

REFERENCES

- Anandan, S. (2007) Recent improvements and arising challenges in dye-sensitized solar cells. Solar Energy Materials and Solar Cells, 91(9), 843-846.
- Calzaferri, G., Lanz, M., and Li, J.W. (1995) Methyl viologen-zeolite electrodes: Intrazeolite charge transfer. Journal of the Chemical Society, Chemical - Communications(13), 1313-1314.
- Carpiné, D., Dagostin, J.L. A., Silva, V.R., Igarashi-Mafra, L., and Mafra, M.R. (2013) Adsorption of volatile aroma compound 2-phenylethanol from synthetic solution onto granular activated carbon in batch and continuous modes. Journal of Food Engineering, 117, 370-377.
- Chang, B.Y.S., Huang, N.M., An'amt, M.N., Marlinda, A.R., Norazrienã, Y., Muhamad, M.R., Harrison, I., Lim, H.N., and Chia, C.H. (2012) Facile hydrothermal preparation of titanium dioxide decorated reduced graphene oxide nanocomposite. International Journal of Nanomedicine, 7, 3379-3387.
- Chang, H. and Lo, Y.J. (2010) Pomegranate leaves and mulberry fruit as natural sensitizers for dye-sensitized solar cells. Solar Energy, 84(10), 1833-1837.
- Chethankumar, M. (2011) Role of lutein in inhibiting endogenous oxidative DNA damage in lymphocytes with the occurrence of age-related macular degeneration. International of Pharmaceutical Science Review and Research, 7, 59-65.
- Choi, H., Nahm, C., Kim, J., Kim, C., Kang, S., Hwang, T., and Park, B. (2013) Review paper: Toward highly efficient quantum-dot- and dye-sensitized solar cells. Current Applied Physics, 13, S2-S13.
- Chou, T.P., Zhang, Q.F., and Cao, G.Z. (2007) Effects of dye loading conditions on the energy conversion efficiency of ZnO and TiO₂ dye-sensitized solar cells. Journal of physical Chemistry C, 111(50), 18804-18811.
- Davies, K.M. (2004). Plant Pigments and their Manipulation. Oxford: Blackwell (pp.342).
- El-Bindary, A.A., Diab, M.A., Hussien, M.A., El-Sonbati, A.Z., and Eessa, A.M. (2014) Adsorption of Acid Red 57 from aqueous solutions onto

- polyacrylonitrile/activated carbon composite. Spectrochimica Acta. Part A: Molecular and Biomolecular Spectroscopy, 124C, 70-77.
- Gorer, S., Albu-Yaron, A., and Hodes, G. (1995) Quantum size effects in chemically deposited, nanocrystalline lead selenide films. Journal of Physical Chemistry, 99(44), 16442-16448.
- Grätzel, M. (1991) The artificial leaf, molecular photovoltaics achieve efficient generation of electricity from sunlight. Coordination Chemistry Reviews, 111, 167-174.
- Grätzel, M. (2003) Dye-sensitized solar cells. Journal of Photochemistry and Photobiology C: Photochemistry Reviews, 4(2), 145-153.
- Grätzel, M. (2005) Solar energy conversion by dye-sensitized photovoltaic cells. Inorganic Chemistry, 44(20), 6841-6851.
- Hao, S., Wu, J., Huang, Y., and Lin, J. (2006) Natural dyes as photosensitizers for dye-sensitized solar cell. Solar Energy, 80(2), 209-216.
- Jacquemin, D., Preat, J., Wathélet, V., and Perpète, E.A. (2006) Substitution and chemical environment effects on the absorption spectrum of indigo. The Journal of Chemical Physics, 124(7), (12 pages).
- Ji, G., Liu, Z., Guan, D., and Yang, Y. (2013) Ag₂S quantum dots and N3 dye co-sensitized TiO₂ nanotube arrays for a solar cell. Applied Surface Science, 282, 695-699.
- Kay, A. and Grätzel, M. (1996) Low cost photovoltaic modules based on dye sensitized nanocrystalline titanium dioxide and carbon powder. Solar Energy Materials and Solar Cells, 44(1), 99-117.
- Kebede, Z. and Lindquist, S. E. (1999) Donor-acceptor interaction between non-aqueous solvents and I₂ to generate I₃⁻, and its implication in dye sensitized solar cells. Solar Energy Materials and Solar Cells, 57(3), 259-275.
- Klimov, V.I. (2010). Nanocrystal Quantum Dots Boca Raton, FL: (pp.466)
- Lee, D.K., Cho, D.H., Lee, J.H., and Shin, H.Y. (2008) Fabrication of nontoxic natural dye from sappan wood. Korean Journal of Chemical Engineering, 25(2), 354-358.

- Li, P., Wu, J., Lin, J., Huang, M., Huang, Y., and Li, Q. (2009) High-performance and low platinum loading Pt/Carbon black counter electrode for dye-sensitized solar cells. Solar Energy, 83(6), 845-849.
- Liao, J.Y., and Ho, K.C. (2005) A photovoltaic cell incorporating a dye-sensitized ZnS/ZnO composite thin film and a hole injecting PEDOT layer. Solar Energy Materials and Solar Cells, 86, 229-241.
- Ludin, N.A., Al-Alwani Mahmoud, A.M., Bakar Mohamad, A., Kadhum, A.A., Sopian, K., and Abdul Karim, N.S. (2014) Review on the development of natural dye photosensitizer for dye-sensitized solar cells. Renewable and Sustainable Energy Reviews, 31, 386-396.
- Luque, A. and Hegedus, S. (2003). Handbook of Photovoltaic Science and Engineering. Hoboken, NJ: Wiley.
- Monari, A., Assfeld, X., Beley, M., and Gros, P.C. (2011) Theoretical study of new ruthenium-based dyes for dye-sensitized solar cells. Journal of Physical Chemistry A, 115(15), 3596-3603.
- Narayan, M.R. (2011) Review: Dye sensitized solar cells based on natural photosensitizers. Renewable and Sustainable Energy Reviews, 16(1), 208-215.
- Nazeeruddin, M.K., Humphry-Baker, R., Liska, P. and Grätzel, M. (2003) Investigation of sensitizer adsorption and the influence of protons on current and voltage of a dye-sensitized nanocrystalline TiO₂ solar cell. Journal of Physical Chemistry B, 107(34), 8981-8987.
- Nelson, J. and Chandler, R.E. (2004) Random walk models of charge-transfer and transport in dye sensitized systems. Coordination Chemistry Reviews, 248(13-14), 1181-1194.
- Osterloh, F.E. (2008) Inorganic materials as catalysts for photochemical splitting of water. Chemistry of Materials, 20(1), 35-54.
- Pathan, H.M., Salunkhe, P.V., Sankapal, B.R., and Lokhande, C.D. (2001) Photoelectrochemical investigation of Ag₂S thin films deposited by SILAR method. Materials Chemistry and Physics, 72(1), 105-108.
- Raghavarao, G. (2007) Aqueous two phase extraction for purification of C-phycoyanin. Biochemical Engineering Journal, 34, 154-164.

