

REFERENCES

- Andari, M.K., Abu-Seedo, F., Stanislaus, A., and Qabazard, H.M. (1996). Kinetics of individual sulfur compounds in deep hydrodesulfurization of Kuwait diesel oil. Fuel 75(14), 1664-1670.
- Bianchini, C., Meli, A., and Vizza, F., 2004, Role of single-site catalysts in the hydrogenation of thiophenes: from models systems to effective HDS catalysts. Journal of Organometallic Chemistry, 689 (24), 4277-4290.
- Boxiong, S., Chunfei, W., Cai, L., Binbin, G., and Rui, W. (2007). Pyrolysis of waste tyres: The influence of USY catalyst/tyre ratio on products. Journal of Analytical and Applied Pyrolysis 78(2), 243-249.
- Brown, C.A., 1973. Kalliation. I. Remarkable fast reaction of potassium hydride with amines and other feeble organic acids. Convenient rapid route to elusive new superbases. Journal of the American Chemical Society, 95 (3), 982-983.
- Cunliffe, A.M. and Williams, P.T. (1998). Composition of oils derived from the batch pyrolysis of tyres. Journal of Analytical and Applied Pyrolysis 44(2), 131-152.
- Dai, X., Yin, X., Wu, C., Zhang, W., and Chen, Y. (2001). Pyrolysis of waste tires in a circulating fluidized-bed reactor. Energy 26(4), 385-399.
- Dallüge, J., van Stee, L.L.P., Xu, X., Williams, J., Beens, J., Vreuls, R.J.J., and Brinkman, U.A.T. (2002). Unravelling the composition of very complex samples by comprehensive gas chromatography coupled to time-of-flight mass spectrometry: Cigarette smoke. Journal of Chromatography A 974(1-2), 169-184.
- Dallüge, J., Vreuls, R.J.J., Beens, J., and Brinkman, U.A.T. (2002). Optimization and characterization of comprehensive two-dimensional gas chromatography with time-of-flight mass spectrometric detection (GC×GC-TOF MS). Journal of Separation Science 25(4), 201-214.
- Düng, N.A., Wongkasemjit, S. and Jitkarnka, S. (2009). Effects of pyrolysis temperature and Pt-loaded catalysts on polar-aromatic content in tire-derived oil. Applied

Catalysis B: Environmental, 91(1–2), 300-307.

- Girgis, M.J. and Gates, B.C. (1991) Reactivities, reaction networks, and kinetics in high-pressure catalytic hydroprocessing, Industrial and Engineering Chemistry, 30, 2021.
- Gutiérrez, O.Y. and Klimova, T. (2011). Effect of the support on the high activity of the (Ni)Mo/ZrO₂-SBA-15 catalyst in the simultaneous hydrodesulfurization of DBT and 4,6-DMDBT. Journal of Catalysis 281(1), 50-62.
- Kilanowski, D.R., Teeuwen, H., De Beer, V.H.J., Gates, B.C., Schuit, B.C.A., and Kwart, H. (1978) Hydrodesulfurization of thiophene, benzothiophene, dibenzothiophene, and related compounds catalyzed by sulfided CoO-MoO₃/□-Al₂O₃: low-pressure reactivity studies, Journal of Catalysis 55, 129
- Laresgoiti, M.F., Caballero, B.M., de Marco, I., Torres, A., Cabrero, M.A., and Chomón, M.J. (2004). Characterization of the liquid products obtained in tyre pyrolysis. Journal of Analytical and Applied Pyrolysis 71(2), 917-934.
- Manzano, C., Hoh, E., and Simonich, S.L.M. (2012). improved separation of complex polycyclic aromatic hydrocarbon mixtures using novel column combinations in GC × GC/ToF-MS. Environmental Science & Technology 46(14), 7677-7684.
- Marriott, P. and Shellie, R. (2002). Principles and applications of comprehensive two-dimensional gas chromatography. TrAC Trends in Analytical Chemistry 21(9–10), 573-583.
- Melbye, A.G., Brakstad, O.G., Hokstad, J.N., Gregersen, I.K., Hansen, B.H., Booth, A.M., Rowland, S.J., and Tollefsen, K.E. (2009). Chemical and toxicological characterization of an unresolved complex mixture-rich biodegraded crude oil. Environmental Toxicology and Chemistry 28(9), 1815-1824.
- Nelson, R.K., Kile, B.M., Plata, D.L., Sylva, S.P., Xu, L., Reddy, C.M., Gaines, R.B., Frysinger, G.S., and Reichenbach, S.E. (2006). Tracking the weathering of an oil spill with comprehensive two-dimensional gas chromatography. Environmental Forensics 7(1), 33-44.

- Pakdel, H., Pantea, D.M., and Roy, C. (2001). Production of dl-limonene by vacuum pyrolysis of used tires. Journal of Analytical and Applied Pyrolysis 57(1), 91-107.
- Phillips, J.B. and Beens, J. (1999). Comprehensive two-dimensional gas chromatography: a hyphenated method with strong coupling between the two dimensions. Journal of Chromatography A 856(1-2), 331-347.
- Pines, H. and Eschinazi, H.E., 1955, Studies in the Terpene Series. XXIV.1 Sodium-catalyzed double bonds migration and dehydrogenation of d-limonene, 1- α -phellandrene and of 2,4(8)- and 3,8(9)-p-menthadiene2,2a. Journal of the American Chemical Society, 77 (23), 6314-6321.
- Roy, C., Labrecque, B., and de Caumia, B. (1990). Recycling of scrap tires to oil and carbon black by vacuum pyrolysis. Resources, Conservation and Recycling 4(3), 203-213.
- Sato, T., Kunimori, K., and Hayashi, S. 1999. Dynamics of benzene, cyclohexane and n-hexane in KL zeolite studied by ^2H NMR. Physical Chemistry Chemical Physics, 1(16), 3839-3843.
- Slaugh, L.H., 1967, Metal hydrides. Hydrogenation and isomerization catalysts. Journal of Organic Chemistry, 32 (1), 108-113.
- Slaugh, L.H., 1968, Hydrogenation of benzene to phenylcyclohexane with supported alkali metal catalysts. Tetrahedron, 24 (12), 4525-4533.
- Stapp, P.R. and Kleinschmidt, R.F., 1965, The Isomerization of cyclooctadienes to cis-bicyclo[3.3.0]oct-2-ene. Journal of Organic Chemistry, 30 (9), 3006-3009.
- Williams, P.T. and Bottrill, R.P. (1995). Sulfur-polycyclic aromatic hydrocarbons in tyre pyrolysis oil. Fuel 74(5), 736-742.
- Yuan, T. and Marshall, W.D., 2005, Catalytic hydrogenation of polycyclic aromatic hydrocarbons over palladium/ γ -Al₂O₃ under mild conditions. Journal of Hazardous Materials, 126 (1-3), 149-157.

