

การสังเคราะห์และสมบัติของเปปไทค์นิวคลีอิกแอนด์ชานิดใหม่ที่มีโซ่อ้างแบบไฮโดรฟลิก

นาย ชาญชัย คงดีเสมอ



วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิทยาศาสตรมหาบัณฑิต

สาขาวิชาเคมี ภาควิชาเคมี

คณะวิทยาศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย

ปีการศึกษา 2542

ISBN 974-334-224-9

ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

SYNTHESIS OF NOVEL PEPTIDE NUCLEIC ACIDS CARRYING HYDROPHILIC  
SIDE-CHAIN AND THEIR PROPERTIES

Mr. Chanchai Khongdeesameor

A Thesis Submitted in Partial Fulfillment of the Requirements  
for the Degree of Master of Science in Chemistry

Department of Chemistry

Faculty of Science

Chulalongkorn University

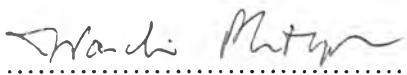
Academic Year 1999

ISBN 974-334-224-9

**Thesis Title** SYNTHESIS OF NOVEL PEPTIDE NUCLEIC ACIDS CARRYING HYDROPHILIC SIDE-CHAIN AND THEIR PROPERTIES  
**By** Mr. Chanchai Khongdeesameor  
**Department** Chemistry  
**Thesis Advisor** Assistant Professor Tirayut Vilaivan, D. Phil.

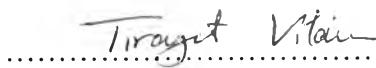
---

Accepted by the Faculty of science, Chulalongkorn University in Partial Fulfillment of the Requirements for the Master's Degree

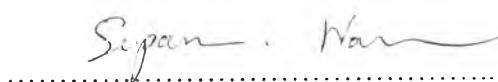
 ..... Dean of Faculty of science  
(Associate Professor Wanchai Phothiphichitr, Ph. D.)

Thesis Committee

 ..... Chairman  
(Associate Professor Udom Kokpol, Ph. D.)

 ..... Thesis Advisor  
(Assistant Professor Tirayut Vilaivan, D. Phil.)

 ..... Member  
(Buncha Pulpoka, Ph. D.)

 ..... Member  
(Assistant Professor Supasorn Wanichwecharungruang, Ph. D.)

ชาญชัย คงคีเสมอ : การสังเคราะห์และสมบูติของเปปไทด์นิวคลีอิกแอซิดชนิดใหม่ที่มีโซ่อ้าง  
แบบไฮdroฟิลิก : (SYNTHESIS OF NOVEL PEPTIDE NUCLEIC ACIDS CARRYING  
HYDROPHILIC SIDE-CHAIN AND THEIR PROPERTIES): ผศ.ดร. ธีรยุทธ วิไลวัลย์;  
197หน้า. ISBN 974-334-224-9

ได้สังเคราะห์เปปไทด์นิวคลีอิกแอซิดแบบมีคอนฟอร์เมชันที่ถูกจำกัด(ซีพีเอ็นเอ)ที่นี่  
โครงสร้างต่างกัน 3 แบบ ซีพีเอ็นเอนี้เป็นสารเดี่ยวนแบบดีเอ็นเอโดยแทนที่ดีอโกรซีไรโนสฟอสเฟต  
ของดีเอ็นเอหง磋商 ด้วยไดเปปไทด์โนโนเมอร์ของดีและแอลซีริลโปรดีน หรือดีอโกรซีไกลซิล  
โปรดีนที่เขื่อมต่อกับไทมินที่เป็นนิวคลีโอเบสที่ตำแหน่งที่ 4 ของวง โครงสร้างที่แข็งเกร็งของ  
ซีพีเอ็นเอไดจากการเชื่อมวงโปรดีนที่มีสเตอโรโลเคมีแบบ “ซีส-ดี” หรือ 2R, 4R สถาบันกรดอะมิโน  
ซีรีนทั้งชนิดแอลหรือดีหรือหมู่อะมิโนเอชิลด้วยวัตถุประสงค์ที่จะเพิ่มความสามารถในการละลายน้ำ  
ของซีพีเอ็นเอ ไดเปปไทด์โนโนเมอร์นี้ถูกสังเคราะห์ขึ้นจาก ซีส-4-ไฮดรอกซี-ดี-โปรดีน ที่ถูกแทนที่  
ด้วยไทมินโดยอาศัยปฏิกิริยามิชูโนบุและจึงนำไปทำปฏิกิริยาคู่ความต่อเข้ากับชิ้นส่วนสเปเซอร์  
ในกรณีของซีรีนจะใช้ Fmoc-Ser(O'Bu)-OH ซึ่งสามารถถูกควบเข้ากับสารมัธยัณฑ์ไดโดยตรง ในขณะ  
ทำการต่อหมู่ดีอโกรซีไกลซิลสเปเซอร์ทำไดโดยใช้แอลซีริດินที่ถูกกระตุ้น การโอลิโกเมอร์ไรซ์ให้  
ซีพีเอ็นเอมีความยาว 10 หน่วย ทำไดโดยใช้ยุทธวิธีการคู่ความแบบชิ้นส่วนโดยใช้การปักปูองแบบ  
Fmoc/O'Bu และใช้การกระตุ้นด้วยเพนตะฟลูออฟินิลเอสเทอร์ หรือ HBTU นอกจากนี้ยังไดศึกษา  
เสถียรภาพในการจับยึดเบื้องต้นระหว่างซีพีเอ็นเอที่ไดกับโอลิโกแอคตีโนลิกแอซิดที่ผ่านการติดผลัก  
แบบฟลูออเรสเซนต์ ด้วยเทคนิคเจลอิเล็กโทร โฟลิชิสเพลที่ไดแสดงให้เห็นว่าไม่พบรการจับยึดกันแม้จะ  
ใช้ความเข้มข้นซีพีเอ็นเอสูงก็ตาม