- Rani, G., and Sahare, P. D. (2013) Study of the structural and morphological changes during the phase transition of ZnS to ZnO. Applied Physics A, 21, 345-349.
- Ruhle, S., Shalom, M., and Zaban, A. (2010) Quantum-dot-sensitized solar cells. Chemphyschem, 11(11), 2290-2304.
- Ryan, M. (2009) Progress in ruthenium complexes for dye sensitised solar cells. Platinum Metals Review, 53, 216-218.
- Saito, M., and Fujihara, H. (2009) Fabrication and photovoltaic properties of dye-⁻ sensitized ZnO thick films by a facile doctor-blade printing method using nanocrystalline pastes. Journal of the Ceramic Society of Japan, 117(7), 823-827.
- Sambur, J.B., Novet, T., and Parkinson, B.A. (2010) Multiple exciton collection in a sensitized photovoltaic system. Science, 330(6000), 63-66.
- Satyanarayana, J., Murthy, G.S., and Sasidhar, P. (1999) Adsorption studies of caesium on zirconium molybdoarsenate (ZrMAs). Waste Management, 19(6), 427-432.
- Serpone, N., Lawless, D., Disdier, J., and Herrmann, J.M. (1994) Spectroscopic, photoconductivity, and photocatalytic studies of TiO₂ colloids. Naked and with the lattice doped with Cr³⁺, Fe³⁺, and V⁵⁺ cations. Langmuir, 10(3), 643-652.
- Smestad, G.P. (1998) Education and solar conversion: Demonstrating electron transfer. Solar Energy Materials and Solar Cells, 55(1-2), 157-178.
- Subash, B., Krishnakumar, B., Swaminathan, M., and Shanthi, M. (2013) Ag₂S-ZnO-⁻an efficient photocatalyst for the mineralization of Acid Black 1 with UV light. Spectrochimica Acta. Part A: Molecular and Biomolecular Spectroscopy, 105, 314-319.
- Tatsuzawa, F., Saito, N., Seki, H., Yokoi, M., Yukawa, T., Shinoda, K., and Honda, T. (2004) Acylated anthocyanins in the flowers of *Vanda* (Orchidaceae). Biochemical Systematics and Ecology, 32, 651-664.
- Tributsch, H. and Gerischer, H. (1969) The use of semiconductor electrodes in the study of photochemical reactions. Berichte der Bunsen-Gesellschaft für Physikalische Chemie, 73, 850 – 854.

- Tripathi, M., Upadhyay, R., Pandey, A., and Dubey, P.K. (2012) Natural dye-based photoelectrode for improvement of solar cell performance. Ionics, 19(8), 1179-1183.
- Wongcharee, K., Meeyoo, V., and Chavadej, S. (2007) Dye-sensitized solar cell using natural dyes extracted from rosella and blue pea flowers. Solar Energy Materials and Solar Cells, 91(7), 566-571.
- Ye, T.X., Ye, S.L., Chen, D.M., Chen, Q.A., Qiu, B., and Chen, X. (2012) Spectroscopic characterization of tetracationic porphyrins and their noncovalent functionalization with graphene. Spectrochimica Acta. Part A: Molecular and Biomolecular Spectroscopy, 86, 467-471.
- Zou, Z., Xie, C., Zhang, S., Yang, C., Zhang, G., and Yang, L. (2013) CdS/ZnO nanocomposite film and its enhanced photoelectric response to UV and visible lights at low bias. Sensors and Actuators B: Chemical, 188, 1158-1166.

APPENDICES

Appendix A Extinction Coefficient of Natural Dyes

Table A1 Extinction coefficient of natural dyes

Natural dyes	Extinction coefficient ($\text{L g}^{-1} \text{cm}^{-1}$)
Indigo	0.078
Red orchid	0.076
Spirulina	0.107
Yellow cotton flower	0.098

From the Beer-Lambert Law relationship for absorbance of light:

$$A = \epsilon cl,$$

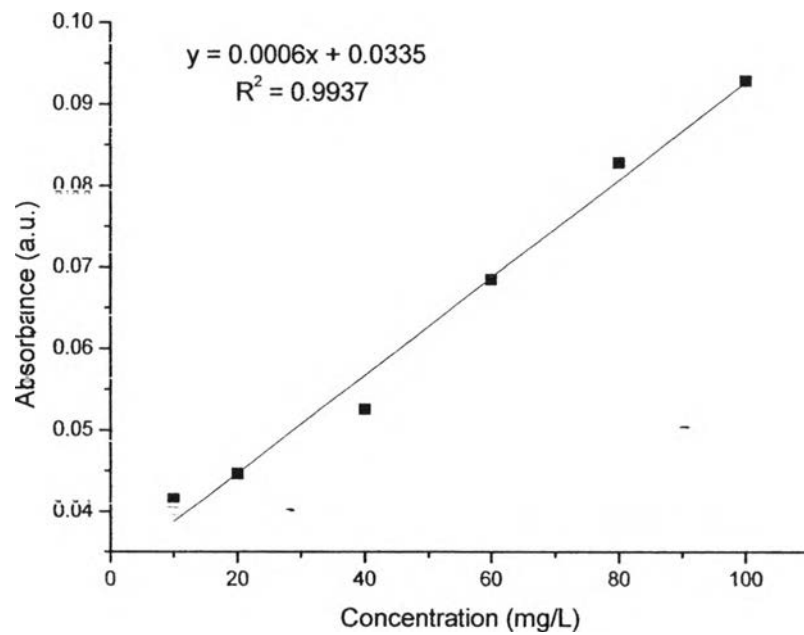
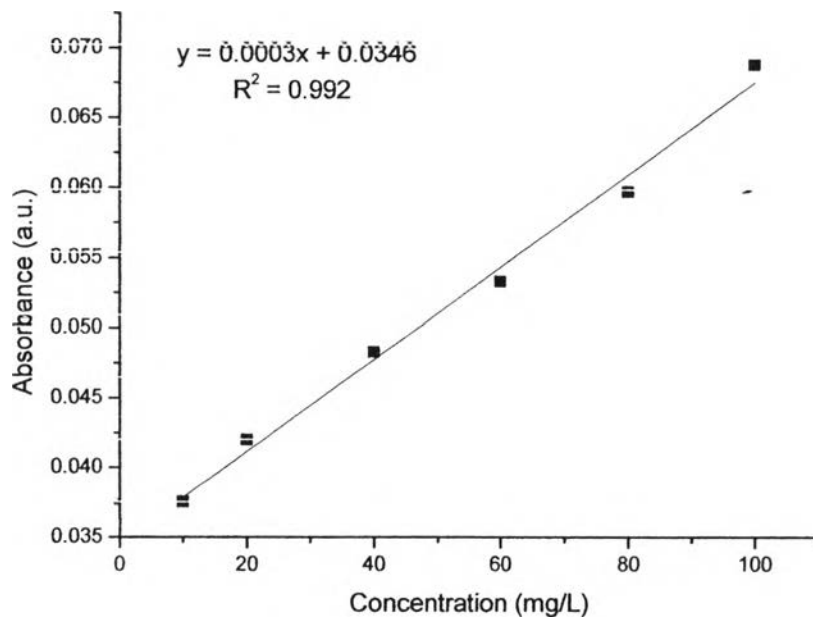
Where A = absorbance

ϵ = extinction coefficient ($\text{L g}^{-1} \text{cm}^{-1}$)

c = concentration of the dye (g/L)

l = path length of the cell, which is normally 1cm

The absorbance of each dye was obtained using UV-Visible spectrometer and the concentration of dye equaled to 2.5 g/L then put all values in Beer-Lambert equation to get the extinction coefficient ($\text{L g}^{-1} \text{cm}^{-1}$)

Appendix B Calibration Curve of Kinetic Adsorption Studies**Figure B1** Standard calibration curve of indigo extract.**Figure B2** Standard calibration curve of red orchid extract.

