boonlaksiri <i>et al.</i> , 2000
4-Prenyloxyresveratrol [276] ( <i>trans</i> -4-Isopentenyl-3,5,2',4'-tetrahydroxy stilbene)	Aerial part	Boonlaksiri <i>et al.</i> , 2000
<i>trans</i> -4-(3-Methyl- <i>E</i> -but-1-enyl)-3,5,2',4'-tetrahydroxystilbene [300] (Artoindonesianin F)	Aerial part	Boonlaksiri <i>et al.</i> , 2000
<i>Artocarpus lakoocha</i>		
Artolakoochol [301]	Root bark	Sritularak <i>et al.</i> , 2010
cycloartolakoochol [302]	Root bark	Sritularak <i>et al.</i> , 2010

**Table 3 (continued)**

Plant and chemical compound	Plant part	Reference
4-Hydroxyartolakochole [303]	Root bark	Sritularak <i>et al.</i> , 2010
Lakoochin A [304]	Root	Puntumchai <i>et al.</i> , 2004
Lakoochin B [305]	Root	Puntumchai <i>et al.</i> , 2004
Oxyresveratrol [285]	Heartwood	Venkataraman, 1972; Likhitwitayawuid <i>et al.</i> , 2001
Resveratrol [286]	Heartwood	Venkataraman, 1972
<b><i>Artocarpus nitidus</i></b>		
Artonitidin A [306]	Stem	Zhao <i>et al.</i> , 2009
Artonitidin B [307]	Stem	Zhao <i>et al.</i> , 2009
<b><i>Artocarpus nobilis</i></b>		
<i>trans</i> -4-(3-Methyl-E-but-1-enyl)-3,5,2',4'-tetrahydroxystilbene [300] (Artoindonesianin F)	Stem bark	Jayasinghe <i>et al.</i> , 2004b
<i>trans</i> -4-Isopentenyl-3,5,2',4'-tetrahydroxy stilbene [276] (4-prenyloxyresveratrol)	Stem bark	Jayasinghe <i>et al.</i> , 2004b
<b><i>Artocarpus petelotii</i></b>		
Artopetelin L [308]	Root bark	Shen and Hou, 2008
Artopetelin M [309]	Root bark	Shen and Hou, 2008
<b><i>Artocarpus rigida</i></b>		
Artocarpol A [310]	Root bark	Ko, Lin and Yang, 2001
Artocarpol C [311]	Root bark	Ko <i>et al.</i> , 2001
Artocarpol D [312]	Root bark	Ko <i>et al.</i> , 2001
Artocarpol E [313]	Root bark	Ko <i>et al.</i> , 2001
Artocarpol F [314]	Root bark	Ko <i>et al.</i> , 2001
<b><i>Artocarpus tonkinensis</i></b>		
Artotonkin [315]	Stem bark	Lien <i>et al.</i> , 1998

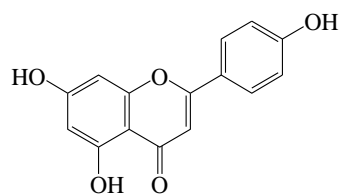


**Table 4** Distribution of miscellaneous constituents in the genus *Artocarpus*

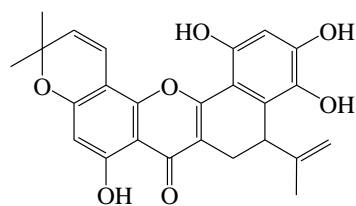
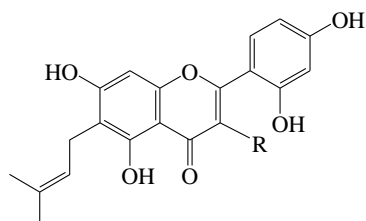
Plant and chemical compound	Category	Plant part	Reference
<i>Artocarpus altilis</i> $\gamma$ -Aminobutyric acid [316]	Amino acid	Leaves	Durand <i>et al.</i> , 1962
<i>Artocarpus chaplasha</i> Resorcinol [317] $\beta$ -Resorcylaldehyde [318]	Benzenoid Benzenoid	Heartwood Heartwood	Rao <i>et al.</i> , 1972 Rao <i>et al.</i> , 1972
<i>Artocarpus dadah</i> Dadahol A [319] Dadahol B [320]	Neolignan Neolignan	Twig Twig	Su <i>et al.</i> , 2002 Su <i>et al.</i> , 2002
<i>Artocarpus gomezianus</i> Arbutin [321] 1-Dotriacontanol [322] Phenyl- $\beta$ -naphthylamine [323] Resorcinol [317]	Phenolic glycoside Alcohol Naphthalene Benzenoid	Leaves Leaves Root Root	Kingroungpet, 1994 Kingroungpet, 1994 Likhitwitayawuid and Sritularak, 2001 Sritularak, 1998
<i>Artocarpus heterophyllus</i> Acetylcholine [324] Artocarpus integra $\alpha$ -D-Galactose specific lectin [325] Artocarpus integrifolia lectin [326] Artocarpus lectin CE-A-I [327] Aurantiamide acetate [328] 9-Hydroxytridecyl docosanoate [329]	Amine Lectin Lectin Lectin Protein Fatty acid	Seed Seed Seed Seed Seed Latex	Pereira <i>et al.</i> , 1962 Suresh, Appukuttan and Basu, 1982 Chatterjee <i>et al.</i> , 1982; Namjuntra and Culavatnatol, 1984 Ferreira <i>et al.</i> , 1992 Chakraborty and Mandal, 1981 Pant and Chaturvedi, 1989

**Table 4 (continued)**

Plant and chemical compound	Category	Plant part	Reference
4-Hydroxyundecyl docosanoate [330]	Fatty acid	Latex	Pant and Chaturvedi, 1989
Jacalin [331]	Lectin	Seed	Hagiwara <i>et al.</i> , 1988
Lymphoagglutinin [332]	Lectin	Seed	Ferreira <i>et al.</i> , 1992 Arora <i>et al.</i> , 1987
Recinoleic acid [333]	Fatty acid	Seed oil	Daulatabad and Mirajkar, 1989
<i>Artocarpus hirsuta</i>			
Lymphoagglutinin [332]	Lectin	Seed	Arora <i>et al.</i> , 1987
<i>Artocarpus integer</i>			
Artocarpus lectin C [334]	Lectin	Seed	Hashim, Gendeh and Jaafar, 1992
4-Hydroxyundecyl docosanoate [330]	Fatty acid	Latex	Pant and Chaturvedi, 1989
<i>Artocarpus lakoocha</i>			
ALA-I [335]	Isolectin	Seed	Wongkham <i>et al.</i> , 1995
ALA-II [336]	Isolectin	Seed	Wongkham <i>et al.</i> , 1995
ALA-III [337]	Isolectin	Seed	Promdee, 1996
Artocarpus lakoocha lectin [338]	Lectin	Seed	Chatterjee <i>et al.</i> , 1982
Lymphoagglutinin [332]	Lectin	Seed	Arora <i>et al.</i> , 1987
Resorcinol [317]	Benzenoid	Heartwood	Venkataraman, 1972
<i>Artocarpus lignanensis</i>			
Artocarpus lectin [339]	Lectin	Seed	Zhang <i>et al.</i> , 1999
<i>Artocarpus masticatus</i>			
Artocarpus lectin AM [340]	Lectin	Seed	Blasco <i>et al.</i> , 1996
<i>Artocarpus melinoxylus</i>			
Artocarpus lectin AME [341]	Lectin	Seed	Blasco <i>et al.</i> , 1996

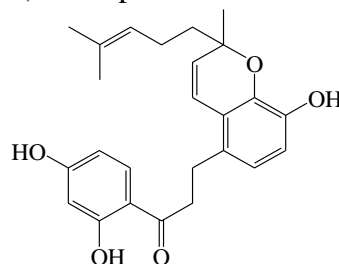


[1] Apigenin

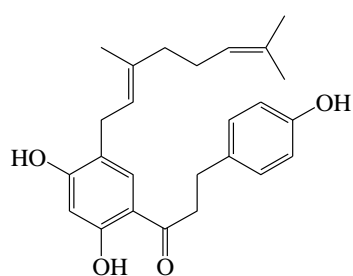
[2] Artobiloxanthone  
(artocarpus flavone KB-1)

[3] Artocarpesin R=H

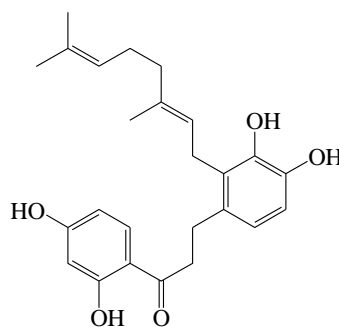
[4] Artocarpin R=



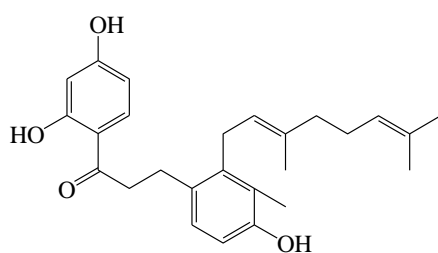
[5] Artocarpus chalcone AC-3-1



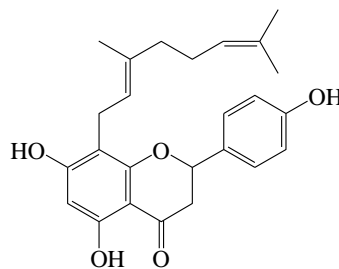
[6] Artocarpus chalcone AC-3-3



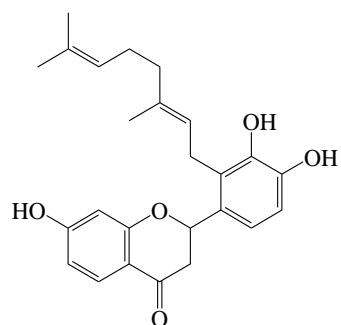
[7] Artocarpus chalcone AC-5-1



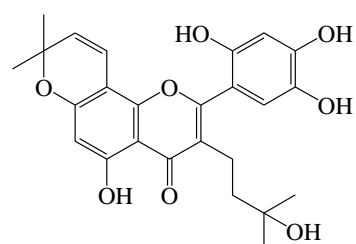
[8] Artocarpus chalcone I



[9] Artocarpus flavanone AC-3-2

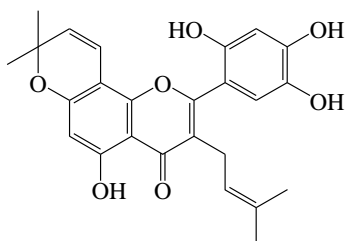


[10] Artocarpus flavanone AC-5-2

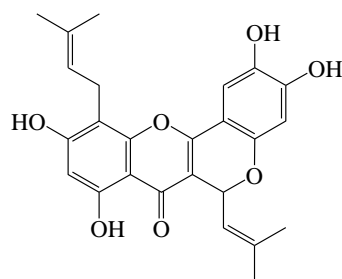


[11] Artocarpus flavone KB-2

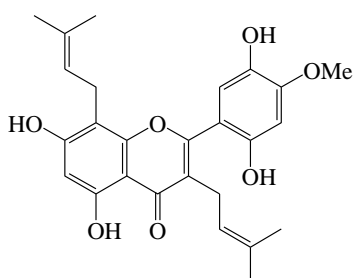
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp.



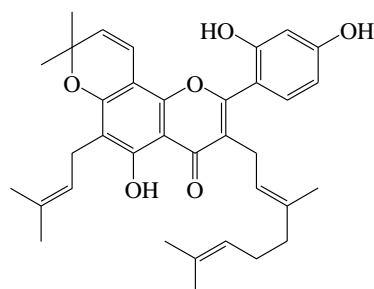
[12] Artocarpus flavone KB-3 (Artonin E)



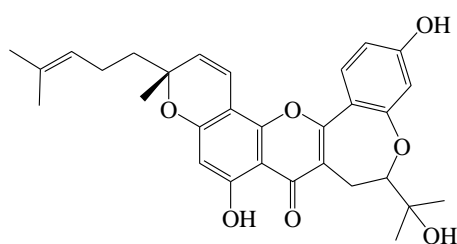
[13] Artochamin B



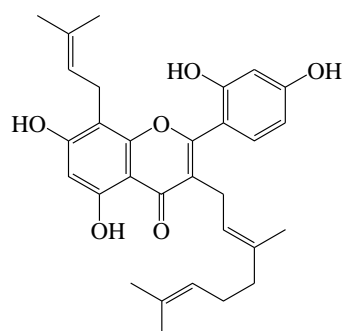
[14] Artochamin D



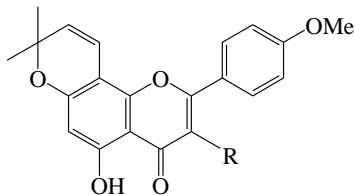
[15] Artocommunol CB



[16] Artocommunol CC

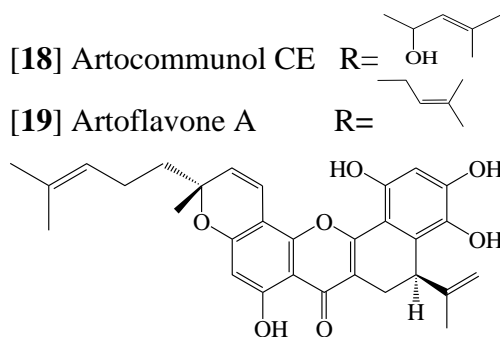


[17] Artocommunol CD

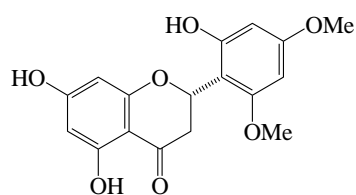


[18] Artocommunol CE R=

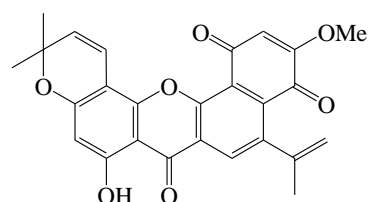
[19] Artoflavone A R=



[21] Artomunisoaxanthone

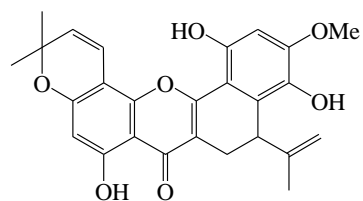


[20] Artomunoflavanone

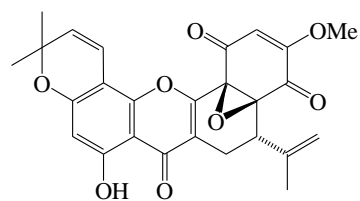


[22] Artomunoxanthentrione

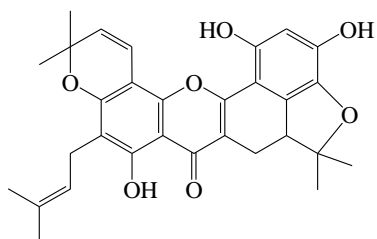
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



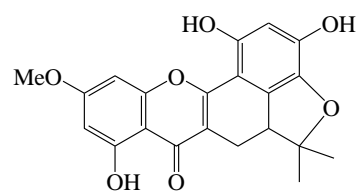
[23] Artomunoxanthone



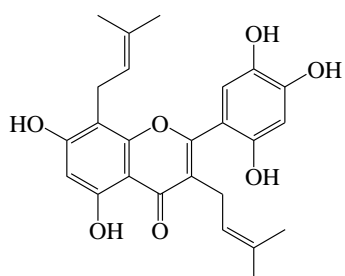
[24] Artomunoxanthotriene epoxide



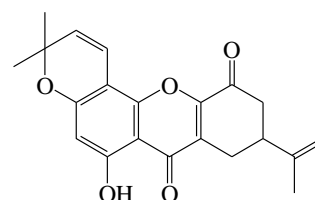
[25] Artonin F



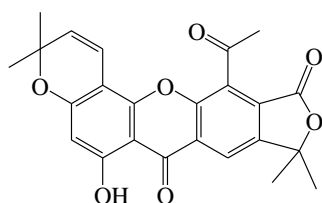
[26] Artonin K



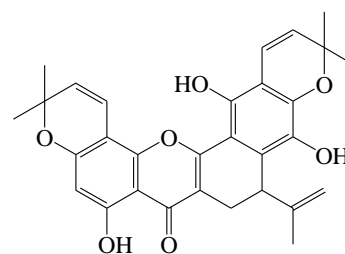
[27] Artonin V



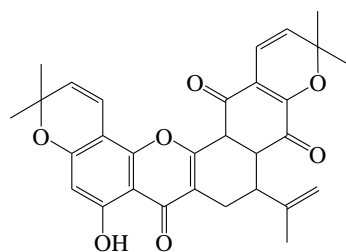
[28] Artonol A



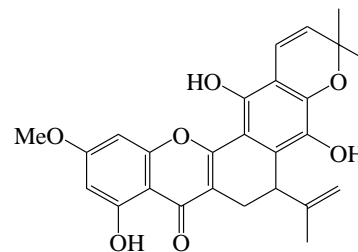
[29] Artonol B



[30] Artonol C

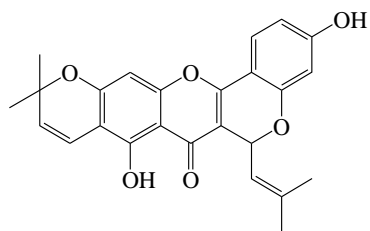


[31] Artonol D

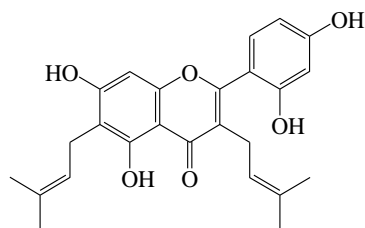


[32] Artonol E

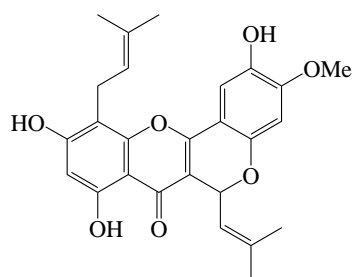
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



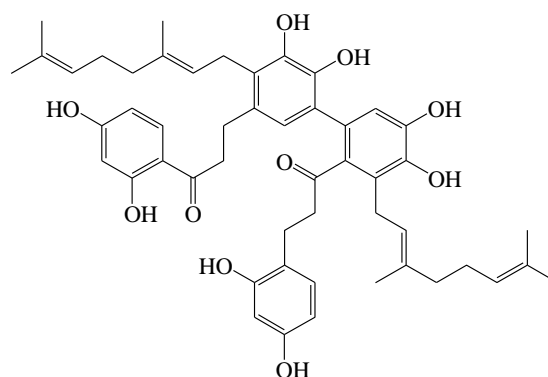
[33] Cudraflavone A  
(Isocyclomorusin)



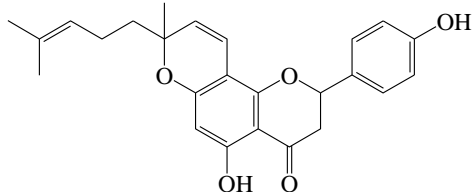
[34] Cudraflavone C



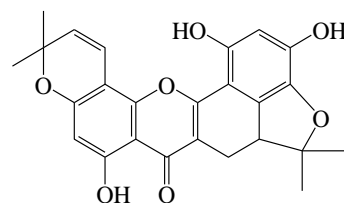
[35] Cycloaltilisin



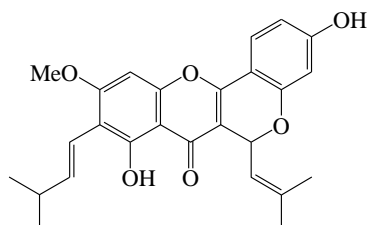
[36] Cycloaltilisin 6



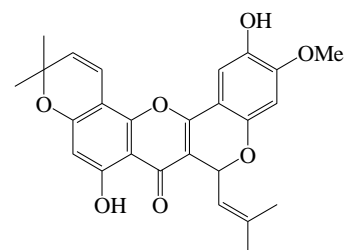
[37] Cycloaltilisin 7



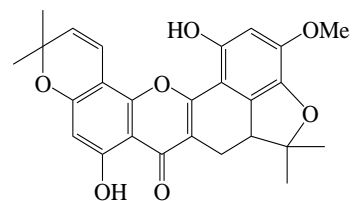
[38] Cycloartobiloxanthone



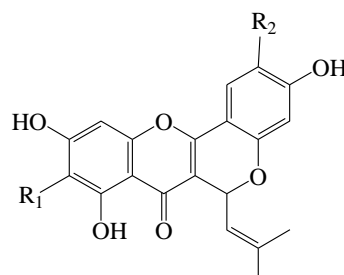
[39] Cycloartocarpin




[40] Cycloartomunin



[41] Cycloartomunoxanthone

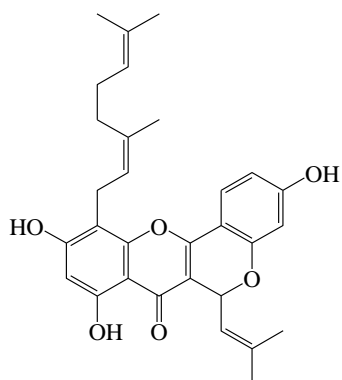


[42] Cyclocommunin R<sub>1</sub> =  R<sub>2</sub> = H

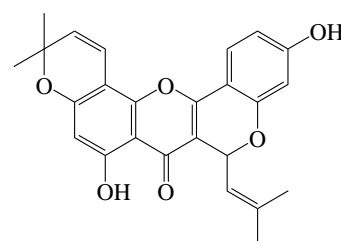
[43] Cyclocommunol R<sub>1</sub> = R<sub>2</sub> = H

[44] Cyclocommunomethonol R<sub>1</sub> = H R<sub>2</sub> = OMe

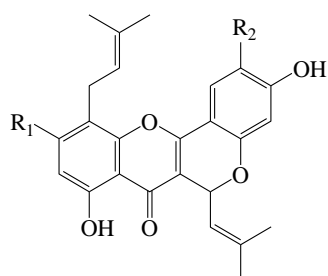
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



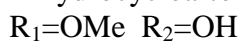
[45] Cyclogeracommunin



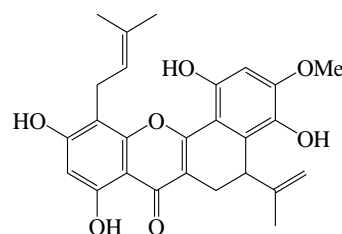
[46] Cyclomorusin

[47] Cyclomulberrin R<sub>1</sub>=OH R<sub>2</sub>=H

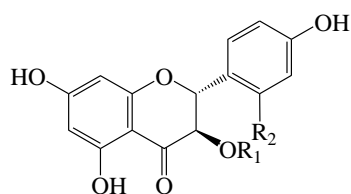
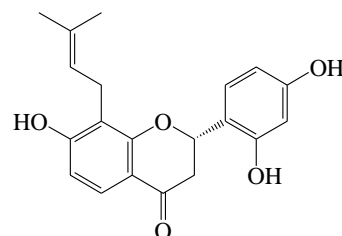
[49] Dihydrocycloartomunin



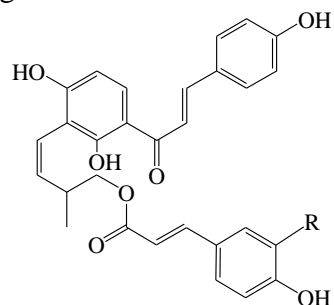
[50] Dihydroisocycloartomunin



[48] Dihydroartomunoxanthone

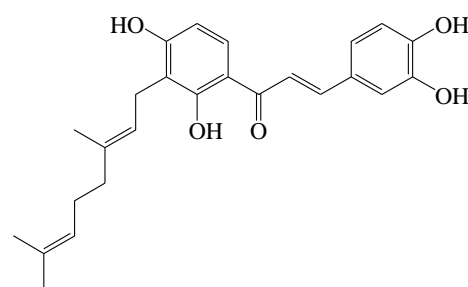
[51] Dihydromorin R<sub>1</sub>=H R<sub>2</sub>=OH[52] Engeletin R<sub>1</sub>=rhamnose R<sub>2</sub>=H

[53] (2S)-Euchrenone a7



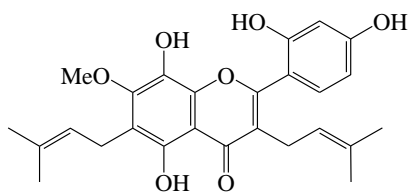
[54] Gemichalcone B R=H

[55] Gemichalcone C R=OMe

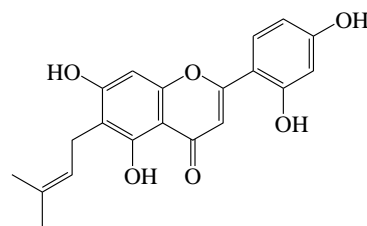


[56] 3'-Geranyl-2',3,4,4'-tetrahydroxychalcone

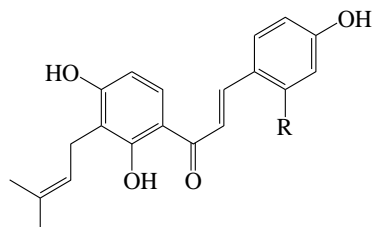
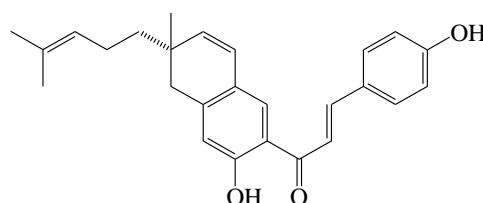
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



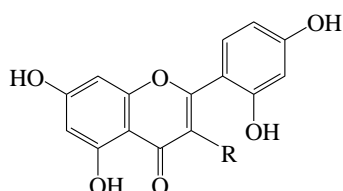
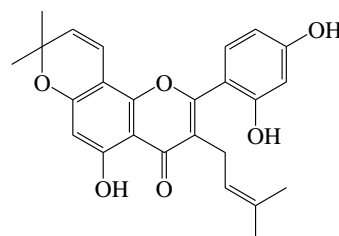
[57] Hydroxyartocarpin



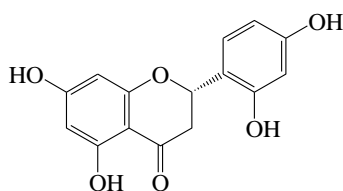
[58] Isoartocarpesin

[59] Isobacachalcone R=H  
[62] Morachalcone A R=OH

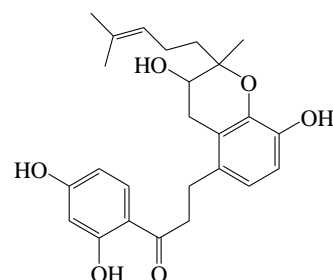
[60] Isolespeol

[61] Morin R=OH  
[65] Norartocarpetin R=H

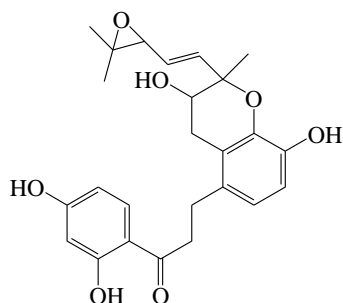
[63] Morusin



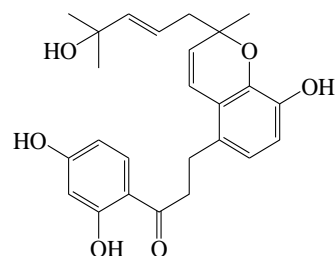
[64] (+)-Norartocarpanone



[66] 1-(2,4-Dihydroxyphenyl)-3-[3,4-dihydro-3,8-dihydroxy-2-methyl-2-(4-methyl-3-pentenyl)-2H-1-benzopyran-5-yl]-1-propanone



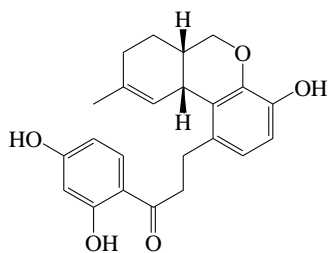
[67] 1-(2,4-Dihydroxyphenyl)-3-[8-hydroxy-2-methyl-2-(3,4-epoxy-4-methyl-1-pentenyl)-2H-1-benzopyran-5-yl]-1-propanone



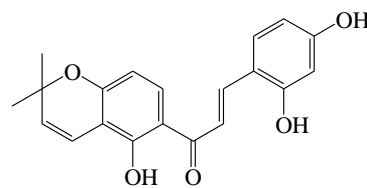
[68] 1-(2,4-Dihydroxyphenyl)-3-[8-hydroxy-2-methyl-2-(4-hydroxy-4-methyl-2-pentenyl)-2H-1-benzopyran-5-yl]-1-propanone

Figure 2 Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

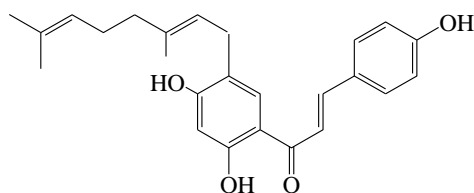




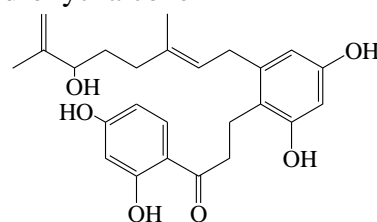
**[69]** 1-(2,4-Dihydroxyphenyl)-3-{4-hydroxy-6,6,9-trimethyl-6a,7,8,10a-tetrahydro-6H-dibenzo [b,d]pyran-5-yl}-1-propanone



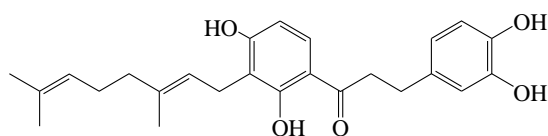
**[70]** 3'',3''-Dimethylpyrano[3',4']2,4,2'-trihydroxychalcone



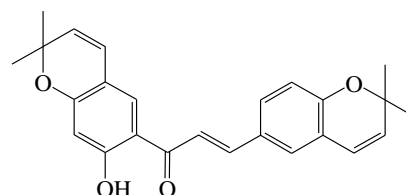
**[71]** 5'-Geranyl-2',4',4'-trihydroxychalcone



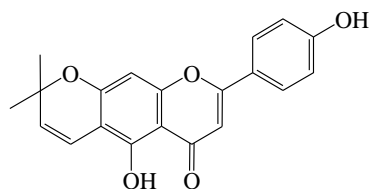
**[72]** 2-[6-Hydroxy-3,7-dimethylocta-2(E),7-dienyl]-2',3,4,4'-tetrahydroxy dihydrochalcone



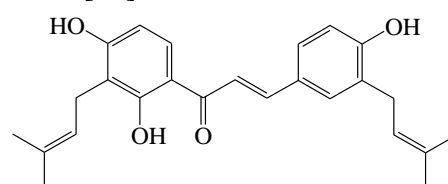
**[73]** 3,4,2',4'-Tetrahydroxy-3'-geranyl dihydrochalcone



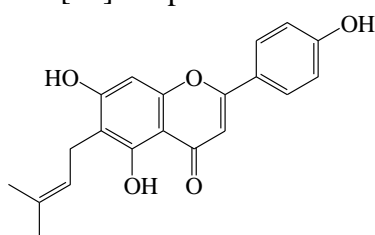
**[74]** Artoindonesianin J



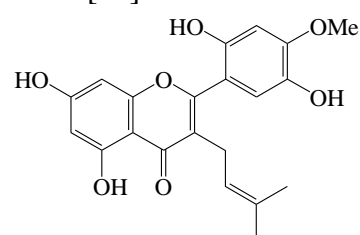
**[75]** Carpachromene



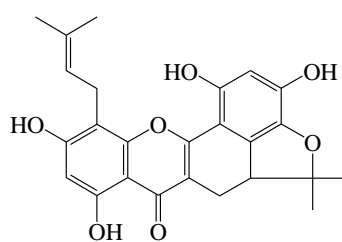
**[76]** Kozonol C



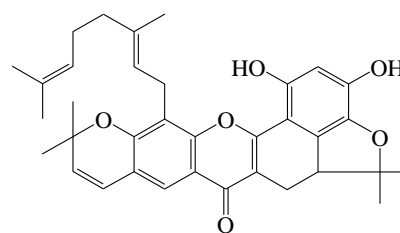
**[77]** 6-Prenylalpigenin



**[78]** Artocarpone A

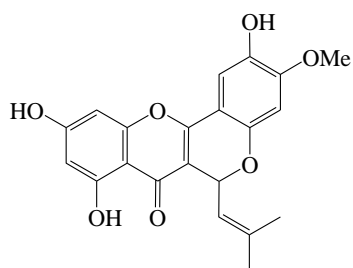


**[79]** Artocarpone B

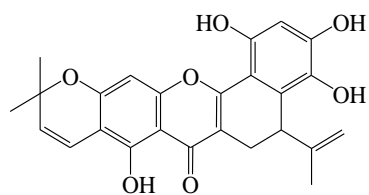


**[80]** Artoindonesianin A

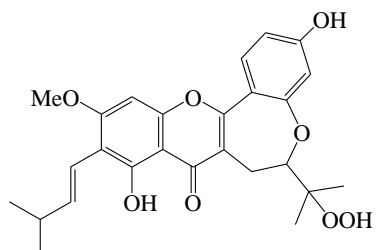
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



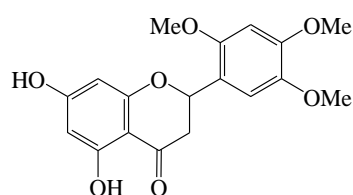
[81] Artoindonesianin A-2



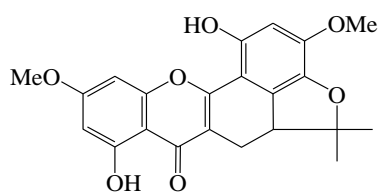
[82] Artoindonesianin A-3



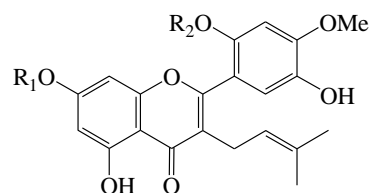
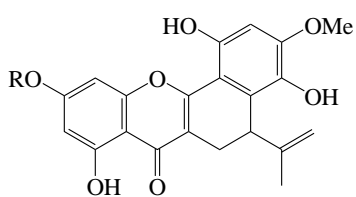
[83] Artoindonesianin B



[84] Artoindonesianin E

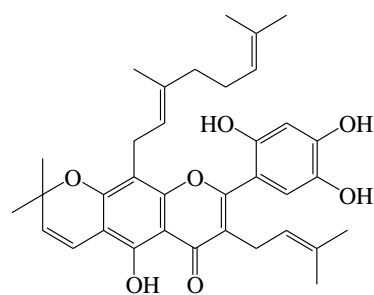


[85] Artoindonesianin M

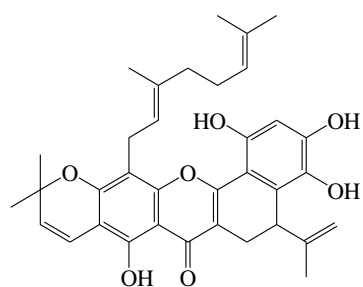
[86] Artoindonesianin Q R<sub>1</sub>=Me R<sub>2</sub>=H[87] Artoindonesianin R R<sub>1</sub>=H R<sub>2</sub>=Me

[88] Artoindonesianin S R=Me

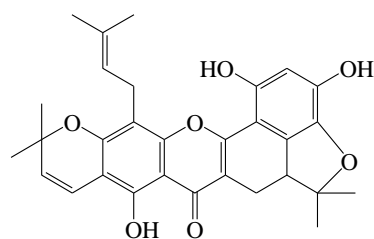
[89] Artoindonesianin T R=H



[90] Artoindonesianin U

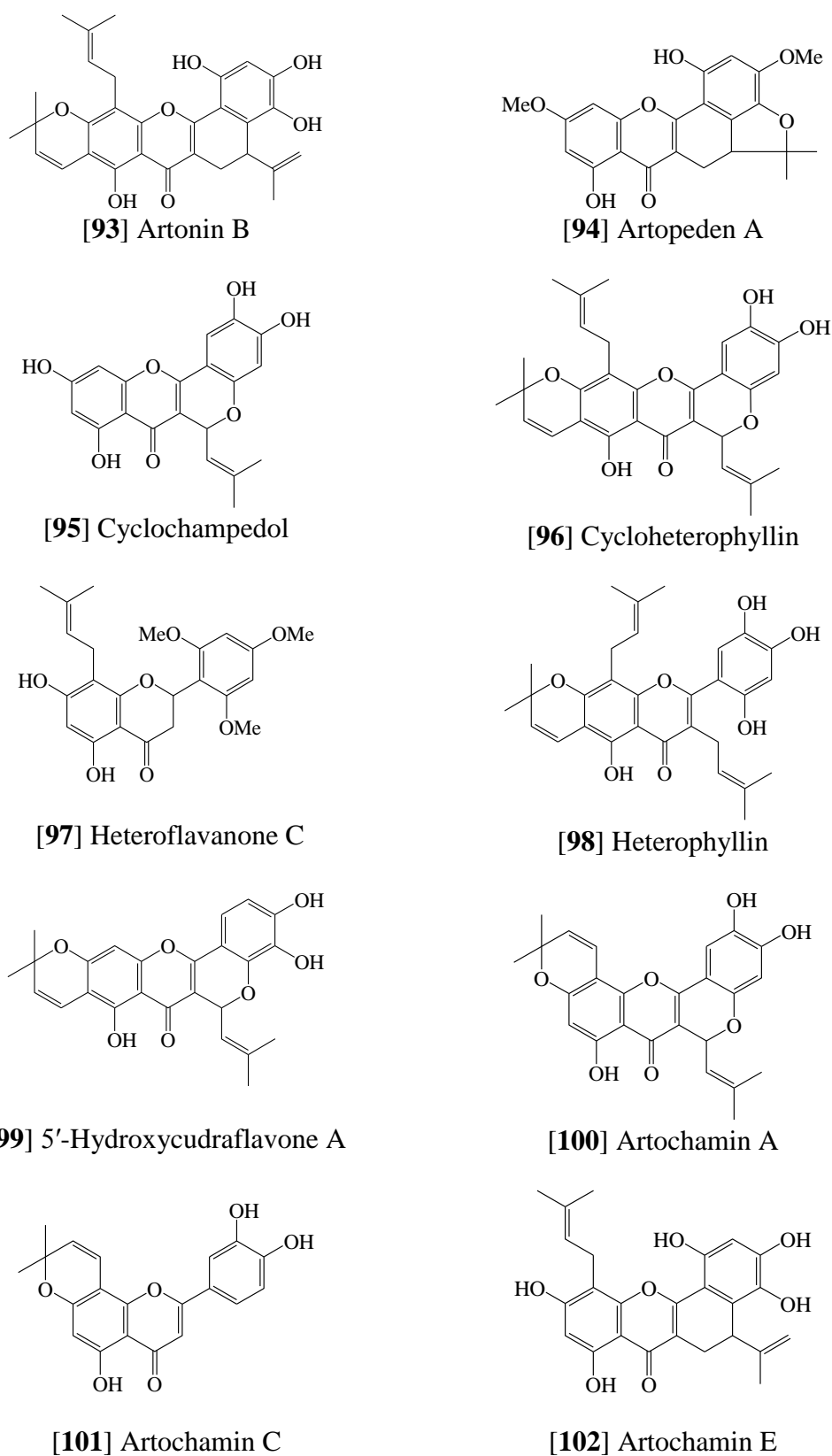


[91] Artoindonesianin V

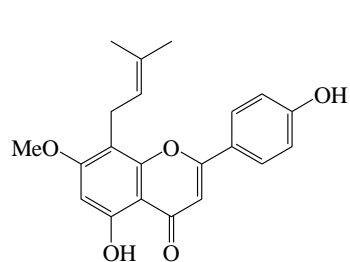


[92] Artonin A

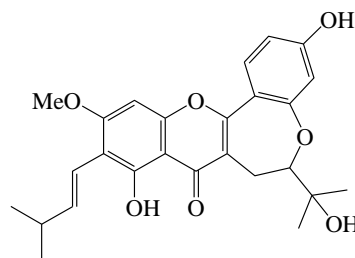
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



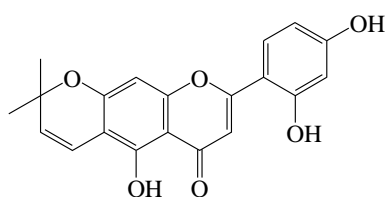
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



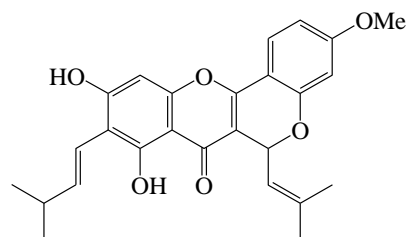
[103] Artonin U



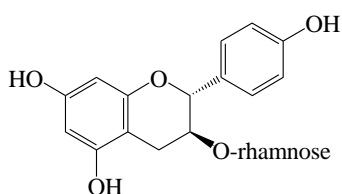
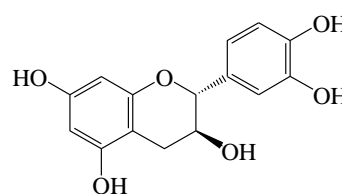
[104] Chaplashin



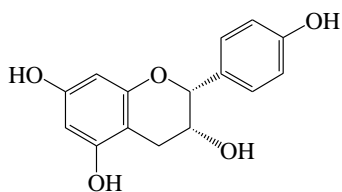
[105] Cycloartocarpesin



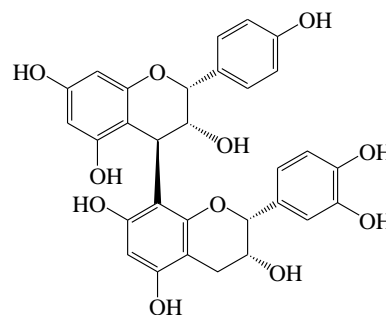
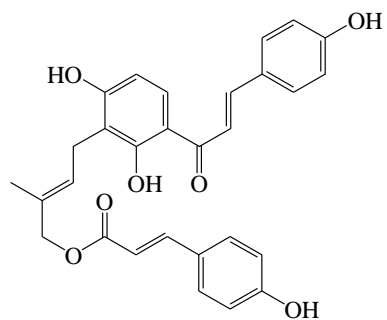
[106] Cycloartocarpin A

[107] Afzelechin-3-O- $\alpha$ -L-rhamnoside

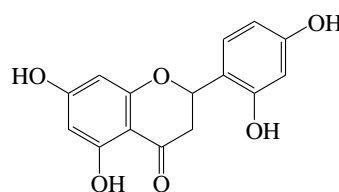
[108] (+)-Catechin



[109] (+)-Epiafzelechin

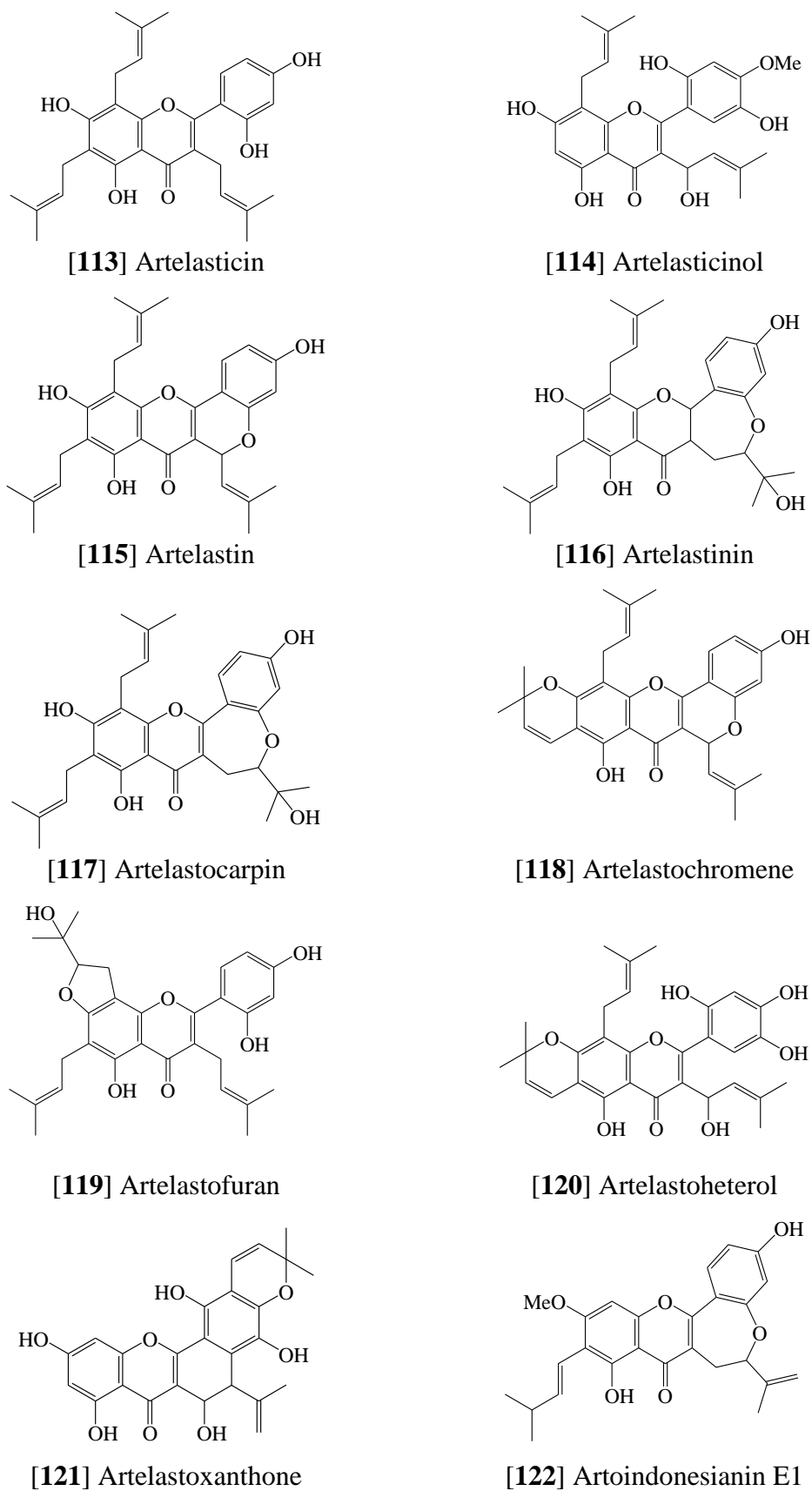
[110] (-)-Epiafzelechin-(4 $\beta$ →8)-epicatechin

[111] Isogemichalcone B

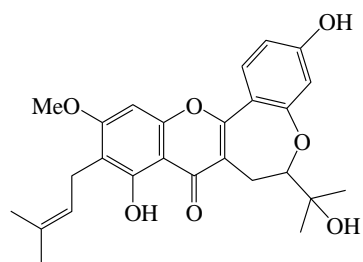


[112] Steppogenin

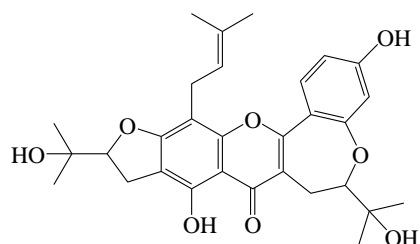
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



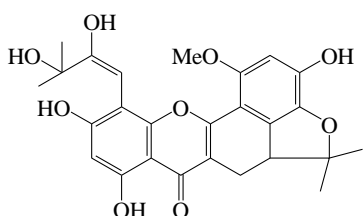
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



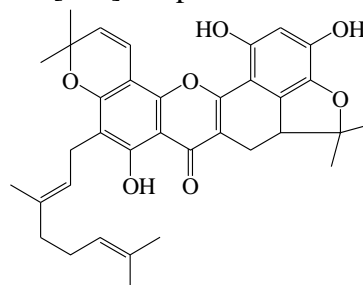
[123] Artonin S



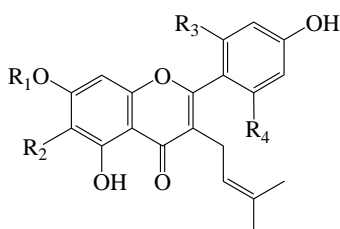
[124] Carpelastofuran



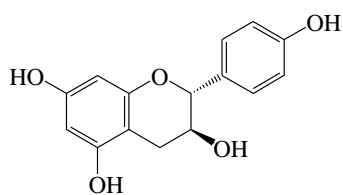
[125] Cycloartelastoxanthendiol



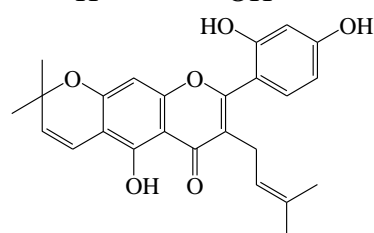
[126] Cycloartelastoxanthone



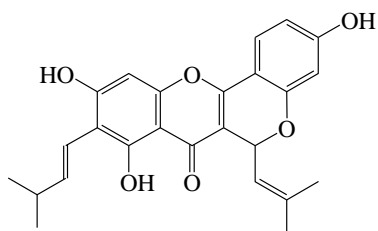
	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
[127] Integrin	Me	H	H	OH
[128] Norartocarpin	H		H	OH
[130] Mulberrin	H		OH	H
[132] Albanin A	H	H	H	OH



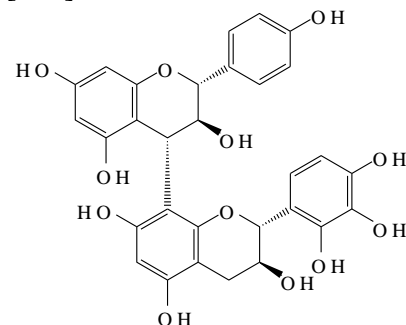
[129] Afzelechin



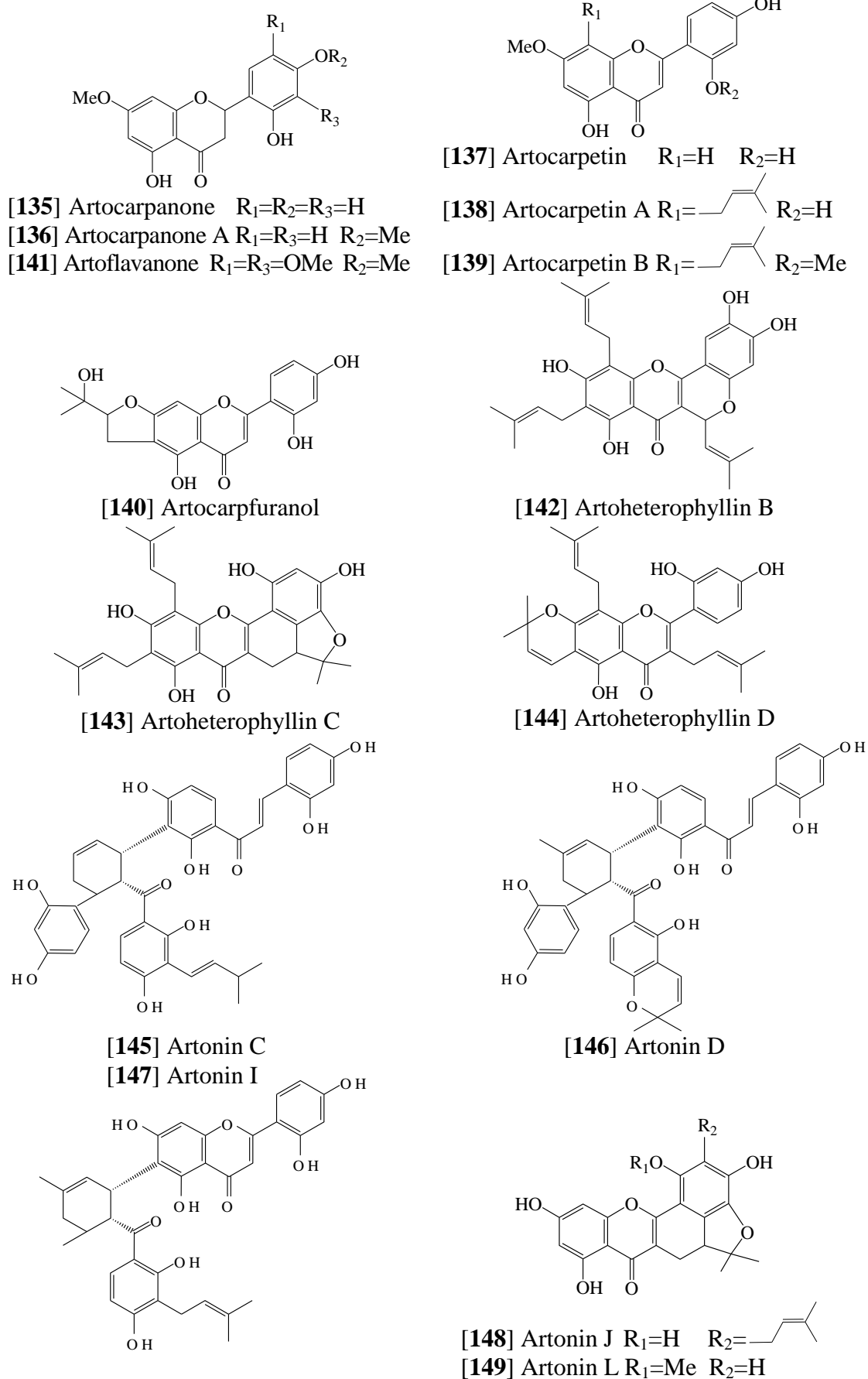
[131] Mulberrochromene



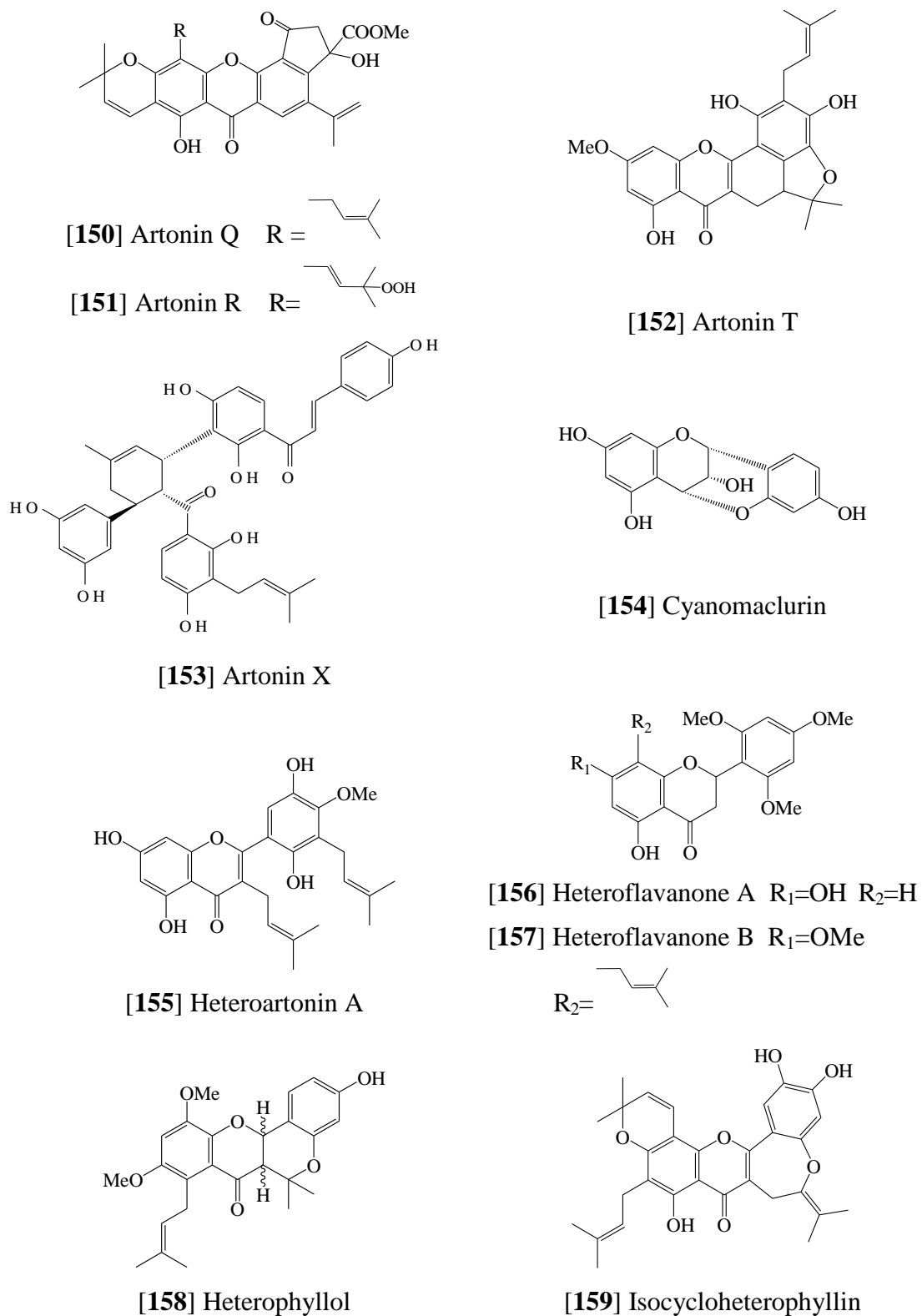
[133] Norcycloartocarpin

[134] Afzelechin-(4 $\alpha$ →8)-catechin

**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

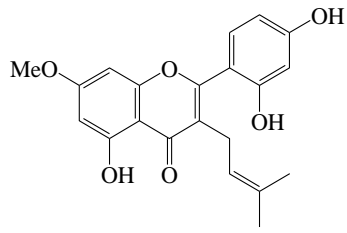


**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

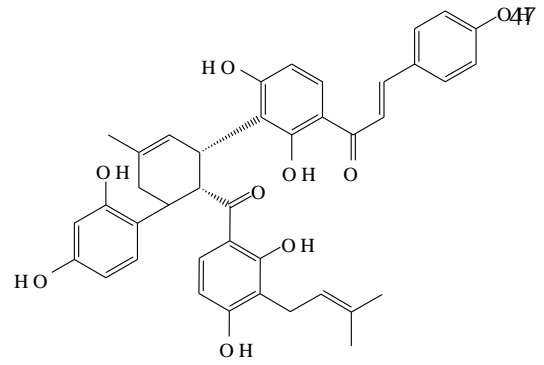


**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

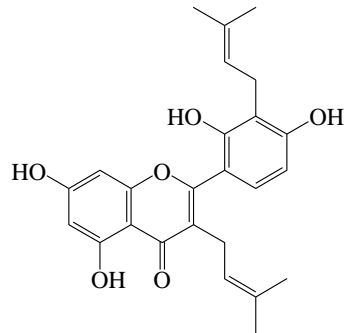




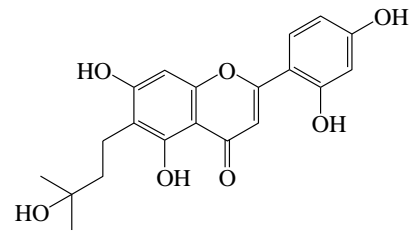
[160] Kuwanon C



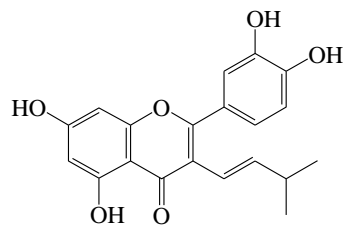
[161] Kuwanon R



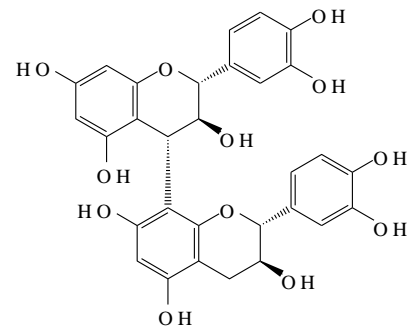
[162] Kuwanon T



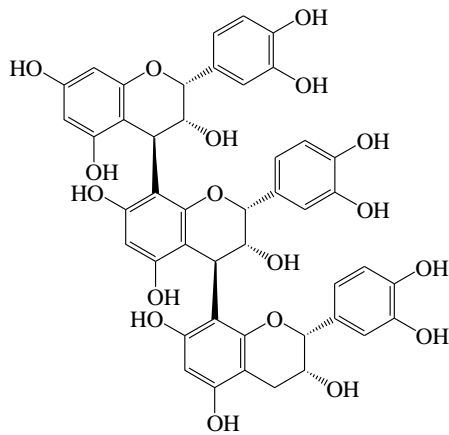
[163] Oxydihydroartocarpesin



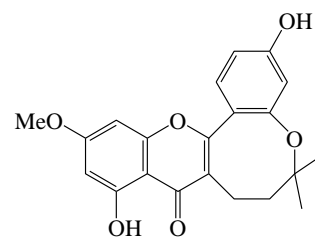
[164] 3-Prenyl luteolin



[165] Procyanidin B-3

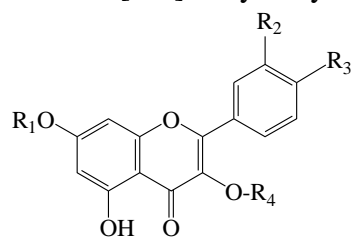
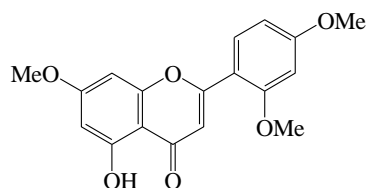
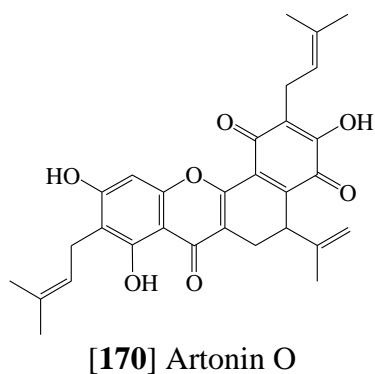
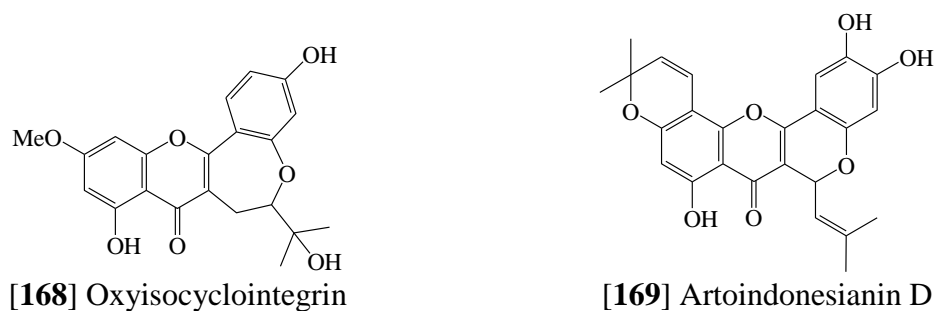


