

### Chapter 1

### Introduction.

In this chapter I will review the history of superconductivity, and its basic properties. The discovery of high temperature superconductor follows and the chapter ends with a section on the structure of these new superconductors.

Superconductors are a class of materials whose dc resistivity goes to zero below a certain temperature T<sub>c</sub>, called the critical temperature. This incredible behavior, called superconductivity, was discovered in 1911 by the Dutch physicist

H. Kamerlingh Onnes (1).He and one of his graduate students found that the dc resistance of purified mercury dropped abruptly to zero at a temperature of 4.15 K, as shown in Figure 1.1. Above this temperature the resistivity is small, but finite, while the resistivity below this point is so small that it is essentially zero. When it was cooled through the transition temperature in the absence of any external magnetic field, it became superconducting. Onnes won the Nobel Prize in 1913 for the discovery of superconductivity, and for the liquefaction of helium.

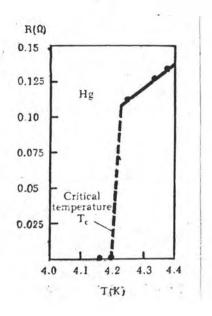


Fig.1.1 Resistance versus temperature for mercury. The graph follows that of a normal metal above the critical temperature. The dc resistance drops abruptly to zero at the critical temperature, which is 4.15 K for mercury (2).

Following Onnes's discovery, many new materials (both elements and compounds), were found to be superconductors, each with own critical temperature  $\mathbf{T}_{\mathbf{c}}$ .

Over the years after Onnes's discovery, intense theoretical and experimental efforts have been devoted to study superconductivity. It is found that when a normal metal becomes superconducting, its superconducting properties are very different from normal metal behaviors.

## Basic Properties of Conventional Superconductors

Various experiments on superconductivity in many metals and compounds have shown that the nature of the superconducting state is very different from that of metals. For example,

- 1. A superconductor exhibits infinite conductivity (2).
- 2. A bulk specimen of metal in the superconducting state exhibits perfect diamagnetism, with the magnetic induction  $\stackrel{\rightharpoonup}{B}=0$ . This is the Meissner effect (2). The external magnetic field will penetrate the surface of the specimen over a distance determined by the penetration depth,  $\lambda$ (2).
- 3. There are two types of superconductors, I and II.In a bulk specimen of type I superconductor, the superconducting state is destroyed and the normal state is restored by application of an external magnetic field in excess of a critical value H<sub>c</sub>. A type II superconductor has two critical fields, a vortex state exists in the range between H<sub>c1</sub> and H<sub>c2</sub> (2).
- 4. The charge carriers in the superconducting state have charge -2e and are hence formed by pairs of electron. The average separation of the electrons making up the pairs is given by the intrinsic coherence length  $\S$ . Type II superconductors have  $\S$ . (2).
- 5. In the superconducting state, an energy gap  $E_{\tt g} = 3.5 \; k_{\tt B} T_{\tt c} \; {\tt separates} \; {\tt superconducting} \; {\tt electrons} \; {\tt below} \; {\tt from} \; {\tt normal} \; {\tt electrons} \; {\tt above} \; {\tt the} \; {\tt gap} \; {\tt is} \; {\tt detected} \; {\tt in} \; {\tt experiments} \; {\tt on} \; {\tt heat} \; {\tt capacity,} \; {\tt infrared} \; {\tt absorption,} \; {\tt and} \; {\tt tunneling} \; (2).$
- 6. In many superconductors, the critical temperature varies with isotopic mass M as  $T_{\rm c} \propto M^{-1/2}$  (2).

7. Magnetic impurities such as Gd in LaAl<sub>2</sub> destroy the superconductivity at very small concentrations (2).

Since the discovery of superconductivity in metals by Onnes, there has been a substantial effort to increase the transition temperature by alloying intermetallic compounds in many laboratories.

For many years, all efforts to enhance T<sub>c</sub> over the 23.3 K reached by Gavaler (3) and Testardi et al.(4) in thin films of Nb<sub>3</sub>Ge failed. This situation led one to the conviction that the efforts in intermetallic compounds should not be pursued further.

