

គោលការណ៍នេះមិនអាចបង្កើតឡើងបានដូចជាអាជីវកម្មទាំងនេះ

ໂຄຍພວລິຄອນເຄນເຊັ້ນແບບວິນເຄວ່າ ເພເຊີຍລ



นาย ลธน ชื่นโชคสันติ

วิทยานพนธน์เป็นล้วนหนึ่งของการศึกษาด้านหลักสูตรปริญญาที่ศาสตร์มหาบัณฑิต

សណ្ឋាគារិចាបិໂទរ- ឪលិមេវី

ប៊ងកិច្ចិកម្មាលី ជាំរាយក្រសួងអាជីវកម្ម

พ.ศ. 2533

ISBN 974-577-270-4

ລະສັບອີງບໍ່ເກາທຍາລຍ ຈຳລັງກຽມແກ້ທຍາລຍ

11-3946-1

**FACTORS INFLUENCING PARTICLE SIZE OF POLYAMIDE MICROCAPSULES
SYNTHESIZED VIA INTERFACIAL POLYCONDENSATION**

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A Thesis Submitted in Partial Fulfillment of the Requirements
for the Degree of Master of Science
Multidisciplinary Program of Petrochemical and Polymer
Graduate School
Chulalongkorn University
1990
ISBN 974-577-270-4



Thesis Title : Factors Influencing Particle Size of Polyamide Microcapsules Synthesized via Interfacial Polycondensation
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พิมพ์ครั้งที่หนึบแก้ดีปัจจุบันนี้ในวิชาการวิเคราะห์เชิงทางเคมีชีวภาพ



ส่วน ชื่น โชคสันต์ : ตัวแปรที่มีอิทธิพลต่อขนาดของพอลิเอโอม์ไดโน่ไมโครแคปซูลที่สั่งเคราะห์โดย พอลิค่อนเดนเซชันแบบอินเตอร์เฟเชียล (FACTORS INFLUENCING PARTICLE SIZE OF POLYAMIDE MICROCAPSULES SYNTHESIZED VIA INTERFACIAL POLYCONDENSATION) อ.ที่ปรึกษา : รศ.ดร. สุชา เกียรติกาจรงค์, อ.ที่ปรึกษาร่วม : ดร. สุพัล โชคธรรม, 292 หน้า. ISBN 974-577-270-4

พอลิเอโอม์ไดโน่ไมโครแคปซูลที่มีขนาดระหว่าง 1-100 ไมโครเมตร สามารถสั่งเคราะห์ได้โดย กระบวนการพอลิค่อนเดนเซชันแบบอินเตอร์เฟเชียล ระหว่างไดออก็อกโลไรด์กับไดเอมีน การสั่งเคราะห์นั้นกระทำโดยละเอียดได้เชิงกลอไรด์ในสารที่ต้องการบรรจุอยู่ในไมโครแคปซูล ซึ่งมีลักษณะเป็นผังผืด นำสารละลายผสมกับสารอิมัลชันให้อย่าง เอเจนต์ละลายอย่างรวดเร็วผ่านผังผืดไปเป็นที่เป็นอิมัลชันค้ายเครื่อง ไซโนจีไนเซอร์ที่มีความเร็วในการเดินทาง 2,000-10,000 รอบต่อนาที และใช้ระยะเวลาบันทึกแค่ 30-150 วินาที แล้วนำอิมัลชันที่ได้เทเพสมลงในผังผืดที่มีไดเอมีนและสารรับกรดละลายอยู่ ปฏิกิริยาพอลิค่อนเดนเซชันจะเกิดขึ้นตรงผิวสัมผัสของหยดน้ำที่แขวนอยู่ในผ่านสารละลายในผ่าน ทำให้เกิดเยื่อบางๆ ของพอลิเอโอม์เป็นเปลือกของไมโครแคปซูลห่อหุ้นสารที่มีลักษณะเป็นผังผืดบรรจุอยู่ในกลางของแคปซูล

การสั่งเคราะห์พอลิเอโอม์ไดโน่ไมโครแคปซูลได้ทดลองเลือก ชีบาร์โซลิคโลไรด์, เทเรพทาโลอลิคโลไรด์, เยกซ์ เมทิลลินไดเอมีน และพาราฟีนลลินไดเอมีน เป็นโนโนเมอร์ในการสั่งเคราะห์เปลือก พอลิเอโอม์ โดยที่มีบทบาทหลักที่เกิดหรือควบคุมพากาเลตเป็นสารบรรจุอยู่ในไมโครแคปซูล สารอิมัลชันให้อย่าง เอเจนต์เลือกใช้ 2 ชนิด คือ โซเดียมโคดิชลเบนซินชัล โพเนตและพอลิออกซิเอทิลลูริลอีเทอร์ เป็นสารอิมัลชันไฟเวอร์ และโซเดียมของลิตรีพาราลิอิกเกนไฮไดรคิดและพอลิไวนิลแอลกอฮอล์ เป็นสารโปรเทกท์พคอลลอยด์ เลือกโซเดียมไฮดรอกโซไซด์เป็นตัวรับกรดที่เกิดจากปฏิกิริยาพอลิค่อนเดนเซชัน