[166] Procyanidin C-1

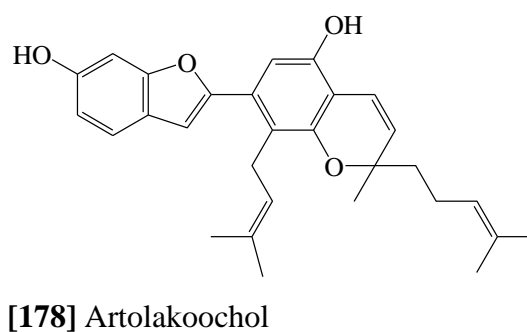
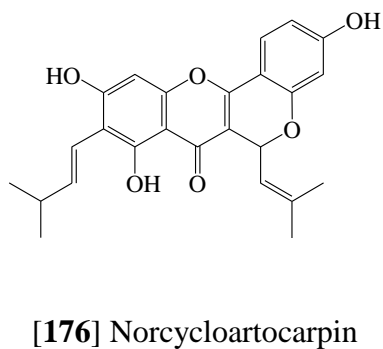


[167] Cyclointegrin

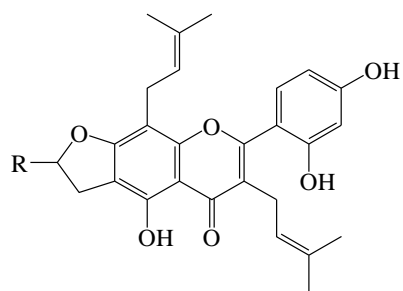
Figure 2 Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
[171] 5,7-Dihydroxyflavone-3- <i>O</i> - $\alpha$ -L-rhamnoside	H	H	H	rhamnose
[173] Galangin-3- <i>O</i> - $\alpha$ -L-rhamnoside	Me	H	H	rhamnose
[174] Galangin-3- <i>O</i> - $\beta$ -D-galactopyranosyl-(1 $\rightarrow$ 4)- $\alpha$ -L-rhamnoside	Me	H	H	galactose-rhamnose
[175] Kaempferol-3- <i>O</i> - $\beta$ -D-xyloside	H	H	OH	xylose
[177] Quercetin-3- <i>O</i> - $\alpha$ -L-rhamnoside	H	OH	OH	rhamnose



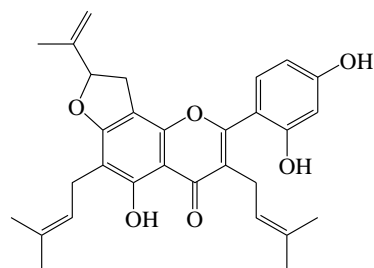
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



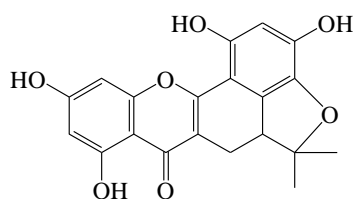
[179] Artoindonesianin G R=



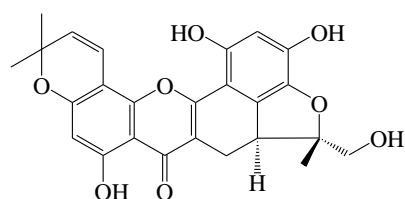
[181] Artoindonesianin I R=



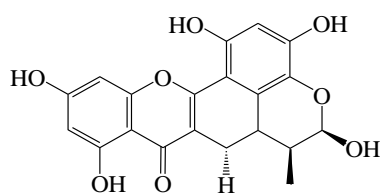
[180] Artoindonesianin H



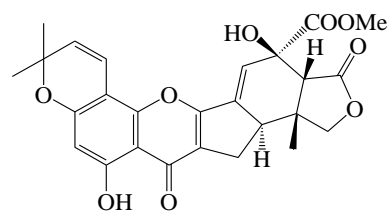
[182] Artoindonesianin P



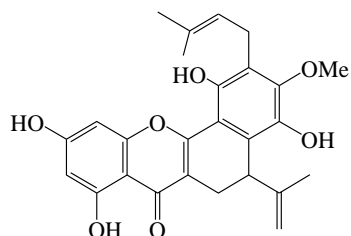
[183] Artoindonesianin Z-1



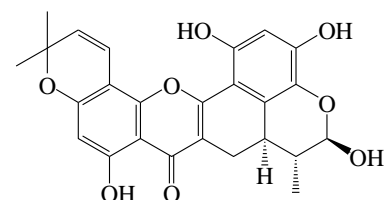
[184] Artoindonesianin Z-2



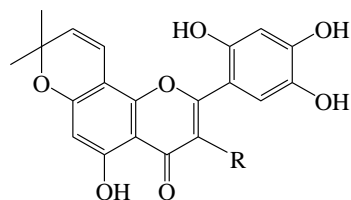
[185] Artoindonesianin Z-3



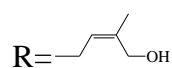
[186] Artoindonesianin Z-4



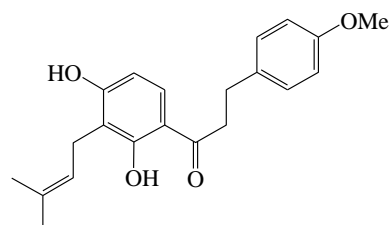
[187] Artoindonesianin Z-5



[188] 14-Hydroxyartonin E

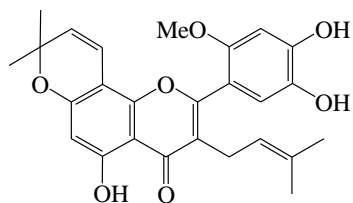


[190] Artobilochromen R=

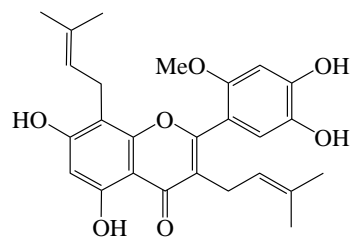


[189] 2',4'-dihydroxy-4-methoxy-3'-prenyl dihydrochalcone

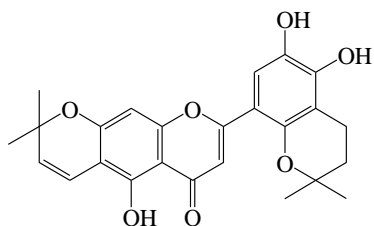
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



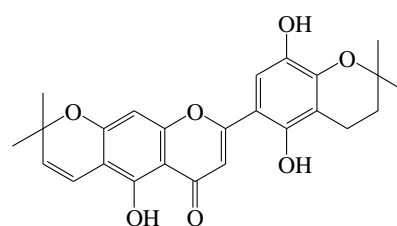
[191] Artonin E 2'-methylether



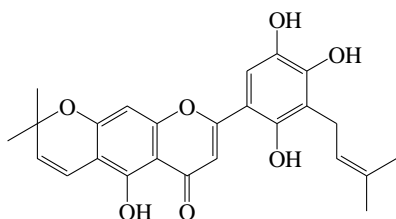
[192] Artonin V 2'-methylether



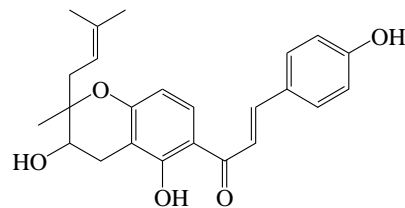
[193] Chromanoartobilochromen A



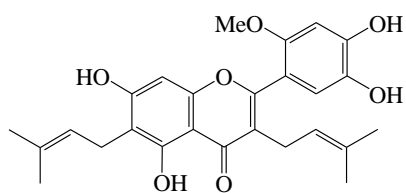
[194] Chromanoartobilochromen B



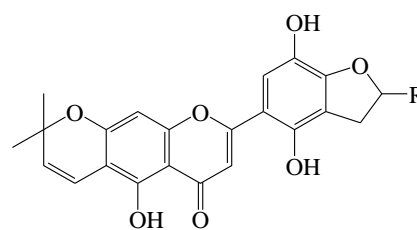
[195] Chromanoartobilochromene



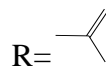
[196] 1-(3,4-Dihydro-3,5-dihydroxy-2-methyl-2-(3-methyl-2-butenyl)-2H-1-benzopyran-6-yl-3-(4-hydroxy phenyl)-2(E)-propen-1-one



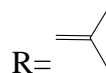
[200] Dihydroisoartonin E 2'-methylether

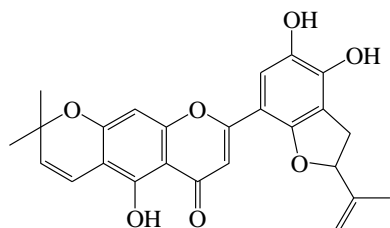


[198] (-)-Dihydrofuranoartobilochromen B-1

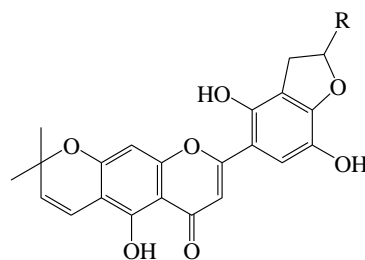


[199] (-)-Dihydrofuranoartobilochromen B-2

Figure 2 Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

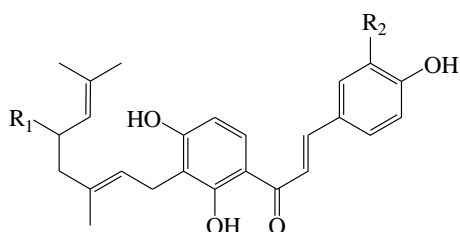
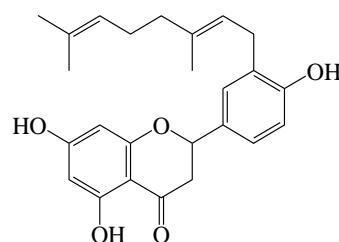


[201] Furanoartobilochromen A

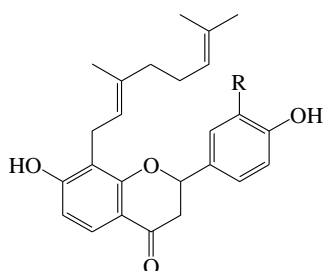
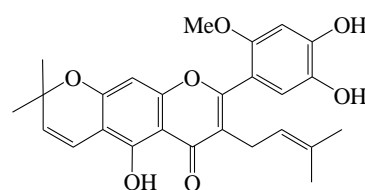


[202] Furanoartobilochromen B-1 R=

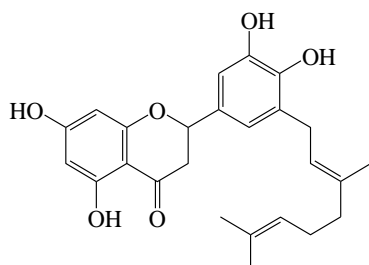
[203] Furanoartobilochromen B-2 R=

[204] 3'-Geranyl-2',3,4,4'-tetrahydroxy  
chalcone R<sub>1</sub>=H R<sub>2</sub>=OH[211] 2,4,4'-Trihydroxy-3-[(2E)-5-methoxy-  
3,7-dimethylocta-2,6-dienyl]chalcone  
R<sub>1</sub>=OMe R<sub>2</sub>=H

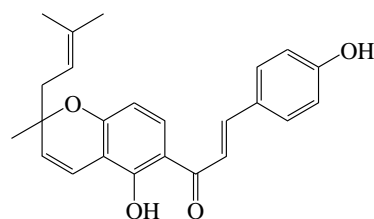
[205] 3'-Geranyl-4',5,7-trihydroxyflavanone

[206] 8-Geranyl-3',4',7-trihydroxy  
flavanone R=OH[207] 8-Geranyl-4',7-dihydroxyflavanone  
R=H

[208] Isoartoinin E 2'-methylether

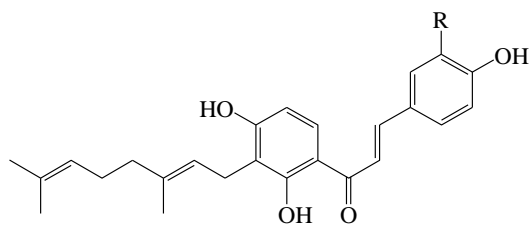


[209] Isonymphaeol-B



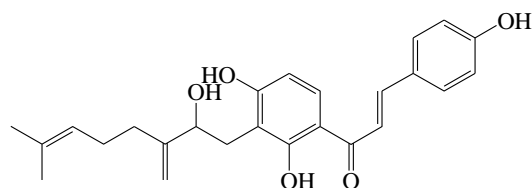
[210] Lespeol

Figure 2 Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

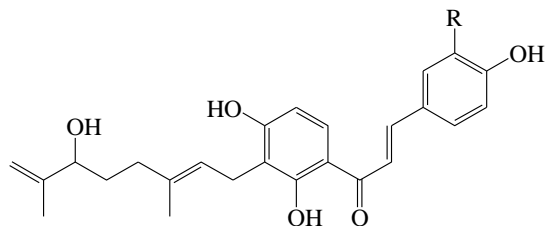


[212] 2',4',4'-Trihydroxy-3'-geranylchalcone  
R=H

[215] 2',3,4,4'-Tetrahydroxy-3'-geranyl  
chalcone R=OH

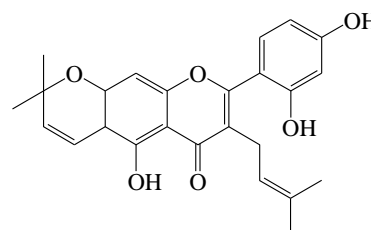


[213] 2',4',4'-Trihydroxy-3'-[6-hydroxy-3,7-  
dimethyl-2(E),7-octadiethyl]chalcone

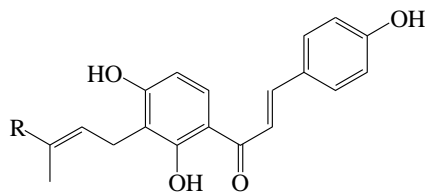


[214] 2',4',4'-Trihydroxy-3'-[2-hydroxy-7-  
methyl-3-methylene-6-octaethyl]  
chalcone R=H

[216] 2',3,4,4'-Tetrahydroxy-3'-[6-hydroxy-  
3,7-dimethyl-2(E),7-octadiethyl]  
chalcone R=OH

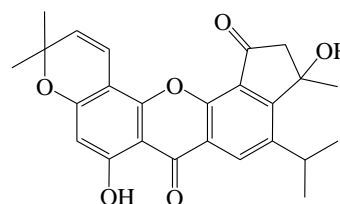


[217] Oxydihydromorusin

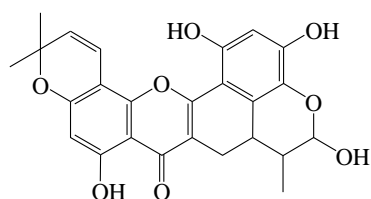


[218] Xanthoangelol R=

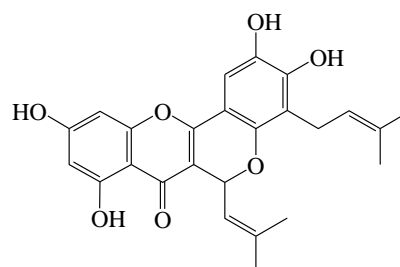
[219] Xanthoangelol B R=



[220] Dihydroartoindolesianin C

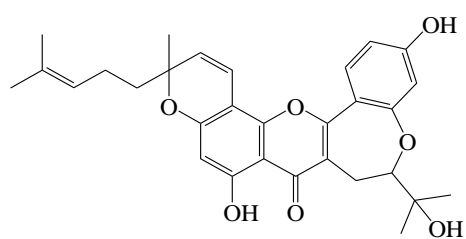


[221] Pyranocycloartobiloxanthone A

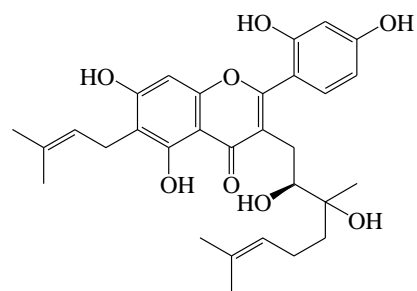


[222] Artosimmin

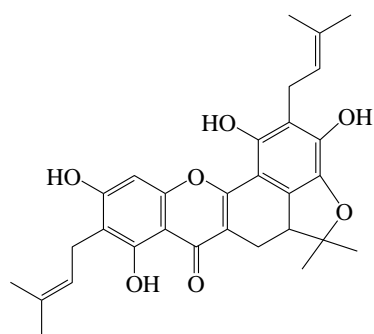
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)



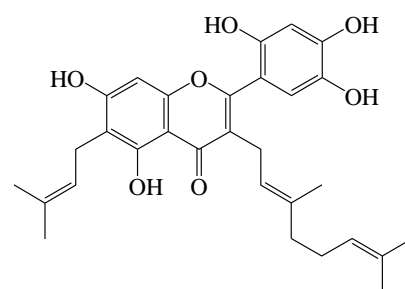
[223] Artocarpol B



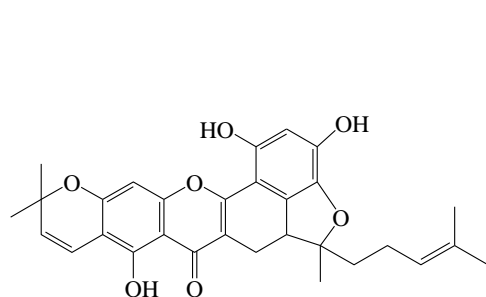
[224] Artocarpol H



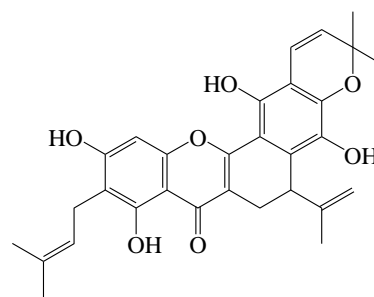
[225] Artonin G



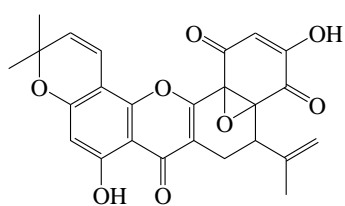
[226] Artonin H



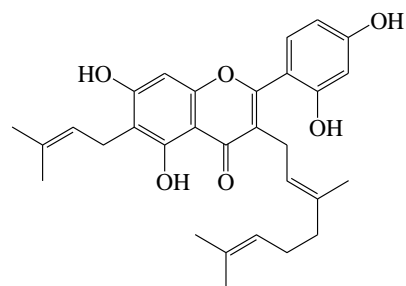
[227] Artonin M



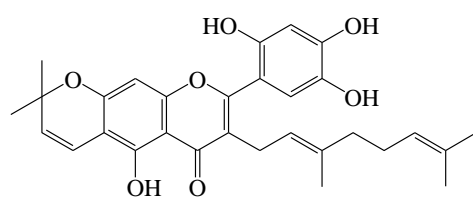
[228] Artonin N



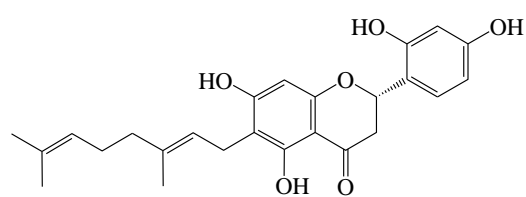
[229] Artonin P



[230] Rubraflavone C

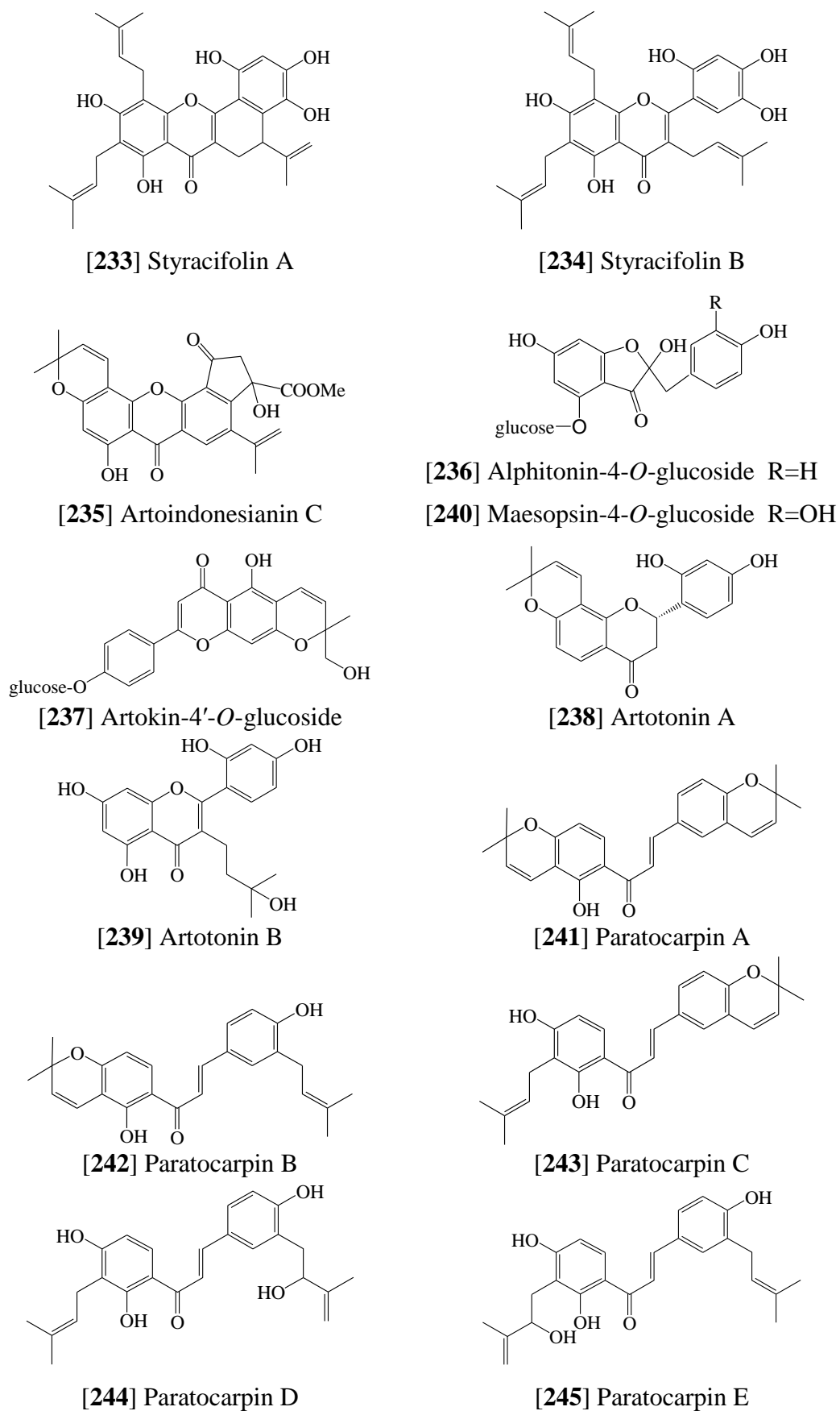


[231] Artoindonesianin L



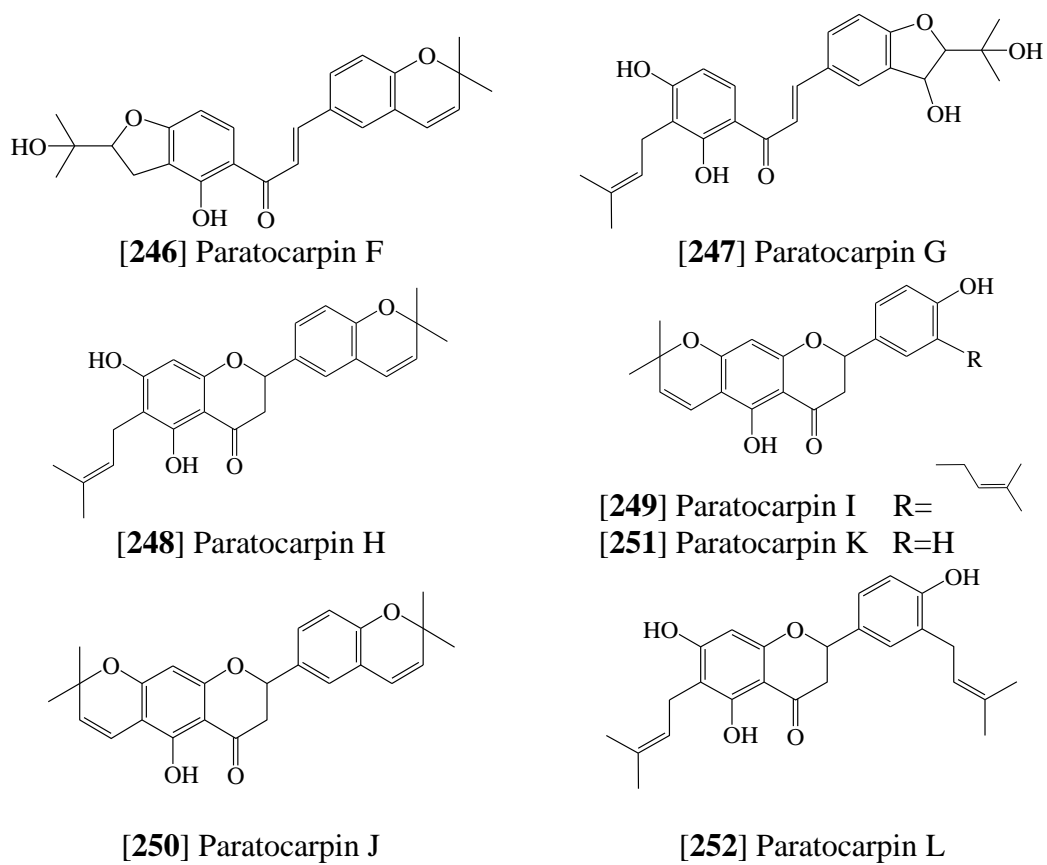
[232] Sepicanin A

**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

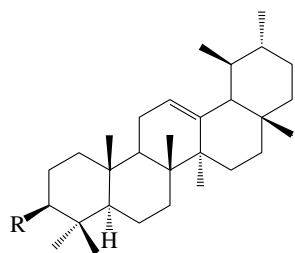
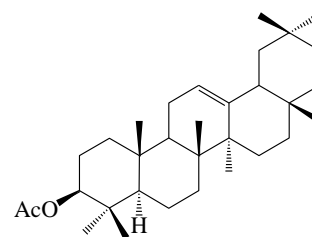
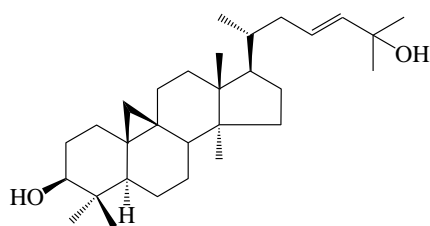
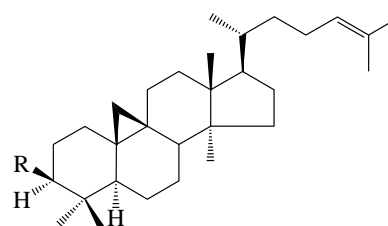
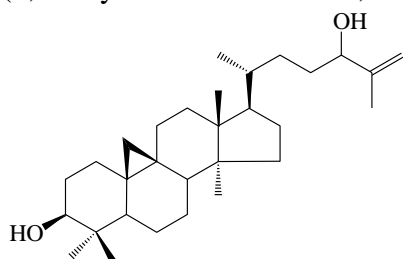
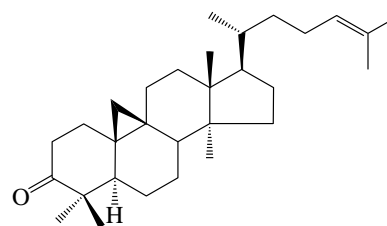


**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

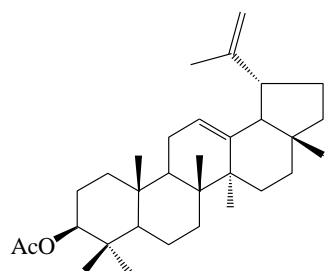




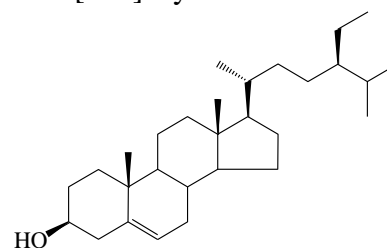
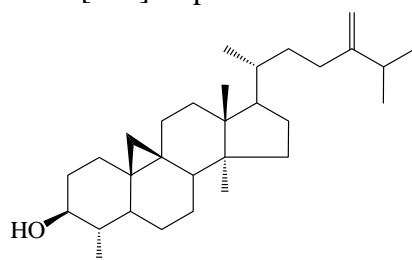
**Figure 2** Structures of flavonoids previously isolated from *Artocarpus* spp. (continued)

[253]  $\alpha$ -Amyrin R=OH[254]  $\alpha$ -Amyrin acetate R=OAc[255]  $\beta$ -Amyrin acetate[256] Cycloart-23-ene-3 $\beta$ ,25-diol  
(9,19-Cyclolanost-3-one-24,25-diol)[257] Cycloartenol  
(Cycloart-24-ene-3 $\beta$ -ol) R=OH  
[260] Cycloartenyl acetate R=OAc[258] Cycloart-25-ene-3 $\beta$ ,24-diol  
(9,19-Cyclolanost-23-ene-3 $\beta$ ,25-diol)

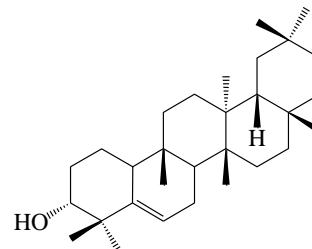
[259] Cycloartenone



[261] Lupeol acetate

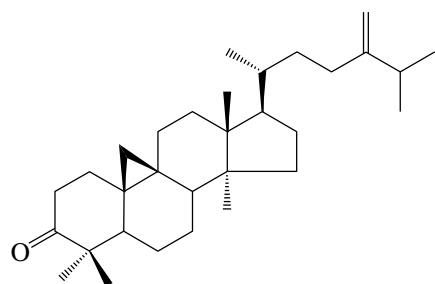
[262]  $\beta$ -Sitosterol

[263] Cycloeucalenol

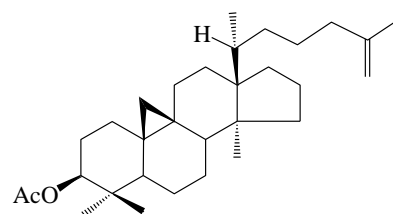


[264] Glutinol

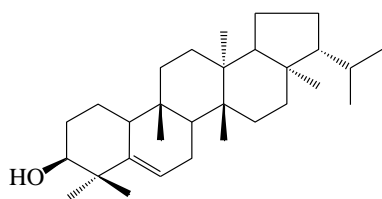
**Figure 3** Structures of triterpenoids and steroids previously isolated from *Artocarpus* spp.



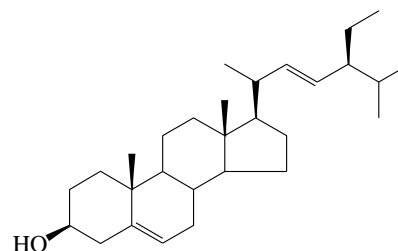
[265] 24-Methylenecycloartenone



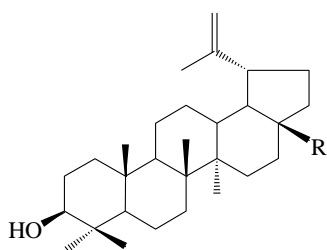
[266] Isocycloartenol acetate



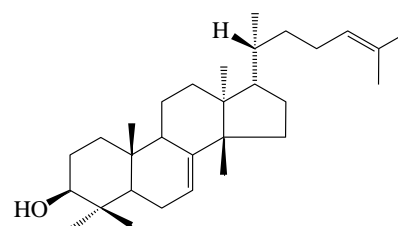
[267] Simiarenol



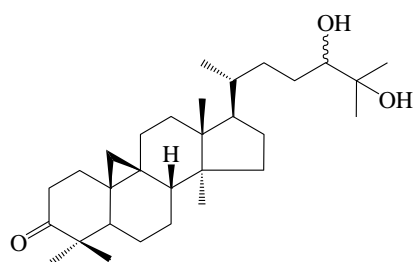
[268] Stigmasterol

[269] Betulin R=CH<sub>2</sub>OH

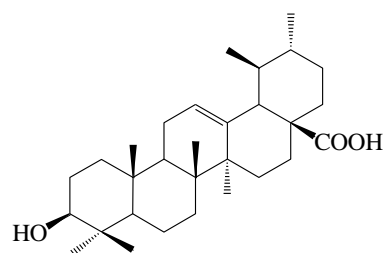
[270] Betulinic acid R=COOH

[274] Lupeol R=CH<sub>3</sub>

[271] Butyrospermol

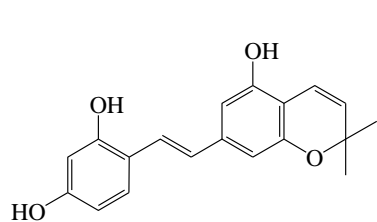


[272] (9,19-Cyclolanost-3-one-24,25-diol)

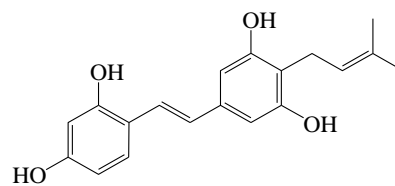
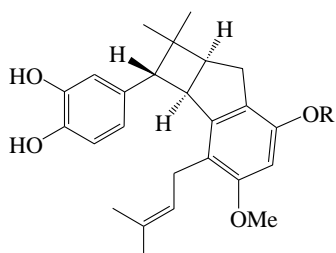
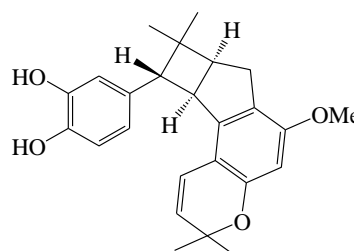


[273] Ursolic acid

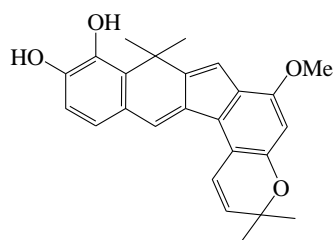
**Figure 3** Structures of triterpenoids previously isolated from *Artocarpus* spp. (continued)



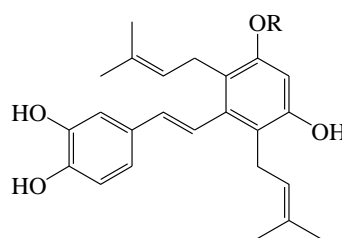
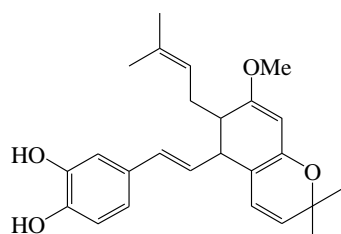
[275] Artocarbene

[276] 4-Prenyloxysesveratrol (*trans*-4-Isopentenyl-3,5,2',4'tetrahydroxystilbene)[277] Artochamin H R=H  
[278] Artochamin I R=Me

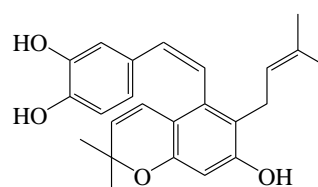
[279] Artochamin J



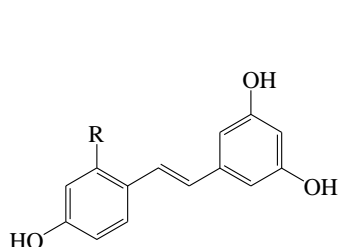
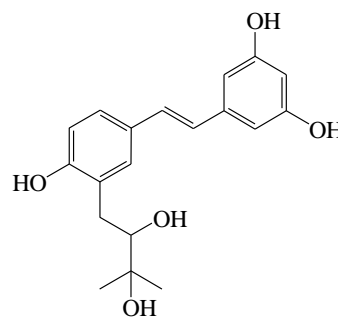
[280] Artochamin K

[281] Artochamin F R=H  
[282] Artochamin G R=Me

[283] Artostilbene A

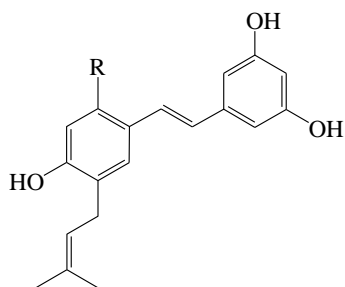


[284] Artostilbene B

[285] Oxyresveratrol R=OH  
[286] Resveratrol R=H

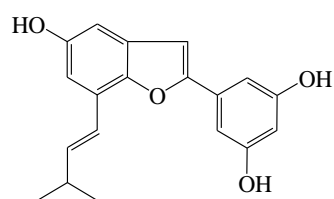
[287] 3-(2,3-Dihydroxy-3-methylbutyl)-resveratrol

**Figure 4** Structures of stilbenoids previously Isolated from *Artocarpus* spp.

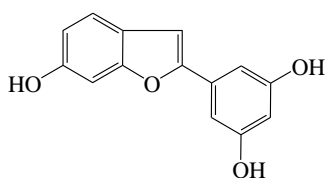


[288] 3-( $\gamma,\gamma$ -Dimethylallyl) oxyresveratrol  
R=OH

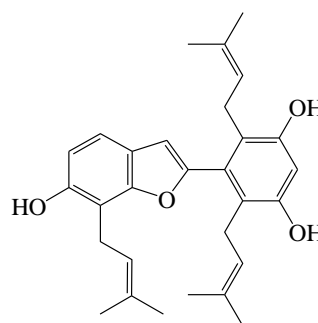
[289] 3-( $\gamma,\gamma$ -Dimethylallyl) resveratrol  
R=H



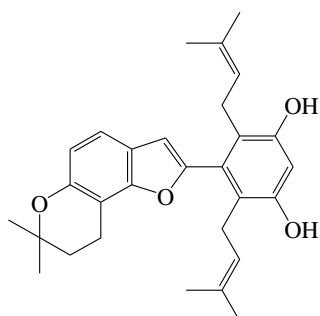
[290] 3-( $\gamma,\gamma$ -Dimethylpropenyl)  
Moracin M



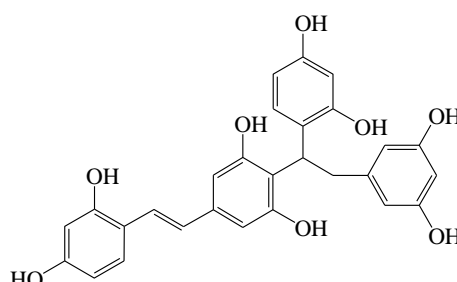
[291] Moracin M



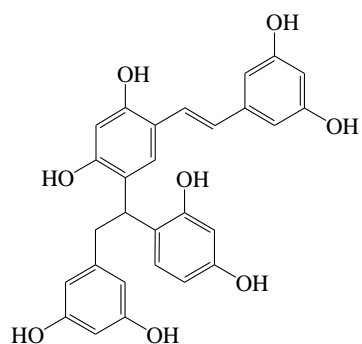
[292] Artoindonesianin X



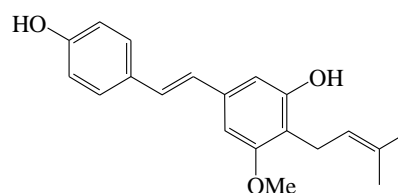
[293] Artoindonesianin Y



[294] Andalasin

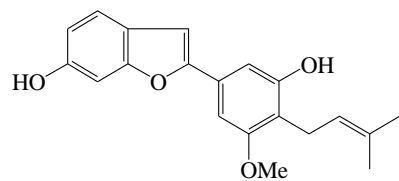


[295] Artogomezianol

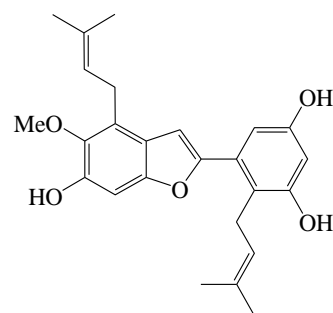


[296] Artoindonesianin N

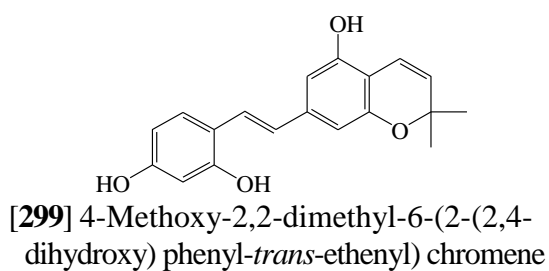
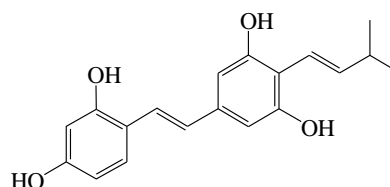
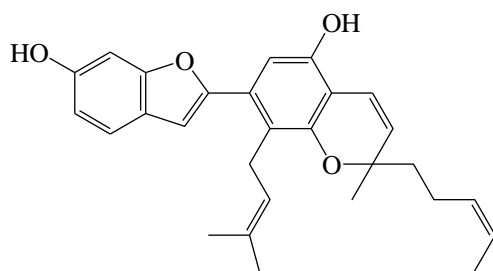
**Figure 4** Structures of stilbenoids previously Isolated from *Artocarpus* spp.



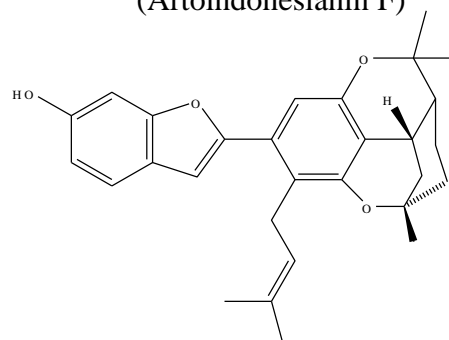
[297] Artoindonesianin O



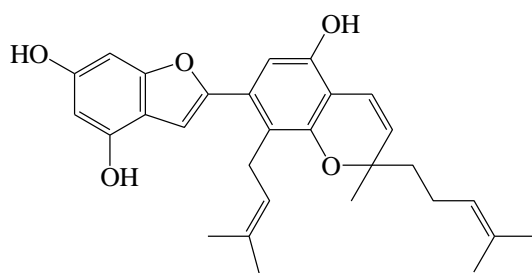
[298] Artoheterophyllin A

[299] 4-Methoxy-2,2-dimethyl-6-(2-(2,4-dihydroxy) phenyl-*trans*-ethenyl) chromene[300] *trans*-4-(3-Methyl-*E*-but-1-enyl)-3,5,2',4'-tetrahydroxystilbene (Artoindonesianin F)

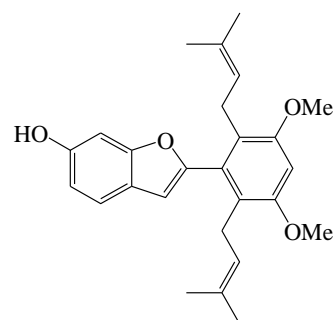
[301] Artolakoochol



[302] Cycloartolakoochol

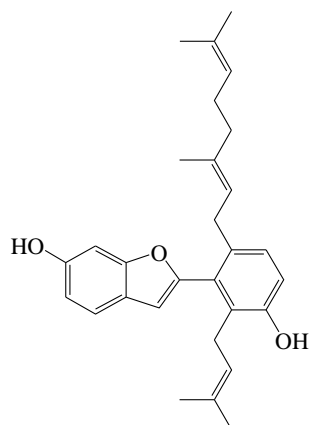


[303] 4-Hydroxyartolakoochol

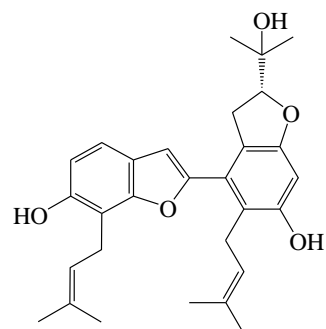


[304] Lakoochin A

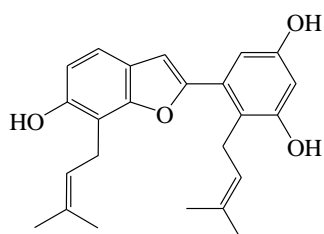
**Figure 4** Structures of stilbenoids previously isolated from *Artocarpus* spp. (continued)



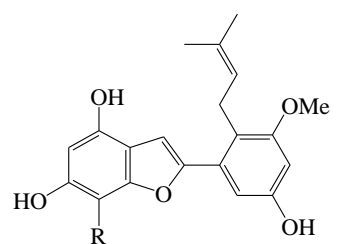
[305] Lakoochin B



[306] Artonitidin A



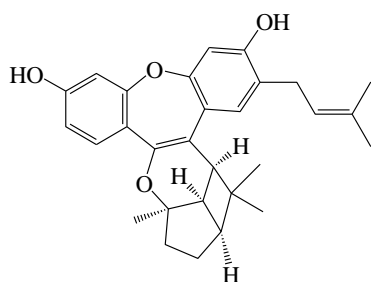
[307] Artonitidin B



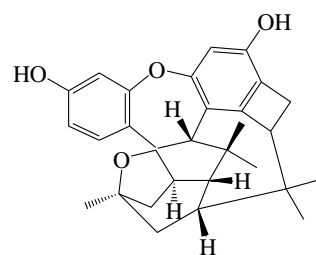
[308] Artopetelin L R=



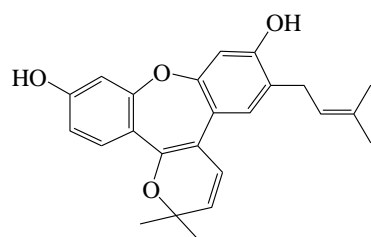
[309] Artopetelin M R=



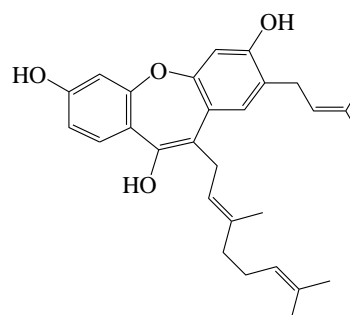
[310] Artocarpol A



[311] Artocarpol C

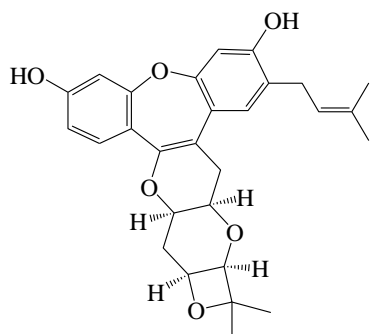


[312] Artocarpol D

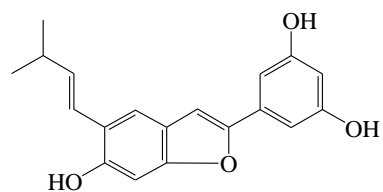


[313] Artocarpol E

**Figure 4** Structures of stilbenoids previously isolated from *Artocarpus* spp.  
(continued)



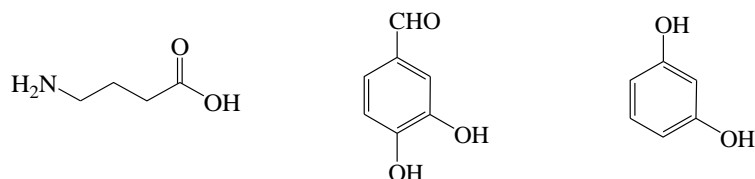
[314] Artocarpol F



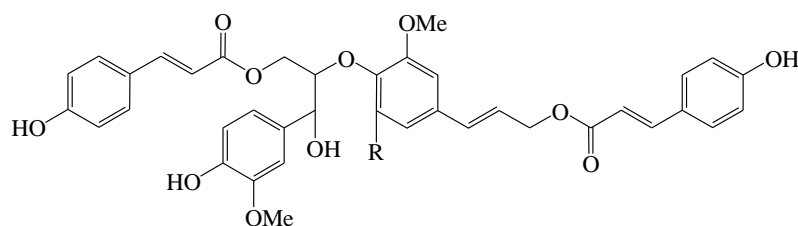
[315] Artotonkin

**Figure 4** Structures of stilbenoids previously isolated from *Artocarpus* spp.  
(continued)



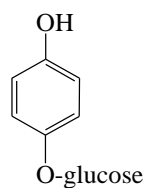


[316]  $\gamma$ -Aminobutyric acid [317] Resorcinol [318]  $\beta$  Resorcylaldehyde

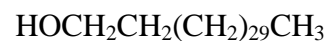


[319] Dadahol A R=OMe

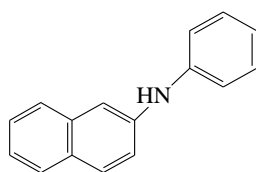
[320] Dadahol B R=H



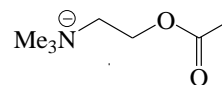
[321] Arbutin



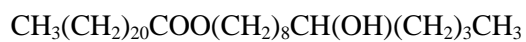
[322] 1-Dotriacontanol



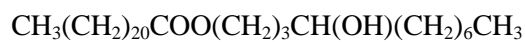
[323] Phenyl- $\beta$ -naphthylamine



[324] Acetylcholine



[329] 9-Hydroxytridecyl docosanoate



[330] 4-Hydroxyundecyl docosanoate



[333] Ricinoleic acid

**Figure 5** Structures of miscellaneous compounds previously isolated from *Artocarpus* spp.

## 2. Traditional Uses and Biological Activities of *Artocarpus* spp.

### 2.1 Traditional uses of *Artocarpus* spp.

Many members of the genus *Artocarpus* have been used in traditional medicine in South-East Asia for the treatment of several diseases and disorders, such as inflammation, malarial fever, ulcer, abscess and diarrhea. *A. heterophyllus* is used in the treatment of various symptoms. For example, the pulp and seeds are used as a cooling tonic and pectoral. The roots are used to treat diarrhea and skin disease, whereas its leaves are used to promote lactation in women. Its latex is mixed with vinegar as a treatment for abscess, and the wood, which is used by monks to dye their robes, has sedative effect in convulsions. (Jagtap and Bapat, 2010). The leaves of *A. altilis* have been used for the treatment of liver cirrhosis, hypertension and diabetes (Wang *et al.*, 2006). In Indonesia, the seeds of *A. champeden* have been used against diarrhea while its roots are used against malaria fever. In western Java, *A. elasticus* has been used to treat inflammation. The bark of this plant is used as a contraceptive, while the young leaves are used to treat tuberculosis (Jagtap and Bapat, 2010). The sap from *A. lowii* is traditionally used as an ointment and as cooking oil (Jamil *et al.*, 2008). Wood is used in making furniture, oars and musical instruments, in building construction and boat, and as a yellow dye for dyeing cotton and silk. (Rujinirum *et al.*, 2005)

*Artocarpus lakoocha* Roxb. is a valuable tropical tree species native to India and used for fruit, furniture, timber, and feed. The lakoocha fruits are generally eaten fresh. The edible fruit pulp is believed to act as a tonic for the liver. (Joshee *et al.*, 2002). The brown powder from aqueous extraction of its wood, called Puag-Haad in Thai, has been used as a traditional anthelmintic drug to treat tapeworm infection. In addition, *A. lakoocha* roots yield a color dye (Charoenlarp *et al.*, 1981; Jagtap and Bapat, 2010).

### 2.2 Biological activities of *Artocarpus* spp.

Compounds from *Artocarpus* plants showed various activities such as antimicrobial, cytotoxic, anti-inflammatory, anti-platelet aggregation and antioxidant properties.

Artocarpin [4] and artocarpesin [3], isolated from *A. heterophyllus*, inhibited the growth of cariogenic bacteria such as *Streptococci mutans*, *Actinomyces viscosus*, *Lactobacillus casei* (Sato *et al.*, 1996). Artonin E [12], isolated from the bark of *A. rigida*,

showed antimicrobial activity against the gram-negative bacteria *Escherichia coli* and *Bacillus subtilis* (Suhartati and Yandri, 2008). Lakoochins A [327] and B [328], from the root of *A. lakoocha*, displayed anti-tuberculosis activity by inhibiting the growth of *Mycobacterium tuberculosis* (Puntumchai *et al.*, 2004). Oxyresveratrol [285] from *A. lakoocha* possessed moderate activity against herpes simplex virus (HSV) type I and type II and against a wild-type human immunodeficiency virus type I (HIV-1/LAI) (Likhitwitayawuid *et al.*, 2005). Morusin [63], cycloartobiloxanthone [38] and artonin E [12] from the root bark of *A. altilis* exhibited moderate antiplasmodial activity against *Plasmodium falciparum* (Boonphong *et al.*, 2007). Cycloartocarpin [39], isocyclomorusin [33] and norartocarpetin [65], compounds isolated from *A. gomezianus*, showed moderate activity against both types of HSV (Likhitwitayawuid *et al.*, 2006).

Prenylated flavones such as artoindonesianins U [90] and V [91] (Syah *et al.*, 2004), artoindonesianins A-2 [81] artoindonesianins A-3 [82] and artoindonesianins T [89], heterophyllin [98] and cudraflavone C [34] from the heartwood of *A. champeden* (Syah *et al.*, 2006b) and cycloartobiloxanthone [38] and artonin E [12] from *A. rigida* (Suhartati and Yandri, 2008) showed strong cytotoxicity against the murine leukemia P-388 cell line. In another report, artosimmin [221], isolated from *A. odoratissimus*, exhibited significant cytotoxicity against HL-60 and MCF-7 cancer cell lines (Ee *et al.*, 2010).

Dihydroartomunoxanthone [48], artochamin B [13] and artocommunol CC [16], isolated from the roots of *A. communis*, significantly inhibited secondary platelet aggregation induced by adrenaline, resulting in the inhibitory effect on thromboxane formation (Weng *et al.*, 2006).

The anti-inflammatory activity of artocarpesin [3], norartocarpetin [65] and oxyresveratrol [284] from *A. heterophyllus* has been demonstrated against production of proinflammatory mediators in lipopolysaccharide (LPS)-activated RAW 264.7 murine macrophage cells (Fang, Hsu, and Yen, 2008a).

Artocarpin [4], isolated from a diethyl ether extract of heartwood of *A. incisus*, possessed potent 5 $\alpha$ -reductase inhibitory effect resulting in the inhibition of the conversion of testosterone into 5 $\alpha$ -dihydrotestosterone. The 5 $\alpha$ -reductase inhibitor acts on androgen receptors which are found in both preputial skin and nongenital skin. Therefore artocarpin might be useful in selective treatment of androgen-dependent disorders such as male pattern alopecia and acne (Shimizu *et al.*, 2000).

Norartocarpin [128], cudraflavone C [34], artotonkin [337], albanin A [132] and artopetelin M [331], isolated from *A. nitidus*, showed inhibitory effect on pancreatic lipase with IC<sub>50</sub> values ranging from 1.8 ± 0.1 to 63.8 ± 3.6 µM (Zhao *et al.*, 2009).

Several phenolic compounds found in plants of this genus were shown to be inhibitors of tyrosinase activity. For example, norartocarpetin [65] and resveratrol [287] from *A. gomezianus* exhibited potent tyrosinase inhibitory activity (Likhitwitayawuid *et al.*, 2000). Artocarpanone [135] inhibited both mushroom tyrosinase enzyme activity and melanin production in B16 melanoma cells (Arung *et al.*, 2006a). Artocarpin [4], cudraflavone C [34], 6-prenylapigenin [77], kuwanon C [160], norartocarpin [128] and albanin A [132] inhibited melanin biosynthesis in B16 melanoma cells without inhibiting the tyrosinase enzyme (Arung *et al.*, 2006b). The heartwood extract of *A. lakoocha* showed both tyrosinase-inhibitory activity *in vitro* and melanin-reducing efficacy in humans (Tengamnuay *et al.*, 2006).

The prenylated flavones cycloheterophyllin [96] and artonins A [92] and B [93] inhibited lipid peroxidation and were antioxidants against DPPH, peroxy and hydroxyl radicals and H<sub>2</sub>O<sub>2</sub> (Ko *et al.*, 1998). Artelastin [115], another prenylated flavone isolated from *A. elasticus*, showed strong O<sub>2</sub><sup>•-</sup> scavenging activity (Cerqueira *et al.*, 2008). In another study, cyclogeracommunin [45], artoflavone A [19], artomunioisoxanthone [21], artocommunol CC [16], artochamin D [14], artochamin B [13] and dihydroartomunoxanthone [48] isolated from the root cortex of *A. communis*, and cycloartelastoxanthone [126], artelastoheterol [120], cycloartobiloxanthone [38] and artonol A [28] isolated from *A. elasticus* all showed inhibition of oxidative DNA damage. Compounds [19], [126], [120] and [38] showed significant DPPH-scavenging activity, while compounds [45] and [28] significantly inhibited xanthine oxidase (XO) activity (Lin *et al.*, 2009). The xanthone pyranocycloartobiloxanthone A [220], isolated from the stem bark of *A. obtusus*, showed strong DPPH-scavenging activity (Hashim *et al.*, 2010).

### 3. Tissue Cultures of *Artocarpus* spp.

With regard to cell cultures of plants of the genus *Artocarpus*, there have been a few reports which focused on micropropagation. For example, a micropropagation protocol for *Artocarpus lakoocha* using leaf disc, nodal segment, and shoot tip explants was described. Explants taken from seedling by air dried seeds of this plant were germinated on sterile MS medium containing 0.8% agar and 0.5% sucrose. Optimum results for

multiple shoot induction were obtained with MS medium supplemented with 1.0 to 2.0 mg/l BA.(Joshee *et al.*, 2002). For *A. altilis*, shoot tip explants taken from mature trees were used. Cultures were initiated from shoot tip explants maintained on MS medium supplemented with 6- benzyladenine (BA, 4.4 mM) (Rouse-Miller and Duncan, 2000). A micropropagation protocol of *A. heterophyllum* using shoot explants cultured on MS medium supplement with BA 4.5 mg/l was reported.

