พฤกษเคมีและฤทธิ์ทางชีวภาพของลำต้นตากวางและม้ากระทืบโรง

นายปฐม โสมวงศ์

วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรดุษฎีบัณฑิต สาขาวิชาเภสัชเวท ภาควิชาเภสัชเวทและเภสัชพฤกษศาสตร์ คณะเภสัชศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย ปีการศึกษา 2555 ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย บทคัดย่อและแฟ้มข้อมูลฉบับเต็มของวิทยานิพนธ์ตั้งแต่ปีการศึกษา 2554 ที่ให้บริการในคลังปัญญาจุฬาฯ (CUIR)

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PHYTOCHEMISTRY AND BIOACTIVITIES OF SALACIA VERRUCOSA AND FICUS FOVEOLATA STEMS

Mr. Pathom Somwong

A Dissertation submitted in Partial Fulfillment of the Requirements for the Degree of Doctor of Philosophy Program in Pharmacognosy Department of Pharmacognosy and Pharmaceutical Botany Faculty of Pharmaceutical Sciences Chulalongkorn University Academic Year 2012 Copyright of Chulalongkorn University

PHYTOCHEMISTRY AND BIOACTIVITIES OF SALACIA
VERRUCOSA AND FICUS FOVEOLATA STEMS
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ปฐม โสมวงศ์: พฤกษเคมีและฤทธิ์ทางชีวภาพของลำด้นตากวางและม้ากระทืบโรง (PHYTOCHEMISTRY AND BIOACTIVITIES OF *SALACIA VERRUCOSA* AND *FICUS FOVEOLATA* STEMS) อ. ที่ปรึกษาวิทยานิพนธ์หลัก : รศ.คร.รุทธ์ สุทธิศรี, 262 หน้า.

การศึกษาองค์ประกอบทางเคมีของลำด้นตากวาง (Salacia verrucosa Wight) วงศ์ Celastraceae สามารถแยกสารใหม่ในกลุ่ม 1,3-diketofriedelane triterpeneได้ 1 ชนิดคือ 21α-hydroxyfriedelane-1,3dione และที่เคยมีรายงานแล้ว 3 ชนิดคือ friedelane-1,3-dione, 26-hydroxyfriedelane-1,3-dione และ 30 -hydroxyfriedelane-1,3-dione, สารในกลุ่ม friedelane triterpene 3 ชนิดคือ friedelin, kokoonol และ 21α -hydroxyfriedelan-3-one และ สารในกลุ่ม oleanane triterpene 1 ชนิดคือ 3β,22α-dihydroxyolean-12-en-29-oic acid สำหรับการศึกษาองค์ประกอบทางเคมีของลำด้นม้ากระทีบโรง [*Ficus foveolata* (Wall. ex Miq.) Miq.] วงศ์ Moraceae สามารถแยกสารใหม่ในกลุ่ม eudesmane sesquiterpene ได้ 2 ชนิดคือ foveolide A และ foveoeudesmenone และที่เคยมีรายงานแล้ว 2 ชนิดคือ 4(15)-eudesmene-1β,6α-diol และ 4(15)-eudesmene-1β,5α-diol, สารใหม่ในกลุ่ม sesquiterpenoid dimer 1 ชนิดคือ foveolide B, สารใหม่ในกลุ่ม phenolic 1 ชนิดคือ foveospirolide และที่เคยมีรายงานแล้ว 1 ชนิดคือ ethyl rosmarinate นอกจากนี้ยังพบสารในกลุ่ม triterpene 3 ชนิดคือ friedelin, taraxerol และ betulin พิสูจน์โครงสร้างทาง เคมีขององค์ประกอบของพืชเหล่านี้โดยอาศัยการวิเคราะห์เชิงสเปกตรัมด้วย UV, IR, MS และ NMR ร่วมกับการเปรียบเทียบข้อมูลที่เคยมีรายงานแล้ว

สาร friedelane-1,3-dione มีความเป็นพิษระดับสูงต่อเซลล์มะเร็งลำใส้ของมนุษย์ (SW620) ด้วย ค่า IC₅₀ 2.02 ใมโครโมลาร์ ในขณะที่สาร 26-hydroxyfriedelane-1,3-dione และ 21α-hydroxyfriedelan-3-one มีความเป็นพิษในระดับปานกลางต่อเซลล์มะเร็งลำใส้, เซลล์มะเร็งตับ (HepG2) และ เซลล์มะเร็ง กระเพาะอาหาร (KATO-III) นอกจากนี้สาร 26-hydroxyfriedelane-1,3-dione ยังแสดงความเป็นพิษระดับ ปานกลางต่อเซลล์มะเร็งปอด (CHAGO) และเซลล์มะเร็งเด้านม (BT474) สาร foveolide A แสดงความ เป็นพิษระดับปานกลางต่อเซลล์มะเร็งลำใส้, เซลล์มะเร็งตับ, เซลล์มะเร็งเด้านม และเซลล์มะเร็งกระเพาะ อาหาร ขณะที่สาร foveolide B แสดงความเป็นพิษระดับปานกลางเฉพาะต่อเซลล์มะเร็งลำใส้ นอกจากนี้ สาร foveolide A ยังแสดงฤทธิ์ด้านเชื้อวัณโรค *Mycobacterium tuberculosis* โดยมีก่าความเข้มข้นต่ำสุดที่ สามารถยับยั้งเชื้อได้คือ 200 ไมโครโมลาร์

ภาควิชา เภสัชเวทและเภสัชพฤกษศาสตร์	ลายมือชื่อนิสิต
สาขาวิชา เฏสัชเวท	ลายมือชื่อ อ.ที่ปรึกษาวิทยานิพนธ์หลัก
ปีการศึกษา <u>2555</u>	

KEYWORDS: *SALACIA VERRUCOSA* / CELASTRACEAE / *FICUS FOVEOLATA* / MORACEAE / CYTOTOXICITY

PATHOM SOMWONG: PHYTOCHEMISTRY AND BIOACTIVITIES OF *SALACIA VERRUCOSA* AND *FICUS FOVEOLATA* STEMS. ADVISOR: ASSOC.PROF. RUTT SUTTISRI, Ph.D., 262 PP.

Chemical investigation of the stems of *Salacia verrucosa* Wight (family Celastraceae) led to the isolation of one new (21 α -hydroxyfriedelane-1,3-dione) and three known 1,3-diketofriedelane triterpenes (friedelane-1,3-dione, 26-hydroxyfriedelane-1,3-dione and 30-hydroxyfriedelane-1,3-dione), three friedelane-type triterpenes (friedelin, kokoonol and 21 α -hydroxyfriedelan-3-one), and one oleanane-type triterpene (3 β ,22 α -dihydroxyolean-12-en-29-oic acid). From the stems of *Ficus foveolata* (Wall. ex Miq.) Miq. (family Moraceae), two new (foveolide A and foveoeudesmenone) and two known eudesmane-type sesquiterpenes [4(15)-eudesmene-1 β ,6 α -diol and 4(15)-eudesmene-1 β ,5 α -diol], a new sesquiterpenoid dimer (foveolide B), one new (foveospirolide) and one known phenolic compound (ethyl rosmarinate), together with three known triterpenes (friedelin, taraxerol and betulin) were isolated. The chemical structures of these plant constituents were determined by spectroscopic analyses, including UV, IR, MS and NMR, and comparison with previously reported data.

Friedelane-1,3-dione was strongly cytotoxic against human colon cancer cell line (SW620) with an IC₅₀ value of 2.02 μ M, whereas 26-hydroxyfriedelane-1,3-dione and 21 α -hydroxyfriedelan-3-one were moderately cytotoxic against colon, liver (HepG2) and gastric (KATO-III) cancer cell lines. 26-Hydroxyfriedelane-1,3-dione also exhibited moderate cytotoxicity against lung (CHAGO) and breast (BT474) cancer cell lines. Foveolide A was moderately cytotoxic against colon, liver, breast and gastric cancer cell lines, while foveolide B was specifically cytotoxic toward colon cancer cell line. In addition, foveolide A exhibited anti-tuberculosis activity against *Mycobacterium tuberculosis* with a minimum inhibitory concentration of 200 μ M.

Department: <u>Pharmacognosy and Pharmaceutical Botany</u> Student's signature	
Field of Study: <u>Pharmacognosy</u>	Advisor's signature
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LIST OF ABBREVIATIONS

acetone- d_6	=	Deuterated acetone
α	=	Alpha
β	=	Beta
br s	=	Broad singlet (for NMR spectra)
°C	=	Degree Celsius
Calc	=	Calculate
CC	=	Column chromatography
CDCl ₃	=	Deuterated chloroform
CD ₃ OD	=	Deuterated methanol
CH ₂ Cl ₂	=	Dichloromethane
cm	=	Centimeter
cm ⁻¹	=	Reciprocal centimeter (unit of wave number)
¹³ C NMR	=	Carbon-13 Nuclear Magnetic Resonance
DEPT	=	Distortionless Enhancement by Polarization Transfer
2D NMR	=	Two dimensional Nuclear Magnetic Resonance
d	=	Doublet (for NMR spectra)
dd	=	Doublet of doublets (for NMR spectra)
ddd	=	Doublet of doublets of doublets (for NMR spectra)
dddd	=	Doublet of doublets of doublets of doublets (for NMR spectra)
DMSO	=	Dimethyl sulfoxide
δ	=	Chemical shift
3	=	Molar absorptivity
ESI-MS	=	Electrospray Ionization Mass Spectrometry
EtOAc	=	Ethyl acetate
Fu _c	=	Fluorescent unit from untreated
Fu _t	=	Fluorescent unit from treated
g	=	Gram
h	=	Hour

¹ H NMR	=	Proton Nuclear Magnetic Resonance
¹ H- ¹ H COSY	=	Homonuclear (Proton-Proton) Correlation Spectroscopy
HMBC	=	¹ H-detected Heteronuclear Multiple Bond Coherence
HR	=	High Resolution
HSQC	=	Heteronuclear Single Quantum Correlation
Hz	=	Hertz
IC 50	=	Median Inhibitory Concentration
IR	=	Infrared
J	=	Coupling constant
KBr	=	Potassium bromide
Kg	=	Kilogram
L	=	Liter
λ_{max}	=	Wavelength at maximal absorption
μg	=	Microgram
µg/ml	=	Microgram per milliliter
μl	=	Microliter
$[\mathbf{M}]^{+}$	=	Molecular ion
$[M+H]^+$	=	Pseudomolecular ion
$[M+Na]^+$	=	Pseudomolecular ion
[M+H+Na] ⁺	=	Pseudomolecular ion
т	=	Multiplet (for NMR spectra)
MeOH	=	Methanol
mg	=	Milligram
MHz	=	Megahertz
MIC	=	Minimum inhibitory concentration
min	=	Minute
ml	=	Milliliter
mm	=	Millimeter
mp	=	Melting point
MS	=	Mass Spectrometry

MW	=	Molecular weight
m/z	=	Mass to charge ratio
Na	=	Sodium
ν_{max}	=	Wave number at maximal absorption
nm	=	Nanometer
NMR	=	Nuclear Magnetic Resonance
NOESY	=	Nuclear Overhauser Enhancement Spectroscopy
OD	=	Optical density
PBS	=	Phosphate buffered saline
ppm	=	Part-per-million
q	=	Quartet (for NMR spectra)
rel int	=	Relative intensity
S	=	Singlet (for NMR spectra)
t	=	Triplet
td	=	Triplet of doublets (for NMR spectra)
TLC	=	Thin Layer Chromatography
TOF	=	Time of flight
UV	=	Ultraviolet

CHAPTER I

INTRODUCTION

Several members of the family Celastraceae have been traditionally used as medicinal plants. Many of them belong to the genus Salacia which has been found throughout. For example, the decoction of S. reticulata roots is used in the treatment of gonorrhea, rheumatism, skin diseases, itching and swelling, hemorrhoids, asthma, amenorrhoea and dysmenorrhoea (Arunakumara and Subasinghe, 2010). Salacia species (e.g. S. chinensis, S. oblonga, S. reticulata), known as 'Ponkoranti' in Ayurvedic medicine, are widely distributed in India and Sri Lanka (Li, Huang, and Yamahara, 2008) and these species have been used in the treatment of diabetes (Matsuda et al., 2005; Mukherjee et al., 2006; Li et al., 2008). Further investigation into the anti-diabetic activity of the extract of S. oblonga in patients with type 2 diabetes showed that it could lower acute glycemia and insulinemia after high carbohydrate consumption (Williams et al., 2007). In addition, S. oblonga also displayed anti-inflammatory activity in vivo by inhibiting the formation of mediators of inflammation (Ismail et al., 1997). The chloroform and alcoholic extract of S. chinensis roots possessed significant antihyperlipidemic activity in triton-induced and atherogenic diet-induced hyperlipidemic rats (Sikarwar and Patil, 2012). These Salacia species are an important source of medicinal compounds, which are effective as antidiabetic, antiobesity, hypolipidemic, hepatoprotective and antioxidant agents (Paarakh, Patil, and Thanga, 2008).

Moraceae is another important plant family of which several members of its genera have been utilized as medicinal plants; some with similar indications as plants belonging to the family Celastraceae. Members of this family that have been used traditionally include plants in the genus *Ficus* (fig). A number of *Ficus* species are generally used as food and for their medicinal properties in Ayurvedic and traditional Chinese medicine (Lansky *et al.*, 2008). In India, decoction of *F. religiosa* bark has been used as a treatment for various disorders such as fever, gonorrhea and skin diseases; its leaf juice has also been used to treat asthma and cough (Makhija, Sharma, and Khamar, 2010). The plant is also a good source of traditional medicine for the treatment of diabetes, diarrhea, epilepsy, gastric problems, inflammatory and infectious disorders (Singh, Singh, and Goel, 2011). Powdered *F. racemosa* leaves, mixed with honey, are given in bilious infections and its fruits are useful in regulating diarrhea and constipation (Joseph and Raj, 2010). In Nepal, powdered *F. foveolata* bark is taken as a cure for boils and to promote milk secretion during child birth (Kunwar and Bussmann, 2006).

In Thailand, several plants belonging to both families have been employed as medicinal plants, especially in the northeastern part. Celastraceous plants that have been utilized in traditional medicine include *Euonymus cochinchinensis* Craib (Thai name: Nang yai, Maha kanang), *Siphonodon celastrineus* Griff. (Thai name: Duk khao, Ma duk), *Salacia chinensis* L. (Thai name: Ta kai, Kamphaeng chet chan) and *Salacia verrucosa* Wight (Thai name: Ta kwang, Kamphaeng kao chan), whereas members of the family Moraceae popular in herbal medicine are *Ficus hispida* L.f. (Thai name: Ma duea plong), *Ficus racemosa* L. (Thai name: Ma duea uthumphon), *Ficus foveolata* (Wall. ex Miq.) Miq. (Thai name: Ma kra thuep rong) and *Maclura cochinchinensis* Corner (Thai name: Khe, Wua thaloeng). Different parts of these medicinal plants (e.g. stems, leaves and roots) are used as crude drugs and ingredients in medicinal preparations (Chuakul, Saralamp, and Boonpleng, 2002). In northeastern Thailand, crude drugs derived from the stems of these plants are frequently used. For example, the decoctions of *S. chinensis* and *S. verrucosa* stems are prescribed as laxative, whereas the alcoholic maceration of *F. foveolata* stems, prepared solely or mixed with other herbal ingredients, is taken as a tonic (Suksri et al., 2005).

Based on traditional uses of *Salacia* species, many researches were conducted in order to investigate their active compounds and biological activities. Potent α -glucosidase inhibitors, which are thiosugars sulfonium sulfate such as kotalanol (Matsuda *et al.*, 1999; Yoshikawa *et al.*, 2008a), ponkoranol (Yoshikawa *et al.*, 2008a), salacinol (Matsuda *et al.*, 1999; Yoshikawa *et al.*, 1997, 2008a) and salaprinol (Yoshikawa *et al.*, 2008a), were isolated from *S. oblonga*, *S. prinoides* and *S. reticulata*. Several investigated *Salacia* species yielded various triterpenoid compounds, which possessed significant biological activities including antimalarial (Figueiredo, Raz, and Sequin, 1998), antileishmanial (Thiem *et al.*, 2005), antileukemic (Sneden, 1981), cytotoxic (Setzer *et al.*, 2001; Minh *et al.*, 2010), aldose reductase (Matsuda *et al.*, 1999; Morikawa *et al.*, 2003), and α -glucosidase inhibitory (Gao *et al.*, 2010; Huang *et al.*, 2012) and radical scavenging activities (Kishi *et al.*, 2003). Furthermore, phytochemical studies of *S.* *chinensis* have revealed various types of triterpenoid constituents such as the salasones (friedelane-type) (Kishi *et al.*, 2003; Morikawa *et al.*, 2003), foliasalacin A (dammarane-type), foliasalacin B (lupane-type), foliasalacin C (oleanane-type) (Yoshikawa *et al.*, 2008b) and foliasalacin D (baccharane-type) (Zhang *et al.*, 2008b). In addition, eudesmane-type sesquiterpenes, i.e. the salasols (Kishi *et al.*, 2003; Morikawa *et al.*, 2003), megastigmane glycosides, i.e. the foliasalaciosides (Nakamura *et al.*, 2008a, 2011; Zhang *et al.*, 2008a) and phenolic glycosides, i.e. the foliachinenosides (Nakamura *et al.*, 2008b), were also isolated from this plant.

A series of works on *Ficus* species demonstrated various bioactive chemical constituents including antimalarial macrocyclic trichothecene sesquiterpenoids from the leaves and stem bark of *F. fistulosa* (Zhang *et al.*, 2002), anti-inflammatory phenolic glucosides from the bark of *F. racemosa* (Li *et al.*, 2004), antimicrobial triterpenes and isoflavones from the stem bark of *F. ovata* (Kuete *et al.*, 2009), insecticidal and antifungal flavonoids and coumarins from the leaves and stems of *F. sarmentosa* var. *henryi* (Wang *et al.*, 2010, 2011b). Several of these studies on *Ficus* species focused on anticancer and anti-inflammatory activities (Lansky *et al.*, 2008). Different types of compounds such as triterpenes from *F. microcarpa* (Chiang *et al.*, 2005), flavonoids and chromenes from *F. formosana* f. *formosana* (Sheu *et al.*, 2005), phenanthroindolizidine alkaloids from *F. septica* (Wu *et al.*, 2002; Damu *et al.*, 2005, 2009) and a ceramide from *F. elastica* (Mbosso *et al.*, 2012) demonstrated significant cytotoxicity against cancer cell lines *in vitro*. Recently, stilbenoid compounds isolated from the methanolic extract of *F. foveolata* stems were able to potently inhibit butyrylcholinesterase, an enzyme involved in the etiology of Alzheimer's disease (Sermboonpaisarn and Sawasdee, 2012).

From these literature data, it can be observed that members of *Salacia* and *Ficus* genera are important Asian medicinal plants and their utilization in traditional medicine is also widespread throughout Thailand. However, although chemical and biological studies of these plants have been widely performed, phytochemical study of two well-known Thai medicinal plants such as *S. verrucosa* and *F. foveolata* has rarely been conducted. A previous study of *S. verrucosa* stems reported the presence of three friedelane-type triterpenes and one sugar (Jangruang *et al.*, 2009), whereas the similar investigation of *F. foveolata* stems demonstrated four stilbenes, four flavonoids, one alkyl ferulate and a mixture of two alkyl diferulates

(Sermboonpaisarn and Sawasdee, 2012). Therefore, these two plants were selected for further investigation of their chemical constituents and biological activities. The phytochemical data obtained in this study would contribute to the knowledge on chemotaxonomy of these plant families and would be valuable information in the studies of Thai medicinal plants.

The purposes of this research were as follows:

- 1. Isolation and purification of compounds from the stems of *S. verrucosa* and *F. foveolata*.
- 2. Determination of chemical structures and physical properties of each isolated compound.
- 3. Evaluation of biological activities of the isolated compounds

CHAPTER II

HISTORICAL

1. Family Celastraceae

Celastraceae is the dicotyledonous plants belonging to the order Celastrales (including Hippocrateaceae). Members of plants in this family can be found in both hemispheres except the arctic regions, predominantly distributed in the tropics and subtropics. Plants in the family of Celastraceae, including trees and erect or scandent shrubs, were classified into 90 genera and over 1,000 species (Mabberley, 1990). Fruit of the family is often colourful. Leaves are frequently leathery and flowers are small, with four to five sepals and petals; alternating between the petals, stamens rise from a usually conspicuous nectar disk (Celastraceae, 2013). In Thailand, there are 16 genera consisting of 58 species with several subspecies and varieties (Hou, Savinov, and Van Welzen, 2010). Examples of plant genera belonging to this family that can be found are *Euonymus, Siphonodon, Celastrus, Kokoona* and *Salacia*. Several plants of this family have been used in traditional medicine of different countries and one of them, i.e. the genus *Salacia*, has been reported as Thai medicinal plants (Chuakul *et al.*, 2002; Suksri *et al.*, 2005).

2. Genus Salacia

The genus *Salacia*, native to tropical Asia and India, comprises about 200 species. Eleven of these species can be found in evergreen and deciduous forests throughout Thailand. *Salacia* plant is usually climbing shrub or, rarely, small tree. Their leaves are opposite decussate, glabrous and entire. The flowers are in axillary fascicles or on very short bracteate tubercles, in cymes, panicles or thyrses and are bisexual. The calyx is deeply 5-lobed. The numbers of petals are 4-7, imbricate or erect. The stamens are 3 or 2, with slender filaments tapering to the anthers, which mostly open transversely. The ovary, partly or completely immersed in the disc, has 3 or 2 locules with 2-8 ovules in each locule. The subglobose or drupaceous fruits have 1-3 locules with coriaceous pericarp when dried. The seeds usually are 1 to several, embedded in mucilaginous pulp (Hou *et al.*, 2010).

Salacia species found growing in Thailand have been recorded as follows (Royal Forest Department, 2001; Hou *et al.*, 2010).

- 1. S. chinensis L. (Thai name: Kamphaeng chet chan)
- 2. S. dongnaiensis Pierre
- 3. S. grandiflora Kurz (Thai name: Ta som)
- 4. S. korthalsiana Miq. (Thai name: Kamphaeng talang)
- 5. S. laotica Pit.
- 6. S. macrophylla Blume (Thai name: Sadao yen)
- 7. S. noronhioides Pierre (Thai name: Krado yen)
- 8. S. oblongifolia Blume (Thai name: Sadao hot)
- 9. S. rostrata Pierre
- 10. S. verrucosa Wight (Thai name: Ta kwang)
- 11. S. viminea Wall. ex M.A. Lawson (Thai name: Sadao tan)

3. Salacia verrucosa Wight

The plant is a straggling shrub (up to 4 m tall) or, rarely, small tree (up to 9 m tall). Its branches are usually densely covered with lenticels. The leaves are opposite or opposite decussate. The leaf texture is usually subcoriaceous. Their shape is elliptic or obovate, 8-20 by 4-7.5 cm. The leaf base is cuneate or obtuse, while the apex is short-acuminate. The leaf margin is slightly crenulate with 6-10 veins per side. The petiole is 3-10 mm long. The flowers are several to many on short, bracteate tubercles. The pedicels are 9-14.5 mm long. The calyx lobes are slightly erose or shortly fimbriate, about 1 mm long. The petals are broad-elliptic or obovate, 2-3 by 1.5-2 mm. The receptacle disk is round, flat and slightly concave at the center, with the margin slightly extended outward into a narrow membranous rim, about 1.5 mm in diameter. The stamens are 3, about 0.5 mm long. The ovary has 3 locules with 2 ovules in each locule. The subglobose fruits are 2.5-3.5 cm in diameter with orange or red colors. The ellipsoid seeds are about 1.5-1.8 by 1-1.5 cm (Hou *et al.*, 2010).

The habitat of *S. verrucosa* is in the evergreen and deciduous forests of India, Myanmar, Laos, Cambodia, Vietnam, Malaysia, Indonesia and Philippines. In Thailand, this plant is

distributed in the northern and the northeastern parts, for example, Lampang, Phrae, Nakhon Phanom and Nong Khai (Royal Forest Department, 2001).



В





Figure 1. Salacia verrucosa Wight A) Habit, B) Stem, C) Leaves, D) Flowers, E) Cross sections of stem

4. Family Moraceae

Moraceae, the mulberry family of the rose order (Rosales), consists of about 40 genera and over 1,000 species. Most are widely distributed in tropical and subtropical regions, less common in temperate areas (Mabberley, 1990). Plants of the family include trees, shrubs, vines, or rarely herbs often with milky or watery latex and sometimes spines and have small alternate or opposite leaves, along with petalless male or female flowers. The fruits of many species are multiple because fruits from different flowers become joined together (Moraceae, 2013). Examples of the plant genera in this family are *Antiaris*, *Artocarpus*, *Broussonetia*, *Ficus*, *Maclura* and *Morus*. Parts of these plants have been utilized as food and also important in various industries. In India, fruits of these plants are eaten raw and cooked or pickled, while woods are used as fuel woods and furniture. Young leaves of *Morus spp*. are used for rearing silkworms (Chhetri, 2010). In China, the most economically important species are those of *Morus* and *Maclura* associated with the production of silk. Some species in *Broussonetia*, *Maclura* and *Morus* are used for paper making, whereas some species in *Artocarpus*, *Ficus* and *Morus* are edible fruits (Zhekun and Gilbert, 2003).

In Thailand, the members of plants in genus *Artocarpus* and *Ficus* are predominantly used as food and also described as medicinal plants (Suksri *et al.*, 2005).

5. Genus Ficus

Ficus, commonly known as 'Fig', is a plant genus easily characterized by its very distinctive inflorescence. It is one of the largest plant genera consisting of about 1,000 species that can be found mainly in tropical and subtropical regions throughout the world. The genus can be divided into six subgenera including *Urostigma*, *Pharmacosycea*, *Sycomorus*, *Ficus*, *Sycidium* and *Synoecia* (Chaudhary *et al.*, 2012). The genus *Ficus* includes plants like trees, shrubs, climbers, stragglers and woody epiphytes with latex. The leaves are usually alternate. The leaf margin is entire to lobed, rarely palmate. The leaf blade is glabrous or hairy and contained pinnate veins. The inflorescences are axillary or on specialized cauliflorous branches. *Ficus* plant is either monoecious with male, gall (sterile female) and female flowers in each fig, or dioecious with either male and gall flowers or only female flowers in each fig. The male flowers have 2-6 calyx lobes with 1-3 stamens, while the female flowers have 0-6 calyx lobes with a free ovary.

The gall flowers are similar to female flowers but do not produce seeds. The fruits are a seed-like achene, usually enclosed within a syconium formed from the enlarged hollow fleshy receptacle (hypanthodium). The seeds are pendulous with small amount of endosperm (Zhekun and Gilbert, 2003).

In Thailand, about 77 *Ficus* species have been recorded as follows (Royal Forest Department, 2001; Berg, 2007).

- 1. F. acuminata Roxb. (Thai name: Duea)
- 2. F. albipila (Miq.) King (Thai name: Liang phueng)
- 3. F. altissima Blume (Thai name: Krang)
- 4. F. annulata Blume (Thai name: Sai)
- 5. F. anserina (Corner) C.C. Berg
- 6. F. arnottiana (Miq.) Miq.
- 7. F. aurantiaca Griff. (Thai name: Duea thao)
- 8. F. auriculata Lour. (Thai name: Duea wa)
- 9. F. benghalensis L. (Thai name: Ni khrot)
- 10. F. benjamina L. (Thai name: Sai yoi)
- 11. F. bistipulata Griff.
- 12. F. botryocarpa Miq. (Thai name: Duea ching)
- 13. F. calcicola Corner (Thai name: Hai)
- 14. F. callosa Willd. (Thai name: Ma duea kwang)
- 15. F. capillipes Gagnep. (Thai name: Kariang)
- 16. F. carica L. (Thai name: Ma duea farang)
- F. chartacea Wall. ex King var. chartacea (Thai name: Ma duea khi nok)

var. torulosa Wall. (Thai name: Sai nok)

- 18. F. consociata Blume var. murtonii King (Thai name: Sai yai)
- 19. F. curtipes Corner (Thai name: Sai hin)
- 20. F. drupacea Thunb.

var. drupacea (Thai name: Lung khon)

var. pubescens Corner (Thai name: Sai)

- 21. F. elastica Roxb. ex Hornem. (Thai name: Yang india)
- 22. F. fistulosa Reinw. ex Blume (Thai name: Ching khao)
- 23. F. foveolata (Wall. ex Miq.) Miq. (Thai name: Ma kra thuep rong)
- 24. F. geniculata Kurz (Thai name: Hai)
- 25. F. glabella Blume var. concinna Miq. (Thai name: Krai)
- 26. F. glaberrima Blume

var. glaberrima (Thai name: Duea sai)

var. siamensis Corner (Thai name: Pho hin)

- 27. F. globosa Blume (Thai name: Sai luk klom)
- 28. F. griffithii (Miq.) Miq.
- 29. F. heterophylla L.f. (Thai name: Salot nam)
- 30. F. heteropleura Blume (Thai name: Sai)
- 31. F. heterostyla Merr.
- 32. F. hirta Vahl (Thai name: Ma duea hom)
- 33. F. hispida L.f. (Thai name: Ma duea plong)
- 34. F. indica L. (Thai name: Sai tok)
- 35. F. infectoria Roxb. (Thai name: Liap)
- 36. F. laevis Blume (Thai name: Ma duea thao)
- 37. F. lepicarpa Blume (Thai name: Cha luk pho)
- 38. F. lyrata Warb. ex De Wild. & Durand (Thai name: Yang bai so)
- 39. F. malayana C.C.Berg & Chantaras.
- 40. F. montana Burm.f. (Thai name: Ma duea hin)
- 41. F. nervosa Heyne (Thai name: Pho kha nun)
- 42. F. obpyramidata King (Thai name: Pho)
- 43. F. obscura Blume

var. *borneensis* (Miq.) Corner (Thai name: Sai chueak) var. *obscura* (Thai name: Sai tok)

- 44. F. oligodon Miq. (Thai name: Duea wa)
- 45. F. parietalis Blume (Thai name: Ma duea khon)
- 46. F. pellucidopunctata Griff. (Thai name: Sai tok)

- 47. F. praetermissa Corner (Thai name: Duea nam)
- 48. F. prostrata Wall. (Thai name: Ma not)
- 49. F. pubilimba Merr. (Thai name: Sai)
- 50. F. pumila L. (Thai name: Ma duea thao)
- 51. F. punctata Thunb. (Thai name: Duea thao bai yai)
- 52. F. pyriformis Hook. & Arn. (Thai name: Luk khlai)
- 53. F. racemosa L. (Thai name: Ma duea uthumphon)
- 54. F. religiosa L. (Thai name: Pho si maha pho)
- 55. F. retusa L. var. retusa (Thai name: Sai yoi bai thu)
- 56. F. ribes Reinw. ex Blume (Thai name: Ma not)
- 57. F. ridleyana C.C.Berg & Chantaras.
- 58. F. rostrata Lam. (Thai name: Salot hin)
- 59. F. rumphii Blume (Thai name: Pho khi nok)
- 60. F. saemocarpa Miq. (Thai name: Duea pha)
- 61. F. saxophila Blume subsp. cardiophylla (Merr.) C.C. Berg
- 62. F. schwarzii Koord. (Thai name: Duea pho)
- 63. F. scortechinii King (Thai name: Duea din)
- 64. F. semicordata Buch.-Ham. ex Sm. (Thai name: Duea plong hin)
- 65. F. subcordata Blume (Thai name: Sai)
- 66. F. subgelderi Corner var. rigida (Miq.) Corner (Thai name: Sai krang)
- 67. F. subincisa Sm. (Thai name: Ma duea noi)
- 68. F. subpisocarpa Gagnep.
- 69. F. superba (Miq.) Miq. var. superba (Thai name: Krai)
- 70. F. talbotii King (Thai name: Khan laen)
- 71. F. tinctoria G.Forst.

subsp. *gibbosa* (Blume) Corner (Thai name: Krang) subsp. *parasitica* Willd. (Thai name: Sai krang)

- 72. F. triloba Buch.-Ham. ex Voigt
- 73. F. tuphapensis Drake var. annamensis Corner (Thai name: Ma duea khon)
- 74. F. variegata Blume (Thai name: Phuk)

- 75. F. vasculosa Wall. ex Miq. (Thai name: Ma duea thong)
- 76. F. virens Aiton

var. glabella (Blume) Corner (Thai name: Krang)var. virens (Thai name: Phak lueat)

77. F. viridicarpa Corner (Thai name: Duea phlong)

6. Ficus foveolata (Wall. ex Miq.) Miq.

A synonym of this plant is F. sarmentosa Buch.-Ham. ex Sm. The plant is a climber or woody vine belonging to the subgenus Synoecia. Its branchlets are grayish white when dried, wrinkled, glabrous or densely white-hairy. The thinly membranous stipules are lanceolate-ovate, about 8 mm long. The leaves are arranged alternately in 2 vertical rows on opposite sides of the stem; the petiole is about 1 cm long. The leaf blade is ovate, ovate-elliptic, elliptic-lanceolate, lanceolate or oblong, 8-12 by 3-4 cm; both surfaces are glabrous or sparsely brown pubescent on the lower surface. The leaf base is rounded to broadly cuneate, while the apex is acute to acuminate. The secondary veins are 4-12 on each side of the midvein. The globose fruits are axillary on leafy or on leafless branchlets, solitary or occasionally paired, blackish purple when matured. They are 0.5-2 cm in diameter and are densely covered with brown hairs. The peduncle is 0.5-1.5 cm or shorter along with triangular involucral bracts. The male flowers are pedicellate with 3-4 oblanceolate calyx lobes, 2 stamens with short filaments and mucronate anthers. The gall flowers are pedicellate with 4 obovate-spatulate calyx lobes, elliptic ovary and shallowly funnelform stigma. The female flowers are pedicellate with spatulate calyx lobes, obovate ovary and thin and long stigma. The achenes are ovoid-ellipsoid with sticky liquid (Zhekun and Gilbert, 2003).

F. foveolata has been found growing in the forests of East, South and South-East Asia, including the northern and northeastern parts of Thailand, at 600-2,500 m above sea level (Royal Forest Department, 2001; Zhekun and Gilbert, 2003).







Figure 2. *Ficus foveolata* (Wall. ex Miq.) Miq. A) Habit, showing young stem and leaves, B) Stem, C) Cross section of stem

7. Friedelane and Quinone-methide Triterpenes in the Family Celastraceae

Quinone-methide triterpenes (celastroloids) are secondary metabolites related to friedelane-type triterpenes and are restricted to the higher plant families including Celastraceae and Hippocrateaceae. The biosynthesis of quinone-methides requires an oxidosqualene as a central intermediate, which would be converted by a cyclase into the first cyclic intermediate, 3β -friedelanol and then, by an oxidoreductase, into friedelin. Transformation of friedelin to quinone-methide required many biosynthetic steps involving various enzymes. Due to the enzyme activities and their localizations, friedelin and its derivatives usually accumulate in the leaves and

stems, whereas the quinone-methides are mostly found in the root bark. These compounds have been predominantly isolated from the plant genera *Maytenus* and *Salacia* (Corsino *et al.*, 2000).



Friedelin

A quinone-methide

Maytenus species constitute a rich source of friedelane-type triterpenes. Compounds isolated from these plants have shown a variety of biological activities such as antitumour (Nozaki *et al.*, 1986), antiulcerogenic (Andrade *et al.*, 2008), cytotoxicity (Oramas-Royo *et al.*, 2010) and antidiabetic activities (Ardiles *et al.*, 2012). For example, maytenfoliol (54), an antileukemic agent, was isolated along with canophyllal (3), canophyllol (4), friedelin (27), 29-hydroxyfriedelan-3-one (34), 30-hydroxyfriedelan-3-one (35), maytensifolins A-C (56-58) and pachysonol (67) from *M. diversifolia* (Nozaki *et al.*, 1986, 1991). 3,15-Dioxo-21α-hydroxyfriedelane (18), isolated from *M. robusta*, exhibited antiulcerogenic activity in mice afflicted with gastric ulcers comparable with omeprazole as the positive control (54.6 and 57.9% inhibition, respectively) (Andrade *et al.*, 2008). The aromatic and quinone-methide triterpenes isolated from *M. retusa*, such as 23-*nor*-blepharodol (1), 6,23-dioxo-7,8-dihydropristimerol-23-oic acid (17), 3-methoxy-6-oxotingenol-23-oic acid (60), 7-oxo-7,8-dihydroscutione (62) and 21-oxopristimerine (65), showed cytotoxic activity against the human tumor cell lines HL-60 and MCF-7 (Oramas-Royo *et al.*, 2010). 7β,29-Dihydroxy-D:A-friedooleanan-3-one (13) and 7β-hydroxy-3-oxo-D:A-friedooleanan-28-oic acid (40) from the root bark of *M. jelskii* appeared to
increase insulin-mediated signaling, suggesting their potential in the treatment of type 2 diabetes mellitus (Ardiles *et al.*, 2012).

The friedelane-type triterpenes were also isolated from plants in the genera *Celastrus*, *Euonymus* and *Kokoona*. Cassinolide (5), a triterpenoid lactone, was isolated along with 7 β ,29dihydroxy-D:A-friedooleanan-3-one (13), friedelin (27), 29-hydroxyfriedelan-3-one (34) and 7 β hydroxy-3-oxo-D:A-friedooleanan-28-oic acid (40) from *C. vulcanicola* (Nunez *et al.*, 2012; Torres-Romero *et al.*, 2010). 30-Hydroxyfriedelan-3-one (35), isolated from *E. alatus*, exhibited cytotoxic activity against MDA-MB-435 cell line with an inhibition percentage of 57.38% at the concentration of 10 µg/ml (Tu *et al.*, 2011). Demethylzeylasterone (7), a 6-oxophenolic triterpenoid from *K. zeylanica*, was found to be an inhibitor of the enzyme topoisomerase II α (IC₅₀ = 17.6 µM). The compound was also selectively cytotoxic against MCF-7 cancer cell line (IC₅₀ = 12.5 µM) (Furbacher and Gunatilaka, 2001). The same plant also contains other friedelane-type triterpenoids including D:A-friedo-olean-3,21-dione (28), 21 α -hydroxyfriedelan-3-one (33), zeylandiol (82), zeylanol (83) and zeylanonol (84) (Gunatilaka, Nanayakkara, and Sultanbawa, 1979; Gunatilaka *et al.*, 1982).

The presence of friedelane-type triterpenes and quinone-methides in *Salacia* species has been reported and some of them have been shown to be bioactive. For example, 17methoxycarbonyl-28-*nor*-isoiguesterin (**59**) from the roots of *S. kraussii* displayed antimalarial activity against the multidrug-resistant strain K1 and the drug-sensitive strain NF₅₄ of *Plasmodium falciparum*, with IC₅₀ values of 27.6 and 37.1 ng/ml, respectively (Figueiredo *et al.*, 1998). Regeol A (**69**) and tingenone (**81**), from *S. chinensis* stems, showed an inhibitory effect on rat lens aldose reductase, with IC₅₀ values of 30 and 13 μ M, respectively (Morikawa *et al.*, 2003). Two triterpenoids isolated from the roots of *S. madagascariensis*, isoiguesterin (**46**) and 20-*epi*isoiguesterinol (**49**), were potently active against the protozoa *Leishmania* (Thiem *et al.*, 2005).

Friedelane triterpenoids isolated from *Salacia* species also include celastrol (6), 30hydroxypristimerin (42), 22 β -hydroxytingenone (44), isoiguesterin (46), isoiguesterinol (48), netzahualcoyene (61), pristimerin (68), salacenonal (70), salaciquinone (71) and tingenone (81) found in the root bark of *S. reticulata* (Dhanabalasingham *et al.*, 1996), friedelane-1,3-dione (24) from the stems of *S. verrucosa* (Jangruang *et al.*, 2009), 15 α -hydroxyfriedelan-1,3-dione (30) from the stem bark of *S. beddomei* (Hisham *et al.*, 1996b), 26-hydroxyfriedelan-1,3-dione (31) from the roots of *S. oblonga* (Matsuda *et al.*, 1999) and the stems of *S. verrucosa* (Jangruang *et al.*, 2009), and kokoonol (**52**), salaquinone A (**72**) and salasones A-C (**74-76**) from the stems of *S. chinensis* (Morikawa *et al.*, 2003).

