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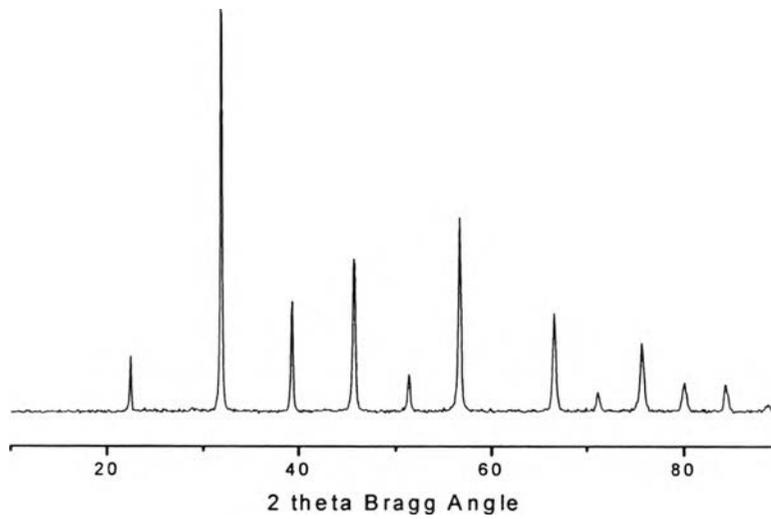
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## APPENDICES

### Appendix A Barium Strontium Titanate Analysis

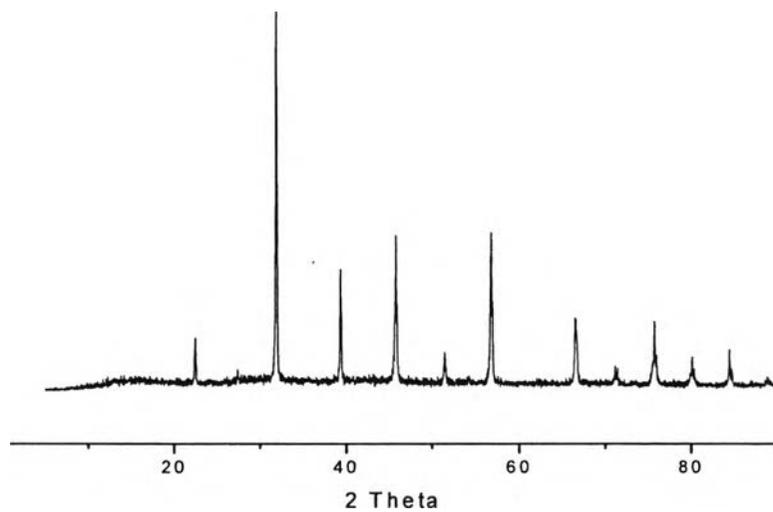
Lattice parameter of the calcined  $Ba_{0.7}Sr_{0.3}TiO_3$  at  $1000^{\circ}C$



**Table A1** Lattice parameter of calcined  $Ba_{0.7}Sr_{0.3}TiO_3$  at  $1000^{\circ}C$

peak No.	$\lambda$ ( nm )	( 1 ~ 3 )	h	k	l	$2\theta$ (degree)
1	0.15418	3	1	0	0	22.33
2	0.15418	3	1	1	0	31.8
3	0.15418	3	1	1	1	39.28
4	0.15418	3	2	0	0	45.66
5	0.15418	3	2	1	1	56.76
6	0.15418	3	2	2	0	66.68
a=	0.3971					
c=	0.39697					
c/a	0.99967					

Lattice parameter of the sintered  $\text{Ba}_{0.7}\text{Sr}_{0.3}\text{TiO}_3$  at  $1330^\circ\text{C}$



**Table A2** Lattice parameter of sintered  $\text{Ba}_{0.7}\text{Sr}_{0.3}\text{TiO}_3$  at  $1330^\circ\text{C}$  for 2 h.

peak No.	$\lambda$ ( nm )	( 1 ~ 3 )	h	k	l	2 $\theta$ (degree)
1	0.15418	3	1	0	0	22.44
2	0.15418	3	1	1	0	31.9
3	0.15418	1	1	1	1	39.3
4	0.15418	1	2	0	0	45.72
5	0.15418	1	2	1	1	56.76
6	0.15418	1	2	2	0	66.52
a=	0.3969					
c=	0.39749					
c/a	1.001325					

## Appendix B Polyvinylidene Fluoride Analysis

The degree of crystallinity was measured as the ratio between  $\Delta H_m$  and  $\Delta H_0$ , as following equation

$$\% \text{ Crystallinity} = \frac{\Delta H_m}{\Delta H_0} \times 100$$

where  $\Delta H_m$  is the melting enthalpy of the material under study and  $\Delta H_0$  is the melting enthalpy of totally crystalline material ( $\Delta H_0 = 104.50 \text{ J/g}$  for PVDF).

