

## REFERENCES

- Broussely, M. and Archdale, G. (2004) Li-ion batteries and portable power source prospects for the next 5-10 years. Journal of Power Sources, 136, 386-394.
- Burger, C., Hsiao, B.S., and Chu, B. (2006) Nanofibrous materials and their applications. Annual Review of Materials Research, 36, 333–368.
- Choi, N.S., Lee, Y.M., Park, J.H., and Park, J.K. (2003) Interfacial enhancement between lithium electrode and polymer electrolytes. Journal of Power Sources, 119-121, 610-616.
- Chronakis, I.S. (2010) Micromanufacturing Engineering and Technology. New York: William Andrew.
- Cubillas, P. and Anderson, M.W. (2010) Zeolites and Catalysis, Synthesis, Reactions and Applications, 1, 1-55. Weinheim: WILEY-VCH.
- Dong, Z., Kennedy, S.J., and Wu, Y. (2011) Electrospinning materials for energy-related applications and devices. Journal of Power Sources, 196, 4886-4904.
- Frenot, A. and Chronakis, I.S. (2003) Polymer nanofibers assembled by electrospinning. Current Opinion in Colloid and Interface Science, 8, 64-75.
- Garg, K. and Bowlin G.L. (2011) Electrospinning jets and nanofibrous structures. Biomicrofluidics, 5, 013403.
- Guo, B., Li, Y., Yao, Y., Lin, Z., Ji, L., Xu, G., Liang, Y., Shi, Q., and Zhang, X. (2011) Electrospun  $\text{Li}_4\text{Ti}_5\text{O}_{12}/\text{C}$  composites for lithium-ion batteries with high rate performance. Solid State Ionics, 204-205, 61-65.
- Hatice, B.-S. and Olgun, G. (2011) Enhancement of conductivity in polyaniline-[poly(vinylidene chloride)-co-(vinyl acetate)] blends by irradiation. Radiation Physics and Chemistry, 80, 153-158.
- Huang, Z.M., Zhang, Y.Z., Kotakic, M., and Ramakrishna, S. (2003) A review on polymer nanofibers by electrospinning and their applications in nanocomposites. Composites Science and Technology, 63, 2223-2253.
- Ji, L., Yao, Y., Toprakci, O., Lin, Z., Liang, Y., Shi, Q., Medford, A.J., Millns, C.R., and Zhang, X. (2010) Fabrication of carbon nanofiber-driven electrodes

- from electrospun polyacrylonitrile/polypyrrole bicomponents for high-performance rechargeable lithium-ion batteries. Journal of Power Sources, 195, 2050-2056.
- Ji, L. and Zhang X. (2009) Fabrication of porous carbon/Si composite nanofibers as high-capacity battery electrodes. Electrochemistry Communications, 11, 1146-1149.
- Kanjwal, M.A., Barakat, N.A.M., Sheikh, F.A., Ganakumar, G., Park, D.K., and Kim, H.Y. (2010) Titanium oxide nanofibers attached to zinc oxide nanobranches as a novel nanostructure for lithium ion batteries applications. Journal of Ceramic Processing Research, 11(4), 437-442.
- Kulkarni, A., Bambole, V.A., and Mahanwar, P.A. (2011) Electrospinning of polymers, their Modeling and applications. Polymer-Plastics Technology and Engineering, 49, 427-441.
- Lee, B.S., Son, S.B., Park, K.M., Seo, J.H., Lee, S.H., Choi, I.S., Oh, K.H., and Yu, W.R. (2012) Fabrication of Si core/C shell nanofibers and their electrochemical performances as a lithium-ion battery anode. Journal of Power Sources, 206, 267-273.
- Lee, B.S., Son, S.B., Park, K.M., Yu, W.R., Oh, K.H., and Lee, S.H. (2012) Anodic properties of hollow carbon nanofibers for Li-ion battery. Journal of Power Sources, 199, 53-60.
- Liu, B., Yu, Y., Chang, J., Yang, X., Wu, D., and Yang, X. (2011) An enhanced stable-structure core-shell coaxial carbon nanofiber web as a direct anode material for lithium-based batteries. Electrochemistry Communications, 13, 558-561.
- Liu, J. and Xue, D. (2010) Hollow nanostructured anode material for lithium-based batteries. Nanoscale Research Letters, 5, 1525-1534.
- Lu, H.W., Zeng, W., Li, Y.S., and Fu, Z.W. (2007) Fabrication and electrochemical properties of three-dimensional net architectures of anatase TiO<sub>2</sub> and spinel Li<sub>4</sub>Ti<sub>5</sub>O<sub>12</sub> nanofibers. Journal of Power Sources, 164, 874-879.
- Mukherjee, R., Krishnanb, R., Lu, T.M., and Koratkar, N. (2012) Nanostructured electrodes for high-power lithium ion batteries. Nano Energy, 1(4), 518-533.

