

## REFERENCES

- Adachi, Y., Komoto, M., Watanabe, I., Ohno, Y., and Fujimoto, K. (2000) Effective utilization of remote coal through dimethyl ether synthesis. *Fuel*, 79(3–4), 229-234.
- Amann, J.M. (2007) Study of CO<sub>2</sub> capture processes in power plants. PhD Dissertation, MINES ParisTech, Paris, France.
- An, X., Li, J., Zuo, Y., Zhang, Q., Wang, D., and Wang, J. (2007) A Cu/Zn/Al/Zr Fibrous Catalyst that is an Improved CO<sub>2</sub> Hydrogenation to Methanol Catalyst. *Catalysis Letters*, 118(3-4), 264-269.
- Anderson, J.J., Drury, D.J., Hamlin, J.E., and Kent, A.G. (1986) U.S. Patent 4855496A.
- Arab Aboosadi, Z., Jahanmiri, A.H., and Rahimpour, M.R. (2011) Optimization of tri-reformer reactor to produce synthesis gas for methanol production using differential evolution (DE) method. *Applied Energy*, 88(8), 2691-2701.
- Arakawa, H. (1998) Research and development on new synthetic routes for basic chemicals by catalytic hydrogenation of CO<sub>2</sub>. Inui, M.A.K.I.S.Y. and Yamaguchi, T., *Advances in Chemical Conversions for Mitigating Carbon Dioxide: Proceedings of the Fourth International Conference on Carbon Dioxide Utilization* (pp. 19-30). New York: Elsevier.
- Arena, F., Barbera, K., Italiano, G., Bonura, G., Spadaro, L., and Frusteri, F. (2007) Synthesis, characterization and activity pattern of Cu-ZnO/ZrO<sub>2</sub> catalysts in the hydrogenation of carbon dioxide to methanol. *Journal of Catalysis*, 249(2), 185-194.
- Aresta, M. (2003) *Carbon Dioxide Recovery and Utilization*. Boston: Kluwer Academic Publishers.
- Aresta, M., Dibenedetto, A., Gianfrate, L., and Pastore, C. (2003) Nb(V) compounds as epoxides carboxylation catalysts: the role of the solvent. *Journal of Molecular Catalysis A: Chemical*, 204–205(0), 245-252.
- Aresta, M., Quaranta, E., and Tommasi, I. (1994) The role of metal centers in reduction and carboxylation reactions utilizing carbon dioxide. *New Journal of Chemistry*, 18, 133-142.

- Aresta, M., Quaranta, E., Liberio, R., Dileo, C., and Tommasi, I. (1998) Enzymatic synthesis of 4-OH-benzoic acid from phenol and CO<sub>2</sub>: the first example of a biotechnological application of a Carboxylase enzyme. *Tetrahedron*, 54(30), 8841-8846.
- Aresta, M., Quaranta, E., Tommasi, I., Giannoccaro, P., and Ciccarese, A.A. (1995) Enzymatic versus chemical carbon dioxide utilization. Part I. The role of metal centres in carboxylation reactions. *Gazzetta Chimica Italiana*, 125(11), 509-538.
- Arpe, H.-J. and Weissermel, K. (2010) *Industrial Organic Chemistry*. Weinheim: Wiley-VCH.
- Baranova, E.A., Fóti, G., and Comninellis, C. (2004) Promotion of Rh catalyst interfaced with TiO<sub>2</sub>. *Electrochemistry Communications*, 6(2), 170-175.
- Barbieri, G., Marigliano, G., Golemme, G., and Drioli, E. (2002) Simulation of CO<sub>2</sub> hydrogenation with CH<sub>3</sub>OH removal in a zeolite membrane reactor. *Chemical Engineering Journal*, 85(1), 53-59.
- Bartels, J.R., Pate, M.B., and Olson, N.K. (2010) An economic survey of hydrogen production from conventional and alternative energy sources. *International Journal of Hydrogen Energy*, 35(16), 8371-8384.
- Bhattacharya, A.K., Breach, J.A., Chand, S., Ghorai, D.K., Hartridge, A., Keary, J., and Mallick, K.K. (1992) Selective oxidation of methane to carbon monoxide on supported palladium catalyst. *Applied Catalysis A: General*, 80(1), L1-L5.
- Bonivardi, A.L., Chiavassa, D.L., Querini, C.A., and Baltanás, M.A. (2000) Enhancement of the catalytic performance to methanol synthesis from CO<sub>2</sub>/H<sub>2</sub> by gallium addition to palladium/silica catalysts. Avelino Corma, F.V.M.S.M. and José Luis, G.F., *12th International Congress on Catalysis: Proceedings of the 12th ICC* (pp. 3747-3752). New York: Elsevier.
- Bradford, M.C.J. and Vannice, M.A. (1999) CO<sub>2</sub> Reforming of CH<sub>4</sub>. *Catalysis Reviews*, 41(1), 1-42.
- Bussche, K.M.V. and Froment, G.F. (1996) A Steady-State Kinetic Model for Methanol Synthesis and the Water Gas Shift Reaction on a Commercial Cu/ZnO/Al<sub>2</sub>O<sub>3</sub>Catalyst. *Journal of Catalysis*, 161(1), 1-10.

- Cai, Q., Jin, C., Lu, B., Tangbo, H., and Shan, Y. (2005) Synthesis of Dimethyl Carbonate from Methanol and Carbon dioxide using Potassium Methoxide as Catalyst under Mild Conditions. *Catalysis Letters*. 103(3-4), 225-228.
- Challand, N., Sava, X., and Roeper, M. (2008) U.S. Patent 20100063320 A1.
- Chang, C.D., Jiang, Z., LaPierre, R.B., Schramm, S.E., and Timken, H.K.C. (2002) U.S. Patent 6365767 B1.
- Cheng, W.-H. (1994) Methanol Production and Use. New York: Marcel Dekker.
- Choudhary, T.V. and Choudhary, V.R. (2008) Energy-Efficient Syngas Production through Catalytic Oxy-Methane Reforming Reactions. *Angewandte Chemie International Edition*, 47(10), 1828-1847.
- Choudhary, V.R., Rajput, A.M., and Prabhakar, B. (1994) NiO/CaO-Catalyzed Formation of Syngas by Coupled Exothermic Oxidative Conversion and Endothermic CO<sub>2</sub> and Steam Reforming of Methane. *Angewandte Chemie International Edition*, 33(20), 2104-2106.
- Choudhary, V.R., Uphade, B.S., and Mamman, A.S. (1998) Simultaneous steam and CO<sub>2</sub> reforming of methane to syngas over NiO/MgO/SA-5205 in presence and absence of oxygen. *Applied Catalysis A: General*, 168(1), 33-46.
- Collins, S.E., Baltanás, M.A., and Bonivardi, A.L. (2004) An infrared study of the intermediates of methanol synthesis from carbon dioxide over Pd/β-Ga<sub>2</sub>O<sub>3</sub>. *Journal of Catalysis*, 226(2), 410-421.
- Darensbourg, D.J. and Holtcamp, M.W. (1996) Catalysts for the reactions of epoxides and carbon dioxide. *Coordination Chemistry Reviews*, 153(0), 155-174.
- De Pasquale, R.J. (1973) Unusual catalysis with nickel(0) complexes. *Journal of the Chemical Society. Chemical Communications*, (5), 157-158.
- Dong, W.S., Roh, H.S., Jun, K.W., Park, S.E., and Oh, Y.S. (2002) Methane reforming over Ni/Ce-ZrO<sub>2</sub> catalysts: effect of nickel content. *Applied Catalysis a-General*, 226(1-2), 63-72.
- Drury, D.J. and Hamlin, J.E. (1983) U.S. Patent 0095321 B1.

- Elek, J., Nádasdi, L., Papp, G., Laurenczy, G., and Joó, F. (2003) Homogeneous hydrogenation of carbon dioxide and bicarbonate in aqueous solution catalyzed by water-soluble ruthenium(II) phosphine complexes. Applied Catalysis A: General, 255(1), 59-67.
- Evans, T., Brunelle, D., Salem, A., and Stewart, K. (1991) Developments in the Chemistry of Oligocyclic Carbonates for Use in Structural Composites. Polymer Preprints (USA), 32(2), 176-177.
- Farlow, M.W. and Adkins, H. (1935) The Hydrogenation of Carbon Dioxide and a Correction of the Reported Synthesis of Urethans. Journal of the American Chemical Society, 57(11), 2222-2223.
- Fathi, M., Bjorgum, E., Viig, T., and Rokstad, O.A. (2000) Partial oxidation of methane to synthesis gas: Elimination of gas phase oxygen. Catalysis Today, 63(2-4), 489-497.
- Fornero, E.L., Chiavassa, D.L., Bonivardi, A.L., and Baltanás, M.A. (2011) CO<sub>2</sub> capture via catalytic hydrogenation to methanol: Thermodynamic limit vs. 'kinetic limit'. Catalysis Today, 172(1), 158-165.
- Frevel, L.K. and Gilpin, J.A. (197?) U.S. Patent 3642858 A.
- Gadalla, A.M. and Bower, B. (1988) The role of catalyst support on the activity of nickel for reforming methane with CO<sub>2</sub>. Chemical Engineering Science, 43(11), 3049-3062.
- Gallucci, F., Paturzo, L., and Basile, A. (2004) An experimental study of CO<sub>2</sub> hydrogenation into methanol involving a zeolite membrane reactor. Chemical Engineering and Processing: Process Intensification, 43(8), 1029-1036.
- Gao, Y., Kuncheria, J.K., Jenkins, H.A., Puddephatt, R.J., and Yap, G.P.A. (2000). The interconversion of formic acid and hydrogen/carbon dioxide using a binuclear ruthenium complex catalyst. Journal of the Chemical Society, Dalton Transactions, (18), 3212-3217.
- Goehna, H. and Koenig, P. (1994) Producing methanol from CO<sub>2</sub>. ChemTech, 24(6), 36-39.

- Gómez, J.P., Jiménez, J.M., Vic, S., Lezaun, J., Terreros, P., Cabrera, I., Peña, M.A., and Fierro, J.L.G. (1997) Hydrogen production on nickel-monolith structures by partial oxidation of methane at high pressure. M. de Pontes, R.L.E.C.P.N.J.H.S. and Scurrell, M.S., Natural Gas Conversion IV (pp. 397-402). New York: Elsevier.
- Graaf, G.H., Sijtsema, P.J.J.M., Stamhuis, E.J., and Joosten, G.E.H. (1986) Chemical equilibria in methanol synthesis. Chemical Engineering Science, 41(11), 2883-2890.
- Green, C.J., Cockshutt, N.A., and King, L. (1990) Dimethyl ether as a methanol ignition improver: substitution requirements and exhaust emissions impact. SAE Technical Paper.
- Gressin, J.C., Michelet, D., Nadjo, L., and Saveant, J.M. (1979) Electrochemical reduction of carbon dioxide in weakly protic medium. Nouveau Journal de Chimie, 3(8-9), 545-554.
- Guerrero-Ruiz, A., Rodriguez-Ramos, I., and Sepulveda-Escribano, A. (1993) Effect of the basic function in Co, MgO/C catalysts on the selective oxidation of methane by carbon dioxide. Journal of the Chemical Society, Chemical Communications, (5), 487-488.
- Guo, X., Mao, D., Wang, S., Wu, G., and Lu, G. (2009) Combustion synthesis of CuO-ZnO-ZrO<sub>2</sub> catalysts for the hydrogenation of carbon dioxide to methanol. Catalysis Communications, 10(13), 1661-1664.
- Hallgren, J.E. and Lucas, G.M. (1981) The palladium-catalyzed synthesis of diphenyl carbonate from phenol, carbon monoxide, and oxygen: II. Aqueous sodium hydroxide as a base. Journal of Organometallic Chemistry, 212(1), 135-139.
- Halmann, M.M. and Steinberg, M. (1998) Greenhouse Gas Carbon Dioxide Mitigation: Science and Technology. Florida: CRC press.
- Hammond, G.P., Akwe, S.S.O., and Williams, S. (2011) Techno-economic appraisal of fossil-fuelled power generation systems with carbon dioxide capture and storage. Energy, 36(2), 975-984.

- Hansen, J.B. and Højlund Nielsen, P.E. (2008) Methanol Synthesis. Gerhard Ertl, Helmut Knözinger, Ferdi Schuth, Jens Weitkamp, Handbook of Heterogeneous Catalysis. Weinheim: Wiley-VCH.
- Hegarty, M.E.S., O'Connor, A.M., and Ross, J.R.H. (1998) Syngas production from natural gas using ZrO<sub>2</sub>-supported metals. Catalysis Today, 42(3), 225-232.
- Hoberg, H., Schaefer, D., and Oster, B.W. (1984) Diencarbonsäuren aus 1,3-dienen und CO<sub>2</sub> durch C-C-verknüpfung an nickel(0). Journal of Organometallic Chemistry, 266(3), 313-320.
- Holladay, J.D., Hu, J., King, D.L., and Wang, Y. (2009) An overview of hydrogen production technologies. Catalysis Today, 139(4), 244-260.
- Holm-Larsen, H. (2001) CO<sub>2</sub> reforming for large scale methanol plants - an actual case. E. Iglesia, J.J.S. and Fleisch, T.H., Natural Gas Conversion VI (pp. 441-446). New York: Elsevier.
- Hou, K. and Hughes, R. (2001) The kinetics of methane steam reforming over a Ni/ $\alpha$ -Al<sub>2</sub>O catalyst. Chemical Engineering Journal, 82(1–3), 311-328.
- Huang, C.H. and Tan, C.S. (2014) A Review: CO<sub>2</sub> Utilization. Aerosol and Air Quality Research, 14(2), 480-499.
- IEA (2013a) CO<sub>2</sub> Emissions from Fuel Combustion 2013. Paris: OECD/IEA.
- IEA (2013b) Carbon Capture and Storage. Paris: OECD/IEA.
- Inoue, S., Koinuma, H., and Tsuruta, T. (1969) Copolymerization of carbon dioxide and epoxide. Journal of Polymer Science Part B: Polymer Letters, 7(4), 287-292.
- Institute, M. "Methanol Fuel in the Environment." Methanol Fuels. 27 March 2015 <<http://www.methanolfuels.org/about-methanol/environment/>>
- International, C.R. "First Commercial Plant." Carbon Recycling International (CRI). 25 June 2014 <<http://www.carbonrecycling.is/>>
- Inui, T., Hara, H., Takeguchi, T., and Kim, J.-B. (1997) Structure and function of Cu-based composite catalysts for highly effective synthesis of methanol by hydrogenation of CO<sub>2</sub> and CO. Catalysis Today, 36(1), 25-32.
- Inui, T., Saigo, K., Fujii, Y., and Fujioka, K. (1995) Catalytic combustion of natural gas as the role of on-site heat supply in rapid catalytic CO<sub>2</sub>/H<sub>2</sub>O reforming of methane. Catalysis Today, 26(3–4), 295-302.

- Iwasaki, N.-o., Miyake, T., Yagasaki, E., and Suzuki, T. (2006) Partial oxidation of ethane to synthesis gas over Co-loaded catalysts. *Catalysis Today*, 111(3–4), 391-397.
- Jessop, P.G., Hsiao, Y., Ikariya, T., and Noyori, R. (1996) Homogeneous Catalysis in Supercritical Fluids: Hydrogenation of Supercritical Carbon Dioxide to Formic Acid, Alkyl Formates, and Formamides. *Journal of the American Chemical Society*, 118(2), 344-355.
- Jessop, P.G., Ikariya, T., and Noyori, R. (1994) Homogeneous catalytic hydrogenation of supercritical carbon dioxide. *Nature*, 368(6468), 231-233.
- Jessop, P.G., Ikariya, T., and Noyori, R. (1995) Homogeneous Hydrogenation of Carbon Dioxide. *Chemical Reviews*, 95(2), 259-272.
- Ji, D., Lu, X., and He, R. (2000) Syntheses of cyclic carbonates from carbon dioxide and epoxides with metal phthalocyanines as catalyst. *Applied Catalysis A: General*, 203(2), 329-333.
- Jitaru, M., Lowy, D.A., Toma, M., Toma, B.C., and Oniciu, L. (1997) Electrochemical reduction of carbon dioxide on flat metallic cathodes. *Journal of Applied Electrochemistry*, 27(8), 875-889.
- Kangwanwatana, W., Saiwan, C., and Tontiwachwuthikul, P. (2013) Study of CO<sub>2</sub> adsorption using adsorbent modified with piperazine. *Chemical Engineering Transactions*, 35, 403-408.
- Keeling, C.D. and Whorf, T.P. (2005) Atmospheric CO<sub>2</sub> records from sites in the SIO air sampling network. *Trends: a compendium of data on global change*, 16-26.
- Kim, G., Cho, D.-S., Kim, K.-H., and Kim, J.-H. (1994) The reaction of CO<sub>2</sub> with CH<sub>4</sub> to synthesize H<sub>2</sub> and CO over nickel-loaded Y-zeolites. *Catalysis Letters*, 28(1), 41-52.
- Kim, W.B. and Lee, J.S. (2002) Comparison of polycarbonate precursors synthesized from catalytic reactions of bisphenol-A with diphenyl carbonate, dimethyl carbonate, or carbon monoxide. *Journal of Applied Polymer Science*, 86(4), 937-947.

- Kim, W.S., Yang, D.R., Moon, D.J., and Ahn, B.S. (2014) The process design and simulation for the methanol production on the FPSO (floating production, storage and off-loading) system. Chemical Engineering Research and Design, 92(5), 931-940.
- Klier, K. (1982) Methanol Synthesis. D.D. Eley, H.P. and Paul, B.W., Advances in Catalysis (pp. 243-313). New York: Academic Press.
- Knifton, J.F. and Duranleau, R.G. (1991) Ethylene glycol—dimethyl carbonate cogeneration. Journal of Molecular Catalysis, 67(3), 389-399.
- Koo, K.Y., Roh, H.-S., Jung, U.H., and Yoon, W.L. (2012) Combined H<sub>2</sub>O and CO<sub>2</sub> reforming of CH<sub>4</sub> over Ce-promoted Ni/Al<sub>2</sub>O<sub>3</sub> catalyst for gas to liquid (GTL) process: Enhancement of Ni–CeO<sub>2</sub> interaction. Catalysis Today, 185(1), 126-130.
- Kozole, H., K., Wallace, and S., J. (1989) The use of dimethyl ether as a starting aid for methanol-fueled SI engines at low temperatures. SAE Technical Paper.
- Lachowska, M. and Skrzypek, J. (2004) Methanol synthesis from carbon dioxide and hydrogen over Mn-promoted copper/zinc/zirconia catalysts. Reaction Kinetics and Catalysis Letters, 83(?), 269-273.
- Le Quéré, C., Andres, R.J., Boden, T., Conway, T., Houghton, R.A., House, J.I., Marland, G., Peters, G.P., van der Werf, G.R., Ahlström; A., Andrew, R.M., Bopp, L., Canadell, J.G., Ciais, P., Doney, S.C., Enright, C., Friedlingstein, P., Huntingford, C., Jain, A.K., Jourdain, C., Kato, E., Keeling, R.F., Klein Goldewijk, K., Levis, S., Levy, P., Lomas, M., Poulter, B., Raupach, M.R., Schwinger, J., Sitch, S., Stocker, B.D., Viovy, N., Zaehle, S., and Zeng, N. (2013) The global carbon budget 1959–2011. Earth System Science Data, 5(1), 165-185.
- Lee, J.K., Ko, J.B., and Kim, D.H. (2004) Methanol steam reforming over Cu/ZnO/Al<sub>2</sub>O<sub>3</sub> catalyst: kinetics and effectiveness factor. Applied Catalysis A: General, 278(1), 25-35.

- Li, Y., Markley, B., Mohan, A.R., Rodriguez-Santiago, V., Thompson, D., and Niekerk, D. "Utilization of carbon dioxide from coal-fired power plant for the production of value-added products." College of Earth and Mineral Sciences, The Pennsylvania State University. 2 June 2014 <[http://www.ems.psu.edu/~elsworth/courses/egee580/Utilization\\_final\\_report.pdf](http://www.ems.psu.edu/~elsworth/courses/egee580/Utilization_final_report.pdf)>
- Liang, X.-L., Dong, X., Lin, G.-D., and Zhang, H.-B. (2009) Carbon nanotube-supported Pd–ZnO catalyst for hydrogenation of CO<sub>2</sub> to methanol. Applied Catalysis B: Environmental. 88(3–4), 315-322.
- Limura, N., Takagi, M., Iwane, H., and Ookago, J. (1995) J.P. Patent 07267944 A2.
- Liu, J., Shi, J., He, D., Zhang, Q., Wu, X., Liang, Y., and Zhu, Q. (2001) Surface active structure of ultra-fine Cu/ZrO<sub>2</sub> catalysts used for the CO<sub>2</sub>+H<sub>2</sub> to methanol reaction. Applied Catalysis A: General. 218(1–2), 113-119.
- Liu, X.-M., Lu, G.Q., and Yan, Z.-F. (2005) Nanocrystalline zirconia as catalyst support in methanol synthesis. Applied Catalysis A: General. 279(1–2), 241-245.
- Machado, C.F., de Medeiros, J.L., Araújo, O.F., de Janeiro, R., and Alves, B.R.M. (2014) A comparative analysis of methanol production routes: synthesis gas versus CO<sub>2</sub> hydrogenation. Paper presented at The 2014 International Conference on Industrial Engineering and Operations Management, Bali, Indonesia.
- Mahajan, D. and Goland, A.N. (2003) Integrating low-temperature methanol synthesis and CO<sub>2</sub> sequestration technologies: application to IGCC plants. Catalysis Today. 84(1–2), 71-81.
- Maroto-Valer, M.M., Song, C., and Soong, Y. (2002) Environmental challenges and greenhouse gas control for fossil fuel utilization in the 21st century. Boston: Springer.
- Matson, P., Dietz, T., Abdalati, W., Busalacchi, A., Caldeira, K., Corell, R., Defries, R., Fung, I., Gaines, S., and Hornberger, G. (2010) Advancing the Science of Climate Change. Washington, D.C.: National Academies Press.

- Melián-Cabrera, I., López Granados, M., and Fierro, J.L.G. (2002) Reverse Topotactic Transformation of a Cu–Zn–Al Catalyst during Wet Pd Impregnation: Relevance for the Performance in Methanol Synthesis from CO<sub>2</sub>/H<sub>2</sub> Mixtures. *Journal of Catalysis*, 210(2), 273-284.
- Mikkelsen, M., Jorgensen, M., and Krebs, F.C. (2010) The teraton challenge. A review of fixation and transformation of carbon dioxide. *Energy & Environmental Science*, 3(1), 43-81.
- Minutillo, M., and Perna, A. (2010) A novel approach for treatment of CO<sub>2</sub> from fossil fired power plants. Part B: The energy suitability of integrated tri-reforming power plants (ITRPPs) for methanol production. *International Journal of Hydrogen Energy*, 35(13), 7012-7020.
- Moradi, G.R., Ahmadpour, J., and Yaripour, F. (2008) Intrinsic kinetics study of LPDME process from syngas over bi-functional catalyst. *Chemical Engineering Journal*, 144(1), 88-95.
- Munshi, P., Main, A.D., Linehan, J.C., Tai, C.-C., and Jessop, P.G. (2002) Hydrogenation of Carbon Dioxide Catalyzed by Ruthenium Trimethylphosphine Complexes: The Accelerating Effect of Certain Alcohols and Amines. *Journal of the American Chemical Society*, 124(27), 7963-7971.
- Nakamura, J., Aikawa, K., Sato, K., and Uchijima, T. (1994) Role of support in reforming of CH<sub>4</sub> with CO<sub>2</sub> over Rh catalysts. *Catalysis Letters*, 25(3-4), 265-270.
- Nowell, G.P. "On the Road with Methanol: The Present and Future Benefits of Methanol Fuel." Methanol Institute. 2 June 2014  
[<http://www.methanol.org>](http://www.methanol.org)
- Oh, Y.S., Roh, H.S., Jun, K.W., and Baek, Y.S. (2003) A highly active catalyst, Ni/Ce-ZrO<sub>2</sub>/theta-Al<sub>2</sub>O<sub>3</sub>, for on-site H-2 generation by steam methane reforming: pretreatment effect. *International Journal of Hydrogen Energy*, 28(12), 1387-1392.
- Olah, G.A. (2005) Beyond Oil and Gas: The Methanol Economy. *Angewandte Chemie International Edition*, 44(18), 2636-2639.

