A NOVEL ROUTE TO SYNTHESIS OF LEAD GLYCOLATE AND PEROVSKITE LEAD TITANATE, LEAD ZIRCONATE, AND LEAD ZIRCONATE TITANATE (PZT) VIA SOL-GEL PROCESS

Nuchnapa Tangboriboon

A Dissertation Submitted in Partial Fulfilment of the Requirements
for the Degree of Philosophy

The Petroleum and Petrochemical College, Chulalongkorn University
in Academic Partnership with

The University of Michigan, The University of Oklahoma,
and Case Western Reserve University

2006

ISBN 974-9990-14-5

Thesis Title:

A Novel Route to Synthesis of Lead Glycolate and Perovskite

Lead Titanate, Lead Zirconate, and Lead Zirconate Titanate

(PZT) via Sol-Gel Process

By:

Nuchnapa Tangboriboon

Program:

Polymer Science

Thesis Advisors:

Assoc. Prof. Sujitra Wongkasemjit

Prof. Alexander M. Jameison

Assoc. Prof. Anuvat Sirivat

Accepted by the Petroleum and Petrochemicai College, Chulalongkorn University, in partial fulfilment of the requirements for the Degree of Doctor of Philosophy.

> Nantrya Yanumit College Director (Assoc. Prof. Nantaya Yanumet)

Thesis Committee:

(Assoc. Prof. Nantaya Yanumet)

(Prof. Alexander M. Jamieson)

(Assoc. Prof. Anuvat Sirivat)

(Assoc. Prof. Sujitra Wongkasemjit)

Ratoma Rujiravanit

(Asst. Prof. Ratana Rujirawanit)

(Assoc. Prof. Supa Hannongbua)

Lupa Hammingbue_

ABSTRACT

4592003063:

Polymer Science Program

Nuchnapa Tangboriboon: A Novel Route to Synthesis of Lead

Glycolate and Perovskite Lead Titanate, Lead Zirconate, and Lead

Zirconate Titanate (PZT) via Sol-Gel Process

Thesis Advisors: Assoc. Prof. Sujitra Wongkasemjit, Prof.

Alexander M. Jameison, and Assoc. Prof. Anuvat Sirivat, 160 pp.

ISBN 974-9990-14-5

Keywords:

Phase Transformations / Sol-Gel Process / Dielectric Materials/

Perovskite Structures

The reaction of lead acetate trihydrate Pb(CH₃COO)₂.3H₂O and ethylene glycol, using triethylenetetramine (TETA) as a catalyst, provides, in one step, an access to a polymer-like precursor of lead glycolate [-PbOCH₂CH₂O-] via oxide one pot synthesis (OOPS). The lead glycolate precursor has superior electrical properties than lead acetate trihydrate, suggesting that the lead glycolate precursor can possibly be used as a starting material mixed with other precursors such as titanium glycolate and sodium tris (glycozirconate). The lead-titanium glycolates underwent the sol-gel transition through the formation of Pb-O-Ti bonds, for use as electronic-grade PbTiO₃. Perovskite lead zirconate was synthesized, using lead glycolate and sodium tris (glycozirconate) as the starting precursors. Moreover, the perovskite lead zirconate titanate was synthesized, using lead glycolate, sodium tris (glycozirconate), and titanium glycolate. The obtained molar ratio Pb:Zr:Ti of Pb(Zr_xTi_{1-x})O₃: Pb(Zr_{0.5}Ti_{0.5})O₃ was 1:0.5:0.5. Our synthesized material appears to be a suitable candidate for producing electrical and semiconducting ceramics, viz. ferroelectric, anti-ferroelectric, and piezoelectric materials.

บทคัดย่อ

นุชนภา ตั้งบริบูรณ์ : นวัตกรรมของกระบวนการสังเคราะห์เลดไกลโคเลคและ โครงสร้างแบบเพอรอฟสไกต์ของเลดไททาเนต เลดเซอร์โคเนต และ เลดเซอร์โคเนตไททาเนต ผ่านกระบวนการโซล-เจล (A Novel Route to Synthesis of Lead Glycolate and Perovskite Lead Titanate, Lead Zirconate, and Lead Zirconate Titanate (PZT) via Sol-Gel Process) อ. ที่ปรึกษา: รองศาสตราจารย์. คร. สุจิตรา วงศ์เกษมจิตต์, ศาสตราจารย์ คร. อเล็กซานเดอร์ เอ็ม เจมิสัน และ รองศาสตราจารย์. คร. อนุวัฒน์ ศิริวัฒน์ 160 หน้า ISBN 974-9990-14-5

