

## REFERENCES

- Abdulsalam, M., and Stanley, J.T. (1992). Steady-state model for erosion-corrosion of feedwater piping. Corrosion, 48(7), 587-593.
- ASTM, (1994). ASTM Designation, Standard A 106-94, 26-37.
- Balakrishnan, P.V. (1977). A radiochemical technique for the study of corrosion products in high-temperature water. Canadian Journal of Chemical Engineering, 55, 357-360.
- Barber, D., and Lister, D.H. (1982, 22-26 November). Chemistry of the water circuits of CANDU reactors. Water Chemistry and Corrosion Problems in Nuclear Power Plants, Proceedings of A Symposium, Vienna
- Basque, F. (1997). Passivation of stainless steels in reactor coolants. M.S. Thesis in Engineering, Faculty of Engineering, University of New Brunswick.
- Berge, P., Ribon, C., and Saint-Paul, P. (1976). Effect of hydrogen on the corrosion of steels in high temperature water. Corrosion, 32(6), 223-228.
- Bignold, G.J., DeWhalley, C.H., Garbett, K., Woolsey, I.S., Mechanistic Aspects of Erosion-Corrosion Under Boiler Feedwater Conditions, Water Chemistry 3, BNES, London, 1983, pp.219-226.
- Castle, J.E., and Mann, G.M.C. (1966). The mechanism of formation of a porous oxide film on steel. Corrosion Science, 6, 253-262.
- Castle, J.E., and Masterson, H.G. (1966). The role of diffusion in the oxidation of mild steel in high temperature aqueous solutions. Corrosion Science, 6, 93-104.
- Effertz, P.H. (1972). Morphology and composition of magnetite layers in boiler tubes following long exposure. 5<sup>th</sup> International Congress on Metallic Corrosion, Tokyo, pages 920-924.

- Ferrell, R.T., and Himmelblau, D.M. (1967). Diffusion coefficients of hydrogen and helium in water. AIChE Journal, 13(4), 702-708.
- Fontana, M.G. (1986). Corrosion Engineering. Singapore: McGraw-Hill, Inc. page91.
- Foster, A.R., and Wright, R.L. (3<sup>rd</sup> edition). (1977). Basic Nuclear Engineering. Boston: Allyn and Bacon, Inc.
- Heitz, E. (1991). Chemo-mechanical effects of flow on corrosion. Corrosion, 42(2), 135-145.
- Heitz, E. (1996). Mechanistically based prevention strategies of flow-induced corrosion. Electrochimica Acta, 41(4), 503-509.
- Henzel, N., Kastner, W., Stellwag, B. (1988). Erosion corrosion in power plants under single- and two-phase flow conditions – updated experience and proven counteractions, Process American Power Conference, 50, pp.992-1000.
- Johari, J.M.C. (1994). Modelling corrosion for activity transport in CANDU reactors and PWRs. M.S. Thesis in Engineering, Faculty of Engineering, University of New Brunswick.
- Lang, L.C. (2000). Modelling the corrosion of carbon steel feeder pipes in CANDU reactors. M.S. Thesis in Engineering, Faculty of Engineering, University of New Brunswick.
- Lister, D.H., Arbeau, N., and Johari, J.M.C. (1994). Erosion and cavitation in the CANDU primary heat transport system report, University of New Brunswick.
- Lister, D.H., Gauthier, P., Goszczynski, J., and Slade, J. (1998). The accelerated corrosion of CANDU primary piping. Paper presented at the 1998 JAIF International Conference, Japan Atomic Industrial Forum on Water Chemistry in Nuclear Power Plants.
- Lister, D.H., Slade, J., and Arbeau, N. (1997). The accelerated corrosion of CANDU outlet feeders-observations, possible mechanisms and

- potential remedies. Proc. Canadian Nuclear Society Annual Conference.
- MacEwan, J.R., Notley, M.J.F., Wood, J.C., Gacesa, M. (1983). CANDU Fuel : Past, Present and Future, International Atomic Entergy Agency (IAWA), Vienna.
- Mancey, D.S. (1997, November). Private communication, AECL.
- Nešić, S., and Postlethwaite, J. (1991). Hydrodynamics of disturbed flow and erosion-corrosion :part I-single-phase flow study. The Canadian Journal of Chemical Engineering, 69, 698-703.
- Olson, R.M., and Sparrow, E.M. (1963). Measurements of turbulent flow development in tubes and annuli with square or rounded entrances. AIChE Journal, 9(6), 766-769.
- Postlethwaite, J., and Lotz, U. (1988). Mass transfer at erosion-corrosion roughened surfaces. The Canadian Journal of Chemical Engineering, 66, 75-78.
- Potter, E.C., and Mann, G.M.W. (1962). Oxidation of mild steel in high-temperature aqueous systems. First International Congress on Metallic Corrosion, Butterworth, London.
- Poulson, B., and Robinson, R. (1988). The local enhancement of mass transfer at 180° bends. International Journal of Heat and Mass Transfer, 31(6), 1289-1297.
- Robertson, J. (1989). The mechanism of high temperature aqueous corrosion of steel. Corrosion Science, 29(11/12), 1275-1291.
- Robertson, J., and Forrest, E. (1991). Corrosion of carbon steels in high temperature acid chloride solution. Corrosion Science, 32(5/6), 521-540.
- Sweeton, F.H., and Baes, C.F. (1970). The solubility of magnetite and hydrolysis of ferrous ion in aqueous solutions at elevated temperature. Journal of Chem. Thermodynamics, 2, 479-500.

- Sanchez-Caldera , L.E., Griffith, P., and Rabinowicz, E. (1988). The mechanism of corrosion-erosion in steam extraction lines of power stations. Journal of Engineering for Gas Turbines and Power, 110, 180-184.
- Stack, M.M., Chacon-Nava, J., and Stott, F.H. (1995). Relationship between the effects of velocity and alloy corrosion resistance in erosion-corrosion environments at elevated temperatures. Wear, 180, 91-99.
- Steward, F.R. (2000, December). The simpler model for outlet feeder pipes. Private Communication, CNER Canada.
- Supa-Amornkul, S. (2000). Numerical analysis of thermalhydraulics for flow-assisted corrosion in CANDU outlet feeder pipes. M.S. Thesis in Engineering, Faculty of Engineering, University of New Brunswick.
- Tomlinson, L. (1981). Mechanism of corrosion of carbon and low alloy ferritic steels by high temperature water. Corrosion-NACE, 39(10), 591-596.
- Tomlinson, L., Hurdus, M., and Silver, P.J.B. (1981). The effect of hydrogen on the corrosion of a 2.25%Cr ferritic steel by high temperature water. Corrosion Science, 21(5), 369-380.
- Tremaine, P.R., and Leblanc, J.C. (1980). The solubility of magnetite and the hydrolysis and oxidation of Fe<sup>2+</sup> in water at 300°C. Journal of Solution Chemistry, 9(6), 415-442.
- Videm, K., Aas, S. (1963, February). Mild Steel in Primary Circuits in Water Cooled Power Reactors, Kjeller Report, Institutt for Atomenergi Kjeller Research Establishment, Kjeller, Norway.
- Wells, A.F.( 4<sup>th</sup> edition). 1975. Structural Inorganic Chemistry. London: Oxford University Press.

## APPENDICES

### Appendix A Input for the program

#### A.1 Sheet "pipe"

##### **Inlet Section**

Number of Nodes	7						
Diameter of Piping (cm)	44.3	50	106	5.68	5.68	5.68	5.68
Velocity of Coolant (cm/s)	1530	1200	270	985	985	985	985
Length of Piping (cm)	477.6	281.8	78.6	350	350	350	350
Surface Area (cm <sup>2</sup> )	6.65E+04	4.43E+04	2.62E+04	5.93E+05	5.93E+05	5.93E+05	5.93E+05
Temperature (°C)	265	265	265	265	265	265	265
Solubility (g/cm <sup>3</sup> )	7.48E-10						

##### **Outlet Section-Feeders**

Number of Nodes	7						
Temperature (°C)	312	311	311	310	310	310	310
Solubility (g/cm <sup>3</sup> )	1.32E-09						

##### **Steam Generator Section**

Number of Nodes	18						
Surface Area	5.30E+04	3.20E+04	6.00E+05	2.30E+06	2.30E+06	3.20E+06	
	3.20E+06	3.20E+06	3.20E+06	3.20E+06	3.20E+06	9.00E+05	
	9.00E+05	9.00E+05	9.00E+05	9.00E+05	6.00E+05	3.20E+04	

