

References

- [1.] Hoogers, G., Fuel Cell Technology Handbook. 2003, CRC Press LLC.
- [2.] EG&G Services, P., Inc., Fuel Cell Handbook. Fifth Edition. 2000, Science Application International Corporation.
- [3.] Potter, M.C., Electrical Effects accompanying the Decomposition of Organic Compounds. Proc. Roy. Soc. (London) Ser. B, 1911. 84: 260-276.
- [4.] Ieropoulos, I.A., et al., Comparative Study of Three Types of Microbial Fuel Cell. Enzyme and Microbial Technology, 2005. 37(2): 238-245.
- [5.] Kim, H.J., et al., A Mediator-Less Microbial Fuel Cell using a Metal Reducing Bacterium, *Shewanella putrefaciens*. Enzyme and Microbial Technology, 2002. 30(2): 145-152.
- [6.] Min, B., Cheng, S., and Logan, B.E., Electricity Generation using Membrane and Salt Bridge Microbial Fuel Cells. Water Research, 2005. 39(9): 1675-1686.
- [7.] Oh, S.-E. and Logan, B.E., Proton Exchange Membrane and Electrode Surface Areas as Factors that Affect Power Generation in Microbial Fuel Cells. Appl Microbiol Biotechnol, 2006. 70(2): 162-169.
- [8.] Tanaka, K., Vega, C.A., and Tamamushi, R., Thionine and Ferric Chelate Compounds as Coupled Mediators in Microbial Fuel Cells. Bioelectrochemistry and Bioenergetics, 1983. 11(4-6): 289-297.
- [9.] Tanaka, K., Vega, C.A., and Tamamushi, R., Mediating Effects of Ferric Chelate Compound in Microbial Fuel Cells. Bioelectrochemistry and Bioenergetics, 1983. 11(2-3): 135-143.
- [10.] Park, D.H. and Zeikus, J.G., Electricity Generation in Microbial Fuel Cells Using Neutral Red as an Electronophore. Appl. Environ. Microbiol., 2000. 66(4): 1292-1297.

- [11.] Zhao, F., et al., Application of Pyrolysed Iron(II) Phthalocyanine and CoTMPP based Oxygen Reduction Catalysts as Cathode Materials in Microbial Fuel Cells. Electrochemistry Communications, 2005. 7(12): 1405-1410.
- [12.] Viedla, H.A. and Arvia, A.J., The Response of a Bioelectrochemical Cell with *Saccharomyces cerevisiae* Metabolizing Glucose under Various Fermentation Conditions. Biotechnol. Bioeng., 1975. 17: 1529-1543.
- [13.] Ardeleanu, I., Margineanu, D.-G., and Vais, H., Electrochemical Conversion in Biofuel Cells using *Clostridium butyricum* or *Stapylococcus aureus oxford*. Bioelectrochem. and Bioenerg., 1983. 11: 273-277.
- [14.] Bennetto, H.P., Electricity Generation by Microorganisms. Biotechnol. Educ., 1990. 1(4): 163-168.
- [15.] Zhang, T., et al., Improved Performances of *E. coli*-Catalyzed Microbial Fuel Cells with Composite Graphite/PTFE Anodes. Electrochemistry Communications, 2007. 9(3): 349-353.
- [16.] Tanisho, S., Kamiya, N., and Wakao, N., Microbial Fuel Cell Using *Enterobacter aerogenes*. Bioelectrochemistry and Bioenergetics, 1989. 21(1): 25-32.
- [17.] Wingard, L.B., Shaw, C.H., and Castner, J.F., Bioelectrochemical Fuel Cells. Enzyme and Microbial Technology, 1982. 4(3): 137-142.
- [18.] Disalvo, E.A. and Videla, H.A., Relation Between Biological Parameters and the Bioelectrochemical Fuel-Cell Response. Bioelectrochem. and Bioenerg., 1979. 6: 185-195.
- [19.] Chaudhuri, S.K. and Lovley, D.R., Electricity Generation by Direct Oxidation of Glucose in Mediatorless Microbial Fuel Cells. Nat. Biotechnol., 2003. 21(10): 1229-1232.