#### **4. Quantitative Analysis of Chemical Constituents of *Artocarpus* spp.**

With regard to analysis of oxyresveratrol in *A. lakoocha*, a few methods for the quantitation of oxyresveratrol in *A. lakoocha* heartwood have been described, and no quantitative analyses of the traditional drug Puag-Haad have been reported. The previously reported analytical methods include gravimetric, high performance liquid chromatographic (HPLC) and capillary zone electrophoretic techniques. These procedures, however, have some disadvantages. The gravimetric method is a laborious process and has a wide margin of errors. The high-performance liquid chromatography (HPLC) and the electrophoresis methods require careful optimization of chromatographic conditions, such as well chosen stationary and mobile phases and proper treatments of the sample and solvent.

The determination of other compounds such as  $\beta$ -sitosterol and Lupeol in *A. lakoocha* has been reported. Quantitative analysis of  $\beta$ -sitosterol and Lupeol in leaf powder was performed on silica gel 60 F<sub>254</sub> HPTLC (High-performance thin layer chromatographic) plate, with toluene: methanol: formic acid, 7.0:2.0:0.3 (v/v/v), as mobile phase and treated with methanolic sulphuric acid reagent. The  $\beta$ - sitosterol and Lupeol contents in leaf powder were found to be 0.176 and Lupeol 0.217 mg/ 500 mg plant power (Vaidya *et al.*, 2011).

## CHAPTER III

### EXPERIMENTAL

#### 1. Plant Materials

##### 1.1 Sample source

Natural samples of *Artocarpus lakoocha* heartwood (ALH1-ALH3) were collected from natural habitats: Chiang Rai (Northern province); Kanchanaburi (Western) and Bangkok (Central). Commercial samples (ALH-4 to ALH-14) were purchased from drugstores as indicated in Table 11. Two samples of the traditional drug Puag-Haad (PH1 and PH2) were obtained from drugstores (Table 11).

##### 1.2 Plant cell cultures

The seeds of *A. lakoocha* Roxb. were collected from the Eastern Botanical Garden (Khao Hin Son) Chachoengsao Province, Thailand in February 2007. The plant was identified by officials of the Eastern Botanical Garden. Authentication was performed by comparison with herbarium specimens (BKF 29303) at the Royal Forest Department, Ministry of Natural Resources and Environment of Thailand. A voucher specimen (SS00101) has been deposited at the Department of Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Chulalongkorn University.

#### 2. Quantitative Analysis of Oxyresveratrol in *A. lakoocha* and 'Puag-Haad'

##### 2.1 Sample Preparation

Each sample of dried heartwood of *A. lakoocha* (5 g) was extracted with 250 ml ethanol in a Soxhlet apparatus for 8 h. The organic layer, after removal of the organic solvent and drying in a desiccator, gave a dried extract. This extract (10 mg) was dissolved in 100 ml methanol to give a sample solution. For Puag-Haad, a sample solution was prepared by dissolving Puag-Haad (1 mg) in 10 ml methanol.

##### 2.2 Preparation of Standard Solutions

Oxyresveratrol was obtained from a previous study (Likhitwittayawut, *et al.* 2005). A standard solution with an accurate concentration of 0.16 mg/ml was prepared by dissolving oxyresveratrol (16 mg) in methanol (100 ml). From this solution, 4 additional standard solutions (0.08, 0.04, 0.02 and 0.01 mg/ml) were prepared by serial dilution.

## **2.3 Instruments and Chromatographic Conditions**

### **2.3.1 HPLC method**

The reversed-phase HPLC system consisted of a pump equipped with a UV detector Shimadzu SPD-10 A $\lambda$ ) and a Phenomenex Luna Phenyl-Hexyl (150 $\times$ 4.60 mm) column connected to security guard (HPLC Guard Cartridge System). The mobile phase contained MeOH:H<sub>2</sub>O (80: 20 v/v) and the flow rate was 0.1 ml/min. The mobile phase was filtered through Millipore filter 0.45 mm, white nylon HNWP 47mm and was degassed before use. The wavelength was set at 254 nm. The samples were manually injected (20  $\mu$ l) and the concentration of oxyresveratrol in the samples was calculated using the average peak area compared between standard and sample after triplicate injections.

### **2.3.2 TLC densitometric method**

A Camag TLC system (Linomat, Switzerland), which included an automatic TLC sampler, a TLC scanner and a CATS software, was used. Chromatography was performed on a TLC aluminum sheet (silica gel 60 F 254 plate, 20  $\times$  10 cm) using methylene chloride/ methanol (85: 15) as the mobile phase. Each sample (20  $\mu$ l) was applied as a band (5  $\times$  0.30 mm) in triplicate. Plates were developed in a glass tank preequilibrated with the mobile phase. The solvent was allowed to run up the plate to a height of 8 cm. Chromatograms were evaluated by measuring the peak area with the TLC scanner in the absorbance mode at 254 nm.

## **2.4 Method Validation**

### **2.4.1. Linearity**

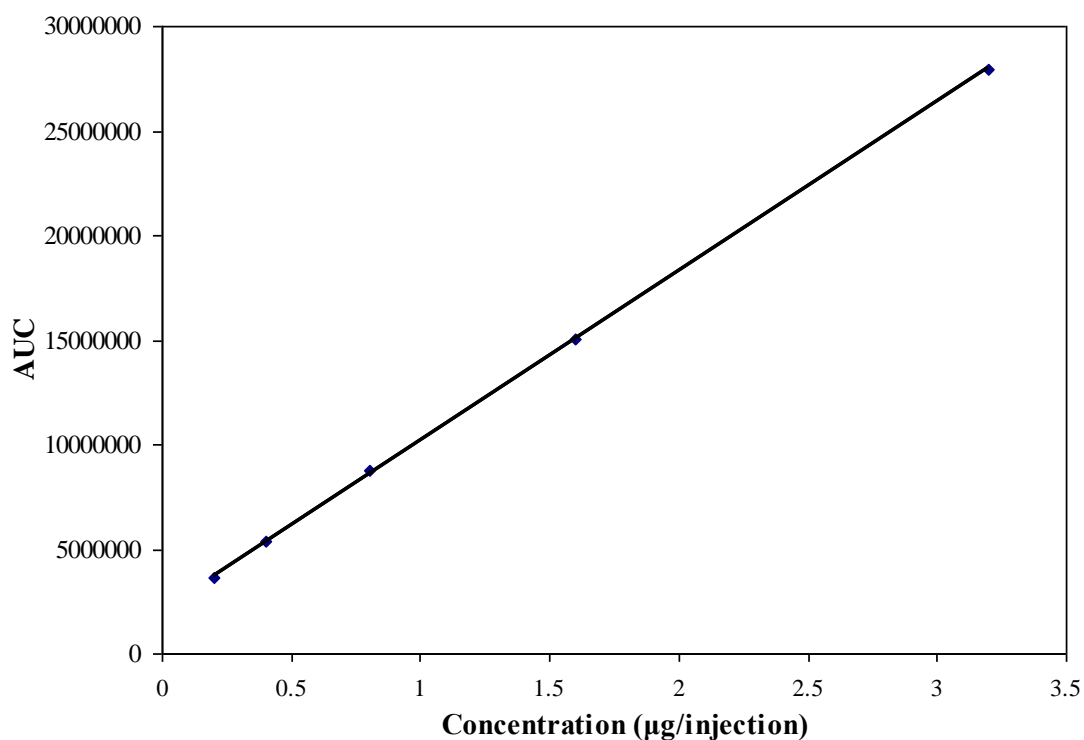
For the HPLC method, linearity was determined over the range of 0.2–3.2  $\mu$ g/injection. Five standard oxyresveratrol solutions with different concentration were prepared by serial dilution. Each of the standard solutions was evaluated by triplicate. For the TLC densitometric method, linearity was determined over the range of 0.2–3.2  $\mu$ g/spot. For both methods, a plot of average area under curve (AUC) versus concentration was obtained. Linearity was expressed as correlation coefficient ( $r^2$ ).

**Table 5** Concentrations of standard solution (HPLC method)

Standard	Concentration		First	Second	Third	Average of AUC
	mg/ml	µg/injection	AUC	AUC	AUC	
S1	0.16	3.2	27863410	27856466	28162246	27960707
S2	0.08	1.6	15101118	15223273	14769249	15031213
S3	0.04	0.8	8859345	8772674	8679064	8770361
S4	0.02	0.4	5402098	5455891	5311082	5389690
S5	0.01	0.2	3562417	3630490	3660043	3617650

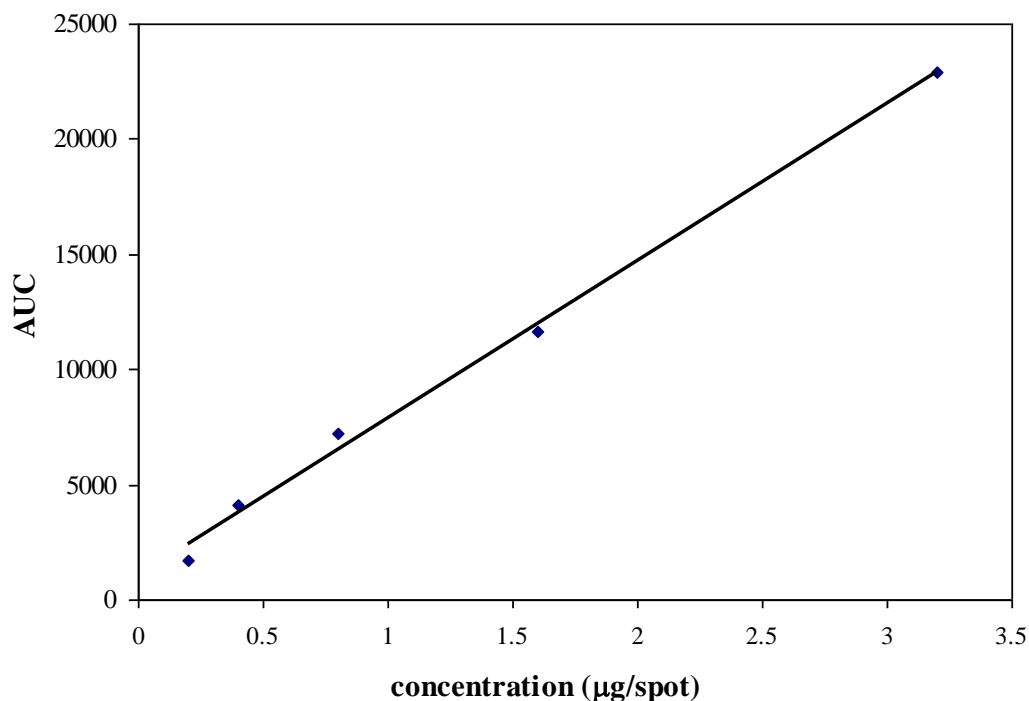
**Table 6** Concentrations of standard solution (TLC densitometric method).

Standard	Concentration		First	Second	Third	Average of AUC
	mg/ml	µg/spot	AUC	AUC	AUC	
S1	0.16	3.2	24056	21779	22895	22910
S2	0.08	1.6	12029	11825	11027	11627
S3	0.04	0.8	7619	7118	6991	7243
S4	0.02	0.4	4194	4165	3929	4096
S5	0.01	0.2	1777	1746	1699	1741

**Figure 6** Calibration curve for standard solution of oxyresveratrol (HPLC method)

$$y = 8075532.25x + 2140264.21 \quad (r^2 = 1.00)$$





**Figure 7** Calibration curve for standard solution of oxyresveratrol (TLC densitometric method)  $Y = 6829.825x + 1054.417$  ( $r^2 = 0.996$ )

#### 2.4.2 Accuracy

Determination of accuracy was done using the recovery method. One milliliter of the *A. lakoocha* heartwood sample solution was spiked with 0.05, 0.25, 0.45, 0.65, 0.85 and 1.05 ml of oxyresveratrol solution in methanol (0.20 mg/ml). Each of these solutions (20 µl/spot for TLC densitometric method; 20 µl/injection for HPLC method) was then analyzed, and the percentage recovery of oxyresveratrol was determined. The theoretical content of oxyresveratrol in each spot/injection was determined from the equation: theoretical content (µg/spot or injection =  $20(8.4 + 0.2a) / (1,000 + a)$ ), where 8.4 is the initial amount of oxyresveratrol (µg) in 1,000 µl of sample solution (from 18 determinations), and a is the volume (µl) of oxyresveratrol solution added.

**Table 7** Accuracy of HPLC method.

Sample number	Oxyresveratrol added (µg)	Theoretical content (µg/injection)	HPLC	
			Experimental content (ug/injection) ± SD	Recovery (%) ± SD
1	10	0.3505	0.3594 ± 0.0171	102.54 ± 4.75
2	50	0.9344	0.9273 ± 0.0389	99.24 ± 4.19
3	90	1.2682	1.2241 ± 0.0533	96.52 ± 4.36
4	130	1.6776	1.6238 ± 0.0555	97.24 ± 3.40
5	170	1.9287	1.9320 ± 0.0782	100.17 ± 4.05
6	210	2.1307	2.0921 ± 0.0847	98.19 ± 4.05

**Table 8** Accuracy of TLC densitometric method.

Sample number	Oxyresveratrol added ( $\mu\text{g}$ )	Theoretical content( $\mu\text{g}/\text{spot}$ )	TLC	
			Experimental content ( $\mu\text{g}/\text{spot}$ ) $\pm$ SD	Recovery (%) $\pm$ SD
1	10	0.3505	0.3493 $\pm$ 0.0143	99.65 $\pm$ 4.07
2	50	0.9344	0.9349 $\pm$ 0.0300	100.05 $\pm$ 3.21
3	90	1.3572	1.37458 $\pm$ 0.0467	101.288 $\pm$ 3.44
4	130	1.6776	1.7204 $\pm$ 0.0540	102.55 $\pm$ 3.22
5	170	1.9287	1.9116 $\pm$ 0.0589	99.11 $\pm$ 3.05
6	210	2.1307	2.1860 $\pm$ 0.0643	102.60 $\pm$ 3.02

### 2.4.3 Precision

The extract (1 mg) was dissolved in the methanol (10 ml) to give a solution with an accurate concentration of 0.1 mg/ml (SM1). The AUC and concentration of oxyresveratrol was calculated from the standard curve. The mean concentration was then calculated from 8 successive injections of the sample (HPLC method). The mean concentration was then calculated from 9 spots of the sample (TLC densitometric method). The precision was expressed as percent coefficient of variation (% CV), which was obtained from the standard deviation (SD).

**Table 9** Precision of HPLC method (n=8)

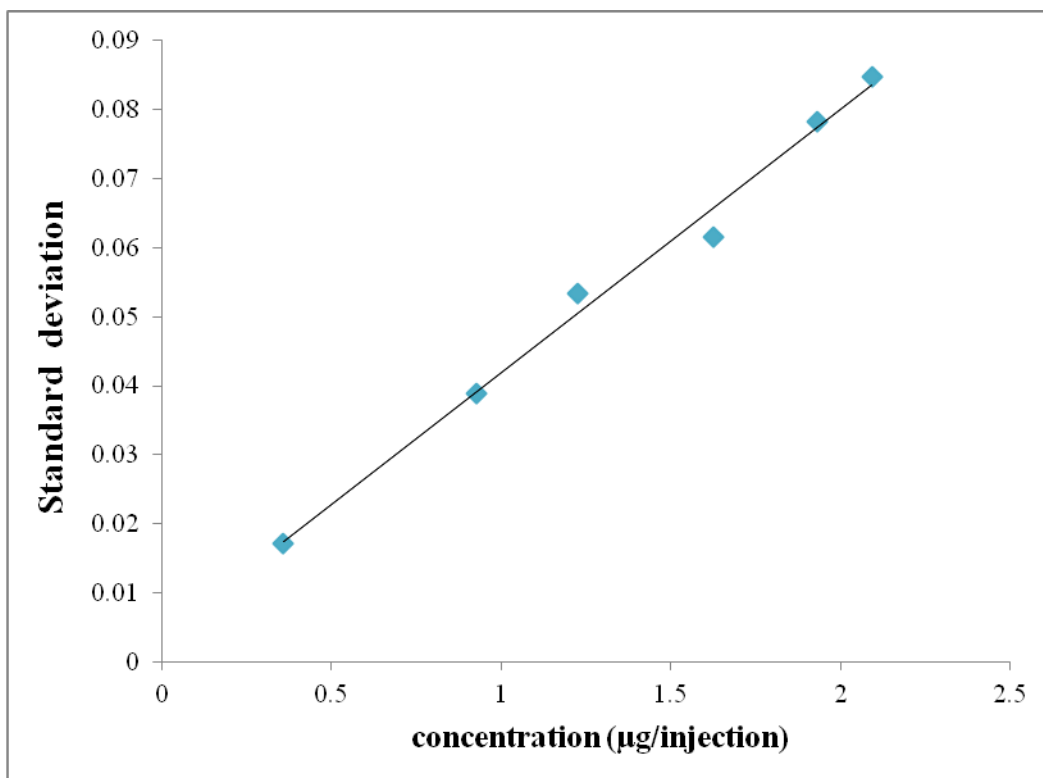
Intra day			Inter day		
Spot number	AUC of SM1 (0.1 mg/ml) 2 $\mu\text{g}$	Concentration ( $\mu\text{g}/\text{injection}$ )	Spot number	AUC of SM1 (0.1 mg/ml) 2 $\mu\text{g}$	Concentration ( $\mu\text{g}/\text{injection}$ )
1	14580816	1.540524	1	14763915	1.563197
2	14504556	1.531081	2	16168672	1.737150
3	16163526	1.736512	3	16069212	1.724833
4	15143544	1.610207	4	15101928	1.605054
5	15658686	1.673998	5	15658835	1.674016
6	15455945	1.648892	6	16163528	1.736513
7	16037244	1.720875	7	14964828	1.588077
8	15285222	1.627751	8	16469443	1.774394
Mean		1.63623	Mean		1.675404
SD		0.07537	SD		0.080139
% CV		4.61	% CV		4.78
Total Mean (intra+inter day)			1.655817		
SD			0.077832		
% CV			4.70		

**Table 10** Precision of TLC densitometric method (n=9)

Intra day			Inter day		
Spot number	AUC of SM1 (0.1 mg/ml) 2 µg	Concentration (µg/spot)	Spot number	AUC of SM1 (0.1 mg/ml) 2 µg	Concentration (µg/spot)
1	11893.44	1.587013	1	11778.51	1.570186
2	12208.91	1.633203	2	12294.66	1.645759
3	12287.44	1.644701	3	12621.54	1.693619
4	12388.02	1.659428	4	12761.39	1.714096
5	12222.95	1.635259	5	13220.68	1.781343
6	12099.19	1.617139	6	13073.30	1.759764
7	12238.30	1.637507	7	13083.51	1.761259
8	12495.18	1.675118	8	13152.39	1.771345
9	12461.05	1.670121	9	13229.23	1.782595
Mean		1.64	Mean		1.72
SD		0.0273	SD		0.0727
% CV		1.66	% CV		4.23
Total Mean (intra+inter day)			1.68		
SD			0.0673		
% CV			4.01		

#### 2.4.4 Limit of Detection and Limit of Quantitation

A plot of experimental concentration against SD was performed to reflect a linear correlation, and the value of the SD on the Y-axis intercept was obtained. This intercept value was then multiplied by a factor of 3 to give the limit of detection, or by 10 to give the limit of quantitation according to the protocols described by Taylor (Taylor, J.K. 1987).



**Figure 8** Plot of oxyresveratrol content in spiked samples vs. SD (HPLC method)

$$Y = 0.0383x + 0.0036 \quad (r^2 = 0.9913)$$

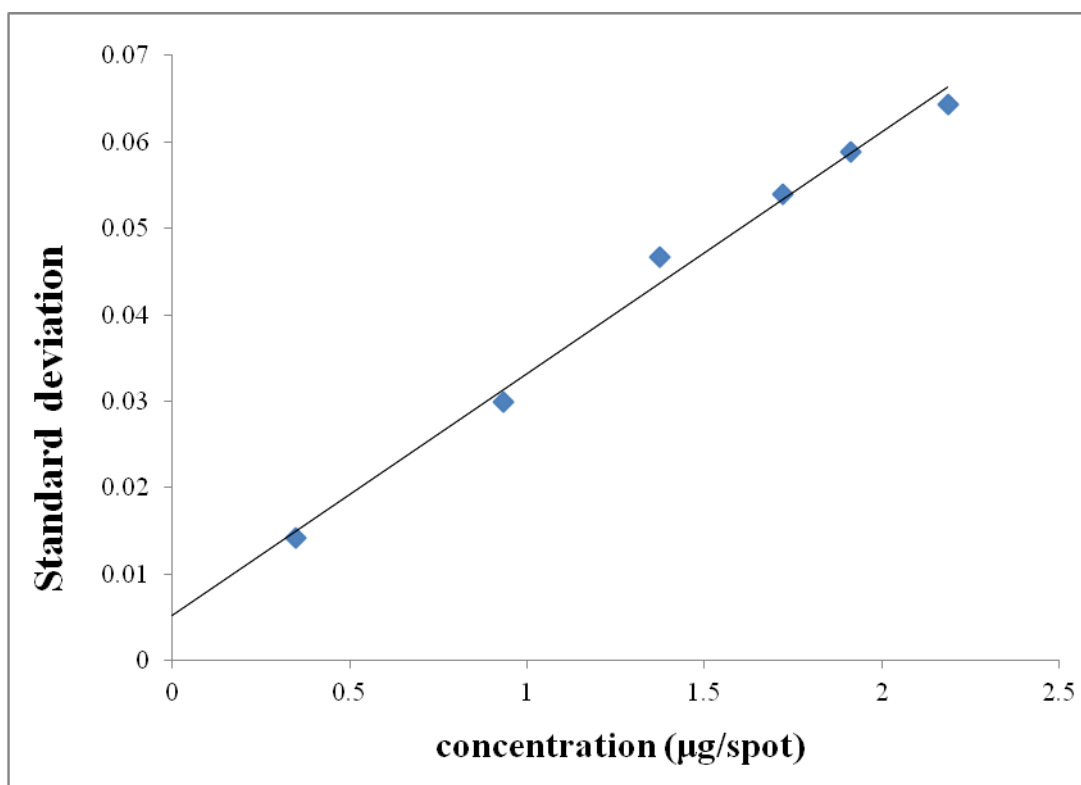
$$Y\text{-intercept} = 0.0036 \text{ } \mu\text{g/injection}$$

$$\text{LOD} = 3 \times Y\text{-intercept}$$

$$\text{LOD} = 3 \times 0.0036 = 0.0108 \text{ } \mu\text{g/injection}$$

$$\text{LOQ} = 10 \times Y\text{-intercept}$$

$$\text{LOQ} = 10 \times 0.0036 = 0.036 \text{ } \mu\text{g/injection}$$



**Figure 9** Plot of oxyresveratrol content in spiked samples vs. SD (TLC densitometric method)

$$Y = 0.0280x + 0.0052 \quad (r^2 = 0.9909)$$

$$Y\text{-intercept} = 0.0052 \text{ } \mu\text{g/spot}$$

$$\text{LOD} = 3 \times Y\text{-intercept}$$

$$\text{LOD} = 3 \times 0.0052 = 0.0156 \text{ } \mu\text{g/spot}$$

$$\text{LOQ} = 10 \times Y\text{-intercept}$$

$$\text{LOQ} = 10 \times 0.0052 = 0.052 \text{ } \mu\text{g/spot}$$

### 3. *In Vitro* Cultures of *A. lakoocha*

#### 3.1 Callus cultures

The seeds of *A. lakoocha* were surface sterilized and germinated on free hormone MS medium (Murashige and skoogs, 1962), B5 medium (Gamborg, 1968), SH medium (Schenk and Hildebrandt 1939) and Woody plant medium (Lloyd and Mc. Cown, 1980). After seedling 1-2 months, the calli derived on the explants were transferred to the same medium with various combinations of 2,4-dichlorophenoxyacetic acid (2,4-D) and benzyladenine (BA); 1-Naphthaleneacetic acid (NAA) and 6-Furfurylaminopurine (kinetin). Explants from 1–2 months old

sterile seedlings were used for induction of callus cultures. Callus formed after 3 weeks from explants was then subcultured onto fresh medium every 3 weeks. WPM medium was found to be most suitable for callus induction and seed germination. Explants were inoculated on agar solidified (0.8%, w/v) WPM basal medium with 2% sucrose and incorporated with 1 mg/l 2,4-dichlorophenoxyacetic acid (2,4-D) and 1 mg/l benzyladenine (BA). All the cultures were incubated at  $25 \pm 2$  °C under fluorescent light (16/8 photoperiod). Cultures were maintained by subculturing and transferring after every 25 days. The induced calli were subcultured on the same medium in 3 months for mass production and then transferred to the dark for over 6 months for accumulated secondary metabolites.

### **3.2 Cell Suspension Cultures**

Cell suspension cultures were initiated by inoculation of 1 g fresh weight of friable callus into a 125 ml Erlenmeyer flask containing 25 ml of liquid WPM medium supplemented with 1 mg/l 2,4-dichlorophenoxyacetic acid (2,4-D), 1 mg/l benzyladenine (BA) and 20 g/l sucrose. The flasks were placed on the rotary shaker (100 rpm) at  $25 \pm 2$  °C in darkness. Subculturing was done every 3 weeks.

### **3.3 Study on the Relationship between the Growth of Callus Cultures and Secondary Metabolites Production in Callus Cultures**

Calli derived from the aerial part of sterilized explants were maintained on WPM medium supplemented with 1 mg/l 2,4-dichlorophenoxyacetic acid (2,4-D), 1 mg/l benzyladenine (BA) and 20 g/l sucrose. Subculturing was done every 3 weeks over 6 months in dark condition. Practically, 1g FW of calli was transferred to the fresh medium. The cultures were grown under the dark condition at  $25 \pm 2$  °C. Each treatment will be done in three replicates.

The dry weight of callus culture were determined at different ages of 5, 10, 12, 15, 17, 20, 22 and 25 day-after-culture. The secondary metabolite contents were also measured using TLC densitometric method as described above.

## **4. General Techniques for Isolation of Secondary Metabolites in Callus Cultures**

### **4.1 Analytical Thin-layer Chromatography (TLC)**

Technique	:	One dimension, ascending
Adsorbent	:	Silica gel 60 F <sub>254</sub> (E. Merck) percoated plate
Layer thickness	:	0.25 mm
Distance	:	8 cm

Temperature	:	Laboratory temperature (30-35 °C)
Detection	:	1. Ultraviolet light (254 and 365 nm) 2. 50% H <sub>2</sub> SO <sub>4</sub> in H <sub>2</sub> O and heating at 105 °C for 10 min.

## 4.2 Column Chromatography

### 4.2.1 Vacuum Liquid Column Chromatography

Adsorbent	:	Silica gel 60 particle size 0.063-0.200 nm (70-230 mesh ASTM) (Merck.)
Packing method	:	Dry packing
Sample loading	:	The sample was dissolved in a small amount of organic solvent, mixed with a small quantity of adsorbent, triturated, dried and then placed gently on top of the column.
Detection	:	Fractions were examined by TLC observing under UV light (254 and 365 nm), and 50% H <sub>2</sub> SO <sub>4</sub> spraying reagent in hot condition.

### 4.2.2 Porus Polymer gel Chromatography

Adsorbent	:	Diaion HP-20
Packing method	:	Wet packing
Sample loading	:	The sample was dissolved in water and then placed on top of the column.
Detection	:	Fractions were examined by TLC observing under UV light (254 and 365 nm) and spraying with sulfuric acid and then heating at 105 °C for 10 min.

### 4.2.3 High Pressure Liquid Chromatography (HPLC)

Semi-preparative column	:	1. TSK-Gel ODS-100V, 5 µm, 2 × 25 cm (Tosoh Chemicals Co., Ltd.) 2. Capcell Pak ODS, 5 µm, 2 × 25 cm (Shiseido Fine Chemical Co., Ltd.)
Analytical column	:	Capcell Pak ODS, 5 µm, 0.46 × 25 cm
Flow rate	:	1. 1 ml/min for analytical column 2. 9 ml/min for semi-preparative column
Mobile phase	:	methanol or acetonitrile in water
Sample preparation	:	The sample was dissolved in a small amount of

		eluent and injected into the column.
Injection volume	:	10 $\mu\text{l}$ for analytical column 500 $\mu\text{l}$ for semi-preparative column
Pump	:	1. 880-PU (JASCO) for semi-preparative column and analytical column 2. PU-2089 Plus (JASCO) for semi-preparative column
Detector	:	UV-Vis detector 875-UV (JASCO) variable- wavelength
Recorder	:	1. Hitachi 561 2. Sekonic SS-250 F
Temperature	:	Room temperature

### 4.3 Spectroscopy

#### 4.3.1 Ultraviolet (UV) Absorption Spectra

UV (in methanol) spectra were obtained on a Milton Roy Spectronic 3000 Array spectrophotometer (Pharmaceutical Research Instrument Center, Faculty of Pharmaceutical Science, Chulalongkorn University).

#### 4.3.2 Mass Spectra

Mass spectra were recorded on a Bruker micro TOF mass spectrometer (National Center for Genetic Engineering and Biotechnology).

#### 4.3.3 Proton and Carbon-13 Nuclear Magnetic Resonance ( $^1\text{H}$ and $^{13}\text{C}$ -NMR) Spectra

$^1\text{H}$ -NMR (400 MHz) and  $^{13}\text{C}$ -NMR (100 MHz) spectra were recorded on a JEOL JMN- $\alpha$  400 instrument. (School of Pharmaceutical Sciences, University of Shizuoka, Shizuoka, Japan)

Solvent for NMR spectra were deuterated acetone, deuterated methanol. Chemical shifts were given in  $\delta$  (ppm) with the solvent as an internal standard.

### 4.4 Optical Rotations

Optical rotations were measured on a JASCO DIP-360 digital polarimeter. (School of Pharmaceutical Sciences, University of Shizuoka, Shizuoka, Japan)

### 4.5 Solvents

Throughout this work, all organic solvents were of commercial grade and were redistilled prior to use.



## 5. Isolation and Extraction of Callus Cultures

A study on the isolation and extraction used callus cultures in day 17 of callus growth. The dried callus of *Artocarpus lakoocha* (40 g) was extracted three times with ethanol (3 × 500 ml) by maceration method. The extracts were combined and evaporated under reduced pressure to yield 8 g of viscous mass.

The water soluble extract was subjected to HP-20 column chromatography, using 50% methanol (4 l), 75 % methanol (7 l) and 100% methanol (7 l), respectively as elutents, to yield 50% methanol fraction (0.597 g), 75 % methanol fraction (1.074 g) and 100% methanol fraction (2.016 g) eluate, respectively after removal of the solvent.

The 50 % methanol fraction (0.597 g) was chromatographed on a silica gel column, using chloroform-methanol (95:5) 500 ml, (9:1) 200 ml, (8:2) 300 ml and then fractionated using chloroform-methanol-water (8:2:0.2) 300 ml. Fractions were collected and pooled by TLC analysis to afford 8 combined fractions.

From the combined fractions, fraction 6 (0.133 g) was subjected to semipreparative HPLC [ODS-EP, 20 × 250 mm; solvent, methanol-water (15:85), UV-detector 205 nm, room temperature] to yield aesculin (**CAL13**) (4 mg,  $t_R$  32 min) and (-)-catechin (**CAL12**) (2 mg,  $t_R$  88 min).

The 75 % methanol fraction (1.074 g) was chromatographed on a silica gel column and fractionated using hexane: chloroform-methanol (95:5); 1:3 400 ml, hexane: chloroform-methanol (92:8); 1:3 400 ml, chloroform-methanol (88:22) 500 ml, chloroform-methanol (8:2) 100 ml and then chloroform-methanol (1:1) 40 ml. Fractions were collected and pooled by TLC analysis to afford 8 combined fractions.

From the combined fractions, fraction 75MF2 (0.046 g) was subjected to semipreparative HPLC [ODS-EP, 20 × 250 mm; solvent, methanol-water (65:35), UV-detector 205 nm, room temperature] to yield compound **CAL1** (2.8 mg,  $t_R$  42 min).

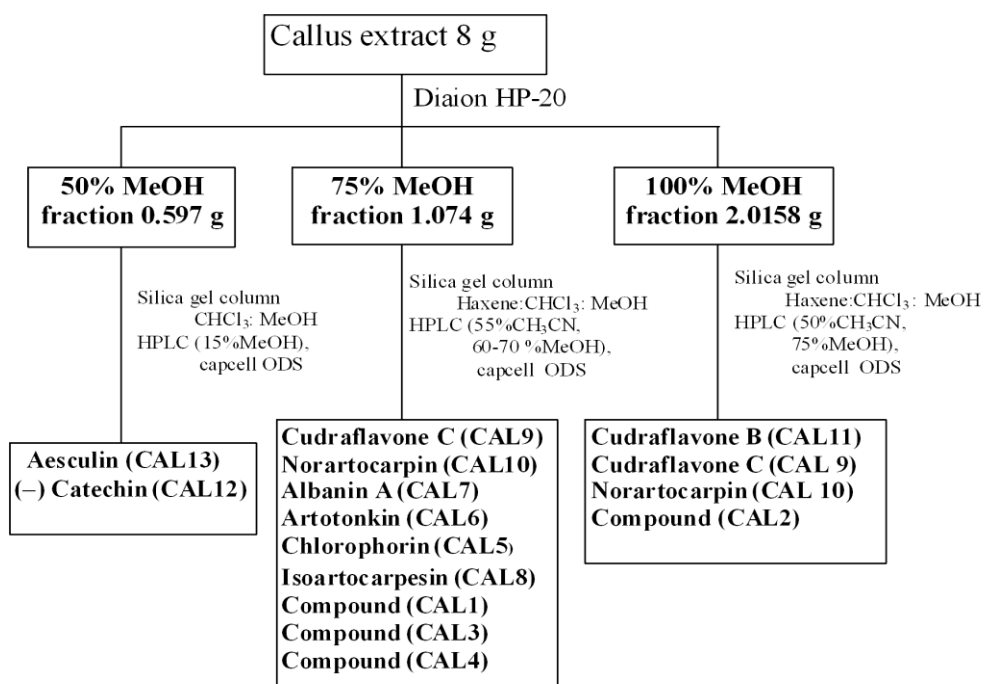
From the combined fractions, fraction 75MF3 (0.267 g) was subjected to semipreparative HPLC [ODS-EP, 20 × 250 mm; solvent, methanol-water (70:30), UV-detector 205 nm, room temperature] to yield compound **CAL3** (5 mg,  $t_R$  25 min), compound **CAL4** (5.8 mg,  $t_R$  28 min) cudraflavone C (**CAL9**) (49 mg,  $t_R$  77 min) and Norartocarpin (**CAL10**) (99 mg,  $t_R$  91 min)

From the combined fractions, fraction 75MF4 (0.102 g) was subjected to semipreparative HPLC [ODS-EP, 20 × 250 mm; solvent, methanol-water (70:30), UV-detector 205 nm, room temperature] to yield albanin A (**CAL7**) (10 mg,  $t_R$  17 min), artotonkin (**CAL6**) (28.7 mg,  $t_R$  23 min), chlorophorin (**CAL5**) (4 mg,  $t_R$  30 min) and isoartocarpesin (**CAL8**) (7 mg,  $t_R$  39 min).

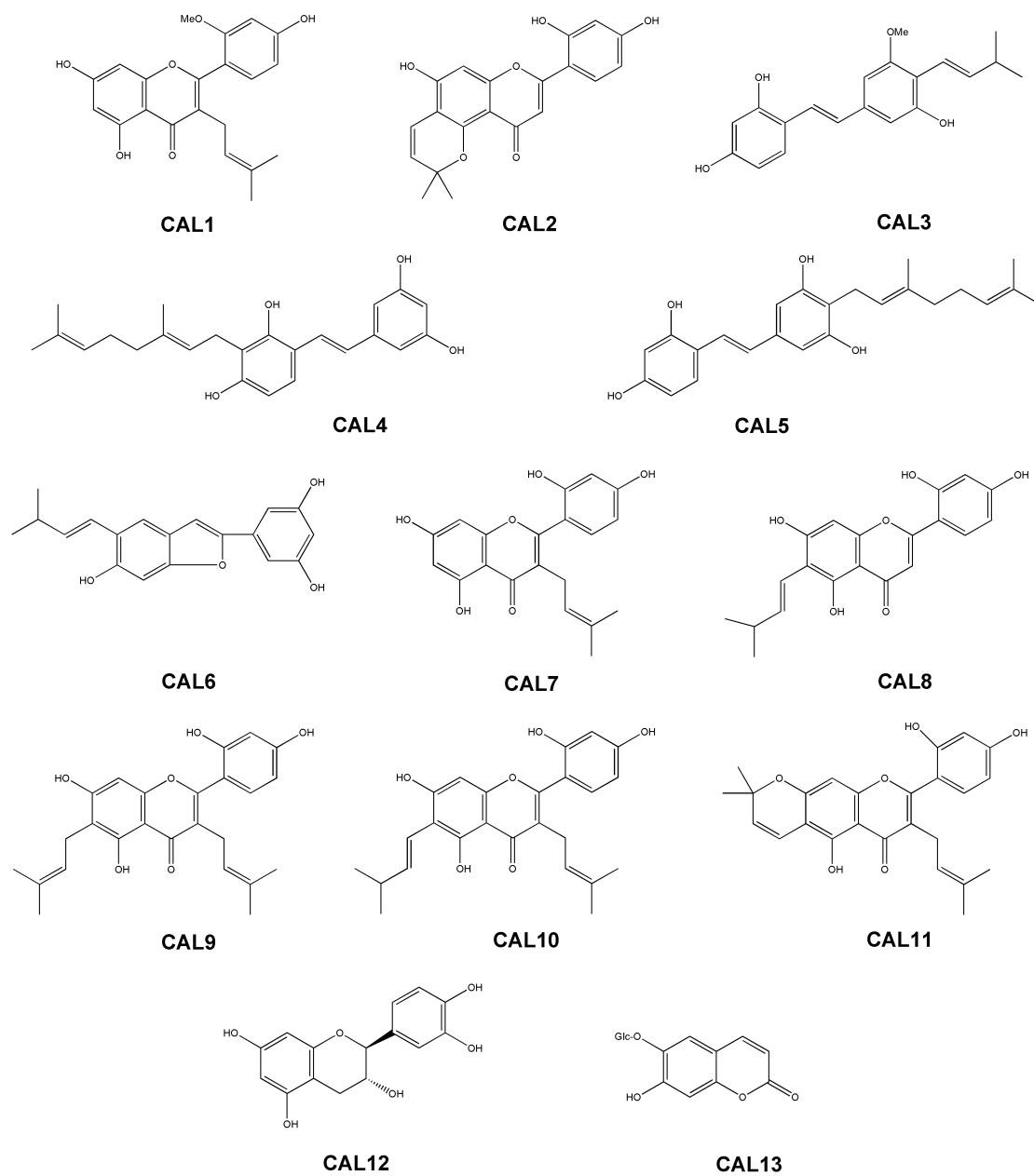
The 100 % methanol fraction (2.016 g) was chromatographed on a silica gel column and eluted with hexane-(chloroform-methanol) [1:3 (95:5)] 600 ml, chloroform-methanol (95:5) 550 ml, (92:8) 300 ml, chloroform-methanol (88:12) 200 ml and (1:1) 200 ml. Fractions were collected and pooled by TLC analysis to afford 9 combined fractions.

From the combined fractions, fraction 100MF3 (0.099 g) was subjected to semipreparative HPLC [ODS-EP, 20 × 250 mm; solvent, methanol-water (75:25), UV-detector 205 nm, room temperature] to yield cudraflavone B (**CAL11**) (52.3 mg,  $t_R$  60 min).

From the combined fractions, fraction 100MF5 (0.149 g) was subjected to semipreparative HPLC [ODS-EP, 20 × 250 mm; solvent, acetonitrile-water (50:50), UV-detector 205 nm, room temperature] to yield compound **CAL2** (1.7 mg,  $t_R$  60 min), cudraflavone C (**CAL9**) (8 mg,  $t_R$  93 min) and norartocarpin (**CAL10**) (19.8 mg,  $t_R$  111 min)



**Scheme 1** Separation of the callus cultures of *Artocarpus lakoocha*



**Figure 10** Compounds from callus cultures of *A. lakoocha*

## 6. Physical and Spectral Data of Isolated Compounds

### 6.1 Compound CAL1

Compound CAL1 was obtained as a yellow amorphous solid, soluble in methanol (2.8 mg,  $7 \times 10^{-3}$  % based on dried weight of callus cultures).

**HRESIMS** :  $[M+H]^+$  at  $m/z$  369.1340 (calcd for  $C_{21}H_{21}O_6$  369.1333), Figure 29

**UV** :  $\lambda_{max}$  nm (log  $\epsilon$ ), in methanol  
209 (4.50), 242(4.28), 258 (4.32), 303 (3.92), Figure 30

**$^1H$  NMR** :  $\delta$  ppm, 400 MHz, in methanol- $d_4$ ; see Table 12, Figure 31

**$^{13}C$  NMR** :  $\delta$  ppm, 100 MHz, in methanol- $d_4$ ; see Table 12, Figure 32

### 6.2 Compound CAL2

Compound CAL2 was obtained as a yellow amorphous solid, soluble in acetone (1.7 mg,  $3.75 \times 10^{-3}$  % based on dried weight of callus cultures).

**HRESIMS** :  $[M+H]^+$  at  $m/z$  353.1015 (calcd for  $C_{20}H_{16}O_6$ , 353.1020), Figure 36

**UV** :  $\lambda_{max}$  nm (log  $\epsilon$ ), in methanol  
221 (4.39), 287 (4.29), 346 (4.29), Figure 37

**$^1H$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 13, Figure 38

**$^{13}C$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 13, Figure 39

### 6.3 Compound CAL3

Compound CAL3 was obtained as a white powder, soluble in acetone (5 mg,  $12.5 \times 10^{-3}$  % based on dried weight of callus cultures).

**HRESIMS** :  $[M+H]^+$  at  $m/z$  327.1621 (calcd for  $C_{20}H_{22}O_4$  327.1597), Figure 42

**UV** :  $\lambda_{max}$  nm (log  $\epsilon$ ), in methanol  
218 (5.38), 346 (5.39), Figure 43

**$^1H$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 14, Figure 44

**$^{13}C$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 14, Figure 45

### 6.4 Compound CAL4

Compound CAL4 was obtained as a white amorphous solid, soluble in acetone (5 mg,  $12.5 \times 10^{-3}$  % based on dried weight of callus cultures).

**HRESIMS** :  $[M+H]^+$  at  $m/z$  381.2076 (calcd for  $C_{24}H_{28}O_4$  381.2067) Figure 49

**UV** :  $\lambda_{max}$  nm (log  $\epsilon$ ), in methanol  
211 (4.58), 330 (4.27), Figure 50

**$^1H$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 15, Figure 51

**$^{13}C$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 15, Figure 52

### 6.5 Compound CAL5

Compound CAL5 was obtained as a white amorphous solid, soluble in acetone (4 mg,  $10 \times 10^{-3}$  % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  at  $m/z$  381.21, Figure 55
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
220 (3.37), 261 (2.73), 314 (3.95), Figure 56
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 16, Figure 57
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 16, Figure 58

### 6.6 Compound CAL6

Compound CAL6 was obtained as a white amorphous solid, soluble in acetone (28.7 mg,  $71.8 \times 10^{-3}$  % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  at  $m/z$  481.21, Figure 59
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
228 (4.57), 302 (4.33), 339 (4.49), Figure 60
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 17, Figure 61
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 17, Figure 62

### 6.7 Compound CAL7

Compound CAL7 was obtained as a yellow amorphous solid, soluble in acetone (10 mg,  $20.5 \times 10^{-3}$  % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  at  $m/z$  355.12, Figure 63
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
210 (4.50), 256 (4.55), Figure 64
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 18, Figure 65
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 18, Figure 66

### 6.8 Compound CAL8

Compound CAL8 was obtained as a yellow amorphous solid, soluble in methanol (7 mg,  $17.5 \times 10^{-3}$  % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  at  $m/z$  355.12, Figure 67
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
211 (4.75), 287 (4.66), 347 (4.74), Figure 68
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in methanol- $d_4$ ; see Table 19, Figure 69
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in methanol- $d_4$ ; see Table 19, Figure 70

### 6.9 Compound CAL9

Compound CAL9 was obtained as a yellow amorphous solid, soluble in acetone (118 mg, 0.295 % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  at  $m/z$  423.18, Figure 71
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
212 (4.50), 262 (4.35), 312 (4.02), Figure 72
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 20, Figure 73
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 20, Figure 74

### 6.10 Compound CAL10

Compound CAL10 was obtained as a yellow amorphous solid, soluble in acetone (57 mg, 0.143 % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  at  $m/z$  423.18, Figure 75
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
210 (4.07), 279 (4.54), 325 (4.14), Figure 76
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 21, Figure 77
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 21, Figure 78

### 6.11 Compound CAL11

Compound CAL11 was obtained as a yellow amorphous solid, soluble in acetone (52.3 mg, 0.131 % based on dried weight of callus cultures).

- ESIMS** :  $[M+Na]^+$  at  $m/z$  423.18, Figure 79
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
210 (4.23), 278 (4.52), 330 (4.05), Figure 80
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in acetone- $d_6$ ; see Table 22, Figure 81
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in acetone- $d_6$ ; see Table 22, Figure 82

### 6.12 Compound CAL12

Compound CAL12 was obtained as a yellow amorphous solid, soluble in methanol (2 mg,  $5 \times 10^{-3}$  % based on dried weight of callus cultures).

- ESIMS** :  $[M+H]^+$  291.09 at  $m/z$ , Figure 83
- $[\alpha]_D^{28}$**  :  $-48.0^\circ$  ( $c$  0.1; MeOH)
- UV** :  $\lambda_{\max}$  nm (log  $\epsilon$ ), in methanol  
211 (3.32), 280 (2.50), Figure 84
- $^1\text{H}$  NMR** :  $\delta$  ppm, 400 MHz, in methanol- $d_4$ ; see Table 23, Figure 85
- $^{13}\text{C}$  NMR** :  $\delta$  ppm, 100 MHz, in methanol- $d_4$ ; see Table 23, Figure 86

### 6.13 Compound CAL13

Compound CAL13 was obtained as a yellow amorphous solid, soluble in methanol (4 mg,  $10 \times 10^{-3}$  % based on dried weight of callus cultures).

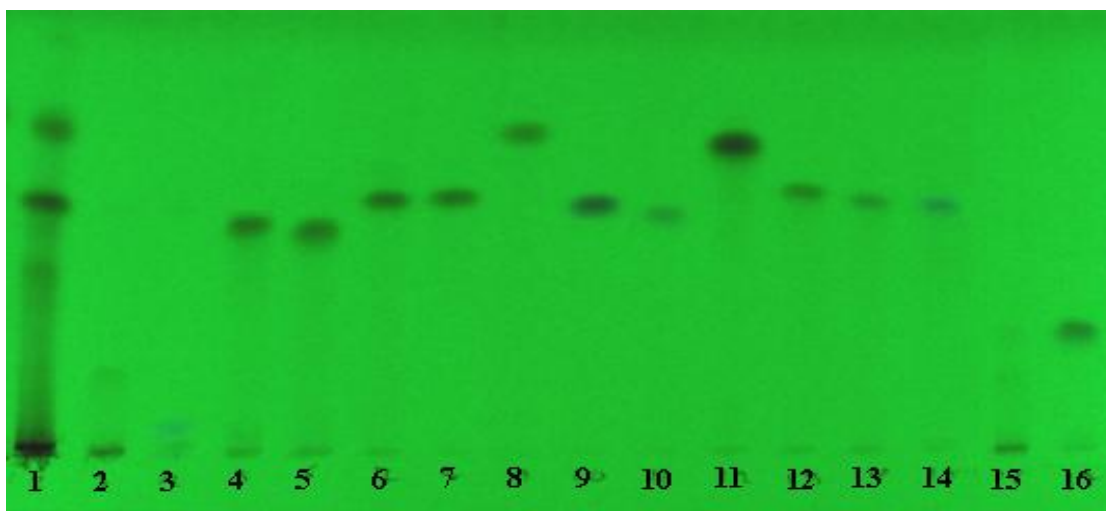
<b>ESIMS</b>	: $[M+H]^+$ at $m/z$ 341.08, Figure 87
<b>UV</b>	: $\lambda_{\max}$ nm (log $\epsilon$ ), in methanol 207 (4.39), 221 (4.35), 331 (3.98), Figure 88
<b><math>^1\text{H}</math> NMR</b>	: $\delta$ ppm, 400 MHz, in methanol- $d_4$ ; see Table 24 , Figure 89
<b><math>^{13}\text{C}</math> NMR</b>	: $\delta$ ppm, 100 MHz, in methanol- $d_4$ ; see Table 24, Figure 90

## 7. Analysis of Secondary Metabolites in the Callus Cultures of *A. lakoocha*.

### 7.1 Preparation of crude extracts of callus cultures

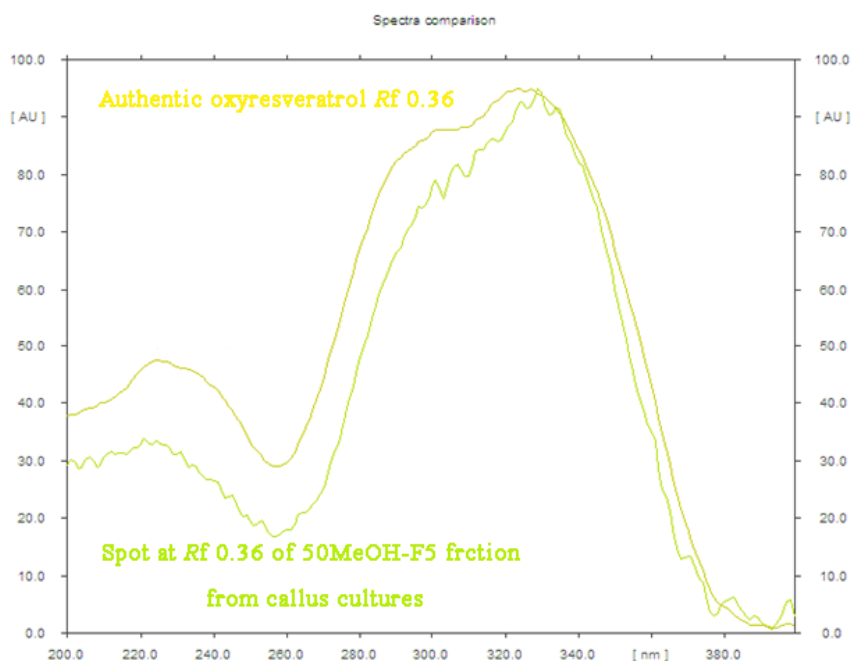
The dried calli of *A. lakoocha* (100 mg) were extracted with 10 ml ethanol by maceration for 24 h, followed by sonicated at 50° C for 20 min and centrifugation at 14,00 rpm for 10 min. The supernatant part was concentrated by evaporator to give a dried extract. This extract was dissolved in 0.5 ml methanol to give a sample solution.

In our cell cultures of *A. lakoocha*, thirteen compounds were isolated and characterized. Only a trace amount of oxyresveratrol was detected on TLC densitometric plate (Fig 11) in the same condition used for analysis of oxyresveratrol in the heartwood of *A. lakoocha* (methylene chloride/ methanol (85: 15)). Comparison of the UV–vis spectra of the callus extract with reference that of standard confirmed the finding. (Fig 12), However, the compound could not be fully identified.



**Figure 11** TLC pattern of isolated 13 compounds from *A. lakoocha* callus cultures

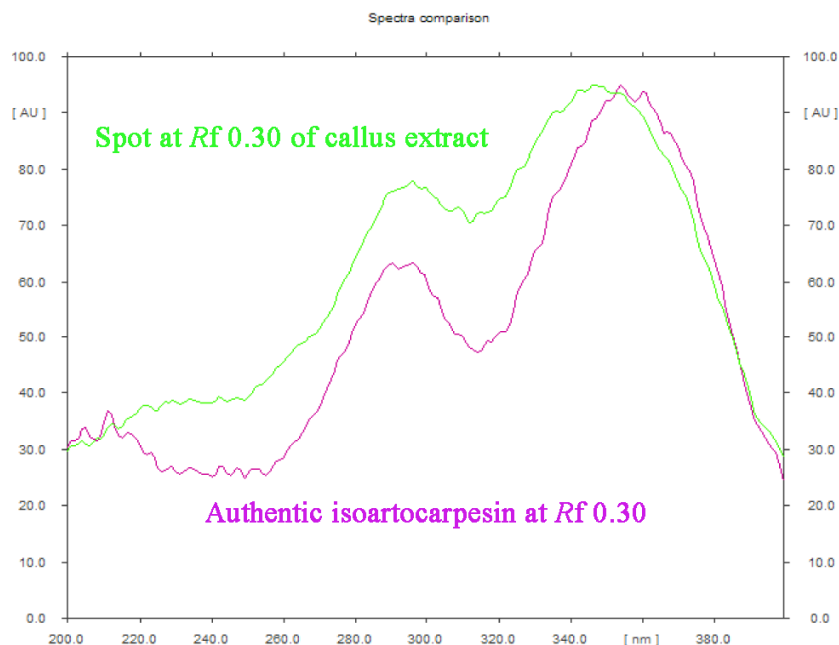
- |                            |                               |
|----------------------------|-------------------------------|
| 1 = callus extract         | 9 = CAL6 (artotonkin)         |
| 2 = CAL12 (-catechin)      | 10 = CAL5 (chlorophorin)      |
| 3 = CAL13 (aesculin)       | 11 = CAL 1                    |
| 4 = CAL7 (albanin A)       | 12 = CAL 2                    |
| 5 = CAL8 (isoartocarpesin) | 13 = CAL 3                    |
| 6 = CAL10 (norartocarpin)  | 14 = CAL 4                    |
| 7 = CAL9 (cudraflavone C)  | 15 = 50 MeOH-F5 fraction      |
| 8 = CAL11 (cudraflavone B) | 16 = authentic oxyresveratrol |



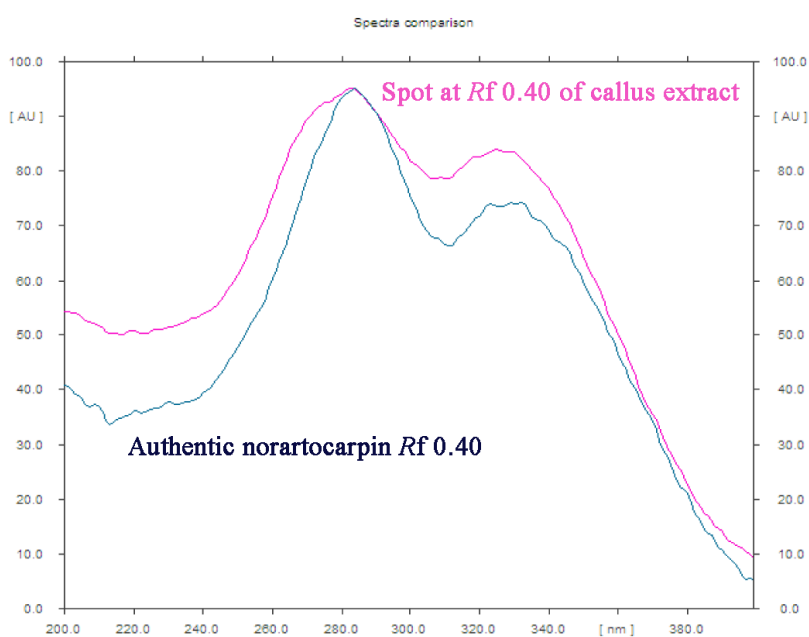
**Figure 12** UV absorption spectrum of authentic oxyresveratrol and 50MeOH-F5 fraction of similar  $R_f$  value to oxyresveratrol which was separated from callus extract



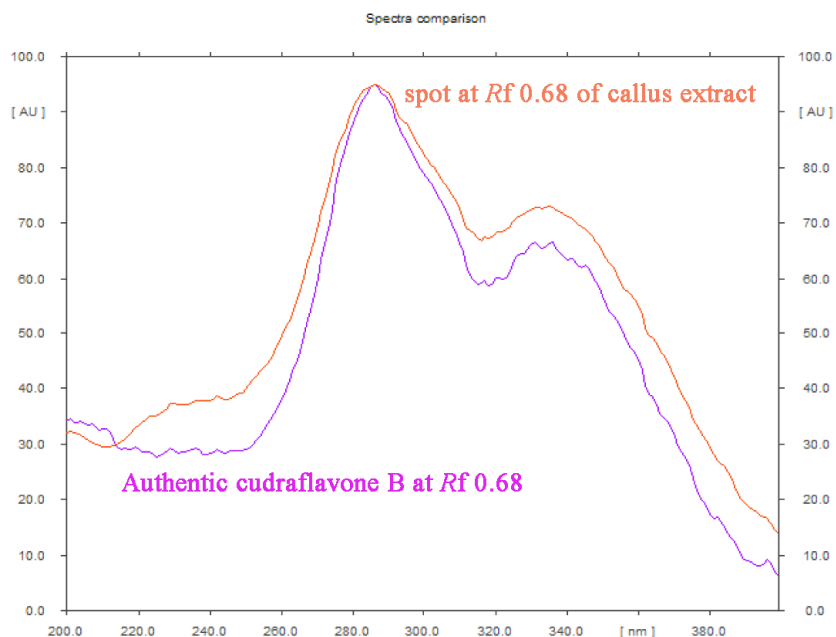
The major metabolites were found to be isoartocarpesin, norartocarpin and cudraflavone B, which were separated from other compounds in callus extract. These compounds were subjected to comparison of their UV–Vis spectra with reference standards. (Figures 13-15).



**Figure 13** UV absorption spectrum of authentic isoartocarpesin and compound of similar *Rf* value to isoartocarpesin which was separated from callus extract



**Figure 14** UV absorption spectrum of authentic norartocarpin and compound of similar *Rf* value to norartocarpin which was separated from callus extract



**Figure 15** UV absorption spectrum of authentic cudraflavone B and compound of similar  $R_f$  value to cudraflavone B which was separated from callus extract

### 7.2 Preparation of Standard Solutions

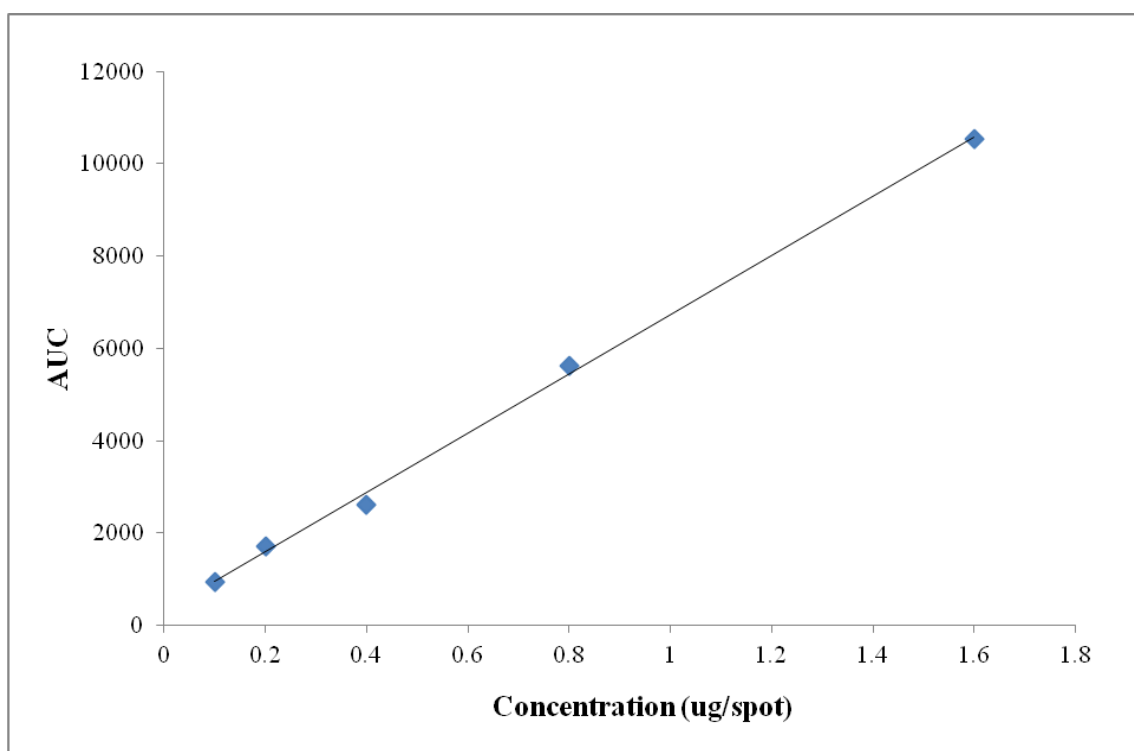
Isoartocarpin, norartocarpin and cudraflavone B were isolated from callus of *A. lakoocha*. For norartocarpin and cudraflavone B, the standard solution of each compound with an accurate concentration of 0.16 mg/ml was prepared by dissolving the compound (1.6 mg) in methanol (10 ml). From this solution, 4 additional standard solutions (0.08, 0.04, 0.02 and 0.01 mg/ml) were prepared by serial dilution. For isoartocarpin, a standard solution with an accurate concentration of 0.16 mg/ml was prepared by dissolving the sample (1.6 mg) in methanol (10 ml). From this solution, 7 additional standard solutions (0.08, 0.04, 0.02, 0.01, 0.005, 0.0025 and 0.00125 mg/ml) were obtained by serial dilution.