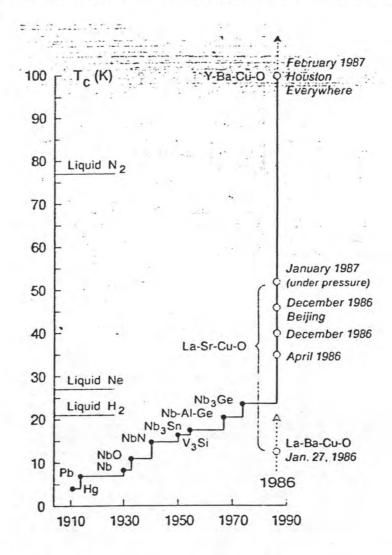
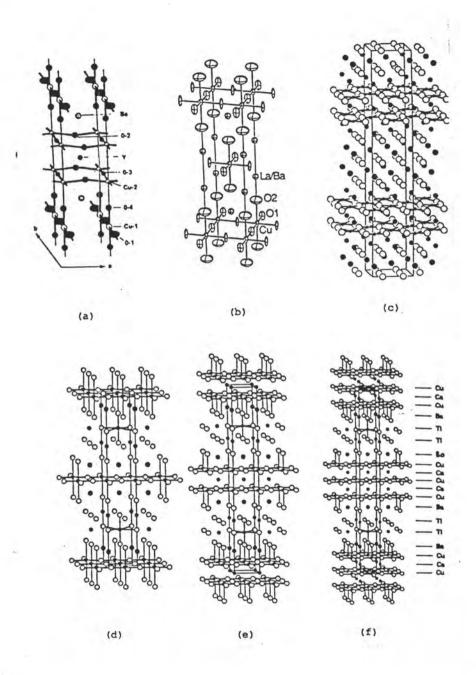


Fig.1.2 Evolution of the superconductive transition temperature(5).

Later studies in niobium-doped SrTiO3, the first superconducting metallic oxide with a T<sub>e</sub> of 0.3K (6) increased T<sub>e</sub> to 0.7K (7,8). For this T<sub>c</sub>, the carrier concentration n was only 2 × 10<sup>20</sup> cm<sup>-3</sup>, two orders of magnitude lower than in a metal, suggesting an extremely large electron-phonon coupling. In 1973, superconductivity in the Li-Ti-O system with onsets as high as 13.7 K was reported by Johnston et al. (9). Two years later, superconductivity in the mixed-valence compound BaPb, Bi,O3, a perovskite, was discovered by Sleight et al. (10). The highest T<sub>c</sub> in homogeneous oxygen-deficient mixed crystals occurs at 13 K, with a comparatively low concentration of carriers, n=2X10<sup>21</sup> to 4X10<sup>21</sup> cm<sup>-3</sup>(11,12). Therefore, according to the Bardeen-Cooper-Schrieffer (BCS) theory (13), a large electron-phonon coupling was present. Thus one could expect to find still higher T<sub>c</sub>'s in other metallic oxides if the electron-phonon interactions and the carrier densities, N(E,), at the Fermi level could be further enhanced.

In 1986 Bednorz and Müller (14) discovered that La<sub>2</sub>CuO<sub>4</sub> when doped with barium, is a superconductor. The dopant replaced La, and the optimum concentration is about 8% of Ba, yielding a superconductor with T<sub>c</sub> of 35 K.

The next major breakthrough was that of Chu's group (15) in 1987. They found T<sub>c</sub> above the liquid nitrogen temperature of 90 K for a new material, comprising the elements barium, yttrium, copper, and oxygen, which is now known to be YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-8</sub> and its crystallography structure is shown in Fig.(1.3).



Chu's material will be referred to henceforth a "YBCO" and the class of materials to which it belongs will be called collectively the "1-2-3" materials.

One of the most surprising subsequent discoveries (19) was that the yttrium in YBCO is a "passenger" for which almost any rare earth element can be substituted, with hardly any change in  $T_{\rm c}$ .