Phytochemical investigation of plants in the family Celastraceae has resulted in the isolation of various friedelane and quinone-methide triterpenes. Distribution of these compounds in this plant family is summarized in **Table 1**, and their chemical structures are shown in **Figure 3**.

Compound	Source	Plant part	References
23-nor-Blepharodol (1)	Maytenus retusa		Oramas-Royo
	Root bark e M. ilicifolia I		et al., 2010
Cangoronine (2)			Itokawa et al., 1991
Canophyllal (3)	M. diversifolia	Stems	Nozaki <i>et al.</i> , 1986
Canophyllol (4)	Celastrus vulcanicola	Stems	Torres-Romero et
			al., 2010
	Lepidobotrys staudtii	Leaves,	Tane et al., 1996
	Stem bark		
	M. diversifolia	Stems	Nozaki <i>et al</i> ., 1986
	M. macrocarpa	Stem bark	Chavez et al., 1998
	Salacia elliptica	Branches	Duarte et al., 2009
Cassinolide (5)	Cassine xylocarpa	Root bark	Nunez et al., 2012
	Celastrus vulcanicola	Stems	Torres-Romero
			et al., 2010
Celastrol (6)	S. kraussii	Roots	Figueiredo et al.,
			1998
	S. reticulata	Root bark	Dhanabalasingham
			<i>et al.</i> , 1996

Table 1. Distribution of friedelane and quinone-methide triterpenes in the family Celastraceae

Compound	Source	Plant part	References
Demethylzeylasterone (7)	Kokoona zeylanica	Kokoona zeylanica Inner bark	
			Gunatilaka, 2001
1β,15α-Dihydroxy-friedelan-	Salacia beddomei	Stem bark	Hisham et al., 1996b
3-one (8)			
2α,29-Dihydroxy-3-			
friedelanone (9)			
2β,21α-Dihydroxy-3-	T . 1 1 1	Leaves,	T (1 100)
friedelanone (10)	Lepiaoboirys siauain	stem bark	1 ane <i>et al.</i> , 1996
6β,21α-Dihydroxy-3-			
friedelanone (11)			
16α,28-Dihydroxyfriedelin	S. elliptica	Branches	Duarte et al., 2009
(12)			
7β,29-Dihydroxy-D:A-	Maytenus jelskii	Root bark	Ardiles et al., 2012
friedooleanan-3-one (13)			
21α,26-Dihydroxy-D:A			Kumar, Wazeer, and
-friedooleanan-3-one (14)			Wijeratne, 1985
21α,30-Dihydroxy-D:A	S. reticulata	Stom harls	Kumar, Wijeratne,
-friedooleanan-3-one (15)	var. <i>diandra</i>	Stem bark	and
			Abeygunawardena,
			1990
28,29-Dihydroxyfriedelan-3-	Elaeodendron balae	Stems	Weeratunga et al.,
one (16)			1982
6,29-Dioxo-7,8-	M. retusa	Root bark	Oramas-Royo et al.,
dihydropristimerol-23			2010
-oic acid (17)			
3,15-Dioxo-21α-	M. robusta	Aerial parts	Andrade et al., 2008
hydroxyfriedelane (18)			

Compound	Source	Plant part	References
Elaeodendradiol (19)	Elaeodendron	Bark	Anjaneyulu and Rao,
Elaeodendrol (20)	glaucum 1		1980
2,4(23)-Friedeladien-22β-			
hydroxy-21-one (21)	Acanthothamnus	Roots	Estrada <i>et al</i> ., 1994
2,4(23)-Friedeladien-29-oic	aphyllus		
acid (22)			
Friedelan-3 β -ol (23)	Celastrus vulcanicola	Stems	Torres-Romero
			et al., 2010
	Pleurostylia opposita	Leaves	Dantanarayana et al.,
			1983
	Salacia elliptica	Branches	Duarte et al., 2009
Friedelane-1,3-dione (24)	Peritassa compta		Klass and Tinto,
		Stems	1992
	S. verrucosa		Jangruang et al.,
			2009
Friedelane-3,15-dione (25)	P. compta	Stems	Klass and Tinto,
			1992
3,4-seco-Friedelan-3-oic acid	Maytenus obtusifolia	Roots	Silva <i>et al.</i> , 2008
(26)			
Friedelin (27)	Celastrus vulcanicola	Stems	Torres-Romero
			<i>et al.</i> , 2010
	Euonymus alatus	Root bark	Tu et al., 2011
	Kokoona zeylanica	Inner bark	Gunatilaka <i>et al</i> .,
			1982
	M. diversifolia	Stems	Nozaki <i>et al</i> ., 1986
	M. macrocarpa	Stem bark	Chavez <i>et al.</i> , 1998

Compound	Source	Plant part	References
Friedelin (27)	Maytenus obtusifolia	Roots	Silva <i>et al.</i> , 2008
	M. salicifolia	Fruits	Valladao et al., 2010
	Peritassa compta	Stems	Klass and Tinto,
			1992
	P. opposita	Leaves	Dantanarayana et al.,
			1983
	Salacia beddomei	Stem bark	Hisham et al., 1996b
	S. elliptica	Branches	Duarte et al., 2009
	S. verrucosa	Stems	Jangruang et al.,
			2009
D:A-Friedo-olean-3,21-dione	Kokoona zeylanica	Inner bark	Gunatilaka <i>et al</i> .,
(28)			1982
15α-Hydroxyfriedelan-3-one	P. compta	Stems	Klass and Tinto,
(29)			1992
	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. beddomei	Stem bark	Hisham et al., 1996b
15α-Hydroxyfriedelan-1,3-	P. compta	Stems	Klass and Tinto,
dione (30)			1992
	S. beddomei	Stem bark	Hisham et al., 1996b
26-Hydroxyfriedelan-1,3-dione	S. oblonga	Roots	Matsuda et al., 1999
(31)	S. reticulata	Roots	Yoshikawa et al.,
			2002
	S. verrucosa	Stems	Jangruang et al.,
			2009
28-Hydroxyfriedelan-1,3-dione	M. macrocarpa	Stem bark	Chavez <i>et al.</i> , 1998
(32)			

Compound	Source	Plant part	References
21α-Hydroxyfriedelan-3-one	Kokoona zeylanica Inner bark		Gunatilaka et al.,
(33)			
	Lepidobotrys staudtii	Leaves,	Tane et al., 1996
		stem bark	
29-Hydroxyfriedelan-3-one	Catha cassinoides	Stems	Betancor et al., 1980
(34)	Celastrus vulcanicola	Stems	Torres-Romero
			et al., 2010
	Euonymus alatus	Root bark	Tu et al., 2011
	L. staudtii	Leaves,	Tane et al., 1996
		stem bark	
	Maytenus diversifolia	Stems	Nozaki <i>et al.</i> , 1986
	M. ilicifolia	Root bark	Itokawa et al., 1991
	M. macrocarpa	Stem bark	Chavez et al., 1998
	M. nemerosa	Stems	Fang et al., 1984
	M. obtusifolia	Roots	Silva et al., 2008
	Salacia chinensis	Stems	Morikawa <i>et al.</i> ,
			2003
	S. reticulata	Root bark	Dhanabalasingham
			<i>et al.</i> , 1996
30-Hydroxyfriedelan-3-one	C. cassinoides	Stems	Betancor et al., 1980
(35)	E. alatus	Root bark	Tu et al., 2011
	L. staudtii	Leaves,	Tane et al., 1996
		stem bark	
	M. diversifolia	Stems	Nozaki <i>et al.</i> , 1986
	M. nemerosa	Stems	Fang <i>et al.</i> , 1984
	S. elliptica	Branches	Duarte et al., 2009
16α-Hydroxyfriedelin (36)	S. elliptica	Branches	Duarte <i>et al.</i> , 2009

Compound	Source	Plant part	References
28-Hydroxyisoiguesterin (37)	Salacia kraussii	Roots	Figueiredo et al.,
			1998
3-Hydroxy-2-oxofriedelan-3-			
en-20α-carboxylic acid (38)	Austroplenckia		G (1 1000
3β-Hydroxy-2-oxofriedelan-	populnea	D (1 1	Sousa <i>et al.</i> , 1990
20α-carboxylic acid (39)		Kool bark	
7β-Hydroxy-3-oxo-D:A	Maytenus jelskii	-	Ardiles et al., 2012
-friedooleanan-28-oic acid (40)			
2α-Hydroxypopulnonic acid	Acanthothamnus	Roots	Estrada et al., 1994
(41)	aphyllus		
30-Hydroxypristimerin (42)	S. reticulata		Dhanabalasingham
		Root bark	et al., 1996
20α- Hydroxytingenone (43)	A. populnea		Sousa et al., 1990
22β-Hydroxytingenone (44)	A. aphyllus	Roots	Estrada et al., 1994
			Dhanabalasingham
	S. reticulata	Root bark	<i>et al.</i> , 1996
Ilicifoline (45)	M. ilicifolia	Root bark	Itokawa <i>et al</i> ., 1991
Isoiguesterin (46)	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. madagascariensis	Roots	Thiem <i>et al.</i> , 2005
	S. reticulata	Root bark	Dhanabalasingham
			<i>et al.</i> , 1996
28-nor-Isoiguesterin-17	S. kraussii	Roots	Figueiredo et al.,
-carbaldehyde (47)			1998
Isoiguesterinol (48)	S. madagascariensis	Roots	Thiem <i>et al.</i> , 2005
	S. reticulata	Root bark	Dhanabalasingham
			et al., 1996

Compound	Source	Plant part	References
20-epi-Isoiguesterinol (49)	Salacia	Roots	Thiem <i>et al.</i> , 2005
	madagascariensis		
Isopristimerin III (50)		D (1 1	
Isotingenone III (51)	Maytenus ilicijolia	Koot bark	Itokawa <i>et al.</i> , 1991
Kokoonol (52)	Kokoona zeylanica	Stems	Gunatilaka <i>et al</i> .,
			1983a
	S. chinensis	Stems	Morikawa et al.,
			2003
Kotalagenin 16-acetate (53)	S. oblonga	Roots	Matsuda et al., 1999
Maytenfoliol (54)	M. diversifolia	Stems	Nozaki <i>et al</i> ., 1986
Maytenoic acid (55)	Austroplenckia	Root bark	Sousa et al., 1990
	populnea		
	Catha cassinoides	Stems	Betancor et al.,
			1980
	Celastrus vulcanicola	Stems	Torres-Romero
			et al., 2010
	Gymnosporia	Bark	Ramaiah et al., 1984
	emarginata		
	M. ilicifolia	Root bark	Itokawa <i>et al.</i> , 1991
Maytensifolin-A (56)			Nagali et al. 1096
Maytensifolin-B (57)	M. diversifolia	Stems	Nozaki <i>el al.</i> , 1986
Maytensifolin-C (58)			Nozaki <i>et al</i> ., 1991
17-Methoxycarbonyl-28-nor	S. kraussii	Roots	Figueiredo et al.,
-isoiguesterin (59)			1998
3-Methoxy-6-oxotingenol-23-	M. retusa	Root bark	Oramas-Royo et al.,
oic acid (60)			2010

Compound	Source	Plant part	References
Netzahualcoyene (61)	Salacia amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. reticulata	Root bark	Dhanabalasingham
			<i>et al.</i> , 1996
7-Oxo-7,8-dihydroscutione	Maytenus retusa	Root bark	Oramas-Royo et al.,
(62)			2010
7-Oxofriedelin (63)	M. obtusifolia	Dooto	Silva <i>et al.</i> , 2008
6-Oxoisoiguesterin (64)	S. madagascariensis	ROOIS	Thiem <i>et al.</i> , 2005
21-Oxopristimerine (65)	M. retusa	Root bark	Oramas-Royo et al.,
			2010
Pachysandiol-B (66)	M. acanthophylla	Leaves	Oliveira et al., 2009
Pachysonol (67)	M. diversifolia	Stems	Nozaki <i>et al.</i> , 1986
Pristimerin (68)	Acanthothamnus	Roots	Estrada et al., 1994
	aphyllus		
	Austroplenckia	Root bark	Sousa et al., 1990
	populnea		
	M. ilicifolia	Root bark	Itokawa et al., 1991
	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. beddomei	Stem bark	Hisham, 1995
	S. kraussii	Roots	Figueiredo et al.,
			1998
	S. reticulata	Root bark	Dhanabalasingham
			et al., 1996
Regeol A (69)	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. chinensis	Stems	Morikawa <i>et al.</i> ,
			2003
Salacenonal (70)	S. reticulata	Root bark	Dhanabalasingham
			<i>et al.</i> , 1996

Compound	Source	Plant part	References	
Salaciquinone (71)	Salacia reticulata Root bark		Dhanabalasingham	
			<i>et al.</i> , 1996	
Salaquinone A (72)			Morikawa <i>et al.</i> ,	
			2003	
Salaquinone B (73)			Kishi et al., 2003	
Salasone A (74)				
Salasone B (75)	S. chinensis	Stems	Morikawa <i>et al.</i> ,	
Salasone C (76)			2003	
Salasone D (77)			Kishi <i>et al.</i> , 2003	
Salasone E (78)				
Salaspermic acid (79)	Maytenus ilicifolia	Root bark	Itokawa et al., 1991	
7-Tetraene-24-nor-friedelane-	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a	
29-oic acid methylester (80)				
Tingenone (81)	Acanthothamnus	Roots	Estrada et al., 1994	
	aphyllus			
	M. nemerosa	Stems	Fang <i>et al.</i> , 1984	
	S. chinensis	Stems	Morikawa <i>et al</i> .,	
			2003	
	S. reticulata	Root bark	Dhanabalasingham	
			<i>et al.</i> , 1996	
Zeylandiol (82)			Gunatilaka <i>at al</i>	
Zeylanol (83)	Kokoona zeylanica	n.i.	1979	
Zeylanonol (84)			1717	

n.i. = not indicated



6,29-Dioxo-7,8-dihydropristimerol-23-oic acid (17) $R = CO_2H$



	\mathbf{R}_1	R_2	R ₃	R_4	R ₅
Canophyllal (3)	H_2	H_2	H_2	СНО	CH ₃
Maytenfoliol (54)	H ₂	H_2	H_2	CH ₂ OH	CH ₂ OH
Maytensifolin-B (57)	H_2	Ο	H_2	CH ₃	CH ₃
Maytensifolin-C (58)	β-ΟΗ, α-Η	Ο	0	CH ₃	CH ₃
Pachysonol (67)	H_2	β-ΟΗ, α-Η	H_2	CH ₃	CH ₃

Figure 3. Friedelane and quinone-methide triterpenes isolated from plants in the family Celastraceae

 R_3



	R_1	R ₂	R ₃	R_4	R ₅	R ₆	R ₇
Canophyllol (4)	Н	Н	Н	Н	CH ₃	CH ₂ OH	CH ₃
6β,21α-Dihydroxy-3-	OH	Н	Н	OH	CH ₃	CH ₃	CH ₃
friedelanone (11)							
16α,28-Dihydroxy	Н	Н	OH	Н	CH ₃	CH ₂ OH	CH ₃
friedelin (12)							
7β,29-Dihydroxy-	Н	OH	Н	Н	CH ₃	CH ₃	CH ₂ OH
D:A-friedooleanan-3-							
one (13)							
21α,26-Dihydroxy-	Н	Н	Н	OH	CH ₂ OH	CH ₃	CH_3
D:A-friedooleanan-3-							
one (14)							
28,29-Dihydroxy	Н	Н	Н	Н	CH ₃	CH ₂ OH	CH ₂ OH
friedelan-3-one (16)							
16α-Hydroxy	Н	Н	OH	Н	CH ₃	CH ₃	CH ₃
friedelin (36)							
7β-Hydroxy-3-oxo-	Н	OH	Н	Н	CH ₃	CO ₂ H	CH ₃
D:A-friedooleanan-							
28-oic acid (40)							
Kokoonol (52)	Н	Н	Н	Н	CH ₂ OH	CH ₃	CH ₃



	R_1	R_2	R ₃
Celastrol (6)	H_2	CO ₂ H	CH ₃
30-Hydroxypristimerin (42)	H_2	CO ₂ CH ₃	CH ₂ OH
Isoiguesterinol (48)	H_2	CH ₂ OH	Н
20-epi-Isoiguesterinol (49)	H_2	Н	CH ₂ OH
21-Oxopristimerine (65)	0	CO ₂ CH ₃	CH ₃
Pristimerin (68)	H_2	CO ₂ CH ₃	CH ₃



	R_1	R ₂	R ₃
Demethylzeylasterone (7)	Н	H_2	CO ₂ H
3-Methoxy-6-oxotingenol-23-oic acid (60)	CH ₃	0	Н



	R ₁	R ₂	R ₃	R_4	R ₅	R ₆	R ₇
1β,15α-Dihydroxy-friedelan-3-one (8)	OH	Н	Н	ОН	Н	CH ₃	CH ₃
2α ,29-Dihydroxy-3-friedelanone (9)	Н	OH	Н	Н	Н	CH ₂ OH	CH ₃
2β ,21 α -Dihydroxy-3-friedelanone	Н	Н	OH	Н	OH	CH ₃	CH ₃
(10)21α,30-Dihydroxy-D:A-friedooleanan-3-one (15)	Н	Н	Н	Н	ОН	CH ₃	CH ₂ OH
Friedelin (27)	Н	Н	Н	Н	Н	CH ₃	CH ₃
15α -Hydroxyfriedelan-3-one (29)	Н	Н	Н	OH	Н	CH ₃	CH ₃
21 α -Hydroxyfriedelan-3-one (33)	Н	Н	Н	Н	OH	CH ₃	CH ₃
29-Hydroxyfriedelan-3-one (34)	Н	Н	Н	Н	Н	CH ₂ OH	CH ₃
30-Hydroxyfriedelan-3-one (35)	Н	Н	Н	Н	Н	CH ₃	CH ₂ OH
2α-Hydroxypopulnonic acid (41)	Н	OH	Н	Н	Н	CO ₂ H	CH ₃
Maytenoic acid (55)	Н	Н	Н	Н	Н	CO ₂ H	CH ₃



3,15-Dioxo-21 α -hydroxyfriedelane (18) R = OH Friedelane-3,15-dione (25) R = H

Elaeodendradiol (19) $R = CH_2OH$ Elaeodendrol (20) $R = CH_3$

R ₁
R ₂

	R_1	R_2	R ₃
2,4(23)-Friedeladien-22β-	0	OH	CH ₃
hydroxy-21-one (21)			
2,4(23)-Friedeladien-29-	H_2	Н	CO ₂ H
oic acid (22)			



	R ₁	R_2	R ₃
Friedelan-3β-ol (23)	H_2	Н	CH ₃
3β-Hydroxy-2-oxofriedelan	0	Н	CO ₂ H
-20α-carboxylic acid (39)			
Pachysandiol-B (66)	H_2	OH	CH ₃



	R_1	R_2	R ₃	R_4
Friedelane-1,3-dione (24)	Н	Н	CH ₃	CH ₃
15α -Hydroxyfriedelan-1,3-dione (30)	ОН	Н	CH ₃	CH ₃
26-Hydroxyfriedelan-1,3-dione (31)	Н	Н	CH ₂ OH	CH ₃
28-Hydroxyfriedelan-1,3-dione (32)	Н	Н	CH ₃	CH ₂ OH
Kotalagenin 16-acetate (53)	Н	OAc	CH ₂ OH	CH ₃



3,4-seco-Friedelan-3-oic acid (26)



- R₁ R₂
- D:A-Friedo-olean-3,21-dione H_2 O
- 7-Oxofriedelin (63) O H_2

Figure 3. (continued)



28-Hydroxyisoiguesterin (37) $R = CH_2OH$

Isoiguesterin (46) $R = CH_3$

28-nor-Isoiguesterin-17-carbaldehyde (47) R = CHO

17-Methoxycarbonyl-28-*nor*-isoiguesterin (59) $R = CO_2CH_3$





Figure 3. (continued)



6-Oxoisoiguesterin (64)

	R_1	R ₂	
pristimerin III (50)	H_2	$\rm CO_2 CH_3$	
ingenone III (51)	0	Н	

Maytensifolin-A (56)



7-Oxo-7,8-dihydroscutione (62)



Regeol A (69)



Salaciquinone (71)

Salaquinone B (73)



	R_1	R_2	R ₃	R_4
Salasone A (74)	Н	0	CH ₂ OH	CH ₃
Salasone B (75)	OH	0	CH ₃	CH ₃
Salasone C (76)	Н	α-ΟΗ, β-Η	CH ₃	CH ₂ OH
Salasone D (77)	Н	α-ОН, β-Н	CH ₂ OH	CH ₃
Salasone E (78)	OH	H_2	CH ₂ OH	CH ₃

Figure 3. (continued)

ΌΗ



Figure 3. (continued)

8. Chemical Constituents of Plants in the Genus Salacia

Plants in the genus *Salacia* such as *S. reticulata* and *S. oblonga* have been widely used in the treatment for the initial stages of diabetes in the Ayurvedic system of Indian traditional medicine. The mode of administration usually consists of an aqueous decoction prepared from the roots of these plants (Matsuda, Morikawa, and Yoshikawa, 2002). Investigations of the bioactive chemical constituents of these plants led to the identification of a xanthone glucoside, mangiferin (**153**), from the root bark of *S. reticulata* (Karunanayake and Sirimanne, 1985) and the stems of *S. chinensis* (Kishi *et al.*, 2003), as the hypoglycemic and radical scavenging principle.

Various studies on *Salacia* species have reported the isolation of α -glucosidase inhibitors involved in the treatment of diabetes mellitus. Thiosugar sulfonium salts, including kotalanol (145), ponkoranol (162), salaprinol (167) and salacinol (166) (Yoshikawa et al., 1997, 2008a; Matsuda et al., 1999), isolated from the roots of S. oblonga, S. prinoides and S. reticulata, exhibited potent inhibitory activity on the enzyme α -glucosidase (Matsuda et al., 1999). Neoponkoranol (157) and Neosalaprinol (158), the de-o-sulfonate derivatives of 162 and 167 respectively, obtained from the stems of S. chinensis, also showed inhibitory activity toward this enzyme(Xie et al., 2011). A polyhydroxylated cyclic 13-membered sulfoxide (161) found in the stems of S. reticulata showed even higher α -glucosidase inhibitory activity than the thiosugar sulfonium salts kotalanol (145) and salacinol (166) (Ozaki, Oe, and Kitamura, 2008). Furthermore, four lupane-type triterpenes i.e. $2\beta_3\beta$ -dihydroxylup-20(29)-ene (91), $3\alpha_2$ dihydroxylup-20(29)-en-2-one (92), 3α -hydroxylup-20(29)-en-2-one (141) and 2,3-seco-lup-20(29)-en-2,3-dioic acid (152), along with three dammarane-type triterpenes, 24S,25dihydroxytirucall-7-en-3-one (95), olibanumol (159) and 24,25,26-trihydroxytirucall-7-en-3-one (175), from the roots of S. hainanensis, showed stronger inhibiting activity on α -glucosidase than the anti-diabetic drug acarbose (Gao et al., 2010; Huang et al., 2012).

Phytochemical studies of *Salacia* species have yielded various types of chemical constituents. For example, lupane-type triterpenes such as betulin (87), 15,28-dihydroxylup-20(29)-en-3-one (93), 28-hydroxylup-20(29)-en-3-one (142), and 30-hydroxylup-20(29)-en-3-one (143), were isolated from the stem bark of *S. cordata* (Tinto, Blair, and Alli, 1992), while 2α ,3β-dihydroxylup-20(29)-ene (90), 20,29-epoxysalacianone (97), 6β-hydroxysalacianone

(144), salacianol (164) and salacianone (165) were isolated from the stem bark of *S. beddomei* (Hisham *et al.*, 1995, 1996a).

Several types of glycosides named foliachinenosides A_1 - A_3 , B_1 - B_2 , C-I (**99-110**) (Nakamura *et al.*, 2008b, 2011), along with the megastigmane glycosides foliasalaciosides A_1 - A_2 , B_1 - B_2 , C-D, E_1 - E_3 , F-L (**122-137**) (Nakamura *et al.*, 2008a, 2011; Zhang *et al.*, 2008a), the dammarane-type triterpenes foliasalacins A_1 - A_4 (**111-114**), the lupane-type triterpenes foliasalacins B_1 - B_3 (**115-117**), the oleanane-type triterpene foliasalacin C (**118**) (Yoshikawa *et al.*, 2008b), the baccharane triterpenes foliasalacins D_1 - D_3 (**119-121**) (Zhang *et al.*, 2008b) and the acylated eudesmane-type sesquiterpenes salasols A-B (**168-169**) (Kishi *et al.*, 2003; Morikawa *et al.*, 2003), were isolated from the leaves of *S. chinensis* collected in Thailand.

A number of stilbene derivatives, named lehmbachols A-D (**147-150**), have been isolated from the bark of *S. lehmbachii*, collected in Papua New Guinea (Kawazoe *et al.*, 1997).

Other chemical constituents reported as constituents of members of the genus *Salacia* are shown in **Table 2**, and their chemical structures are presented in **Figure 4**.

Compound	Source	Plant part	References
α-Amyrin (85)	Salacia amplifolia		Wang <i>et al.</i> , 2011a
β-Amyrin (86)	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. hainanensis		Gao et al., 2010
Betulin (87)	S. beddomei	Stem bark	Hisham <i>et al.</i> , 1996a
	S. cordata		Tinto <i>et al.</i> , 1992
Coniferaldehyde (88)	C. martifalia	Deste	Wang of al. 2011.
Dibutyl phthalate (89)	S. ampiijolia	Rools	wang <i>et al.</i> , 2011a
2α,3β-Dihydroxylup-20(29)-	S. beddomei	Stem bark	Hisham <i>et al.</i> , 1996a
ene (90)			
2β,3β-Dihydroxylup-20(29)-			Huang <i>et al.</i> , 2012
ene (91)			
3α,28-Dihydroxylup-20(29)-	S. hainanensis	Roots	Gao et al., 2010
en-2-one (92)			
15,28-Dihydroxylup-20(29)-	S. cordata	Stem bark	Tinto et al., 1992
en-3-one (93)			
3β,22β -Dihydroxyolean-12-	S. oblonga		Matsuda et al., 1999
en-29-oic acid (94)		Deete	
24S,25-Dihydroxytirucall-7-	S. hainanensis	Roots	Gao et al., 2010
en-3-one (95)			
Dulcitol (96)	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
	S. elliptica	Leaves	Duarte et al., 2010
	S. oblonga	Roots	Matsuda et al., 1999
	S. verrucosa	Stems	Jangruang et al., 2009
20,29-Epoxysalacianone (97)	S. beddomei	Stem bark	Hisham et al., 1996a
Ethyl glucopyranoside (98)	S. elliptica	Leaves	Duarte et al., 2010

Table 2. Chemical constituents of plants in the genus Salacia

Compound	Source	Plant part	References
Foliachinenoside A_1 (99)			
Foliachinenoside A_2 (100)			
Foliachinenoside A ₃ (101)			
Foliachinenoside B_1 (102)			Nakamura <i>el al.</i> ,
Foliachinenoside B_2 (103)			20080
Foliachinenoside C (104)			
Foliachinenoside D (105)			
Foliachinenoside E (106)			
Foliachinenoside F (107)			
Foliachinenoside G (108)			Nakamura et al., 2011
Foliachinenoside H (109)			
Foliachinenoside I (110)			
Foliasalacin A ₁ (111)		т	
Foliasalacin A_2 (112)	Salacia chinensis	Leaves	
Foliasalacin A ₃ (113)			
Foliasalacin A_4 (114)			Yoshikawa et al.,
Foliasalacin B_1 (115)			2008b
Foliasalacin B_2 (116)			
Foliasalacin B ₃ (117)			
Foliasalacin C (118)			
Foliasalacin D ₁ (119)			
Foliasalacin D_2 (120)			Zhang et al., 2008b
Foliasalacin D_3 (121)			
Foliasalacioside A_1 (122)			
Foliasalacioside A_2 (123)			Nakamura <i>et al.</i> ,
Foliasalacioside B_1 (124)			2000a

Compound	Source	Plant part	References
Foliasalacioside B2 (125)Foliasalacioside C (126)Foliasalacioside D (127)			Nakamura <i>et al.</i> , 2008a
Foliasalacioside E_1 (128)Foliasalacioside E_2 (129)Foliasalacioside E_3 (130)Foliasalacioside F (131)Foliasalacioside G (132)Foliasalacioside H (133)Foliasalacioside I (134)	Salacia chinensis	Leaves	Zhang <i>et al.</i> , 2008a
Foliasalacioside J (135) Foliasalacioside K (136) Foliasalacioside L (137)	-		Nakamura <i>et al.</i> , 2011
Gult-5-en-3β-ol (138) (-)-Gynuraone (139)	S. amplifolia		Wang et al., 2011a
19-Hydroxyferruginol (140)	S. oblonga	Roots	Matsuda et al., 1999
3α-Hydroxylup-20(29)-en-2- one (141)	S. hainanensis		Gao et al., 2010
28-Hydroxylup-20(29)-en-3- one (142) 30-Hydroxylup-20(29)-en-3- one (143)	S. cordata	Stem bark	Tinto <i>et al.</i> , 1992
6β-Hydroxysalacianone (144)	S. beddomei		Hisham <i>et al.</i> , 1996a
Kotalanol (145)	S. oblonga S. prinoides	Roots Roots, Stems	Matsuda <i>et al.</i> , 1999 Yoshikawa <i>et al.</i> , 2008a

Compound	Source	Plant part	References
Lambertic acid (146)	Salacia oblonga	Roots	Matsuda et al., 1999
Lehmbachol A (147)			
Lehmbachol B (148)	C. I. have been bit	Devle	V
Lehmbachol C (149)	S. lenmbachti	Bark	Kawazoe <i>et al.</i> , 1997
Lehmbachol D (150)			
Lup-20(29)-en-3-one (151)	S. beddomei	Stem bark	Hisham <i>et al.</i> , 1995
2,3-seco-Lup-20(29)-en-2,3-	S. hainanensis	Roots	Gao et al., 2010
dioic acid (152)			
Mangiferin (153)	S. chinensis	Stems	Kishi <i>et al.</i> , 2003
	S. reticulata	Root bark	Karunanayake and
			Sirimanne, 1985
Maytenfolic acid (154)	S. oblonga	Roots	Matsuda et al., 1999
Methyl-2,4-dihydroxy	S. elliptica	Branches	Duarte et al., 2010
-3,6-dimethylbenzoate (155)			
(-)-4'-0-	S. oblonga	Roots	Matsuda et al., 1999
Methylepigallocatechin (156)			
Neoponkoranol (157)	a li i	G.	X: (1 2011
Neosalaprinol (158)	S. chinensis	Stems	Xie <i>et al.</i> , 2011
Olibanumol J (159)	S. hainanensis	Roots	Huang <i>et al.</i> , 2012
Palmitic acid (160)	S. elliptica	Leaves	Duarte et al., 2010
Polyhydroxylated cyclic	S. reticulata	Stems	Ozaki <i>et al.</i> , 2008
-13-membered sulfoxide (161)			
Ponkoranol (162)	S. prinoides	Roots,	Yoshikawa <i>et al.</i> ,
		Stems	2008a
Pyracrenic acid (163)	S. cordata	Stone hards	Tinto <i>et al.</i> , 1992
Salacianol (164)	S. beddomei	Stem Dark	Hisham <i>et al.</i> , 1995

Compound	Source	Plant part	References
Salacianone (165)	Salacia beddomei	Stem bark	Hisham <i>et al.</i> , 1995
	S. hainanensis	Roots	Gao et al., 2010
Salacinol (166)	S. oblonga	Roots	Matsuda et al., 1999
	S. prinoides	Roots,	Yoshikawa et al.,
		Stems	2008a
	S. reticulata	Roots	Yoshikawa et al.,
			1997
Salaprinol (167)	S. prinoides	Roots,	Yoshikawa et al.,
		Stems	2008a
Salasol A (168)		C.	Morikawa et al., 2003
Salasol B (169)	S. chinensis	Stems	Kishi et al., 2003
Sinapic aldehyde (170)	S. amplifolia	Roots	Wang <i>et al.</i> , 2011a
β -Sitosterol (171)			
β -Sitosterol glucoside (172)			
3β-Stearyloxyolean-12-en	S. elliptica	Leaves	Duarte et al., 2010
(173)			
3β-Stearyloxyurs-12-en (174)			
24,25,26-Trihydroxytirucall-7-	S. hainanensis		Huang <i>et al.</i> , 2012
en-3-one (175)			
Wilforlide A (176)		Koots	Wenne et al. 2011
Wilforlide B (177)	s. атријона		wang <i>et at.</i> , 2011a







Dibutyl phthalate (89)

Figure 4. Chemical constituents of plants in the genus Salacia



	R_1	R ₂	R ₃	R_4	R ₅
Betulin (87)	H_2	β-ΟΗ, α-Η	Н	CH ₂ OH	CH ₃
2α,3β-Dihydroxylup-20(29)-ene	α-ОН, β-Н	β-ΟΗ, α-Η	Н	CH ₃	CH ₃
(90)					
2β,3β-Dihydroxylup-20(29)-ene	β-ΟΗ, α-Η	β-ΟΗ, α-Η	Н	CH ₃	CH ₃
(91)					
3α,28-Dihydroxylup-20(29)-en-2-	0	α-ОН, β-Н	Н	CH ₂ OH	CH ₃
one (92)					
15,28-Dihydroxylup-20(29)-en-3-	H_2	0	OH	CH ₂ OH	CH ₃
one (93)					
3α-Hydroxylup-20(29)-en-2-one	О	α-ОН, β-Н	Н	CH ₃	CH ₃
(141)					
28-Hydroxylup-20(29)-en-3-one	H_2	0	Н	CH ₂ OH	CH ₃
(142)					
30-Hydroxylup-20(29)-en-3-one	H_2	0	Н	CH ₃	CH ₂ OH
(143)					
Lup-20(29)-en-3-one (151)	H_2	0	Н	CH ₃	CH ₃
Pyracrenic acid (163)	H_2	β-OCaf*, α-H	Η	CO_2H	CH_3

*Caf = 3,4-dihydroxycinnamoyl





24*S*,25-Dihydroxytirucall-7-en-3-one (95) $R = CH_3$

24,25,26-Trihydroxytirucall-7-en-3-one (175) $R = CH_2OH$





20,29-Epoxysalacianone (97)



Ethyl glucopyranoside (98)













Foliachinenoside G (108)



Foliachinenoside I (110)



Foliachinenoside D (105)



Foliachinenoside F (107)



Foliachinenoside H (109)



R_2

Foliasalacin A_1 (111) OH CH_3

Foliasalacin A_2 (112) CH_3 OH





Foliasalacin A₃ (113)

Foliasalacin A_4 (114)



Foliasalacin C (118)

	R_1	R_2
Foliasalacin B_1 (115)	CH ₂ OH	CH ₃
Foliasalacin B_2 (116)	CH ₃	CH ₂ OH
Foliasalacin B ₃ (117)	CO ₂ H	CH ₃



Foliasalacin D_1 (119)









Foliasalacioside A_1 (122) $R_1 = Glc$, $R_2 = H$ Foliasalacioside A_2 (123) $R_1 = H$, $R_2 = Glc$

Foliasalacioside B_1 (124) $R = Glc^6$ -Ara(p) Foliasalacioside B_2 (125) $R = Glc^6$ -Ara(f)

		R_1	R_2	R ₃
R1 OR3	Foliasalacioside C (126)	0	H_2	Glc ⁶ -Ara(p)
	Foliasalacioside D (127)	H_2	0	Glc ⁶ -Ara(f)
 R ₂				

		R_1	R_2
۰. ۱۰	Foliasalacioside E_1 (128)	Glc ⁶ -Ara(p)	Н
OR ₂	Foliasalacioside E_2 (129)	Н	Glc ⁶ -Ara(p)
R ₁ O ^{µµµ}	Foliasalacioside E_3 (130)	Н	Glc ⁶ -Ara(f)

















Foliasalacioside J (135)



Foliasalacioside L (137)



(-)-Gynuraone (**139**)



Gult-5-en-3 β -ol (138)



19-Hydroxyferruginol (140) $R = CH_2OH$

Lambertic acid (146) $R = CO_2H$









Salacinol (166)







Lehmbachol D (150)



2,3-seco-Lup-20(29)-en-2,3-dioic acid (152)







Methyl-2,4-dihydroxy-3,6-dimethylbenzoate (155)



Neoponkoranol (157) R = H



Olibanumol J (159)

(-)-4'-O-Methylepigallocatechin (156)



Neosalaprinol (158) R = HSalaprinol (167) $R = SO_3^-$



Palmitic acid (160)




Polyhydroxylated cyclic 13-membered sulfoxide (161)







Salasol B (169)





 $\begin{array}{ccc} R_1 & R_2 \\ 3\beta \text{-Stearyloxyolean-12-en (173)} & H & CH_3 \end{array}$

3 β -Stearyloxyurs-12-en (174) CH₃ H





9. Eudesmane-type Sesquiterpenes in Higher Plants

Eudesmane-type sesquiterpenes are widely distributed in the dicotyledonous families of flowering plants such as Apocynaceae, Asteraceae, Chloranthaceae, Euphorbiaceae, Lamiaceae, Lauraceae, Leguminosae, Moraceae, Sapindaceae and Umbelliferae, and can also be found in some monocotyledonous families such as Gramineae, Liliaceae and Zingiberaceae. However, the plant family Asteraceae is the largest source of these compounds.

Several genera in the family Asteraceae including Artemisia, Chrysanthemum, Inula and Varthemia yielded various bioactive eudesmane sesquiterpenes and eudesmane sesquiterpene lactones (eudesmanolides). Artanoate (185) and eudesmanomolide (219), from the aerial parts of Artemisia anomala, were cytotoxic against cancer cell lines in vitro. Artanoate was cytotoxic against HCT-8 cell line (IC₅₀ = 9.13 μ M) and eudesmanomolide exhibited inhibitory activities against both HCT-8 (IC₅₀ = 3.76 μ M) and A549 cell lines (IC₅₀ = 5.49 μ M) (Zan *et al.*, 2012). Methanolic extract from the flowers of Chrysanthemum indicum showed inhibitory activity against rat lens aldose reductase (IC₅₀ = 3.5 μ g/ml), then the active components, kikkanols A-C (259-261), were isolated by bioassay-guided fractionation (Yoshikawa et al., 1999). Essential oil from the roots of *Inula helenium*, containing the eudesmanolides alantolactone (184), diplophyllin (208) and isoalantolactone (256) as major constituents, exhibited very potent antistaphylococcal activity through membrane-damaging effects (Stojanovic-Radic et al., 2012). Similarly, the eudesmane-12,8-olides 11α ,13-dihydro-2 α -hydroxyalantolactone (198), 11,13dihydroivalin (199) and septuplinolide (300), from the roots of Inula racemosa, exhibited moderate cytotoxicity against a number of human cancer cell lines, with IC_{50} values ranging from 25.2 to 39.3 µM (Zhang et al., 2012). 3-Oxocostusic acid [selina-4,11(13)-dien-on-12-oic acid] (281), a constituent of Artemisia altaiensis (Khanina et al., 1998), Nectandra cissiflora (Garcez, Garcez, and Miranda, 2010) and Varthemia iphionoides (Al-Dabbas et al., 2005), was shown to be a potent antimicrobial agent against six bacterial species including *Staphylococcus aureus*, Bacillus subtilis, Micrococcus luteus, Escherichia coli, Bacillus cereus and Salmonella enteritides with MIC values from 250 to 500 µg/ml (Al-Dabbas et al., 2005).