### 7.3 TLC-densitometric Conditions

Chromatography was performed on a TLC aluminum sheet (silica gel 60 F<sub>254</sub> plate, 20 × 10 cm) using methylene chloride/ methanol/toluene (75: 7: 18) with sulfuric acid 0.1 % as the mobile phase. Each sample (10 μl) was applied as a band (5 × 0.30 mm) in triplicate. Plates were developed in a glass tank preequilibrated with the mobile phase. The solvent was allowed to run up the plate to a height of 8 cm. Chromatograms were evaluated by measuring the peak area with the TLC scanner in the absorbance mode at 254 nm.

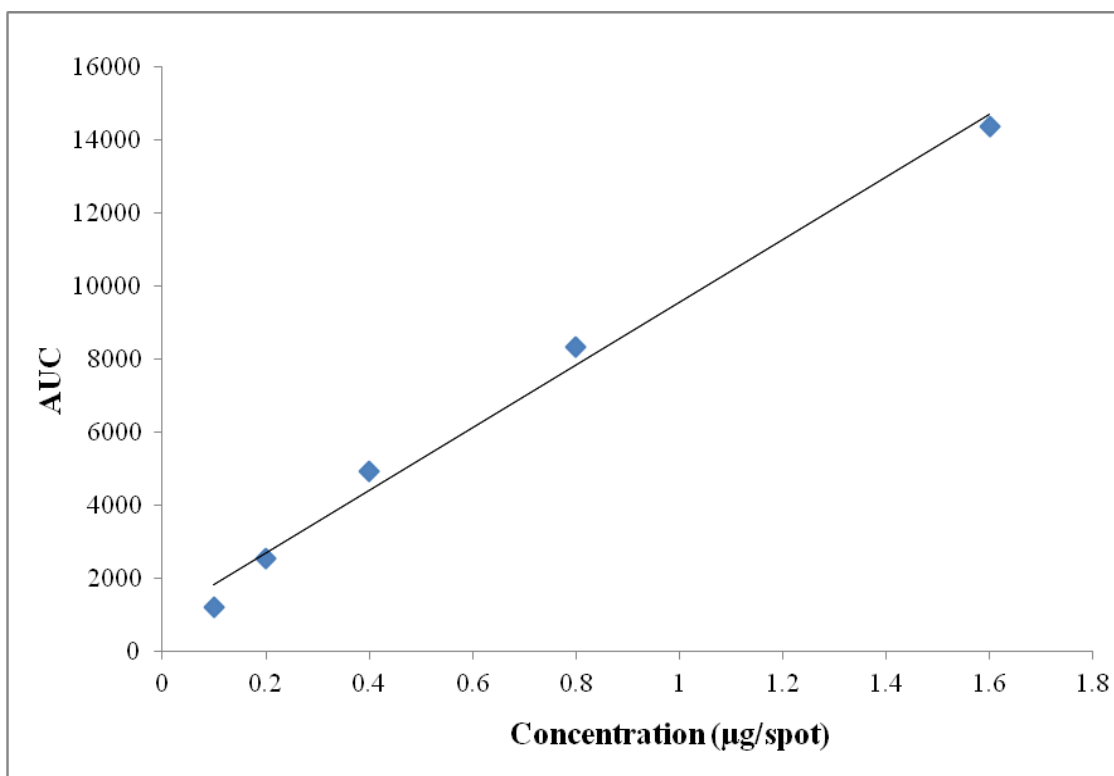
#### 7.4 Standard Calibration Curve

A standard calibration curve of each compound was produced for TLC densitometric analysis. The calibration curve of standard norartocarpin showed linearity of the relationship from 0.01-0.16 mg/ml, and the correlation coefficient was 0.9981. The calibration curve of standard cudraflavone B showed linearity of the relationship from 0.1-1.6 mg/ml, and the correlation coefficient was 0.9906. The calibration curve of isoartocarpesin showed linearity of the relationship from 0.00125-0.04 mg/ml, and the correlation coefficient was 0.9957.



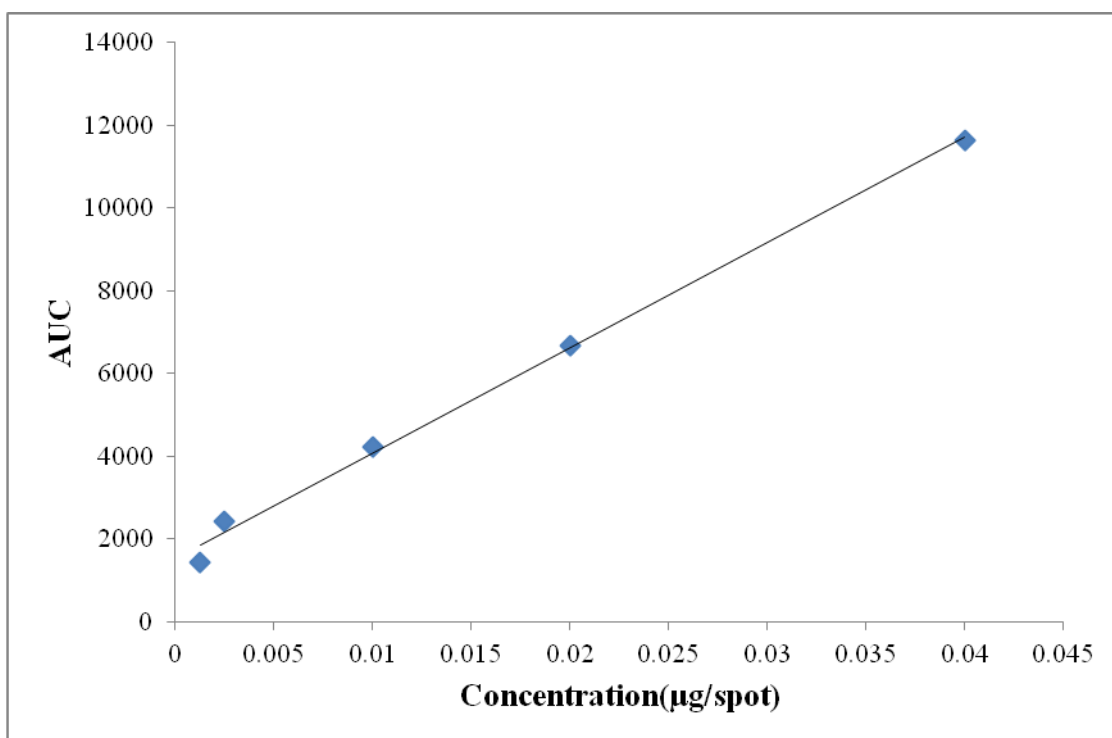
**Figure 16** The Calibration curve of Norartocarpin

$$Y = 6414.4x + 305.22 \quad (r^2=0.9981)$$



**Figure 17** The calibration curve of Cudraflavone B

$$Y = 8586.40x + 950.31 \quad (r^2=0.9906)$$



**Figure 18** The calibration curve of Isoartocarpesin

$$Y = 254617x + 1527.40 \quad (r^2=0.9958)$$

## 8. Determination of Free Radical Scavenging Activity

### 8.1 Free Radical Scavenging Activity Assay

The assay was performed according to an established protocol (Braca *et al.*, 2002).

### 8.2 Preparation of Test Sample

The test compound (1 mg) was dissolved in 1 ml of methanol (or suitable solvent) and diluted with methanol until a suitable range of concentration (mg/ml) was obtained. The concentration was expressed as  $\mu\text{M}$  in final concentration. For example, CAL7 (MW 354) at 1 mg/1ml was equal to 2825  $\mu\text{M}$  [(1 mg  $\times$  10<sup>3</sup>  $\times$  1000 ml) /354]. For each well, 20  $\mu\text{l}$  of test solution was added to the reaction mixture to furnish the total volume of 200  $\mu\text{l}$ . The final concentration was calculated by the formula below.

$$N_1 V_1 = N_2 V_2$$

$N_1$  = Beginning concentration ( $\mu\text{M}$ )

$V_1$  = Beginning volume ( $\mu\text{l}$ )

$N_2$  = Final concentration ( $\mu\text{M}$ )

$V_2$  = Final volume ( $\mu\text{l}$ )

Thus, the final concentration of CAL7 solution = 2825  $\mu\text{M}$   $\times$  20  $\mu\text{l}$  / 200  $\mu\text{l}$   
= 282.5  $\mu\text{M}$

### 8.3 Preparation of DPPH Solution (100 $\mu\text{M}$ )

DPPH (2 mg) was dissolved in 100 ml of methanol, and the solution was stirred for 30 min.

### 8.4 Measurement of Activity

The test sample (20  $\mu\text{l}$ ) was added to 180  $\mu\text{l}$  of DPPH solution (100  $\mu\text{M}$ ) in a 96-well plate. The solution mixture was incubated at 37°C for 30 min, and then the absorbance of each well was measured at 510 nm. The DPPH solution (180  $\mu\text{l}$ ) mixed with methanol (20  $\mu\text{l}$ ) was used as negative control and quercetin as a reference compound.

### 8.5 Calculation of Percent Inhibition of DPPH Free Radical Scavenging Activity

The percentage of DPPH reduction was calculated as follows.

$$\% \text{ DPPH reduction} = (A - B) \times 100 / A$$

A = The absorbance of DPPH solution after incubation at 510 nm

B = The absorbance of the reaction mixture after incubation at 510 nm

For IC<sub>50</sub> evaluation of pure compounds, a graph showing concentration versus % DPPH reduction was plotted. The IC<sub>50</sub> was calculated from the graph.

## 9. Determination of Tyrosinase Inhibitory Activity

In this study, tyrosinase inhibitory activity was determined by the dopachrome method using L-DOPA as the substrate. Dopachrome is one of the intermediate substances in the melanin biosynthesis. The red color of dopachrome can be detected by visible light. In this experiment a microplate reader (BIO-RAD, model 450) with 492 nm interference filter was used for detection. The potential tyrosinase inhibitor would show minimal dopachrome absorption. This method was modified from the methods of Masamoto (Masamoto, Iida, and Kubo, 1980; Iida *et al.*, 1995 and Morita *et al.*, 1994)

### 9.1 Preparation of the Reaction Mixture

#### 9.1.1 Preparation of 20 mM phosphate buffer (pH 6.8)

Solution A: NaH<sub>2</sub>PO<sub>4</sub>·2H<sub>2</sub>O (312 mg) was dissolved in 100 ml of H<sub>2</sub>O.

Solution B: Na<sub>2</sub>HPO<sub>4</sub> (284 mg) was dissolved in 100 ml of H<sub>2</sub>O.

Then, solutions A and B were mixed until pH 6.8 was reached.

#### 9.1.2 Preparation of 0.8 mM L-DOPA

L-DOPA (0.8 mg) was dissolved in 5 ml of 20 mM phosphate buffer (pH 6.8).

#### 9.1.3 Preparation of tyrosinase solution

Tyrosinase enzyme (0.5 mg) was dissolved in 5 ml of 20 mM phosphate buffer (pH 6.8).

#### 9.1.4 Preparation of the test sample

One milligram of the test compound was dissolved in 3 ml of ethanol (or suitable solvent) and diluted with ethanol until a suitable range of concentrations (mg/ml) was obtained. The concentration was expressed as μM in the final calculation. For example, the concentration of compound CAL7 (MW 354) at 1 mg/3ml was equal to 942 μM [(1 mg × 10<sup>3</sup> × 1000 ml) / (3 × 354)]. For each well, 20 μl of test solution was added to the reaction mixture to furnish the total volume of 200 μl. The final concentration was calculated by the formula below.

$$N_1V_1 = N_2V_2$$

N<sub>1</sub> = Beginning concentration (μM)

V<sub>1</sub> = Beginning volume (μl)

N<sub>2</sub> = Final concentration (μM)

V<sub>2</sub> = Final volume (μl)

Thus, the final concentration of CAL7 solution = 942 μM × 20 μl / 200 μl = 94.2 μM

## 9.2 Measurement of Activity

The reaction mixture (200  $\mu$ l) was measured in four wells (A, B, C and D). In each well, the substance was added in the order of mixing, as follows;

A (control)	20 $\mu$ l of mushroom tyrosinase solution (48 unit/ml)
	140 $\mu$ l of 20 mM phosphate buffer (pH 6.8)
	20 $\mu$ l of ethanol
B (blank of A)	160 $\mu$ l of 20 mM phosphate buffer (pH 6.8)
	20 $\mu$ l of ethanol
C (test sample)	20 $\mu$ l of mushroom tyrosinase solution (48 unit/ml)
	140 $\mu$ l of 20 mM phosphate buffer (pH 6.8)
	20 $\mu$ l of test sample in ethanol
D (blank of C)	160 $\mu$ l of 20 mM phosphate buffer (pH 6.8)
	20 $\mu$ l of test sample in ethanol

After each well was mixed and preincubated at 25 °C for 10 min, 20  $\mu$ l of 0.85  $\mu$ M L-DOPA was added, and the mixture was incubated at 25 °C for 20 min. The absorbance of each well was measured at 492 nm with the microplate reader both before and after incubation.

## 9.3 Calculation of the Percent Inhibition of Tyrosinase Enzyme

The percent inhibition of tyrosinase reaction was calculated as follows.

$$\% \text{ Tyrosinase inhibition} = 100 \times [(A-B)-(C-D)] / (A-B)$$

A = The difference of optical density before and after incubation at 492 nm without test sample

B = The difference of optical density before and after incubation at 492 nm without test sample and enzyme

C = The difference of optical density before and after incubation at 492 nm with test sample

D = The difference of optical density before and after incubation at 492 nm with test sample, but without enzyme

## 9.4 Calculation of IC<sub>50</sub>

After the % tyrosinase inhibition of the test solution in each concentration was obtained, a graph showing concentration against % tyrosinase inhibition was plotted. The IC<sub>50</sub> (concentration at 50% tyrosinase inhibition) of each pure compound was then obtained from the graph.

## CHAPTER IV

### RESULTS AND DISCUSSION

#### 1. Quantitative Analysis of Oxyresveratrol Content in *A. lakoocha* Heartwood and Puag-haad

Three samples of *A. lakoocha* heartwood (ALH-1 to ALH-3) were collected from natural habitats, while 11 samples (ALH-4 to ALH-14) were purchased from drugstores as indicated in Table 11. Two samples of the traditional drug Puag-Haad (PH-1 and PH-2) were obtained from drugstores (Table 11).

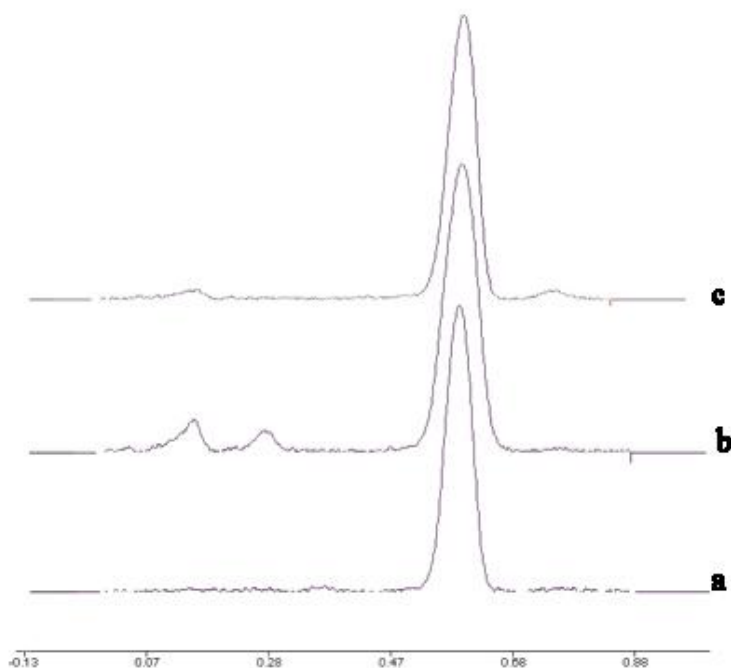
**Table 11** Oxyresveratrol content in *A. lakoocha* heartwood and Puag-haad

Sample No.	Locality (site of collection)	Oxyresveratrol (mg/g) $\pm$ SD
Natural samples of <i>A. lakoocha</i> heartwood		
ALH-1	Chiang Rai (Northern province)	182.3 $\pm$ 0.18
ALH-2	Kanchanaburi (Western)	166.0 $\pm$ 0.80
ALH-3	Bangkok (Central)	49.0 $\pm$ 0.24
Commercial samples of <i>A. lakoocha</i> heartwood		
ALH-4	Nakhon Sawan (Central)	29.6 $\pm$ 0.11
ALH-5	Kamphaeng phet (Central)	31.6 $\pm$ 0.14
ALH-6	Nakhon Prathom (Central)	28.6 $\pm$ 0.12
ALH-7	Bangkok (Central)	45.2 $\pm$ 0.20
ALH-8	Bangkok (Central)	50.4 $\pm$ 0.23
ALH-9	Trang (Southern)	26.2 $\pm$ 0.11
ALH-10	Songkhla (Southern)	33.4 $\pm$ 0.17
ALH-11	Nakhon Sri Thammarat (Southern)	51.0 $\pm$ 0.18
ALH-12	Chon Buri (Eastern)	24.6 $\pm$ 0.12
ALH-13	Nakhon Ratchasima (Northeastern)	23.4 $\pm$ 0.11
ALH-14	Ubon Ratchathani (Northeastern)	69.6 $\pm$ 0.33
Samples of traditional drug 'Puag-Haad'		
PH-1	Chiang Mai (Northern)	837.5 $\pm$ 4.04
PH-2	Chiang Rai (Northern)	780.1 $\pm$ 3.52

SD = standard deviation

The amounts of oxyresveratrol in 3 samples of *A. lakoocha* heartwood collected from natural habitats were 49.0–182.3 mg/g, whereas those in 11 commercial samples were in the range of 23.4–69.6 mg/g. The oxyresveratrol contents in 2 samples of traditional drug Puag-Haad were 780.1 and 837.5 mg/g





**Figure 19** Representative TLC densitometric chromatograms.

a Standard sample of oxyresveratrol (0.80  $\mu\text{g}/\text{spot}$ ).

b Sample of *A. lakoocha* heartwood extract (1.60  $\mu\text{g}/\text{spot}$ ).

c Sample of Puag-Haad (2.00  $\mu\text{g}/\text{spot}$ ).

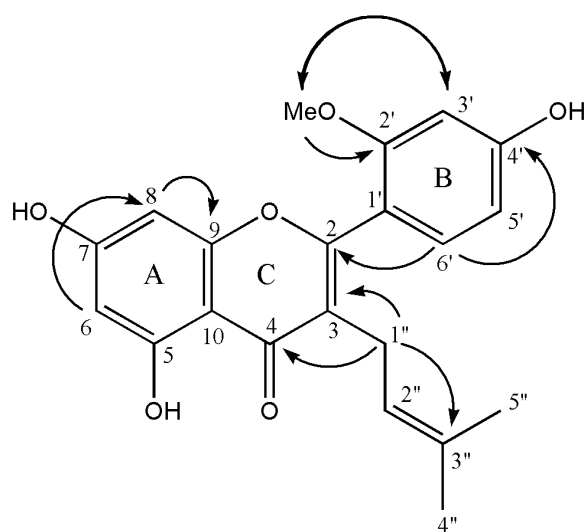
## 2. Structure Determination of Isolated Compounds

The structure determination of each isolates compound was performed by interpretation of their UV, MS and NMR data, and then confirmed by comparison with previously reported values.

### 2.1 Structure Determination of Compound CAL1

Compound CAL1, a yellow amorphous powder, was analyzed for  $\text{C}_{21}\text{H}_{20}\text{O}_6$  from its  $[\text{M}+\text{H}]^+$  at  $m/z$  369.1340 (calcd for  $\text{C}_{21}\text{H}_{21}\text{O}_6$ , 369.1333) in the HR-ESI-MS (Figure 29). The UV spectrum exhibited maxima at 209, 242, 258 and 303 nm, suggestive of a flavone skeleton (Figure 30). The  $^1\text{H}$  NMR spectrum (Figure 31) showed the presence of an isoprenyl group with two methyl groups at  $\delta$  1.58 (3H, s, H-4''),  $\delta$  1.36 (3H, s, H-5''), an olefinic proton at  $\delta$  5.04 (1H, m, H-2'') and a pair of methylene protons at  $\delta$  3.00 (2H, d,  $J = 7$  Hz, H-1''). It also showed a methoxy signal at  $\delta$  3.77 ( $\text{CH}_3\text{O}-\text{C}-2'$ ). In addition, it exhibited an ABX-spin system of aromatic protons (B-ring) at  $\delta$  6.54 (1H, d,  $J = 2$  Hz, H-3'),  $\delta$  6.48 (1H, dd,  $J = 8, 2$  Hz, H-5') and  $\delta$  7.12 (1H, d,  $J = 8$  Hz, H-6'), and an AX-spin system of *meta*-coupled aromatic protons (A-ring) at  $\delta$  6.18 (1H, d,  $J = 2$  Hz, H-8) and  $\delta$  6.23 (1H, d,  $J = 2$  Hz, H-6).

The  $^{13}\text{C}$  NMR spectrum contained a signal for a ketone carbonyl carbon at  $\delta$  183.6 (Figure 32). Interpretation of the HMBC and HMQC spectra (Figures 33 and 34) revealed the substitution pattern. The HMBC correlations from H-6' ( $\delta$  7.12) to C-2 ( $\delta$  163.3), C-4' ( $\delta$  162.3) and C-2' ( $\delta$  159.8), from H-1'' ( $\delta$  3.0) to C-2 (163.3), C-3 ( $\delta$  122.0), C-4 ( $\delta$  183.6) and C-3'' ( $\delta$  132.6), supported the location of the prenyl group at C-3. The HMBC experiment also showed the connectivity between the methoxy protons at  $\delta$  3.77 (MeO-2') and C-2, and an NOE experiment further assigned the methoxy group at C- 2', since irradiation of these protons ( $\delta$  3.78) enhanced H-3' (Figure 35).



[342]

**Figure 20** Important NOE (H↔H) and HMBC (H→C) correlations of CAL1

Based on the above spectral evidence, the structure of CAL1 was characterized as a new compound, namely 5,7,4'-trihydroxy-2'-methoxy-3-( $\gamma,\gamma$ -dimethylallyl)-flavone [342].

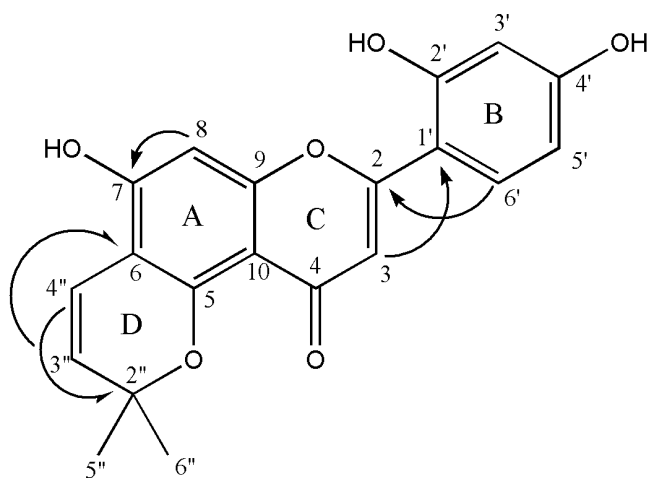
**Table 12** NMR Spectral data of compound CAL1 (methanol- $d_4$ )

position	compound CAL1		HMBC
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	(correlation with $^1\text{H}$ )
2		163.3	6', 1''
3		122.0	1''
4		183.6	1''
5		160.0	-
6	6.18, d (2.0)	99.6	8
7		165.6	-
8	6.23, d (2.0)	94.5	6
9		159.8	8
10		105.4	6, 8
1'		114.7	3', 5'
2'		159.8	6', OMe
3'	6.54, d (2.4)	100.2	5'
4'		162.3	3', 6'
5'	6.48, dd (8.4, 2.4)	108.3	3'
6'	7.11, d (8.4)	132.3	-
1''	3.00, d (7), 2H	24.8	-
2''	5.04, m	122.8	4'', 5''
3''		132.6	1'', 4'', 5''
4''	1.58, s	25.8	5''
5''	1.36, s	17.6	4''
OMe	3.77	56.1	-

## 2.2 Structure Determination of Compound CAL2

Compound CAL2, a pale yellow amorphous powder, was analyzed for  $\text{C}_{20}\text{H}_{16}\text{O}_6$  from its  $[\text{M}+\text{H}]^+$  ion at  $m/z$  353.1015 (calcd for 353.1020) in the HR-ESI-MS (Figure 36). The UV spectrum exhibited absorptions at 221, 287 and 346 nm (Figure 37). The  $^1\text{H}$  NMR spectrum (Figure 38) showed the presence of a 2,2-dimethyl-3,4-dehydropyran ring at  $\delta$  1.46 (6H, s, H-5'', H-6''),  $\delta$  5.73 (1H, d,  $J = 10$  Hz, H-3'') and  $\delta$  6.66 (1H, d,  $J = 10$  Hz, H-4''). It also exhibited an ABX-spin system of aromatic protons at  $\delta$  6.60 (1H, d,  $J = 2$  Hz, H-3'),  $\delta$  6.56 (1H, dd,  $J = 9, 2$  Hz, H-5') and  $\delta$  7.83 (1H, d,  $J = 9$  Hz, H-6'), and an aromatic proton at  $\delta$  6.43 (1H, br s, H-8). The  $^{13}\text{C}$  NMR spectrum contained a signal for a ketone carbonyl carbon at  $\delta$  183.6 (Figure 39). Examination of the HMBC and HMQC spectra (Figures 40 and 41) revealed the substitution pattern. The HMBC correlations from H-4'' ( $\delta$  5.73) to C-7 ( $\delta$  159.3) and H-3'' ( $\delta$  6.66) to C-6 ( $\delta$  105.8) supported the location of the 2,2-dimethyl-

3,4-dehydropyran ring at C-5 and C-6. The HMBC experiment showed the connectivities from the proton at C-6' ( $\delta$  7.83) to C-2 ( $\delta$  108.6), and from H-3 ( $\delta$  7.07) to C-1' ( $\delta$  110.8).



[343]

**Figure 21** Important HMBC (H→C) correlations of CAL2

Based on the above spectral evidence, the structure of compound CAL2 was elucidated as 7,2',4'-trihydroxy-3'',4''-dehydropyrano [1'',4'':5,6] flavone.

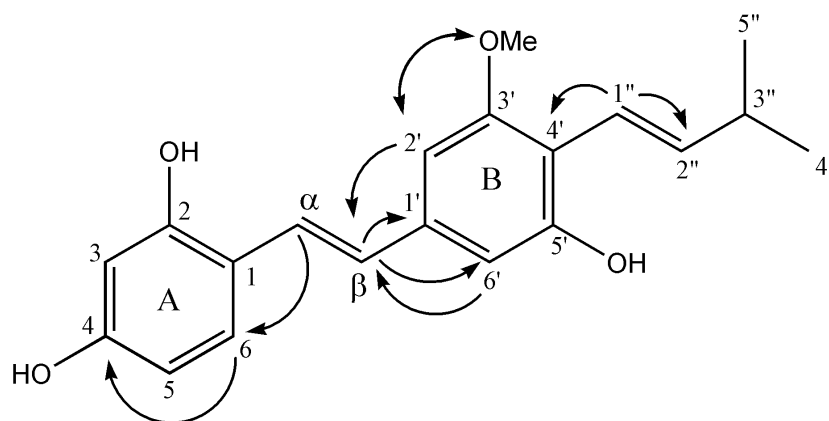
**Table 13** NMR Spectral data of compound CAL2 (acetone- $d_6$ )

position	compound CAL2		HMBC
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	(correlation with $^1\text{H}$ )
2	7.07, s	163.0	6'
3		108.6	1'
4		183.6	-
5		158.0	-
6		105.8	8, 3''
7		159.3	8
8	6.43, s	95.5	7
9		157.4	-
10		105.8	-
1'		110.8	3
2'		159.3	-
3'		104.4	-
4'	6.60, br s	162.6	-
5'		109.2	-
6'		131.0	2
1''	-	-	-
2''	6.66, d (10)	78.6	-
3''		115.9	3''
4''		129.2	3'', 4''
5''		28.4	-
6''		28.4	2'', 3''
		1.46, s	
	1.46, s		

### 2.3 Structure Determination of Compound CAL3

Compound CAL3, a yellow amorphous solid, was analyzed for  $\text{C}_{20}\text{H}_{22}\text{O}_4$  from its  $[\text{M}+\text{H}]^+$  ion at  $m/z$  327.1621 (calcd for 327.1597) in the HR-ESI-MS (Figure 42). The UV spectrum exhibited absorptions at 218 and 346 nm (Figure 43). The  $^1\text{H}$  NMR spectrum (Figure 44) showed the presence of a *trans*-olefin at  $\delta$  7.36 and 6.93 (1H each, d,  $J = 16$  Hz, H- $\alpha$  and H- $\beta$ ). The  $^1\text{H}$  NMR spectrum showed signals for an ABX-spin system of aromatic protons (A-ring) at  $\delta$  6.45 (1H, d,  $J = 2$  Hz, H-3),  $\delta$  6.38 (1H, dd,  $J = 9, 2$  Hz, H-5) and  $\delta$  7.41 (1H, d,  $J = 9$  Hz, H-6) typical of substitution at the 2- and 4-positions on the A ring. An AX-spin system of *meta*-coupled aromatic protons (B-ring) at  $\delta$  6.71 (1H, d,  $J = 2$  Hz, H-2') and  $\delta$  6.67 (1H, d,  $J = 2$  Hz, H-6') suggested that the ring B was tetrasubstituted. The  $^1\text{H}$  NMR spectrum showed the presence of a prenyl group at  $\delta$  1.07 (6H, d,  $J = 6.8$  Hz, H-4'', 5''),  $\delta$  6.64 (1H, d,  $J = 8$  Hz, H-1''),  $\delta$  6.63 (1H, d,  $J = 8$  Hz, H-2'') and  $\delta$  2.40 (1H, m, H-3''). In addition, the  $^1\text{H}$  NMR

spectrum showed a signal for a methoxyl group at  $\delta$  3.86 (3H, s). Interpretation of the  $^{13}\text{C}$  NMR, HMBC and HMQC spectra (Figures 45, 46 and 47) revealed the substitution for this compound. The HMBC correlations from H-6' ( $\delta$  6.67) to C- $\beta$  ( $\delta$  126.3), and H-2' ( $\delta$  6.71) to C- $\beta$  ( $\delta$  126.3), and H- $\beta$  (6.93) to C-1' ( $\delta$  138.7), C-6' ( $\delta$  101.8), and H- $\alpha$  ( $\delta$  7.36) to C-1 ( $\delta$  117.4), C-6 ( $\delta$  128.3), and H-1'' ( $\delta$  6.64) to C-4' ( $\delta$  113.4), and C-2'' ( $\delta$  141.4), supported the presence of the prenyl group at C-4'. The HMBC experiment showed the connectivity between the methoxy protons at  $\delta$  3.86 (MeO-3') and C-3'. The result from an NOE experiment assigned the methoxy group at C-3' since irradiation the proton at  $\delta$  3.86 enhanced the H-2' signal (Figure 48). These results provided support for the presence of a prenyl substituent at C-4' and the methoxyl group at C-3'.



[344]

**Figure 22** NOE experiment (H $\leftrightarrow$ H) and important HMBC (H $\rightarrow$ C) correlations of CAL3

Compound CAL3 was characterized as a new compound, namely *trans*-2,4,5'-trihydroxy-3'-methoxy-4'-(3-methyl-*E*-but-1-enyl)-stilbene.

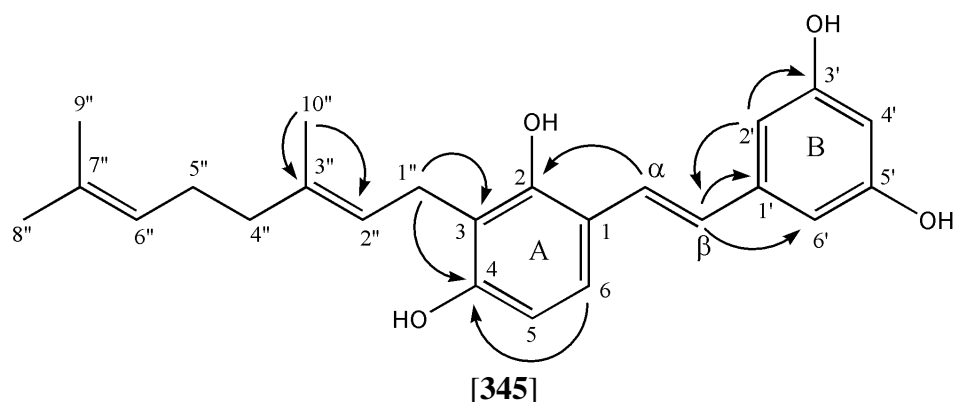
**Table 14** NMR Spectral data of compound CAL3 (acetone-*d*<sub>6</sub>)

position	compound CAL3		HMBC
	<sup>1</sup> H (mult., <i>J</i> in Hz)	<sup>13</sup> C	(correlation with <sup>1</sup> H)
1		117.4	3, 5, β
2		156.9	6
3	6.45, d (2)	103.7	5
4		159.1	3, 6
5	6.38, dd (9,2)	108.5	3
6	7.41, d (9)	128.3	α
α	7.36, d (16)	124.3	6
β	6.93, d (16)	126.3	6'
1'		138.7	α, β
2'	6.71, d (2)	107.1	β, 6'
3'		156.8	OMe
4'		113.4	1''
5'		159.6	6'
6'	6.67, d (2)	101.8	β
1''	6.64, br s	119.0	2''
2''	6.64, br s	141.4	1''
3''	2.40, m	34.0	1'', 4'', 5''
4''	1.07, d (6.8)	23.2	5'', 3'', 2''
5''	1.07, d (6.8)	23.2	4'', 3''
OMe	3.86, s	56.0	-

#### 2.4 Structure Determination of Compound CAL4

Compound CAL4 was obtained as a yellow amorphous solid. It was analyzed for C<sub>24</sub>H<sub>28</sub>O<sub>4</sub> from its [M+H]<sup>+</sup> ion at *m/z* 381.2076 (calcd for 381.2067) in the HR-ESI-MS (Figure 49). The UV spectrum exhibited absorptions at 209, 242, 258 and 303 nm, suggestive of a flavone skeleton (Figure 50). Its <sup>1</sup>H NMR spectrum (Figure 51) also showed the presence of a *trans*-olefin at δ 6.79, 7.36 (1H each, d, *J* = 16 Hz, H-α and H-β). The <sup>1</sup>H NMR spectrum showed signals for an AX-spin system of *ortho*-coupled aromatic protons (A-ring) at δ 6.48 (1H, d, *J* = 8 Hz, H-5), δ 7.25 (1H, d, *J* = 8 Hz, H-6) typical of 1,2,3,4-tetrasubstitution. An ABX-spin system of aromatic protons (B-ring) at δ 6.52 (2H, d, *J* = 2 Hz, H-2', H-6') and δ 6.25 (1H, t, H-4') suggested that ring B was symmetrically trisubstituted. The <sup>1</sup>H NMR spectrum also showed the presence of a geranyl group [ δ 3.44 (2H, d, *J* = 7 Hz, H-1''), δ 5.30 (1H, m, H-2''), δ 2.08 (1H, m, H-4''), δ 1.98 (1H, m, H-5''), δ 5.09 (1H, m, H-6''), δ 1.57 (3H, s, H-8''), δ 1.63 (3H, s, H-9'') and δ 1.79 (3H, s, H-10'')]. Interpretation of the <sup>13</sup>C NMR, HMBC and HMQC spectra (Figures 52, 53 and 54) revealed the substitution pattern. The HMBC correlations from H-2' (δ 6.52) to C-β (δ 127.2), C-3'

( $\delta$  159.5); H- $\beta$  (6.79) to C-1' ( $\delta$  141.5), C-6' ( $\delta$  105.7); H- $\alpha$  ( $\delta$  7.36) to C-2 ( $\delta$  154.4); H-1'' ( $\delta$  3.44) to C-3 ( $\delta$  116.4), C-4 ( $\delta$  156.7); H-10'' ( $\delta$  1.79) to C-2'' ( $\delta$  123.5) and C-3'' ( $\delta$  135.7) supported the location of geranyl substituted at C-3.



**Figure 23** Important HMBC (H→C) correlations of CAL4

Compound CAL4 was characterized as a new compound, namely 3-geranyl-2,4,3',5'-tetrahydroxystilbene.

**Table 15** NMR Spectral data of compound CAL4 (acetone- $d_6$ )

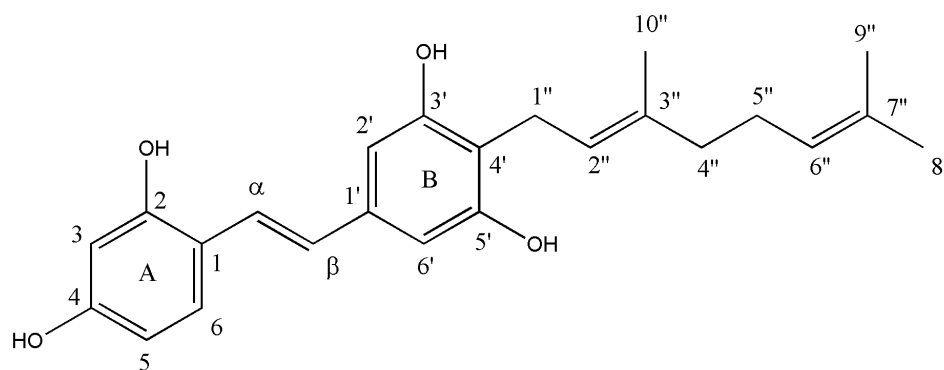
position	compound CAL 4		HMBC (correlation with $^1\text{H}$ )
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	
1		118.2	5, $\beta$
2		154.4	6, $\alpha$ , 1''
3		116.4	1''
4		156.7	6, 1''
5	6.48, d (8)	108.9	-
6	7.25, d (8)	124.8	$\alpha$
$\alpha$	7.36, d (16)	124.6	6
$\beta$	6.79, d (16)	127.2	2', 6'
1'		141.5	$\alpha$ , $\beta$
2'	6.52, d (2)	105.7	$\beta$
3'		159.5	2'
4'	6.25, t, 2	102.4	2', 6'
5'		159.5	2', 6'
6'	6.52, d (2)	105.7	$\beta$
1''	3.44, d (7), 2H	23.1	2''
2''	5.30, m	123.5	10''
3''		135.7	10''
4''	1.98, m	40.6	5''
5''	2.08, m	27.4	4''
6''	5.09, m	125.2	9''
7''		131.7	-
8''	1.63, s	25.8	6'', 9'', 10''
9''	1.57, s	17.7	6'', 8''
10''	1.79, s	16.3	4'', 8''



## 2.5 Structure Determination of Compound CAL5

Compound CAL5 was obtained as a yellow amorphous solid. The ESI-MS revealed a molecular ion at  $m/z$  381.21 (Figure 55). The UV spectrum exhibited absorptions at 220, 261 and 328 nm (Figure 56). The  $^1\text{H}$  NMR spectrum (Figure 57) showed the presence of a *trans*-olefin at  $\delta$  6.82, 7.37 (1H each,  $J = 16$  Hz, H- $\alpha$  and H- $\beta$ ). The  $^1\text{H}$  NMR spectrum showed signals for an ABX-spin system of aromatic protons (A-ring) at  $\delta$  6.42 (1H, d,  $J = 2$  Hz, H-3),  $\delta$  6.36 (1H, dd,  $J = 8, 2$  Hz, H-5) and  $\delta$  7.37 (1H, d,  $J = 8$  Hz, H-6) typical of 2, 4- trisubstitution on the A ring. The aromatic protons (B-ring) at  $\delta$  6.56 (2H, s, H-2', 6') suggested that ring B was symmetrically tetrasubstituted. The  $^1\text{H}$  NMR spectrum showed signals for the presence of a geranyl group at  $\delta$  3.36 (2H, d,  $J = 7$  Hz, H-1''),  $\delta$  5.33 (1H, m, H-2''),  $\delta$  1.98 (2H, m, H-4''),  $\delta$  2.10 (2H, m, H-5''),  $\delta$  5.10 (1H, m, H-6''),  $\delta$  1.57 (3H, s, H-8''),  $\delta$  1.62 (3H, s, H-9'') and  $\delta$  1.78 (3H, s, H-10'').

This compound was identified as chlorophorin [346] through comparison of its  $^1\text{H}$  and  $^{13}\text{C}$ -NMR (Figure 58) and MS data with those reported in the literature (Table 16). The compound was first isolated from *Chlorophora excels* (Christensen *et al.*, 1987).



Chlorophorin [346]

**Table 16** NMR Spectral data of compound CAL5 (acetone- $d_6$ ) and chlorophorin

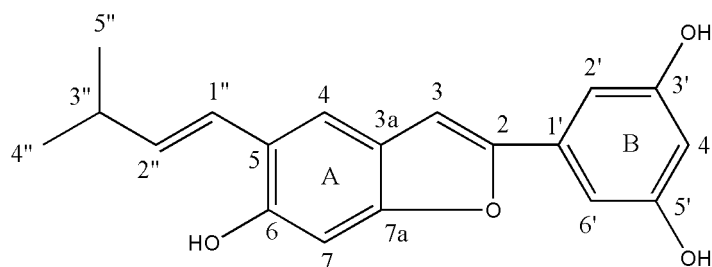
Position	compound CAL5		chlorophorin in CD <sub>3</sub> CN	
	<sup>1</sup> H (mult., <i>J</i> in Hz)	<sup>13</sup> C	<sup>1</sup> H (mult., <i>J</i> in Hz)	<sup>13</sup> C
1		117.5		115.9
2		156.9		154.5
3	6.42, d (2)	103.6	6.44, d (2.4)	101.8
4		158.9		156.7
5	6.36, dd (2, 8)	108.4	6.38, dd (2.4, 8.5)	107.1
6	7.37, d (8)	123.5	7.39, d (8.5)	121.6
1'		134.5		134.1
2'	6.56, s	105.6	6.58, s	104.1
3'		156.7		154.8
4'		114.8		113.2
5'		156.7		154.8
6'	6.56, s	105.6	6.58, s	104.1
α	6.82, d (16)	126.5	6.83, d (16.4)	124.9
β	7.25, d (16)	128.0	7.27, d (16.4)	126.7
1''	3.36, m	23.0	3.37, d (7)	21.3
2''	5.33, m	124.3	5.34, m	121.9
3''		138.1		136.4
4''	1.98, m	40.6	1.93, m	38.8
5''		27.5		25.8
6''	5.10, m	124.3	5.10, m	123.5
7''		131.6		130.4
8''	1.62, s	25.8	1.62, s	24.3
9''	1.57, s	17.7	1.57, s	16.2
10''	1.78, s	16.3	1.79, s	14.7

## 2.6 Structure Determination of Compound CAL6

Compound CAL6 was obtained as a yellow amorphous solid and its molecular formula was analyzed as C<sub>19</sub>H<sub>18</sub>O<sub>4</sub> from its [M+H]<sup>+</sup> ion at *m/z* 311.13 in the ESI-MS (Figure 59). The UV absorptions at 228 and 339 nm (Figure 60) were indicative of a 2-arylbenzofuran skeleton (Yenesew *et al.*, 2002; Kapche *et al.*, 2009), and this was supported by the <sup>1</sup>H-NMR signal at δ 7.01 (1H, br s, H-3) and the <sup>13</sup>C-NMR signals at δ 102.3 (C-3) and δ 155.8 (C-2) (Figures 61 and 62) (Yenesew *et al.*, 2002). The <sup>1</sup>H NMR spectrum showed the presence of an isoprenyl group [two methyl groups at δ 1.10, 1.12 (6H, d, *J* = 6 Hz, H-4'', H-5''), an olefinic proton at δ 2.48 (1H, m, H-3'') and a pair of methylene protons at δ 6.77 (1H, dd, *J* = 16, 1 Hz, H-1''), 6.21 (1H, dd, *J* = 16, 2 Hz, H-2'')]. An AX<sub>2</sub>-spin system of aromatic protons at δ 6.86 (2H, d, *J* = 2

Hz, H-2' and H-6'),  $\delta$  6.32 (1H, d,  $J = 2$  Hz, H-4') was observed for ring B. Two aromatic protons of ring A were observed at  $\delta$  7.60 (1H, s, H-4) and  $\delta$  7.01 (1H, s, H-7). The  $^{13}\text{C}$  NMR spectrum showed that there were five oxygenated  $\text{sp}^2$  carbons  $\delta$  155.8 (C-2),  $\delta$  153.8 (C-6),  $\delta$  155.7 (C-7a),  $\delta$  159.8 (C-3', 5'). The signals for the prenyl group appeared in the  $^{13}\text{C}$  NMR spectrum at  $\delta$  123.0 (C-1''),  $\delta$  137.5 (C-2''),  $\delta$  32.6 (C-3''),  $\delta$  22.9 (C-4'', 5'').

This compound was identified as artotonkin [**341**] through comparison of its  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR and MS data with those reported in the literature (Table 17). The flavonoid was first isolated from *Artocarpus tonkinensis* (Moraceae) (Lien *et al.*, 1997).



[341]

**Table 17** NMR Spectral data of compound CAL6 and artotonkin (acetone- $d_6$ )

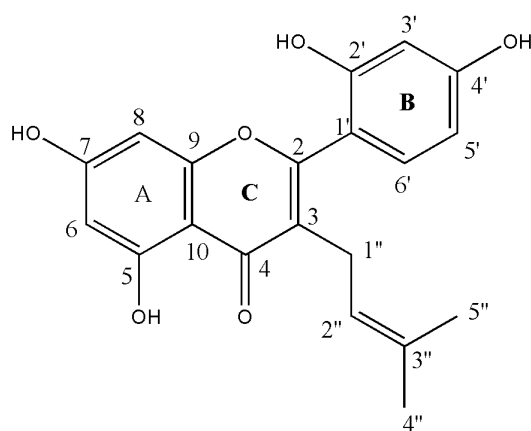
Position	compound CAL6		artotonkin	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2	-	155.8	-	155.6
3	7.01, s	102.3	7.03, br s	102.3
3a	-	133.4	-	133.2
4	7.60, s	118.5	7.62, s	118.3
5	-	123.4	-	123.3
6	-	153.8	-	153.7
7	7.01, s	98.3	7.01, br s	98.3
7a	-	155.7	-	155.7
1'	-	122.9	-	122.9
2'	6.86, d (2)	103.8	6.85, d (2.2)	103.8
3'	-	159.8	-	159.8
4'	6.37, t (2)	103.6	6.37, t (2.2)	103.8
5'	-	159.8	-	159.8
6'	6.86, d (2)	102.3	6.85, d (2.2)	103.5
1''	6.77, dd (16, 1)	123.0	6.78, dd (15.9, 1.1)	123.0
2''	6.21, dd (16, 7)	137.5	6.22, dd (16.1, 7.0)	137.3
3''	2.48, m	32.6	2.49, dd (1.2, 6.7)	32.6
4''	1.10, d (6)	22.9	1.11, d (6.9)	23.0
5''	1.12, d (6)	22.9	1.11, d (6.9)	23.0

## 2.7 Structure Determination of Compound CAL7

Compound CAL7 was obtained as a yellow amorphous solid and its molecular formula was analyzed as  $C_{20}H_{18}O_6$  from its  $[M+H]^+$  ion at  $m/z$  355.12 in the ESI-MS (Figure 63). The UV spectrum exhibited absorptions at 210, 241 and 256 nm, suggestive of a flavone skeleton (Figure 64).

The  $^1H$  NMR spectrum (Figure 65) showed the presence of an isoprenyl group [two methyl groups at  $\delta$  1.43 (3H, s, H-4''),  $\delta$  1.57 (3H, s, H-5''), an olefinic proton at  $\delta$  5.12 (1H, m, H-2'') and a pair of methylene protons at  $\delta$  3.11 (2H, d,  $J = 7$  Hz, H-1'')]. The AX-spin system of *meta*-coupled aromatic protons (A-ring) at  $\delta$  6.32 (1H, d,  $J = 2$  Hz, H-6),  $\delta$  6.24 (1H, d,  $J = 2$  Hz, H-8) suggested that the A ring was tetrasubstituted. An ABX-spin system of aromatic protons at  $\delta$  6.55 (1H, d,  $J = 2$  Hz, H-3'),  $\delta$  6.50 (1H, dd,  $J = 8, 2$  Hz, H-5') and  $\delta$  7.18 (1H, d,  $J = 8$  Hz, H-6) was observed for the B ring.

This compound was identified as albanin A [132] through comparison of its  $^1H$ - and  $^{13}C$ -NMR (Figure 66) and MS data with those reported in the literature (Table 18). (Ferrari *et al.*, 1989).



[132]

**Table 18** NMR Spectral data of compound CAL7 (acetone- $d_6$ ) and albanin A (methanol- $d_4$ )

Position	compound CAL7		albanin A	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2		162.4		163.3
3		121.8		121.6
4		183.0		182.9
5		163.4		162.0
6	6.24, d (2)	99.2	6.25, d (2)	99.1
7		164.7		164.6
8	6.32, d (2)	94.2	6.32, d (2)	94.0
9		159.3		159.2
10		104.0		105.3
1'		113.1		113.0
2'		157.2		157.1
3'	6.55, d (2)	103.1	6.56, d (2.5)	103.9
4'		161.5		161.3
5'	6.50, dd (2, 8)	108.1	6.51, dd (2.5, 8)	108.0
6'	7.18, d (8)	132.1	7.18, d (8)	132.1
1''	3.11, d (7.1)	24.6	3.10, d (7)	24.4
2''	5.12, m	122.7	5.13, m	122.5
3''		132.3		131.9
4''	1.57, s	25.8	1.43, s	25.6
5''	1.43, s	17.6	1.57, s	17.5

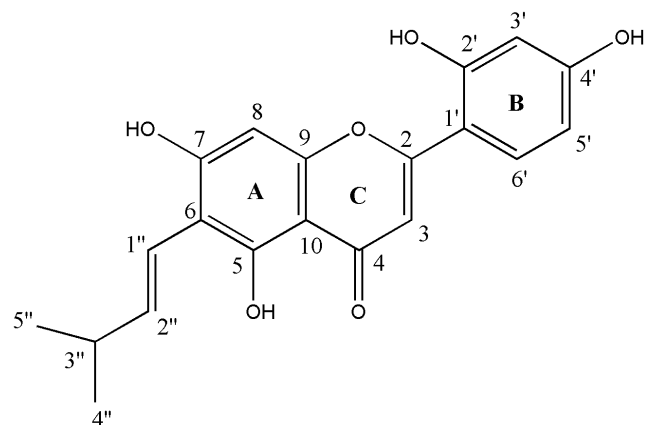
## 2.8 Structure Determination of Compound CAL8

Compound CAL8 was obtained as a yellow amorphous solid and its molecular formula was analyzed as  $\text{C}_{20}\text{H}_{18}\text{O}_6$  from its  $[\text{M}+\text{H}]^+$  ion at  $m/z$  355.12 in the ESI-MS (Figure 67). The UV absorptions at 211, 287 and 347 nm (Figure 68) were suggestive of a flavone skeleton.

The  $^1\text{H}$  NMR spectrum (Figure 69) showed the presence of an isoprenyl group [two methyl groups at  $\delta$  1.11 (3H, s, H-4''),  $\delta$  1.09 (3H, s, H-5''), an methine proton at  $\delta$  2.43 (1H, m, H-3'') and a pair of olefinic protons at  $\delta$  6.55 (1H, d,  $J = 16$  Hz, H-1''), 6.69 (1H, d,  $J = 16$  Hz, H-2'')]. An ABX -spin system of aromatic protons at  $\delta$  6.42 (1H, d,  $J = 2$  Hz, H-3'),  $\delta$  6.45 (1H, dd,  $J = 9, 2$  Hz, H-5') and  $\delta$  7.73 (1H, d,  $J = 9$  Hz, H-6') was observed for ring B. Aromatic protons at  $\delta$  6.46 (1H, s, H-8) and  $\delta$  7.12 (1H, s, H-3) were also observed.

This compound was identified as isoartocarpesin [58] through comparison of its  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR (Figure 70) and MS data with those reported in the literature

(Table 19). The flavonoid was first isolated from *Artocarpus incisus* (Moraceae) (Shimizu *et al.*, 1998).



[58]

**Table 19** NMR Spectral data of compound CAL8 (methanol-*d*<sub>4</sub>) and isoartocarpesin (acetone-*d*<sub>6</sub>)

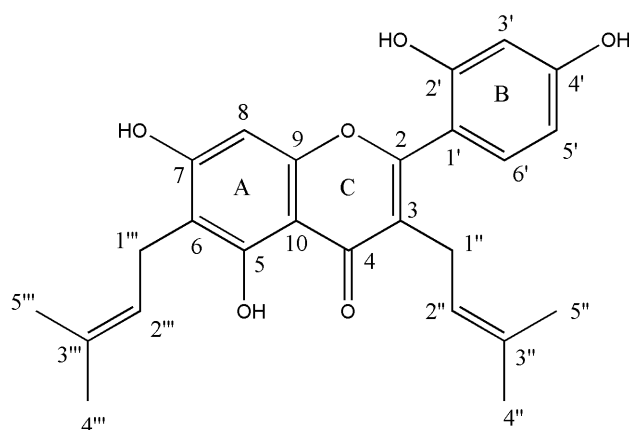
Position	compound CAL8		isoartocarpesin	
	<sup>1</sup> H (mult., <i>J</i> in Hz)	<sup>13</sup> C	<sup>1</sup> H (mult., <i>J</i> in Hz)	<sup>13</sup> C
2		163.86		163.09
3	7.12, s	108.32	7.03, s	108.91
4		184.52		184.14
5		163.24		161.19
6		110.19		109.88
7		163.26		162.65
8	6.46, s	94.25	6.54, s	94.63
9		157.28		157.0
10		105.02		105.48
1'		110.84		111.17
2'		160.31		159.72
3'	6.42, d (2)	104.22	6.56	104.78
4'		163.31		162.99
5'	6.45, dd (9, 2.5)	109.09	6.51, dd (8.55, 2.45)	109.60
6'	7.73, d (9)	130.90	7.77, d (8.55)	131.36
1''	6.55, d (16)	117.41	6.58, dd (0.78, 16.12)	117.75
2''	6.69, dd (16, 7)	142.63	6.72, dd (7.08, 16.12)	142.40
3''	2.43, m	34.31	2.39, m	34.38
4''	1.10, d (6.4)	23.20	1.04, d (6.8)	23.61
5''	1.10, d (6.4)	23.20	1.04, d (6.8)	23.61

## 2.9 Structure Determination of Compound CAL9

Compound CAL9 was obtained as a yellow amorphous solid and its molecular formula was analyzed as  $C_{25}H_{26}O_6$  from its  $[M+H]^+$  ion at  $m/z$  423.18 in the ESI-MS (Figure 71). The UV absorptions at 212, 262 and 312 nm (Figure 72) were suggestive of a flavone skeleton.

The  $^1H$ -NMR spectrum (Figure 73) exhibited signals for two prenyl groups. The first prenyl group [ $\delta$  1.77 (3H, br s, H-5'''),  $\delta$  1.64 (3H, br s, H-4'''),  $\delta$  3.36 (2H, m, H-1''') and  $\delta$  5.29 (1H, m, H-2''')] was located on C-6 of ring A. An aromatic proton signal for H-8 was observed at  $\delta$  6.39. The second prenyl group [ $\delta$  1.43 (3H, s, H-4''),  $\delta$  1.56 (3H, s, H-5''),  $\delta$  3.11 (2H, d,  $J = 6.9$  Hz, H-1'') and  $\delta$  5.14 (1H, m, H-2'')] was situated at C-3 of ring C. For ring B, the presence of an ABM spin system at  $\delta$  6.52 (1H, dd,  $J = 2, 8$  Hz, H-5'),  $\delta$  6.57 (1H, d,  $J = 2$  Hz, H-3'), 7.19 (1H, d,  $J = 8$  Hz, H-6') indicated 2',4'-dihydroxy substitution. The  $^{13}C$ -NMR spectrum (Figure 74) displayed 25 signals for one carbonyl group, four methyl groups, six methine carbons, two methylene carbons and twelve quaternary carbons. The signals  $\delta$  162.3 (C-2),  $\delta$  121.6 (C-3) and  $\delta$  183.1 (C-4) confirmed the flavone structure.

This compound was identified as cudraflavone C [**34**] through comparison of its  $^1H$ - and  $^{13}C$ -NMR and MS data with those reported in the literature (Table 20). The flavonoid was first isolated from *Cudrania tricuspidata* (Moraceae) (Hano *et al.*, 1990c).



[34]

**Table 20** NMR Spectral data of compound CAL9 and cudraflavone C (acetone- $d_6$ )

Position	compound CAL9		cudraflavone C	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2	-	162.3	-	162.0
3	-	121.6	-	121.5
4	-	183.1	-	183.0
5	-	160.1	-	160.0
6	-	111.8	-	111.8
7	-	162.3	-	162.3
8	6.39, s	93.6	6.40, s	93.5
9	-	157.1	-	157.0
10	-	105.1	-	105.0
1'	-	113.2	-	113.1
2'	-	157.1	-	157.1
3'	6.57, d (2)	103.9	6.57, d, (2.0)	103.9
4'	-	161.4	-	161.3
5'	6.52, dd (2, 8)	108.1	6.52, dd, (8.0, 2.0)	108.1
6'	7.19, d (8)	132.8	7.19, d (8.0)	132.2
1''	3.11, d (7), 2H	24.7	3.12, d (7.0), 2H	24.6
2''	5.14, m	122.8	5.14, m	122.7
3''	-	131.9	-	132.0
4''	1.43, s	17.6	1.43, s	17.9
5''	1.56, s	25.8	1.57, s	25.8
1'''	3.36, d (7), 2H	22.0	3.41, d (6.9), 2H	22.0
2'''	5.29, m	123.5	5.29, m	123.3
3'''	-	131.5	-	131.4
4'''	1.64, s	25.9	1.65, s	25.9
5'''	1.77, s	17.6	1.78, s	17.6

### 2.10 Structure Determination of Compound CAL10

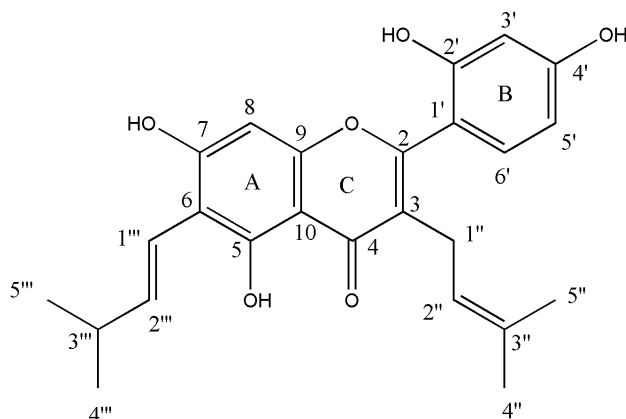
Compound CAL10 was obtained as a yellow amorphous solid and its molecular formula was analyzed as  $\text{C}_{25}\text{H}_{26}\text{O}_6$  from its  $[\text{M}+\text{H}]^+$  ion at  $m/z$  423.18 in the ESI-MS (Figure 75). The UV spectrum showed absorptions at 210, 279 and 325 nm (Figure 76).

