การศึกษาและประคิษฐ์โครงสร้างนาโนของอินเดียมฟอสไฟด์ ด้วยวิธีดรอปเลทอิพิแทกซีจากลำโมเลกุล



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วิทยานิพนธ์นี้เป็นส่วนหนึ่งของการศึกษาตามหลักสูตรปริญญาวิศวกรรมศาสตรคุษฎีบัณฑิต สาขาวิชาวิศวกรรมไฟฟ้า ภาควิชาวิศวกรรมไฟฟ้า คณะวิศวกรรมศาสตร์ จุฬาลงกรณ์มหาวิทยาลัย ปีการศึกษา 2552 ลิขสิทธิ์ของจุฬาลงกรณ์มหาวิทยาลัย

STUDY AND FABRICATION OF InP NANOSTRUCTURES GROWN BY DROPLET MOLECULAR BEAM EPITAXY

Miss Wipakorn Jevasuwan

14.1

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วิภากร จีวะสุวรรณ : การศึกษาและประคิษฐ์โครงสร้างนาโนของอินเดียมฟอสไฟด์ด้วยวิธีดรอปเลทอิพิ แทกซี่งากลำโมเลกุล. (STUDY AND FABRICATION OF InP NANOSTRUCTURES GROWN BY DROPLET MOLECULAR BEAM EPITAXY) อ. ที่ปรึกษาวิทยานิพนธ์หลัก : รศ. ดร. สมชัย รัตนธรรมพันธ์, อ. ที่ปรึกษาวิทยานิพนธ์ร่วม : ศ.ดร. สมศักดิ์ ปัญญาแก้ว, 113 หน้า.

การประดิษฐ์ควอนดัมดอดโมเลกุลรูปวงแหวนของอินเดียมฟอสไฟด์ที่ก่อตัวขึ้นเอง ถูกนำเสนอขึ้นเป็นกรั้ง แรกในวิทยานิพนธ์นี้ อิทธิพลของด้วแปรในการปลกผลึกที่มีต่อการก่อด้วและคุณสมบัติของควอนดัมดอด โมเลกลรปวงแหวนของอินเดียมฟอสไฟด์ได้มีการศึกษาอย่างละเอียด ด้วอย่างควอนดัมดอดโมเลกุลรูปวงแหวน ้ของอินเดียมฟอสไฟด์ในเมตริกซ์ของอินเดียมแกลเลียมฟอสไฟด์ที่มีส่วนผสมของอินเดียมเท่ากับ 0.5 บนแผ่น ้จานแกลเลียมอาร์เซไนด์ชนิดไม่มีการโด๊ป และมีระนาบ 001 ถกเตรียมขึ้นด้วยเครื่องปลกผลึกด้วยลำโมเลกลที่มี ้ชนิคสารตั้งค้นเป็นของแข็งและใช้เทคนิคการปลกครอปเลทอิพิแทกซี ตัวแปรที่ทำการศึกษาสัมพันธ์โดยตรงกับ ้กระบวนการสร้างอินเดียมครอปเลท และกระบวนการทำอินเดียมครอปเลทให้เป็นผลึกอินเดียมฟอสไฟค์ ซึ่ง ้ได้แก่ อุณหภูมิการสร้างอินเดียมครอปเลท, อุณหภูมิการทำให้เป็นผลึก, อัตราการสร้างอินเดียมครอปเลท และ ปริมาณอินเดียม คุณสมบัติของควอนตัมดอต โมเลกุลถูกทำการติดตามผลอย่างละเอียดระหว่างกระบวนการ และภายหลังเสร็จสิ้นกระบวนการปลกผลึก โดยการสังเกตรปแบบการเลี้ยวเบนของการสะท้อน ปลกผลึก อิเล็กตรอนพลังงานสูง (Reflection Eigh Energy Electron Diffraction), การวัดด้วยแรงอะตอม (Atomic Microscopy), การวัดด้วยกล้องจุลทรรศน์อิเล็กตรอนแบบส่องผ่าน Force (Transmission Electron Microscopy) และ การวัดโฟโคลูมิเนสเซนด์ (Photoluminescence Spectroscopy)

