REFERENCES

- Ahel, M., Giger, W., and Koch, M. (1994). Behavior of alkylphenol polyethoxylate surfactants in the aquatic Environment-I. occurrence and transformation in sewage treatment. Water Research, 28(5), 1131-1142.
- Ashworth, R.A., Howe, G.B., Mullins, M.E., and Rogers, T.N. (1988). Air-water partitioning coefficients of organics in dilute aqueous solutions. <u>Hazardous Materials</u>, 18, 25-36.
- Barber, L.B., Brown, G.K., and Zaugg, S.D., (2000). <u>Potential endocrine disrupting organic chemicals in treated municipal wastewater and river water. In:</u>

 <u>Analysis of Environmental Endocrine Disruptors, American Chemical Society Symposium Series 747.</u> Washington, DC, 97–123.
- Barber, L.B., Kathy E. Lee, Deborah L. Swackhamer, Heiko L. Schoenfuss. (2007). Reproductive responses of male fathead minnows exposed to wastewater treatment plant effluent, effluent treated with XAD8 resin, and an environmentally relevant mixture of alkylphenol compounds. <u>Aquatic Toxicology</u>, 82, 36–46.
- Choori, U.N., Scamehorn, J.F., O'Haver, J.F., and Harwell, J.H. (1998). Removal of volatile organic compounds from surfactant solutions by flash vacuum stripping in a packed column. Ground Water Monitor & Remediation, 18(4), 157-165.
- Dunaway, C.N., Christian, S.D., and Scamehorn, J.H. (Eds.). (1995). <u>Solubilization</u> in <u>Surfactant Aggregates</u>, New York: Marcel Dekker.
- Giesy, J.P., Pierens, S.L., Snyder, E.M., M.-R., S., Kramer, V.J., Snyder, S.A., Nichols, K.M., and Villeneuve, D.A. (2000). Effects of 4-nonylphenol on fecundity and biomarkers of estrogenicity in fathead minnows (*Pimephales promelas*). Environment Toxicology and Chemistry, 19, 1368–1377.
- Hitchens, L., Vane, L.M., and Alvarez, F.R. (2001). VOC removal from water and surfactant solutions by pervaporation: A pilot study. Separation and Purification Technology, 24, 67-84.

- Huibers, P.D.T., Shah, D.O., and Katritzky, A.R. (1997). Predicting surfactant cloud point from molecular structure. <u>Journal of Colloid and Interface</u>
 <u>Science</u>, 193(1), 132-136.
- Hung, K.-C., Chen, B.-H., and Yu, L. E. (2007). Cloud-point extraction of selected polycyclic aromatic hydrocarbons by nonionic surfactants. <u>Seperation and Purification Technology</u>, 57, 1-10.
- Jiang, J., Vane, L.M., and Sikdar, S.K. (1997). Recovery of VOCs from surfactant solutions by pervaporation. <u>Journal of Membrane Science</u>, 136(1-2), 233-247.
- Jobling, S., Sheahan, D., Osborne, J.A., Matthiessen, P., and Sumpter, J.P. (1996).

 Inhibition of testicular growth in rainbow trout (*Oncorhynchus mykiss*)

 Exposed to Estrogenic Alkylphenolic Chemicals. Environment Toxicology
 and Chemistry, 194-202.
- Kouloheris, A.P. (1989). Surfactants: important tools in chemical processing. Chemical Engineering, 96, 130-136.
- Kungsanant, S., Kitiyanan, B., Rirksomboon, T., Osuwan, S., and Scamehorn, J.F. (2008). Toluene removal from nonionic surfactant coacervate phase solutions by vacuum stripping. <u>Seperation and Purification Technology</u>, 63(2), 370-378.
- Kungsanant, S., Kitiyanan, B., Rirksomboon, T., Osuwan, S., and Scamehorn, J.F. Volatile organic compound removal from nonionic surfactant coacervate phase solutions by co-current vacuum stripping: Effect of surfactant concentration, temperature, and solute type. <u>Seperation and Purification</u> Technology, 66, 510-516.
- Lemos, V.A., Franc, R.S., Moreira, B.O. (2007). Cloud point extraction for Co and Ni determination in water samples by flame atomic absorption spectrometry. Separation and Purification Technology, 54, 349–354.
- Li, J.-L., and Chen B.-H. (2002). Solubilization of model polycyclic aromatic hydrocarbons by nonionic surfactants. <u>Journal of Chemical Engineering</u> Science, 57, 2825-2835.