The  $^1\text{H}$ -NMR spectrum (Figure 77) exhibited signals for two prenyl groups. The first prenyl group [ $\delta$  1.08 (3H, d,  $J = 6$  Hz, H-4'''),  $\delta$  1.10 (3H, d,  $J = 6$  Hz, H-5'''),  $\delta$  2.45 (1H, m, H-3''')] was located on ring A. The second prenyl group [ $\delta$  1.43 (3H, s, H-4''),  $\delta$  1.57 (3H, s, H-5''),  $\delta$  3.11 (2H, d,  $J = 7$ , H-1'') and  $\delta$  5.12 (1H, m, H-2'')] was situated at C-3 of ring C. An aromatic singlet signal appeared at  $\delta$  6.41 for H-8. For ring B, the presence of an ABM spin system at  $\delta$  6.50 (1H, dd,  $J = 2, 8$  Hz, H-5'),  $\delta$  6.57 (1H, d,  $J = 2$  Hz, H-3'),  $\delta$  7.18 (1H, d,  $J = 8$  Hz, H-6') indicated 2',4'-dihydroxy



substitution. The  $^{13}\text{C}$ -NMR spectrum (Figure 78) showed signals at  $\delta$  162.12 (C-2),  $\delta$  121.62 (C-3) and  $\delta$  183.28 (C-4) confirmed flavone structure.

This compound was identified as norartocarpin [**128**] through comparison of its  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR and MS data with those reported in the literature (Table 21).



[**128**]

**Table 21** NMR Spectral data of compound CAL10 and norartocarpin(acetone- $d_6$ )

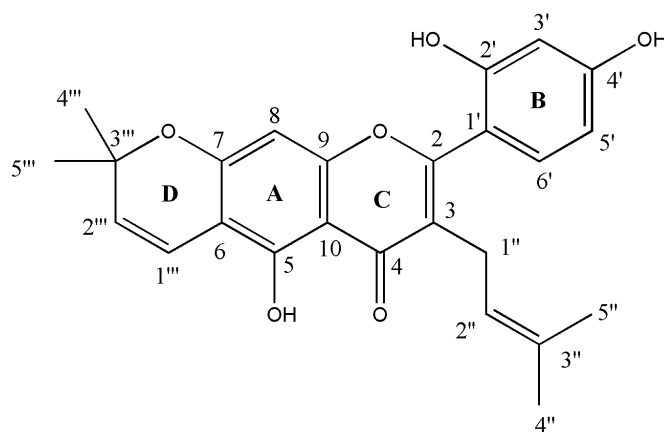
Position	compound CAL10		norartocarpin in methanol- $d_4$	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2		162.1		163.2
3		121.6		121.7
4		183.3		183.8
5		160.8		160.8
6		109.2		109.7
7		163.2		163.1
8	6.41, s	93.8	6.30, s	93.8
9		157.0		157.7
10		105.1		105.1
1'		113.0		113.3
2'		161.8		161.8
3'	6.57, d (2)	103.9	6.58, d (2)	103.7
4'		157.0		157.5
5'	6.50, dd (8, 2)	108.1	6.38, dd (8, 2)	107.9
6'	7.18, d (8)	132.1	7.04, d (8)	132.3
1''	3.11, d (7), 2H	24.6	3.07, d (6.8), 2H	24.2
2''	5.12, m	122.7	5.08, m	122.9
3''		132.3		132.6
4''	1.38, s	17.6	1.43, s	17.6
5''	1.54, s	25.8	1.57, s	25.8
1'''	6.64, d (16)	117.3	6.56, d (16)	117.5
2'''	6.76, dd (16, 7)	141.9	6.69, d (16, 7.04)	142.2
3'''	2.45, m	33.9	2.41, m	34.4
4'''	1.09, d (6)	23.1	1.08, d (6)	23.2
5'''	1.11, d (6)	23.1	1.10, d (6)	23.2

## 2.11 Structure Determination of Compound CAL11

Compound CAL11 was obtained as a yellow amorphous solid and its molecular formula was analyzed as  $C_{25}H_{24}O_6$  from its  $[M+Na]^+$  ion at  $m/z$  423.18 in the ESI-MS (Figure 79). The UV spectrum showed absorptions at 223, 262 and 314 nm (Figure 80).

The  $^1H$ -NMR spectrum (Figure 81) exhibited the signals of prenyl groups signals at [ $\delta$  1.45 (3H, s, H-4''),  $\delta$  1.57 (3H, s, H-5''),  $\delta$  3.12 (2H, d,  $J = 7$  Hz, H-1'') and  $\delta$  5.12 (1H, m, H-2'')] which was located at C-3 of ring C. The second prenyl group [ $\delta$  1.43 (6H, s, H-4''', 5'''),  $\delta$  5.73 (1H, d,  $J = 10$  Hz, H-1''') and 6.68 (1H, d,  $J = 10$  Hz, H-2''') was substituted at C-6 of ring D. An aromatic singlet signal appeared at  $\delta$  6.26 (H-8). For ring B, the presence of an ABM spin system at  $\delta$  6.51 (1H, dd,  $J = 2, 8$  Hz, H-5'),  $\delta$  6.56 (1H, d,  $J = 2$ , H-3'),  $\delta$  7.19 (1H, d,  $J = 8$  Hz, H-6') indicated 2',4'-dihydroxy substitution. The  $^{13}C$ -NMR spectrum (Figure 82) showed signals at  $\delta$  158.5 (C-2),  $\delta$  121.9 (C-3) and  $\delta$  183.2 (C-4) confirmed a flavone structure.

This compound was identified as cudraflavone B [347] through comparison of its  $^1H$ - and  $^{13}C$ -NMR and MS data with those reported in the literature (Table 22). The flavonoid was first reported from *Cudrania tricuspidata* (Moraceae) (Fujimoto *et al.*, 1984).



[347]

**Table 22** NMR Spectral data of compound CAL11 and cudraflavone B (acetone- $d_6$ )

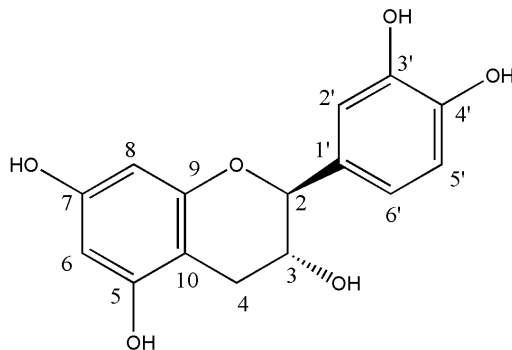
Position	compound CAL11		cudraflavone B	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2		158.5		158.4
3		121.9		121.7
4		183.2		183.2
5		157.2		157.2
6		105.7		105.7
7		162.7		162.8
8	6.26, s	95.2	6.24, s	95.2
9		160.1		160.6
10		105.7		105.6
1'		112.9		112.7
2'		157.3		157.3
3'	6.56, d (2)	103.9	6.55, d (2)	103.8
4'		161.5		161.6
5'	6.51, dd (2, 8)	108.1	6.49, dd (2, 8)	108.0
6'	7.19, d (8)	132.3	7.16, d (8)	132.2
1''	3.12, d (7), 2H	24.6	3.15, d (6.5), 2H	24.6
2''	5.12, t (7)	122.6	5.08, t	122.5
3''		132.2		132.2
4''	1.45, s	25.8	1.42, s	25.8
5''	1.57, s	17.6	1.52, s	17.7
1'''	5.73, d (10)	115.9	5.71, d (10)	115.9
2'''	6.68, d (10)	129.1	6.66, d (10)	129.1
3'''		78.6		78.6
4'''	1.43, s	28.4	1.42, s	28.3
5'''	1.43, s	28.4	1.42, s	28.3

### 2.12 Structure determination of compound CAL12

Compound CAL12 was obtained as a brown amorphous powder and its molecular formula was analyzed as  $\text{C}_{15}\text{H}_{14}\text{O}_6$  from its  $[\text{M}+\text{H}]^+$  ion at  $m/z$  291.09 in the ESI-MS (Figure 83). The UV spectrum exhibited absorption at 211 and 280 nm (Figure 84).

The  $^1\text{H}$ -NMR spectrum (Figure 85) showed proton signals for a flavan skeleton at  $\delta$  2.85 (1H, dd,  $J = 16, 7$  Hz, H-4a) and  $\delta$  2.52 (1H, dd,  $J = 16, 6$  Hz, H-4b), aromatic proton doublets at 5.93 (1H, d,  $J = 2$  Hz, H-6),  $\delta$  5.86 (1H, d,  $J = 2$  Hz, H-8) and an ABX system at  $\delta$  6.84 (1H, d,  $J = 2$  Hz, H-2'),  $\delta$  6.71 (1H, dd,  $J = 8, 2$  Hz, H-6').

This compound was identified as (–) catechin [108] through comparison of its  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR (Figure 86), optical rotation (– 48) and MS data with those reported in the literature (Table 23). The compound was isolated from *Carapa guianensis* (Meliaceae) (Shu-Hua *et al.*, 2003).



[108]

**Table 23** NMR Spectral data of compound CAL12 and (–) catechin (methanol- $d_4$ )

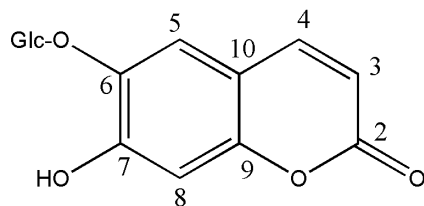
Position	compound CAL12		(–) catechin	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2	4.75, d (7.5)	82.9	4.57, d (7.6)	82.3
3	3.97, ddd (5, 8, 7)	68.8	4.01, ddd (5.4, 8.2, 7.6)	68.3
4a	2.85, dd (16, 7)	28.5	2.88, dd (16.1, 5.4)	28.2
4b	2.51, dd (16, 6)	28.5	2.52, dd (16, 2)	28.2
5		157.6		156.9
6	5.93, d (2)	96.4	5.95, d (1.8)	96.2
7		157.8		157.0
8	5.86, d (2)	95.6	5.88, d (1.8)	95.4
9		156.9		156.3
10		100.9		100.7
1'		132.3		131.6
2'	6.84, d (2)	115.3	6.83, d (1.8)	115.0
3'		146.3		145.6
4'		146.3		145.7
5'	6.76, d (8)	116.1	6.77, d (8)	116.0
6'	6.71, dd (8, 2)	120.0	6.72, dd (8.0, 1.8)	119.9

### 2.13 Structure determination of compound CAL13

Compound CAL6 was obtained as a yellow amorphous solid and its molecular formula was analyzed as  $\text{C}_{15}\text{H}_{16}\text{O}_9$  from its  $[\text{M}+\text{H}]^+$  ion at  $m/z$  341.08 in the ESI-MS (Figure 87). The UV spectrum exhibited absorptions at 207, 221 and 331 nm (Figure 88), characteristic of a coumarin structure. (Matsuda *et al.*, 1995).

The  $^1\text{H}$  and  $^{13}\text{C}$ -NMR spectra (Figures 89 and 90) indicated the presence of a coumarin skeleton by the signals at  $\delta$  6.21 (1H, d,  $J=10$  Hz, H-3),  $\delta$  7.83 (1H, d,  $J=10$  Hz, H-4) and  $\delta$  113.0 (C-3),  $\delta$  146.0 (C-4). The two signals at  $\delta$  6.80 and  $\delta$  7.43, showed that substituents were present at C-6 and C-7 of the B-ring.

This compound was identified as aesculin [348] through comparison of its  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR and MS data with those reported in the literature (Table 24) (Matsuda *et al.*, 1995).



[348]

**Table 24** NMR Spectral data of compound CAL13 and aesculin (methanol- $d_4$ )

Position	compound CAL13		aesculin	
	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$	$^1\text{H}$ (mult., $J$ in Hz)	$^{13}\text{C}$
2		163.7		163.7
3	7.83, d (10)	113.0	7.89, d (9.5)	113.1
4	6.21, d (10)	146.0	6.17, d (9.5)	145.9
5	7.43, s	116.9	7.39, d (2)	116.9
6		153.7		153.4
7		144.5		144.4
8	6.80, s	104.6	6.77, d (2)	104.6
9		152.7		152.6
10		112.8		112.8
Glc-1		104.4		104.4
Glc-2		74.9		74.8
Glc-3		78.5		78.5
Glc-4		71.4		71.4
Glc-5		77.7		77.7
Glc-6		62.6		62.6

### 3 Tissue Culture of *A. lakoocha*

#### 3.1 Establishment of callus cultures

Callus formation was induced successfully on WPM (Woody Plant Medium) supplemented with 1 mg/l 2,4-dichlorophenoxyacetic acid (2,4-D) and 1 mg/l benzyladenine (BA).



24 a



24 b



24 c

**Figure 24 Induction of callus from the explants of *A. lakoocha* on WPM agar medium contain 1 mg/l 2, 4-D and 1 mg/l BA**

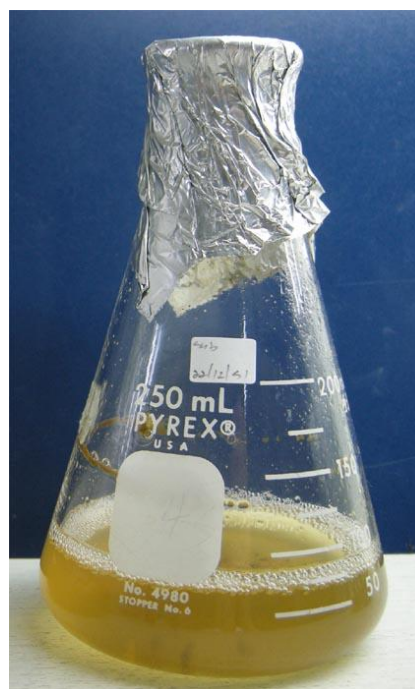
- 24 a Induction of callus from the cotyledon explants
- 24 b Induction of callus from the stem explants
- 24 c Induction of callus from the leaf explants

#### 3.2 Establishment of cell suspension cultures

Cell suspension cultures of *A lakoocha* were obtained from the callus which were maintained by a regular subculturing as described earlier. The callus tissues were separated into small aggregates before transfers into WPM liquid medium culture and rotated at 120 rpm. on a rotary shaker. Cell suspension was maintained in the same medium by subculture every three weeks.



25 a



25 b

**Figure 25 Cell suspension cultures of *A. lakoocha* maintained in WPM liquid medium contain 1 mg/l 2,4-D and 1 mg/l BA.**

**25 a** The apparent friable callus cultures

**25 b** cell suspension cultures

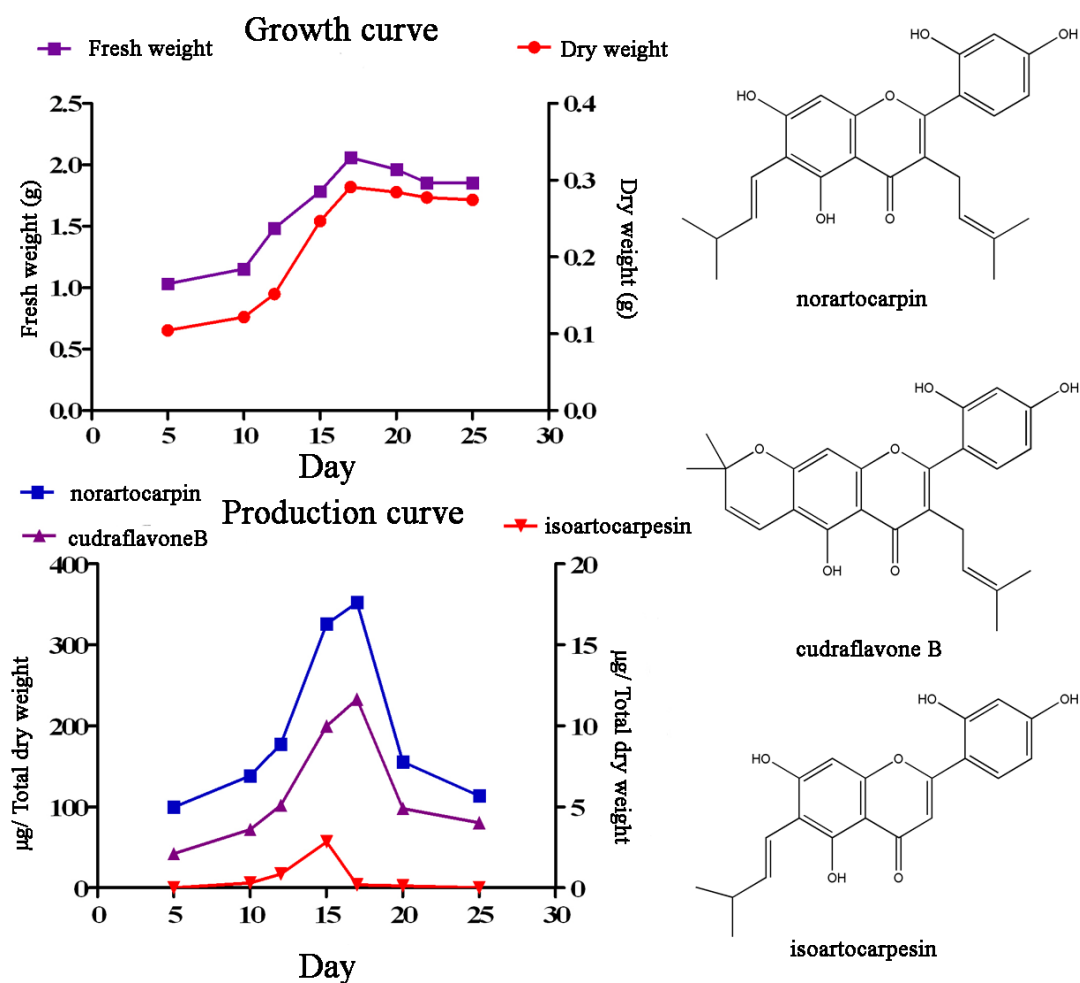
### **3.3 Relationship between cell growth and secondary metabolites production**

Figure 26 shows the culture growth of the callus cultures of *A. lakoocha* during a period of 25 days. It can be seen that the starting callus cultures had a lag phase of almost 10 days before entering a rapid growth for a period of 7 days (day 10-day 17) to obtain its highest peak at day 17, with about 0.2 g dry weight.

During this growth cycle, the formation of secondary metabolites was monitored by TLC-densitometric method (Figure 27). Various compounds isolated from the callus cultures with structures elucidated were used as standard. Figure 27 shows that the TLC pattern of the ethanolic extract obtained from various ages of the callus tissue. The standard mixture and individual 11 pure compounds were also chromatographed in the same TLC condition. It can be seen that three major secondary products, i.e. isoartocarpesin, norartocarpin and cudraflavone B were present all the period of the culture growth, gradually increased during the first 17 days and declined rapidly when the culture entered the stationary phase until day 25.

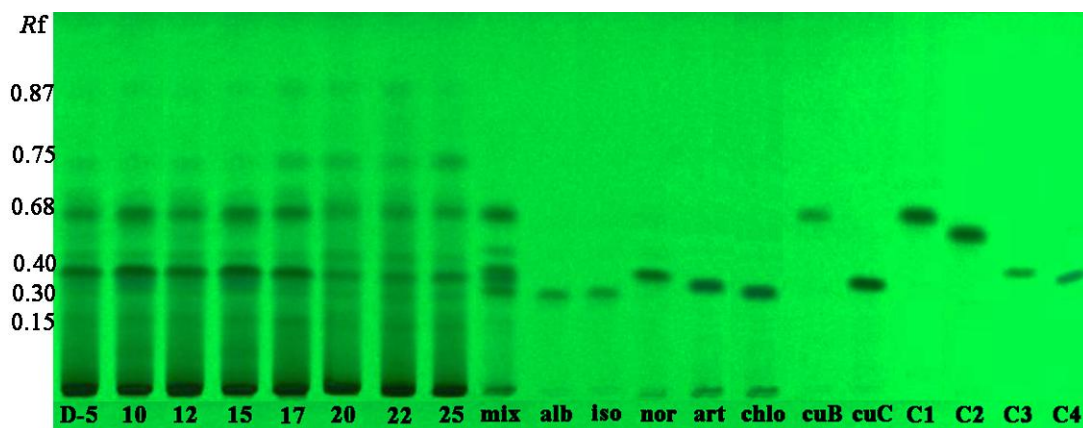
It should be noted that no sequential formation of these metabolites was observed. All of the compounds were formed and changed parallel and in the same period of growth curve.

Based on these results, it was concluded that the whole profile of secondary metabolites of *A. lakoocha* was expressed all over the growth period of the callus culture until entering the stationary phase of the fully mature tissues.



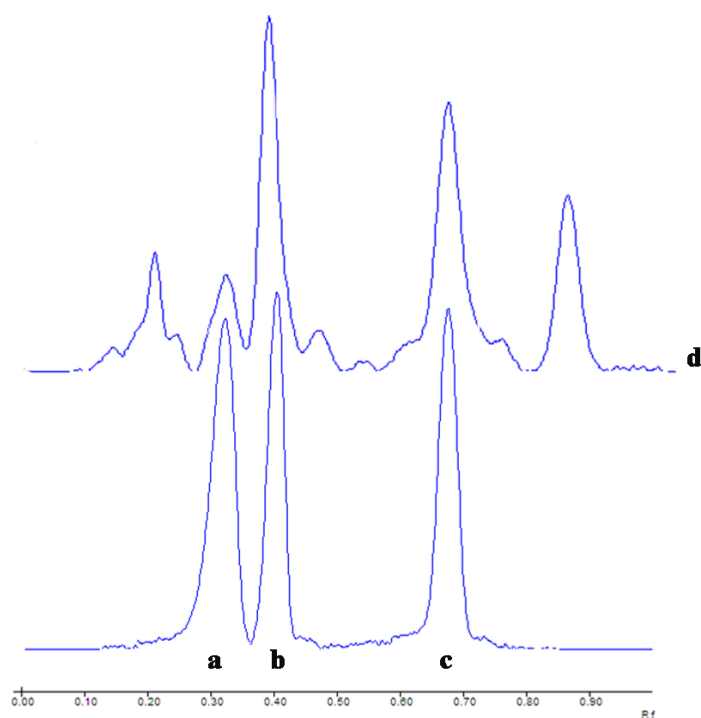
**Figure 26** Growth curve and production curve of *A. lakoocha* callus culture





**Figure 27** TLC pattern of ethanol extract from *A. lakoocha* callus cultures and time-courses of cell growth under 254 nm UVlight with authentic compounds

D-5 = callus day 5	iso = isoartocarpesin
D-10 = callus day 10	nor = norartocarpin
D-12 = callus day 12	art = artotonkin
D-15 = callus day 15	chlo = chlorophorin
D-17 = callus day 17	cuB = cudraflavone B
D-20 = callus day 20	cuC = cudraflavone C
D-22 = callus day 22	C1 = CAL 1
D-25 = callus day 25	C2 = CAL 2
Mix = mix standard	C3 = CAL 3
alb = albanin A	C4 = CAL 4



**Figure 28** Representative TLC densitometric chromatograms.

- |                   |                                  |
|-------------------|----------------------------------|
| a isoartocarpesin | b norartocarpin                  |
| c cudraflavone B  | d chromatogram of callus extract |

#### 4. Free Radical Scavenging Activity

Free radicals are products of normal cellular metabolism that can be defined as an atom or molecules containing one or more unpaired electrons in its atomic or molecular orbitals. Reactive oxygen species (ROS) represent the most important class of radical species generated in living systems. They include superoxide ( $O_2^{\cdot-}$ ), peroxy ( $ROO^{\cdot}$ ), alkoxy ( $RO^{\cdot}$ ), hydroxyl ( $HO^{\cdot}$ ) and hydrogen peroxide ( $H_2O_2$ ). ROS can be either deleterious or beneficial, but the excess ROS can damage cellular lipids, proteins and DNA (Valko *et al.*, 2007). These pathophysiological conditions can cause violent chronic diseases including cancer, arteriosclerosis, stroke, rheumatoid arthritis, neurodegeneration and diabetes (Fang, Yang, and Wu, 2002). In this study, ten compounds from *A. lakoocha* cultures were evaluated for their DPPH scavenging activity. Quercetin was employed as positive control. The results are summarized in Table 25.

**Table 25** Percentage of DPPH reduction by compounds isolated from *A. lakoocha* calli.

compound	DPPH 1 mg/ml	DPPH IC <sub>50</sub> ( $\mu$ M)
1. norartocarpin (CAL10)	46.52738	-
2. cudraflavone C (CAL9)	34.18487	-
3. albanin A (CAL7)	42.73605	-
4. aesculin (CAL13)	17.95114	-
5. artotonkin (CAL6)	95.34351	12.5
6. chlorophorin (CAL5)	95.53859	38.63
7. cudraflavone B (CAL11)	29.2228	-
8. isoartocarpesin (CAL8)	38.89575	-
9. <i>trans</i> -2,4,5'-trihydroxy-3'-methoxy-4'-(3-methyl- <i>E</i> -but-1-enyl)-stilbene (CAL3)	95.6152	37.47
10. 3-geranyl-2,4,3',5'-tetrahydroxystilbene (CAL4)	94.48479	34.29
11. quercetin*	95.67253	10.3

\* positive control

As shown in Table 25, ten pure compounds were tested for their DPPH free radical scavenging activity. It can be seen that all compounds exhibited recognizable DPPH scavenging activity. The most active compound was artotonkin [315], but its activity was lower than that of the positive control quercetin.

#### 5. Tyrosinase Inhibitory Activity

Tyrosinase is a copper monooxygenase enzyme widely distributed in nature. It has been found in plants, fungi, insects and animals. A number of physiological functions of this

enzyme have been studied (Gelder *et al.*, 1997). Tyrosinase is one of the important key enzymes involved in the molting process of insects (Kubo, Yokokawa and Kinst-Hori, 1995). A search for its inhibitors may therefore lead to discovery of insect control agents. In plants, tyrosinase has been found to be responsible for browning in fruits and vegetables. In mammals and humans, the biosynthesis of melanin has been studied intensively by Raper, and subsequently by Manson (Britton, 1983), which led to the proposal of Raper-Manson scheme of melanogenesis and the function of tyrosinase in the biosynthesis of the skin pigment melanin is well-established (Gelder *et al.*, 1997). Thus, the study of tyrosinase inhibitors should be useful for the treatment of localized hyperpigmentation in human. Moreover, tyrosinase inhibitors are becoming more popular in the development of cosmetic products (Kubo *et al.*, 1995).

In this study, the tyrosinase inhibitory activity of each pure compound was determined by a modified dopachrome method (Masamoto *et al.*, 1980; Iida *et al.*, 1995 and Morita *et al.*, 1994). The activity of each of these compounds was expressed as IC<sub>50</sub> value (concentration of 50% inhibition) in comparison with oxyresveratrol and kojic acid, known inhibitors of tyrosinase. The results are summarized in Table 26.

**Table 26** Percentage of tyrosinase inhibition by compounds isolated from *A. lakoocha* calli.

compound	Tyrosinase Inhibitory 0.33 mg/ml	Tyrosinase Inhibitory IC <sub>50</sub> (μM)
1. norartocarpin (CAL10)	22.68657	-
2. cudraflavone C (CAL9)	35.52239	-
3. albanin A (CAL7)	11.64179	-
4. aesculin (CAL13)	43.58209	-
5. artotonkin (CAL6)	79.40299	46.2
6. chlorophorin (CAL5)	85.97015	7.90
7. cudraflavone B (CAL11)	20.59701	-
8. isoartocarpesin (CAL8)	26.56716	-
9. <i>trans</i> -2,4,5'-trihydroxy-3'-methoxy-4'-(3-methyl- <i>E</i> -but-1-enyl)-stilbene (CAL3)	87.16418	12.8
10. 3-geranyl-2,4,3',5'-tetrahydroxystilbene (CAL4)	66.26866	66.0
11. Kojic acid*	57.31343	62.29
12. Oxyresveratrol*	83.28358	1.73

\* positive control

Among the ten pure compounds tested for tyrosinase inhibitory activity, it can be seen that stilbenoids showed recognizable tyrosinase inhibitory activity, whereas other compounds showed weak activity.

## CHAPTER V

### CONCLUSION

The TLC densitometric method developed in this study is a simple, convenient, sensitive and reliable procedure. It was an effective analytical tool for the evaluation of oxyresveratrol content in both *A. lakoocha* heartwood and the traditional drug Puag-Haad and was modified for analysis of secondary metabolite production in cell cultures.

Callus cultures of *A. lakoocha* can be established from aerial part explants on WPM medium which contained 20 g/l sucrose, 1 mg/l 2,4-D, 1 mg/l BA and 0.8 % (w/v) agar and subculture on this medium.

From cell cultures of *A. lakoocha*, thirteen compounds were characterized. Most of them found in Genus *Artocarpus*, but thirteen isolates except norartocarpin were found for the first time from this plant. It should be noted that a trace amount of oxyresveratrol was detected by TLC densitometry and comparison of the UV–Vis spectra with reference standards. However, full identification could not be done. The major metabolites were isoartocarpesin, norartocarpin and cudraflavone B. These compounds were present all the period of the culture growth, gradually increased during the first 17 days and declined rapidly when the culture entered the stationary phase until day 25. All of the compounds were formed and changed parallel and in the same period of growth curve. Based on these results, it was concluded that the whole profile of secondary metabolites of *A. lakoocha* was expressed all over the growth period of the callus culture until entering the stationary phase of the fully mature tissues.

The study also focused on the production of secondary metabolites from callus cultures of *A. lakoocha*, and this led to the isolation of four new compounds together with the nine known compounds. The isolated compounds were evaluated for their free radical scavenging and antityrosinase activity. Artotonkin, chlorophorin, *trans*-2,4,5'-trihydroxy-3'-methoxy-4'-(3-methyl-*E*-but-1-enyl)-stilbene and 3-geranyl-2,4,3',5'-tetrahydroxystilbene showed appreciable free radical scavenging activity and antityrosinase activity, while the other compounds were weakly active. In previous studies, norartocarpin, cudraflavone C, artotonkin and albanin A showed a strong inhibitory

effect on pancreatic lipase (Zhao *et al.*, 2009), whereas cudraflavone C had a strong inhibitory effect on murine leukemia P-388 (Syah *et al.*, 2006). In this study cudraflavone C and norartocarpin were found as main compounds from callus cultures. Further studies should also be carried out on developing techniques for large scale production. These bioactive compounds from *A. lakoocha* provide possibilities for application in medicine and cosmetics.

## References

- Abou-karam, M., and Shier, W. T. 1990. A simplified plaque reduction assay for antiviral agents from plants. Demonstration of frequent occurrence of antiviral activity in higher plants. J. Nat. Prod. 53: 340-344.
- Achmad, S. A., *et al.* 1996. A new prenylated flavone from *Artocarpus champeden*. J. Nat. Prod. 59: 878-879.
- Aida, M., Shinomiya, K., Hano, Y., and Nomura, T. 1993. Artonins J, K and L, three new isoprenylated flavones from the root bark of *Artocarpus heterophyllus* Lamk. Heterocycles 36: 575-583.
- Aida, M., Shinomiya, K., Matsuzawa, K., Hano, Y., and Nomura, T. 1994. Artonins Q, R, S, T, and U, five new isoprenylated phenols from the root bark of *Artocarpus heterophyllus* Lamk. Heterocycles 39: 847-858.
- Aida, M., Yamaguchi, N., Hano, Y., and Nomura, T. 1997. Artonols A, B, C, D, and E, five new isoprenylated phenols from the bark of *Artocarpus communis* Forst. Heterocycles 45: 163-175.
- Altman, L. J., and Zito, S. W. 1976. Sterols and triterpenes from the fruit of *Artocarpus altilis*. Phytochemistry 15: 829-830.
- An, B. J., Choi, J. Y., Kwon, I. B., Nishioka, I., and Choi, C. 1992. Structure and isolation of glucosyl transferase inhibitor from jackfruit. Han'guk Saenghwa Hakhoechi 25: 347-353. (CA. 118: 87474j).
- Arora, J. S., Sanhu, R. S., Kamboj, S. S., and Chopra, S. K. 1987. Occurrence and characterization of lympho-agglutinins in Indian plants. Vox Sang 52: 134-137. (CA. 106: 211018g).
- Arung, E.T., Shimizu, K., and Kondo, R. 2006a. Inhibitory effect of artocarpanone from *Artocarpus heterophyllus* on melanin biosynthesis. Biol. Pharm. Bull. 29: 1966-1969.
- Arung, E.T., Shimizu, K., and Kondo, R. 2006b. Inhibitory effect of isoprenoid-substituted flavonoids isolated from *Artocarpus heterophyllus* on melanin biosynthesis. Planta Med. 72: 847-850.
- Arung, E.T., Shimizu, K. and Kondo, R. 2007. Structure– Activity Relationship of Prenyl-Substituted Polyphenols from *Artocarpus heterophyllus* as Inhibitors of

- Melanin Biosynthesis in Cultured Melanoma Cells. Chem. Biodiv. 4: 2166-2171.
- Arung, E.T., Shimizu, K., Tanaka, H., and Kondo, R. 2010. 3-Prenyl luteolin, a new prenylated flavone with melanin biosynthesis inhibitory activity from *Artocarpus heterophyllus*. Fitoterapia 81: 640-643.
- Arung, E.T., Yoshikawa, K., Shimizu, K. and Kondo, R. 2010. Isoprenoid-substituted flavonoids from wood of *Artocarpus heterophyllus* on B16 melanoma cells: Cytotoxicity and structural criteria. Fitoterapia. (81): 120–123.
- Barik, B. R., Bhaumik, T., Dey, A. K., and Kundu, A. B. 1994. Triterpenoids from *Artocarpus heterophyllus*. Phytochemistry 35: 1001-1004.
- Barton, D. H. R. 1951. Cycloartenone, a triterpenoid ketone. J. Indian Chem. Soc. 74: 1444-1451.
- Blasco, E., Ngoc, L. D., Aucouturier, P., Preudhomme, J. L., and Barra, A. 1996. Mitogenic activity of new lectins from seeds of wild *Artocarpus* species from Vietnam. CR Acad. Sci. 319: 405-409.
- Boonlaksiri, C., *et al.* 2000. An antimalarial stilbene from *Artocarpus integer*. Phytochemistry 54: 415-417.
- Boonphong, S., Baramée, A., Kittakoop, P., and Puangsombat, P. 2007. Antitubercular and antiplasmodial prenylated flavones from the roots of *Artocarpus altilis*. Chiang Mai J. sci. 34: 339-344.
- Bourjot, M., *et al.* 2010. Antiplasmodial, antitrypanosomal, and cytotoxic activities of prenylated flavonoids isolated from the stem bark of *Artocarpus styracifolius*. Planta Med. 76: 1600-1604.
- Braca, A., Sortino, C., Politi, M., Morelli, I., and Mendez, J. 2002. Antioxidant activity of flavonoids from *Licania licaniaeflora*. J. Ethnopharmacol. 79: 379-381.
- Britton, G. 1983. The biochemistry of natural pigments. pp. 222-239. London : Cambridge University Press.
- Cao, S., Butler, M. S., and Buss, A. D. 2003. Flavonoids from *Artocarpus lanceifolius*. Nat. Prod. Res. 17: 79-81.
- Cerqueira, F., *et al.* 2008. The natural prenylated flavone artelastin is an inhibitor of ROS and NO production. Int. Immunopharmacol. 8: 597-602.
- Chakraborty, D. P., and Mandal, A. K. 1981. Aurantinamide acetate from *Artocarpus integrifolia* Linn. J. Indian. Chem. Soc. 58: 103.

- Chan, S. C., Ko, H. H., and Lin, C. N. 2003. New prenylflavonoids from *Artocarpus communis*. J. Nat. Prod. 66: 427-430.
- Chander, V., Singh, D. and Chopra, K. 2003. Catechin, a natural antioxidant protects against rhabdomyolysis-induced myoglobinuric acute renal failure. Pharmacological Research. (48): 503–509.
- Charoenlarp, P., Radomyos, P., and Harinasuta, T. 1981. Treatment of taeniasis with Puag-Haad: a crude extract of *Artocarpus lakoocha* wood. Southeast Asian J. Trop. Med. Pub. Hlth. 12: 568-570.
- Chatterjee, B. P., Sarkar, N., and Rao, A. S. 1982. Serological and chemical investigations of the anomeric configuration of the sugar units in the D-galacto-D-mannan of fenugreek (*Trigonella foenum-graecum*) seed. Carbohydr. Res. 104: 348-353.
- Chauhan, J. S., and Kumari, G. 1979. A new glycoflavonol from the root bark of *Artocarpus lakoocha*. Planta Med. 37: 86-88.
- Chauhan, J. S., Kumari, G., and Saraswat, M. 1979. A new flavonol glycoside from the root bark of *Artocarpus lakoocha*. Indian J. Chem. 18B: 473-475.
- Chauhan, J. S., Kumari, G., Kumar, S., and Chaturvedi, R. 1982. Chemical examination of the root bark of *Artocarpus lakoocha*. Proc. Natl. Acad. Sci. India, sect. A 52: 217-218. (CA. 99: 102301k).
- Chen, C.-C., Huang, Y. L., Ou, J. C., Lin, C. F., and Pan, T. M. 1993. Three new prenylflavones from *Artocarpus altilis*. J. Nat. Prod. 56: 1594-1597.
- Chung, M. I., Lu, C. M., Huang, P. L., and Lin, C. N. 1995. Prenylflavonoids of *Artocarpus heterophyllus*. Phytochemistry 40 : 1279-1281.
- Cidade, H. M., *et al.* 2001. Artelastocarpin and carpelastofuran, two new flavones and cytotoxicities of prenylflavonoids from *Artocarpus elasticus* against three cancer cell lines. Planta Med. 67: 867-870.
- Dang, D. T. N., *et al.* 2009. A novel anti-inflammatory compound, artonkin-4'-O-glucoside, from the leaves of *Artocarpus tonkinensis* suppress experimentally induced arthritis. J. Immunol. 69: 110-118.
- Daulatabad, C. D., and Mirajkar, A. M. 1989. Ricinoleic acid in *Artocarpus integrifolia* seed oil. J. Am. Oil Chem. Soc. 66: 1631.
- Dayal, R., and Seshadri, T. R. 1974. Colourless components of the roots of *Artocarpus heterophyllus*: Isolation of a new compound, artoflavone. Indian J. Chem. 12: 895-896.



- Durand, E., *et al.* 1962. Simple hypotensive and hypertensive principle from some West Indian medicinal plants. J. Pharm. Pharmacol. 14: 562-566.
- Ee, G. C. L., *et al.* 2010. Artosimmin-a potential anti-cancer lead compound from *Artocarpus odoratissimus*. Lett. Org. Chem. 7: 240-244.
- Ersam, T., *et al.* 2002. A new isoprenylated chalcone, artoindonesianin J, from the root and tree bark of *Artocarpus bracteata*. J. Chem. Res. Synop. 4: 186-187.
- Fang, Y. Z., Yang, S., and Wu, G. 2002. Free radicals, antioxidants, and nutrition. Nutrition 18: 872-879.
- Fang, S. C., Hsu, C. L., and Yen, G. C. 2008a. Anti-inflammatory effects of phenolic compounds isolated from the fruits of *Artocarpus heterophyllus*. J. Agric. Food Chem. 56: 4463-4468.
- Fang, S. C., Hsu, C. L., Yu, Y. S., and Yen, G. C. 2008b. Cytotoxic effects of new geranylchalcone derivatives isolated from the leaves of *Artocarpus communis* in SW 872 human liposarcoma cells. J. Agric. Food Chem. 56: 8859-8868.
- Ferreira De Miranda-Santos, I. K. Delgado, M., Bonini, P. V., Bunn, M. M., and Campos, N. A. 1992. A crude extract of *Artocarpus integrifolia* contains two lectins with distinct biological activities. Immunol. Lett. 31: 65-71.
- Fujimoto, T., Hano, Y., Nomura, T. and Uzawa, J. 1984. Components of Root Bark of *Cudrania tricuspidata* 2. Structures of Two new Isoprenylated Flavones, Cudraflavone A and B. Planta Med. 161-163.
- Fujimoto, Y., Agusutein, S., and Made, S. 1987a. Isolation of a chalcone derivative and antitumor compositions containing it. Jpn. Kokai. Tokkyo. Koho. JP 62270544. (CA. 110: 13561y).
- Fujimoto, Y., *et al.* 1987b. Isolation and structural elucidation of new lipoxygenase inhibitors from Indonesian *Artocarpus communis*. Tennen Yuki Kagobutsu Toronkai Koen Yoshishu 29: 721-728. (CA. 108: 164664u).
- Fujimoto, Y., Zhang, X. X., Kirisawa, M., Uzawa, J., and Sumatra, M. 1990. New flavones from *Artocarpus communis* Forst. Chem. Pharm. Bull. 38: 1787-1789.
- Fujimoto, Y., Jayasinghe, L., Hara, N., and Chintha, B. W. 2004. Geranyl chalcone derivatives with antifungal and radical scavenging properties from the leaves of *Artocarpus nobilis*. Phytochemistry 65: 1287-1290.
- Fukai, T., and Nomura, T. 1993. <sup>1</sup>H NMR spectra of prenylated flavonoids and pyranoflavonoids. Heterocycles 36: 329-343.

- Gardner, S., Sidisunthorn, P., and Anusarnsunthorn, V. 2000. A field guide to forest trees of Northern Thailand. pp 329. Bangkok : Kobfai Publishing Project.
- Gelder, C. W. G., Flurkey, W. H., and Wicher, H. J. 1997. Sequence and structural features of plant and fungal tyrosinases. Phytochemistry 45: 1309-1323.
- Hagiwara, K., Collet, C. D., Kobayashi, K., and Vaerman, J. P. 1988. Jacalin: isolation, characterization and influence of various factors on its interaction with human IGAI, as assessed by precipitation and latex agglutination. Mole. Immunol. 25: 69-83.
- Hakim, E. H., *et al.* 1999. Artoindonesianins A and B, two new prenylated flavones from the root of *Artocarpus champeden*. J. Nat. Prod. 62: 613-615.
- Hakim, E. H., *et al.* 2002a. Artoindonesianin P, a new prenylated flavones with cytotoxicity from *Artocarpus lanceifolius*. Fitoterapia 73: 668-673.
- Hakim, E. H., Ulinnuha, U. Z., Syah, Y. M., and Ghisalberti, E. L. 2002b. Artoindonesianins N and O, new prenylated stilbene and prenylated arylbenzofuran derivatives from *Artocarpus gomezianus*. Fitoterapia 73: 597-603.
- Hakim, E. H., *et al.* 2006. Prenylated flavonoids and related compounds of the Indonesian *Artocarpus* (Moraceae). J. Nat. Med. 60: 161-184.
- Han, A.-R., Kang, Y.-J., Windono, T., Lee, S.-K., and Seo, E.-K. 2006. Prenylated flavonoids from the heartwood of *Artocarpus communis* with inhibitory activity on lipopolysaccharide-induced nitric oxide production. J. Nat. Prod. 69: 719-721.
- Hano, Y., *et al.* 1989. Artonins A and B, two new prenylflavones from the root bark of *Artocarpus heterophyllus* Lamk. Heterocycles 29: 1447-1453.
- Hano, Y., Aida, M., and Nomura, T. 1990a. Two new natural Diels-Alder-type adducts from the root bark of *Artocarpus heterophyllus*. J. Nat. Prod. 53: 391-395.
- Hano, Y., Inami, R., and Nomura, T. 1990b. Components of the *Artocarpus rigida* Bl. Structures of two new isoprenylated flavones, artonins G and H. Heterocycles 31: 2137-2179.
- Hano, Y., Matsumoto, Y., Shinohara, K., Sun, J.-Y., and Nomura, T. 1990c. Cudraflavones C and D, two new prenylflavones from the root bark of *Cudrania tricuspidata* (Carr.) Bur. Heterocycles 31: 1339-1344.
- Hano, Y., Yamagami, Y., Kobayashi, M., Isohata, R., and Nomura, T. 1990d. Artonins E and F, two new prenylated flavones from the bark of *Artocarpus communis* Forst. Heterocycles 31: 877-882.

- Hano, Y., Inami, R., and Nomura, T. 1994. Constituents of Moraceae plants. 20. A novel flavone, artonin V, from the root bark of *Artocarpus altilis*. J. Chem. Res. Synop. 348-349. (CA. 251263z).
- Hashim, O. H., Gendeh, G. S., and Jaafar, M. I. N. 1992. Lectin extracts of champedak seeds demonstrate selective stimulation of T lymphocyte proliferation. Biochem. Int. 27: 139-143.
- Hashim, N., *et al.* 2010. Two xanthones from *Artocarpus obtusus*. J. Asian Nat. Prod. Res. 12: 106-112.
- Hooker, J.D. 1890. Flora of British India. pp. 539-544. Lreeve & Co. London.
- Hua, S. Z., Luo, J. G., Wang, X. B., Wang, J. S., and Kong, L. Y. 2009. Two novel monoterpene-chalcone conjugates isolated from the seeds of *Alpinia katsumadai*. Bioorg. Med. Chem. Lett. 19: 2728-2730.
- Iida, K., *et al.* 1995. Potent inhibitors of tyrosinase activity and melanin biosynthesis from *Rheum officinale*. Planta Med. 61: 425-428.
- Jagtap, U. B., and Bapat, V. A. 2010. *Artocarpus* : a review of its traditional uses, phytochemistry and pharmacology. J. Ethnopharmacol. 129: 142-166.
- Jamil, S., Sirat, H. M., Jantan, I., Aimi, N., and Kitajima, M. 2008. A new prenylated dihydrochalcone from the leaves of *Artocarpus lowii*. J. Nat. Med. 62: 321-324.
- Jayasinghe, L., Balasooriya, B. A. I. S., Padmini, W. C., Hara, N., and Fujimoto, Y. 2004a. Geranylchalcone derivatives with antifungal and free radical properties from the leaves of *Artocarpus nobilis*. Phytochemistry 65: 1287-1290.
- Jayasinghe, U. L. B., Puvanendran, S., Hara, N., and Fujimoto, Y. 2004b. Stilbene derivatives with antifungal and radical scavenging properties from the stem bark of *Artocarpus nobilis*. Nat. Prod. Res. 18: 571-574.
- Jayasinghe, L., Rupasinghe, G. K., Hara, N., and Fujimoto, Y. 2006. Geranylated phenolic constituents from the fruits of *Artocarpus nobilis*. Phytochemistry 67: 1353-1358.
- Jayasinghe, U. L. B., Samarakoon, T. B., Kumarihamy, B. M. M., Hara, N., and Fujimoto, Y. 2008. Four new prenylated flavonoids and xanthones from the root bark of *Artocarpus nobilis*. Fitoterapia 79: 37-41.
- Kapche, G. D. W. F., *et al.* 2009. Prenylated arylbenzofuran derivatives from *Morus mesozygia* with antioxidant activity. Phytochemistry 70: 216-221.
- Kapil, R. S., and Joshi, S. S. 1960. Chemical constituents of *Artocarpus lakoocha*. J. Sci. Ind. Res. (Indian) 19B: 498.

- Kielland, I. S., and Malterud, K. E. 1994. Triterpenoids from *Artocarpus integrifolia* fruits. Planta Med. 60: 196.
- Kijjoa, A., *et al.* 1996. Prenylflavonoids from *Artocarpus elasticus*. Phytochemistry 43: 691-694.
- Kijjoa, A., *et al.* 1998. Further prenylflavonoids from *Artocarpus elasticus*. Phytochemistry 47: 875-878.
- Kingroungpet, B. 1994. Phytochemical study of *Artocarpus gomezianus* Wall. Ex Tréc. leaves. Master's thesis, Department of Pharmacognosy, Graduate School, Chulalongkorn University.
- Ko, H. H., Yu, S. M., Ko, F. N., Teng, C. M., and Lin, C. N. 1997. Bioactive constituents of *Morus australis* and *Broussonetia papyrifera*. J. Nat. Prod. 60: 1008-1011.
- Ko, F. N., Cheng, Z. J., Lin, C. N., and Teng, C. M. 1998. Scavenger and antioxidant properties of prenylflavones isolated from *Artocarpus heterophyllus*. Free radic. Biol. Med. 25: 160-168.
- Ko, H.-H., Lin, C.-N., and Yang, S.-Z. 2000. New constituents of *Artocarpus rigida*. Helv. Chim. Acta 83: 3000-3005.
- Ko, H.-H., Lin, C.-N., and Yang, S.-Z. 2001. Artocarpol F, a phenolic compound with a novel skeleton, isolated from *Artocarpus rigida*. Tetrahedron Lett. 42: 5269-5270.
- Ko, H.-H., Lu, Y.-H., Yang, S.-Z., Won, S.-J., and Lin, C.-N. 2005. Cytotoxic prenyl flavonoids from *Artocarpus elasticus*. J. Nat. Prod. 68: 1692-1695.
- Kochummen, K. M. 1978. Tree flora of Malaya vol.3. pp. 119-133. Kuala Lumpur: Malaysia: Forest Department, Ministry of Primary Industries.
- Kong, L., Zhou, J., Wen, Y, Li, J. and Cheng, C.H. 2002 . Aesculin possesses potent hypouricemic action in rodents but is devoid of xanthine oxidase/dehydrogenase inhibitory activity. Planta Med. 68(2):175-8.
- Kubo, I., Yokokawa, Y., and Kinst-Hori, I. 1995. Tyrosinase inhibitors from Bolivian medicinal plants. J. Nat. Prod. 58: 734-739.
- Kumar, N. S., Pavanassasivam, G., Sultanbawa, M. U. S., and Mageswaran, R. 1977. Chemical investigation of Ceylonese plants. Part 24. New chromenoflavonoids from the bark of *Artocarpus nobilis* Thw. (Moraceae). J. Chem. Soc. Perkin Trans I : 1243-1251.
- Lee, C.L. and Keng, C.L. 2005. The micropropagation of *Artocarpus heterophyllus* Lamk. Planter, Kuala Lumpur 8(956): 689-695.

- Lien, H., Ripperger, A., Porzel, T. V., and Adam, G. 1998. Constituents of *Artocarpus tonkinensis*. Pharmazie 53: 353.
- Likhitwitayawuid, K., Sritularak, B., and De-Eknamkul, W. 2000. Tyrosinase inhibitors from *Artocarpus gomezianus*. Planta Med. 66: 275-277.
- Likhitwitayawuid, K., and Sritularak, B. 2001. A new dimeric stilbene with tyrosinase inhibitory activity from *Artocarpus gomezianus*. J. Nat. Prod. 64: 1457-1459.
- Likhitwitayawuid, K., *et al.* 2005. Phenolics with antiviral activity from *Millettia erythrocalyx* and *Artocarpus lakoocha*. Nat. Prod. Res. 19: 177-182.
- Likhitwitayawuid, K., Chaiwiriyaya, S., Sritularak, B. and Lipipun, V. 2006. Antiherpetic flavones from the heartwood of *Artocarpus gomezianus*. Chem. Biodivers. 3: 1138-1143.
- Lin, C. N., Lu, C. M., and Huang, P. L. 1995. Flavonoids from *Artocarpus heterophyllus*. Phytochemistry 39: 1447-1451.
- Lin, C. N., and Shieh, W. L. 1991. Prenylflavonoids and pyranodihydrobenzoxanthone from *Artocarpus communis*. Phytochemistry 30: 1669-1671.
- Lin, C. N., and Shieh, W. L. 1992. Pyranoflavonoids from *Artocarpus communis*. Phytochemistry 31: 2922-2924.
- Lin, C. N., Shieh, W. L., and Jong, T.T. 1992. Pyranoflavonoids from *Artocarpus communis*. Phytochemistry 31: 2563-2564.
- Lin, C. N., Ko, H. H., Weng, J. R., Chan, S. C., and Lu, Y. H. 2006. Antiplatelet prenylflavonoids from *Artocarpus communis*. Phytochemistry 67: 824-829.
- Lin, K.-W., Liu, C.-H., Tu, H.-Y., Ko, H.-H., and Wei, B.-L. 2009. Antioxidant prenylflavonoids from *Artocarpus communis* and *Artocarpus elasticus*. Food Chem. 115: 558-562.
- Lu, C. M., and Lin, C. N. 1993. Two 2',4',6'-trioxygenated flavones from *Artocarpus heterophyllus*. Phytochemistry 33: 909-911.
- Lu, C. M., and Lin, C. N. 1994. Flavonoids and 9-hydroxytridecyl docosanoate from *Artocarpus heterophyllus*. Phytochemistry 35: 781-783.
- Lu, Y.-H., *et al.* 2002. Two novel and anti-inflammatory constituents of *Artocarpus rigida*. Helv. Chim. Acta 85: 1626-1632.
- Ma, J.-P., *et al.* 2010. New isoprenylated flavonoids and cytotoxic constituents from *Artocarpus tonkinensis*. J. Asian Nat. Prod. Res. 12: 586-592.
- Mahato, S. B., Banerjee, S. K., and Chakravarti, R. N. 1971. Triterpenoids of the stem bark of *Artocarpus chaplasha*. Phytochemistry 10: 1351-1354.

- Markham, K. R. 1982. Techniques of flavonoid identification. London : Academic Press.
- Makmur, L., *et al.* 2000. Artoindonesianin C, a new xanthone derivative from *Artocarpus teysmanii*. J. Nat. Prod. 63: 243-244.
- Masamoto, Y., Iida, S., and Kubo, M. 1980. Inhibitory effect of Chinese crude drugs on tyrosinase. Planta Med. 40: 361-365.
- Matsuda, N. and Kikuchi, M. 1995. A coumarin glycoside from *Lonicera gracilipes* var. *glandulosa*. Phytochemistry. 38(3). 803-804.
- Morita, H., *et al.* 1994. Pseudostellalins A-C, new tyrosinase inhibitory cyclic peptides from *Pseudostellaria heterophylla*. Tetrahedron 50: 6797-6804.
- Mu, Q. Z., and Li, Q. X. 1982. The isolation and identification of morin-calcium chelate compound from *Artocarpus pithecoallus* C. Y. Wu and *Artocarpus heterophyllus* Lamk. Chih Wu Hsueh Pao 24: 147-153. (CA. 97: 107043w).
- Musthapa, I., *et al.* 2009. An oxepinoflavone from *Artocarpus lanceifolius* with cytotoxic activity against P-388 cells. Arch. Pharm. Res. 32: 191-194.
- Namjuntra, P., and Culavatnatol, M. 1984. Purification and characterization of a lectin from seeds of jackfruit (*Artocarpus heterophyllus*). 10<sup>th</sup> Conference of Science and Technology, Thailand. Chiang Mai university, Chiang Mai : 356-357.
- Nath, M. C., and Mukherjee, M. C. 1939. Crystallographic investigation of artostenone, the stenone isolated from the Indian summer fruit, *Artocarpus integrifolia* by means of Goniometer and X-rays. J. Indian Chem. Soc. 16: 229-234.
- Nogueira, P. L., and Correia, A. A. 1958. Chemical study of a tropical wood. Garcia Orta 6: 679-688. (CA. 53: 20792a).
- Nomura, T., Hano, Y., and Aida, M. 1998. Isoprenoid-substituted flavonoids from *Artocarpus* plants (Moraceae). Heterocycles 47: 1179-1205.
- Pant, T., and Chaturvedi, K. 1989. 4-Hydroxyundecyl docosanoate and cycloartenone in *Artocarpus integra* latex. Phytochemistry 28: 2197-2199.
- Parenti, P., *et al.* 1998. A new prenylated flavone from *Artocarpus champeden* inhibits the K<sup>+</sup>-dependent amino acid transport in *Bombyx mori* midgut. Biochem. Biophys. Res. Commun. 244: 445-448. (CA.128: 306539w).
- Pathasarathy, P. C., Radhakrishnan, P. V., Rathi, S. S., and Venkataraman, K. 1969. Coloring matters of the wood of *Artocarpus heterophyllus*. V. Cycloartocarpesin and oxydihydroartocarpesin, two new flavones. Indian J. Chem. 7: 101-102.

- Patil, A. D., *et al.* 2002. A new dimeric dihydrochalcone and a new prenylated flavone from the bud covers of *Artocarpus altilis*: potent inhibitors of cathepsin K. J. Nat. Prod. 65: 624-627.
- Pavanasasivum, G., and Sultanbawa, M. U. S. 1973. Cycloartenyl acetate, cycloartenol and cycloartenone in the bark of *Artocarpus* species. Phytochemistry 12: 2725-2726.
- Pavanasasivam, G., Sultanbawa, M. U. S., and Mageswaran, R. 1974. Chemical investigation of Ceylonese plants. X. New chromenoflavonoids from the bark of *Artocarpus nobilis* (Moraceae). Chem. Ind. (London) : 875-876.
- Pavaro, J., and Reutrakul, V. 1976. New flavones in *Artocarpus lakoocha*. Mahidol Univ. J. Pharm. Sci. 3: 161-164.
- Pendse, A. D., Pendse, R., Rao, A. V. R., and Venkataraman, K. 1976. Integrin, cyclointegrin and oxyisocyclointegrin, three new flavones from the heartwood of *Artocarpus integer*. Indian J. Chem. 14B: 69-72.
- Pereira, J. R., Medina, H., and Bustos, R. E. 1962. On the presence of acetylcholine in seeds of *Artocarpus integrifolia* (Moraceae) and of *Annona squamosa* (Annonaceae). An. Fac. Med. Univ. Parana 5: 45-47. (Biological Abstracts 46: 12891).
- Promdee, L., 1996. Purification and applications of *Artocarpus lakoocha* agglutinin. Master's thesis, Department of Medical Biochemistry, Graduate School, Khonkaen University.
- Puntumchai, A., *et al.* 2004. Lakoochins A and B, new antimycobacterial stilbene derivatives from *Artocarpus lakoocha*. J. Nat. Prod. 67: 485-486.
- Radhakrishnan, P. V., Rao, A. V. R., and Venkataraman, K. 1965. Two new flavones from *Artocarpus heterophyllus*. Tetrahedron Lett. 663-667.
- Radwan, M. M., Guzman, R. R., Manly, S. P., Jacob, M., and Ross, S. A. 2009. Sepicanin A-a new geranyl flavanone from *Artocarpus sepicanus* with activity against methicillin-resistant *Staphylococcus aureus* (MRSA). Phytochemistry Lett. 2: 141-143.
- Rao, A. V. R., Varadan, M., and Venkataraman, K. 1971. Coloring matters of the wood of *Artocarpus heterophyllus*: part VI. Cycloheterophyllin, a flavone linked to three isoprenoid groups. Indian J. Chem. 9: 7-13.

