THE MODIFICATIONS OF COTTON PROPERTIES BY ADMICELLAR POLYMERIZATION

Mr. Chaiyapat Pisuntornsug

A Thesis Submitted in Partial Fulfilment of the Requirements
for the Degree of Master of Science

The Petroleum and Petrochemical College, Chulalongkorn University
in Academic Partnership with

The University of Michigan, The University of Oklahoma,
and Case Western Reserve University

2001
ISBN 974-13-0690-3

Thesis Title:

The Modification of Cotton Properties by Admicellar

Polymerization

By:

Mr. Chaiyapat Pisuntornsug

Program:

Petrochemical Technology

Thesis Advisor:

Prof. Edgar A. O'Rear

Asst. Prof. Nantaya Yanumet

Accepted by the Petroleum and Petrochemical College, Chulalongkorn University, in partial fulfilment of the requirements for the Degree of Master of Science

K. Bunyahat.

College Director

(Assoc. Prof. Kunchana Bunyakiat)

Thesis Committee:

(Prof. Edgar A. O'Rear)

(Asst. Prof. Nantaya Yanumet)

N. Yanumet.

(Dr. Pramoch Rangsunvigit)

ABSTRACT

4271002063: PETROCHEMICAL TECHNOLOGY PROGRAM

Chaiyapat Pisuntornsug: The Modification of Cotton Properties

by Admicellar Polymerization.

Thesis Advisors: Prof. Edgar A. O'Rear and Asst. Prof. Nantaya

Yanumet, 46 pp ISBN 974-13-0690-3

Keywords: Surfactant/Thermal stability/Admicelle/Dye ability/Admicellar

Polymerization

The modification of cotton properties has been developed because cotton has deficiency in certain properties such as thermal stability and dyeing ability. Conventional methods for modifications of cotton properties employ the coating technique, which also closes the gaps between fibers and yarns resulting in fabric with poor moisture permeability. Admicellar polymerization is a technique which can be used to produce cotton with good ventilation. In the present work, two different monomers, 3,4-dichloro-1-butene and sodium styrene sulfonate, which can improve the thermal stability and dyeing ability of cotton, were used respectively. The results showed that the treated cotton has improved thermal stability and dyeing ability. Moreover, the dyeing of treated cotton can be done at room temperature with no addition of salt.

บทคัดย่อ

นายชัยภัทร ไพสุนทรสุข: การคัดแปลงคุณสมบัติของผ้าฝ้ายโดยกระบวนการแอคไม เซลลาล์พอลิเมอร์ไรซ์เซชั่น (The Modification of Cotton Properties by Admicellar Polymerization) อ. ที่ปรึกษา ศ. คร. เอ็คการ์ โอเรียร์ และ ผศ. คร. นันทยา ยานุเมศ 46 หน้า ISBN 974-13-0690-3

การคัดแปลงคุณสมบัติของผ้าฝ้ายได้มีการพัฒนาขึ้น เพื่อปรับปรุงข้อบกพร่องที่มีอยู่ใน ฝ้าย เช่น ความคงทนต่อความร้อน และ คุณสมบัติในการติดสี วิธีที่ใช้คัดแปลงคุณสมบัติของผ้า ฝ้ายโดยทั่วไปคือการเคลือบสารเคมีลงบนผ้า ทำให้ผ้าไม่สามารถระบายอากาศได้เนื่องจากเกิดการ อุคตันในร่องผ้า กระบวนการแอดไมเซลลาพอลิเมอร์ไรซ์เซชั่นสามารถคัดแปลงคุณสมบัติของผ้า ฝ้ายโดยที่คุณสมบัติการระบายอากาศยังคงอยู่ ในงานวิจัยนี้ใช้ 3,4-ไคคลอโร-1-บิวทีน และ โซเดียม สไตรีน ซัลโฟเนต เพื่อพัฒนาคุณสมบัติความคงทนต่อความร้อน และ คุณสมบัติในการ ติดสี จากการศึกษาพบว่า ผ้าฝ้ายที่ได้รับการเคลือบผิวมีความคงทนต่อความร้อนและคุณสมบัติใน การติดสีที่ดีขึ้น นอกจากนี้การย้อมผ้ายังสามารถกระทำได้ที่อุณหภูมิห้องโคยไม่ต้องมีการใส่เกลือ

ACKNOWLEDGEMENTS

I would like to give my deepest sincere gratitude to my advisors, Prof. Edgar A. O'Rear for being my inspirator. I wish to express my thankfulness to Asst. Prof. Nanyata Yanumet for being my mentor and for her support and suggestion. Without her, I still remain in the whirlpool of work.