APPENDICES

APPENDIX A Temperature Profiles

Table A1 Pyrolysis conditions: Non-catalytic Pyrolysis

Tire = 30 g, N₂ flow = 30 ml/min

Pyrolysis Temperature (T2) = 500 °C

Catalytic Temperature (T1) = 350 °C

Time (min)	T1	T2	Time (min)	T1	T2	Time (min)	T1	T2	Time (min)	T1	T2
2	33.4	31.1	32	367.9	499.2	62	361.2	506.7	92	361.2	496.6
4	53.6	52.8	34	370.2	496.9	64	356.4	502.8	94	359.9	507.8
6	78.4	81.5	36	367.4	499.9	66	352.4	491.8	96	354.5	502.6
8	120.4	125.2	38	366.4	496.5	68	353.4	502.7	98	350.1	500.9
10	145.7	158.0	40	364.2	498.3	70	346.2	500.1	100	353.5	502.7
12	186.0	205.0	42	365.5	495.2	72	355.2	502.3	102	350.5	494.0
14	266.2	302.6	44	357.2	502.7	74	351.4	496.2	104	351.2	502.2
16	277.5	325.5	46	359.1	502.3	76	355.8	508.7	106	353.1	498.8
18	296.5	412.3	48	353.9	506.1	78	352.4	504.2	108	352.4	505.7
20	324.4	429.8	50	352.7	500.1	80	352.0	505.7	110	350.2	495.8
22	331.9	470.8	52	355.6	502.9	82	352.9	497.9	112	349.2	494.8
24	349.1	500.2	54	358.8	497.0	84	351.4	501.5	114	356.1	499.8
26	360.1	501.5	56	359.1	500.9	86	350.3	508.0	116	351.3	499.0
28	363.2	486.0	58	355.9	502.8	88	342.5	496.4	118	350.2	499.2
30	368.4	501.2	60	351.9	501.1	90	361.1	506.3	120	349.4	498.9

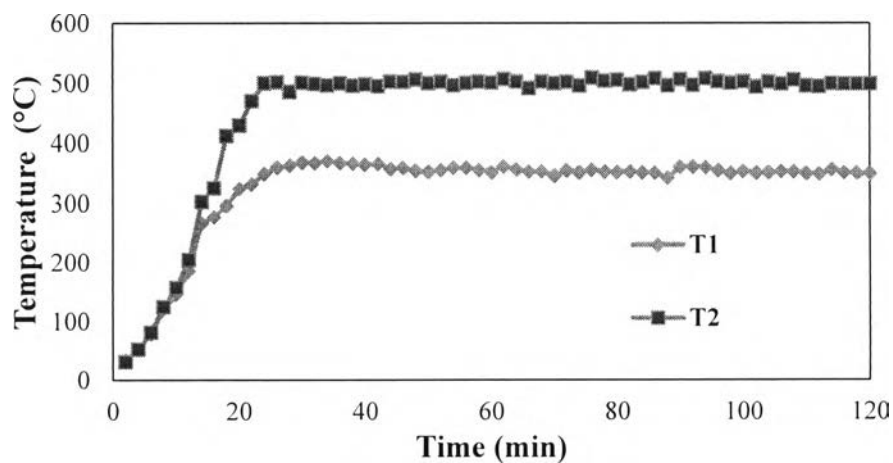


Figure A1 Temperature profiles of non-catalytic pyrolysis.

Table A2 Pyrolysis conditions: KL catalystTire = 30 g, KL = 7.5g, N₂ flow = 30 ml/min

Pyrolysis Temperature (T2) = 500 °C

Catalytic Temperature (T1) = 350 °C

Time (min)	T1	T2	Time (min)	T1	T2	Time (min)	T1	T2	Time (min)	T1	T2
2	30.2	33.1	32	358.0	504.6	62	354.7	506.3	92	352.3	507.8
4	49.4	55.8	34	354.9	497.4	64	357.8	499.1	94	350.0	496.8
6	58.1	67.7	36	363.1	496.2	66	356.6	506.9	96	352.9	506.5
8	88.3	110.9	38	360.9	506.4	68	354.2	496.4	98	356.9	506.0
10	115.2	148.0	40	357.1	500.4	70	350.4	497.1	100	353.9	498.7
12	154.1	199.5	42	355.6	506.0	72	348.6	506.9	102	352.4	503.1
14	194.4	252.9	44	354.6	496.6	74	355.1	499.2	104	347.1	499.7
16	249.1	342.2	46	353.0	507.1	76	356.1	502.1	106	352.6	504.4
18	289.9	411.8	48	351.3	497.3	78	354.5	494.5	108	356.0	500.0
20	322.8	454.9	50	348.5	502.8	80	351.6	507.2	110	353.4	503.2
22	324.6	513.3	52	354.4	495.5	82	349.0	499.5	112	348.9	493.6
24	334.4	503.8	54	353.8	495.9	84	346.9	508.4	114	355.1	507.1
26	336.9	493.5	56	352.4	503.2	86	357.5	498.5	116	354.1	500.1
28	344.1	507.5	58	350.0	500.3	88	357.2	505.6	118	351.6	505.7
30	361.5	499.5	60	347.8	497.7	90	355.4	497.7	120	347.2	498.2