งานวิจัยนี้ได้ศึกษาการเกิดปฏิกิริยาระหว่างเลดอะซีเตดไตรไฮเดรต และเอทิลีนไกลดอล โดยใช้สารไตรเอทิลีนเตตระมีนเป็นคะตะลีสต์เพื่อใช้ในการผลิตสารตั้งต้นโพลิเมอร์คือเลกไกลโค เลด [-PBOCH₂CH₂O-] ที่สำคัญชนิคหนึ่งต่อขบวนการนำไปใช้ในการสังเคราะห์ สารไคอิเลก ตริกชนิคต่างๆ ได้แก่ สารเฟอร์โรไดอิเลคตริก สารแอนไทเฟอร์โรอิเลคตริก และสารเพียโช ใดอิเลคตริก สารตั้งต้นสามารถสังเคราะห์ได้ จากจากสารประกอบออกไซค์เพียงขั้นตอนเคียวที่ เรียกว่า Oxide One Pot Synthesis (OOPS) ทำให้ได้สารตั้งต้นเลดไกลโคเลคมีสมบัติทางไฟฟ้า ที่ดีกว่าเลคอะซีเตคไตรไฮเครต นอกจากนี้ยังสามารถนำไปใช้ผสมกับสารตั้งต้นชนิคอื่นๆ เช่น ไท ทาเนียมไกลโกเลดและโซเคียมทรีสไกลโกเซอร์โกเนตสามารถสังเกราะห์เลดไททาเนตได้โดยผ่าน ระหว่างเลคไกลโคเลคและไททาเนียมไกลโคเลค กระบวนการ โซล-เจลทรานสิชั่น โครงสร้างพันธะเป็น Pb-O-Ti สำหรับใช้ในการผลิตวัสคุไคอิเลคตริกเลคไททาเนต งานวิจัยนี้ สามารถเตรียมเพอรอพสไกต์เลคเซอร์โคเนตโคยการใช้ เลคไกลโคเลคและโซเคียมทรีสไกลโคเลค เซอร์โกเนตเป็นสารตั้งต้นผ่านกระบวนการโซล-เจลเช่นเคียวกันนอกจากนี้เลคเซอร์โกเนตไททา เนตสามารถสังเคราะห์ได้จากสารตั้งต้นเลคไกลโคเลค โซเคียมทรีสไกลโคเซอร์โคเนตและไททา เนียมใกลโคเลค ด้วยอัตราส่วนโมลระหว่าง Pb:Zr:Ti ของ Pb(Zr_XTi_{1-X})O₃: Pb(Zr_{0.5}Ti_{0.5})O₃ คือ 1:0.5:0.5 สารสังเคราะห์ที่ผลิตได้สามารถนำไปใช้ในการผลิตวัสคุไดอิเลคตริกและวัสคุเซรา มิกส์กึ่งตัวนำ ได้แก่ สารเฟอร์โรอิเลคตริก แอนติเฟอร์โรอิเลคตริก และเพียโซอิเลคตริก

ACKNOWLEDGEMENTS

This work would not have been possible without the assistance of the following individuals.

First of all, the author would like to express her appreciation to her advisors, Assoc.Prof. Sujitra Wongkasemjit, Assoc. Prof. Anuvat Sirivat, and Prof. Alexander M. Jamieson, for their kind supports, recommendations, suggestions, opportunities to express her ideas, and a chance to learn how to solve the problems, arrange data, and write the papers.

The author would like to give her thankfulness to Assoc. Prof. Nantaya Yanumet, for providing useful recommendations, kindness, creative comments, and encouragement throughout the course of her work.

The author would like to thank all of dissertation committee for their kind advices.

Special thanks go to all of the Petroleum and Petrochemical College's staff who helped with typing various reports.