จากผลการทดลองแสดงให้เห็นว่า การก่อตัวของควอนดัมดอดโมเลกุลรูปวงแหวนของอินเดียมฟอสไฟด์ เกิดขึ้นเนื่องจากความสามารถของเทคนิคครอปเลทอิพิแทกซี และค่าคงดัวผลึกที่แตกค่างกัน 3.8 เปอร์เซ็นด์ของ อินเดียมฟอสไฟด์และอินเดียมแกลเลียมฟอสไฟด์ที่มีส่วนผสมของอินเดียมเท่ากับ 0.5 ความสม่ำเสมอของขนาด ควอนดัมดอดที่ดีที่สุด และจำนวนสูงสุดของควอนดัมดอดโมเลกุลเท่ากับ 46 เปอร์เซ็นด์ ที่ประกอบด้วยแปด ควอนดัมดอดที่ดีที่สุด และจำนวนสูงสุดของควอนดัมดอดโมเลกุลเท่ากับ 46 เปอร์เซ็นด์ ที่ประกอบด้วยแปด ควอนดัมดอดค่อโมเลกุล สัมฤทธิ์ผลได้โดยการใช้ อุณหภูมิการสร้างอินเดียมดรอปเลท, อุณหภูมิการทำให้เป็น ผลึก, อัตราการสร้างอินเดียมดรอปเลท และ ปริมาณอินเดียมเท่ากับ 250 °C, 200 °C, 1.6 ML/s และ 3.2 ML ตามลำดับ ค่ายอดและค่าความกว้างของสเปกตรัมโฟโตลูมิเนสเซนต์ของควอนดัมดอดโมเลกุลรูปทรงวงแหวน ของอินเดียมฟอสไฟด์ที่ได้จากเงื่อนไขการปลูกนี้มีค่าเท่ากับ 1.68 eV และ 43 meV ที่ 20 K และ 1.61 eV และ 60 meV ที่อุณหภูมิห้อง

ควอนดัมดอดโมเลกุลรูปทรงวงแหวนของอินเดียมฟอสไฟต์ที่สร้างขึ้นโดยเทคนิดดรอปเลทอิพิแทกซีถูกทำ การศึกษา และแสดงให้เห็นความเป็นไปได้ที่จะเป็นทางเลือกหนึ่ง ในการนำไปใช้กับเทคโนโลยีควอนดัม คอมพิวเตอร์ ความสำเร็จที่ได้จากการทดลองประดิษฐ์ควอนดัมดอดโมเลกุลรูปทรงวงแหวนของอินเดียมฟอส ไฟด์ในวิทยานิพนธ์ฉบับนี้ นับเป็นจุดเริ่มด้นที่น่าสนใจในฐานะระบบสารที่ให้โครงสร้างควอนดัมดอดโมเลกุล รูปทรงวงแหวน ซึ่งจะเป็นด้วหลักสำคัญในการพัฒนาโครงสร้างดังกล่าวต่อไปในอนาดด

ภาควิชา	วิศวกรรมไฟฟ้า	ลายมือชื่อนิสิด ว็ภาทร จีวะสวรรณ
สาขาวิชา	วิศวกรรมไฟฟ้า	ลายมือซื้ออ. ที่ปรึกษาวิทยานิพนธ์หลัก
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WIPAKORN JEVASUWAN : STUDY AND FABRICATION OF INP NANOSTRUCTURES GROWN BY DROPLET MOLECULAR BEAM EPITAXY. THESIS ADVISOR : ASSOC. PROF. SOMCHAI RATANATHAMMAPHAN, D. Eng., THESIS CO-ADVISOR : PROF. SOMSAK PANYAKEOW, D. Eng., 113 pp.

The fabrication of self-assembled InP ring-shaped QDMs has been first proposed in this thesis. The influences of growth parameters on the formation and characteristics of InP ring-shaped QDMs have been intensively studied. The InP ringshaped QDMs samples were prepared using solid-source MBE via droplet epitaxy technique in $In_{0.5}Ga_{0.5}P$ matrices on semi-insulating GaAs (001) substrate. The investigated parameters relating to the In droplet deposition and crystallization process were deposition temperature, crystallization temperature, In deposition rate and In amount. The InP ring-shaped QDMs properties were thoroughly evaluated by both *in situ* and *ex situ* monitoring including Reflection High Energy Electron Diffraction (RHEED), Atomic Force Microscopy (AFM), Transmission Electron Microscopy (TEM), and Photoluminescence Spectroscopy (PL).