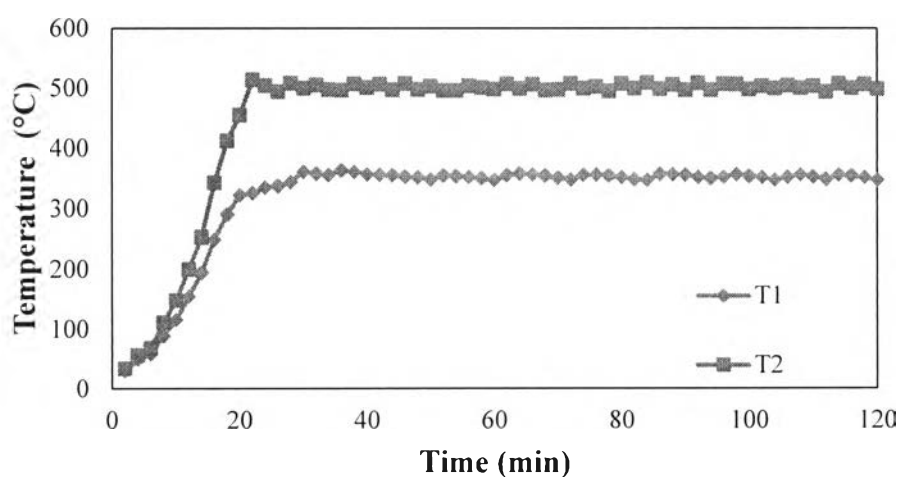
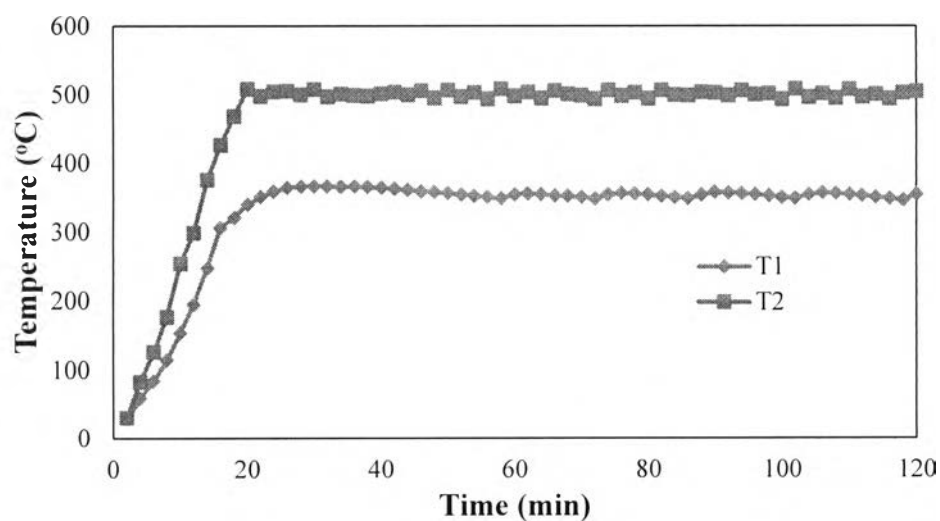
**Figure A2** Temperature profiles of KL pyrolysis.

Table A3 Pyrolysis conditions: 1% Rh/KL catalystTire = 30 g, N₂ flow = 30 ml/min

Pyrolysis Temperature (T2) = 500 °C

Catalytic Temperature (T1) = 350 °C

Time (min)	T1	T2	Time (min)	T1	T2	Time (min)	T1	T2	Time (min)	T1	T2
2	28.5	30.3	32	366.4	496.9	62	356.3	503.4	92	357.6	498.7
4	59.1	82.4	34	365.8	500.6	64	355.0	494.4	94	356.3	505.7
6	83.8	125.8	36	366.4	499.8	66	353.1	505.3	96	355.0	500.4
8	114.8	176.3	38	365.0	498.9	68	352.5	500.5	98	353.2	501.5
10	153.8	253.6	40	364.5	501.2	70	350.8	498.8	100	351.1	493.3
12	194.8	298.2	42	362.6	503.4	72	348.9	493.6	102	349.3	508.2
14	247.1	375.3	44	361.2	499.8	74	354.7	506.1	104	355.1	496.3
16	305.8	426.2	46	359.5	505.6	76	356.5	498.1	106	357.3	501.7
18	320.8	468.3	48	358.1	494.9	78	356.5	503.1	108	356.7	495.1
20	340.3	507.5	50	356.5	506.2	80	354.4	494.1	110	355.1	507.7
22	350.5	497.5	52	354.6	496.8	82	352.4	506.5	112	353.3	497.1
24	358.4	503.8	54	352.9	502.9	84	351.3	499.5	114	351.1	500.5
26	364.7	505.1	56	350.9	493.5	86	349.1	499.7	116	349.1	493.9
28	366.2	499.5	58	348.9	508.1	88	353.9	503.4	118	347.1	502.5
30	367.1	507.1	60	354.9	498.9	90	358.2	502.4	120	355.7	504.3

**Figure A3** Temperature profiles of 1% Rh/KL pyrolysis.

APPENDIX B Yields of Pyrolysis Products**Table B1** Yield of products obtained from using KL and 1% Rh/KL catalysts

Sample	%Metal loading	Yield (%)		
		Gas	Liquid	Solid
Non-catalyst	-	10.19	45.81	44.00
KL	0	13.24	42.98	43.78
Rh/KL	1	17.10	41.57	41.33

APPENDIX C Gas Yield (%)

Table C1 Yield of gases obtained from pyrolysis with 1%Rh/KL

Sample Component	Non-Catalyst	KL	1%Rh/KL
Methane	21.122	19.058	18.891
Ethylene	9.744	9.156	9.201
Ethane	16.516	15.529	15.832
Propylene	12.685	12.060	12.403
Propane	9.399	8.893	9.125
C4	19.627	19.377	19.895
C5	10.903	15.923	14.650
C6	0	0	0
C7	0	0	0
C8	0	0	0
Total	100	100	100