The author would like to express her sincere gratitude to Prof. Nontawat Junjareon, the Dean of Engineering Faculty, and her colleagues at Materials Engineering Department for supporting her Ph.D. study. Furthermore, the work would not be carried out successfully without financial supports from: the Postgraduate Education and Research Program in Petroleum and Petrochemical Technology (ADB) Fund, Ratchadapisake Sompoch Fund, Chulalongkorn University and the Faculty of Engineering, Kasetsart University. Thanks are extended to the Department of Materials Engineering, the Chemical Engineering Department, and the Physics Department of Kasetsart University, for X-ray microscan, X-ray diffraction, and the electrical property measurements.

Finally, the author would like to take this opportunity to thank PPC Ph.D. students and all her PPC friends for their friendly assistances, cheerfulness, creative suggestions, and encouragements. The author had the most enjoyable time working with all of them. Also, the author is greatly indebted to her parents and her family for their support, love, constant inspiration and understanding.

TABLE OF CONTENTS

				I	PAGI
	Title P	age		•	ĭ
	Abstra	ct (i	n Engl	ish)	iii
	Abstra	ct (ii	n Thai)	iv
	Ackno	wled	lgemer	nts	v
	Table	of Co	ontents	3	vi
	List of	Tab	les		X
	List of	Figu	res		xii
	Abbrev	riatio	ons		xvii
СП	APTER				
CHA	I	INT	rod	UCTION	1
	п	BA	CKGF	ROUND AND LITERATURE SURVEY	5
		2.1	Synth	nesis of Metal Alkoxide	5
		2.2	Sol-G	del Process of Metal Alkoxide	6
		2.3	Diele	ctric Material Properties	8
		2.4	Litera	ature Review	14
	Ш	EX	PERI	MENTAL	21
		3.1	Mater	rials	21
		3.2	Precu	rsor synthesis	21
		3.3	Instru	ments	24
			3.2.1	FAB ⁺ -MS Spectroscopy (MS)	24
			3.2.2	Elemental Analysis (EA)	24
			3.2.3	Themogravimetric Analysis (TGA)	24
			3.2.4	Nuclear Magnetic Resonance Spectroscopy (NMR)	24
			3.2.5	Fourier Transform Infrared Spectroscopy (FTIR)	25
			3.2.6	X-Ray Diffraction	25

CHAPTER				PAGE
		3.2.7	Scanning Electron Microscopy (SEM)	25
		3.2.8		25
		3.2.9	200 SI SI SI SI	25
		3.2.10) Raman Spectroscopy	26
	3.3		odology	26
		3.3.2	Synthesis of Lead Titanate	26
		3.3.3	Synthesis of Lead Zirconate	26
		3.3.4	Synthesis of Lead Zirconate Titanate	27
IV	EL	ECTRI	CAL PROPERTIES OF A NOVEL LEAD	
	AL	KOXID	E PRECURSOR: LEAD GLYCOLATE	28
	4.1	Abstra	ct	28
	4.2	Introdu	action	29
	4.3	Experi	mental	30
	4.4	Result	s and Discussion	32
	4.5	Conclu	isions	35
	4.6	Ackno	wledgements	35
	4.7	Refere	nces	36
v	A N	OVEL	ROUTE PEROVSKITE LEAD TITANATE	
	FRO	OM LE	AD AND TITANIUM GLYCOLATES VIA	
	SOI	L-GEL	PROCESS	50
	5.1	Abstra	ct	50
	5.2	Introdu	action	51
	5.3	Experi	mental	52
	5.4	Result	s and Discussion	55
	5.5	Conclu	isions	59
	5.6	Ackno	wledgements	59
	5.7	Refere	nces	59

CHAPTER			PAGE
VI	AN	OVEL ROUTE PEROVSKITE LEAD ZIRCONATE	
	FR	OM LEAD GLYCOLATES AND SODIUM TRIS	
	GL	YCO ZIRNATE) VIA SOL-GEL PROCESS	74
	6.1	Abstract	74
	6.2	Introduction	75
	6.3	Experimental	76
	6.4	Results and Discussion	79
	6.5	Conclusions	82
	6.6	Acknowledgements	83
	6.7	References	83
VII		OVEL ROUTE PEROVSKITE LEAD ZIRCONATE	
	TIT	TANATE (PZT) FROM LEAD GLYCOLATES,	
	TIT	CANIUM GLYCOLATE AND SODIUM TRIS GLYCO	O
	ZIF	RCONATE) VIA SOL-GEL PROCESS	99
	7.1	Abstract	99
	7.2	Introduction	100
	7.3	Experimental	101
	7.4	Results and Discussion	106
	7.5	Conclusions	110
	7.6	Acknowledgements	110
	7.7	References	111
VIII	CO	NCLUSIONS AND RECOMMENDATIONS	131
	RE	FERENCES	133
	AP	PENDICES	140
	App	pendix A The FTIR standard patterns of starting	
		materials	140

