EFFECT OF CARBON DIOXIDE ON METHANE CONVERSION UNDER ELECTRIC DISCHARGE

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ABSTRACT

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The behavior of direct methane conversion processed in an applied AC electric discharge at ambient conditions was investigated by using a Dielectric-Barrier Discharge (DBD) reactor. The main objectives of this study were to determine the feasibility of DBD for conversion of methane to higher hydrocarbons under in the presence of carbon dioxide and to determine the effects of carbon dioxide and helium on methane conversion. The studies were conducted by operating a reactor in the range of 5,500 to 7,700 V at which the space time was 4 to 8 minutes. CO₂:CH₄ ratios of 0:1, 1:1, 1:2 and 2:1 were used. Helium was also added in the methane feed at different concentrations. It was found that methane conversion increased remarkably with increases in CO₂:CH₄ ratio, voltage, helium concentration and space time. The C₂ products selectivities decreased but the carbon monoxide and hydrogen selectivities increased when voltage, helium concentration and space time were increased. Under the condition studied, small amounts of methanol were produced while most reaction products were ethane, ethylene, acetylene, carbon monoxide and hydrogen. It can be concluded that under the conditions studied, most of CO₂ is simply converted into CO but very low amount of methanol is produced.

บทคัดย่อ

มาลินี ลีโทชวลิต : ผลกระทบของก๊าซการ์บอนไดออกไซค์ต่อการเปลี่ยนแปลงทางเคมี ของก๊าซมีเธนภายใต้สนามไฟฟ้าแรงสูง(Effect of Carbon Dioxide on Methane Conversion Under Electric Discharge) อ. ที่ปรึกษา : รศ.คร. แลนซ์ ลอบแบน (Assoc. Prof. Lance Lobban) และ คร. สูเมธ ชวเคช 92 หน้า ISBN 974-638-540-2

การศึกษาพฤติกรรมการเปลี่ยนแปลงทางเคมีของก๊าซมีเธน ภายใต้สภาวะไฟฟ้าแรงสง ได้ถูกดำเนินการโดยใช้เครื่องปฏิกรณ์เคมีแบบ DBD วัตถุประสงค์หลักของการศึกษานี้เพื่อศึกษา ความเป็นไปได้ในการเปลี่ยนก๊าซมีเธนภายใต้สภาวะที่มีก๊าซคาร์บอนไดออกไซด์ร่วมอย่ให้เป็น ที่มีน้ำหนักโมเลกุลสูงขึ้นและศึกษาถึงผลกระทบของก๊าซ ก๊าซไฮโครคาร์บอนชนิคอื่น ๆ คาร์บอนไดออกไซด์ และ ก๊าซฮีเลียมที่มีต่อการเปลี่ยนแปลงก๊าซมีเทน การศึกษานี้ควบคุมเครื่อง ปฏิกรณ์เคมีที่ความต่างศักย์อยู่ในช่วงระหว่าง 5,500 ถึง 7,700 โวลต์ เวลาในเครื่องปฏิกรณ์เคมี ของก๊าซมีค่าอยู่ในช่วง 4 ถึง 8 นาที อัตราส่วนระหว่างก๊าซคาร์บอนไดออกไซค์และก๊าซมีเธนเป็น 0:1 1:1 1:2 และ 2:1 โดยมีก๊าซฮีเลียมร่วมอยู่ในสารตั้งต้นที่ปริมาณต่าง ๆ กัน จากการศึกษาพบ การเปลี่ยนแปลงของก๊าซมีเธนเพิ่มสูงขึ้นเมื่อปริมาณก๊าซคาร์บอนไดออกไซด์ในสารตั้งต้น ความต่างศักย์ไฟฟ้า เวลาในเครื่องปฏิกรณ์ของก๊าซ ตลอคจนปริมาณก๊าซฮีเลียมที่ร่วมในปฏิกริยา เพิ่มสูงขึ้น โดยทั่วไปอัตราการเลือกเกิดของก๊าซไฮโดรคาร์บอนที่มีคาร์บอน 2 อะตอมลดต่ำลง แต่ อัตราการเลือกเกิดของก๊าซคาร์บอนมอนอกไซด์และก๊าซไฮโครเจนเพิ่มสูงขึ้น ศักย์ไฟฟ้า เวลาในเครื่องปฏิกรณ์ของก๊าซ และปริมาณฮีเลียมในสารตั้งตั้น ภายใต้สภาวะที่ทำการ ศึกษาพบว่ามีเมทานอลเกิดขึ้นในปริมาณน้อย ในขณะที่ผลิตภัณฑ์ส่วนใหญ่ที่เกิดขึ้น ได้แก่ อีเทน เอทิลีน อะเซทิลีน คาร์บอนมอนอกไซค์ และ ก๊าซไฮโครเจน