- [20.] Rabaey, K., et al., A Microbial Fuel Cell Capable of Converting Glucose to Electricity at High Rate and Efficiency. Biotechnology Letters 2003. 25: 1531-1535.
- [21.] Mano, N., et al., A Miniature Biofuel Cell Operating at 0.78 V. Chem. Commun 2003: 518-519.
- [22.] Niessen, J., Schroder, U., and Scholz, F., Exploiting Complex Carbohydrates for Microbial Electricity Generation - a bacterial fuel cell operating on starch. Electrochemistry Communications, 2004. 6(9): 955-958.
- [23.] Ieropoulos, I., et al., Energy Accumulation and Improved Performance in Microbial Fuel Cells. Journal of Power Sources, 2005. 145(2): 253-256.
- [24.] He, Z., Minteer, S.D., and Angenent, L.T., Electricity Generation from Artificial Wastewater Using an Upflow Microbial Fuel Cell. Environ. Sci. Technol., 2005. 39(14): 5262-5267.
- [25.] He, Z., et al., An Upflow Microbial Fuel Cell with an Interior Cathode: Assessment of the Internal Resistance by Impedance Spectroscopy. Environ. Sci. Technol., 2006. 40(17): 5212-5217.
- [26.] Bond, D.R., Lovley, D.R., Electricity Production by *Geobacter sulfurreducens* Attached to Electrodes. Appl. Environ. Microbiol., 2003. 69(3): 1548-1555.
- [27.] Liu, H., Ramnarayanan, R., and Logan, B.E., Production of Electricity during Wastewater Treatment Using a Single Chamber Microbial Fuel Cell. Environ. Sci. Technol., 2004. 38(7): 2281-2285.
- [28.] Min, B., et al., Electricity Generation from Swine Wastewater Using Microbial Fuel Cells. Water Research, 2005. 39(20): 4961-4968.
- [29.] Liu, H., Cheng, S., and Logan, B.E., Production of Electricity from Acetate or Butyrate Using a Single-Chamber Microbial Fuel Cell. Environ. Sci. Technol., 2005. 39(2): 658-662.

- [30.] Min, B. and Logan, B.E., Continuous Electricity Generation from Domestic Wastewater and Organic Substrates in a Flat Plate Microbial Fuel Cell. Environ. Sci. Technol., 2004. 38(21): 5809-5814.
- [31.] Rabaey, K. and Verstraete, W., Microbial Fuel Cells: Novel Biotechnology for Energy Generation. Trends in Biotechnology, 2005. 23(6): 291-298.
- [32.] Park, H.S., et al., A Novel Electrochemically Active and Fe(III)-Reducing Bacterium Phylogenetically Related to *Clostridium butyricum* Isolated from a Microbial Fuel Cell. Anaerobe, 2001. 7(6): 297-306.
- [33.] Park, D.H., Zeikus, J. G., Impact of Electrode Composition on Electricity Generation in a Single-Compartment Fuel Cell using *Shewanella putrefaciens*. Appl. Microbiol. Biotechnol., 2002. 59: 58-61.
- [34.] Bond, D.R., et al., Electrode-Reducing Microorganisms That Harvest Energy from Marine Sediments. Science 2002. 295: 483-485.
- [35.] Pham, C.A., et al., A Novel Electrochemically Active and Fe(III)-Reducing Bacterium Phylogenetically Related to *Aeromonas hydrophila*, isolated from a microbial fuel cell. FEMS Microbiology Letters, 2003. 223(1): 129-134.
- [36.] Liu, H. and Logan, B.E., Electricity Generation Using an Air-Cathode Single Chamber Microbial Fuel Cell in the Presence and Absence of a Proton Exchange Membrane. Environ. Sci. Technol., 2004. 38(14): 4040-4046.
- [37.] Oh, S. and Logan, B.E., Hydrogen and Electricity Production from a Food Processing Wastewater Using Fermentation and Microbial Fuel Cell Technologies. Water Research, 2005. 39(19): 4673-4682.
- [38.] Kim, J.R., Min, B., and Logan, B.E., Evaluation of Procedures to Acclimate a Microbial Fuel Cell for Electricity Production. Appl. Microbiol. Biotechnol., 2005. 68(1): 23-30.