ภาควิชา.....เคมี..... ลายมือชื่อนิสิต..... จำกัด.....  
สาขาวิชา.....เคมี..... ลายมือชื่ออาจารย์ที่ปรึกษา.....  
ปีการศึกษา.....2542..... ลายมือชื่ออาจารย์ที่ปรึกษาร่วม.....

## 4072241923 : MAJOR CHEMISTRY

KEY WORD: PNA / peptide nucleic acid / DNA mimics

CHANChAI KHONGDEESAMEOR : SYNTHESIS OF NOVEL PEPTIDE NUCLEIC ACIDS CARRYING HYDROPHILIC SIDE-CHAIN AND THEIR PROPERTIES.

THESIS ADVISOR: ASSIST. PROF. TIRAYUT VILAIVAN, D. Phil.

197 pp. ISBN 974-334-224-9

Three different conformationally constraint peptide nucleic acids (cPNA) have been synthesized. These cPNA are DNA mimics in which the entire deoxyribose phosphate is replaced with L- and D-serylproline or deoxyglycylproline dipeptide units bearing thymine as nucleobase at C-4 position. Their rigid backbones were derived from "cis-D" or (2R, 4R) absolute stereochemistry on proline ring alternating with D or L serine or aminoethyl spacers with the aim to improve water solubility of the resulting cPNA. Dipeptide monomers were synthesized from *cis*-4-hydroxy-D-proline derivatized with thymine by Mitsunobu reaction followed by coupling with the spacer moiety. The serine spacer were introduced as Fmoc-Ser(O<sup>t</sup>Bu)-OH which could be directly coupled to the intermediate by DCC coupling whereas the deoxyglycyl spacer was introduced by alkylation of the intermediate with activated aziridine. Oligomerization of the cPNA up to 10-mers were achieved by solid phase peptide synthesis methodology employing Fmoc/O<sup>t</sup>Bu fragment coupling strategy using pentafluorophenyl ester or HBTU activation. Preliminary binding stability of these cPNA toward fluorescence labelled oligoadenylic acid F(dA<sub>10</sub>) were studied by gel electrophoresis. Lacking of binding of all cPNA oligomers was observed even at high PNA concentration.

ภาควิชา.....เคมี..... ลายมือชื่อนิสิต.....  
สาขาวิชา.....เคมี..... ลายมือชื่ออาจารย์ที่ปรึกษา.....  
ปีการศึกษา.....2542..... ลายมือชื่ออาจารย์ที่ปรึกษาร่วม.....