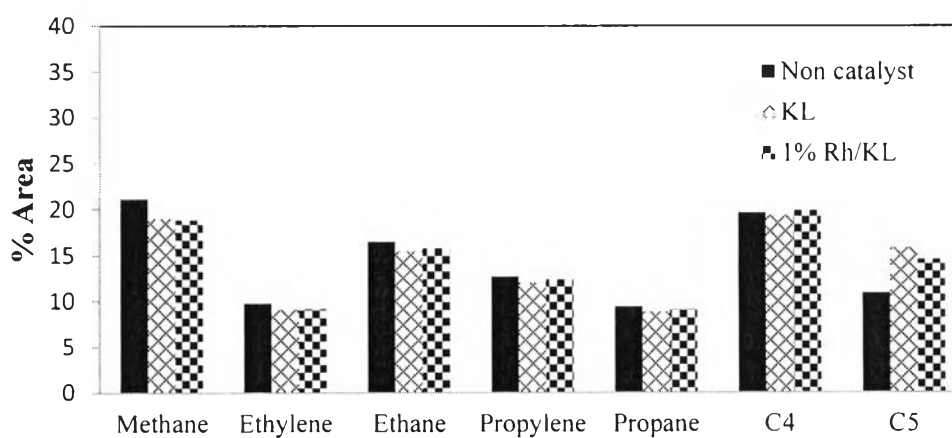


Figure C1 Distribution of gases obtained from pyrolysis with 1%Rh/KL catalyst.

APPENDIX D Hydrocarbon species in Maltenes

Table D1 Hydrocarbon species in maltene from using the non-catalyst batch

Group	Class	Name	%Area		
SATs		Tridecane	0.2234		
		3-Tridecene, (E)-	0.8763		
		Tetradecane	0.6476		
		1-Dodecen-3-yne	0.6411		
		1-Dodecene	1.1833		
		Tetradecane	0.7016		
		3-Tetradecen-5-yne, (Z)-	0.7764		
		Hexadecane	1.7856		
		Hexadecane	0.3343		
		Hexadecane	0.7313		
	OLEs		1-Dodecene	1.1833	
			3-Tridecene, (E)-	0.8763	
			3-Tetradecen-5-yne, (Z)-	0.7764	
		1-Dodecen-3-yne	0.6411		
		1,5-Cyclooctadiene, 3-(1-methyl-2-propenyl)-	0.5943		
		2-Dodecen-4-yne, (Z)-	0.5333		
		1-Dodecene	0.5088		
MAHs	Indanes	1H-Indene, 2,3-dihydro-1,6-dimethyl-	1.1668		
		1H-Indene, 2,3-dihydro-1,6-dimethyl-	1.0663		
		1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	0.8600		
		Indan, 1-methyl-	0.8237		
		1H-Indene, 2,3-dihydro-1,6-dimethyl-	0.8124		
		1H-Indene, 2,3-dihydro-4-propyl-	0.6059		
		1H-Indene, 2,3-dihydro-1,1,5-trimethyl-	0.5506		
		Indan, 1-methyl-	0.5415		
		1H-Indene, 2,3-dihydro-4-methyl-	0.4816		
		1H-Indene, 2,3-dihydro-4-propyl-	0.4578		
		Indane	0.3930		
		1H-Indene, 2,3-dihydro-4,7-dimethyl-	0.3836		
		1H-Indene, 2,3-dihydro-4-propyl-	0.3478		
		1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	0.2934		
		1H-Indene, 2,3-dihydro-4,7-dimethyl-	0.2830		
		1H-Indene, 2,3-dihydro-1,1,5-trimethyl-	0.2269		
		1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	0.2040		
		1H-Indene, 2,3-dihydro-1,1,5-trimethyl-	0.1781		
		1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	0.1640		
		1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	0.1606		
		Alkylbenzenes		Benzene, 3-cyclohexen-1-yl-	0.8825
				Benzene, cyclopentyl-	0.7508
				Benzene, hexyl-	0.7192
	Benzene, cyclopentyl-		0.6010		
	Benzene, cyclohexyl-		0.4476		
	Benzene, heptyl-		0.4374		

Group	Class	Name	%Area
MAHs	Alkylbenzenes	Benzene, 1-methyl-2-(1-methylethyl)-	0.3731
		Benzene, (1,3-dimethylbutyl)-	0.3308
		Benzene, pentyl-	0.2969
		Benzene, (1,3-dimethylbutyl)-	0.2794
		Benzene, cyclohexyl-	0.2469
		Cyclohexene, 1-phenyl-	0.2220
		Benzene, 1,4-diethyl-	0.2156
		Benzene, (1,1-diethylpropyl)-	0.2054
		1-Phenyl-5-methylheptane	0.2037
		Benzene, (3-cyclopentylpropyl)-	0.1933
		Benzene, cyclohexyl-	0.1509
		Benzene, (1-methylbutyl)-	0.1488
		Tetralins	Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-
	Naphthalene, 5-ethyl-1,2,3,4-tetrahydro-		0.3645
	Naphthalene, 1,2-dihydro-3-methyl-		0.2983
	Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-		0.2977
	Naphthalene, 1,2,3,4-tetrahydro-5-methyl-		0.2748
	Naphthalene, 1,2,3,4-tetrahydro-5-methyl-		0.2616
	Naphthalene, 1,2,3,4-tetrahydro-6-propyl-		0.2298
	Naphthalene, 1,2,3,4-tetrahydro-1-methyl-		0.1664
	Alkenylbenzenes	Benzene, (1-ethyl-1-propenyl)-	1.5108
		Benzene, (2-cyclopropylethenyl)-	1.1233
		Hexa-2,4-dienylbenzene	0.9598
		Benzene, (1-methyl-2-cyclopropen-1-yl)-	0.5550
		Benzene, 1-(1-buten-3-yl)-2-vinyl-	0.5513
		Benzene, 4-hexenyl-	0.5068
		Benzene, 1-heptenyl-	0.3053
		3-Ethyl-3-phenyl-1-pentene	0.2693
		Benzene, 4-hexenyl-	0.2669
		Benzene, 1,2-bis(1-buten-3-yl)-	0.2498
		Benzene, (1,2,3-trimethyl-2-cyclopropen-1-yl)-	0.2478
		Benzene, (1,1-dimethyl-2-butenyl)-	0.2319
		Benzene, 2-(1,3-butadienyl)-1,3,5-trimethyl-	0.2223
Benzene, 1-(1-buten-3-yl)-2-vinyl-		0.2026	
Benzene, 1-(1-buten-3-yl)-2-vinyl-		0.1691	
3-Ethyl-3-phenyl-1-pentene		0.1633	
Benzene, [(1-methylethylidene)cyclopropyl]-, (R)-		0.1618	
Hex-1-enylbenzene		0.1533	
Benzene, (2-cyclopropylethenyl)-		0.1530	
Indenes	1H-Indene, 2,3-dimethyl-	1.9680	
	(1-Methylenepent-2-enyl)benzene	0.9372	
	1H-Indene, 2,3-dimethyl-	0.9320	
	(1-Methylpenta-2,4-dienyl)benzene	0.9027	
	1,2,3,6,7,8-Hexahydro-as-indacene	0.7143	
	1H-Indene, 2,3-dimethyl-	0.6121	
	(1-Methylpenta-2,4-dienyl)benzene	0.5530	
	1,2,3-Trimethylindene	0.5470	
	(1-Methylpenta-2,4-dienyl)benzene	0.5452	
	1H-Indene, 1-methyl-3-propyl-	0.5005	

