

REFERENCES

1. Kuminek G, Rauber GS, Riekes MK, Campos CEMd, Monti GA, Bortoluzzi AJ, et al. Single crystal structure, solid state characterization and dissolution rate of terbinafine hydrochloride. *Journal of Pharmaceutical and Biomedical Analysis*. 2013 5/5/;78–79(0):105-11.
2. Karmwar P, Boetker JP, Graeser KA, Strachan CJ, Rantanen J, Rades T. Investigations on the effect of different cooling rates on the stability of amorphous indomethacin. *European Journal of Pharmaceutical Sciences*. 2011 10/9/;44(3):341-50.
3. Kamada K, Yoshimura S, Murata M, Murata H, Nagai H, Ushio H, et al. Characterization and monitoring of pseudo-polymorphs in manufacturing process by NIR. *International Journal of Pharmaceutics*. 2009 2/23/;368(1–2):103-8.
4. Chieng N, Aaltonen J, Saville D, Rades T. Physical characterization and stability of amorphous indomethacin and ranitidine hydrochloride binary systems prepared by mechanical activation. *European Journal of Pharmaceutics and Biopharmaceutics*. 2009 1//;71(1):47-54.
5. Kao JY, McGoverin CM, Graeser KA, Rades T, Gordon KC. Measurement of amorphous indomethacin stability with NIR and Raman spectroscopy. *Vibrational Spectroscopy*. 2012 1//;58(0):19-26.
6. Rajjada DK, Prasad B, Paudel A, Shah RP, Singh S. Characterization of degradation products of amorphous and polymorphic forms of clopidogrel bisulphate under solid state stress conditions. *Journal of Pharmaceutical and Biomedical Analysis*. 2010 7/8/;52(3):332-44.
7. Srivastava A, Mishra S, Tandon P, Patel S, Ayala AP, Bansal AK, et al. Molecular structure and vibrational spectroscopic analysis of an antiplatelet drug; clopidogrel hydrogen sulphate (form 2) – A combined experimental and quantum chemical approach. *Journal of Molecular Structure*. 2010 2/14/;964(1–3):88-96.
8. 123RF Limited. Stock Photo - Clopidogrel antiplatelet agent, chemical structure [Online] 2005-2014 [2013, May 15]. Available from: http://www.123rf.com/photo_21198617_clopidogrel-antiplatelet-agent-chemical-structure-inhibits-blood-clotting-atoms-are-represented-as-s.html.
9. Vinko Z, Matej S, Promo B, Igor S, Miha P, Gregor R, et al. Preformulation Investigation of Some Clopidogrel Addition Salts. *Acta chimica slovenica (Print)*. 2010;57(2):376-85.
10. Bousquet A, Castro B, Saint-Germain J. Polymorphic form of clopidogrel hydrogen sulphate. US6504030 B1: Google Patents; Jan 7, 2003.



11. Rajjada D, Singh S, Bansal A. Influence of Microenvironment pH, Humidity, and Temperature on the Stability of Polymorphic and Amorphous Forms of Clopidogrel Bisulfate. *AAPS PharmSciTech*. 2010 2010/03/01;11(1):197-203. English.
12. Uvarov V, Popov I. Development and metrological characterization of quantitative X-ray diffraction phase analysis for the mixtures of clopidogrel bisulphate polymorphs. *Journal of Pharmaceutical and Biomedical Analysis*. 2008 3/13;46(4):676-82.
13. United State Pharmacopoeia 35 (USP 35). Official Monographs. Clopidogrel Bisulfate 2012. p. 2733-4.
14. Mohan A, Hariharan M, Vikraman E, Subbaiah G, Venkataraman BR, Saravanan D. Identification and characterization of a principal oxidation impurity in clopidogrel drug substance and drug product. *Journal of Pharmaceutical and Biomedical Analysis*. 2008 5/12;47(1):183-9.
15. Lohray BB, Lohray VB, Pandey B, Dave MG. Polymorphs and amorphous form of (S)-(+)-clopidogrel bisulfate. Google Patents; 2011.
16. Hilfiker R. Polymorphism in the Pharmaceutical Industry. Edited by Rolf Hilfiker: WILEY-VCH Verlag; 2006.
17. Karmwar P, Graeser K, Gordon KC, Strachan CJ, Rades T. Effect of different preparation methods on the dissolution behaviour of amorphous indomethacin. *European Journal of Pharmaceutics and Biopharmaceutics*. 2012 2//;80(2):459-64.
18. Kim M-S, Jin S-J, Kim J-S, Park HJ, Song H-S, Neubert RHH, et al. Preparation, characterization and in vivo evaluation of amorphous atorvastatin calcium nanoparticles using supercritical antisolvent (SAS) process. *European Journal of Pharmaceutics and Biopharmaceutics*. 2008 6//;69(2):454-65.
19. Pralhad T, Rajendrakumar K. Study of freeze-dried quercetin-cyclodextrin binary systems by DSC, FT-IR, X-ray diffraction and SEM analysis. *Journal of Pharmaceutical and Biomedical Analysis*. 2004 2/4;34(2):333-9.
20. Ohta M, Buckton G. A study of the differences between two amorphous spray-dried samples of cefditoren pivoxil which exhibited different physical stabilities. *International Journal of Pharmaceutics*. 2005 1/31;289(1-2):31-8.
21. Karmwar P, Graeser K, Gordon KC, Strachan CJ, Rades T. Investigation of properties and recrystallisation behaviour of amorphous indomethacin samples prepared by different methods. *International Journal of Pharmaceutics*. 2011 9/30;417(1-2):94-100.
22. Zhang F, Aaltonen J, Tian F, Saville DJ, Rades T. Influence of particle size and preparation methods on the physical and chemical stability of amorphous