I am so deeply indebted to Mr. Thirawudh Pongprayoon and Mr. Siriphong Roatluechai for invaluable comments, creative suggestion, and intensive guidance and for their sentence "Return to basic".

I would like to thank my entire friends for their encouragement and their invaluable friendship that drive me to my goal.

Finally, I would like to express my deepest thankfulness to my beloved parents for their love, kindness and inspiration and to my brother who disturbs me from dusk till dawn.

TABLE OF CONTENTS

		PAGE
	Title Page	i
	Abstratct (in English)	iii
	Abstratct (in Thai)	iv
	Acknowledgements	v
	Table of Contents	vi
	List of Tables	ix
	List of Figures	xi
СНАРТЕ	ER	
I	INTRODUCTION	1
	1.1 Surfactant	3
	1.2 Adsorption Isotherm	4
	1.3 Admicellar Polymerization	5
	1.4 Effect of pH	6
	1.5 Effect of Salt	7
	1.6 Cotton Fabric	8
II	LITERATURE SURVEY	10
III	EXPERIMENTAL	13
	3.1 Materials	13
	3.2 Instruments	13
	3.3 Methodology	14
	3.3.1 Preparation of Cotton Fabric	14
	3.3.2 Preparation of Monomer	14

СНАРТЕ	CHAPTER	
	3.3.3 Admicellar Polymerization	14
	3.3.4 Washing Procedure	15
	3.3.5 Hydrophobic Test	16
	3.3.6 Dyeing Test	16
	3.3.7 Color Strength Analysis	16
	3.3.8 Washing Fastness Testing	17
	3.3.9 Rubbing Fastness Testing	17
IV	RESULTS AND DISCUSSION	19
	4.1 Admicellar Polymerization of 3,4-Dichloro-1-	
	Butene	19
	4.1.1 Effect of Adsolubilization Time	19
	4.1.2 Effect of Polymerization Time	20
	4.1.3 Effect of LAS:monomer	21
	4.1.4 Effect of Initiator:monomer Ratio	21
	4.1.5 Characterization of 3,4-Dichloro-1-	
	Butene Coated Cotton by SEM	22
	4.1.6 Thermal Stability of Treated Cotton	
	by TGA	23
	4.2 Admicellar Polymerization of 4-Chloromethyl	
	Styrene	24
	4.3 Admicellar Polymerization of Sodium Styrene	
	Sulfonate	25
	4.3.1 Effect of NaSS Monomer	25
	4.3.2 Effect of Initiator	26
	4.3.3 Effect of Salt	28
	4.3.4 Effect of pH	28

СНАРТЕ	₹	PAGE
	4.3.5 Effect of Polymerization Time	29
	4.3.6 Effect of Dyeing Temperature	30
	4.3.7 Effect of Dye Concentration	31
	4.3.8 Effect of NaSS Concentration on	
	Washing Fastness	32
	4.3.9 Effect of %Dye on Washing Fastness	33
	4.1.10 Effect of NaSS Concentration on	
	Rubbing Fastness	33
	4.1.11 Effect of %Dye on Rubbing	
	Fastness	34
	4.1.12 Characterization of Cotton Surface	35
V	CONCLUSIONS AND RECOMMENDATION	36
	REFERENCES	37
	APPENDICES	39
	Appendix A Admicellar polymerization of 3,4-	
	dichloro-1-butene	39
	Appendix B Admicellar polymerization of	
	sodium styrene sulfonate	42
	CURRICULUM VITAE	46

1

LIST OF TABLES

ABLE	
4.1 The varied parameter in the admicellar	
polymerization of 4-chloromethyl styrene.	24
4.2 Washing fastness of dyed cotton at different	
concentrations of NaSS.	32
4.3 Washing fastness of dyed cotton at different dye	
concentration.	33
4.4 Rubbing fastness of dyed cotton at different	
concentration of NaSS.	34
4.5 Washing fastness of dyed cotton at different dye	
concentration.	35
A.1 Results of dropped test of treated cotton from	
experiments that determined the adsolubilization time.	39
A.2 Results of dropped test of treated cotton from	
experiments that determined the polymerization time.	39
A.3 Results of dropped test of treated cotton from	
experiments that determined LAS:Monomer ratio.	40
A.4 Results of dropped test of treated cotton from	
experiments that determined Monomer:Initiator ratio.	40
A.5 Results of untreated and treated cotton from	
experiments that determined by TGA.	40
B.1 Results of color strength of treated cotton from	
experiments that determined the polymerization time.	42
B.2 Results of color strength of treated cotton from	
experiments that determined sodium styrene	
sulfonate concentration.	42