##### **Constants**

Outer Oxide Porosity	0.3
Tortuosity	1.15
Fraction of Fe in Fe <sub>3</sub> O <sub>4</sub>	0.724
Surface Area Factor	1.732
Density of Metal (g/cm <sup>3</sup> )	7.86
Density of Oxide (g/cm <sup>3</sup> )	5.2
Period (hours)	100
Time Increment (seconds)	3600
Symmetry Coefficient	0.5
Iron Molecular Weight (g/mol)	55.9
Magnetite Molecular Weight (g/mol)	231.7
Activation Energy for Fe = Fe <sup>2+</sup> + 2e	140800
Activation Energy for FeOH = Fe <sub>3</sub> O <sub>4</sub>	120000
Activation Energy for Fe <sup>2+</sup> = Fe <sub>3</sub> O <sub>4</sub>	125000
Activation Energy for H <sup>+</sup> = H <sub>2</sub>	90000

**Inlet Section**

Inlet Iron Diffusivity (cm <sup>2</sup> /s)	3.20E-04
Inlet Iron Transport Coefficient (cm <sup>2</sup> /s)	1.71E-04
Inlet Inner Oxide Porosity	0.1
Inlet Inner Oxide Spalling Constant	2.5E+14
Inlet Outer Spalling Constant	6.0E+13
Inlet Density of Water (g/cm <sup>3</sup> )	0.783
Inlet Viscosity of Water (g/cm s)	1.01E-03
Inlet Hydrogen Concentration (g/cm <sup>3</sup> )	2.00E-11
Inlet Hydrogen Diffusivity (cm <sup>2</sup> /s)	7.45E-04
Inlet Supersaturation Factor	1.1

**Outlet Section**

Outlet Iron Diffusivity (cm <sup>2</sup> /s)	4.10E-04
Outlet Iron Transport Coefficient (cm <sup>2</sup> /s)	1.65E-04
Outlet Inner Oxide Porosity	0.1
Outlet Inner Oxide Spalling Constant	0.224
Outlet Outer Spalling Constant	0.01
Outlet Density of Water (g/cm <sup>3</sup> )	0.691
Outlet Viscosity of Water (g/cm s)	8.22E-04
Outlet Hydrogen Concentration (g/cm <sup>3</sup> )	8.00E-10
Outlet Hydrogen Diffusivity (cm <sup>2</sup> /s)	9.93E-04
Outlet Supersaturation Factor	1.1
Favourite outlet feeder 1	84
Favourite outlet feeder 2	40
Favourite outlet feeder 3	21
Favourite outlet feeder 4	86
Favourite outlet feeder 5	64
Print Frequency ratio	5
pH at room temperature	10.2
pH at high temperature	7.71

### A.2 Velocity of outlet feeders

<b>Feeder</b>	<b>Node 1</b>	<b>Feeder</b>	<b>Node 1</b>	<b>Feeder</b>	<b>Node 1</b>
	V (cm/s)		V (cm/s)		V (cm/s)
A09	1229	J01	1302	O10	1678
A11	1378	J03	1425	P03	1352
B06	1187	J05	1617	P05	1625
B08	1578	J07	1667	P07	1725
B10	1199	J09	1657	P09	1682
C05	1305	J11	1663	P11	1708
C07	1180	K02	1228	Q02	1410
C09	1412	K04	1636	Q04	1399
C11	1441	K06	1653	Q06	1623
D04	1333	K08	1647	Q08	1688
D06	1244	K10	1641	Q10	1662
D08	1533	L01	1393	R03	1054
D10	1603	L03	1535	R05	1377
E03	1307	L05	1664	R07	1619
E05	1569	L07	1682	R09	1662
E07	1299	L09	1652	R11	1680
E09	1569	L11	1619	S04	1052
E11	1621	M02	1290	S06	1388
F04	1281	M04	1686	S08	1619
F06	1563	M06	1715	S10	1669
F08	1651	M08	1670	T05	1035
F10	1618	M10	1639	T07	1369
G03	1213	N01	1367	T09	1548
G05	1519	N03	1516	T11	1552
G07	1666	N05	1705	U06	1384
G09	1642	N07	1714	U08	1249
G11	1654	N09	1667	U10	1379
H02	1584	N11	1650	V07	1212
H04	1512	O02	1194	V09	987
H06	1653	O04	1634	V11	1149
H08	1659	O06	1749	W10	1178
H10	1645	O08	1710		

### A.3 Diameter of outlet feeders

<b>Feeder</b>	<b>Node 1 D (cm)</b>	<b>Node 2 D (cm)</b>	<b>Node 3 D (cm)</b>	<b>Node 4 D (cm)</b>	<b>Node 5 D (cm)</b>	<b>Node 6 D (cm)</b>	<b>Node 7 D (cm)</b>
<b>F11</b>	<b>6.4</b>						
A09	5.1	5.1	5.1	5.1	5.1	5.1	5.1
A11	5.1	5.1	5.1	5.1	5.1	5.1	5.1
B06	5.1	5.1	5.1	5.1	5.1	5.1	5.1
B08	5.1	5.1	5.1	6.4	6.4	6.4	6.4
B10	6.4	6.4	6.4	6.4	6.4	6.4	6.4
C05	5.1	5.1	5.1	5.1	5.1	5.1	5.1
C07	6.4	6.4	6.4	6.4	6.4	6.4	6.4
C09	6.4	6.4	6.4	6.4	6.4	6.4	6.4
C11	6.4	6.4	6.4	6.4	6.4	7.6	7.6
D04	5.1	5.1	5.1	5.1	5.1	5.1	5.1
D06	6.4	6.4	6.4	6.4	6.4	6.4	6.4
D08	6.4	6.4	6.4	6.4	6.4	7.6	7.6
D10	6.4	6.4	6.4	6.4	7.6	7.6	7.6
E03	5.1	5.1	5.1	5.1	6.4	6.4	6.4
E05	6.4	6.4	6.4	6.4	6.4	6.4	6.4
E07	6.4	6.4	6.4	6.4	6.4	7.6	7.6
E09	6.4	6.4	6.4	6.4	6.4	8.9	8.9
E11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
F04	6.4	6.4	6.4	6.4	6.4	6.4	6.4
F06	6.4	6.4	6.4	6.4	6.4	7.6	7.6
F08	6.4	6.4	6.4	6.4	6.4	8.9	8.9
F10	6.4	6.4	6.4	6.4	6.4	8.9	8.9
G03	6.4	6.4	6.4	6.4	6.4	6.4	6.4
G05	6.4	6.4	6.4	6.4	6.4	8.9	8.9
G07	6.4	6.4	6.4	6.4	6.4	8.9	8.9
G09	6.4	6.4	6.4	6.4	6.4	8.9	8.9
G11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
H02	6.4	6.4	6.4	6.4	6.4	6.4	6.4
H04	6.4	6.4	6.4	6.4	7.6	7.6	7.6
H06	6.4	6.4	6.4	6.4	6.4	8.9	8.9
H08	6.4	6.4	6.4	6.4	6.4	8.9	8.9
H10	6.4	6.4	6.4	6.4	6.4	8.9	8.9
J01	5.1	5.1	5.1	5.1	5.1	6.4	8.9
J03	6.4	6.4	6.4	6.4	8.9	8.9	8.9
J05	6.4	6.4	6.4	6.4	8.9	8.9	8.9
J07	6.4	6.4	6.4	6.4	8.9	8.9	8.9
J09	6.4	6.4	6.4	6.4	8.9	8.9	8.9
J11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
K02	6.4	6.4	6.4	6.4	6.4	6.4	6.4
K04	6.4	6.4	6.4	6.4	8.9	8.9	8.9
K06	6.4	6.4	6.4	6.4	8.9	8.9	8.9
K08	6.4	6.4	6.4	6.4	8.9	8.9	8.9
K10	6.4	6.4	6.4	6.4	8.9	8.9	8.9
L01	5.1	5.1	5.1	5.1	6.4	6.4	6.4
L03	6.4	6.4	6.4	6.4	7.6	7.6	7.6
L05	6.4	6.4	6.4	8.9	8.9	8.9	8.9

Diameter of outlet feeders (continued)