CHAPTER			PAGE
	Appendix B	The XRD standard peak patterns of	
		synthesis lead glycolate, lead titanate,	
		lead zirconate, and lead zirconate titanate	141
	Appendix C	The SEM microstructure of dried lead	
		zirconate gel and calined at 200°C, 300°C,	
		400°C, 500°C, 600°C, and 700°C for 1h	
		at magnification of 5000	156
	Appendix D	The SEM microstructure of dried lead	
		zirconate titanate gel of PZT 1; PZT 2;	
		PZT3; PZT 4; PZT 5; PZT 6; PZT 7 at	
		magnification of 1500	157
	Appendix E	The electrical property of preparation and	
		calculation	158
	CURRICULU	M VITAE	159

LIST OF TABLES

TABLE	PA	AGE
	CHAPTER IV	
4.1	The proposed structures and fragments pattern of lead glycolate	39
4.2	The percentages of C and H present in the synthesized lead glycolate	40
4.3	The physical properties of lead glycolate and lead acetate trihydrate	41
4.4	The electrical properties of lead glycolate and lead acetate trihydrate	42
	CHAPTER V	
5.1	The proposed structures and fragmentation pattern of dried lead	
	titanate gel by mass spectroscopy	62
5.2	The percentages of chemical compositions of calcined powder and	
	mole ratio of Pb:Ti.	63
5.3	The electrical properties of lead glycolate, titanium glycolate and	
	lead titanate synthesized were measured at 1000 Hz. and at 25°C.	64
	CHAPTER VI	
6.1	The percentages of chemical compositions of dried lead zirconate	
	gel and calcined lead zirconate samples with the Pb:Zr mole ratio.	87
6.2	The proposed structure and the percentage of carbon content of lead	
	zirconate.	88
6.3	The dielectric properties (1000 Hz., 27°C) and DC electrical conduc	
0.0	tivity of lead glycolate, sodium tris (glycozirconate), dried lead	
	zirconate gel, and calcined lead zirconate samples.	89
	Ziroonate get, and eaterned read ziroonate samples.	0,
	CHAPTER VII	
7.1	The percentages of PbO, ZrO2, and TiO2 in dried lead zirconate	
	titanate gels: PZT1; PZT2; PZT3; PZT4; PZT5; PZT6; PZT7 and	
	calcined lead zirconate titanate PZT4 Pb(Zr _x Ti _{1-x})O ₃ of various	

FABLE		PAGE
	calcination temperatures with corresponding Pb:Zr:Ti	
	mole ratios.	115
7.2	The proposed structure and the percentage of carbon content in	
	Dried lead zirconate titanate PZT4 gel.	116
7.3	The dielectric properties (1000 Hz., 27°C) and DC electrical	
	conductivity of lead glycolate precursor, sodium tris (glycol	
	zirconate) precursor, titanium glycolate precursor, lead zirconate	
	titanate of Pb(Zr _x Ti _{1-x})O ₃ dried gels: PZT1; PZT2; PZT3; PZT4;	
	PZT5; PZT6; PZT7 and calcined lead zirconate titanate PZT4 of	
	various calcination temperatures.	117

LIST OF FIGURES

FIGURE		PAGE
	CHAPTER I	
1.1	Classification of technologically useful electronic materials	3
	CHAPTER II	
2.1	The possible structures of gel formation	6
2.2	Idealized Perovskite structure	8
2.3	The Curie temperature of dielectric materials	9
2.4	Structural phase diagram of PZT-lead zirconate titanate	10
2.5	Piezoelectric coupling coefficient and chemical composition	11
2.6	The hysteresis loop of papraelectric and ferroelectric	12
	CHAPTER IV	
4.1	FTIR spectra of the lead acetate trihydrat and lead glycolate	
	precursor	43
4.2	¹³ C-Solid state NMR spectrum of the synthesized lead glycolate	44
4.3	TGA curve of lead glycolate	45
4.4	The SEM micrographs of lead acetate trihydrate (figures a-d)	
	and lead glycolate (figures a'-d') pyrolyzed at: a) room tempe	
	rature; b) 400°C; c) 600°C; and d) 800°C	46
4.5	The XRD patterns of lead acetate trihydrate (figures a-d) and	
	lead glycolate (figures a'-d') at: (a) room temp; (b) 400°C;	
	(c) 600°C; and (d) 800°C	47
4.6	The dielectric constants of lead acetate trihydrate and lead	
	glycolate precursor as functions of calcined temperature	
	at room temp;400°C;600°C; and 800°C	48
4.7	The conductivity values of lead acetate trihydrate and lead	
	glycolate precursor as functions of calcined temperature at	
	room temp; 400°C;600°C; and 800°C	49