- Olah, G.A., Goeppert, A., Czaun, M., and Prakash, G.K.S. (2013) Bi-reforming of Methane from Any Source with Steam and Carbon Dioxide Exclusively to Metgas (CO-2H(2)) for Methanol and Hydrocarbon Synthesis. *Journal of the American Chemical Society*, 135(2), 648-650.
- Olah, G.A., Goeppert, A., and Prakash, G.K.S. (2009a) *Beyond Oil and Gas: The Methanol Economy*. Weinheim : Wiley-VCH.
- Olah, G.A., Goeppert, A., and Prakash, G.K.S. (2009b) Chemical Recycling off Carbon Dioxide to Methanol and Dimethyl Ether: From Greenhouse Gas to Renewable, Environmentally Carbon Neutral Fuels and Synthetic Hydrocarbons. *Journal of Organic Chemistry*, 74(2), 487-498.
- Olah, G.A., Prakash, G.K.S., and Goeppert, A. (2011) Anthropogenic Chemical Carbon Cycle for a Sustainable Future. *Journal of the American Chemical Society*, 133(33), 12881-12898.
- Ostrovskii, V.E. (2002). Mechanisms of methanol synthesis from hydrogen and carbon oxides at Cu-Zn-containing catalysts in the context of some fundamental problems of heterogeneous catalysis. *Catalysis Today*, 77(3), 141-160.
- Ott, J., Gronemann, V., Pontzen, F., Fiedler, E., Grossmann, G., Kersebohm, D.B., Weiss, G., and Witte, C. (2000) Methanol. *Ullmann's Encyclopedia of Industrial Chemistry*, 2012.
- Özkara-Aydinoğlu, S. (2010) Thermodynamic equilibrium analysis of combined carbon dioxide reforming with steam reforming of methane to synthesis gas. *International Journal of Hydrogen Energy*, 35(23), 12821-12828.
- Park, D., Moon, D.J., and Kim, T. (2013) Steam-CO<sub>2</sub> reforming of methane on Ni/γ-Al<sub>2</sub>O<sub>3</sub>-deposited metallic foam catalyst for GTL-FPSO process. *Fuel Processing Technology*, 112(0), 28-34.
- Park, S.E., Nam, S.S., Choi, M.J., and Lee, K.W. (1995) Catalytic reduction of carbon dioxide. The effects of catalysts and reductants. *Energy Conversion and Management*, 36(6–9), 573-576.
- Passos, F., Oliveira, E., Mattos, L., and Noronha, F. (2006) Effect of the support on the mechanism of partial oxidation of methane on platinum catalysts. *Catalysis Letters*, 110(1-2), 161-167.

- Pierantozzi, R. (2000) Carbon Dioxide. Kirk-Othmer Encyclopedia of Chemical Technology, 4, 803-822.
- Qin, D., Lapszewicz, J., and Jiang, X. (1996) Comparison of Partial Oxidation and Steam-CO<sub>2</sub> Mixed Reforming of CH<sub>4</sub> to Syngas on MgO-Supported Metals. Journal of Catalysis, 159(1), 140-149.
- Quadrelli, E.A., Centi, G., Duplan, J.-L., and Perathoner, S. (2011) Carbon Dioxide Recycling: Emerging Large-Scale Technologies with Industrial Potential. ChemSusChem, 4(9), 1194-1215.
- Rabe, S., Truong, T.-B., and Vogel, F. (2005) Low temperature catalytic partial oxidation of methane for gas-to-liquids applications. Applied Catalysis A: General, 292(0), 177-188.
- Rahimpour, M.R. (2008) A two-stage catalyst bed concept for conversion of carbon dioxide into methanol. Fuel Processing Technology, 89(5), 556-566.
- Raudaskoski, R., Turpeinen, E., Lenkkeri, R., Pongrácz, E., and Keiski, R.L. (2009) Catalytic activation of CO<sub>2</sub>: Use of secondary CO<sub>2</sub> for the production of synthesis gas and for methanol synthesis over copper-based zirconia-containing catalysts. Catalysis Today, 144(3-4), 318-323.
- Reed, T.B. and Lerner, R.M. (1973) Methanol: A Versatile Fuel for Immediate Use: Methanol can be made from gas, coal, or wood. It is stored and used in existing equipment. Science, 182(4119), 1299-1304.
- Romano, U., Tesel, R., Mauri, M.M., and Rebora, P. (1980) Synthesis of Dimethyl Carbonate from Methanol, Carbon Monoxide, and Oxygen Catalyzed by Copper Compounds. Industrial & Engineering Chemistry Product Research and Development, 19(3), 396-403.
- Rostrup-Nielsen, J.R. (2008) Steam Reforming. Handbook of Heterogeneous Catalysis, 2008, 2882-2905.
- Rozovskii, A. and Lin, G. (2003) Fundamentals of Methanol Synthesis and Decomposition. Topics in Catalysis, 22(3-4), 137-150.
- Saito, M. (1998) R&D activities in Japan on methanol synthesis from CO<sub>2</sub> and H<sub>2</sub>. Catalysis Surveys from Asia, 2(2), 175-184.

- Saito, M., Fujitani, T., Takeuchi, M., and Watanabe, T. (1996) Development of copper/zinc oxide-based multicomponent catalysts for methanol synthesis from carbon dioxide and hydrogen. *Applied Catalysis A: General*, 138(2), 311-318.
- Saito, M. and Murata, K. (2004) Development of high performance Cu/ZnO-based catalysts for methanol synthesis and the water-gas shift reaction. *Catalysis Surveys from Asia*, 8(4), 285-294.
- Sakai, T., Kihara, N., and Endo, T. (1995) Polymer Reaction of Epoxide and Carbon Dioxide. Incorporation of Carbon Dioxide into Epoxide Polymers. *Macromolecules*, 28(13), 4701-4706.
- Shulenberger, A.M., Jonsson, F.R., Ingolfsson, O., and Tran, K.C. (2007) U.S. Patent 8198338 B2.
- Skrzypek, J., Lachowska, M., and Serafin, D. (1990) Methanol synthesis from CO<sub>2</sub> and H<sub>2</sub>: dependence of equilibrium conversions and exit equilibrium concentrations of components on the main process variables. *Chemical Engineering Science*, 45(1), 89-96.
- Sokolovskii, V.D., Coville, N.J., Parmaliana, A., Eskendirov, I., and Makoa, M. (1998) Methane partial oxidation. Challenge and perspective. *Catalysis Today*, 42(3), 191-195.
- Song, C. (2006) Global challenges and strategies for control, conversion and utilization of CO<sub>2</sub> for sustainable development involving energy, catalysis, adsorption and chemical processing. *Catalysis Today*, 115(1-4), 2-32.
- Song, C., Gaffney, A.F., and Fujimoto, K. (2002). *CO<sub>2</sub> Conversion and Utilization*. Washington, D.C.: American Chemical Society.
- Song, C.S. (2001) Tri-reforming: A new process for reducing CO<sub>2</sub> emissions. *Chemical Innovation*, 31(1), 21-26.
- Song, C.S. and Pan, W. (2004) Tri-reforming of methane: A novel concept for synthesis of industrially useful syngas with desired H<sub>2</sub>/CO ratios using flue gas of power plants without CO<sub>2</sub> pre-separation. *American Chemical Society, Division of Fuel Chemistry*, 49(1), 128-131.

- Song, H.Y., Park, E.D., and Lee, J.S. (2000) Oxidative carbonylation of phenol to diphenyl carbonate over supported palladium catalysts. *Journal of Molecular Catalysis A: Chemical*, 154(1–2), 243-250.
- Stiles, A.B. (1977) Methanol, past, present, and speculation on the future. *AIChE Journal*, 23(3), 362-375.
- Super, M., Berluche, E., Costello, C., and Beckman, E. (1997) Copolymerization of 1,2-Epoxyhexane and Carbon Dioxide Using Carbon Dioxide as Both Reactant and Solvent. *Macromolecules*, 30(3), 368-372.
- Takahashi, R., Sato, S., Sodesawa, T., Yoshida, M., and Tomiyama, S. (2004) Addition of zirconia in Ni/SiO<sub>2</sub> catalyst for improvement of steam resistance. *Applied Catalysis A: General*, 273(1-2), 211-215.
- Taniguchi, Y., Hayashida, T., Kitamura, T., and Fujiwara, Y. (1998) Vanadium-catalyzed acetic acid synthesis from methane and carbon dioxide. T. Inui, M. Anpo, K. Izui, S. Yanagida and T. Yamaguchi, *Advances in Chemical Conversions for Mitigating Carbon Dioxide, Proceedings of the Fourth International Conference on Carbon Dioxide Utilization* (pp. 439-442). New York: Elsevier.
- Tatsumi, T., Watanabe, Y., and Koyano, K.A. (1996) Synthesis of dimethyl carbonate from ethylene carbonate and methanol using TS-1 as solid base catalyst. *Chemical Communications*, (19), 2281-2282.
- Tomishige, K., Sakaihori, T., Ikeda, Y., and Fujimoto, K. (1999) A novel method of direct synthesis of dimethyl carbonate from methanol and carbon dioxide catalyzed by zirconia. *Catalysis Letters*, 58(4), 225-229.
- Tommasi, I., Aresta, M., Giannoccaro, P., Quaranta, E., and Fragale, C. (1998) Bioinorganic chemistry of nickel and carbon dioxide: A Ni complex behaving as a model system for carbon monoxide dehydrogenase enzyme. *Inorganica Chimica Acta*, 272(1–2), 38-42.
- Toyir, J., de la Piscina, P.R.r., Fierro, J.L.G., and Homs, N.s. (2001) Highly effective conversion of CO<sub>2</sub> to methanol over supported and promoted copper-based catalysts: influence of support and promoter. *Applied Catalysis B: Environmental*, 29(3), 207-215.

- Tremblay, J.-F. (2008) CO AS FEEDSTOCK Mitsui will make methanol from the greenhouse gas. *Chemical & Engineering News*, 86(35), 13.
- Uchiumi, S.-i., Ataka, K., and Matsuzaki, T. (1999) Oxidative reactions by a palladium–alkyl nitrite system. *Journal of Organometallic Chemistry*, 576(1–2), 279–289.
- Ugwu, C.U., Ogbonna, J.C., and Tanaka, H. (2005) Characterization of light utilization and biomass yields of Chlorella sorokiniana in inclined outdoor tubular photobioreactors equipped with static mixers. *Process Biochemistry*, 40(11), 3406–3411.
- Van-Dal, E.S. and Bouallou, C. (2013) Design and simulation of a methanol production plant from CO<sub>2</sub> hydrogenation. *Journal of Cleaner Production*, 57, 38–45.
- Wang, D., Dewaele, O., Groote, A.M.D., and Froment, G.F. (1996) Reaction Mechanism and Role of the Support in the Partial Oxidation of Methane on Rh/Al<sub>2</sub>O<sub>3</sub>. *Journal of Catalysis*, 159(2), 418–426.
- Watanabe, Y. and Tatsumi, T. (1998) Hydrotalcite-type materials as catalysts for the synthesis of dimethyl carbonate from ethylene carbonate and methanol. *Microporous and Mesoporous Materials*, 22(1–3), 399–407.
- Wu, J., Saito, M., Takeuchi, M., and Watanabe, T. (2001) The stability of Cu/ZnO-based catalysts in methanol synthesis from a CO<sub>2</sub>-rich feed and from a CO-rich feed. *Applied Catalysis A: General*, 218(1–2), 235–240.
- Xu, J. and Moulijn, J.A. (1996) Mitigation of CO<sub>2</sub> by Chemical Conversion: Plausible Chemical Reactions and Promising Products. *Energy & Fuels*, 10(2), 305–325.
- Xu, J. and Froment, G.F. (1989) Methane steam reforming, methanation and water-gas shift: I. Intrinsic kinetics. *AIChE Journal*, 35(1), 88–96.
- Yamaguchi, K., Ebitani, K., Yoshida, T., Yoshida, H., and Kaneda, K. (1999) Mg–Al Mixed Oxides as Highly Active Acid–Base Catalysts for Cycloaddition of Carbon Dioxide to Epoxides. *Journal of the American Chemical Society*, 121(18), 4526–4527.

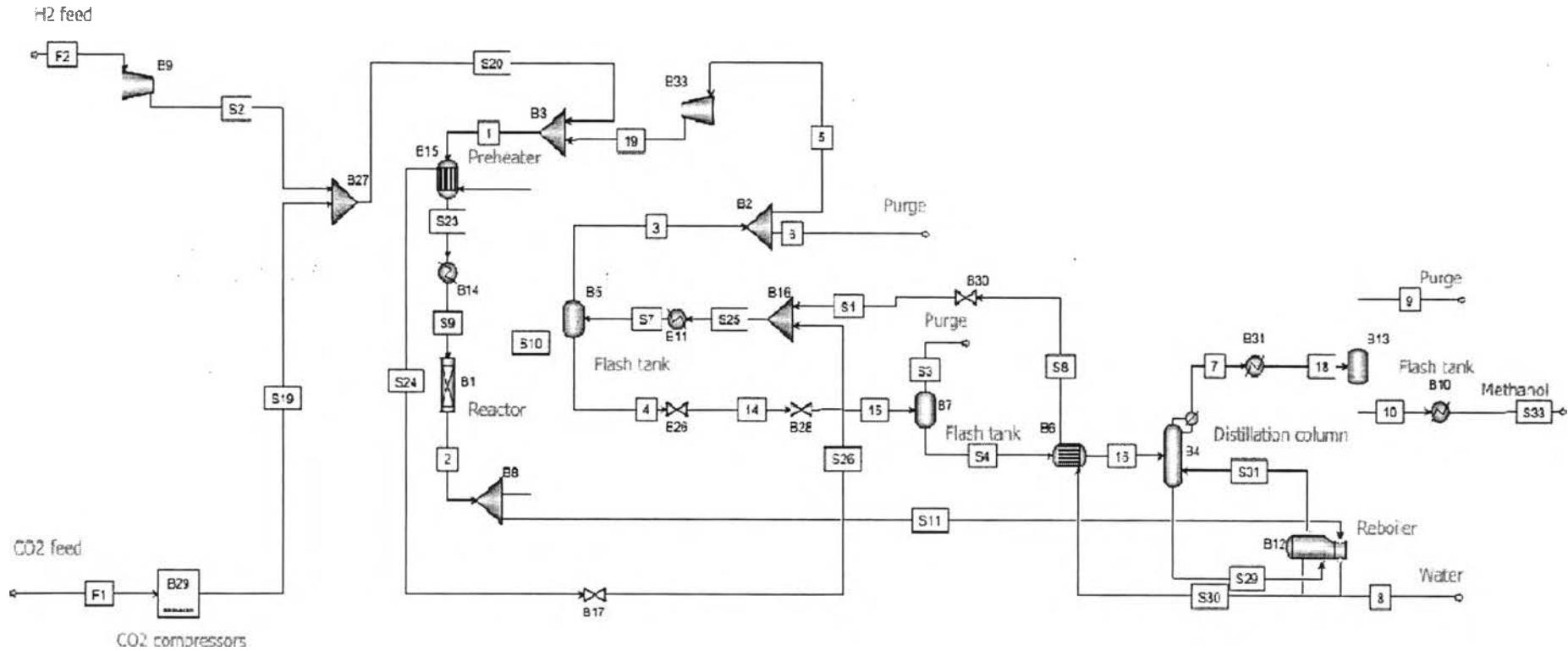
- Yang, C., Ma, Z., Zhao, N., Wei, W., Hu, T., and Sun, Y. (2006) Methanol synthesis from CO<sub>2</sub>-rich syngas over a ZrO<sub>2</sub> doped CuZnO catalyst. Catalysis Today, 115(1–4), 222-227.
- Yanji, W., Xinjiang, Z., Baoguo, Y., Bingchang, Z., and Jinsheng, C. (1998) Synthesis of dimethyl carbonate by gas-phase oxidative carbonylation of methanol on the supported solid catalyst I. Catalyst preparation and catalytic properties. Applied Catalysis A: General, 171(2), 255-260.
- Yano, T., Matsui, H., Koike, T., Ishiguro, H., Fujihara, H., Yoshihara, M., and Maeshima, T. (1997) Magnesium oxide-catalysed reaction of carbon dioxide with an epoxide with retention of stereochemistry. Chemical Communications, (12), 1129-1130.
- Yoneda, N., Kusano, S., Yasui, M., Pujado, P., and Wilcher, S. (2001) Recent advances in processes and catalysts for the production of acetic acid. Applied Catalysis A: General, 221(1–2), 253-265.
- York, A.E., Xiao, T., and Green, M.H. (2003) Brief Overview of the Partial Oxidation of Methane to Synthesis Gas. Topics in Catalysis, 22(3-4), 345-358.
- Yu, K.M.K., Curcic, I., Gabriel, J., and Tsang, S.C.E. (2008) Recent Advances in CO<sub>2</sub> Capture and Utilization. ChemSusChem, 1(11), 893-899.
- Yumurtaci, Z. and Bilgen, E. (2004) Hydrogen production from excess power in small hydroelectric installations. International Journal of Hydrogen Energy, 29(7), 687-693.
- Zhang, Y., Cruz, J., Zhang, S., Lou, H.H., and Benson, T.J. (2013) Process simulation and optimization of methanol production coupled to tri-reforming process. International Journal of Hydrogen Energy, 38(31), 13617-13630.
- Zhang, Y., Fei, J., Yu, Y., and Zheng, X. (2006) Methanol synthesis from CO<sub>2</sub> hydrogenation over Cu based catalyst supported on zirconia modified  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>. Energy Conversion and Management, 47(18–19), 3360-3367.
- Zhang, Y., Li, Z., Wen, X., and Liu, Y. (2006) Partial oxidation of methane over Ni/Ce-Ti-O catalysts. Chemical Engineering Journal, 121(2–3), 115-123.

Zicha, J. "Methanol, the Cheaper Gasoline Substitute." Spend Matters. 12 December 2014 <<http://spendmatters.com/2014/06/23/methanol-the-cheaper-gasoline-substitute/>>

## **APPENDICES**

**Appendix A CO<sub>2</sub> Conversion Process Flowsheet and Steam Tables Implemented by Aspen Plus 8.6**

### A.1 Hydrogenation of CO<sub>2</sub> into Methanol



**Figure A1.1** Flowsheet of the hydrogenation of CO<sub>2</sub> into methanol for the base case design.

**Table A1.1** Stream table of the hydrogenation of CO<sub>2</sub> into methanol for the base case design

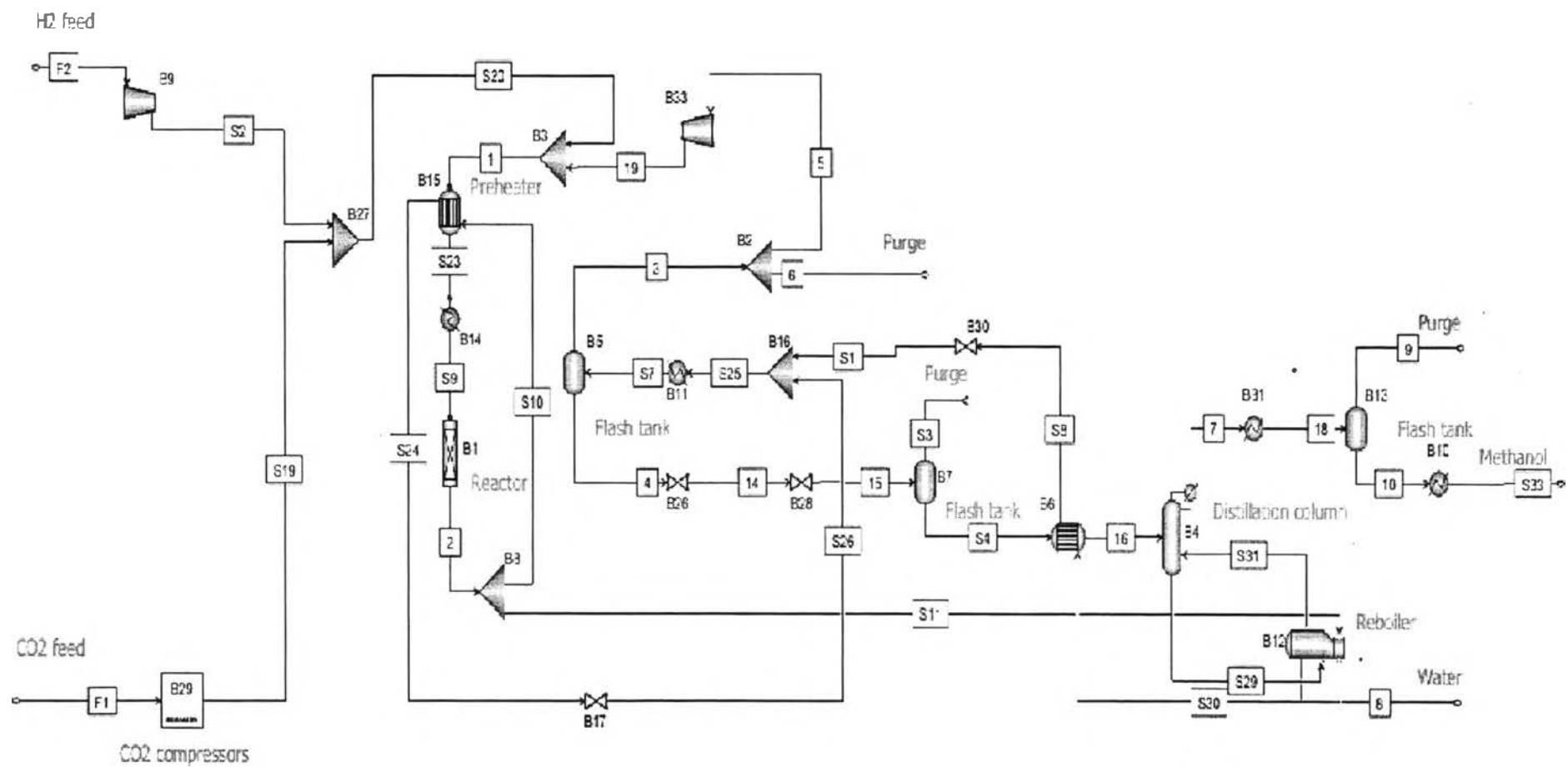
	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b>	<b>10</b>	<b>14</b>	<b>15</b>
<b>Temperature, °C</b>	58.3	284.5	40	40	40	40	64.3	102.3	50	50	30.2	27.5
<b>Pressure, bar</b>	76	74.36	73	73	73	73	1	1.1	1	1	10	1.2
<b>Vapor Frac</b>	1	1	1	0	1	1	1	0	1	0	0.03	0.041
<b>Mole Flow, kmol/hr</b>	17741.2	16476.1	15157.3	1318.74	15081.5	75.787	631.45	632.573	1.341	630.109	1318.74	1318.74
<b>Mass Flow, kg/hr</b>	122488	122488	89665.8	32822.7	89217.5	448.329	20216.5	11396	46.592	20169.9	32822.7	32822.7
<b>Volume Flow, cum/hr</b>	6596.32	10513.6	5557.47	51.195	5529.68	27.787	17406.9	16.12	35.654	34.871	148.913	1185.86
<b>Enthalpy, Gcal/hr</b>	-174.7	-153.84	-115.26	-81.761	-114.68	-0.576	-30.153	-42.315	-0.084	-35.788	-81.761	-81.761
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	76156.7	48164.9	47143.4	1021.83	46907.7	235.717	71.818	0	22.437	49.381	1021.83	1021.83
<b>H<sub>2</sub></b>	31194.3	27361.8	27309.1	52.701	27172.5	136.545	0.183	0	0.164	0.019	52.701	52.701
<b>WATER</b>	221.184	11679.5	222.295	11457.2	221.184	1.111	42.764	11395.9	0.019	42.745	11457.2	11457.2
<b>CO</b>	13059.5	13157.4	13125.1	32.373	13059.5	65.626	0.125	0	0.109	0.016	32.373	32.373
<b>METHANOL</b>	1856.62	22124.5	1865.95	20258.6	1856.62	9.33	20101.6	0.138	23.863	20077.7	20258.6	20258.6

**Table A1.1** Stream table of the hydrogenation of CO<sub>2</sub> into methanol for the base case design (con't.)

	<b>16</b>	<b>18</b>	<b>19</b>	<b>F1</b>	<b>F2</b>	<b>S1</b>	<b>S2</b>	<b>S3</b>	<b>S4</b>	<b>S7</b>	<b>S8</b>	<b>S9</b>
<b>Temperature, °C</b>	80	50	44.5	25	25	84.6	139.2	27.5	27.5	40	84.8	215
<b>Pressure, bar</b>	1.1	1	76	1	30	73	76	1.2	1.2	73	74.36	75.7
<b>Vapor Frac</b>	0.501	0.002	1	1	1	0.943	1	1	0	0.92	0.943	1
<b>Mole Flow, kmol/hr</b>	1264.02	631.45	15081.5	664.602	1995.03	6590.44	1995.03	54.715	1264.02	16476.1	6590.44	17741.2
<b>Mass Flow, kg/hr</b>	31612.5	20216.5	89217.5	29249	4021.74	48995.3	4021.74	1210.17	31612.5	122489	48995.3	122488
<b>Volume Flow, cum/hr</b>	16676.1	70.525	5395.87	16393.3	1679.54	2610.58	934.313	1137.05	48.812	5608.66	2565.54	9780.52
<b>Enthalpy, Gcal/hr</b>	-72.313	-35.872	-114.19	-62.471	0.012	-75.127	1.631	-2.353	-79.408	-197.02	-75.127	-153.84
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	71.817	71.818	46907.7	29249	0	19266	0	950.007	71.817	48165.2	19266	76156.7
<b>H<sub>2</sub></b>	0.183	0.183	27172.5	0	4021.74	10944.7	4021.74	52.518	0.183	27361.8	10944.7	31194.3
<b>WATER</b>	11438.7	42.764	221.184	0	0	4671.81	0	18.574	11438.7	11679.5	4671.81	221.184
<b>CO</b>	0.125	0.125	13059.5	0	0	5262.98	0	32.248	0.125	13157.5	5262.98	13059.5
<b>METHANOL</b>	20101.8	20101.6	1856.62	0	0	8849.81	0	156.818	20101.8	22124.5	8849.81	1856.62