Group	Class	Name	%Area		
MAHs	Indenes	1,2,3-Trimethylindene	0.4750		
		1,2,3-Trimethylindene	0.4153		
		1H-Indene, 1-methyl-3-propyl-	0.4127		
		1H-Indene, 1-ethenyl-2,3-dihydro-	0.3883		
		1,2,3-Trimethylindene	0.3836		
		1H-Indene, 3-ethyl-1-(1-methylethyl)-	0.3465		
		1,2,3-Trimethylindene	0.3426		
		1H-Indene, 2,3-dimethyl-	0.3360		
		1,2,3-Trimethylindene	0.3032		
		1H-Indene, 2,3-dimethyl-	0.2719		
		1H-Indene,2,3-dihydro-2,2-dimethyl-	0.2386		
		1H-Indene, 3-ethyl-1-(1-methylethyl)-	0.2332		
		1H-Indene, 5,5'-(1,10-decanediyl)bis[octahydro-	0.2078		
		1H-Indene, 3-ethyl-1-(1-methylethyl)-	0.2004		
		1H-Indene, 3-ethyl-1-(1-methylethyl)-	0.2001		
		1H-Indene, 1-methyl-3-propyl-	0.1923		
		1H-Indene, 2,3-dimethyl-	0.1601		
			0.0000		
		DAHs		Naphthalene, 1-methyl-	3.7545
				Naphthalene, 1-methyl-	3.0762
h-Naphthalene, 1,8-dimethyl-	2.1437				
Naphthalene, 1-ethyl-	1.7838				
h-Naphthalene, 1,8-dimethyl-	1.7601				
h-Naphthalene, 1,8-dimethyl-	1.1602				
h-Naphthalene, 1,4,6-trimethyl-	1.0148				
Naphthalene, 1,3-dimethyl-	0.9774				
Naphthalene, 2-(1-methylethyl)-	0.9140				
h-Naphthalene, 1,4,6-trimethyl-	0.7005				
Naphthalene, 2-ethyl-	0.5093				
h-Naphthalene, 1,6,7-trimethyl-	0.4834				
h-Naphthalene, 1,4,5-trimethyl-	0.3858				
Naphthalene, 1-ethyl-	0.3748				
Naphthalene, 2-(1-methylethyl)-	0.2593				
Naphthalene, 1-propyl-	0.2471				
Naphthalene, 1-(2-propenyl)-	0.2316				
h-Naphthalene, 1,4,5-trimethyl-	0.2089				
h-Naphthalene, 1,2,4a,5,6,8a-hexahydro-4,7-dimethyl-1-(1-methylethyl)-	0.1689				
A-Naphthalene, 1,2-dihydro-	0.3929				
A-Naphthalene, 1,2-dihydro-	0.1646				
A-Naphthalene	1.7932				
1-Isopropenyl-naphthalene	0.1717				
PAHs		Biphenyl	1.6582		
		B-1,1'-Biphenyl, 4-methyl-	0.9182		
		B-1,1'-Biphenyl, 4-methyl-	0.8579		
		E-Benzene, 1,1'-(1,3-propanediyl)bis-	0.5603		
		E-Benzene, (1,2-dicyclopropyl-2-phenylethyl)-	0.5437		
		z-Acenaphthene	0.5184		
		B-1,1'-Biphenyl, 4-methyl-	0.5081		
		B-2,2'-Dimethylbiphenyl	0.4371		

Group	Class	Name	%Area
PAHs		z-9H-Fluorene, 2-methyl-	0.3671
		z-Fluorene	0.2335
		B-4-Ethylbiphenyl	0.2010
		z-Phenanthrene	0.1714
		B-1,1'-Biphenyl, 3,4'-dimethyl-	0.1570
PPAHs		Benzo[b]thiophene, 2,7-dimethyl-	0.1476
		Benzo[b]thiophene, 2-ethyl-7-methyl-	0.0786
		Benzo[b]thiophene, 4-methyl-	0.0668
		b-3-Methylbenzothiophene	0.0648
		Benzo[b]thiophene, 2,5-dimethyl-	0.0530
		Benzo[b]thiophene	0.0367
		Benzo[b]thiophene, 2-methyl-	0.0242
		y-Thiophene, 2-phenyl-	0.0215
		D-1,7-Dimethyldibenzothiophene	0.0187
		z-4-Methylnaphtho[1,2-b]thiophene	0.0187
		Dibenzothiophene	0.0186
		z-4-Methylnaphtho[1,2-b]thiophene	0.0163
		Thiophene, 2-(1-methylethyl)-	0.0152
		z-Naphtho[2,1-b]thiophene	0.0080
		D-2,8-Dimethyldibenzo(b,d)thiophene	0.0080
		z-Phenaleno[1,9-bc]thiophene	0.0080
		D-Thioxanthene	0.0057
		y-Thiophene, 3-phenyl-	0.0052
		Benzo[b]naphtho[2,1-d]thiophene	0.0045
		Benzo[b]thiophene, 2-methyl-	0.0019
	Benzo[b]thiophene, 7-ethyl-	0.0012	
TERs		Limonene	16.1588
		Bi-2,4,6-cycloheptatrien-1-yl	1.2898
		Cyclohexene, 1-methyl-4-(1-methylethenyl)-, (S)-	0.2368
		Bicyclo[3.2.1]oct-2-ene, 3-methyl-4-methylene-	0.1584
		3-Carene	0.1082
	Dispiro[4.2.4.2]tetradeca-6,13-diene	4.8490	