- simvastatin. *European Journal of Pharmaceutics and Biopharmaceutics*. 2009 1//;71(1):64-70.
23. Newman AW, Byrn SR. Solid-state analysis of the active pharmaceutical ingredient in drug products. *Drug Discovery Today*. 2003 10/1//;8(19):898-905.
 24. Yu L. Amorphous pharmaceutical solids: preparation, characterization and stabilization. *Advanced Drug Delivery Reviews*. 2001 5/16//;48(1):27-42.
 25. Gary D. Christian JEOR. *Instrumental Analysis*. Edition 5, editor 1986.
 26. UC Davis GeoWiki by University of California. Powder X-ray Diffraction (PXRD): Instrumentation [Online] [2013, June 9]. Available from: http://chemwiki.ucdavis.edu/Analytical_Chemistry/Instrumental_Analysis/Diffraction/Powder_X-ray_Diffraction.
 27. Thornton, Steven T., Rex A. *Modern Physics for Scientists and Engineers: Bragg's Law* [Online]: Saunders College Publishing; 1993 [2013, June 9]. Available from: <http://www.mwit.ac.th/~physicslab/hbase/quantum/bragg.html>.
 28. National Nanotechnology Center (NANOTEC). Power Compensated DSC and Heat Flux DSC [Online]: National Science and Technology Development Agency, THAILAND; [2013, June 10]. Available from: <http://www.nanotec.or.th/lab/viewTool.php?toolId=17>.
 29. Wynnyckyj C, Omelon S, Willett TL, Kyle K, Goldberg H, Grynypas MD. Mechanism of bone collagen degradation due to KOH treatment. *Biochimica et Biophysica Acta (BBA) - General Subjects*. 2011 2//;1810(2):192-201.
 30. วีระพงศ์ สัจवाल. Thermogravimetric Analysis (TGA) instrument [Online]: ศูนย์เครื่องมือวิทยาศาสตร์และเทคโนโลยี มหาวิทยาลัยแม่ฟ้าหลวง; 2012, Feb 12 [2013, June 10]. Available from: <http://www.mfu.ac.th/center/stic/index.php/thermal-analysis-instrument-menu/item/111-thermogravimetric-analysis-tga.html>.
 31. The "AZo Journal of Materials Online" by AZoM™.com. Dynamic Vapour Sorption for Organic Solvent Sorption by Surface Measurement Systems: DVS instrument [Online] 2000-2014 [2013, June 10]. Available from: <http://www.azom.com/article.aspx?ArticleID=5181>.
 32. Shinya Inoué. *Introduction to Biological Polarization Microscopy: The Polarizing Microscope* [Online]: Marine Biological Laboratory; 2005, Jul 26 [2013, June 12]. Available from: <http://micro.magnet.fsu.edu/primer/techniques/polarized/biologicalparttwo.html>.
 33. Hédoux A, Guinet Y, Descamps M. The contribution of Raman spectroscopy to the analysis of phase transformations in pharmaceutical compounds. *International Journal of Pharmaceutics*. 2011 9/30//;417(1-2):17-31.



34. Wartewig S, Neubert RHH. Pharmaceutical applications of Mid-IR and Raman spectroscopy. *Advanced Drug Delivery Reviews*. 2005 6/15/;57(8):1144-70.
35. Findlay WP, Bugay DE. Utilization of Fourier transform-Raman spectroscopy for the study of pharmaceutical crystal forms. *Journal of Pharmaceutical and Biomedical Analysis*. 1998 2//;16(6):921-30.
36. Hausman DS, Cambron RT, Sakr A. Application of on-line Raman spectroscopy for characterizing relationships between drug hydration state and tablet physical stability. *International Journal of Pharmaceutics*. 2005 8/11/;299(1-2):19-33.
37. Douglas A. Skoog FJH, Stanley R. Crouch. *Principles of Instrumental Analysis*. Sixth Edition ed. Canada 2007.
38. Second Edition ed: A subsidiary of Harcourt Brace Jovanovich; 1975. *Introduction to Infrared and Raman Spectroscopy*. Academic Press, New York-San Francisco-London 1975.; p. 57-68.
39. SURF Research Group Electrochemical And Surface Engineering. Raman spectroscopy [Online] [2013, June 15]. Available from: <http://surfgroup.be/raman>.
40. Michael K. Denk. Raman Spectroscopy Principles: Stokes line and Anti-Stokes line [Online]: University of GUELPH; 2005 [2013, July 4]. Available from: http://131.104.156.23/Lectures/CHEM_207/vibrational_spectroscopy.
41. United State Pharmacopoeia (USP32) General Chapter. Lasers Used in Pharmaceutical Applications [Online] p. 626 [2013, July 4]. Available from: http://www.drugfuture.com/Pharmacopoeia/USP32/pub/data/v32270/usp32nf27s0_c1120.html.
42. Vankeirsbilck T, Vercauteren A, Baeyens W, Van der Weken G, Verpoort F, Vergote G, et al. Applications of Raman spectroscopy in pharmaceutical analysis. *TrAC Trends in Analytical Chemistry*. 2002 12//;21(12):869-77.
43. Thermo Scientific: Part of Thermo Fisher Scientific. Dispersive Raman spectroscopy [Online] [2013, July 10]. Available from: <http://www.biotechprofiles.com/companyfiles/madisonnetwork/5bdd0a9f37694d6c9fb6e62db5049477.pdf>.
44. R. J. H. Clark. Schematic representation of an FT-based Raman spectrometer [Online] [2013, July 11]. Available from: http://www.medicinescomplete.com/mc/clarke/current/images/Clkraman_spectroscopyF004_default.png.
45. Das RS, Agrawal YK. Raman spectroscopy: Recent advancements, techniques and applications. *Vibrational Spectroscopy*. 2011 11//;57(2):163-76.