The experimental results have shown that the formation of InP ring-shaped quantum dot molecules (QDMs) is owing to the performance of droplet epitaxy technique and the lattice mismatched between InP and $In_{0.5}Ga_{0.5}P$ of 3.8%. The best QD size homogeneity and the highest number of QDMs (46%) which consist of eight QDs per QDM was achieved with the deposition temperature, crystallization temperatures, In deposition rate and In coverage of 250 °C, 200 °C, 1.6 ML/s and 3.2 ML, respectively. The PL peaks and FWHMs of these InP ring-shaped QDMs are 1.68 eV and 43 meV at 20 K and 1.61 eV and 60 meV at room temperature.

InP ring-shaped QDMs grown by droplet epitaxy technique has been investigated and shows the feasibility as an alternative material system for quantum computing technologies. The achievement on this thesis should renew the interest of InP ring-shaped QDMs as a promising system and also be an important mile stone in the development of ring-shaped QDM structure.

Department: Electrical Engineering Field of study: Electrical Engineering Academic year: 2009 Student's signature <u>Wipakorn</u> jevasuwan Advisor's signature <u>Survay</u> Co-advisor's signature

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LIST OF SYMBOLS

∇^2	Laplacian operator
α	control parameter
\mathbf{a}_0	space between surface sites
a //	parallel lattice constant
a⊥	perpendicular lattice constant
a _e	lattice constant of deposited material
as	lattice constant of substrate material
A	surface area
AFM	atomic force microscopy
AICL	Analytical Instrument Center and Laboratory
Ar ⁺	argon
As	arsenic
AsBr ₃	arsenic tribromide
AsH ₃	arsine
BEP	beam equivalent pressure
C ₀	concentration of group III atoms on the droplet boundary
C ₁₁ , C ₁₂	elastic constant of epitaxial layer
CB	conduction band
CCD	charge-coupled device
CdSe	cadmium selenide
CEO	cleaved-edge overgrowth
CRT	cathode ray tube
0D	zero dimension
1D	one dimension
2D	two dimension
3D	three dimension
δ	delta function
∆h	height of nanostructure
Δt	short time
	or limit time
Δγ	change of surface free energy

d ₀	strained epitaxial film thickness
d ₁	substrate thickness
D _{0,111}	prefactor
$D_{bulk}(E)$	bulk density of state
D _{III}	diffusion coefficient of group III atoms
D _{QW} (E)	quantum well density of state
$D_{QWR}(E)$	quantum wire density of state
$D_{\mathrm{QD}}(E)$	quantum dot density of state
D.O.S.	density of state
ε	lattice mismatch or misfit strain
Exx	axial strain in x-direction
Еуу	axial strain in y-direction
EZZ	axial strain in z-direction
Exz	shear strain on x-plane directed through z-direction
Exy	shear strain on x-plane directed through y-direction
Eyz	shear strain on y-plane directed through z-direction
E //	in-plane strain
\mathcal{E}_{ot}	strain in perpendicular to the growth direction
E	carrier energy
	or total energy per unit cell
E(L)	total energy per unit volume
E ₀	characteristic energy
Ea	adsorption energy and the
E _{bulk}	carrier energy of bulk
Ed	energy barrier for the hopping between surface sites
Eg	band gap energy
E _{l,x}	quantized energy in x-direction
E _{m, y}	quantized energy in y-direction
E _{n,z}	quantized energy in z-direction
E _{QW}	carrier energy of quantum well
E _{QWR}	carrier energy of quantum wire

E _{QD}	carrier energy of quantum dot
E _{WL}	energy of wetting layer
E _{rip}	energy of ripened island
Eisland	energy of single island
Eelastic	elastic strain energy
E _{st}	strain energy
E _{surface}	island surface energy
E _{edge}	island edge energy
EQCA	extended quantum dot cellular automata
F(r)	envelope wave function
FM	Frank van der Merwe
FWHM	full width at half maximum
γe	surface free energy of the epilayer/vacuum interface
γ_i	surface free energy of epilayer/substrate interface
γs	surface free energy of the substrate/vacuum interface
G	reciprocal lattice vector
Ga	gallium
GaAs	gallium arsenide
GaP	gallium phosphide
GaSb	gallium antimony
Ge	germanium
GS-MBE	gas-source molecular beam epitaxy
h	Planck's constant
	or height of pyramidal quantum dot
	or final height
ħ	reduced Planck's constant
h ₀	thickness of monolayer
h _c	critical thickness of strained layer
НН	heavy-hole band
InAs	indium arsenide
InGaAs	indium gallium arsenide