- Nam, S.H., Shim, H.S., Kim, Y.S., Dar, M.A., Kim, J.G., and Kim, W.B. (2010) Ag or Au Nanoparticle-embedded one-dimensional composite TiO<sub>2</sub> nanofibers prepared via electrospinning for use in lithium-ion batteries. ACS applied materials & interfaces, 2(7), 2046-2052.
- Ning, G., Haran, B., and Popov, B.N. (2003) Capacity fade study of lithium-ion batteries cycled at high discharge rates. Journal of Power Sources, 117, 160–169.
- Niu, H., Zhang, J., Xie, Z., Wang, X., and Lin, T. (2011) Preparation, structure and supercapacitance of bonded carbon nanofiber electrode materials. Cabon, 49, 2380-2388.
- Nwanya, A.C., Ugwuoke, P.E., Ezekoye, B.A., Osuji, R.U. and Ezema, F.I. (2013) Structural and optical properties of chemical bath deposited silver oxide thin films: role of deposition time. Advances in Materials Science and Engineering, 2013, 1-8.
- Reddy, M.V., Jose, R., Teng, T.H., Chowdari, B.V.R., and Ramakrishna, S. (2010) Preparation and electrochemical studies of electrospun TiO<sub>2</sub> nanofibers and molten salt method nanoparticles. Electrochimica Acta, 55, 3109-3117.
- Riman, R.E., Suchanek, W.L., and Lencka, M.M. (2002) Hydrothermal crystallization of ceramics. Annales de Chimie Science des Matériaux, 27(6), 15-36.
- Song, M.K., Park S., Alamgir F.M., Cho, J., and Liu M. (2011) Nanostructured electrodes for lithium-ion and lithium-air batteries: the latest developments, challenges, and perspectives. Materials Science and Engineering R, 72, 203-252.
- Suresh, K.P., Dhayal, R.A., Mangalaraj, D., and Nataraj, D. (2008) Growth and characterization of ZnO nanostructured thin films by a two step chemical method. Applied Surface Science, 255, 2382-2387.
- Tasaki, K., Goldberg, A., Lian, J.J., Walker, M., Timmons, A., and Harris, S.J (2009) Solubility of lithium salts formed on the lithium-ion battery and negative electrode surface in organic solvents. Journal of the Electrochemical Society, 156(12), A1019-A1027.

- Verma, P., Mairel, P., and Novak P. (2011) A review of the features and analyses of the solid electrolyte interphase in Li-ion batteries. Electrochimica Acta, 55, 6332-6341.
- Wakihara, M. (2001) Recent development in lithium ion batteries. Materials Science and Engineering, R33, 109-134.
- Wang, L., Ding, C.X., Zhang, L.C., Xu, H.W., Zhang, D.W., Cheng, T., and Chen, C.H. (2010) A novel carbon-silicon composite nanofiber prepared via electrospinning as anode material for high energy-density lithium ion batteries. Journal of Power Sources, 195, 5052-5056.
- Wang, L., Yu, Y., Chen, P.C., and Chen, C.H. (2008) Electrospun carbon-cobalt composite nanofiber as an anode material for lithium ion batteries. Scripta Materialia, 58, 405-408.
- Wang, L., Yu, Y., Chen, P.C., Zhang, D.W., and Chen, C.H. (2008) Electrospinning synthesis of C/Fe<sub>3</sub>O<sub>4</sub> composite nanofibers and their application for high performance lithium-ion batteries. Journal of Power Sources, 183, 717-723.
- Wang, Z., Zhou, L., and Lou, X.W. (2012) Metal oxide hollow nanostructures for lithium-ion batteries. Advanced materials, 24, 1903-1911.
- Wattanaarun, J., Pavrajarn, V., and Supaphol, P. (2005) Titanium (IV) oxide nanofibers by combined sol-gel and electrospinning techniques: preliminary report on effects of preparation conditions and secondary metal dopant. Science and Technology of Advanced Materials, 6, 240-245.
- Wu, C.H. (2004) Comparison of azo dye degradation efficiency using UV/single semiconductor and UV/coupled semiconductor systems. Chemosphere, 57, 601-608.
- Yang, X., Teng, D., Liu, B., Yu, Y., and Yang, X. (2011) Nanosized anatase titanium dioxide loaded porous carbon nanofiber webs as anode materials for lithium-ion batteries. Electrochemistry Communications, 13, 1098-1101.
- Yang, Z., Du, G., Feng, C., Li, S., Chen, Z., Zhang, P., Guoa, Z., Yu, X., Chen, G., Huang, S., and Liu, H. (2010) Synthesis of uniform polycrystalline tin dioxide nanofibers and electrochemical application in lithium-ion batteries. Electrochimica Acta, 55, 5485-5491.