46. De Grauw CJ, Sijtsema NM, Otto C, Greve J. Axial resolution of confocal Raman microscopes: Gaussian beam theory and practice. *Journal of Microscopy*. 1997;188(3):273-9.
47. Princeton Instruments. Confocal Raman Microscopy: General Overview [Online] [2013, July 12]. Available from: http://content.piacton.com/Uploads/Princeton/Documents/Library/UpdatedLibrary/Confocal_raman_microscopy_note.pdf.
48. MathWorks: Accelerating the pace of engineering and science. R2014a Documentation: Hierarchical Clustering [Online] [2013, July 13]. Available from: <http://www.mathworks.com/help/stats/hierarchical-clustering.html>.
49. Eilers P. CHEMOMETRICS. DATA ANALYSIS FOR THE LABORATORY AND CHEMICAL PLANT, R. G. Brereton, Wiley, Chichester, 2003, (hardback) ISBN 0-417-48977-8, \$130.00, (paperback) ISBN 0-471-48978-6, \$55.00. *Journal of Chemometrics*. 2003;17(6):360-1.
50. Ma L, Shao X, Wang Y, Yang Y, Bai Z, Zhao Y, et al. Molecular cloning, characterization and expression of myoglobin in Tibetan antelope (*Pantholops hodgsonii*), a species with hypoxic tolerance. *Gene*. 2014 1/10/;533(2):532-7.
51. Lima dSHT, Wagner F, Oliveira AMd, Gontijo dMP. Chemometrics: Theory and Application [Online] 2013, Sep 01 [2013, Oct. 11]. Available from: <http://www.intechopen.com/books/multivariate-analysis-in-management-engineering-and-the-sciences/chemometrics-theory-and-application>.
52. Marcos M. Campos. PCA: Coordinates for a point on the original axes and on the PC axes [Online] 2007, June 04 [2013, Sep. 27]. Available from: <http://oracledmt.blogspot.com/2007/06/way-cooler-pca-and-visualization-linear.html>.
53. Geladi P. Chemometrics in spectroscopy. Part 1. Classical chemometrics. *Spectrochimica Acta Part B: Atomic Spectroscopy*. 2003 5/30/;58(5):767-82.
54. Henning Risvik. Graphical Representation of Scores and Loadings: Score plots [Online]: University of Oslo; 2007, Oct. 01 [2013, Nov. 13]. Available from: http://folk.uio.no/henninri/pca_module/plot_example.html.
55. Dukeck R, Sieger P, Karmwar P. Investigation and correlation of physical stability, dissolution behaviour and interaction parameter of amorphous solid dispersions of telmisartan: A drug development perspective. *European Journal of Pharmaceutical Sciences*. 2013 7/16/;49(4):723-31.
56. Dixon SJ, Heinrich N, Holmboe M, Schaefer ML, Reed RR, Trevejo J, et al. Use of cluster separation indices and the influence of outliers: application of two new separation indices, the modified silhouette index and the overlap coefficient to

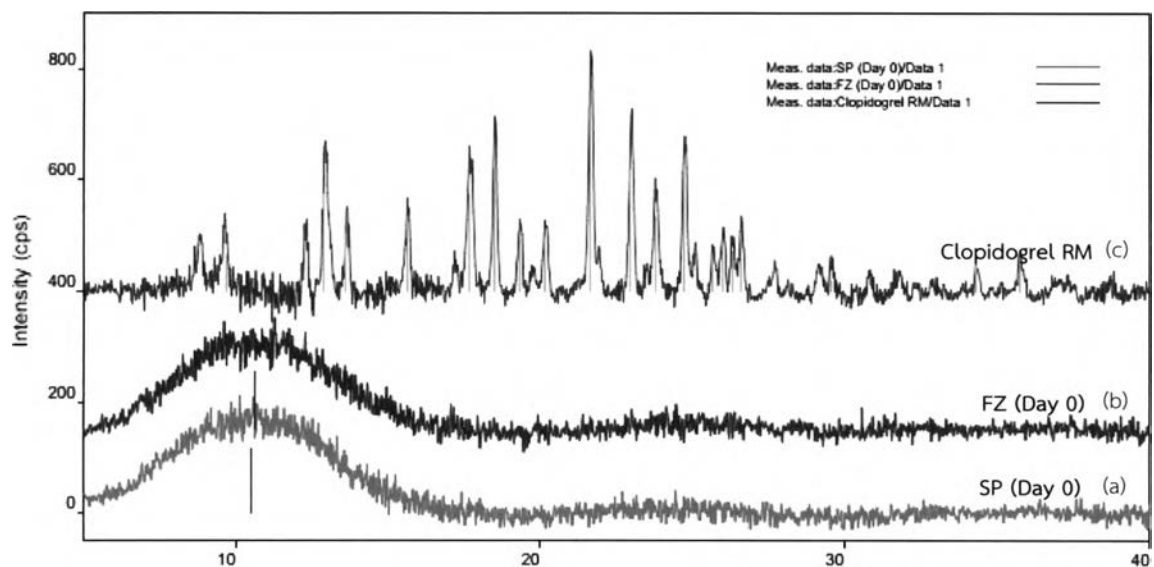


- simulated data and mouse urine metabolomic profiles. *Journal of Chemometrics*. 2009;23(1):19-31.
57. Shin E-C, Craft BD, Pegg RB, Phillips RD, Eitenmiller RR. Chemometric approach to fatty acid profiles in Runner-type peanut cultivars by principal component analysis (PCA). *Food Chemistry*. 2010 4/1/;119(3):1262-70.
58. Sanofi aventis Canada Inc. PRODUCT MONOGRAPH "PLAVIX" [Online] 2013 [2013, Nov. 25]. Available from: <http://products.sanofi.ca/en/plavix.pdf>.
59. Haque MK, Roos YH. Crystallization and X-ray diffraction of spray-dried and freeze-dried amorphous lactose. *Carbohydrate Research*. 2005 2/7/;340(2):293-301.