InGaP	indium gallium phosphide
InP	indium phosphide
k	Boltzmann constant
	or amplitude of wave vector
K	Knudsen
k _B	Boltzmann's constant
k _{//}	amplitude of in-plane (y-z) wave vector
\mathbf{k}_{\perp}	amplitude of wave vector in x-direction
$\mathbf{k}=(\mathbf{k}_{\mathbf{x}},\mathbf{k}_{\mathbf{y}},\mathbf{k}_{\mathbf{z}})$	carrier wave vector
k _{in}	wave vectors of incident electron
k _{diff}	wave vectors of diffraction electron
λ	elastic modulus
	or wavelength of the x-ray
$\lambda_{de Broglie}$	de Broglie wavelength
1	quantum number in x-direction
L	macroscopic length scale
	or base size of pyramidal quantum dot
L ₀	characteristic length
LaB ₆	Lanthanum Hexaboride
L _{opt}	optimal island size
LH	light-hole band
LN_2	liquid nitrogen
LPE	liquid phase epitaxy
m	quantum number in y-direction
	or mass of an group V atom
m	mass of an electron, 9.11×10^{-31} kg
m _e	effective electron mass
m _h	effective electron mass
m*	carrier effective mass
MBE	molecular beam epitaxial or molecular beam epitaxy
ML	monolayer

Mo	molybdenum
MOCVD	metalorganic chemical vapour deposition
MO-MBE	metal-organic molecular beam epitaxy
υ_{PR}	Poisson's ratio
n	quantum number in z-direction
	or an integer representing the order of the diffraction peak
N _A	Avogadro constant
N _D	volume density of quantum dot
N _{(i)III-V}	final shape of nanostructures
NIII	amount of diffused group III atoms
N _{III-V}	total amount per unit time
Nv	amount of the trapped group V atoms
Nwi	area density of the quantum wires
р	carrier momentum
Р	intensity of group V flux
P ₂	phosphorus dimers
P ₄	phosphorous tetramers
PBN	pyrolytic boron nitride
PH ₃	phosphine
PL	photoluminescence
Q	total deposited material (monolayer)
Q ₁	deposited material that form wetting layer
Q ₂	deposited material that form coherent 3D island
QCA	quantum dot cellular automata
QW	quantum well
QWR	quantum wire
QD	quantum dot
QDM	quantum dot molecule
QR	quantum ring
r _c	trapping radius of droplet
r _{GaAs}	growth rate of gallium arsenide

r _{III}	atomic radius of group III atoms
۲ _{InAs}	growth rate of indium arsenide
r _{InGaAs}	growth rates of indium gallium arsenide
r = (x , y , z)	carrier position vector
R	amount of impacting group V atoms on
	the substrate per unit time
Re	rhenium
RHEED	reflection high-electron energy diffraction
RT	room temperature
SDRL	semiconductor devices research laboratory
SEM	scanning electron microscopy
SK	Stranski Krastanow
SO	spin-split-off band
SS-MBE	solid-source molecular beam epitaxy
STM	scanning tunneling microscope
Θ	Heaviside's unit step function
θ	scattering angle
t	film thickness
t _m	time of crystallization process
Т	temperature
T ₁	first temperature
T ₂	second temperature
T ₃	third temperature
T ₄	fourth temperature
T _{sub}	substrate temperature
T _{buffer} growth	buffer growth temperature
T _{transition}	transition temperature
TEGa	triethylgallium
TEM	transmission electron microscopy
Ti	titanium
TMIn	trimethylindium

UHV	ultra-high vacuum
ν_1	thermal vibration frequencies for the upward direction
ν_0	thermal vibration frequencies for the lateral direction
V _{droplet}	volume of droplet
V _{mIII}	molar volume of droplet
V(r)	confinement potential
VPE	vapor phase epitaxy
VW	Volmer Weber
W	tungsten
WL	wetting layer
X _{in}	In composition

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