Chemical investigation of the chloroform extract of *Caragana intermedia* aerial part (family Leguminosae) yielded several 4(15)-eudesmene sesquiterpenes, including eudesma-4(15),7-dien-1 β -ol (**218**), 4(15)-eudesmene-1 β ,5 α -diol (**221**), 4(15)-eudesmene-1 β ,6 α -diol (**222**),

4(15)-eudesmene-1 β ,7 α -diol (223), 4(15)-eudesmene-1 β ,7 β -diol (224), 5-*epi*-eudesma-4(15)ene-1 β ,6 β -diol (225) and 7-trinoreudesma-4(15),8-dien-1 β -ol-7-one (308). The bioassay showed that compounds 221, 223 and 224 exhibited antifungal activity against the rice blast fungus *Pyricularia oryzae* P-2b, with MIC values of 20, 12, and 16 µg/ml, respectively. 4(15)-Eudesmene-1 β ,5 α -diol also showed effect on energy metabolism by stimulating glucose consumption in C₂C₁₂ skeletal muscle cells with an IC₅₀ value of 10.7 µg/ml and in db/db mice with an MIC value of 100 mg/kg. 5-*epi*-Eudesma-4(15)-ene-1 β ,6 β -diol exhibited weak anti-HIV activity with an IC₅₀ value of 10 µg/ml (Sun *et al.*, 2004).

Two trinoreudesmane sesquiterpenes, oxyphyllanenes A-B (**285-286**), a noreudesmane sesquiterpene, oxyphyllanene C (**287**), and four eudesmane sesquiterpenes, oxyphyllanenes D-G (**288-291**), were isolated from the fruits of *Alpinia oxyphylla* (family Zingiberaceae). Among them, compounds **287-290** showed inhibitory activity against nitric oxide production in lipopolysaccharide and interferon- γ -induced RAW 264.7 murine macrophages (Xu *et al.*, 2012).

Eudesmane sesquiterpene glucosides, litchiosides A-B (263-264) and pumilaside A (299), were isolated from the seeds of *Litchi chinensis* (family Sapindaceae). Pumilaside A exhibited significant cytotoxicity against cancer cell lines (A549, LAC, Hela and HepG2), with IC_{50} values of 0.012-6.29 μ M (Xu et al., 2010). This compound was firstly isolated from the fruits of Ficus pumila in the family Moraceae (Kitajima, Kimizuka, and Tanaka, 2000). A sesquiterpene lactone glucoside, $1-O-\beta$ -D-glucopyranosyl-9 β , 15-dihydroxy-5 α , 6 β H-eudesma-3ene- 6α ,12-olide (233), from the seed pods of *Bauhinia retusa* (family Leguminosae), displayed moderate antibacterial activity against *Pseudomonas aeruginosa* and *Escherichia coli* in the disc diffusion method (Semwal and Sharma, 2011). Other eudesmane sesquiterpene glucosides such as 1β -O- β -D-glucopyranosyl- 4α -hydroxyl- 5α , 6β , 11β H-eudesma-12, 6α -olide (**234**) from Lactuca sativa var. anagustata (family Asteraceae) (Han, Cao, and Xia, 2009), hierapolitanins C-D (236-237) from Centaurea hierapolitana (family Asteraceae) (Karamenderes et al., 2007), liriopeoside A (262) and ophiopogonosides A-B (279-280) from Liriope muscari and Ophiopogon japonicus (family Liliaceae) (Cheng et al., 2004; Jiang et al., 2012), and pterodontriol D-6-O- β -Dglucopyranoside (298) from Parepigynum funingense (family Apocynaceae) (Hua et al., 2004), have also been reported.

Eudesmane sesquiterpenes with rearranged carbon skeletons, such as diversifolol (209) and macrophyllic acids A-E (265-269), were reported as constituents of plants in the family Asteraceae, including *Tithonia diversifolia* (Kuo and Chen, 1997) and *Inula macrophylla* (Su *et al.*, 2000b), respectively. This family also contains sesquiterpenoid dimers, consisting of two eudesmanolide units, such as biatractylolide (188) from *Atractylodes macrocephala* (Lin *et al.*, 1997) and hydroxy-bis-dihydroencelin (239) from *Montanoa speciosa* (Quijano *et al.*, 1991). Dimers of two different sesquiterpenoid units, such as fruticolide (230) from *Ferreyranthus fruticosus* (Jakupovic *et al.*, 1988) and macrophyllidimer A (270) from *Inula macrophylla* (Su *et al.*, 2000a), have also been reported.

Eudesmane sesquiterpenes have been found in other flowering plants including *Chloranthus henryi* (Wu *et al.*, 2007), *Chloranthus japonicus* (Fang, Liu, and Zhong, 2012), *Cymbopogon proximus* (El-Askary, Meselhy, and Galal, 2003), *Dittrichia graveolens* (Abou-Douh, 2008), *Isodon grandifolia* var. *atuntzensis* (Wu *et al.*, 1993), *Laggera alata* (Raharivelomanana *et al.*, 1998; Zheng *et al.*, 2003), *Laggera pterodonta* (Zhao *et al.*, 1997; Liu *et al.*, 2007), *Ligularia dentata* (Naya *et al.*, 1990), *Melampodium camphoratum* (Chaturvedula *et al.*, 2004), *Plectranthus cylindraceus* (Orabi *et al.*, 2000), *Pluchea arguta* (Ahmad, Fizza, and Amber, 1989), *Pluchea dioscoridis* (Mahmoud, 1997), *Sambucus williamsii* (Yang *et al.*, 2006), *Tanacetum praeteritum* subsp. *praeteritum* (Goren, 1996), *Teucrium heterophyllum* (Fraga *et al.*, 1993), *Teucrium polium* (Kamel, 1995) and *Torilis japonica* (Kitajima *et al.*, 2002).

Eudesmane sesquiterpenes found in these flowering plants are summarized in **Table 3** and their chemical structures are shown in **Figure 5**.





Eudesmane



Compound	Source	Family	Plant part	References
6α -({4'-Acetoxy}-7'Z	Melampodium		Leaves	Chaturvedula
-coumaryloxy)eudesm	camphoratum			et al., 2004
-4(14)-ene (178)				
1α-Acetoxy-3α-hydroxy		Astorococo		
-5,7 α ,6,11 β (H)-eudesm		Asteraceae	l	Maalala a waa 41ala a
-4,15-en-6,12-olide (179)	Artemisia		A amial manta	
1α-Acetoxy-4α-hydroxy	lehmanniana		Aerial parts	-nova <i>el al.</i> ,
-5,7 α ,6,11 β (H)-eudesm				2004
-2,3-en-6,12-olide (180)				
11-Acetoxy-4α	Allophylus	Sapindaceae	Fruits	David,
-methoxyeudesmane	laevigatus			Santos, and
(181)				David, 2004
13-Acetyloxy-5,7(11)	Inula	Asteraceae		Zhang et al.,
-eudesmadien-12,8-olide	racemosa			2012
(182)				
Ainsliaside E (183)	Parepigynum	Apocynaceae	Poots	Hua et al.,
	funingense		KOOIS	2004
Alantolactone (184)	I. helenium			Stojanovic-
				Radic et al.,
				2012
Artanoate (185)	Artemisia			Zan <i>et al</i> .,
	anomala	Asteraceae	Aerial parts	2012
	Tanacetum			Goren, 1996
	praeteritum			
Artapshin (186)	Artemisia			Sanz and
	caerulescens			Marco, 1990

Table 3. Distribution of eudesmane-type sesquiterpenes in higher plants

Compound	Source	Family	Plant part	References
Atractylenolide III (187)	Chloranthus	Chloranthaceae	Aerial parts	Fang et al.,
	japonicus			2012
Biatractylolide (188)	Atractylodes	Asteraceae	Branches	Lin et al.,
	macrocephala			1997
12-Carboxyeudesman	Nectandra	Lauraceae	Bark	Garcez et al.,
-3,11(13)-diene (189)	cissiflora			2010
Chlojaponilactone A (190)	C. japonicus	Chloranthaceae	Aerial parts	Fang <i>et al</i> .,
				2012
Costic acid (191)	Nectandra	Lauraceae	Bark	Garcez et al.,
	cissiflora			2010
6α -(7 ['] Z-Coumaryloxy)	Melampodium	Asteraceae	- Leaves	Chaturvedula
-eudesm-4(14)-ene (192)	camphoratum			et al., 2004
Crytomeridiol (193)	Isodon	Lamiaceae		Wu et al.,
	grandifolia			1993
1,2-Dehydro-3-epi				
-isotelekin (194)				
1,2-Dehydro-3-epi	Montanoa	A - 4 - 11		Quijano et al.,
-isotelekin acetate (195)	speciosa	Asteraceae	A 11 /	1991
1,2-Dehydro-3-oxo-costic			Aerial parts	
acid (196)				
Desacetylplectranthone	Plectranthus	Lamiaceae		Orabi <i>et al.</i> ,
(197)	cylindraceus			2000
11α,13-Dihydro-2α				
-hydroxyalantolactone	Inula	A - 4 - 11	Desta	Zhang et al.,
(198)	racemosa	Asteraceae	KOOIS	2012
11,13-Dihydroivalin (199)				

Compound	Source	Family	Plant part	References
4β,11-Dihydroxy	Laggera			Zhao et al.,
-enantioeudesmane-1-one	pterodonta			1997
(200)				
5α,11-Dihydroxy-3-ene	Artemisia	Asteraceae	Aerial parts	Hu, Bai, and
-eudesman-2-one (201)	eriopoda			Jia, 1996
	L. pterodonta			Zhao et al.,
				1997
1β,6α-Dihydroxy-7- <i>epi</i>	Croton	Euphorbiaceae	Aerial	Aguilar-
-eudesm-3-ene (202)	arboreous		parts	Guadarrama
				and Rios,
				2004
	Pluchea	Asteraceae	Leaves	Mahmoud,
	dioscoridis			1997
1β,6β-Dihydroxy-7-epi	P. dioscoridis		Leaves	Mahmoud,
-eudesm-3-ene (203)				1997
4α,13-Dihydroxy-5,7(11)	Inula		Roots	Zhang et al.,
-eudesmadien-12,8-olide	racemosa			2012
(204)				
1β,9α-Dihydroxyeudesm		Astaragaaa		
-4(15),11(13)-dien-		Asteraceae		
5α,7αΗ				Mahmand
-12,6α-olide (205)	P. dioscoridis		Leaves	
4β,5α-Dihydroxy-15-oxo				177/
-eudesm-11(13)-en-12-oic				
acid (206)				

Compound	Source	Family	Plant part	References
4β,5β-Dihydroxy-15-oxo	Pluchea		Leaves	Mahmoud,
-eudesm-11(13)-en-12-oic	dioscoridis			1997
acid (207)				
Diplophyllin (208)	Inula	Asterosooo	Roots	Stojanovic-
	helenium	Asteraceae		Radic et al.,
				2012
Diversifolol (209)	Tithonia		Leaves	Kuo and
	diversifolia			Chen, 1997
4β,7β,11-Enantio	Coleus	Lamiaceae		Shan <i>et al.</i> ,
-eudesmantriol (210)	forskohlii			2007
Encelin (211)	Montanoa			Quijano et al.,
	speciosa	- Asteraceae	1991	
Ent-7(11)-selinen-4-ol	Laggera			Zhao et al.,
(212)	pterodonta			1997
10- <i>epi</i> -5α-hydroperoxy-β				
-eudesmol (213)				Itokawa,
10- <i>epi</i> -5β-hydroperoxy-β	Alpinia	7 . 1	Rhizomes	Morita, and
-eudesmol (214)	japonica	Zingiberaceae		Watanabe,
4,10- <i>epi</i> -5β-hydroxy				1987
-dihydroeudesmol (215)				
4α,15-Epoxyeudesmane	Torilis	Umbelliferae	Fruits	Kitajima <i>et</i>
-1β,6α-diol (216)	japonica			al., 2002
Epoxysantamarin (217)	Tanacetum	Asteraceae		Goren, 1996
	praeteritum			
Eudesma-4(15),7-dien -1 β	Caragana	Leguminosae	Aerial parts	Sun et al.,
-ol (218)	intermedia			2004

Compound	Source	Family	Plant part	References
Eudesmanomolide (219)	Artemisia	Asteraceae		Zan <i>et al.</i> ,
	anomala			2012
5α,10β -4(15)-	Croton	Euphorbiaceae	A anial manta	Aguilar-
Eudesmen-1β,6β-diol	arboreous		Aeriai paris	Guadarrama
(220)				and Rios,
				2004
4(15)-Eudesmene-1 β ,5 α	Caragana	Leguminosae	Aerial parts	Sun <i>et al.</i> ,
-diol (221)	intermedia			2004
	Torilis japonica	Umbelliferae	Fruits	Kitajima <i>et</i>
				al., 2002
4(15)-Eudesmene-1 β ,6 α	Ageratina	Asteraceae	Aerial parts	Gonzalez et
-diol (222)	glechonophylla			al., 1989
	C. intermedia	Leguminosae	Aerial parts	Sun et al.,
				2004
	Torilis japonica	Umbelliferae	Fruits	Kitajima <i>et</i>
				al., 2002
4(15)-Eudesmene-1 β ,7 α				
-diol (223)				
4(15)-Eudesmene-1 β ,7 β	C internetin	T	A	Sun <i>et al.</i> ,
-diol (224)	C. intermeata	Leguminosae	Aerial parts	2004
5-epi-Eudesma-4(15)-				
ene-1β,6β-diol (225)				
7-epi-Eudesm-4(15)-ene				
-1β,6α-diol (226)	Teucrium	- ·	-	
7-epi-Eudesm-4(15)-ene	polium	Lamiaceae	Leaves	Kamel, 1995
-1β,6β-diol (227)				

Compound	Source	Family	Plant part	References
7- <i>epi</i> -γ-Eudesmol (228)	Turner and a last a			Raharivelomana
7- <i>epi</i> -β-Eudesmol (229)	Laggera aiaia			-na <i>et al.</i> , 1998
Fruticolide (230)	Ferreyranthus	Astensoro	Aerial	Jakupovic et al.,
	fruticosus	Asteraceae	parts	1988
Gargantolide (231)	Artemisia			Sanz and Marco,
1-epi-Gargantolide (232)	caerulescens			1990
1-О-β-D-	Bauhinia	Leguminosae	Seed pods	Semwal and
Glucopyranosyl	retusa			Sharma, 2011
-9β,15-dihydroxy-				
5α,6βH-eudesma-3-ene-				
6α,12 -olide (233)				
1β- <i>О</i> -β-D-	Lactuca		Stalk	Han <i>et al.</i> , 2009
Glucopyranosyl	<i>sativa</i> var.			
-4α-hydroxyl-	anagustata L.			
$5\alpha, 6\beta, 11\beta$ H-eudesma-				
12,6α-olide (234)		Asteraceae		
$6\alpha - (\{4' - O - [9''Z$	Melampodium		Leaves	Chaturvedula
-Hexadecenoyl]}-7'E	camphoratum			<i>et al.</i> , 2004
-coumaryloxy)eudesm				
-4(15)-ene (235)				
Hierapolitanin C (236)	Centaurea	Asteraceae		Karamenderes
Hierapolitanin D (237)	hierapolitana			et al., 2007
5α-Hydroperoxy-β	Cymbopogon	Gramineae	Aerial	El-Askary et al.,
-eudesmol (238)	proximus		parts	2003
Hydroxy-bis-	Montanoa	A = 4 = 11 = 1		Quijano et al.,
dihydroencelin (239)	speciosa	Asteraceae		1991

Compound	Source	Family	Plant part	References
1β-Hydroxy-α-cyperone	Artemisia		Aerial	Sanz and Marco,
(240)	caerulescens		parts	1990
	Ligularia		Rhizomes	Naya <i>et al.</i> , 1990
	dentata			
2α-Hydroxy-4- <i>epi</i> -ilicic	Dittrichia		Epigeal	Abou-Douh,
acid (241)	graveolens		parts	2008
5α-Hydroxy-4- <i>epi</i> -ilicic	Laggera alata			Zheng et al.,
acid methyl ester (242)		Asteraceae		2003
8α-Hydroxy-1,4β-epoxy	A. caerulescens		Aerial	Sanz and Marco,
-10-epieudesman			parts	1990
-5α,6β,7α,11βΗ				
-12,6-olide (243)				
5α-Hydroxy-eudesma	A. annua		Leaves	Sy and Brown,
-4(15),11-diene (244)				1998
1β-Hydroxy-α-eudesmol				
(245)				
1β-Hydroxy-β-eudesmol	Cymbopogon	c ·		El-Askary et al.,
(246)	proximus	Gramineae		2003
5α-Hydroxy-β-eudesmol			Aerial	
(247)			parts	
3α-Hydroxyilicic acid				
(248)	T I (Zheng et al.,
5β-Hydroxyilicic acid	L. alata	Asteraceae		2003
(249)				
3α-Hydroxyilicic acid	D. graveolens		Epigeal	Abou-Douh,
methyl ester (250)			parts	2008

Compound	Source	Family	Plant part	References
1β-Hydroxy-15- <i>O</i> -(<i>p</i> -	Lactuca sativa		Stalks	Han <i>et al.</i> , 2009
methoxyphenylacetyl)	var. anagustata			
-5α,6β,11βH-eudesma-3	L.			
-en-12,6α-olide (251)				
1α-Hydroxypinnatifidin	Ferreyranthus			Jakupovic et al.,
(252)	fruticosus			1988
2α-Hydroxypterodontic	Laggera	Asteraceae		Liu et al., 2007
acid (253)	pterodonta		A	
8α-Hydroxytaurin (254)	Artemisia	Asteraceae	Aerial parts	Sanz and
	caerulescens			Marco, 1990
Ilicic acid (255)	Aster			Xie et al., 2010
	himalaicus			
Isoalantolactone (256)	Inula helenium		Roots	Stojanovic-
				Radic et al.,
				2012
Isodeacetylplectranthone	Plectranthus		Aerial parts	Orabi <i>et al.</i> ,
(257)	cylindraceus	Lamiaaaa		2000
Isodonsesquitin A (258)	Isodon	Lamiaceae	Leaves	Wu et al., 1993
	grandifolia			
Kikkanol A (259)				X7 1 '1
Kikkanol B (260)	Chrysanthemum	Asteraceae	Flowers	Y oshikawa
Kikkanol C (261)	ากลาะนท			<i>et al.</i> , 1999
Liriopeoside A (262)	Liriope muscari	Liliaceae	Tubers	Cheng et al.,
				2004
Litchiosides A (263)	T 1 . 1	0 1	0 1	X / 1 2010
Litchiosides B (264)	Litchi chinensis	Sapindaceae	Seeds	Au et al., 2010

Compound	Source	Family	Plant part	References
Macrophyllic acid A (265) Macrophyllic acid B (266)	-			
Macrophyllic acid C (267)	Inula			Su <i>et al.</i> , 2000b
Macrophyllic acid D (268)	macrophylla		Bark	
Macrophyllic acid E (269)				
Macrophyllidimer A (270)				Su <i>et al.</i> , 2000a
Macrophyllilactone E	I. macrophylla		Bark	Fu et al., 2001
(271)	I. racemosa	Asteraceae	Roots	Zhang et al.,
				2012
Macrophyllilactone F	I. macrophylla			
(272)			D 1	E (1 2001
Macrophyllilactone G			Bark	Fu <i>et al.</i> , 2001
(273)				
6α -({4'-O-Methyl}-7'E	Melampodium	-	Leaves	Chaturvedula
-coumaryloxy)eudesm	camphoratum			et al., 2004
-4(15)-ene (274)				
Neolitacumone B (275)	Chloranthus	Chlorantha	Aerial parts	Fang <i>et al.</i> ,
	japonicus	-ceae		2012
Odonticin (276)			Whole	Ahmad et al.,
Odontin (277)	Pluchea arguta		plants	1989
1-One-4-epi-alantolactone	I. racemosa	Asteraceae		Zhang et al.,
(278)			Desta	2012
Ophiopogonoside A (279)	Liriope muscari	Liliaceae	KOOIS	Cheng et al.,
				2004

Compound	Source	Family	Plant part	References
Ophiopogonoside B	Liriope	Liliaceae	Roots	Jiang <i>et al.</i> ,
(280)	muscari			2012
3-Oxocostusic acid	Artemisia	Asteraceae	Epigeal part	Khanina
(281)	altaiensis			<i>et al.</i> , 1998
	Nectandra	Lauraceae	Trunk bark	Garcez
	cissiflora			et al., 2010
	Varthemia	Asteraceae	Aerial parts	Al-Dabbas
	iphionoides			et al., 2005
1-Oxo-cryptomeridiol	Artemisia	Asteraceae	Aerial parts	Hu et al., 1996
(282)	eriopoda			
(7α)-8-Oxoeudesm-	Chloranthus	Chloranthaceae	Leaves,	Wu et al., 2007
4(15)-en-12-oic acid	henryi		Stems	
(283)				
1-Oxoeudesm-11(13)-	Aster	Asteraceae	Aerial parts	Xie et al., 2010
eno-12,8α-lactone (284)	himalaicus			
Oxyphyllanene A (285)				
Oxyphyllanene B (286)				
Oxyphyllanene C (287)				
Oxyphyllanene D (288)	Alpinia	Zingiberaceae	Fruits	Xu et al., 2012
Oxyphyllanene E (289)	oxypnylla			
Oxyphyllanene F (290)				
Oxyphyllanene G (291)				
6α -({4'-O-Palmityl}-	Melampodium	Asteraceae	Leaves	Chaturvedula
7' <i>E</i> -coumaryloxy)	camphoratum			et al., 2004
eudesm-4(15)-ene (292)				

Compound	Source	Family	Plant part	References
2α,5α-Peroxyeudesma	Artemisia	Asteraceae		Sanz and
-3,11-dien-1-one (293)	caerulescens			Marco, 1990
Plectranthone (294)	Plectranthus	Lamiaceae		Orabi <i>et al.</i> ,
	cylindraceus		Aerial parts	2000
Pterodolide (295)	T			Liu et al., 2007
Pterodontriol C (296)	Laggera	Asteraceae		Zhao et al.,
Pterodontriol D (297)	pterodonta			1997
Pterodontriol D-6- <i>O</i> -β-D	Parepigynum	Apocynaceae	Roots	Hua et al., 2004
-glucopyranoside (298)	funingense			
Pumilaside A (299)	Ficus pumila	Moraceae	Fruits	Kitajima <i>et al</i> .,
				2000
	Litchi chinensis	Sapindaceae	Seeds	Xu et al., 2010
Septuplinolide (300)	Inula racemosa		Roots	Zhang et al.,
				2012
6α -({4'-O-Stearyl}-7'E	Melampodium		Leaves	Chaturvedula et
-coumaryloxy)eudesm	camphoratum	Asteraceae		al., 2004
-4(15)-ene (301)				
Tanapraetenolide (302)	Tanacetum			Goren, 1996
	praeteritum			
Teucdiol A (303)	Tauanium			Erece et al
Teucdiol B (304)	Leterorehullum	Lamiaceae	Aerial parts	1002
Teucrenone (305)	neteropnytium			1995
Tourneforin (306)	Artemisia	Asteraceae		Talzhanov et
	tournefortiana			al., 2007
1,4,13-Trihydroxy	Sambucus	Adoxaceae	Stems	Yang <i>et al.</i> ,
eudesm-11(12)-ene(307)	williamsii			2006

Compound	Source	Family	Plant part	References
7-Trinoreudesma-	Caragana	Leguminosae	Aerial parts	Sun et al., 2004
4(15),8-dien-1β-ol-7-	intermedia			
one (308)				
Viscic acid (309)	Nectandra	Lauraceae	Bark	Garcez
	cissiflora			et al., 2010
Yomogin (310)	Ferreyranthus	Asteraceae	Aerial parts	Jakupovic et al.,
	fruticosus			1988





 6α -({4'-Acetoxy}-7'Z-coumaryloxy) eudesm-4(14)-ene (178) R = COCH₃

1α-Acetoxy-3α-hydroxy-5,7α,6,11β(H) -eudesm-4,15-en-6,12-olide (**179**)

 6α -(7[']Z-Coumaryloxy) eudesm-4(14)-ene (192) R = H



1α-Acetoxy-4α-hydroxy-5,7α,6,11β(H) -eudesm-2,3-en-6,12-olide (**180**)



11-Acetoxy- 4α -methoxyeudesmane (181)



Figure 5. Eudesmane-type sesquiterpenes in higher plants



	R_1	R ₂	R ₃	R_4
Ainsliaside E (183)	β-ΟΗ, α-Η	α -OH, β -CH ₃	Н	OGlc
Crytomeridiol (193)	H_2	α -OH, β -CH ₃	Н	Н
5α -Hydroperoxy- β –eudesmol (238)	H_2	CH ₂	OOH	Н
1β -Hydroxy- β -eudesmol (246)	β-ΟΗ, α-Η	CH ₂	Н	Н
5α -Hydroxy- β -eudesmol (247)	H_2	CH ₂	ОН	Н
1-Oxo-cryptomeridiol (282)	0	α-OH, β-CH ₃	Н	Н







Biatractylolide (188)

Figure 5. (continued)









12-Carboxyeudesman-3,11(13)-diene (189)



	\mathbf{R}_1	R_2	R ₃
Costic acid (191)	Н	Н	CO ₂ H
5α-Hydroxy-eudesma	Н	OH	CH ₃
-4(15),11-diene (244)			
Viscic acid (309)	OH	Н	CO ₂ H



Chlojaponilactone A (190)



1,2-Dehydro-3-*epi*-isotelekin (194) $R = \beta$ -OH, α -H 1,2-Dehydro-3-*epi*-isotelekin acetate (195) $R = \beta$ -OAc, α -H Encelin (211) R = O

Figure 5. (continued)



1,2-Dehydro-3-oxo-costic acid (196)



11,13-Dihydroivalin (**199**)

OR₂

OR₁

H

0



11α,13-Dihydro-2α-hydroxyalantolactone (**198**)



5α,11-Dihydroxy-3-ene-eudesman-2-one (201)

	R_1	R_2
Desacetylplectranthone (197)	COCH ₃	Н
Isodeacetylplectranthone (257)	Н	COCH ₃
Plectranthone (294)	COCH ₃	COCH ₃

		R ₁	R ₂	R ₃
R1	4β ,11-Dihydroxy enantioeudesmane-1-one (200)	0	Н	Н
	4β , 7β ,11-Enantioeudesmantriol (210)	H_2	ОН	Н
	Pterodontriol C (296)	H_2	Н	OH

Figure 5. (continued)





	R_1	R_2	R ₃	R_4	R ₅
4β,5α-Dihydroxy-15-oxo-eudesm-	Н	Н	СНО	OH	α-OH
11(13)-en-12-oic acid (206)					
4β,5β-Dihydroxy-15-oxo-eudesm-	Н	Н	СНО	OH	β-ОН
11(13)-en-12-oic acid (207)					
2α-Hydroxy-4- <i>epi</i> -ilicic acid (241)	OH	Н	CH ₃	OH	α-Н
3α-Hydroxyilicic acid (248)	Н	ОН	ОН	CH ₃	α-Н
5β-Hydroxyilicic acid (249)	Н	Н	OH	CH ₃	β-ОН
Ilicic acid (255)	Н	Н	OH	CH ₃	α-Н





Diplophyllin (208)

Diversifolol (209)

Figure 5. (continued)

71



Ent-7(11)-selinen-4-ol (212)



 4α ,15-Epoxyeudesmane-1 β ,6 α -diol (**216**)

 R_1

 R_2



10- <i>epi</i> -5α-hydroperoxy-β	CH ₂	α-ООН
-eudesmol (213)		
10- <i>epi</i> -5β-hydroperoxy-β	CH ₂	β-ООН
-eudesmol (214)		
4,10- <i>epi</i> -5β-hydroxy	α-CH ₃ , β-H	β-ОН
-dihydroeudesmol (215)		



Epoxysantamarin (217)

ΟН

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011111



ОН

Eudesma-4(15),7-dien -1β-ol (218)



5-epi-Eudesma-4(15)-ene-1 β , 6β -diol (225)

οн

ő Eudesmanomolide (219)

Figure 5. (continued)



	R_1	R_2	R ₃	R_4
5α ,10 β -4(15)-Eudesmen-1 β ,6 β -diol (220)	Н	Н	OH	Н
4(15)-Eudesmene-1β,5α-diol (221)	ОН	Н	Н	Н
4(15)-Eudesmene-1 β ,6 α -diol (222)	Н	ОН	Н	Н
4(15)-Eudesmene-1 β ,7 α -diol (223)	Н	Н	Н	OH
Kikkanol A (259)	OH	Н	OH	Н

ОН		R_1	R ₂	R ₃
	4(15)-Eudesmene-1 β ,7 β -diol (224)	Н	Н	OH
	7- <i>epi</i> -Eudesm-4(15)-ene-1β,6α-diol (226)	OH	Н	Н
$\ R_1 R_2 \ $	7- <i>epi</i> -Eudesm-4(15)-ene-1β,6β-diol (227)	Н	OH	Н







7-*epi*-β-Eudesmol (**229**) $R = CH_2$ Isodonsesquitin A (**258**) $R = \alpha$ -OH, β-CH₃

Figure 5. (continued)





Fruticolide (230)

 $1\text{-}O\text{-}\beta\text{-}D\text{-}Glucopyranosyl-9\beta,15\text{-}dihydroxy}$ $-5\alpha,6\beta\text{H-eudesma-3-ene-}6\alpha,12\text{-}olide\text{ (233)}$



	R_1	R_2
Gargantolide (231)	β-ΟΗ, α-Η	OAc
1-epi-Gargantolide (232)	$\alpha\text{-OH},\beta\text{-H}$	OAc
8α-Hydroxytaurin (254)	0	OH



 1β -*O*- β -D-Glucopyranosyl-4 α -hydroxyl -5 α ,6 β ,11 β H-eudesma-12,6 α -olide (**234**)



Hierapolitanin C (236) $R = CH_2$ Hierapolitanin D (237) $R = \beta$ -OH, α -CH₃



Figure 5. (continued)



 6α -({4'-O-[9''Z-Hexadecenoyl]}-7'E-coumaryloxy) eudesm-4(15)-ene (235) R = CO(CH₂)₇CH=CH(CH₂)₅CH₃

 6α -({4'-O-Methyl}-7'E-coumaryloxy) eudesm-4(15)-ene (274) R = CH₃

 6α -({4'-O-Palmityl}-7'E-coumaryloxy) eudesm-4(15)-ene (**292**) R = CO(CH₂)₁₄CH₃

 6α -({4'-O-Stearyl}-7'E-coumaryloxy) eudesm-4(15)-ene (**301**) R = CO(CH₂)₁₆CH₃

=0



Hydroxy-bis-dihydroencelin (239)



8 α -Hydroxy-1,4 β -epoxy-10-epieudesman -5 α ,6 β ,7 α ,11 β H-12,6-olide (**243**)



	R_1	R_2	R ₃	R ₄
5α-Hydroxy-4- <i>epi</i> -ilicic acid methyl ester (242)	Н	CH ₃	OH	OH
3α -Hydroxyilicic acid methyl ester (250)	OH	OH	CH ₃	Н

Figure 5. (continued)



 1β -Hydroxy- α -eudesmol (245)



1α-Hydroxypinnatifidin (252)



1β-Hydroxy-15-*O*-(*p*-methoxyphenylacetyl) $-5\alpha,6\beta,11\beta$ H-eudesma-3-en-12,6 α -olide (251)



 2α -Hydroxypterodontic acid (253)



Kikkanol B (260)



Liriopeoside A (262)



0



Kikkanol C (261)

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Figure 5. (continued)







Litchioside B (264)



Macrophyllic acid A (265)



Macrophyllic acid D (268)



Macrophyllic acid B (**266**) $R = \beta$ -OH Macrophyllic acid C (**267**) $R = \alpha$ -OH



Macrophyllic acid E (269)

Figure 5. (continued)



Macrophyllidimer A (270)



Macrophyllilactone G (273)



Macrophyllilactone F (272)



Odonticin (276)

Ē

1-One-4-epi-alantolactone (278)

=0



Odontin (277)



	R_1	R ₂
Ophiopogonoside A (279)	β-ОН	β-н
Ophiopogonoside B (280)	α-ОН	α-Н

Figure 5. (continued)





1-Oxoeudesm-11(13)-eno-12,8α-lactone (284)

(7α)-8-Oxoeudesm-4(15)-en-12-oic acid (283)



Oxyphyllanene A (285)



Oxyphyllanene C (287)



Oxyphyllanene B (286)



Oxyphyllanene D (288)



Oxyphyllanene E (289)



Oxyphyllanenes F, G (290, 291) R = OH (diastereomers)



 2α , 5α -Peroxyeudesma-3, 11-dien-1-one (293)



Pterodolide (295)

Figure 5. (continued)



	R_1	R_2
Pterodontriol D (297)	β-ОН	OH
Pterodontriol D-6- O - β -D-glucopyranoside (298)	β-ОН	OGlc
Pumilaside A (299)	α-OH	OGlc



Septuplinolide (300)



Teucdiol A (303)



Teucrenone (305)



Tanapraetenolide (302)



Teucdiol B (304)



Tourneforin (306)

Figure 5. (continued)



1,4,13-Trihydroxyeudesm-11(12)-ene (**307**)



7-Trinoreudesma-4(15),8-dien-1β-ol-7-one (**308**)



Yomogin (310)

10. Chemical Constituents of Plants in the Genus Ficus

Several *Ficus* species are used ethanomedicinally in Indian traditional medicine. Every part of these plants has been also utilized. For example, the bark of *F. racemosa* has been used as antiseptic, antipyretic and vermicidal, and its decoction is used in the treatment of various skin diseases, ulcers and diabetes (Joseph and Raj, 2010). The stem bark of *F. religiosa* has been used in the treatment of diabetes, gynaecological problems, dysentery, diarrhea, and as antipyretic, antibacterial, tonic and astringent (Makhija *et al.*, 2010). In Nepal, the powdered bark of *F. foveolata* is taken as a cure for boils and given to women during child birth to promote milk secretion (Kunwar and Bussmann, 2006). In addition, a number of *Ficus* species have also been reported as the sources of anticancer and anti-inflammatory agents (Lansky *et al.*, 2008). Based on their biological acitivites, phytochemical investigations of these plants revealed the presence of triterpenoids, sesquiterpenoids, alkaloids, flavonoids, stilbenes, coumarins, lignans and phenolic compounds.

Triterpenes, including friedelin (27), betulinic acid (323), oleanoic acid (384) and taraxeryl acetate (394), were isolated from the stem bark of *F. ovata*; the latter compound showed antimicrobial activity against half of the tested organisms (Kuete *et al.*, 2009). Betulonic acid (324), together with oleanonic acid (385), 3-oxofriedelan-28-oic acid (386) and ursolic acid (401) from the roots of *F. microcarpa* exhibited significant cytotoxic activity against HONE-1, KB and HT29 cancer cell lines with IC₅₀ values in the range of 4.0-9.4 μ M (Chiang *et al.*, 2005). A triterpenoid saponin, elasticoside (337), from the bark of aerial roots of *F. elastica*, was potently active against *Enterococcus faecalis* with an MIC value of 0.03 mg/ml (Mbosso *et al.*, 2012).

Sesquiterpenes, including alloaromadendrene (**316**), β -bourbonene (**325**), α caryophyllene (**327**), β -caryophyllene (**328**), 1,8-cineole (**329**), α -cubenene (**331**), β -cubenene (**332**), β -elemene (**338**), geranyl acetone (**366**), germacrene D (**367**), α -guaiene (**369**), α gurjunene (**370**), heneicosene (**371**), (+)-ledene (**380**), 6,10,14-trimethyl-2-pentadecanone (**400**) and α -ylangene (**403**), were characterized in the essential oils from the leaves of *F. carica* (Oliveira *et al.*, 2010), *F. elastica* and *F. mucosa* (Ogunwande *et al.*, 2009, 2011). δ -Cadinene (**326**) obtained from the bark of *F. religiosa* (Makhija *et al.*, 2010), α -copaene (**330**) from the wood of *F. benjamina* (Niogret et al., 2011), along with compounds **328**, **331** and **338**, were also found in the floral fragrances and released by receptive figs in plant-pollinator interactions (Grison, Edwards, and Hossaert-McKey, 1999). A macrocyclic trichothecene sesquiterpene, named verrucarin L acetate (**402**), from the leaves and stem bark of *F. fistulosa*, showed strong antimalarial activity by inhibiting the growth of *Plasmodium falciparum* with an IC₅₀ value of lower than 1 ng/ml (Zhang *et al.*, 2002). Glucosides of sesquiterpenes, such as pumilasides A, B and C (**299**, **388** and **389**), were also found in the fruits of *F. pumila* (Kitajima *et al.*, 2000).

A series of phenanthroindolizidine alkaloids found in *F. septica*, called ficuseptines A-N (**345-358**), were strongly cytotoxic against two human cancer cell lines, NUGC and HONE-1 (Wu *et al.*, 2002; Damu *et al.*, 2005, 2009). Another alkaloid of the same type, (-)-13a α -antofine (**317**) from the stem bark of *F. fistulosa*, exhibited antifungal activity against *Aspergillus fumigatus* and *Candida albicans* with an IC₅₀ value of 4 μ M in both assays (Subramaniam *et al.*, 2009).

The flavonoids dihydroquercetin (**333**), eriodictyol (**339**), homoeriodictyol (**372**) and luteolin (**381**), isolated from the stems and leaves of *F. sarmentosa* var. *henryi*, displayed inhibitory activity against the pathogenic fungi *Fusarium graminearum* and *Septoria zeicola*, with luteolin showing the strongest inhibitory activity (Wang *et al.*, 2010). Apigenin (**318**), found in *F. formosana* f. *formosana* (Sheu *et al.*, 2005) and *F. sarmentosa* var. *henryi* (Wang *et al.*, 2011b), exhibited significant cytotoxicity against HepG2, PLC/PRF/5 and Raji cancer cell lines *in vitro* with IC₅₀ values of 0.045, 0.066 and 0.066 μ M, respectively (Sheu *et al.*, 2005). Another phenolic constituent from the latter *Ficus* species, 7-hydroxycoumarin (**374**), has also been reported as insecticidal (Wang *et al.*, 2011b).

The stilbenes gnetol (368), isorhapontigenin (378), pinosylvin (387) and resveratrol (392), the flavonoids galangin 3-methyl ether (365), 5-hydroxy-3,7-dimethoxyflavone (375), quercetin (390) and tectochrysin (395), along with the phenylpropanoids 1,22-docosanediol diferulate (336), 24'-hydroxy-tetracosyl ferulate (377) and 1,24-tetracosanediol diferulate (396), were reported as constituents of the stems of *F. foveolata*. The isolated stilbenes displayed high inhibitory activity towards the enzyme butyrylcholinesterase, with gnetol as the most active compound (IC₅₀ = 1.3 μ M) (Sermboonpaisarn and Sawasdee, 2012).

Two phenolic compounds, bergenin (322) and racemosic acid (391), were isolated from the bark of *F. racemosa*. Racemosic acid potently inhibited the enzymes COX-1 and 5-LOX *in vitro* with IC₅₀ values of 90 and 18 μ M, respectively, and also demonstrated strong antioxidant activity in scavenging ABTS free radical cations with an IC₅₀ value of 19 μ M (Li *et al.*, 2004).

Chemical constituents of plants in the genus *Ficus* have been summarized in **Table 4**. Their chemical structures are presented in **Figure 6**.