APPENDICES

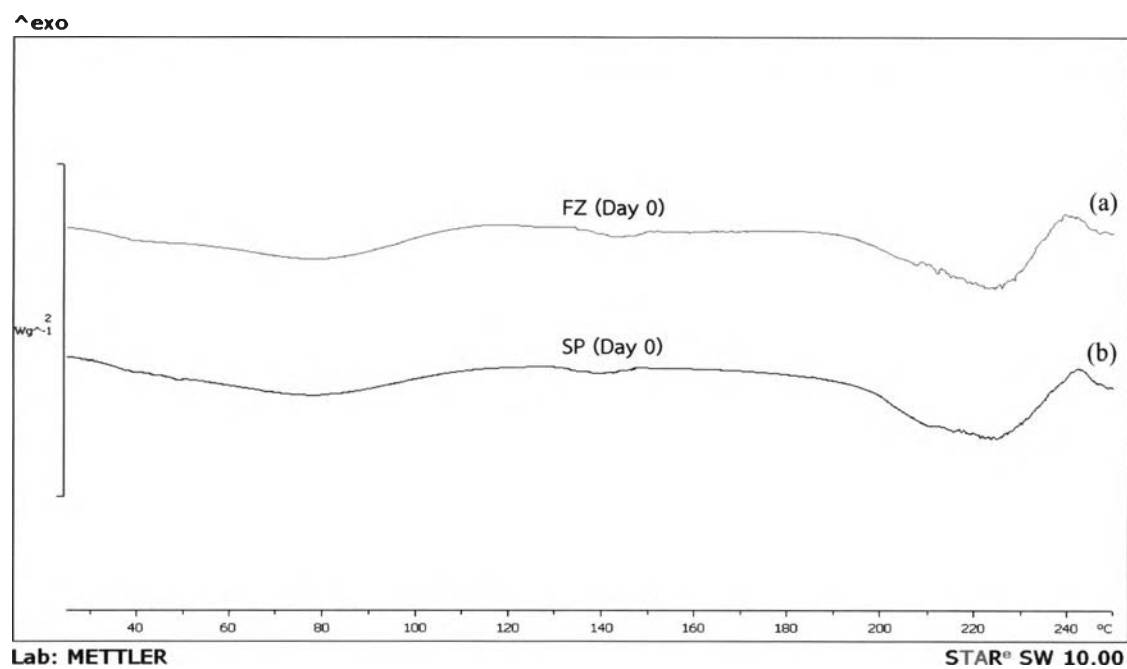
APPENDIX A



PXRD diffractograms of amorphous clopidogrel 2nd batch prepared by spray drying method (a) and freeze drying method (b) compare to clopidogrel RM (c)



APPENDIX B

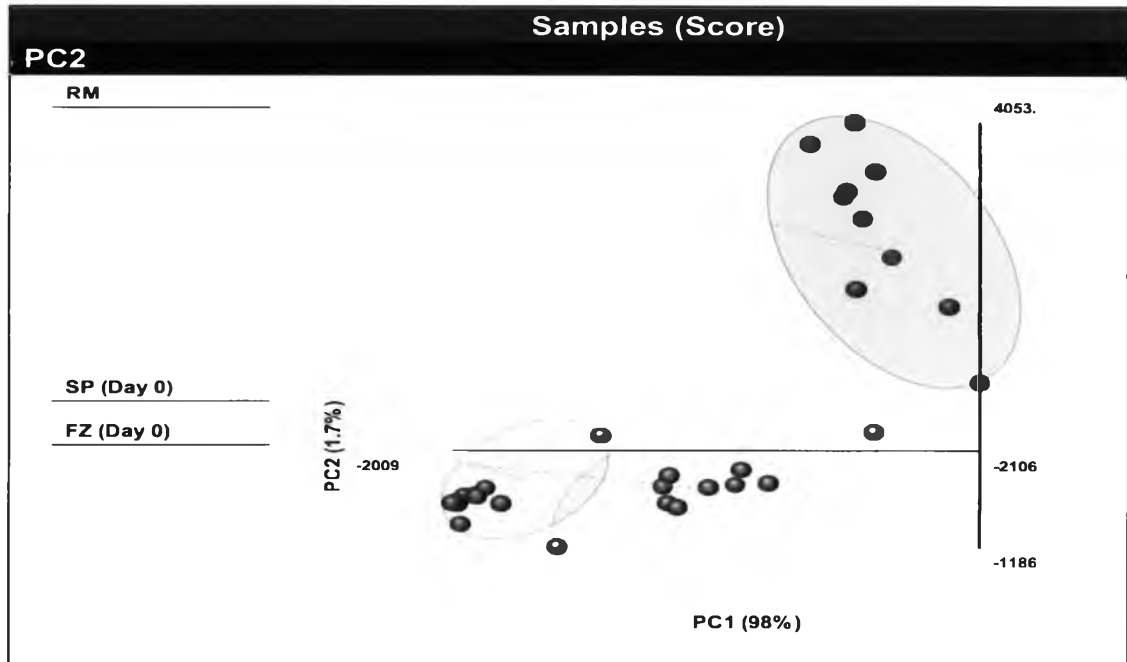


DSC thermograms of initially prepared spray dried (a) and freeze dried clopidogrel (b)
(2nd batch)