จากการทดลองพบว่า พอลิไวนิลแอลกอฮอล์มีประสิทธิภาพในการเป็นสารอิมัลชันให้อย่าง เอเจนต์มากที่สุด โดยที่สามารถใช้โซเดียมโคดิชลเบนซินชัล โพเนตเป็นตัวช่วยเสริมประสิทธิภาพในการเกิดอิมัลชันของการสั่งเคราะห์ในไมโครแคปซูลที่มีขนาดต่างๆ กัน โนโนเมอร์ที่ใช้ในการสั่งเคราะห์พอลิเอโอม์สามารถสัลบค์โนโนเมอร์ได้ทุกด้า ซึ่งผลต่อสมบัติทางกายภาพและเคมีของไมโครแคปซูลที่สั่งเคราะห์ได้ ความเร็วในการเดินทางของไซโนจีไนเซอร์ ระยะเวลาที่บันทึก ปริมาณของพอลิไวนิลแอลกอฮอล์ และปริมาณของไดออก็อกโลไรด์มีผลต่อลักษณะของไมโครแคปซูลได้ โดยที่ความเร็วในการเดินทางจะมีผลต่อสัมผัสนายกลาง เฉลี่ยของไมโครแคปซูลลดลง ระยะเวลาที่ใช้ในการเดินทางจะมีผลต่อขนาดของไมโครแคปซูลมากขึ้น การเพิ่มปริมาณของพอลิไวนิลแอลกอฮอล์จะทำให้ไมโครแคปซูลอยู่ในสภาพอิมัลชันได้นานและมีผลทำให้เส้นผ่าศูนย์กลางเฉลี่ยของไมโครแคปซูลลดลง ปริมาณสารโซเดียมไฮดรอกโซไซด์และพอลิอิกเกนไฮไดรคิดจะมีผลต่อประสิทธิภาพในการห่อหุ้นจุลภาค ซึ่งจะต้องปรับความเข้มข้นให้เหมาะสม เพื่อจะได้ไมโครแคปซูลที่เปลือกห่อหุ้นที่สมบูรณ์

ภาควิชา สหสาขาวิชาปีโตร-โพลิเมอร์
สาขาวิชา วิทยาศาสตร์โพลิเมอร์
ปีการศึกษา 2532

ลายมือชื่อนักศึกษา ๑๗๘๙๒๖๓๔๗๘
ลายมือชื่ออาจารย์ที่ปรึกษา อ.ดร. นพดล ใจดี
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ลายมือชื่ออาจารย์ที่ปรึกษา อ.ดร. นพดล ใจดี



พิมพ์โดยบดีบุ๊กเซ็ป วิทยานิพนธ์ภารกิจในกรอบวิชาชีววิทยาและเคมีชีวภาพ

SUTHON CHUENCHOKESANT : FACTORS INFLUENCING PARTICLE SIZE OF POLYAMIDE MICROCAPSULES SYNTHESIZED VIA INTERFACIAL POLYCONDENSATION.
THESIS ADVISOR : ASSOC. PROF. SUDA KIATKAMJORNWONG, Ph.D.; THESIS CO-ADVISOR : SUPON CHOTIWANA, Ph.D., 292 pp. ISBN 974-577-270-4

Synthesis of 1-100 micrometers of polyamide microcapsules by interfacial polycondensation between diacid chlorides and diamines was carried out. The synthesis was started by dissolving the diacid chloride in an oily-like material to be encapsulated in microcapsules and then mixing this solution in an aqueous solution of an emulsifier or a protective colloid. The mixture was agitated into emulsion by a high-speed homogenizer at the propeller speeds between 2,000-10,000 rpm., and the agitation times between 30-150 sec. This emulsion was poured into the aqueous solution of a diamine and an acid-receiver. Polycondensation occurred at the interface between the tiny oil droplets and another liquid layers and produced a thin film of polyamide as a shell of the microcapsules which encapsulated the oily-like material in the core.

Sebacoyl chloride and terephthaloyl chloride as diacid chlorides; and hexamethylene diamine and para-phenylene diamine as diamines were selected as monomer pairs to synthesize the polyamide microcapsule shell. Sodium dodecyl benzene sulfonate and poly(oxyethylene lauryl ether) as surface active agents; and sodium salt of styrene maleic anhydride and poly(vinyl alcohol) as protective colloids were selected to be emulsifying agents. Sodium hydroxide was used as an acid-receiver during polycondensation. Butyl acetate and dibutyl phthalate were used as encapsulated materials in microcapsules.