**Table A1.1** Stream table of the hydrogenation of CO<sub>2</sub> into methanol for the base case design (con't.)

	S10	S11	S19	S20	S23	S24	S25	S26	S29	S30	S31	S33
<b>Temperature, °C</b>	284.5	284.5	127	129.9	215	81.7	82.7	81.5	102.3	159.9	102.3	40
<b>Pressure, bar</b>	74.36	74.36	76	76	75.7	74.36	73	73	1.1	74.36	1.1	1
<b>Vapor Frac</b>	1	1	1	1	1	0.94	0.941	0.94	0	1	1	0
<b>Mole Flow, kmol/hr</b>	9885.64	6590.44	664.602	2659.63	17741.2	9885.64	16476.1	9885.64	1299.23	6590.44	666.656	630.109
<b>Mass Flow, kg/hr</b>	73493.1	48995.3	29249	33270.8	122488	73493.1	122489	73493.1	23406.5	48995.3	12010.5	20169.9
<b>Volume Flow, cum/hr</b>	6308.2	4205.46	258.219	1187.74	9780.52	3806.9	6484.51	3873.78	33.109	3253.8	18747.8	34.436
<b>Enthalpy, Gcal/hr</b>	-92.305	-61.537	-62.137	-60.506	-153.84	-113.16	-188.29	-113.16	-86.91	-68.032	-38.1	-35.928
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	28899.1	19266	29249	29249	76156.7	28899.1	48165.2	28899.1	0	19266	0	49.381
<b>H<sub>2</sub></b>	16417.1	10944.7	0	4021.74	31194.3	16417.1	27361.8	16417.1	0	10944.7	0	0.019
<b>WATER</b>	7007.72	4671.81	0	0	221.184	7007.72	11679.5	7007.72	23405.3	4671.81	12009.4	42.745
<b>CO</b>	7894.49	5262.98	0	0	13059.5	7894.49	13157.5	7894.49	0	5262.98	0	0.016
<b>METHANOL</b>	13274.7	8849.81	0	0	1856.62	13274.7	22124.5	13274.7	1.217	8849.81	1.08	20077.7



**Figure A1.2** Flowsheet of the hydrogenation of CO<sub>2</sub> into methanol for the optimized case design.

**Table A1.2** Stream table of the hydrogenation of CO<sub>2</sub> into methanol for the optimized case design

	1	2	3	4	5	6	7	8	9	10	14	15
<b>Temperature, °C</b>	53.1	264.9	40	40	40	40	64.3	102.3	50	50	30.2	27.8
<b>Pressure, bar</b>	55.3	52.925	52.3	52.3	52.3	52.3	1	1.1	1	1	10	1.2
<b>Vapor Frac</b>	1	1	1	0	1	1	1	0	1	0	0.022	0.033
<b>Mole Flow, kmol/hr</b>	18740.2	17472.1	16161.4	1310.68	16080.6	80.807	633.702	634.122	1.665	632.038	1310.68	1310.68
<b>Mass Flow, kg/hr</b>	135943	135943	103189	32755	102673	515.943	20296.4	11423.9	58.953	20237.5	32755	32755
<b>Volume Flow, cum/hr</b>	9334.1	15009.2	8182.16	51.006	8141.24	40.911	17467.4	16.159	44.258	34.99	120.558	940.035
<b>Enthalpy, Gcal/hr</b>	-201.18	-180.55	-141.32	-81.682	-140.61	-0.707	-30.266	-42.419	-0.106	-35.896	-81.682	-81.682
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	88004.3	59968.4	59050.6	918.023	58755.3	295.253	80.742	0	29.045	51.697	918.023	918.023
<b>H<sub>2</sub></b>	32658	28817.4	28780.2	37.201	28636.3	143.901	0.166	0	0.151	0.014	37.201	37.201
<b>WATER</b>	308.377	11784.8	309.927	11474.9	308.377	1.55	36.299	11423.8	0.02	36.279	11474.9	11474.9
<b>CO</b>	12425.2	12508.5	12487.7	20.884	12425.2	62.438	0.103	0	0.093	0.011	20.884	20.884
<b>METHANOL</b>	2547.52	22864.4	2560.32	20304	2547.52	12.802	20179.1	0.149	29.643	20149.5	20304	20304

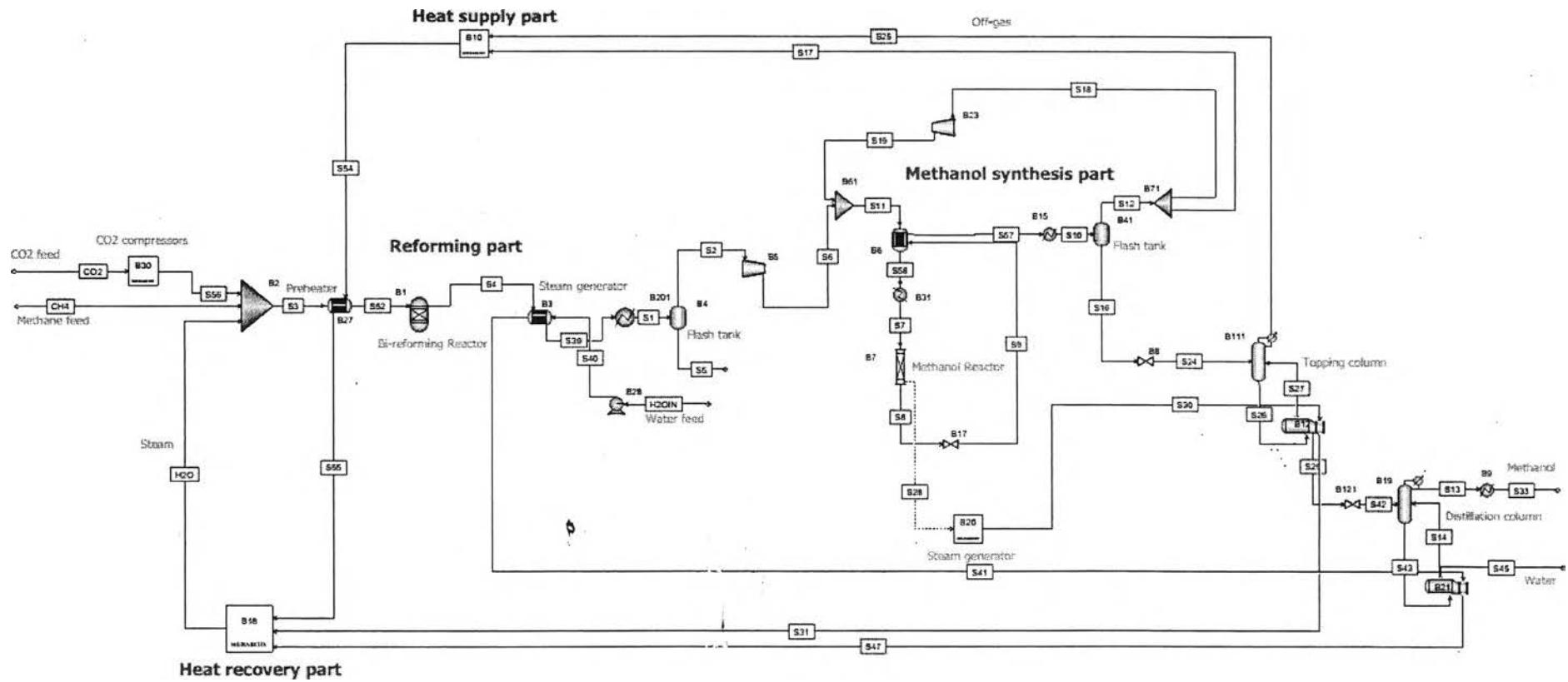
**Table A1.2** Stream table of the hydrogenation of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	16	18	19	F1	F2	S1	S2	S3	S4	S7	S8	S9
<b>Temperature, °C</b>	80	50	46.2	25	25	76.8	96.8	27.8	27.8	40	76.9	200
<b>Pressure, bar</b>	1.1	1	55.3	1	30	52.3	55.3	1.2	1.2	52.3	52.925	55
<b>Vapor Frac</b>	0.503	0.003	1	1	1	0.946	1	1	0	0.925	0.946	1
<b>Mole Flow, kmol/hr</b>	1267.83	633.702	16080.6	664.602	1995.03	6988.84	1995.03	42.853	1267.83	17472.1	6988.84	18740.2
<b>Mass Flow, kg/hr</b>	31720.4	20296.4	102673	29249	4021.74	54377.4	4021.74	1034.64	31720.4	135944	54377.4	135943
<b>Volume Flow, cum/hr</b>	16795.4	79.248	7864.03	16393.3	1679.54	3749.45	1143.24	891.044	48.991	8233.16	3707.27	13673.1
<b>Enthalpy, Gcal/hr</b>	-72.509	-36.002	-139.9	-62.471	0.012	-85.846	1.024	-2.041	-79.64	-223	-85.846	-180.55
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	80.742	80.742	58755.3	29249	0	23987.3	0	837.281	80.742	59968.6	23987.3	88004.3
<b>H<sub>2</sub></b>	0.166	0.166	28636.3	0	4021.74	11527	4021.74	37.035	0.166	28817.4	11527	32658
<b>WATER</b>	11460.1	36.299	308.377	0	0	4713.93	0	14.785	11460.1	11784.8	4713.93	308.377
<b>CO</b>	0.103	0.103	12425.2	0	0	5003.41	0	20.781	0.103	12508.6	5003.41	12425.2
<b>METHANOL</b>	20179.3	20179.1	2547.52	0	0	9145.75	0	124.76	20179.3	22864.3	9145.75	2547.52

**Table A1.2** Stream table of the hydrogenation of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	S10	S11	S19	S20	S23	S24	S25	S26	S29	S30	S31	S33
<b>Temperature, °C</b>	264.9	264.9	95.1	91.7	200	75.8	76.1	75.7	102.3	146.6	102.3	40
<b>Pressure, bar</b>	52.925	52.925	55.3	55.3	55	52.925	52.3	52.3	1.1	52.925	1.1	1
<b>Vapor Frac</b>	1	1	1	1	1	0.945	0.945	0.945	0	1	1	0
<b>Mole Flow, kmol/hr</b>	10483.2	6988.84	664.602	2659.63	18740.2	10483.2	17472.1	10483.2	1300.78	6988.84	666.655	632.038
<b>Mass Flow, kg/hr</b>	81566.1	54377.4	29249	33270.8	135943	81566.1	135944	81566.1	23434.4	54377.4	12010.5	20237.5
<b>Volume Flow, cum/hr</b>	9005.52	6003.68	321.186	1462.83	13673.1	5537.87	9350.39	5600.9	33.149	4664.51	18747.8	34.552
<b>Enthalpy, Gcal/hr</b>	-108.33	-72.219	-62.307	-61.283	-180.55	-128.96	-214.81	-128.96	-87.014	-78.714	-38.1	-36.036
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	35981.1	23987.3	29249	29249	88004.3	35981.1	59968.6	35981.1	0	23987.3	0	51.697
<b>H<sub>2</sub></b>	17290.4	11527	0	4021.74	32658	17290.4	28817.4	17290.4	0	11527	0	0.014
<b>WATER</b>	7070.89	4713.93	0	0	308.377	7070.89	11784.8	7070.89	23433.1	4713.93	12009.3	36.279
<b>CO</b>	7505.14	5003.41	0	0	12425.2	7505.14	12508.6	7505.14	0	5003.41	0	0.011
<b>METHANOL</b>	13718.6	9145.75	0	0	2547.52	13718.6	22864.3	13718.6	1.314	9145.75	1.165	20149.5

## A.2 Bi-reforming of CO<sub>2</sub> into Methanol



**Figure A2.1** Flowsheet of the bi-reforming of CO<sub>2</sub> into methanol for the base case design.

**Table A2.1** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the base case design

	CH <sub>4</sub>	CO <sub>2</sub>	H <sub>2</sub> O	H <sub>2</sub> OIN	S1	S2	S3	S4	S5	S6	S7	S8
<b>Temperature, °C</b>	25	50	227	25	55	55	199.4	920	55	173.5	220	246
<b>Pressure, bar</b>	25	1	25	1	25	25	25	25	25	63.3	63	61.376
<b>Vapor Frac</b>	1	1	1	0	0.729	1	0.938	1	0	1	1	1
<b>Mole Flow, kmol/hr</b>	589.66	152.722	1474.1	2000	3270.96	2383.76	2216.48	3270.96	887.2	2383.76	12001.8	10725
<b>Mass Flow, kg/hr</b>	9459.78	6721.27	26556.3	36030.6	42737.4	26753	42737.4	42737.4	15984.4	26753	108523	108523
<b>Volume Flow, cum/hr</b>	560.782	4087.81	2145.93	36.137	2649.36	2632.79	3065.72	13044.1	16.572	1440.61	7992.71	7665.15
<b>Enthalpy, Gcal/hr</b>	-10.554	-14.321	-83.527	-136.62	-93.476	-32.787	-108.29	-60.678	-60.689	-30.689	-118.23	-129.25
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	0	6721.27	0	0	8775.92	8773.67	6721.27	8775.92	2.253	8773.67	35695.5	28387.7
<b>CH<sub>4</sub></b>	9459.78	0	0	0	1001.4	1001.39	9459.78	1001.4	0.014	1001.39	32652.4	32652.1
<b>CO</b>	0	0	0	0	13460.5	13460.5	0	13460.5	0.019	13460.5	21473.5	8243.49
<b>H<sub>2</sub></b>	0	0	0	0	3282.67	3282.66	0	3282.67	0.008	3282.66	16774.8	13866.2
<b>WATER</b>	0	0	26556.3	36030.6	16216.9	234.774	26556.3	16216.9	15982.1	234.774	301.82	3293.22
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	0	0	0	0	0	0	0	1625.02	22079.9
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.1** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the base case design (con't.)

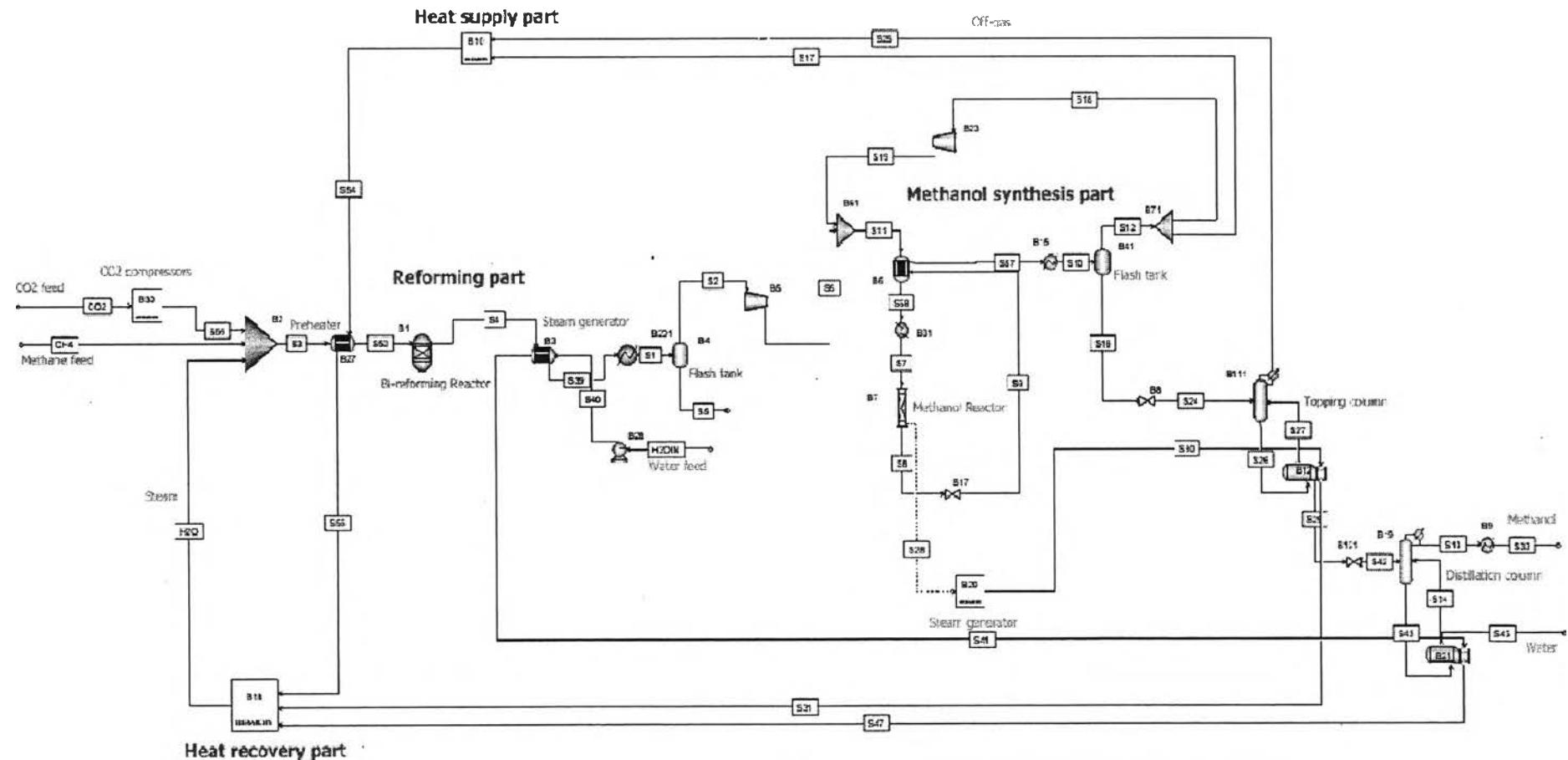
	<b>S9</b>	<b>S10</b>	<b>S11</b>	<b>S12</b>	<b>S13</b>	<b>S14</b>	<b>S16</b>	<b>S17</b>	<b>S18</b>	<b>S19</b>	<b>S24</b>	<b>S25</b>
<b>Temperature, °C</b>	246	35	66.1	35	63.9	99.5	35	35	35	40.1	34.8	51.7
<b>Pressure, bar</b>	60.3	60.3	63.3	60.3	1	1.1	60.3	60.3	60.3	63.3	17.5	17
<b>Vapor Frac</b>	1	0.92	1	1	0	1	0	1	1	1	0.025	1
<b>Mole Flow, kmol/hr</b>	10725	10725.1	12001.8	9864.68	635.286	792.714	860.387	246.617	9618.06	9618.06	860.387	46.103
<b>Mass Flow, kg/hr</b>	108523	108523	108523	83866.8	20316.2	15474.2	24656.1	2096.67	81770.1	81770.1	24656.1	1074.24
<b>Volume Flow, cum/hr</b>	7799.23	4259.62	5427.82	4219.18	35.823	22108.7	40.436	105.48	4113.7	3990.89	71.442	71.692
<b>Enthalpy, Gcal/hr</b>	-129.25	-155.82	-132.96	-105.27	-35.845	-44.474	-50.556	-2.632	-102.63	-102.27	-50.556	-1.966
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	28387.7	28387.8	35695.5	27612.1	1.734	0	775.616	690.303	26921.8	26921.8	775.616	773.882
<b>CH<sub>4</sub></b>	32652.1	32652.3	32652.4	32462.6	0.006	0	189.694	811.564	31651	31651	189.694	189.688
<b>CO</b>	8243.49	8243.5	21473.5	8218.54	0	0	24.962	205.464	8013.08	8013.08	24.962	24.962
<b>H<sub>2</sub></b>	13866.2	13866.3	16774.8	13838.1	0	0	28.202	345.952	13492.1	13492.1	28.202	28.202
<b>WATER</b>	3293.22	3293.22	301.82	68.765	51.677	12748.4	3224.46	1.719	67.046	67.046	3224.46	0.8
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	22079.9	22079.9	1625.02	1666.69	20262.8	2725.81	20413.2	41.667	1625.02	1625.02	20413.2	56.701
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.1** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the base case design (con't.)

	S26	S27	S29	S30	S31	S33	S39	S40	S41	S42	S43	S45
<b>Temperature, °C</b>	162.9	164.5	164.5	253.3	253.3	40	150	261	286.7	70.9	91.1	99.5
<b>Pressure, bar</b>	17.5	17.5	17.5	42	42	1	25	42	42	1.1	1.1	1.1
<b>Vapor Frac</b>	0	1	0	1	0.54	0	0.898	0	1	0.276	0	0
<b>Mole Flow, kmol/hr</b>	1253.43	439.15	814.284	938.117	938.117	635.286	3270.96	2000	2000	814.284	971.711	178.998
<b>Mass Flow, kg/hr</b>	36831.7	13249.8	23581.9	16900.4	16900.4	20316.2	42737.4	36030.6	36030.6	23581.9	18740	3265.74
<b>Volume Flow, cum/hr</b>	76.675	780.373	48.893	799.841	441.324	34.715	4149.36	36.081	1926.35	5766.83	27.288	4.643
<b>Enthalpy, Gcal/hr</b>	-70.151	-21.186	-45.808	-53.206	-56.362	-36.186	-84.776	-136.55	-112.45	-45.808	-64.194	-11.949
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	32.891	31.158	1.734	0	0	1.734	8775.92	0	0	1.734	0	0
<b>CH<sub>4</sub></b>	0.321	0.315	0.006	0	0	0.006	1001.4	0	0	0.006	0	0
<b>CO</b>	0.012	0.012	0	0	0	0	13460.5	0	0	0	0	0
<b>H<sub>2</sub></b>	0.011	0.01	0	0	0	0	3282.67	0	0	0	0	0
<b>WATER</b>	4289.07	1065.41	3223.66	16900.4	16900.4	51.677	16216.9	36030.6	36030.6	3223.66	15920.4	3171.98
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	32509.4	12152.9	20356.5	0	0	20262.8	0	0	0	20356.5	2819.57	93.758
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.1** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the base case design (con't.)

	S47	S52	S54	S55	S56	S57	S58
<b>Temperature, °C</b>	253.3	650	1000	659.4	137.7	101.6	220
<b>Pressure, bar</b>	42	25	1	1	25	60.3	63
<b>Vapor Frac</b>	0.535	1	1	1	1	0.986	1
<b>Mole Flow, kmol/hr</b>	2000	2216.48	3979.29	3979.29	152.722	10725	12001.8
<b>Mass Flow, kg/hr</b>	36030.6	42737.4	109395	109395	6721.27	108523	108523
<b>Volume Flow, cum/hr</b>	933.871	6796.17	421326	308633	200.92	5491.19	7992.71
<b>Enthalpy, Gcal/hr</b>	-120.22	-96.333	-44.43	-56.389	-14.211	-143.98	-118.23
<b>Mass Flow, kg/hr</b>							
<b>CO<sub>2</sub></b>	0	6721.27	14808.4	14808.4	6721.27	28387.7	35695.5
<b>CH<sub>4</sub></b>	0	9459.78	0	0	0	32652.1	32652.4
<b>CO</b>	0	0	0.006	0.006	0	8243.49	21473.5
<b>H<sub>2</sub></b>	0	0	0.001	0.001	0	13866.2	16774.8
<b>WATER</b>	36030.6	26556.3	13934.1	13934.1	0	3293.22	301.82
<b>O<sub>2</sub></b>	0	0	2185.84	2185.84	0	0	0
<b>C</b>	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	0	0	22079.9	1625.02
<b>N<sub>2</sub></b>	0	0	78466.9	78466.9	0	0	0



**Figure A2.2** Flowsheet of the bi-reforming of CO<sub>2</sub> into methanol for the optimized case design.

**Table A2.2** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the optimized case design

	<b>CH<sub>4</sub></b>	<b>CO<sub>2</sub></b>	<b>H<sub>2</sub>O</b>	<b>H<sub>2</sub>OIN</b>	<b>S1</b>	<b>S2</b>	<b>S3</b>	<b>S4</b>	<b>S5</b>	<b>S6</b>	<b>S7</b>	<b>S8</b>
<b>Temperature, °C</b>	25	50	227	25	55	55	199.3	915	55	168.8	215	241.3
<b>Pressure, bar</b>	25	1	25	1	25	25	25	25	25	61.3	61	59.28
<b>Vapor Frac</b>	1	1	1	0	0.728	1	0.938	1	0	1	1	1
<b>Mole Flow, kmol/hr</b>	589.66	152.722	1469.95	2000	3257.71	2371.55	2212.34	3257.71	886.162	2371.55	12105.7	10839.1
<b>Mass Flow, kg/hr</b>	9459.78	6721.27	26481.6	36030.6	42662.7	26696.9	42662.7	42662.7	15965.7	26696.9	112149	112149
<b>Volume Flow, cum/hr</b>	560.782	4087.81	2139.89	36.137	2635.68	2619.13	3059.7	12937.1	16.553	1463.24	8233.92	7941.43
<b>Enthalpy, Gcal/hr</b>	-10.554	-14.321	-83.292	-136.62	-93.466	-32.849	-108.06	-60.876	-60.618	-30.844	-124.15	-134.96
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	0	6721.27	0	0	8842.06	8839.78	6721.27	8842.06	2.279	8839.78	37027.4	29685.2
<b>CH<sub>4</sub></b>	9459.78	0	0	0	1074.42	1074.41	9459.78	1074.42	0.015	1074.41	35355.3	35355
<b>CO</b>	0	0	0	0	13290.9	13290.9	0	13290.9	0.019	13290.9	21173.7	8108.25
<b>H<sub>2</sub></b>	0	0	0	0	3258.17	3258.16	0	3258.17	0.008	3258.16	16600.9	13711.3
<b>WATER</b>	0	0	26481.6	36030.6	16197.1	233.699	26481.6	16197.1	15963.4	233.699	304.036	3309.52
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	0	0	0	0	0	0	0	1687.75	21979.4
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.2** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the optimized case design (con't.)