- Yu, Y., Yang, Q., Teng, D., Yang, X., and Ryu, S. (2010) Reticular Sn nanoparticle-dispersed PAN-based carbon nanofibers for anode material in rechargeable lithium-ion batteries. Electrochemistry Communications, 12, 1187–1190.
- Yuan, T., Zhao, B., Cai, R., Zhou, Y., and Shao, Z. (2011) Electrospinning based fabrication and performance improvement of film electrodes for lithium-ion batteries composed of TiO<sub>2</sub> hollow fibers. Journal of Materials Chemistry, 21, 15041.
- Zhu, P., Wu, Y., Reddy, M.V., Nair, A.S., Chowdari, B.V.R., and Ramakrishna, S. (2012) Long term cycling studies of electrospun TiO<sub>2</sub> nanostructures and their composites with MWCNTs for rechargeable Li-ion batteries. RSC Advances, 2, 531-537.
- Ziabari, M., Mottaghitalab, V., and Haghi, A.K. (2009) Application of direct tracking method of measuring electrospun nanofiber diameter. Brazilian Journal of Chemical Engineering, 26(1), 53-62.

## APPENDIX

### Appendix A Experimental Data of BET Surface Area Analysis

Ideally five data points from the isotherm, with a minimum of three data points, in the  $P/P_0$  range 0.025 to 0.30 should be used to successfully determine the surface area using the BET equation. The computer program takes over and a least-squares linear regression is used to fit the best straight line through a transformed data set consisting of the following pairs of values:  $1 / [ W((P_0/P) - 1) ]$  and  $P/P_0$ . The total surface area can be calculated using the following equation.

The monolayer capacity,  $V_m$ , is calculated from the slope,  $s$ , and the intercept,  $i$ , of the straight line which can be obtained using least squares regression.

$$V_m = \frac{1}{s + i}$$

Once  $X_m$  is determined, the total surface area  $S_t$  can be calculated with the following equation.

$$S_t = \frac{V_m L_{AV} A_m}{M_v}$$

Where  $L_{av}$  is Avogadro's number and equals  $6.02 \times 10^{23}$ ,  $A_m$  is the cross sectional area of the adsorbate and equals  $0.162 \text{ nm}^2$  for an adsorbed nitrogen molecule, and  $M_v$  is the molar volume and equals 22414 mL. All surface area results are finally reported normalized by sample weight, or mass, as square meters per gram, written  $\text{m}^2/\text{g}$  or  $\text{m}^2\text{g}^{-1}$ .

**Table A1** Raw data of multi-point BET of TiO<sub>2</sub> fibers

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
4.98112e-02	3.9903	1.0512e+01
1.00416e-01	5.2218	1.7104e+01
1.50344e-01	6.5068	2.1759e+01
2.00353e-01	7.4241	2.7003e+01
2.50309e-01	8.2890	3.2229e+01
3.00271e-01	9.0302	3.8023e+01

Slope	107.414
Intercept	5.614
Correlation coefficient, r	0.999195
C constant	20.134

**Table A2** Raw data of multi-point BET of TiO<sub>2</sub> hollow fibers

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
4.96553e-02	4.6058	9.0769e+00
1.00324e-01	6.3354	1.4083e+01
1.50328e-01	7.7396	1.8291e+01
2.00255e-01	9.2026	2.1771e+01
2.50210e-01	10.5306	2.5355e+01
3.00294e-01	11.0683	3.1024e+01

Slope	83.894
Intercept	5.237
Correlation coefficient, r	0.997304
C constant	17.019

**Table A3** Raw data of multi-point BET of calcined Zn/TiO<sub>2</sub> composite hollow fibers

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.02964e-02	8.5101	4.9793e+00
1.00064e-01	10.2497	8.6798e+00
1.50028e-01	11.9278	1.1840e+01
2.00014e-01	13.4326	1.4893e+01
2.50052e-01	14.7656	1.8068e+01
2.99966e-01	16.0970	2.1299e+01

Slope	64.528
Intercept	1.996
Correlation coefficient, r	0.999577
C constant	33.323