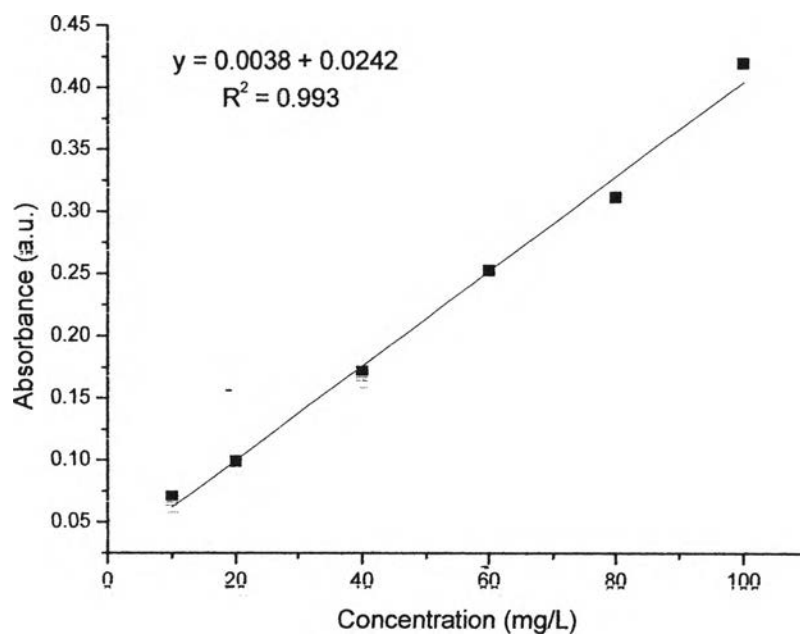


Figure B3 Standard calibration curve of spirulina extract.

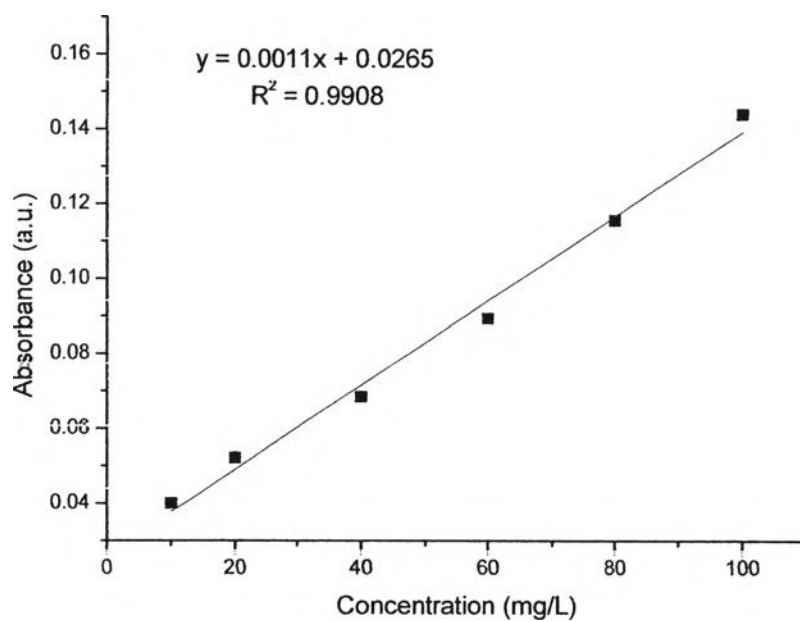


Figure B4 Standard calibration curve of yellow cotton extract.

Appendix C Calibration Curve of Isothermal Adsorption Studies

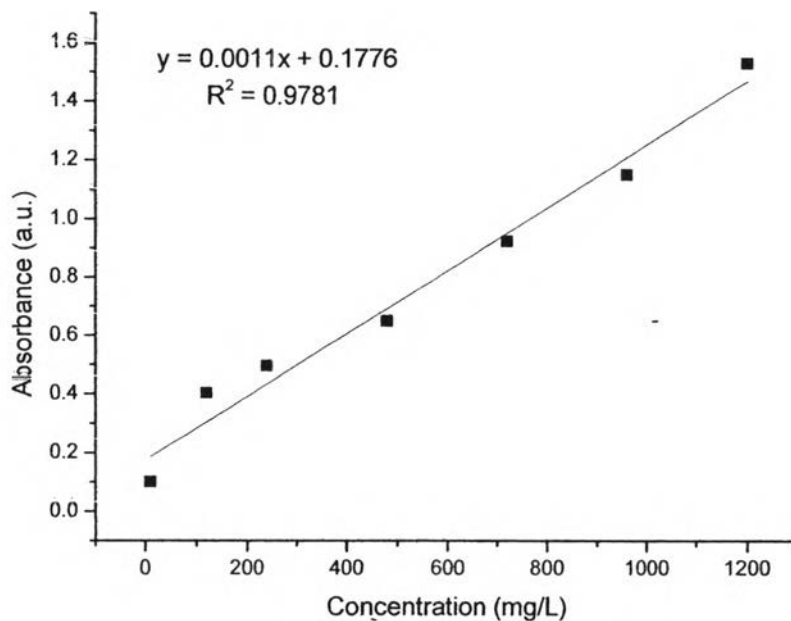


Figure C1 Standard calibration curve of indigo extract.

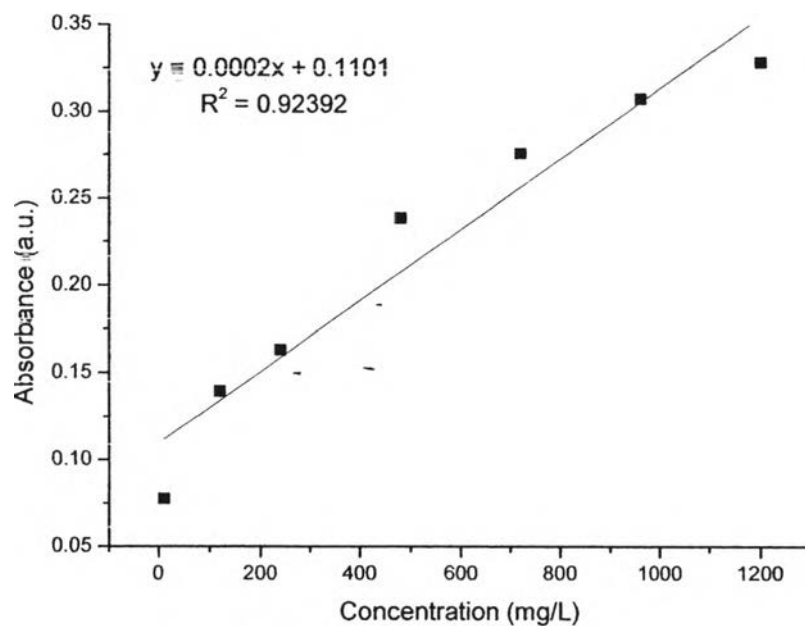


Figure C2 Standard calibration curve of red orchid extract.

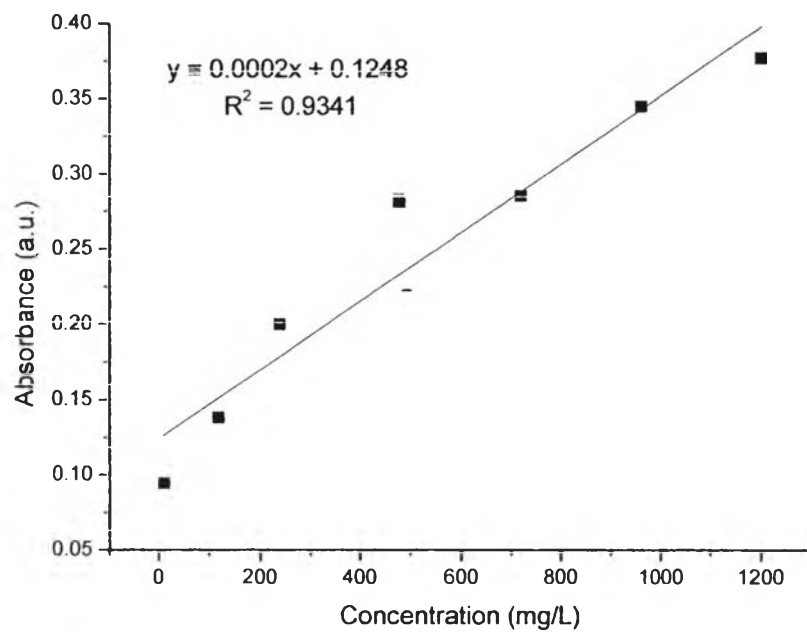


Figure C3 Standard calibration curve of spirulina extract.