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TABLE OF CONTENTS

		PAGE
	Title Page	i
	Abstract	iii
	Acknowledgments	v
	Table of Contents	vi
	List of Tables	ix
	List of Figures	xi
CHAPTER		
I	INTRODUCTION	1
II	LITERATURE SURVEY	je.
	2.1 Physical and Chemical Properties of Methane	6
	2.2 Gaseous Plasmas for Activating Methane	8
	Molecules	
	2.2.1 Fundamental properties of plasma	9
	2.2.2 Generation of plasma	9
	2.3 Type of Non-Equilibrium Plasma	14
	2.3.1 Radio frequency (RF) discharge	14
	2.3.2 Microwave discharge	15
	2.3.3 Glow discharge	16
	2.3.4 Corona discharge	17
	2.3.5 The Dielectric-barrier discharge (DBD)	18

CHAPTER		PAGE
III	EXPERIMENTAL SECTION	
	3.1 Experimental Setup	23
	3.1.1 The reactant make-up section	23
	3.1.2 The reaction section	25
	3.1.3 Sample collecting and analysis section	27
	3.2 Experimental Procedure	30
	3.2.1 Influence of carbon dioxide-methane ratio	30
	3.2.2 Influence of helium	30
IV	RESULTS AND DISCUSSION	
	4.1 Effect of CO ₂ :CH ₄ Ratio	32
	4.1.1 Effect of CO ₂ :CH ₄ ratio on methane	32
	conversion	
	4.1.2 Effect of CO ₂ :CH ₄ ratio on carbon dioxide	34
	conversion	
	4.1.3 Effect of CO ₂ :CH ₄ ratio on product	34
	selectivity	
	4.2 Effect of Helium	39
	4.2.1 Effect of helium on methane conversion	39
	4.2.2 Effect of helium on product selectivity	39
	4.3 Effect of Applied Voltage	47
	4.3.1 Effect of applied voltage on methane	47
	conversion	
	4.3.2 Effect of applied voltage on product	49
	selectivity	

CHAPTER		PAGE
	4.4 Effect of Space Time4.4.1 Effect of space time on methane conversion	56 56
	4.4.2 Effect of space time on product selectivity	57
${f v}$	CONCLUSIONS AND RECOMMENDATIONS	
	5.1 Conclusions	61
	5.2 Recommendations	62
	REFERENCES	63
	APPENDICES	67
	APPENDIX A Experimental Data	68
	APPENDIX B Calculation Procedures	76
	CURRICULUM VITAE	78

LIST OF TABLES

TABLE		PAGE
1.1	Approximate composition of wet and dry natural gas	2
2.1	Average chemical bond energy of some covalent bonds	7
2.2	The first ionization potential of some common gases	8
2.3	Collision mechanisms in the gases	12
3.1	Specification of gases used in the experiment	25
A.1	Methane conversion, carbon dioxide conversion and	68
	product selectivities at different voltage (CO ₂ : CH ₄	
	ratio of 1:1, 80% helium concentration and space time	
	of 4 minutes)	
A.2	Methane conversion, carbon dioxide conversion and	69
	product selectivities at different voltage (CO ₂ : CH ₄	
	ratio of 1:1, 80% helium concentration and space time	
	of 6 minutes)	
A.3	Methane conversion, carbon dioxide conversion and	69
	product selectivities at different voltage (CO ₂ : CH ₄	
	ratio of 1:2, 80% helium concentration and space time	
	of 4 minutes)	
A.4	Methane conversion, carbon dioxide conversion and	70
	product selectivities at different voltage (CO ₂ : CH ₄	
	ratio of 1:2, 80% helium concentration and space time	
	of 6 minutes)	

TABI	TABLE	
A.5	Methane conversion, carbon dioxide conversion and	70
	product selectivities at different voltage (CO ₂ : CH ₄	
	ratio of 2:1, 80% helium concentration and space time	
	of 4 minutes)	
A.6	Methane conversion and product selectivities at voltage	71
	of $7,700 \text{ V}$ (CO_2 : CH_4 ratio of $0:1$, 80% helium	
	concentration and space time of 4 minutes)	
A.7	Methane conversion and product selectivities at	72
	different voltage (CO_2 : CH_4 ratio of 1:1, 50% helium	
	concentration and space time of 4 minutes)	
A.8	Methane conversion and product selectivities at	72
	different voltage (CO ₂ : CH ₄ ratio of 1:1, 50% helium	
	concentration and space time of 6 minutes)	
A.9	Methane conversion and product selectivities at	73
	different voltage (CO_2 : CH_4 ratio of 1:1, 50% helium	
	concentration and space time of 8 minutes)	
A.10	Methane conversion and product selectivities at	74
	different voltage (CO ₂ : CH ₄ ratio of 1:1, 20% helium	
	concentration and space time of 4 minutes)	
A.11	Methane conversion and product selectivities at	74
	different voltage (CO ₂ : CH ₄ ratio of 1:1, 20% helium	
	concentration and space time of 6 minutes)	
A.12	Methane conversion and product selectivities at	75
	different voltage (CO2: CH4 ratio of 1:1, free helium	
	in feed and space time of 6 minutes)	