Compound	Source	Plant part	References
3β-Acetoxy-12,19-dioxo-13(18)			
-oleanene (311)			
3β-Acetoxy-11α,12α-epoxy-16-			
oxo-14-taraxerene (312)	Ficus microcarpa	Roots	Chiang <i>et al.</i> , 2005
3β-Acetoxy-21α,22α-epoxy			
taraxastan-20α-ol (313)			
3β-Acetoxy-25-methoxylanosta-			
8,23-diene (314)			
3β-Acetoxy-19(29)-taraxasten-			
20α-ol (315)			
Alloaromadendrene (316)	F. carica	Leaves	Oliveira et al., 2010
(-)-13aα-Antofine (317)	F. fistulosa	Stem bark	Subramaniam et al.,
			2009
Apigenin (318)	F. formosana f.	Stems	Sheu <i>et al.</i> , 2005
	formosana		
	F. sarmentosa var.	Stems,	Wang et al.,2011b
	henryi	leaves	
Aripuanin (319)	F. aripuanensis	Leaves	Nascimento et al.,
			1999
Bergapten (320)	E valiaiosa		Makhija <i>et al.</i> , 2010
Bergaptol (321)	r. reugiosa	Bark	
Bergenin (322)	F. racemosa		Li et al., 2004

Table 4. Chemical constituents of plants in the genus Ficus

Compound	Source	Plant part	References
Betulinic acid (323)	Ficus ovata	Stem bark	Kuete et al., 2009
Betulonic acid (324)	F. microcarpa	Roots	Chiang et al., 2005
β -Bourbonene (325)	F. carica	Leaves	Oliveira et al., 2010
δ-Cadinene (326)	F. religiosa	Bark	Makhija <i>et al.</i> , 2010
α-Caryophyllene (327)	F. carica	Leaves	Oliveira et al., 2010
β-Caryophyllene (328)	F. carica		Oliveira et al., 2010
	F. mucosa	Lagyas	Ogunwande et al.,
		Leaves	2009
1,8-Cineole (329)	F. elastica		Ogunwande et al.,
			2011
α-Copaene (330)	F. benjamina	Wood	Niogret et al., 2011
α-Cubenene (331)	E	Leaves	Oliveira et al., 2010
β -Cubenene (332)	F. curica		
Dihydroquercetin (333)	F. sarmentosa var.	Stems,	Wang et al., 2010
	henryi	leaves	
2,3-Dihydroxy-1-(4-hydroxy-3,5	F. beecheyana		Lee et al., 2002
-dimethoxyphenyl)-1-propanone		Deste	
(334)		ROOIS	
3,22-Dioxo-20-taraxastene (335)	F. microcarpa		Chiang et al., 2005
1,22-Docosanediol diferulate (336)	F. foveolata	Vines	Sermboonpaisarn
			and Sawasdee, 2012
Elasticoside (337)	F. elastica	Aerial root	Mbosso et al., 2012
		bark	
β -Elemene (338)	F. carica	Leaves	Oliveira et al., 2010
Eriodictyol (339)	F. sarmentosa var.	Stems,	Wang et al.,2011b
	henryi	leaves	

Compound	Source	Plant part	References
erythro-2,3-Bis(4-hydroxy-3	Ficus beecheyana	Roots	Lee et al., 2002
-methoxyphenyl)-3-ethoxypropan-			
1-ol (340)			
Ficuformodiol A (341)	F. formosana f.	Stems	Sheu <i>et al.</i> , 2005
Ficuformodiol B (342)	formosana		
Ficusal (343)	F. microcarpa	Wood	Li and Kuo, 2000
Ficusamide (344)	F. elastica	Aerial root	Mbosso et al., 2012
		bark	
Ficuseptine A (345)		Leaves	Wu et al., 2002
Ficuseptine B (346)		Stems	Damu <i>et al.</i> , 2005
Ficuseptine C (347)			
Ficuseptine D (348)			
Ficuseptine E (349)		Roots	Damu <i>et al.</i> , 2009
Ficuseptine F (350)			
Ficuseptine G (351)	Figure conting		
Ficuseptine H (352)	Ficus septica		
Ficuseptine I (353)			
Ficuseptine J (354)			
Ficuseptine K (355)			
Ficuseptine L (356)			
Ficuseptine M (357)			
Ficuseptine N (358)			
Ficusesquilignan A (359)			
Ficusesquilignan B (360)	F. microcarpa	Wood	Li and Kuo, 2000
Ficusolide diacetate (361)			
Table 4. (continued)

Compound	Source	Plant part	References
Ficusoside (362)	Ficus elastica	Aerial root	Mbosso et al., 2012
		bark	
Ficustriol (363)	F. hispida	Leaves,	Peraza-Sanchez
		twigs	et al., 2002
Fistulosine (364)	F. fistulosa		Subramaniam et al.,
		Stem bark	2009
Friedelin (27)	F. ovata		Kuete et al., 2009
Galangin 3-methyl ether (365)	F. foveolata	Vines	Sermboonpaisarn
			and Sawasdee, 2012
Geranyl acetone (366)	F. elastica		Ogunwande et al.,
		Leaves	2011
Germacrene D (367)	F. carica		Oliveira et al., 2010
Gnetol (368)	F. foveolata	Vines	Sermboonpaisarn
			and Sawasdee, 2012
α-Guaiene (369)			
α-Gurjunene (370)	F. carica	T	Oliveira <i>et al.</i> , 2010
Heneicosene (371)	F. elastica	Leaves	Ogunwande et al.,
			2011
Homoeriodictyol (372)	<i>F. sarmentosa</i> var.	Stems,	Wang <i>et al.</i> , 2010
	henryi	leaves	
(-)-14β-Hydroxyantofine (373)	F. fistulosa	Stem bark	Subramaniam et al.,
			2009
7-Hydroxycoumarin (374)	F. sarmentosa var.	Stems,	Wang et al.,2011b
	henryi	leaves	
5-Hydroxy-3,7-dimethoxyflavone	F. foveolata	Vines	Sermboonpaisarn
(375)			and Sawasdee, 2012

Table 4. (continued)

Compound	Source	Plant part	References	
3-Hydroxy-1-(4-hydroxy-3,5	Ficus beecheyana	Roots	Lee et al., 2002	
-dimethoxyphenyl)-1-propanone				
(376)				
24'-Hydroxy-tetracosyl ferulate			Sermboonpaisarn	
(377)	F. foveolata	Vines	and Sawasdee, 2012	
Isorhapontigenin (378)				
Lanosterol (379)	F. religiosa	Bark	Makhija <i>et al.</i> , 2010	
(+)-Ledene (380)	F. carica	Leaves	Oliveira et al., 2010	
Luteolin (381)	<i>F. sarmentosa</i> var.	Stems,	Wang <i>et al.</i> , 2010	
	henryi	leaves		
<i>O</i> -Methyltylophorinidine (382)	F. hispida	Leaves,	Peraza-Sanchez et	
		twigs	al., 2002	
α-Muurolene (383)	F. carica	Leaves	Oliveira et al., 2010	
Oleanoic acid (384)	F. ovata	Stem bark	Kuete et al., 2009	
Oleanonic acid (385)		Deete	Chines et al. 2005	
3-Oxofriedelan-28-oic acid (386)	F. microcarpa	KOOIS	Chiang <i>et al.</i> , 2005	
Pinosylvin (387)	F. foveolata	Vines	Sermboonpaisarn	
			and Sawasdee, 2012	
Pumilaside B (388)	Eile	E secito	Kitaling of al. 2000	
Pumilaside C (389)	F. pumua	Fruits	Kitajima <i>et al.</i> , 2000	
Quercetin (390)	F. foveolata	Vines	Sermboonpaisarn	
			and Sawasdee, 2012	
Racemosic acid (391)	F. racemosa	Bark	Li et al., 2004	
Resveratrol (392)	F. foveolata	Vines	Sermboonpaisarn	
			and Sawasdee, 2012	
(-)-13aα-Secoantofine (393)	F. fistulosa	Stem bark	Subramaniam et al.,	
			2009	

Table 4. (continued)

Compound	Source Plant part		References	
β-Sitosterol (171)	Ficus religiosa	Ficus religiosa Bark Makhija		
Taraxeryl acetate (394)	F. ovata	Stem bark	Kuete et al., 2009	
Tectochrysin (395)			Sermboonpaisarn	
1,24-Tetracosanediol diferulate	F. foveolata	Vines	and Sawasdee, 2012	
(396)				
threo-2,3-Bis(4-hydroxy-3				
-methoxyphenyl)-3-ethoxypropan				
-1-ol (397)		Desta		
threo-3-(4-Hydroxy-3,5				
-dimethoxyphenyl)-3-	Eboobara		Les et al. 2002	
ethoxypropan	F. beecneyana	KOOIS	Lee <i>et ut.</i> , 2002	
-1,2-diol (398)				
trans-4,5-Bis(4-hydroxy-3				
-methoxyphenyl)				
-1,3-dioxacyclohexane (399)				
6,10,14-Trimethyl-2-	F. elastica	Leaves	Ogunwande et al.,	
pentadecanone (400)			2011	
Ursolic acid (401)	F. microcarpa	Roots	Chiang et al., 2005	
Verrucarin L acetate (402)	F. fistulosa	Leaves,	Zhang <i>et al.</i> , 2002	
		stem bark		
α-Ylangene (403)	F. carica	Leaves	Oliveira et al., 2010	





3β-Acetoxy-12,19-dioxo-13(18)-oleanene (**311**)

3β-Acetoxy-11α,12α-epoxy-16-oxo-14 -taraxerene (**312**)





 3β -Acetoxy- 21α , 22α -epoxy taraxastan- 20α -ol (**313**)





 3β -Acetoxy-19(29)-taraxasten-20 α -ol (**315**)



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Figure 6. Chemical constituents of plants in the genus Ficus



(-)-13a α -Antofine (**317**) R = H (-)-14 β -Hydroxyantofine (**373**) R = OH



	R_1	R ₂	R ₃	R_4	R ₅
Apigenin (318)	Н	OH	Н	OH	Н
Galangin 3-methyl ether (365)	OCH ₃	OH	Н	Н	Н
5-Hydroxy-3,7-dimethoxyflavone (375)	OCH ₃	OCH ₃	Н	Н	Н
Luteolin (381)	Н	ОН	Н	OH	OH
Quercetin (390)	ОН	ОН	ОН	OH	Н
Tectochrysin (395)	Н	OCH ₃	Н	Н	Н



Aripuanin (319)



Bergenin (322)



 β -Bourbonene (325)



 α -Caryophyllene (327)



Bergapten (**320**) $R = OCH_3$ Bergaptol (**321**) R = OH



Betulinic acid (323) $R = \beta$ -OH, α -H

Betulonic acid (324) R = O



 δ -Cadinene (326)



 β -Caryophyllene (328)

92



α-Copaene (**330**)



1,8-Cineole (329)





 β -Cubenene (332)





2,3-Dihydroxy-1-(4-hydroxy-3,5-dimethoxyphenyl)-1-propanone (**334**) R = OH3-Hydroxy-1-(4-hydroxy-3,5-dimethoxyphenyl)-1-propanone (**376**) R = H



3,22-Dioxo-20-taraxastene (335)





1,22-Docosanediol diferulate (336) n = 20

1,24-Tetracosanediol diferulate (**396**) n = 22



 β -Elemene (338)



erythro-2,3-Bis(4-hydroxy-3-methoxyphenyl)
-3-ethoxypropan-1-ol (340) R = H, erythro
threo-2,3-Bis(4-hydroxy-3-methoxyphenyl)
-3-ethoxypropan-1-ol (397) R = H, threo



Ficuformodiol A (341)

Ficuformodiol B (342)











Figure 6. (continued)



	\mathbf{R}_1	R ₂	R ₃	R_4	R ₅	R_6
Ficuseptine B (346)	-OC]	H ₂ O-	Н	OCH ₃	OCH ₃	Н
Ficuseptine C (347)	-OC	H ₂ O-	Н	Н	OCH ₃	Н
Ficuseptine D (348)	Н	OCH ₃	OCH ₃	Н	OCH ₃	Н
Ficuseptine E (349)	OH	OCH ₃	OH	Н	OCH ₃	Н
Ficuseptine F (350)	OCH ₃	OCH ₃	OCH ₃	OCH ₃	OCH ₃	Н
Ficuseptine G (351)	Н	OCH ₃	OCH ₃	Н	OH	OCH ₃
Ficuseptine H (352)	OCH ₃	OH	Н	OCH ₃	OCH ₃	Н
Ficuseptine J (354)	OCH ₃	OCH ₃	Н	OCH ₃	OH	Н





Ficuseptine I (353)

Ficuseptine N (358)





Ficusesquilignan A (359)

97

Figure 6. (continued)







Ficusolide diacetate (361)

Ficusoside (362)







Fistulosine (364)





Geranyl acetone (366)

Germacrene D (367)



	\mathbf{R}_1	R ₂	R ₃	R_4
Gnetol (368)	OH	Н	Н	OH
Isorhapontigenin (378)	Н	ОН	OCH ₃	Н
Pinosylvin (387)	Н	Н	Н	Н
Resveratrol (392)	Н	OH	Н	Н





α-Guaiene (369)

 α -Gurjunene (370)



Heneicosene (371)







24[′]-Hydroxy-tetracosyl ferulate (377)



Lanosterol (379)



O-Methyltylophorinidine (**382**)



(+)-Ledene (380)



 α -Muurolene (383)





Oleanoic acid (**384**) $R = \beta$ -OH, α -H Oleanonic acid (**385**) R = O



Pumilaside B (388)



Racemosic acid (391)

(-)-13a α -Secoantofine (393)





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3-Oxofriedelan-28-oic acid (386)



Pumilaside C (389)







Taraxeryl acetate (394)

threo-3-(4-Hydroxy-3,5-dimethoxyphenyl) -3-ethoxypropan-1,2-diol (**398**)



trans-4,5-Bis(4-hydroxy-3-methoxyphenyl)

-1,3-dioxacyclohexane (399)



6,10,14-Trimethyl-2-pentadecanone (400)



Ursolic acid (401)



Verrucarin L acetate (402)



 α -Ylangene (403)

CHAPTER III

EXPERIMENTAL

1. Source of Plant Materials

The stems of *Salacia verrucosa* and *Ficus foveolata* were collected in Bueng Kan, Thailand, in January 2010. Voucher specimens of both species have been deposited at the herbarium of the Faculty of Pharmaceutical Sciences, Chulalongkorn University, Bangkok, Thailand.

2. General Techniques

2.1 Solvents

Throughout this work, all organic solvents were of commercial grade and were redistilled prior to use.

2.2 Analytical Thin-Layer Chromatography (TLC)

Technique:	One dimension, ascending
Adsorbent:	Silica gel 60 F_{254} (E. Merck) pre-coated plates
Layer thickness:	0.2 mm
Distance:	5 cm
Temperature:	Laboratory temperature (30-35 °C)
Detection:	1. Ultraviolet light (254 and 365 nm)
	2. 10% Sulfuric acid and heating at 110 °C for 10 minutes

2.3 Column Chromatography

2.3.1 Conventional Column Chromatography

Gel filter:	Silica gel 60 number 9385 (particle size 0.040-0.063 nm) and
	number 7734 (particle size 0.063-0.200 nm) (E. Merck)
Packing method:	Wet packing: The adsorbent was mixed with the eluent into
	slurry, then poured into a column and allowed to settle.
Sample loading:	The sample was dissolved in a small amount of the eluent, and
	then applied gently on top of the column.

Detection:	Fractions	were	examined	by	TLC	technique	in	the	same
	manner as	descri	bed in secti	ion 2	2.2.				

2.3.2 Size-Exclusion Column Chromatography

Gel filter:	Sephadex LH-20 (Pharmacia Biotech AB)
Packing method:	Gel filter was suspended in the eluent and left standing to
	swell for 24 hours prior to use. It was then poured into the
	column and allowed to set tightly.
Sample loading:	The sample was dissolved in a small amount of eluent, and then
	applied gently on top of the column.
Detection:	Fractions were examined by TLC technique in the same
	manner as described in section 2.2.

2.4 Spectroscopy

2.4.1 Ultraviolet (UV) Spectra

UV spectra were obtained on a Shimadzu UV-160A spectrophotometer (Pharmaceutical Research Instrument Center, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

2.4.2 Infrared (IR) Spectra

IR spectra (KBR disc) were recorded on a Perkin Elmer FT-IR 1760X spectrometer (Scientific and Technological Research Equipment Center, Chulalongkorn University).

2.4.3 Mass Spectra

Electrospray Ionization (ESI) mass spectra were obtained on a Bruker micrOTOF mass spectrometer (National Center for Genetic Engineering and Biotechnology, BIOTEC, Thailand).

2.4.4 Proton and Carbon-13 Nuclear Magnetic Resonance (¹H and ¹³C) Spectra

¹H (300 MHz) and ¹³C (75 MHz) NMR spectra were recorded on a Bruker DPX-300 FT-NMR spectrometer (Pharmaceutical Research Instrument Center, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

¹H (500 MHz) and ¹³C (125 MHz) NMR spectra were obtained on a JEOL JMN-A500, Varian Unity INOVA (Scientific and Technological Research Equipment Center, Chulalongkorn University).

2.5 Physical Properties

2.5.1 Melting Points

Melting points were obtained on a Fisher-John melting point apparatus (Department of Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

2.5.2 Optical rotations

Optical rotations were measured on a Perkin-Elmer 314 polarimeter using a sodium lamp operating at 589 nm (Pharmaceutical Research Instrument Center, Faculty of Pharmaceutical Sciences, Chulalongkorn University).

3. Extraction and Isolation of Compounds from the Stems of Salacia verrucosa

3.1 Extraction of the Stems of S. verrucosa

Dried stems of *S. verrucosa* (1.2 kg) were ground and then sequentially macerated with *n*-hexane (5 × 4 l), CH_2Cl_2 (5 × 4 l) and MeOH (5 × 4 l). The *n*-hexane extract was concentrated under reduced pressure at 45 °C, then acetone was added to exclude the gummy material, whereas the CH_2Cl_2 extract was evaporated to dryness and then reconstituted with EtOAc to separate the elastic gum from the crude extract. Each solvent extract was evaporated to give *n*-hexane-acetone (18.7 g, 1.56 % yield), EtOAc (6.1 g, 0.51 % yield) and MeOH extracts (55.1 g, 4.59 % yield), as presented in **Scheme 1**.

3.2 Isolation of Compounds from the *n*-Hexane-Acetone Extract of *S. verrucosa* Stems

The extract (18.7 g) was re-dissolved in a small volume of *n*-hexane and then subjected to silica gel column chromatography (500 g, 9.5×15 cm). The column was eluted with a gradient mixture of *n*-hexane and CH_2Cl_2 (3:7 \rightarrow 0:1). One hundred and sixty fractions (50 ml each) were collected and combined according to their TLC profiles by using mixtures of *n*hexane and CH_2Cl_2 as mobile phases into 8 fractions (H1-H8), as shown in **Table 5**.



Scheme 1. Extraction of S. verrucosa stems

Table 5. Combined fractions from the *n*-hexane-acetone extract of *S. verrucosa* stems

Fraction Code	Weight (g)
H1	2.01
H2	0.32
Н3	0.73
H4	0.63
H5	0.55
H6	0.75
H7	0.93
H8	1.65

3.2.1 Isolation of Compound SV-1 (Friedelin)

Fraction H4 (0.63 g) was further separated on a silica gel column (40 g, 3×15 cm) using *n*-hexane-CH₂Cl₂ (7:13) as the mobile phase. One hundred and fifteen collected fractions (10 ml each) were examined by TLC using *n*-hexane-CH₂Cl₂ (2:3) as the mobile phase, and then combined to yield 4 subfractions (H4A-H4D). Purification of subfraction H4B (50 mg) on a column of silica gel (12 g, 2×12 cm), washed down with *n*-hexane-EtOAc (39:1), gave compound SV-1 as colorless needles (4.0 mg, 0.0003 % yield).

3.2.2 Isolation of Compound SV-2 (Friedelane-1,3-dione)

Subfraction H4D (66 mg) was further purified by recrystallization in MeOH to afford compound SV-2 as colorless needles (25.5 mg, 0.002 % yield).

3.3 Isolation of Compounds from the EtOAc Extract of S. verrucosa Stems

The EtOAc extract (6.1 g) was dissolved in a small amount of CH_2Cl_2 and applied to the top of a silica gel column (150 g, 4.5 × 20 cm) eluted with gradient mixtures of CH_2Cl_2 -EtOAc (19:1 \rightarrow 0:1). One hundred and fifty fractions (25 ml each) were collected. According to their chemical profiles by TLC using CH_2Cl_2 -EtOAc (19:1) as the mobile phase, these fractions were combined into 10 fractions (E1-E10), as shown in **Table 6**.

Table 6. Combined fractions from the EtOAc extract of S. verrucosa stems

Fraction Code	Weight (g)
E1	0.06
E2	0.05
E3	0.10
E4	0.28
E5	0.12
E6	0.09
E7	0.07
E8	0.02
E9	0.19
E10	0.23

3.3.1 Isolation of Compound SV-3 (Kokoonol)

Fraction E3 (0.1 g) was further purified on a silica gel column (20 g, 2.5×12 cm), using *n*-hexane-EtOAc (17:3) as the mobile phase, to yield compound SV-3 as a white, amorphous powder (7.4 mg, 0.0006 % yield) after recrystallization in MeOH.

3.3.2 Isolation of Compound SV-4 (26-Hydroxyfriedelane-1,3-dione)

Fraction E4 (0.28 g) was chromatographed on a silica gel column (25 g, 2.5×15 cm), eluted with a solvent system of CH₂Cl₂-EtOAc (97:3). Eighty fractions (10 ml each) were collected and combined into 3 subfractions (E4A-E4C). Subfraction E4A (64 mg) was further purified on another silica gel column (20 g, 2.5×12 cm) eluting with *n*-hexane-EtOAc (87:13) to afford compound SV-4 as a white, amorphous powder (7.4 mg, 0.0006 % yield).

3.3.3 Isolation of Compound SV-5 (21a-Hydroxyfriedelan-3-one)

Subfraction E4B (54 mg), after being subjected to a silica gel column (20 g, 2.5 \times 12 cm) eluting with *n*-hexane-EtOAc (17:3), was divided into 47 collected fractions (10 ml each). These fractions were pooled into 2 subfractions (E4B1-E4B2) depending on their TLC profiles. Compound SV-5 was obtained as colorless cubic crystals (2.3 mg, 0.0002 % yield) upon recrystallizing subfraction E4B2 with MeOH.

3.3.4 Isolation of Compound SV-6 $(3\beta,22\alpha$ -Dihydroxyolean-12-en-29-oic acid)

Subfraction E4C (45 mg) was purified on a silica gel column (20 g, 2.5×12 cm) eluting with *n*-hexane-EtOAc (4:1). Fifty fractions (10 ml each) were collected and combined into 2 subfractions (E4C1-E4C2) according to their TLC data. Compound SV-6 was obtained as colorless prisms (6.2 mg, 0.0005 % yield) by recrystallizing subfraction E4C2 with *n*-hexane.

3.3.5 Isolation of Compound SV-7 (30-Hydroxyfriedelane-1,3-dione)

Fraction E6 (0.09 g) was subjected to gel filtration chromatography using a Sephadex LH-20 column (100 g, 2.3×82 cm) and a solvent system of CH₂Cl₂-MeOH (1:1) as the mobile phase. Selected fractions were pooled and further chromatographed on a silica gel column (10 g, 2×10 cm) eluting with CH₂Cl₂-MeOH (99:1). Sixty fractions (10 ml each) were collected and combined to give 3 subfractions (E6A-E6C). Subfraction E6C (20 mg) was further separated on a silica gel column (10 g, 2×10 cm), eluted with *n*-hexane-EtOAc (4:1), into 2

subfractions (E6C1-E6C2). Subfraction E6C1 was purified by recrystallization with MeOH to afford compound SV-7 as a white amorphous powder (5.6 mg, 0.0005 % yield).

3.3.6 Isolation of Compound SV-8 (21α-Hydroxyfriedelane-1,3-dione)

After the separation of subfraction E6C (20 mg) into subfractions E6C1 and E6C2, Compound SV-8 was obtained as a white amorphous powder (4.2 mg, 0.0004 % yield) from the recrystallization of subfraction E6C2 with MeOH.

The isolation of compounds from the *n*-hexane-acetone and EtOAc extracts of *S*. *verrucosa* stems is presented in **Schemes 2** and **3**, respectively.



n-Hexane-acetone extract of *S. verrucosa* stems (18.7 g)

Scheme 2. Isolation of compounds from the *n*-hexane-acetone extract of *S. verrucosa* stems



EtOAc extract of S. verrucosa stems (6.1 g)

Scheme 3. Isolation of compounds from the EtOAc extract of S. verrucosa stems



Scheme 3. (continued)

4. Extraction and Isolation of Compounds from the Stems of Ficus foveolata

4.1 Extraction of the Stems of F. foveolata

Dried, powdered *F. foveolata* stems (2.0 kg) were macerated with 95% ethanol (3×8 l) and the combined ethanolic extract was concentrated under reduced pressure at 45°C by rotary evaporator. The aqueous ethanol extract was then sequentially partitioned with *n*-hexane and CH₂Cl₂ to yield *n*-hexane (7.0 g, 0.35 % yield), CH₂Cl₂ (28.9 g, 1.45 % yield) and aqueous ethanol (77.3 g, 3.87 % yield) extracts upon evaporation, as shown in **Scheme 4.**



Scheme 4. Extraction of F. foveolata stems

4.2 Isolation of Compounds from the *n*-Hexane Extract of *F. foveolata* Stems

The *n*-hexane extract (7.0 g) was reconstituted in a small volume of a mixture of *n*-hexane and $CH_2Cl_2(1:1)$, then applied on top of a silica gel column (200 g, 5.0×26 cm) eluted with gradient mixtures of *n*-hexane-CH₂Cl₂ (1:1 \rightarrow 1:3) to yield nine fractions (FH1-FH9), as shown in **Table 7**.

Fraction Code	Weight (g)
FH1	0.74
FH2	0.08
FH3	0.03
FH4	0.14
FH5	0.04
FH6	0.24
FH7	0.19
FH8	0.15
FH9	0.27

Table 7. Combined fractions from the *n*-hexane extract of *F. foveolata* stems

4.2.1 Isolation of Compound FF-1 (Friedelin)

Fraction FH4 (0.14 g) was purified on a silica gel column (20 g, 2.5×12 cm) using *n*-hexane-EtOAc (24:1) as the mobile phase to yield compound FF-1 as colorless needles (24.5 mg, 0.0013 % yield) after crystallization with MeOH.

4.2.2 Isolation of Compound FF-2 (Taraxerol)

Fraction FH7 (0.19 g) was chromatographed on a column of silica gel (25 g, 2.5 \times 13 cm) eluting with *n*-hexane-EtOAc (19:1). Sixty-eight fractions (10 ml each) were collected and verified by TLC using *n*-hexane-EtOAc (19:1) as the solvent system. These fractions were combined into 2 subfractions (FH7A-FH7B) based on their TLC profiles. Upon evaporation, subfraction FH7B afforded compound FF-2 as a white amorphous powder (5.8 mg, 0.0003 % yield).

4.3 Isolation of Compounds from the CH₂Cl₂ Extract of F. foveolata Stems

The CH₂Cl₂ extract (28.9 g) was re-dissolved with a small volume of *n*-hexane-EtOAc (3:7) mixture and applied to the top of a silica gel column (500 g, 9.5×15 cm). Elution was done with gradient mixtures of *n*-hexane-EtOAc-MeOH (3:7:0 \rightarrow 0:1:0 \rightarrow 0:7:3) to give three

hundred fractions (25 ml each). These collected fractions were combined according to their TLC profiles into nine fractions (C1-C9), as shown in **Table 8**.

Fraction Code	Weight (g)
C1	0.98
C2	0.58
C3	0.98
C4	1.87
C5	0.35
C6	0.37
C7	0.57
C8	1.99
C9	6.58

Table 8. Combined fractions from the CH₂Cl₂ extract of *F. foveolata* stems

4.3.1 Isolation of Compound FF-3 (Betulin)

Fraction C1 (0.98 g) was purified over a silica gel column (20 g, 2.5×12 cm) using *n*-hexane-EtOAc (7:3) as the mobile phase to yield compound FF-3 as a white amorphous powder (13.8 mg, 0.0007 % yield).

4.3.2 Isolation of Compound FF-4 [4(15)-Eudesmene-1β,6α-diol]

Fraction C2 (0.58 g) was separated on a silica gel column (40 g, 3×15 cm) eluting with *n*-hexane-EtOAc (7:3) into five subfractions (C2A-C2E) based on their TLC profiles. The subfraction C2B (217.8 mg) was further purified by silica gel column chromatography (20 g, 2.5×12 cm) using CH₂Cl₂-EtOAc (9:1) as the mobile phase. The eluates could be combined into 2 subfractions (C2B1-C2B2) and evaporation of the first subfraction yielded compound FF-4 as colorless needles (8.5 mg, 0.0004 % yield).

4.3.3 Isolation of Compound FF-5 [4(15)-Eudesmene-1β,5α-diol]

Subfraction C2B2, which was obtained from further separation of subfraction C2B, gave compound FF-5 as colorless needles (5.1 mg, 0.0003 % yield) upon removal of the solvent.

4.3.4 Isolation of Compound FF-6 (Foveolide B)

Subfraction C2D (31.8 mg) was further purified on a silica gel column (10 g, 2 \times 10 cm) using a mixture of CH₂Cl₂-EtOAc (9:1) as the mobile phase. The eluates (5 ml each) were collected and verified by TLC. Fractions containing the same major spot were combined and evaporated to obtain compound FF-6 as a colorless gum (7.9 mg, 0.0004 % yield).

4.3.5 Isolation of Compound FF-7 (Foveolide A)

Fraction C3 (0.98 g) was separated on a silica gel column (40 g, 3×15 cm) using CH₂Cl₂-EtOAc (9:1) as the mobile phase. One hundred and twenty fractions (10 ml each) were collected and combined according to their TLC data into four subfractions (C3A-C3D). Subfraction C3C (169.9 mg) was further purified on another silica gel column (20 g, 2.5×12 cm) eluting with *n*-hexane-EtOAc (3:2) to furnish compound FF-7 as a yellow oil (16.2 mg, 0.0008 % yield).

4.3.6 Isolation of Compound FF-8b (Foveoeudesmenone)

Fraction C4 (1.87 g) was applied to a silica gel column (40 g, 3×15 cm) washed down with *n*-hexane-acetone (13:7). Seventy fractions (25 ml each) were collected, verified by TLC-and then combined into four subfractions (C4A-C4D). Subfraction C4A (92 mg) was further chromatographed over a silica gel column (25 g, 2.5×15 cm) eluted with CH₂Cl₂-EtOAc (3:2), followed by another silica gel column (10 g, 2×10 cm) eluted with *n*-hexane-EtOAc (1:1) to afford compound FF-8a as a colorless oil (14.7 mg, 0.0007 % yield). After that, the compound was spontaneously converted into FF-8b by oxidation and reduction.

4.3.7 Isolation of Compound FF-9 (Ethyl rosmarinate)

Fraction C6 (0.37 g) was subjected to silica gel column chromatography (20 g, 2.5×12 cm) eluted with *n*-hexane-EtOAc (1:9). Forty fractions (10 ml each) were collected and combined into two subfractions (C6A-C6B). Subsequent purification of subfraction C6A over another silica gel column (15 g, 2.5×12 cm), using CH₂Cl₂-acetone (3:2) as the mobile phase, afforded compound FF-9 as a yellow gum (15.7 mg, 0.0008 % yield).

4.3.8 Isolation of Compound FF-10 (Foveospirolide)

Fraction C7 (0.57 g) was separated over a silica gel column (40 g, 3×15 cm) eluting with EtOAc-MeOH (49:1). Fifty fractions (10 ml each) were collected, inspected by TLC and pooled into two subfractions (C7A-C7B). Subfraction C7A (140 mg) was purified by gel filtration through a Sephadex LH-20 column (100 g, 2.3×82 cm) eluted with CH₂Cl₂-MeOH (1:1) to give compound FF-10 as a yellow oil (12.5 mg, 0.0006 % yield).

The isolation of chemical constituents from the *n*-hexane and CH_2Cl_2 extracts of *F. foveolata* stems is presented in **Schemes 5** and **6**, respectively.



Scheme 5. Isolation of compounds from the *n*-hexane extract of *F*. foveolata stems



CH₂Cl₂ extract of *F. foveolata* stems (28.9 g)

Scheme 6. Isolation of compounds from the CH₂Cl₂ extract of *F. foveolata* stems



CH₂Cl₂ extract of *F. foveolata* stems (28.9 g)

Scheme 6. (continued)

5. Physical and Spectral Data of Isolated Compounds

5.1 Compound SV-1 (Friedelin)

Compound SV-1 was obtained as colorless needles (4.0 mg, 0.0003 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp:	260-263 °C

ESI-MS: m/z (% rel. int.): 427 [M + H]⁺(14); Figure 7.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3426, 2927, 1716, 1463, 1390; Figure 8.

¹H NMR: δ ppm, 300 MHz, in CDCl₃; 0.71 (3H, *s*), 0.85 (3H, *s*), 0.86 (3H, *d*, *J* = 5.4 Hz), 0.93 (3H, *s*), 0.98 (3H, *s*), 0.99 (3H, *s*), 1.03 (3H, *s*) and 1.15 (3H, *s*); **Table 9** and

Figure 9.

¹³C NMR: δ ppm, 75 MHz, in CDCl₃; 6.8, 14.7, 17.9, 18.3, 18.6, 20.3, 22.3, 28.2, 30.0, 30.5, 31.8, 32.1, 32.5, 32.8, 35.0, 35.4, 35.7, 36.0, 37.5, 38.3, 39.3, 39.7, 41.3, 41.5, 42.2, 42.9, 53.1, 58.3, 59.5 and 213.0; Table 9 and Figure 10.

5.2 Compound SV-2 (Friedelane-1,3-dione)

Compound SV-2 was obtained as colorless needles (25.5 mg, 0.002 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp:	270-272 °C
ESI-MS:	m/z (% rel. int.): 467 [M + 4H + Na] ⁺ (9); Figure 11.
IR:	v _{max} cm ⁻¹ (KBr): 3417, 2943, 1732, 1705, 1459, 1392; Figure 12.
¹ H NMR:	δ ppm, 300 MHz, in CDCl ₃ ; 0.67 (3H, s), 0.93 (3H, s), 0.98 (3H, s), 1.00 (3H,
	<i>s</i>), 1.01 (3H, <i>s</i>), 1.03 (3H, <i>d</i> , <i>J</i> = 8.4 Hz), 1.16 (3H, <i>s</i>), 1.18 (3H, <i>s</i>), 1.88 (1H, <i>br</i>
	<i>d</i> , <i>J</i> = 12.0 Hz), 2.13 (1H, <i>br d</i> , <i>J</i> = 12.0 Hz), 2.36 (1H, <i>s</i>), 2.56 (1H, <i>q</i> , <i>J</i> = 6.6
	Hz), 3.22 (1H, $d, J = 15.9$ Hz), 3.44 (1H, $d, J = 15.9$ Hz) and 1.15 (3H, s); Table
	10 and Figure 13.
¹³ C NMR:	δ ppm, 75 MHz, in CDCl ₃ ; 7.3, 16.0, 18.0, 18.1, 18.8, 20.4, 28.2, 30.0, 30.2,
	31.8, 32.1, 32.5, 32.8, 34.6, 35.0, 35.3, 36.0, 37.3, 37.9, 38.3, 39.3, 39.6, 40.7,

42.8, 52.2, 59.1, 60.7, 72.0, 202.8 and 204.1; Table 10 and Figure 14.

5.3 Compound SV-3 (Kokoonol)

Compound SV-3 was obtained as a white, amorphous powder (7.4 mg, 0.0006 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp: 270-271 °C

ESI-MS: m/z (% rel. int.): 443 [M + H]⁺(9); Figure 15.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3543, 2946, 1706, 1630, 1460, 1388; Figure 16.

- ¹H NMR: δ ppm, 300 MHz, in CDCl₃; 0.71 (3H, *s*), 0.86 (3H, *d*, *J* = 6.3 Hz), 0.93 (3H, *s*), 1.07 (3H, *s*), 1.23 (3H, *s*), 1.67 (1H, *m*), 1.95 (1H, *m*), 2.20 (1H, *q*, *J* = 6.3 Hz), 2.29 (1H, *m*), 2.36 (1H, *m*), 4.01 (1H, *d*, *J* = 11.6 Hz) and 4.11 (1H, *d*, *J* = 11.6 Hz); **Table 11** and **Figure 17**.
- ¹³C NMR: δ ppm, 75 MHz, in CDCl₃; 7.2, 14.9, 18.1, 21.0, 22.5, 22.7, 24.4, 28.7, 30.8, 31.0, 32.4, 33.0, 33.3, 35.9, 36.4, 37.4, 37.9, 37.9, 38.6, 39.5, 41.9, 42.6, 42.8, 43.9, 45.7, 53.4, 58.5, 60.3, 64.7 and 213.4; Table 11 and Figure 18.

5.4 Compound SV-4 (26-Hydroxyfriedelane-1,3-dione)

Compound SV-4 was obtained as a white, amorphous powder (7.4 mg, 0.0006 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp: 266-268 C	Mp:	266-268°	С
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ESI-MS:	m/z (% rel. int.): 457 [M + H] ⁺ (21); Figure 19.
IR:	v _{max} cm ⁻¹ (KBr): 3575, 2945, 1728, 1703, 1462, 1391; Figure 20.
¹ H NMR:	δ ppm, 300 MHz, in CDCl ₃ ; 0.68 (3H, s), 0.92 (3H, s), 0.95 (3H, s), 1.02 (3H, d,
	<i>J</i> = 6.8 Hz), 1.04 (3H, <i>s</i>), 1.12 (3H, <i>s</i>), 1.36 (3H, <i>s</i>), 2.38 (1H, <i>s</i>), 2.54 (1H, <i>q</i> , <i>J</i>
	= 6.8 Hz), 3.23 (1H, <i>d</i> , <i>J</i> = 15.9 Hz), 3.44 (1H, <i>d</i> , <i>J</i> = 15.9 Hz), 4.04 (1H, <i>d</i> , <i>J</i> =
	11.6 Hz) and 4.14 (1H, <i>d</i> , <i>J</i> = 11.6 Hz); Table 12 and Figure 21 .
13 C NMD.	S mm 75 MHz in CDC1, 77 161 182 201 208 245 287 202 208

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<sup>15</sup>C NMR: δ ppm, 75 MHz, in CDCl<sub>3</sub>; 7.7, 16.1, 18.2, 20.1, 20.8, 24.5, 28.7, 30.3, 30.8,
32.1, 32.4, 33.3, 34.9, 35.4, 35.8, 35.9, 37.6, 38.5, 39.5, 40.1, 42.2, 42.5, 43.9,
52.5, 59.3, 61.0, 64.4, 72.8, 203.2 and 204.5; Table 12 and Figure 22.
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5.5 Compound SV-5 (21α-Hydroxyfriedelan-3-one)

Compound SV-5 was obtained as colorless cubic crystals (2.3 mg, 0.0002 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp: 266-268 °C

ESI-MS: m/z (% rel. int.): 443 [M + H]⁺(20); Figure 23.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3429, 2931, 1714, 1628, 1452, 1389; Figure 24.

¹H NMR: δ ppm, 300 MHz, in CDCl₃; 0.70 (3H, s), 0.85 (3H, s), 0.85 (3H, d, J = 6.8 Hz), 0.89 (3H, s), 0.97 (3H, s), 1.05 (3H, s), 1.09 (3H, s), 1.17 (3H, s), 2.24 (1H, q, J = 6.8 Hz), 2.30 (1H, m), 2.38 (1H, m) and 3.68 (1H, dd, J = 12.0, 4.2 Hz); **Table** 13 and Figure 25.

¹³C NMR: δ ppm, 75 MHz, in CDCl₃; 7.2, 15.1, 18.2, 18.6, 18.7, 19.7, 22.7, 25.3, 30.6, 30.9, 32.3, 32.9, 33.6, 34.8, 35.7, 36.4, 36.5, 37.9, 39.2, 39.5, 41.7, 41.9, 42.5, 44.7, 47.5, 52.0, 58.6, 59.9, 74.7 and 213.0; Table 13 and Figure 26.