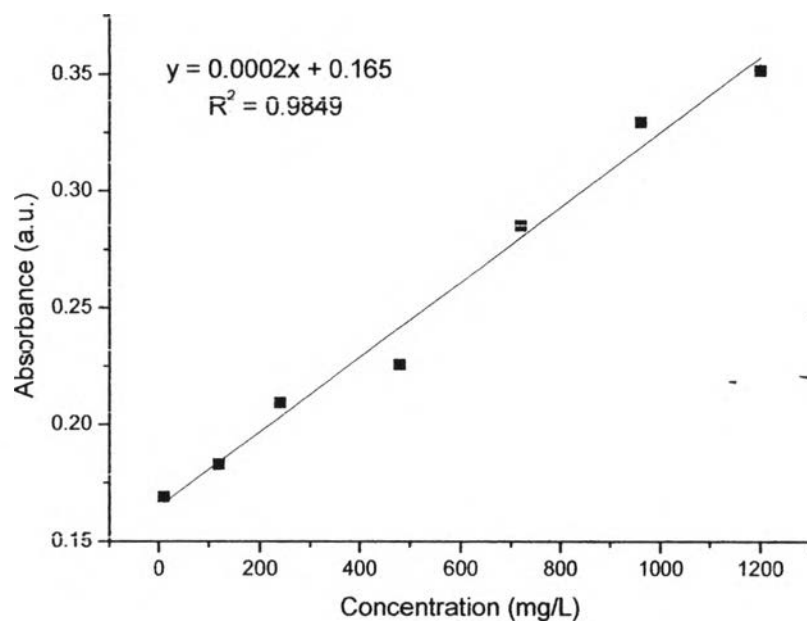


Figure C4 Standard calibration curve of yellow cotton extract.

Appendix D FE-SEM Cross-section Images with Particle Size Analysis of ZnO and ZnO/QDs

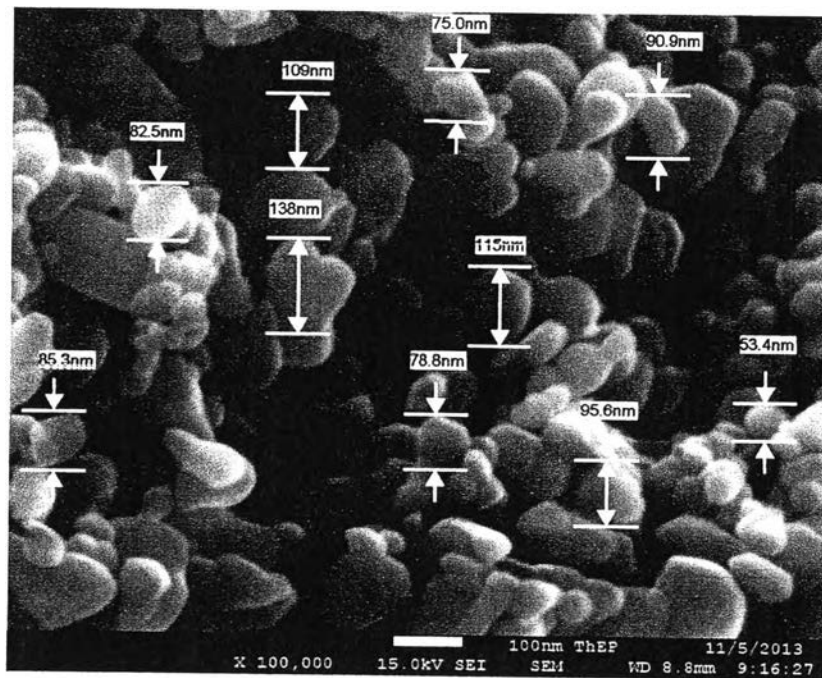


Figure D1 ZnO.

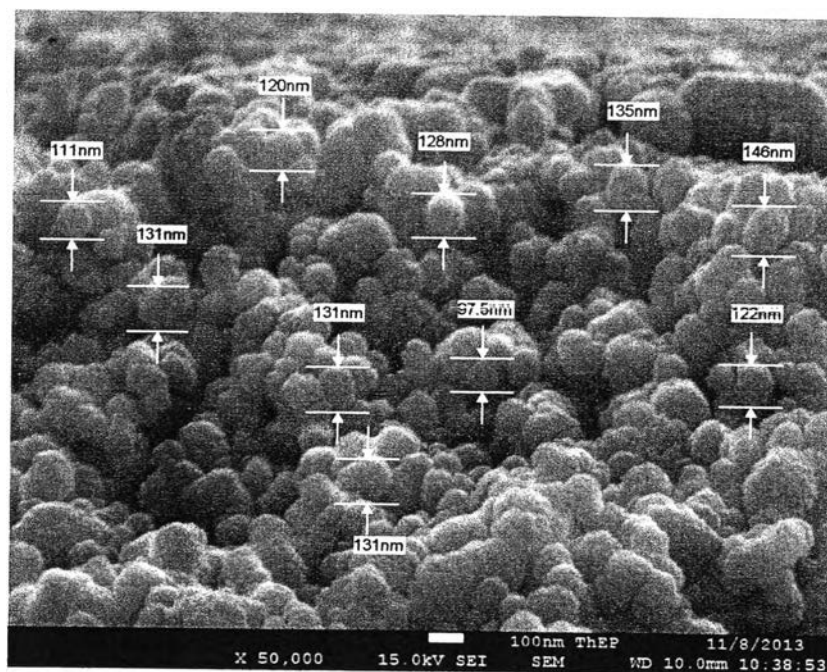


Figure D2 ZnO/CdS QD at 1 min.

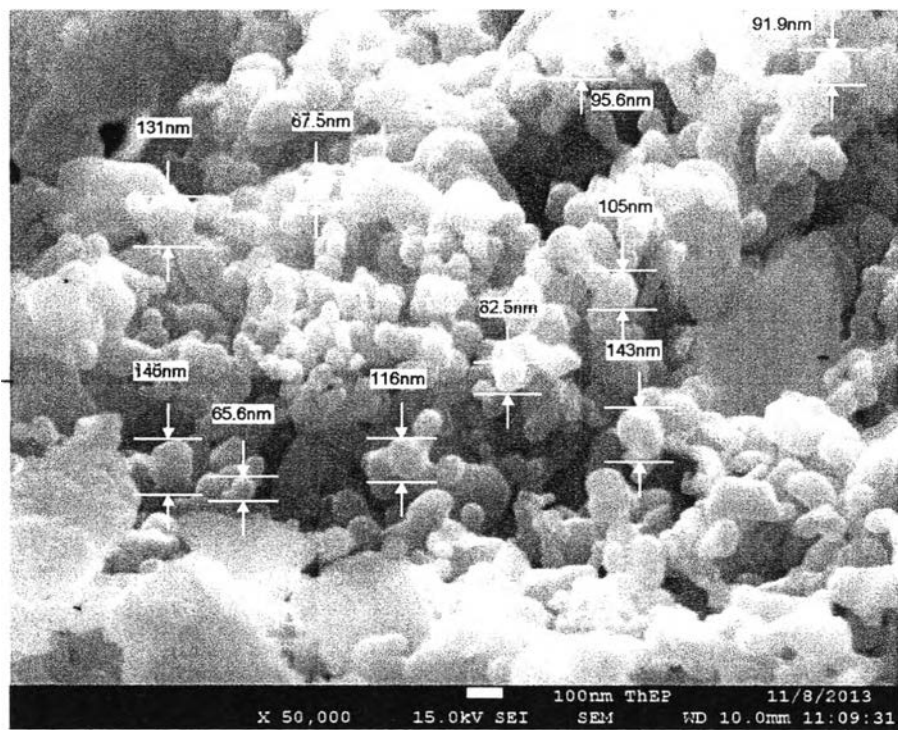


Figure D3 ZnO/CdS QD at 3 min.

