

การวิเคราะห์สารประกอบคลอรีเนตเตตไฮโดรคาร์บอนบางชนิดในน้ำ

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วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต

ภาควิชาเคมี

บัณฑิตวิทยาลัย จุฬาลงกรณ์มหาวิทยาลัย

พ. ศ. 2535

ISBN 974-582-090-3

ลิขสิทธิ์ของบัณฑิตวิทยาลัย จุฬาลงกรณ์มหาวิทยาลัย

019588 1171114080

ANALYSIS OF SOME CHLORINATED HYDROCARBONS IN WATER BY

HEADSPACE TECHNIQUE



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A thesis Submitted in Partial Fulfillment of the Requirements

for the Degree of Master of Science

Department of Chemistry

Graduate School

Chulalongkorn University

1992

ISBN 974-582-090-3

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Thesis Title Analysis of Some Chlorinated Hydrocarbons in Water
by Headspace Technique.
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วุฒิชัย เย็นธงชัย : การวิเคราะห์ สารประกอบคลอรีเนเตด ไฮโดรคาร์บอนบางชนิด
ในน้ำ โดยเทคนิคเฮดสเปส (Analysis of some chlorinated hydrocarbons
in water by headspace technique) อาจารย์ที่ปรึกษา :
ดร. สิทธิชัย สิทธิพัฒน์ไพบูลย์, 179 หน้า ISBN 974-582-090-3



เทคนิคเฮดสเปส ได้รับการพัฒนาขึ้นเพื่อใช้วิเคราะห์ สารประกอบคลอรีเนเตด ไฮโดร-
คาร์บอน ที่ระเหยได้ง่ายในตัวอย่างน้ำ เช่น เมธิลคลอไรด์ คลอโรฟอร์ม คาร์บอนเตตระคลอไรด์
1,1,1-ไตรคลอโรเอเทน และ ไตรคลอโรเอธิลีน โดยทำการศึกษาและประเมินค่าปัจจัยต่างๆ ที่มี
ผลต่อความไว (sensitivity) และ ประสิทธิภาพของการสกัด (percent
recovery) ได้แก่ เวลาที่ระบบเข้าสู่สมดุล อุณหภูมิ อัตราส่วนของปริมาตรระหว่างน้ำต่อ
อากาศ ปริมาณของเฮดสเปสแก๊สที่ใช้ในการวิเคราะห์และการเติมเกลือโซเดียมคลอไรด์ โซเดียมซัลเฟต
และ แคลเซียมคาร์บอเนต ผลการศึกษาพบว่า การใช้อุณหภูมิ 60 °C เป็นเวลา 60 นาที ด้วย
อัตราส่วนน้ำต่ออากาศ 30 : 30 ในขวดขนาด 60 มิลลิลิตร ปริมาตรของเฮดสเปสแก๊สที่นำเข้าไป
สู่เครื่องแก๊สโครมาโตกราฟ 1.50 มิลลิลิตร และใช้เกลือโซเดียมซัลเฟต 13.00 กรัม เป็นสภาวะ
ที่เหมาะสมที่สุดของการวิเคราะห์โดยเทคนิคเฮดสเปส ซึ่งสามารถที่จะนำมาใช้กับการวิเคราะห์ สารประกอบ
คลอรีเนเตดไฮโดรคาร์บอน ที่ระเหยได้ง่ายในตัวอย่างน้ำได้ เทคนิคนี้สามารถตรวจวัดสารประเภทนี้ใน
ตัวอย่างน้ำได้ที่ระดับต่ำที่สุดซึ่งต่ำกว่า 0.50 ไมโครกรัมต่อลิตร (ppb) โดยมีประสิทธิภาพการสกัด
สารประกอบคลอรีเนเตด ไฮโดรคาร์บอนนี้ อยู่ในช่วง 54.22 ถึง 90.81 เปอร์เซ็นต์ และมีค่าเบี่ยง
เบนมาตรฐานสัมพัทธ์ (%RSD) ระหว่าง 1.04 ถึง 4.81 เปอร์เซ็นต์ นอกจากนี้ยังได้ศึกษา
ความถูกต้องในการวิเคราะห์ด้วยเทคนิคนี้โดยวิธีอินเทอร์เนลสแตนดาร์ดไอเซชัน (internal stander-
dization method) พบว่า เปอร์เซ็นต์ของความผิดพลาดอยู่ในช่วง 0.31 ถึง 6.44 เปอร์เซ็นต์
จากการเก็บตัวอย่างน้ำประปาและน้ำจากสระน้ำ 6 แห่ง มาตรวจวิเคราะห์ด้วยเทคนิคนี้พบว่า ตัวอย่าง
น้ำ ที่เก็บมาทั้งหมด มีสาร เมธิลคลอไรด์ และ คลอโรฟอร์ม อยู่ในช่วง 10.22 - 448.68 และ
0.22 - 31.86 ส่วนในพันล้านส่วน ตามลำดับ