5.6 Compound SV-6 (3β,22α-Dihydroxyolean-12-en-29-oic acid)

Compound SV-6 was obtained as colorless prisms (6.2 mg, 0.0005 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp: 293-295 °C

ESI-MS: m/z (% rel. int.): 429 [M + H -COO]⁺(26); Figure 27.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3491, 2948, 1749, 1650, 1456, 1390, 1363; **Figure 28.**

- ¹H NMR: δ ppm, 500 MHz, in CDC1₃; 0.72 (1H, dd, J = 11.5, 1.5 Hz), 0.77 (3H, s), 0.85 (3H, s), 0.91 (3H, s), 0.92 (3H, s), 0.97 (3H, s), 1.05 (3H, s), 1.19 (3H, s), 1.36 (1H, ddd, J = 12.5, 5.0, 3.0 Hz), 1.50 (1H, dd, J = 13.0, 8.0 Hz), 1.68 (1H, ddd, J = 14.0, 10.5, 3.5 Hz), 2.11 (1H, dd, J = 12.0, 8.5 Hz), 2.24 (1H, d, J = 12.0 Hz), 3.20 (1H, dd, J = 11.5, 4.5 Hz), 4.13 (1H, d, J = 5.5 Hz) and 5.28 (1H, dd, J = 3.5, 2.5 Hz); **Table 14** and **Figure 29.**
- ¹³C NMR: δ ppm, 125 MHz, in CDCl₃; 15.6, 15.6, 17.0, 18.3, 21.0, 23.5, 24.0, 24.3, 25.0, 25.2, 27.2, 28.1, 33.1, 33.8, 35.2, 37.0, 38.6, 38.7, 39.3, 39.5, 39.8, 42.5, 43.4, 47.5, 55.2, 78.9, 83.1, 124.6, 140.2 and 182.4; Table 14 and Figure 30.

5.7 Compound SV-7 (30-Hydroxyfriedelane-1,3-dione)

Compound SV-7 was obtained as a white amorphous powder (5.6 mg, 0.0005 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp: 165-167 °C

HR-ESI-MS: m/z (% rel. int.): 457.3681 [M + H]⁺ (100), calc. for C₃₀H₄₉O₃, 457.3682; Figure 33.
IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3435, 2940, 1729, 1707, 1632, 1458, 1392; **Figure 34.**

- ¹H NMR: δ ppm, 500 MHz, in CDCl₃; 0.67 (3H, *s*), 0.97 (3H, *s*), 0.99 (3H, *s*), 1.02 (3H, *s*), 1.03 (3H, *d*, *J* = 6.5 Hz), 1.13 (3H, *s*), 1.18 (3H, *s*), 1.88 (1H, *br dd*, *J* = 12.5, 3.5 Hz), 2.14 (1H, *ddd*, *J* = 13.5, 7.0, 3.5 Hz), 2.36 (1H, *s*), 2.56 (1H, *q*, *J* = 6.5 Hz), 3.23 (1H, *d*, *J* = 15.8 Hz), 3.34 (1H, *d*, *J* = 10.5 Hz), 3.40 (1H, *d*, *J* = 10.5 Hz) and 3.43 (1H, *d*, *J* = 15.8 Hz); **Table 15** and **Figure 35**.
- ¹³C NMR: δ ppm, 125 MHz, in CDCl₃; 7.3, 16.0, 18.0, 18.1, 18.6, 20.0, 28.1, 28.9, 29.4, 30.0, 30.1, 32.1, 32.1, 33.4, 34.5, 35.8, 37.2, 37.8, 38.1, 38.3, 39.6, 40.6, 42.6, 52.0, 59.1, 60.6, 71.9, 72.0, 202.7 and 204.1; Table 15 and Figure 36.

5.8 Compound SV-8 (21α-Hydroxyfriedelane-1,3-dione)

Compound SV-8 was obtained as a white amorphous powder (4.2 mg, 0.0004 % based on dried weight of *S. verrucosa* stems). The compound is soluble in CH_2Cl_2 .

Mp:	160-162	Ċ

HR-ESI-MS: m/z (% rel. int.): 457.3671 [M + H]⁺ (100), calc. for C₃₀H₄₉O₃, 457.3682; Figure 38.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3436, 2941, 1731, 1705, 1619, 1462, 1392; Figure 39.

- ¹H NMR: δ ppm, 500 MHz, in CDCl₃; 0.67 (3H, *s*), 0.92 (3H, *s*), 0.96 (3H, *s*), 1.03 (3H, *d*, J = 7.0 Hz), 1.05 (3H, *s*), 1.07 (3H, *s*), 1.18 (3H, *s*), 1.19 (3H, *s*), 1.21 (1H, *dd*, *J* = 12.0, 4.5 Hz), 1.60 (1H, *dd*, *J* = 13.8, 4.8 Hz), 1.68 (1H, *dd*, *J* = 12.5, 12.0 Hz), 1.87 (1H, *br dd*, *J* = 9.5, 3.0 Hz), 2.14 (1H, *ddd*, *J* = 13.5, 7.0, 3.5 Hz), 2.37 (1H, *s*), 2.55 (1H, *q*, *J* = 7.0 Hz), 3.23 (1H, *d*, *J* = 15.8 Hz), 3.44 (1H, *d*, *J* = 15.8 Hz) and 3.67 (1H, *dd*, *J* = 12.0, 4.5 Hz); **Table 16** and **Figure 40**.
- ¹³C NMR: δ ppm, 125 MHz, in CDCl₃; 7.3, 16.0, 17.8, 18.1, 18.2, 19.3, 24.9, 29.8, 30.5, 31.8, 32.5, 33.1, 34.2, 34.3, 35.9, 36.0, 37.3, 37.8, 38.7, 38.8, 40.6, 44.2, 47.0, 50.6, 59.1, 60.6, 71.9, 74.3, 202.7 and 204.1; Table 16 and Figure 41.

5.9 Compound FF-1 (Friedelin)

Compound FF-1 was obtained as colorless needles (24.5 mg, 0.0013 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 .

Mp: 260-263 °C

ESI-MS: m/z (% rel. int.): 449 [M + Na]⁺(40); Figure 44.

IR:	v _{max} cm ⁻¹ (KBr): 3431, 2927, 1715, 1463, 1390; Figure 45.
¹ H NMR:	δ ppm, 300 MHz, in CDCl ₃ ; 0.70 (3H, s), 0.85 (3H, s), 0.87 (3H, d, J = 5.4 Hz),
	0.93 (3H, s), 0.99 (3H, s), 0.99 (3H, s), 1.03 (3H, s) and 1.16 (3H, s); Figure 46.
¹³ C NMR:	δ ppm, 75 MHz, in CDCl ₃ ; 6.8, 14.7, 17.9, 18.2, 18.6, 20.3, 22.3, 28.2, 30.0,
	30.5, 31.8, 32.1, 32.4, 32.8, 35.0, 35.3, 35.6, 36.0, 37.5, 38.3, 39.2, 39.7, 41.3,
	41.5, 42.1, 42.8, 53.1, 58.2, 59.5 and 213.1; Figure 47.

5.10 Compound FF-2 (Taraxerol)

Compound FF-2 was obtained as a white amorphous powder (5.8 mg, 0.0003 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 .

Mp:	282-285 °C
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ESI-MS: m/z (% rel. int.): 427 [M + H]⁺(14); Figure 48.

IR∙	$v \text{ cm}^{-1}$ (KBr): 3484–2934–1641–1464	1385. Figure 49
IK.	v_{max} CIII (KDI). 5464, 2954, 1041, 1404,	1365, Figure 49 .

- ¹H NMR: δ ppm, 300 MHz, in CDCl₃; 0.78 (3H, s), 0.80 (3H, s), 0.89 (3H, s), 0.89 (3H, s), 0.91 (3H, s), 0.93 (3H, s), 0.96 (3H, s), 1.07 (3H, s), 3.17 (1H, dd, J = 10.1, 4.7 Hz) and 5.51 (1H, dd, J = 6.3, 3.3 Hz); **Table 17** and **Figure 50**.
- ¹³C NMR: δ ppm, 75 MHz, in CDCl₃; 15.9, 15.9, 17.9, 19.2, 21.7, 26.3, 27.6, 28.4, 29.2, 30.3, 30.3, 33.5, 33.8, 34.1, 35.6, 36.2, 37.1, 38.0, 38.0, 38.1, 38.4, 39.2, 39.4, 41.8, 49.2, 49.7, 56.0, 79.5, 117.3 and 158.5; Table 17 and Figures 51a-51b.

5.11 Compound FF-3 (Betulin)

Compound FF-3 was obtained as a white amorphous powder (13.8 mg, 0.0007 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 .

Mp: 236-238 °C

ESI-MS: m/z (% rel. int.): 466 $[M + H + Na]^+$ (18); Figure 52.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3400, 2942, 1645, 1454, 1376; Figure 53.

¹H NMR: δ ppm, 300 MHz, in CDCl₃; 0.74 (3H, s), 0.80 (3H, s), 0.95 (3H, s), 0.96 (3H, s), 1.00 (3H, s), 1.66 (3H, s), 3.16 (1H, dd, J = 10.8, 5.4 Hz), 3.31 (1H, d, J = 10.8 Hz), 3.77 (1H, d, J = 10.8 Hz), 4.56 (1H, br s) and 4.66 (1H, br s); **Table 18** and **Figure 54**.

¹³C NMR: δ ppm, 75 MHz, in CDCl₃; 14.8, 15.4, 16.0, 16.1, 18.3, 19.1, 20.9, 25.3, 27.0, 27.2, 28.0, 29.2, 29.8, 34.0, 34.3, 37.2, 37.3, 38.8, 38.9, 40.9, 42.7, 47.8, 47.8, 48.8, 50.4, 55.3, 60.2, 78.9, 109.6 and 150.6; Table 18 and Figures 55a-55b.

5.12 Compound FF-4 [4(15)-Eudesmene-1β,6α-diol]

Compound FF-4 was obtained as colorless needles (8.5 mg, 0.0004 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 and MeOH.

Mp:	129-130 °C
$[\alpha]^{20}_{\ D}$:	$+50^{\circ}$
ESI-MS:	m/z (% rel. int.): 239 [M + H] ⁺ (60), 221 [M - H ₂ O] ⁺ (36), 203 [M + H - 2H ₂ O] ⁺
	(30); Figure 56.
IR:	v _{max} cm ⁻¹ (KBr): 3589, 3274, 2937, 1647, 1451, 1386; Figure 57.
¹ H NMR:	δ ppm, 300 MHz, in CDCl ₃ ; 0.68 (3H, s), 0.85 (3H, br s), 0.93 (3H, br s), 1.72
	(1H, d, J = 8.4 Hz), 2.04 (1H, m), 2.30 (1H, m), 3.39 (1H, dd, J = 10.2, 4.2 Hz),
	3.69 (1H, <i>br s</i>), 4.72 and 5.00 (2H, each <i>br s</i>); Table 19 and Figure 58 .
¹³ C NMR:	δ ppm, 75 MHz, in CDCl_3; 12.0, 16.6, 18.6, 21.5, 26.4, 32.3, 35.5, 36.7, 42.1,

49.8, 56.3, 67.4, 79.4, 108.2 and 146.6; Table 19 and Figure 59.

5.13 Compound FF-5 [4(15)-Eudesmene-1β,5α-diol]

Compound FF-5 was obtained as colorless needles (5.1 mg, 0.0003 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 and MeOH.

Mp:	108-110 °C
$[\alpha]^{20}_{\ D}$:	+124°
ESI-MS:	m/z (% rel. int.): 239 [M + H] ⁺ (100), 221 [M + H - H ₂ O] ⁺ (75), 203 [M + H -
	2H ₂ O] ⁺ (55); Figure 60.
IR:	v_{max} cm ⁻¹ (KBr): 3627, 3414, 2952, 1645, 1435, 1368; Figure 61.
¹ H NMR:	δ ppm, 300 MHz, in CDCl ₃ ; 0.74 (3H, <i>s</i>), 0.88 (3H, <i>d</i> , J = 6.6 Hz), 0.90 (3H, <i>d</i> , <i>d</i>)
	<i>J</i> = 6.6 Hz), 2.13 (1H, <i>dd</i> , <i>J</i> = 13.5, 5.1 Hz), 2.68 (1H, <i>ddd</i> , <i>J</i> = 13.5, 8.1, 5.7 Hz)
), 4.03 (1H, dd , $J = 10.4$, 4.8 Hz), 4.73 and 4.83 (2H, each $br s$); Table 20 and
	Figure 62.
¹³ C NMR:	δ ppm, 75 MHz, in CDCl_3; 13.1, 20.1, 20.4, 24.1, 30.2, 30.4, 31.0, 33.2, 34.7,

38.7, 42.7, 73.5, 76.6, 109.0 and 151.1; Table 20 and Figure 63.

5.14 Compound FF-7 (Foveolide A)

Compound FF-7 was obtained as yellow oil (16.2 mg, 0.0008 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 and MeOH.

 $[\alpha]_{D}^{20}$: -120° (*c* 0.05, MeOH)

UV:
$$\lambda_{max}$$
 (MeOH) nm (log ε): 225 (3.09), 275 (2.23); Figure 64.

HR-ESI-MS: m/z (% rel. int.): 251.1621 [M + H]⁺ (100), calc. for C₁₅H₂₃O₃, 251.1647; Figure 65.

IR:
$$v_{max} \operatorname{cm}^{-1}$$
 (KBr): 3430, 2937, 1749, 1682, 1652, 1466, 1390, 1371; **Figure 66.**

¹³C NMR: δ ppm, 125 MHz, in CDCl₃; 9.8, 19.8, 20.0, 24.8, 30.6, 34.0, 35.1, 36.8, 50.3, 54.2, 75.3, 79.6, 128.1, 131.2 and 169.0; **Table 21** and **Figure 68**.

5.15 Compound FF-6 (Foveolide B)

Compound FF-6 was obtained as colorless gum (7.9 mg, 0.0004 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 .

 $[\alpha]_{0}^{20}$: -52° (c 0.05, CHCl₃)

HR-ESI-MS: m/z (% rel. int.): 469.3304 [M + H]⁺ (100), calc. for C₃₀H₄₅O₄, 469.3317; Figure 71.

IR: $v_{max} \text{ cm}^{-1}$ (KBr): 3443, 2933, 1753, 1699, 1464, 1443, 1371; **Figure 72.** ¹H NMR: δ ppm, 500 MHz, in CDCl₃; 0.86 (3H, *d*, *J* = 7.0 Hz), 0.88 (3H, *d*, *J* = 6.5 Hz), 0.97 (3H, *d*, *J* = 6.5 Hz), 1.00 (3H, *d*, *J* = 7.0 Hz), 1.09 (3H, *s*), 1.20 (3H, *s*), 1.59 (1H, *d*, *J* = 17.0 Hz), 2.07 (1H, *d*, *J* = 10.8 Hz), 2.35 (1H, *ddd*, *J* = 13.9, 5.0, 3.0 Hz), 2.64 (1H, *dd*, *J* = 17.0, 3.5 Hz), 2.69 (1H, *ddd*, *J* = 13.9, 12.5, 7.0 Hz), 3.83 (1H, *t*, *J* = 8.0 Hz) and 4.25 (1H, *t*, *J* = 10.8 Hz); **Table 22** and **Figure 73.** ¹³C NMR: δ ppm, 125 MHz, in CDCl₃; 15.1, 16.4, 19.6, 20.2, 20.5, 20.5, 21.6, 25.6, 26.1, 30.5, 30.5, 30.7, 30.9, 34.7, 35.1, 35.9, 39.9, 42.0, 43.2, 45.7, 46.5, 48.8, 52.0, 57.1, 74.2, 77.4, 126.5, 134.2, 178.2 and 215.5; Table 22 and Figure 74.

5.16 Compound FF-8b (Foveoeudesmenone)

Compound FF-8b, the product of the conversion of FF-8a, was initially obtained as colorless oil (14.7 mg, 0.0007 % based on dried weight of *F. foveolata* stems). The compound is soluble in CH_2Cl_2 and MeOH.

$$[\alpha]_{D}^{20}$$
: -50° (*c* 0.01, MeOH)

UV: λ_{max} (MeOH) nm (log ϵ): 245 (3.17); Figure 80.

HR-ESI-MS: m/z (% rel. int.): 237.1842 [M + H]⁺ (100), calc. for C₁₅H₂₅O₂, 237.1849; Figure 81.

	-1									
IR:	v cm	(KBr): 3423,	2958.	1709.	1655.	1604.	1465.	1386.	1370: Figure 8	32.

¹H NMR: δ ppm, 500 MHz, in CDCl₃; 0.91 (3H, *d*, *J* = 7.0 Hz), 0.92 (3H, *d*, *J* = 7.0 Hz), 1.12 (3H, *s*), 1.74 (3H, *d*, *J* = 1.5 Hz), 2.10 (1H, *dt*, *J* = 13.0, 3.0 Hz), 2.53 (1H, *dd*, *J* = 16.5, 12.8 Hz), 2.61 (1H, *dd*, *J* = 16.5, 5.5 Hz), 2.72 (1H, *ddd*, *J* = 15.0, 5.5, 1.5 Hz) and 3.77 (1H, *dd*, *J* = 12.8, 5.5 Hz); **Table 24** and **Figures 83a-83b.** δ ppm, 125 MHz, in CDCl₃; 10.9, 16.2, 19.3, 19.7, 24.1, 31.4, 32.8, 37.8, 41.5, 42.3, 44.2, 74.6, 129.2, 162.8 and 197.4; **Table 24** and **Figure 84.**

5.17 Compound FF-9 (Ethyl rosmarinate)

Compound FF-9 was obtained as yellow gum (15.7 mg, 0.0008 % based on dried weight of *F. foveolata* stems). The compound is soluble in acetone and MeOH.

$[\alpha]^{20}_{\ D}$:	$+112^{\circ}$ (<i>c</i> 0.05, MeOH)
UV:	$λ_{max}$ (MeOH) nm (log ε): 232 (3.56), 290 (3.50), 315 (3.51); Figure 87.
ESI-MS:	m/z (% rel. int.): 388 [M] ⁺ (100), 389 [M + H] ⁺ (30); Figure 88.
IR:	v _{max} cm ⁻¹ (KBr): 3391, 1724, 1653, 1601, 1516, 1445, 1375, 1357; Figure 89.
¹ H NMR:	δ ppm, 500 MHz, in acetone- d_6 ; 1.19 (3H, t , J = 7.0 Hz), 2.99 (1H, dd , J = 11.6,
	7.0 Hz), 4.11 (2H, <i>q</i> , <i>J</i> = 7.0 Hz), 4.75 (1H, <i>dd</i> , <i>J</i> = 13.9, 7.6 Hz), 6.54 (1H, <i>dd</i> , <i>J</i>
	= 7.9, 2.1 Hz), 6.55 (1H, <i>d</i> , <i>J</i> = 15.6 Hz), 6.71 (1H, <i>d</i> , <i>J</i> = 2.1 Hz), 6.71 (1H, <i>d</i> , <i>J</i>
	= 7.9 Hz), 6.82 (1H, <i>d</i> , <i>J</i> = 8.1 Hz), 6.93 (1H, <i>dd</i> , <i>J</i> = 8.1, 1.8 Hz), 7.06 (1H, <i>d</i> , <i>J</i>
	= 1.8 Hz) and 7.40 (1H, d, J = 15.6 Hz); Table 25 and Figure 90.

¹³C NMR: δ ppm, 125 MHz, in acetone-d₆; 14.4, 37.8, 54.9, 61.4, 114.9, 115.9, 116.3, 117.1, 118.9, 121.5, 121.7, 128.3, 129.3, 141.4, 144.8, 145.7, 146.2, 147.9, 166.4 and 172.4; Table 25 and Figure 91.

5.18 Compound FF-10 (Foveospirolide)

Compound FF-10 was obtained as yellow oil (12.5 mg, 0.0006 % based on dried weight of *F. foveolata* stems). The compound is soluble in MeOH.

$$[\alpha]_{D}^{20}$$
: -24° (c 0.05, MeOH)

UV: λ_{max} (MeOH) nm (log ε): 241 (3.22), 276 (2.90), 325 (2.00); Figure 92.

HR-ESI-MS: m/z (% rel. int.): 327.1072 [M + H]⁺ (100), calc. for C₁₅H₁₉O₈, 327.1074; Figure 93.

IR:	$v_{\rm max} {\rm cm}^{-1}$	(KBr): 3366, 29	944, 1778,	1605, 1520,	1455, 1435,	1372; Figure 94.
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¹ H NMR:	δ ppm, 500 MHz, in CD ₃ OD; 2.71 (1H, dd , J = 17.1, 8.4 Hz), 3.03 (1H, dd , J =
	17.1, 12.5 Hz), 3.33 (1H, d, J = 9.3 Hz), 3.33 (1H, ddd, J = 10.9, 9.3, 5.7 Hz),
	3.49 (1H, <i>t</i> , <i>J</i> = 10.9 Hz), 3.58 (1H, <i>t</i> , <i>J</i> = 9.3 Hz), 3.66 (1H, <i>dd</i> , <i>J</i> = 10.9, 5.7
	Hz), 3.82 (3H, <i>s</i>), 4.03 (1H, <i>dd</i> , <i>J</i> = 12.5, 8.4 Hz), 6.75 (1H, <i>d</i> , <i>J</i> = 8.1 Hz), 6.79
	(1H, <i>dd</i> , <i>J</i> = 8.1, 1.8 Hz) and 6.96 (1H, <i>d</i> , <i>J</i> = 1.8 Hz); Table 26 and Figure 95 .
¹³ C NMR:	δ ppm, 125 MHz, in CD ₃ OD; 33.9, 45.2, 56.3, 65.0, 70.2, 71.1, 75.9, 109.2,
	113.6, 115.1, 122.9, 126.8, 146.7, 147.8 and 175.0; Table 26 and Figure 96.

6. Evaluation of Biological Activities

6.1 Determination of Cytotoxic Activity against SW620, CHAGO, HepG2, BT474 and KATO-III Cell Lines

Bioassay of cytotoxicity against human cell cultures *in vitro* was performed by the MTT [3-(4,5)-dimethylthiazol-2,5-diphenyltetrazolium bromide] colorimetric method (Carmichael *et al.*, 1987; Twentyman and Luscombe, 1987). The isolated compounds from *S. verrucosa* and *F. foveolata* stems were evaluated for cytotoxicity against colon carcinoma (SW620), lung carcinoma (CHAGO), human hepatocarcinoma (HepG2), breast carcinoma (BT474) and gastric carcinoma (KATO-III) cell lines.

All cell lines were grown in Roswell Park Memorial Institute (RPMI) medium 1640 containing 5% foetal calf serum (FCS) and incubated at 37 °C in humidified atmosphere of 5%

 CO_2 . Exponentially growing cells were seeded in 96-well plates (200 µl/well at a density of 2.5 × 10^4 cells/ml). The cells were incubated in growth media for 24 h. The tested compounds at various concentrations were added (2 µl/well) and incubated for up to 72 h. At the end of this incubation period, 10 µl of MTT solution (5 mg/ml) were added and the plate was further incubated for 4h at 37 °C. After removal of the culture supernatants, 150 µl of DMSO and 25 µl of glycine buffer (pH 10.4) were added to each well sequentially in order to facilitate solubilization of the formazan product. The plate was shaken, and the absorbance was measured at 540 nm in a SynergyTM Absorbance Microplate Reader (Bio-Tek). The percentage of cell survival was calculated as followed.

% Cell survival = $[OD \text{ test}/OD \text{ control}] \times 100$

Whereas OD test and OD control are the absorbance from treated condition and untreated condition, respectively.

Dose response curves were plotted from 5 concentrations (μ g/ml) of 10-fold serially diluted test compounds against their percentage of cell survival. The concentration of each compound tested that reduces the effect by 50 % was also calculated from these curves and reported as the median inhibitory concentration (IC₅₀ value). The cytotoxicity of the compounds tested was compared with that of the anticancer antibiotic doxorubicin, which was used as the positive control.

6.2 Determination of Anti-Mycobacterium tuberculosis Activity

The bioassay was performed by using green fluorescent protein (GFP)-expressing *Mycobacterium tuberculosis* strain H_{37} Ra (Changsen, Franzblau, and Palittapongarnpim, 2003). The mycobacteria were cultivated on 7H10 agar containing 30 µg/ml kanamycin at 37°C for 4 weeks or until growth was observed. Starter culture was prepared by fully looping 2-3 single colonies into 7H9 broth supplemented with 0.2% v/v glycerol, 0.1% w/v solution (BD Biosciences) and 30 µg/ml of kanamycin. The mixture was then incubated at 37°C in 200 rpm shaker incubator until the optical density (OD) at 550 nm was between 0.5 and 1.

For batch cultivation, the starter cultures were transferred at the rate of 1/10 volume to the 7H9 broth and incubated at 37° C in 200 rpm shaker incubator until the OD at 550 nm was approximately 0.5 to 1. Cells were pelleted, washed and suspended in PBS buffer, and then sonicated 8 times for 15 seconds each. Sonicated samples were then aliquoted and frozen at -80° C

prior to use. Titer stocks were determined by colony forming unit (cfu) assay and the seeding density. For assay in 384-well format, the seeding was approximately 2×10^4 to 1×10^5 cfu/ml/well.

The assay was performed in duplicate. Each well contained 5 μ l of test samples serially diluted in 5% DMSO, followed by 45 μ l of cell suspension prepared as described above. Plates were incubated at 37°C for 7 days and fluorescence was measured at the excitation and emission wavelengths of 485 and 535 nm. Fluorescent signals on day zero were used as background to subtract from the signals on day 7. The percentage of growth inhibition is calculated from the mean of fluorescent unit of cells treated with sample (Fu_t) and untreated cells (Fu_s), as in the following equation:

% Inhibition = $[1 - (Fu_t / Fu_c)] \times 100$

The lowest concentration of the compound that inhibits cell growth by 90% is reported as the minimum inhibitory concentration (MIC). Streptomycin, isoniazid, ofloxacin and ethambutol are used as positive controls, and 0.5% DMSO is used as a negative control.

CHAPTER IV

RESULTS AND DISCUSSION

Chemical constituents of the *n*-hexane-acetone and EtOAc extracts of *S. verrucosa* stems and the *n*-hexane and CH_2Cl_2 extracts of *F. foveolata* stems were investigated. Chromatographic techniques were employed in order to isolate eight compounds (SV-1 to SV-8) from *S. verrucosa* and ten compounds (FF-1 to FF-10) from *F. foveolata* as described in Chapter III. Identification and structure elucidation of these compounds were achieved through spectroscopic techniques, including UV, IR, MS and NMR. Isolated compounds from both plants were examined for their biological activities, including cytotoxicity against five human cancer cell lines and anti-*Mycobacterium tuberculosis* activity.

1. Identification of Compound SV-1 and FF-1 (Friedelin)

Compound SV-1, obtained as colorless needles (4.0 mg, 0.0003 % yield), appeared as a purple spot upon spraying with 10% sulfuric acid and heated. According to the $[M + H]^+$ peak at m/z 427 in the mass spectrum (**Figure 7**), its molecular formula could be identified as $C_{30}H_{50}O$. Its IR absorption bands at 1716, 1463 and 1390 cm⁻¹ (**Figure 8**) indicated the presence of ketone, methylene and methyl groups in the molecule, respectively.

The ¹H-NMR spectrum of compound SV-1 (**Figure 9**) showed eight methyl signals including seven methyl singlets at δ 0.71 (H₃-24), 0.85 (H₃-25), 0.93 (H₃-30), 0.98 (H₃-26), 0.99 (H₃-29), 1.03 (H₃-27) and 1.15 ppm (H₃-28) and one methyl doublet at δ 0.86 ppm (J = 5.4 Hz, H₃-23). These methyl signals are characteristic of friedelane-type triterpenoid skeleton.

The ¹³C-NMR spectrum of this compound (**Figure 10**) displayed 30 carbon signals, including those of eight methyl carbons at δ 6.8 (C-23), 14.7 (C-24), 17.9 (C-25), 18.6 (C-27), 20.3 (C-26), 31.8 (C-29), 32.1 (C-28) and 35.0 ppm (C-30), eleven methylene carbons at δ 18.3 (C-7), 22.3 (C-1), 30.5 (C-12), 32.5 (C-21), 32.8 (C-15), 35.4 (C-19), 35.7 (C-11), 36.0 (C-16), 39.3 (C-22), 41.3 (C-6) and 41.5 ppm (C-2), four methine carbons at δ 42.9 (C-18), 53.1 (C-8), 58.3 (C-4) and 59.5 ppm (C-10), six quaternary carbons at δ 28.2 (C-20), 30.0 (C-17), 37.5 (C-9), 38.3 (C-14), 39.7 (C-13), 42.2 (C-5) and one keto-carbonyl at δ 213.0 ppm (C-3).

Comparison of the NMR data of compound SV-1 with previous reports (Gunatilaka *et al.*, 1983b; Klass and Tinto, 1992) (**Table 9**) led to the identification of this compound as friedelin, which is one of the most common friedelane-type triterpenes. The compound is widely distributed in higher plants and can be frequently found in members of the family Celastraceae, for example, in the inner bark of *Kokoona zeylanica* (Gunatilaka et al., 1982), in the stems of *Maytenus diversifolia* (Nozaki *et al.*, 1986), in the stem bark of *M. macrocarpa* (Chavez *et al.*, 1998), in the roots of *M. obtusifolia* (Silva *et al.*, 2008) and in the fruits of *M. salicifolia* (Valladao *et al.*, 2010). Furthermore, this triterpenoid has also been reported as a constituent of *Salacia beddomei* (Hisham *et al.*, 1996b), *S. elliptica* (Duarte *et al.*, 2009) and *S. verrucosa* (Jangruang *et al.*, 2009).

Compound FF-1, obtained as colorless needles (24.5 mg, 0.0013% yield), appeared as a purple spot on TLC upon spraying with 10% sulfuric acid and heated. The mass spectrum (**Figure 44**) showed $[M + Na]^+$ peak at m/z 449, suggesting the molecular formula $C_{30}H_{50}O$. IR and NMR spectral data of compound FF-1 (**Figures 45-47**) are similar to those of compound SV-1. Therefore, this compound was determined as friedelin.



Friedelin has been reported to possess various biological activities such as cytotoxic (Zheng, 1994), antifungal (Duraipandiyan, Gnanasekar, and Ignacimuthu, 2010), antiinflammatory, analgesic and antipyretic activities (Antonisamy, Duraipandiyan, and Ignacimuthu, 2011).

Desition	SV-1			Friedelin*		
FOSITION	$\delta_{\rm c}$	$\delta_{_{\rm H}}$	δ_{c}	$\delta_{_{ m H}}$		
1	22.3	-	22.3	-		
2	41.5	-	41.5	-		
3	213.0	-	213.0	-		
4	58.3	-	58.2	-		
5	42.2	-	42.1	-		
6	41.3	-	41.3	-		
7	18.3	-	18.2	-		
8	53.1	-	53.1	-		
9	37.5	-	37.4	-		
10	59.5	-	59.5	-		
11	35.7	-	35.6	-		
12	30.5	-	30.5	-		
13	39.7	-	39.7	-		
14	38.3	-	38.3	-		
15	32.8	-	32.8	-		
16	36.0	-	36.0	-		
17	30.0	-	30.0	-		
18	42.9	-	42.8	-		
19	35.4	-	35.3	-		
20	28.2	-	28.1	-		
21	32.5	-	32.4	-		
22	39.3	-	39.2	-		
23	6.8	0.86 (3H, <i>d</i> , <i>J</i> = 5.4 Hz)	6.8	0.87 (3H, <i>d</i>)		
24	14.7	0.71 (3H, s)	14.7	0.71 (3H, <i>s</i>)		
25	17.9	0.85 (3H, <i>s</i>)	17.9	0.86 (3H, <i>s</i>)		
26	20.3	0.98 (3H, <i>s</i>)	20.3	1.00 (3H, <i>s</i>)		
27	18.6	1.03 (3H, <i>s</i>)	18.7	1.05 (3H, <i>s</i>)		
28	32.1	1.15 (3H, <i>s</i>)	32.1	1.17 (3H, <i>s</i>)		
29	31.8	0.99 (3H, s)	31.8	1.00 (3H, <i>s</i>)		
30	35.0	0.93 (3H, s)	35.0	0.95 (3H, <i>s</i>)		

Table 9. ¹H- and ¹³C-NMR spectral data for compound SV-1 and friedelin (300 and 75 MHz, in CDCl₃).

*Gunatilaka et al., 1983b and Klass and Tinto, 1992

2. Identification of Compound SV-2 (Friedelane-1,3-dione)

Compound SV-2, obtained as colorless needles (25.5 mg, 0.002 % yield), appeared as a yellow spot upon spraying with 10% sulfuric acid and heated. According to its $[M + 4H + Na]^+$ peak at m/z 467 in the mass spectrum (**Figure 11**), the molecular formula was determined as $C_{30}H_{48}O_2$. The IR spectrum (**Figure 12**) showed absorption bands at 1732 and 1705 cm⁻¹, indicating the presence of keto carbonyls in the molecule.

The ¹H-NMR spectrum of compound SV-2 (**Figure 13**) showed eight methyl signals, which could be categorized into seven methyl singlets at δ 0.67 (H₃-24), 0.93 (H₃-30), 0.98 (H₃-29), 1.00 (H₃-27), 1.01 (H₃-26), 1.16 (H₃-28) and 1.18 ppm (H₃-25) and one methyl doublet at δ 1.03 ppm (J = 8.4 Hz, H₃-23). Based on the evidence of both methyl signals and a methine quartet at δ 2.56 ppm (J = 6.6 Hz, H-4), the compound could be identified as a friedelane-type triterpene (Gunatilaka *et al.*, 1982). In addition, two methylene doublets at δ 3.22 and 3.44 ppm (each 1H, J = 15.9 Hz, H₂-2) and a methine singlet at δ 2.36 ppm (H-10) were also characteristic of a 1,3-diketotriterpene (Klass and Tinto, 1992).

Thirty carbon signals could be observed in the ¹³C-NMR spectrum of this compound (**Figure 14**), supporting its identification as a triterpenoid. These signals represent eight methyl carbons at δ 7.3 (C-23), 16.0 (C-24), 18.0 (C-25), 18.8 (C-27), 20.4 (C-26), 31.8 (C-29), 32.1 (C-28) and 35.0 ppm (C-30), ten methylene carbons at δ 18.1 (C-7), 30.2 (C-12), 32.5 (C-15), 32.8 (C-21), 34.6 (C-11), 35.3 (C-19), 36.0 (C-16), 39.3 (C-22), 40.7 (C-6) and 60.7 ppm (C-2), four methine carbons at δ 42.8 (C-18), 52.2 (C-8), 59.1 (C-4) and 72.0 ppm (C-10), six quaternary carbons at δ 28.2 (C-20), 30.0 (C-17), 37.3 (C-9), 37.9 (C-5), 38.3 (C-14), 39.6 (C-13), and two keto-carbonyl carbons at δ 202.8 (C-1) and 204.1 ppm (C-3). Both downfield methine carbon resonances at δ 59.1 (C-4) and 72.0 (C-10) were suggestive of their positions adjacent to two keto carbonyls in ring A of a friedelane-type triterpene.

NMR data of compound SV-2 are identical to those of friedelane-1,3-dione (Klass and Tinto, 1992), as shown in **Table 10**. This triterpenoids has been isolated from a Guyanese medicinal plant, *Peritassa compta* (Klass and Tinto, 1992), and from members of the genus *Salacia* such as *S. beddomei* (Hisham *et al.*, 1996b), *S. chinensis* (Kishi *et al.*, 2003) and *S. campestris* (Paulo *et al.*, 2005). It was also found as the major constituent of *S. verrucosa* (Jangruang *et al.*, 2009).



Friedelane-1,3-dione

Table 10. ¹H- and ¹³C-NMR spectral data for compound SV-2 and friedelane-1,3-dione (300 and 75 MHz, in CDCl₃).

Position SV-2		Friedelane-1,3-dione*		
rosition	δ_{c} δ_{H}		δ_{c}	$\delta_{_{ m H}}$
1	202.8	-	202.8	-
2	60.7	3.22 (1H, <i>d</i> , <i>J</i> = 15.9 Hz)	60.7	3.24, 3.46
		3.44 (1H, <i>d</i> , <i>J</i> = 15.9 Hz)		
3	204.1	_	204.2	-
4	59.1	2.56 (1H, $q, J = 6.6$ Hz)	59.1	2.58
5	37.9	_	37.9	-
6	40.7	1.38 (1H, <i>m</i>), 1.88 (1H, <i>m</i>)	40.7	1.39, 1.90
7	18.1	-	18.1	-
8	52.2	-	52.2	-
9	37.3	-	37.2	-
10	72.0	2.36 (1H, <i>s</i>)	71.9	2.38
11	34.6	1.13 (1H, <i>m</i>), 2.13 (1H, <i>m</i>)	34.6	1.14, 2.15
12	30.2	-	30.2	-
13	39.6	-	39.5	-
14	38.3	_	38.3	-
15	32.5	-	32.4	-

D:4:		SV-2		Friedelane-1,3-dione*
Position	δ_{c}	$\delta_{_{ m H}}$	$\delta_{\rm c}$	$\delta_{_{ m H}}$
16	36.0	-	35.9	-
17	30.0	-	30.0	-
18	42.8	-	42.7	-
19	35.3	-	35.3	-
20	28.2	-	28.2	-
21	32.8	-	32.8	-
22	39.3	-	39.3	-
23	7.3	1.03 (3H, <i>d</i> , <i>J</i> = 8.4 Hz)	7.3	1.05 (3H, <i>d</i>)
24	16.0	0.67 (3H, s)	16.0	0.69 (3H, s)
25	18.0	1.18 (3H, s)	18.0	1.20 (3H, s)
26	20.4	1.01 (3H, s)	20.3	1.03 (3H, s)
27	18.8	1.00 (3H, s)	18.8	1.02 (3H, s)
28	32.1	1.16 (3H, <i>s</i>)	32.1	1.18 (3H, s)
29	31.8	0.98 (3H, s)	31.8	1.00 (3H, s)
30	35.0	0.93 (3H, s)	35.0	0.94 (3H, s)

* Klass and Tinto, 1992

3. Identification of Compound SV-3 (Kokoonol)

Compound SV-3, which was obtained as an amorphous powder (7.4 mg, 0.0006 % yield) after recrystallization, gave a purple spot upon spraying with 10% sulfuric acid and heated. The ESI-MS spectrum (**Figure 15**) showed a quasi-molecular $[M + H]^+$ ion peak at m/z 443, suggesting the molecular formula $C_{30}H_{50}O_2$. Its IR spectrum (**Figure 16**) displayed absorption bands at 3543 and 1706 cm⁻¹ corresponding to hydroxyl and carbonyl groups, respectively.

The ¹H-NMR spectrum (**Figure 17**) showed six methyl singlets at δ 0.71 (H₃-24), 0.93 (H₃-25), 0.95 (H₃-29), 1.03 (H₃-30), 1.07 (H₃-26) and 1.23 ppm (H₃-28) and one methyl doublet at δ 0.86 ppm (J = 6.3 Hz, H₃-23) of a friedelane triterpenoid. Two pairs of methylene protons resonated at δ 1.67 and 1.95 ppm (H₂-1) and at δ 2.29 and 2.36 ppm (H₂-2), whereas a pair of oxygenated methylene signals appeared at δ 4.01 and 4.11 ppm (J = 11.6 Hz, H₂-27). In addition, a methine quartet at δ 2.20 ppm (J = 6.3 Hz) could be assigned as H-4. Therefore, this compound

could be a friedelane-type triterpene in which one methyl group has been replaced by a hydroxymethylene moiety.