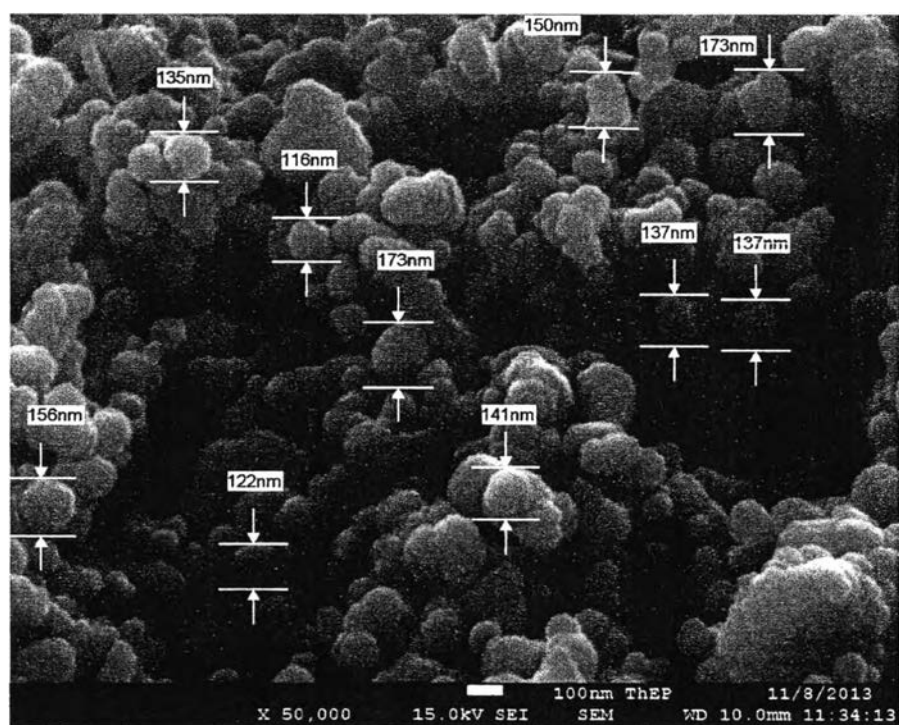


Figure D4 ZnO/CdS QD at 5 min.

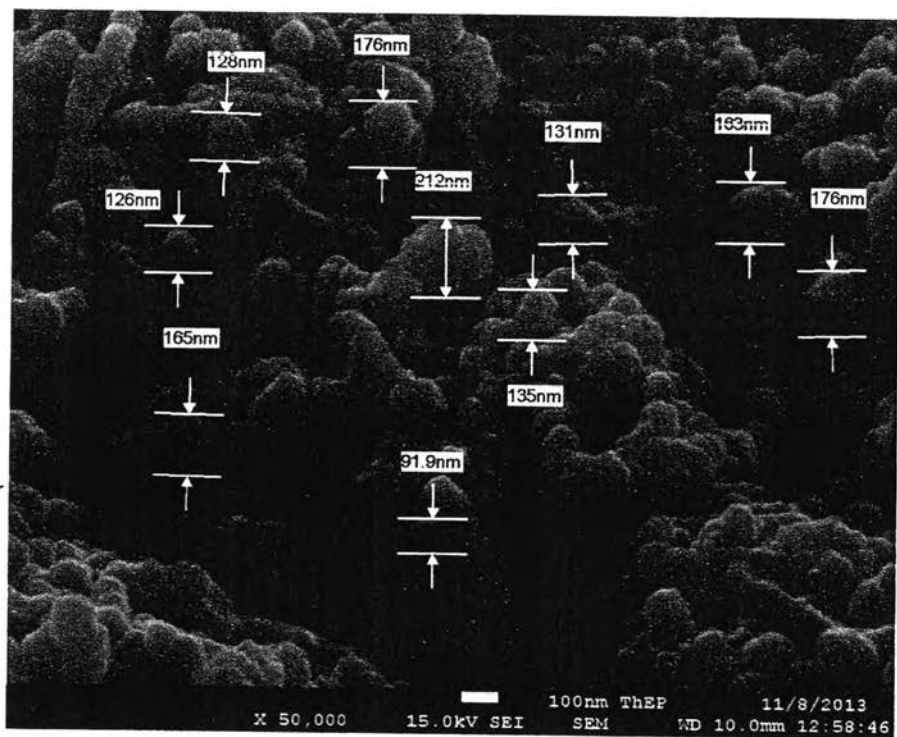


Figure D5 ZnO/CdS QD at 7 min.

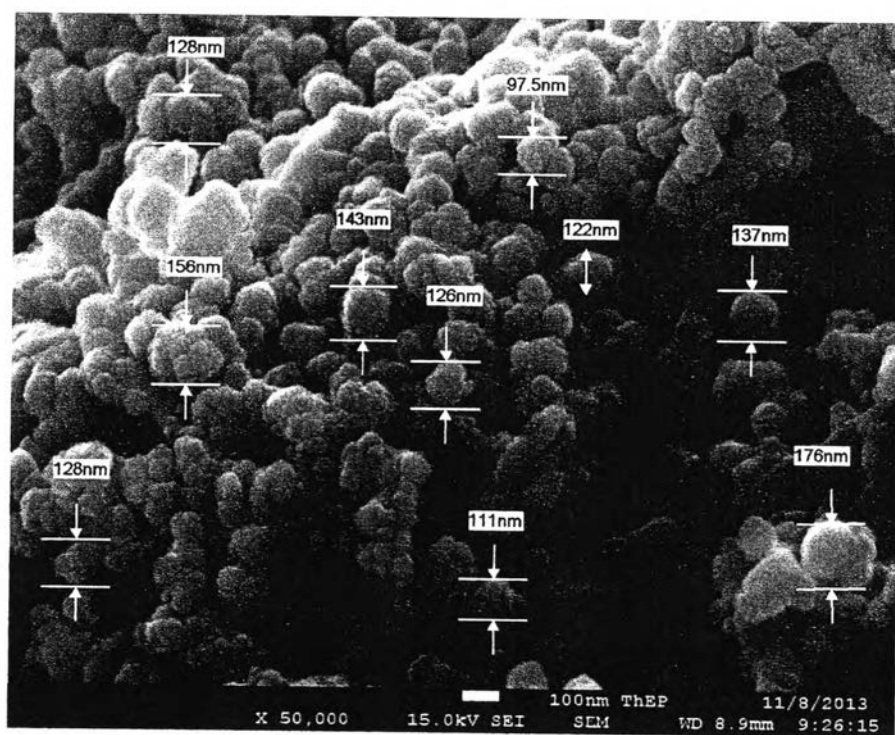


Figure D6 ZnO/CdS QD at 9 min.