- Li, J.-L., and Chen B.-H. (2003). Equilibrium partition of polycyclic aromatic hydrocarbons in a cloud-point extraction process. <u>Journal of Colloid and</u> Interface Science, 263, 625–632
- Lipe, K.M., Sabatini, D.A., Hasegawa, M.A., and Harwell, J.H. (1996). Micellar-enhanced ultrafiltration and air stripping for surfactant-contaminant separation and surfactant reuse. <u>Ground Water Monitor & Remediation</u>, 16(1), 85-92.
- Lopes, A.S., Garcia, J.S., Catharino, R.R., Santos, L.S., Eberlin, M.N., and Arruda, M.A.Z. (2007). Cloud point extraction applied to casein proteins of cow milk and their identification by mass spectrometry. <u>Analytica Chimica</u> Acta, 590(2), 166–172.
- Mackay, D. (2006) <u>Handbook of physical-chemical properties and environmental</u> fate for organic chemicals, 2nd Ed.; CRC/Taylor & Francis: Florida.
- Mar., M. S., Liya E. Y., Kun-Chilh., H., and Bing-Hung, C. (2006). Solubilization of selected polycyclic compounds by nonionic surfactant. <u>Journal of Surfactants and Detergents</u>, 9(3), 237-244.
- Nimrod, A.C., and Benson, W.H. (1996). Environmental estrogenic effects of alkylphenol ethoxylates. <u>Critical Reviews in Toxicology</u>, 26(3), 335–364.
- Ohashi, A., Hashimoto, T., Imura, H., Ohashi, K. (2007). Cloud point extraction equilibrium of Lanthanum(III), Europium(III), and Lutetium(III) using di(2-ethylhexyl) phosphoric acid and Triton X-100. Talanta, 73, 893–898.
- Pedersen, S.N., Christiansen, L.B., Pedersen, K.L., Korsgaard, B., and Bjerregaard, P. (1999). In vivo estrogenic activity of branched and linear alkylphenols in rainbow trout (*Oncorhynchus mykiss*). The Science of The Total Environment, 233(1-3), 89–96.
- Pramauro, E., and Pelizzetti, E. (1990). The effect of surface active compounds on chemical processes occurring in aquatic environments. <u>Journal of Colloid and Interface Science</u>, 48, 193-208.
- Robbins, G.A., Wang, S., and Stuart, J.D. (1993). Using the static headspace method to determine Henry's law constants. <u>Analitical Chemistry</u>, 65(21), 3113-3118.

- Rosen, J.M. (2004). <u>Surfactants and Interfacial Phenomena.</u> 3rd ed., New York: John Wiley & Sons.
- Scamehorn, J.F., and Harwell, J.H. (2000). <u>Surfactant-Based Separations: Science and Technology</u>. Washington, DC: American Chemical Society.
- Scott, M.H., and Jones, M.N. (2000). The biodegradation of surfactants in the environment. <u>Biochimica et Biophysica Acta (BBA)/Biomembranes</u>, 1508(1), 235-251
- Staudinger, J., and Roberts, P.V. (2001). A critical complication of Henry' law constant temperature dependence relations for organic compounds in dilute aqueous solutions, <u>Chemmosphere</u>, 445, 61-576.
- Staudinger, J., P.V. Roberts. (2001). A critical complication of Henry's law constant temperature dependence relations for organic compounds in diute aqueous solutions. Chemosphere, 44, 561-576.
- Taechangam, P., Scamehom., J.F., Osuwan, S., and Rirksomboon, T. Continuous cloud point extraction of volatile organic contaminants from wastewater in a multi-stage rotating disc contractor: Effect of structure and concentration of solutes. Seperation and Purification Technology, 43(14), 3601-3623.
- Taechangam, P., Scamehorn., J.F., Osuwan, S., and Rirksomboon, T. Effect of nonionic surfactant molecular structure on cloud point extraction of phenol from wastewater. <u>Submitted to Seperation and Purification Technology</u>.
- Talmage, S.S. (1994). <u>Environmental and Human Safety of Major Surfactants. Alcohol Ethoxylates and Alkylphenol Ethoxylates.</u> Boca Raton, FL: Lewis Publishers,
- Trakultamupatam, P., Scamehorn, J.F., Osuwan, S. (2002). Removal of volatile aromatic contaminants from wastewater by cloud point extraction.

 Seperation and Purification Technology, 37, 1291-1305.
- Trakultamupatam, P., Scamehorn, J.F., Osuwan, S. (2004). Scaling up cloud point extraction of aromatic contaminants from wastewater in a continuous rotating disk contactor. I. Effect of disk rotation speed and wastewater to surfactant ratio. Separation and Purification Technology, 39, 479-499.
- Trakultamupatam, P., Scamehorn, J.F., Osuwan, S. (2004). Scaling up cloud point extraction of aromatic contaminants from wastewater in a continuous