L07	6.4	6.4	6.4	6.4	8.9	8.9	8.9
L09	6.4	6.4	6.4	6.4	8.9	8.9	8.9
L11	6.4	6.4	6.4	6.4	6.4	8.9	8.9
M02	6.4	6.4	6.4	6.4	6.4	6.4	6.4
M04	6.4	6.4	6.4	6.4	8.9	8.9	8.9
M06	6.4	6.4	6.4	6.4	8.9	8.9	8.9
M08	6.4	6.4	6.4	6.4	8.9	8.9	8.9
M10	6.4	6.4	6.4	6.4	8.9	8.9	8.9
N01	5.1	5.1	5.1	5.1	6.4	6.4	6.4
N03	6.4	6.4	6.4	6.4	7.6	7.6	7.6
N05	6.4	6.4	6.4	6.4	8.9	8.9	8.9
N07	6.4	6.4	6.4	6.4	8.9	8.9	8.9
N09	6.4	6.4	6.4	6.4	8.9	8.9	8.9
N11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
O02	6.4	6.4	6.4	6.4	6.4	6.4	6.4
O04	6.4	6.4	6.4	6.4	8.9	8.9	8.9
O06	6.4	6.4	6.4	6.4	8.9	8.9	8.9
O08	6.4	6.4	6.4	6.4	8.9	8.9	8.9
O10	6.4	6.4	6.4	6.4	8.9	8.9	8.9
P03	6.4	6.4	6.4	7.6	7.6	7.6	7.6
P05	6.4	6.4	6.4	6.4	8.9	8.9	8.9
P07	6.4	6.4	6.4	6.4	8.9	8.9	8.9
P09	6.4	6.4	6.4	6.4	8.9	8.9	8.9
P11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
Q02	5.1	5.1	5.1	6.4	6.4	6.4	6.4
Q04	6.4	6.4	6.4	6.4	7.6	7.6	7.6
Q06	6.4	6.4	6.4	6.4	8.9	8.9	8.9
Q08	6.4	6.4	6.4	6.4	8.9	8.9	8.9
Q10	6.4	6.4	6.4	6.4	8.9	8.9	8.9
R03	6.4	6.4	6.4	6.4	6.4	6.4	6.4
R05	6.4	6.4	6.4	6.4	7.6	7.6	7.6
R07	6.4	6.4	6.4	6.4	8.9	8.9	8.9
R09	6.4	6.4	6.4	6.4	8.9	8.9	8.9
R11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
S04	6.4	6.4	6.4	6.4	6.4	6.4	6.4
S06	6.4	6.4	6.4	6.4	7.6	7.6	7.6
S08	6.4	6.4	6.4	6.4	8.9	8.9	8.9
S10	6.4	6.4	6.4	8.9	8.9	8.9	8.9
T05	6.4	6.4	6.4	6.4	6.4	6.4	6.4
T07	6.4	6.4	6.4	6.4	7.6	7.6	7.6
T09	6.4	6.4	6.4	6.4	8.9	8.9	8.9
T11	6.4	6.4	6.4	6.4	8.9	8.9	8.9
U06	5.1	5.1	5.1	5.1	6.4	6.4	6.4
U08	6.4	6.4	6.4	6.4	7.6	7.6	7.6
U10	6.4	6.4	6.4	6.4	7.6	7.6	7.6
V07	5.1	5.1	5.1	5.1	6.4	6.4	6.4
V09	6.4	6.4	6.4	6.4	6.4	6.4	6.4
V11	6.4	6.4	6.4	6.4	6.4	6.4	6.4
W10	5.1	5.1	5.1	5.1	6.4	6.4	6.4

#### A.4 Length of outlet feeders

<b>Feeder</b>	<b>Node 1 L (cm)</b>	<b>Node 2 L (cm)</b>	<b>Node 3 L (cm)</b>	<b>Node 4 L (cm)</b>	<b>Node 5 L (cm)</b>	<b>Node 6 L (cm)</b>	<b>Node 7 L (cm)</b>
F11	17	3.5	1	1	1	1	1
A09	17	3.5	107.0	188.2	188.2	188.2	188.2
A11	17	3.5	107.0	189.4	189.4	189.4	189.4
B06	17	3.5	107.0	189.4	189.4	189.4	189.4
B08	17	3.5	107.0	160.0	225.5	225.5	225.5
B10	17	3.5	139.5	197.5	197.5	197.5	197.5
C05	17	3.5	107.0	206.9	206.9	206.9	206.9
C07	17	3.5	139.5	202.5	202.5	202.5	202.5
C09	17	3.5	139.5	208.8	208.8	208.8	208.8
C11	17	3.5	139.5	229.0	229.0	190.0	176.0
D04	17	3.5	107.0	215.7	215.7	215.7	215.7
D06	17	3.5	139.5	215.0	215.0	215.0	215.0
D08	17	3.5	139.5	176.5	176.5	190.0	326.0
D10	17	3.5	139.5	170.0	190.0	248.5	248.5
E03	17	3.5	107.0	226.5	160.0	271.0	271.0
E05	17	3.5	139.5	212.5	212.5	212.5	212.5
E07	17	3.5	139.5	208.0	208.0	190.0	240.0
E09	17	3.5	139.5	208.0	208.0	222.5	211.5
E11	17	3.5	139.5	210.0	222.5	208.3	208.3
F04	17	3.5	139.5	231.3	231.3	231.3	231.3
F06	17	3.5	139.5	227.5	227.5	190.0	246.0
F08	17	3.5	139.5	227.5	227.5	222.5	213.5
F10	17	3.5	139.5	227.5	227.5	222.5	213.5
G03	17	3.5	139.5	238.8	238.8	238.8	238.8
G05	17	3.5	139.5	263.5	263.5	222.5	141.5
G07	17	3.5	139.5	259.0	259.0	222.5	173.5
G09	17	3.5	139.5	259.0	259.0	222.5	173.5
G11	17	3.5	139.5	274.0	222.5	193.3	193.3
H02	17	3.5	139.5	230.0	230.0	230.0	230.0
H04	17	3.5	139.5	292.0	190.0	223.0	223.0
H06	17	3.5	139.5	284.5	284.5	222.5	171.5
H08	17	3.5	139.5	284.5	284.5	222.5	179.5
H10	17	3.5	139.5	284.5	284.5	222.5	171.5
J01	17	3.5	107.0	497.3	497.3	506.0	637.0
J03	17	3.5	139.5	334.0	222.5	206.8	206.8
J05	17	3.5	139.5	346.0	222.5	206.8	206.8
J07	17	3.5	139.5	346.0	222.5	206.8	206.8
J09	17	3.5	139.5	346.0	222.5	206.8	206.8
J11	17	3.5	139.5	346.0	222.5	206.8	206.8
K02	17	3.5	139.5	247.5	247.5	247.5	247.5
K04	17	3.5	139.5	364.0	222.5	208.3	208.3
K06	17	3.5	139.5	380.0	222.5	209.8	209.8
K08	17	3.5	139.5	380.0	222.5	209.8	209.8
K10	17	3.5	139.5	380.0	222.5	209.8	209.8
L01	17	3.5	107.0	186.5	160.0	360.0	360.0
L03	17	3.5	139.5	221.0	190.0	351.0	351.0

## Length of outlet feeders (continued)

L05	17	3.5	139.5	222.5	299.8	299.8	299.8
L07	17	3.5	139.5	230.0	222.5	418.8	418.8
L09	17	3.5	139.5	230.0	222.5	418.8	418.8
L11	17	3.5	139.5	319.0	319.0	222.5	143.5
M02	17	3.5	139.5	296.3	296.3	296.3	296.3
M04	17	3.5	139.5	245.0	222.5	372.8	372.8
M06	17	3.5	139.5	151.0	222.5	440.8	440.8
M08	17	3.5	139.5	221.0	222.5	438.8	438.8
M10	17	3.5	139.5	376.0	222.5	395.8	395.8
N01	17	3.5	107.0	290.5	160.0	368.5	368.5
N03	17	3.5	139.5	236.0	190.0	409.0	409.0
N05	17	3.5	139.5	157.0	222.5	434.8	434.8
N07	17	3.5	139.5	332.0	222.5	365.8	365.8
N09	17	3.5	139.5	303.0	222.5	428.8	428.8
N11	17	3.5	139.5	462.0	222.5	206.8	206.8
O02	17	3.5	139.5	313.8	313.8	313.8	313.8
O04	17	3.5	139.5	352.0	222.5	359.8	359.8
O06	17	3.5	139.5	358.0	222.5	376.3	376.3
O08	17	3.5	139.5	294.0	222.5	442.8	442.8
O10	17	3.5	139.5	356.0	222.5	443.8	443.8
P03	17	3.5	139.5	190.0	380.0	380.0	380.0
P05	17	3.5	139.5	358.0	222.5	395.8	395.8
P07	17	3.5	139.5	358.0	222.5	413.8	413.8
P09	17	3.5	139.5	358.0	222.5	449.8	449.8
P11	17	3.5	139.5	434.0	222.5	446.8	446.8
Q02	17	3.5	107.0	160.0	409.9	409.9	409.9
Q04	17	3.5	139.5	416.0	190.0	410.5	410.5
Q06	17	3.5	139.5	136.0	222.5	522.8	522.8
Q08	17	3.5	139.5	352.0	222.5	452.8	452.8
Q10	17	3.5	139.5	306.0	222.5	506.8	506.8
R03	17	3.5	139.5	353.8	353.8	353.8	353.8
R05	17	3.5	139.5	294.0	190.0	517.0	517.0
R07	17	3.5	139.5	440.0	222.5	413.8	413.8
R09	17	3.5	139.5	349.0	222.5	498.8	498.8
R11	17	3.5	139.5	312.0	222.5	544.8	544.8
S04	17	3.5	139.5	361.3	361.3	361.3	361.3
S06	17	3.5	139.5	380.0	190.0	464.5	464.5
S08	17	3.5	139.5	432.0	222.5	460.3	460.3
S10	17	3.5	139.5	222.5	468.8	468.8	468.8
T05	17	3.5	139.5	388.8	388.8	388.8	388.8
T07	17	3.5	139.5	702.0	190.0	331.0	331.0
T09	17	3.5	139.5	267.0	222.5	568.8	568.8
T11	17	3.5	139.5	410.0	222.5	544.3	544.3
U06	17	3.5	107.0	375.5	160.0	544.0	544.0
U08	17	3.5	139.5	236.0	222.5	596.8	596.8
U10	17	3.5	139.5	306.0	222.5	596.8	596.8
V07	17	3.5	107.0	430.5	160.0	556.0	556.0
V09	17	3.5	139.5	430.0	430.0	430.0	430.0
V11	17	3.5	139.5	448.8	448.8	448.8	448.8
W10	17	3.5	107.0	570.5	160.0	539.5	539.5