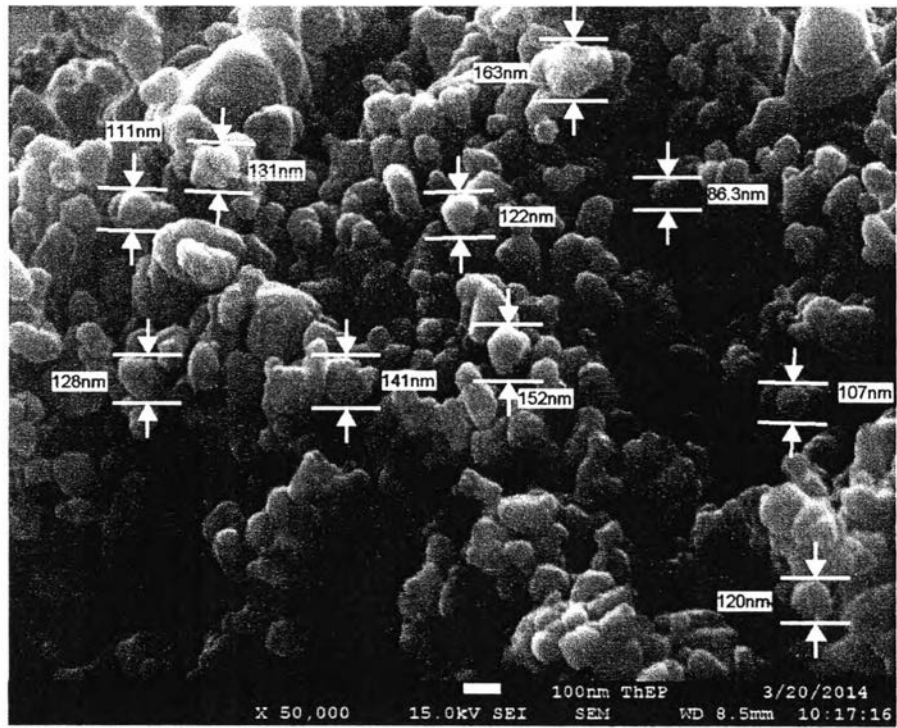


Figure D7 ZnO/ZnS QD at 1 min.

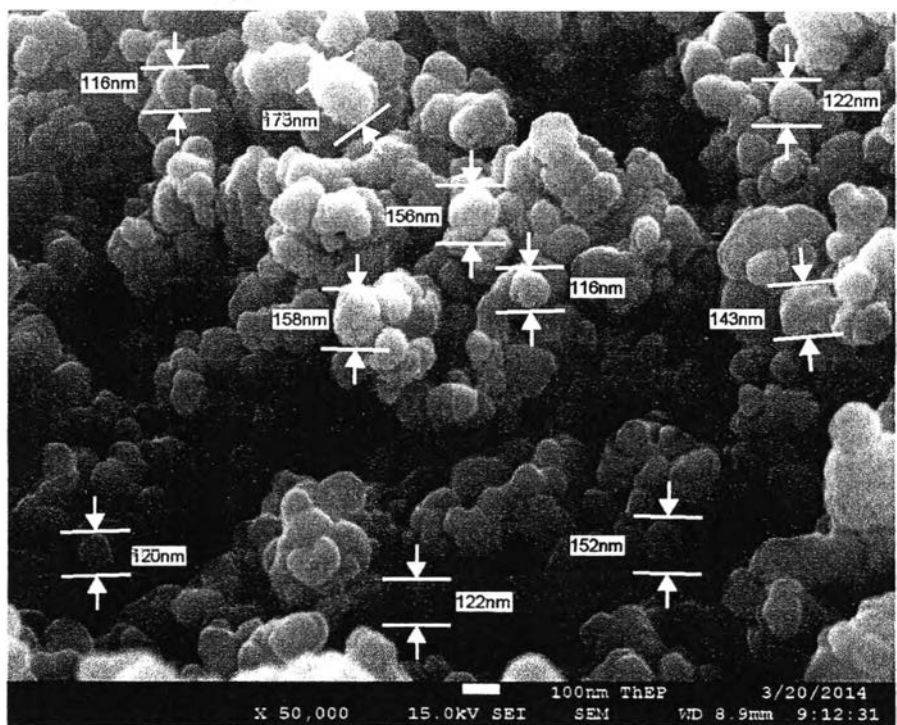


Figure D8 ZnO/ZnS QD at 3 min.

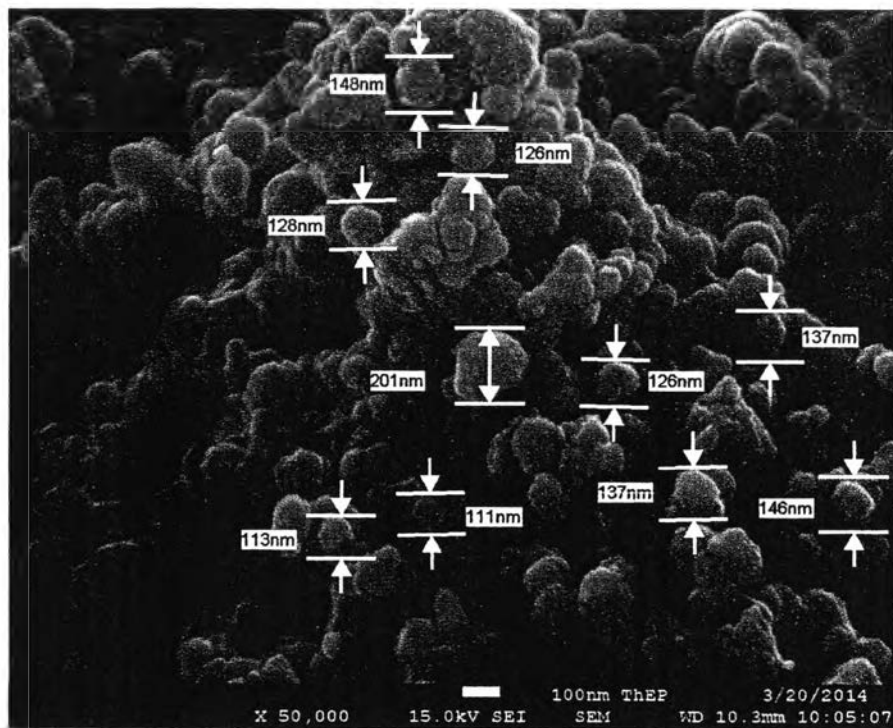


Figure D9 ZnO/ZnS QD at 5 min.

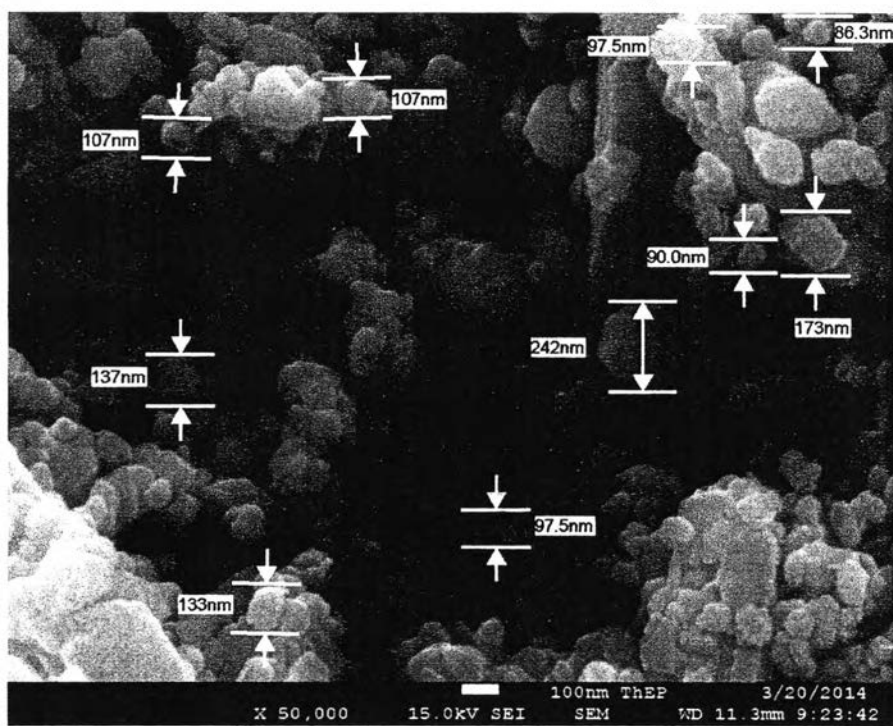


Figure D10 ZnO/ZnS QD at 7 min.