For almost a year, the 1-2-3 materials have the entire stage to themselves. But some new classes(20) of copper-oxide-based materials have been discovered more recently in particular(16,17,18) the series of compounds Bi<sub>2</sub>Sr<sub>2</sub>Ca<sub>n-1</sub>Cu<sub>n</sub>O<sub>2n</sub> with T<sub>c</sub>'s of 7 K, 80 K, and 110 K for n=1,2,3 respectively and the series of Tl<sub>2</sub>Ba<sub>2</sub>Ca<sub>n-1</sub>Cu<sub>n</sub>O<sub>2n+4</sub> with T<sub>c</sub>'s of 80 K, 110 K and 125 K for n=1,2,3 respectively. In both these systems, T<sub>c</sub> was seen to increase with the number n of CuO<sub>2</sub> planes per unit cell, at least in the range n=1 to 3. This has led to a suggestion that increasing n still further might provide a route to higher T<sub>c</sub>.

Recently Ihara et al.(21)find that  $T_c$  has a maximum around n=4 in the series of compound  $TlBa_2Ca_{n-1}Cu_nO_{2n+3}$  for n=1,2,3,4 and 5 they report values of  $T_c$  respectively, 17 K, 91 K, 116 K, 122 K and less than 120 K.

## Basic Structure of High-temperature Superconductors.

High-temperature superconductor has now been observed in four families of copper oxide compounds.

- 1. The  $\rm La_{z-x}AE_zCuO_4$  phases (14) where AE is a alkaline earth atom Ba, Sr or Ca, with a maximum  $\rm T_c$  of 40 K.
- 2. The  $YBa_2Cu_3O_7$  (1-2-3) phases (15) where Y can be replaced by other rare earths, with a maximum  $T_c$  of 93 K.

- 3. The Bi<sub>2</sub>Ca<sub>1</sub>Sr<sub>2</sub>Cu<sub>2</sub>O<sub>8</sub> (2-1-2-2) phases(22,23), where Bi can be replaced by Tl and Sr by Ba, with a maximum T<sub>c</sub> around 90 K.
- 4. The  $\mathrm{Tl_2Ca_2Ba_2Cu_3O_{10}}$  phases (2-2-2-3)(24) with a maximum T of 125 K.

The most fundamental structure of superconductors is the  ${\rm CuO}_2$  plane as shown in Fig.1.4 The  ${\rm CuO}_2$  plane is a two dimensional plane formed by combining square planar units, each of which is surrounded by oxygen and has a Cu at its center. The square planars are combined through the oxygen at their apices. This plane seems to be the most important structure factor to produce superconductivity. Any of the substances such as  $Y_1 {\rm Ba}_2 {\rm Cu}_3 {\rm O}_7$ ,  $({\rm BiO})_2 {\rm Sr}_2 {\rm Ca}_{n-1} {\rm Cu}_n {\rm O}_{2n+2}$ ,  $({\rm TlO})_2 {\rm Ba}_2 {\rm Ca}_{n-1} {\rm Cu}_{n-1} {\rm Cu}_n {\rm O}_{2n+2}$ ,  $({\rm Nd}, {\rm Ce}({\rm Sr}))_2 {\rm CuO}_4$ , and  $({\rm La}, {\rm Ba})_2 {\rm CuO}_4$  discoverd by Bednorz-Muller, have this basic structure.

# Some features of La 2-x Ba (Sr ) CuO4 and LaBa Cu3O7

 ${\rm La_{2-x}Ba_x(Sr_x)CuO_4}$  compounds (Fig.1.5) have the quasi two-dimensional  ${\rm K_2NiF_4}$  structure with tetragonally elongated Cuoxygen octahedra. These oxides have a tetragonal structure at room temperature when x>0.05 and become orthorhombic around 180 K, well above the superconducting transition temperature.  ${\rm T_c}$  in these oxides shows a maximum around a specific value of x (~0.15 and 0.2 respectively in the case of Ba and Sr). The La ion in  ${\rm La_{2-x}Ba_xSr_x}$   ${\rm CuO_4}$  can be replaced by other rare earth ions upto a point (<10%) without losing superconductivity. Substitution of Cu partly by Ni, Zn and such ions generally lower the  ${\rm T_c}$ .