## **Appendix B Program code for feeder pipes**

It is similar to the previous program[Lang, 2000], but some parts in the program were added. See more in attached CD.

The solubility computation based on Tremaine and Leblanc's work [1980] was added. The expression for dissolution rate constant was corrected to be more accurate.

## Appendix C Program code for test probe

" Probe variables

Dim di, Velo, dis, ARE As Single

" To calculate solubility

Dim TC1 As Variant 'Inlet temperature

Dim TC2 As Variant 'Outlet temperature

Dim ph1 As Variant

Dim sol1 As Double 'solubility of magnetite at inlet

Dim sol2 As Double 'solubility of magnetite at outlet

Dim delta As Double 'the difference of solubility between inlet and outlet

Dim Sat As Single

" Variables and constants for T&L model

Dim KW As Double 'Dissociation constant of water

Dim KFE2 As Double 'Solubility constant of Fe<sup>2+</sup>

Dim KFEOH As Double 'T&L: Hydrolysis constant of FeOH<sup>+</sup>

Dim KFEOH2 As Double 'T&L: Hydrolysis constant of Fe(OH)<sup>2</sup>

Dim KFEOH3\_ As Double 'T&L: Hydrolysis constant of Fe(OH)<sup>3-</sup>

Dim KFEOH3 As Double 'Solubility constant of Fe(OH)<sup>3</sup>

Dim KFEOH4 As Double 'Solubility constant of Fe(OH)<sup>4-</sup>

\*\*\*\* FOR T&L: The solubility/hydrolysis constants are

'evaluated as the 4th order polynomial fit of the Tremaine and LeBlanc (1980) 'data at 100, 150, 200, 250 and 300 deg C.

Dim MFE2 As Double 'Molal concentration of Fe<sup>2+</sup>

Dim MFEOH As Double 'Molal concentration of FeOH<sup>+</sup>

Dim MFEOH2 As Double 'Molal concentration of Fe(OH)<sup>2</sup>

Dim MFEOH3\_ As Double 'Molal concentration of Fe(OH)<sup>3-</sup>

Dim MFEOH3 As Double 'Molal concentration of Fe(OH)<sup>3</sup>

Dim MFEOH4 As Double 'Molal concentration of Fe(OH)<sup>4-</sup>

Dim MFESAT As Double 'Solubility of Magnetite

Dim MH As Double 'Molal concentration of acid at Temp = TC

Dim mHCL As Double 'Molal concentration of HCl at beginning of rxn  
 Dim Value As Double 'The answer that appears in the text box  
 Dim mhtemp As Double 'Variable used in Do Loop (see functions)  
 Dim TK As Variant 'temperature in Kelvin  
 Dim MatDen  
 Dim sf 'scaling factor

Const pi = 22 / 7  
 Const Tconst = 273.15

#### ' COMMON VARIABLES

Dim INCR, PERIOD As Long  
 Dim OP2, TOR, FE, SA, FDEN, OXDEN, H2, PORCON As Single  
 Dim BETA, MW, MWFE3O4  
 Dim AEFEFE, AEFEFEO, AEHH, AEFEHOH  
 Dim CounterParticle(125), Counter(125) As Long  
 Dim CorrRate(210000), OxideThick(210000), OSSolubility(210000),  
 Solubility(210000)

#### ' Variables for Spalling

Dim SIZE(125), OuterSize(125), ISize(125), IOuterSize(125) As Long  
 Dim LABEL(125, 7), OuterLabel(125, 7), ILabel(125, 7), IOuterLabel(125, 7)  
 Dim Num, OuterNum, INum  
 Dim InnerCon, OuterCon

#### ' OUTLET SECTION VARIABLES

Dim OFDIF, OTRANS, OH2DIF, oop1, OVIS, OWDEN, OF, OH2  
 Dim HD, HDH, KD, ELEKD  
 Dim OSAT, CW, CB, CS, H2W, H2OS  
 Dim CP, DC, MET, OAveCorr  
 Dim OX1, OX2, OAveOx  
 Dim POT, OSPOT, MOEXFE, EeFEFE, MOEeH2H, EeFEOHFEO,

MOEXFEOH, MOEXH

```
Private Sub CommandButton1_Click()
```

```
End Sub
```

```
Private Sub Area_Click()
```

```
End Sub
```

```
Private Sub AreaBox_Change()
```

```
End Sub
```

```
Private Sub cmdstart_Click()
```

```
    PERIOD = Val(PERIODBox.Text)
```

```
    DriveSpec = "C:"
```

```
    DirectorySpec = "C:\Users\Orawee\Input"
```

```
    FileSpec = "Microsoft Excel Files (*.xls),*.xls"
```

```
    Call fileFind(fileToOpen, DriveSpec, DirectorySpec, FileSpec)
```

```
    Call DataLoad(fileToOpen, PERIOD)
```

```
    Call ITERATE(PERIOD)
```

```
End Sub
```

```
Private Sub DelSol_Click()
```

```
End Sub
```

```
Private Sub DelSolBox_Change()
```

```
End Sub
```

```
Private Sub Dia_Click()
```

```
End Sub
```

```
Private Sub DiaBox_Change()
    di = Val(DiaBox.Text)
    dis = Val(LenBox.Text)
    ARE = pi * di * dis
    AreaBox.Text = Format(ARE, "#####.##")
End Sub
```

```
Private Sub Frame1_Click()
```

```
End Sub
```

```
Private Sub InSol_Click()
```

```
End Sub
```

```
Private Sub InSolBox_Change()
```

```
End Sub
```

```
Private Sub lblspall_Click()
```

```
End Sub
```

```
Private Sub LenBox_Change()
```

```
    di = Val(DiaBox.Text)
    dis = Val(LenBox.Text)
    ARE = pi * di * dis
    AreaBox.Text = Format(ARE, "#####.##")
End Sub
```

```
Private Sub Length_Click()
```

```
End Sub
```

```
Private Sub Note_Click()
```

```
End Sub
```

```
Private Sub OutSol_Click()
```

```
End Sub
```

```
Private Sub OutSolBox_Change()
```

```
End Sub
```

```
Private Sub PERIODBox_Change()
```

```
End Sub
```

```
Private Sub pH_Click()
```

```
End Sub
```

```
Private Sub pHBox_Change()
```

```
    ph1 = Val(pHBox.Text)
```

```
    Call TC1Box_Change
```

```
    Call TC2Box_Change
```

```
End Sub
```

```
Private Sub SFBox_Change()
```

```
    sf = Val(SFBox.Text)
```

```
End Sub
```

```
Private Sub SpallBox_Change()
```

```
End Sub
```

```
Private Sub SSENHBox_Change()
    SSENH = Val(SSENHBox.Text)
End Sub
```

```
Private Sub TC1Box_Change()
    TC1 = Val(TC1Box.Text)
    Call TL_solubility(TC1, ph1, sol1)
    InSolBox.Text = sol1
End Sub
```

```
Private Sub TC2Box_Change()
    TC2 = Val(TC2Box.Text)
    Call TL_solubility(TC2, ph1, sol2)
    OutSolBox.Text = sol2
    delta = Abs(sol2 - sol1)
    DelSolBox.Text = delta
End Sub
```

```
Private Sub Temp1_Click()
End Sub
```

```
Private Sub Temp2_Click()
End Sub
```

```
Sub fileFind(fileToOpen, DriveSpec, DirectorySpec, FileSpec)
    ChDrive DriveSpec
    ChDir DirectorySpec
    fileToOpen = Application.GetOpenFilename(FileSpec)
    If fileToOpen <> False Then
        End If
End Sub
```

```
Private Sub Userform_Activate()

End Sub

Sub DataLoad(fileToOpen, PERIOD)

    ' OPEN APPROPRIATE WORKBOOK
    Workbooks.Open FileName:=fileToOpen
    DataWorkbook = ActiveWorkbook.Name
    UserForm1.Repaint

    ' LOAD MISCELLANEOUS CONSTANTS AND VARIABLES
    Sheets("probe").Select
    OP2 = Range("B2").Value
    TOR = Range("B3").Value
    FE = Range("B4").Value
    SA = Range("B5").Value
    FDEN = Range("B6").Value
    OXDEN = Range("B7").Value
    INCR = Range("B8").Value
    BETA = Range("B9").Value
    MW = Range("B10").Value
    MWFE3O4 = Range("B11").Value
    AEEFEFE = Range("B12").Value
    AEFEEOH = Range("B13").Value
    AEFEFEO = Range("B14").Value
    AEHH = Range("B15").Value
    IFDIF = Range("B17").Value
    ITRANS = Range("B18").Value
    iop1 = Range("B19").Value
    IIInnerCon = Range("B20").Value
    IOOuterCon = Range("B21").Value
    iwden = Range("B22").Value
    ivis = Range("B23").Value