FIGURE		PAGE
	CHAPTER V	
5.1	FTIR spectra of lead glycolate, titanium glycolate and	
	dried lead titanate gel	65
5.2	FTIR spectra of lead titanate samples prepared at various	
	molar Pb/Ti ratios: 1:1; 1:2; 1:3; 1:4; 2:1; 3:1; and 4:1	66
5.3	DTA and TGA thermograms of dried lead titanate gel	67
5.4	XRD diffraction patterns of calcined lead titanate powders at:	
	300°C; 400°C; 600°C, 800°C; and 1000°C for 1; 3 and 5h	
	(* tetragonal perovskite, Δ pyrochlore, + cubic perovskite)	68
5.5	Raman spectra of lead titanates at: a)300°C, 400°C, 600°C,	
	800°C, and 1000°C for 1 h; and b)1000°C for 1,3,5 h	69
5.6	SEM micrographs (magnification = 500, size bar = 50 μ m)	
	showing the phase transformation of lead titanate calcined	
	for 1 (column 1); 3(column 2) and (column 3) h at: a) 300°C;	
	b) 800°C; and c) 1000°C	70
5.7	SEM micrographs (magnification = $10,000$, size bar = $1 \mu m$)	
	showing agglomeration and crystal growth of lead titanate	
	powders calcined from 25°C (sphere shape) to 1000°C	
	(fiberlike): a) 25°C; b) 300°C; c) 400°C; d) 600°C; e) 800°C;	
	and f) 1000°C for 3 h	71
5.8	The dielectric constant and dielectric loss tangent (tan δ) vs.	
	frequency of lead glycolate, titanium glycolate and dried lead	
	titanate gel measured at 25°C	72
5.9	The dielectric constant and dielectric loss tangent (tan δ) measured	
	at 25 °C of the lead titanate powders calcined at 300°C, 400°C,	
	$600^{\rm o}{\rm C},800^{\rm o}{\rm C}$ and $1000^{\rm o}{\rm C}$ for 3 h. from 1000 to 1000000 Hz.	73

CHAPTER VI

6.1 FTIR spectra of lead glycolate precursor, sodium tris

FIGURE		
	(glycozirconate) precursor, and dried lead zirconate gel at the	
	mole Pb/Zr ratio 1:1	90
6.2	FTIR spectra of the lead zirconate gels at the mole ratio 1:1;	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
0.2	1:2; 1:3; 1:4; 2:1; 3:1; and 4:1	91
6.3	The TGA-DSC thermograms of dried lead zirconate gel from	71
0.5	25°C to 1000°C	92
6.4	Raman spectra of lead glycolate precursor, sodium tris	
	(glycozirconate) precursor, and dried lead zirconate gel	93
6.5	Raman spectra of dried lead zirconate gel, calcined lead	
	zirconates at 200°C; 300°C; 400°C; 500°C; 600°C; 700°C	
	for 1h. from 200-1000 cm ⁻¹	94
6.6	SEM micrographs showing the phase transformation of dried	
	lead zirconate gel and calcined lead zirconates at: a) 25°C;	
	b)200°C; c).300°C; d) 400°C; e). 500°C; f) 600°C; and	
	g) 700°C for 1 (column1), 2 (column2), and 3 (column3) h	
	at the magnification of 1,500	95
6.7	XRD diffraction patterns of dried lead zirconate gel, and	
	calcined lead zirconates at calcination temperatures of 200°C;	
	300°C; 400°C; 500°C; 600°C; 700°C for durations of 1, 2, and 3 h.	96
6.8	Dielectric constant and tan delta of lead glycolate precursor,	
	sodium tris (glycozirconate), and dried lead zirconate gel vs.	
	frequency measured at room temperature	97
6.9	Dielectric constant and tan delta of calcined temperature lead	
	zirconates at 200°C, 300°C, 400°C, 500°C, 600°C, and 700°C	
	for 1h.at various frequencies	98
200	CHAPTER VII	
7.1	FTIR spectra: lead glycolate precursor; titanium glycolate	
	precursor; sodium tris (glycozirconate) precursor, dried PbTiO ₃	4004.4950
	gel, dried PbZrO ₃ gel, and dried lead zirconate titanate PZT4 gel	118