Table D2 Hydrocarbon species in maltene from using KL batch

Group	Class	Name	%Area	
SATs		Tridecane	0.8166	
		Hexadecane	0.4426	
		Hexadecane	0.2509	
		Hexadecane	1.5143	
		Hexadecane	0.5149	
		Heptacosane	0.7392	
OLEs		3-Tridecene, (E)-	0.6011	
		3-Tetradecene, (Z)-	1.6277	
		1-Octene, 3,7-dimethyl-	0.6117	
MAHs	Indanes	Indan, 1-methyl-	6.6773	
		Indane	3.2006	
		1H-Indene, 2,3-dihydro-1,3-dimethyl-	1.0603	
		1H-Indene, 2,3-dihydro-1,6-dimethyl-	0.9654	
		1H-Indene, 2,3-dihydro-1,1-dimethyl-	0.9421	
		Indan, 1-methyl-	0.6267	
		1H-Indene, 2,3-dihydro-1,1-dimethyl-	0.3404	
		1H-Indene, 2,3-dihydro-1,6-dimethyl-	0.3203	
		Alkylbenzenes	Benzene, cyclohexyl-	2.6988
			Benzene, (1,3-dimethylbutyl)-	2.1738
	Benzene, (1-methylethyl)-		1.9362	
	Benzene, cyclopentyl-		1.7966	
	Benzene, pentyl-		1.2433	
	Benzene, heptyl-		0.8220	
	Benzene, cyclohexyl-		0.7768	
	Benzene, hexyl-		0.5863	
	Benzene, 1-methyl-4-(1-methylpropyl)-		0.4571	
	Benzene, 1-methyl-2-(1-methylethyl)-		0.3931	
	Benzene, octyl-		0.2653	
	Benzene		0.1458	
	Tetralins		Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	9.5421
		Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	3.5944	
		Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	0.2743	
Naphthalene, 5-ethyl-1,2,3,4-tetrahydro-		1.7201		
Naphthalene, 1-ethyl-1,2,3,4-tetrahydro-		0.4863		
Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-		0.3462		
Naphthalene, 5-ethyl-1,2,3,4-tetrahydro-		0.5287		
Naphthalene, 1,2,3,4-tetrahydro-5-methyl-		1.0950		
Naphthalene, 1,2,3,4-tetrahydro-1,4-dimethyl-		0.2489		
Naphthalene, 1,2,3,4-tetrahydro-		0.9951		
Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-		0.3462		
Naphthalene, 5-ethyl-1,2,3,4-tetrahydro-		0.5287		
Alkenylbenzenes		Benzene, 3-hexenyl-	0.6911	
	Benzene, 3-cyclohexen-1-yl-	1.7074		
	Benzene, (1-ethyl-1-propenyl)-	0.8004		
Indenes		2-Methylindene	1.0739	

Group	Class	Name	%Area
MAHs	Indenes	2-Methylindene	0.4566
		1H-Indene, 3-ethyl-1-(1-methylethyl)-	0.2426
		1H-Indene, 2,3-dimethyl-	0.9084
		1H-Indene, 1,3-dimethyl-	1.2139
		1H-Indene, 1-ethylidene-	3.1652
		1,2,3,6,7,8-Hexahydro-as-indacene	0.7226
		1,2,3-Trimethylindene	0.5399
		1,2,3-Trimethylindene	0.6284
		1,2,3-Trimethylindene	0.7309
		DAHs	
Naphthalene, 1-(2-propenyl)-	1.5212		
H-Naphthalene, 2,3,6-trimethyl-	1.2468		
Naphthalene, 1-methyl-	1.2421		
Naphthalene, 1-ethyl-	1.1037		
H-Naphthalene, 1,6-dimethyl-	0.8722		
Naphthalene, 1-propyl-	0.7585		
H-Naphthalene, 1,7-dimethyl-	0.6977		
Naphthalene, 1-ethyl-	0.6848		
H-Naphthalene, 1,6,7-trimethyl-	0.6032		
Naphthalene, 2-ethyl-	0.5869		
H-Naphthalene, 1,4,5-trimethyl-	0.5219		
Naphthalene, 2-(1-methylethyl)-	0.5162		
Naphthalene, 1-(2-propenyl)-	0.4863		
H-Naphthalene, 1,4,5-trimethyl-	0.4125		
Naphthalene, 2-methyl-1-propyl-	0.4035		
Naphthalene, 1-methyl-	0.3670		
H-Naphthalene, 1,6,7-trimethyl-	0.2638		
1-Isopropenylnaphthalene	0.2394		
Naphthalene, 2-(1-methylethyl)-	0.2149		
P-Azulene	0.1696		
PAHs		Benzene, 1,1'-(1,3-propanediyl)bis-	2.3762
		B-4-Ethylbiphenyl	1.5341
		B-2,2'-Dimethylbiphenyl	1.2323
		z-Phenanthrene	0.7948
		z-9H-Fluorene, 9-methyl-	0.7200
		B-1,1'-Biphenyl, 3-methyl-	0.6011
		Benzene, 1,1',1'',1'''-(1,6-hexanediyliidene)tetrakis-	0.5987
		B-1,1'-Biphenyl, 4-methyl-	0.5262
		z-1H-Phenylene	0.4493
		Biphenyl	0.3883
		z-9H-Fluorene, 2-methyl-	0.2496
		z-Acenaphthene	0.1434
		PPAHs	
Benzo[b]thiophene, 2-methyl-	0.0691		
Benzo[b]thiophene	0.0530		
Benzo[b]thiophene, 6-methyl-	0.0398		
Benzo[b]thiophene, 2,7-dimethyl-	0.0385		
b3-Methylbenzothiophene	0.0294		
Thiophene, 2,5-diethyl-	0.0261		