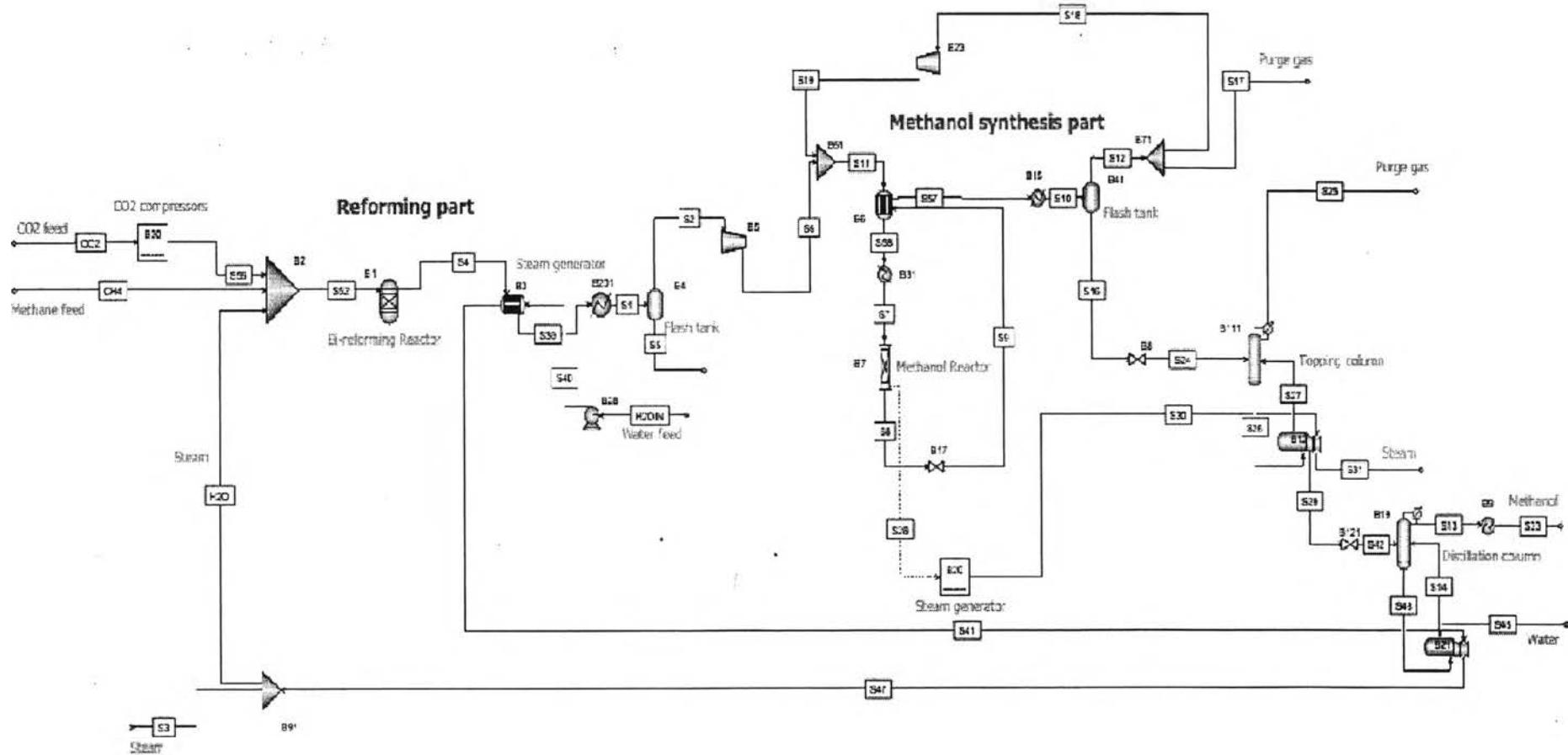
	S9	S10	S11	S12	S13	S14	S16	S17	S18	S19	S24	S25
<b>Temperature, °C</b>	241.3	35	64.8	35	63.9	102.3	35	35	35	40.2	34.8	70.6
<b>Pressure, bar</b>	58.3	58.3	61.3	58.3	1	1.1	58.3	58.3	58.3	61.3	17.5	17
<b>Vapor Frac</b>	1	0.921	1	1	0	1	0	1	1	1	0.025	1
<b>Mole Flow, kmol/hr</b>	10839.1	10839.1	12105.7	9983.74	633.401	797.549	855.391	249.593	9734.15	9734.15	855.391	47.666
<b>Mass Flow, kg/hr</b>	112149	112149	112149	87643.2	20222.2	14368.1	24505.7	2191.08	85452.1	85452.1	24505.7	1143.1
<b>Volume Flow, cum/hr</b>	8072.33	4447.19	5622.36	4407.03	35.64	22428.9	40.164	110.176	4296.86	4163.89	70.052	78.174
<b>Enthalpy, Gcal/hr</b>	-134.96	-161.35	-138.72	-111.03	-35.764	-45.581	-50.317	-2.776	-108.25	-107.88	-50.317	-2.064
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	29685.2	29685.2	37027.4	28910.3	1.625	0	774.857	722.759	28187.6	28187.6	774.857	773.231
<b>CH<sub>4</sub></b>	35355	35355.2	35355.3	35159.9	0.006	0	195.327	878.997	34280.9	34280.9	195.327	195.321
<b>CO</b>	8108.25	8108.27	21173.7	8084.95	0	0	23.326	202.124	7882.82	7882.82	23.326	23.325
<b>H<sub>2</sub></b>	13711.3	13711.4	16600.9	13684.9	0	0	26.501	342.121	13342.7	13342.7	26.501	26.5
<b>WATER</b>	3309.52	3309.52	304.036	72.14	94.816	14368	3237.38	1.804	70.337	70.337	3237.38	2.086
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	21979.4	21979.4	1687.75	1731.02	20125.7	0.132	20248.3	43.276	1687.75	1687.75	20248.3	122.639
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.2** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	S26	S27	S29	S30	S31	S33	S39	S40	S41	S42	S43	S45
<b>Temperature, °C</b>	163	164.6	164.6	253.3	253.3	40	150	26.1	280.3	71	102.3	102.3
<b>Pressure, bar</b>	17.5	17.5	17.5	42	42	1	25	42	42	1.1	1.1	1.1
<b>Vapor Frac</b>	0	1	0	1	0.525	0	0.897	0	1	0.276	0	0
<b>Mole Flow, kmol/hr</b>	1246.92	439.199	807.724	909.122	909.122	633.401	3257.71	2000	2000	807.724	971.873	174.323
<b>Mass Flow, kg/hr</b>	36603.9	13241.3	23362.6	16378.1	16378.1	20222.2	42662.7	36030.6	36030.6	23362.6	17508.6	3140.48
<b>Volume Flow, cum/hr</b>	76.174	780.69	48.408	775.119	416.603	34.535	4128.11	36.081	1886.81	5716.99	24.766	4.442
<b>Enthalpy, Gcal/hr</b>	-69.813	-21.194	-45.463	-51.561	-54.718	-36.105	-84.801	-136.55	-112.63	-45.463	-65.013	-11.661
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	31.205	29.58	1.625	0	0	1.625	8842.06	0	0	1.625	0	0
<b>CH<sub>4</sub></b>	0.317	0.311	0.006	0	0	0.006	1074.42	0	0	0.006	0	0
<b>CO</b>	0.011	0.011	0	0	0	0	13290.9	0	0	0	0	0
<b>H<sub>2</sub></b>	0.01	0.009	0	0	0	0	3258.17	0	0	0	0	0
<b>WATER</b>	4313.11	1077.82	3235.3	16378.1	16378.1	94.816	16197.1	36030.6	36030.6	3235.3	17508.5	3140.48
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	32259.2	12133.5	20125.7	0	0	20125.7	0	0	0	20125.7	0.136	0.004
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.2** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	S47	S52	S54	S55	S56	S57	S58
<b>Temperature, °C</b>	253.3	650	1000	656	137.7	100.4	215
<b>Pressure, bar</b>	42	25	1	1	25	58.3	61
<b>Vapor Frac</b>	0.523	1	1	1	1	0.986	1
<b>Mole Flow, kmol/hr</b>	2000	2212.34	3934.36	3934.36	152.722	10839.1	12105.7
<b>Mass Flow, kg/hr</b>	36030.6	42662.7	108180	108180	6721.27	112149	112149
<b>Volume Flow, cum/hr</b>	914.155	6783.54	416568	304034	200.92	5717.45	8233.92
<b>Enthalpy, Gcal/hr</b>	-120.4	-96.115	-44.086	-56.028	-14.211	-149.53	-124.15
<b>Mass Flow, kg/hr</b>							
<b>CO<sub>2</sub></b>	0	6721.27	14707.4	14707.4	6721.27	29685.2	37027.4
<b>CH<sub>4</sub></b>	0	9459.78	0	0	0	35355	35355.3
<b>CO</b>	0	0	0.006	0.006	0	8108.25	21173.7
<b>H<sub>2</sub></b>	0	0	0.001	0.001	0	13711.3	16600.9
<b>WATER</b>	36030.6	26481.6	13784.2	13784.2	0	3309.52	304.036
<b>O<sub>2</sub></b>	0	0	2159.7	2159.7	0	0	0
<b>C</b>	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	0	0	21979.4	1687.75
<b>N<sub>2</sub></b>	0	0	77528.5	77528.5	0	0	0



**Figure A2.3** Flowsheet of the bi-reforming of  $\text{CO}_2$  into methanol for the alternative case design.

**Table A2.3** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the alternative case design

	CH <sub>4</sub>	CO <sub>2</sub>	H <sub>2</sub> O	H <sub>2</sub> OIN	S1	S2	S3	S4	S5	S6	S7	S8
<b>Temperature, °C</b>	25	50	252	25	55	55	252	915	55	168.8	215	241.3
<b>Pressure, bar</b>	25	1	42	1	25	25	42	25	25	61.3	61	59.28
<b>Vapor Frac</b>	1	1	0.5	0	0.728	1	0.5	1	0	1	1	1
<b>Mole Flow, kmol/hr</b>	589.66	152.722	1470	2000	3257.77	2371.57	530	3257.77	886.199	2371.57	12105.2	10838.6
<b>Mass Flow, kg/hr</b>	9459.78	6721.27	26482.5	36030.6	42663.5	26697.1	9548.1	42663.5	15966.4	26697.1	112141	112141
<b>Volume Flow, cum/hr</b>	560.782	4087.81	669.909	36.137	2635.7	2619.15	241.532	12937.3	16.553	1463.25	8233.6	7941.04
<b>Enthalpy, Gcal/hr</b>	-10.554	-14.321	-88.49	-136.62	-93.469	-32.849	-31.905	-60.879	-60.62	-30.844	-124.14	-134.95
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	0	6721.27	0	0	8842.3	8840.02	0	8842.3	2.279	8840.02	37026.6	29684.2
<b>CH<sub>4</sub></b>	9459.78	0	0	0	1074.37	1074.35	0	1074.37	0.015	1074.35	35349.1	35348.8
<b>CO</b>	0	0	0	0	13290.8	13290.8	0	13290.8	0.019	13290.8	21173.1	8107.65
<b>H<sub>2</sub></b>	0	0	0	0	3258.2	3258.2	0	3258.2	0.008	3258.2	16600.8	13711.2
<b>WATER</b>	0	0	26482.5	36030.6	16197.8	233.701	9548.1	16197.8	15964.1	233.701	304.035	3309.61
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	0	0	0	0	0	0	1687.64	21979.3	
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

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**Table A2.3** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the alternative case design (con<sup>\*\*</sup>.)

	S9	S10	S11	S12	S13	S14	S16	S17	S18	S19	S24	S25
<b>Temperature, °C</b>	241.3	35	64.8	35	63.9	102.3	35	35	35	40.2	34.8	70.5
<b>Pressure, bar</b>	58.3	58.3	61.3	58.3	1	1.1	58.3	58.3	58.3	61.3	17.5	17
<b>Vapor Frac</b>	1	0.921	1	1	0	1	0	1	1	1	0.025	1
<b>Mole Flow, kmol/hr</b>	10838.6	10838.6	12105.2	9983.23	633.403	797.549	855.398	249.581	9733.65	9733.65	855.398	47.664
<b>Mass Flow, kg/hr</b>	112141	112141	112141	87635.1	20222.3	14368.1	24505.9	2190.88	85444.2	85444.2	24505.9	1143.03
<b>Volume Flow, cum/hr</b>	8071.96	4446.97	5622.17	4406.82	35.64	22428.9	40.164	110.17	4296.64	4163.68	70.051	78.166
<b>Enthalpy, Gcal/hr</b>	-134.95	-161.34	-138.76	-111.02	-35.765	-45.581	-50.317	-2.775	-108.24	-107.87	-50.317	-2.064
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	29684.2	29684.2	37026.6	28909.3	1.626	0	774.875	722.733	28186.6	28186.6	774.875	773.25
<b>CH<sub>4</sub></b>	35348.8	35348.8	35349.1	35153.6	0.006	0	195.303	878.841	34274.8	34274.8	195.303	195.297
<b>CO</b>	8107.65	8107.65	21173.1	8084.34	0	0	23.325	202.109	7882.23	7882.23	23.325	23.325
<b>H<sub>2</sub></b>	13711.2	13711.2	16600.8	13684.7	0	0	26.502	342.118	13342.6	13342.6	26.502	26.502
<b>WATER</b>	3309.61	3309.61	304.035	72.137	94.757	14368	3237.47	1.803	70.334	70.334	3237.47	2.085
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	21979.3	21979.3	1687.64	1730.91	20125.9	0.132	20248.4	43.273	1687.64	1687.64	20248.4	122.569
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

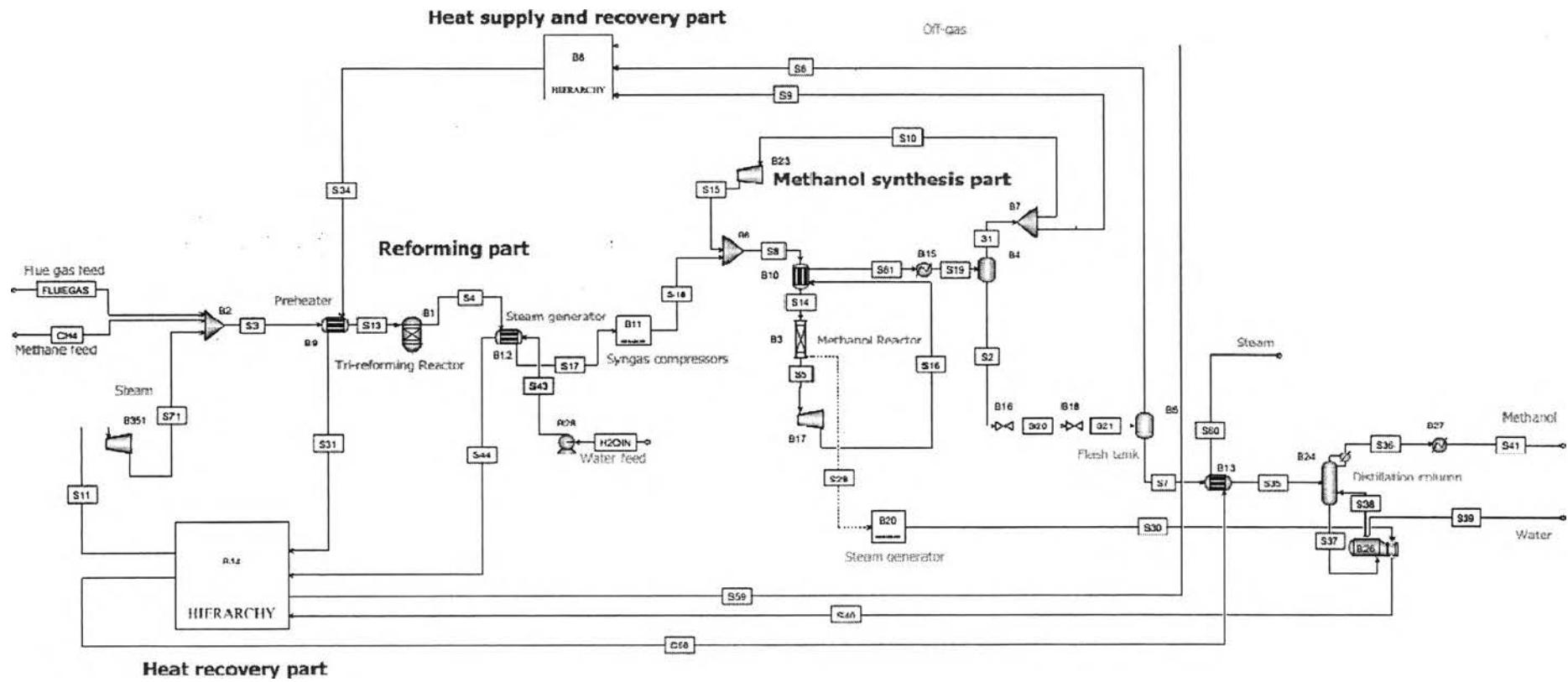
**Table A2.3** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the alternative case design (con't.)

	S26	S27	S29	S30	S31	S33	S39	S40	S41	S42	S43	S45
<b>Temperature, °C</b>	163	164.6	164.6	253.3	253.3	40	150	26.1	280.3	71	102.3	102.3
<b>Pressure, bar</b>	17.5	17.5	17.5	42	42	1	25	42	42	1.1	1.1	1.1
<b>Vapor Frac</b>	0	1	0	1	0.525	0	0.897	0	1	0.276	0	0
<b>Mole Flow, kmol/hr</b>	1246.93	439.198	807.735	909.15	909.15	633.403	3257.77	2000	2000	807.735	971.881	174.332
<b>Mass Flow, kg/hr</b>	36604.1	13241.2	23362.9	16378.6	16378.6	20222.3	42663.5	36030.6	36030.6	23362.9	17508.8	3140.64
<b>Volume Flow, cum/hr</b>	76.174	780.689	48.408	775.143	416.627	34.535	4128.13	36.081	1886.96	5717.06	24.766	4.442
<b>Enthalpy, Gcal/hr</b>	-69.814	-21.193	-45.464	-51.563	-54.719	-36.105	-84.804	-136.55	-112.62	-45.464	-65.013	-11.662
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	31.209	29.583	1.626	0	0	1.626	8842.3	0	0	1.626	0	0
<b>CH<sub>4</sub></b>	0.317	0.311	0.006	0	0	0.006	1074.37	0	0	0.006	0	0
<b>CO</b>	0.011	0.011	0	0	0	0	13290.8	0	0	0	0	0
<b>H<sub>2</sub></b>	0.01	0.009	0	0	0	0	3258.2	0	0	0	0	0
<b>WATER</b>	4313.22	1077.83	3235.39	16378.6	16378.6	94.757	16197.8	36030.6	36030.6	3235.39	17508.6	3140.63
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	32259.4	12133.5	20125.9	0	0	20125.9	0	0	0	20125.9	0.136	0.004
<b>N<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0

**Table A2.3** Stream table of bi-reforming of CO<sub>2</sub> into methanol for the alternative case design (con't.)

	S47	S52	S56	S57	S58
<b>Temperature, °C</b>	253.3	188.2	137.7	100.4	215
<b>Pressure, bar</b>	42	25	25	58.3	61
<b>Vapor Frac</b>	0.523	0.684	1	0.986	1
<b>Mole Flow, kmol/hr</b>	2000	2212.38	152.722	10838.6	12105.2
<b>Mass Flow, kg/hr</b>	36030.6	42663.5	6721.27	112141	112141
<b>Volume Flow, cum/hr</b>	914.232	2221.62	200.92	5717.21	8233.6
<b>Enthalpy, Gcal/hr</b>	-120.4	-113.26	-14.211	-149.52	-124.14
<b>Mass Flow, kg/hr</b>					
<b>CO<sub>2</sub></b>	0	6721.27	6721.27	29684.2	37026.6
<b>CH<sub>4</sub></b>	0	9459.78	0	35348.8	35349.1
<b>CO</b>	0	0	0	8107.65	21173.1
<b>H<sub>2</sub></b>	0	0	0	13711.2	16600.8
<b>WATER</b>	36030.6	26482.5	0	3309.61	304.035
<b>O<sub>2</sub></b>	0	0	0	0	0
<b>C</b>	0	0	0	0	0
<b>METHANOL</b>	0	0	0	21979.3	1687.64
<b>N<sub>2</sub></b>	0	0	0	0	0

### A.3 Tri-reforming of CO<sub>2</sub> into Methanol



**Table A3.1** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the base case design

	CH <sub>4</sub>	FLUEGAS	H <sub>2</sub> OIN	S1	S2	S3	S4	S5	S6	S7	S8	S9
<b>Temperature, °C</b>	25	150	25	35	35	123.1	850	235.1	31	31	48.9	35
<b>Pressure, bar</b>	1	1	1	62.3	62.3	1	1	61.357	1.2	1.2	65.3	62.3
<b>Vapor Frac</b>	1	1	0	1	0	1	1	1	1	0	1	1
<b>Mole Flow, kmol/hr</b>	569.212	1003.748	1414.03	40890.9	680.937	1966.24	3054.82	41571.8	31.954	648.983	42850.5	1022.27
<b>Mass Flow, kg/hr</b>	9131.73	29856.014	25474.1	935695	21367.6	46072.7	46072.7	957063	871.354	20496.2	957062	23392.4
<b>Volume Flow, cum/hr</b>	14086.4	35316.302	25.549	17170.1	35.911	64709.3	285329	28584.2	671.505	34.496	18030.5	429.253
<b>Enthalpy, Gcal/hr</b>	-10.134	-16.105	-96.592	-118.55	-37.955	-48.342	-7.005	-87.955	-0.641	-37.313	-132.32	-2.964
<b>Mass Flow, kg/hr</b>												
CO <sub>2</sub>	0	6184.466	0	25616.8	68.462	6184.47	1451.31	25785.3	147.08	21.382	26427.6	640.421
CH <sub>4</sub>	9131.73	0	0	3547.24	5.096	9131.73	93.794	3552.34	4.999	0.097	3552.37	88.681
CO	0	0	0	50215.3	36.74	0	18792.5	50252.1	36.491	0.249	67752.4	1255.38
H <sub>2</sub>	0	0	0	16866.1	7.543	0	3036.45	16873.6	7.503	0.04	19480.9	421.653
WATER	0	1193.465	25474.1	48.362	390.741	8278.45	1441.07	439.103	1.163	389.578	176.131	1.209
O <sub>2</sub>	0	1220.512	0	0	0	1220.51	0	0	0	0	0	0
C	0	0	0	0	0	0	0	0	0	0	0	0
METHANOL	0	0	0	8593.35	20272.2	0	0	28865.6	189.999	20082.2	8378.5	214.834
N <sub>2</sub>	0	21257.571	0	830808	486.752	21257.6	21257.6	831295	484.118	2.634	831294	20770.2