- rotating disk contactor. II. Effect of operating temperature and added electrolyte. Separation and Purification Technology, 39, 501-516.
- Turner, L.H., Chiew, Y.C., Ahlert, R.C., Kosson K.S. (1996) Measuring vapor-liquid equilibrium for aqueous-organic systems: review and a new technique. AIChE Journal, 42(6), 1772-1788.
- U.S. Environmental Protection Agency, (2002).
- Vane L.M., Giroux E.L., Henry's law constants and micellar partitioning of volatile organic compounds in surfactant solutions. <u>Journal of Chemical & Engineering Data</u> 45(1), 38-47.
- Vane, L.M., and Alvarez, F.R. (2002). Full-scale vibrating pervaporation membrane unit: VOC removal from water and surfactant solutions. <u>Journal of</u> Membrane Science, 202(1-2), 177-193.
- Yalkowsky, S.H. (1999) <u>Solubility and Solubilization in Aqueous Media</u>; American Chemical Society: Washington, DC, Chapter 7.
- Zhou, J., Wang, S.W., and Sun, X.L. (2008). Determination of osthole in rat plasma by high-performance liquid chromatograph using cloud-point extraction.

 <u>Analytica Chimica Acta</u>, 608(2), 158-164.
- Zhu, H.Z., Liu, W., Mao, J.W., and Yang, M.M. (2008). Cloud point extraction and determination of trace trichlorofrom by high performance liquid chromatography with ultraviolet-detection based on its catalytic effect on benzidine oxidizing. Analytica Chimica Acta, 614(1), 58-62.
- Http://www.biryong.co.kr/datacenter/chemistry/15-S-7.pdf.

APPENDICES

Appendix A Surfactant Properties

Tergitol 15-S-7 is a secondary alcohol ethoxylate non-ionic surfactant. It is under a trademark of Union Carbide. It has an alcohol group located at various positions along a chain of 11–15 carbon atoms with an average ethylene oxide number of 7.3. It is used to formulate high performance cleaners and detergents, especially hard surface cleaners. It can be used in paper and textile processing, paints and coatings, agrochemical, and oil field chemicals. Moreover, Tergitol 15-S-7 passed OECD 301 test for ready biodegradability that prove the environmentally friendly property.

Properties

Identifiers

Chemical Type: Secondary alcohol ethoxylate nonionic

Structural formula: $C_{11-15}H_{23-31}O(CH_2CH_2O)_{7.3}H$

Typical physical properties

Molecular weight: 515

Density at 20°C: 0.992 g/ml

Appearance: Transparent, colorless liquid

Cloud point temperature: 37°C

Viscosity at 25°C: 51 cP

Typical performance properties

CMC at 25°C: 39 ppm

Surface tension at 25°C: 43 dynes/cm, 1 wt%

Foam height at 25°C: 117/28 mm Ross-Miles method, at 0.1 wt% initial / 5

min

Appendix B Solute properties

The solute can be classified into 2 types; aromatic and chlorinated hydrocarbon series. The aromatic hydrocarbon series are benzene (BEN), toluene (TOL), and ethylbenzene (ETB). While the chlorinated hydrocarbon consist of 1,1-dichloroetyhlene (1,1-DCE), trichloroethylene (TCE), and tetrachloroethylene (PCE).

Table B1 Physical and chemical properties of all solutes

Solutes	Molecular	Density	Water	Log octanol-	Molar	Henry's
	weight	(g/cm ³)	solubility	water	volume,	law
			(mg/l)	partition	V_x	constant
				coefficient,	(cm3/mol)*	(Pa
				log Kow		m ³ /mol)
BEN	78.11	0.8786	1770	2.13	89.1	576
TOL	92.14	0.8669	533	2.69	106.3	673
ETB	106.18	0.867	169	3.15	122.4	854
1,1-DCE	96.94	1.213	2500	1.32	79.46	2675.0
TCE	131.39	1.465	1370	2.29	89.02	970.7
PCE	165.83	1.623	151	3.4	102	1852

^{*}observed at 20 °C

References: U.S. Environmental Protection Agency, 2002.

Yalkowsky, 1999.

Turner et al., 1996.

Mackay, 2006.

Staudinger et al., 2001.

CURRICULUM VITAE

Name:

Ms. Sirinthip Kittisrisawai

Date of Birth:

October 27, 1984

Nationality:

Thai

University Education:

2003-2006

(1st class honors) Bachelor Degree of Chemical Engineering,

Faculty of Engineering, Mahidol University, Thailand.

Presentations and Proceedings:

- Kittisrisawai, S., Kungsanant, S., Kitiyanan, B., Osuwan, S., and Scamehorn, J.F. (2008, October 21-22) Removal of Toluene from Coacervate Phase Nonionic Surfactant Solutions by Vacuum Stripping. Poster Presented at 18th Thailand Chemical Engineering and Applied Chemistry Conference, TIChE 18, Chonburi, Thailand.
- Kittisrisawai, S., Kitiyanan, B., Osuwan, S., and Scamehorn, J.F. (2009, April 22) Separation of VOCs from the Coacervate Phase of Alcohol Ethoxylate Surfactant by Vacuum Stripping. Proceedings of <u>The 15th PPC Symposium on Petroleum</u>, Petrochemicals, and Polymers. Bangkok, Thailand.