- Rao, A. V. R., Rathi, S. S., and Venkataraman, K. 1972. Chaplashin, a flavone containing an oxepine ring from the heartwood of *Artocarpus chaplasha* Roxb. Indian J. Chem. 10: 905-907.
- Rao, A. V. R., Varadan, M., and Venkataraman, K. 1973. Coloring matters of the wood of *Artocarpus heterophyllus*: part VII. Isocycloheterophyllin, a new flavone. Indian J. Chem. 11: 298-299.
- Rouse-Miller, J. and Duncan, J.E. 2000. In vitro propagation of *Artocarpus altilis* (Park.) Fosberg (Breadfruit) From mature plant material. In Vitro Cellular & Developmental Biology – Plant. 36: 115–117
- Royal Forest Department, The Forest Herbarium. 2001. Thai plant names Tem Smitinand revised edition. pp. 56-57. Bangkok : Prachachon.
- Sato, M., *et al.* 1996. Flavones with antibacterial activity against cariogenic bacteria. J. Ethnopharmacol. 54: 171-176.
- Seo, E.-K., *et al.* 2003. Bioactive prenylated flavonoids from the stem bark of *Artocarpus kemando*. Arch. Pharm. Res. 26: 124-127.
- Shamaun, S. S., *et al.* 2010. Prenylated flavones from *Artocarpus altilis*. J. Nat. Med. 64: 478-481.
- Shieh, W. L., and Lin, C. N. 1992. A quinonoid pyranobenzoxanthone and pyranodihydrobenzoxanthone from *Artocarpus communis*. Phytochemistry 31: 364-367.
- Shen, H., and Hou, A.-J. 2008. Prenylated 2-arylbenzofurans from *Artocarpus petelotii*. Nat. Prod. Res. 22: 1451-1456.
- Shimizu, K., Kondo, R., and Sakai, K. 1997. A stilbene derivative from *Artocarpus incisus*. Phytochemistry 45: 1297-1298.
- Shimizu, K., Kondo, R., Sakai, K., Lee, S. H., and Sato, H. 1998. The inhibitory components from *Artocarpus incisus* on melanin biosynthesis. Planta Med. 64: 408-412.
- Shimizu, K., Kondo, R., Sakai, K., and Buabarn, S. 2000. A geranylated chalcone with 5 $\alpha$ -reductase inhibitory properties from *Artocarpus incisus*. Phytochemistry 54: 737-739.
- Shinomiya, K., Aida, M., Hano, Y., and Nomura, T. 1995. A Diels-Alder-type adduct from *Artocarpus heterophyllus*. Phytochemistry 40: 1317-1319.
- Shu-Hua, Q., Da-Gang, W., Yun-Bao, M. and Xiao-Dong, L. 2003. A Novel Flavane from *Carapa guianensis*. Acta Bot. Sin. 45(9): 1129-1133



- Smitinand, T. 2001. Thai plant names (botanical names-vernacular names) revised edition. pp. 56-57 and 359-360. Bangkok: The Forest Herbarium, Royal Forest Department.
- Soekamto, N. H., Achmad, S. A., Ghisalberti, E. L., Hakim, E. H., and Syah, Y. M. 2003. Artoindonesianins X and Y, two isoprenylated 2-arylbenzofurans, from *Artocarpus fretessi* (Moraceae). Phytochemistry 64: 831-834.
- Sritularak, B. 1998. Chemical constituents of *Artocarpus lakoocha* and *A. gomezianus*. Master's Thesis, Department of Pharmacognosy, Graduate School, Chulalongkorn University.
- Sritularak, B., Tantrakarnsakul, K., Likhitwitayawuid, K., and Lipipun, V. 2010. New 2-arylbenzofuran from the root bark of *Artocarpus lakoocha*. Molecules. 15: 6548-6558.
- Su, B.-N., *et al.* 2002. Constituents of the bark and twigs of *Artocarpus dadah* with cyclooxygenase inhibitory activity. J. Nat Prod. 65: 163-169.
- Suhartati, T., *et al.* 2001. Artoindonesianin L, a new prenylated flavone with cytotoxic activity from *Artocarpus rotundo*. Fitoterapia 72: 912-918.
- Suhartati, T., and Yandri, H. S. 2008. The bioactivity test of artonin E from the bark of *Artocarpus rigida* Blume. Eur. J. Sci. Res. 23: 330-337.
- Sultanbawa, M. U. S., and Surendrakumar, S. 1989. Two pyranodihydrobenzoxanthones from *Artocarpus nobilis*. Phytochemistry 28: 599-605.
- Suresh, G. K., Appukuttan, P. S. and Basu, D. K. 1982.  $\alpha$ -D-Galactose specific lectin from jackfruit (*Artocarpus integra*) seed. J. Biol. Sci. 4: 257-261.
- Syah, Y. M., *et al.* 2001. Artoindonesianins G-I, three new prenylated flavones from *Artocarpus lanceifolius*. Fitoterapia 72: 765-773.
- Syah, Y. M., *et al.* 2002a. Artoindonesianin M, a new prenylated flavone from *Artocarpus champeden*. Bull. Soc. Nat. Prod. Chem. (Indonesia) 2: 31-36.
- Syah, Y. M., *et al.* 2002b. Artoindonesianins Q-T, four new prenylated flavones from *Artocarpus champeden* Spreng. (Moraceae). Phytochemistry 61: 949-953.
- Syah, Y. M., Achmad, S. A., Ghisalberti, E. L., Hakim, E. H., and Mujahidin, D. 2004. Two new cytotoxic isoprenylated flavones, Artoindonesianins U and V, from the heartwood of *Artocarpus champeden*. Fitoterapia 75: 134-140.
- Syah, Y. M., *et al.* 2006a. Two prenylated flavones from the root bark of *Artocarpus lanceifolius*. Z. Naturforsch. 61: 1134-1137.

- Syah, Y. M., Juliawaty, L. D., Achmad, S. A., Hakim, E. H., and Ghisalberti, E. L. 2006b. Cytotoxic prenylated flavones from *Artocarpus champeden*. J. Nat. Med. 60: 308-312.
- Takao, T., Kitatani, F., Watanabe, N., Yaki, A., and Sakata, K. 1994. A simple screening method for antioxidants and isolation of several antioxidants produced by marine bacteria from fish and shellfish. Biosci. Biotech. Biochem. 58: 1780-1783.
- Tantrakarnsakul, K. 2010. Bioactive compounds from the root bark of *Artocarpus lakoocha*. Master's Thesis, Department of Pharmacognosy, Graduate School, Chulalongkorn University.
- Taylor, J. K. 1987. Quality Assurance of Chemical Measurements. Principles of Measurement. pp. 79–81. Chelsea: Lewis Publishers.
- Tengamnuay, P., Pengrungruangwong, K., Pheansri, I., and Likhitwitayawuid, K. 2006. *Artocarpus lakoocha* heartwood extract as a novel cosmetic ingredient: evaluation of the *in vitro* anti-tyrosinase and *in vivo* skin whitening activities. Int. J. Cosm. Sci. 28: 269-276.
- Thuy, T. T., *et al.* 2004. Immonosuppressive auronol glycosides from *Artocarpus tonkinensis*. Pharmazie 59: 297-300.
- Ultee, A. J. 1949. Components of samples of latex (from a number of tropical trees). Pharm. Weekbl. 84: 65-70. (CA. 43: 4041i).
- Valko, M., *et al.* 2007. Free radicals and antioxidants in normal physiological functions and human disease. Int. J. Biochem. Cell Biol. 36: 44-84.
- Vaidya, V., Raote, A., Jangid, A. and Tale, R. 2010. Estimation of two bioactive compounds from *Artocarpus lakoocha* Roxb. Leaves using HPTLC. Int. J. Pharm. Bio Sci. 2: 860-866
- Venkataraman, K. 1972. Wood phenolics in the chemotaxonomy of the Moraceae. Phytochemistry 11: 1571-1586.
- Wahyuni, T. S., *et al.* 2009. Artopenen A, a new antiplasmodial isoprenylated flavone from *Artocarpus champeden*. Heterocycles 79: 1121-1126.
- Wang, Y.-H., *et al.* 2004. New isoprenylated flavones, artochamins A-E and cytotoxic principles from *Artocarpus chama*. J. Nat. Prod. 67: 757-761.
- Wang, Y.-H., *et al.* 2006. Prenylated stilbenes and their novel biogenetic derivatives from *Artocarpus chama*. Eur. J. Org. Chem. 15: 3457-3463.

- Wang, Y.-H., Hou, A.-J., and Chen, D.-F. 2007a. Two new isoprenylated stilbenes from *Artocarpus chama*. J. Integ. Pl. Biol. 49: 605-608.
- Wang, Y., Xu, K., Lin, L., Pan, Y., and Zheng, X. 2007b. Geranyl flavonoids from the leaves of *Artocarpus altilis*. Phytochemistry 68 : 1300-1306.
- Weng, J. R., *et al.* 2006. Antiplatelet prenylflavonoids from *Artocarpus communis*. Phytochemistry 67: 824-829.
- Widyawaruyanti, A., *et al.* 2007. New prenylated flavones from *Artocarpus champeden* and their antimalarial activity *in vitro*. J. Nat. Med. 61: 410-413.
- Wongkham, S., *et al.* 1995. Isolectins from seeds of *Artocarpus lakoocha*. Phytochemistry 40: 1331-1334.
- Wu, Z. Y., Raven, P. H., and Hong, D.Y. 2003. Flora of China vol. 5 (Ulmaceae through Basellaceae). : 30-34. Beijing: Science Press.
- Yamazaki, M., Okuyama, E., Matsudo, T., Takamaru, T., and Kaneko, T. 1987. Principle of Indonesian herbal drugs having antiulcerogenic activity I, isolation and identification of ( ± )-catechin from *Artocarpus integra* Merr. Yakugaku Zasshi 107: 914-916. (CA. 108: 68693g).
- Yenesew, A., Midiwo, J. O., Guchu, S. M., Heydenreich, M., and Peter, M. G. 2002. Three isoflavan-3-ones and a 2-arylbenzofuran from the root bark of *Erythrina burttii*. Phytochemistry 59: 337-341.
- Zhang, C. L., Zhou, S. F., Wu, Y. S., Deng, Y. Z., and Zhou, D. Y. 1999. Purification and characterization of lectin from the seeds of *Artocarpus lignanensis*. Zhongguo Shengwu Huaxue Yu Fenzi Shengwu Xuebao 15: 142-144. (CA.131: 2788k)
- Zhao, T., Yan, G.-R., Pan, S.-L., Wang, H.-Y., and Hou, A.-J. 2009. New isoprenylated 2-arylbenzofurans and pancreatic lipase inhibitory constituents from *Artocarpus nitidus*. Chem. Biodiv. 6: 2209-2216.
- Zheng, Z.-P., Cheng, K.-W., To, J. T.-K., Li, H., and Wang, M. 2008. Isolation of tyrosinase inhibitors from *Artocarpus heterophyllus* and use of its extract as antibrowning agent. Mol. Nutr. Food Res. 52: 1530-1538.
- Zheng, Z.-P., *et al.* 2009. Chemical components and tyrosinase inhibitors from the twigs of *Artocarpus heterophyllus*. J. Agric. Food Chem. 57: 6649-6655.
- Zhong, J.-J. and Yue, C.-J. 2005. Plant Cells: Secondary Metabolite Heterogeneity and Its Manipulation. Adv Biochem Engin/Biotechnol. 100: 53–88

## **APPENDIX**

## High resolution report

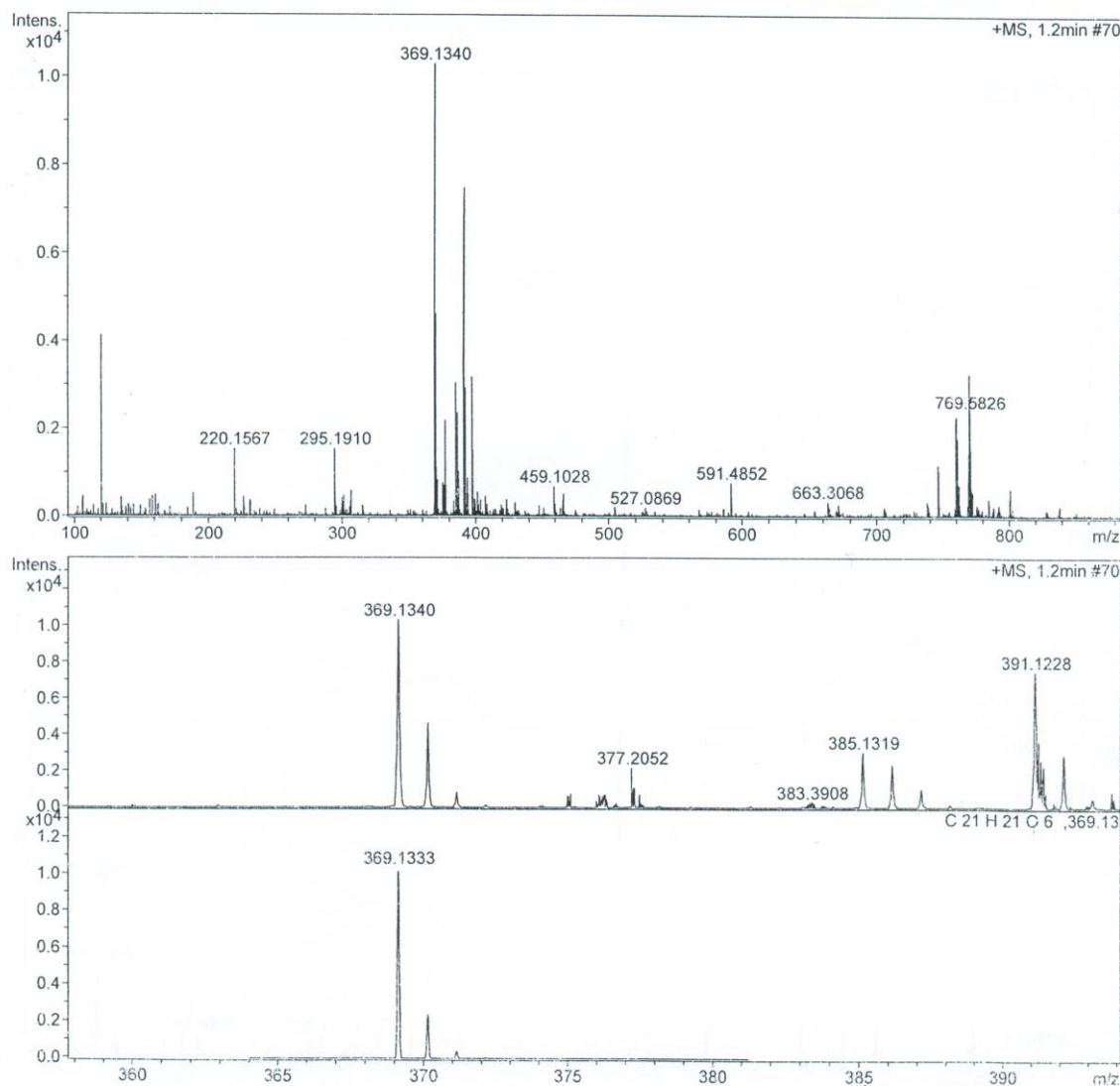
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Sample Name 75MF2\_5\_II

Acquisition Date 5/3/2011 2:11:24 PM

Operator Sutichai Ext: 3560  
Instrument micrOTOF Bruker  
Calibrate by Sodium Formate

## Acquisition Parameter

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Focus	Not active			Set Dry Heater	150 °C
Scan Begin	100 m/z	Set Capillary	5000 V	Set Dry Gas	2.0 l/min
Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source



**Figure 29** HRESI Mass Spectrum of compound CAL1

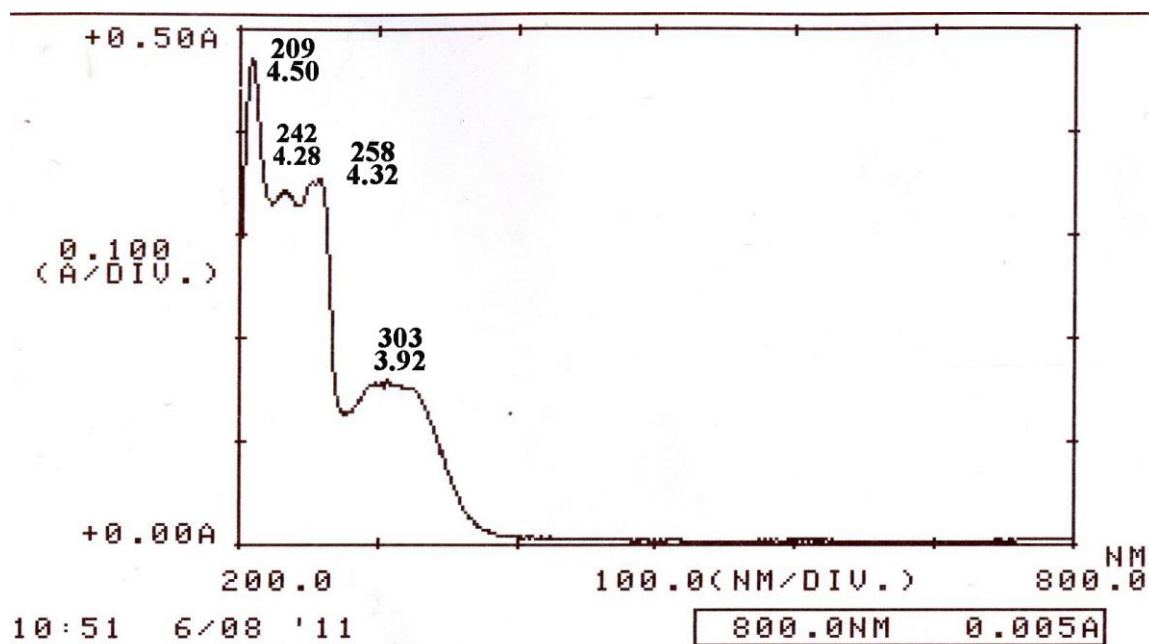


Figure 30 UV Spectrum of compound CAL1 (methanol)

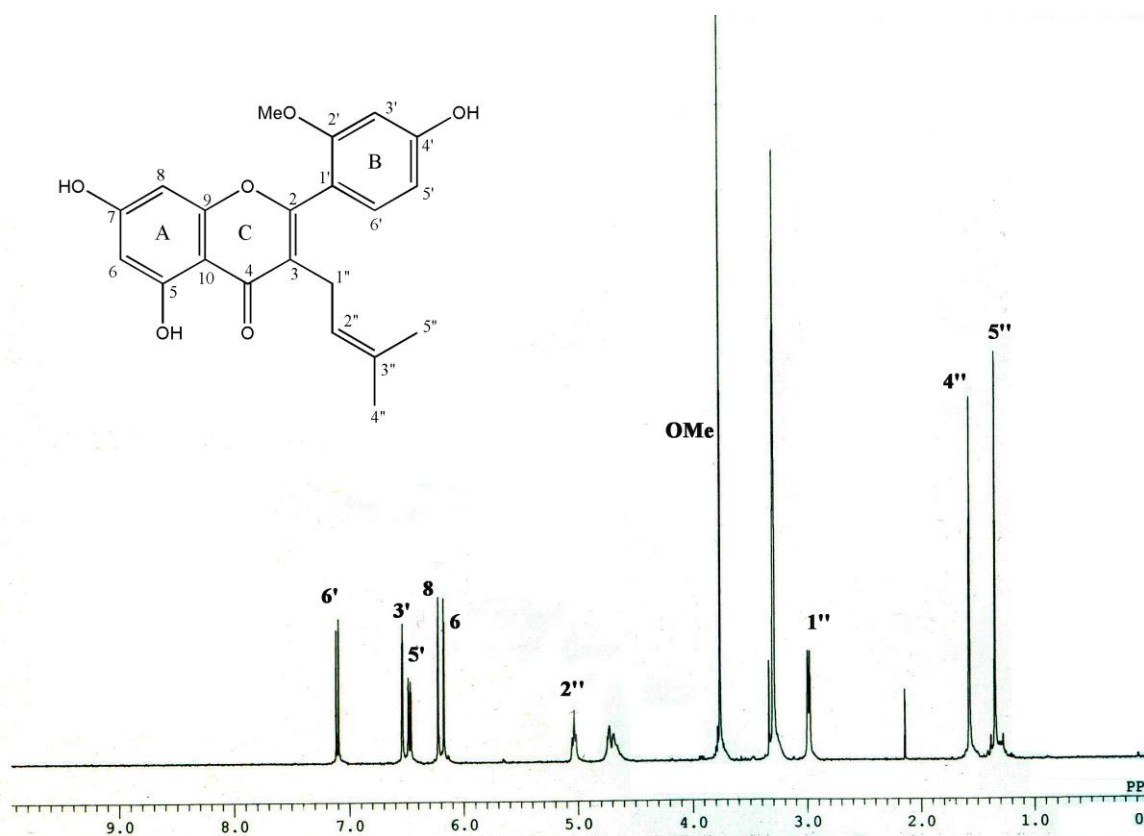
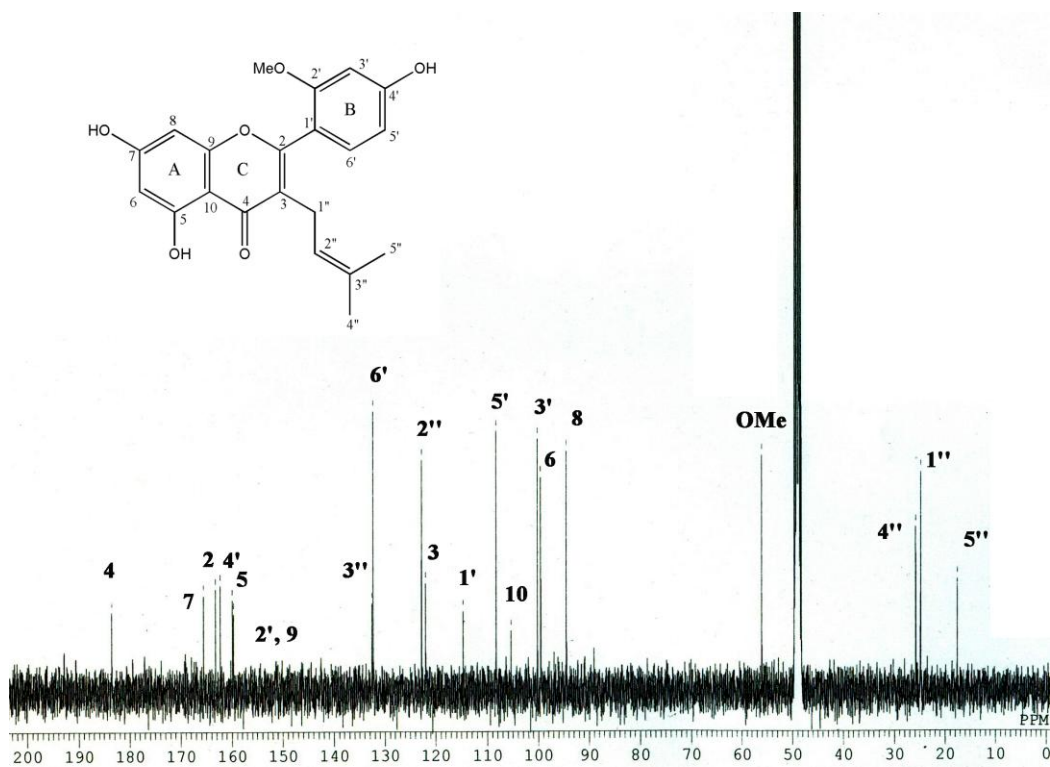
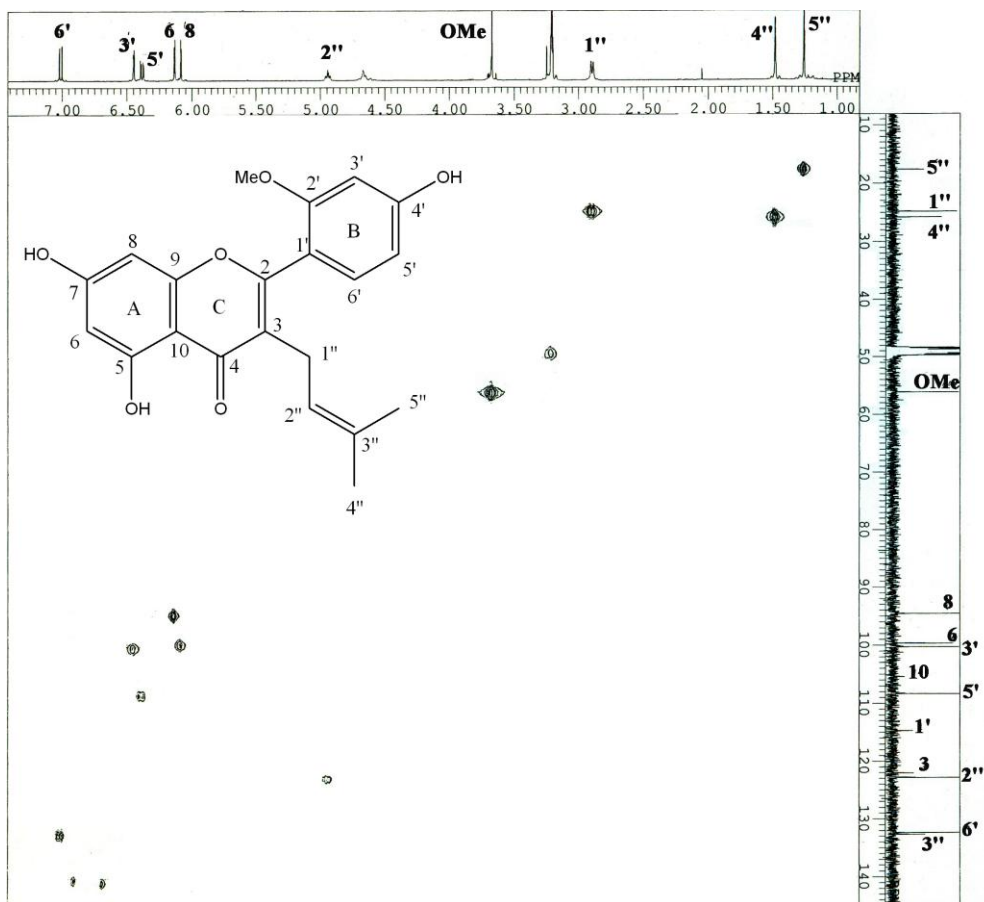


Figure 31 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL1 (methanol-*d*<sub>4</sub>)



**Figure 32**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL1 (methanol- $d_4$ )



**Figure 33** HMBC Spectrum of compound CAL1 (methanol- $d_4$ )



Figure 34 HMBC Spectrum of compound CAL1 (methanol- $d_4$ )

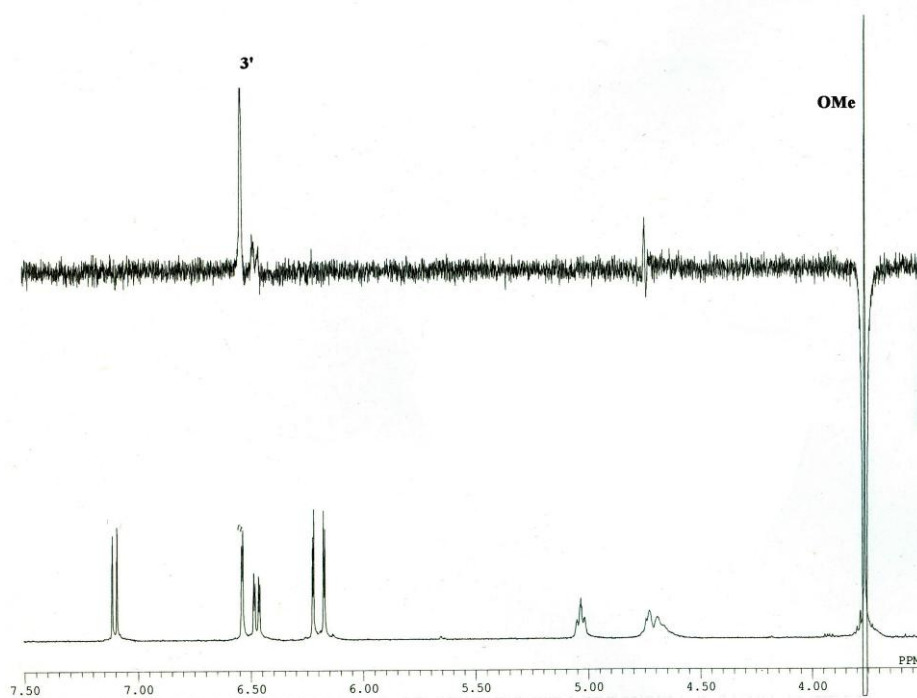


Figure 35 NOE Spectrum of compound CAL1 (methanol- $d_4$ )



## High resolution report

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Method NaFormate\_pos\_infusion .m  
Sample Name 100MF5\_2\_1

Acquisition Date 5/3/2011 2:24:11 PM

Operator Sutichai Ext: 3560  
Instrument micrOTOF Bruker  
Calibrate by Sodium Formate

## Acquisition Parameter

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Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

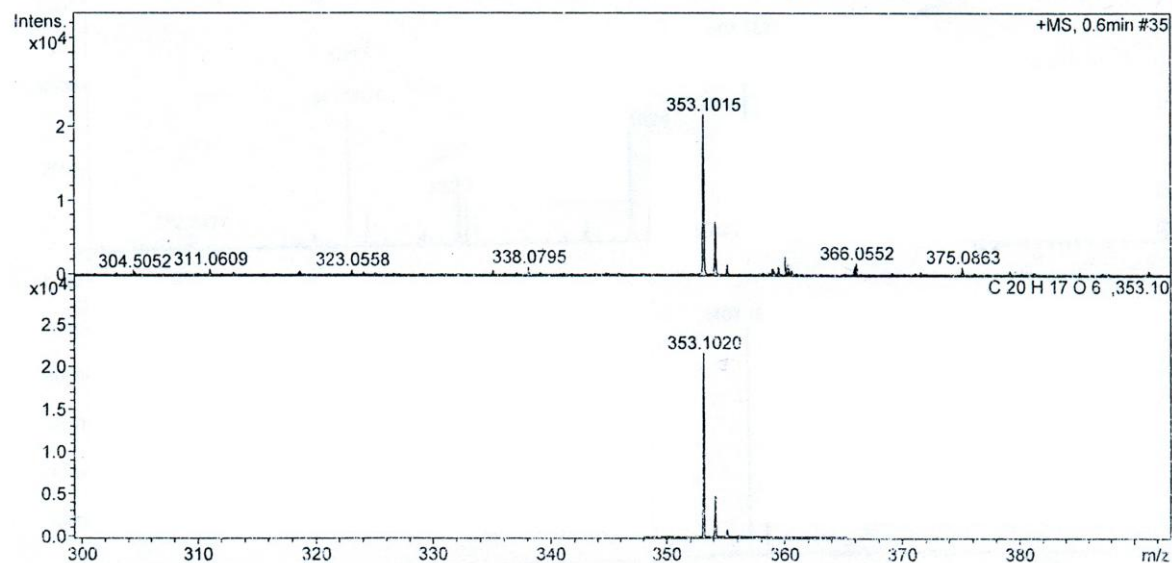
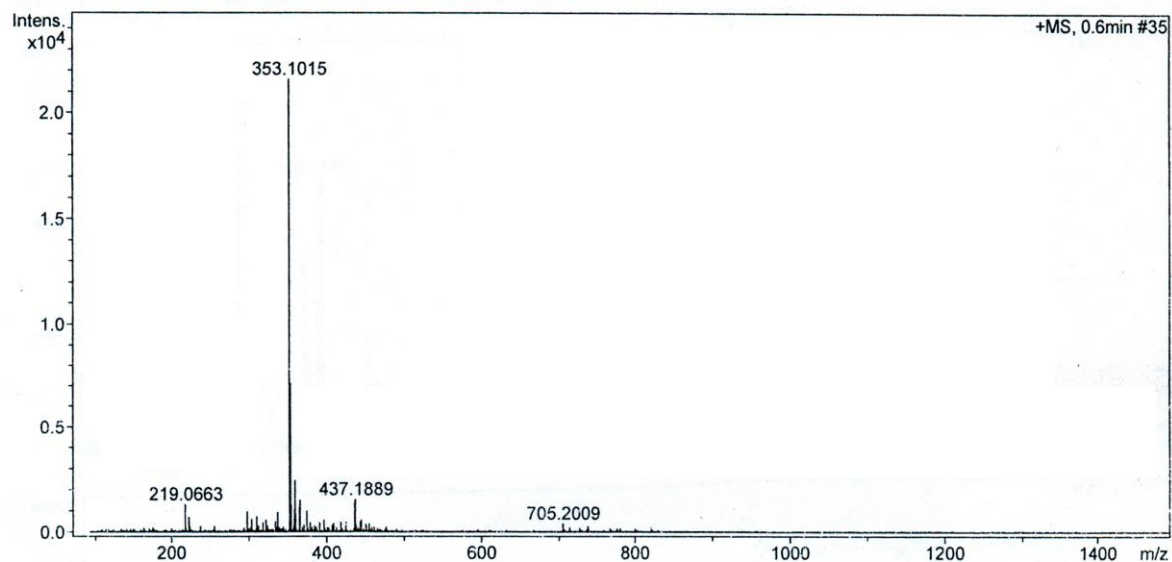


Figure 36 HRESI Mass Spectrum of compound CAL2

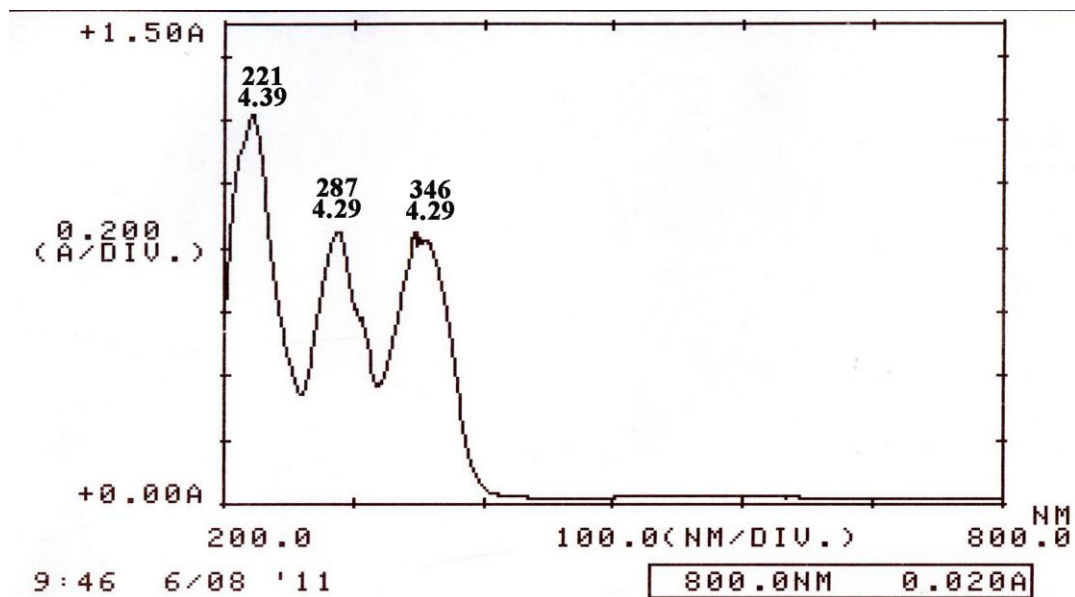


Figure 37 UV Spectrum of compound CAL2 (methanol)

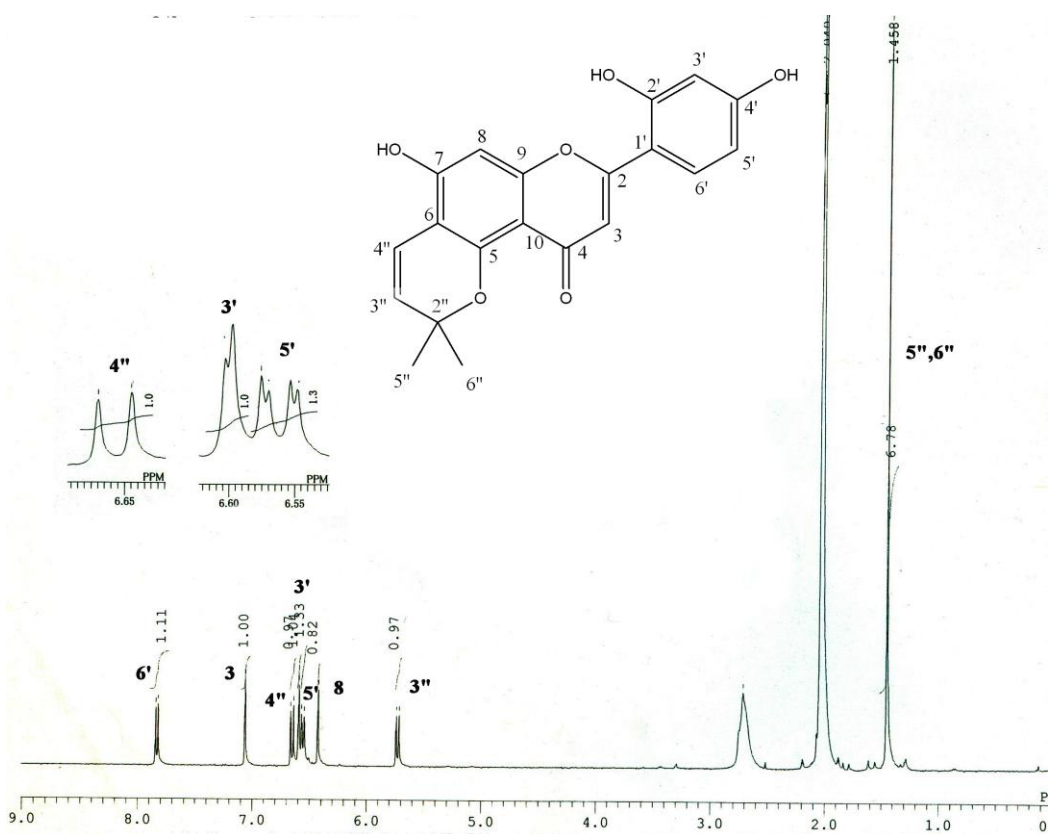
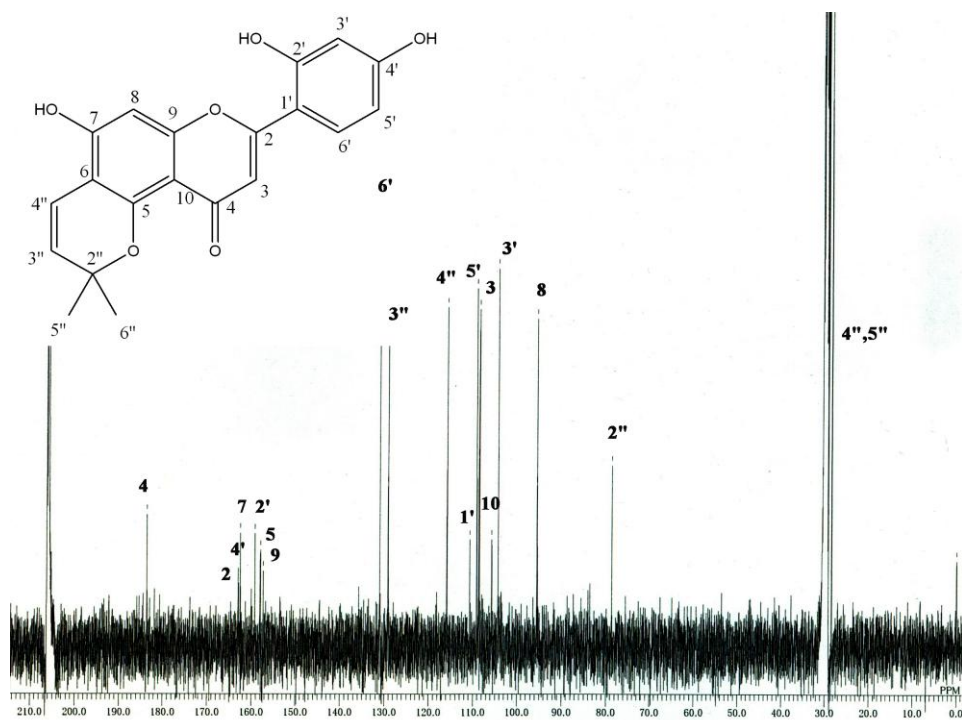
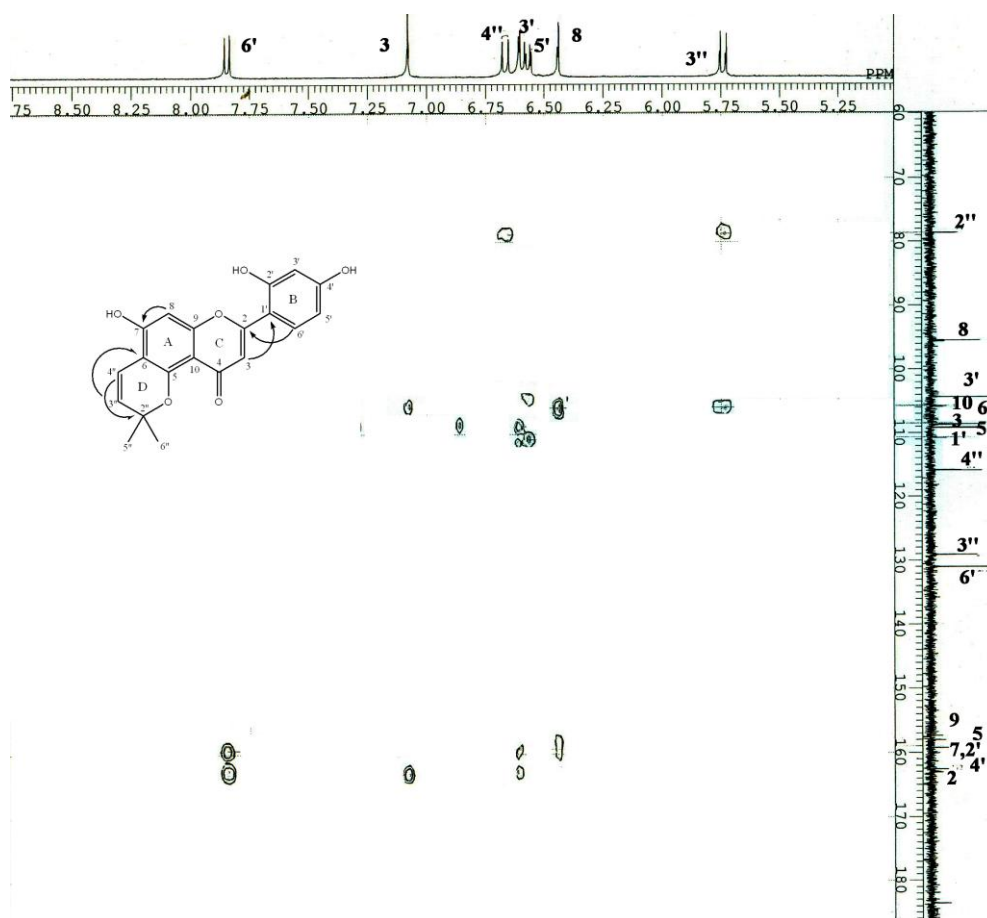


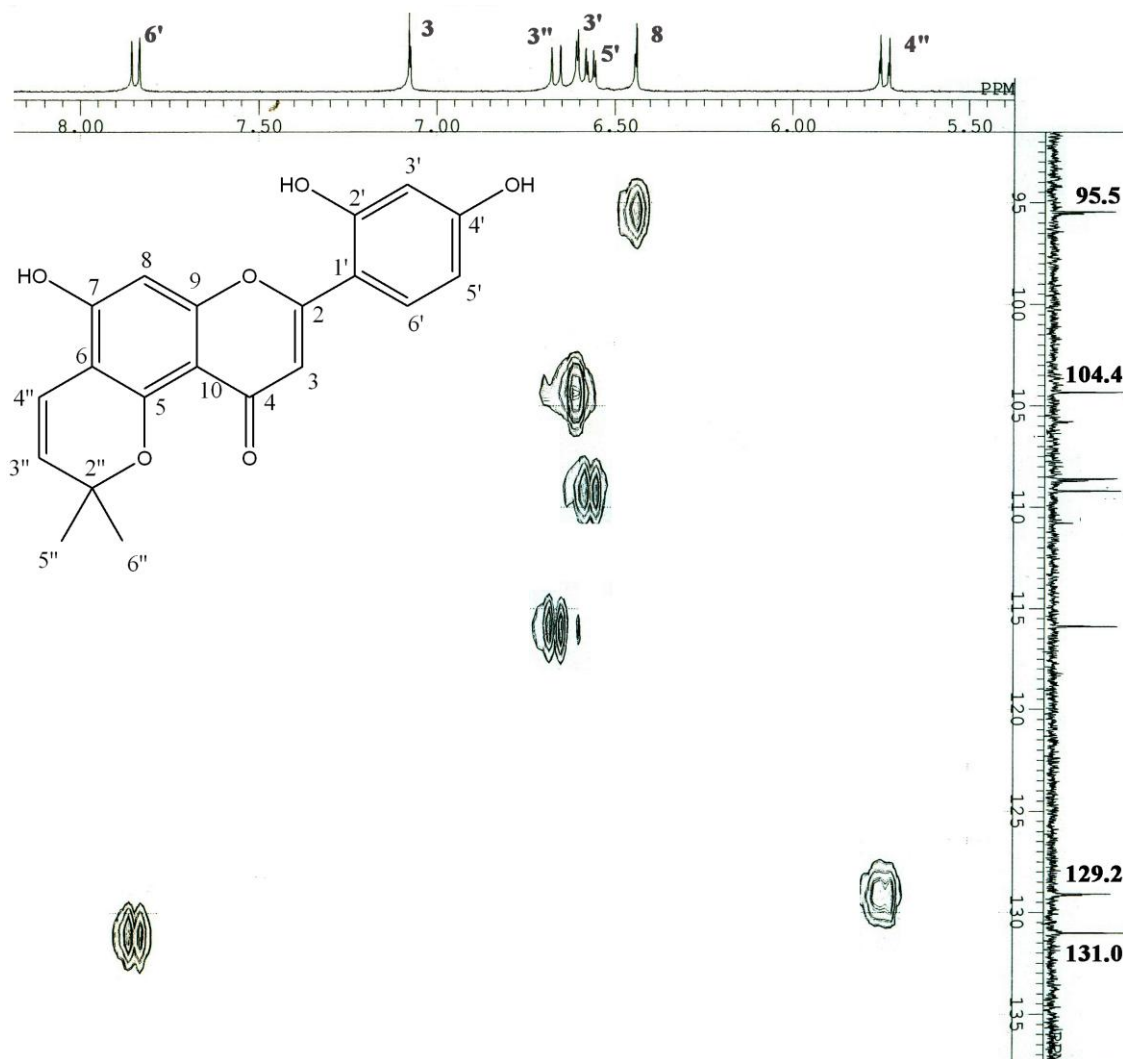
Figure 38 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL2 (acetone-*d*<sub>6</sub>)



**Figure 39**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL2 (acetone- $d_6$ )



**Figure 40** HMBC Spectrum of compound CAL2 (acetone- $d_6$ )



**Figure 41** HMQC Spectrum of compound CAL2 (acetone- $d_6$ )

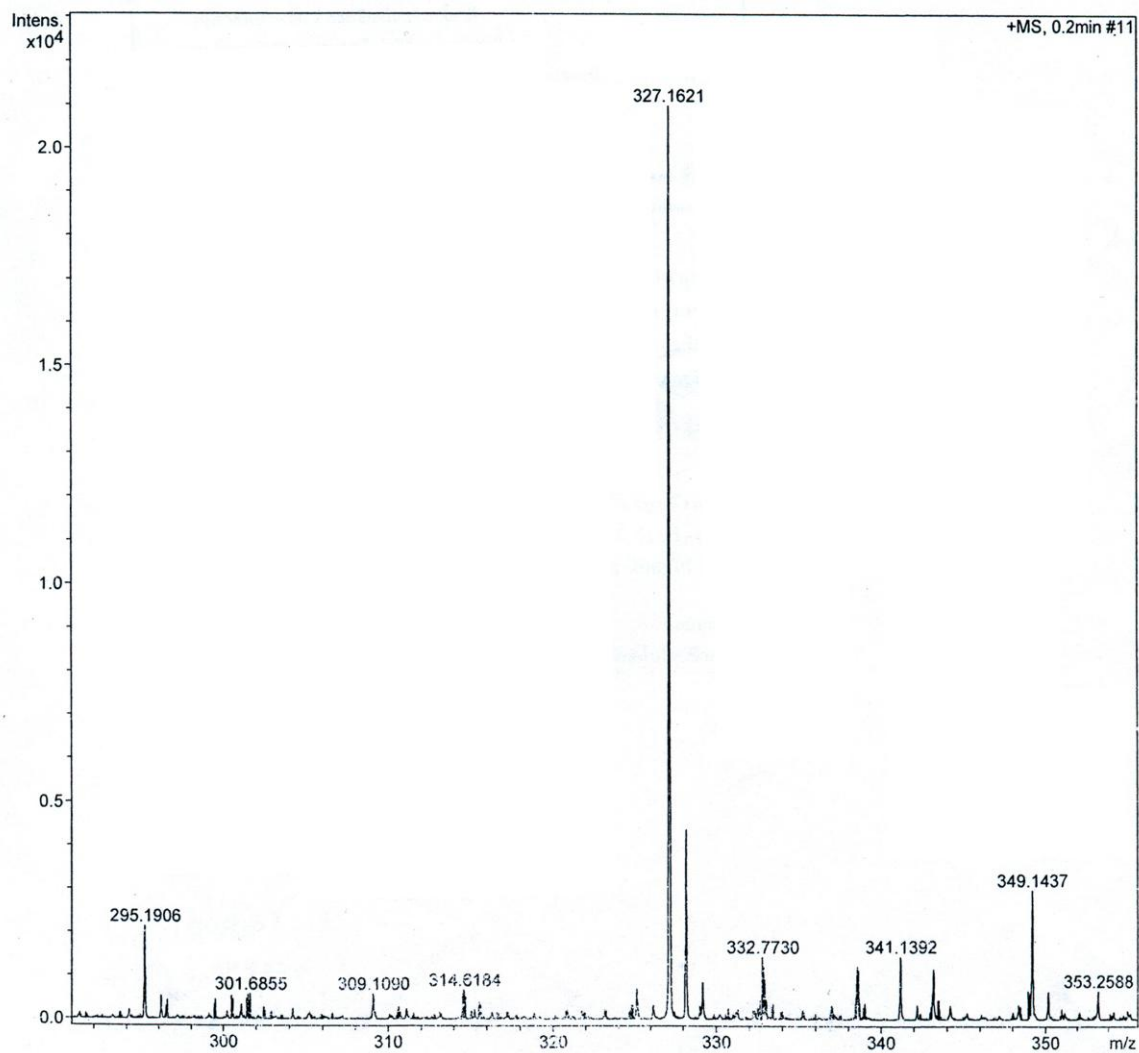
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Sample Name 75MF3\_2\_III

Acquisition Date 5/3/2011 2:14:37 PM

Operator Sutichai Ext: 3560  
Instrument micrOTOF Bruker

**Acquisition Parameter**

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Not active			Set Dry Heater	150 °C
Scan Begin	100 m/z	Set Capillary	5000 V	Set Dry Gas	2.0 l/min
Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source



**Figure 42** HRESI Mass Spectrum of compound CAL3

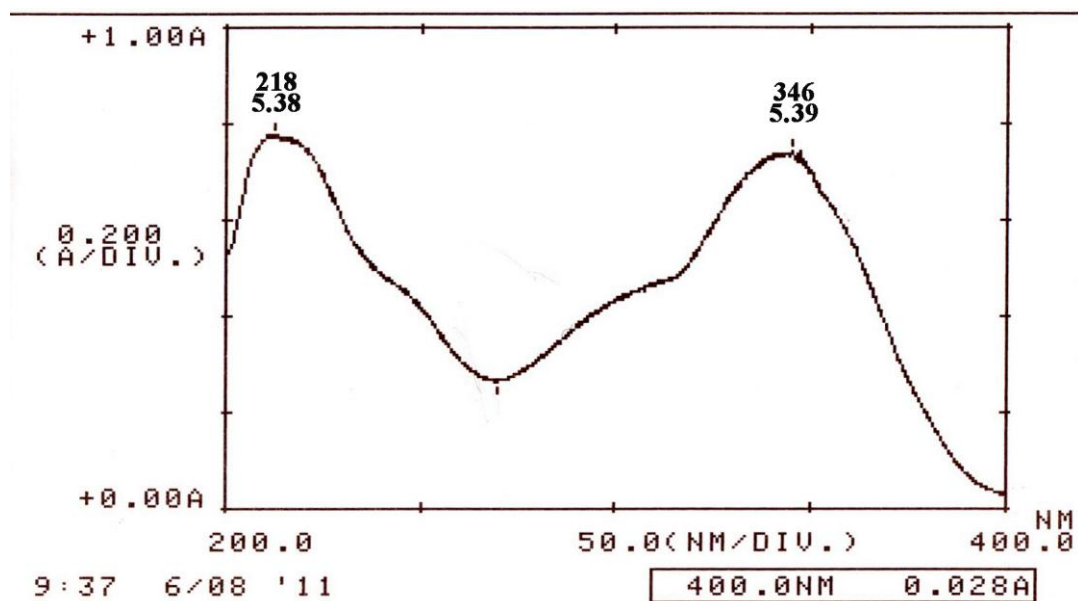


Figure 43 UV Spectrum of compound CAL3 (methanol)

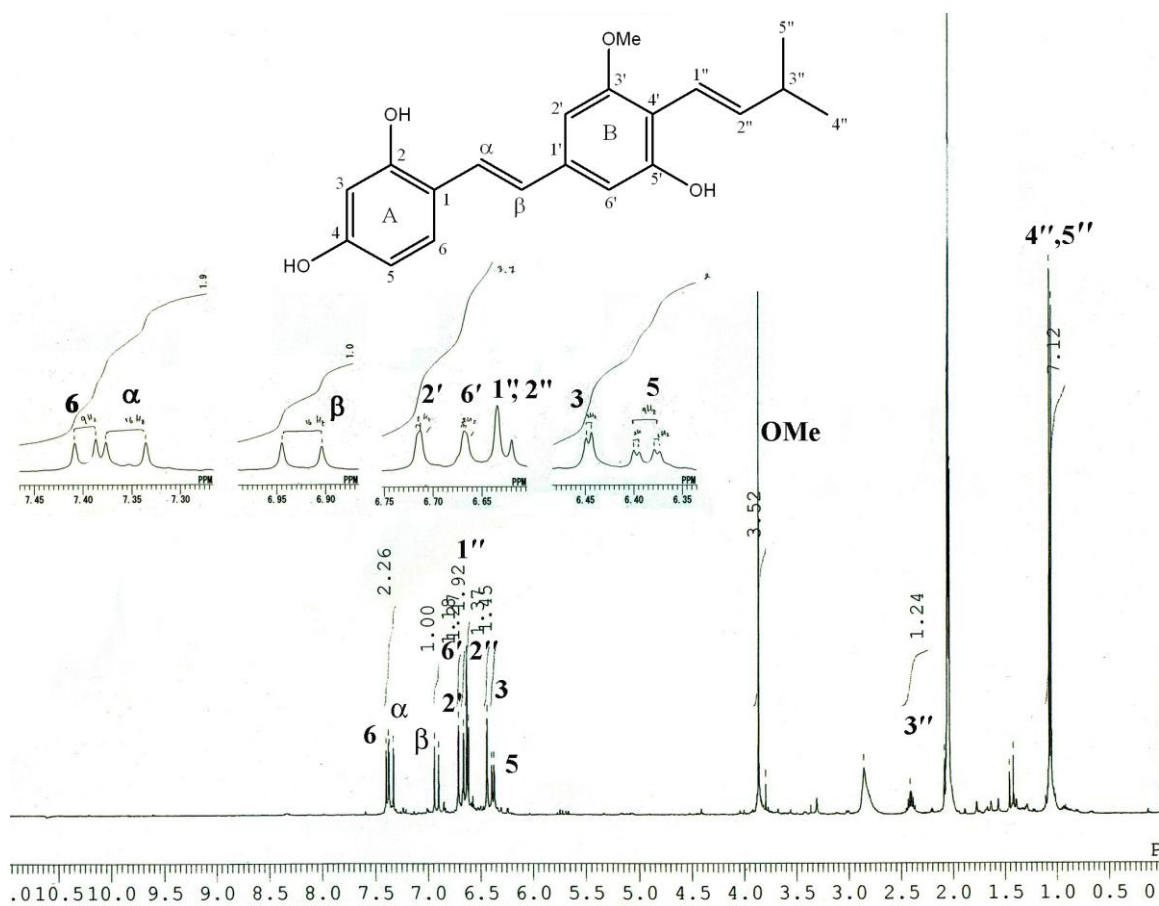


Figure 44  $^1\text{H-NMR}$  (400 MHz) Spectrum of compound CAL3 (acetone- $d_6$ )

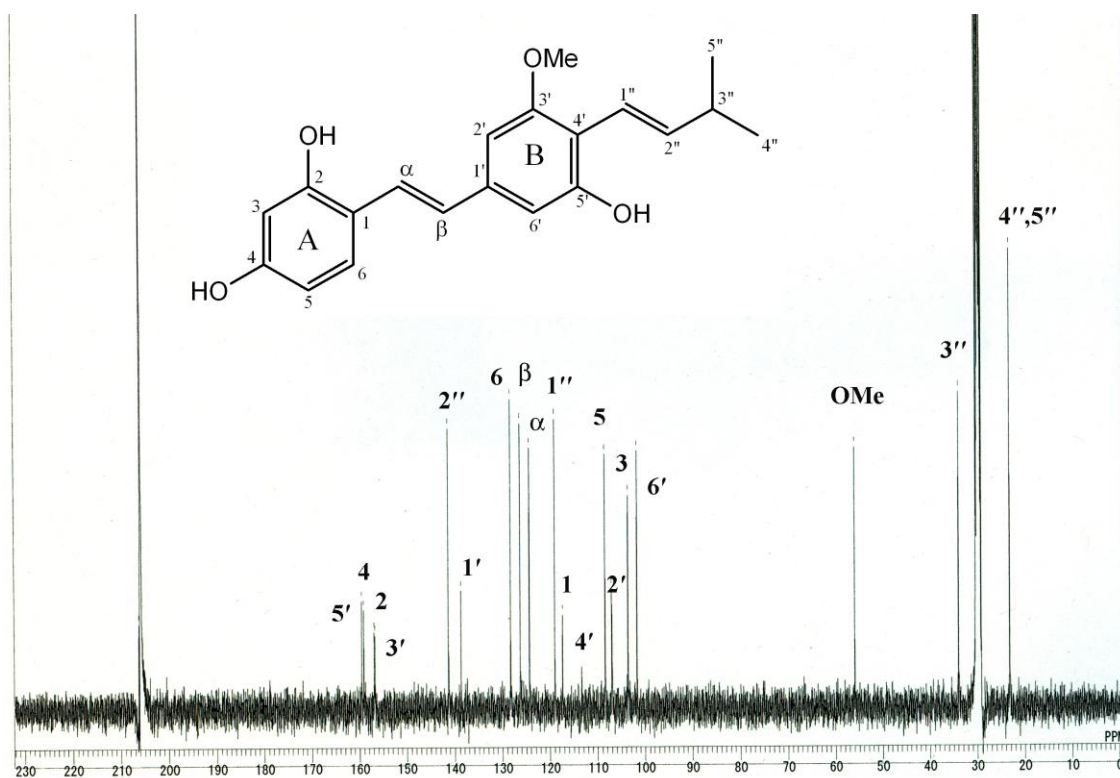


Figure 45  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL3 ( $\text{acetone-}d_6$ )

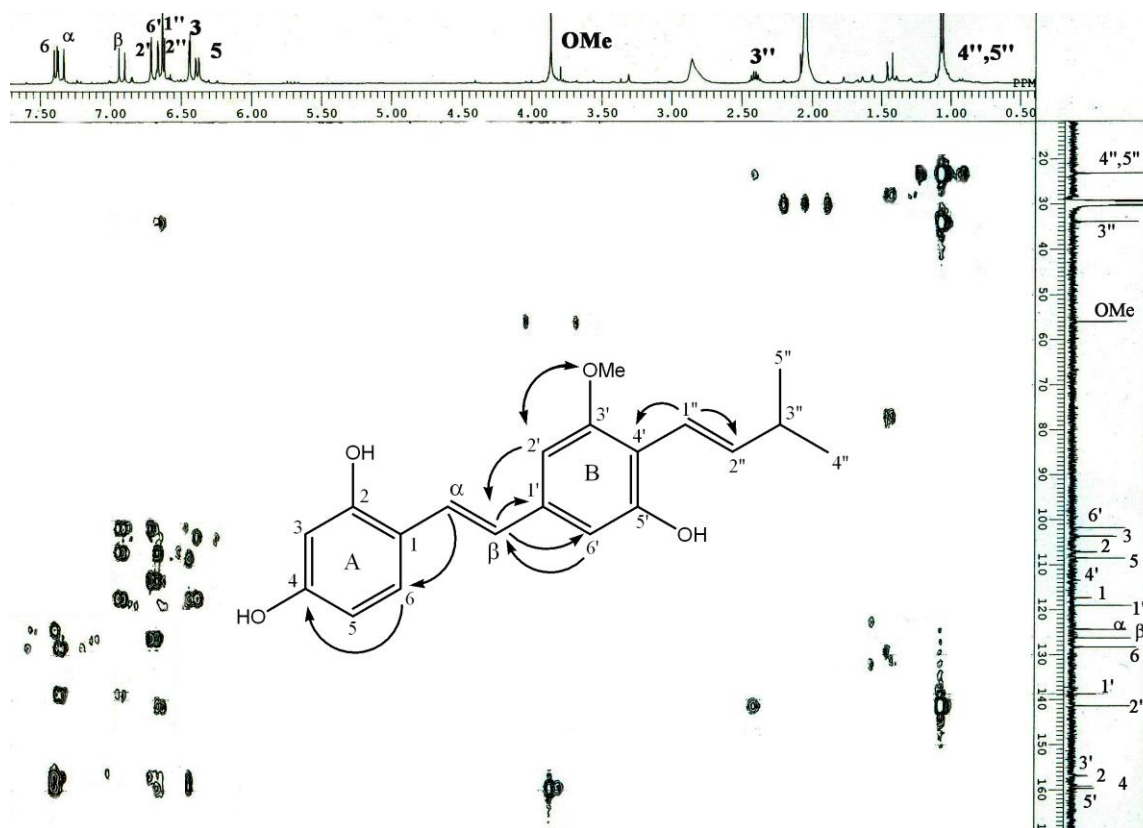


Figure 46 HMBC Spectrum of compound CAL3 ( $\text{acetone-}d_6$ )