**Table A3.1** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the base case design (con't.)

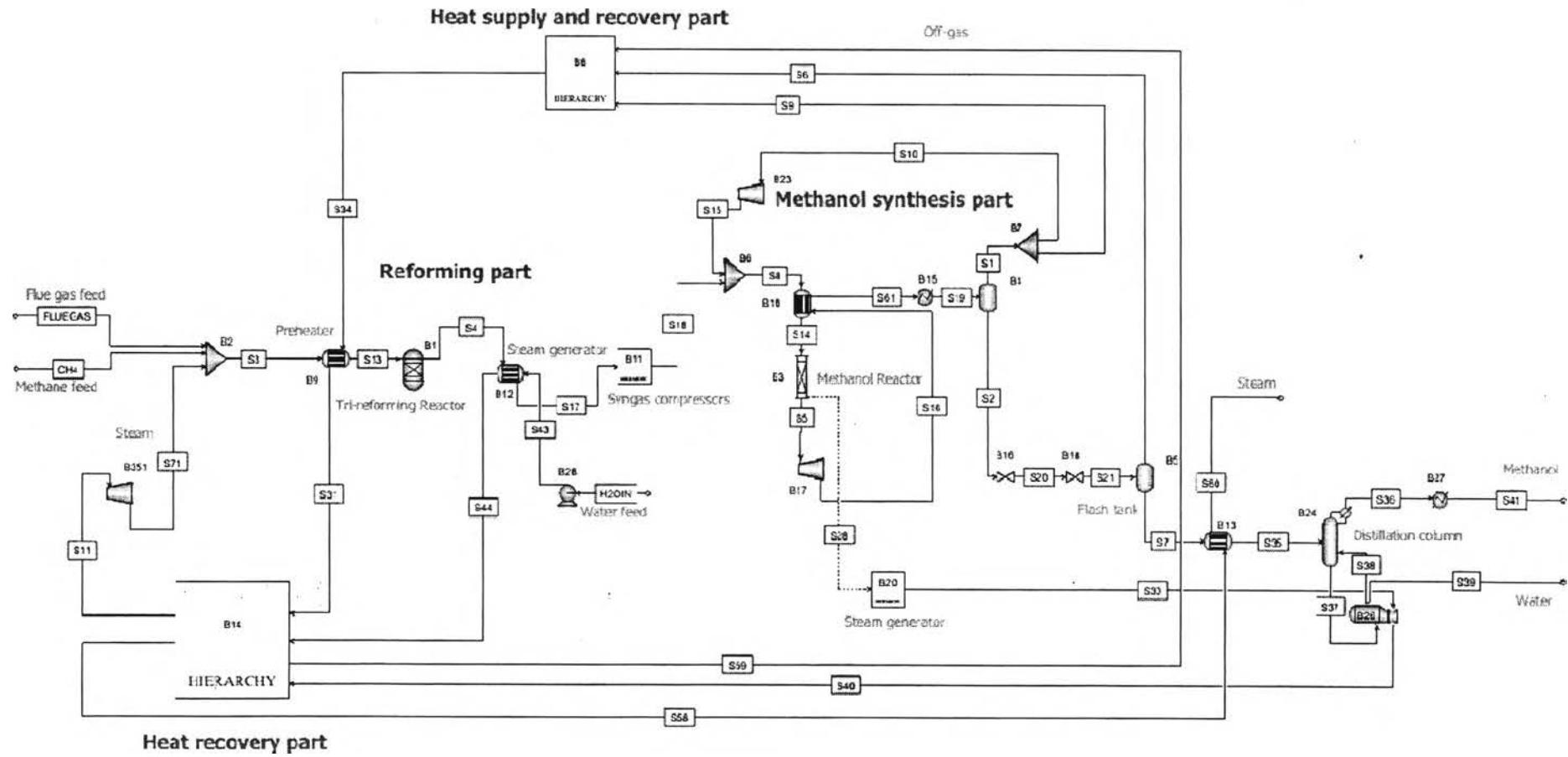
	S10	S11	S13	S14	S15	S16	S17	S18	S19	S20	S21	S30
<b>Temperature, °C</b>	35	391.7	700	220	40.1	233.4	55	172.5	35	34.4	31	253.3
<b>Pressure, bar</b>	62.3	6.51	1	65	65.3	62.3	1	65.3	62.3	10	1.2	42
<b>Vapor Frac</b>	1	1	1	1	1	1	1	1	0.984	0.035	0.047	1
<b>Mole Flow, kmol/hr</b>	39868.5	393.276	1966.24	42850.5	39868.5	41571.8	3054.82	2981.99	41571.8	680.937	680.937	876.928
<b>Mass Flow, kg/hr</b>	912301	7084.986	46072.7	957062	912301	957063	46072.7	44760.6	957063	21367.6	21367.6	15798.1
<b>Volume Flow, cum/hr</b>	16740.8	3309.316	159132	27927.1	16279.4	28961.1	83382.6	1749.04	17206	95.248	706.001	747.671
<b>Enthalpy, Gcal/hr</b>	-115.59	-21.54	-37.182	-77.762	-114.11	-88.478	-24.897	-18.205	-156.51	-37.955	-37.955	-49.735
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	24976.3	0	6184.47	26427.6	24976.3	25785.3	1451.31	1451.3	25785.3	168.462	168.462	0
<b>CH<sub>4</sub></b>	3458.58	0	9131.73	3552.37	3458.58	3552.34	93.794	93.794	3552.34	5.096	5.096	0
<b>CO</b>	48959.9	0	0	67752.4	48959.9	50252.1	18792.5	18792.5	50252.1	36.74	36.74	0
<b>H<sub>2</sub></b>	16444.4	0	0	19480.9	16444.4	16873.6	3036.45	3036.45	16873.6	7.543	7.543	0
<b>WATER</b>	47.153	7084.986	8278.45	176.131	47.153	439.103	1441.07	128.978	439.103	390.741	390.741	15798.1
<b>O<sub>2</sub></b>	0	0	1220.51	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	8378.5	0	0	8378.5	8378.5	28865.6	0	0	28865.6	20272.2	20272.2	0
<b>N<sub>2</sub></b>	810036	0	21257.6	831294	810036	831295	21257.6	21257.6	831295	486.752	486.752	0

**Table A3.1** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the base case design (con't.)

	S31	S34	S35	S36	S37	S38	S39	S40	S41	S43	S44	S58
<b>Temperature, °C</b>	722	1000	80	64.4	75.5	86.1	86.1	253.3	40	26.1	337	232.1
<b>Pressure, bar</b>	1	1	1.1	1	1.1	1.1	1.1	42	1	42	42	1
<b>Vapor Frac</b>	1	1	1	1	0	1	0	0.534	0	0	1	1
<b>Mole Flow, kmol/hr</b>	4593.35	4593.352	648.983	627.299	339.104	317.42	21.684	876.928	627.299	1414.03	1414.03	1897.68
<b>Mass Flow, kg/hr</b>	126783	126782.56	20496.2	20055	8498.69	8057.48	441.211	15798.1	20055	25474.1	25474.1	34187.2
<b>Volume Flow, cum/hr</b>	380164	486343.44	17034.7	17298.3	13.832	8502.17	0.658	408.369	36.801	25.509	1557.77	79393.9
<b>Enthalpy, Gcal/hr</b>	-49.14	-37.98	-31.009	-29.912	-20.931	-16.528	-1.416	-52.723	-35.749	-96.542	-78.651	-106.5
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	14596.2	14596.217	21.382	21.382	0	0	0	0	21.382	0	0	0
<b>CH<sub>4</sub></b>	0	0	0.097	0.097	0	0	0	0	0.097	0	0	0
<b>CO</b>	0.007	0.007	0.249	0.249	0	0	0	0	0.249	0	0	0
<b>H<sub>2</sub></b>	0.001	0.001	0.04	0.04	0	0	0	0	0.04	0	0	0
<b>WATER</b>	13442	13441.96	389.578	63.882	3039.93	2714.24	325.695	15798.1	63.882	25474.1	25474.1	34187.2
<b>O<sub>2</sub></b>	2100.12	2100.124	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	20082.2	19966.7	5458.76	5343.24	115.516	0	19966.7	0	0	0
<b>N<sub>2</sub></b>	96644.3	96644.252	2.634	2.634	0	0	0	0	2.634	0	0	0

**Table A3.1** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the base case design (con't.)

	S59	S60	S61	S71
<b>Temperature, °C</b>	473.4	99.6	61.9	222.1
<b>Pressure, bar</b>	1	1	62.3	1
<b>Vapor Frac</b>	1	0.775	0.998	1
<b>Mole Flow, kmol/hr</b>	4593.35	1897.678	41571.8	393.276
<b>Mass Flow, kg/hr</b>	126782	34187.203	957063	7084.99
<b>Volume Flow, cum/hr</b>	285201	44874.626	18984.8	16122.5
<b>Enthalpy, Gcal/hr</b>	-58.592	-112.799	-143.04	-22.104
<b>Mass Flow, kg/hr</b>				
<b>CO<sub>2</sub></b>	14596.2	0	25785.3	0
<b>CH<sub>4</sub></b>	0	0	3552.34	0
<b>CO</b>	0.007	0	50252.1	0
<b>H<sub>2</sub></b>	0.001	0	16873.6	0
<b>WATER</b>	13442	34187.203	439.103	7084.99
<b>O<sub>2</sub></b>	2100.12	0	0	0
<b>C</b>	0	0	0	0
<b>METHANOL</b>	0	0	28865.6	0
<b>N<sub>2</sub></b>	96644.2	0	831295	0



**Figure A3.2** Flowsheet of the tri-reforming of CO<sub>2</sub> into methanol for the optimized case design.

**Table A3.2** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the optimized case design

	CH <sub>4</sub>	FLUEGAS	H <sub>2</sub> OIN	S1	S2	S3	S4	S5	S6	S7	S8	S9
<b>Temperature, °C</b>	25	150	25	35	35	122.5	840	225.2	31	31	48.6	35
<b>Pressure, bar</b>	1	1	1	56.3	56.3	1	1	.57.25	1.2	1.2	59.3	56.3
<b>Vapor Frac</b>	1	1	0	1	0	1	1	1	1	0	1	1
<b>Mole Flow, kmol/hr</b>	569.212	1003.748	1414.03	40419.3	684.008	1972.88	3059.56	41103.4	29.642	654.366	42390.7	1010.48
<b>Mass Flow, kg/hr</b>	9131.73	29856.014	25474.1	932705	21474.3	46192.4	46192.4	954179	818.569	20655.8	954179	23317.6
<b>Volume Flow, cum/hr</b>	14086.4	35316.302	25.549	18721.5	36.087	64827	283228	30580.5	623.007	34.76	19555.8	468.037
<b>Enthalpy, Gcal/hr</b>	-10.134	-16.105	-96.592	-118.96	-38.261	-48.73	-7.654	-92.48	-0.627	-37.633	-133.08	-2.974
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	0	6184.466	0	28614.5	177.422	6184.47	1578.31	28791.9	153.272	24.149	29477.2	715.361
<b>CH<sub>4</sub></b>	9131.73	0	0	4142.34	5.528	9131.73	109.116	4147.86	5.414	0.115	4147.86	103.558
<b>CO</b>	0	0	0	42492.3	28.696	0	18685	42521	28.485	0.211	60114.7	1062.31
<b>H<sub>2</sub></b>	0	0	0	16128.5	6.633	0	3036.49	16135.1	6.595	0.038	18761.7	403.211
<b>WATER</b>	0	1193.465	25474.1	53.6	408.293	8398.15	1525.99	461.892	1.123	407.169	181.262	1.34
<b>O<sub>2</sub></b>	0	1220.512	0	0	0	1220.51	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	9086.46	20398.2	0	0	29484.6	176.722	20221.4	8859.26	227.162
<b>N<sub>2</sub></b>	0	21257.571	0	832187	449.599	21257.6	21257.6	832637	446.958	2.642	832637	20804.7

**Table A3.2** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	S10	S11	S13	S14	S15	S16	S17	S18	S19	S20	S21	S30
<b>Temperature, °C</b>	35	386.1	700	210	40.6	223.6	55	158.4	35	34.3	31	253.3
<b>Pressure, bar</b>	56.3	6.51	1	59	59.3	56.3	1	59.3	56.3	10	1.2	42
<b>Vapor Frac</b>	1	1	1	1	1	1	1	1	0.983	0.031	0.043	1
<b>Mole Flow, kmol/hr</b>	39408.7	399.921	1972.88	42390.7	39408.7	41103.4	3059.56	2982.01	41103.4	684.008	684.008	886.507
<b>Mass Flow, kg/hr</b>	909383	7204.685	46192.4	954179	909383	954179	46192.4	44795.4	954179	21474.3	21474.3	15970.7
<b>Volume Flow, cum/hr</b>	18253.4	3335.981	159669	29734.1	17694.2	30982.6	83511.2	1860.25	18757.6	90.016	657.767	755.838
<b>Enthalpy, Gcal/hr</b>	-115.99	-21.924	-37.527	-82.217	-114.38	-92.983	-25.354	-18.697	-157.22	-38.261	-38.261	-50.278
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	27898.9	0	6184.47	29477.2	27898.9	28791.9	1578.31	1578.3	28791.9	177.422	177.422	0
<b>CH<sub>4</sub></b>	4038.75	0	9131.73	4147.86	4038.75	4147.86	109.116	109.116	4147.86	5.528	5.528	0
<b>CO</b>	41429.8	0	0	60114.7	41429.8	42521	18685	18685	42521	28.696	28.696	0
<b>H<sub>2</sub></b>	15725.2	0	0	18761.7	15725.2	16135.1	3036.49	3036.49	16135.1	6.633	6.633	0
<b>WATER</b>	52.259	7204.685	8398.15	181.262	52.259	461.892	1525.99	129.003	461.892	408.293	408.293	15970.7
<b>O<sub>2</sub></b>	0	0	1220.51	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	8859.26	0	0	8859.26	8859.26	29484.6	0	0	29484.6	20398.2	20398.2	0
<b>N<sub>2</sub></b>	811379	0	21257.6	832637	811379	832637	21257.6	21257.6	832637	449.599	449.599	0

**Table A3.2** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	S31	S34	S35	S36	S37	S38	S39	S40	S41	S43	S44	S58
<b>Temperature, °C</b>	719.8	1000	80	64.4	74.7	84.5	84.5	253.3	40	26.1	325.1	227.6
<b>Pressure, bar</b>	1	1	1.1	1	1.1	1.1	1.1	42	1	42	42	1
<b>Vapor Frac</b>	1	1	1	1	0	1	0	0.539	0	0	1	1
<b>Mole Flow, kmol/hr</b>	4577.44	4577.437	654.366	630.767	342.842	319.244	23.598	886.507	630.767	1414.03	1414.03	1900.61
<b>Mass Flow, kg/hr</b>	126334	126334.25	20655.8	20164.7	8780.37	8289.31	491.066	15970.7	20164.7	25474.1	25474.1	34240.1
<b>Volume Flow, cum/hr</b>	378013	484658.32	17176.1	17394.3	14.408	8509.11	0.74	416.536	37.117	25.509	1513.96	78810.1
<b>Enthalpy, Gcal/hr</b>	-48.891	-37.687	-31.277	-30.081	-21.017	-16.496	-1.533	-53.266	-35.951	-96.542	-78.842	-106.73
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	14493.6	14493.588	24.149	24.149	0	0	0	0	24.149	0	0	0
<b>CH<sub>4</sub></b>	0	0	0.115	0.115	0	0	0	0	0.115	0	0	0
<b>CO</b>	0.007	0.007	0.211	0.211	0	0	0	0	0.211	0	0	0
<b>H<sub>2</sub></b>	0.001	0.001	0.038	0.038	0	0	0	0	0.038	0	0	0
<b>WATER</b>	13377.5	13377.492	407.169	66.71	2832.01	2491.56	340.453	15970.7	66.71	25474.1	25474.1	34240.1
<b>O<sub>2</sub></b>	2092.58	2092.575	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	20221.4	20070.8	5948.36	5797.75	150.613	0	20070.8	0	0	0
<b>N<sub>2</sub></b>	96370.6	96370.589	2.642	2.642	0	0	0	0	2.642	0	0	0

**Table A3.2** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the optimized case design (con't.)

	S59	S60	S61	S71
<b>Temperature, °C</b>	468.3	99.6	61.4	217.6
<b>Pressure, bar</b>	1	1	56.3	1
<b>Vapor Frac</b>	1	0.768	0.999	1
<b>Mole Flow, kmol/hr</b>	4577.43	1900.612	41103.4	399.921
<b>Mass Flow, kg/hr</b>	126334	34240.063	954179	7204.69
<b>Volume Flow, cum/hr</b>	282291	44586.836	20676.6	16246
<b>Enthalpy, Gcal/hr</b>	-58.408	-113.087	-143.85	-22.492
<b>Mass Flow, kg/hr</b>				
CO <sub>2</sub>	14493.6	0	28791.9	0
CH <sub>4</sub>	0	0	4147.86	0
CO	0.007	0	42521	0
H <sub>2</sub>	0.001	0	16135.1	0
<b>WATER</b>	13377.5	34240.063	461.892	7204.69
O <sub>2</sub>	2092.57	0	0	0
C	0	0	0	0
<b>METHANOL</b>	0	0	29484.6	0
N <sub>2</sub>	96370.4	0	832637	0

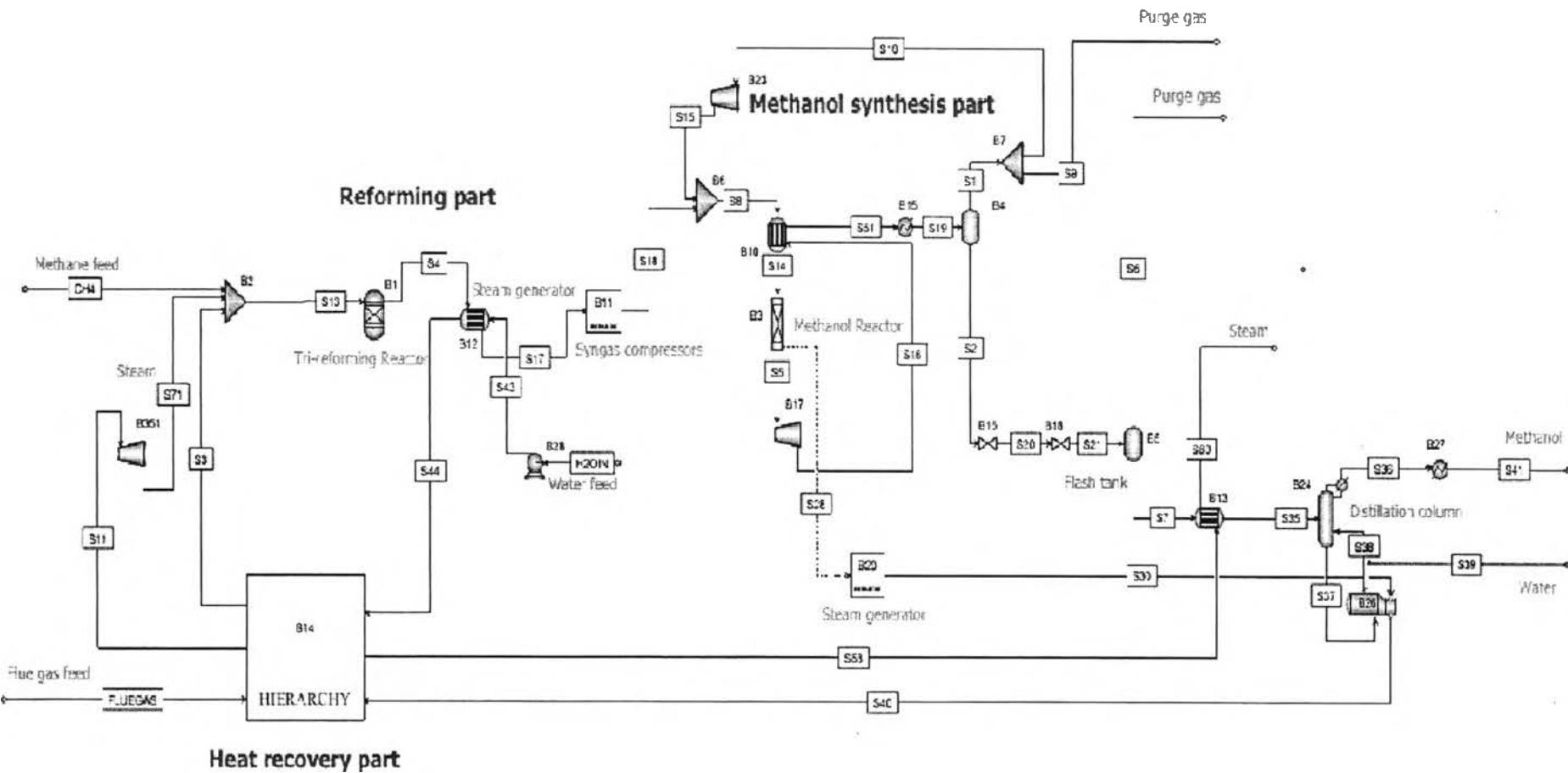


Figure A3.3 Flowsheet of the tri-reforming of  $\text{CO}_2$  into methanol for the alternative case design.

**Table A3.3** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the alternative case design

	CH <sub>4</sub>	FLUEGAS	H <sub>2</sub> OIN	S1	S2	S3	S4	S5	S6	S7	S8	S9
<b>Temperature, °C</b>	25	1000	25	35	35	207.1	840	225.2	31	31	48.6	35
<b>Pressure, bar</b>	1	1	1	56.3	56.3	1	1	57.25	1.2	1.2	59.3	56.3
<b>Vapor Frac</b>	1	1	0	1	0	1	1	1	1	0	1	1
<b>Mole Flow, kmol/hr</b>	569.212	1003.748	1414.03	40423.2	684.021	1003.75	3059.65	41107.1	29.643	654.379	42394.7	1010.58
<b>Mass Flow, kg/hr</b>	9131.73	29856.014	25474.1	932793	21474.6	29856	46193.9	954265	818.614	20656	954269	23319.8
<b>Volume Flow, cum/hr</b>	14086.4	106277.36	25.549	18723.2	36.087	40086	283236	30583.5	623.026	34.76	19557.5	468.081
<b>Enthalpy, Gcal/hr</b>	-10.134	-8.967	-96.592	-118.99	-38.262	-15.67	-7.658	-92.5	-0.627	-37.634	-133.11	-2.975
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	0	6184.466	0	28627.1	177.484	6184.47	1579.26	28804.6	153.327	24.157	29490.7	715.677
<b>CH<sub>4</sub></b>	9131.73	0	0	4140.64	5.525	0	109.043	4146.16	5.411	0.114	4146.17	103.516
<b>CO</b>	0	0	0	42490.2	28.692	0	18684.5	42519.1	28.481	0.211	60112.4	1062.26
<b>H<sub>2</sub></b>	0	0	0	16130.5	6.634	0	3036.56	16137	6.596	0.038	18763.8	403.261
<b>WATER</b>	0	1193.465	25474.1	53.632	408.512	1193.47	1526.99	462.143	1.124	407.388	181.296	1.341
<b>O<sub>2</sub></b>	0	1220.512	0	0	0	1220.51	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	0	0	0	9087.16	20398.2	0	0	29485.3	176.719	20221.4	8859.98	227.179
<b>N<sub>2</sub></b>	0	21257.571	0	832264	449.599	21257.6	21257.6	832711	446.957	2.642	832715	20806.6

**Table A3.3** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the alternative case design (con't.)

	<b>S10</b>	<b>S11</b>	<b>S13</b>	<b>S14</b>	<b>S15</b>	<b>S16</b>	<b>S17</b>	<b>S18</b>	<b>S19</b>	<b>S20</b>	<b>S21</b>	<b>S30</b>
<b>Temperature, °C</b>	35	266.9	129.4	210	40.6	223.6	55	158.4	35	34.3	31	253.3
<b>Pressure, bar</b>	56.3	6.51	1	59	59.3	56.3	1	59.3	56.3	10	1.2	42
<b>Vapor Frac</b>	1	1	1	1	1	1	1	1	0.983	0.031	0.043	1
<b>Mole Flow, kmol/hr</b>	39412.6	400.003	1972.96	42394.7	39412.6	41107.1	3059.65	2982.05	41107.1	684.021	684.021	886.444
<b>Mass Flow, kg/hr</b>	909473	7206.158	46193.9	954269	909473	954265	46193.9	44795.9	954265	21474.6	21474.6	15969.5
<b>Volume Flow, cum/hr</b>	18255.2	2702.558	65969.2	29736.9	17695.9	30985.4	83513.7	1860.27	18759.3	90.017	657.786	755.784
<b>Enthalpy, Gcal/hr</b>	-116.01	-22.358	-48.622	-82.238	-114.41	-93.003	-25.359	-18.699	-157.25	-38.262	-38.262	-50.275
<b>Mass Flow, kg/hr</b>												
<b>CO<sub>2</sub></b>	27911.4	0	6184.47	29490.7	27911.4	28804.6	1579.26	1579.26	28804.6	177.484	177.484	0
<b>CH<sub>4</sub></b>	4037.13	0	9131.73	4146.17	4037.13	4146.16	109.043	109.043	4146.16	5.525	5.525	0
<b>CO</b>	41428	0	0	60112.4	41428	42519.1	18684.5	18684.5	42519.1	28.692	28.692	0
<b>H<sub>2</sub></b>	15727.2	0	0	18763.8	15727.2	16137	3036.56	3036.56	16137	6.634	6.634	0
<b>WATER</b>	52.291	7206.158	8399.63	181.296	52.291	462.143	1526.99	129.004	462.143	408.512	408.512	15969.5
<b>O<sub>2</sub></b>	0	0	1220.51	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	8859.98	0	0	8859.98	8859.98	29485.3	0	0	29485.3	20398.2	20398.2	0
<b>N<sub>2</sub></b>	811457	0	21257.6	832715	811457	832711	21257.6	21257.6	832711	449.599	449.599	0