ภาควิชาเคมี.....
สาขาวิชาเคมีวิเคราะห์.....
ปีการศึกษา2535.....

ลายมือชื่อนิสิต
ลายมือชื่ออาจารย์ที่ปรึกษา
ลายมือชื่ออาจารย์ที่ปรึกษาร่วม



C225251 : MAJOR CHEMISTRY
KEY WORD: HEADSPACE / CHLORINATED HYDROCARBON

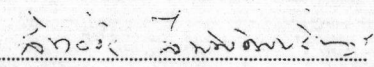
WUTICHAI YENTONGCHAI : ANALYSIS OF SOME CHLORINATED HYDROCARBONS
IN WATER BY HEADSPACE TECHNIQUE. THESIS ADVISER :
Dr. SITTICHAI LEEPIPATPIBOON, Ph.D. 179 pp. ISBN 974-582-090-3

In this study, a headspace technique was developed for the determination of some volatile chlorinated hydrocarbons i.e., methylene chloride, chloroform, carbon tetrachloride, 1,1,1-trichloroethane and trichloroethylene in water samples. Various factors having effect on the sensitivity and the percent recovery i.e., equilibration time, temperature, liquid to gas phase volume ratio, injection volume and salting out with sodium chloride, sodium sulfate and calcium carbonate were studied and evaluated. The temperature of 60 °C, equilibration time of 60 minutes, the liquid to gas phase volume ratio of 30 : 30 in 60 mL serum vial, 1.50 mL of injection volume, and salting out with 13.00 g of anhydrous sodium sulfate were chosen as the optimal headspace analysis condition for the analysis of the volatile chlorinated hydrocarbons in water samples. The minimum detectable level of this technique was lower than 0.50 ppb ($\mu\text{g/L}$) for all studied compounds and the percent recoveries of the chlorinated hydrocarbons were in the range of 54.22 - 90.81 % with $\pm 1.04 - 4.81$ % RSD. The accuracy of this technique was also studied by means of internal standardization method and the percent errors were in the range of 0.31 - 6.44 % at the ppb level of concentration. Moreover, the developed technique was also applied to analyze tap water and pool water collected from six different places. Results showed that all water samples appeared to have methylene chloride and chloroform in the range of 10.22 - 448.68 and 0.22 - 31.86 ppb, respectively.

ภาควิชา.....เคมี.....

ลายมือชื่อผู้นิสิต.....

สาขาวิชา.....เคมีวิเคราะห์.....

ลายมือชื่ออาจารย์ที่ปรึกษา.....

ปีการศึกษา.....2535.....

ลายมือชื่ออาจารย์ที่ปรึกษาร่วม.....

ACKNOWLEDGEMENTS

The author would like to take this opportunity to convey the most sincere thanks and appreciation to his adviser Dr. Sittichai Leepipatpiboon. This thesis would not be completed without his kindness, guidance, and assistance.

He is also grateful to Dr. Varaporn Leepipatpiboon for her helpful and invaluable suggestions. In addition, appreciations are also expressed to his colleagues for their encouragement.

He would also like to thank the Graduate School, Chulalongkorn University for the financial support for the preparation of this thesis, to the thesis committee for their comments and to the staff of the instrumental laboratory in Chemistry Building 3, Department of Chemistry, Faculty of Science, Chulalongkorn University for his permission of using the gas chromatograph.

Finally, the author is greatly indebted and deeply grateful to his parents and the family members for their encouragement and understanding throughout the entire course of study.



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