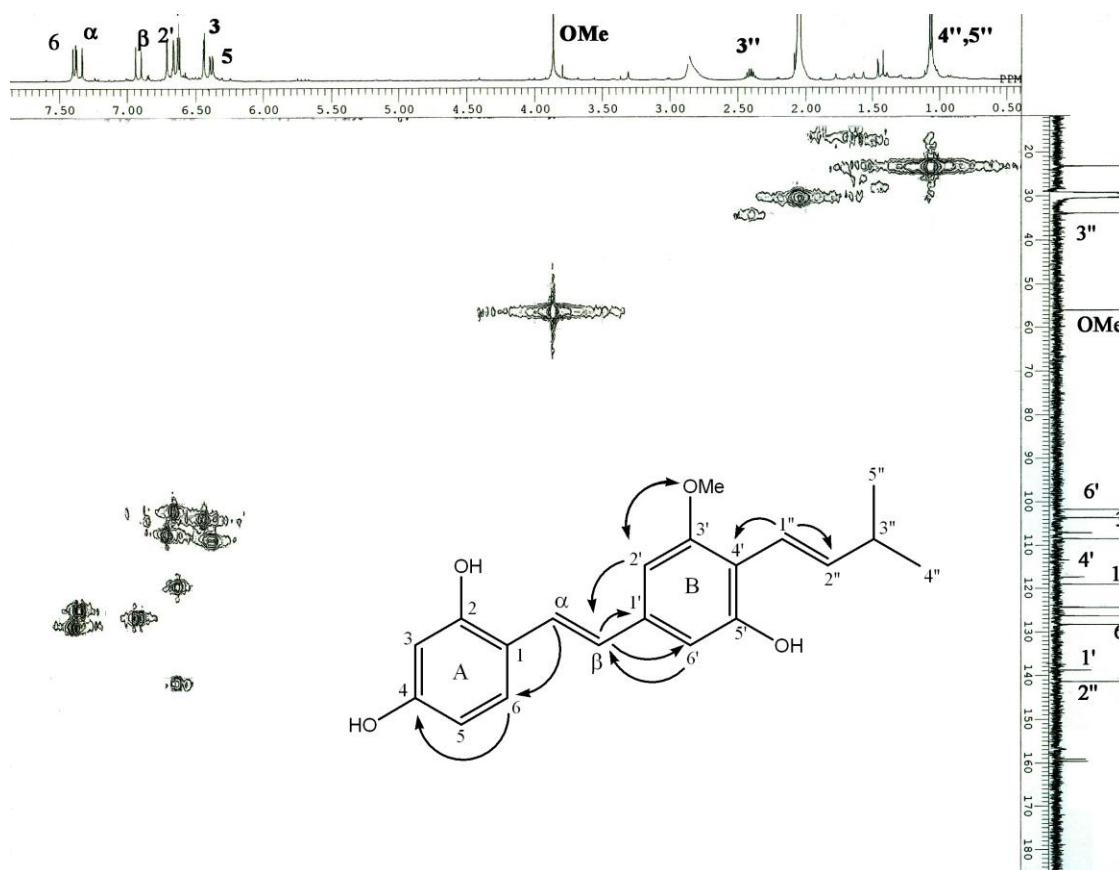


Figure 47 HMQC Spectrum of compound CAL 3 (acetone- $d_6$ )

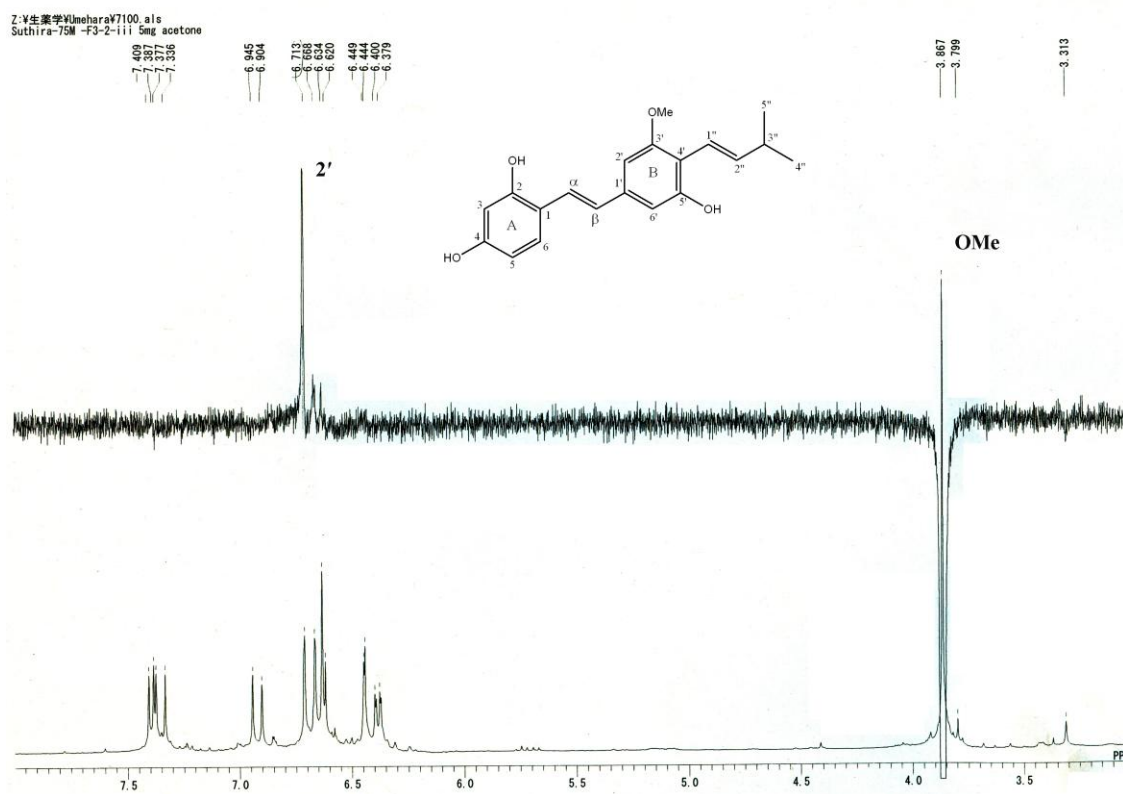


Figure 48 NOE Spectrum of compound CAL3 (acetone- $d_6$ )



## High resolution report

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Sample Name 75MF3\_2\_1V

Acquisition Date 5/3/2011 2:22:39 PM

Operator Sutichai Ext: 3560  
Instrument micrOTOF Bruker  
Calibrate by Sodium Formate

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Not active			Set Dry Heater	150 °C
Scan Begin	100 m/z	Set Capillary	5000 V	Set Dry Gas	2.0 l/min
Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

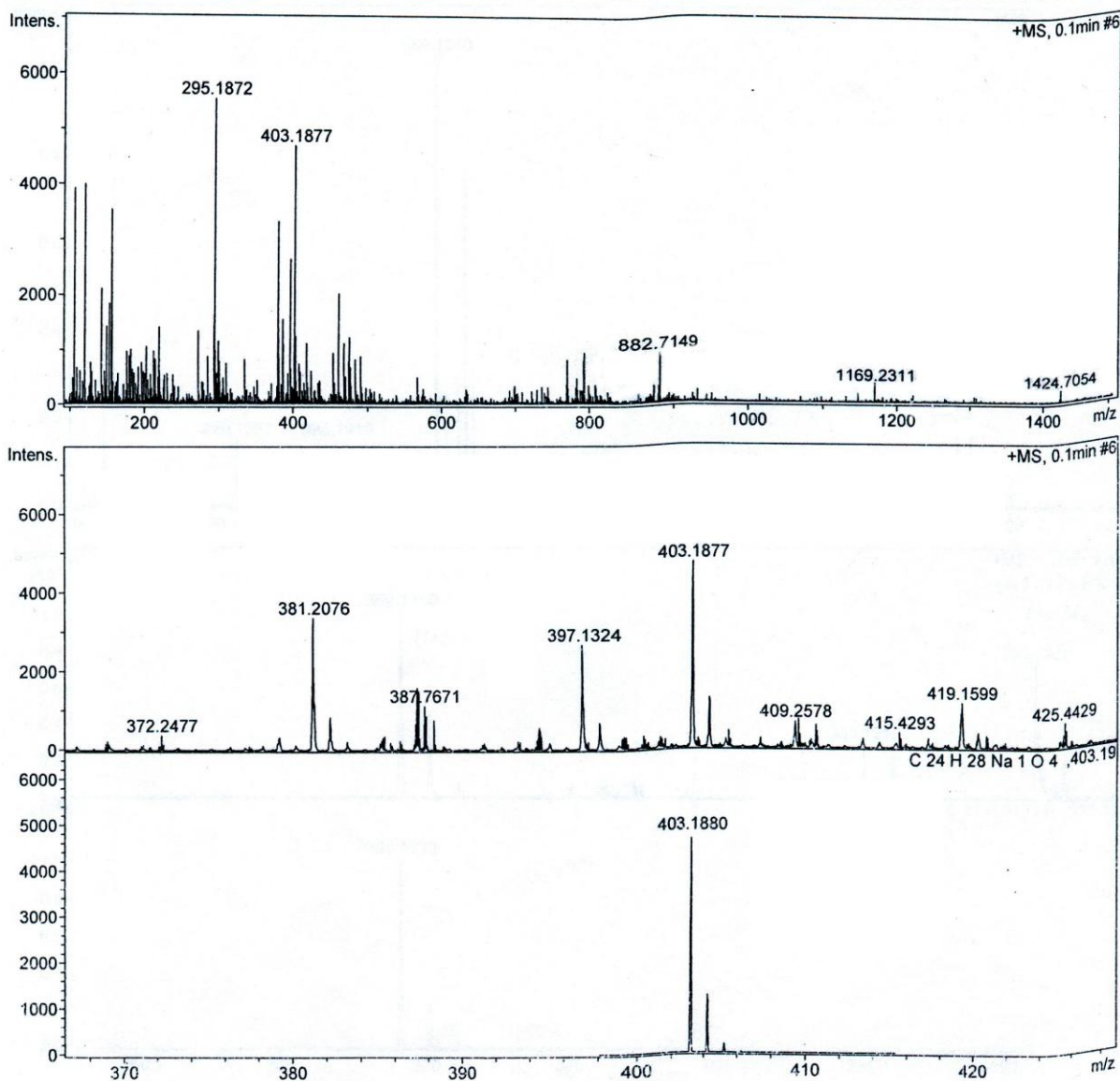


Figure 49 HRESI Mass Spectrum of compound CAL4

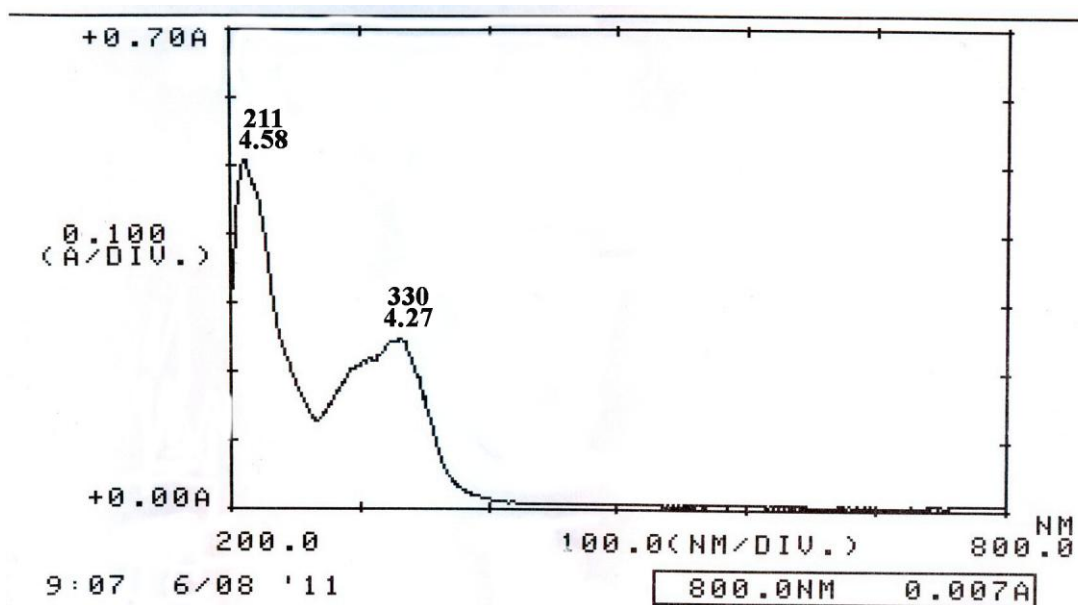


Figure 50 UV Spectrum of compound CAL4 (methanol)

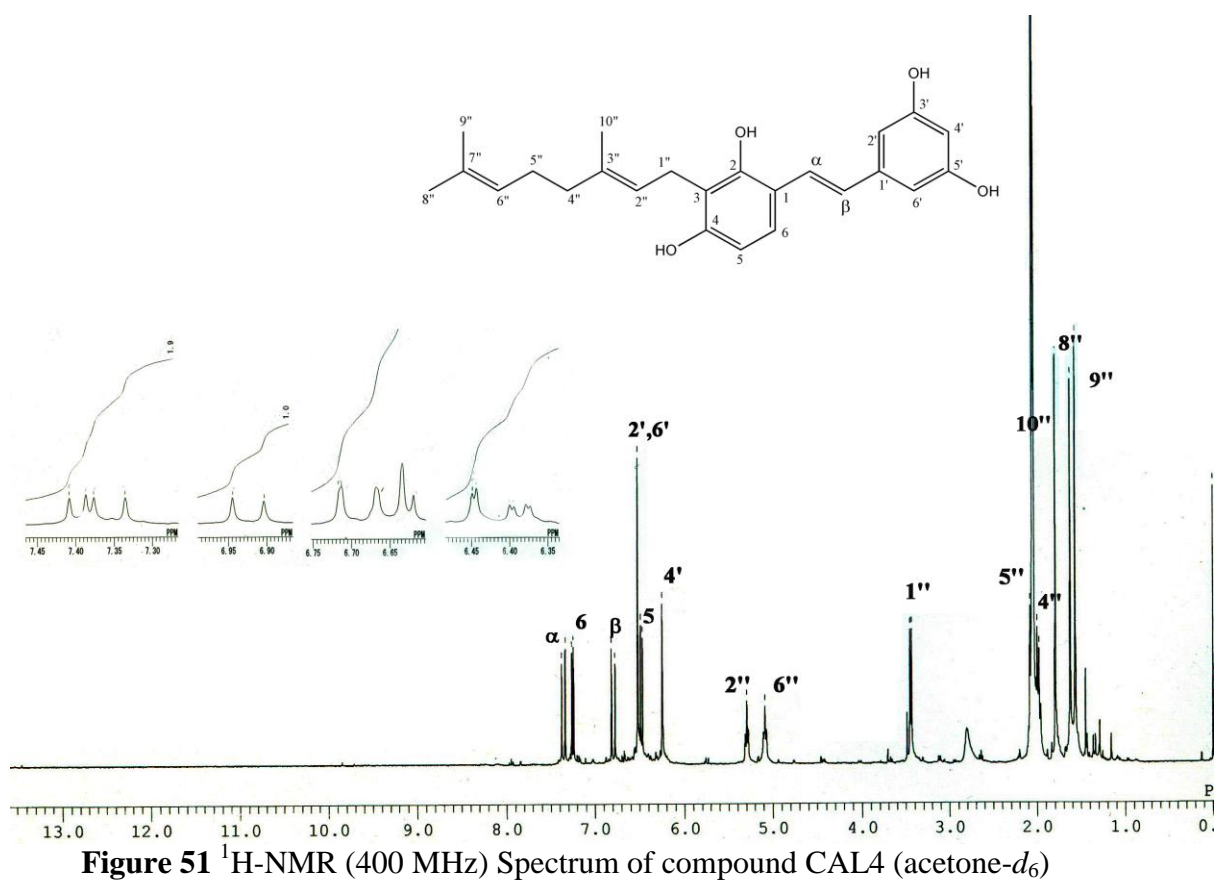


Figure 51 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL4 (acetone-*d*<sub>6</sub>)

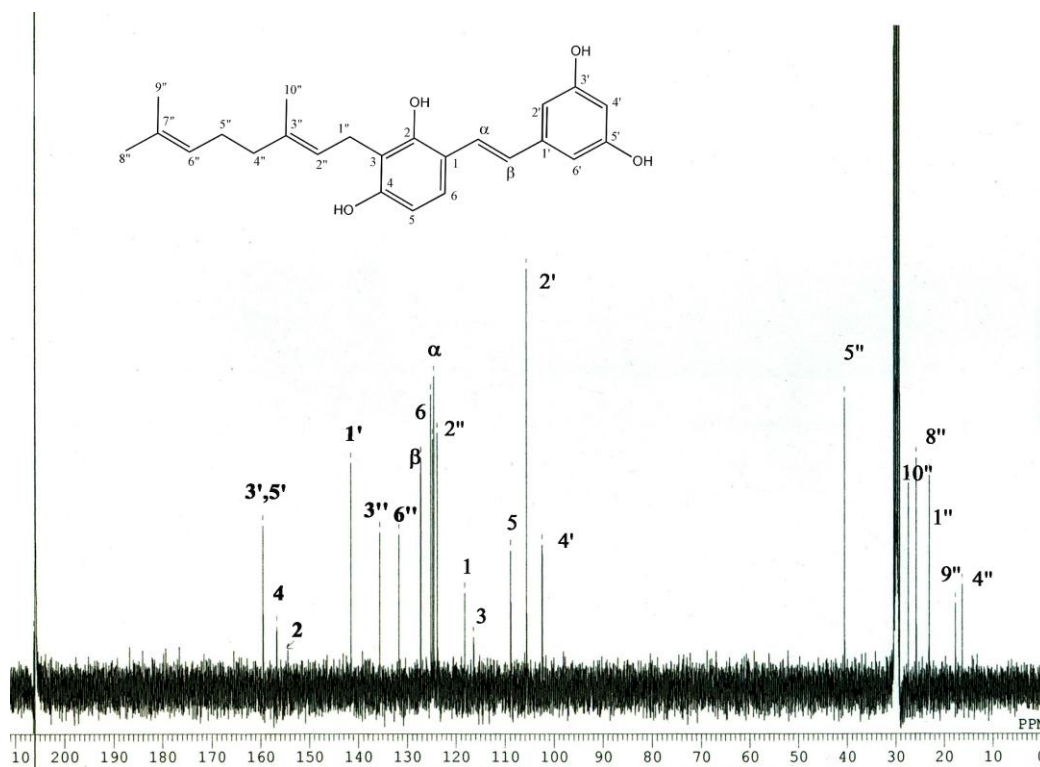


Figure 52  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL4 (acetone- $d_6$ )

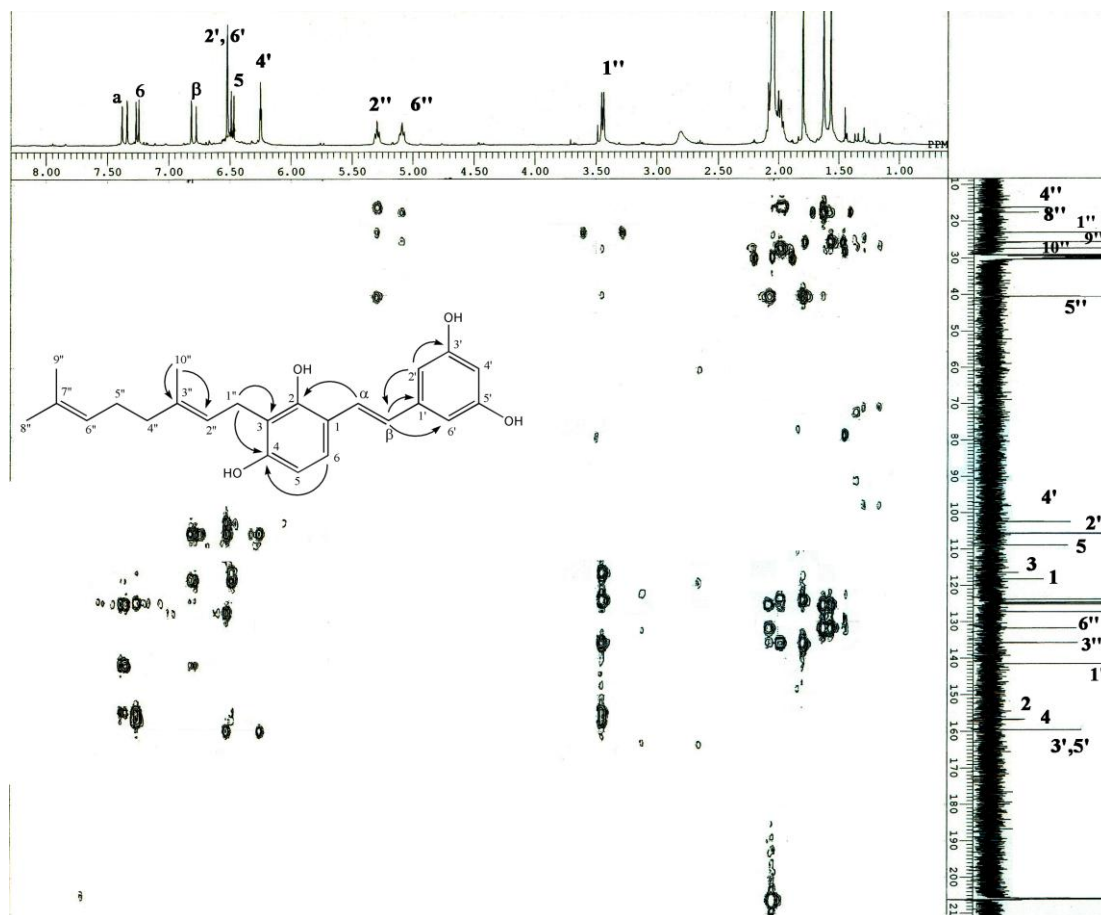


Figure 53 HMBC Spectrum of compound CAL4 (acetone- $d_6$ )

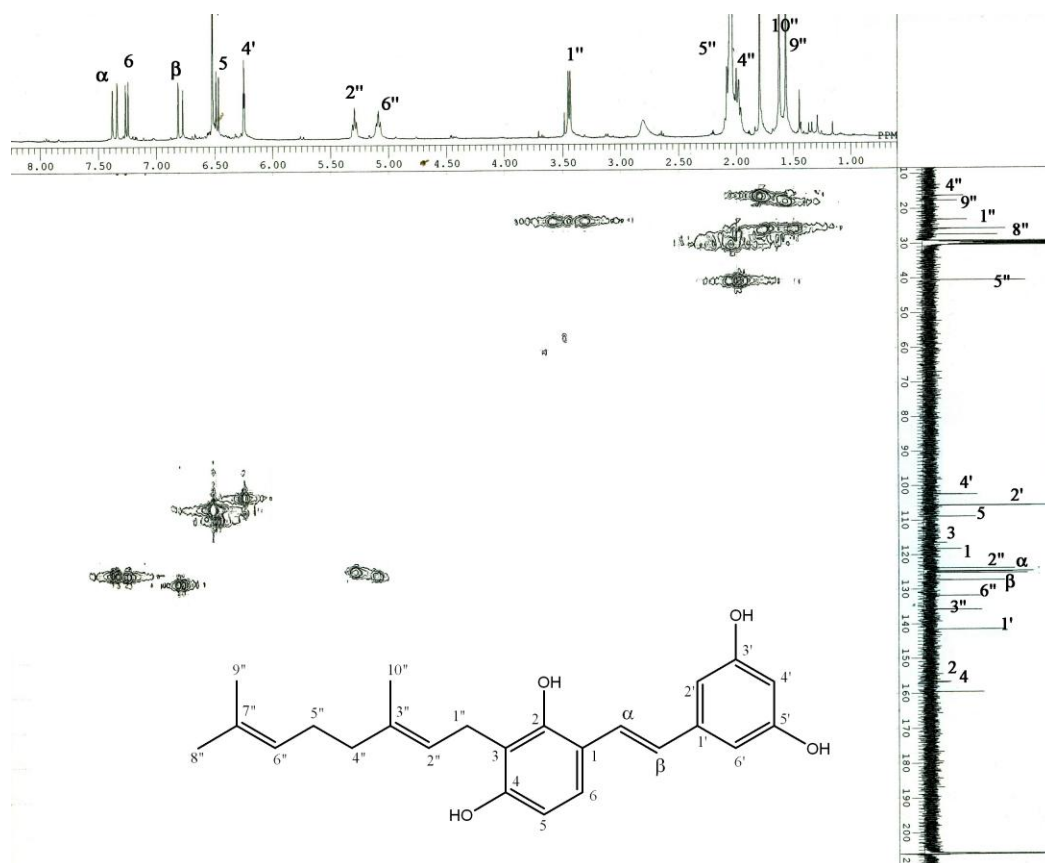


Figure 54 HMQC Spectrum of compound CAL 4 (acetone- $d_6$ )

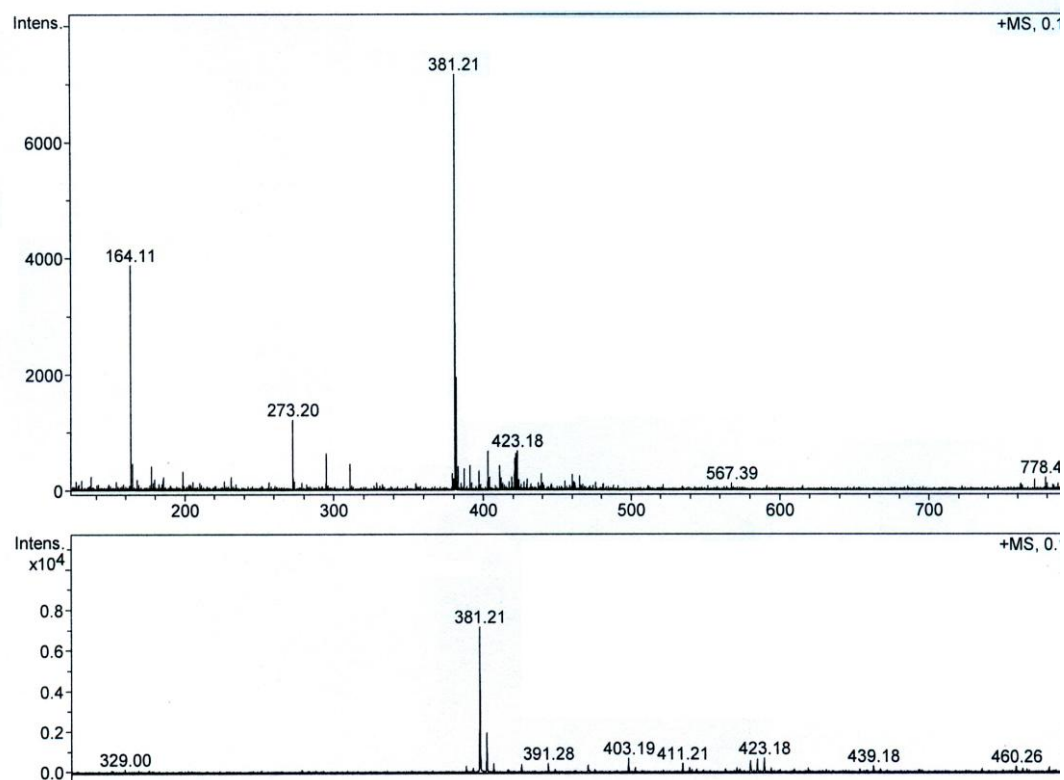
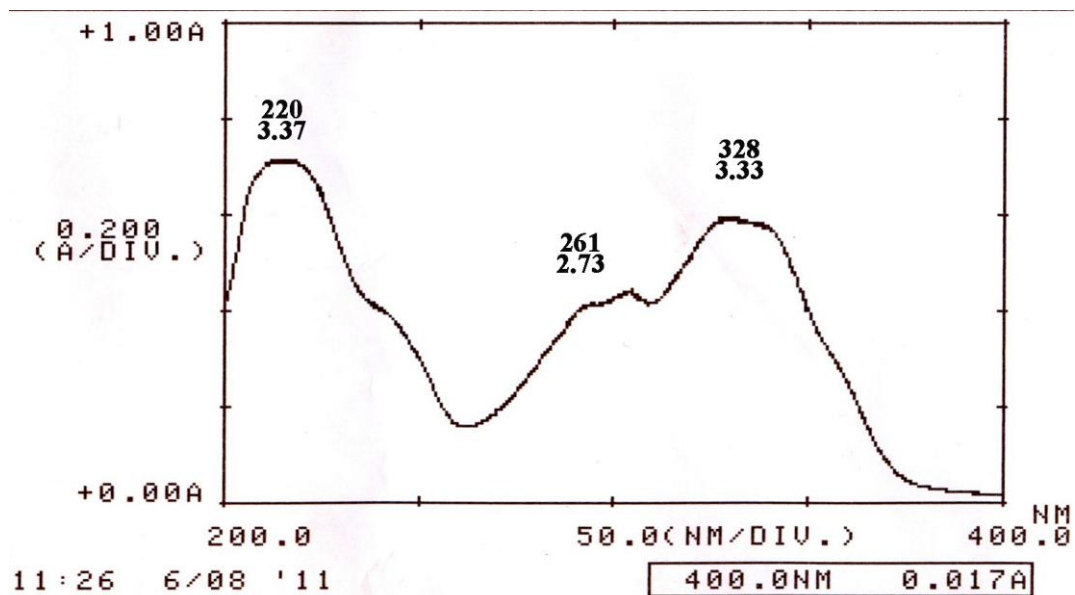
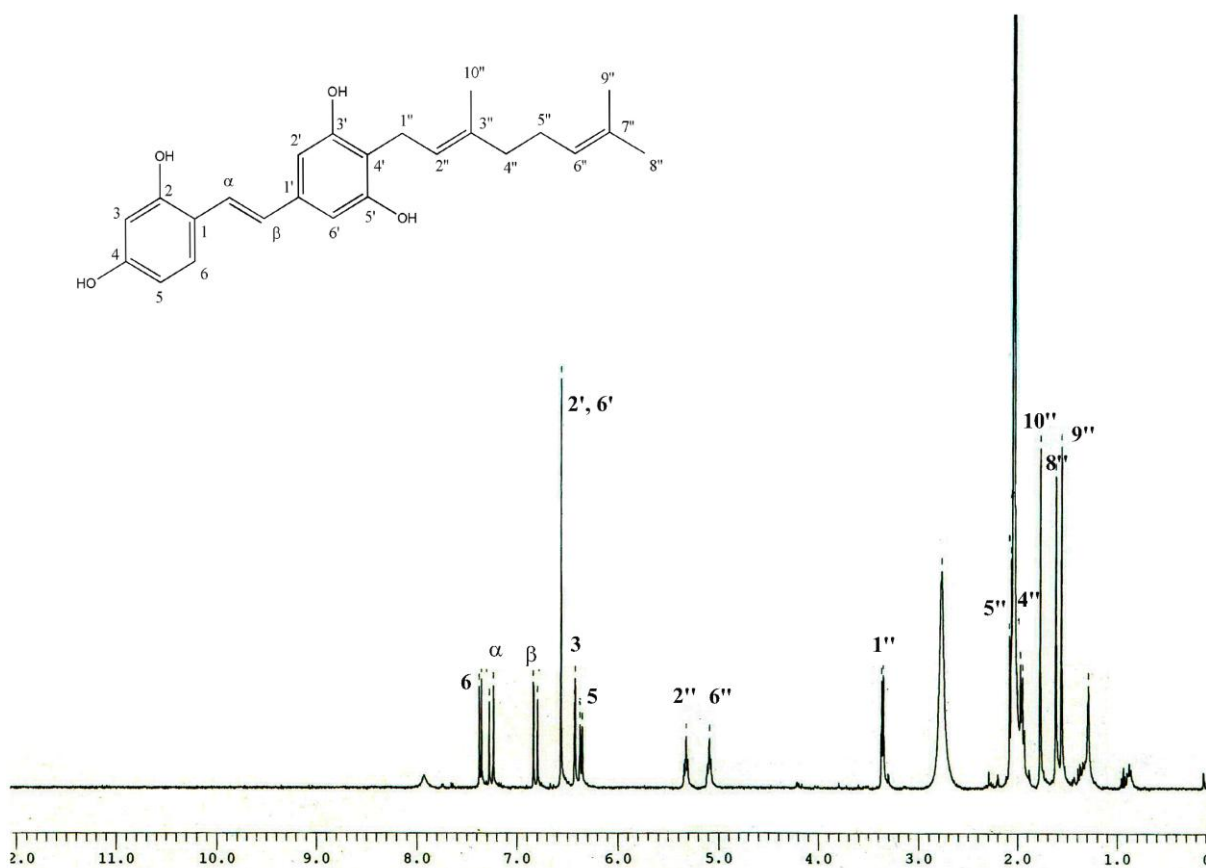


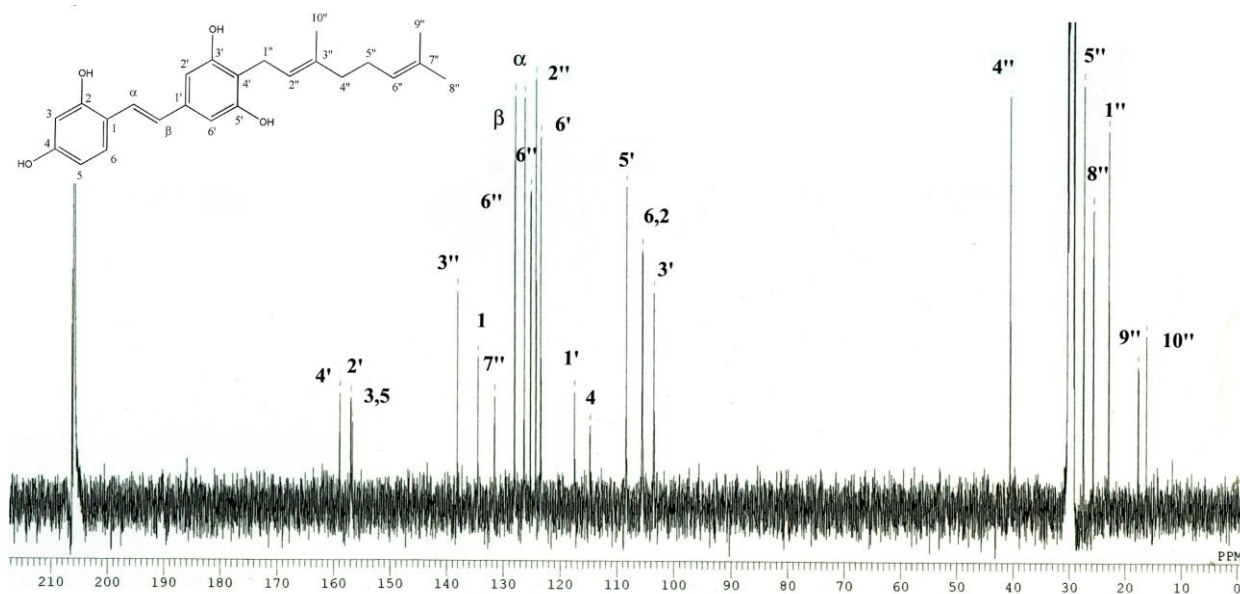
Figure 55 ESI Mass Spectrum of compound CAL5



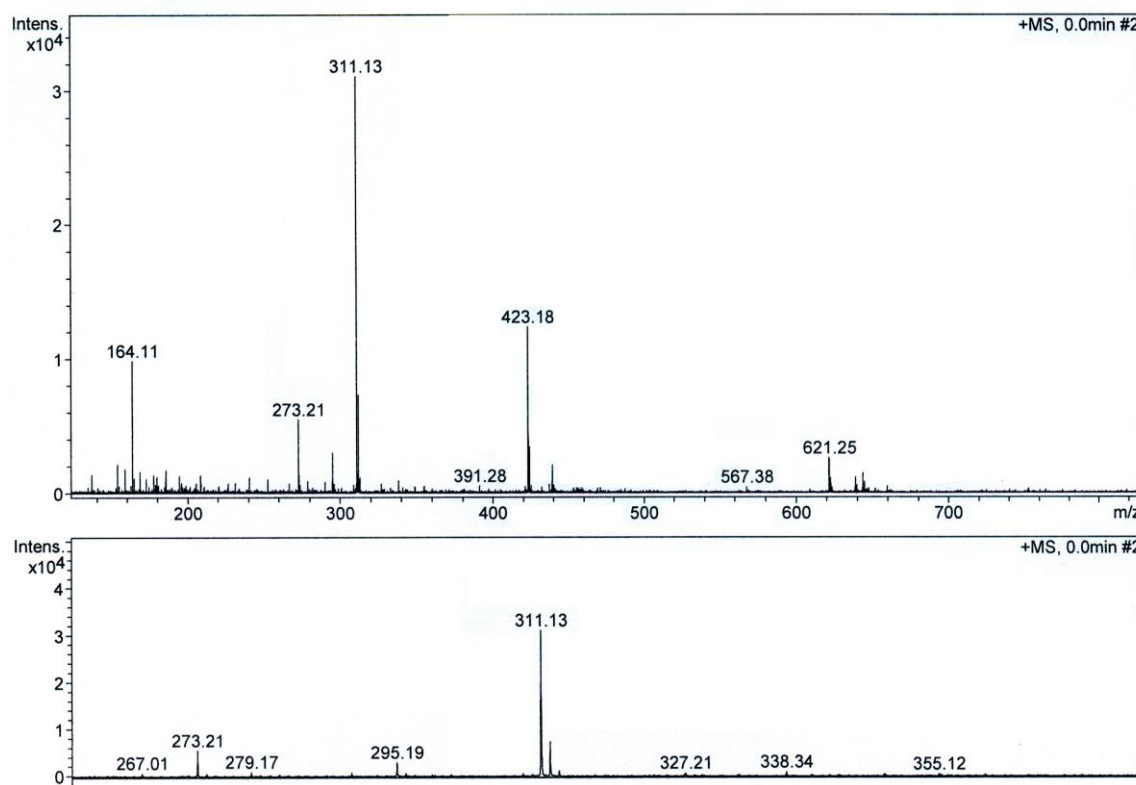
**Figure 56** UV Spectrum of compound CAL5 (methanol)



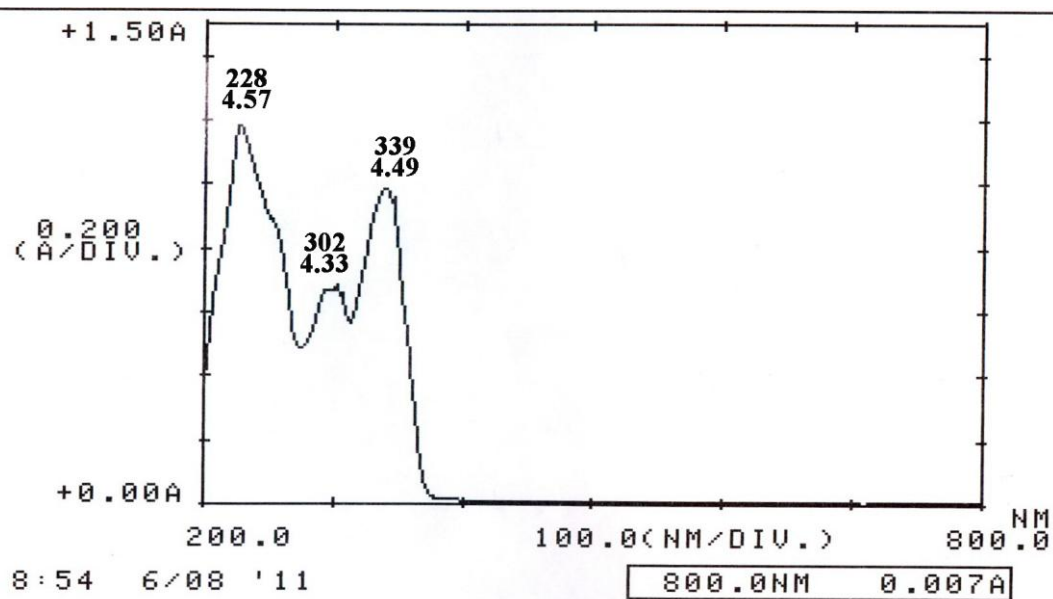
**Figure 57**  $^1\text{H-NMR}$  (400 MHz) Spectrum of compound CAL5 (acetone- $d_6$ )



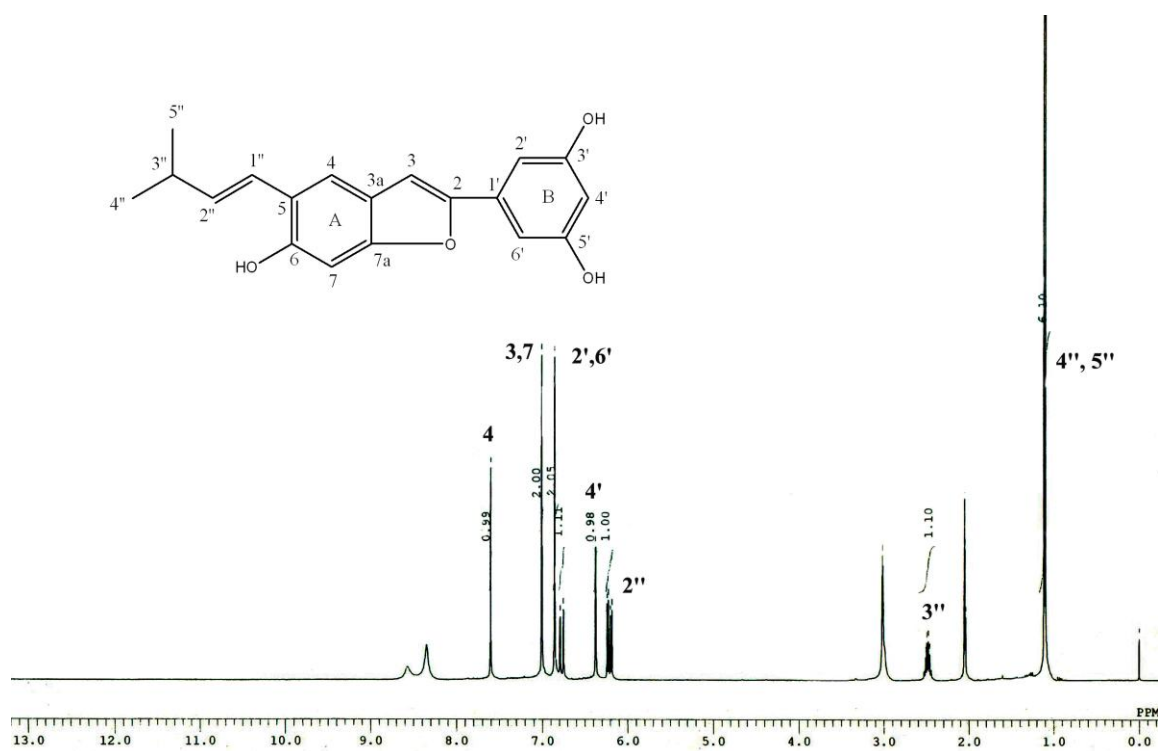
**Figure 58**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL5 (acetone- $d_6$ )



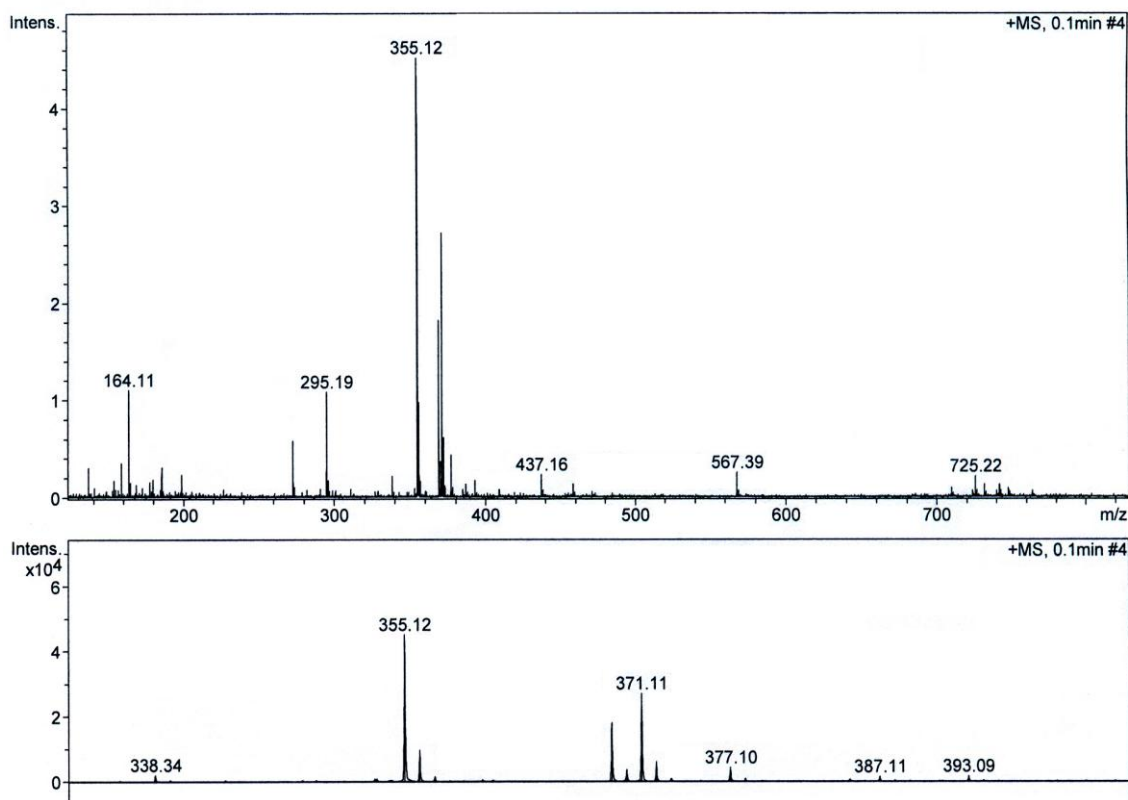
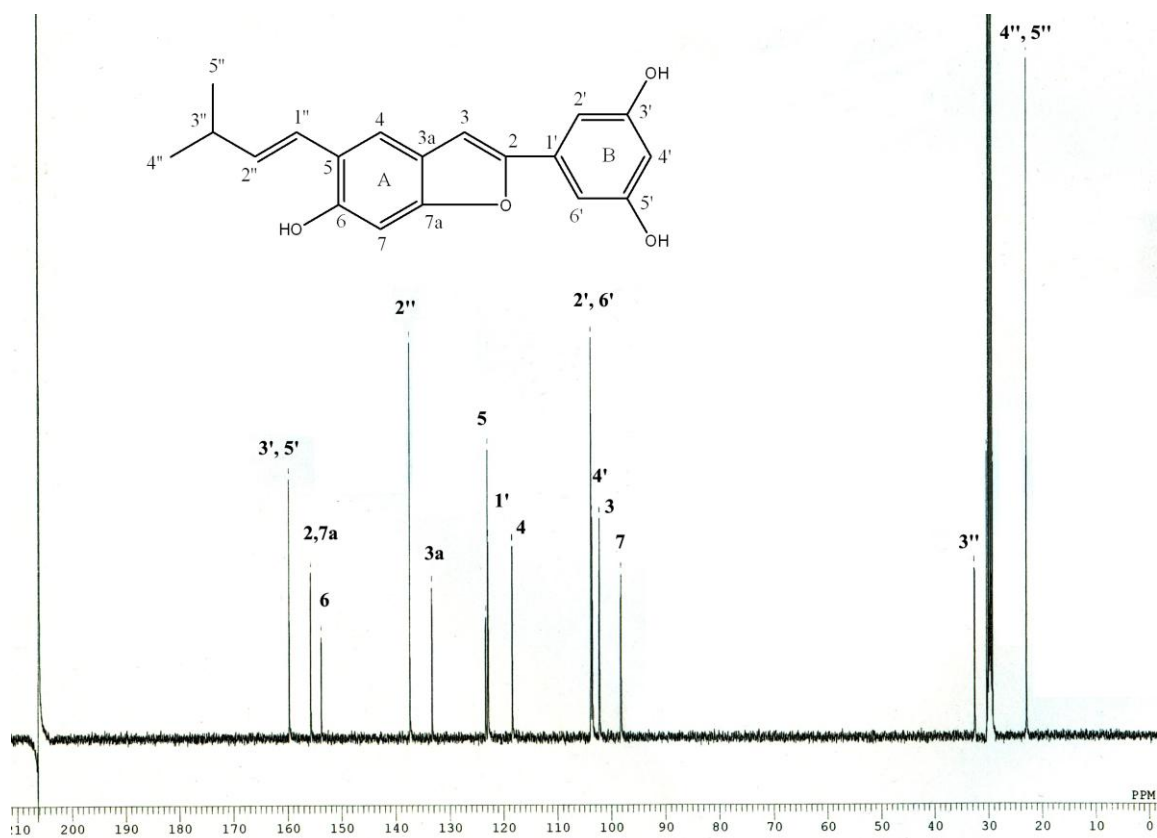
**Figure 59** ESI Mass Spectrum of compound CAL6



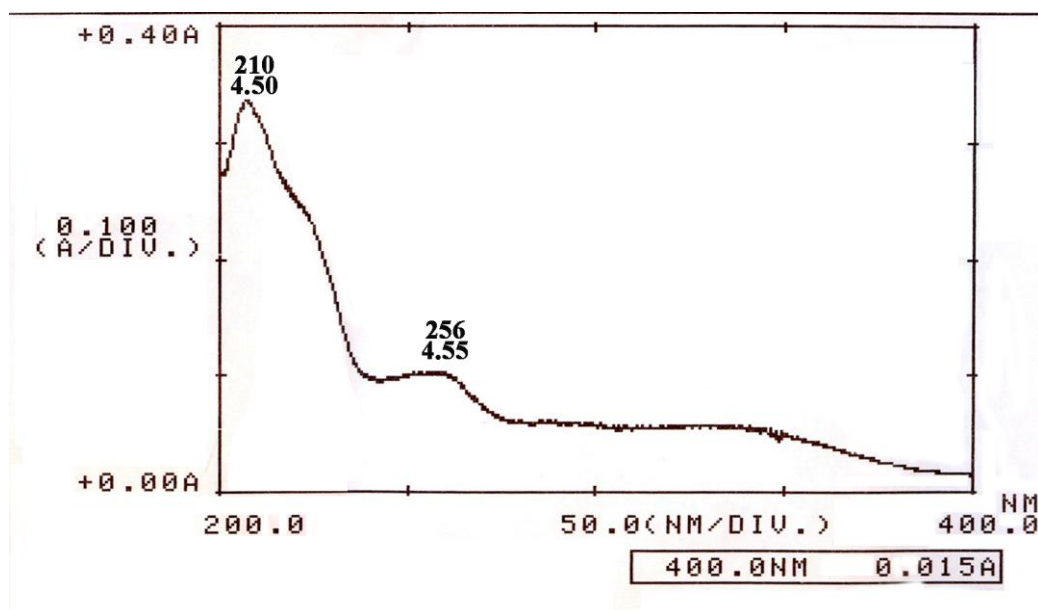
**Figure 60** UV Spectrum of compound CAL6 (methanol)



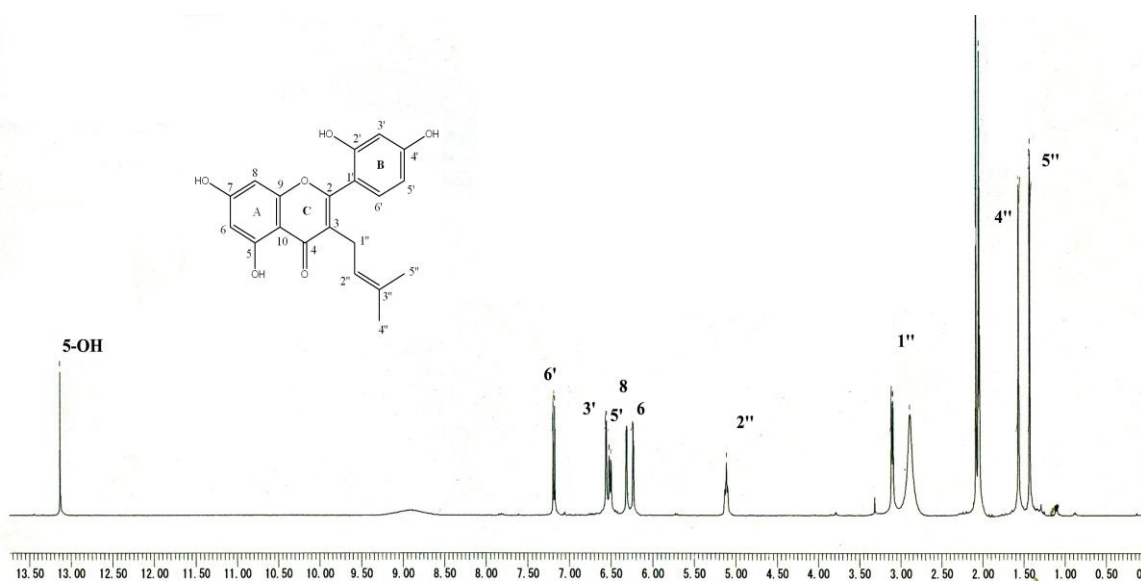
**Figure 61**  $^1\text{H-NMR}$  (400 MHz) Spectrum of compound CAL6 (acetone- $d_6$ )



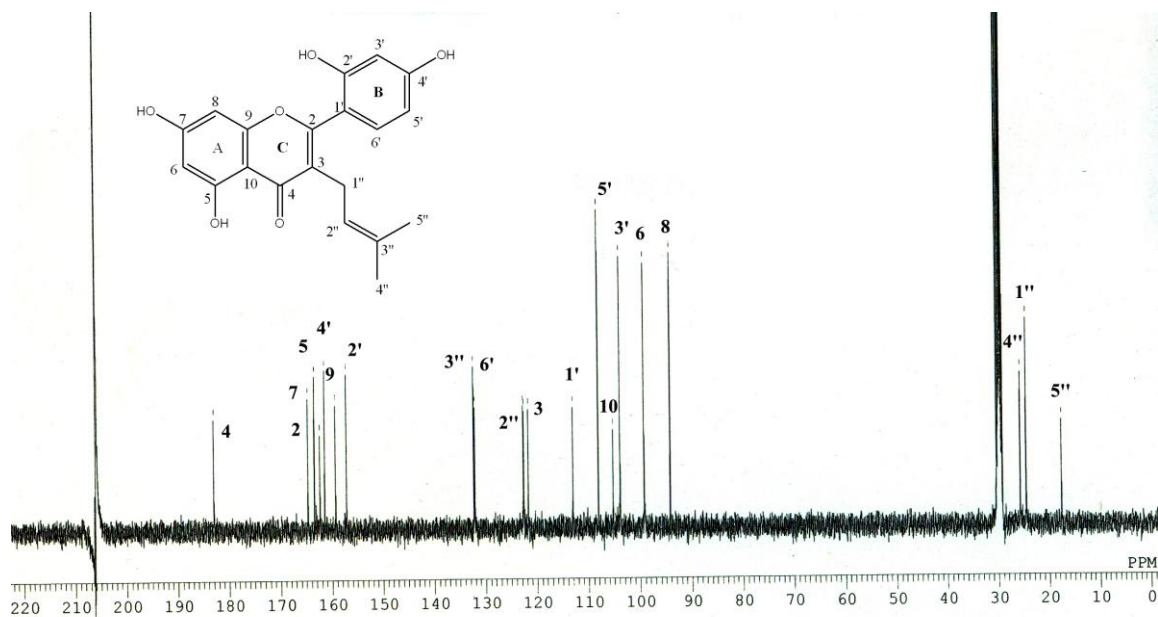




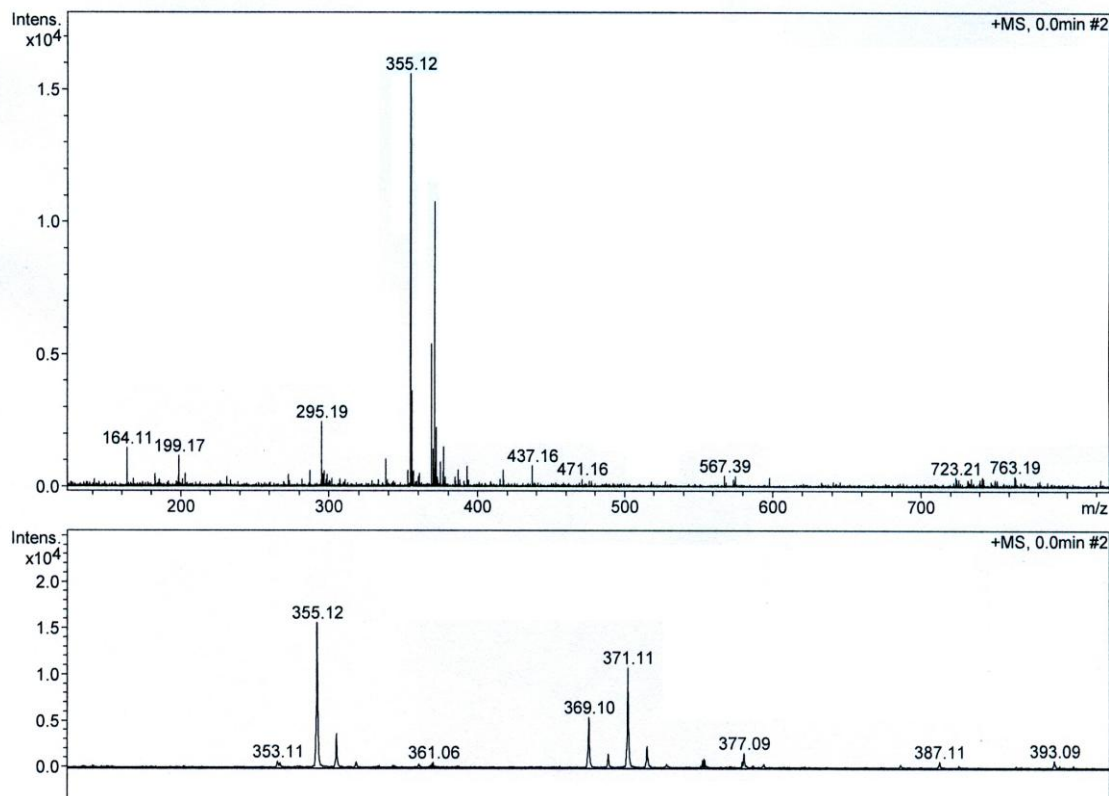
**Figure 64** UV Spectrum of compound CAL7 (methanol)



**Figure 65** <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL7 (acetone-d<sub>6</sub>)



**Figure 66**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL7 ( $\text{acetone-}d_6$ )



**Figure 67** ESI Mass Spectrum of compound CAL8

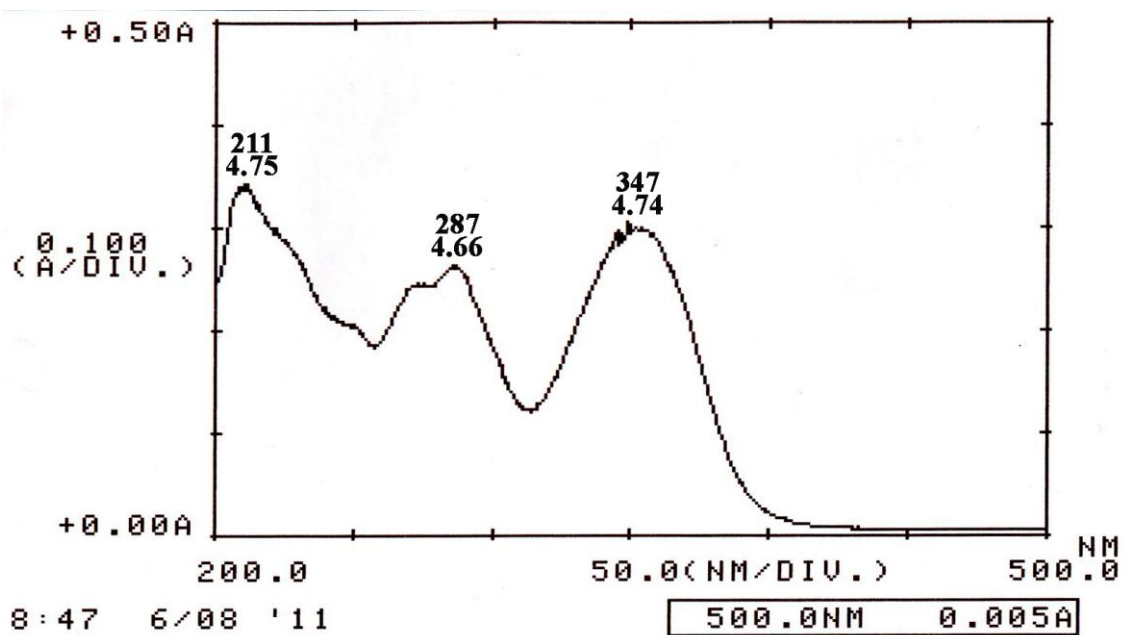


Figure 68 UV Spectrum of compound CAL8 (methanol)