**Table B1** % Crystallinity of PVDF

Material	T <sub>m</sub> Onset (°C)	T <sub>m</sub> Peak (°C)	heat of fusion $\Delta H_m$ , J/g	Crystallinity %
solution-casted PVDF	162.3	172.3	65.5	62.6
stretched PVDF X2	162.2	170.1	65.9	63.1
stretched PVDF X4	161.9	170.3	64.4	61.6
stretched PVDF X6	165.3	169.1	66.3	63.4

$F(\beta)$  of PVDF system containing  $\alpha$  and  $\beta$  phase the relative fraction of the  $\beta$  phase,  $F(\beta)$ , was calculated by the following equation:

$$F(\beta) = \frac{X_\beta}{X_\alpha + X_\beta} = \frac{A_\beta}{1.26A_\alpha + A_\beta}$$

Where  $X_\alpha$  and  $X_\beta$  are the degree of crystallinity of  $\alpha$  and  $\beta$  phases,  $A_\alpha$  and  $A_\beta$  are the absorbances of  $\alpha$  and  $\beta$  phases at  $763$  and  $840 \text{ cm}^{-1}$ , respectively.

**Table B2**  $F(\beta)$  of PVDF

<b>Material</b>	<b><math>A_\alpha</math> (normalized 763 <math>\text{cm}^{-1}</math>)</b>	<b><math>A_\beta</math> (normalized 840 <math>\text{cm}^{-1}</math>)</b>	<b>% <math>\beta</math> phase content <math>F(\beta)</math>,</b>
solution-casted PVDF	0.01695	0.42586	95.22
stretched PVDF X2	0.02108	0.33825	92.72
stretched PVDF X4	0.027384	0.43282	92.62
stretched PVDF X6	0.030339	0.446089	92.11

Table B3 Physical properties of PVDF (SOLEF 1008).

SOLEF®

**SOLEF® 1008**

PVDF Homopolymer

Low viscosity - Injection

Physical properties	Standards	Units	
Density	ISO 1183	g/cm <sup>3</sup>	1.78
Water absorption (24 h at 23°C)	ISO 62 (method 1)	%	< 0.04
Melt Flow Index	ASTM D 1238		
	230°C, 10 kg	g/10 min	-
	230°C, 5 kg	g/10 min	24
	230°C, 2.16 kg	g/10 min	8

**Mechanical properties**

Tensile	ASTM D 638		
Tensile stress at yield		MPa/psi	53 - 57/7685 - 8265
Tensile stress at break	23 °C, 50 mm/min	MPa/psi	35 - 50/5076 - 7250
Elongation at yield		%	5 - 10
Elongation at break		%	5 - 10
Modulus	23 °C, 1 mm/min	MPa/psi	2600/377000
Flexion	ASTM D 790		
Maximum load	23 °C	MPa/psi	78/11310
Modulus	2 mm/min	MPa/psi	2200/319000
IZOD impact (notched V 10 mm - at 23 °C - 4 mm thick)	ASTM D 256	J/m	55
Shore D Hardness (2 mm thick)	ASTM D 2240	-	78
Abrasion resistance	TABER CS 10/1 kg	mg/1000 rev	5 - 10
Friction coefficient: static	ASTM D 1894	-	0.2 - 0.4
dynamic		-	0.2 - 0.3

**Thermal properties**

Crystallinity by DSC	ASTM D 3418		
Melting point		°C/°F	174/345
Heat of fusion (80 °C to end of melting)		J/g	67
Crystallizing point		°C/°F	140/284
Crystallizing heat		J/g	59
VICAT point (4 mm thick) load 1 kg	ISO 306	°C/°F	171/340
Deflection temperature (4 mm thick)	ASTM D 648		
load 0.48 MPa	after annealing	°C/°F	148/298
load 1.82 MPa		°C/°F	115/239
Glass transition (T <sub>g</sub> )	DMTA	°C/°F	- 30/ -22
Brittleness temperature (on 2 mm pressed sheet)	ASTM D 746 A	°C/°F	0 - 10/32 - 50