LIST OF FIGURE

FIGURE		PAGE
2.1	The alternative methods of charged particles generation.	10
3.1	The schematic diagram of experimental equipment.	24
3.2	Configuration of DBD reactor.	26
3.3	Power supply diagram.	28
4.1	Effect of CO ₂ :CH ₄ ratio on methane conversion at three	33
	different applied voltage 80% helium concentration	
	and space time of 4 minutes.	
4.2	Effect of CO ₂ :CH ₄ ratio on carbon dioxide conversion	33
	at three different applied voltage 80% helium	
	concentration and space time of 4 minutes.	
4.3	Effect of CO ₂ :CH ₄ ratio on ethylene selectivity at three	35
	different applied voltage, 80% helium concentration	
	and space time of 4 minutes.	
4.4	Effect of CO ₂ :CH ₄ ratio on acetylene selectivity at three	35
	different applied voltage, 80% helium concentration	
	and space time of 4 minutes.	
4.5	Effect of CO ₂ :CH ₄ ratio on ethane selectivity at three	37
	different applied voltage ,80% helium concentration	
	and space time of 4 minutes.	
4.6	Effect of CO ₂ :CH ₄ ratio on carbon monoxide selectivity	37
	at three different applied voltage, 80% helium	
	concentration and space time of 4 minutes.	

FIGUR	TIGURE	
4.7	Effect of CO ₂ :CH ₄ ratio on hydrogen selectivity at three	38
	different applied voltage, 80% helium concentration	
	and space time of 4 minutes.	
4.8	Effect of CO ₂ :CH ₄ ratio on methanol selectivity at three	38
	different applied voltage, 80% helium concentration	
	and space time of 4 minutes.	
4.9	Effect of helium concentration on methane conversion	40
	at three different applied voltage, space time of 4 and	
	6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.10	Effect of helium concentration on ethylene	41
	conversion at three different applied voltage, space time	
	of 4 and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.11	Effect of helium concentration on acetylene	42
	conversion at three different applied voltage, space time	
	of 4 and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.12	Effect of helium concentration on ethane selectivity at	43
	three different applied voltage, space time of 4 and 6	
	minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.13	Effect of helium concentration on cabon monoxide	44
	selectivity at three different applied voltage, space time	
	of 4 and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.14	Effect of helium concentration on hydrogen	45
	selectivity at three different applied voltage, space time	
	of 4 and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	

FIGU	RE	PAGE
4.15	Effect of helium concentration on methanol selectivity	46
	at three different applied voltage, space time of 4 and	
	6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.16	Effect of applied voltage on methane conversion at	48
	three different helium concentration, space time of 4	
	and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.17	Effect of applied voltage on ethylene selectivity at	50
	three different helium concentration, space time of 4	
	and 6 minutes and CO_2 : CH_4 ratio of 1: 1.	
4.18	Effect of applied voltage on acetylene selectivity at	51
	three different helium concentration, space time of 4	
	and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.19	Effect of applied voltage on ethane selectivity at three	52
	different helium concentration, space time of 4 and 6	
	minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.20	Effect of applied voltage on cabon monoxide	53
	selectivity at three different helium concentration,	
	space time of 4 and 6 minutes and CO ₂ : CH ₄ ratio of	
	1: 1.	
4.21	Effect of applied voltage on hydrogen selectivity	54
	at three different helium concentration, space time of 4	
	and 6 minutes and CO ₂ : CH ₄ ratio of 1: 1.	
4.22	Effect of applied voltage on methanol selectivity at	55
	three different helium concentration, space time of 4	
	and 6 minutes and CO ₂ : CH ₄ ratio of 1:1.	

FIGU	URE	PAGE
4.23	Effect of space time on methane conversion at three	57
	different applied voltage, 50 % helium concentration	
	and CO ₂ : CH ₄ ratio of 1:1.	
4.24	Effect of space time on acetylene selectivity at three	58
	different applied voltage, 50% helium concentration	
	and CO ₂ : CH ₄ ratio of 1:1.	
4.25	Effect of space time on ethylene selectivity at three	58
	different applied voltage ,50 % helium concentration	
	and CO ₂ : CH ₄ ratio of 1:1.	
4.26	Effect of space time on ethane selectivity at three	59
	different applied voltag, 50 % helium concentration	
	and CO ₂ : CH ₄ ratio of 1:1.	
4.27	Effect of space time on carbon monoxide selectivity at	59
	three different applied voltage, 50 % helium	
	concentration and CO ₂ : CH ₄ ratio of 1:1.	
4.28	Effect of space time on hydrogen selectivity at three	60
	different applied voltage 50 % helium concentration	
	and CO ₂ : CH ₄ ratio of 1:1.	
4.29	Effect of space time on methanol selectivity at three	60
	different applied voltage, 50 % helium concentration	
	and CO ₂ : CH ₄ ratio of 1:1.	