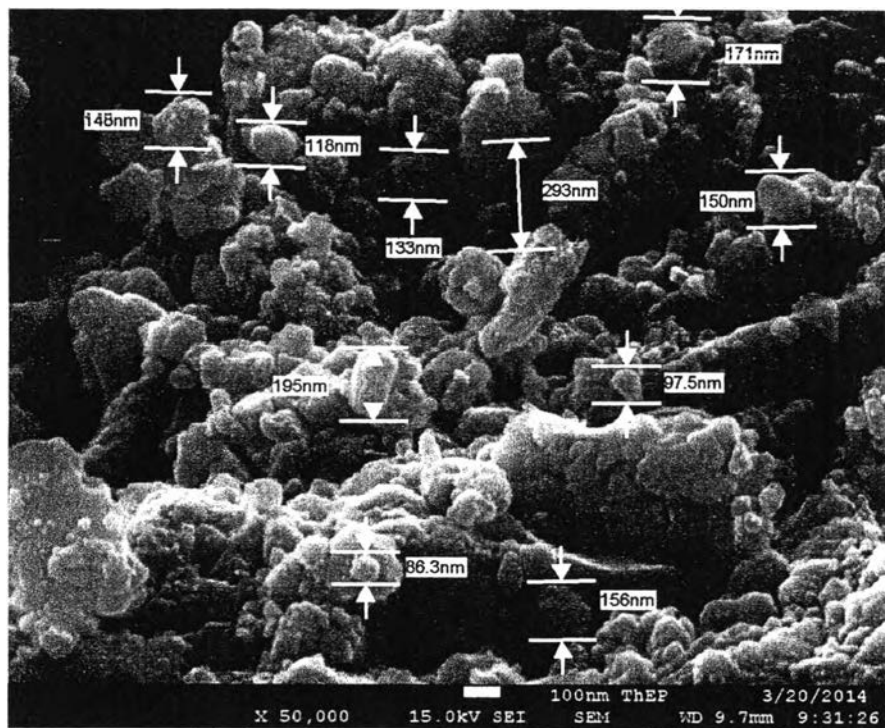


Figure D11 ZnO/ZnS QD at 9 min.

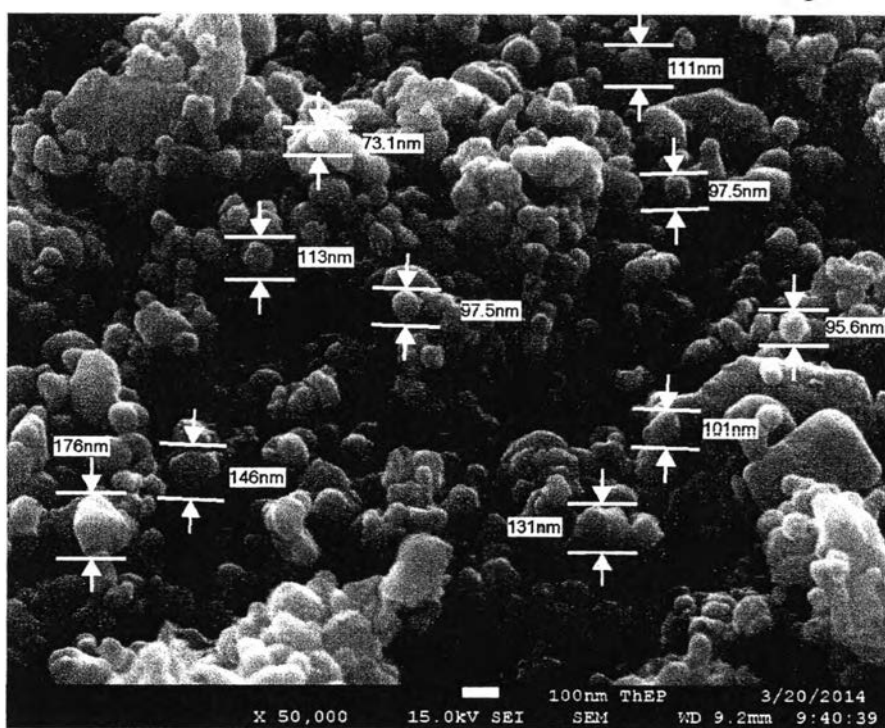


Figure D12 ZnO/Ag₂S QD at 1 min.

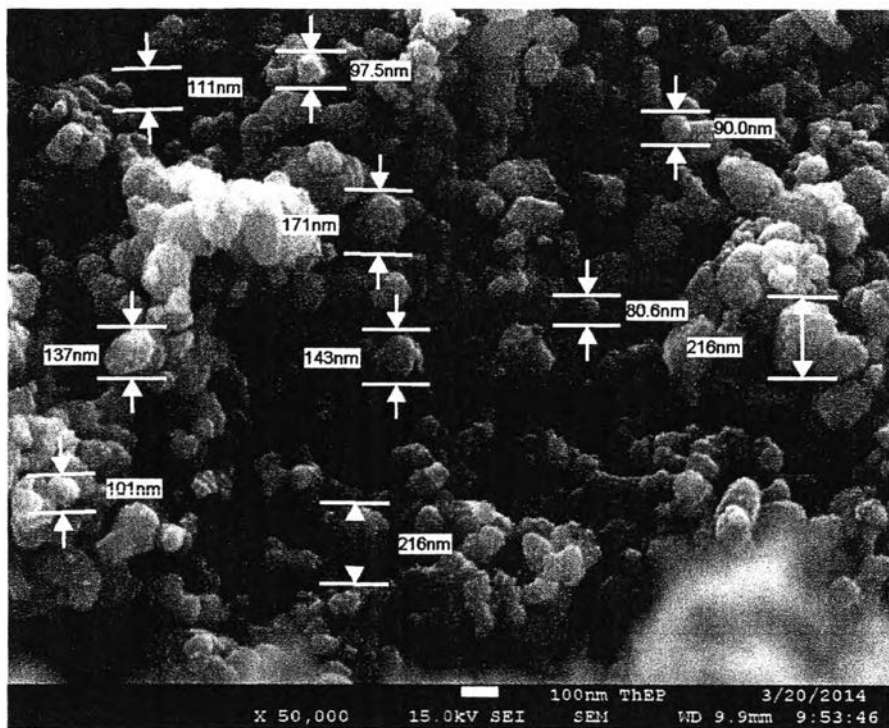


Figure D13 ZnO/Ag₂S QD at 3 min.