**Table A3.3** Stream table of tri-reforming of CO<sub>2</sub> into methanol for the alternative case design (con't.)

	S35	S36	S37	S38	S39	S40	S41	S43	S44	S58	S60	S61	S71
<b>Temperature, °C</b>	80	64.4	74.7	84.5	84.5	253.3	40	26.1	325.1	133.2	99.6	61.4	123.2
<b>Pressure, bar</b>	1.1	1	1.1	1.1	1.1	42	1	42	42	1	1	56.3	1
<b>Vapor Frac</b>	1	1	0	1	0	0.539	0	0	1	1	0.686	0.999	1
<b>Mole Flow, kmol/hr</b>	654.379	630.776	342.833	319.229	23.603	886.444	630.776	1414.03	1414.03	1900.47	1900.47	41107.1	400.003
<b>Mass Flow, kg/hr</b>	20656	20164.92	8778.53	8287.46	491.069	15969.5	20164.9	25474.1	25474.1	34237.5	34237.5	954265	7206.16
<b>Volume Flow, cum/hr</b>	17176.4	17394.565	14.404	8509.07	0.74	416.482	37.118	25.509	1514.11	63558.1	39784.6	20678.4	13032.1
<b>Enthalpy, Gcal/hr</b>	-31.278	-30.082	-21.017	-16.497	-1.533	-53.262	-35.951	-96.542	-78.841	-108.25	-114.61	-143.87	-22.818
<b>Mass Flow, kg/hr</b>													
<b>CO<sub>2</sub></b>	24.157	24.157	0	0	0	0	24.157	0	0	0	0	28804.6	0
<b>CH<sub>4</sub></b>	0.114	0.114	0	0	0	0	0.114	0	0	0	0	4146.16	0
<b>CO</b>	0.211	0.211	0	0	0	0	0.211	0	0	0	0	42519.1	0
<b>H<sub>2</sub></b>	0.038	0.038	0	0	0	0	0.038	0	0	0	0	16137	0
<b>WATER</b>	407.388	66.748	2833.99	2493.35	340.643	15969.5	66.748	25474.1	25474.1	34237.5	34237.5	462.143	7206.16
<b>O<sub>2</sub></b>	0	0	0	0	0	0	0	0	0	0	0	0	0
<b>C</b>	0	0	0	0	0	0	0	0	0	0	0	0	0
<b>METHANOL</b>	20221.4	20071.01	5944.54	5794.11	150.426	0	20071	0	0	0	0	29485.3	0
<b>N<sub>2</sub></b>	2.642	2.642	0	0	0	0	2.642	0	0	0	0	832711	0

## Appendix B Economic Evaluation For Each Process

### B.1 Hydrogenation of CO<sub>2</sub> into Methanol

#### B.1.1 Hydrogenation of CO<sub>2</sub> into Methanol for the Base Case Design

##### B.1.1.1 Raw Material, Product and Utilities Prices

**Table B1.1** Raw material and product prices

Raw Material Price		
Raw Material	Value	Unit
Hydrogen method <sup>[1]</sup>		
<i>Methane steam reforming</i>	2.885	\$/kgH <sub>2</sub>
<i>Wind/electric</i>	4.14	\$/kgH <sub>2</sub>
<i>Nuclear/steam electrolysis</i>	6.42	\$/kgH <sub>2</sub>
<i>Solar thermal</i>	3.07	\$/kgH <sub>2</sub>
<i>Biomass</i>	1.59	\$/kgH <sub>2</sub>
<i>Hydroelectric</i>	1.28	\$/kgH <sub>2</sub>
CO <sub>2</sub> <sup>[2]</sup>	35.3	\$/tCO <sub>2</sub>
Product Price		
Product	Value	Unit
Methanol <sup>[3]</sup>	0.48	\$/kgMeOH

References:

- [1] Yumurtaci *et al.* (2004); Bartels *et al.* (2010)
- [2] Amann (2007)
- [3] [www.icis.com](http://www.icis.com) (12/2014)

**Table B1.2 Utility price**

<b>Cooling Water</b>		
	Value	Unit
Cooling water 28-40 °C*	0.35	\$/GJ
<b>Electricity</b>		
	Value	Unit
Electricity*	18.42	\$/GJ

\* 2013 U.S. average cost

#### B.1.1.2 Raw Materials and Product Annual Prices

**Table B1.3 Raw materials annual price**

<b>Raw Material</b>	<b>Quantity (kg/year)</b>	<b>Annual Price (\$/year)</b>
Hydrogen method		
<i>Methane steam reforming</i>	32141742.00	92,729,812.72
CO <sub>2</sub>	233758128.00	8,251,661.92

**Table B1.4 Products annual price**

<b>Product</b>	<b>Quantity (kg/day)</b>	<b>Quantity (kg/year)</b>	<b>Annual Price (\$/year)</b>
Methanol	484077.77	161197904.00	77,374,993.92

### B.1.1.3 Annual Utility Cost

**Table B1.5** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-29-10	451.09	239,018.55
C-29-19	756.26	400,719.49
C-29-21	692.60	366,986.74
C-29-23	651.19	345,044.72
C-29-24	591.81	313,583.27
C-33	631.29	334,502.06
C-9	2091.52	1,108,232.58
<b>Total</b>		<b>3,108,087.41</b>

**Table B1.6** Annual cooling water cost

Equipment	Quantity (kW)	Annual Price (\$/year)
E-29-18	320.69	3,266.00
E-29-20	701.80	7,148.00
E-29-22	678.26	6,908.00
E-29-25	739.04	7,527.00
E-11	10152.48	103,403.00
E-31	6651.53	67,746.00
E-19	162.30	1,653.00
<b>Total</b>		<b>197,651.00</b>

#### B.1.1.4 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B1.7 Equipment sizing and purchase cost**

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase Cost (\$)</b>
V-5	Vessel	5.58	m	316 Stainless Steel	134,340.82
V-7	Vessel	5.49	m	316 Stainless Steel	46,350.02
V-13	Vessel	1.95	m	316 Stainless Steel	31,011.25
E-29-18	Heat Exchanger	57.16	m <sup>2</sup>	SS shell&SS tube	15,398.62
E-29-20	Heat Exchanger	76.24	m <sup>2</sup>	SS shell&SS tube	18,875.53
E-29-22	Heat Exchanger	76.93	m <sup>2</sup>	SS shell&SS tube	20,331.00
E-29-25	Heat Exchanger	84.30	m <sup>2</sup>	SS shell&SS tube	21,766.19
E-15	Heat Exchanger	3,369.44	m <sup>2</sup>	CS shell&SS tube	412,116.00
E-11	Heat Exchanger	1,677.22	m <sup>2</sup>	SS shell&SS tube	352,563.00
E-6	Heat Exchanger	34.69	m <sup>2</sup>	SS shell&SS tube	12,436.96
E-31	Heat Exchanger	253.07	m <sup>2</sup>	SS shell&SS tube	51,090.71
E-19	Heat Exchanger	13.01	m <sup>2</sup>	SS shell&SS tube	7,355.47
R-1	Reactor	5,095.63	m <sup>2</sup>	SS shell&SS tube	3,355,746.00
T-4	Tower Unit	20.62	m	316 Stainless Steel	753,791.88
tT-4	Valvetray	25.00	trays	Stainless Steel	136,648.28
cT-4	Heat Exchanger	464.06	m <sup>2</sup>	SS shell&SS tube	171,051.24
E-12	Heat Exchanger	40.96	m <sup>2</sup>	SS shell&SS tube	13,692.62
C-29-10	Compressor	451.09	kW	Carbon Steel	286,282.24
C-29-19	Compressor	756.26	kW	Carbon Steel	405,509.46
C-29-21	Compressor	692.60	kW	Carbon Steel	382,181.29
C-29-23	Compressor	651.19	kW	Carbon Steel	366,630.53
C-29-24	Compressor	591.81	kW	Carbon Steel	343,755.74
C-33	Compressor	631.29	kW	Carbon Steel	359,043.81
C-9	Compressor	2,091.52	kW	Carbon Steel	804,785.22
<b>Total</b>					<b>8,502,753.91</b>

### B.1.1.5 Capital Cost Analysis

**Table B1.8** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	9,353,029.30
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	4,395,923.77
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	3,367,090.55
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	6,360,059.92
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	1,028,833.22
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	1,683,545.27
Yard Improvement (% of Purchased Equipment Delivered)	10.00	935,302.93
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	6,547,120.51
<b>Total Direct Cost</b>		33,670,905.48
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	3,086,499.67
Construction Expenses	41.00	3,834,742.01
Legal Expenses	4.00	374,121.17
Contractor's Fees	22.00	2,057,666.45
Contingency	44.00	4,115,332.89
<b>Total Indirect Cost</b>		13,468,362.19
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		47,139,267.68
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	8,324,196.08
<b>V. Total Capital Investment (TCI) = Fixed-Capital Investment + Working Capital</b>		55,463,463.75

### B.1.1.6 Production Cost Analysis

**Table B1.9 Breakdown of production cost**

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	100,981,474.64
Operating Labor	5.00	Fixed Capital Investment	2,356,963.50
Operating Supervision	15.00	Operating Labor	353,544.53
Utilities	-	-	3,305,738.41
Maintenance and Repairs	6.00	Fixed Capital Investment	2,828,356.06
Operating Supplies	15.00	Maintenance and Supplies	424,253.40
Laboratory Charges	15.00	Operating Labor	353,544.53
Royalties	1.00	Total Product Cost	1,282,891.20
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			112,180,720.52
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	942,785.35
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	471,392.68
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,414,178.03
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,323,318.40
<b>Total Manufacturing Cost</b>			113,594,898.55
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,107,772.80
Distribution & selling	4.00	Total Product Cost	5,131,564.80
Research & Development	4.00	Total Product Cost	5,131,564.80
<b>General Expense</b>			11,370,902.40
<b>V. Total Product Cost with Out Depreciation</b>			128,289,120.00

### B.1.1.7 Profitability Assumptions

**Table B1.10** Profitability conditions

<b>Project Details</b>	
Project Life Time	20 years
Land Cost	\$500,000.00
Construction Inflation Rate	2%
Product Inflation Rate	10%
TPC Inflation Rate	10%
Minimum Rate of Return, Mar	15%
Income Tax Rate	25%
Type of Depreciation	MACRS

B.1.2 Hydrogenation of CO<sub>2</sub> into Methanol for the Optimized and Alternative Case Designs

B.1.2.1 Raw Materials and Product Annual Prices

**Table B1.11** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	485,698.92	161,737,744.00	77,634,117.12

**Table B1.12** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Hydrogen method		
<i>Methane steam reforming</i>	32141742.00	92,729,812.72
<i>Wind/electric</i>	32141742.00	132,915,666.98
<i>Nuclear/steam electrolysis</i>	32141742.00	206,418,500.49
<i>Solar thermal</i>	32141742.00	98,630,000.23
<i>Biomass</i>	32141742.00	51,252,375.12
<i>Hydroelectric</i>	32141742.00	41,168,729.12
CO <sub>2</sub>	233758128.00	8,251,661.92

### B.1.2.2 Annual Utility Cost

**Table B1.13** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-29-10	451.09	239,018.55
C-29-19	756.26	400,719.49
C-29-21	692.60	366,986.74
C-29-23	651.19	345,044.72
C-29-24	371.74	196,972.98
C-33	1,307.49	692,799.75
C-9	924.19	489,700.49
<b>Total</b>		<b>2,731,242.72</b>

**Table B1.14** Annual cooling water cost

Equipment	Quantity (kW)	Annual Price (\$/year)
E-29-18	320.69	3,266.00
E-29-20	701.80	7,148.00
E-29-22	678.26	6,908.00
E-29-25	739.04	7,527.00
E-11	9,528.82	97,051.00
E-31	6,670.78	67,942.00
E-19	162.80	1,658.00
<b>Total</b>		<b>191,500.00</b>

### B.1.2.3 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B1.15** Equipment sizing and purchase cost

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase Cost (\$)</b>
V-5	Vessel	5.57	m	316 Stainless Steel	134,255.63
V-7	Vessel	5.50	m	316 Stainless Steel	46,378.42
V-13	Vessel	1.95	m	316 Stainless Steel	20,583.56
E-29-18	Heat Exchanger	57.16	m <sup>2</sup>	SS shell&SS tube	15,398.62
E-29-20	Heat Exchanger	76.24	m <sup>2</sup>	SS shell&SS tube	18,875.53
E-29-22	Heat Exchanger	76.93	m <sup>2</sup>	SS shell&SS tube	20,331.00
E-29-25	Heat Exchanger	84.30	m <sup>2</sup>	SS shell&SS tube	21,766.19
E-15	Heat Exchanger	3,510.37	m <sup>2</sup>	CS shell&SS tube	427,488.00
E-11	Heat Exchanger	1,740.51	m <sup>2</sup>	SS shell&SS tube	365,247.00
E-6	Heat Exchanger	41.27	m <sup>2</sup>	SS shell&SS tube	13,755.50
E-31	Heat Exchanger	253.97	m <sup>2</sup>	SS shell&SS tube	51,253.00
C-29-10	Compressor	451.09	kW	Carbon Steel	286,282.24
C-29-19	Compressor	756.26	kW	Carbon Steel	405,509.46
C-29-21	Compressor	692.60	kW	Carbon Steel	382,181.29
C-29-23	Compressor	651.19	kW	Carbon Steel	366,630.53
C-29-24	Compressor	371.74	kW	Carbon Steel	251,292.02
C-9	Compressor	1,307.49	kW	Carbon Steel	586,420.40
C-33	Compressor	924.19	kW	Carbon Steel	464,175.74
R-1	Reactor	5,095.63	m <sup>2</sup>	SS shell&SS tube	3,355,746.00
T-4	Tower Unit	20.62	m	316 Stainless Steel	753,791.88
tT-4	Valvetray	25.00	trays	Stainless Steel	137,174.69
cT-4	Heat Exchanger	466.15	m <sup>2</sup>	SS shell&SS tube	171,375.81
E-12	Heat Exchanger	48.72	m <sup>2</sup>	SS shell&SS tube	15,246.48
E-19	Heat Exchanger	13.05	m <sup>2</sup>	SS shell&SS tube	7,362.57
<b>Total</b>					<b>8,318,521.56</b>

#### B.1.2.4 Capital Cost Analysis

**Table B1.16** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	9,150,373.72
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	4,300,675.65
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	3,294,134.54
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	6,222,254.13
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	1,006,541.11
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	1,647,067.27
Yard Improvement (% of Purchased Equipment Delivered)	10.00	915,037.37
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	6,405,261.60
<b>Total Direct Cost</b>		32,941,345.38
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	3,019,623.33
Construction Expenses	41.00	3,751,653.22
Legal Expenses	4.00	366,014.95
Contractor's Fees	22.00	2,013,082.22
Contingency	44.00	4,026,164.44
<b>Total Indirect Cost</b>		13,176,538.15
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		46,117,883.53
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	8,143,832.61
<b>V. Total Capital Investment (TCI) = Fixed-Capital Investment + Working Capital</b>		54,261,716.14

### B.1.2.5 Production Cost Analysis

**Table B1.17** Breakdown of production cost for the methane steam reforming case

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	100,981,474.64
Operating Labor	5.00	Fixed Capital Investment	2,305,894.25
Operating Supervision	15.00	Operating Labor	345,884.14
Utilities	-	-	2,922,742.72
Maintenance and Repairs	6.00	Fixed Capital Investment	2,767,073.01
Operating Supplies	15.00	Maintenance and Supplies	415,060.95
Laboratory Charges	15.00	Operating Labor	345,884.14
Royalties	1.00	Total Product Cost	1,275,786.64
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			111,653,754.75
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	922,357.67
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	461,178.84
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,383,536.51
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,251,310.90
<b>Total Manufacturing Cost</b>			113,037,291.25
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,083,770.30
Distribution & selling	4.00	Total Product Cost	5,103,146.56
Research & Development	4.00	Total Product Cost	5,103,146.56
<b>General Expense</b>			11,290,063.42
<b>V. Total Product Cost with Out Depreciation</b>			127,578,664.00

### B.1.2.6 Production Cost Analysis

**Table B1.18** Breakdown of production cost for the wind/electric case

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	141,167,328.90
Operating Labor	5.00	Fixed Capital Investment	2,305,894.25
Operating Supervision	15.00	Operating Labor	345,884.14
Utilities	-	-	2,922,742.72
Maintenance and Repairs	6.00	Fixed Capital Investment	2,767,073.01
Operating Supplies	15.00	Maintenance and Supplies	415,060.95
Laboratory Charges	15.00	Operating Labor	345,884.14
Royalties	1.00	Total Product Cost	1,717,389.44
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			152,281,211.81
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	922,357.67
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	461,178.84
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,383,536.51
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,251,310.90
<b>Total Manufacturing Cost</b>			153,664,748.31
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,083,770.30
Distribution & selling	4.00	Total Product Cost	6,869,557.76
Research & Development	4.00	Total Product Cost	6,869,557.76
<b>General Expense</b>			14,822,885.82
<b>V. Total Product Cost with Out Depreciation</b>			171,738,944.00

**Table B1.19** Breakdown of production cost for the nuclear/steam electrolysis

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	214,670,162.40
Operating Labor	5.00	Fixed Capital Investment	2,305,894.25
Operating Supervision	15.00	Operating Labor	345,884.14
Utilities	-	-	2,922,742.72
Maintenance and Repairs	6.00	Fixed Capital Investment	2,767,073.01
Operating Supplies	15.00	Maintenance and Supplies	415,060.95
Laboratory Charges	15.00	Operating Labor	345,884.14
Royalties	1.00	Total Product Cost	2,525,112.80
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			226,591,768.67
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	922,357.67
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	461,178.84
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,383,536.51
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,251,310.90
<b>Total Manufacturing Cost</b>			227,975,305.18
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,083,770.30
Distribution & selling	4.00	Total Product Cost	10,100,451.20
Research & Development	4.00	Total Product Cost	10,100,451.20
<b>General Expense</b>			21,284,672.70
<b>V. Total Product Cost with Out Depreciation</b>			252,511,280.00

**Table B1.20** Breakdown of production cost for the solar thermal case

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	106,881,662.15
Operating Labor	5.00	Fixed Capital Investment	2,305,894.25
Operating Supervision	15.00	Operating Labor	345,884.14
Utilities	-	-	2,922,742.72
Maintenance and Repairs	6.00	Fixed Capital Investment	2,767,073.01
Operating Supplies	15.00	Maintenance and Supplies	415,060.95
Laboratory Charges	15.00	Operating Labor	345,884.14
Royalties	1.00	Total Product Cost	1,340,623.92
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			117,618,779.54
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	922,357.67
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	461,178.84
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,383,536.51
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,251,310.90
<b>Total Manufacturing Cost</b>			119,002,316.05
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,083,770.30
Distribution & selling	4.00	Total Product Cost	5,362,495.68
Research & Development	4.00	Total Product Cost	5,362,495.68
<b>General Expense</b>			11,808,761.66
<b>V. Total Product Cost with Out Depreciation</b>			134,062,392.00

**Table B1.21** Breakdown of production cost for the biomass case

<b>Items of Production Cost</b>	<b>% of Basis</b>	<b>Basis</b>	<b>Cost, \$/year</b>
<b>I. Variable Cost</b>			
Raw Material	-	-	59,504,037.04
Operating Labor	5.00	Fixed Capital Investment	2,305,894.25
Operating Supervision	15.00	Operating Labor	345,884.14
Utilities	-	-	2,922,742.72
Maintenance and Repairs	6.00	Fixed Capital Investment	2,767,073.01
Operating Supplies	15.00	Maintenance and Supplies	415,060.95
Laboratory Charges	15.00	Operating Labor	345,884.14
Royalties	1.00	Total Product Cost	819,990.56
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			69,720,521.07
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	922,357.67
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	461,178.84
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,383,536.51
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,251,310.90
<b>Total Manufacturing Cost</b>			71,104,057.58
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,083,770.30
Distribution & selling	4.00	Total Product Cost	3,279,962.24
Research & Development	4.00	Total Product Cost	3,279,962.24
<b>General Expense</b>			7,643,694.78
<b>V. Total Product Cost with Out Depreciation</b>			81,999,056.00

**Table B1.22** Breakdown of production cost for the hydroelectric case

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	49,420,391.03
Operating Labor	5.00	Fixed Capital Investment	2,305,894.25
Operating Supervision	15.00	Operating Labor	345,884.14
Utilities	-	-	2,922,742.72
Maintenance and Repairs	6.00	Fixed Capital Investment	2,767,073.01
Operating Supplies	15.00	Maintenance and Supplies	415,060.95
Laboratory Charges	15.00	Operating Labor	345,884.14
Royalties	1.00	Total Product Cost	709,181.36
Catalysts	-	-	293,954.26
<b>Total Variable Cost</b>			59,526,065.86
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	922,357.67
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	461,178.84
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,383,536.51
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	3,251,310.90
<b>Total Manufacturing Cost</b>			60,909,602.37
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	1,083,770.30
Distribution & selling	4.00	Total Product Cost	2,836,725.44
Research & Development	4.00	Total Product Cost	2,836,725.44
<b>General Expense</b>			6,757,221.18
<b>V. Total Product Cost with Out Depreciation</b>			70,918,136.00

## B.2 Bi-reforming of CO<sub>2</sub> into Methanol

### B.2.1 Bi-reforming of CO<sub>2</sub> into Methanol for the Base Case Design

#### B.2.1.1 Raw Material, Product and Utilities Prices

**Table B2.1** Raw material and product prices

<b>Raw Material Price</b>		
<b>Raw Material</b>	<b>Value</b>	<b>Unit</b>
Methane <sup>[1]</sup>	198	\$/tCH <sub>4</sub>
CO <sub>2</sub> <sup>[2]</sup>	35.3	\$/tCO <sub>2</sub>
Demineralized water <sup>[3]</sup>	1.62	\$/tH <sub>2</sub> O
<b>Product Price</b>		
<b>Product</b>	<b>Value</b>	<b>Unit</b>
Methanol <sup>[4]</sup>	0.48	\$/kgMeOH

References:

- [1] [http://www.eia.gov/dnav/ng/ng\\_pri\\_sum\\_dcu\\_nus\\_m.htm](http://www.eia.gov/dnav/ng/ng_pri_sum_dcu_nus_m.htm) (12/2014)
- [2] Amann (2007)
- [3] [www.alibaba.com](http://www.alibaba.com) (12/2014)
- [4] [www.icis.com](http://www.icis.com) (12/2014)

**Table B2.2** Utility price

<b>Cooling Water</b>		
	<b>Value</b>	<b>Unit</b>
Cooling water 28-40 °C*	0.35	\$/GJ
<b>Electricity</b>		
	<b>Value</b>	<b>Unit</b>
Electricity*	18.42	\$/GJ

\* 2013 U.S. average cost

### B.2.1.2 Raw Materials and Product Annual Price

**Table B2.3** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Methane	104,883,620.00	20,766,956.65
CO <sub>2</sub>	53,716,348.00	1,896,187.08
Demineralized water	429,054,784.00	695,068.75

**Table B2.4** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	487,588.01	162,366,800.00	77,936,064.00

### B.2.1.3 Annual Utility Cost

**Table B2.5** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-30-10	189.79	100,562.54
C-30-19	181.44	96,138.50
C-30-23	177.02	93,799.89
C-5	2,710.86	1,436,401.68
C-23	468.16	248,062.45
PD-20-1	44.81	23,745.62
PD-28	91.46	48,460.45
<b>Total</b>		2,047,171.13

**Table B2.6** Annual generated electricity cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
TB-18-33	-935.64	-495,766.00
TB-18-35	-969.78	-513,859.00
TB-18-38	-1,879.65	-995,969.00
TB-18-351	-535.28	-283,628.00
<b>Total</b>		<b>-2,289,222.00</b>

**Table B2.7** Annual cooling water

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
E-30-1	193.42	1,970.00
E-30-18	172.52	1,757.00
E-201	10,117.96	103,051.00
E-15	13,768.33	140,230.00
E-9	396.03	4,034.00
<b>Total</b>		<b>251,042.00</b>