```

```

IH2 = Range("B24").Value
IH2DIF = Range("B25").Value
IFactor = Range("B26").Value
OFDIF = Range("B28").Value
OTRANS = Range("B29").Value
oop1 = Range("B30").Value
InnerCon = Range("B31").Value
OuterCon = Range("B32").Value
OWDEN = Range("B33").Value
OVIS = Range("B34").Value
OH2 = Range("B35").Value
OH2DIF = Range("B36").Value
OF = Range("B37").Value
pht = range ("B39").Value

```

#### ' OPTIONAL LOADING OF SPALLING DATA

```

Sheets("spall").Select
For a = 1 To 125
    SIZE(a) = Range(Cells(a + 3, 1), Cells(a + 3, 1)).Value
    For B = 1 To 7
        LABEL(a, B) = Range(Cells(a + 3, B + 1), Cells(a + 3, B + 1)).Value
    Next B
    OuterSize(a) = Range(Cells(a + 3, 9), Cells(a + 3, 9)).Value
    OuterLabel(a, 1) = Range(Cells(a + 3, 10), Cells(a + 3, 10)).Value
    ISize(a) = Range(Cells(a + 3, 11), Cells(a + 3, 11)).Value
    ILabel(a, 1) = Range(Cells(a + 3, 12), Cells(a + 3, 12)).Value
    IOuterSize(a) = Range(Cells(a + 3, 13), Cells(a + 3, 13)).Value
    IOuterLabel(a, 1) = Range(Cells(a + 3, 14), Cells(a + 3, 14)).Value
    CounterParticle(a) = Range(Cells(a + 3, 15), Cells(a + 3, 15)).Value
Next a
PORCON = 1#
Call DataInitialize(PERIOD)
End Sub
Sub DataInitialize(PERIOD)
    Call COEFF(di, Velo, OH2DIF, OWDEN, OVIS, HD, KD, HDH, TC2)

```

```
Call initials(OH2, sol2, OF, KD, Num, OuterNum, Sat, H2W, CW, CB, CS, CP, OX1,
OX2, MET, ELEKD)
```

```
End Sub
```

```
Sub COEFF(di, Velo, H2DIF, WDEN, VIS, HD, KD, HDH, TC)
```

```
V = Velo * 0.01
```

```
B = di * 0.01
```

```
MTENH = 1.3 * 31.2 * VIS ^ 0.24 / (WDEN ^ 0.24 * di ^ 0.24 * Velo ^ 0.24)
```

```
HD = MTENH * 100 * 0.000637 * V ^ (0.82) / B ^ (0.18)
```

```
KD = sf * Exp(0.0879 - 3569 / (TC + 273))
```

```
HDH = 0.02 * H2DIF ^ 0.667 * Velo ^ 0.83 * WDEN ^ 0.5 / (di ^ 0.17 * VIS ^ 0.5)
```

```
End Sub
```

```
Sub initials(H2, Sol, Factor, KD, Num, OuterNum, Sat, H2W, CW, CB, CS, CP, OX1,
OX2, MET, ELEKD)
```

```
Randomize
```

```
Num = Rnd()
```

```
OuterNum = Rnd()
```

```
Sat = (10 / 18) ^ (1# / 3#) * Sol
```

```
H2W = H2
```

```
CW = Factor * Sat
```

```
CB = Sat
```

```
CS = Sat
```

```
CP = 0#
```

```
OX1 = 0.00025
```

```
OX2 = 0.00025
```

```
MET = 0#
```

```
ELEKD = KD
```

```
End Sub
```

```
Sub ITERATE(PERIOD)
```

```
Dim K As Long
```

```
PrintIter = 0
```

```

For K = 1 To PERIOD + 1
    UserForm1.lblIterate.Caption = K
    UserForm1.Repaint
    CorrosionTime = (K - 1)
    CBtot = 0
    CPtot = 0
    Flowtot = 0
    Isat = (10 / 18) ^ (1# / 3#) * sol1
    ICB = Isat
    CB = ICB

    InnerCon = Val(SpallBox.Text)
    If (K - 1) < 0# Then
        Call SPALL(LABEL, OuterLabel, SIZE, OuterSize, Num, OuterNum,
                   TIMECOU, OUTERTIMECOU, di, dis, Velo, Sat, CS, CP,
                   OX1, OX2, ELEKD, InnerCon, OuterCon, CounterParticle,
                   Counter, oop1)
    End If
    Call corrode(PERIOD, TC2, HD, KD, ELEKD, HDH, di, Velo, dis, Sat, OF, DC,
                 CW, CB, CS, OX1, OX2, MET, OFDIF, OTRANS, OH2DIF, oop1,
                 OH2W, OH2OS, OAveOx, OAveCorr, OH2, POT, OSPOT, MOEXFE,
                 EeFEFE, MOEeH2H, EeFEOHFEO, MOEXFEOH, MOEXH)
    If (CorrosionTime = PERIOD) Then
        OAveOx = OAveOx / 2000
        OAveCorr = OAveCorr / 2000
    End If

    If (K < 1 And K < PERIOD) Then
        CorrRate(K) = DC * 8640000000#
        OxideThick(K) = OX1 * 10000# / (OXDEN * (1 - (oop1 + OOX1 * PORCON)))
        OSSolubility(K) = Sat
        Solubility(K) = CW
    End If

```

```

If (K = PERIOD) Then
    Results3 = MET * 10000# / (FDEN * (K - 1) / 24# / 365#)
End If

```

' ASSIGN DATA TO BE SAVED TO RESULTS ARRAY

```
If (K - 1) Mod (Int(PERIOD / 100) + 1) = 0 And K <> 1 Then
```

```
    PrintIter = PrintIter + 1
```

```
    Call OutputDataVariables(DC, CS, OX1, OX2, POT, OSPOT)
```

```
End If
```

```
Next K
```

```
End Sub
```

```
Sub SPALL(ILabel, OLABEL, ISize, OSIZE, INum, ONum, ITIMEcou, OTIMEcou,
di, dis, Velo, Sat, CS, CP, IOX, OOX, KD, ISPALLCON, OSPALLCON,
CounterParticle, Counter, op1)
```

' SPALLING PORTION OF THE PROGRAM

```
Dim OXIDE, OOXIDE, IPARSIZE, OPARSIZE, ISPALLT, OSPALLT
```

```
Dim StuckParticle, IPAR, OPAR, PAR, SSENH
```

' ASSIGNING THE APPROPRIATE PARTICLE SIZE DISTRIBUTION TO  
' THE APPROPRIATE VELOCITY PROBE

```
E = 1
```

```
F = 125
```

```
C = 125
```

```
SSENH = Val(SSENHBox.Text)
```

```
If (Velo <= 1099) Then
```

```
    B = 1
```

```
ElseIf (Velo > 1100 And Velo <= 1199) Then
```

```

B = 2
ElseIf (Velo >= 1200 And Velo <= 1299) Then
    B = 3
ElseIf (Velo >= 1300 And Velo <= 1399) Then
    B = 4
ElseIf (Velo >= 1400 And Velo <= 1499) Then
    B = 5
ElseIf (Velo >= 1500 And Velo <= 1599) Then
    B = 6
ElseIf (Velo >= 1600) Then
    B = 7
End If