Group	Class	Name	%Area
PPAHs		Benzo[b]thiophene, 2-ethyl-7-methyl-	0.0250
		Benzo[b]thiophene, 7-ethyl-	0.0201
		Thiophene, 2-(1-methylethyl)-	0.0175
		Benzo[b]thiophene, 3,5-dimethyl-	0.0123
		γ -Thiophene, 2-phenyl-	0.0113
		Thiophene, 2-hexyl-	0.0103
		α -Naphtho[2,3-b]thiophene, 4,9-dimethyl-	0.0087
		Thiophene, 2-ethyl-	0.0087
		Thiophene, 3-(2-butanyl)-, (E)-	0.0064
		D-Thioxanthene	0.0063
		D-Thioxanthene	0.0061
		b1-Methylidibenzothiophene	0.0060
		Dibenzothiophene	0.0050
		Benzo[b]thiophene, 2-methyl-	0.0049
		T3,4-Dimethylthiophene	0.0047
		Thiophene, 2-propyl-	0.0035
		γ -Thiophene, 3-phenyl-	0.0033
TERs		Bicyclo[6.1.0]nonane, 9-(1-methylethylidene)-	1.0431
		Bicyclo[3.2.1]oct-2-ene, 3-methyl-4-methylene-	0.3604
		Cyclohexene, 1-methyl-4-(1-methylethenyl)-, (S)-	0.8275
		Bicyclo[6.1.0]nonane, 9-(1-methylethylidene)-	1.1538
		Bicyclo[3.2.1]oct-2-ene, 3-methyl-4-methylene-	0.3986
	Cyclohexene, 1-methyl-4-(1-methylethenyl)-, (S)-	0.9153	

Table D3 Hydrocarbon species in maltene from using Rh/KL batch

Group	Class	Name	%Area
SATs		Tridecane, 2-methyl-2-phenyl-	0.2131
		Tridecane	0.7961
		Tridecane	1.1066
		Hexadecane	0.1804
		Hexadecane	3.4135
		Hexadecane	0.8768
		Hexadecane	2.3057
		Hexadecane	0.5097
		Hexadecane	0.2542
		Heptacosane	0.1876
OLEs		3-Tetradecene, (Z)-	1.0028
		3-Tetradecene, (Z)-	0.5471
		3-Tetradecene, (Z)-	0.5074
		Cyclopropane, 1-heptyl-2-methyl-	0.7970
		3-Tridecene, (E)-	1.1305
	ç-Elementene	1.3017	
MAHs	Indanes	Indane	0.4161
		Indan, 1-methyl-	0.8946
		Indan, 1-methyl-	0.5671
		Indan, 1-methyl-	1.9447
		1H-Indene, 2,3-dihydro-1,6-dimethyl-	0.3955
		1H-Indene, 2,3-dihydro-1,6-dimethyl-	1.1920
		1H-Indene, 2,3-dihydro-1,3-dimethyl-	1.3091
		1H-Indene, 2,3-dihydro-1,1-dimethyl-	1.1632
		1H-Indene, 2,3-dihydro-1,1-dimethyl-	0.4202
	Alkylbenzenes	Diphenylmethane	0.1695
		Benzene, pentyl-	0.7512
		Benzene, nonyl-	0.1558
		Benzene, hexyl-	1.1409
		Benzene, heptyl-	0.8825
		Benzene, heptyl-	0.3651
		Benzene, cyclopentyl-	0.7577
		Benzene, cyclohexyl-	0.3211
		Benzene, cyclohexyl-	0.6826
		Benzene, 1-methyl-4-(1-methylethyl)-	0.4110
		Benzene, (3-cyclopentylpropyl)-	0.4301
		Benzene, (2-cyclohexylethyl)-	0.2499
		Benzene, (1,3-dimethylbutyl)-	0.3583
		Benzene, (1-methylhexyl)-	0.4630
	Tetralins	Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	1.9305
		Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	1.1915
		Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	0.2806
		Naphthalene, 6-ethyl-1,2,3,4-tetrahydro-	0.6144
		H-Naphthalene, 1,2,3,4-tetrahydro-1,8-dimethyl-	1.0983
		H-Naphthalene, 1,2,3,4-tetrahydro-1,1,6-trimethyl-	0.8392
		Naphthalene, 1,2,3,4-tetrahydro-6-propyl-	0.5114
		Naphthalene, 1,2,3,4-tetrahydro-5-methyl-	0.6170
		Naphthalene, 1,2,3,4-tetrahydro-5-methyl-	0.3270
Naphthalene, 1,2,3,4-tetrahydro-1,4-dimethyl-		2.1167	
Naphthalene, 1,2,3,4-tetrahydro-1,4-dimethyl-		0.1364	
Naphthalene, 1,2,3,4-tetrahydro-1,1-dimethyl-		0.3887	
Naphthalene, 1,2,3,4-tetrahydro-1-propyl-		0.5344	
Naphthalene, 1,2,3,4-tetrahydro-1-nonyl-		0.4702	
Naphthalene, 2-ethyl-1,2,3,4-tetrahydro-	0.3714		
Alkenylbenzenes	Cyclopropene, 2,3-dimethyl-3-phenyl-	0.9304	
	Cyclohexene, 1-phenyl-	0.5382	
	Benzene, 4-hexenyl-	0.3987	
	Benzene, 3-cyclohexen-1-yl-	1.4738	
	Benzene, 2-heptenyl-	0.2321	
	Benzene, 1-heptenyl-	0.6502	
	Benzene, 1-(1-buten-3-yl)-2-vinyl-	0.6274	
	Benzene, (2-cyclopropylethenyl)-	0.9131	