- [39.] Park, D.H., Zeikus, J. G., Improved Fuel Cell and Electrode Designs for Producing Electricity from Microbial Degradation. Biotechnol. Bioeng., 2003. 81(3, Feb): 348-355.
- [40.] Cheng, S., Liu, H., and Logan, B.E., Increased Performance of Single-Chamber Microbial Fuel Cells Using an Improved Cathode Structure. Electrochemistry Communications, 2006. 8(3): 489-494.
- [41.] <http://users.rcn.com/jkimball.ma.ultranet/BiologyPages/R/RedoxPotentials.html>.
- [42.] Bagotsky, V.S., Fundamentals of Electrochemistry. Second edition. 2006, John Wiley & Sons, Inc.
- [43.] <http://www.gamry.com>.
- [44.] Nelson, D.L. and Lehninger, C., Principles of Biochemistry. Fourth edition. 2004, W.H. Freeman and company.
- [45.] Rao, P.S. and Hayon, E., Redox Potentials of Free Radicals. I. Simple Organic Radicals. J. Am. Chem. Soc., 1974. 96: 1287-1294.
- [46.] Bard, A.J. and Faulkner, L.R., Electrochemical Methods : Fundamentals and Applications. 1980, New York : Wiley.

APPENDIX

APPENDIX A

STUDY THE USED COMPONENT IN THE BIO-FUEL CELL

This section will describe the effect of each component such as yeast, glucose, methylene blue and ferricyanide (FeCN). The 2515010 compartment was used for studying these components. Each electrode had a surface area of 50 cm^2 . The concentration of yeast, glucose, MB and ferricyanide that dissolved in 0.1 M, pH 7.0 PB were 10 mg/ml, 0.1M, 1mM and 1mM, respectively. Yeast, glucose and MB were added in the anodic compartment while the ferricyanide was added in the cathodic compartment.

Fig. A1 shows the voltage-current density characteristic of one added component in the bio-fuel cell chamber. The result shows that the open circuit voltage generation of MB or glucose was lower than 10 mV. Thirty mV and 82 mV of open circuit voltage when adding yeast and ferricyanide, respectively, were obtained. The current density of yeast and ferricyanide adding were about 0.47 mA/m^2 ($10 \text{ k}\Omega$) and 1 mA/m^2 ($5.1 \text{ k}\Omega$), respectively. It shows that only yeast and ferricyanide affects the small voltage and current generation of the BFC.

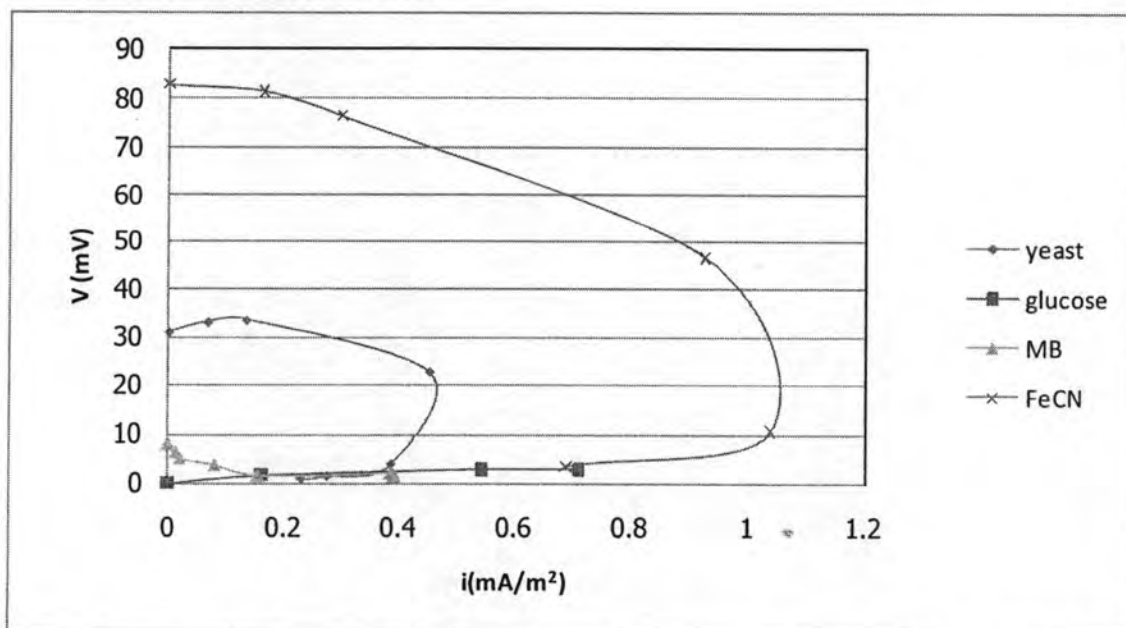


Fig. A1 Voltage-current density generation of each component in the bio-fuel cell