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SOLEF®

## SOLEF® 1008

PVDF Homopolymer

Thermal properties (continuation)	Standards	Units	
Molding shrinkage		%	2 - 3
Thermal stability	TGA beginning- and at 1% weight loss in air	°C/°F	375 - 400/707 - 752
Linear thermal expansion coefficient	ASTM D 696	10 <sup>-6</sup> K <sup>-1</sup>	120 - 140
Thermal conductance at 23°C	ASTM C 177	W/m.K	0.2
Specific heat	23 °C & 100 °C	J/g K	1.2 - 1.6

### Electrical properties

Surface resistivity	ASTM D 257		
Voltage < 1 V, after 2 min - 500 V at 23 °C	DIN 53483	ohm/square	≥ 1.10 <sup>14</sup>
Volume resistivity	ASTM D 257		
Intensity = 10 mA, after 2 min at 23 °C	DIN 53483	ohm.cm	≥ 1.10 <sup>14</sup>

### Fire resistance

UL-94 Flammability test	UL-94	Class	V-0
Limiting Oxygen Index (sheet 3 mm thick)	ASTM D 2863	%	44

The data and numerical results contained in this document are provided for the sake of general information and are given in good faith. They reflect the state of our knowledge at the time of publication. Because the possibilities and application conditions of our product are many and varied, and lie beyond our control, we can in no event be held responsible if all the necessary information on planned applications have not been formally brought to our attention. The information presented here cannot be considered as a suggestion to use our products without taking into account existing patents, or legal provisions or regulations, whether national or local. The purchaser is obliged to verify whether the possession, use or marketing of our products is subject within his territory to particular rules, especially with respect to public health, hygiene and worker and/or consumer safety. The purchaser alone assumes the duties of information and advice for the ultimate user. Solvay Solexis can in no event be held responsible for a possible failure on the part of the purchaser to respect these regulations, provisions and duties.

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### Appendix C Lamination Film Analysis

The effective dielectric constant of the multilayer structure is a series connection model of PVDF and PVDF/BST composite layer dielectrics. the effective dielectric constant,  $\epsilon_{eff}$ , can theoretically be expressed as:

$$\frac{1}{\epsilon_{eff}} = \left[ \frac{1}{\epsilon_1} + \frac{1}{\epsilon_2} + \frac{1}{\epsilon_3} + \dots \right] / (t_1 + t_2 + t_3 + \dots) \quad \text{where } t \text{ is thickness of each layer}$$

**Table C1** Dielectric constant of PVDF x4 and PVDF/BST40% composite

Material	Thickness ( $\mu\text{m}$ )	Dielectric constant ( $\epsilon_r$ )			
		1 kHz	10 kHz	100 kHz	1 MHz
PVDF x4	40	4.77	4.60	4.37	3.89
composite	280	23.03	22.53	21.62	18.41

**Table C2** Dielectric constant of Laminate films

Material	Dielectric constant							
	1 kHz		10 kHz		100 kHz		1 MHz	
	$\epsilon_r$	$\epsilon_{eff}$	$\epsilon_r$	$\epsilon_{eff}$	$\epsilon_r$	$\epsilon_{eff}$	$\epsilon_r$	$\epsilon_{eff}$
P-C	19.01	15.58	18.63	15.15	17.8	14.48	15.03	12.55
P-C-P	17.22	12.445	16.84	12.07	16.18	11.52	13.64	10.06
C-P-C	18.48	18.35	18.09	17.88	17.45	17.12	14.9	14.74

**Remark** P-C is PVDF-Composite bilayer film  
P-C-P is PVDF-Composite-PVDF trilayer film  
C-P-C is PVDF-Composite-PVDF trilayer film  
 $\epsilon_r$  is experimental dielectric constant

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2. Sodsong, T.; and Manuspiya, H. (2009, March 22-27) Improvement in dielectric properties by using lamination technique of stretched polyvinylidene fluoride and polyvinylidene fluoride/barium strontium titanate composite film. Proceedings of the NanoThailand Symposium 2008, Bangkok, Thailand.
3. Sodsong, T.; and Manuspiya, H. (2009, April 22) Piezoelectric Polymer for Mechanical Sensor in Smart Card Application. Proceedings of the 15<sup>th</sup> PPC Symposium on Petroleum, Petrochems, and Polymers, Bangkok, Thailand.

**Presentations:**

1. Suwansumpan, D.; Sodsong, T.; Manuspiya H.; Laoratanakul P.; and Amar S. Bhalla. (2008, August 19-20) Porous PVDF Film: Effect of bubble shapes to Piezoelectric Properties. Paper presented at the Thai-Japan Joint Symposium on Advances in Materials Science and Environmental Technology 2008, Bangkok, Thailand.
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