632193480

APPENDIX C



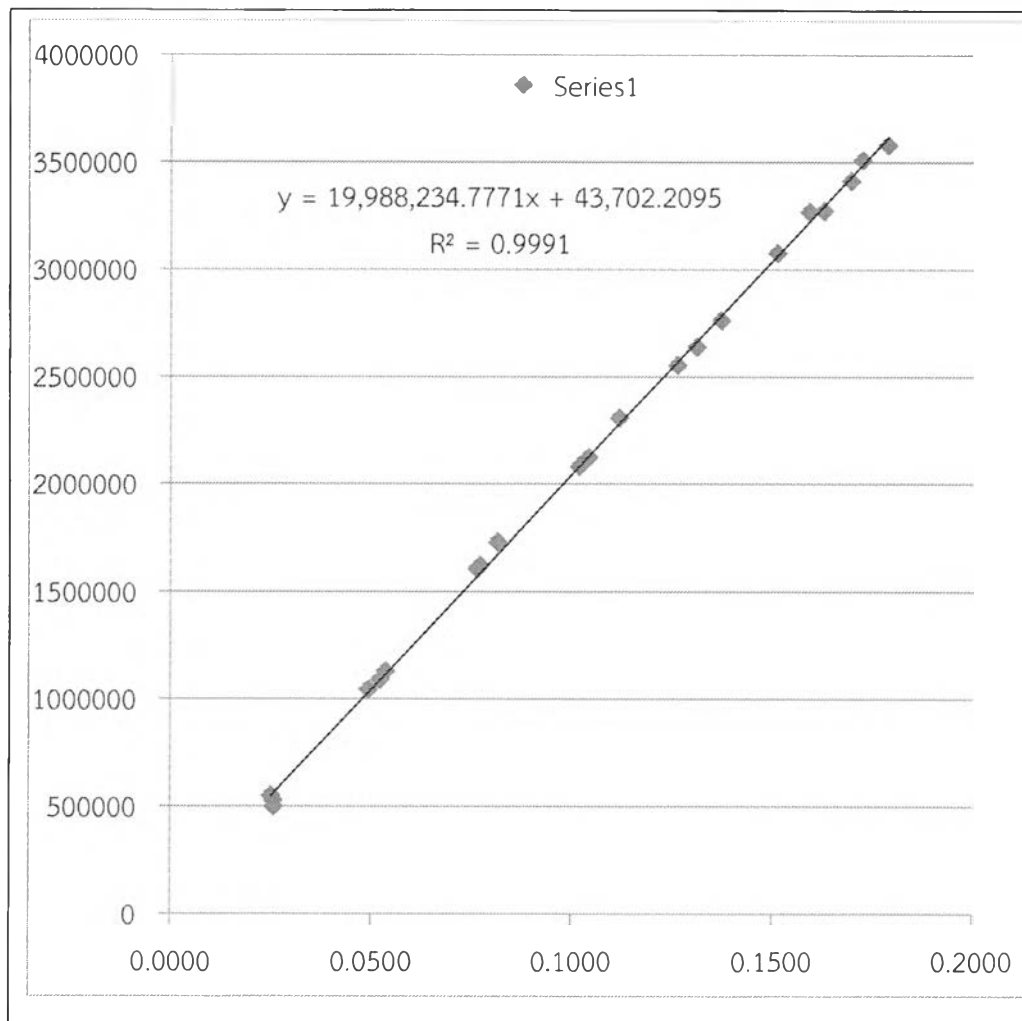
PCA of initial samples prepared by spray drying and freeze drying compare to clopidogrel RM obtained by Multibase program (2nd batch)





Linearity and Range

Std-Clopidogrel		
Conc. (mg/ml)		Area
0.025	0.0259	527753
	0.0253	548043
	0.0260	498857
0.050	0.0494	1042992
	0.0536	1129362
	0.0522	1090972
0.075	0.0772	1622496
	0.0816	1730037
	0.0764	1607193
0.100	0.1044	2124783
	0.1120	2307230
	0.1020	2081571
0.125	0.1264	2553025
	0.1312	2638394
	0.1372	2762477
0.150	0.1512	3075976
	0.1591	3269942
	0.1628	3272746
0.175	0.1788	3580481
	0.1696	3412437
	0.1724	3509795



Regression equation:

$$y = (19,988,234.7771)x + 43702.2095$$

The correlation coefficient:

$$R^2 = 0.9991$$



Accuracy

Sample solution	No.	Weight (mg)	Conc. Add (mg/ml)	Area	Conc. found (mg/ml)	% Recovery
80% of the test concentration	1	3.98	0.07960	1646570	0.08019	100.74
	2	4.13	0.08260	1675031	0.08161	98.81
	3	3.88	0.07760	1619781	0.07885	101.61
100% of the test concentration	1	5.15	0.10300	2118533	0.10380	100.78
	2	5.12	0.10240	2128358	0.10429	101.85
	3	5.01	0.10020	2080935	0.10192	101.72
120% of the test concentration	1	5.96	0.11920	2457759	0.12077	101.32
	2	6.08	0.12160	2463165	0.12104	99.54
	3	6.11	0.12220	2533967	0.12459	101.95
					Mean	100.92

Average %Recovery = 100.92%

Specification:

- Average %Recovery between 98 and 102%



Precision (Related substances)

Std-Clopidogrel Related Cpd A			Std-Clopidogrel Related Cpd B1 & B2				Std-Clopidogrel			Std-Clopidogrel Related Cpd C			
Injection No.	t _R (mins)	Area (r _s)	Injection No.	t _R (mins)	Area (r _s)	t _R (mins)	Area (r _s)	Injection No.	t _R (mins)	Area (r _s)	Injection No.	t _R (mins)	Area (r _s)
1	3.875	58088	1	8.000	46710	10.983	45261	1	9.917	10579	1	17.150	78043
2	3.867	59816	2	8.075	45292	11.167	48920	2	9.900	11097	2	17.158	78011
3	3.867	58158	3	8.142	43288	11.250	48901	3	9.892	11173	3	17.208	77500
4	3.858	58531	4	8.125	46026	11.200	49522	4	9.892	11324	4	17.208	77436
5	3.867	58735	5	8.125	43634	11.275	49071	5	9.900	11479	5	17.258	77751
6	3.858	58735	6	8.158	43693	11.317	49560	6	9.917	11695	6	17.092	77679
Average	3.865	58677.2	Average	8.104	44773.8	11.199	48539.2	Average	9.903	11224.5	Average	17.179	77736.7
SD	-	623.16	SD	-	1432.50	-	1631.76	SD	-	382.52	SD	-	252.60
%RSD	-	1.06	%RSD	-	3.20	-	3.36	%RSD	-	3.41	%RSD	-	0.32

Item / Compound	Related A	Related B1	Related B2	Clopidogrel	Related C
Retention time (t _R) (mins)	3.865	8.104	11.199	9.903	17.179
Relative retention time (RRT)	0.39	0.82	1.13	1.00	1.73
Resolution (R _s) between B1 and clopidogrel: NLT 2.5	2.61				
Relative standard deviation (RSD) : NMT 15% for each peak	1.06	3.20	3.36	3.41	0.32

Specification (USP 35)

System suitability solution::

- The relative retention times are about 0.5 for clopidogrel related cpd A, 0.8 and 1.2 for the two enantiomers of clopidogrel related cpd B, 1.0 for clopidogrel and 2.0 for clopidogrel related cpd C
- The resolution (R) between clopidogrel and the first enantiomer of clopidogrel related cpd B is greater than 2.5

Standard solution:

- The relative standard deviation (RSD) for replicate injections is not more than 15% for each peak



Precision (Assay)