Fig. A2 shows the power density-load characteristic of each component. The result shows that yeast and ferricyanide can give the small power density about $10 \mu\text{W/m}^2$ ($10 \text{ k}\Omega$) and $45 \mu\text{W/m}^2$ ($10 \text{ k}\Omega$), respectively.

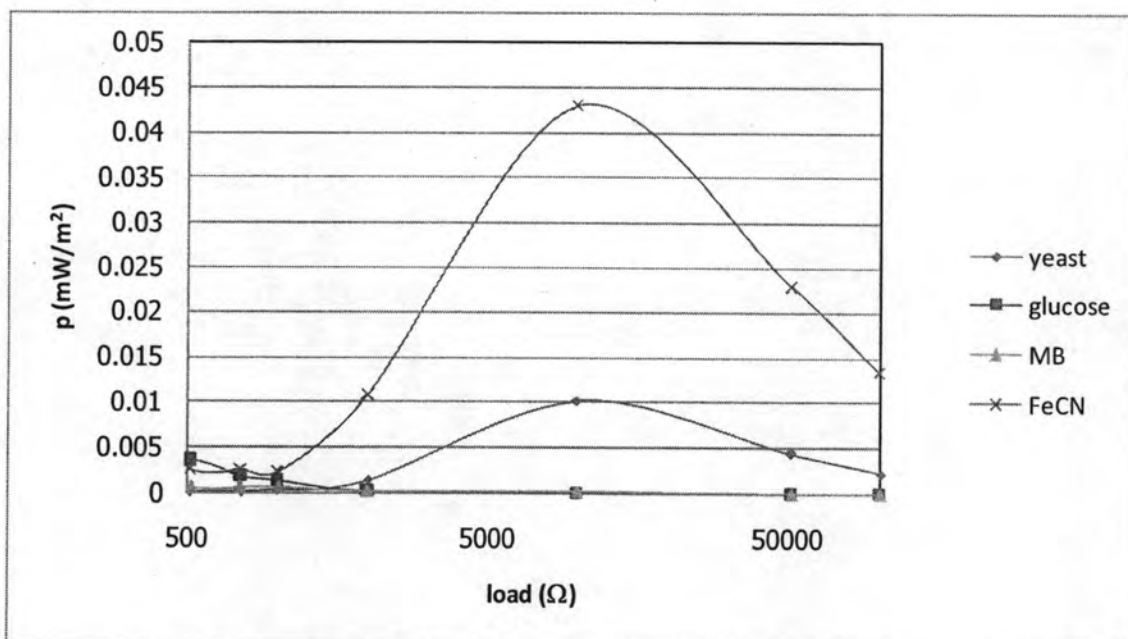


Fig. A2 Power density-load of each component in the bio-fuel cell

Fig. A3 shows voltage-current generation characteristic when adding two components in the BFC. It was found that the voltage generation from yeast and glucose was zero. This result shows that if the BFC consists of yeast and glucose, the BFC can not generate the voltage and current. The effect of combination of glucose and MB gave 37 mV of open circuit voltage and low current density ($<1\text{mA}/\text{m}^2$). When using ferricyanide with glucose or methylene blue, the open circuit voltage and current density of the BFC were in the range of 145-152 mV and $1.5\text{-}2\text{ mA}/\text{m}^2$ ($5.1\text{-}10\text{k}\Omega$), respectively.