**Table A4** Raw data of multi-point BET of ZnO-TiO<sub>2</sub> composite hollow fibers at 115 °C 1 h

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.02237e-02	0.9421	4.4909e+01
1.00349e-01	1.6241	5.4951e+01
1.50304e-01	2.2411	6.3154e+01
2.00239e-01	2.8888	6.9346e+01
2.50133e-01	3.5504	7.5174e+01
3.00029e-01	4.2820	8.0092e+01

Slope	138.877
Intercept	4.027e+01
Correlation coefficient, r	0.991158
C constant	4.449



**Table A5** Raw data of multi-point BET of ZnO-TiO<sub>2</sub> composite hollow fibers at 115 °C 0.75 h

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.11496e-02	1.3029	3.3106e+01
1.00923e-01	2.1646	4.1492e+01
1.50896e-01	3.0260	4.6990e+01
2.00840e-01	3.8336	5.2453e+01
2.50899e-01	4.6178	5.8034e+01
3.00824e-01	5.2209	6.5938e+01

Slope	125.405
Intercept	2.761e+01
Correlation coefficient, r	0.997014
C constant	5.542

**Table A6** Raw data of multi-point BET of ZnO-TiO<sub>2</sub> composite hollow fibers at 115 °C 0.5 h

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.08183e-02	1.6215	2.6418e+01
1.00621e-01	2.5440	3.5187e+01
1.50649e-01	3.3939	4.1816e+01
2.00633e-01	4.2630	4.7108e+01
2.50577e-01	5.1710	5.1736e+01
3.00458e-01	6.1502	5.5878e+01

Slope	115.692
Intercept	2.271e+01
Correlation coefficient, r	0.989745
C constant	6.095

**Table A7** Raw data of multi-point BET of calcined Ag/TiO<sub>2</sub> composite hollow fibers

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
4.95494e-02	6.2422	6.6823e+00
1.00403e-01	7.7345	1.1546e+01
1.50459e-01	8.8637	1.5987e+01
2.00377e-01	9.8586	2.0338e+01
2.50334e-01	10.7297	2.4901e+01
3.00318e-01	11.3185	3.0342e+01

Slope	92.792
Intercept	2.039
Correlation coefficient, r	0.999406
C constant	46.517

**Table A8** Raw data of multi-point BET of Ag<sub>2</sub>O-TiO<sub>2</sub> composite hollow fibers at 110 °C 1 h

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.02959e-02	1.3926	3.0427e+01
1.00315e-01	2.5547	3.4922e+01
1.50235e-01	3.7525	3.7697e+01
2.00085e-01	5.2237	3.8314e+01
2.50118e-01	6.4059	4.1661e+01
2.99985e-01	7.9532	4.3113e+01

Slope	48.217
Intercept	2.924e+01
Correlation coefficient, r	0.978400
C constant	2.649

**Table A9** Raw data of multi-point BET of Ag<sub>2</sub>O-TiO<sub>2</sub> composite hollow fibers at 115 °C 1 h

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.06183e-02	1.1475	3.7177e+01
1.00252e-01	2.3419	3.8069e+01
1.50237e-01	3.4730	4.0731e+01
2.00285e-01	4.5120	4.4413e+01
2.50195e-01	5.7250	4.6635e+01
3.00216e-01	6.8014	5.0469e+01

Slope	54.842
Intercept	3.330e+01
Correlation coefficient, r	0.988545
C constant	2.647

**Table A10** Raw data of multi-point BET of Ag<sub>2</sub>O-TiO<sub>2</sub> composite hollow fibers at 120 °C 1 h

Relative pressure [P/P <sub>0</sub> ]	Volume @ STP [cc/g]	1 / [ W((P <sub>0</sub> /P) - 1) ]
5.05097e-02	0.7749	5.4929e+01
1.00515e-01	1.3752	6.5017e+01
1.50471e-01	1.8850	7.5183e+01
2.00451e-01	2.2950	8.7404e+01
2.50373e-01	2.8035	9.5323e+01
3.00357e-01	3.1184	1.1015e+02

Slope	216.860
Intercept	4.329e+01
Correlation coefficient, r	0.997689
C constant	6.010

## CURRICULUM VITAE

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**Presentations:**

1. Sirimekanont, T.; Supaphol, P.; and Sombatmankhong, K. (2013, March 11-15) Novel electrospun nanofibrous materials as anode in lithium-ion batteries. Paper presented at POLYCHAR 21 World Forum on Advance Materials, Gwangju, Republic of Korea.
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