Std-Clopidogrel			Std-Clopidogrel Related Cpd B1 & B2					Std-Clopidogrel		
Injection No.	t _R (mins)	Area (r _s)	Injection No.	t _R (mins)	Area (r _s)	t _R (mins)	Area (r _s)	Injection No.	t _R (mins)	Area (r _s)
1	8.819	2408533	1	6.917	28424	9.042	29588	1	8.000	11287
2	8.801	2429964	2	6.917	28450	9.025	29711	2	7.992	11514
3	8.803	2428358	3	6.900	28486	9.008	29574	3	8.000	11218
4	8.789	2430935	4	6.883	28599	9.000	28995	4	7.992	10907
5	8.784	2441902	5	6.892	28310	8.992	29434	5	7.992	11330
6	8.764	2430549	6	6.883	28591	9.008	29491	6	7.975	11162
Average	8.793	2428373.5	Average	6.899	28476.7	9.013	29465.5	Average	7.992	11236.3
SD	-	10867.29	SD	-	108.98	-	249.02	SD	-	201.38
%RSD	-	0.45	%RSD	-	0.38	-	0.85	%RSD	-	1.79

Item / Compound	Related B1	Related B2	Clopidogrel
Retention time (t _R) (mins)	6.899	9.013	7.992
Relative retention time (RRT)	0.86	1.13	1.00
Resolution (R _s) between B1 and clopidogrel: NLT 2.5	2.91		
Relative standard deviation (RSD) : NMT 1.0%			0.45

Specification (USP 35)

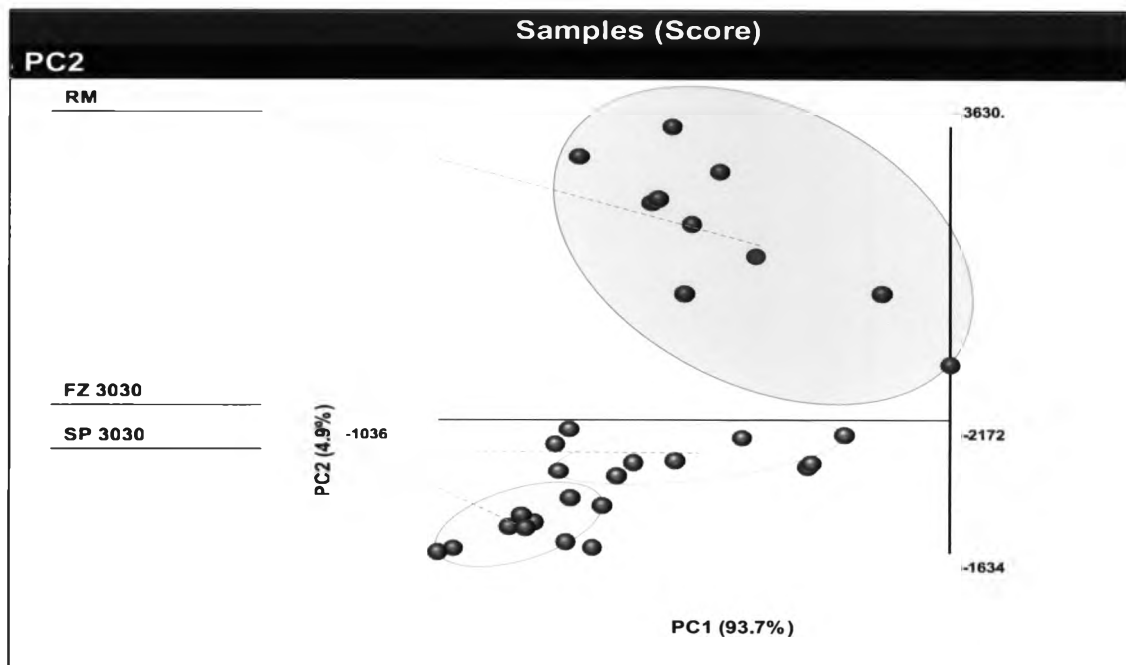
System suitability solution::

- The relative retention times are about 0.8 and 1.2 for the two enantiomers of clopidogrel related cpd B and 1.0 for clopidogrel
- The resolution (R) between clopidogrel and the first enantiomer of clopidogrel related cpd B is greater than 2.5

Standard preparation:

- The relative standard deviation (RSD) for replicate injections determined from clopidogrel bisulfate is not more than 1.0%