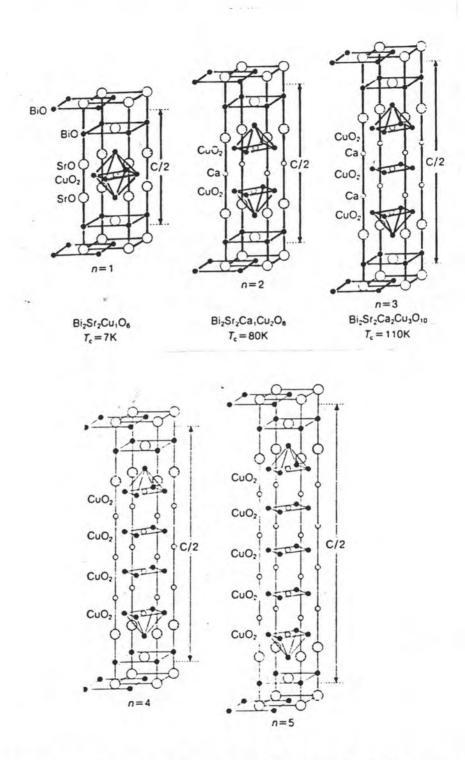


Fig. 1.4 Crystal structure of a Bi-Sr-Ca-Cu-O system (25).

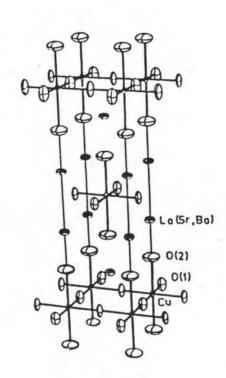


Fig. 1.5 Structure of  $La_{2-x}Ba_x(Sr_x)CuO_4$  (26).

 ${\rm YBa_2Cu_3O_7}$  and related 123 oxides are orthorhombic (Fig.1.6), by virtue of the preferential population of the 01 sites (along the b-axis) giving rise to the Cu-O chains. Depletion of 01 oxygens or disorder between the 01 and 05 site gives rise to a tetragonal structure (Fig.1.6). The orthorhombic structure is responsible for the formation of twins; across the twin boundary, the <u>a</u> and <u>b</u> parameters seem to interchange.  ${\rm YBa_2Cu_3O_{7-5}}$  shows a progressive lowering of T<sub>c</sub> as  ${\cal S}$  increases until the superconductivity is lost at  ${\cal S}\approx 0.6$  when the structure is tetragonal. The Cu ion in  ${\rm YBa_2Cu_3O_7}$  can be replaced by cations such as Zn, Fe and Ca and these substitions generally lower the T<sub>c</sub> because of a change in oxygen stoichiometry.

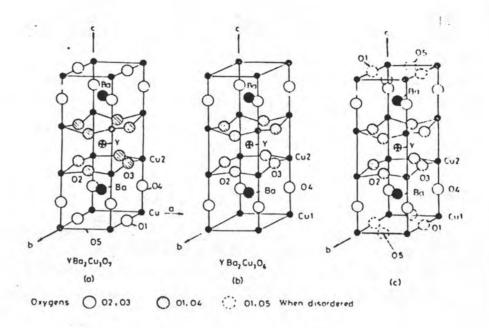


Fig. 1.6 Structure of YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7- $\delta$ </sub>
(a)  $\delta$  = 0.0, orthorhombic; (b)  $\delta$  =1.0, tetragonal; (c) disordered tetragonal structure (26).

### Bi-Ca-Sr-Cu-O and Tl-Ca-Ba-Cu-O systems

Several compositions of the Bi-Ca-Sr-Cu-O system have been studied, the member which has reproducibility shows high  $T_c$  superconductivity (zero-resistance 90 K and onset 112 K is  $Bi_2(Ca,Sr)_3$   $Cu_2O_{8+\delta}$ . By varying the Ca/Sr ratio, it is found that compositions close to  $Bi_2Ca_{1.5}Sr_{1.5}Cu_2O_{8+\delta}$  with a slight Bi excess are ideal. In Fig.(1.7), electrical resistivity data of the n=2 member of the  $Bi_2Ca_xSr_{n+1-x}Cu_nO_{2n+4}$  series is shown along with that of n=3 member. The n=1 member without Ca (27) has very low  $T_c$ , but with x=1, a  $T_c$  of around 80 K seems to be observed (Fig. 1.7), this n=1 member,  $Bi_2CaSrCuO_{6+\delta}$ , is generally in mixture with the 2122 phases. In n=3 member can be prepared in mixture with the n=2 member. The n=3 member  $Bi_2Ca_2Sr_2Cu_3O_{10+\delta}$  shows a  $T_c$  of 110 K with

a still higher onset temperature. The presence of the n=3 member was probably responsible for the 110 K superconductivity found in other compositions. Partial substitution of Bi by Pb upto 25% is possible, with a marginal increase in  $T_{\rm c}$  in the case of the n=2 member.