## ACKNOWLEDGEMENTS

I would like to thank the following: the National Science and Technology Development Agency (NSTDA) for 1997 Local Graduate Scholarship (LGS) and central office project RDE-Co grant number: CO-B-06-22-09-001 for 1999 research assistant scholarship; Assistant Professor Dr. Tirayut Vilaivan, my thesis advisor, for good advice and some financial support throughout the course of this research; Associate Professor Dr. Udom Kokpol, Dr. Buncha Pulpoka and Assistant Professor Dr. Supasorn Wanichwecharungruang, thesis examiners, for all valuable comments regarding this thesis; Dr. Aroonsiri Shitangkoon for her help involving HPLC experiments; Dr. Pongchai Hanyuthanakorn for some chemicals and materials for the gel electrophoresis experiment and invaluable suggestion on biological assay; Chulalongkorn Research Equipment Centre, Ms. Amporn Ungpakornkaew for Elemental analysis, Mrs Wanna Sririnnut and Ms Wanwimon Thabdee for 500 MHz NMR experiment; Institute of Biotechnology and Genetic Engineering for UV-spectrophotometer service and Ms. Nanthiga Panchan for MALDI-TOF mass spectrometry. Department of Chemistry, Chulalongkorn University for laboratory facilities, chemicals and equipment. I wish to express my special thank also to Mr. Sutirawat Samingprai for some parts of laboratory setup, without his help I would not have a chance to be here and writing this thesis. Finally, I would like to pay obeisance to my parent who had encouraged me with sincere throughout the research courses and to my family for introducing the will power. Thanks God for giving power, hope and confidence during the research.

## CONTENTS

	page
Abstract in Thai.....	iv
Abstract in English.....	v
Acknowledgements.....	vi
List of Figures.....	ix
List of Tables.....	xviii
List of Abbreviations.....	xix
CHAPTER I INTRODUCTION.....	1
1.1 Literature reviews.....	1
1.1.1 Modified oligonucleotides.....	2
1.1.2 Backbone modifications.....	4
1.1.3 Peptide Nucleic Acid (PNA).....	5
1.1.4 Hybridizations and their secondary structures.....	6
1.1.5 PNA structural modifications.....	9
1.1.6 PNA applications.....	15
1.1.7 Problems of PNA and their solutions.....	18
1.2 Objectives of this research.....	20
CHAPTER II EXPERIMENTAL.....	21
2.1 General.....	21
2.2 Materials.....	22
2.3 Synthesis of cPNA monomers.....	23
2.3.1 Synthesis of intermediate.....	23
2.3.2 Synthesis of serylproline cPNA monomers.....	31
2.3.3 Synthesis of deoxyglycylproline cPNA monomer.....	42
2.4 Oligomerization of cPNA.....	52
2.4.1 Preparation of the reaction pipette and apparatus for solid phase synthesis.....	52

	page
2.4.2 Solid phase peptide synthesis of both L- and D-serylproline cPNA according to Fmoc/OtBu dipeptide fragment strategy.....	53
2.4.2.1 The decamers of L-serylproline cPNA.....	53
2.4.2.2 The decamers of D-serylproline cPNA.....	56
2.4.3 Solid phase peptide synthesis of deoxyglycylproline cPNA according to Fmoc Chemistry.....	57
2.3 Biophysical studies.....	59
CHAPTER III RESULTS AND DISCUSSION.....	61
3.1 Synthesis of cPNA monomers.....	61
3.1.1 Synthesis of intermediate.....	61
3.1.2 Synthesis of serylproline cPNA monomers.....	71
3.1.3 Synthesis of deoxyglycylproline cPNA monomer.....	75
3.2 Oligomerization of cPNA.....	87
3.2.1 The decamers of deoxyglycylproline cPNA.....	87
3.2.2 The decamers of serylproline cPNAs.....	96
3.3 Biophysical studies.....	104
CHAPTER IV CONCLUSION.....	108
REFERENCES.....	111
APPENDIX.....	120
VITA.....	197

## LIST OF FIGURES

	page
Figure 1.1 A schematic diagram illustrates the gene expression inhibition by antisense or antigene approach.....	1
Figure 1.2 Watson-Crick base pairing AT and CG.....	2
Figure 1.3 Structures of modified phosphorus-containing oligonucleotides.....	3
Figure 1.4 Some modified backbone oligonucleotides.....	4
Figure 1.5 Structures of Peptide Nucleic Acid.....	5
Figure 1.6 Hydrogen bonding via Watson-Crick and Hoogsteen base pairing and “P-loop” triple helix formation of (PNA) <sub>2</sub> DNA .....	7
Figure 1.7 Schematic drawing of complexes formed upon treating double strand DNA with PNAs.....	8
Figure 1.8 Three dimensional structures of PNA complexes.....	9
Figure 1.9 Structures of extended modified PNAs, PNA with ethylene linker from base to backbone, Reversed amide PNA and “retro-inverso PNA” PNA with moving carbonyl of acetyl linker to backbone.....	10
Figure 1.10 Structures of non-natural nucleobases.....	11
Figure 1.11 Structures of some chiral PNAs.....	13
Figure 1.12 Structures of some conformationally constraint PNAs and PHONA.....	14
Figure 1.13 Effect of PNA to RNA transcription.....	16
Figure 1.14 Some applications of PNA in diagnostics and biotechnology.....	17
Figure 1.15 Schematic of the “Achilles heel” approach or genome rare cutter.....	18
Figure 1.16 Structures of DNA, glycylproline cPNA, serylproline cPNA and deoxy-glycylproline cPNA.....	20
Figure 2.1 A diagram showing the manual technique for solid phase peptide synthesis.....	52
Figure 3.1 Synthesis of intermediate [Boc-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-ODpm, (7)].....	61
Figure 3.2 Epimerization of <i>trans</i> -4-hydroxy-L-proline into <i>cis</i> -4-hydroxy-D-proline via enolization mechanism.....	62