The ¹³C-NMR spectrum of this compound (**Figure 18**) displayed 30 carbon signals from seven methyl carbons at δ 7.3 (C-23), 14.9 (C-24), 18.1 (C-25), 22.5 (C-26), 31.0 (C-30), 33.0 (C-28) and 35.9 ppm (C-29), twelve methylene carbons at δ 21.0 (C-7), 22.7 (C-1), 24.4 (C-12), 32.4 (C-15), 33.3 (C-21), 36.4 (C-16), 37.4 (C-19), 37.9 (C-11), 39.5 (C-22), 41.9 (C-2), 42.8 (C-6) and 64.7 ppm (C-27), four methine carbons at δ 43.9 (C-18), 53.4 (C-8), 58.5 (C-4) and 60.3 ppm (C-10), six quaternary carbons at δ 28.7 (C-20), 30.8 (C-17), 37.9 (C-9), 38.6 (C-14), 42.6 (C-5) and 45.7 ppm (C-13) and one keto-carbonyl carbon at δ 213.4 ppm (C-3). These ¹³C-NMR data are similar to those of friedelin; the major differences are the downfield shift of the C-13 signal and a methyl signal of friedelin has been replaced by an oxygenated methylene one. Thus, these data help identify compound SV-3 as the 27-hydroxy derivative of friedelin, named kokoonol (27-hydroxyfriedelan-3-one).

Comparison of the NMR data of compound SV-3 and kokoonol isolated from the stem bark of a flacourtiaceous plant, *Caloncoba glauca* (Giner *et al.*, 1993) is shown in **Table 11**. Kokoonol was first reported as a constituent of *Kokoona zeylanica* (Gunatilaka *et al.*, 1983a) and, later, was also found in *Salacia chinensis* stems (Morikawa *et al.*, 2003); both plants belong to major genera of the family Celastraceae, of which various friedelane-type triterpenes were found.



Kokoonol

Deeitien	SV-3		Kokoonol*	
Position	δ_{c}	$\delta_{_{\rm H}}$	$\delta_{\rm c}$	$\delta_{_{ m H}}$
1	22.7	1.67 (1H, <i>m</i>)	22.5	1.69 (1H, <i>ddd</i> , <i>J</i> = 12.8, 5.1, 5.1 Hz)
		1.95 (1H, <i>m</i>)		1.97 (1H, <i>m</i>)
2	41.9	2.29 (1H, <i>m</i>)	41.6	2.24 (1H, <i>ddd</i> , <i>J</i> = 13.7, 13.7, 7.3 Hz)
		2.36 (1H, <i>m</i>)		2.40 (1H, <i>ddd</i> , <i>J</i> = 13.7, 5.1, 2.1 Hz)
3	213.4	-	213.3	-
4	58.5	2.20 (1H, q, J = 6.3 Hz)	58.4	2.23 (1H, q , $J = 6.4$ Hz)
5	42.6	-	42.3	-
6	42.8	-	41.5	1.35 (1H, <i>m</i>), 1.75 (1H, <i>m</i>)
7	21.0	-	18.6	-
8	53.4	-	53.8	-
9	37.9	-	37.6	-
10	60.3	-	59.7	1.48 (1H, <i>m</i>)
11	37.9	-	37.8	-
12	24.4	-	24.1	-
13	45.7	-	45.4	-
14	38.6	-	38.4	-
15	32.4	-	32.2	-
16	36.4	-	36.3	-
17	30.8	-	30.3	-
18	43.9	-	43.3	-
19	37.4	-	37.1	-
20	28.7	-	28.5	-
21	33.3	-	32.6	-
22	39.5	-	40.1	-
23	7.2	0.86 (3H, d, J = 6.3 Hz)	7.1	0.89 (3H, <i>d</i> , <i>J</i> = 6.6 Hz)
24	14.9	0.71 (3H, s)	14.9	0.74 (3H, <i>s</i>)
25	18.1	0.93 (3H, <i>s</i>)	18.2	0.93 (3H, <i>s</i>)
26	22.5	1.07 (3H, <i>s</i>)	22.3	1.12 (3H, <i>s</i>)
27	64.7	4.01 (1H, d, J = 11.6 Hz)	63.4	4.02 (1H, br d)
		4.11 (1H, <i>d</i> , <i>J</i> = 11.6 Hz)		4.06 (1H, <i>d</i> , <i>J</i> = 11.5 Hz)
28	33.0	1.23 (3H, s)	32.8	1.24 (3H, <i>s</i>)
29	35.9	0.95 (3H, s)	35.8	0.98 (3H, s)
30	31.0	1.03 (3H, s)	30.6	1.01 (3H, <i>s</i>)

Table 11. ¹H- and ¹³C-NMR spectral data for compound SV-3 and kokoonol (300 and 75 MHz, in $CDCl_3$).

*Giner et al., 1993

4. Identification of Compound SV-4 (26-Hydroxyfriedelane-1,3-dione)

Compound SV-4, obtained as a white, amorphous powder (7.4 mg, 0.0006 % yield), appeared as a purple spot upon spraying with 10% sulfuric acid and heated. According to quasimolecular $[M + H]^+$ ion peak at m/z 457 in the mass spectrum (**Figure 19**), its molecular formula was calculated as $C_{30}H_{48}O_3$. The absorption band at 3575 cm⁻¹ of its IR spectrum (**Figure 20**) showed that the compound contains hydroxyl group, in addition to absorption bands at 1728 and 1703 cm⁻¹ that indicated the presence of diketone group in the molecule.

The ¹H-NMR spectrum of compound SV-4 (**Figure 21**) showed several prominent signals, including six methyl singlets at δ 0.68 (H₃-24), 0.92 (H₃-29), 0.95 (H₃-30), 1.04 (H₃-27), 1.12 (H₃-28) and 1.36 ppm (H₃-25), one methyl doublet at δ 1.02 ppm (J = 6.8 Hz, H₃-23), two methylene doublets of a 1,3-diketone at δ 3.23 and 3.44 ppm (each 1H, J = 15.9 Hz, H₂-2), two oxymethylene doublets at δ 4.04 and 4.14 ppm (each 1H, J = 11.6 Hz, H₂-26), one methine singlet at δ 2.38 ppm (H-10) and a methine quartet at δ 2.54 ppm (J = 6.8 Hz, H-4). According to chemical shifts and splitting patterns of these protons, compound SV-4 was determined as a derivative of friedelane-1,3-dione with one additional hydroxyl group.

The ¹³C-NMR spectrum (**Figure 22**) displayed 30 carbon signals, including those of seven methyl carbons at δ 7.7 (C-23), 16.1 (C-24), 18.2 (C-25), 20.1 (C-27), 32.1 (C-28), 32.4 (C-30) and 34.9 ppm (C-29), eleven methylene carbons at δ 20.8 (C-7), 24.5 (C-15), 30.3 (C-12), 33.3 (C-21), 35.4 (C-11), 35.8 (C-16), 35.9 (C-19), 39.5 (C-22), 42.2 (C-6), 61.0 (C-2) and 64.4 ppm (C-26), four methine carbons at δ 43.9 (C-18), 52.5 (C-8), 59.3 (C-4) and 72.8 ppm (C-10), six quaternary carbons at δ 28.7 (C-20), 30.8 (C-17), 37.6 (C-9), 38.5 (C-5), 40.1 (C-13) and 42.5 ppm (C-14), and two keto-carbonyl carbons at δ 203.2 (C-1) and 204.5 ppm (C-3). Its 1,3-diketone skeleton was supported by the chemical shifts of C-1, C-3, C-4 and C-10. Compound SV-4 possesses one more methylene carbon (attached to an oxygen) and one less methyl carbon than friedelane-1,3-dione. When the ¹³C-NMR data of both compounds were compared, the more downfield shift of C-14 of this compound could clearly be observed, suggesting that there was a hydroxyl substituent at C-26. Comparison of its NMR data with those of 26-hydroxyfriedelane-1,3-dione. (Matsuda *et al.*, 1999) (**Table 12**) confirmed this. Therefore, compound SV-4 was identified as 26-hydroxyfriedelane-1,3-dione.

26-Hydroxyfriedelane-1,3-dione has been reported as a constituent of plants in the genus *Salacia* such as *S. oblonga* (Matsuda *et al.*, 1999), *S. reticulata* (Yoshikawa *et al.*, 2002) and *S. verrucosa* (Jangruang *et al.*, 2009). The compound, isolated from the roots of *S. oblonga*, weakly inhibited rat lens aldose reductase, a key enzyme in the development of eye and nerve damage in diabetic patients (Matsuda *et al.*, 1999).



26-Hydroxyfriedelane-1,3-dione

Table 12. ¹H- and ¹³C-NMR spectral data for compound SV-4 and 26-hydroxyfriedelane-1,3dione (300 and 75 MHz, in CDCl₃).

Desition	SV-4		26-Hydroxyfriedelane-1,3-dione*	
Position	$\delta_{\rm c}$	$\delta_{_{ m H}}$	$\delta_{\rm c}$	$\delta_{_{ m H}}$
1	203.2	-	202.8	-
2	61.0	3.23 (1H, <i>d</i> , <i>J</i> = 15.9 Hz)	60.6	3.23 (1H, <i>d</i> , <i>J</i> = 15.9 Hz)
		3.44 (1H, <i>d</i> , <i>J</i> = 15.9 Hz)		3.44 (1H, <i>d</i> , <i>J</i> = 15.9 Hz)
3	204.5	-	204.1	-
4	59.3	2.54 (1H, $q, J = 6.8$ Hz)	58.9	2.54 (1H, q, J = 6.5 Hz)
5	38.5	-	38.1	-
6	42.2	1.21 (1H, <i>m</i>), 1.85 (1H, <i>m</i>)	41.7	1.20, 1.85
7	20.8	1.63 (1H, <i>m</i>), 1.83 (1H, <i>m</i>)	20.4	1.60, 1.80
8	52.5	-	52.0	1.35

D :/:		SV-4	26-Hydroxyfriedelane-1,3-dione*	
Position	$\delta_{\rm C}$	$\delta_{_{ m H}}$	δ _c	$\delta_{_{ m H}}$
9	37.6	-	37.1	-
10	72.8	2.38 (1H, s)	72.4	2.38 (1H, s)
11	35.4	1.68 (1H, <i>m</i>), 2.19 (1H, <i>m</i>)	35.0	1.65, 2.16
12	30.3	-	29.9	1.10, 1.30
13	40.1	-	39.7	-
14	42.5	-	42.0	-
15	24.5	1.16 (1H, <i>m</i>), 2.09 (1H, <i>m</i>)	24.1	1.15, 2.10
16	35.8	-	35.3 1.65	
17	30.8	-	30.4	-
18	43.9	1.47	43.5	1.46
19	35.9	-	35.4	1.65
20	28.7	-	28.3	-
21	33.3	-	32.9 0.95, 1.35	
22	39.5	-	39.1 0.90, 1.50	
23	7.7	1.02 (3H, d, J = 6.8 Hz)	7.3	1.00 (3H, d, J = 6.5 Hz)
24	16.1	0.68 (3H, s)	15.7	0.67 (3H, s)
25	18.2	1.36 (3H, s)	17.8	1.35 (3H, s)
26	64.4	4.04 (1H, <i>d</i> , <i>J</i> = 11.6 Hz)	64.0	4.03 (1H, <i>d</i> , <i>J</i> = 11.5 Hz)
		4.14 (1H, <i>d</i> , <i>J</i> = 11.6 Hz)		4.13 (1H, <i>d</i> , <i>J</i> = 11.5 Hz)
27	20.1	1.04 (3H, <i>s</i>)	19.7	1.03 (3H, s)
28	32.1	1.12 (3H, <i>s</i>)	31.7	1.11 (3H, s)
29	34.9	0.92 (3H, <i>s</i>)	34.5	0.91 (3H, s)
30	32.4	0.95 (3H, s)	32.0	0.94 (3H, s)

*Matsuda et al., 1999

5. Identification of Compound SV-5 (21α-Hydroxyfriedelan-3-one)

Compound SV-5, obtained as colorless cubic crystals (2.3 mg, 0.0002 % yield), appeared as a purple spot upon spraying with 10% sulfuric acid and heated. Its ESI-MS spectrum (**Figure 23**) presented a quasi-molecular $[M + H]^+$ ion peak at m/z 443, suggesting that the compound was an isomer of kokoonol, with the same molecular formula of $C_{30}H_{50}O_2$. The IR spectrum (**Figure** 24) showed absorption bands of hydroxyl and carbonyl groups at 3429 and 1714 cm^{-1} , respectively.

The ¹H-NMR spectrum of compound SV-5 (**Figure 25**) showed seven methyl singlets at $\delta 0.70$ (H₃-24), 0.85 (H₃-25), 0.89 (H₃-26), 0.97 (H₃-30), 1.05 (H₃-29), 1.09 (H₃-27) and 1.17 ppm (H₃-28), and one methyl doublet at $\delta 0.85$ ppm (J = 6.8 Hz, H₃-23) of a friedelane skeleton. Similar to kokoonol, the methylene multiplets at $\delta 2.30$ and 2.38 ppm (1H each) could be characterized as those of H₂-2. Similarly, a one-proton quartet at $\delta 2.24$ ppm (J = 6.8 Hz) could be assigned to the methine H-4, while a doublet of doublets at $\delta 3.68$ ppm (J = 12.0, 4.2 Hz) suggested the presence of an oxymethine proton in this triterpenoid molecule.

Thirty signals could be observed in its ¹³C-NMR spectrum (**Figure 26**), including those of eight methyl carbons at δ 7.2 (C-23), 15.1 (C-24), 18.2 (C-26), 18.7 (C-25), 19.7 (C-27), 25.3 (C-30), 32.3 (C-29) and 33.6 ppm (C-28), ten methylene carbons at δ 18.6 (C-7), 22.7 (C-1), 30.6 (C-15), 30.9 (C-12), 35.7 (C-11), 36.4 (C-16), 36.5 (C-19), 41.7 (C-2), 41.9 (C-6) and 47.5 ppm (C-22), five methine carbons at δ 44.7 (C-18), 52.0 (C-8), 58.6 (C-4), 59.9 (C-10) and 74.7 ppm (C-21), six quaternary carbons at δ 32.9 (C-17), 34.8 (C-20), 37.9 (C-9), 39.2 (C-14), 39.5 (C-13) and 42.5 ppm (C-5) and one keto-carbonyl carbon at δ 213.0 ppm (C-3). Compared with the ¹³C-NMR data of friedelin, this triterpene exhibited one less methylene signals while one additional oxymethine signal was observed at δ 74.7 ppm. The signals of C-20 quaternary carbon and C-22 methylene carbon were both significantly shifted downfield. These data indicated the position of the oxymethine carbon as at C-21.

The ¹H-NMR data of compound SV-5 were compared with those of 21α -hydroxyfriedelan-3-one isolated from the inner bark of *Kokoona zeylanica* (Gunatilaka *et al.*, 1982), while its ¹³C-NMR data were compared with those of 6β , 21α -dihydroxy-3-friedelanone, a structurally related friedelane-type triterpene from *Lepidobotrys staudtii* (Tane *et al.*, 1996) (**Table 13**). The structure of compound SV-5 was thus confirmed as 21α -hydroxyfriedelan-3-one. The triterpene is a significant chemotaxonomic marker of plants in the family Celastraceae. In the genus *Salacia*, it has been reported from *S. reticulata* stem bark (Kumar *et al.*, 1990) and was also shown to exhibit antiprotozoal activity against *Giardia intestinalis* with an IC₅₀ value of 19.8 μ M (Mena-Rejon *et al.*, 2007).



21α-Hydroxyfriedelan-3-one

Table 13. ¹H- and ¹³C-NMR spectral data for compound SV-5 and 6β ,21 α -dihydroxy-3-friedelanone (300 and 75 MHz, in CDCl₃).

Desition	SV-5		6β,21α-Dihydroxy-3-friedelanone*	
Position	$\delta_{\rm c}$	$\delta_{_{\rm H}}$	$\delta_{\rm c}$	$\delta_{_{ m H}}$
1	22.7	1.95 (1H, <i>m</i>)	21.9	1.76 (1H, <i>m</i>), 2.00 (1H, <i>m</i>)
2	41.7	2.30 (1H, <i>m</i>)	41.2	2.27 (1H, dd, J = 13.2 Hz)
		2.38 (1H, <i>m</i>)		2.39 (1H, <i>m</i>)
3	213.0	-	212.5	-
4	58.6	2.24 (1H, $q, J = 6.8$ Hz)	58.3	2.41 (1H, q , J = 6.7 Hz)
5	42.5	-	47.6	-
6	41.9	-	79.4	3.66 (1H, <i>dd</i> , <i>J</i> = 10.7, 4.7 Hz)
7	18.6	-	29.3	-
8	52.0	-	48.3	-
9	37.9	-	37.5	-
10	59.9	-	58.5	1.50 (1H, <i>m</i>)
11	35.7	-	35.2	-
12	30.9	-	30.1	-
13	39.5	-	39.1	-
14	39.2	-	38.6	-
15	30.6	-	30.4	-
16	36.4	-	36.0	-
17	32.9	-	32.5	-

Desition	SV-5		6β,21α-Dihydroxy-3-friedelanone*	
Position	$\delta_{\rm c}$	$\delta_{_{\rm H}}$	$\delta_{\rm c}$	$\delta_{_{\rm H}}$
18	44.7	-	44.3	1.58 (1H, <i>m</i>)
19	36.5	-	36.0	-
20	34.8	-	34.4	-
21	74.7	3.68 (1H, <i>dd</i> , <i>J</i> = 12.0, 4.2 Hz)	74.3	3.69 (1H, <i>m</i>)
22	47.5	-	47.0	-
23	7.2	0.85 (3H, d, J = 6.8 Hz)	10.4	1.11 (3H, <i>d</i> , <i>J</i> = 6.7 Hz)
24	15.1	0.70 (3H, s)	9.1	0.75 (3H, s)
25	18.7	0.85 (3H, s)	17.8	0.86 (3H, s)
26	18.2	0.89 (3H, s)	17.7	0.90 (3H, s)
27	19.7	1.09 (3H, s)	19.2	1.10 (3H, <i>s</i>)
28	33.6	1.17 (3H, s)	33.1	1.19 (3H, s)
29	32.3	1.05 (3H, s)	31.9	1.06 (3H, s)
30	25.3	0.97 (3H, s)	24.9	0.98 (3H, s)

*Tane et al., 1996

6. Identification of Compound SV-6 (3β,22α-Dihydroxyolean-12-en-29-oic acid)

Compound SV-6 was obtained as colorless prisms (6.2 mg, 0.0005 % yield) which gave a purple spot upon spraying with 10% sulfuric acid and heated. The mass spectrum (**Figure 27**) showed a prominent fragment ion peak at m/z 429 [M + H - COO]⁺, suggesting its molecular formula as $C_{30}H_{48}O_4$. The IR spectrum (**Figure 28**) showed absorption bands at 3491, 1749 and 1650 cm⁻¹, representing hydroxyl, carboxylic and olefinic groups in the molecule, respectively.

The ¹H-NMR spectrum of this compound (**Figure 29**) showed seven methyl singlets at δ 0.77 (H₃-24), 0.85 (H₃-28), 0.91 (H₃-26), 0.92 (H₃-25), 0.97 (H₃-23), 1.05 (H₃-27) and 1.19 ppm (H₃-30). The H₂-19 methylene protons resonated separately as a multiplet at δ 1.88 (1H, H-19_b) and a doublet at δ 2.24 ppm (1H, J = 12.0 Hz, H-19_a). Three doublets of doublets at δ 0.72 (1H, J = 11.5, 1.5 Hz), 1.50 (1H, J = 13.0, 8.0 Hz) and 2.11 ppm (1H, J = 12.0, 8.5 Hz) could be assigned to the methine H-5, H-9 and H-18, respectively. Two oxymethine protons gave a double doublets at δ 3.20 (1H, J = 11.5, 4.5 Hz, H-3) and a doublet at δ 4.13 ppm (1H, J = 5.5 Hz, H-22). In addition, an olefinic proton resonated as the most downfield doublet of doublets at δ 5.28 ppm (1H, J = 3.5, 2.5 Hz, H-12). Based on these data, compound SV-6 was deduced to be different

from the previously identified triterpenoids and that its basic skeleton should be the oleanane-type triterpene with one double bond and two hydroxyl substitutions.

Its ¹³C-NMR spectrum (Figure 30), with the aid of DEPT and HSOC experiments, showed 30 carbon signals from seven tertiary methyls at δ 15.6 (C-24 and C-25), 17.0 (C-26), 21.0 (C-30), 24.0 (C-27), 25.0 (C-28) and 28.1 ppm (C-23), nine methylenes at δ 18.3 (C-6), 23.5 (C-11), 24.3 (C-15), 25.2 (C-16), 27.2 (C-2), 33.1 (C-7), 33.8 (C-19), 38.6 (C-1) and 39.8 ppm (C-21), three methines at δ 43.4 (C-18), 47.5 (C-9) and 55.2 ppm (C-5), two oxymethines at δ 78.9 (C-3) and 83.1 ppm (C-22), one olefinic methine at δ 124.6 ppm (C-12), six quaternary carbons at δ 35.2 (C-17), 37.0 (C-10), 38.7 (C-4), 39.3 (C-8), 39.5 (C-20) and 42.5 ppm (C-14), one olefinic quaternary carbon at δ 140.2 ppm (C-13) and one carboxylic carbon at δ 182.4 ppm (C-29). The last signal suggested that the compound was the acid derivative of an oleanane-type triterpene. HMBC correlations (Figures 31a-31c, Table 14) observed between the signal of H-3 and those of C-1, C-2, C-23 and C-24, as well as between H-22 resonance and those of C-16, C-18, C-20 and C-28, indicated that these oxymethine protons located at C-3 and C-22, respectively. The correlations observed between H-12 resonance and those of C-9, C-11, C-14 and C-18 supported the assignment of one double bond between C-12 and C-13. Finally, one carboxylic group was located at C-29 according to correlations between the signal of H₃-30 and those of C-19, C-20 and C-29. NOE correlations, observed in the NOESY spectrum (Figure 32) between the α -oriented H₃-23 and H-3 as well as the β -oriented H₃-28 and H-22, indicated the β - and α orientation of the hydroxyl groups at C-3 and C-22, respectively. Therefore, compound SV-6 was identified as $3\beta_{22\alpha}$ -dihydroxyolean-12-en-29-oic acid. This compound has been given two trivial names. It was firstly reported as an antileukemic constituent of Maytenus diversifolia named maytenfolic acid (Nozaki et al., 1982, 1986). It was later found in the woody part of the roots and in the cell culture of Tripterygium wilfordii and shown to be anti-inflammatory. The compound was also given another trivial name as triptotriterpenic acid A (Kutney et al., 1992, 1993).

Maytenfolic acid (triptotriterpenic acid A) is widely distributed in plants of the family Celastraceae such as *Salacia oblonga* (Matsuda *et al.*, 1999), *S. chinensis* (Morikawa *et al.*, 2003), *Maytenus laevis* (Nakagawa *et al.*, 2004) and *Celastrus hypoleucus* (Wang *et al.*, 2005). The triterpenoid exhibited various biological activities including inhibitory effect against rat lens aldose reductase with an IC₅₀ value of 72 μ M (Morikawa *et al.*, 2003), cytokine-inducing activity on human peripheral blood mononuclear cells (PBMCs) (Nakagawa *et al.*, 2004) and significant activity against HIV replication in H9 lymphocyte cells with an EC₅₀ value of 5.65 μ g/ml (Wu *et al.*, 2004).



3β,22α-Dihydroxyolean-12-en-29-oic acid

Table 14. ¹H- and ¹³C-NMR spectral data for 3β ,22 α -dihydroxyolean-12-en-29-oic acid (500 and 125 MHz, in CDCl₃).

Position	$\delta_{\rm c}$	δ_{H}	НМВС (Н→С)
1	38.6	0.96 (1H, <i>m</i>)	2, 5
		1.59 (1H, <i>m</i>)	2, 3, 5, 9, 10
2	27.2	0.96 (1H, <i>m</i>)	
		1.58 (1H, m)	3, 4, 10
3	78.9	3.20 (1H, <i>dd</i> , <i>J</i> = 11.5, 4.5 Hz)	1, 2, 23, 24
4	38.7	-	
5	55.2	0.72 (1H, dd, J = 11.5, 1.5 Hz)	4, 6, 7, 10, 23, 24, 25
6	18.3	1.35 (1H, <i>m</i>)	5, 7, 10
		1.53 (1H, <i>m</i>)	5, 10
7	33.1	1.36 (1H, ddd , $J = 12.5, 5.0, 3.0$ Hz)	5, 6, 8, 9, 26
		1.46 (1H, <i>m</i>)	8, 14
8	39.3	-	

Position	$\delta_{\rm c}$	$\delta_{_{ m H}}$	НМВС (Н→С)
9	47.5	1.50 (1H, <i>dd</i> , <i>J</i> = 13.0, 8.0 Hz)	8, 10, 11, 14, 25, 26
10	37.0	-	
11	23.5	1.85 (2H, <i>m</i>)	8, 9, 12, 13
12	124.6	5.28 (1H, <i>dd</i> , <i>J</i> = 3.5, 2.5 Hz)	9, 11, 14, 18
13	140.2	_	
14	42.5	_	
15	24.3	1.08 (1H, <i>m</i>)	13, 14, 27
		1.68 (1H, <i>ddd</i> , <i>J</i> = 14.0, 10.5, 3.5 Hz)	8, 13, 14, 16, 27
16	25.2	1.16 (1H, <i>m</i>)	
		1.87 (1H, <i>m</i>)	14, 15, 17, 18, 28
17	35.2	-	
18	43.4	2.11 (1H, <i>dd</i> , <i>J</i> = 12.0, 8.5 Hz)	12, 13, 14, 17, 28
19	33.8	1.88 (1H, <i>m</i>)	13, 17, 29, 30
		2.24 (1H, <i>d</i> , <i>J</i> = 12.0 Hz)	13, 17, 18, 20, 21, 29, 30
20	39.5	-	
21	39.8	1.46 (1H, <i>m</i>)	17, 20, 29, 30
		1.88 (1H, <i>m</i>)	17, 20, 22, 29, 30
22	83.1	4.13 (1H, <i>d</i> , <i>J</i> = 5.5 Hz)	16, 18, 20, 28
23	28.1	0.97 (3H, <i>s</i>)	3, 4, 5, 24
24	15.6	0.77 (3H, <i>s</i>)	3, 4, 5, 23
25	15.6	0.92 (3H, <i>s</i>)	1, 5, 9, 10
26	17.0	0.91 (3H, <i>s</i>)	7, 8, 9, 14
27	24.0	1.05 (3H, s)	8, 13, 14, 15
28	25.0	0.85 (3H, s)	16, 17, 18, 22
29	182.4	-	
30	21.0	1.19 (3H, <i>s</i>)	19, 20, 29

7. Identification of Compound SV-7 (30-Hydroxyfriedelane-1,3-dione)

Compound SV-7 was isolated as a white amorphous powder (5.6 mg, 0.0005 % yield) with the same molecular formula $C_{30}H_{48}O_3$ as compound SV-4. This molecular formula was analyzed from the $[M + H]^+$ ion peak at m/z 457.3681 in its high resolution ESI-TOF mass spectrum (**Figure 33**). The IR spectrum showed absorption bands of hydroxyl group at 3435 cm⁻¹

and carbonyl groups at 1729 and 1707 cm⁻¹ (**Figure 34**). Its ¹H NMR spectrum (**Figure 35**, **Table 15**) was characteristic of a 1,3-diketofriedelane triterpenoid, showing six methyl singlets at δ 0.67 (H₃-24), 0.97 (H₃-29), 0.99 (H₃-26), 1.02 (H₃-27), 1.13 (H₃-28) and 1.18 ppm (H₃-25), one methyl doublet at δ 1.03 ppm (J = 6.5 Hz, H₃-23) along with two methylene doublets of a 1,3-diketone at δ 3.23 and 3.43 ppm (each 1H, J = 15.8 Hz, H₂-2). A pair of doublet signals of an oxymethylene protons resonated at δ 3.34 and 3.40 ppm (1H each, both J = 10.5 Hz, H₂-30). The more upfield shift of these doublets, when compared to those of compound SV-4, indicated the different location of the hydroxymethyl group in the molecule.

¹³C-NMR data, together with DEPT and HSQC experiments (**Figure 36**), helped categorizing the rest of its carbon signals as those of seven methyls at δ 7.3 (C-23), 16.0 (C-24), 18.1 (C-25), 18.6 (C-27), 20.0 (C-26), 28.9 (C-29) and 32.1 ppm (C-28), ten methylenes at δ 18.0 (C-7), 28.1 (C-21), 29.4 (C-19), 30.1 (C-12), 32.1 (C-15), 34.5 (C-11), 35.8 (C-16), 38.1 (C-22), 40.6 (C-6) and 60.6 ppm (C-2), one oxymethylene at δ 72.0 ppm (C-30), four methines at δ 42.6 (C-18), 52.0 (C-8), 59.1 (C-4) and 71.9 ppm (C-10), six quaternary carbons at δ 30.0 (C-17), 33.4 (C-20), 37.2 (C-9), 37.8 (C-5), 38.3 (C-14) and 39.6 ppm (C-13), and two keto-carbonyl signals at δ 202.7 (C-1) and 204.1 ppm (C-3).

HMBC correlations (**Figures 37a-37c**) from both oxymethylene doublets to C-20, C-21 and C-29, as well as from the signals of H_2 -19 and H_3 -29 to C-30, indicated the position of the hydroxyl group to be at C-30. The more upfield resonance of the hydroxy-substituted methylene carbon (at δ 72.0 instead of at around δ 74-75 ppm for C-29) also supported the C-30 substitution (Nozaki *et al.*, 1986). Consequently, the structure of compound SV-7 was characterized as 30-hydroxyfriedelane-1,3-dione.

This triterpene has previously been isolated from the stem bark of *Salacia reticulata* var. β -*diandra* (Gunatilaka, 1986). However, no spectral data of the compound were presented in the literature. Therefore, this is the first report of its NMR assignments.



30-Hydroxyfriedelane-1,3-dione

Table 15.	1 H- and 1	³ C-NMR spectral d	ata for 30-hy	droxyfriedel	ane-1,3-dione	(500 and	125
MHz, in C	DCl ₃).						

Position	$\delta_{\rm c}$	$\delta_{_{ m H}}$	HMBC (H \rightarrow C)
1	202.7	-	
2	60.6	3.23 (1H, <i>d</i> , <i>J</i> = 15.8 Hz)	1, 3, 4, 10
		3.43 (1H, <i>d</i> , <i>J</i> = 15.8 Hz)	1,3
3	204.1	-	
4	59.1	2.56 (1H, $q, J = 6.5$ Hz)	3, 5, 6, 10, 23, 24
5	37.8	-	
6	40.6	1.37 (1H, <i>m</i>)	4, 5, 8, 10, 24
		1.88 (1H, <i>br dd</i> , <i>J</i> = 12.5, 3.5 Hz)	5, 7, 8, 10, 24
7	18.0	1.44 (1H, <i>m</i>)	5, 14
		1.51 (1H, <i>m</i>)	5, 14
8	52.0	1.23 (1H, <i>m</i>)	9, 11, 26
9	37.2	-	
10	71.9	2.36 (1H, s)	1, 4, 5, 6, 8, 9, 11, 24, 25
11	34.5	1.13 (1H, <i>m</i>)	9, 12, 13
		2.14 (1H, <i>ddd</i> , <i>J</i> = 13.5, 7.0, 3.5 Hz)	8, 9, 12, 13, 25

Position	δ_{c}	$\delta_{_{ m H}}$	HMBC (H \rightarrow C)
12	30.1	1.30 (1H, <i>m</i>)	11, 27
		1.39 (1H, <i>m</i>)	14, 27
13	39.6	-	
14	38.3	-	
15	32.1	1.25 (1H, <i>m</i>)	13, 14, 17, 26
		1.49 (1H, <i>m</i>)	16, 17, 26
16	35.8	1.38 (1H, <i>m</i>)	14, 15, 18, 22
		1.53 (1H, <i>m</i>)	15, 17, 22, 28
17	30.0	-	
18	42.6	1.49 (1H, <i>m</i>)	16, 17, 20, 27
19	29.4	1.20 (1H, <i>m</i>)	13, 17, 18, 20, 30
		1.46 (1H, <i>m</i>)	17, 18, 20, 21, 30
20	33.4	-	
21	28.1	1.14 (1H, <i>m</i>)	17, 19, 20, 29
		1.42 (1H, <i>m</i>)	17, 19, 20, 22, 30
22	38.1	0.97 (1H, <i>m</i>)	17, 18, 20, 28
		1.55 (1H, <i>m</i>)	17, 21, 28
23	7.3	1.03 (3H, <i>d</i> , <i>J</i> = 6.5 Hz)	3, 4, 5
24	16.0	0.67 (3H, <i>s</i>)	4, 5, 6, 10
25	18.1	1.18 (3H, <i>s</i>)	8, 9, 10, 11
26	20.0	0.99 (3H, s)	8, 13, 14, 15
27	18.6	1.02 (3H, <i>s</i>)	12, 13, 14, 18
28	32.1	1.13 (3H, s)	16, 17, 18, 22
29	28.9	0.97 (3H, s)	19, 20, 21, 30
30	72.0	3.34 (1H, <i>d</i> , <i>J</i> = 10.5 Hz)	20, 21, 29
		3.40 (1H, d, J = 10.5 Hz)	20, 21, 29

8. Structure Elucidation of Compound SV-8 (21α-Hydroxyfriedelane-1,3-dione)

Compound SV-8 was obtained as a white amorphous powder (4.2 mg, 0.0004 % yield). The molecular formula of this compound was determined as $C_{30}H_{48}O_3$ from its quasi-molecular $[M + H]^+$ ion peak at m/z 457.3671 in the high resolution ESI-TOF mass spectrum (**Figure 38**). Its IR spectrum displayed absorption bands at 3436, 1731 and 1705 cm⁻¹ (**Figure 39**) ascribable to hydroxyl and carbonyl functions in the structure. The general appearance of the ¹H- and ¹³C-NMR spectra, in addition to the information obtained from its mass spectrum, suggested its structure to be a friedelane-type triterpenoid isomer of compounds SV-4 and SV-7.

The ¹H-NMR spectrum of compound SV-8 (**Figure 40**) showed seven methyl singlets at δ 0.67 (H₃-24), 0.92 (H₃-26), 0.96 (H₃-29), 1.05 (H₃-30), 1.07 (H₃-27), 1.18 (H₃-28) and 1.19 ppm (H₃-25), a methyl doublet at δ 1.03 ppm (H₃-23), two methylene doublets of 1,3-diketone at δ 3.23 and 3.44 ppm (each 1H, *J* = 15.8 Hz, H₂-2) along with an oxymethine double doublet at δ 3.67 ppm (*J* = 12.0, 4.5 Hz, H-21).

Its ¹³C-NMR spectrum (Figure 41) exhibited signals of eight methyl carbons at δ 7.3 (C-23), 16.0 (C-24), 17.8 (C-26), 18.2 (C-25), 19.3 (C-27), 24.9 (C-29), 31.8 (C-30) and 33.1 ppm (C-28), nine methylenes at δ 18.1 (C-7), 29.8 (C-12), 30.5 (C-15), 34.2 (C-11), 35.9 (C-16), 36.0 (C-19), 40.6 (C-6), 47.0 (C-22) and 60.6 ppm (C-2), four methines at δ 44.2 (C-18), 50.6 (C-8), 59.1 (C-4) and 71.9 ppm (C-10), an oxymethine carbon at δ 74.3 ppm (C-21), six quaternary carbons at δ 32.5 (C-17), 34.3 (C-20), 37.3 (C-9), 37.8 (C-5), 38.7 (C-14) and 38.8 ppm (C-13), and two keto-carbonyl carbons at δ 202.7 (C-1) and 204.1 ppm (C-3). These spectral data are similar to those of friedelane-1,3-dione (compound SV-2) except for the signals due to the hydroxyl substitution. These NMR data (Table 16) were assigned with the aid of DEPT, HSQC and HMBC experiments. The position of the hydroxyl group at C-21 was established by HMBC experiment (Table 16), in which correlations could be observed between the oxymethine proton signal at δ 3.67 ppm and those of C-20, C-22, C-29 and C-30 (Figures 42a-42b). The orientation of this hydroxyl group was determined as α by NOESY experiment, which showed NOE correlations between H-21 β and H₂-28 as well as H₂-30 (Figure 43), and by comparison of its NMR data with those of related friedelane triterpene (Tane et al., 1996). The molecular structure of compound SV-8 was therefore determined as a new 1,3-diketofriedelane triterpene, 21ahydroxyfriedelane-1,3-dione.



21α-Hydroxyfriedelane-1,3-dione

Table 16. ¹H- and ¹³C-NMR spectral data for 21α -hydroxyfriedelane-1,3-dione (500 and 125 MHz, in CDCl₃).

Position	$\delta_{\rm C}$	$\delta_{_{ m H}}$	НМВС (Н→С)
1	202.7	-	
2	60.6	3.23 (1H, <i>d</i> , <i>J</i> = 15.8 Hz)	1, 3, 4, 10
		3.44 (1H, d, J = 15.8 Hz)	1,3
3	204.1	-	
4	59.1	2.55 (1H, q, J = 7.0 Hz)	3, 5, 6, 10, 23, 24
5	37.8	-	
6	40.6	1.39 (1H, <i>m</i>)	7, 8, 24
		1.87 (1H, <i>br dd</i> , <i>J</i> = 9.5, 3.0 Hz)	5, 7, 8, 10, 24
7	18.1	1.41 (1H, <i>m</i>)	
		1.52 (1H, <i>m</i>)	8
8	50.6	1.28 (1H, <i>m</i>)	7, 13, 14, 26
9	37.3	-	
10	71.9	2.37 (1H, s)	1, 4, 5, 8, 9, 11, 24, 25
11	34.2	1.18 (1H, <i>m</i>)	25
		2.14 (1H, <i>ddd</i> , <i>J</i> = 13.5, 7.0, 3.5 Hz)	8, 9, 12, 13, 25

Position	$\delta_{\rm c}$	$\delta_{_{ m H}}$	HMBC (H \rightarrow C)	
12	29.8	1.26 (1H, <i>m</i>)	9, 13, 14	
		1.34 (1H, <i>m</i>)	13, 14	
13	38.8	-		
14	38.7	-		
15	30.5	1.27 (1H, <i>m</i>)	13, 14, 26	
		1.44 (1H, <i>m</i>)	13, 14, 17, 26	
16	35.9	1.53 (1H, <i>m</i>)	17, 22	
		1.60 (1H, <i>m</i>)	22	
17	32.5	-		
18	44.2	1.51 (1H, <i>m</i>)	13, 14, 17, 19, 27	
19	36.0	1.53 (1H, <i>m</i>)	13, 17, 18, 20, 21, 29	
		1.60 (1H, <i>dd</i> , <i>J</i> = 13.8, 4.8 Hz)	13, 17, 18, 20, 21, 29	
20	34.3	-		
21	74.3	3.67 (1H, <i>dd</i> , <i>J</i> = 12.0, 4.5 Hz)	20, 22, 29, 30	
22	47.0	1.21 (1H, <i>dd</i> , <i>J</i> = 12.0, 4.5 Hz)	18, 20, 21	
		1.68 (1H, <i>dd</i> , <i>J</i> = 12.5, 12.0 Hz)	16, 17, 21, 28	
23	7.3	1.03 (3H, d, J = 7.0 Hz)	3, 4, 5	
24	16.0	0.67 (3H, <i>s</i>)	4, 5, 6, 10	
25	18.2	1.19 (3H, <i>s</i>)	8, 9, 10, 11	
26	17.8	0.92 (3H, s)	8, 13, 14, 15	
27	19.3	1.07 (3H, s)	12, 13, 14, 18	
28	33.1	1.18 (3H, s)	16, 17, 18, 22	
29	24.9	0.96 (3H, s)	19, 20, 21, 30	
30	31.8	1.05 (3H, <i>s</i>)	19, 20, 21, 29	

9. Identification of Compound FF-2 (Taraxerol)

Compound FF-2, obtained as a white amorphous powder (5.8 mg, 0.0003 % yield), appeared as a purple spot on TLC upon spraying with 10% sulfuric acid and heated. The mass spectrum (**Figure 48**) exhibited a quasi-molecular $[M + H]^+$ ion peak at m/z 427, consistent with the formula molecule $C_{30}H_{50}O$. Its IR absorption bands at 3484 and 1641 cm⁻¹ (**Figure 49**) indicated the presence of hydroxyl and olefinic functions in the molecule, respectively.