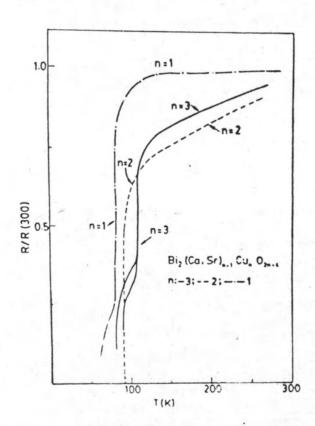


Fig. 1.7 Resistivity data of the Bi-Ca-Sr-Cu-O system (26).

The bismuth cuprates have structures similar to the family of oxides,  $(Bi_2O_2)^{2+}(A_{n-1}B_nO_{3n+1})^{2-}$ . The main difference is that we do not have  $Bi_2O_2$  layers in the oxides. The actual structure of  $Bi_2(Ca,Sr)_3Cu_{8+\S}$  is shown in Fig.(1.8).

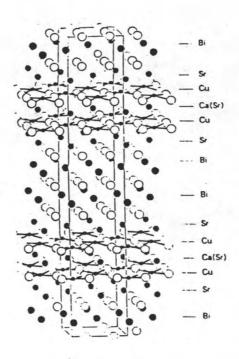


Fig. 1.8 Structure of Bi<sub>2</sub>(Ca,Sr)<sub>3</sub>Cu<sub>8+8</sub> (26).

In the Tl-Ca-Ba-Cu-O system, the TlBa $_2$ CuO $_6$ , TlCaBa $_2$ Cu $_2$ O $_8$  and TlCa $_2$ Ba $_2$ Cu $_3$ O $_{10}$  are well-characterized (28,29,30) with zero-resistance T $_c$ 's around 80, 110 and 125 K for n=1,2 and 3 respectively. The Tl cuprates give very sharp resistivity and susceptibility anomalies at T $_c$ . The structures of Tl-Ca-Ba-Cu-O superconductors are similar to those of the Bi-Ca-Sr-Cu-O superconductors. Fig. (1.9), shows the structure of Tl $_2$ CaBa $_2$ Cu $_2$ O $_8$ .

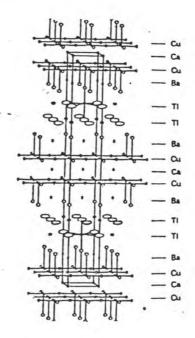


Fig. 1.9 Structure of  $Tl_2CaBa_2Cu_2O_{8+3}$  (26).

The  $T_c$  in the Tl-Ca-Ba-Cu-O system also increases with the number of Cu-O layers. The n=4 and 5 members of this family would be expected to show  $T_c$ 's of around 140 and 160 K respectively. Fig.(1.10) shows the schematic structure of the different members of Bi and Tl cuprate families to point out their similarity.

				TI
			11	——— Во
				Cu
	——Ві		Ва	Ca
	DI.		Cu	Cu
——Ві	Sr	—— Ва	—— Са	Ca
0.	Cu	Cu	Cu	Cu
Co	Ca.Sr	Са	——Со	Ca
Cu	Cu	Cu	Cu	Cu
Sr	Sr	Ва	——— Ва	——— Ва
—— Ві	Bi	TI	T1	ti
c = 24 Å	c=309Å	c= 29 8Å	c = 36 Å	c = 42 Å
B, Co S CuO.	B <sub>12</sub> (Ca.Sr) <sub>3</sub> Cu <sub>2</sub> O <sub>8-4</sub>	TI,CaBa,Cu,Oa.	TI,Ca,Ba,Cu,O,	A,M,M, CU, D, 1.6
n=1(80K)	n= 2(90K)	n= 2(110 K)	n=3(125K)	(Proposed structure)

Fig. 1.10 Schematic structure of the Bi-Ca-Si-Cu-O and Tl-Ca-Ba-Ca-O families. The  $T_{\rm c}$  values are shown in parentheses next to the number of Cu-O layers(n) (26).