Figure 3.3 Comparison of $^1\text{H}$ NMR spectra ( $\text{D}_2\text{O}$ , 200 MHz) between <i>trans</i> -4-hydroxy-L-proline and <i>cis</i> -4-hydroxy-D-proline.....	64
Figure 3.4 Reaction mechanism for synthesis of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-hydroxyproline ( <b>2</b> ).....	66
Figure 3.5 The two resonance forms of amide bond and restricted rotation around C-N bond in Boc-D-Pro( <i>cis</i> -4-OH)-OH ( <b>2</b> ).....	66
Figure 3.6 Reaction mechanism of Boc-D-Pro( <i>cis</i> -4-OH)-OH ( <b>2</b> ) with diphenyldiazomethane ( <b>3</b> ).....	67
Figure 3.7 Introducing nucleobase into proline ring <i>via</i> tosylation and $\text{S}_{\text{N}}^2$ displacement.....	68
Figure 3.8 Synthesis of <i>trans</i> -4-formate ester ( <b>5a</b> ) and Boc-D-Pro( <i>trans</i> -4-OH)-ODpm ( <b>5b</b> ).....	69
Figure 3.9 Synthesis of intermediate ( <b>7</b> ) <i>via</i> Mitsunobu reaction.....	70
Figure 3.10 Two possible ways to join intermediate ( <b>7</b> ) and a spacer amino acid together.....	71
Figure 3.11 Racemerization of a peptide by oxazolone mechanism.....	71
Figure 3.12 Coupling mechanism for synthesis of serylproline dipeptide monomers [( <b>8</b> ) and ( <b>9</b> )].....	73
Figure 3.13 Selective deprotection of Dpm ester [( <b>8</b> ) or ( <b>9</b> )] into free acid derivative [( <b>10</b> ) or ( <b>11</b> )] and synthesis of pentafluorophenyl ester [( <b>12</b> ) or ( <b>13</b> )].....	74
Figure 3.14 Possible routes to synthesize dipeptide carrying reduced amide bond.....	76
Figure 3.15 Synthesis of <i>N</i> -nosylethanolamine ( <b>17</b> ) and <i>N</i> -nosylaziridine ( <b>18</b> ).....	79
Figure 3.16 Synthesis of Ns-( $\psi$ -CH <sub>2</sub> )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-ODpm ( <b>19</b> ).....	80
Figure 3.17 Synthesis of Boc/Ns-( $\psi$ -CH <sub>2</sub> )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-Dpm ( <b>20</b> ).....	80
Figure 3.18 Synthesis of Boc-( $\psi$ -CH <sub>2</sub> )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-ODpm ( <b>21</b> ).....	81
Figure 3.19 Synthesis of Fmoc-( $\psi$ -CH <sub>2</sub> )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-ODpm ( <b>22</b> ).....	82
Figure 3.20 Synthesis of Fmoc-( $\psi$ -CH <sub>2</sub> )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-OH.HCl ( <b>23</b> ).....	83