```

## ' SPALLING FOR THE OUTER OXIDE

```

If (OOX > 0#) Then
22   For D = 1 To F
      If (ONUM < OLABEL(D, E) And ONUM >= OLABEL((D - 1), E)) Then
          OPARSIZE = OSIZE(D - 1)
      End If
      Next D
      OOXIDE = OOX * 10000 / OXDEN - OPARSIZE
      If (OOXIDE < 0.1 And OOXIDE > 0#) Then
          Randomize
          ONUM = Rnd()
          GoTo 22
      Else
          GoTo 23
      End If
23   OSPALLT = OSPALLCON * OPARSIZE / (SSENH * Velo ^ 2 * KD * (Sat -
                           CS))
      OTIMECOU = OTIMECOU + 3600
      If (OTIMECOU + 1800 >= OSPALLT And OSPALLT > 0) Then
          OOXIDE = OOX * 10000 / OXDEN - OPARSIZE
          StuckParticle = OPARSIZE - OOX * 10000 / OXDEN

```

```

OOX = OOXIDE * 0.0001 * OXDEN
If (OOX < 0# And StuckParticle > 0#) Then
    OOX = 0#
    IOXIDE = IOX * 10000 / OXDEN - StuckParticle
    IOX = IOXIDE * 0.0001 * OXDEN
End If
If (OPARSIZE > 0.2) Then
    OPAR = 0.000000111 * OPARSIZE * OXDEN * dis / (di * Velo)
Else
    OPAR = 0#
End If
For Y = 1 To 125
    If (CounterParticle(Y) = OPARSIZE) Then
        Counter(Y) = Counter(Y) + 1
    End If
    Next Y
    Randomize
    ONUM = Rnd()
    OTIMECOU = 0#
End If
Else
    OPAR = 0#
End If

```

## ' SPALLING FOR THE INNER OXIDE

```

For a = 1 To C
    If (INum < ILabel(a, B) And INum >= ILabel((a - 1), B)) Then
        IPARSIZE = ISize(a - 1)
    End If
    Next a
    ISPALLT = ISPALLCON * IPARSIZE / (SSENH * Velo ^ 2 * KD * (Sat - CS) * (op1
        + IOX * PORCON))
    ITimeCou = ITimeCou + 3600
    If (ITimeCou + 1800 >= ISPALLT And OOX = 0# And ISPALLT > 0) Then

```

```

IOXIDE = IOX * 10000 / OXDEN - IPARSIZE
IOX = IOXIDE * 0.0001 * OXDEN
If (IOX < 0#) Then
    IOX = 0#
End If
If (IPARSIZE > 0.2) Then
    IPAR = 0.000001111 * IPARSIZE * OXDEN * dis / (di * Velo)
Else
    IPAR = 0#
End If
For W = 1 To 125
    If (CounterParticle(W) = IPARSIZE) Then
        Counter(W) = Counter(W) + 1
    End If
    Next W
    Randomize
    INum = Rnd()
    ITIMEcou = 0#
Else
    IPAR = 0#
End If

PAR = IPAR + OPAR
Call Particulate(CP, PAR)

End Sub

Sub Particulate(CP, Particulate)
    CP = CP + Particulate
End Sub

Sub corrode(PERIOD, TC, HD, CKD, ELEKD, HDH, di, Velo, dis, Sat, F, DC, CW, CB,
           OX1, OX2, MET, FDIF, TRANS, H2DIF, op1, H2W, H2OS, AveOx, AveCorr,
           H2, POT, OSPOT, MOEXFE, EeFEFE, MOEeH2H, EeFEOHFEO,
           MOEXFEOH, MOEXH)

```

```

Dim ITER As Integer
Dim DD, MM, SOX, DD2, SOX2, SMET, AA(10)
Dim KK(4), KK2(4), HH(4)

CBB = CB
SOX = OX1
SOX2 = OX2
SMET = MET
' CORROSION RATE
ITER = 0

49 ITER = ITER + 1
AA(1) = 0.476 * (1.101 + (op1 + SOX * PORCON)) / (HD + ELEKD * SA)
AA(2) = 0.476 * (1.101 + (op1 + SOX * PORCON)) * TOR / (FDIF * OXDEN * (op1
+ SOX * PORCON) * (1 - (op1 + SOX * PORCON)))
AA(3) = ELEKD * SA / (HD + ELEKD * SA)
AA(4) = 0.476 * (1.101 + (op1 + SOX * PORCON)) * TOR / (FDIF * OXDEN * OP2
* (1 - OP2))

40 MM = 1 / (AA(1) + AA(2) * SOX + AA(4) * SOX2) * (CW - AA(3) * Sat - (1 - AA
(3)) * CBB) + (TRANS * CW * 2 * 96500 * (POT - OSPOT) / (8.314 * (TC
+ 273)) / (AA(1) * FDIF * op1 / TOR + AA(2) * op1 * FDIF / TOR * SOX))

If (MM <= 0) Then
    MM = 0
End If

If (CBB >= Sat) Then
    CS = 1 / (HD - ELEKD * SA) * (0.476 * (1.101 + (op1 + SOX * PORCON)) * MM
+ HD * CBB - ELEKD * SA * Sat)
    ' Falling if keeps CS negative
    If (CS <= 0) Then
        CS = 1E-300
    End If

    DD = 0.658 * (1 - (op1 + SOX * PORCON)) * MM
    DD2 = (-ELEKD * 35 * SA * (Sat - CS)) / FE
Else

```

```

CS = 1 / (HD + ELEKD * SA) * (0.476 * (1.101 + (op1 + SOX * PORCON)) * MM +
    ELEKD * SA * Sat + HD * CBB)

' Falling if keeps CS negative
If (CS <= 0) Then
    CS = 1E-300
End If

If (SOX <> 0 And CS < Sat And SOX2 = 0) Then
    LAMBDA = 1#
Else
    LAMBDA = 0#
End If

DD = 0.658 * (1 - (op1 + SOX * PORCON)) * MM - LAMBDA / FE * ELEKD * SA *
    (Sat - CS)
If (SOX2 > 0) Then
    DD2 = (-ELEKD * SA * (Sat - CS)) / FE
End If
End If

H2W = H2 + 0.1 * (0.00566 * MM * (7.32 - (op1 + SOX * PORCON)) * OX1 * TOR /
    (OXDEN * (op1 + SOX * PORCON) * (1 - (op1 + SOX * PORCON))) / H2DIF)
H2OS = H2 + 0.1 * 0.00566 * MM * (7.32 - op1) / HDH

' TOTAL INNER LAYER AND METAL LOSS
KK(ITER) = INCR * DD
KK2(ITER) = INCR * DD2
HH(ITER) = INCR * MM

If ITER < 4 Then
    SOX = OX1 + KK(ITER) / 2#
    SOX2 = OX2 + KK2(ITER) / 2#
    If (SOX2 < 0) Then
        SOX2 = 0
    End If
    SMET = MET + HH(ITER) / 2#
End If

```

Select Case ITER

Case Is = 1

DOX = DD

DC = MM

GoTo 49

Case Is = 2

GoTo 49

Case Is = 3

GoTo 49

Case Is = 4

OX1 = OX1 + (KK(1) + 2 \* KK(2) + 2 \* KK(3) + KK(4)) / 6#

If (OX1 < 0) Then

OX1 = 0

End If

OX2 = OX2 + (KK2(1) + 2 \* KK2(2) + 2 \* KK2(3) + KK2(4)) / 6#

If (OX2 < 0) Then

OX2 = 0

End If

MET = MET + (HH(1) + 2 \* HH(2) + 2 \* HH(3) + HH(4)) / 6#

If (MET < 0) Then

MET = 0

End If

DC = (HH(1) + 2 \* HH(2) + 2 \* HH(3) + HH(4)) / 6# / 3600

If (CorrosionTime > (PERIOD - 2000)) Then

AveOx = AveOx + OX1

AveCorr = AveCorr + (HH(1) + 2 \* HH(2) + 2 \* HH(3) + HH(4)) / 6#

End If

End Select

If (K <> 1) Then

Call Potential(DC, OX1, op1, H2, H2W, H2OS, CKD, ELEKD, TC, di, CW,

CB, CS, Sat, POT, OSPOT, MOEXFE, EeFEFE, MOEeH2H,

EeFEOHFE, MOEXFEOH, MOEXH)

End If

Call Bulk(di, Velo, dis, HD, CB, CS)