The ¹H-NMR spectrum of compound FF-2 (**Figure 50**) showed eight methyl singlets at δ 0.78 (H₃-25), 0.80 (H₃-28), 0.89 (H₃-26 and H₃-30), 0.91 (H₃-24), 0.93 (H₃-29), 0.96 (H₃-23) and 1.07 ppm (H₃-27), an oxymethine double doublet at δ 3.17 ppm (J = 10.1, 4.7 Hz, H-3) and one double doublet of an olefinic proton at δ 5.51 ppm (J = 6.3, 3.3 Hz, H-15). Its ¹³C-NMR spectrum (**Figures 51a-51b**) displayed 30 carbon signals, including those of eight methyl carbons at δ 15.9 (C-24 and C-25), 21.7 (C-30), 26.3 (C-27), 28.4 (C-23), 30.3 (C-26 and C-28) and 33.8 ppm (C-29), ten methylene carbons at δ 17.9 (C-11), 19.2 (C-6), 27.6 (C-2), 33.5 (C-22), 34.1 (C-21), 35.6 (C-7), 36.2 (C-12), 37.1 (C-16), 38.4 (C-1) and 41.8 ppm (C-19), three methine carbons at δ 49.2 (C-9), 49.7 (C-18) and 56.0 ppm (C-5), an oxymethine carbon at δ 79.5 ppm (C-3), six quaternary carbons at δ 29.2 (C-20), 38.0 (C-13 and C-17), 38.1 (C-10), 39.2 (C-4) and 39.4 ppm (C-8), one olefinic methine carbon at δ 117.3 ppm (C-15) and one olefinic quaternary carbon at δ 158.5 ppm (C-14). These NMR data (**Table 17**) were characteristic of a taraxerane-type triterpene having one double bond at C-14/C-15, as well as one β-hydroxyl substituent at C-3. Therefore, compound FF-2 was identified as taraxerol. This is the first report of the triterpene from this plant.

Taraxerol is widely distributed and has been found in various plant parts such as the leaves of *Rhizophora apiculata* (Rhizophoraceae) (Kokpol and Chavasiri, 1990), the stem of *Opuntia dillenii* (Cactaceae) (Jiang *et al.*, 2006) and the root bark of *Phyllanthus columnaris* (Euphorbiaceae) (Jamal, Yaacob, and Din, 2009). Taraxerol, produced by *Agrobacterium*-transformed root cultures of butterfly pea *Clitoria ternatea* (Fabaceae), was reported to be anti-cancer (Swain, Rout, and Chand, 2012).



Taraxerol

Table 17. ¹H- and ¹³C-NMR spectral data for compound FF-2 and taraxerol (300 and 75 MHz, in $CDCl_3$).

Position	FF-2		Taraxerol*		
	δ_{c}	$\delta_{_{\rm H}}$	$\delta_{\rm c}$	$\delta_{_{\rm H}}$	
1	38.4	-	38.2	-	
2	27.6	-	27.4	-	
3	79.5	3.17 (1H, dd, J = 10.1, 4.7 Hz)	79.3	3.20 (1H, <i>dd</i> , <i>J</i> = 9.5, 4.0 Hz)	
4	39.2	-	39.0	-	
5	56.0	-	55.7	-	
6	19.2	-	19.0	-	
7	35.6	-	35.3	-	
8	39.4	-	39.2	-	
9	49.2	-	48.9	-	
10	38.1	-	37.9	-	
11	17.9	-	17.7	-	
12	36.2	-	36.0	-	
13	38.0	-	37.9	-	
14	158.5	-	158.3	-	
15	117.3	5.51 (1H, <i>dd</i> , <i>J</i> = 6.3, 3.3 Hz)	117.1	5.53 (1H, <i>dd</i> , <i>J</i> = 5.5, 2.6 Hz)	

Position	FF-2		Taraxerol*		
	$\delta_{\rm C}$	$\delta_{_{ m H}}$	δ_{c}	$\delta_{_{ m H}}$	
16	37.1	-	36.9	-	
17	38.0	-	37.9	-	
18	49.7	-	49.5	-	
19	41.8	-	41.5	-	
20	29.2	-	29.0	-	
21	34.1	-	33.9	-	
22	33.5	-	33.3	-	
23	28.4	0.96 (3H, s)	28.2	0.98 (3H, s)	
24	15.9	0.91 (3H, s)	15.6	0.93 (3H, s)	
25	15.9	0.78 (3H, s)	16.0	0.81 (3H, s)	
26	30.3	0.89 (3H, s)	30.1	0.91 (3H, s)	
27	26.3	1.07 (3H, s)	26.1	1.09 (3H, s)	
28	30.3	0.80 (3H, s)	30.0	0.83 (3H, s)	
29	33.8	0.93 (3H, s)	33.6	0.96 (3H, s)	
30	21.7	0.89 (3H, s)	21.5	0.91 (3H, s)	

*Jamal et al., 2009

10. Identification of Compound FF-3 (Betulin)

Compound FF-3, obtained as a white amorphous powder (13.8 mg, 0.0007% yield), appeared as a purple spot on TLC upon spraying with 10% sulfuric acid and heated. The mass spectrum (**Figure 52**) showed a quasi-molecular $[M + H + Na]^+$ ion peak at m/z 466 consistent with the molecular formula $C_{30}H_{50}O_2$. Its IR absorption bands at 3400 and 1645 cm⁻¹ (**Figure 53**) indicated the presence of hydroxyl and olefinic functions in the molecule, respectively.

The ¹H-NMR spectrum of compound FF-3 (**Figure 54**) showed five methyl singlets at δ 0.74 (H₃-24), 0.80 (H₃-25), 0.95 (H₃-23), 0.96 (H₃-27) and 1.00 ppm (H₃-26), two broad exomethylene singlets and one methyl singlet of an isopropenyl group at δ 4.56, 4.66 (each 1H, H₂-29) and 1.66 ppm (H₃-30), a pair of oxymethylene doublets at δ 3.31 and 3.77 ppm (each 1H, *J* = 10.8 Hz, H₂-28), and one oxymethine double doublet at δ 3.16 ppm (*J* = 10.8, 5.4 Hz, H-3). Its

¹³C-NMR spectrum (**Figures 55a-55b**) displayed 30 carbon signals, including those of six methyl carbons at δ 14.8 (C-27), 15.4 (C-24), 16.0 (C-26), 16.1 (C-25), 19.1 (C-30) and 28.0 ppm (C-23), ten methylene carbons at δ 18.3 (C-6), 20.9 (C-11), 25.3 (C-12), 27.0 (C-15), 27.2 (C-2), 29.2 (C-16), 29.8 (C-21), 34.0 (C-22), 34.3 (C-7) and 38.8 ppm (C-1), one oxymethylene carbon at δ 60.6 ppm (C-28), five methine carbons at δ 37.3 (C-13), 47.8 (C-19), 48.8 (C-18), 50.4 (C-9) and 55.3 ppm (C-5), one oxymethine carbon at δ 78.9 ppm (C-3), five quaternary carbons at δ 37.2 (C-10), 38.9 (C-4), 40.9 (C-8), 42.7 (C-14), 47.8 ppm (C-17), one olefinic methylene carbon and one olefinic quaternary carbon at δ 109.7 (C-29) and 150.5 ppm (C-20), respectively.

These spectral data indicated that compound FF-3 was a lupane-type triterpene possessing an isopropenyl group and two hydroxyl substitutions at C-3 and C-28. Therefore, it was identified as the known triterpenoid, betulin. Comparison of its NMR data with those of betulin isolated from the stem bark of *Salacia cordata* (Tinto *et al.*, 1992) (**Table 18**) confirmed this assignment.



Betulin

Betulin is one of the lupane-type triterpenoids frequently found in plants and especially in *Diospyros* species (Ebenaceae). The compound has been shown to exhibit various biological activities, for example, anti-inflammatory (Del Recio *et al.*, 1995), antimycobacterial (Cantrell, Franzblau, and Fisher, 2001) and antitumor activities (Rzeski *et al.*, 2009). However, this is the first isolation of the compound from *Ficus foveolata*.

Position	FF-3		Betulin*	
	δ_{c}	$\delta_{_{\rm H}}$	δ _c	$\delta_{_{\rm H}}$
1	38.8	-	38.8	0.89, 1.65
2	27.2	-	27.2	1.58
3	78.9	3.16 (1H, <i>dd</i> , <i>J</i> = 10.8, 5.4 Hz)	78.9	3.18
4	38.9	-	38.9	-
5	55.3	0.66 (1H, br d, J = 9.6 Hz)	55.3	0.67
6	18.3	-	18.3	1.38, 1.52
7	34.3	-	34.3	1.39
8	40.9	-	40.9	-
9	50.4	-	50.4	1.27
10	37.2	-	37.2	-
11	20.9	-	20.9	1.19, 1.41
12	25.3	-	25.3	1.03, 1.63
13	37.3	-	37.3	1.64
14	42.7	-	42.7	-
15	27.0	-	27.0	1.04, 1.70
16	29.2	-	29.2	1.20, 1.93
17	47.8	-	47.8	-
18	48.8	-	48.8	1.57
19	47.8	2.36 (1H, <i>m</i>)	47.8	2.38
20	150.5	-	150.6	-
21	29.8	-	29.8	1.40, 1.95
22	34.0	-	34.0	1.02, 1.86
23	28.0	0.95 (3H, <i>s</i>)	28.0	0.96
24	15.4	0.74 (3H, <i>s</i>)	15.4	0.76
25	16.1	0.80 (3H, <i>s</i>)	16.1	0.82
26	16.0	1.00 (3H, <i>s</i>)	16.0	1.02
27	14.8	0.96 (3H, s)	14.8	0.98
28	60.6	3.31 (1H, d, J = 10.7 Hz)	60.2	3.31
		3.77 (1H, d, J = 10.7 Hz)		3.77
29	109.7	4.56 (1H, <i>br s</i>), 4.66 (1H, <i>br s</i>)	109.6	4.58, 4.68
30	19.1	1.66 (3H, s)	19.1	1.68

Table 18. ¹H- and ¹³C-NMR spectral data for compound FF-3 and betulin (300 and 75 MHz, in $CDCl_3$).

*Tinto et al., 1992
11. Identification of Compound FF-4 [4(15)-Eudesmene-1β,6α-diol]

Compound FF-4, obtained as colorless needles (8.5 mg, 0.0004 % yield), appeared as a purple spot on TLC upon spraying with 10% sulfuric acid and heated. Its mass spectrum (**Figure 56**) showed a quasi-molecular $[M + H]^+$ ion peak at m/z 239, suggesting the molecular formula of $C_{15}H_{26}O_2$. Its IR absorption bands at 3589 and 1647 cm⁻¹ (**Figure 57**) indicated the presence of hydroxyl and olefinic groups in the molecule, respectively.

The ¹H-NMR spectrum of compound FF-4 (**Figure 58**) showed one prominent methyl singlet at δ 0.68 (H₃-14), two methyl broad singlets of an isopropyl group at δ 0.85 (H₃-12) and 0.93 ppm (H₃-13), a pair of methylene multiplets at δ 2.04 (1H, H-3 β) and 2.30 ppm (1H, H-3 α) and two exocyclic methylene broad singlets at δ 4.72 (1H, H-15b) and 5.00 ppm (1H, H-15a). One oxymethine proton resonated as a doublet of doublets at δ 3.39 ppm (*J* = 10.2, 4.2 Hz, H-1), whereas another oxymethine proton gave a broad singlet at δ 3.69 ppm (H-6). The ¹³C-NMR spectrum (**Figure 59**) displayed 15 carbon signals of a sesquiterpene skeleton, which could be differentiated as those of one tertiary methyl at δ 12.0 ppm (C-14), two secondary methyls at δ 16.6 (C-12) and 21.5 ppm (C-13), four methylenes at δ 18.6 (C-8), 32.3 (C-2), 35.5 (C-3) and 36.7 ppm (C-9), an exomethylene at δ 108.2 ppm (C-15), three methines at δ 26.4 (C-11), 49.8 (C-7) and 56.3 ppm (C-5), two oxymethines at δ 67.4 (C-6) and 79.4 ppm (C-1), two quaternary carbons at δ 42.1 (C-10) and 146.6 ppm (C-4). These data suggested a bicyclic sesquiterpenoid structure with one exocyclic double bond, one isopropyl group and two hydroxyl substituents.

Careful analysis of these spectral data indicated that the compound possessed eudesmane-type sesquiterpenoid skeleton with two hydroxyl groups located at C-1 and C-6 and one double bond between at C-4 and C-15. The sesquiterpene FF-4 was thus identified as 4(15)-eudesmene-1 β , 6α -diol and confirmed by comparison of its NMR data (**Table 19**) and specific rotation ($[\alpha]_{p}^{20}$ +50°) with published values (Kitajima *et al.*, 2002).

4(15)-Eudesmene-1 β ,6 α -diol is a constituent of the aerial parts of several asteraceous plants including *Senecio microglossus* (Bohlmann *et al.*, 1983), *Helianthus microcephalus* (Gutierrez and Herz, 1988) and *Ageratina glechonophylla* (Gonzalez *et al.*, 1989). The terpenoid was also found in other flowering plant species such as in the leaves of *Teucrium polium* (family Labiatae) (Kamel, 1995) and in the fruits of *Torilis japonica* (family Umbelliferae) (Kitajima *et al.*, 2002). However, this is the first report of its occurrence in plants of the family Moraceae.



4(15)-Eudesmene-1 β ,6 α -diol

Table 19. ¹H- and ¹³C-NMR spectral data for compound FF-4 and 4(15)-eudesmene-1 β ,6 α -diol (300 and 75 MHz, in CDCl₃).

Desition		FF-4		4(15)-Eudesmene-1β,6α-diol*
Position	$\delta_{\rm c}$	$\delta_{_{\rm H}}$	$\delta_{\rm c}$	δ _н
1	79.4	3.39 (1H, <i>dd</i> , <i>J</i> = 10.2, 4.2 Hz)	79.0	3.42 (1H, <i>dd</i> , <i>J</i> = 11.5, 4.5 Hz)
2	32.3	1.50 (1H, <i>m</i>)	31.9	1.54 (1H, <i>m</i>)
		1.82 (1H, <i>m</i>)		1.86 (1H, <i>dddd</i> , <i>J</i> = 12.5, 5.0, 5.0, 4.5 Hz)
3	35.5	2.04 (1H, <i>m</i>)	35.1	2.07 (1H, <i>br ddd</i> , <i>J</i> = 13.5, 12.5, 5.0 Hz)
		2.30 (1H, <i>m</i>)		2.33 (1H, <i>ddd</i> , <i>J</i> = 13.5, 5.0, 3.5 Hz)
4	146.6	-	146.2	-
5	56.3	1.72 (1H, d, J = 8.4 Hz)	55.9	1.75 (1H, <i>d</i> , <i>J</i> = 10.0 Hz)
6	67.4	3.69 (1H, <i>br s</i>)	67.0	3.72 (1H, <i>dd</i> , <i>J</i> = 10.0, 10.0 Hz)
7	49.8	1.26 (1H, <i>m</i>)	49.3	1.30 (1H, <i>m</i>)
8	18.6	1.26 (1H, <i>m</i>)	18.1	1.23 (1H, <i>br ddd</i> , <i>J</i> = 12.0, 12.0, 3.0 Hz)
		1.50 (1H, <i>m</i>)		1.54 (1H, <i>m</i>)
9	36.7	1.18 (1H, <i>m</i>)	36.3	1.17 (1H, ddd, J = 13.0, 12.0, 3.0 Hz)
		1.91 (1H, <i>m</i>)		1.93 (1H, <i>br dd</i> , <i>J</i> = 13.0, 3.0 Hz)
10	42.1	-	41.7	-
11	26.4	2.22 (1H, <i>m</i>)	26.0	2.25 (1H, <i>m</i>)
12	16.6	0.85 (3H, <i>s</i>)	16.2	0.87 (3H, s)
13	21.5	0.93 (3H, <i>s</i>)	21.1	0.96 (3H, s)
14	12.0	0.68 (3H, s)	11.6	0.71 (3H, <i>s</i>)
15	108.2	4.72 (1H, <i>br s</i>)	107.8	4.75 (1H, <i>br d</i> , <i>J</i> = 1.0 Hz)
		5.00 (1H, <i>br s</i>)		5.02 (1H, br d, J = 1.0 Hz)

*Kitajima et al., 2002

12. Identification of Compound FF-5 [4(15)-Eudesmene-1β,5α-diol]

Compound FF-5, obtained as colorless needles (5.1 mg, 0.0003% yield), appeared as a purple spot on TLC upon spraying with 10% sulfuric acid and heated. Its quasi-molecular $[M + H]^+$ ion peak at m/z 239 in the ESI mass spectrum (**Figure 60**) corresponded to the molecular formula $C_{15}H_{26}O_2$ and suggested that the compound was an isomer of compound FF-4. Its IR absorption bands at 3627 and 1645 cm⁻¹ (**Figure 61**) were ascribable to hydroxyl and olefinic groups in the structure, respectively.

The NMR data of compounds FF-4 and FF-5 were also similar. The ¹H-NMR spectrum of compound FF-5 (Figure 62) displayed one methyl singlets at δ 0.74 ppm (H₃-14), two methyl doublets of an isopropyl group at δ 0.88 (J = 6.6 Hz, H₂-13) and 0.90 ppm (J = 6.6 Hz, H₂-12). A pair of methylene protons resonated as a doublet of doublets at δ 2.13 ppm (1H, J = 13.7, 5.5 Hz, H-3 β) and a doublet of doublet of doublets at δ 2.68 ppm (1H, J = 13.7, 8.1, 5.5 Hz, H-3 α). Two prominent exocyclic methylene singlets appeared at δ 4.73 and 4.83 ppm (each 1H, H₂-15) and one oxymethine doublet of doublets at δ 4.03 ppm (J = 10.4, 4.8 Hz, H-1). The ¹³C-NMR spectrum (Figure 63) showed 15 carbon signals of a sesquiterpene, including those of one tertiary methyl at δ 13.1 ppm (C-14), two secondary methyls of an isopropyl side-chain at δ 20.1 (C-12) and 20.4 ppm (C-13), five methylenes at δ 24.1 (C-8), 30.2 (C-3), 30.4 (C-9), 31.0 (C-2) and 34.7 ppm (C-6), two methines at δ 33.2 (C-11) and 38.7 ppm (C-7), one oxymethine at δ 73.5 ppm (C-1), a quaternary carbon at δ 42.7 ppm (C-10), an oxygenated quaternary carbon at δ 76.6 ppm (C-5) and two olefinic carbons of an exocyclic double bond at δ 109.0 (C-15) and 151.1 ppm (C-4). Analysis of these spectral data confirmed the compound to be the same eudesmane-type sesquiterpenoid as compound FF-4, with identical exocyclic double bond and a hydroxyl substitution at C1. However, the presence of one oxygenated quaternary carbon while there was one less oxymethine carbon when compared to the previous compound indicated the shift of another hydroxyl group from C-6 to C-5 position. Consequently, compound FF-5 was identified as 4(15)-eudesmene-1 β ,5 α -diol and confirmed by comparison of its ¹H and ¹³C NMR data with those previously reported (Sun et al., 2004) (Table 20). This is the first report of the presence of this compound in F. foveolata.

4(15)-Eudesmene-1 β ,5 α -diol, together with 4(15)-eudesmene-1 β ,6 α -diol, have been isolated from the fruits of *Torilis japonica* (family Umbelliferae) (Kitajima *et al.*, 2002). The sesquiterpene, obtained from *Caragana intermedia* (family Leguminosae), showed anti-diabetic activity at a concentration of 10.7 µg/ml by improving glucose transformation. In db/db mice, the compound produced the same effect on oral glucose tolerance as the anti-diabetic drug metformin, with the MIC value of 100 mg/ml (Sun *et al.*, 2004).



4(15)-Eudesmene-1 β ,5 α -diol

Desition		FF-5	4	(15)-Eudesmene-1β,5α-diol*
Position	δ_{c}	$\delta_{_{ m H}}$	$\delta_{\rm c}$	$\delta_{_{ m H}}$
1	73.5	4.03 (1H, <i>dd</i> , <i>J</i> = 10.4, 4.8 Hz)	73.1	4.05 (1H, <i>dd</i> , <i>J</i> = 11.6, 5.0 Hz)
2	31.0	1.52 (1H, <i>m</i>)	30.6	1.55 (1H, <i>m</i>)
		1.83 (1H, <i>m</i>)		1.85 (1H, <i>m</i>)
3	30.2	2.13 (1H, <i>dd</i> , <i>J</i> = 13.7, 5.5 Hz)	29.8	2.15 (1H, <i>ddd</i> , <i>J</i> = 13.8, 5.3, 1.8 Hz)
		2.68 (1H, <i>ddd</i> , <i>J</i> = 13.7, 8.1, 5.5 Hz)		2.70 (1H, <i>m</i>)
4	151.1	-	150.7	-
5	76.6	-	76.2	-
6	34.7	1.54 (1H, <i>m</i>)	34.3	1.54 (1H, <i>m</i>)
		1.57 (1H, <i>m</i>)		1.57 (1H, <i>dd</i> , <i>J</i> = 13.2, 4.1 Hz)
7	38.7	1.57 (1H, <i>m</i>)	38.3	1.58 (1H, <i>m</i>)
8	24.1	1.23 (1H, <i>m</i>)	23.7	1.25 (1H, <i>m</i>)
		1.59 (1H, <i>m</i>)		1.60 (1H, <i>m</i>)
9	30.4	1.68 (1H, <i>m</i>)	30.0	1.68 (1H, <i>m</i>)
		1.68 (1H, <i>m</i>)		1.68 (1H, <i>dd</i> , <i>J</i> = 13.2, 4.0 Hz)
10	42.7	-	42.3	-
11	33.2	1.48 (1H, <i>m</i>)	32.8	1.50 (1H, <i>m</i>)
12	20.1	0.90 (3H, <i>d</i> , <i>J</i> = 6.6 Hz)	19.7	0.92 (3H, <i>d</i> , <i>J</i> = 6.8 Hz)
13	20.4	0.88 (3H, <i>d</i> , <i>J</i> = 6.6 Hz)	20.0	0.90 (3H, <i>d</i> , <i>J</i> = 6.8 Hz)
14	13.1	0.74 (3H, <i>s</i>)	12.7	0.76 (3H, s)
15	109.0	4.73 (1H, s), 4.83 (1H, s)	108.6	4.74(1H, <i>s</i>), 4.85 (1H, <i>s</i>)

Table 20. ¹H- and ¹³C-NMR spectral data for compound FF-5 and 4(15)-eudesmene-1 β ,5 α -diol (300 and 75 MHz, in CDCl₃).

* Sun et al., 2004

13. Structure Elucidation of Compound FF-7 (Foveolide A)

Compound FF-7 was obtained as yellow oil (16.2 mg, 0.0008% yield). The molecular formula was determined as $C_{15}H_{22}O_3$ based on its quasi-molecular $[M + H]^+$ ion peak at m/z 251.1621 in the high resolution ESI-TOF mass spectrum (**Figure 65**). IR spectrum (**Figure 66**) showed absorption bands representing hydroxyl group at 3430 cm⁻¹, ester-carbonyl at 1749 cm⁻¹ and gem-dimethyl at 1390 and 1371 cm⁻¹. Careful examination of its ¹H- and ¹³C-NMR spectra (**Table 21** and **Figures 67-68**), with the aid of DEPT and HSQC experiments, showed the presence of one tertiary methyl singlet at δ 0.80 ppm (H₃-14), two isopropyl methyl doublets at δ 0.94 (J = 6.5 Hz, H₃-12) and 0.98 ppm (J = 7.0 Hz, H₃-13), three methylenes, three methines, two oxymethine doublets of doublets at δ 3.72 (J = 9.5, 6.7 Hz, H-1) and 3.91 ppm (J = 11.0, 10.0 Hz, H-6), one olefinic methine doublet of doublets at δ 6.54 ppm (J = 6.8, 3.2 Hz, H-3), one quaternary carbon at δ 36.8 ppm (C-10), one olefinic quaternary carbon at δ 128.1 ppm (C-4) and one ester carbonyl carbon at δ 169.0 ppm (C-15). The structure of compound FF-7 was therefore deduced to be a eudesmane sesquiterpenoid having similar partial structure as compound FF-4.

HMBC correlations (**Table 21** and **Figures 69a-69e**) observed between the signal of H-1 and carbon signals at δ 9.8 (C-14), 35.1 (C-9) and 54.2 ppm (C-5), as well as between H-6 resonance and carbon signals at δ 24.8 (C-8), 30.6 (C-11) and 36.8 ppm (C-10), confirmed that these oxymethine protons were located at the same positions as in compound FF-4. However, in this sesquiterpene, the exomethylene C-15 of compound FF-4 has been replaced by an estercarbonyl which formed a γ -lactone ring with C-6, whereas the double bond has shifted to between C-3 and C-4. These locations were established through the HMBC correlations observed between H-3 resonance and carbon signals at δ 54.2 (C-5), 75.3 (C-1) and 169 ppm (C-15), as well as between the signals of H-6 and C-4 (δ 128.1 ppm). NOE correlations, observed in the NOESY spectrum (**Figures 70a-70d**), between the signal of the α-oriented H-5 (δ 2.26 ppm) and both H-1 and H-7 signals (δ 1.69 ppm) indicated the β-orientation of the hydroxyl group at C-1. Furthermore, the presence of NOE correlations between H-6 and all three methyl groups demonstrated that the γ -lactone connection at this position was α-oriented. Thus, the structure of compound FF-7 was elucidated as 1β-hydroxyeudesm-3-en-15,6α-olide and was trivially named foveolide A.



Foveolide A

Table 21. ¹H- and ¹³C-NMR spectral data for foveolide A (500 and 125 MHz, in CDCl₃).

Position	$\delta_{_{\rm H}}$	$\delta_{\rm C}$	HMBC (H→C)
1	3.72 (1H, <i>dd</i> , <i>J</i> = 9.5, 6.7 Hz)	75.3	5, 9, 10, 14
2	α : 2.60 (1H, <i>dddd</i> , <i>J</i> = 19.3, 6.7, 4.0, 3.2 Hz)	34.0	1, 3, 4, 10
	β : 2.19 (1H, <i>dddd</i> , J = 19.3, 9.5, 5.0, 3.2 Hz)		1, 3, 4, 10
3	6.54 (1H, <i>dd</i> , <i>J</i> = 6.8, 3.2 Hz)	131.2	1, 2, 4, 5, 15
4	-	128.1	
5	2.26 (1H, <i>ddd</i> , <i>J</i> = 11.0, 4.8, 3.3 Hz)	54.2	1, 3, 4, 6, 7, 9, 10, 14, 15
6	3.91 (1H, dd, J = 11.0, 10.0 Hz)	79.6	4, 5, 8, 10, 11
7	1.69 (1H, <i>m</i>)	50.3	5, 6, 8, 11, 12, 13
8	α : 1.85 (1H, <i>dtd</i> , <i>J</i> = 14.5, 4.5, 2.0 Hz)	24.8	6, 7, 9, 11
	β: 1.42 (1H, <i>m</i>)		6, 7, 9, 11
9	α : 1.27 (1H, td, J = 13.3, 4.8 Hz)	35.1	1, 5, 7, 8, 10, 14
	β : 1.81 (1H, <i>ddd</i> , <i>J</i> = 13.3, 4.8, 2.0 Hz)		5, 7, 8, 10, 14
10	-	36.8	
11	1.72 (1H, <i>m</i>)	30.6	6, 7, 8, 12, 13
12*	0.94 (3H, <i>d</i> , <i>J</i> = 6.5 Hz)	19.8	7, 11, 13
13*	0.98 (3H, d, J = 7.0 Hz)	20.0	7, 11, 12
14	0.80 (3H, s)	9.8	1, 5, 9, 10
15	-	169.0	

* These assignments are interchangeable.

14. Structure Elucidation of Compound FF-6 (Foveolide B)

Compound FF-6, a colorless gum (7.9 mg, 0.0004 % yield), gave a purple spot on TLC upon spraying with 10% sulfuric acid and heated. Its molecular formula was deduced from the pseudomolecular $[M + H]^+$ ion peak at m/z 469.3304 in its high resolution mass spectrum (Figure 71) as $C_{30}H_{44}O_4$, representing nine degrees of unsaturation. The IR spectrum (Figure 72) displayed absorption bands at 3443, 1753 and 1699 cm⁻¹ ascribable to hydroxyl, ester carbonyl and keto-carbonyl functional groups in the structure.

Analysis of the ¹H- and ¹³C-NMR spectra of compound FF-6 (**Table 22** and **Figures 73-74**), with the aid of DEPT and HSQC experiments, revealed the presence of two tertiary methyl singlets at δ 1.09 (H₃-14) and 1.20 ppm (H₃-14'), two pairs of isopropyl methyl doublets at δ 0.86 (*J* = 7.0 Hz, H₃-12') and 0.88 ppm (*J* = 6.5 Hz, H₃-13') along with δ 0.97 (*J* = 6.5 Hz, H₃-13) and 1.00 ppm (*J* = 7.0 Hz, H₃-12), eight methylenes, seven methines, two oxymethine triplets at δ 3.83 (*J* = 8.0 Hz, H-1) and 4.25 ppm (*J* = 10.8 Hz, H-6), three quaternary carbons at δ 42.0 (C-10), 46.5 (C-4) and 48.8 ppm (C-10'), two olefinic carbons of a tetra-substituted double bond at δ 126.5 (C-4') and 134.2 ppm (C-5'), one ester carbonyl carbon at δ 178.2 ppm (C-15) and one keto-carbonyl carbon at δ 215.5 ppm (C-1'). The presence of six rings in the structure of compound FF-6 could be inferred from these data. Furthermore, partial similarity of the NMR data to those of compound FF-4 and the number of carbon atoms indicated that compound FF-6 is a dimer of two different eudesmane sesquiterpenoid units.

As shown in **Scheme 7**, oxidation and dehydration of compound FF-4 could produce a hypothetical monomer, eudesm-4(15),5-dien-1-one, which could subsequently form two C-C bonds by Diels-Alder reaction (Oikawa and Tokiwano, 2004) with foveolide A (compound FF-7), linking them directly through positions 3/15' and 4/6' and thus creating compound FF-6 as a true disesquiterpenoid (Zhan *et al.*, 2011). These connectivities were confirmed by HMBC cross-peaks (**Figures 75a-75e**) from signals at δ 1.59 and 2.64 ppm (H₂-15') to those of δ 34.7 (C-2), 30.5 (C-3), 46.5 (C-4), 30.5 (C-3'), 126.5 (C-4') and 134.2 ppm (C-5'), from both resonances at δ 2.38 (H-3) and 2.07 (H-5) to the signal of C-6' at δ 43.2 ppm, and from H-7' signal at δ 1.48 ppm to that of C-4 at δ 46.5 ppm. Furthermore, HMBC correlations between signals at δ 2.35 and 2.69 (H₂-2'), 2.23 and 2.28 (H₂-3'), 1.49 and 1.50 (H₂-9'), 1.20 ppm (H₃-14') and the keto-carbonyl signal at δ 215.5 ppm established the position of keto-carbonyl as at C-1'.



Scheme 7. Postulated biogenesis of compound FF-6

NOE correlations between H-6 and H₃-14, as well as between H-1, H-5 and H-7, in the NOESY spectrum (**Figures 76a-76b**) indicated the relative configuration of these protons to be identical to those of compound FF-4. The presence of NOESY cross-peaks between H-1 and H-6' and between H-6' and H₃-12', H₃-13' and H₃-14' established the orientation of H-6', H-7' and H₃-14' to be α , β and α , respectively. Accordingly, H-3 was assigned as β -oriented based on NOE correlations between its proton signal at δ 2.38 ppm and those of H-6 and H₃-14. The orientation of H-3 and configuration at C-4 indicated that the compound is an intermolecular [4+2] *endo*-Diels-Alder adduct. Therefore, compound FF-6 was characterized as a new sesquiterpenoid dimer, trivially named foveolide B, and is the first compound of this type found in *Ficus* species.



Foveolide B

Table 22. ¹H- and ¹³C-NMR spectral data for foveolide B (500 and 125 MHz, in CDCl₃).

Position	$\delta_{\rm H}$	δ_{c}	HMBC (H→C)
1	3.83 (1H, t, J = 8.0 Hz)	74.2	2, 5, 9, 10, 14
2	α : 1.96 (1H, <i>ddd</i> , <i>J</i> = 14.0, 10.5, 8.0 Hz)	34.7	3, 4, 10, 15'
	β : 1.41 (1H, dd, J = 14.0, 8.0 Hz)		3, 4, 15'
3	2.38 (1H, <i>m</i>)	30.5	2, 4, 4', 6', 15, 15'
4	-	46.5	
5	2.07 (1H, $d, J = 10.8$ Hz)	57.1	1, 3, 4, 6, 6', 7, 9, 10,14
6	4.25 (1H, $t, J = 10.8$ Hz)	77.4	5, 10, 11
7	1.75 (1H, <i>m</i>)	52.0	5, 6, 8, 11, 12, 13
8	α : 1.83 (1H, <i>dtd</i> , J = 14.5, 5.0, 1.5 Hz)	25.6	6, 7, 9, 10, 11
	β: 1.56 (1H, <i>m</i>)		9
9	α : 1.30 (1H, <i>td</i> , J = 11.8, 5.0 Hz)	39.9	1, 5, 8, 10, 14
	β : 1.70 (1H, dd, J = 13.8, 5.0 Hz)		5, 7, 8, 10, 14
10	-	42.0	
11	1.75 (1H, <i>m</i>)	30.9	6, 7, 8, 12, 13
12*	1.00 (3H, d, J = 7.0 Hz)	19.6	7, 11, 13
13*	0.97 (3H, d, J = 6.5 Hz)	20.2	7, 11, 12
14	1.09 (3H, <i>s</i>)	15.1	1, 5, 9, 10
15	-	178.2	

Position	$\delta_{_{ m H}}$	δ _c	HMBC (H→C)
1'	-	215.5	
2'	α : 2.35 (1H, ddd, J = 13.9, 5.0, 3.0 Hz)	35.9	1', 3', 4', 10'
	β : 2.69 (1H, <i>ddd</i> , <i>J</i> = 13.9, 12.5, 7.0 Hz)		1', 3', 4'
3'	α: 2.23 (1H, <i>m</i>)	30.5	1', 2', 4', 5', 15'
	β: 2.28 (1H, <i>m</i>)		1', 2', 4', 5', 15'
4′	-	126.5	
5'	-	134.2	
6'	2.17 (1H, <i>d</i> , <i>J</i> = 12.0 Hz)	43.2	4', 5', 15
7'	1.48 (1H, <i>m</i>)	45.7	4, 6', 9', 12'
8'	α: 1.42 (1H, <i>m</i>)	20.5	6', 9', 10'
	β: 1.55 (1H, <i>m</i>)		
9'	α: 1.50 (1H, <i>m</i>)	35.1	1', 5', 7', 8', 10'
	β: 1.49 (1H, <i>m</i>)		1', 5', 7', 8', 10'
10'	-	48.8	
11'	2.18 (1H, <i>m</i>)	26.1	6', 7', 8', 12', 13'
12'**	0.86 (3H, <i>d</i> , <i>J</i> = 7.0 Hz)	21.6	7', 11', 13'
13'**	0.88 (3H, d, J = 6.5 Hz)	16.4	7', 11', 12'
14'	1.20 (3H, <i>s</i>)	20.5	1', 5', 9', 10'
15'	α : 1.59 (1H, $d, J = 17.0$ Hz)	30.7	2, 3, 3', 4, 4', 5'
	β : 2.64 (1H, dd, J = 17.0, 3.5 Hz)		2, 3, 3', 4', 5'

*, ** These assignments are interchangeable.

15. Structure Elucidation of Compound FF-8a [Eudesm-4(15),5-dien-1β,3β-diol]

Compound FF-8a was obtained as colorless oil (14.7 mg, 0.0007% yield). Its NMR data (**Figures 77-78, Table 23**) displayed signals of one tertiary methyl at δ 0.89 ppm (3H, *s*, H₃-14; $\delta_{\rm C}$ 17.4), two isopropyl methyls at δ 0.87 (3H, *d*, *J* = 7.0 Hz, H₃-12; $\delta_{\rm C}$ 19.0) and 0.89 ppm (3H, *d*, *J* = 7.0 Hz, H₃-13; $\delta_{\rm C}$ 19.5), one exocyclic methylene at δ 4.96 ppm (2H, *dt*, *J* = 19.0, 2.0 Hz, H₂-15; $\delta_{\rm C}$ 106.6), one olefinic methine at δ 5.61 ppm (1H, *t*, *J* = 1.5 Hz, H-6; $\delta_{\rm C}$ 129.3), two oxygenated methine at δ 3.46 (1H, *dd*, *J* = 12.0, 4.0 Hz, H-1; $\delta_{\rm C}$ 76.1) and 4.10 ppm (1H, *ddt*, *J* = 12.0, 5.3, 2.0 Hz, H-3; $\delta_{\rm C}$ 69.6), three methylene carbons at δ 20.8 (C-8), 35.0 (C-9) and 39.7 ppm (C-2), two methine carbons at δ 32.0 (C-11) and 42.4 ppm (C-7), one quaternary carbon at δ

40.3 ppm (C-10) and two olefinic quaternary carbons at δ 142.1 (C-5) and 151.3 ppm (C-4). These data suggested a eudesmane sesquiterpenoid structure similar to compound FF-4; the differences being the location of the second hydroxyl group at C-3 instead of C-6 and the presence of another double bond between C-5 and C-6. NOE correlations (**Figure 79**) between H₃-14 and H-2 β (δ 1.63 ppm), as well as between H-1 and H-3 with H-2 α (δ 2.19 ppm) indicated that both hydroxyl groups at C-1 and C-3 were β -oriented. Therefore, the structure of this compound was determined to be a new eudesmane sesquiterpenoid, eudesm-4(15),5-dien-1 β ,3 β -diol through analysis of its spectral data and comparison with those of known compounds having similar substructures. However, this sesquiterpene was unstable and could be converted through oxidation-reduction interactions spontaneously and completely into another sesquiterpenoid (compound FF-8b), as shown in **Scheme 8**.



FF-8a [eudesm-4(15),5-dien-1 β ,3 β -diol]

FF-8b (foveoeudesmenone)

Scheme 8. Postulated conversion of eudesm-4(15),5-dien-1 β ,3 β -diol into foveoeudesmenone

Table 23. ¹H- and ¹³C-NMR spectral data for eudesm-4(15),5-dien-1 β ,3 β -diol (300 and 75 MHz, in CDCl₃).