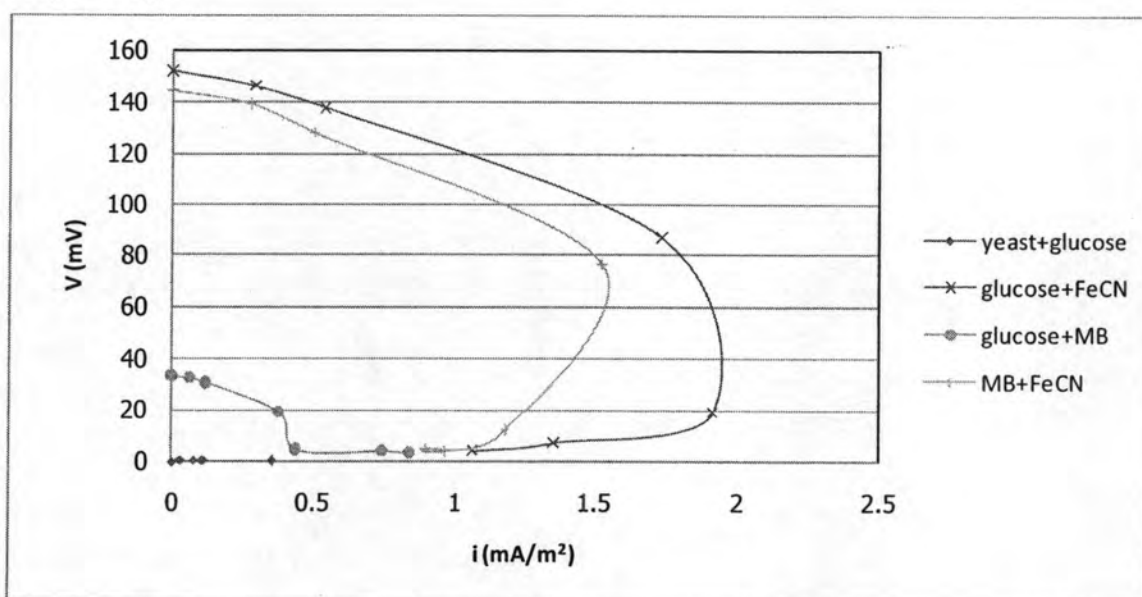


Fig. A3 Voltage-current density generation when adding two components in the bio-fuel cell

Fig. A4 shows power density-load when adding two components in the BFC. It shows that two components with ferricyanide gave the power density in the range of $0.12\text{--}0.15\text{ mW/m}^2$ ($10\text{k}\Omega$). For other components (without ferricyanide) gave the power density lower than 0.01 mW/m^2 ($10\text{k}\Omega$). Moreover, the effect of two components (with ferricyanide) increased the power density higher than only used ferricyanide.