End Sub

Sub Potential(CR, OX1, op1, H2, H2W, H2OS, CKD, KD, TC, di, CW, CB, CS, Sat,  
POT, OSPOT, MOEXFE, EeFEFE, MOEeH2H, EeFEOHFEO,  
MOEXFEOH, MOEXH)

Dim OSEXFE, OSEXH As Double

Dim EeFEOF, EeHH2, Csol, KDMO

Dim Tafel1, FEFEOH, FEFEOHR, FEOHFEO, FEOHFEOR, Tafel2, CorrCur,  
EeH2H, EeFEFEO

Dim ExchangeCon, T, pHCal

Dim PrecCorrCur, PrecPot

If (CR > 0) Then

'COMPUTING THE EFFECT OF POTENTIAL AT THE METAL-OXIDE INTERFACE

'COMPUTING THE EQUILIBRIUM POTENTIAL FOR EACH REACTION AT M-O  
INTERFACE

$$\text{EeFEFE} = 0.14 - 0.000198 * (\text{TC} + 273) * \text{pht} + 0.0000431 * (\text{TC} + 273) * \text{Log}(11 * \text{CW})$$

$$\text{EeFEOHFEO} = -0.75 - 0.000198 * (\text{TC} + 273) * \text{pht} - 0.0001292 * \text{Log}(\text{CW} * 11) * (\text{TC} + 273)$$

$$\text{MOEeH2H} = -0.0000431 * (\text{TC} + 273) * \text{Log}(\text{H2W}) - 0.000198 * (\text{TC} + 273) * \text{pht}$$

'COMPUTING THE EXCHANGE CURRENT FOR EACH REACTION AT THE M-O  
INTERFACE

$$\text{MOEXFE} = 2.011\text{E+15} * (\text{TC} + 273) * \text{Exp}(-\text{AEFEFE} / 8.314 / (\text{TC} + 273)) * \text{CW} * 1000 / 89.9 * \text{Exp}(11607 * \text{EeFEFE} / (\text{TC} + 273))$$

$$\text{MOEXFEOH} = 2.011\text{E+15} * (\text{TC} + 273) * \text{Exp}(-\text{AEFEOH} / 8.314 / (\text{TC} + 273)) * \text{CW} * 1000 / 89.9 * \text{Exp}(1934 * \text{EeFEOHFEO} / (\text{TC} + 273))$$

$$\text{MOEXH} = 2.011\text{E+15} * (\text{TC} + 273) * \text{Exp}(-\text{AEHH} / 8.314 / (\text{TC} + 273)) * \text{H2W} * 1000 / 2.016 * \text{Exp}(5803 / 3 * 7 * \text{MOEeH2H} / (\text{TC} + 273))$$

```

'SETTING UP CONSTANTS FOR BUTLER-VOLMER EXPRESSION
CorrCur = 2 * 96500 * CR / MW
PrecCorrCur = 2 / 3 * 96500 * 0.476 * (1 - (op1 + OX1 * PORCON)) * CR / 89.9
FEFEOH = MOEXFE / Exp(11607 / (TC + 273) * EeFEFE)
FEFEOHR = MOEXFE * Exp(11607 / (TC + 273) * EeFEFE)
Tafel1 = 11607 / (TC + 273)
FEOHFEO = MOEXFEOH / Exp(1934 / (TC + 273) * EeFEOHFEO)
FEOHFEO = MOEXFEOH * Exp(1934 / (TC + 273) * EeFEOHFEO)
Tafel2 = 1934 / (TC + 273)
Call BisectPotential(FEFEOH, FEFEOHR, Tafel1, CorrCur, POT)
End If

'COMPUTE THE EFFECT OF POTENTIAL AT THE OXIDE-SOLUTION INTERFACE
EeH2H = -0.0000431 * (TC + 273) * Log(H2OS) - 0.000198 * (TC + 273) * pht
EeFEFEO = -0.75 - 0.000198 * (TC + 273) * pht - 0.000129 * (TC + 273) * Log(11 * CS)
OSEXFE = 2.011E+15 * (TC + 273) * Exp(-AEFEEFO / 8.314 / (TC + 273)) * CS * 1000
    / 89.9 * di / 4 * Exp(5803 / 3 * EeFEFEO / (TC + 273))
OSEXH = 2.011E+15 * (TC + 273) * Exp(-AEHH / 8.314 / (TC + 273)) * H2OS * 1000 /
    2.016 * di / 4 * Exp(5803 / 3 * EeH2H / (TC + 273))
OSPOT = 0.000258 * (TC + 273) * Log((OSEXFE * Exp(1934 / (TC + 273) * EeFEFEO)
    + OSEXH * Exp(1934 / (TC + 273) * EeH2H)) / (OSEXFE * Exp(-1934 / (TC +
    273) * EeFEFEO) + OSEXH * Exp(-1934 / (TC + 273) * EeH2H)))

'CALCULATIONS FOR PRECIPITATION
If (CB >= Sat) Then
    KD = CKD * Exp(-1934 * OSPOT / (TC + 273))
Else
    'CALCULATIONS FOR DISSOLUTION
    EeFEOF = -EeFEFEO
    KD = CKD * Exp(1934 * OSPOT / (TC + 273))
End If

'DETERMINE NEW ELECTROCHEMICAL SOLUBILITIES AT EACH INTERFACE
KDMO = CKD * Exp(-1934 * POT / (TC + 273))
Csol = (Exp((POT + 0.75 + 0.000198 * (TC + 273) * pht) / (-0.000129 * (TC + 273))) / 11)
CW = Csol + 0.476 * (1 - (op1 + OX1 * PORCON)) * CR / KDMO / 90 / SA

```

```
Sat = (Exp((OSPOT + 0.75 + 0.000198 * (TC + 273) * pht) / (-0.000129 * (TC+
273)))/11)
```

End Sub

```
Sub BisectPotential(Forward, Reverse, ALPHA, ICorr, ECorr)
```

```
Dim MAXITER, MAXERR, MID, EA, ITERATION, TEST
```

```
MAXITER = 10000
```

```
MAXERR = 0.000005
```

```
LOWER = -1.5
```

```
UPPER = 1#
```

```
ITERATION = 0
```

```
EA = 1000000 * MAXERR
```

```
10 If (EA > MAXERR And ITERATION < MAXITER) Then
```

```
    MID = (LOWER + UPPER) / 2
```

```
    ITERATION = ITERATION + 1
```

```
    If (LOWER + UPPER <> 0) Then
```

```
        EA = Abs((UPPER - LOWER) / (UPPER + LOWER)) * 100
```

```
    End If
```

```
    TEST = MOPOT(LOWER, Forward, Reverse, ALPHA, ICorr) * MOPOT(MID,
        Forward, Reverse, ALPHA, ICorr)
```

```
    If (TEST = 0) Then
```

```
        EA = 0
```

```
    Else
```

```
        If (TEST < 0) Then
```

```
            UPPER = MID
```

```
        Else
```

```
            LOWER = MID
```

```
        End If
```

```
    End If
```

```
    GoTo 10
```

End If

ECorr = MID

End Sub

Function MOPOT(P, Forward, Reverse, ALPHA, ICorr)

MOPOT = Forward \* Exp(ALPHA \* P) - Reverse \* Exp(-ALPHA \* P) - ICorr

End Function

Sub Bulk(di, Velo, dis, HD, CB, CS)

' SUB BULK: CALCULATE THE IRON BULK CONCENTRATION FOR

' INLET AND OUTLET SECTIONS

Z = 4 \* HD / (Velo \* di) \* (CS - CB)

CB = CB + Z \* dis

End Sub

Sub OutputDataVariables(DC, CS, OX1, OX2, POT, OSPOT)

MatDen = 7.86

CRBox.Text = (DC \* 8640000000#) \* 36.5 / MatDen

CSBox.Text = CS

Thick1Box.Text = OX1 \* 1000# / (OXDEN \* (1 - (op1 + OX1 \* PORCON)))

Thick2Box.Text = OX2 \* 1000# / (OXDEN \* (1 - (op1 + OX1 \* PORCON)))

MOPotBox.Text = POT

OSPotBox.Text = OSPOT

End Sub

## Appendix D Samples of calculations

### D.1 Dissolution rate constant calculations

At pH10.2, and T=250°C

$$\text{Dissolution rate from Balakrishnan's work} = 7.30 \times 10^{-13} \text{ g/cm}^2 \text{s} \quad (\text{D.1})$$

$$\text{Oxide solubility from Tremaine and Leblanc's work} = 6.20 \times 10^{-10} \text{ g/cm}^3 \quad (\text{D.2})$$

$$\text{Dissolution rate constant} = \frac{7.30 \times 10^{-13} \text{ g/cm}^2 \text{s}}{6.20 \times 10^{-10} \text{ g/cm}^3} \quad (\text{D.3})$$