### B.2.1.4 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B2.8 Equipment sizing and purchase cost**

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase Cost (\$)</b>
V-4	Vessel	3.83	m	316 Stainless Steel	139,152.51
V-41	Vessel	5.16	m	316 Stainless Steel	129,052.43
V-20-2	Vessel	1.48	m	316 Stainless Steel	55,402.37
E-30-1	Heat Exchanger	18.56	m <sup>2</sup>	SS shell&SS tube	8,367.71
E-30-18	Heat Exchanger	18.12	m <sup>2</sup>	SS shell&SS tube	8,286.57
E-27	Heat Exchanger	203.27	m <sup>2</sup>	SS shell&SS tube	44,958.45
E-3	Heat Exchanger	359.02	m <sup>2</sup>	SS shell&SS tube	81,652.64
E-201	Heat Exchanger	684.81	m <sup>2</sup>	SS shell&SS tube	144,160.00
E-6	Heat Exchanger	3,304.96	m <sup>2</sup>	CS shell&SS tube	405,090.00
E-15	Heat Exchanger	2,193.89	m <sup>2</sup>	SS shell&SS tube	461,588.00
E-18-1	Heat Exchanger	41.42	m <sup>2</sup>	SS shell&SS tube	13,407.61
E-18-2	Heat Exchanger	170.72	m <sup>2</sup>	SS shell&SS tube	38,612.17
E-18-3	Heat Exchanger	19.93	m <sup>2</sup>	SS shell&SS tube	9,219.70
E-18-4	Heat Exchanger	58.78	m <sup>2</sup>	SS shell&SS tube	16,792.22
E-18-5	Heat Exchanger	78.74	m <sup>2</sup>	SS shell&SS tube	19,329.92
C-30-10	Compressor	189.79	kW	Carbon Steel	159,753.31
C-30-19	Compressor	181.44	kW	Carbon Steel	154,983.21
C-30-23	Compressor	177.02	kW	Carbon Steel	152,432.32
C-5	Compressor	2,710.86	kW	Carbon Steel	958,473.21
C-23	Compressor	468.16	kW	Carbon Steel	293,537.30
PD-20-1	Pump include drive	20.17	cubic.m /s*kPa	Stainless Steel	17,371.37
PD-28	Pump include drive	41.16	cubic.m /s*kPa	Stainless Steel	23,254.12
TB-18-33	Turbine	935.64	kW	Stainless Steel	549,673.47
TB-18-35	Turbine	969.78	kW	Stainless Steel	561,400.44
TB-18-38	Turbine	1,879.65	kW	Stainless Steel	828,940.02
TB-18-351	Turbine	535.28	kW	Stainless Steel	395,622.37
T-111	Tower Unit	7.22	m	316 Stainless Steel	424,543.21
tT-111	valvetray	3.00	trays	Stainless Steel	13,467.45

**Table B2.8** Equipment sizing and purchase cost (con't.)

Equipment		Size	Unit	Material	Purchase Cost (\$)
cT-111	Heat Exchanger	17.20	m <sup>2</sup>	Carbon Steel	18,536.76
E-12	Heat Exchanger	24.05	m <sup>2</sup>	SS shell&SS tube	10,865.85
T-19	Tower Unit	26.71	m	316 Stainless Steel	926,452.77
tT-19	valvetray	35.00	trays	Stainless Steel	132,385.32
cT-19	Heat Exchanger	686.40	m <sup>2</sup>	SS shell&SS tube	201,744.01
E-21	Heat Exchanger	30.41	m <sup>2</sup>	SS shell&SS tube	11,262.43
E-9	Heat Exchanger	20.20	m <sup>2</sup>	SS shell&SS tube	8,664.89
R-7	Methanol synthesis reactor	2,261.95	m <sup>2</sup>	SS shell&SS tube	1,488,477.00
R-10-29	Vessel	10.00	m	316 Stainless Steel	171,952.00
R-1	Bi-reforming reactor	1,456.49	m <sup>2</sup>	SS shell&SS tube	751,608.00
<b>Total</b>					<b>9,830,473.15</b>

### B.2.1.5 Capital Cost Analysis

**Table B2.9** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	10,813,520.47
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	5,082,354.62
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	3,892,867.37
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	7,353,193.92
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	1,189,487.25
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	1,946,433.68
Yard Improvement (% of Purchased Equipment Delivered)	10.00	1,081,352.05
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	7,569,464.33
<b>Total Direct Cost</b>		38,928,673.67
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	3,568,461.75
Construction Expenses	41.00	4,433,543.39
Legal Expenses	4.00	432,540.82
Contractor's Fees	22.00	2,378,974.50
Contingency	44.00	4,757,949.00
<b>Total Indirect Cost</b>		15,571,469.47
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		54,500,143.14
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	9,624,033.21
<b>V. Total Capital Investment (TCI) = Fixed-Capital Investment + Working Capital</b>		64,124,176.36

### B.2.1.6 Production Cost Analysis

**Table B2.10** Breakdown of production cost

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	23,358,212.48
Operating Labor	25.00	Fixed Capital Investment	13,625,036.00
Operating Supervision	15.00	Operating Labor	2,043,755.40
Utilities	-	-	8,991.13
Maintenance and Repairs	6.00	Fixed Capital Investment	3,270,008.59
Operating Supplies	15.00	Maintenance and Supplies	490,501.28
Laboratory Charges	15.00	Operating Labor	2,043,755.40
Royalties	1.00	Total Product Cost	678,866.80
Catalysts	-	-	150,572.50
<b>Total Variable Cost</b>			45,669,699.57
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	1,090,002.86
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	545,001.43
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,635,004.29
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	•Labor + Supervision + Maintenance	11,363,280.00
<b>Total Manufacturing Cost</b>			47,304,703.87
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	3,787,760.00
Distribution & selling	4.00	Total Product Cost	2,715,467.20
Research & Development	4.00	Total Product Cost	2,715,467.20
<b>General Expense</b>			9,218,694.40
<b>V. Total Product Cost with Out Depreciation</b>			67,886,680.00

**B.2.1.7 Profitability Assumptions****Table B2.11 Profitability conditions**

<b>Project Details</b>	
Project Life Time	20 years
Land Cost	\$1,000,000.00
Construction Inflation Rate	2%
Product Inflation Rate	10%
TPC Inflation Rate	10%
Minimum Rate of Return, Mar	15%
Income Tax Rate	30%
Type of Depreciation	MACRS

### B.2.2 Bi-reforming of CO<sub>2</sub> into Methanol for the Optimized Case

#### B.2.2.1 Raw Materials and Product Annual Price

**Table B2.12** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Methane	103,666,926.00	20,526,051.24
CO <sub>2</sub>	53,716,348.00	1,896,187.08
Demineralized water	429,054,784.00	695,068.75

**Table B2.13** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	485,331.60	161,615,424.00	77,575,403.52

#### B.2.2.2 Annual Utility Cost

**Table B2.14** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-5	2,590.50	1,372,626.17
C-23	488.70	258,946.90
PD-20-1	44.81	23,745.62
PD-28	91.46	48,460.45
C-30-10	189.79	100,562.54
C-30-19	181.44	96,138.50
C-30-23	177.02	93,799.89
<b>Total</b>		1,994,280.07

**Table B2.15** Annual generated electricity cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
TB-18-33	-919.76	-487,352.00
TB-18-35	-953.32	-505,138.00
TB-18-38	-1,847.75	-979,064.00
TB-18-351	-533.77	-282,830.00
<b>Total</b>		<b>-2,254,384.00</b>

**Table B2.16** Annual cooling water

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
E-30-1	193.42	1,970.00
E-30-18	172.52	1,757.00
E-201	10,077.79	102,642.00
E-15	13,742.55	139,968.00
E-9	396.07	4,034.00
<b>Total</b>		<b>250,371.00</b>

**B.2.2.3 Equipment Sizing and Purchase Cost**

Chemical engineering plant cost index (12/2014): 575.8

**Table B2.17 Equipment sizing and purchase cost**

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase Cost (\$)</b>
V-4	Vessel	3.83	m	316 Stainless Steel	139,106.87
V-41	Vessel	5.15	m	316 Stainless Steel	128,904.35
V-20-2	Vessel	1.76	m	316 Stainless Steel	61,454.50
E-30-1	Heat Exchanger	18.56	m <sup>2</sup>	SS shell&SS tube	8,367.71
E-30-18	Heat Exchanger	18.12	m <sup>2</sup>	SS shell&SS tube	8,286.57
E-27	Heat Exchanger	203.75	m <sup>2</sup>	SS shell&SS tube	45,051.76
E-3	Heat Exchanger	355.91	m <sup>2</sup>	SS shell&SS tube	80,996.41
E-201	Heat Exchanger	682.09	m <sup>2</sup>	SS shell&SS tube	143,630.00
E-6	Heat Exchanger	3,246.69	m <sup>2</sup>	CS shell&SS tube	398,730.00
E-15	Heat Exchanger	2,216.61	m <sup>2</sup>	SS shell&SS tube	466,140.00
E-18-1	Heat Exchanger	41.89	m <sup>2</sup>	SS shell&SS tube	13,498.89
E-18-2	Heat Exchanger	186.63	m <sup>2</sup>	SS shell&SS tube	41,714.82
E-18-3	Heat Exchanger	22.28	m <sup>2</sup>	SS shell&SS tube	9,678.15
E-18-4	Heat Exchanger	66.50	m <sup>2</sup>	SS shell&SS tube	18,297.40
E-18-5	Heat Exchanger	94.60	m <sup>2</sup>	SS shell&SS tube	22,218.56
C-5	Compressor	2,590.50	kW	Carbon Steel	929,587.87
C-23	Compressor	488.70	kW	Carbon Steel	302,154.52
PD-20-1	Pump include drive	20.17	cubic.m /s*kPa	Stainless Steel	17,371.37
PD-28	Pump include drive	41.16	cubic.m /s*kPa	Stainless Steel	23,254.12
TB-18-33	Turbine	919.76	kW	Stainless Steel	544,160.92
TB-18-35	Turbine	953.32	kW	Stainless Steel	555,769.22
TB-18-38	Turbine	1,847.75	kW	Stainless Steel	820,626.06
TB-18-351	Turbine	533.77	kW	Stainless Steel	394,966.14
T-111	Tower Unit	7.22	m	316 Stainless Steel	424,543.21
tT-111	valvetray	3.00	trays	Stainless Steel	13,438.04
cT-111	Heat Exchanger	12.86	m <sup>2</sup>	Carbon Steel	16,400.72
E-12	Heat Exchanger	24.07	m <sup>2</sup>	SS shell&SS tube	10,869.91
T-19	Tower Unit	26.71	m	316 Stainless Steel	926,452.77

**Table B2.17** Equipment sizing and purchase cost (con't.)

Equipment		Size	Unit	Material	Purchase Cost (\$)
tT-19	valvetray	35.00	trays	Stainless Steel	131,791.97
cT-19	Heat Exchanger	682.27	m <sup>2</sup>	SS shell&SS tube	201,231.81
E-21	Heat Exchanger	32.32	m <sup>2</sup>	SS shell&SS tube	11,633.66
E-9	Heat Exchanger	20.16	m <sup>2</sup>	SS shell&SS tube	8,658.81
C-30-10	Compressor	189.79	kW	Carbon Steel	159,753.00
C-30-19	Compressor	181.44	kW	Carbon Steel	154,983.00
C-30-23	Compressor	177.02	kW	Carbon Steel	152,433.00
R-7	Methanol synthesis reactor	2,261.95	m <sup>2</sup>	SS shell&SS tube	1,488,477.00
R-10-29	Vessel	10.00	m	316 Stainless Steel	171,952.00
R-1	Bi-reforming reactor	1,439.50	m <sup>2</sup>	SS shell&SS tube	743,328.00
<b>Total</b>					<b>9,789,913.09</b>

#### B.2.2.4 Capital Cost Analysis

**Table B2.18** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	10,768,904.40
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	5,061,385.07
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	3,876,805.58
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	7,322,854.99
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	1,184,579.48
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	1,938,402.79
Yard Improvement (% of Purchased Equipment Delivered)	10.00	1,076,890.44
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	7,538,233.08
<b>Total Direct Cost</b>		38,768,055.84
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	3,553,738.45
Construction Expenses	41.00	4,415,250.80
Legal Expenses	4.00	430,756.18
Contractor's Fees	22.00	2,369,158.97
Contingency	44.00	4,738,317.94
<b>Total Indirect Cost</b>		15,507,222.33
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		54,275,278.17
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	9,584,324.92
<b>V. Total Capital Investment (TCI) = Fixed-Capital Investment + Working Capital</b>		63,859,603.09

### B.2.2.5 Production Cost Analysis

**Table B2.19** Breakdown of production cost

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	23,117,307.07
Operating Labor	25.00	Fixed Capital Investment	13,568,820.00
Operating Supervision	15.00	Operating Labor	2,035,323.00
Utilities	-	-	-9,732.93
Maintenance and Repairs	6.00	Fixed Capital Investment	3,256,516.69
Operating Supplies	15.00	Maintenance and Supplies	488,477.51
Laboratory Charges	15.00	Operating Labor	2,035,323.00
Royalties	1.00	Total Product Cost	674,274.80
Catalysts	-	-	150,186.05
<b>Total Variable Cost</b>			45,316,495.19
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	1,085,505.56
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	542,752.78
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,628,258.35
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	11,316,396.00
<b>Total Manufacturing Cost</b>			46,944,753.53
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	3,772,132.00
Distribution & selling	4.00	Total Product Cost	2,697,099.20
Research & Development	4.00	Total Product Cost	2,697,099.20
<b>General Expense</b>			9,166,330.40
<b>V. Total Product Cost with Out Depreciation</b>			67,427,480.00

### B.2.3 Bi-reforming of CO<sub>2</sub> into Methanol for the Alternative Case

#### B.2.3.1 Raw Materials and Product Annual Price

**Table B2.20** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Methane	75,602,544.00	14,969,303.63
CO <sub>2</sub>	53,716,348.00	1,896,187.17
Demineralized water	429,054,784.00	695,068.75

**Table B2.21** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	485,334.14	161,616,272.00	77,575,810.56

#### B.2.3.2 Annual Utility Cost

**Table B2.22** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-30-10	189.79	100,562.54
C-30-19	181.44	96,138.50
C-30-23	177.02	93,799.89
C-5	2,590.51	1,372,635.14
C-23	488.68	258,934.16
PD-20-1	44.81	23,745.62
PD-28	91.46	48,460.45
<b>Total</b>		<b>1,994,276.30</b>

**Table B2.23** Annual cooling water

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
E-30-1	193.42	1,970.00
E-30-18	172.52	1,757.00
E-201	10,077.93	102,644.00
E-15	13,742.29	139,965.00
E-9	396.07	4,034.00
<b>Total</b>		<b>250,370.00</b>

### B.2.3.3 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B2.24 Equipment sizing and purchase cost**

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase Cost (\$)</b>
V-4	Vessel	3.83	m	316 Stainless Steel	139,109.00
V-41	Vessel	5.15	m	316 Stainless Steel	128,904.00
V-20-2	Vessel	1.76	m	316 Stainless Steel	61,449.00
E-30-1	Heat Exchanger	18.56	m <sup>2</sup>	SS shell&SS tube	8,367.00
E-30-18	Heat Exchanger	18.12	m <sup>2</sup>	SS shell&SS tube	8,287.00
E-3	Heat Exchanger	355.93	m <sup>2</sup>	SS shell&SS tube	81,000.00
E-201	Heat Exchanger	682.10	m <sup>2</sup>	SS shell&SS tube	143,632.00
E-6	Heat Exchanger	3,246.51	m <sup>2</sup>	CS shell&SS tube	398,712.00
E-15	Heat Exchanger	2,216.54	m <sup>2</sup>	SS shell&SS tube	466,124.00
E-91	Heat Exchanger	20.16	m <sup>2</sup>	SS shell&SS tube	8,659.00
C-30-10	Compressor	189.79	kW	Carbon Steel	159,753.00
C-30-19	Compressor	181.44	kW	Carbon Steel	154,983.00
C-30-23	Compressor	177.02	kW	Carbon Steel	152,433.00
C-5	Compressor	2,590.51	kW	Carbon Steel	929,592.00
C-23	Compressor	488.68	kW	Carbon Steel	302,144.00
PD-20-1	Pump include drive	20.17	cubic.m /s*kPa	Stainless Steel	17,372.00
PD-28	Pump include drive	41.16	cubic.m /s*kPa	Stainless Steel	23,254.00
T-111	Tower Unit	7.22	m	316 Stainless Steel	424,543.00
tT-111	valvetray	3.00	trays	Stainless Steel	13,438.00
cT-111	Heat Exchanger	12.86	m <sup>2</sup>	SS shell&SS tube	16,402.00
E-12	Heat Exchanger	24.07	m <sup>2</sup>	SS shell&SS tube	10,870.00
T-19	Tower Unit	26.71	m	316 Stainless Steel	926,453.00
tT-19	valvetray	35.00	trays	Stainless Steel	131,793.00
cT-19	Heat Exchanger	682.28	m <sup>2</sup>	SS shell&SS tube	201,233.00
E-21	Heat Exchanger	32.32	m <sup>2</sup>	SS shell&SS tube	11,634.00
R-7	Methanol synthesis reactor	2,261.95	m <sup>2</sup>	SS shell&SS tube	1,488,477.00
R-1	Bi-reforming reactor	2,139.57	m <sup>2</sup>	SS shell&SS tube	1,105,452.00
<b>Total</b>					<b>7,514,069.00</b>

### B.2.3.4 Capital Cost Analysis

**Table B2.25** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	8,265,475.90
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	3,884,773.67
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	2,975,571.32
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	5,620,523.61
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	909,202.35
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	1,487,785.66
Yard Improvement (% of Purchased Equipment Delivered)	10.00	826,547.59
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	5,785,833.13
<b>Total Direct Cost</b>		29,755,713.24
<b>II. Indirect costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	2,727,607.05
Construction Expenses	41.00	3,388,845.12
Legal Expenses	4.00	330,619.04
Contractor's Fees	22.00	1,818,404.70
Contingency	44.00	3,636,809.40
<b>Total Indirect Cost</b>		11,902,285.30
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		41,657,998.54
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	7,356,273.55
<b>V. Total Capital Investment (TCI) = Fixed-capital Investment + Working Capital</b>		49,014,272.09

### B.2.3.5. Production Cost Analysis

**Table B2.26** Breakdown of production cost

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	17,560,559.54
Operating Labor	25.00	Fixed Capital Investment	10,414,500.00
Operating Supervision	15.00	Operating Labor	1,562,175.00
Utilities	-	-	2,244,646.30
Maintenance and Repairs	6.00	Fixed Capital Investment	2,499,479.91
Operating Supplies	15.00	Maintenance and Supplies	374,922.00
Laboratory Charges	15.00	Operating Labor	1,562,175.00
Royalties	1.00	Total Product Cost	540,826.76
Catalysts	-	-	166,112.55
<b>Total Variable Cost</b>			36,925,397.07
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	833,159.97
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	416,579.99
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			1,249,739.96
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	8,685,693.00
<b>Total Manufacturing Cost</b>			38,175,137.02
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	2,895,231.00
Distribution & selling	4.00	Total Product Cost	2,163,307.04
Research & Development	4.00	Total Product Cost	2,163,307.04
<b>General Expense</b>			7,221,845.08
<b>V. Total Product Cost with Out Depreciation</b>			54,082,676.00

### B.3 Tri-reforming of CO<sub>2</sub> into Methanol

#### B.3.1 Tri-reforming of CO<sub>2</sub> into Methanol for the Base Case

##### B.3.1.1 Raw Material, Product and Utilities Prices

**Table B3.1** Raw material and product prices

<b>Raw Material Price</b>		
<b>Raw Material</b>	<b>Value</b>	<b>Unit</b>
Methane <sup>[1]</sup>	198	\$/tCH <sub>4</sub>
Demineralized water <sup>[2]</sup>	1.62	\$/tH <sub>2</sub> O
<b>Product Price</b>		
<b>Product</b>	<b>Value</b>	<b>Unit</b>
Methanol <sup>[3]</sup>	0.48	\$/kgMeOH

References:

- [1] [http://www.eia.gov/dnav/ng/ng\\_pri\\_sum\\_dcu\\_nus\\_m.htm](http://www.eia.gov/dnav/ng/ng_pri_sum_dcu_nus_m.htm) (12/2014)
- [2] [www.alibaba.com](http://www.alibaba.com) (12/2014)
- [3] [www.icis.com](http://www.icis.com) (12/2014)

**Table B3.2** Utility price

<b>Cooling Water</b>		
	<b>Value</b>	<b>Unit</b>
Cooling water 28-40 °C*	0.35	\$/GJ
<b>Electricity</b>		
	<b>Value</b>	<b>Unit</b>
Electricity*	18.42	\$/GJ

\* 2013 U.S. average cost

### B.3.1.2 Raw Materials and Product Annual Price

**Table B3.3** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Methane	104,789,038.00	20,748,229.41
Demineralized water	331,729,408.00	537,401.64

**Table B3.4** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	481,319.64	160,279,440.00	76,934,131.20

### B.3.1.3 Annual Utility Cost

**Table B3.5** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-11-10	3,706.28	1,963,843.32
C-11-19	3,684.25	1,952,172.59
C-11-23	3,661.22	1,939,971.24
C-11-24	3,709.67	1,965,640.18
C-23	1,907.37	1,010,656.23
PD-20-1	40.70	21,564.90
PD-28	64.66	34,262.18
<b>Total</b>		<b>8,888,110.64</b>

**Table B3.6** Annual generated electricity cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
TB-14-33	-1,963.54	-1,040,422.00
TB-14-35	-1,980.07	-1,049,180.00
TB-14-38	-3,165.39	-1,677,245.00
TB-351	-656.00	-347,594.00
TB-17	-608.02	-322,171.00
TB-8-261	-703.69	-372,862.00
TB-8-171	-587.63	-311,368.00
TB-8-271	-1,357.80	-719,456.00
<b>Total</b>		<b>-5,840,298.00</b>

**Table B3.7** Annual cooling water cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
E-11-1	422.44	4,303.00
E-11-18	3,671.14	37,391.00
E-11-20	3,771.68	38,415.00
E-11-4	3,462.64	35,267.00
E-15	15,670.34	159,603.00
E-27	6,788.86	69,145.00
<b>Total</b>		<b>344,124.00</b>

### B.3.1.4 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B3.8** Equipment sizing and purchase cost

Equipment	Size	Unit	Material	Purchase Cost (\$)	
V-11-2	Vessel	1.17	m	316 Stainless Steel	27,701.69
V-11-3	Vessel	1.30	m	316 Stainless Steel	27,701.69
V-11-5	Vessel	0.91	m	316 Stainless Steel	55,402.37
V-4	Vessel	4.96	m	316 Stainless Steel	126,471.12
V-5	Vessel	4.89	m	316 Stainless Steel	43,689.60
V-20-2	Vessel	1.00	m	316 Stainless Steel	55,402.37
E-9	Heat Exchanger	176.57	m <sup>2</sup>	SS shell&SS tube	37,152.64
E-12	Heat Exchanger	494.33	m <sup>2</sup>	SS shell&SS tube	110,248.91
E-11-1	Heat Exchanger	136.88	m <sup>2</sup>	SS shell&SS tube	29,922.94
E-11-18	Heat Exchanger	311.61	m <sup>2</sup>	SS shell&SS tube	61,754.73
E-11-20	Heat Exchanger	319.63	m <sup>2</sup>	SS shell&SS tube	63,215.27
E-11-4	Heat Exchanger	292.93	m <sup>2</sup>	SS shell&SS tube	67,688.20
E-10	Heat Exchanger	28,258.70	m <sup>2</sup>	CS shell&SS tube	3,416,715.00
E-15	Heat Exchanger	4,796.43	m <sup>2</sup>	SS shell&SS tube	1,005,064.00
E-13	Heat Exchanger	19.97	m <sup>2</sup>	SS shell&SS tube	8,623.31
E-27	Heat Exchanger	341.79	m <sup>2</sup>	SS shell&SS tube	67,253.07
E-14-30	Heat Exchanger	32.60	m <sup>2</sup>	SS shell&SS tube	11,689.44
E-14-32	Heat Exchanger	34.60	m <sup>2</sup>	SS shell&SS tube	12,079.93
E-14-34	Heat Exchanger	33.56	m <sup>2</sup>	SS shell&SS tube	11,876.07
E-14-36	Heat Exchanger	64.74	m <sup>2</sup>	SS shell&SS tube	16,779.04
E-14-39	Heat Exchanger	3.03	m <sup>2</sup>	SS shell&SS tube	5,627.16
E-8-131	Heat Exchanger	15.63	m <sup>2</sup>	SS shell&SS tube	8,117.19
E-8-141	Heat Exchanger	13.62	m <sup>2</sup>	SS shell&SS tube	7,706.41
E-8-191	Heat Exchanger	29.39	m <sup>2</sup>	SS shell&SS tube	8,589.84
C-11-10	Compressor	3,706.28	kW	Carbon Steel	1,183,322.25
C-11-19	Compressor	3,684.25	kW	Carbon Steel	1,178,579.53
C-11-23	Compressor	3,661.22	kW	Carbon Steel	1,173,610.63
C-11-24	Compressor	3,709.67	kW	Carbon Steel	1,184,051.51