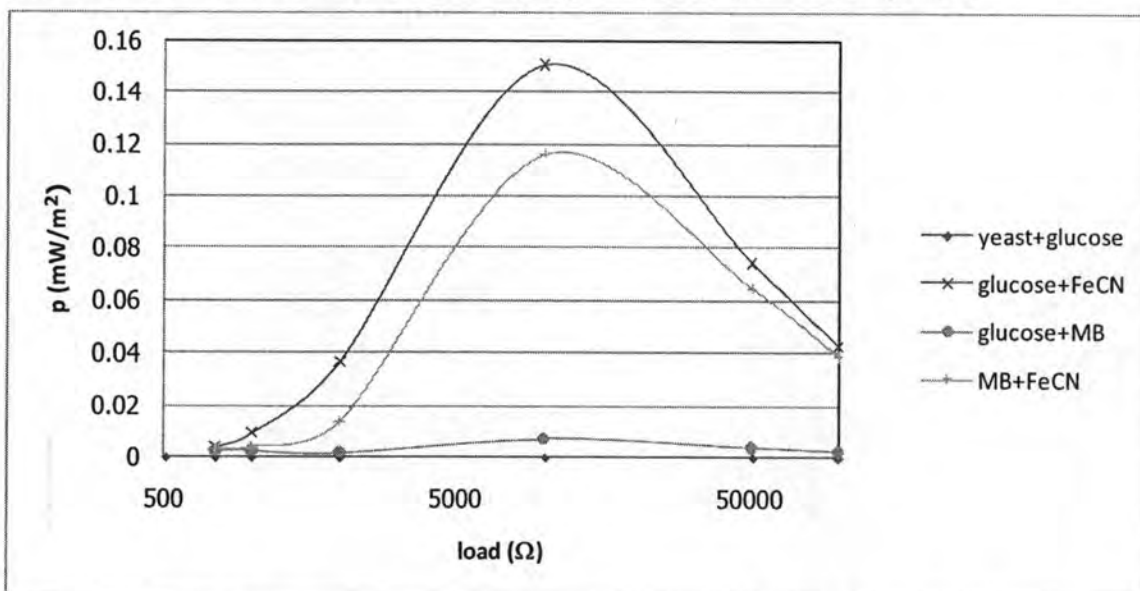


Fig. A4 Power density-load of two components in the bio-fuel cell

Fig. A5 shows voltage-current generation of 3-4 components in the BFC. In the study of 3 components, it was found that 3 components (with ferricyanide) gave the open circuit voltage and current density generation about 135 mV and 2 mA/m^2 ($5.1\text{--}10\text{k}\Omega$) that were nearly the effect of one and two components (with ferricyanide) as mentioned above. Therefore, the effect of ferricyanide increases the open circuit of the BFC. Considering 3 components (without ferricyanide), the result which is the effect of combination from glucose, yeast and MB shows that the open circuit voltage and current density generation of 3 components were up to 230 mV and 15 mA/m^2 (100Ω), respectively. It shows that MB can transfer electron from yeast to electrode. Therefore, the open circuit voltage and current density of the BFC in the anodic compartment depend on the glucose, yeast and MB. When ferricyanide (4 components) was added in the cathodic compartment, the open circuit voltage and current density of the BFC were up to 300 mV and 18 mA/m^2 (510Ω). It shows that the open circuit voltage generation of the BFC increased 70 mV (near the open circuit voltage of one component (with ferricyanide)) higher than 3 components (with glucose, yeast and MB). Therefore,

the open circuit voltage was the summation of 3 components (with glucose, yeast and MB) and ferricyanide in the anodic and cathodic compartment, respectively.

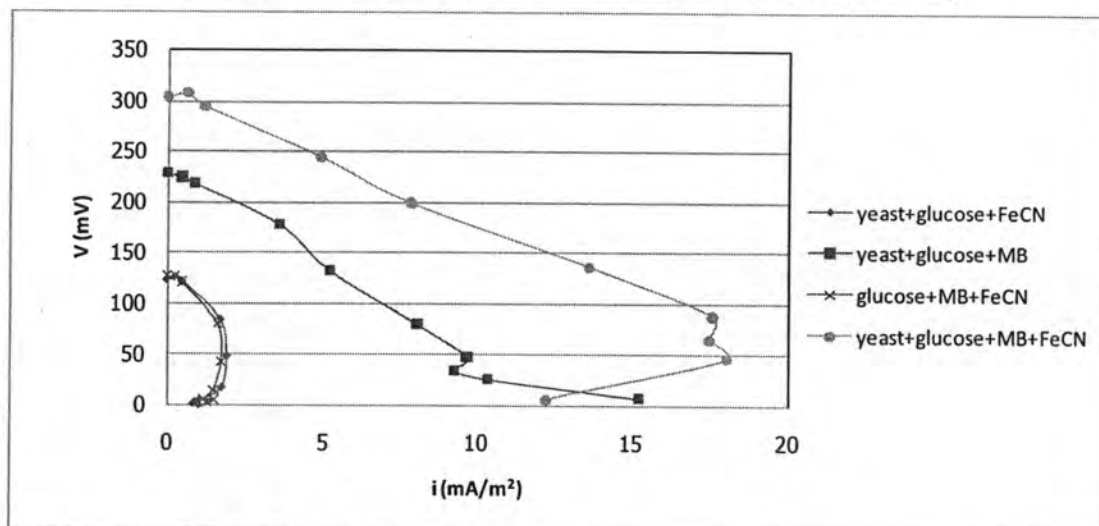


Fig. A5 Voltage-current density generation of 3-4 components in the bio-fuel cell

Fig. A6 shows power density-load of 3-4 components in the BFC. Three components (with ferricyanide) gave the power density in the range of 0.17 mW/m^2 ($10\text{k}\Omega$). This range of power density was nearly one and two components (with ferricyanide). The power density of 3 components (with glucose, yeast and MB) increased the power density generation up to 0.7 mW/m^2 ($5\text{k}\Omega$). When ferricyanide was added in these conditions, 1.8 mW/m^2 of power density was obtained from the BFC.