	page
Figure 3.21 $^1\text{H}$ NMR spectrum of Fmoc-( $\psi\text{-CH}_2$ )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-OH.HCl <b>(23)</b> .....	84
Figure 3.22 HPLC chromatogram of Fmoc-( $\psi\text{-CH}_2$ )Gly-D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-OH.HCl <b>(23)</b> .....	85
Figure 3.23 A possible mechanism of cyclization of Pfp ester of Fmoc-( $\psi\text{-CH}_2$ )Gly- D-Pro( <i>cis</i> -4-T <sup>Bz</sup> )-OH.HCl <b>(23)</b> .....	86
Figure 3.24 The protocol for solid phase synthesis of deoxyglycylproline cPNA <b>(24)</b> ....	88
Figure 3.25 The mechanism for coupling or anchoring <i>via</i> HBTU activation.....	89
Figure 3.26 Mechanism for deprotection of Fmoc protecting group resin-bound peptide.	90
Figure 3.27 A possible mechanism for cyclizing of deoxyglycylproline monomer <b>(23)</b> into ketopiperazine.....	93
Figure 3.28 Reverse phase HPLC chromatogram of deoxyglycylproline cPNA(24).....	94
Figure 3.29 A possible mechanism for debenzoylation of $N^3$ -benzoylthymine by piperidine.....	95
Figure 3.30 The protocol for solid phase synthesis of serylproline cPNA <b>(25)</b> and <b>(26)</b> ... 97	97
Figure 3.31 Coupling mechanism of Pfp active ester <b>(12)</b> and <b>(13)</b> .....	98
Figure 3.32 Reverse phase HPLC chromatogram of 10-mer L-serylproline cPNA <b>(25)</b> ..	101
Figure 3.33 Reverse phase HPLC chromatogram of 10-mer D-serylproline cPNA <b>(26)</b> ..	102
Figure 3.34 Gel electrophoresis results of 1:1 stoichiometric ratio binding.....	105
Figure 3.35 Gel electrophoresis result with stoichiometric variation.....	106
Figure 1 $^1\text{H}$ NMR spectrum ( $\text{D}_2\text{O}$ , 200 MHz) of <i>cis</i> -4-hydroxy-D-proline <b>(1b)</b> .....	121
Figure 2 $^{13}\text{C}$ NMR spectrum ( $\text{D}_2\text{O}$ , 50.28 MHz) of <i>cis</i> -4-hydroxy-D-proline <b>(1b)</b> .....	122
Figure 3 IR spectrum (KBr) of <i>cis</i> -4-hydroxy-D-proline <b>(1b)</b> .....	123
Figure 4 $^1\text{H}$ NMR spectrum ( $\text{DMSO-}d_6$ , 200 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4- hydroxy-D-proline <b>(2)</b> .....	124
Figure 5 $^{13}\text{C}$ NMR spectrum ( $\text{DMSO-}d_6$ , 50.28 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-hydroxy-D-proline <b>(2)</b> .....	125
Figure 6 IR spectrum (KBr) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-hydroxy-D-proline <b>(2)</b> ....	126

page

Figure 7 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 200 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-hydroxy-D-proline diphenylmethyl ester ( <b>4</b> ).....	127
Figure 8 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 50.28 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-hydroxy-D-proline diphenylmethyl ester ( <b>4</b> ).....	128
Figure 9 IR spectrum (KBr) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-hydroxy-D-proline diphenylmethyl ester ( <b>4</b> ).....	129
Figure 10 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 200 MHz) of <i>trans</i> -4-formate ester ( <b>5a</b> ).....	130
Figure 11 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 50.28 MHz) of <i>trans</i> -4-formate ester ( <b>5a</b> ).....	131
Figure 12 IR spectrum (KBr) of <i>trans</i> -4-formate ester ( <b>5a</b> ).....	132
Figure 13 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 200 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>trans</i> -4-hydroxy-D-proline diphenylmethyl ester ( <b>5b</b> ).....	133
Figure 14 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 50.28 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>trans</i> -4-hydroxy-D-proline diphenylmethyl ester ( <b>5b</b> ).....	134
Figure 15 IR spectrum (KBr) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>trans</i> -4-hydroxy-D-proline diphenylmethyl ester ( <b>5b</b> ).....	135
Figure 16 $^1\text{H}$ NMR spectrum ( $\text{DMSO}-d_6$ , 200 MHz) of 3-benzoyl-5-methyl-( <i>1H</i> )-pyrimidine-2,4-dione ( $N^3$ -benzoylthymine) ( <b>6</b> ).....	136
Figure 17 $^{13}\text{C}$ NMR spectrum ( $\text{DMSO}-d_6$ , 50.28 MHz) of 3-benzoyl-5-methyl-( <i>1H</i> )-pyrimidine-2,4-dione ( $N^3$ -benzoylthymine) ( <b>6</b> ).....	137
Figure 18 IR spectrum (KBr) of 3-benzoyl-5-methyl-( <i>1H</i> )-pyrimidine-2,4-dione ( $N^3$ -benzoylthymine) ( <b>6</b> ).....	138
Figure 19 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester (Intermediate) ( <b>7</b> ).....	139
Figure 20 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester (Intermediate) ( <b>7</b> ).....	140
Figure 21 IR spectrum (KBr) of <i>N</i> - <i>tert</i> -butoxycarbonyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester (Intermediate) ( <b>7</b> ).....	141