Position	$\delta_{_{ m H}}$	δ_{c}	НМВС (Н→С)
1	3.46 (1H, <i>dd</i> , <i>J</i> = 12.0, 4.0 Hz)	76.1	2, 3, 9, 14
	α : 2.19 (1H, <i>ddd</i> , <i>J</i> = 12.0, 5.3, 4.0 Hz)	39.7	1, 3, 4, 10
2	β: 1.63 (1H, q , J = 12.0 Hz)		1, 3, 4, 10
3	4.10 (1H, <i>ddt</i> , <i>J</i> = 12.0, 5.3, 2.0 Hz)	69.6	2, 4, 15
4	-	151.3	
5	-	142.1	

Position	$\delta_{_{ m H}}$	$\delta_{\rm C}$	HMBC (H→C)
6	5.61 (1H, <i>br s</i>)	129.3	4, 5, 7, 8, 10, 11
7	1.99 (1H, <i>m</i>)	42.4	5, 6, 8, 11
8	1.31 (1H, <i>m</i>)	20.8	6, 7, 9, 10, 11
	1.59 (1H, <i>m</i>)		6, 7, 9, 10, 11
9	1.31 (1H, <i>m</i>)	35.0	5, 7, 8, 10, 14
	1.87 (1H, <i>m</i>)		5, 7, 8, 10, 14
10	-	40.3	
11	1.62 (1H, <i>m</i>)	32.0	6, 7, 8, 12, 13
12*	0.87 (3H, <i>d</i> , <i>J</i> = 7.0 Hz)	19.0	7, 11, 13
13*	0.89 (3H, <i>d</i> , <i>J</i> = 7.0 Hz)	19.5	7, 11, 12
14	0.89 (3H, s)	17.4	1, 5, 9, 10
15	4.93 (1H, <i>br s</i>)	106.6	3, 4, 5
	4.97 (1H, <i>br s</i>)		3, 4, 5

* These assignments are interchangeable.

16. Structure Elucidation of Compound FF-8b (Foveoeudesmenone)

The molecular formula of compound FF-8b, as the product of the conversion, was determined as $C_{15}H_{24}O_2$ from its $[M + H]^+$ ion peak at *m/z* 237.1842 in the high resolution mass spectrum (**Figure 81**). The IR spectrum (**Figure 82**) exhibited absorption bands of hydroxyl group at 3423 cm⁻¹, α , β -unsaturated keto-carbonyl at 1709 and 1655 cm⁻¹, conjugated double bond at 1604 cm⁻¹ and *gem*-dimethyl at 1386 and 1370 cm⁻¹. Its NMR spectra, obtained while the conversion of the natural sesquiterpene into the final product was still in progress, showed mixed signals of both the precursor (minor component) and product (major component). Compound FF-8b, as the final conversion product, displayed the ¹H- and ¹³C-NMR signals (**Figures 83a-83b, 84, Table 24**) of one tertiary methyl singlet at δ 1.12 ppm (H₃-14), one olefinic methyl doublet at δ 1.74 ppm (*J* = 1.5 Hz, H₃-15), two isopropyl methyl doublets at δ 0.91 (*J* = 7.0 Hz, H₃-12) and 0.92 ppm (*J* = 7.0 Hz, H₃-13), four methylenes, two methines, one oxymethine double doublet at δ 3.77 ppm (*J* = 12.8, 5.5 Hz, H-1), one quaternary carbon at δ 41.5 ppm (C-5) and one keto-carbonyl carbon at δ 197.4 ppm (C-3). These data indicated that compound FF-8b was a

eudesmane sesquiterpenoid with C-1 hydroxyl substitution and a keto-carbonyl function at C-3 conjugated to a C-4/C-5 double bond. The evidence of HSQC and HMBC correlations (**Figures 85a-85c**) between the resonance of H₃-15 and those of C-3, C-4 and C-5, as well as between H-1 signal and C-3 and between H₃-14 signal and C-5, supported these assignments. NOE correlations (**Figures 86a-86b**) observed between the signals of H-1 and H-9α (δ 1.30 ppm), as well as between those of H₃-14 and H-9β (δ 2.10 ppm), confirmed the β-orientation of the hydroxyl group at C-1. Therefore, the chemical structure of compound FF-8b was elucidated as a new eudesmane-type sesquiterpene, 1β-hydroxyeudesm-4-en-3-one, and the compound was trivially named foveoeudesmenone. This proposed structure is slightly different from that of 1β-hydroxy-α-cyperone, a constituent of *Artemisia caerulescens* subsp. *gargantae* (family Asteraceae) (Sanz and Marco, 1990), by the replacement of isopropenyl side-chain with an isopropyl group for compound FF-8b.



Foveoeudesmenone

Table 24. ¹H- and ¹³C-NMR spectral data for foveoeudesmenone (500 and 125 MHz, in CDCl₃).

Position	$\delta_{_{ m H}}$	$\delta_{\rm C}$	HMBC (H \rightarrow C)
1	3.77 (1H, <i>dd</i> , <i>J</i> = 12.8, 5.5 Hz)	74.6	2, 3, 9, 10, 14
	α : 2.61 (1H, dd, J = 16.5, 5.5 Hz)	42.3	1, 3, 4, 10
2	β : 2.53 (1H, dd, J = 16.5, 12.8 Hz)		1, 3, 10
3	-	197.4	
4	-	129.2	
5	-	162.8	

Position	$\delta_{_{ m H}}$	$\delta_{\rm C}$	HMBC (H \rightarrow C)
6	α: 1.86 (1H, <i>m</i>)	31.4	4, 5, 7, 8, 11
	β : 2.72 (1H, <i>ddd</i> , $J = 15.0, 5.5, 1.5$ Hz)		4, 5, 7, 8, 10
7	1.23 (1H, <i>m</i>)	44.2	9
8	1.68 (2H, <i>m</i>)	24.1	9
9	α: 1.30 (1H, <i>m</i>)	37.8	7, 8, 10, 14
	β : 2.10 (1H, dt , J = 13.0, 3.0 Hz)		7, 8, 10, 14
10	-	41.5	
11	1.58 (1H, <i>m</i>)	32.8	6, 7, 8, 12, 13
12*	0.91 (3H, <i>d</i> , <i>J</i> = 7.0 Hz)	19.3	7, 11, 13
13*	0.92 (3H, <i>d</i> , <i>J</i> = 7.0 Hz)	19.7	7, 11, 12
14	1.12 (3H, s)	16.2	1, 5, 9, 10
15	1.74 (3H, <i>d</i> , <i>J</i> = 1.5 Hz)	10.9	3, 4, 5

* These assignments are interchangeable.

17. Identification of Compound FF-9 (Ethyl rosmarinate)

Compound FF-9 was obtained as yellow gum (15.7 mg, 0.0008% yield). Its molecular formula was determined as $C_{20}H_{20}O_8$ from its molecular ion peak at m/z 388 in the mass spectrum (**Figure 88**). IR spectrum (**Figure 89**) showed hydroxyl absorption band at 3391 cm⁻¹, estercarbonyl at 1724 cm⁻¹, aliphatic alkene at 1653 cm⁻¹ and aromatic ring at 1445 and 1601 cm⁻¹.

Major signals in the ¹H- and ¹³C-NMR spectra of compound FF-9 (**Figures 90-91**) include those of an ethyl group [one terminal methyl triplet at δ 1.19 ppm (3H, J = 7.0 Hz) and a oxymethylene quartet at δ 4.11 ppm (2H, J = 7.0 Hz)], a *trans*-double bond [two olefinic doublets at δ 6.55 and 7.40 ppm (1H each, J = 15.6 Hz, H-8' and H-7')], two 3,4-dihydroxybenzene moieties [two aromatic double doublets at δ 6.54 (J = 7.9, 2.1 Hz, H-6) and 6.93 ppm (J = 8.1, 1.8 Hz, H-6') along with four aromatic doublets at δ 6.71 (J = 2.1 Hz, H-2), 6.71 (J = 7.9 Hz, H-5), 6.82 (J = 8.1 Hz, H-5') and 7.06 ppm (J = 1.8 Hz, H-2')], one oxymethine doublet of doublets at δ 4.75 ppm (J = 7.6, 6.1 Hz, H-8), one methylene carbon at δ 37.8 ppm (C-7) and two ester-carbonyl carbons at δ 166.4 (C-9') and 172.4 ppm (C-9). Based on these NMR spectral data, it could be concluded that compound FF-9 was an ethyl ester of two phenylpropanoid units: one

unit with the *trans*-double bond was caffeic acid, while another unit was 3,4-dihydroxyphenyl lactic acid. One natural example of esters formed from these two units is rosmarinic acid. Comparison of these data with literature subsequently revealed compound FF-9 to be the phenolic compound ethyl rosmarinate (Woo and Piao, 2004) (**Table 25**).

Ethyl rosmarinate has been found in plants belonging to Family Labiatae i.e. *Prunella vulgaris* (Wang *et al.*, 2000), *Nepeta prattii* (Hou, Tu, and Li, 2002) and *Lycopus lucidus* (Woo and Piao, 2004). The compound exhibited potent antioxidative activity in the nitroblue tetrazolium (NBT) superoxide scavenging assay with an IC₅₀ value of 0.78 μ g/ml (Woo and Piao, 2004). Although previous phytochemical study of *Ficus foveolata* has reported the presence of some phenolic compounds i.e. alkyl ferulate and alkyl diferulate in its stems (Sermboonpaisarn and Sawasdee, 2012), this is the first report of ethyl rosmarinate in the medicinal plant.



Ethyl rosmarinate

Desition		FF-9		Ethyl rosmarinate*
Position	δ_{c}	$\delta_{_{ m H}}$	δ _c	$\delta_{_{ m H}}$
1	129.3	-	128.9	-
2	117.1	6.71 (1H, $d, J = 2.1$ Hz)	117.8	6.72 (1H, d, J = 2.0 Hz)
3	145.7	-	146.4	-
4	144.8	-	145.6	-
5	115.9	6.71 (1H, <i>d</i> , <i>J</i> = 7.9 Hz)	116.4	6.70 (1H, <i>d</i> , <i>J</i> = 8.0 Hz)
6	121.5	6.54 (1H, <i>dd</i> , <i>J</i> = 7.9, 2.1 Hz)	122.0	6.58 (1H, dd, J = 8.0, 2.0 Hz)
7	37.8	2.92 (1H, <i>dd</i> , <i>J</i> = 13.9, 7.0 Hz)	38.1	3.03 (1H, <i>dd</i> , <i>J</i> = 14.0, 6.6 Hz)
		2.99 (1H, overlapped)		3.31 (1H, <i>dd</i> , <i>J</i> = 14.0, 4.8 Hz)
8	54.9	4.75 (1H, <i>dd</i> , <i>J</i> = 7.6, 6.1 Hz)	55.4	5.05 (1H, dd, J = 7.5, 6.0 Hz)
9	172.4	-	171.9	-
1'	128.3	-	127.7	-
2'	114.9	7.06 (1H, d , J = 1.8 Hz)	115.4	7.05 (1H, d, J = 2.0 Hz)
3'	146.2	-	147.0	-
4'	147.9	-	150.1	-
5'	116.3	6.82 (1H, d, J = 8.1 Hz)	116.7	6.78 (1H, <i>d</i> , <i>J</i> = 8.0 Hz)
6'	121.7	6.93 (1H, <i>dd</i> , <i>J</i> = 8.1, 1.8 Hz)	123.4	6.96 (1H, dd, J = 8.0, 2.0 Hz)
7'	141.4	7.40 (1H, $d, J = 15.6$ Hz)	143.1	7.56 (1H, d, J = 16.0 Hz)
8'	118.9	6.55 (1H, d, J = 15.6 Hz)	119.3	6.27 (1H, d, J = 16.0 Hz)
9'	166.4	-	168.5	-
O <u>CH</u> ₂ CH ₃	61.4	4.11 (2H, q, J = 7.0 Hz)	62.6	4.15 (2H, q, J = 7.5 Hz)
OCH ₂ <u>CH</u> ₃	14.4	1.19 (3H, t, J = 7.0 Hz)	14.5	1.21 (3H, t, J = 7.5 Hz)

Table 25. ¹H- (500 MHz) and ¹³C-NMR (125 MHz) spectral data for compound FF-9 (in acetone- d_6) and ethyl rosmarinate (in CD₃OD).

*Woo and Piao, 2004

18. Structure Elucidation of Compound FF-10 (Foveospirolide)

The molecular formula of compound FF-10, which was obtained as a yellow oil (12.5 mg, 0.0006% yield), was established as $C_{15}H_{18}O_8$ from its quasi-molecular $[M + H]^+$ ion peak at m/z 327.1072 in its HR-ESI-TOF mass spectrum (**Figure 93**). Its IR absorption bands (**Figure 94**) suggested the presence of a hydroxyl (v_{max} 3366 cm⁻¹), an ester carbonyl of a γ -butyrolactone ring (1778 cm⁻¹) and an aromatic ring (1435-1605 cm⁻¹) in the structure. The aromatic region of its

¹H-NMR spectrum (**Table 26** and **Figure 95**) displayed two doublets at δ 6.75 (1H, J = 8.1 Hz, H-5') and 6.96 ppm (1H, J = 1.8 Hz, H-2') and a double doublet at δ 6.79 ppm (1H, J = 8.1, 1.8 Hz, H-6'), indicating the presence of a 1,3,4-trisubstituted benzene ring in the molecule. In addition, the proton spectrum also exhibited a methoxyl singlet at δ 3.82 ppm (3'-OCH₃) and several one-proton resonances i.e. at δ 3.33 (d, J = 9.3 Hz, H-10), 3.33 (ddd, J = 10.9, 9.3, 5.7 Hz, H-8), 3.49 (t, J = 10.9 Hz, H-7 axial), 3.58 (t, J = 9.3 Hz, H-9) and 3.66 ppm (dd, J = 10.9, 5.7 Hz, H-7 equatorial) resembling the NMR pattern of sugar moiety. Its ¹³C-NMR spectrum (**Table 26** and **Figure 96**) exhibited fifteen signals, including those of one methoxyl carbon, one benzene ring, one carbonyl carbon and five oxygen-bearing sp³ carbons.

Analysis of COSY, HSQC and HMBC experiments showed the presence of a γ butyrolactone group [an ester carbonyl at δ 175.0 ppm (C-2); a pair of methylene doublets of doublets at δ 3.03 (1H, J = 17.1, 12.5 Hz, H-3 α) and 2.71 ppm (1H, J = 17.1, 8.4 Hz, H-3 β)/ δ_c 33.9; a methine double doublet at δ 4.03 (1H, J = 12.5, 8.4 Hz, H-4)/ δ_c 45.2; an oxygen-bearing quaternary carbon at δ 109.2 (C-5)]. This lactone ring was connected to a trihydroxytetrahydropyran ring through lactol ether linkage, forming 8,9,10-trihydroxy-1,6dioxaspiro[4,5]decan-2-one skeleton. Attachment of the benzene ring to C-4 position of the γ lactone ring was supported by HMBC correlations (**Table 26** and **Figures 97a-97c**) observed from H-4 to the carbons at δ 113.6 (C-2'), 122.9 (C-6') and 126.8 ppm (C-1'), and also from H-3 α to C-1'. The 3'-methoxyl and 4'-hydroxyl substitutions on the aromatic ring were assigned from the HMBC cross-peaks of carbon at δ 147.8 (C-3') with H-2', H-5' and methoxyl protons, as well as HMBC correlations from carbon at δ 146.7 ppm (C-4') to H-2', H-5' and H-6'. The coupling constants of proton signals on the trihydroxy-tetrahydropyran ring indicated that all three hydroxyl groups were equatorially oriented in a chair conformation.

The NOESY correlation (**Figures 98a-98b**) between axial H-7 and H-9 showed that these two protons have similar orientation on the tetrahydropyran ring, while NOE correlations between both H-2' and H-6' with H-3 α and with H-10 demonstrated that the aromatic ring and H-3 α were located on the same side of the γ -butyrolactone ring. These data showed that compound FF-10 is a derivative of the sawaranospirolides, which were isolated from the heartwood of *Chamaecyparis pisifera* (family Cupressaceae) and were proposed to be biogenetically derived from L-ascorbic acid (Hasegawa, Koyanagi, and Hirose, 1990). The compound could have been biosynthesized via a chemical reaction between ferulic acid (a phenylpropanoid) and ascorbic acid, with an enediol as the intermediate, followed by subsequent formation of the lactone and tetrahydropyran rings, as shown in **Scheme 9**. Relative configuration of both C-4 and the spiro C-5 of compound FF-10 was determined as *S* based on observed NOEs and comparison of its NMR data with the sawaranospirolides. In addition, the configuration of its C-8, C-9 and C-10 (originally C-3, C-4 and C-5 of L-ascorbic acid) should therefore be *S*, *R* and *S*, respectively. Compound FF-10 thus possesses the same 4*S*, 5*S*, 8*S*, 9*R*, 10*S*-configuration as that of sawaranospirolide A, but with one additional methoxyl group at C-3'. Therefore, the structure of compound FF-10 was elucidated as (4*S*, 5*S*, 8*S*, 9*R*, 10*S*)-8,9,10-trihydroxy-4-[4'-hydroxy-3'-methoxyphenyl]-1,6-dioxaspiro[4,5]decan-2-one, and named foveospirolide.



Scheme 9. Postulated biogenesis of compound FF-10



Foveospirolide

Table 26. ¹H- and ¹³C-NMR spectral data for foveospirolide (500 and 125 MHz, in CD₃OD).

Position	$\delta_{\rm H}$	δ_{c}	HMBC (H→C)
2	-	175.0	
3	α : 3.03 (1H, dd, J = 17.1, 12.5 Hz)	33.9	2, 4, 1'
	β : 2.71 (1H, dd, J = 17.1, 8.4 Hz)		2, 4, 5
4	4.03 (1H, <i>dd</i> , <i>J</i> = 12.5, 8.4 Hz)	45.2	3, 5, 10, 1', 2', 6'
5	-	109.2	
7	axial: 3.49 (1H, <i>t</i> , <i>J</i> = 10.9 Hz)	65.0	5, 8, 9
	equatorial: 3.66 (1H, dd, J = 10.9, 5.7 Hz)		5, 8, 9
8	3.33 (1H, <i>ddd</i> , <i>J</i> = 10.9, 9.3, 5.7 Hz)	70.2	7,9
9	3.58 (1H, t, J = 9.3 Hz)	75.9	10
10	3.33 (1H, <i>d</i> , <i>J</i> = 9.3 Hz)	71.1	4,9
1'	-	126.8	
2'	6.96 (1H, <i>d</i> , <i>J</i> = 1.8 Hz)	113.6	1', 3', 4, 4', 6'
3'	-	147.8	
4'	-	146.7	
5'	6.75 (1H, <i>d</i> , <i>J</i> = 8.1 Hz)	115.1	1', 3', 4'
6'	6.79 (1H, dd, J = 8.1, 1.8 Hz)	122.9	1', 2', 4, 4'
3'-OCH ₃	3.82 (3H, <i>s</i>)	56.3	3'

19. Cytotoxicity and Anti-tuberculosis Activity of Compounds Isolated from Salacia verrucosa and Ficus foveolata

The new friedelane-type triterpene, 21 α -hydroxyfriedelane-1,3-dione, along with four known friedelane-type compounds from *S. verrucosa* were investigated for their cytotoxic activities against human colon (SW620), lung (CHAGO), liver (HepG2), breast (BT474) and gastric (KATO-III) cancer cell lines using the MTT method. Doxorubicin was used as the positive control and any tested compound showing the IC₅₀ value of less than 10 μ M was considered to be strongly active. The results (**Table 27**) showed that 26-hydroxyfriedelane-1,3-dione possessed moderate cytotoxicity against all five cancer cell lines with IC₅₀ values in the range of 10.16-17.74 μ M, whereas 21 α -hydroxyfriedelane-1,3-dione and 30-hydroxyfriedelane-1,3-dione were inactive against all cell lines tested. The less polar friedelane-1,3-dione was strongly active specifically against the colon cancer cell line with an IC₅₀ value of 2.02 μ M. The presence of an additional hydroxyl group at C-26 of a 1,3-diketofriedelane skeleton appears to increase the cytotoxicity of these triterpenoids. 21 α -Hydroxyfriedelan-3-one was also moderately cytotoxic to the colon, liver and gastric cancer cell lines with IC₅₀ values of 11.52, 11.27 and 12.81 μ M, respectively, suggesting that the α -oriented hydroxyl substitution at C-21 might be essential to the cytotoxic activity of the 3-ketofriedelane triterpenoids.

These friedelane-type triterpenoids from *S. verrucosa* were also tested against the tuberculosis-causing bacteria, *Mycobacterium tuberculosis*. Four drugs commonly employed in the treatment of tuberculosis, including streptomycin, isoniazid, ofloxacin and ethambutol, were used as the positive controls. None of the assayed plant constituents was active against the pathogen at the maximal concentration of samples tested (50 μ g/ml) (**Table 27**).

		Cytotoxicity (IC_{50})					
Compound	SW620	CHAGO	HepG2	BT474	KATO-III	(MIC)	
Friedelane-1,3-dione	2.02	Inactive	Inactive	Inactive	Inactive	Inactive	
26-Hydroxyfriedelane	11.65	17.74	10.16	13.42	13.95	Inactive	
-1,3-dione							
21α-Hydroxyfriedelan	11.52	Inactive	11.27	Inactive	12.81	Inactive	
-3-one							
30-Hydroxyfriedelane	Inactive	Inactive	Inactive	Inactive	Inactive	Inactive	
-1,3-dione							
21α-Hydroxyfriedelane	Inactive	Inactive	Inactive	Inactive	Inactive	Inactive	
-1,3-dione							
Doxorubicin	0.18	0.15	0.02	1.67	0.96	-	
Streptomycin	-	-	-	-	-	2.15	
Isoniazid	-	-	-	-	-	0.34	
Ofloxacin	-	-	-	-	-	2.16	
Ethambutol	-	-	-	-	-	18.35	

Table 27. Cytotoxicity and anti-tuberculosis activity of compounds isolated from *S. verrucosa* (IC_{50} and MIC in μ M).

From *F. foveolata*, four eudesmane-type sesquiterpenes, including two new compounds, foveolide A and foveoeudesmenone and two known compounds, 4(15)-eudesmene-1 β ,6 α -diol and 4(15)-eudesmene-1 β ,5 α -diol, together with one new sesquiterpenoid dimer, foveolide B, and a new phenolic compound, foveospirolide, were evaluated for their cytotoxic activities against five human cancer cell lines and *Mycobacterium tuberculosis*. The results, as seen in **Table 28**, showed that the α , β -unsaturated sesquiterpene lactone foveolide A displayed moderate cytotoxicity against colon, liver, breast and gastric cancer cell lines with IC₅₀ values of 21.75, 39.97, 31.34 and 30.38 μ M, respectively. The compound also exhibited weak anti-tuberculosis activity against *M. tuberculosis* with an MIC of 200 μ M. Interestingly, foveolide B, which is a sesquiterpenoid dimer with one lactone ring in the skeleton, was specifically cytotoxic toward the

colon cancer cell line with an IC_{50} value of 20.58 μ M. No other tested compounds from this plant were active against the tuberculosis bacterium.

Table 28. Cytotoxicity and anti-tuberculosis activity of compounds isolated from *F. foveolata* (IC_{50} and MIC in μ M).

	Cytotoxicity (IC_{50})					Anti-TB
Compound	SW620	CHAGO	HepG2	BT474	KATO-III	(MIC)
4(15)-Eudesmene-1β,6α	Inactive	Inactive	Inactive	Inactive	Inactive	Inactive
-diol						
4(15)-Eudesmene-1β,5α	Inactive	Inactive	Inactive	Inactive	Inactive	Inactive
-diol						
Foveolide B	20.58	Inactive	Inactive	Inactive	Inactive	Inactive
Foveolide A	21.75	Inactive	39.97	31.34	30.38	200.00
Foveoeudesmenone	Inactive	Inactive	Inactive	Inactive	Inactive	Inactive
Foveospirolide	Inactive	Inactive	Inactive	Inactive	Inactive	Inactive
Doxorubicin	0.11	1.45	0.68	1.45	1.60	-
Streptomycin	-	-	-	-	-	2.15
Isoniazid	-	-	-	-	-	0.34
Ofloxacin	-	-	-	-	-	2.16
Ethambutol	-	-	-	-	-	18.35

CHAPTER V

CONCLUSION

Phytochemical investigation of the stems of *Salacia verrucosa* (Celastraceae), which is used in traditional Thai medicine, led to the isolation of eight triterpenoids, one of which is a new friedelane-type triterpene. Two known compounds (friedelin and friedelane-1,3-dione) were isolated from the *n*-hexane-acetone extract of the stems, whereas a new friedelane-type triterpene, 21 α -hydroxyfriedelane-1,3-dione, along with five known compounds including kokoonol, 26hydroxyfriedelane-1,3-dione, 21 α -hydroxyfriedelan-3-one, 3 β ,22 α -dihydroxyolean-12-en-29-oic acid and 30-hydroxyfriedelane-1,3-dione were obtained from the EtOAc extract of the same plant part. The occurrence of friedelane-type triterpenes as major constituents in the stems of *S. verrucosa* is in accordance with previous chemotaxonomic data of plants in the family Celastraceae. Friedelane-1,3-dione was strongly and selectively cytotoxic to human colon cancer cell line, whereas 26-hydroxyfriedelane-1,3-dione displayed moderate cytotoxic activity against the colon (SW620), lung (CHAGO), liver (HepG2), breast (BT474) and gastric (KATO-III) cancer cell lines. Another friedelane triterpene with moderate cytotoxic activity was 21 α -Hydroxyfriedelan-3-one, which was active against colon, liver and gastric cancer cell lines.

Chemical study of the stems of the other Thai medicinal plant, *Ficus foveolata* (Moraceae), led to the isolation of two new eudesmane-type sesquiterpenes (foveolide A and foveoeudesmenone), a new sesquiterpenoid dimer (foveolide B), and a new phenolic compound (foveospirolide) along with six known compounds. Two known triterpenes (friedelin and taraxerol) were obtained from its *n*-hexane extract, whereas all new compounds along with a known triterpene (betulin), two known eudesmane-type sesquiterpenes [4(15)-eudesmene-1 β ,6 α -diol and 4(15)-eudesmene-1 β ,5 α -diol] and one known phenolic compound (ethyl rosmarinate) were isolated from its CH₂Cl₂ extract. Foveolide A was moderately cytotoxic to colon, liver, breast and gastric cancer cell lines, whereas foveolide B displayed exclusive cytotoxicity against colon cancer cell line. Foveolide A was also weakly active against the tuberculosis-causing *Mycobacterium tuberculosis*. Sesquiterpenoids are rarely found in plants of the genus *Ficus* and the occurrence of a sesquiterpenoid dimer in a member of the family Moraceae is reported herein

for the first time. All known compounds obtained from *F. foveolata* have been reported for the first time from this plant species.

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APPENDIX



Figure 7. ESI Mass spectrum of compound SV-1



Figure 8. IR Spectrum of compound SV-1 (KBr)



Figure 9. ¹H NMR (300 MHz) Spectrum of compound SV-1 (in CDCl₃)



Figure 10. ¹³C NMR (75 MHz) Spectrum of compound SV-1 from 5-60 ppm (in CDCl₃)



Figure 11. ESI Mass spectrum of compound SV-2



Figure 12. IR Spectrum of compound SV-2 (KBr)



Figure 13. ¹H NMR (300 MHz) Spectrum of compound SV-2 (in CDCl₃)



Figure 14. ¹³C NMR (75 MHz) Spectrum of compound SV-2 (in CDCl₃)



Figure 15. ESI Mass spectrum of compound SV-3



Figure 16. IR Spectrum of compound SV-3 (KBr)



Figure 17. ¹H NMR (300 MHz) Spectrum of compound SV-3 (in CDCl₃)



Figure 18. ¹³C NMR (75 MHz) Spectrum of compound SV-3 (in CDCl₃)



Figure 19. ESI Mass spectrum of compound SV-4



Figure 20. IR Spectrum of compound SV-4 (KBr)



Figure 21. ¹H NMR (300 MHz) Spectrum of compound SV-4 (in CDCl₃)



Figure 22. ¹³C NMR (75 MHz) Spectrum of compound SV-4 (in CDCl₃)



Figure 23. ESI Mass spectrum of compound SV-5



Figure 24. IR Spectrum of compound SV-5 (KBr)



Figure 25. ¹H NMR (300 MHz) Spectrum of compound SV-5 (in CDCl₃)



Figure 26. ¹³C NMR (75 MHz) Spectrum of compound SV-5 from 5-80 ppm (in CDCl₃)



Figure 27. ESI Mass spectrum of compound SV-6



Figure 28. IR Spectrum of compound SV-6 (KBr)



Figure 29. ¹H NMR (500 MHz) Spectrum of compound SV-6 (in CDCl₃)



Figure 30. ¹³C NMR (125 MHz) Spectrum of compound SV-6 (in CDCl₃)



Figure 31a. HMBC Spectrum of compound SV-6 (expansion between δ_{H} 3.0-5.4, δ_{C} 13-51 ppm)



Figure 31b. HMBC Spectrum of compound SV-6 (expansion between $\delta_{\rm H}$ 0.6-1.2, $\delta_{\rm C}$ 30-45 ppm)



Figure 31c. HMBC Spectrum of compound SV-6 (expansion between $\delta_{\rm H}$ 1.0-2.3, $\delta_{\rm C}$ 176-187 ppm)



Figure 32. NOESY Spectrum of compound SV-6 (expansion between $\delta_{\rm H}$ 0.5-4.0 ppm)



Figure 33. HR-ESI-TOF Mass spectrum of compound SV-7



Figure 34. IR Spectrum of compound SV-7 (KBr)



Figure 35. ¹H NMR (500 MHz) Spectrum of compound SV-7 (in CDCl₃)



Figure 36. ¹³C NMR (125 MHz) Spectrum of compound SV-7 (in CDCl₃)



Figure 37a. HMBC Spectrum of compound SV-7 (expansion between $\delta_{\rm H}$ 1.6-3.6, $\delta_{\rm C}$ 5-45 ppm)



Figure 37b. HMBC Spectrum of compound SV-7 (expansion between $\delta_{\rm H}$ 1.35-1.65, $\delta_{\rm C}$ 27-35 ppm)



Figure 37c. HMBC Spectrum of compound SV-7 (expansion between $\delta_{\rm H}$ 1.05-1.65, $\delta_{\rm C}$ 70-74 ppm)



Figure 38. HR-ESI-TOF Mass spectrum of compound SV-8



Figure 39. IR Spectrum of compound SV-8 (KBr)



Figure 40. ¹H NMR (500 MHz) Spectrum of compound SV-8 (in CDCl₃)



Figure 41. ¹³C NMR (125 MHz) Spectrum of compound SV-8 (in CDCl₃)



Figure 42a. HMBC Spectrum of compound SV-8 (expansion between δ_{H} 3.1-3.8, δ_{C} 24-38 ppm)



Figure 42b. HMBC Spectrum of compound SV-8 (expansion between $\delta_{\rm H}$ 1.7-3.8, $\delta_{\rm C}$ 45-75 ppm)



Figure 43. NOESY Spectrum of compound SV-8 (expansion between $\delta_{\rm H}$ 0.5-3.8 ppm)



Figure 44. ESI Mass spectrum of compound FF-1



Figure 45. IR Spectrum of compound FF-1 (KBr)



Figure 46. ¹H NMR (300 MHz) Spectrum of compound FF-1 from δ 0.5-2.6 ppm (in CDCl₃)



Figure 47. ¹³C NMR (75 MHz) Spectrum of compound FF-1 (in CDCl₃)



Figure 48. ESI Mass spectrum of compound FF-2



Figure 49. IR Spectrum of compound FF-2 (KBr)



Figure 50. ¹H NMR (300 MHz) Spectrum of compound FF-2 (in CDCl₃)



Figure 51a. ¹³C NMR (75 MHz) Spectrum of compound FF-2 from δ 12-60 ppm (in CDCl₃)



Figure 51b. ¹³C NMR (75 MHz) Spectrum of compound FF-2 from δ 70-160 ppm (in CDCl₃)



Figure 52. ESI Mass spectrum of compound FF-3







Figure 54. ¹H NMR (300 MHz) Spectrum of compound FF-3 from δ 0.1-5.0 ppm (in CDCl₃)



Figure 55a. ¹³C NMR (75 MHz) Spectrum of compound FF-3 from δ 12-65 ppm (in CDCl₃)



Figure 55b. ¹³C NMR (75 MHz) Spectrum of compound FF-3 from δ 70-160 ppm (in CDCl₃)


Figure 56. ESI Mass spectrum of compound FF-4



Figure 57. IR Spectrum of compound FF-4 (KBr)



Figure 58. ¹H NMR (300 MHz) Spectrum of compound FF-4 (in CDCl₃)



Figure 59. ¹³C NMR (75 MHz) Spectrum of compound FF-4 (in CDCl₃)



Figure 60. ESI Mass spectrum of compound FF-5



Figure 61. IR Spectrum of compound FF-5 (KBr)



Figure 62. ¹H NMR (300 MHz) Spectrum of compound FF-5 (in CDCl₃)



Figure 63. ¹³C NMR (75 MHz) Spectrum of compound FF-5 (in CDCl₃)



Figure 64. UV Spectrum of compound FF-7 (in MeOH)



Figure 65. HR-ESI-TOF Mass spectrum of compound FF-7



Figure 66. IR Spectrum of compound FF-7 (KBr)



Figure 67. ¹H NMR (500 MHz) Spectrum of compound FF-7 (in CDCl₃)



Figure 68. ¹³C NMR (125 MHz) Spectrum of compound FF-7 (in CDCl₃)



Figure 69a. HMBC Spectrum of compound FF-7 (in CDCl₃)



Figure 69b. HMBC Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 3.3-4.1, $\delta_{\rm C}$ 5-58 ppm)



Figure 69c. HMBC Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 1.0-4.1, $\delta_{\rm C}$ 120-180 ppm)



Figure 69d. HMBC Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 6.2-6.9, $\delta_{\rm C}$ 32-77 ppm)



Figure 69e. HMBC Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 6.2-6.9, $\delta_{\rm C}$ 125-172 ppm)



Figure 70a. NOESY Spectrum of compound FF-7 (in CDCl₃)



Figure 70b. NOESY Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 1.2-2.8; 3.6-4.0 ppm)



Figure 70c. NOESY Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 1.1-2.4 ppm)



Figure 70d. NOESY Spectrum of compound FF-7 (expansion between $\delta_{\rm H}$ 0.7-1.5; 3.6-4.0 ppm)



Figure 71. HR-ESI-TOF Mass spectrum of compound FF-6



Figure 72. IR Spectrum of compound FF-6 (KBr)



Figure 73. ¹H NMR (500 MHz) Spectrum of compound FF-6 (in CDCl₃)



Figure 74. ¹³C NMR (125 MHz) Spectrum of compound FF-6 (in CDCl₃)



Figure 75a. HMBC Spectrum of compound FF-6 (expansion between $\delta_{\rm H}$ 0.6-2.8, $\delta_{\rm C}$ 10-60 ppm)



Figure 75b. HMBC Spectrum of compound FF-6 (expansion between $\delta_{\rm H}$ 1.1-2.8, $\delta_{\rm C}$ 115-145 ppm)



Figure 75c. HMBC Spectrum of compound FF-6 (expansion between $\delta_{\rm H}$ 1.8-2.5, $\delta_{\rm C}$ 38-56 ppm)



Figure 75d. HMBC Spectrum of compound FF-6 (expansion between $\delta_{\rm H}$ 1.3-1.8, $\delta_{\rm C}$ 38-60 ppm)





Figure 76a. NOESY Spectrum of compound FF-6 (expansion between $\delta_{\rm H}$ 0.8-2.4; 3.7-4.5 ppm)



Figure 76b. NOESY Spectrum of compound FF-6 (expansion between $\delta_{\rm H}$ 0.6-2.8 ppm)



Figure 77. ¹H NMR (300 MHz) Spectrum of compound FF-8a (in CDCl₃)







Figure 79. NOESY Spectrum of compound FF-8a (in CDCl₃)



Figure 80. UV Spectrum of compound FF-8b (in MeOH)



Figure 81. HR-ESI-TOF Mass spectrum of compound FF-8b



Figure 82. IR Spectrum of compound FF-8b (KBr)



Figure 83a. ¹H NMR (500 MHz) Spectrum of compound FF-8b (in CDCl₃)



Figure 83b. ¹H NMR (500 MHz) Spectrum of compound FF-8b (expansion between δ 2.45-2.80 ppm)



Figure 84. ¹³C NMR (125 MHz) Spectrum of compound FF-8b (in CDCl₃)



Figure 85a. HMBC Spectrum of compound FF-8b (in CDCl₃)



Figure 85b. HMBC Spectrum of compound FF-8b (expansion between $\delta_{\rm H}$ 1.5-2.9, $\delta_{\rm C}$ 192-212 ppm)



Figure 85c. HMBC Spectrum of compound FF-8b (expansion between $\delta_{\rm H}$ 0.6-2.9, $\delta_{\rm C}$ 125-170 ppm) H-1



Figure 86a. NOESY Spectrum of compound FF-8b (expansion between $\delta_{\rm H}$ 1.2-2.3; 3.4-4.2 ppm)



Figure 86b. NOESY Spectrum of compound FF-8b (expansion between $\delta_{\rm H}$ 0.8-2.3 ppm)



Figure 87. UV Spectrum of compound FF-9 (in MeOH)



Figure 88. ESI Mass spectrum of compound FF-9



Figure 89. IR Spectrum of compound FF-9 (KBr)



Ire 90. ¹H NMR (500 MHz) Spectrum of compound FF-9 (in acetone- d_6)



Figure 91. ¹³C NMR (125 MHz) Spectrum of compound FF-9 (in acetone- d_6)



Figure 92. UV Spectrum of compound FF-10 (in MeOH)



Figure 93. HR-ESI-TOF Mass spectrum of compound FF-10



Figure 94. IR Spectrum of compound FF-10 (KBr)



Figure 95. ¹H NMR (500 MHz) Spectrum of compound FF-10 (in CD_3OD)



Figure 96. ¹³C NMR (125 MHz) Spectrum of compound FF-10 from δ 30-180 ppm (in CD₃OD)



Figure 97a. HMBC Spectrum of compound FF-10 (expansion between $\delta_{\rm H}$ 2.5-4.2, $\delta_{\rm C}$ 100-180 ppm)



Figure 97b. HMBC Spectrum of compound FF-10 (expansion between $\delta_{\rm H}$ 6.7-7.1, $\delta_{\rm C}$ 110-150 ppm)



Figure 97c. HMBC Spectrum of compound FF-10 (expansion between $\delta_{\rm H}$ 2.6-4.2, $\delta_{\rm C}$ 30-80 ppm)



Figure 98a. NOESY Spectrum of compound FF-10 (in CD_3OD)



Figure 98b. NOESY Spectrum of compound FF-10 (expansion between $\delta_{\rm H}$ 3.0-4.1; 6.7-7.1 ppm)

VITA

Mr. Pathom Somwong was born on September 20, 1978 in Nakhon Ratchasima, Thailand. He received his B.Sc. in Pharmacy (Second class honours) in 2001 from the Faculty of Pharmaceutical Sciences, Khon Kaen University and M.Sc. in Pharmacy in 2007 from the Department of Pharmaceutical Chemistry, Faculty of Pharmaceutical Sciences, Chulalongkorn University. After graduation, he worked as a lecturer at the Faculty of Pharmacy, Rangsit University, Pathum Thani, Thailand. He was granted a Royal Golden Jubilee Ph.D. Scholarship from Thailand Research Fund (TRF) in the year 2009.

Publications

- Somwong, P., Suttisri, R., and Buakeaw, A. 2011. A new 1,3-diketofriedelane triterpene from *Salacia verrucosa*. <u>Fitoterapia</u> 82: 1047-1051.
- Somwong, P., Suttisri, R., and Buakeaw, A. 2013. New sesquiterpenes and phenolic compound from *Ficus foveolata*. <u>Fitoterapia</u> 85: 1-7.

Poster presentations

- Somwong, P. and Suttisri, R. Two new friedelane triterpenoids from *Salacia verrucosa*. Presented at the 9th NRCT-JSPS Joint Seminar "Natural Medicine Research for the Next Decade: New Challenges and Future Collaboration", December 8-9, 2010, Bangkok, Thailand.
- Somwong, P., Suttisri, R., and Buakeaw, A. New eudesmane sesquiterpenes from *Ficus foveolata*. Presented at RGJ-Ph.D. Congress XIV "Basic Research for Sustainable Development", April 5-7, 2013, Jomtien Palm Beach Hotel & Resort, Pattaya, Chonburi, Thailand.