Group	Class	Name	%Area	
MAHs	Alkenylbenzenes	Benzene, (1,1-dimethyl-2-butynyl)-	0.1438	
		Benzene, (1-methylnonadecyl)-	0.3073	
		Benzene, (1-ethyl-1-propenyl)-	1.9163	
			3-Ethyl-3-phenyl-1-pentene	1.1019
	Indenes		2-Methylindene	1.3259
			2-Methylindene	0.5637
			1H-Indene, 3-ethyl-1-(1-methylethyl)-	0.2996
			1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	1.1506
			1H-Indene, 3-ethenyl-2,3-dihydro-1,1-dimethyl-	0.2829
			1H-Indene, 2,3-dimethyl-	1.1216
			1H-Indene, 1,3-dimethyl-	1.4987
			1H-Indene, 1-ethylidene-	3.9080
			1,2,3,6,7,8-Hexahydro-as-indacene	0.8922
			1,2,3-Trimethylindene	0.6666
			1,2,3-Trimethylindene	0.7759
			1,2,3-Trimethylindene	0.9024
DAHs			Naphthalene, 2,3-dimethyl-	1.3206
		Naphthalene, 2-methyl-1-propyl-	0.4376	
		Naphthalene, 2-(1-methylethyl)-	0.3379	
		Naphthalene, 1,4,5-trimethyl-	1.0060	
		Naphthalene, 1,4-dimethyl-	0.5193	
		Naphthalene, 1-propyl-	0.2810	
		Naphthalene, 1-methyl-	4.6290	
		Naphthalene, 1-ethyl-	2.3102	
		Naphthalene, 1-ethyl-	4.3278	
		Naphthalene, 1-ethyl-	1.7529	
		H-Naphthalene, 1,6,7-trimethyl-	0.7029	
		H-Naphthalene, 1,4,6-trimethyl-	1.6983	
		H-Naphthalene, 1,4,5-trimethyl-	0.4819	
		H-Naphthalene, 1,4,5-trimethyl-	0.7832	
	H-Naphthalene, 1,4,5-trimethyl-	1.2475		
	Azulene	1.9984		
PAHs		B-1,1'-Biphenyl, 2-ethyl-	0.3689	
		B-1,1'-Biphenyl, 4-methyl-	1.4101	
		B-2,2'-Dimethylbiphenyl	0.8936	
		B-2,2'-Dimethylbiphenyl	0.1950	
		B-4-Ethylbiphenyl	0.4151	
		Biphenyl	2.4345	
		E-Benzene, 1,1'-ethylidenebis-	0.4887	
		z-1H-Phenylene	0.4793	
		z-9H-Fluorene, 1-methyl-	0.7324	
		z-9H-Fluorene, 2-methyl-	0.3372	
		z-9H-Fluorene, 9-methyl-	0.1635	
		z-9H-Fluorene, 9-methylene-	0.5250	
PPAHs		Thiophene, 3,4-diethyl-	0.0025	
		Thiophene, 3-ethyl-	0.0826	
		Thiophene, 3-(1,1-dimethylethyl)-	0.0975	
		Thiophene, 2,5-dimethyl-	0.0295	
		Thiophene, 2,5-diethyl-	0.0476	
		Thiophene, 2,3,4-trimethyl-	0.0110	
		Thiophene, 2-propyl-	0.0922	
		Thiophene, 2-ethyl-	0.0125	
		Thiophene, 2-(1-methylethyl)-	0.0358	
		T-3,4-Dimethylthiophene	0.0272	
		T-3,4-Dimethylthiophene	0.0455	
		Dibenzothiophene	0.0063	
		D-Thioxanthene	0.0594	
		D-3,7-Dimethyldibenzothiophene	0.0078	
		D-2,6-Dimethyldibenzothiophene	0.0045	
		D-1,7-Dimethyldibenzothiophene	0.0046	
		D-1-Methyldibenzothiophene	0.0042	
		Benzo[b]thiophene, 7-ethyl-2-methyl-	0.0062	
		Benzo[b]thiophene, 2,5-dimethyl-	0.0100	
		Benzo[b]thiophene, 2,3-diethyl-	0.0190	
	Benzo[b]thiophene, 2-methyl-	0.0058		

Group	Class	Name	%Area
PPAHs		Benzo[b]thiophene, 2-methyl-	0.0066
		Benzo[b]thiophene, 2-ethyl-7-methyl-	0.0097
		Benzo[b]thiophene	0.0039
TERs		Bicyclo[6.1.0]nonane, 9-(1-methylethylidene)-	1.0431
		Bicyclo[3.2.1]oct-2-ene, 3-methyl-4-methylene-	0.3604
		Cyclohexene, 1-methyl-4-(1-methylethenyl)-, (S)-	0.8275
		Bicyclo[6.1.0]nonane, 9-(1-methylethylidene)-	1.1538
		Bicyclo[3.2.1]oct-2-ene, 3-methyl-4-methylene-	0.3986
		Cyclohexene, 1-methyl-4-(1-methylethenyl)-, (S)-	0.9153

CURRICULUM VITAE

Name: Mr. Sakollapath Pithakratanayothin

Date of Birth: August 17, 1989

Nationality: Thai

University Education:

2008-2012 Bachelor Degree of Science (Petrochemical Technology),
Faculty of Engineer, King Mongkut's University of Technology Lardkrabang,
Bangkok, Thailand

1. Publications:

1. Pithakratanayothin, S. and Jitkarnka, S. (2014) Comparison of components in oil derived from tyre pyrolysis with and without KL catalyst using GC × GC / TOF-MS. Chemical Engineering Transactions, in press.

2. Proceedings:

1. Pithakratanayothin, S. and Jitkarnka, S. (2014, March 31) analysis of a tire-derived oil using GC × GC – TOF/MS for better identification and grouping of hydrocarbon compounds. Proceeding of the 29th ICSW 2014, Philadelphia, USA.
2. Pithakratanayothin, S. and Jitkarnka, S. (2014, April 24) analysis of a tire-derived oil using GC × GC – TOF/MS for better identification and grouping of sulphur compounds. Proceeding of the 5th Research Symposium on Petrochemical and Materials Technology and the 20th PPC Symposium on Petroleum, Petrochemicals, and Polymers, Bangkok, Thailand.
3. Pithakratanayothin, S. and Jitkarnka, S. (2014, August 23) comparison of components in oil derived from tyre pyrolysis with and without KL catalyst using GC × GC / TOF-MS. Proceeding of the 17th Conference Process Integration, Modelling and Optimisation for Energy Saving and Pollution Reduction, Prague, Czech Republic.