	page
Figure 22 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>8</b> ).....	142
Figure 23 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>8</b> ).....	143
Figure 24 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>8</b> ).....	144
Figure 25 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>9</b> ).....	145
Figure 26 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>9</b> ).....	146
Figure 27 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D- proline diphenylmethyl ester ( <b>9</b> ).....	147
Figure 28 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 200 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D- proline ( <b>10</b> ).....	148
Figure 29 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline ( <b>10</b> ).....	149
Figure 30 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 200 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D- proline ( <b>11</b> ).....	150
Figure 31 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline ( <b>11</b> ).....	151
Figure 32 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester ( <b>12</b> ).....	152

	page
Figure 33 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester ( <b>12</b> ).....	153
Figure 34 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl- <i>O</i> - <i>tert</i> -butyl-L-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester ( <b>12</b> ).....	154
Figure 35 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester ( <b>13</b> ).....	155
Figure 36 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester ( <b>13</b> ).....	156
Figure 37 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl- <i>O</i> - <i>tert</i> -butyl-D-seryl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester ( <b>13</b> ).....	157
Figure 38 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>14</b> ).....	158
Figure 39 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>14</b> ).....	159
Figure 40 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>14</b> ).....	160
Figure 41 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline ( <b>15</b> ).....	161
Figure 42 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline ( <b>15</b> ).....	162
Figure 43 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline ( <b>15</b> ).....	163

	page
Figure 44 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester (16).....	164
Figure 45 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxy-carbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester (16).....	165
Figure 46 IR spectrum (KBr) of <i>N</i> -( <i>N</i> -fluoren-9-ylmethoxycarbonyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline pentafluorophenyl ester (16).....	166
Figure 47 $^1\text{H}$ NMR spectrum ( $\text{DMSO}-d_6$ , 500 MHz) of <i>N</i> -(2-hydroxyethyl)-4-nitrobenzenesulfonamide (17).....	167
Figure 48 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -(2-hydroxyethyl)-4-nitrobenzenesulfonamide (17).....	168
Figure 49 IR spectrum (KBr) of <i>N</i> -(2-hydroxyethyl)-4-nitrobenzenesulfonamide (17).....	169
Figure 50 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -4-nitrobenzenesulfonyl-aziridine (18).....	170
Figure 51 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -4-nitrobenzenesulfonyl-aziridine (18).....	171
Figure 52 IR spectrum (KBr) of <i>N</i> -4-nitrobenzenesulfonylaziridine (18).....	172
Figure 53 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -2-(4'-nitrobenzenesulfonamidoethyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethylester (19)....	173
Figure 54 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -2-(4'-nitrobenzenesulfonamidoethyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester (19).....	174
Figure 55 IR spectrum (KBr) of <i>N</i> -2-(4'-nitrobenzenesulfonamidoethyl)- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester (19).....	175
Figure 56 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -2-( <i>N</i> - <i>tert</i> -butoxycarbonyl-amino, <i>N</i> -4-nitrobenzenesulfonamido)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester (20).....	176