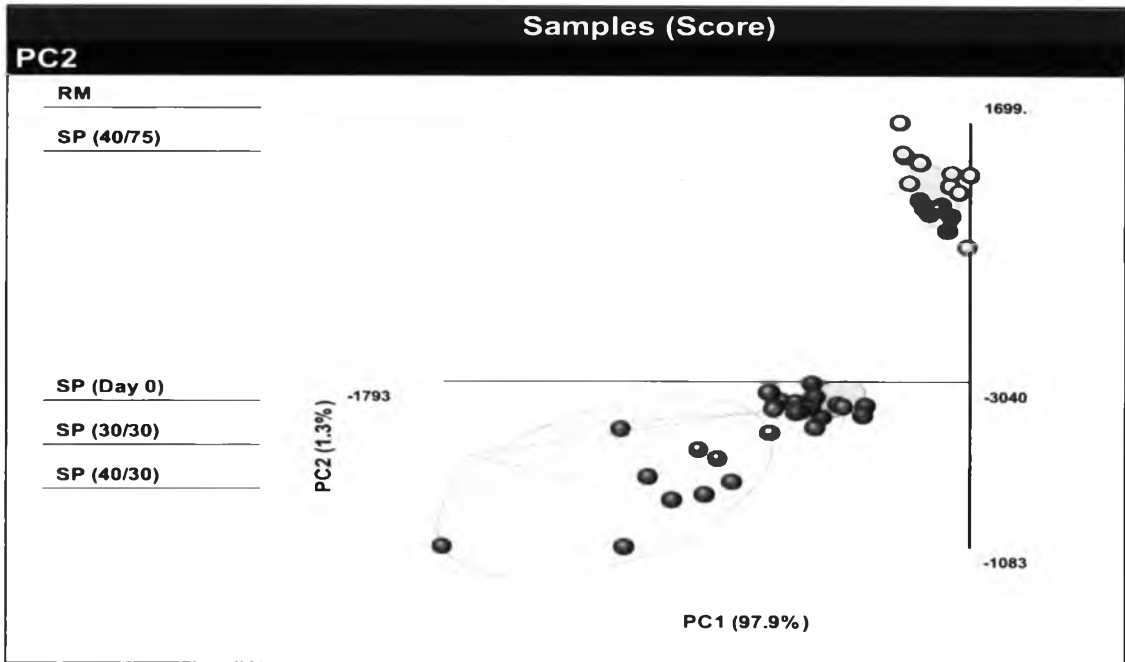
APPENDIX E



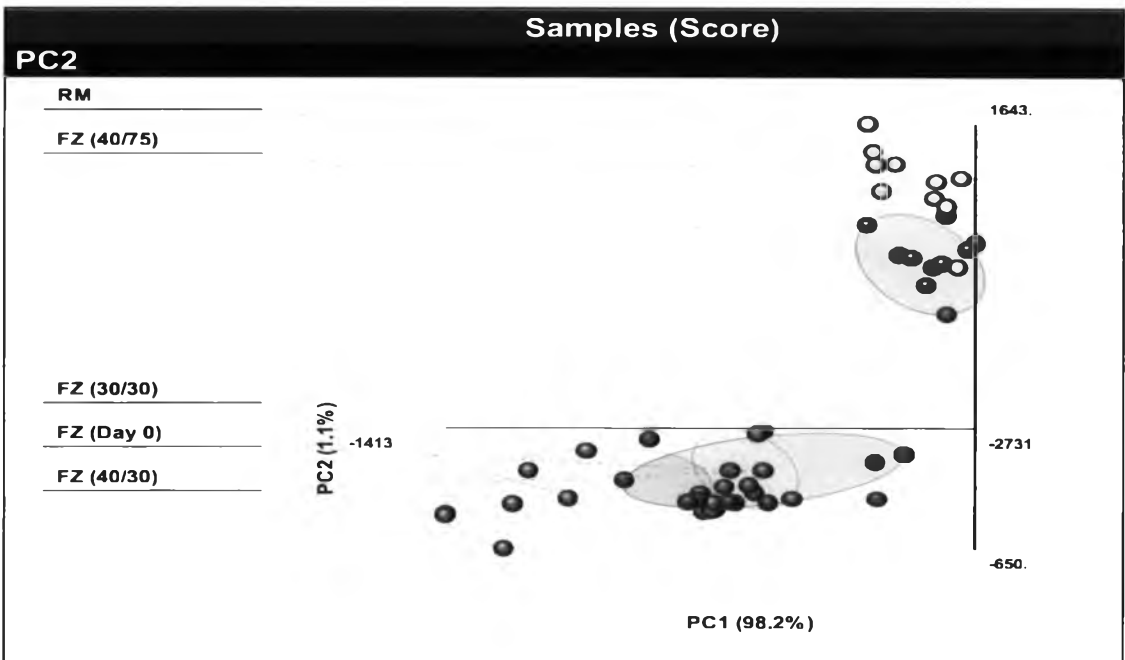
PCA of spray dried and freeze dried samples stored at 30°C 30%RH for 7 days compare to clopidogrel RM obtained by and Multibase program (2nd batch)



APPENDIX F

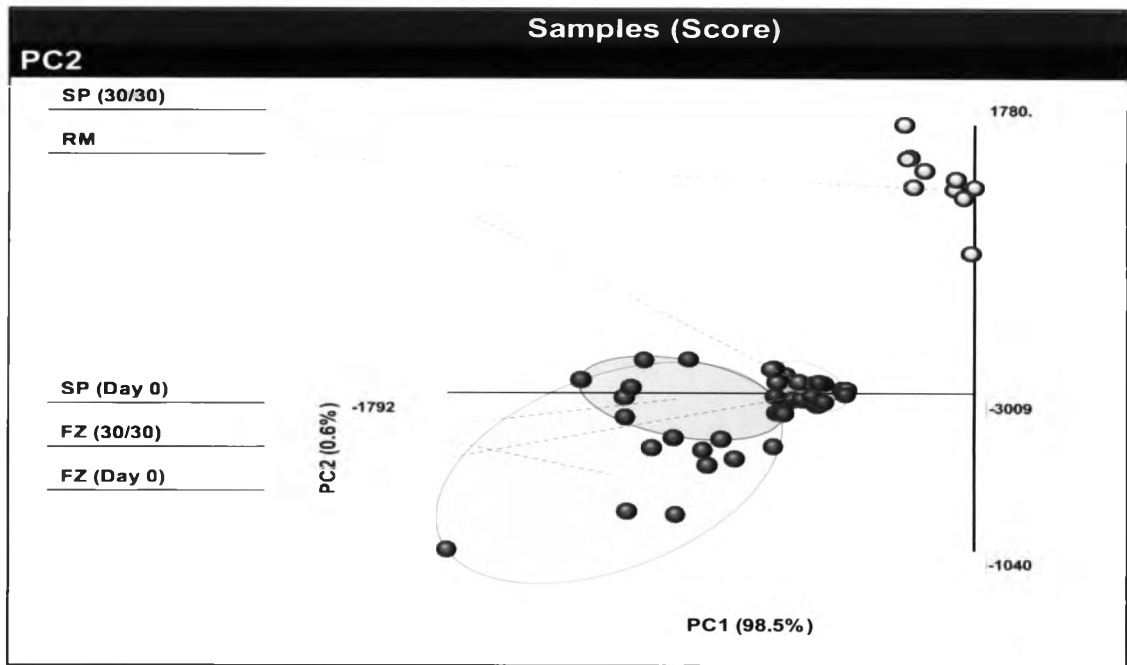


PCA of $SP^7_{30/30}$, $SP^7_{40/30}$, $SP^7_{40/75}$ and RM (Figure 85) compare to SP^0



PCA of $FZ^7_{30/30}$, $FZ^7_{40/30}$, $FZ^7_{40/75}$ and RM (Figure 86) compare to FZ^0





PCA of SP⁷_{30/30}, FZ⁷_{30/30} and RM (Figure 97) compare to SP⁰ and FZ⁰



APPENDIX G

Clopidogrel Related Compound A (Spray dried samples)