### D.2 Electrochemical calculations

For example, H<sub>2</sub> evolution



#### D.2.1 *Nernst equation*

From equation2.39,

$$E_{O/R}^e = E_{O/R}^\circ + \frac{RT}{nF} \ln \frac{a_{Ox}}{a_{Re}} \quad (\text{D.5})$$

$$E_{H^+/H_2}^e = E_{H^+/H_2}^\circ + \frac{RT}{nF} \ln \frac{[H^+]^2}{[H_2]} \quad (\text{D.6})$$

$$= 0 + 2(2.303) \frac{RT}{nF} \log[H^+] - \frac{RT}{nF} \ln[H_2] \quad (\text{D.7})$$

where:

$$R = 8.314 \text{ J/mol K}$$

$$n = 2$$

$$F = \text{Faraday constant (96480C/mol).}$$

$$E_{H^+/H_2}^e = -1.98 \times 10^{-4} T \cdot pH - 4.31 \times 10^{-5} \ln[H_2] \quad (\text{D.8})$$

#### D.2.2 *Exchange current density*

From equation3.22,

$$I_o = nF \frac{k}{h} T \exp\left(-\Delta G_{H_2/H^+}^*/RT\right) C_{H_2} \exp\left(-\frac{(1-\beta)nF}{RT} E_e\right) \quad (\text{D.9})$$

where:

$$\begin{aligned} k &= 1.38 \times 10^{-23} \text{ J K}^{-1} \\ h &= 6.62 \times 10^{-34} \text{ J s} \\ \beta &= 0.5 \end{aligned}$$

$$I_o = 2.011 \times 10^{15} T \cdot \exp\left(-\Delta G_{H_2/H^+}^*/8.314T\right) C_{H_2} \exp(5803/3 \times E_e/T) \quad (\text{D.10})$$

### D.2.3 Mixed potential at oxide/solution interface

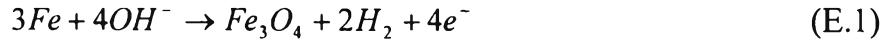
From equation2.84,

$$E_{mixed} = \frac{RT}{nF} \ln \left[ \frac{I_{0,s0} e^{nFE_{e,so}/2RT} + I_{0,M} e^{nFE_{e,M}/2RT}}{I_{0,s0} e^{-nFE_{e,so}/2RT} + I_{0,M} e^{-nFE_{e,M}/2RT}} \right] \quad (\text{D.11})$$

$$E_{mixed} = 4.308 \times 10^{-5} \ln \left[ \frac{I_{0,H^+/H_2} e^{11603E_{e,H^+/H_2}/T} + I_{0,Fe_3O_4/Fe(OH)_2} e^{11603E_{e,Fe_3O_4/Fe(OH)_2}/T}}{I_{0,H^+/H_2} e^{-11603E_{e,H^+/H_2}/T} + I_{0,Fe_3O_4/Fe(OH)_2} e^{-11603E_{e,Fe_3O_4/Fe(OH)_2}/T}} \right] \quad (\text{D.12})$$

## Appendix E A simpler model

The simple model is based on chemical surface reaction and mass transfer. It is used to predict : 1) the corrosion rate, 2) the oxide thickness and 3) the metal penetration rate. The reaction is according to Eq.E.1.



General reaction,



where A = OH<sup>-</sup>

B = Fe

C = Fe<sub>3</sub>O<sub>4</sub>

D = H<sub>2</sub>.

Rate of reaction,

$$-\frac{1}{a} \frac{dN_A}{dt} = -\frac{1}{b} \frac{dN_B}{dt} \quad (E.4)$$

$$-\frac{dN_B}{dt} = -\frac{bdN_A}{dt} = bk_{s,A}C_{Ao} - k_{sB}C_{Bo} \quad (E.5)$$

When C<sub>Bo</sub> equals C<sub>Bosat</sub>, the rate of reaction is zero.

Therefore,

$$bk_{s,A}C_{Ao} = k_{sB}C_{Bosat} \quad (E.6)$$

$$k_{sB} = \frac{bk_{s,A}C_{Ao}}{C_{Bosat}} \quad (E.7)$$

$$-\frac{dN_B}{dt} = -\frac{bdN_A}{dt} = bk_{s,A}C_{Ao} \left( 1 - \frac{C_{Bo}}{C_{Bosat}} \right) \quad (E.8)$$

where  $\frac{dN_B}{dt}$  = reaction rate of Fe (gmoles Fe/m<sup>2</sup> s)  
**b** = stoichiometric coefficient for reaction between Fe and OH<sup>-</sup>  
of boundary layer Fe<sub>3</sub>O<sub>4</sub>  
= 3/4  
C<sub>B0</sub> = concentration of Fe<sup>2+</sup> at the metal/oxide interface  
(gmoles/m<sup>3</sup>)  
C<sub>Bosat</sub> = concentration of Fe<sup>2+</sup> at saturation (gmoles/m<sup>3</sup>)  
C<sub>B2</sub> = concentration of Fe<sup>2+</sup> in the bulk solution (gmoles/m<sup>3</sup>)  
C<sub>A0</sub> = concentration of OH<sup>-</sup> in bulk solution (gmoles/m<sup>3</sup>)  
k<sub>sa</sub> = reaction coefficient for reaction between OH<sup>-</sup> and Fe at the surface (m/s).

Diffusion of Fe<sup>2+</sup> through oxide layer,

$$-(1-f)\frac{dN_B}{dt} = -b(1-f)\frac{dN_A}{dt} = \frac{D\phi}{l}(C_{B0} - C_{B1}) \quad (\text{E.9})$$

where C<sub>B1</sub> = concentration of Fe<sup>2+</sup> at the oxide/fluid interface  
(gmoles/m<sup>3</sup>)  
f̂ = fraction of Fe forming oxide(dimensionless)  
= 0.428  
D = Diffusivity of iron through the oxide layer(m<sup>2</sup>/s)  
f = porosity  
l = oxide layer thickness(m).

Mass transfer of Fe<sup>2+</sup> from the oxide/fluid interface to fluid,

$$-(1-f)\frac{dN_B}{dt} = -b(1-f)\frac{dN_A}{dt} = k_g(C_{B1} - C_{B2}) \quad (\text{E.10})$$

$$\frac{k_g dia}{D} 0.02 \text{Re}^{0.83} \text{Sc}^{1/3} \quad (\text{E.11})$$

where  $k_g$  = mass transfer coefficient between oxide surface and bulk fluid(m/s)  
 $Re$  = Reynolds number  
 $Sc$  = Schmidt number  
 $dia$  = diameter of pipes.

Combine three effect on corrosion rate together,

$$-\frac{dN_B}{dt} = \frac{(C_{B_{sat}} - C_{B_2})}{\frac{C_{B_{sat}}}{bk_{sa}C_{A_0}} + \frac{(1-f')l}{\phi D_{Fe}} + \frac{(1-f')}{k_g}}$$

Chemical reaction      Diffusion through layer      Mass transfer from oxide

Oxide layer growth rate ;

$$\frac{dl}{dt} = -\frac{f'M_{Bo}}{(1-p)\rho_{Bo}} \frac{dN_B}{dt} - kE - kD \quad (E.13)$$

where  $M_{Bo}$  = molecular weight of iron oxide (g/gmole)  
 $\rho_{Bo}$  = density of oxide layer (gmole/m<sup>3</sup>)  
 $kD$  = dissolution rate constant (m/s)  
 $kE$  = erosion rate coefficient (m/s).

$$kE = kE' f \frac{l}{2} \rho V^2 \quad (E.14)$$

where  $f$  = the friction factor for turbulent flow.

For rough approximation,

$$f = 0.04 / Re^{0.16} \dots \text{a rough approximation} \quad (E.15)$$

$$kD = kD'(C_{B_{sat}} - C_{B_2}) \quad (E.16)$$

At pH of 10.2,

$$kD' = \exp(0.0879 - 3569/T) \quad (E.17)$$

where  $T$  = absolute temperature of system(K).

The metal penetration rate ;

$$\frac{dx}{dt} = - \frac{dN_B}{dt} \frac{M_B}{\rho_B} \quad . \quad (\text{E.18})$$

where  $M_B$  = molecular weight of carbon steel

$\rho_B$  = density of carbon steel.

There are three differential equations in the model. These equations can be solved by the mathematical tools such as POLYMATH program, MATLAB, or MAPLE program. The code for POLYMATH solver is in the attached diskette.

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