In the Bi-Ca-Sr-Cu-O and Tl-Ca-Ba-Cu-O systems, the T<sub>c</sub> increases with the number of Cu-O layers(n), as shown in Fig.(1.11). We could expect a T<sub>c</sub> of 200 K only when n is 7 or higher. This is unlikely to be achieved. However, high T<sub>c</sub>'s in the 200-300 K range found in the Bi-Ca-Sr-Cu-O systems could be due to intergrowths with a large number of Cu-O layers in the unit cell(25). It appears that Bi and Tl ions do not play any significant role in the superconductivity of these cuprates, the role is mainly of oxygen and copper ions.

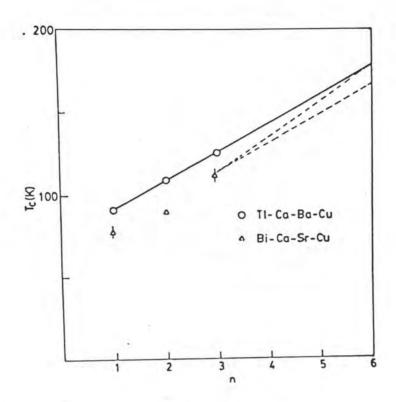


Fig. 1.11 Variation of T<sub>c</sub> with the number of Cu-O layers, in the Bi-Ca-Sr-Cu and Tl-Ca-Ba-Ca systems (26).

### Basic Properties of High Temperature Superconductors.

Some properties of the lamellar CuO<sub>2</sub> superconductors are identical to those of conventional superconductors. Others, however, differ quite fundamentally and this indicates that, at the minimum, the pairing mechanism itself must be quite new. Here, only the essential features are stated.

- 1. In zero field, the resistance is, of course, zero.
- 2. Meissner measurements have turned out to be surprisingly difficult in these materials. Only in high-quality samples at very low fields, typically less than 1 Oe, is a 100% Meissner fraction observed. In higher fields when the sample is cooled through  $T_{\rm c}$  in

the field, strong flux pinning occurs so that there is not a complete flux exclusion. The penetration depth in the in-plane direction at low temperatures is of order 2000 A° (31).

- 3. All of the high- $T_c$  materials are type II superconductors necessitating that  $f_c << \lambda_c$ . The vortices carry one quantum of flux hc/2e implying that the carriers have charge 2e.
- 4. Various measurements, including Josephson tunnel junctions(32), direct measurements of the flux quantum (hc/2e)(33) and the vortex array measurements(34) mentioned above all imply that the carriers are bound in pairs. However the coherence length is only about 15 A° in the plane and less than the interplanar separation perpendicular to the CuO<sub>2</sub> planes (31).
- 5. Measurement of the energy gap have proved to be extremely difficult. Various measurements can be interpreted in terms of a gap with  $E_{\rm g}$  varying between  $2k_{\rm B}T_{\rm c}$  and  $8k_{\rm B}T_{\rm c}$ . However, virtually all spectroscopic studies imply that there are significant excitations with energies less than  $E_{\rm g}$  (35,36).
- 6. The isotope effect in the high-T<sub>c</sub> materials is very small. Indeed, replacement of <sup>16</sup>O by <sup>18</sup>O seems to produce a decrease of T<sub>c</sub> by about 0.3 K independent of T<sub>c</sub> for transition temperature varying from 40 K to 110 K (37).

To date, some relations have been observed between  $T_c$  and the physical properties (25,38). First, the relation between  $T_c$  and the charge of carriers in high temperature, superconductivity is observed when the nominal charge of Cu is +2.1 to +2.3 in the case of the hole conduction. Second, the relation between  $T_c$  and the

number of  ${\rm CuO}_2$  planes is now known. Although the reason is not understood yet, T<sub>c</sub> rises as the number of layers increases from single, to double, to triple. Three layers of  ${\rm CuO}_2$  of the Tl-Ba-Ca-Cu-O system shows the zero resistance temperature, at 125 K. T<sub>c</sub> tends to saturate and decrease as the number of  ${\rm CuO}_2$  increases further.

In the next chapter the BCS theory will be reviewed, and various works done by many researchers using different models will be discussed.