Based on the experiments, it was found that dibutyl phthalate was more appropriate as encapsulated material than butyl acetate; poly(vinyl alcohol) was the most effective emulsifying agent and sodium dodecyl benzene sulfonate could be used as a co-emulsifying agent with poly(vinyl alcohol) to produce microcapsules with various diameters. Furthermore, it was found that all diacid chloride and diamine combinations could be applied as monomer pairs to synthesize the polyamide microcapsule shell having a wide range of chemical and physical properties; the speed of the propeller, the agitation time of the homogenizer, and the amount of diacid chloride and emulsifying agent affected the characteristics of the microcapsules: increasing the speed of propeller decreased the average diameter of the microcapsules; longer agitation time produced more equal-sized microcapsules; increasing the concentration of emulsifying agent produced more stable emulsion which reduced the average diameter of the microcapsules; the amount of diacid chloride had a direct effect on encapsulation efficiency, the concentration of which must be adjusted properly so as to produce a perfect shell of the microcapsules.

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ACKNOWLEDGEMENTS

The author wishes to express his deepest gratitude to his advisor, Assoc. Prof. Dr. Suda Kiatkamjornwong, for her generous guidance, understanding, valuable suggestions, supervision and continuous encouragements throughout the course of this thesis. He also thanks Dr. Supon Chotiwana, his co-advisor ship, and Assoc. Prof. Pattarapan Prasassarakick, and Dr. Nopporn Pramojaney for their valuable comments to the work and serving on the thesis committee. He is grateful to the Scientific and Technological Research Equipment Center (STREC) of Chulalongkorn University for operating his samples via optical microscope and scanning electron microscope. In addition, he is indebted to Asst. Prof. Kemchai Hamajantr, the Department of Materials Science of Chulalongkorn University, for his permission for using the particle analyzer. Finally, the scholarship granted from the Graduate School of Chulalongkorn University is highly appreciated.



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LIST OF ABBREVIATIONS



ABBREVIATIONS	DEFINITION
gm	gram
ml	milliliter
rpm	revolution per minute
sec	second
°C	degree celcius
SBC	sebacoyl chloride
TPC	terephthaloyl chloride
HMDA	hexamethylene diamine
PNDA	para-phenylene diamine
BAT	butyl acetate
DBP	dibutyl phthalate
PVA	poly(vinyl alcohol)
SMA	sodium salt of styrene maleic anhydride copolymer
POLE	poly(oxyethylene lauryl ether)
SDBS	sodium dodecyl benzene sulfonate
NaOH	sodium hydroxide
DIW	deionized water
S	sebacoyl chloride for assigned name
T	terephthaloyl chloride for assigned name
B	butyl acetate for assigned name
D	dibutyl phthalate for assigned name
PV	poly(vinyl alcohol) for assigned name
SM	sodium salt of styrene maleic anhydride copolymer for assigned name
PO	poly(oxyethylene lauryl ether) for assigned name
SD	sodium dodecyl benzene sulfonate for assigned name
SDPxxyyyy	S = sebacoyl chloride D = dibutyl phthalate P = poly(vinyl alcohol) xxx = value of rpm yyy = value of sec
(for section 2.9)	

ABBREVIATIONS	DEFINITION
TDPxxx/yyy (for section 2.9)	T = terephthaloyl chloride D = dibutyl phthalate P = poly(vinyl alcohol) xxx = value of rpm yyy = value of sec
SDPxxxE (for section 2.10)	S = sebacoyl chloride D = dibutyl phthalate P = poly(vinyl alcohol) xxx = value of poly(vinyl alcohol) concentration E = emulsifier
TDPxxxE (for section 2.10)	T = terephthaloyl chloride D = dibutyl phthalate P = poly(vinyl alcohol) xxx = value of poly(vinyl alcohol) concentration E = emulsifier
SDPxxx/yyy (for section 2.11)	S = sebacoyl chloride D = dibutyl phthalate P = poly(vinyl alcohol) xxx = amount of sebacoyl chloride ratio yyy = amount of dibutyl phthalate ratio
TDPxxx/yyy (for section 2.11)	T = terephthaloyl chloride D = dibutyl phthalate P = poly(vinyl alcohol) xxx = amount of terephthaloyl chloride ratio yyy = amount of dibutyl phthalate ratio
STC	sebacoyl chloride and terephthaloyl chloride combination
HMPN	hexamethylene diamine and para-phenylene diamine combination