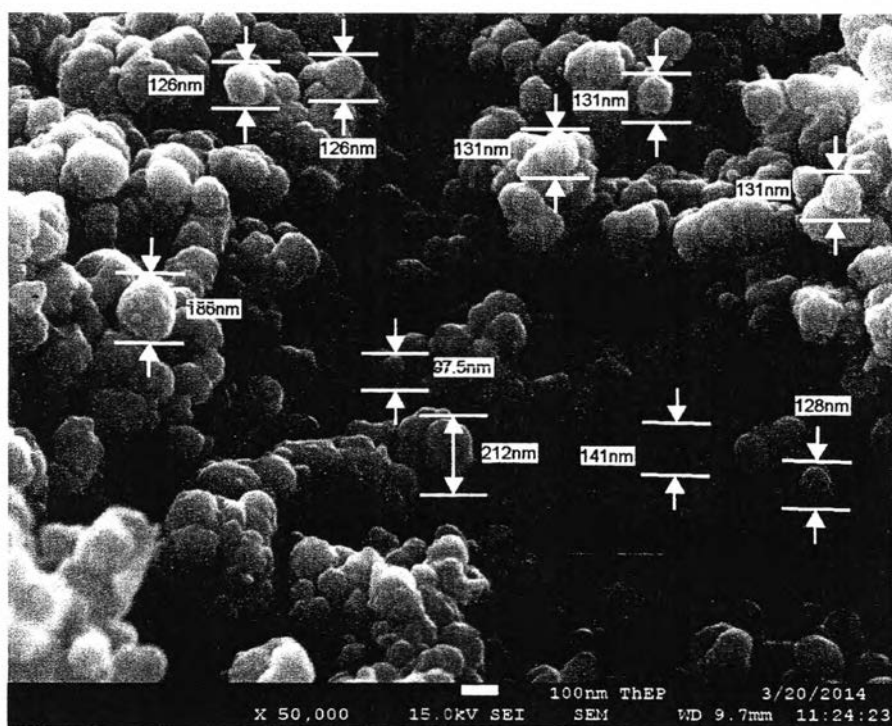


Figure D14 ZnO/Ag₂S QD at 5 min.

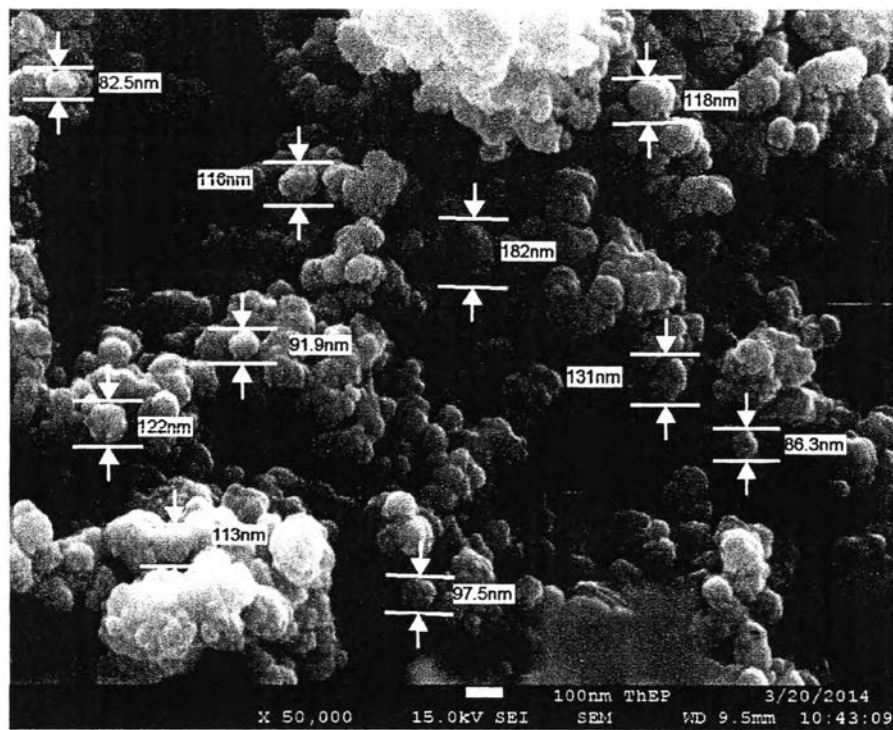


Figure D15 ZnO/Ag₂S QD at 7 min.

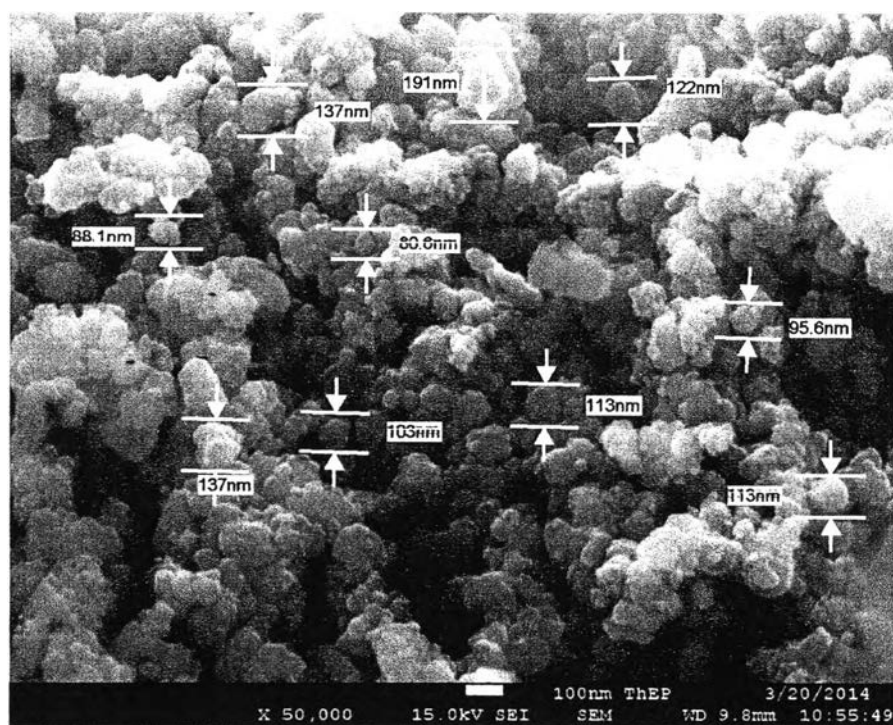


Figure D16 ZnO/Ag₂S QD at 9 min.

CURRICULUM VITAE

Name: Ms. Warunya Junhom

Date of Birth: December 30, 1989

Nationality: Thai

University Education:

2008–2011 Bachelor Degree of Chemistry, Faculty of Science, Chulalongkorn University, Bangkok, Thailand

Proceedings:

1. Junhom, W.; and Magaraphan, R. (2014, April 8--12) Study of different natural pigments in dye-sensitized solar cells. The 5th Research Symposium on Petrochemical and Materials Technology and the 20th PPC Symposium on Petroleum, Petrochemicals, and Polymers. Queen Sirikit National Convention Center, Bangkok, Thailand
2. Junhom, W.; and Magaraphan, R. (2014, June 8-12) Co-sensitization of ZnO by CdS quantum dots in natural dye-sensitized solar cells with polymeric electrolyte to improve the cell stability. The 30th International Conference of the Polymer Processing Society (PPS-30). Cleveland, Ohio, United state

Presentations:

- 1 Junhom, W.; and Magaraphan, R. (2014, April 8--12) Study of different natural pigments in dye-sensitized solar cells. Poster presented at The 5th Research Symposium on Petrochemical and Materials Technology and the 20th PPC Symposium on Petroleum, Petrochemicals, and Polymers. Queen Sirikit National Convention Center, Bangkok, Thailand
2. Junhom, W.; and Magaraphan, R. (2014, June 8-12) Co-sensitization of ZnO by CdS quantum dots in natural dye-sensitized solar cells with polymeric electrolyte to improve the cell stability. Poster presented at The 30th International Conference of The Polymer Processing Society (PPS-30). Cleveland, Ohio, United state