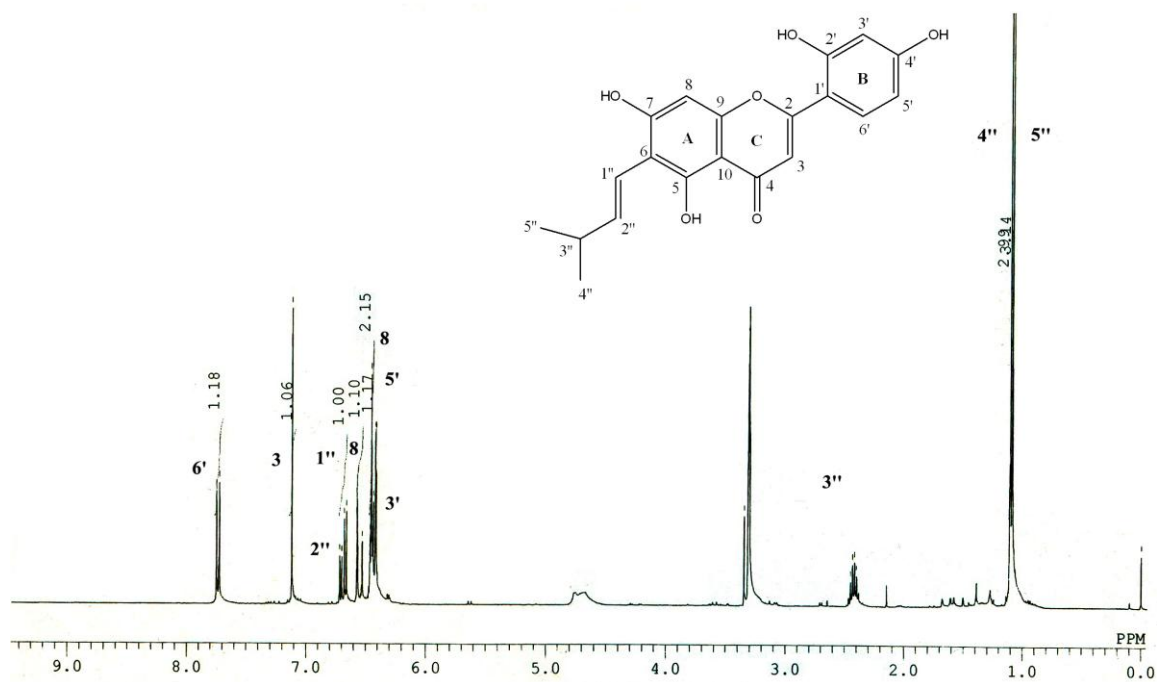
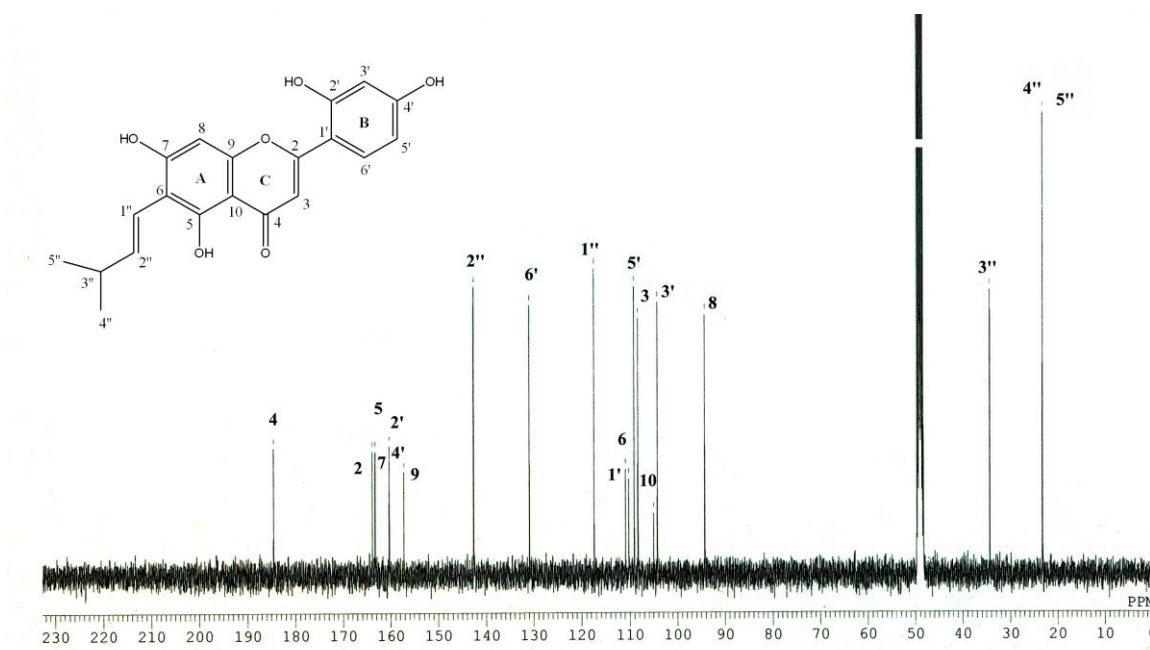
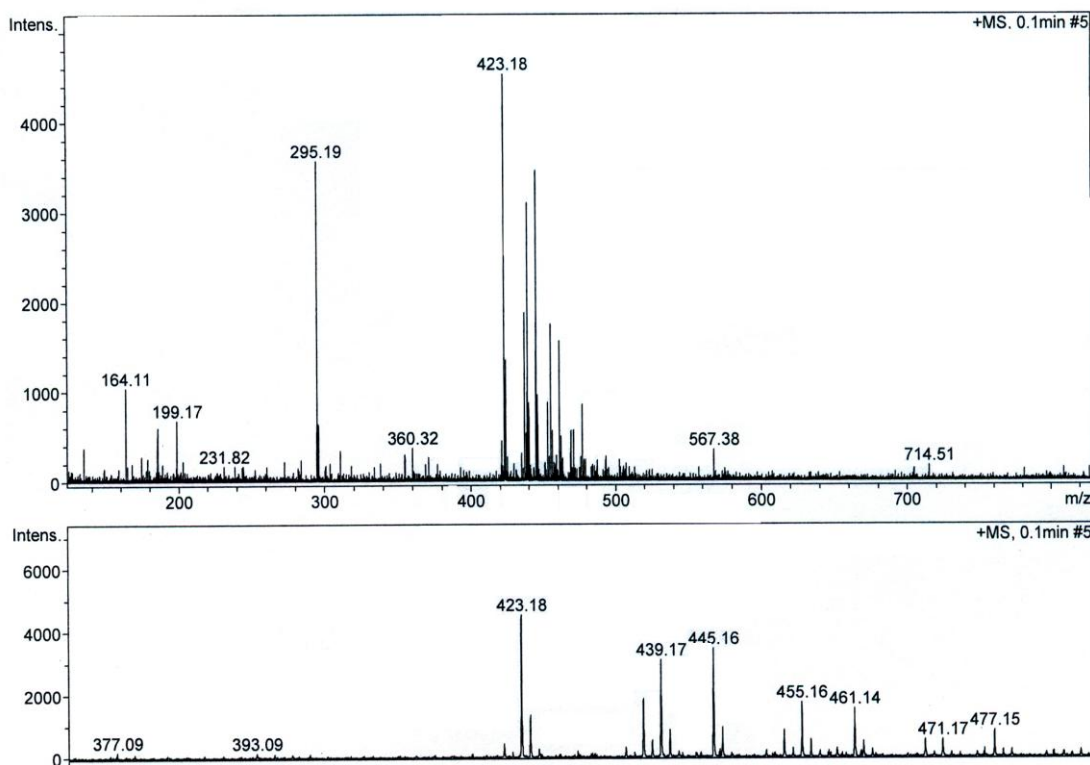


Figure 69 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL8 (methanol-*d*<sub>4</sub>)



**Figure 70**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL8 (methanol- $d_4$ )



**Figure 71** ESI Mass Spectrum of compound CAL9

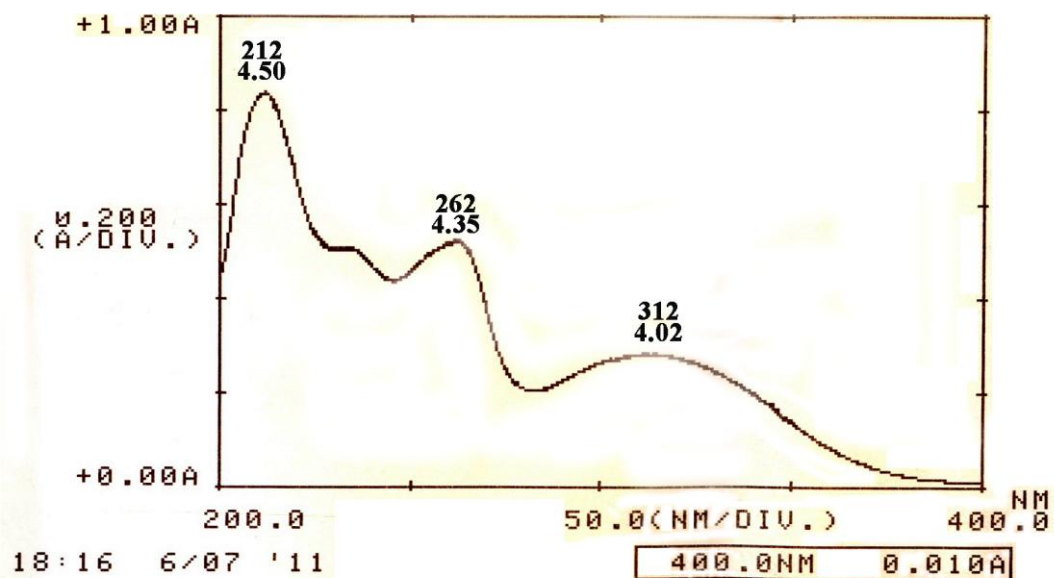


Figure 72 UV Spectrum of compound CAL9 (methanol)

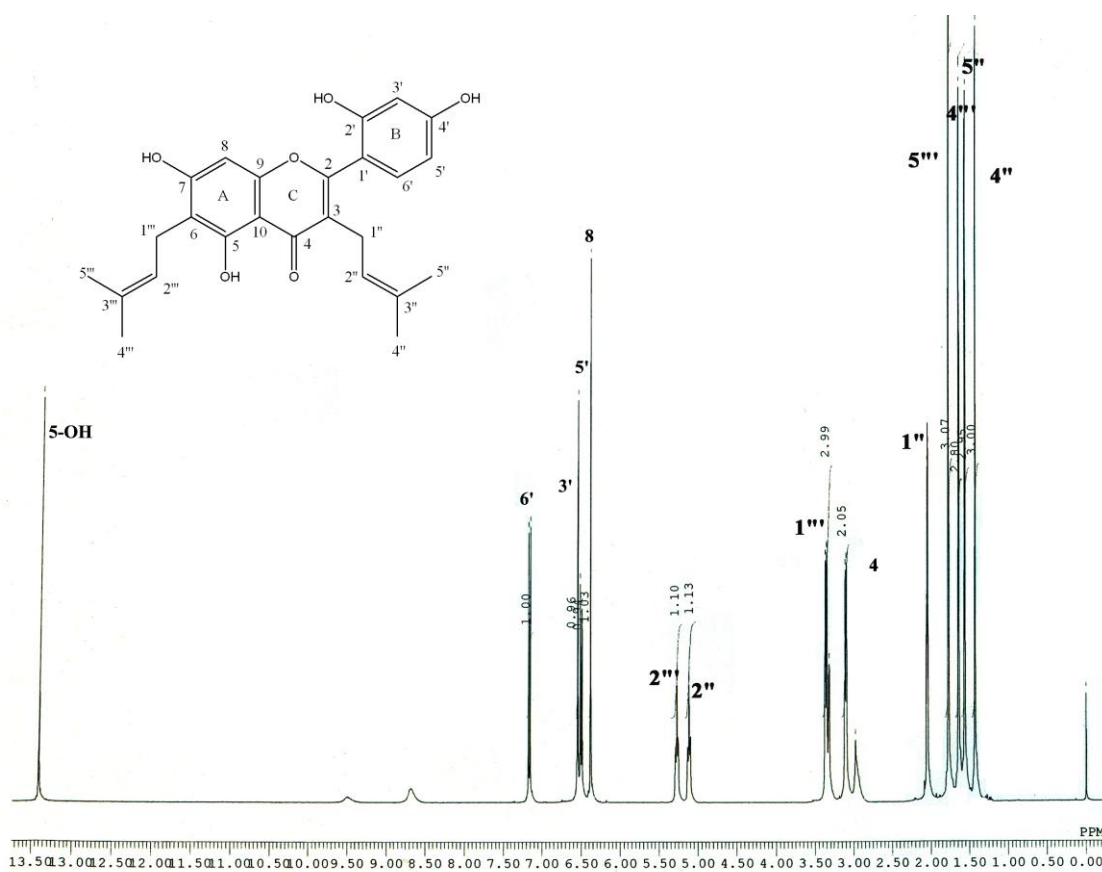
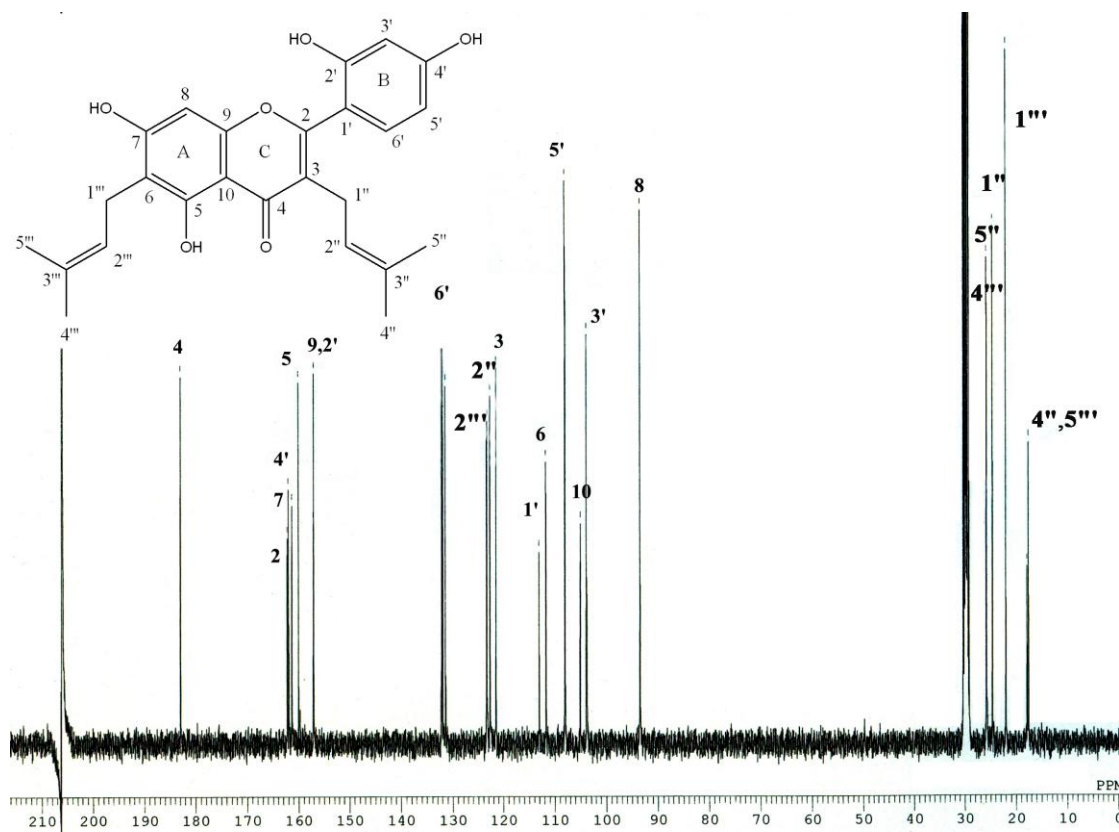
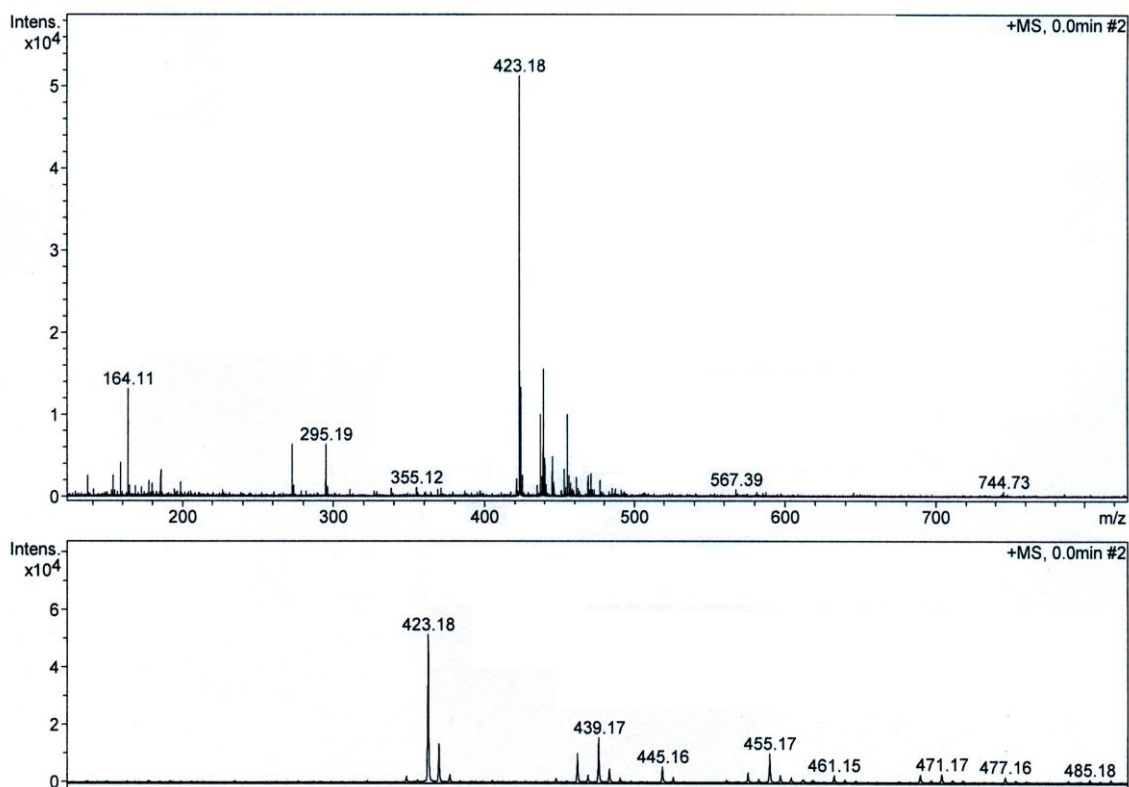


Figure 73 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL9 (acetone-*d*<sub>6</sub>)



**Figure 74**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL9 (acetone- $d_6$ )



**Figure 75** ESI Mass Spectrum of compound CAL10

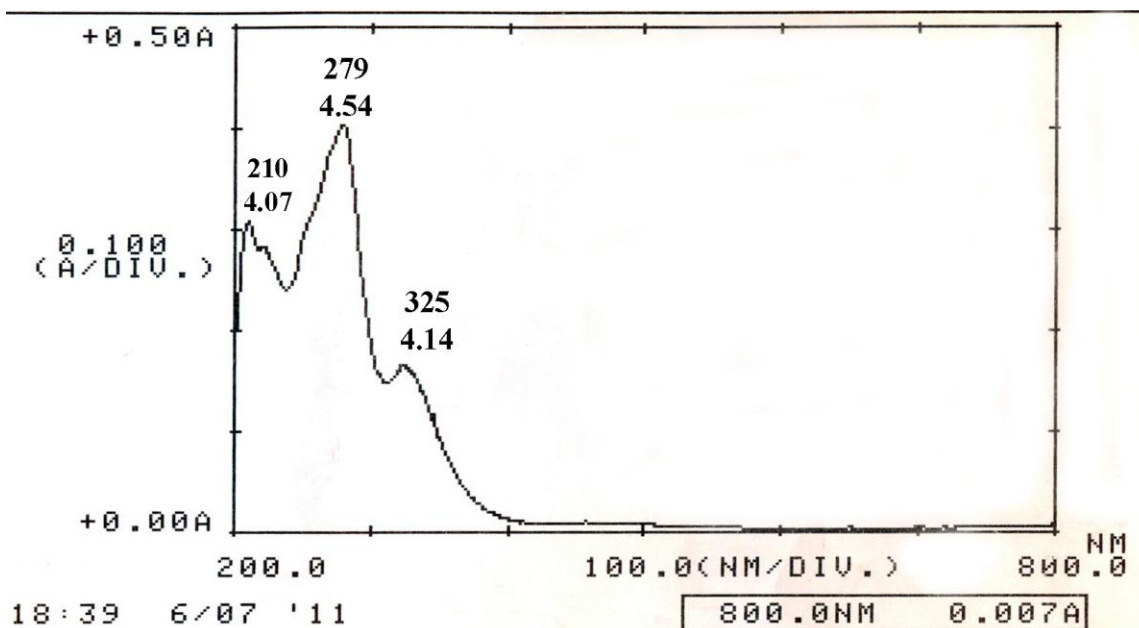


Figure 76 UV Spectrum of compound CAL10 (methanol)

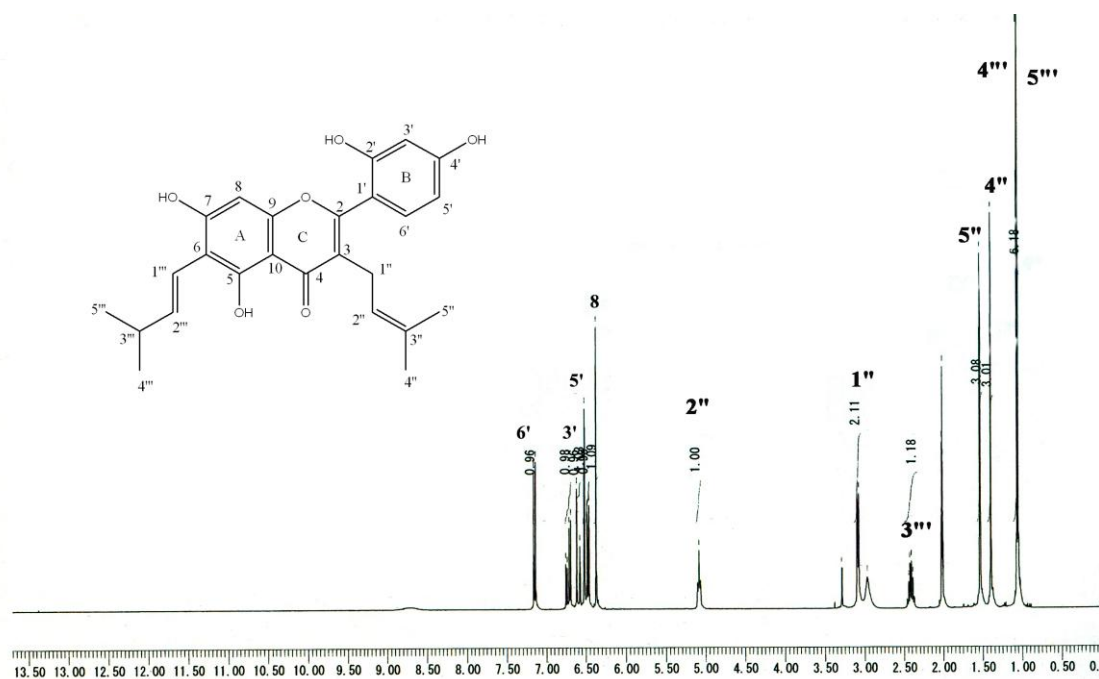
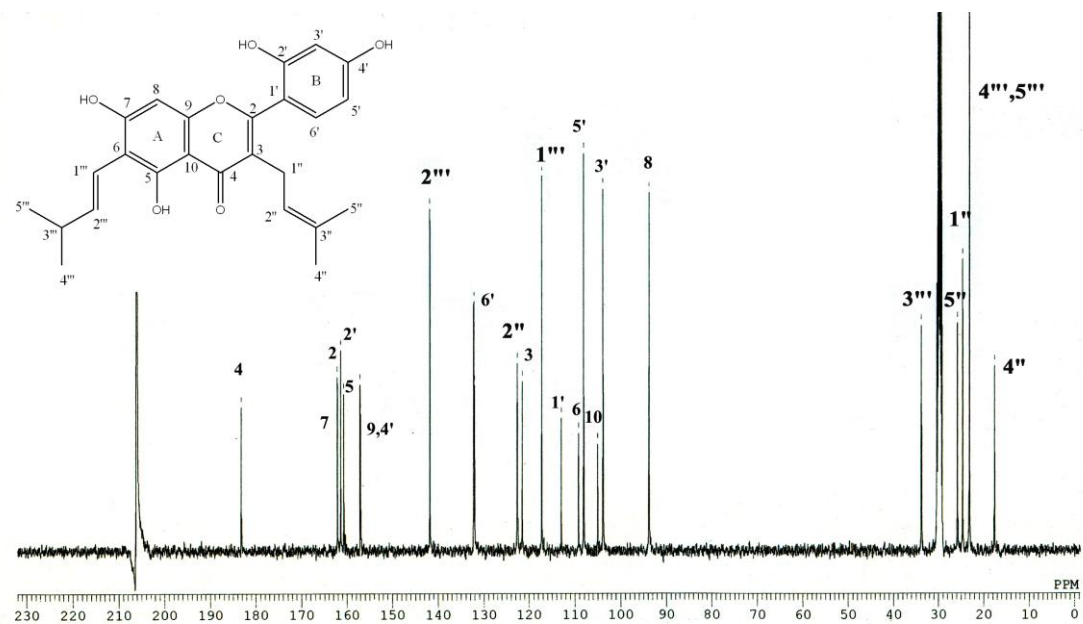
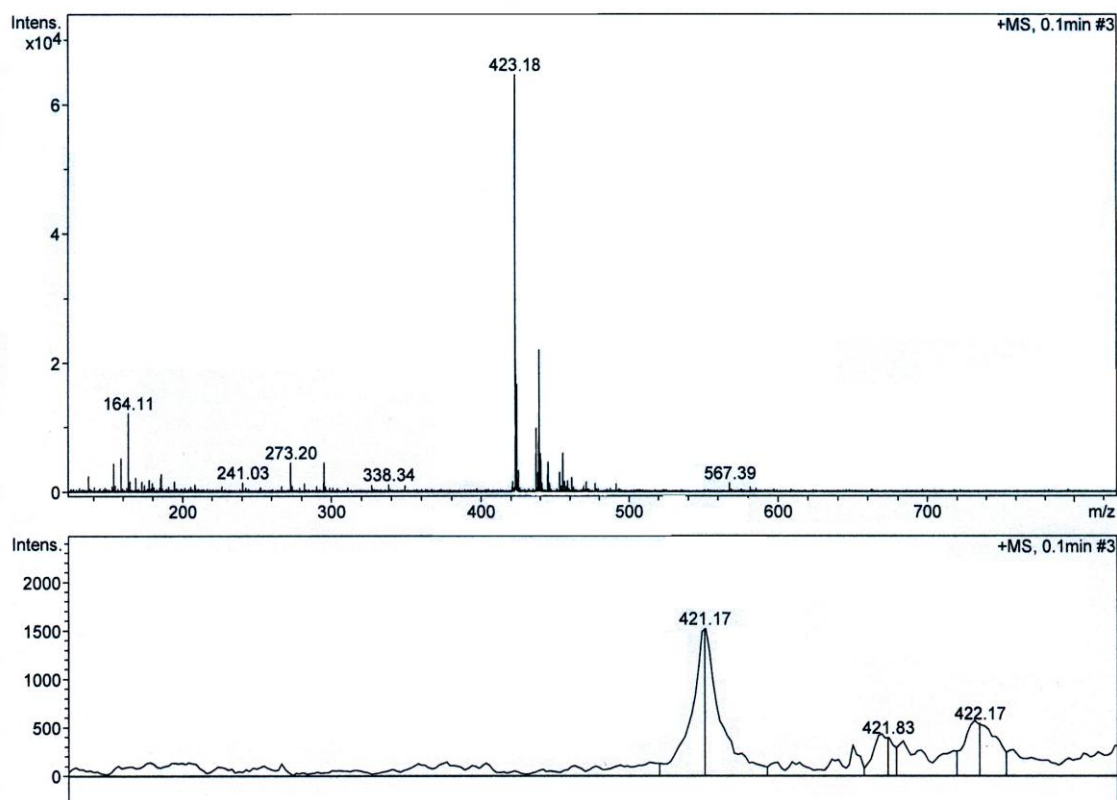


Figure 77 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL10 (acetone-*d*<sub>6</sub>)



**Figure 78**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL10 (acetone- $d_6$ )



**Figure 79** ESI Mass Spectrum of compound CAL11



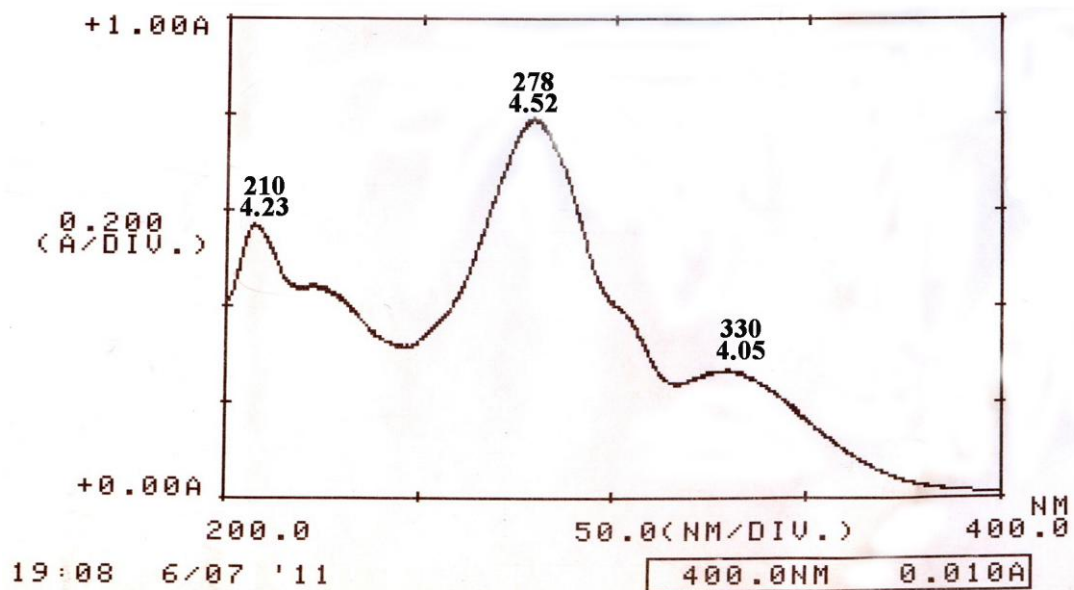


Figure 80 UV Spectrum of compound CAL11 (methanol)

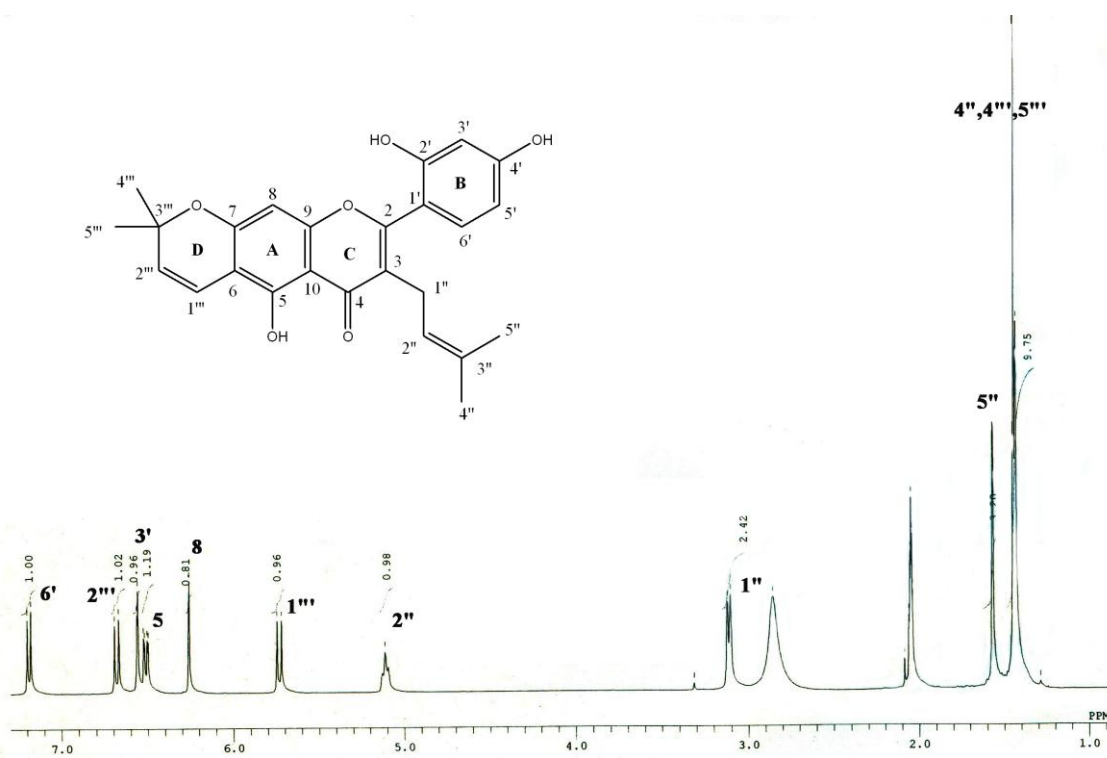


Figure 81 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL11 (acetone-*d*<sub>6</sub>)

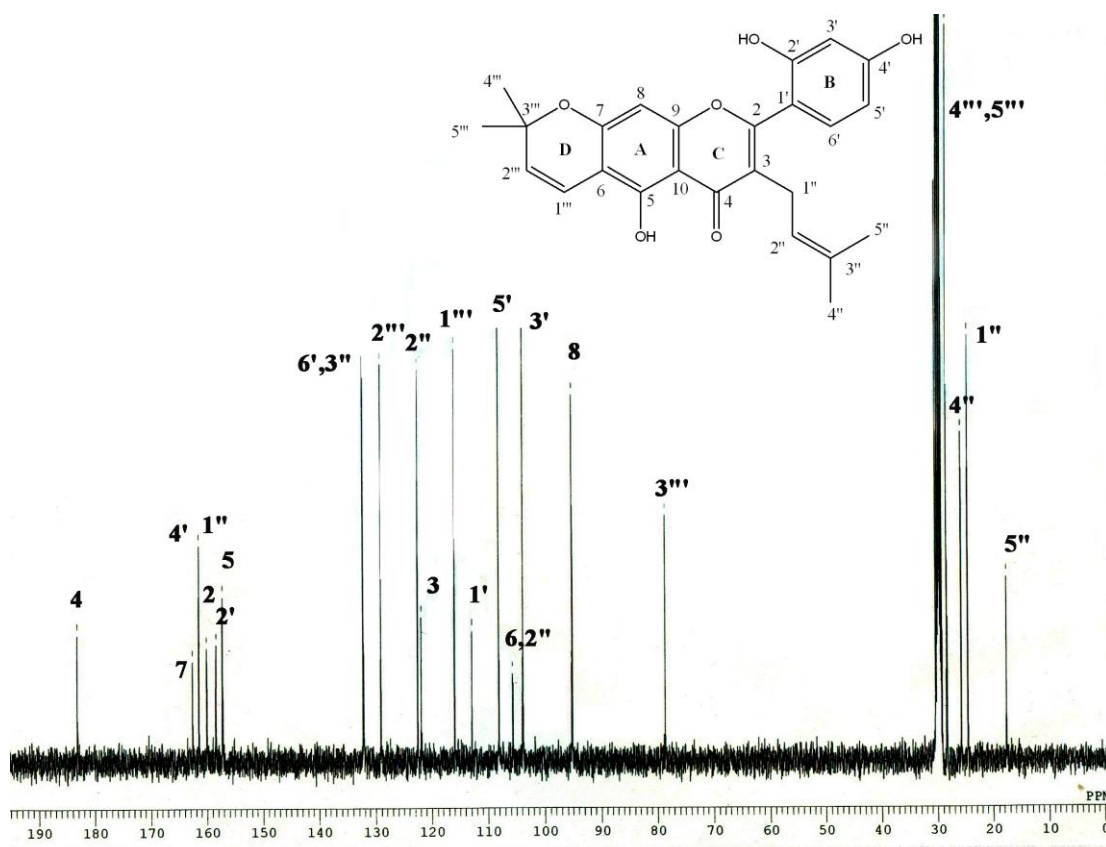


Figure 82  $^{13}\text{C-NMR}$  (100 MHz) Spectrum of compound CAL11 (acetone- $d_6$ )

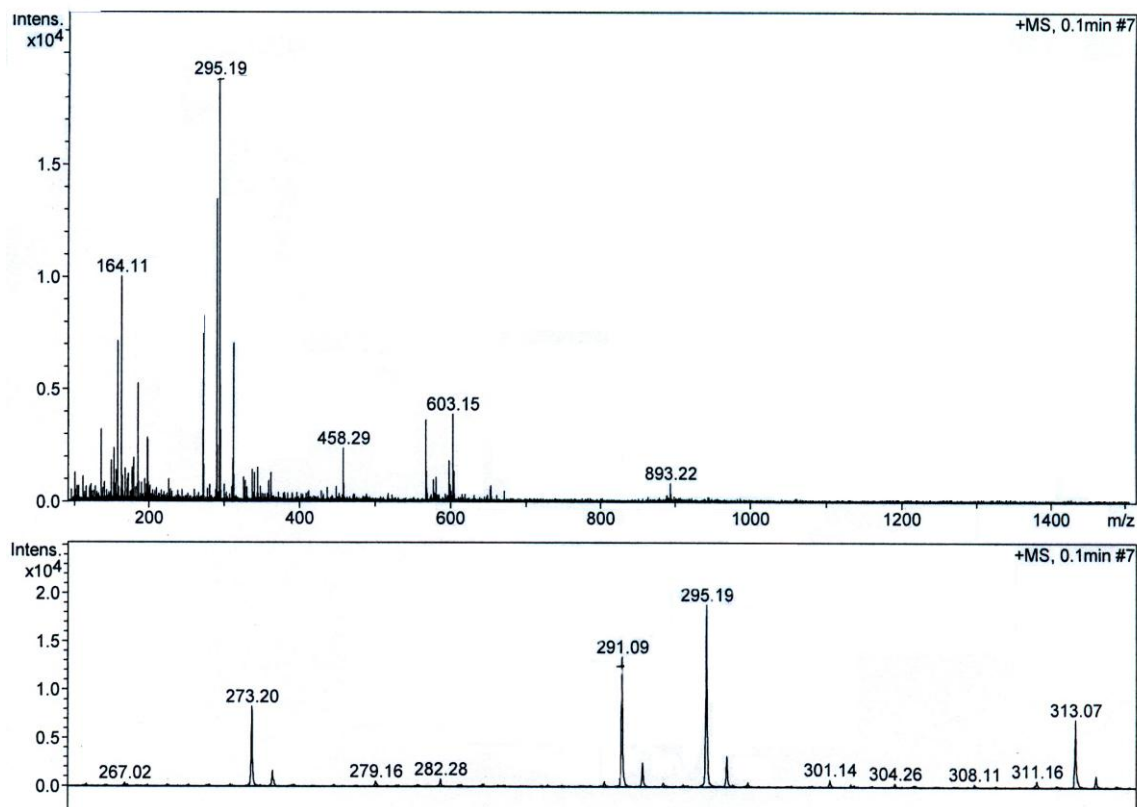


Figure 83 ESI Mass Spectrum of compound CAL12

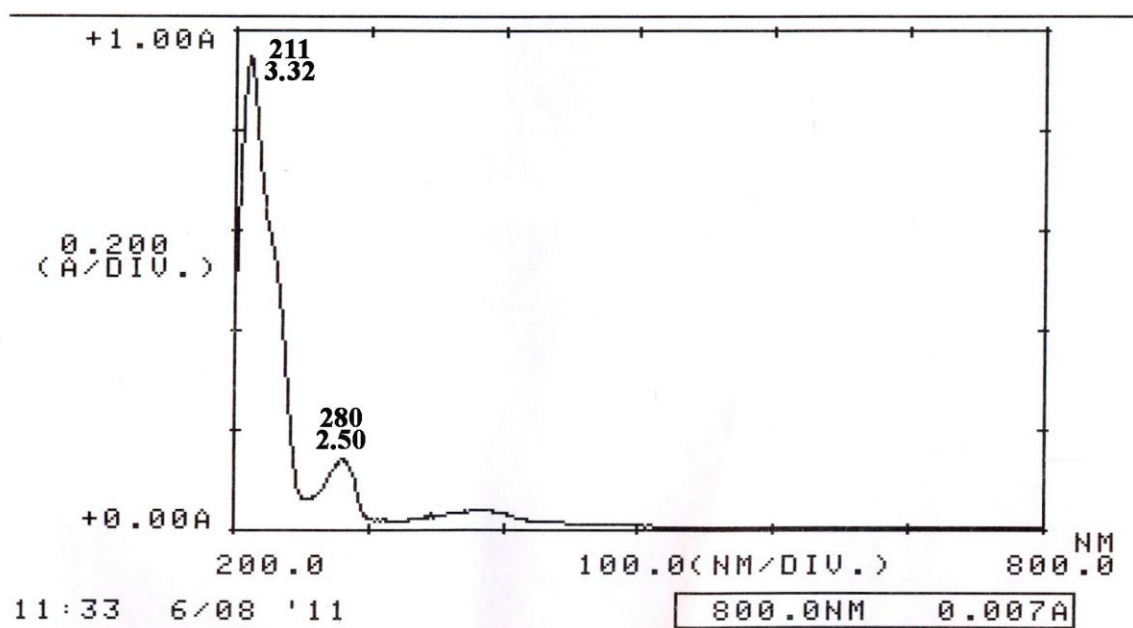


Figure 84 UV Spectrum of compound CAL12 (methanol)

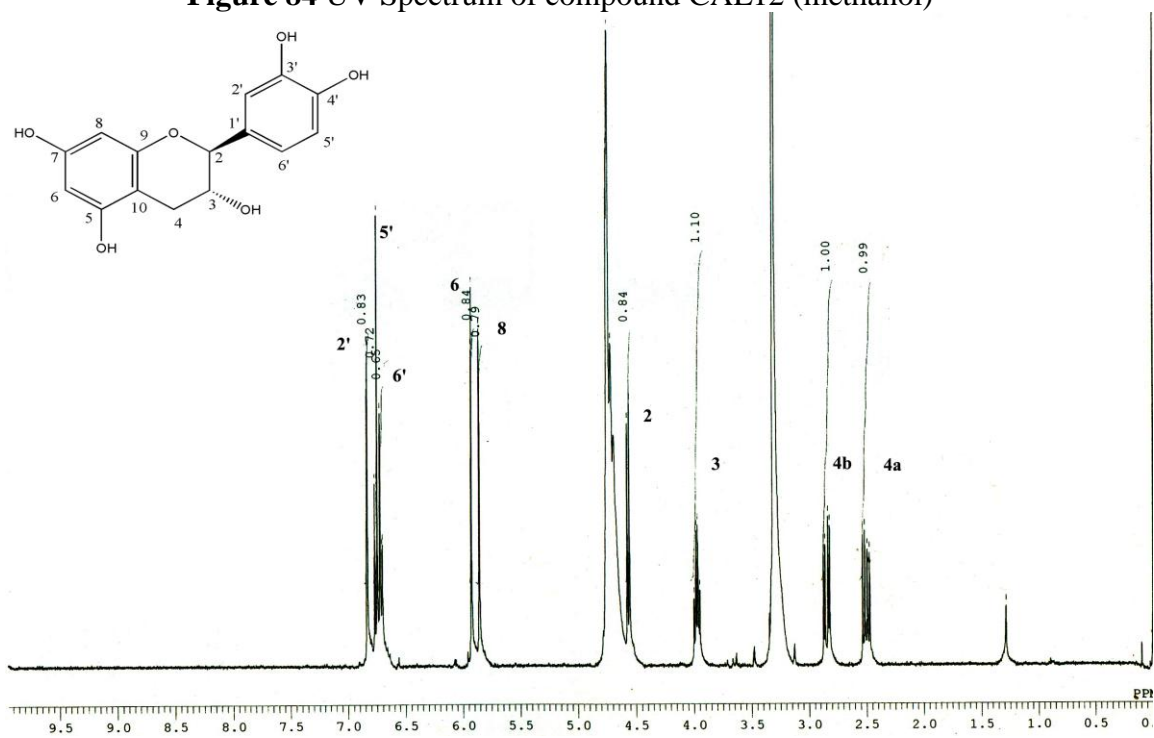
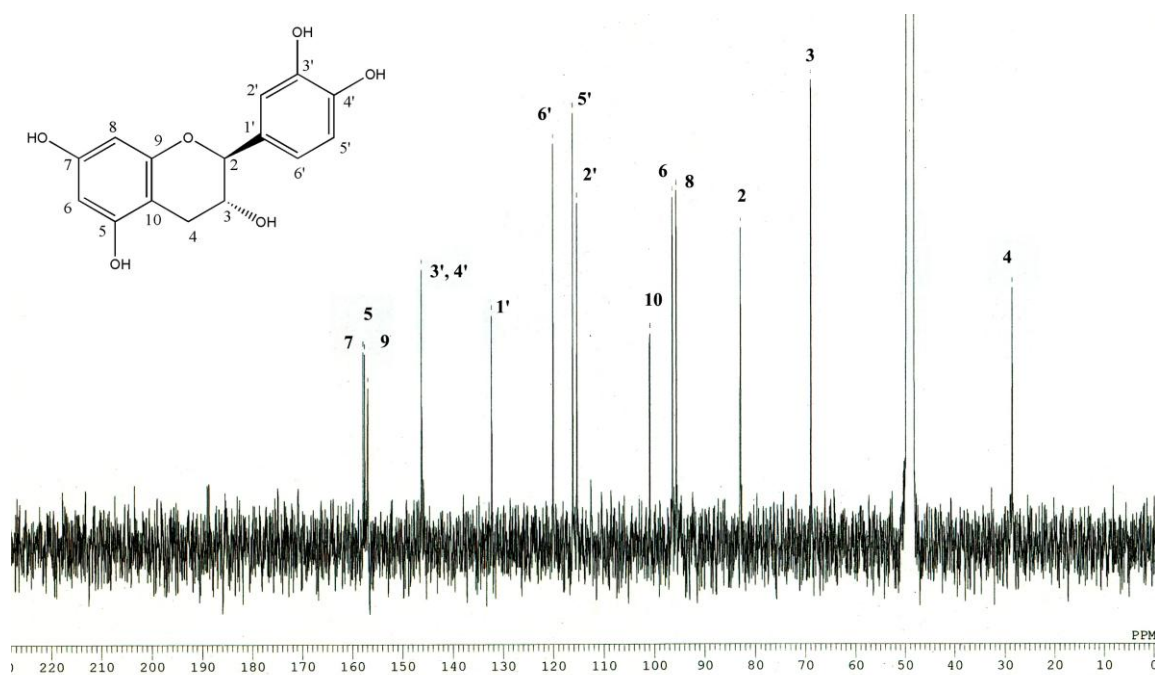
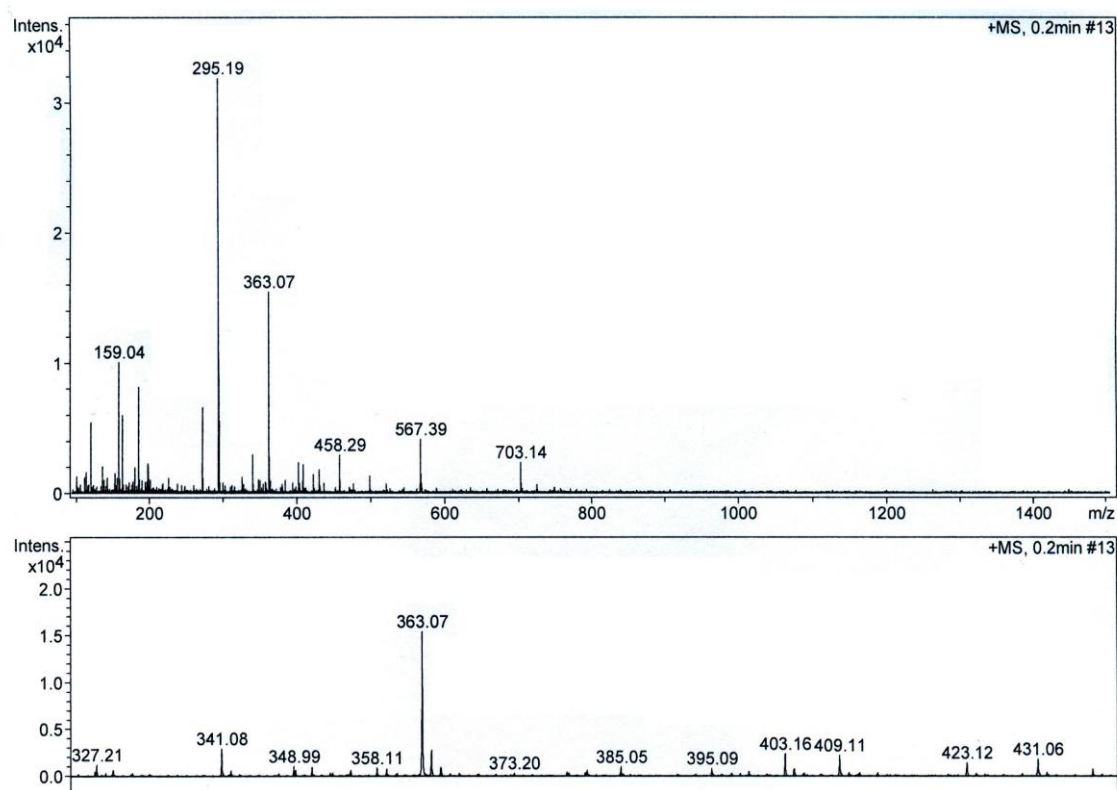


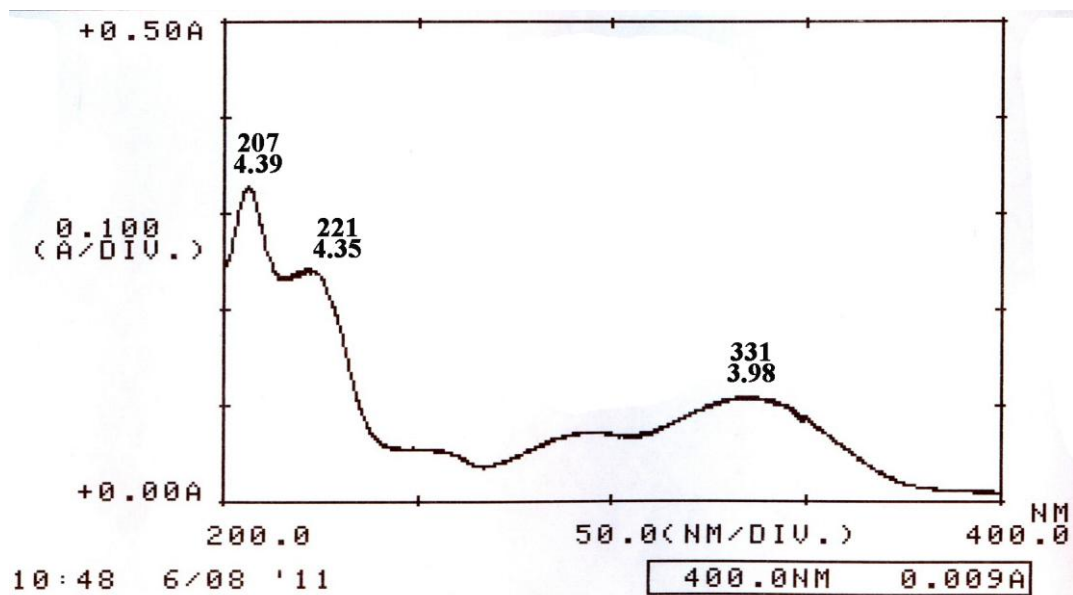
Figure 85 <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL12 (methanol-*d*<sub>4</sub>)



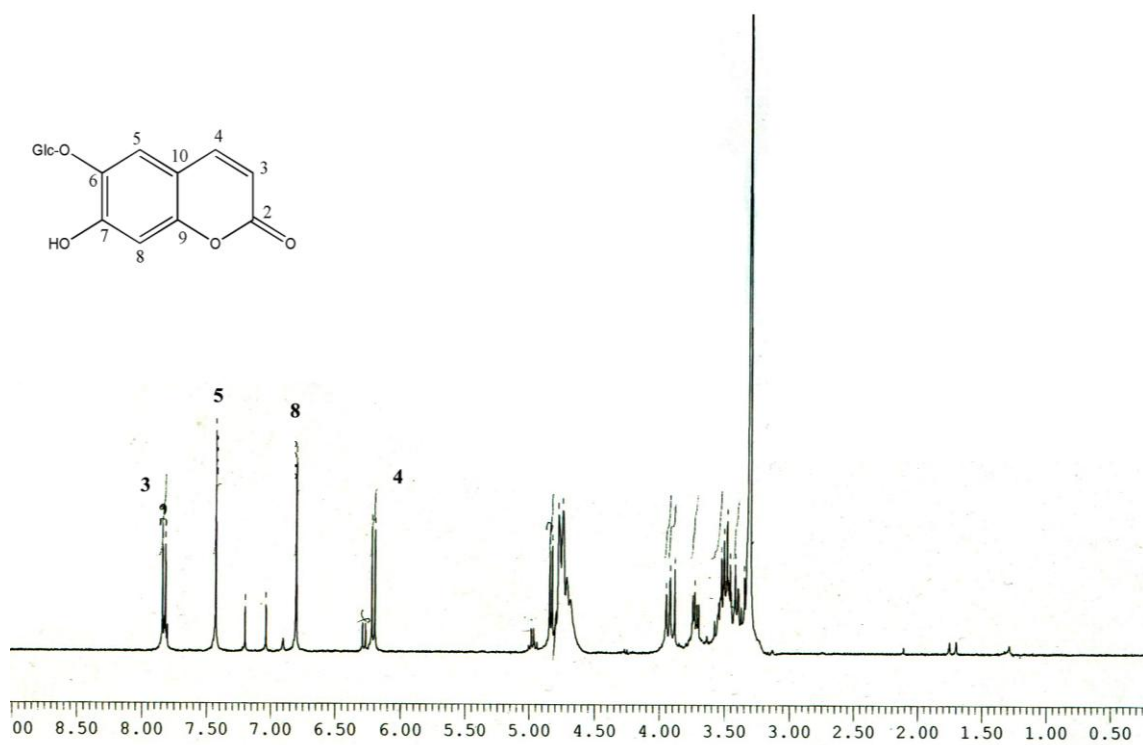
**Figure 86**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL12 (methanol- $d_4$ )



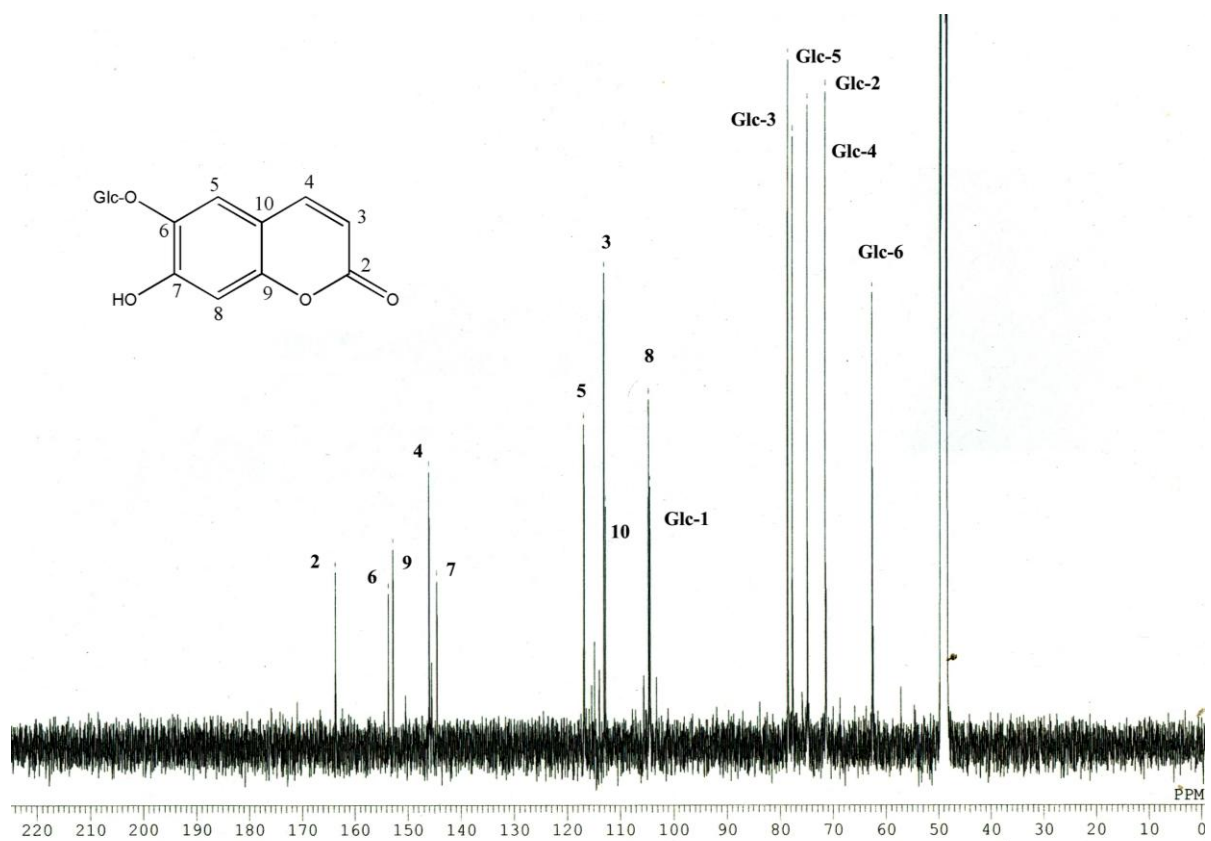
**Figure 87** ESI Mass Spectrum of compound CAL13



**Figure 88** UV Spectrum of compound CAL13 (methanol)



**Figure 89** <sup>1</sup>H-NMR (400 MHz) Spectrum of compound CAL13 (methanol-*d*<sub>4</sub>)



**Figure 90**  $^{13}\text{C}$ -NMR (100 MHz) Spectrum of compound CAL13 ( $\text{methanol-}d_4$ )

## VITA

Mrs. Suthira Maneechai was born on May 18, 1977 in Udonthanee, Thailand. She received her Bachelor's degree of Science in Biology in 2000 from the Faculty of Science, Chiang Mai University, Thailand. She received her Master's degree of Science in Botany in 2003 from the Faculty of Science, Chulalongkorn University, Thailand. After graduation, she worked as a lecturer in Ubon Ratchathani Ratchaphat University.

### Publications

Maneechai, S., Likhitwitayawuid, K., Sritularak, B., Palanuvej, C., Ruangrunsi, N. and Sirisa-ard, P. 2009. Quantitative Analysis of Oxyresveratrol Content in *Artocarpus lakoocha* and 'Puag-Haad'. Medical Principal and Practice. 18: 223–227.