**Table B3.8 Equipment sizing and purchase cost (con't.)**

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase Cost (\$)</b>
C-23	Compressor	1,907.37	kW	Carbon Steel	756,326.54
PD-20-1	Pump include drive	18.31	cubic.m /s*kPa	Stainless Steel	16,817.58
PD-28	Pump include drive	29.10	cubic.m /s*kPa	Stainless Steel	19,938.48
TB-14-33	Turbine	1,963.54	kW	Stainless Steel	850,532.77
TB-14-35	Turbine	1,980.07	kW	Stainless Steel	854,741.98
TB-14-38	Turbine	3,165.39	kW	Stainless Steel	1,126,734.20
TB-351	Turbine	656.00	kW	Stainless Steel	445,957.45
TB-17	Turbine	608.02	kW	Stainless Steel	213,225.53
TB-8-261	Turbine	703.69	kW	Stainless Steel	232,386.06
TB-8-171	Turbine	587.63	kW	Stainless Steel	208,984.87
TB-8-271	Turbine	1,357.80	kW	Stainless Steel	342,228.26
R-3	Methanol synthesis reactor	20,216.15	m <sup>2</sup>	SS shell&SS tube	13,493,550.00
T-24	Tower Unit	21.83	m	316 Stainless Steel	789,313.58
tT-24	valvetray	27.00	trays	Stainless Steel	92,972.89
cT-24	Heat Exchanger	229.95	m <sup>2</sup>	SS shell&SS tube	127,223.70
E-26	Heat Exchanger	11.83	m <sup>2</sup>	SS shell&SS tube	7,639.47
R-1	Tri-reforming reactor	1,232.73	m <sup>2</sup>	SS shell&SS tube	399,640.00
R-8-29	Vessel	10.00	m	316 Stainless Steel	171,952.00
<b>Total</b>					<b>30,156,201.25</b>

### B.3.1.5 Capital Cost Analysis

**Table B3.9 Breakdown of capital cost**

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	33,171,821.38
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	15,590,756.05
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	11,941,855.70
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	22,556,838.54
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	3,648,900.35
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	5,970,927.85
Yard Improvement (% of Purchased Equipment Delivered)	10.00	3,317,182.14
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	23,220,274.96
<b>Total Direct Cost</b>		119,418,556.95
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	10,946,701.05
Construction Expenses	41.00	13,600,446.76
Legal Expenses	4.00	1,326,872.86
Contractor's Fees	22.00	7,297,800.70
Contingency	44.00	14,595,601.41
<b>Total Indirect Cost</b>		47,767,422.78
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		167,185,979.73
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	29,522,921.02
<b>V. Total Capital Investment (TCI) = Fixed-capital Investment + Working Capital</b>		196,708,900.75

### B.3.1.6 Production Cost Analysis

**Table B3.10** Breakdown of production cost

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	21,285,631.05
Operating Labor	10.00	Fixed Capital Investment	16,718,598.00
Operating Supervision	15.00	Operating Labor	2,507,789.70
Utilities	-	-	3,391,936.64
Maintenance and Repairs	6.00	Fixed Capital Investment	10,031,158.78
Operating Supplies	15.00	Maintenance and Supplies	1,504,673.85
Laboratory Charges	15.00	Operating Labor	2,507,789.70
Royalties	1.00	Total Product Cost	960,919.60
Catalysts	-	-	1,074,486.69
<b>Total Variable Cost</b>			59,982,984.01
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	3,343,719.59
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	1,671,859.80
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			5,015,579.39
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	17,554,527.60
<b>Total Manufacturing Cost</b>			64,998,563.40
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	5,851,509.20
Distribution & selling	4.00	Total Product Cost	3,843,678.40
Research & Development	4.00	Total Product Cost	3,843,678.40
<b>General Expense</b>			13,538,866.00
<b>V. Total Product Cost with Out Depreciation</b>			96,091,960.00

**B.3.1.7 Profitability Assumptions****Table B3.11:** Profitability conditions

Project Details	
Project Life Time	20 years
Land Cost	\$1,000,000.00
Construction Inflation Rate	2%
Product Inflation Rate	10%
TPC Inflation Rate	10%
Minimum Rate of Return, Mar	15%
Income Tax Rate	30%
Type of Depreciation	MACRS

### B.3.2 Tri-reforming of CO<sub>2</sub> into Methanol for the Optimized Case

#### B.3.2.1 Raw Materials and Product Annual Price

**Table B3.12** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Methane	105,055,966.00	20,801,081.16
Demineralized water	331,729,408.00	537,401.64

**Table B3.13** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	483,952.75	161,156,272.00	77,355,010.56

#### B.3.2.2 Annual Utility Cost

**Table B3.14** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-11-10	3,711.66	1,966,697.63
C-11-19	3,684.04	1,952,059.72
C-11-23	3,660.98	1,939,843.36
C-11-24	3,315.79	1,756,938.35
C-23	2,076.32	1,100,178.79
PD-20-1	40.70	21,564.90
PD-28	64.66	34,262.18
<b>Total</b>		<b>8,771,544.93</b>

**Table B3.15** Annual generated electricity cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
TB-14-33	-1,944.92	-1,030,553.00
TB-14-35	-1,966.36	-1,041,914.00
TB-14-38	-3,141.67	-1,664,673.00
TB-351	-661.17	-350,332.00
TB-17	-584.44	-309,675.00
TB-8-261	-641.72	-340,027.00
TB-8-171	-589.99	-312,617.00
TB-8-271	-1,357.95	-719,539.00
<b>Total</b>		<b>-5,769,330.00</b>

**Table B3.16** Annual cooling water cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
E-11-1	423.36	4,312.00
E-11-18	3,737.19	38,063.00
E-11-20	3,771.56	38,413.00
E-11-4	3,462.61	35,267.00
E-15	15,561.05	158,489.00
E-27	6,826.24	69,525.00
<b>Total</b>		<b>344,069.00</b>

### B.3.2.3 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B3.17 Equipment sizing and purchase cost**

Equipment	Size	Unit	Material	Purchase cost (\$)	
V-11-2	Vessel	1.24	m	316 Stainless Steel	27,701.69
V-11-3	Vessel	1.30	m	316 Stainless Steel	27,701.69
V-11-5	Vessel	0.91	m	316 Stainless Steel	55,402.37
V-4	Vessel	4.97	m	316 Stainless Steel	126,576.60
V-5	Vessel	4.91	m	316 Stainless Steel	43,746.40
V-20-2	Vessel	0.65	m	316 Stainless Steel	55,402.37
E-9	Heat Exchanger	177.53	m <sup>2</sup>	SS shell&SS tube	37,328.11
E-12	Heat Exchanger	487.81	m <sup>2</sup>	SS shell&SS tube	100,424.71
E-11-1	Heat Exchanger	137.18	m <sup>2</sup>	SS shell&SS tube	29,976.69
E-11-18	Heat Exchanger	317.40	m <sup>2</sup>	SS shell&SS tube	62,808.55
E-11-20	Heat Exchanger	319.74	m <sup>2</sup>	SS shell&SS tube	63,235.56
E-11-4	Heat Exchanger	293.04	m <sup>2</sup>	SS shell&SS tube	58,372.14
E-10	Heat Exchanger	26,393.78	m <sup>2</sup>	CS shell&SS tube	3,191,034.00
E-15	Heat Exchanger	4,828.58	m <sup>2</sup>	SS shell&SS tube	1,011,504.00
E-13	Heat Exchanger	20.48	m <sup>2</sup>	SS shell&SS tube	8,716.62
E-27	Heat Exchanger	343.63	m <sup>2</sup>	SS shell&SS tube	67,586.77
E-14-30	Heat Exchanger	32.78	m <sup>2</sup>	SS shell&SS tube	12,710.81
E-14-32	Heat Exchanger	34.82	m <sup>2</sup>	SS shell&SS tube	13,140.86
E-14-34	Heat Exchanger	33.54	m <sup>2</sup>	SS shell&SS tube	12,870.05
E-14-36	Heat Exchanger	64.68	m <sup>2</sup>	SS shell&SS tube	16,767.88
E-14-39	Heat Exchanger	3.04	m <sup>2</sup>	SS shell&SS tube	5,627.16
E-8-131	Heat Exchanger	15.61	m <sup>2</sup>	SS shell&SS tube	8,115.16
E-8-141	Heat Exchanger	13.96	m <sup>2</sup>	SS shell&SS tube	7,737.85
E-8-191	Heat Exchanger	30.37	m <sup>2</sup>	SS shell&SS tube	8,675.04
C-11-10	Compressor	3,711.66	kW	Carbon Steel	1,184,480.54
C-11-19	Compressor	3,684.04	kW	Carbon Steel	1,178,532.88
C-11-23	Compressor	3,660.98	kW	Carbon Steel	1,173,558.90
C-11-24	Compressor	3,315.79	kW	Carbon Steel	1,097,803.22

**Table B3.17 Equipment sizing and purchase cost (con't.)**

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase cost (\$)</b>
C-23	Compressor	2,076.32	kW	Carbon Steel	800,839.72
PD-20-1	Pump include drive	18.31	cubic.m /s*kPa	Stainless Steel	16,817.58
PD-28	Pump include drive	29.10	cubic.m /s*kPa	Stainless Steel	19,938.48
TB-14-33	Turbine	1,944.92	kW	Stainless Steel	845,772.81
TB-14-35	Turbine	1,966.36	kW	Stainless Steel	851,250.87
TB-14-38	Turbine	3,141.67	kW	Stainless Steel	1,121,752.12
TB-351	Turbine	661.17	kW	Stainless Steel	448,023.52
TB-17	Turbine	584.44	kW	Stainless Steel	208,315.45
TB-8-261	Turbine	641.72	kW	Stainless Steel	220,107.33
TB-8-171	Turbine	589.99	kW	Stainless Steel	209,478.82
TB-8-271	Turbine	1,357.95	kW	Stainless Steel	342,251.58
R-3	Methanol synthesis reactor	20,216.15	m <sup>2</sup>	SS shell&SS tube	13,493,550.00
T-24	Tower Unit	21.83	m	316 Stainless Steel	780,313.58
tT-24	valvetray	27.00	trays	Stainless Steel	93,356.28
cT-24	Heat Exchanger	231.18	m <sup>2</sup>	SS shell&SS tube	127,509.73
E-26	Heat Exchanger	11.74	m <sup>2</sup>	SS shell&SS tube	7,623.24
R-1	Tri-reforming reactor	1,220.31	m <sup>2</sup>	SS shell&SS tube	396,786.00
R-8-29	Vessel	10.00	m	316 Stainless Steel	171,952.00
<b>Total</b>					<b>29,852,177.73</b>

### B.3.2.4 Capital Cost Analysis

**Table B3.18** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	32,837,395.50
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	15,433,575.89
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	11,821,462.38
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	22,329,428.94
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	3,612,113.51
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	5,910,731.19
Yard Improvement (% of Purchased Equipment Delivered)	10.00	3,283,739.55
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	22,986,176.85
<b>Total Direct Cost</b>		118,214,623.81
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	10,836,340.52
Construction Expenses	41.00	13,463,332.16
Legal Expenses	4.00	1,313,495.82
Contractor's Fees	22.00	7,224,227.01
Contingency	44.00	14,448,454.02
<b>Total Indirect Cost</b>		47,285,849.52
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	29,225,282.00
<b>V. Total Capital Investment (TCI) = Fixed-capital Investment + Working Capital</b>		
		194,725,755.33

### B.3.2.5 Production Cost Analysis

**Table B3.19 Breakdown of production cost**

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	21,338,482.79
Operating Labor	10.00	Fixed Capital Investment	16,550,048.00
Operating Supervision	15.00	Operating Labor	2,482,507.20
Utilities	-	-	3,346,283.93
Maintenance and Repairs	6.00	Fixed Capital Investment	9,930,028.40
Operating Supplies	15.00	Maintenance and Supplies	1,489,504.20
Laboratory Charges	15.00	Operating Labor	2,482,507.20
Royalties	1.00	Total Product Cost	942,356.48
Catalysts	-	-	21,338,482.79
<b>Total Variable Cost</b>			58,561,718.20
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	3,310,009.47
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	1,655,004.73
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			4,965,014.20
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	17,377,550.40
<b>Total Manufacturing Cost</b>			63,526,732.40
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	5,792,516.80
Distribution & selling	4.00	Total Product Cost	3,769,425.92
Research & Development	4.00	Total Product Cost	3,769,425.92
<b>General Expense</b>			13,331,368.64
<b>V. Total Product Cost with Out Depreciation</b>			94,235,648.00

### B.3.3 Tri-reforming of CO<sub>2</sub> into Methanol for the Alternative Case

#### B.3.3.1 Raw Materials and Product Annual Price

**Table B3.20** Raw materials annual price

Raw Material	Quantity (kg/year)	Annual Price (\$/year)
Methane	72,980,760.00	14,450,190.40
Demineralized water	331,729,408.00	537,401.64

**Table B3.21** Products annual price

Product	Quantity (kg/day)	Quantity (kg/year)	Annual Price (\$/year)
Methanol	483,958.17	161,158,064.00	77,355,870.72

#### B.3.3.2 Annual Utility Cost

**Table B3.22** Annual electricity cost

Equipment	Quantity (kW)	Annual Price (\$/year)
C-11-10	3,711.73	1,966,732.86
C-11-19	3,684.06	1,952,073.71
C-11-23	3,661.01	1,939,857.22
C-11-24	3,315.82	1,756,950.82
C-23	2,076.23	1,100,131.56
PD-20-1	40.70	21,564.90
PD-28	64.66	34,262.18
<b>Total</b>		<b>8,771,573.25</b>

**Table B3.23** Annual generated electricity cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
TB-14-33	-1,824.42	-966,706.00
TB-14-35	-1,686.78	-893,776.00
TB-14-38	-2,545.38	-1,348,718.00
TB-351	-535.75	-283,877.00
TB-17	-584.51	-309,712.00
<b>Total</b>		<b>-3,802,789.00</b>

**Table B3.24** Annual cooling water cost

<b>Equipment</b>	<b>Quantity (kW)</b>	<b>Annual Price (\$/year)</b>
E-11-1	423.37	4,312.00
E-11-18	3,737.68	38,068.00
E-11-20	3,771.58	38,414.00
E-11-4	3,462.63	35,267.00
E-15	15,560.86	158,487.00
E-27	6,826.32	69,526.00
<b>Total</b>		<b>344,074.00</b>

### B.3.3.3 Equipment Sizing and Purchase Cost

Chemical engineering plant cost index (12/2014): 575.8

**Table B3.25 Equipment sizing and purchase cost**

Equipment	Size	Unit	Material	Purchase cost (\$)	
V-11-2	Vessel	1.24	m	316 Stainless Steel	27,701.69
V-11-3	Vessel	1.30	m	316 Stainless Steel	27,701.69
V-11-5	Vessel	0.91	m	316 Stainless Steel	55,402.37
V-4	Vessel	4.97	m	316 Stainless Steel	126,576.60
V-5	Vessel	4.91	m	316 Stainless Steel	43,746.40
V-20-2	Vessel	0.65	m	316 Stainless Steel	55,402.37
E-12	Heat Exchanger	487.84	m <sup>2</sup>	SS shell&SS tube	100,429.78
E-11-1	Heat Exchanger	137.18	m <sup>2</sup>	SS shell&SS tube	29,977.71
E-11-18	Heat Exchanger	317.44	m <sup>2</sup>	SS shell&SS tube	62,816.67
E-11-20	Heat Exchanger	319.74	m <sup>2</sup>	SS shell&SS tube	63,235.56
E-11-4	Heat Exchanger	293.05	m <sup>2</sup>	SS shell&SS tube	62,458.63
E-10	Heat Exchanger	26,392.87	m <sup>2</sup>	CS shell&SS tube	3,190,908.00
E-15	Heat Exchanger	4,828.43	m <sup>2</sup>	SS shell&SS tube	1,011,472.00
E-13	Heat Exchanger	34.87	m <sup>2</sup>	SS shell&SS tube	11,338.50
E-27	Heat Exchanger	343.63	m <sup>2</sup>	SS shell&SS tube	67,587.78
E-14-30	Heat Exchanger	24.92	m <sup>2</sup>	SS shell&SS tube	10,193.39
E-14-32	Heat Exchanger	24.31	m <sup>2</sup>	SS shell&SS tube	10,072.70
E-14-34	Heat Exchanger	16.88	m <sup>2</sup>	SS shell&SS tube	8,624.32
E-14-36	Heat Exchanger	91.39	m <sup>2</sup>	SS shell&SS tube	21,634.34
E-14-39	Heat Exchanger	8.50	m <sup>2</sup>	SS shell&SS tube	6,532.90
C-11-10	Compressor	3,711.73	kW	Carbon Steel	1,184,494.74
C-11-19	Compressor	3,684.06	kW	Carbon Steel	1,178,538.96
C-11-23	Compressor	3,661.01	kW	Carbon Steel	1,173,563.98
C-11-24	Compressor	3,315.82	kW	Carbon Steel	1,097,808.29
C-23	Compressor	2,076.23	kW	Carbon Steel	800,816.39
PD-20-1	Pump include drive	18.31	cubic.m /s*kPa	Stainless Steel	16,817.58
PD-28	Pump include drive	29.10	cubic.m /s*kPa	Stainless Steel	19,938.48
TB-14-33	Turbine	1,824.42	kW	Stainless Steel	814,510.03

**Table B3.25** Equipment sizing and purchase cost (con't.)

<b>Equipment</b>		<b>Size</b>	<b>Unit</b>	<b>Material</b>	<b>Purchase cost (\$)</b>
TB-14-35	Turbine	1,686.78	kW	Stainless Steel	777,740.78
TB-14-38	Turbine	2,545.38	kW	Stainless Steel	990,984.56
TB-351	Turbine	535.75	kW	Stainless Steel	395,826.24
TB-17	Turbine	584.51	kW	Stainless Steel	416,659.31
R-3	Methanol synthesis reactor	20,216.15	m <sup>2</sup>	SS shell&SS tube	13,493,550.00
T-24	Tower Unit	21.83	m	316 Stainless Steel	789,313.58
tT-24	valvetray	27.00	trays	Stainless Steel	93,357.29
cT-24	Heat Exchanger	231.19	m <sup>2</sup>	SS shell&SS tube	127,510.74
E-26	Heat Exchanger	11.74	m <sup>2</sup>	SS shell&SS tube	7,623.24
R-1	Tri-reforming reactor	1,673.38	m <sup>2</sup>	SS shell&SS tube	503,780.00
<b>Total</b>					<b>28,876,647.59</b>

### B.3.3.4 Capital Cost Analysis

**Table B3.26** Breakdown of capital cost

Description	%	Result (\$)
<b>I. Direct Costs</b>		
Purchased Equipment Delivered (% of Purchase equipment cost)	110.00	31,764,312.35
Purchased Equipment Installation (% of Purchased Equipment Delivered)	47.00	14,929,226.80
Instrumentation and Controls (installed) (% of Purchased Equipment Delivered)	36.00	11,435,152.45
Piping (Installed) (% of Purchased Equipment Delivered)	68.00	21,599,732.40
Electrical Systems (Installed) (% of Purchased Equipment Delivered)	11.00	3,494,074.36
Buildings (Including Services) (% of Purchased Equipment Delivered)	18.00	5,717,576.22
Yard Improvement (% of Purchased Equipment Delivered)	10.00	3,176,431.23
Service Facilities (Installed) (% of Purchased Equipment Delivered)	70.00	22,235,018.64
<b>Total Direct Cost</b>		114,351,524.46
<b>II. Indirect Costs (% of Purchased Equipment Delivered)</b>		
Engineering and Supervision	33.00	10,482,223.08
Construction Expenses	41.00	13,023,368.06
Legal Expenses	4.00	1,270,572.49
Contractor's Fees	22.00	6,988,148.72
Contingency	44.00	13,976,297.43
<b>Total Indirect Cost</b>		45,740,609.78
<b>III. Fixed-capital Investment (FCI) = Direct Cost + Indirect Cost</b>		160,092,134.24
<b>IV. Working Capital Investments (WC) (% of Purchased Equipment Delivered)</b>	89.00	28,270,237.99
<b>V. Total Capital Investment (TCI) = Fixed-capital Investment + Working Capital</b>		188,362,372.23

### B.3.3.5 Production Cost Analysis

**Table B3.27** Breakdown of production cost

Items of Production Cost	% of Basis	Basis	Cost, \$/year
<b>I. Variable Cost</b>			
Raw Material	-	-	14,987,592.04
Operating Labor	10.00	Fixed Capital Investment	16,009,214.00
Operating Supervision	15.00	Operating Labor	2,401,382.10
Utilities	-	-	5,312,858.25
Maintenance and Repairs	6.00	Fixed Capital Investment	9,605,528.05
Operating Supplies	15.00	Maintenance and Supplies	1,440,829.20
Laboratory Charges	15.00	Operating Labor	2,401,382.10
Royalties	1.00	Total Product Cost	885,043.60
Catalysts	-	-	1,164,517.64
<b>Total Variable Cost</b>			54,208,346.99
<b>II. Fixed Charges</b>			
Property Taxes	2.00	Fixed Capital Investment	3,201,842.68
Financing (interest)	0.00	Fixed Capital Investment	0.00
Insurance	1.00	Fixed Capital Investment	1,600,921.34
Rent	0.00	Fixed Capital Investment	0.00
<b>Total Fixed Charges</b>			4,802,764.03
<b>III. Manufacturing Cost</b>			
Plant Overhead	60.00	Labor + Supervision + Maintenance	16,809,674.40
<b>Total Manufacturing Cost</b>			59,011,111.01
<b>IV. General Expense</b>			
Administration	20.00	Labor + Supervision + Maintenance	5,603,224.80
Distribution & selling	4.00	Total Product Cost	3,540,174.40
Research & Development	4.00	Total Product Cost	3,540,174.40
<b>General Expense</b>			12,683,573.60
<b>V. Total Product Cost with Out Depreciation</b>			88,504,360.00

### Appendix C Calculation of Indirect CO<sub>2</sub> Emission

**Table C1** Indirect CO<sub>2</sub> emission from the hydrogenation of CO<sub>2</sub> into methanol for the base case

Electricity Usage (kW)	CO <sub>2</sub> Emission Factor (kgCO <sub>2</sub> /J)	Indirect CO <sub>2</sub> Emission (kgCO <sub>2</sub> /hr)
5865.775	2.008e-07	4240.2

**Table C2** Indirect CO<sub>2</sub> emission from the bi-reforming of CO<sub>2</sub> into methanol for the base case

Electricity Requirement (kW)	Electricity Generated from Turbines (kW)	Electricity Usage (kW)	CO <sub>2</sub> Emission Factor (kgCO <sub>2</sub> /J)	Indirect CO <sub>2</sub> Emission (kgCO <sub>2</sub> /hr)
3863.538	-4320.350	-456.812	2.008e-07	-330.2

**Table C3** Indirect CO<sub>2</sub> emission from the tri-reforming of CO<sub>2</sub> into methanol for the base case

Electricity Requirement (kW)	Electricity Generated from Turbines (kW)	Electricity Usage (kW)	CO <sub>2</sub> Emission Factor (kgCO <sub>2</sub> /J)	Indirect CO <sub>2</sub> Emission (kgCO <sub>2</sub> /hr)
16774.138	-11022.149	5751.989	2.008e-07	4158.0

## CURRICULUM VITAE

**Name:** Mr. Nguyen Bui Huu Tuan

**Date of Birth:** November 03, 1988

**Nationality:** Vietnam

### **University Education:**

2006-2011 Bachelor Degree of Chemical Engineering, Ho Chi Minh City University of Technology, Vietnam

### **Work Experience:**

2011-2013 Position: Researcher

Company name: Ho Chi Minh City University of Technology, Vietnam

### **Proceedings:**

1. Roh, K.; Nguyen, B.H.T.; Suriyaphraphadilok, U.; Jay, H. L., and Rafiqul, G. (2015, May 31) Development of sustainable CO<sub>2</sub> conversion processes for the methanol production. Paper presented at Proceedings of the 12<sup>th</sup> International Symposium on Process Systems Engineering and 25<sup>th</sup> European Symposium of Computer Aided Process Engineering, Copenhagen, Denmark.

### **Presentations:**

1. Nguyen, B.H.T.; Rafiqul G., and Suriyaphraphadilok, U. (2015, April 21) Sustainable treatment for CO<sub>2</sub> from the coal-fired power plant through methanol production. Paper presented at Proceedings of the 21<sup>th</sup> PPC Symposium on Petroleum, Petrochemical, and Polymers, Bangkok, Thailand.