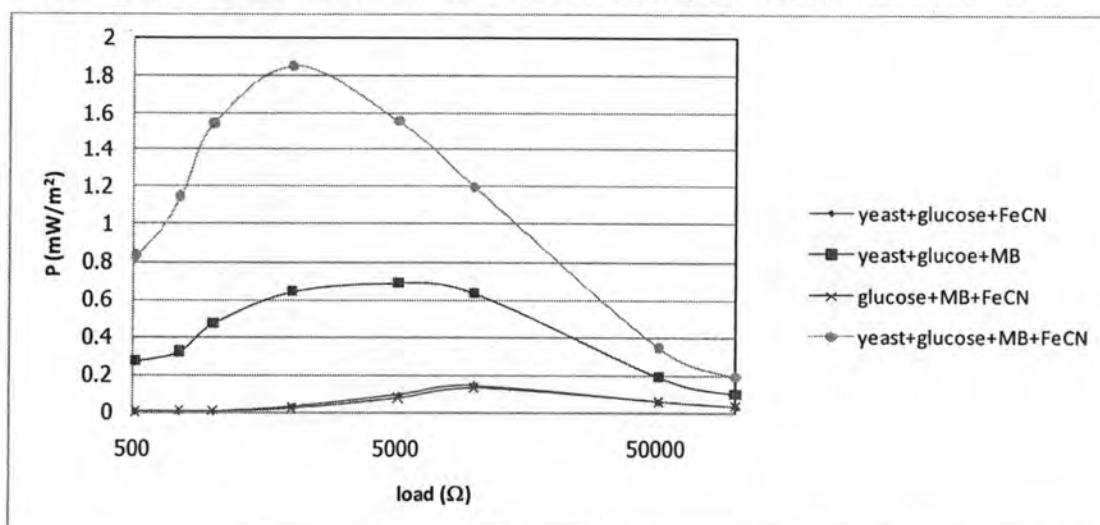


Fig. A6 Power density-load of 3-4 components in the bio-fuel cell

APPENDIX B

PUBLICATIONS AND PRESENTATIONS

- Ouitrakul, S., Sriyudthsak, M., and Kakizono, T., Preliminary Study on Bio-Fuel Cell using Baker Yeast. The 4th JSPS-NRCT Joint Seminar on Development of Thermotolerant Microbial Resources and Their Applications, Kyushu University, Fukuoka, Japan, 2004. November 7-10: 123
- Ouitrakul, S., Sriyudthsak, M., Preliminary Experiment on the Construction of Bio-Fuel Cell using Baker Yeast. 28th Electrical Engineering Conference, PearlVillage hotel, Phuket, Thailand, PW09, 2005. October 20-21: 617-620
- Ouitrakul, S., Sriyudthsak, M., and Kakizono, T., Effect of Electron Acceptor in Bio-Fuel Cell. Proceedings of the 1st IEEE International Conference on Nano/Micro Engineered and Molecular Systems, Zhuhai, China, 2006. January 18-21: 1432-1435
- Ouitrakul, S., Sriyudthsak, M., Effect of Mediator in Bio-Fuel Cell (ผลกระทบของตัวนำพาอิเล็กตรอนที่มีต่อเซลล์เพลิงชีวภาพ). 29th Electrical Engineering Conference, , Ambassador City Jomtien hotel, Pattaya, Thailand, PE30, 2006. November 9-10: 477-480
- Ouitrakul, S., Sriyudthsak, M., Charojrochkul, S. and Kakizono, T., Impedance Analysis of Bio-Fuel Cell Electrodes. Biosensors and Bioelectronics, 2007. 23: 721-727,
- Ouitrakul, S., Charojrochkul, S., Kakizono, T., and Sriyudthsak, M., Comparative Study of Five Electrodes for Bio-Fuel Cell. Proceedings of EEE FWS 2007, Montien Hotel, Bangkok, Thailand, EL06, 2007. November 22-23: 1-7
- Ouitrakul, S., Sriyudthsak, M., Charojrochkul, S. and Kakizono, T., การวิเคราะห์หิมพีแดนซ์ของอิเล็กโทรดในเซลล์เชื้อเพลิงชีวภาพ (Impedance Analysis of Electrodes in Bio-Fuel Cell). วารสารวิจัยและพัฒนา มจร. ปีที่ 31 ฉบับที่ 1 มกราคม – มีนาคม 2551

Vitae

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