FIGUR	FIGURE		
7.2	FTIR spectra of the dried lead zirconate titanate gels at		
	various mole ratios Pb:Zr:Ti: PZT 1 (1:0.2:0.8); PZT 2		
	(1:0.3:0.7); PZT 3 (1:0.4:0.6); PZT 4 (1:0.5:0.5); PZT 5		
	(1:0.6:0.4); PZT 6 (1:0.7:0.3); PZT 7 (1:0.8:0.2)	119	
7.3	Raman spectra of lead glycolate precursor; sodium tris		
	(glycozirconate) precursor; titanium glycolate precursor;		
	dried PbZrO3 gel; dried PbTiO3 gel, and dried PZT4 gel	120	
7.4	Raman spectra of PZT1; PZT2; PZT3; PZT4; PZT5; PZT6;		
	PZT7	121	
7.5	Raman spectra of dried lead zirconate titanate PZT4 gel, and		
	calcined lead zirconate titanates at 200°C; 300°C; 400°C;		
	500°C; 600°C; 700°C; 800°C for 1h	122	
7.6	TGA-DSC thermograms of dried lead zirconate titanate		
	PZT4 gel from 25°C to 1200°C	123	
7.7	XRD diffraction patterns of dried lead zirconate titanate		
	Pb(Zr _x Ti _{1-x})O ₃ gels: PZT1; PZT2; PZT3; PZT4; PZT5;		
	PZT6; PZT7	124	
7.8	XRD diffraction patterns of calcined lead zirconate titanate		
	powders (PZT4) at 200°C; 300°C; 400°C; 500°C; 600°C;		
	700°C; 800°C for durations of 1, 2, and 3 h	125	
7.9	¹³ C-Solid state NMR spectra of the synthesized lead glycolate		
	precursor, sodium tris (glycozirconat) precursor, titanium		
	glycolate precursor, dried PbZrO3 gel, dried PbTiO3 gel, and		
	dried PZT4 gel	126	
7.10	SEM micrographs showing microstructure of dried PZT4 gel		
	and calcined PZT4 at: a) 25°C; b) 200°C; c) 300°C;		
d) 400°C; e). 500°C; f) 600°C; g) 700°C; and h) 800°C		
	for 1 h at the magnification of 350	127	
7.11	Dielectric constant and tan delta of lead glycolate precursor,		

FIGURE		
	sodium tris(glycozirconate) precursor, dried lead zirconate	
	gel, dried lead titanate gel, and dried lead zirconate titanate	
	(PZT4) gel vs. frequency measured at room temperature	128
7.12	Dielectric constant and tan delta of lead zirconate titanate	
	Pb(Zr _x Ti _{1-x})O ₃ dried gels: PZT1 ; PZT2; PZT3; PZT4; PZT5;	
	PZT6; PZT7 vs. frequency measured at room temperature	129
7.13	Dielectric constant and tan delta of calcined lead zirconate	
	titanate Pb(Zr _x Ti _{1-x})O ₃ (PZT4) vs. calcination temperature	
	(200°C, 300°C, 400°C, 500°C, 600°C, 700°C and 800°C)	
	for 1h. at various frequencies	130

ABBREVIATIONS

¹³ C NMR	Carbon Nuclear Magnetic Resonance Spectroscopy
EA	Elemental Analysis
EG	Ethylene Glycol
FAB ⁺ -MS	Fast Atomic Bombardment Mass Spectroscopy
FTIR	Fourier Transform Infrared Spectroscopy
OOPS	Oxide One Pot Synthesis
SEM	Scanning Electron Microscope
SOL-GEL	Sol Gel Process
TETA	Triethylenetetramine
TGA Thermal Gravimetric Ana	
XRD	X-ray Diffraction Spectroscopy