TABL	E	PAGE
B.3	Results of color strength of treated cotton from	
	experiments to determine the optimum	
	monomer:initiator ratio at NaSS concentration of	
	5000μM.	43
B.4	Results of color strength of treated cotton from	
	experiments to determine the optimum	
	monomer:initiator ratio at NaSS concentration of	
	15000μΜ.	43
B.5	Results of color strength of treated cotton from	
	experiments to determine the optimum salt	
	concentration.	44
B.6	Results of color strength of treated cotton from	
	experiments to determine the optimum pH.	44
B.7	Results of color strength of treated cotton from	
	experiments that varied the dyeing time at 30°C.	44
B.8	Results of color strength of treated cotton from	
	experiments that varied the dyeing time at 50°C.	45
B.9	Results of color strength of treated cotton from	
	experiments that varied the dye concentration at	
	30000μM NaSS.	45
B.10	Results of color strength of treated cotton from	
	experiments that varied the dye concentration at	
	60000μM NaSS.	45

LIST OF FIGURES

F)	IGUF	RE	PAGE
	1.1	(a) A typical coating technique in the textile process	
		and (b) A blocking problem in the product due to a	
		technique.	1
	1.2	The types of surfactant formation.	4
	1.3	Typical adsorption isotherm of surfactant on solid	
		surface.	4
	1.4	Admicelle formation.	5
	1.5	Monomer adsolubilization.	6
	1.6	Polymerization of monomers.	6
	1.7	Surfactant removal.	6
	1.8	The point of zero charge.	7
	1.9	Close packing of surfactant by adding salt.	8
	1.10	Cellulose structure.	8
	1.11	Complex structure of cellulose in cotton.	9
	3.1	Dropped test method.	15
	3.2	The rubbing fastness instrument.	18
	4.1	Wetting time of treated cotton with adsolubilization time	
		[$1000\mu M$ LAS, initiator:monomer ratio = 1:4, and	
		polymerization time = 6 hrs at 80°C].	20
	4.2	Wetting time with varying polymerization time	
		[1000µM LAS, LAS:monomer ratio = 1:15,	
		initiator:monomer ratio = 1:4, and polymerization at	
		80°C1.	20

FIGUI	RE	PAGE
4.3	Wetting time of treated cotton with varying	
	LAS:monomer ratio[LAS concentration 1000μM,	
	adsolubilization time =15hrs at 30°C, and	
	polymerization time = 6 hrs at 80°C].	21
4.4	Wetting time of treated cotton with varying	
	initiator:monomer ratio[LAS concentration 1000μM,	
	adsolubilization time = 15 hrs, and polymerization	
	time = 6 hrs at 80°C].	22
4.5	Characterization of cotton surface by SEM (a)	
	untreated cotton and (b) modified cotton.	23
4.6	Thermal stability of untreated and treated cotton by TGA	
	◆ Untreated cotton and ■ Treated cotton.	24
4.7	Color strength of treated cotton with varying NaSS	
	concentration[initiator:monomer ratio = 1:1].	26
4.8	Color strength of treated cotton with varying	
	NaSS:initiator ratio[NaSS concentration (a) 5000μM	
	and (b) $15000\mu M$].	27
4.9	Color strength of treated cotton with varying NaCl	
	concentration [NaSS concentration 20000μM,	
	initiator:monomer ratio = 1:1].	28
4.10	Color strength of treated cotton with varying pH	
	[NaSS concentration 5000 µM, initiator:monomer	
	ratio = 1:1].	29
4.11	Color strength of treated cotton with varying	
	polymerization time [NaSS concentration 15000μM,	
	initiator:monomer ratio = 1:1].	30

FIGURE	PAGI
4.12 Color strength of treated cotton with varying dyeing	
time and temperature [NaSS concentration 30000µM,	
initiator:monomer ratio = 1:1, and polymerization	
time 2 hr].	31
4.13 Color strength of treated cotton with percent dye in solutio	n
[NaSS concentration 30000 and 60000 µM,	
initiator:monomer ratio = 1:1, and polymerization	
time 2 hr].	32
4.14 Characterization of cotton surface by SEM	
(a) untreated cotton and (b) NaSS coated cotton.	35