	page
Figure 57 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -2-( <i>N</i> - <i>tert</i> -butoxycarbonyl-amino, 4-nitrobenzenesulfonamido)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>20</b> ).....	177
Figure 58 IR spectrum (KBr) of <i>N</i> -2-( <i>N</i> - <i>tert</i> -butoxycarbonylamino, <i>N</i> -4-nitrobenzenesulfonamido)ethyl- <i>cis</i> -4-( $N^3$ -benzoyl thymine-1-yl)-D-proline diphenylmethyl ester ( <b>20</b> ).....	178
Figure 59 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -2-( <i>N</i> - <i>tert</i> -butoxycarbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymine-1-yl)-D-proline diphenylmethyl ester ( <b>21</b> )...179	
Figure 60 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65) of <i>N</i> -2-( <i>N</i> - <i>tert</i> -butoxycarbonylamino)-ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymine-1-yl)-D-proline diphenylmethyl ester ( <b>21</b> )....180	
Figure 61 IR spectrum (KBr) of <i>N</i> -2-( <i>N</i> - <i>tert</i> -butoxycarbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymine-1-yl)-D-proline diphenylmethyl ester ( <b>21</b> ).....181	
Figure 62 $^1\text{H}$ NMR spectrum ( $\text{CDCl}_3$ , 500 MHz) of <i>N</i> -2-(9-fluorenylmethoxy-carbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>22</b> ).....182	
Figure 63 $^{13}\text{C}$ NMR spectrum ( $\text{CDCl}_3$ , 125.65 MHz) of <i>N</i> -2-(9-fluorenylmethoxy-carbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester( <b>22</b> ).....183	
Figure 64 IR spectrum (KBr) of <i>N</i> -2-(9-fluorenylmethoxycarbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline diphenylmethyl ester ( <b>22</b> ).....184	
Figure 65 $^1\text{H}$ NMR spectrum ( $\text{DMSO}-d_6$ , 500 MHz) of <i>N</i> -2-(9-fluorenylmethoxy-carbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline hydrochloride ( <b>23</b> ).....185	
Figure 66 $^1\text{H}$ NMR spectrum ( $\text{DMSO}-d_6 + 1 \text{ drop D}_2\text{O}$ , 500 MHz) of <i>N</i> -2-(9-fluorenylmethoxycarbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline hydrochloride ( <b>23</b> ).....186	
Figure 67 $^{13}\text{C}$ NMR spectrum ( $\text{DMSO}-d_6$ , 125.65 MHz) of <i>N</i> -2-(9-fluorenylmethoxy carbonylamino)ethyl- <i>cis</i> -4-( $N^3$ -benzoylthymin-1-yl)-D-proline hydrochloride ( <b>23</b> ).....187	

	page
Figure 68 IR spectrum (KBr) of <i>N</i> -2-(9-fluorenylmethoxycarbonylamino)ethyl- <i>cis</i> -4-( <i>N</i> <sup>3</sup> -benzoylthymin-1-yl)-D-proline hydrochloride ( <b>23</b> ).....	188
Figure 69 MALDI-TOF mass spectrum of Fmoc-[( $\psi$ -CH <sub>2</sub> )Gly-D-Pro(T)] <sub>10</sub> -LysNH <sub>2</sub> or [Fmoc-ON ( <b>24</b> )].....	189
Figure 70 MALDI-TOF mass spectrum of Fmoc-[( $\psi$ -CH <sub>2</sub> )Gly-D-Pro(T <sup>Bz</sup> )]- [( $\psi$ -CH <sub>2</sub> )Gly-D-Pro(T)] <sub>9</sub> -LysNH <sub>2</sub> or [Fmoc/Bz-ON ( <b>24</b> )].....	190
Figure 71 MALDI-TOF mass spectrum of H-[( $\psi$ -CH <sub>2</sub> )Gly-D-Pro(T)] <sub>10</sub> -LysNH <sub>2</sub> or [Fmoc-OFF ( <b>24</b> )].....	191
Figure 72 MALDI-TOF mass spectrum of Fmoc-[L-Ser-D-Pro(T)] <sub>10</sub> -LysNH <sub>2</sub> or [Fmoc-ON ( <b>25</b> )].....	192
Figure 73 MALDI-TOF mass spectrum of H-[L-Ser-D-Pro(T)] <sub>10</sub> -LysNH <sub>2</sub> or [Fmoc-OFF( <b>25</b> )] .....	193
Figure 74 MALDI-TOF mass spectrum of Fmoc-[D-Ser-D-Pro(T)] <sub>10</sub> -LysNH <sub>2</sub> or [Fmoc-ON ( <b>26</b> )].....	194
Figure 75 MALDI-TOF mass spectrum of Fmoc-[D-Ser-D-Pro(T <sup>Bz</sup> )]-[D-ser-D- Pro(T)] <sub>9</sub> -LysNH <sub>2</sub> or [Fmoc/Bz-ON ( <b>26</b> )].....	195
Figure 76 MALDI-TOF mass spectrum of H-[D-Ser-D-Pro(T)] <sub>10</sub> -LysNH <sub>2</sub> or [Fmoc-OFF ( <b>26</b> )].....	196