Descriptives

Data1

	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum
					Lower Bound	Upper Bound		
					1	11		
2	11	.346037	.2403325	.0724630	.184580	.507495	.1596	.9914
3	11	1.009422	.6961938	.2099103	.541712	1.477131	.2066	2.5136
Total	33	.509827	.5516242	.0960255	.314229	.705424	.0840	2.5136

Test of Homogeneity of Variances

Data1

Levene Statistic	df1	df2	Sig.
11.282	2	30	.000

ANOVA

Data1

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	4.281	2	2.141	11.769	.000
Within Groups	5.456	30	.182		
Total	9.737	32			



Multiple Comparisons

Dependent Variable: Data1

	(I) Group	(J) Group	Mean Difference (I-J)	Std. Error	Sig.	95% Confidence Interval	
						Lower Bound	Upper Bound
Dunnett T3	1	2	-.1720155	.0744274	.112	-.378930	.034898
		3	-.8354003*	.2105965	.007	-1.429788	-.241013
	2	1	.1720155	.0744274	.112	-.034898	.378930
		3	-.6633847*	.2220658	.032	-1.270948	-.055822
	3	1	.8354003*	.2105965	.007	.241013	1.429788
		2	.6633847*	.2220658	.032	.055822	1.270948

*. The mean difference is significant at the 0.05 level.

Conclusion (Clopidogrel Related Compound A obtained from spray drying method)

- Storage at 30°C 30%RH and 40°C 75%RH have significant difference (P=0.007)
- Storage at 40°C 30%RH and 40°C 75%RH have significant difference (P=0.032)
- Storage at 30°C 30%RH and 40°C 30%RH have not significant difference

Clopidogrel Related Compound A (Freeze dried samples)

Descriptives

Data1

	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum
					Lower Bound	Upper Bound		
					1	11		
2	11	.645650	.5481467	.1652724	.277400	1.013899	.2370	1.9509
3	11	1.097531	.6831655	.2059822	.638574	1.556487	.2370	2.3252
Total	33	.676553	.5997241	.1043986	.463900	.889206	.1234	2.3252



632193480

Test of Homogeneity of Variances

Data1

Levene Statistic	df1	df2	Sig.
8.348	2	30	.001

ANOVA

Data1

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	3.634	2	1.817	6.921	.003
Within Groups	7.876	30	.263		
Total	11.509	32			

Post Hoc Tests

Multiple Comparisons

Dependent Variable: Data1

	(I) Group	(J) Group	Mean Difference (I-J)	Std. Error	Sig.	95% Confidence Interval	
						Lower Bound	Upper Bound
Dunnett T3	1	2	-.3591707	.1707888	.157	-.832292	.113951
		3	-.8110517*	.2104341	.008	-1.397981	-.224122
	2	1	.3591707	.1707888	.157	-.113951	.832292
		3	-.4518810	.2640902	.270	-1.139702	.235940
	3	1	.8110517*	.2104341	.008	.224122	1.397981
		2	.4518810	.2640902	.270	-.235940	1.139702

*. The mean difference is significant at the 0.05 level.

Conclusion (Clopidogrel Related Compound A obtained from freeze drying method)

- Storage at 30°C 30%RH and 40°C 75%RH have significant difference (P=0.008)
- Storage at 40°C 30%RH and 40°C 75%RH have not significant difference
- Storage at 30°C 30%RH and 40°C 30%RH have not significant difference

Clopidogrel Related Compound C (Spray dried samples)

Descriptives

Data1

	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for Mean		Minimum	Maximum
					Lower Bound	Upper Bound		
					1	11		
2	11	1.295851	.5442203	.1640886	.930239	1.661463	.6675	1.9460
3	11	1.310109	.4625464	.1394630	.999366	1.620852	.7722	1.9254
Total	33	1.305900	.4990152	.0868674	1.128957	1.482843	.6675	1.9460

Test of Homogeneity of Variances

Data1

Levene Statistic	df1	df2	Sig.
1.276	2	30	.294

ANOVA

Data1

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	.002	2	.001	.003	.997
Within Groups	7.967	30	.266		
Total	7.969	32			

Conclusion (Clopidogrel Related Compound C obtained from spray drying method)

- Storage at 30°C 30%RH and 40°C 75%RH have not significant difference
- Storage at 40°C 30%RH and 40°C 75%RH have not significant difference
- Storage at 30°C 30%RH and 40°C 30%RH have not significant difference



632193480

Clopidogrel Related Compound C (Freeze dried samples)

Descriptives

Data1

	N	Mean	Std. Deviation	Std. Error	95% Confidence Interval for		Minimum	Maximum
					Mean			
					Lower Bound	Upper Bound		
1	11	1.296944	.5939246	.1790750	.897940	1.695948	.7107	2.2104
2	11	1.307939	.5796219	.1747626	.918543	1.697334	.6589	2.0857
3	11	1.317583	.5051140	.1522976	.978243	1.656923	.7737	2.0857
Total	33	1.307488	.5431628	.0945525	1.114891	1.500086	.6589	2.2104

Test of Homogeneity of Variances

Data1

Levene Statistic	df1	df2	Sig.
.815	2	30	.452

ANOVA

Data1

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	.002	2	.001	.004	.996
Within Groups	9.438	30	.315		
Total	9.441	32			

Conclusion (Clopidogrel Related Compound C obtained from freeze drying method)

- Storage at 30°C 30%RH and 40°C 75%RH have not significant difference
- Storage at 40°C 30%RH and 40°C 75%RH have not significant difference
- Storage at 30°C 30%RH and 40°C 30%RH have not significant difference



632193480

VITA

Tharinchai Songrojjanawan was born in Bangkok, Thailand, on July 4th 1982. He received Bachelor of Science in Pharmacy degree in 2006 from the Faculty of Pharmaceutical sciences, Chulalongkorn University, Thailand. He presented a poster titled “Solid-state conversion of amorphous clopidogrel generated by spray drying” in The 39th Congress on Science and Technology of Thailand “Innovative Sciences for a Better Life” on October 21 - 23, 2013 at BITEC, Bangkok, Thailand.