## LIST OF TABLES

	page
Table 1.1 Effect on thermal stability per monomer ( $\Delta T_m/^\circ\text{C}$ ) for the PNA sequence H-gtagatcact-NH <sub>2</sub> incorporating three chiral monomers as compared to unmodified PNA.....	12
Table 3.1 Structure of 10-mer cPNAs.....	87
Table 3.2 The UV-absorption data and percent coupling efficiency of 10-mer deoxyglycylproline cPNA ( <b>24</b> ).....	92
Table 3.3 Mass spectral data of Fmoc-ON, Fmoc/Bz-ON and Fmoc-OFF of 10-mer deoxyglycylproline cPNA ( <b>24</b> ).....	96
Table 3.4 The UV-absorption data and percent coupling efficiency of 10-mer L-serylproline cPNA ( <b>25</b> ) and D-serylproline cPNA and ( <b>26</b> ).....	99
Table 3.5 Mass spectral data of Fmoc-ON and Fmoc-OFF of 10-mer L-serylproline cPNA ( <b>25</b> ).....	103
Table 3.6 Mass spectral data of Fmoc-ON, Fmoc/Bz-ON and Fmoc-OFF of 10-mer D-serylproline cPNA ( <b>26</b> ).....	103
Table 3.7 Representative code for 10-mer cPNA.....	104

## LIST OF ABBREVIATIONS

i) Nomenclature and abbreviations of nucleic acids and their constituents

DNA	deoxyribonucleic acids
RNA	ribonucleic acid
mRNA	messenger ribonucleic acid
A	Adenine
T	Thymine
C	Cytosine
G	Guanine
B	unspecified nucleobase
dA	2'-deoxy adenosine
dT	(2'-deoxy)thymine

ii) Nomenclature and Abbreviations of amino acids

Ala	Alanine
Gly	Glycine
Lys	Lysine
Pro	Proline
Ser	Serine

iii) Miscellaneous

Ac <sub>2</sub> O	acetic anhydride
Boc	<i>tert</i> -butoxycarbonyl
Boc <sub>2</sub> O	di- <i>tert</i> -butyl dicarbonate
br	broad
Bz	benzoyl

c	concentration
°C	degree celcius
calcd	calculated
CD	circular dichroism
CDCl <sub>3</sub>	deuterated chloroform
cm <sup>-1</sup>	unit of wave number
<sup>1</sup> H- <sup>1</sup> H COSY	proton-proton correlation spectroscopy
cPNA	chiral or conformationally constraint peptide nucleic acid
d	doublet
dd	doublet of doublet
ddd	doublet of doublet of doublet
dt	doublet of triplet
DC	direct current
DCC	<i>N,N'</i> -dicyclohexylcarbodiimide
DCM	dichloromethane
DCU	dicyclohexyl urea .
DEAD	diethylazodicarboxylate
DIAD	diisopropylazodicarboxylate
DEPT	distortionless enhancement by polarization transfer
DIEA	<i>N,N'</i> -diisopropylethylamine
DMAP	4-dimethylaminopyridine
DMF	<i>N,N'</i> -dimethylformamide
DMSO- <i>d</i> <sub>6</sub>	deuterated dimethylsulfoxide
Dpm	diphenylmethyl
D <sub>2</sub> O	deuterium oxide
EDTA	ethylenediamine tetraacetic acid
eq.	equivalent (s)
Fmoc	9-fluorenylmethoxycarbonyl
FmocCl	9-fluorenylmethylchloroformate
g	gram

HBTU	(1- <i>H</i> -benzotriazol-1-yloxy)-1,1,3,3-tetramethyluronium hexafluorophosphate
HOAc	acetic acid
HOBt	1 <i>H</i> -hydroxybenzotriazole
HPLC	high performance liquid chromatography
hr	hour
IR	infrared
<i>J</i>	coupling constant
m	multiplet
MALDI-TOF	matrix-assisted laser desorption/ionization-time of flight
mg	milligram
MHz	megahertz
min	minute
mL	milliliter
mmol	millimole
mp.	melting point
M <sub>r</sub>	relative molecular mass
NMR	nuclear magnetic resonance
Ns	nosyl or 4-nitrobenzenesulfonyl
m/z	mass per charge ratio
Pfp	pentafluorophenyl
PfpOH	pentafluorophenol
Ph	phenyl
PNA	peptide nucleic acid or polyamide nucleic acid
ppm	part per million
rpm	round per minute
s	singlet
t	triplet
T <sup>Bz</sup>	<i>N</i> <sup>3</sup> -benzoylthymine
TBE	tris-borate EDTA buffer
TFA	trifluoroacetic acid
THF	tetrahydrofuran

TLC	thin layer chromatography
$T_m$	melting temperature
TEMED	$N, N, N', N'$ -tetramethylethylenediamine
$t_R$	retention time
UV	ultraviolet
V	volt
$\nu_{\max}$	the wavelength at maximum absorption
$\mu L$	microliter
$\mu m$	micrometer
$\mu mol$	micromol
$\delta$	chemical shift
$[\alpha]_D$	specific rotation